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Mike Lanagan Penn State University

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Dielectric Properties and Metamaterials

Mike Lanagan

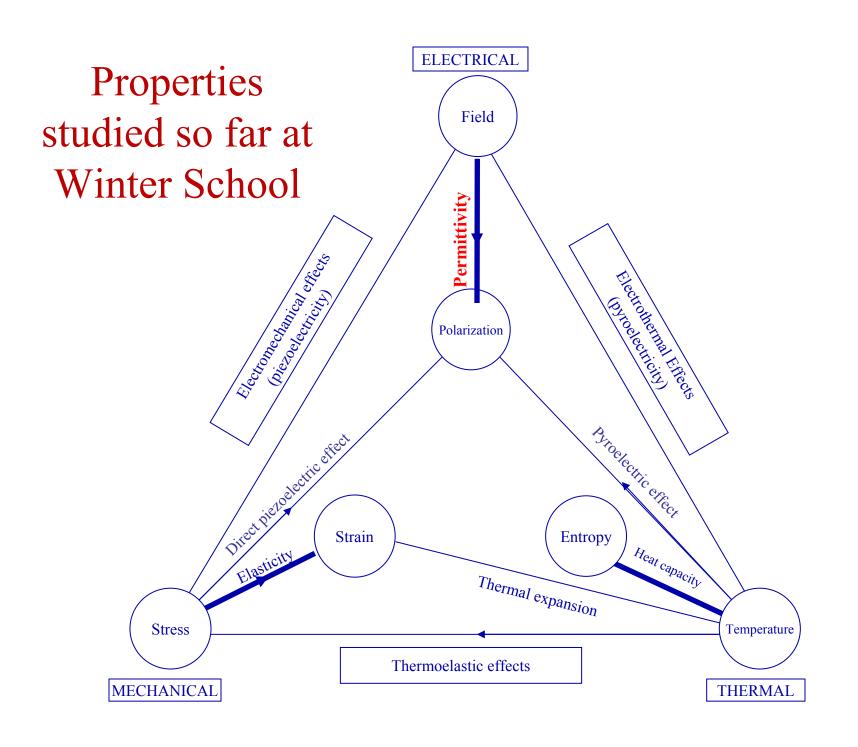
Materials Research Institute

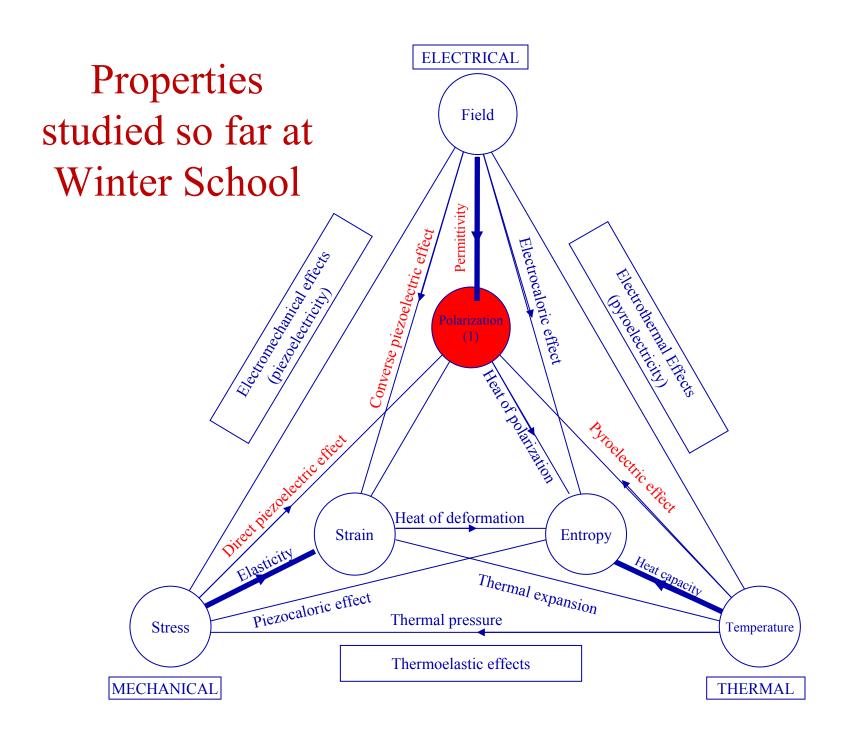
Penn State University

US-Japan Winter School on New Functionality in Glass
January 15, 2008
Kyoto Japan

Dielectric Properties and Metamaterials

- Dielectric Properties (i.e Permittivity)
 - Fundamental frequency dependence
- Metamaterials
 - Negative permittivity and refractive index
 - Based on resonance response
 - Discussion on the potential of glass as a meta-material

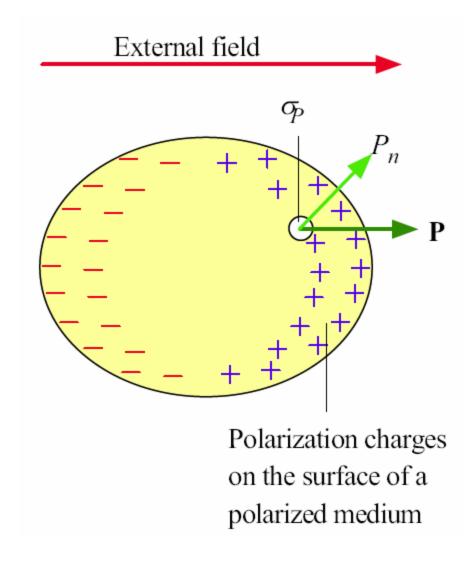




Dielectric Polarization

- Contributes to Permittivity
- 4 basic mechanisms
 - Electronic
 - Ionic
 - Rotational or Dipolar
 - Space charge

External Electric Field Polarizes a Material



Polarization charge density on the surface of a polarized medium is related to the normal component of the polarization vector.

Relative Permittivity and Polarizability

$$\varepsilon_r = 1 + \frac{N\alpha_e}{\varepsilon_o}$$

 ε_r = relative permittivity

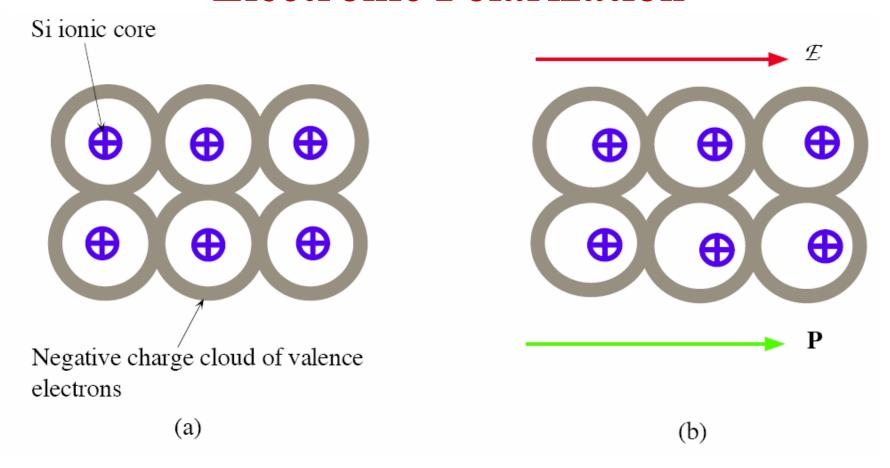
N = number of molecules per unit volume

 α_{e} = electronic polarizability

 ε_o = permittivity of free space

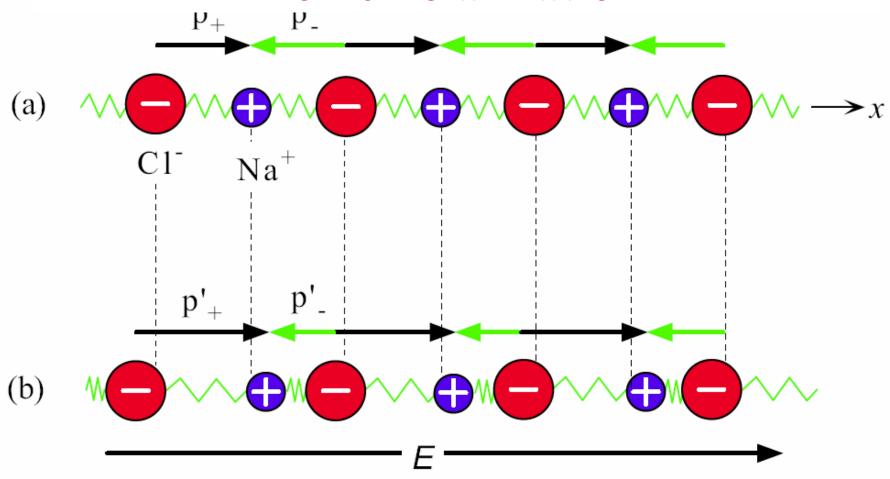
Assumption: Only electronic polarization is present

Electronic Polarization



- (a) Valence electrons in covalent bonds in the absence of an applied field.
- (b) When an electric field is applied to a covalent solid, the valence electrons in the covalent bonds are shifted very easily with respect to the positive ionic cores. The whole solid becomes polarized due to the collective shift in the negative charge distribution of the valence electrons.

Ionic Polarization



moment per 10n 1s zero.

(b) In the presence of an applied field the ions become slightly displaced which leads to a net average dipole moment per ion.

Rotational or **Dipolar** p_o **Polarization** (a) (c) (b) (d)

- (a) A HCl molecule possesses a permanent dipole moment p_0 .
- (b) In the absence of a field, thermal agitation of the molecules results in zero net average dipole moment per molecule.
- (c) A dipole such as HCl placed in a field experiences a torque that tries to rotate it to align p_0 with the field E.
- (d) In the presence of an applied field, the dipoles try to rotate to align with the field against thermal agitation. There is now a net average dipole moment per molecule along the field.

Fig 7.10 From Principles of Electronic Materials and Devices, Third Edition, S.O. Kasap (© McGraw-Hill, 2005)

Complex Relative Permittivity (related to time response)

$$\varepsilon_r = \varepsilon_r' - j\varepsilon_r''$$

 ε_r = dielectric constant

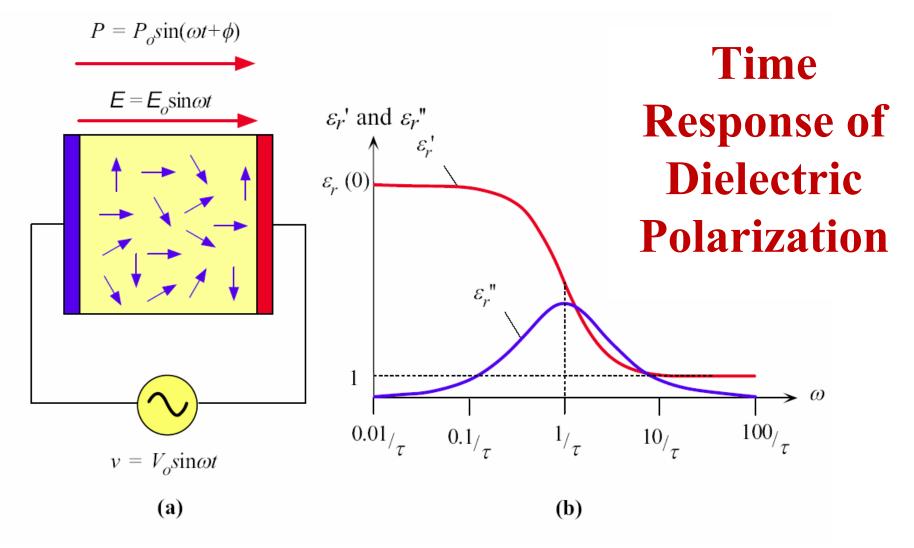
 \mathcal{E}'_r = real part of the complex dielectric constant \mathcal{E}''_r = imaginary part of the complex dielectric constant j = imaginary constant $\sqrt{(-1)}$

Dielectric Loss Factor

Loss Tangent (related to energy loss)

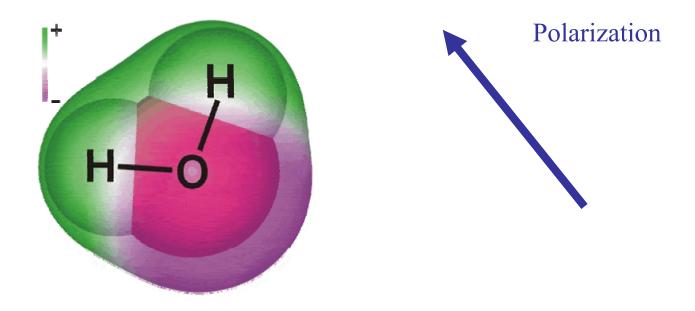
$$\tan \delta = \frac{\mathcal{E}_r^{\prime\prime}}{\mathcal{E}_r^{\prime}}$$

 $\tan \delta = \cos t$ angent or loss factor, $\varepsilon'_r = \text{real part of the complex dielectric constant}$, $\varepsilon''_r = \text{imaginary part of the complex dielectric constant}$



- (a) An ac field is applied to a dipolar medium. The polarization P(P = Np) is out of phase with the ac field.
- (b) The relative permittivity is a complex number with real (\mathcal{E}_r') and imaginary (\mathcal{E}_r'') parts that exhibit frequency dependence.

Example: Water Molecule

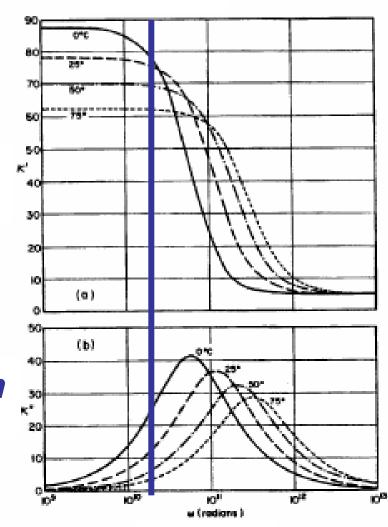


www.lsbu.ac.uk/water/molecule.html

Microwave Dielectric Relaxation of Liquid Water

- Dielectric relaxation indicated by:
 - Decrease in real permittivity
 - Peak in imaginary permittivity
- Maximum ε" corresponds to maximum conversion from EM energy to thermal energy.
- High ε"
 - good for microwave oven
 - Bad for device

2.45 GHz



Frequency

Relaxation spectrum of water as function of frequency and temperature: (a) dispersion; (b) absorption.

A. Von Hipple, IEEE Trans. Insul, 1988

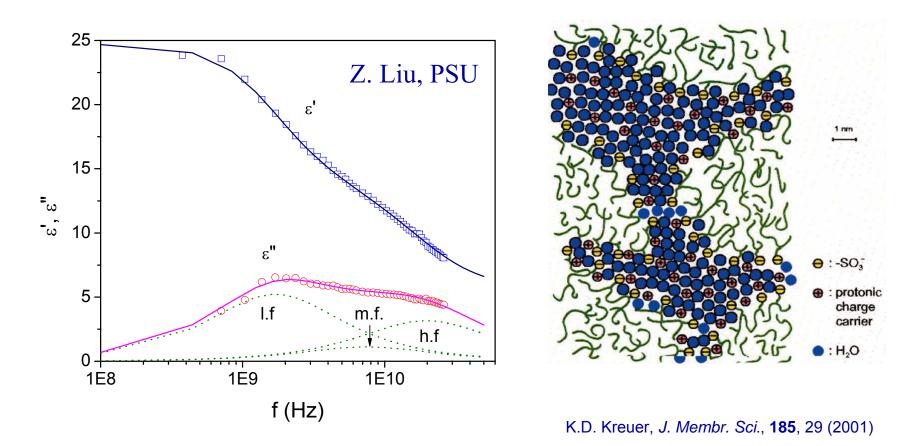
Why not 20 GHz operation for a microwave oven?

- Open bands
 - 915 MHz (not all countries)
 - 2.45 GHz
 - 5.8 GHz
 - 24.1 GHz
- Cost constraints



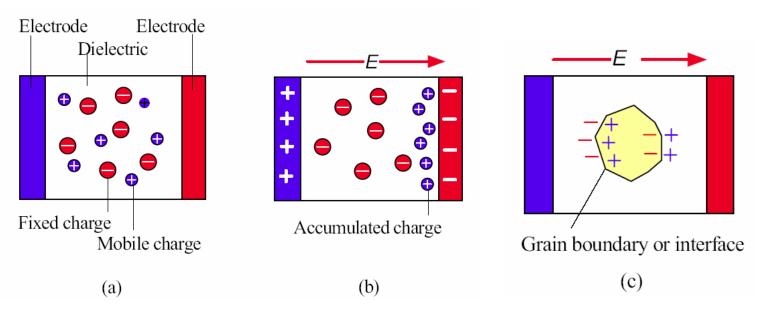
Water in a proton exchange membrane

Current PEM fuel cells are based on PSA membranes, e.g. Nafion.
 The essential feature of Nafion is the nano-separation of hydrophilic/hydrophobic domains



Why not characterize water in porous glass in this way?

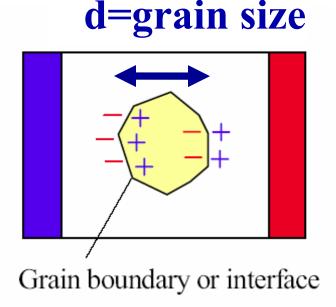
Space Charge Polarizability



- (a) A crystal with equal number of mobile positive ions and fixed negative ions. In the absence of a field, there is no net separation between all the positive charges and all the negative charges.
- (b) In the presence of an applied field, the mobile positive ions migrate toward the negative charges and positive charges in the dielectric. The dielectric therefore exhibits interfacial polarization.
- (c) Grain boundaries and interfaces between different materials frequently give rise to Interfacial polarization.

Space Charge in Ceramic Capacitors with Glass

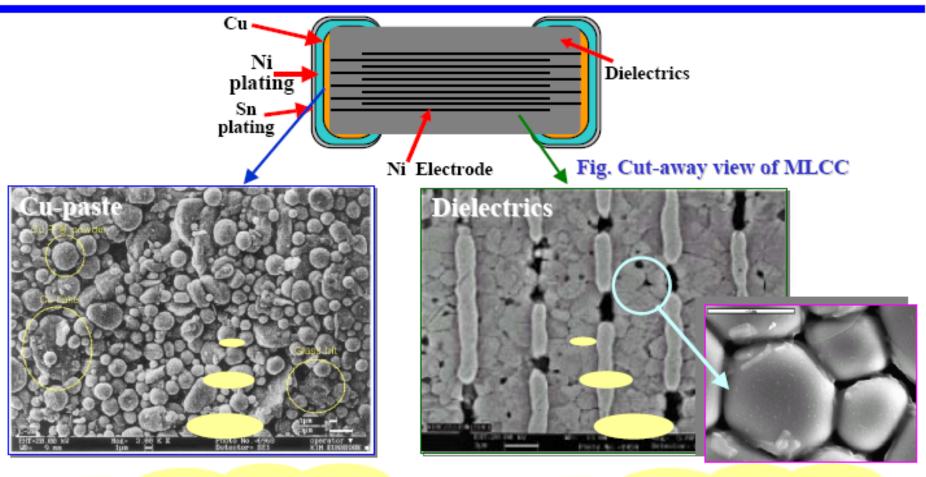
 Need long enough time for charge to move to boundary



Relaxation
$$au \propto \frac{d}{\sigma}$$
 Grain Size Time $au \propto \frac{d}{\sigma}$ Grain Conductivity

Role of Glasses in MLCC





Glass is of major importance as an bond between ceramic and metal, and a filler. Glass is of major importance as an additive to ceramics in order to promote sintering at low temperature.

BaTiO₃ Ceramics with Glass boundaries

Relaxation time = 0.0001 s

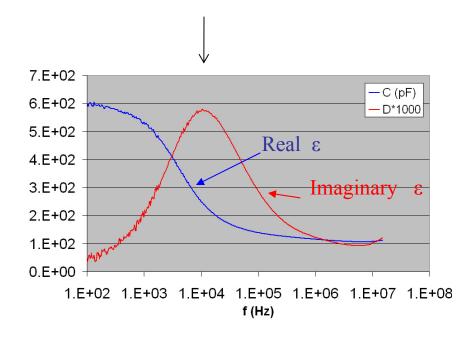
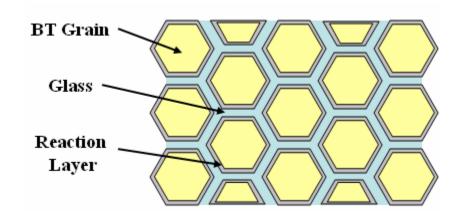


Figure shows dielectric response of a BaTiO3 – Glass composite

Microstructure Schematic



- Conductive grain and insulating grain boundary
- Maxwell-Wagner relaxation

From the relaxation time and microstructure, we can determine the grain conductivity

Frequency Response of Dielectric Polarization

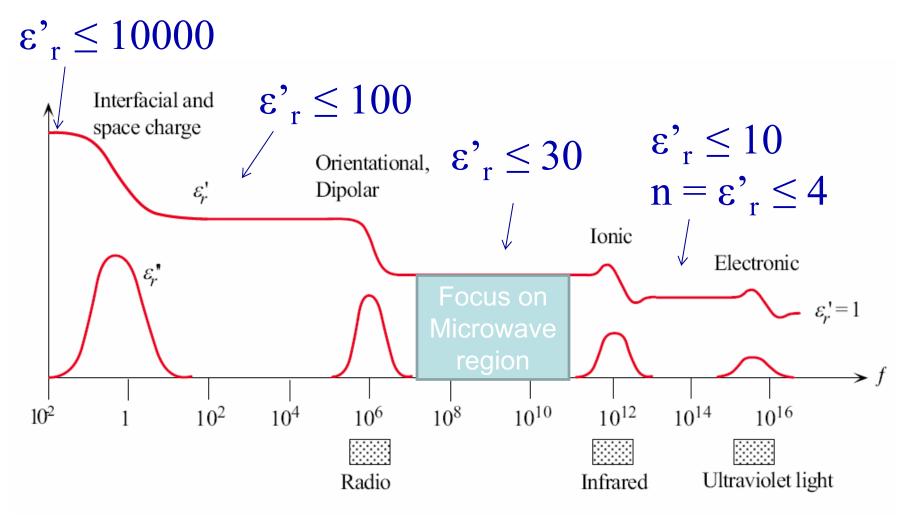


Fig 7.15

Table 7.2 Typical examples of polarization mechanisms

Polarization	Static ε_r	Comment
Electronic	1.0005	Small <i>N</i> in gases: $\varepsilon_r \approx 1$
Electronic	1.53	van der Waals bonding
Electronic polarization due to valence electrons	11.9	Covalent solid; bond polarization
Ionic	5.90	Ionic crystalline solid
Ionic	7.20	Ionic crystalline solid
Orientational	80	Dipolar liquid
Orientational	34	Dipolar liquid
Orientational	7	Dipole orientations partly hindered in the solid
	Electronic Electronic polarization due to valence electrons Ionic Ionic Orientational Orientational	Electronic 1.0005 Electronic 1.53 Electronic polarization 11.9 due to valence electrons Ionic 5.90 Ionic 7.20 Orientational 80 Orientational 34

BaTiO3 permittivity (dielectric constant) = 1,000

Permittivity of Amorphous Materials

Permittivity values are related to the electron density an ionic charge

•
$$SiO_2$$

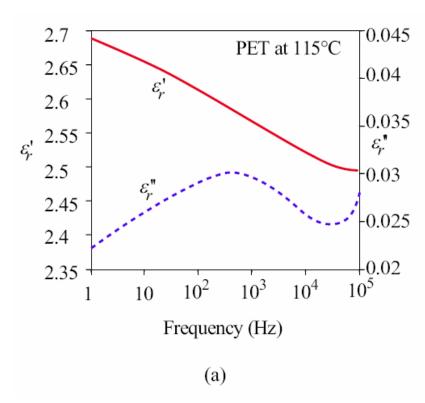
• Commercial flat panel Ba-Si-O ϵ_r = 8
• 40% Ba- 20% Ti- 40% Si-O ϵ_r = 15
• Ta_2O_5 ϵ_r = 25
• Nb_2O_5

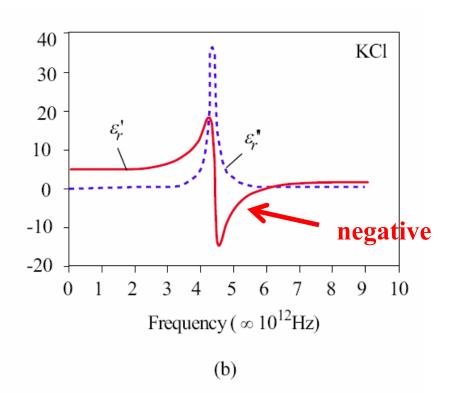
Mainly electronic and ionic contributions

Frequency (or time) Response

- Relaxation Response
 - Based on diffusion mechanisms
 - Significant damping in oscillations
 - Describes Dipolar and Space charge mechanisms
- Resonance Response
 - High frequency response
 - Not as much damping as relaxation response

Relaxation vs Resonant Response





(a) Real and imaginary part is of the dielectric constant, ε_r ' and ε_r " versus frequency for (a) a polymer, PET, at 115 °C and (b) an ionic crystal, KCl, at room temperature. both exhibit relaxation peaks but for different reasons.

SOURCE:

(b) from C. Smart, G.R. Wilkinson, A. M. Karo, and J.R. Hardy, International Conference on lattice Dynamics, Copenhagen, 1963, as quoted by D. G. Martin, "The Study of the Vibration of Crystal Lattices by Far Infra-Red Spectroscopy," *Advances in Physics*, 14, no. 53-56, 1965, pp. 39-100.

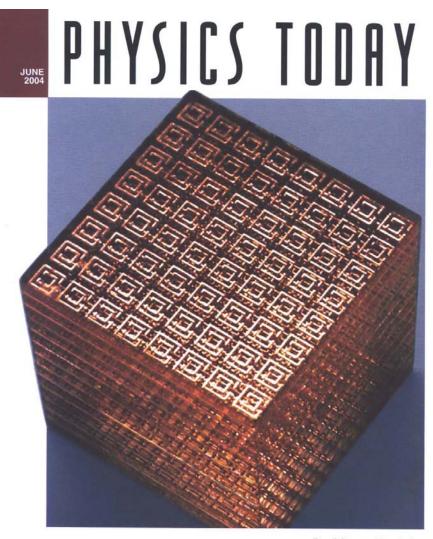
Summary of Dielectric Response

- 4 basic mechanisms with each mechanism having a characteristic frequency response
- Glasses potentially have electronic, ionic and space charge contributions
- Highest permittivity for a glass is less than
 20
- Discussion point is rotational polarization possible in glass?

Metamaterials

Mike Lanagan, Khalid Rajab, Masato Iwasaki, Doug Werner and Elena Semouchkina Materials Research Institute Penn State University

Metamaterials Reading assignment

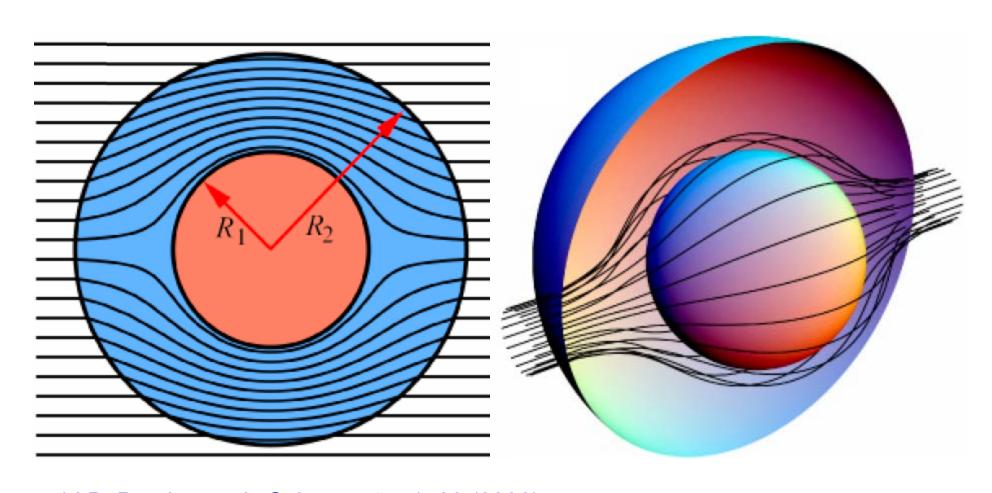


Positive outlook for negative refraction

Metamaterials (Based on negative permittivity)

- Description and Definition of Metamaterials
- Discovery and Application
- Creating materials with a resonant response
 - Plasmonic resonances for optics (not covered here)
 - Dielectric Resonators (interesting for Microwave and THz)
- Why Glass is an Interesting Medium for Metamaterials
 - Low dielectric loss
 - Easy to create spheres and periodic structures
 - Particular interest for mm-wave and THz frequencies

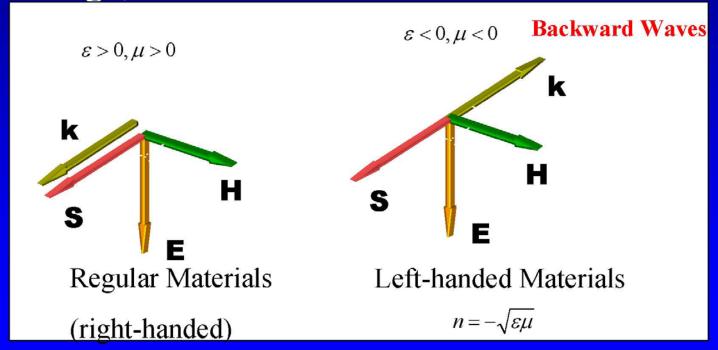
Electromagnetic Cloaking Using Metamaterials



*J.B. Pendry et al., Science **312**, 1780 (2006).

LEFT-HANDED ε<0 AND μ<0 METAMATERIALS

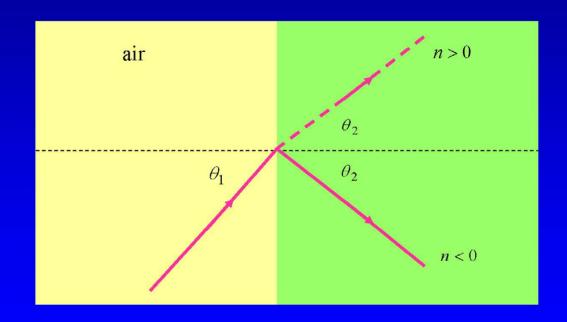
Veselago, 1960s



Negative-Refractive-Index (NRI) Materials

George V. Eleftheriades//University of Toronto

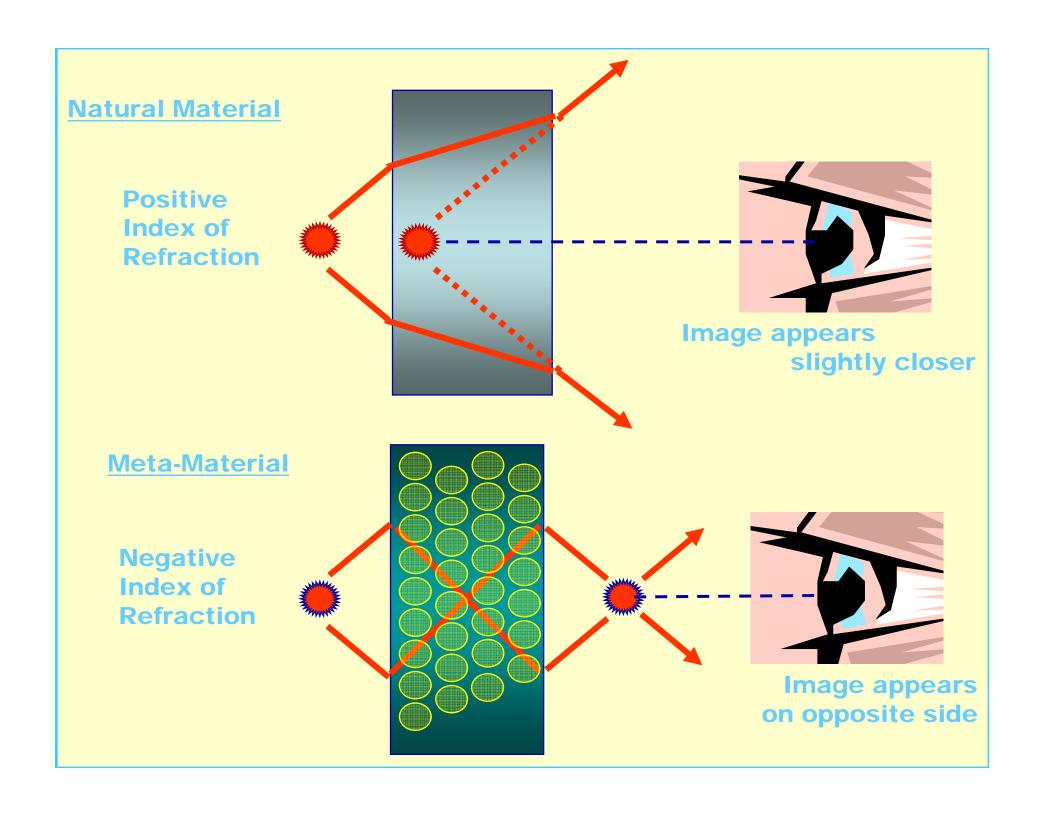
NEGATIVE REFRACTION



$$\frac{\sin \theta_1}{\sin \theta_2} = n$$

Negative-Refractive-Index (NRI) Media

George V. Eleftheriades//University of Toronto



Discovery of Metamaterials

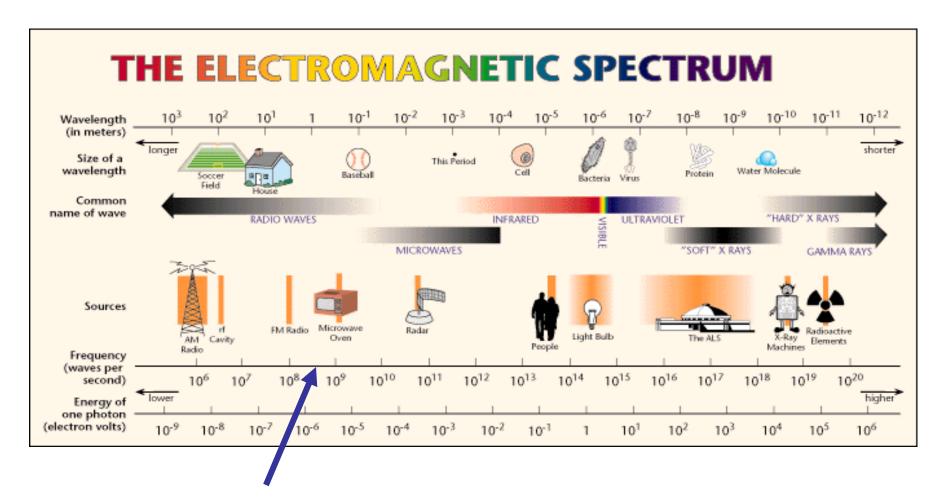
- Predicted by Veselago in 1960s
- First Experiments at UC San Diego in 2000
- Significant interest for applications
 - Magnetic resonant imaging
 - THz imaging
 - Cloaking

How can one make metamaterials?

- Think of resonance
 - Result of standing waves
 - Function of the wavelength and structure size
- We will use ring resonators as a example



Source: Wikipedia



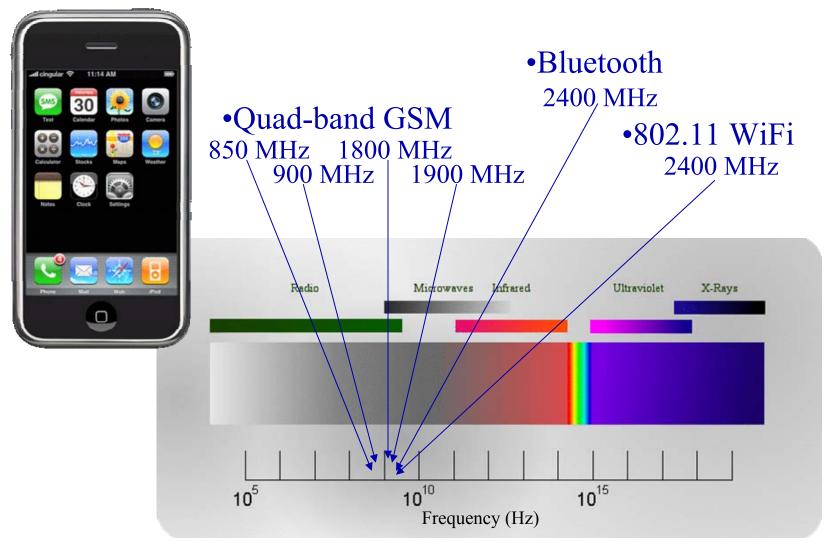
Microwave: Resonator Size should be in centimeters

Resonant frequency

$$f_r \propto \frac{1}{d\varepsilon_r}$$

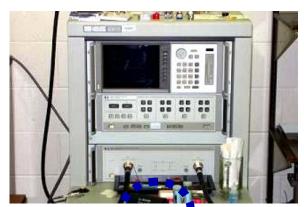
d=Resonator size

Apple iPhone and Microwaves

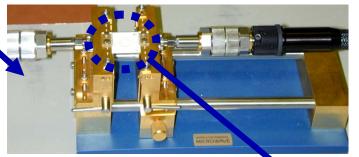


Jeremiah P. Turpin EEREU Symposium 2007

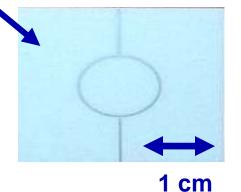
Ring Resonator Measurements



HP8510T Network Analyzer 45 MHz to 26 GHz

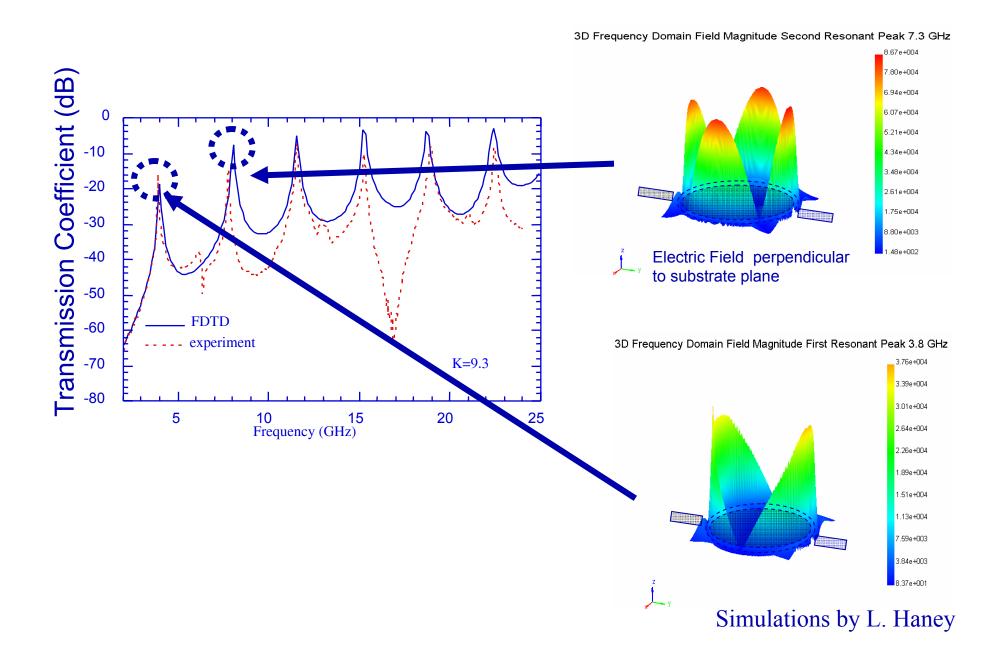


Intercontinental Microwave Fixture

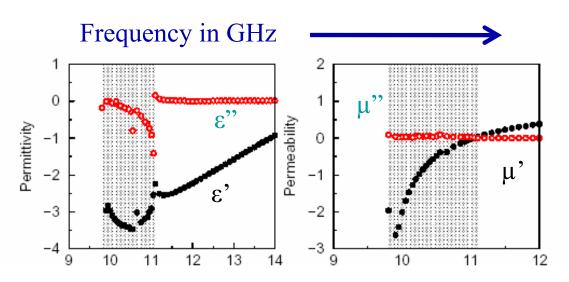


Ring Resonator

Resonant Behavior in Ring Resonators



Double Negative Materials*

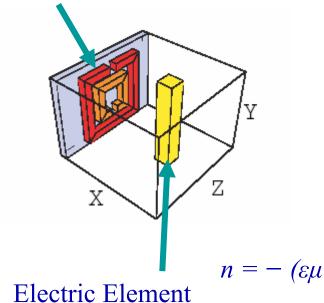


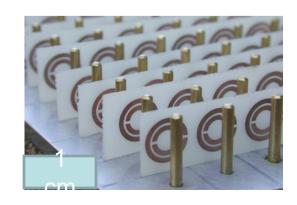
Negative permeability results from the resonating magnetic element

Negative permeability results from the resonating electric element

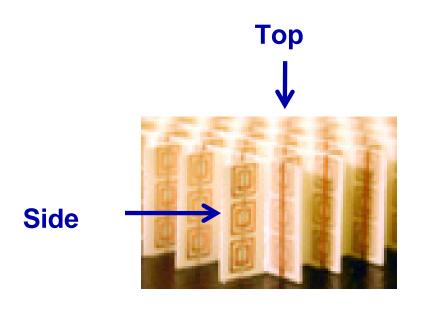
Recall
$$n = -(\epsilon \mu)^{1/2}$$



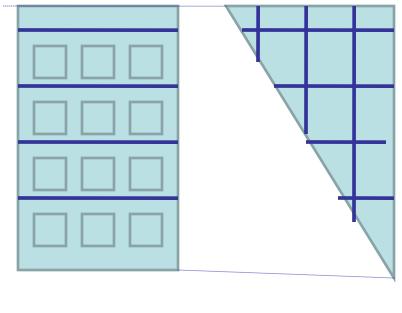




Critical Experiment for Metamaterial



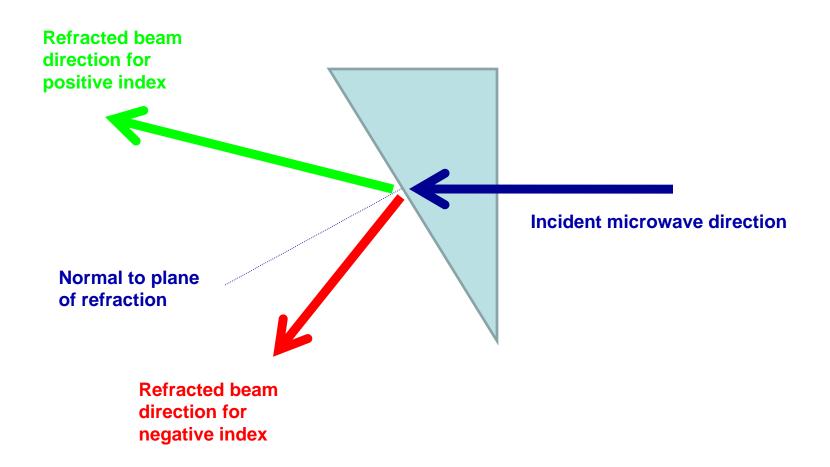
First make a metamaterial prism



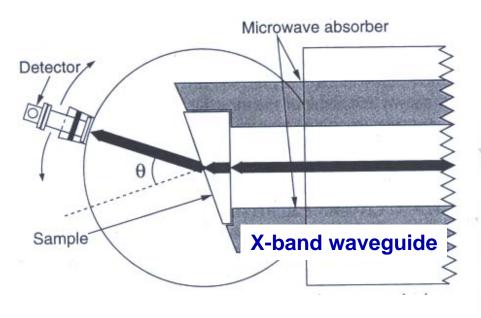
Side View

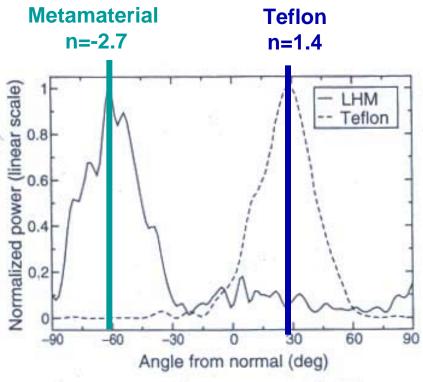
Top View

Critical Experiment for Metamaterial



Experimental Confirmation of a Meta-material*







Why not other resonant structures for Metamaterials?

Ceramic Cylinders



Microwave ceramic resonators made by Murata

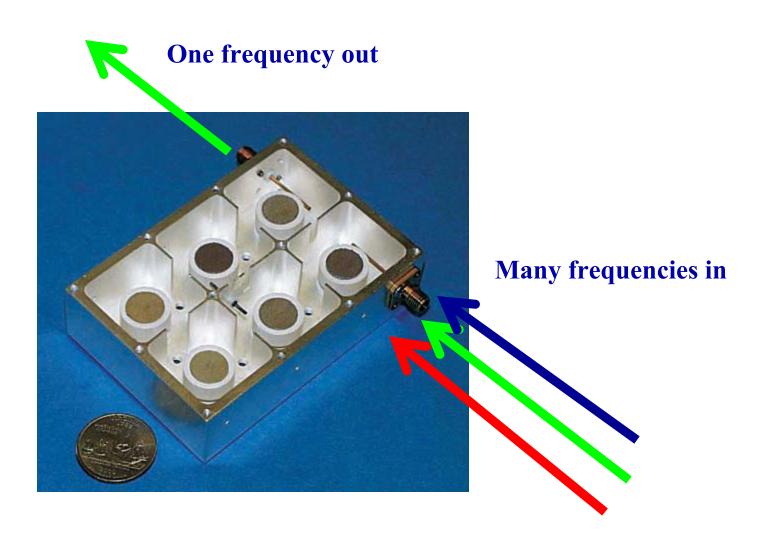
Glass Spheres



1 mm diameter silica spheres. Fabricated by Amanda Baker

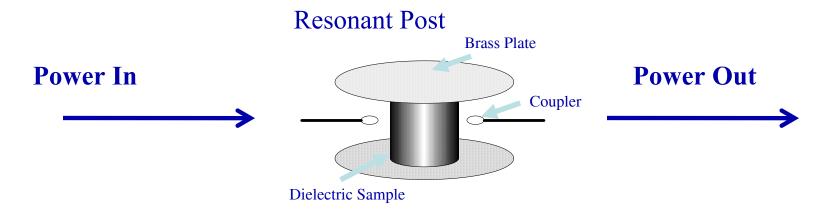
Dielectric properties and geometry are key factors for resonators

Microwave Filter for Cell Phone Base Station: Commercial Application of Ceramic Dielectric Resonators



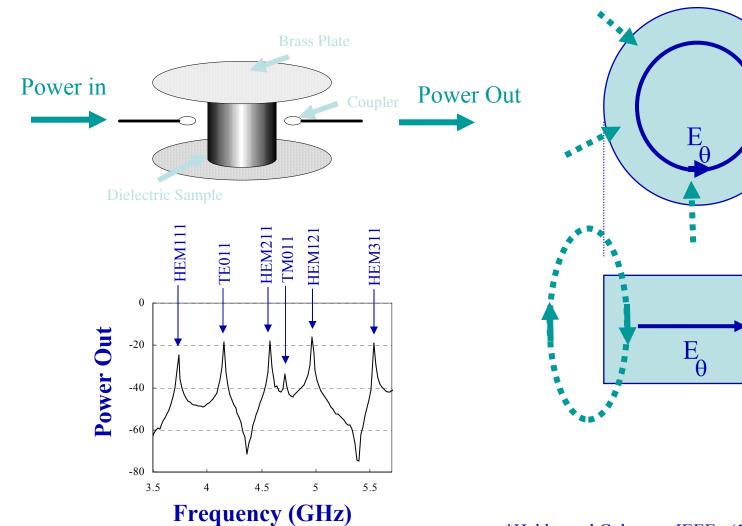
Microwave Characterization





Resonant Post Method*

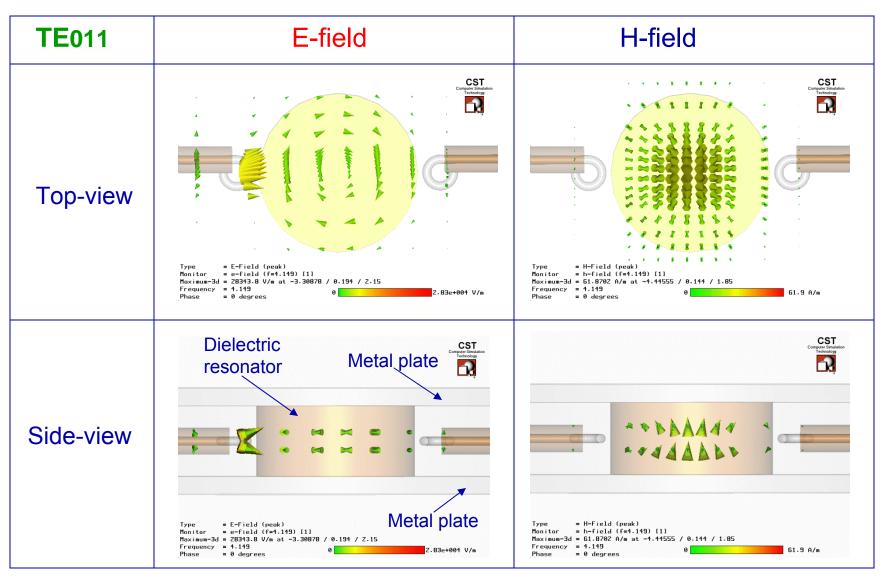




TE₀₁₁ Mode

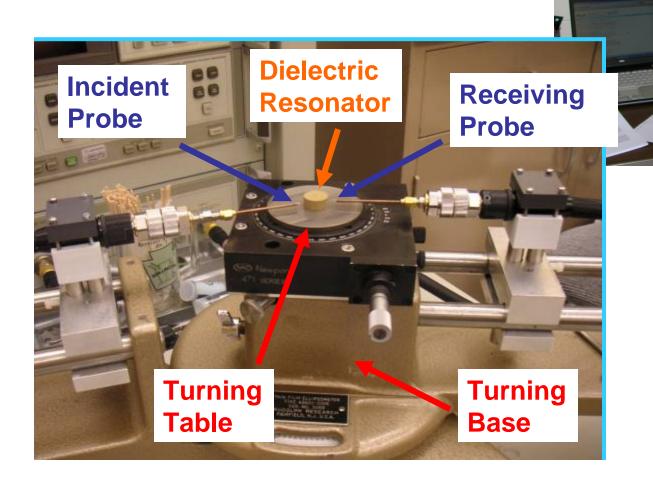
*Hakke and Coleman, IEEE (1960)

Field Distribution of TE₀₁₁ Mode from FDTD Simulation

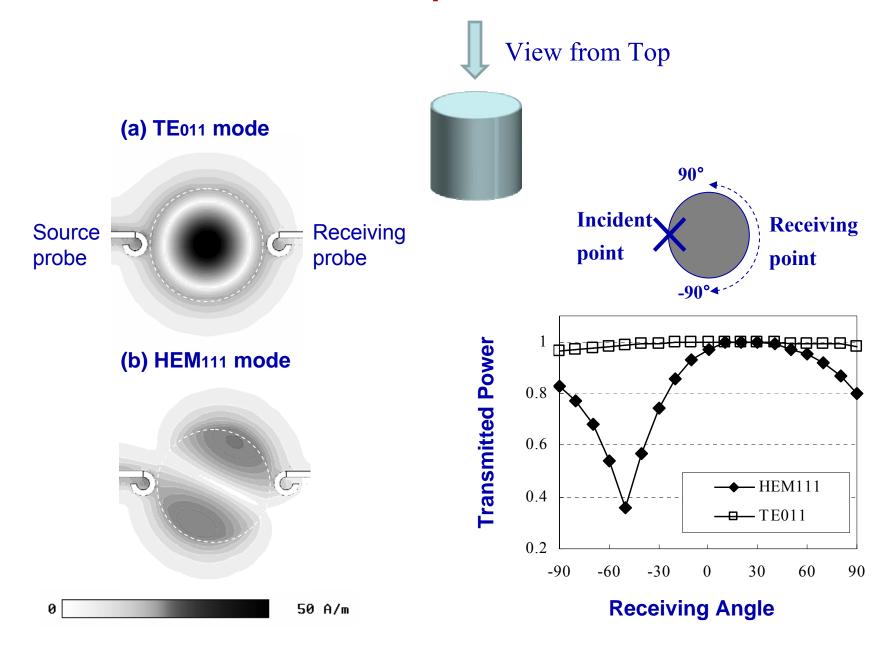


Masato "Mat" Iwasaki, Visiting Scientist NGK Spark Plug

- Meta-materials
- Electromagnetic Simulation

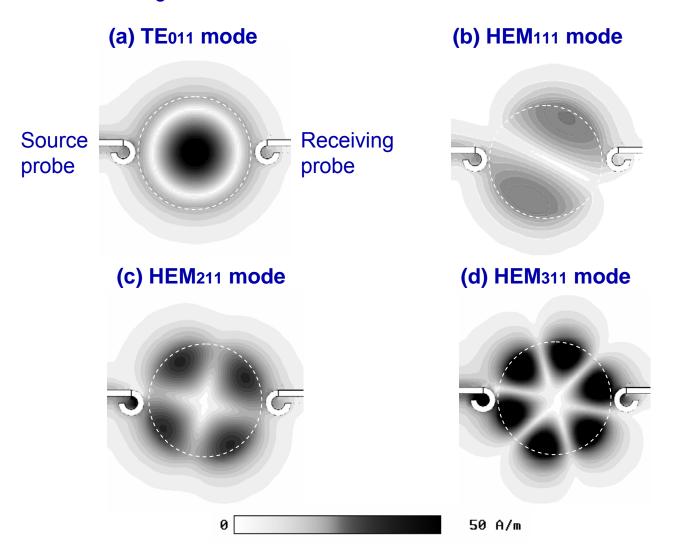


Simulation and Experimental Results



Electric field distribution of single DR

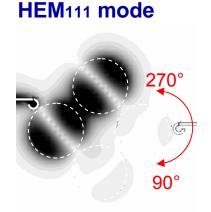
-By simulation results, magnetic field distributions were drawn in longitudinal direction at the half height of DR.



Field Distribution of Square DR Cluster

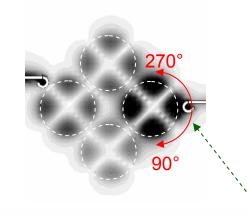
-Magnetic field distributions in longitudinal direction at the half height of DR were drawn.

No HEM 111 mode propagation for square symmetry

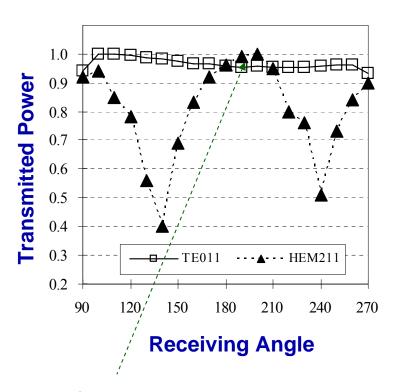


HEM₂₁₁ mode

HEM 211 mode propagation for square symmetry is consistent with lattice symmetry



0

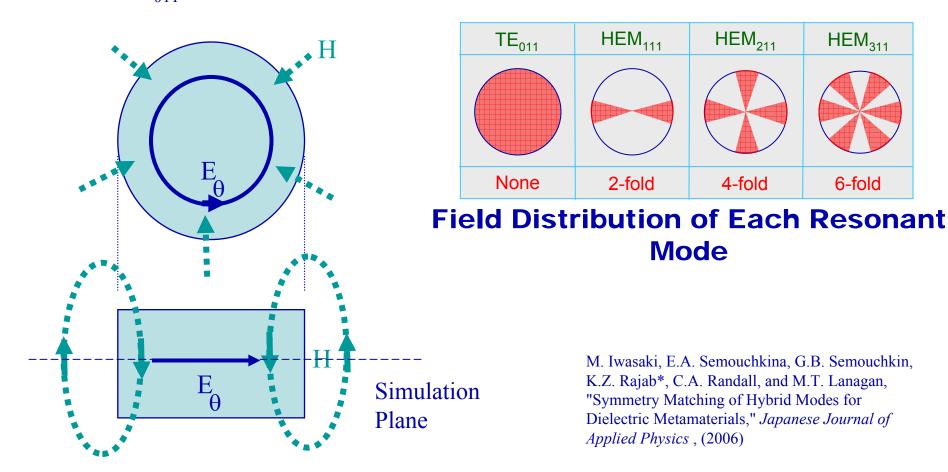


Magnetic field maximum

30 A/m

Magnetic Field Symmetry in Dielectric Resonator Modes

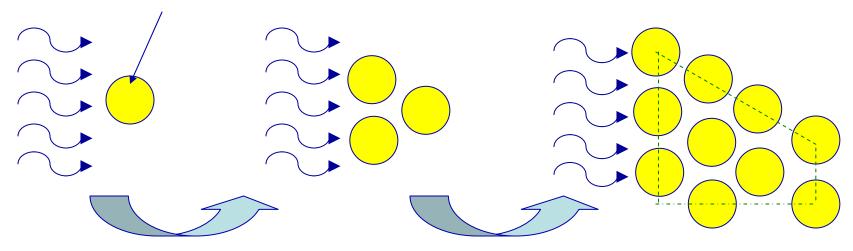
TE₀₁₁ Mode



Moving from individual resonators to clusters to arrays

- -For characterizing the refracted waves through wedge-shaped DR arrays, Simulations and measurements starting from one DR will be performed.
- -Excitation with large area should be employed.

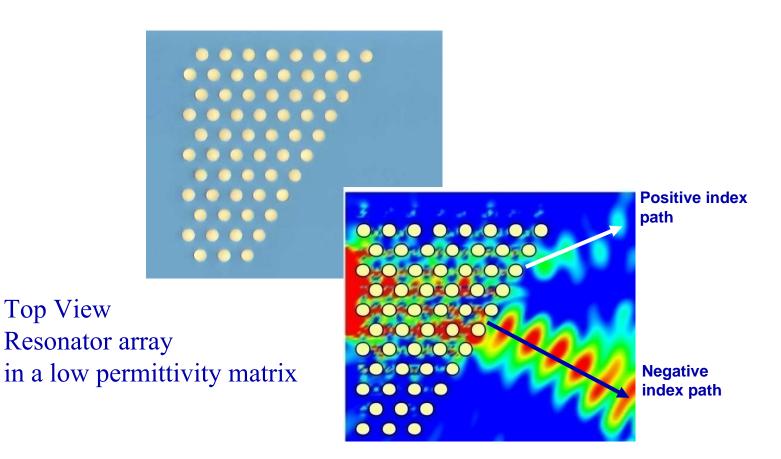
Dielectric Resonator



Increasing the number of DRs to form wedge shape

→ : Incident waves

Ceramic Dielectric Resonator Arrays

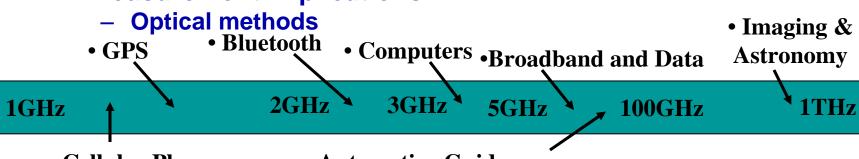


Simulation by Elena Semouchkina

Operating Frequency 15 GHz

Moving Beyond Microwaves

- Materials Trends
 - Higher application frequencies (both communications and computing)
 - Lower permittivity (dielectric constant) and lower loss (higher Q)
 - All dielectric (no metal?) structures
- Design and Process Implications
 - More compact designs
 - Dimensional control becomes more critical
- Measurement Implications

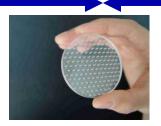


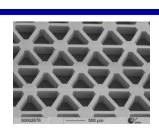
• Cellular Phone

Automotive Guidance

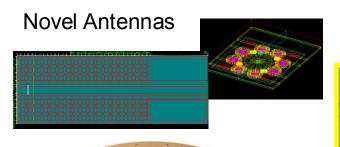


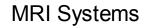
Lumped Elements



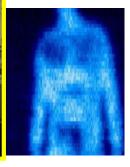


Distributed Elements







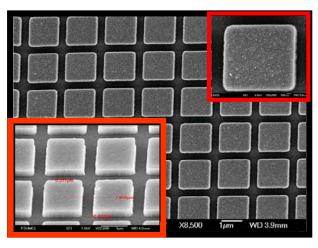


THz Imaging



Systems Level: Innovative devices for precision measurement, shielding, imaging, telecommunications, energy, and biomedicine

Metamaterials for IR Devices



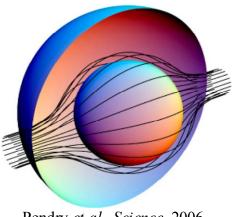
Superlens-based Nanopatterning



Fang et al, Science, 2005



EM Cloaking



Pendry et al., Science, 2006

THz Characterization of Arrays

Materials:

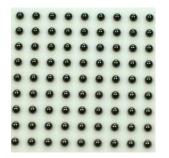
- Silicon Nitride, Si₃N₄
 ε_r≈8.9
- Brass

□ Lattices:

- Square
- Hexagonal

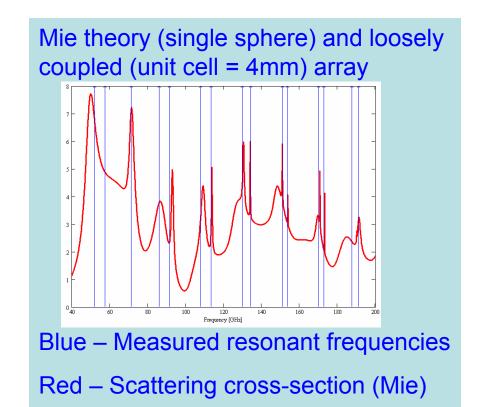
Unit cells:

- 4mm
- 3mm
- 2mm









What's Next for Ceramic Dielectric Materials and Structures?

- Higher Frequencies pushing into the THz range
- What size resonators do we need?
- What types of dielectrics (glass?) do we need?



1 mm diameter silica spheres. Fabricated by Amanda Baker

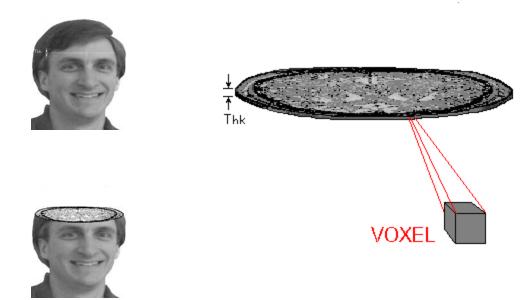
Quiz

- What material property affects the resonator frequency?
- What other parameter affects resonance?
- Why would we NOT want to make the resonator too small?
- What functionalities of glass are potentially important for metamaterials?

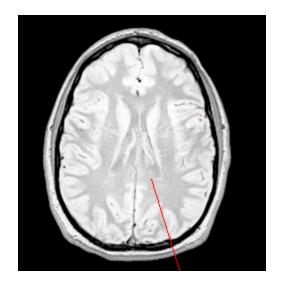
Metamaterials in Magnetic Resonance Imaging?

- Overview of how MRI works
- Use of resonators in MRI (not metamaterials yet)
- Case Study: Glass Metamaterials for MRI

Background on Magnetic Resonance Imaging (MRI)







Magnetic resonance imaging is based on the absorption and emission of energy in the radio frequency range of the electromagnetic spectrum.

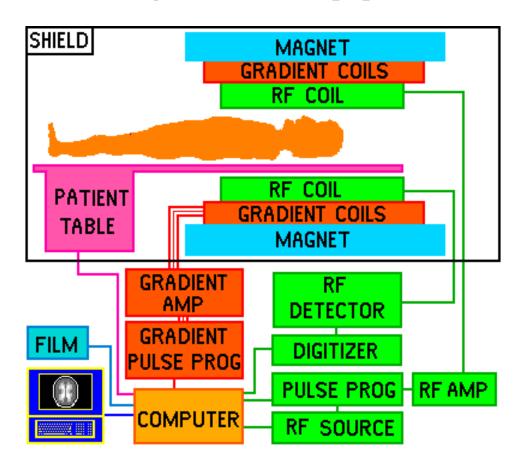
Radio Frequency (RF) Coils are used to transmit and receive energy from the samples.

Background on MRI contd.

- MRI is based on spatial variations in the phase and frequency of the radio frequency energy being absorbed and emitted by the imaged object.
- Important microscopic property responsible for MRI is the spin property within hydrogen nuclei
- The human body is primarily fat and water.
 Fat and water have many hydrogen atoms which make the human body approximately 63%

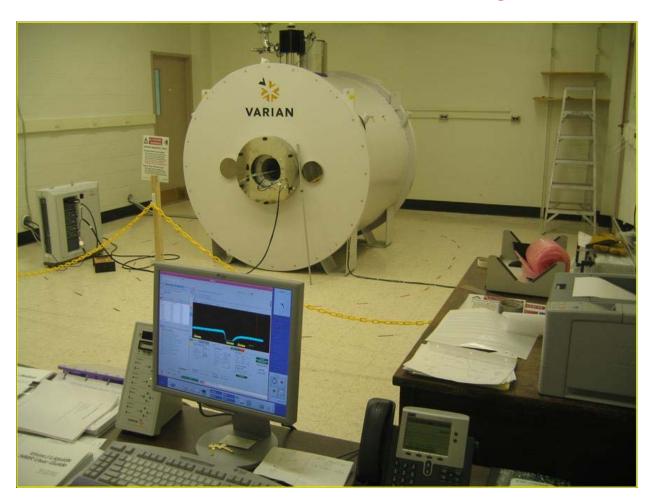
hydrogen atoms

Block diagram of MRI Equipment



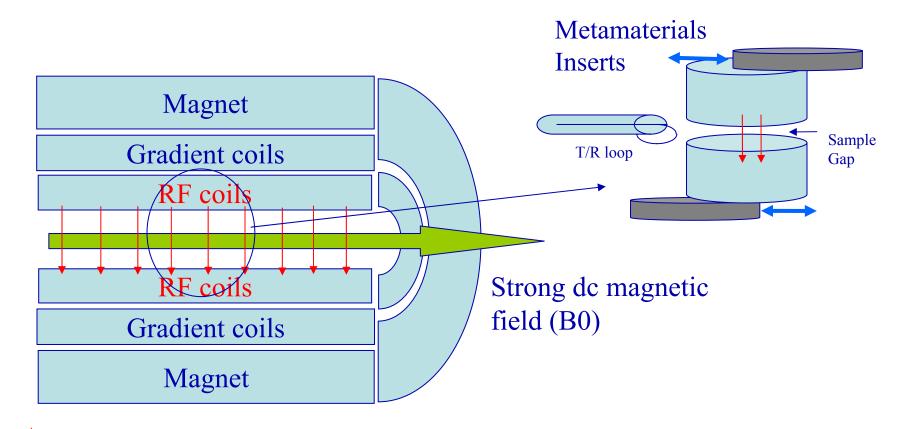
(Adapted from Reference: http://www.cis.rit.edu/htbooks/mri)

7 Tesla MRI device at the NMR spectroscopy lab



Andrew Webb Penn State University

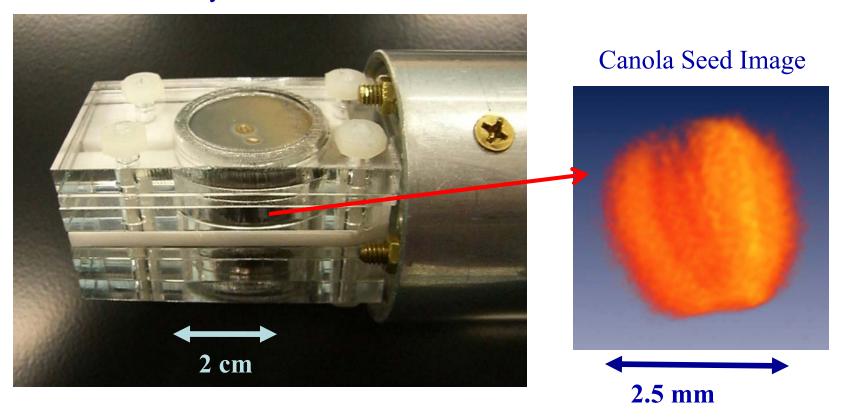
Schematic MRI System



= Time varying transverse field (B1) produced by the RF coils Depending on the magnetic field the rf field varies between 100 and 1,000

Imaging a Canola Seed

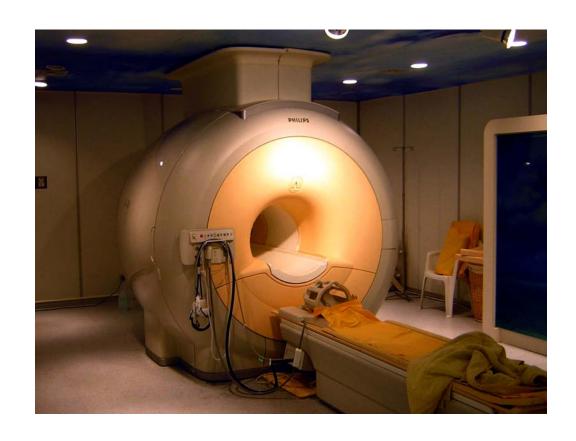
Ceramic Cylinder as an MRI insert



Elena Semouchkina, Varun Tyagi, Michael Lanagan, Amanda Baker, Andrew Webb, Thomas Neuberger

Case Study

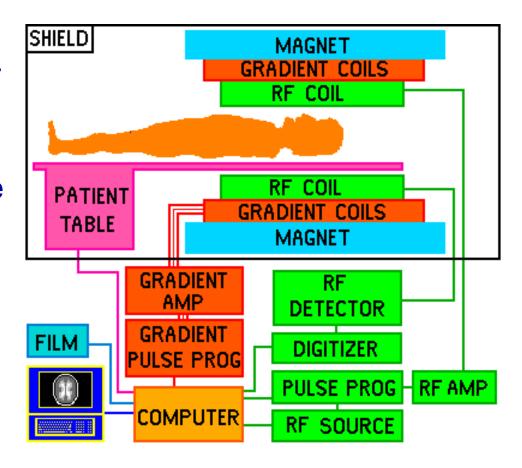
- Design a resonator for a 3Tesla MRI
- Frequency = 300 MHz



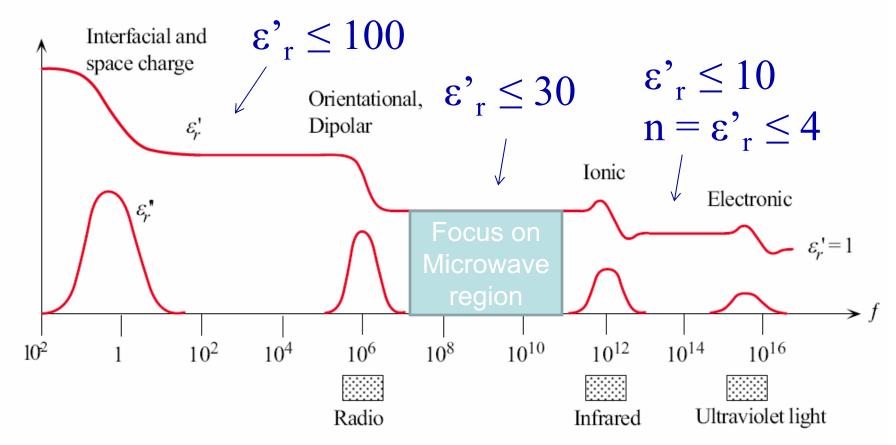
Case Study

- First think of the wavelength for the resonator replacing RF coil insert
- What will be size of the glass resonator
 - How do we shrink the size
 - Do you think that loss is important?

Block diagram of MRI Equipment



Frequency Response of Dielectric Polarization



The frequency dependence of the real and imaginary parts of the dielectric constant in the presence of interfacial, orientational, ionic, and, electronic polarization mechanisms.

How do we make a high permittivity glass?

$$\mathcal{E}_r = 1 + \frac{N\alpha_e}{\mathcal{E}_o}$$

 ε_r = relative permittivity

N = number of molecules per unit volume

 α_{e} = electronic polarizability

 ε_o = permittivity of free space

Assumption: Only ionic and electronic polarization is present

Summary of glass as a dielectric

- Dielectric response for glass occurs over a wide frequency range
- New applications for dielectrics could involve glass
- Functionality of glass
 - Related to dielectric properties (permittivity and loss
 - Formability and cost

Dielectric Loss and Q Factor

Loss Tangent
$$\tan \delta = \frac{\mathcal{E}_r''}{\mathcal{E}_r'}$$

$$Q = \frac{1}{\tan \delta} = \frac{energy_stored}{energy_dissipated}$$

 \mathcal{E}'_r = real part of the complex dielectric constant, \mathcal{E}''_r = imaginary part of the complex dielectric constant