# 1 Modeling NH<sub>4</sub>NO<sub>3</sub> over the San Joaquin Valley during the 2013 DISCOVER-AQ campaign

- 2 James T. Kelly<sup>1</sup>, Caroline L. Parworth<sup>2,3</sup>, Qi Zhang<sup>2,3</sup>, David J. Miller<sup>4</sup>, Kang Sun<sup>5</sup>, Mark A. Zondlo<sup>6</sup>, Kirk R.
- 3 Baker<sup>1</sup>, Armin Wisthaler<sup>7</sup>, John B. Nowak<sup>8</sup>, Sally E. Pusede<sup>9</sup>, Ronald C. Cohen<sup>10</sup>, Andrew J. Weinheimer<sup>11</sup>,
- 4 Andreas J. Beyersdorf<sup>12</sup>, Gail S. Tonnesen<sup>13</sup>, Jesse O. Bash<sup>14</sup>, Luke C. Valin<sup>14</sup>, James H. Crawford<sup>8</sup>, Alan
- 5 Fried<sup>15</sup>, and James G. Walega<sup>15</sup>
- 6 <sup>1</sup>U.S. Environmental Protection Agency, Office of Air Quality Planning & Standards, RTP, NC
- 7 <sup>2</sup>Department of Environmental Toxicology, University of California, Davis, CA
- 8 <sup>3</sup>Agricultural and Environmental Chemistry Graduate Group, University of California, Davis, CA
- 9 <sup>4</sup>Environmental Defense Fund, Boston, MA
- <sup>5</sup>Atomic and Molecular Physics Division, Harvard-Smithsonian Center for Astrophysics, Cambridge, MA
- <sup>6</sup>Department of Civil and Environmental Engineering, Princeton University, Princeton, NJ
- <sup>12</sup> <sup>7</sup>Institute for Ion Physics and Applied Physics, University of Innsbruck, Innsbruck, Austria
- 13 <sup>8</sup>National Aeronautics and Space Administration, Langley Research Center, Hampton, VA
- <sup>9</sup>Department of Environmental Sciences, University of Virginia, Charlottesville, VA
- 15 <sup>10</sup>Department of Earth and Planetary Sciences, University of California at Berkeley, Berkeley, CA
- 16 <sup>11</sup>National Center for Atmospheric Research, Boulder, CO
- 17 <sup>12</sup>Department of Chemistry and Biochemistry, California State University, San Bernardino, CA
- 18 <sup>13</sup>U.S. Environmental Protection Agency, Region 8, Denver, CO
- 19 <sup>14</sup>U.S. Environmental Protection Agency, Office of Research and Development, RTP, NC
- 20 <sup>15</sup>Institute of Arctic and Alpine Research, University of Colorado, Boulder, CO 80309, USA
- 21 Corresponding author: James T. Kelly, Tel: 919-541-0886, Email: <u>kelly.james@epa.gov</u>
- 22 Abstract
- 23 The San Joaquin Valley (SJV) of California experiences high concentrations of PM<sub>2.5</sub> (particulate matter with
- 24 aerodynamic diameter  $\leq$  2.5  $\mu$ m) during episodes of meteorological stagnation in winter. Modeling PM<sub>2.5</sub>
- 25 NH<sub>4</sub>NO<sub>3</sub> during these episodes is challenging because it involves simulating meteorology in complex terrain
- 26 under low wind speed and vertically stratified conditions, representing complex pollutant emissions
- 27 distributions, and simulating daytime and nighttime chemistry that can be influenced by the mixing of
- 28 urban and rural air masses. A rich dataset of observations related to NH<sub>4</sub>NO<sub>3</sub> formation was acquired
- 29 during multiple periods of elevated NH<sub>4</sub>NO<sub>3</sub> during the DISCOVER-AQ (Deriving Information on Surface
- 30 Conditions from Column and Vertically Resolved Observations Relevant to Air Quality) field campaign in SJV
- in January and February 2013. Here, NH<sub>4</sub>NO<sub>3</sub> is simulated during the SJV DISCOVER-AQ study period with
- 32 the Community Multiscale Air Quality (CMAQ) model version 5.1, predictions are evaluated with the
- 33 DISCOVER-AQ dataset, and process analysis modeling is used to quantify HNO<sub>3</sub> production rates. Simulated
- NO<sub>3</sub><sup>-</sup> generally agrees well with routine monitoring of 24-h average NO<sub>3</sub><sup>-</sup>, but comparisons with hourly
- 35 average NO<sub>3</sub><sup>-</sup> measurements in Fresno revealed differences at higher time resolution. Predictions of gas-
- particle partitioning of total nitrate ( $HNO_3 + NO_3^{-1}$ ) and NHx ( $NH_3 + NH_4^{+1}$ ) generally agreed well with
- 37 measurements in Fresno, although partitioning of total nitrate to HNO<sub>3</sub> was sometimes overestimated at
- 38 low relative humidity in afternoon. Gas-particle partitioning results indicate that NH<sub>4</sub>NO<sub>3</sub> formation is

39 limited by HNO<sub>3</sub> availability in both the model and ambient. NH<sub>3</sub> mixing ratios are underestimated,

40 particularly in areas with large agricultural activity, and the spatial allocation of NH<sub>3</sub> emissions could benefit

41 from additional work, especially near Hanford. HNO<sub>3</sub> production via daytime and nighttime pathways is

42 reasonably consistent with the conceptual model of NH<sub>4</sub>NO<sub>3</sub> formation in SJV, and production peaked aloft

43 between about 160 and 240 m in the model. During a period of elevated NH<sub>4</sub>NO<sub>3</sub>, the model predicted that

44 the OH +  $NO_2$  pathway contributed 46% to total HNO<sub>3</sub> production in SJV and the  $N_2O_5$  heterogeneous

45 hydrolysis pathway contributed 54%. The relative importance of the OH + NO<sub>2</sub> pathway for HNO<sub>3</sub>

46 production is predicted to increase as NOx emissions decrease.

47

## 48 1. Introduction

49 The San Joaquin Valley (SJV or Valley) makes up the southern portion of California's Central Valley and 50 is formed by the coastal mountain ranges in the west, the Sierra Nevada mountains in the east, and the 51 convergence of mountain ranges in the south at the Tehachapi mountains. SJV is about 400 km long and 52 60-100 km wide and includes parts or all of eight counties having a combined population of about 4.2 53 million [CDOF, 2017]. The Valley population is projected to increase rapidly in coming decades, by ~60% 54 from 2016 to 2060 [CDOF, 2017], which has implications for air quality planning [Hixson et al., 2012]. SJV 55 contains major cities such as Fresno (pop. ~520,000) and Bakersfield (pop. ~380,000), important oil and gas 56 fields [CDOC, 2015; Gentner et al., 2014], and an extremely productive agricultural region [CDFA, 2016a]. 57 For instance, SJV had about 1.5 million dairy cows and produced about 36 billion pounds of milk in 2016 58 [CDFA, 2016b]. The Valley is also a major north-south corridor for goods transport along Highway 99 in the 59 east and Interstate 5 in the west. SJV's terrain combined with pollutant emissions from the large 60 population and economic activity leads to high concentrations of PM<sub>2.5</sub> (particulate matter with 61 aerodynamic diameter  $\leq 2.5 \,\mu$ m), particularly during periods of stagnant meteorology in winter months. SJV 62 is in nonattainment of U.S. EPA's primary national ambient air quality standards for  $PM_{2.5}$  that are set to protect public health. 63

Air pollution in SJV has been studied for decades, and conceptual models of wintertime PM<sub>2.5</sub> formation in SJV have been developed, largely based on the 1995 Integrated Monitoring Study and the 2000/2001 California Regional PM<sub>10</sub>/PM<sub>2.5</sub> Air Quality Study (CRPAQS) [*Herner et al.*, 2005; *Herner et al.*, 2006; *Pun and Seigneur*, 1999; *Watson and Chow*, 2002; *Watson et al.*, 1998]. Briefly, high pressure systems over the Great Basin lead to subsidence temperature inversions over the Valley that limit daytime mixing heights from less than 400 to ~800 m for periods of days to more than a week. Radiation temperature inversions 70 also form overnight and limit mixing of surface emissions to a ~30-50 m layer that is decoupled from the 71 residual layer above. Primary carbonaceous particles are concentrated in the shallow nighttime surface 72 layer. Between the radiation inversion and the subsidence inversion, air masses rich in oxides of nitrogen, 73 largely from urban areas and major highways, mix with air masses rich in NH<sub>3</sub>, largely from rural agricultural 74 areas, in a valley-wide layer overnight. Ammonium nitrate  $(NH_4NO_3)$  forms in this layer and mixes to the 75 surface in the morning when the radiation inversion breaks. Morning increases in NH<sub>4</sub>NO<sub>3</sub> at the surface 76 therefore tend to coincide with decreases in carbonaceous PM<sub>2.5</sub>. NH<sub>4</sub>NO<sub>3</sub> makes up a large fraction of fine 77 particle mass during major PM<sub>2.5</sub> episodes [e.g., L W A Chen et al., 2007; Chow et al., 2006; Ge et al., 2012a; 78 Herner et al., 2005; SJVAPCD, 2012]. Persistent radiation fogs also occur in SJV in wintertime, and the 79 chemistry of fine particles can be influenced by aqueous-phase processes [e.g., Collett et al., 1999a; Collett 80 et al., 1999b; Ge et al., 2012b; Herckes et al., 2015; Jacob et al., 1986].

81 Air quality models have been used in combination with the CRPAQS dataset to better understand air 82 pollution processes in SJV. Overall, models did a reasonable job of predicting PM during CRPAQS [Kelly et 83 al., 2011; Pun et al., 2009; Ying et al., 2008a; Y Zhang et al., 2010] and were used to provide information on 84 process rates, visibility impairment, and source apportionment and regional contributions to primary and 85 secondary PM [J Chen et al., 2009; 2010; Ying, 2011; Ying and Kleeman, 2009; Ying et al., 2008b; Ying et al., 86 2009]. Air quality models have also been used to understand the impact of precursor emissions on NH<sub>4</sub>NO<sub>3</sub> 87 [Blanchard et al., 2000; J Chen et al., 2014; Kleeman et al., 2005; Livingstone et al., 2009; Pun et al., 2009; 88 Pun and Seigneur, 2001; Stockwell et al., 2000]. Generally, these studies found that NOx (NO + NO<sub>2</sub>) 89 emission reductions would be the most effective emission control for reducing NH<sub>4</sub>NO<sub>3</sub> in SJV. Air quality 90 management strategies based on in part on NOx emission reductions, which are also important for reducing 91 ozone in the Valley, have been implemented [e.g., SJVAPCD, 2012; SJVAPCD, 2016].

92 The studies discussed above were largely based on PM<sub>2.5</sub> episodes that occurred one to two decades 93 ago. Although these studies are still relevant, air quality has improved over time in SJV due to reductions in 94 NOx and other emissions [e.g., McDonald et al., 2012; Pusede and Cohen, 2012; Pusede et al., 2016; Pusede 95 et al., 2014; Russell et al., 2012]. The DISCOVER-AQ (Deriving Information on Surface Conditions from 96 Column and Vertically Resolved Observations Relevant to Air Quality) campaign in January and February of 97 2013 provides a rich dataset for more recent wintertime PM<sub>2.5</sub> episodes in SJV. Peak PM<sub>2.5</sub> concentrations 98 were lower during DISCOVER-AQ than CRPAQS, but  $NH_4NO_3$  still made up a large fraction of fine particle 99 mass consistent with the earlier study. The DISCOVER-AQ dataset has recently been used to investigate 100 PM<sub>2.5</sub> precursor emissions and formation processes in SJV [e.g., Miller et al., 2015; Parworth et al., 2017; 101 Prabhakar et al., 2017; Pusede et al., 2016; Shephard and Cady-Pereira, 2015; Sun et al., 2015; Young et al.,

2016; *X L Zhang et al.*, 2016]. In particular, Pusede et al. [2016] used the DISCOVER-AQ dataset in
 combination with the historical monitoring record to interpret past trends and predict future trends in
 NH<sub>4</sub>NO<sub>3</sub> in SJV. They found that NH<sub>4</sub>NO<sub>3</sub> formation is limited by NOx emissions, both daytime and
 nighttime formation pathways are important, and predict that the daytime pathway will become
 increasingly important in the future.

107 Previous studies using the 2013 SJV DISCOVER-AQ dataset were generally based on conceptual, 108 analytical, and box modeling in combination with the measurements. Compared with earlier field 109 campaigns, limited regional photochemical modeling has been done for SJV DISCOVER-AQ. Regional 110 photochemical modeling is valuable because it provides comprehensive information on key processes in 111 three-dimensions across the entire region for the entire period and is constrained by relatively few 112 assumptions. The lack of constraints in air quality modeling is advantageous for exploring alternative 113 scenarios but requires that models be thoroughly evaluated to insure they adequately reflect ambient 114 processes. Modeling PM<sub>2.5</sub> episodes in SJV is particularly challenging because it involves simulating 115 meteorology in complex terrain under low wind speed and vertically stratified conditions, representing 116 complex pollutant emissions distributions, and simulating daytime and nighttime chemistry that can be 117 influenced by the mixing of urban and rural air masses. Reliable modeling of  $PM_{2.5}$  in SJV is important, however, to help inform air quality management for the highly populated nonattainment area. Here, the 118 119 DISCOVER-AQ dataset is used to perform a thorough evaluation of regional photochemical modeling of 120 NH₄NO₃ in the Valley during January and February 2013. Process analysis modeling is also conducted to 121 help interpret model predictions and contribute to the understanding of air pollution in the Valley.

122 2. Methods

123 2.1 Modeling

124 Photochemical grid modeling was performed with the Community Multiscale Air Quality (CMAQ; 125 www.epa.gov/cmaq) model version 5.1 [Appel et al., 2017] on a domain covering SJV from south of the 126 Tehachapi mountains to north of Sacramento and parts of the Sierra Nevada mountains in the east and 127 Pacific Ocean in the west (Figure S1). The CMAQv5.1 simulations were configured with integrated reaction 128 rate and process analysis [Jang et al., 1995; Jeffries and Tonnesen, 1994; Kim et al., 2014] and covered the 4 129 January – 10 February 2013 period. Horizontal grid resolution of 4 km was used with 35 vertical layers that 130 matched the vertical structure of the meteorological model. Chemical boundary conditions were 131 developed from a CMAQv5.1 simulation that covered the contiguous U.S. and surrounding areas with 12-132 km horizontal resolution. NH<sub>3</sub> surface exchange was simulated with CMAQ's bi-directional exchange

133 parameterization [Bash et al., 2013; Pleim et al., 2013], and gas-phase chemistry was parameterized with 134 the CB05e51 mechanism [Appel et al., 2017]. Inorganic aerosol thermodynamics were simulated with 135 ISORROPIA II [Fountoukis and Nenes, 2007] in metastable mode, where crystallization does not occur. 136 Semi-volatile inorganic particle components (i.e.,  $NO_{3}^{-}$ ,  $NH_{4}^{+}$ ,  $CI^{-}$ ) in the Aitken and accumulation modes are 137 assumed to be in bulk equilibrium with their gas-phase counterparts (i.e., HNO<sub>3</sub>, NH<sub>3</sub>, and HCl) in CMAQ, 138 whereas diffusive mass transfer is explicitly simulated for semi-volatile coarse-mode particle components 139 [Kelly et al., 2010]. Heterogeneous hydrolysis of  $N_2O_5$  on Aitken- and accumulation-mode particles is based 140 on Davis et al. [2008], and N<sub>2</sub>O<sub>5</sub> hydrolysis on coarse-mode particles is based on Bertram and Thornton 141 [2009] as described by Sarwar et al. [2012].

142 Gridded emission fields for CMAQ modeling were developed with the Sparse Matrix Operator Kernel 143 Emissions (SMOKE) model [Houyoux et al., 2000] version 3.7. The emissions modeling procedures used 144 here are similar to those described in detail previously for national 12-km resolution modeling [USEPA, 145 2017b]. Point source emissions were based on 2013 continuous emissions monitoring (CEM) data when available and state submitted data otherwise. Anthropogenic non-point source emissions were based on 146 147 version 2 of the 2011 National Emission Inventory (NEI11v2) [USEPA, 2016]. Onroad mobile source 148 emission totals by county were estimated by interpolating totals from 2011 and 2014 based on EMFAC2014 149 (www.arb.ca.gov/emfac/) modeling by the California Air Resources Board (CARB). The interpolated onroad 150 emission totals were then temporally and spatially allocated using results of a MOVES2014a (Motor Vehicle 151 Emission Simulator; www.epa.gov/moves) simulation according to a hybrid procedure described previously 152 [USEPA, 2012b; 2017b]. Offroad mobile source emission totals were also based on information provided by 153 CARB. The Biogenic Emission Inventory System (BEIS) version 3.61 was used with the Biogenic Emissions 154 Landuse Database (BELD) version 4.1 to estimate biogenic NO and speciated VOC emissions [Bash et al., 155 2016].  $NH_3$  emissions from livestock and fertilizer application were based on NEI11v2 annual county totals 156 that were allocated to hour of day using 2013 temperature data [USEPA, 2016].

The Weather Research and Forecasting (WRF) model [*Skamarock et al.*, 2008] version 3.7 was used to generate gridded meteorological fields for CMAQ and SMOKE. WRFv3.7 was applied with 35 vertical layers from the surface to 50 mb with higher resolution near the surface to better resolve the planetary boundary layer (PBL). Key physics options used in the WRF simulation include the Pleim-Xiu land surface model [*Pleim and Xiu*, 2003], asymmetric convective mixing model version 2 [ACM2; *Pleim*, 2007], RRTMG short and longwave radiation parameterization [*Mlawer et al.*, 1997], and Morrison two-moment microphysics scheme [*Morrison et al.*, 2009].

165 Measurements of NH<sub>3</sub>, TNO3 (HNO<sub>3</sub> + fine particle NO<sub>3</sub><sup>-</sup>), NO, NO<sub>2</sub>, NOy (oxides of nitrogen including NOx, 166 HNO<sub>3</sub>, HNO<sub>4</sub>, HONO, NO<sub>3</sub> radical, organic nitrates, and N<sub>2</sub>O<sub>5</sub>), O<sub>3</sub>, and HCHO made from the NASA P-3B 167 aircraft during daytime flights on 16, 18, 20-22, and 30-31 January and 1, 4, and 6 February 2013 are used 168 to examine model performance. The aircraft flew 2-3 repeated circuits per day over SJV including vertical 169 spiral trajectories with ~5-km diameters over six sites (i.e., Bakersfield, Hanford, Tranquility, Fresno, Huron, 170 and Porterville). NH<sub>3</sub> was measured with a cavity ring down spectrometer (CRDS; G2103, Picarro Inc.) and a 171 proton-transfer-reaction time-of-flight mass spectrometer (PTR-ToF-MS). Measurements from these 172 instruments have been compared previously and were found to provide complementary information [Sun 173 et al., 2015]. Therefore both the CRDS and PTR-ToF-MS measurements are used here. TNO3 was measured 174 on the P-3B aircraft by thermal dissociation of ambient NOy species followed by laser-induced fluorescence 175 of NO<sub>2</sub>. Specifically, TNO3 was calculated as NO<sub>2</sub> measured in the 600°C channel minus that measured in 176 the 400°C channel with correction for slight conversion of HNO<sub>3</sub> in the alkyl nitrate channel [*Pusede et al.*, 177 2016; Womack et al., 2017]. NO, NO<sub>2</sub>, O<sub>3</sub>, and NOy were measured with the National Center for 178 Atmospheric Research (NCAR) four-channel chemiluminescence instrument. The NOy measurement likely 179 includes some contribution from NO<sub>3</sub><sup>-</sup> in sub 1- $\mu$ m particles, although the amount of contribution is 180 uncertain. Airborne size distributions of particles with diameters between 90 and 7500 nm were measured 181 with a Laser Aerosol Spectrometer (LAS, TSI Inc.) calibrated with polystyrene latex spheres. Airborne 182 measurements of aerosol composition by a particle-into-liquid sampler (PILS) and offline ion 183 chromatography (IC) analysis showed that nitrate constituted 53% of the water-soluble aerosol mass. 184 HCHO was measured with difference frequency generation absorption spectroscopy [Weibring et al., 2006]. 185 P-3B measurements were acquired from Revision 4 merged files available in the NASA online database 186 [NASA, 2017].

187 NH<sub>3</sub> was also measured from a mobile ground laboratory that sampled conditions across the Valley 188 during transects on 21-22 and 25-31 January and 1, 3-5, and 7 February 2013. Mobile measurements were 189 performed with an open-path, quantum-cascade laser-based sensor mounted on the roof rack of a sedan 190 passenger car [Miller et al., 2014; Sun et al., 2014] as described previously [Miller et al., 2015; Sun et al., 191 2015]. Mobile laboratory data were acquired from Revision 0 files available online [NASA, 2017]. At the CARB Fresno-Garland site, water soluble inorganic PM3 component ions including NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, SO<sub>4</sub><sup>2-</sup>, and K<sup>+</sup> 192 193 were measured with sub-hourly resolution during 19 January to 10 February using a PILS-IC instrument, and 194 water soluble gases including HNO<sub>3</sub> and NH<sub>3</sub> were collected with ~5-7 h resolution using annular denuders 195 and analyzed offline by IC [Parworth et al., 2017]. These data were acquired directly from the authors 196 Parworth et al. [2017], although the data are also available online [NASA, 2017]. Meteorology

- measurements collected by CARB were acquired from NASA [2017], radar profiler measurements at Visalia
   were obtained from NOAA [2017], and 24-h average PM<sub>2.5</sub> NO<sub>3</sub><sup>-</sup> concentrations at SJV monitoring sites were
   obtained directly from CARB, although routine monitoring data are also available online [USEPA, 2017a].
- 200 2.3 Model-measurement pairing

201 Model predictions were generally paired with measurements according to standard practice by 202 extracting predictions from the grid cell containing the measurement and then averaging the hourly model 203 output to the sampling period of the measurement. To match model predictions with P-3B and mobile 204 laboratory measurements, the grid cell containing the measurement at each second was identified, and 205 predictions from that cell were linearly interpolated to the time of the measurement. The paired 1-s data 206 were then averaged to 10-s resolution for the boxplot comparisons below. For spatial comparisons of 207 CMAQ predictions with mobile laboratory NH<sub>3</sub> measurements, medians of sub-cell median mixing ratios 208 were used to ensure adequate grid cell coverage of the measurements and reduce the influence of near-209 source sampling as follows. First, the 4-km CMAQ grid cells were decomposed into 1-km sub-cells, and grid cells with measurements in at least four sub-cells were selected. Second, median NH<sub>3</sub> mixing ratios in each 210 211 sub-cell were calculated from the 1-s paired model-measurement data. Finally, the median mixing ratio for 212 a 4-km grid cell was calculated as the median of the sub-cell median values. For spatial comparisons of 213 CMAQ predictions and P-3B measurements, mean or median mixing ratios were calculated from the 1-s 214 paired model-measurement data over samples within the modeled PBL during 11-15 PST for grid columns 215 with measurements on at least four days. Modeled PBL heights were well correlated with PBL heights 216 estimated from measurements during P-3B spirals, but predicted values were moderately biased low (12-217 34%; Figure S2).

218

#### 219 3. Results and Discussion

220  $NH_4NO_3$  in fine particles is generally considered to be in thermodynamic equilibrium with  $NH_3$  and 221 HNO<sub>3</sub> for time scales of relevance to regional air quality modeling [e.g., Fountoukis et al., 2009; Meng and 222 Seinfeld, 1996]. Evaluations of model predictions of NH<sub>3</sub>, HNO<sub>3</sub>, NO<sub>x</sub>, and NO<sub>y</sub> are therefore relevant for 223 understanding the model's ability to simulate NH<sub>4</sub>NO<sub>3</sub>. In section 3.1, NH<sub>3</sub> predictions are compared with 224 measurements from the mobile ground laboratory and the P-3B aircraft. In section 3.2, predictions of NOx, 225 NOx/NOy, and TNO3 are compared with measurements from the P-3B aircraft. Routine network 226 observations of  $NO_3^-$  are also used to understand the model's ability to simulate  $NO_3^-$  across the Valley. In 227 section 3.3, the NH<sub>4</sub>NO<sub>3</sub> system is considered at the Fresno site where a comprehensive dataset allows for

228 detailed investigation. Finally, in section 3.4, model predictions of HNO<sub>3</sub> production rates are presented to 229 contribute to understanding of the spatial and temporal patterns of nitrate production in the Valley. The 230 term  $NH_4NO_3$  is used here for convenience and is not meant to imply a solid phase state. For 231 supersaturated conditions and for stable equilibrium conditions at relative humidities (RHs) greater than 232 the mutual deliquescence RH (MDRH) of the inorganic system, NH<sub>4</sub>NO<sub>3</sub> would partially or completely 233 dissociate into NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> ions in aqueous solution [e.g., Kelly et al., 2008; Nenes et al., 1998; Wexler 234 and Seinfeld, 1991]. Since RH is often high in winter in SJV, CMAQ's assumption that NH<sub>4</sub>NO<sub>3</sub> completely 235 dissociates into aqueous solution at all RHs is generally a good one, except possibly during afternoon hours 236 as discussed below.

237 3.1 Examining NH<sub>3</sub> predictions

238 Average modeled NH<sub>3</sub> mixing ratios over SJV during 15 January to 5 February 2013 are shown in Figure 239 1a. Mixing ratios greater than about 7 ppb are predicted throughout SJV, and mixing ratios greater than 20 240 ppb occur in regions just west of Fresno, around Bakersfield, and a large portion of the eastern side of the 241 Valley between Bakersfield and Fresno. The spatial patterns of elevated NH<sub>3</sub> mixing ratios follow the spatial 242 patterns of NH<sub>3</sub> emissions (Figure 1b) closely. NH<sub>3</sub> emissions occur primarily during daytime (Figure S3a) 243 due to the combination of increased emission-related activity and conducive meteorology [e.g., Lonsdale et 244 al., 2017; Zhu et al., 2015]. On average, NH<sub>3</sub> deposition fluxes in the boxed region of Figure 1b were 43% of 245 the emission fluxes during 10-16 PST, and vertical transport of NH<sub>3</sub> from model layer 1 was 55% of the 246 emission fluxes (Figure S3b). This behavior is consistent with a previous study for the eastern U.S. [Dennis 247 et al., 2010] and explains the correspondence in spatial patterns of NH<sub>3</sub> emissions and model surface layer 248 concentrations in Figure 1.

249 Model predictions of NH<sub>3</sub> are compared with measurements from the mobile ground laboratory in 250 Figure 2. These comparisons were done by matching CMAQ predictions in space and time with the 251 measurements for all transects and then calculating the median modeled and measured mixing ratio by 252 CMAQ grid cell from median values in 1-km sub-cells as described above. Model predictions are scaled by 253 two in Figure 2a to better illustrate spatial patterns. The model underestimates mixing ratios considerably 254 in regions where elevated values (> ~20 ppb) were measured (Figure 2b). Yet the model correctly estimates 255 that NH<sub>3</sub> mixing ratios are elevated just southwest of Turlock, near Fresno, and in a region to the southeast 256 of Hanford and that NH<sub>3</sub> mixing ratios are relatively low on the western side of the Valley. The model-257 measurement comparison is complicated by the non-uniform sampling and wide range of scales 258 represented by the high resolution measurements compared with the relatively coarse regional air quality 259 model. The qualitative conclusion of underestimated NH<sub>3</sub> in high emission regions based on the aggregated

results in Figure 2 appears to be robust though based on additional NH<sub>3</sub> evaluation discussed below. Also, previous comparisons of CMAQ predictions with NH<sub>3</sub> measurements in SJV from the NOAA P-3B aircraft during May and June 2010 yielded similar conclusions as here [*Kelly et al.*, 2014]. Model predictions of NO<sub>3</sub><sup>-</sup> appear to be insensitive to the NH<sub>3</sub> underpredictions in the high emission regions though. For instance, in a simulation with NH<sub>3</sub> emissions doubled in the boxed region of Figure 1b, average NO<sub>3</sub><sup>-</sup> concentrations changed by <5% in 93% of SJV grid cells having NO<sub>3</sub><sup>-</sup> concentrations >5 µg m<sup>-3</sup> and the maximum change was 13%.



267

Figure 1. Average NH<sub>3</sub> (a) mixing ratios predicted by CMAQ and (b) gridded emissions during 15 January – 5

269 February 2013 with county lines, box defining region for discussion, and markers for P-3B spiral locations.



Figure 2. (a) Median observed and 2x median modeled NH<sub>3</sub> mixing ratio by CMAQ grid cell over all mobile ground laboratory sampling transects and (b) difference in median values. See text for description of grid cell median calculations.

271

276 Median modeled NH<sub>3</sub> mixing ratios are compared with CRDS and PTR-ToF-MS measurements from the P-3B aircraft in Figure 3. Model results are scaled by three in the figure to better illustrate spatial 277 patterns. Similar to results of the mobile ground laboratory comparison, the spatial patterns of model 278 279 predictions are in general agreement with P-3B NH<sub>3</sub> measurements, but model predictions are too low in 280 areas where elevated mixing ratios were observed. One location where relatively large underpredictions are evident is Hanford, which is located just outside of the high emission and concentration region in the 281 282 model (Figure 1). NH<sub>3</sub> measurements from the CRDS and PTR-ToF-MS were in good agreement near the 283 surface during morning and afternoon P-3B aircraft spirals over Hanford (Figure S4) and indicate that 284 median modeled mixing ratios were underpredicted by a factor of 7-9 in the 0-900 m altitude range. For 285 the mobile laboratory comparison in Figure 2, median NH<sub>3</sub> mixing ratios over Hanford were underpredicted by a factor of 5. Considering that Hanford is located just outside of a high emission region in the model, 286 287 further examination of the spatial allocation of NH<sub>3</sub> emissions in this area is warranted. Modeled PBL 288 heights were in reasonable agreement with empirical estimates at the Hanford site (normalized mean bias: 289 -12%; Figure S2), and so errors in mixing height predictions are unlikely to explain the model-measurement 290 differences.



Figure 3. Median modeled and measured NH₃ within the modeled PBL during 11-15 PST by model grid cell
 for P-3B flights in January and February 2013.

292

### 296 3.2 Examining NOx, NOx/NOy, and NO<sub>3</sub><sup>-</sup> predictions

297 Average model predictions of NOy during 15 January to 5 February are shown in Figure 4a. 298 Relatively high mixing ratios are predicted over Fresno, Bakersfield, and northern SJV cities (e.g., Modesto 299 and Stockton) and along Highway 99 between these cities. Average concentrations of fine particle  $NO_3^{-1}$ 300 (Figure 4b) are more uniformly distributed across the Valley than are mixing ratios of NOy, which are 301 elevated in areas with high NOx emissions. The formation of  $NO_3^-$  requires the oxidation of NOx to HNO<sub>3</sub> 302 and is promoted by the mixing of urban air masses with air rich in  $NH_3$  from surrounding areas. These 303 dependencies help explain the broader average spatial distributions of NO<sub>3</sub><sup>-</sup> than NOy and NH<sub>3</sub> (c.f., Figure 1a). Also, dry deposition velocities of  $HNO_3$  and  $NH_3$  are generally high compared with those of fine particle 304 305 NO<sub>3</sub><sup>-</sup> and contribute to greater spatial gradients in NOy and NH<sub>3</sub>.

NOx and NOy were measured during a series of P-3B spirals over sites including major cities in the 306 307 north (Fresno) and south (Bakersfield), rural locations in the west (Tranguility and Huron), and the Hanford site discussed above. In Figure 5a, model predictions of NOx are compared with measurements during the 308 309 aircraft spirals. The model correctly predicted that the highest mixing ratios occurred in Bakersfield and 310 Fresno, relatively low mixing ratios occurred in Tranquility and Huron, and mixing ratios generally 311 decreased with altitude. Yet NOx predictions were biased high in the 0-300 m bin at Bakersfield, Fresno, and Tranquility. The ratios of NOx-to-NOy are shown in Figure 5b, where modeled NOy was calculated by 312 313 summing gas-phase NOy components and 20% of fine particle NO<sub>3</sub>. Size distribution measurements during 314 the flights indicate that the majority of fine particle  $NO_3^-$  existed in particles with diameters less than 500

nm (Figure S5) and suggest that a significant, although unknown, fraction of fine particle  $NO_3^-$  was 315 measured by the NOy instrument. NOx-to-NOy ratio comparisons based on modeled NOy with 0% and 316 317 100% of modeled  $NO_3^-$  included in the NOv calculation are provided in Figure S6. The model captured the 318 general pattern of relatively high NOx-to-NOy ratios in urban areas (e.g., Bakersfield), where fresh NOx 319 emissions comprise a large fraction of NOy, and relatively low NOx-to-NOy ratios in remote areas (e.g., 320 Huron), where NOx oxidation products comprise a large fraction of NOy (Figure 5b). The model also 321 captured the generally decreasing trends of NOx-to-NOy ratios with altitude. The overestimates of the 322 NOx-to-NOy ratios in Fresno, Bakersfield, and Tranquility in Figure 5b suggest that the overpredictions of 323 NOx in Figure 5a could be due in part to too-low modeled oxidation rates. However, the NOx-to-NOy 324 evaluation is limited by uncertainty in the fraction of particle  $NO_3^-$  measured by the NOy instrument. 325 Underpredictions of HCHO and  $O_3$  during the aircraft spirals suggest that modeled oxidation rates may have 326 been too low over the sites (Figure S7).





- Figure 4. Average (a) NOy (including fine particle NO<sub>3</sub><sup>-</sup>) mixing ratios predicted in SJV with markers for P-3B
- spiral locations and (b) fine particle NO<sub>3</sub><sup>-</sup> concentrations with markers for PM<sub>2.5</sub> monitoring locations during
- 331 15 January 5 February 2013.





Figure 5. Comparison of modeled and measured (a) NOy and (b) NOx/NOy mixing ratio distributions for 335 300-m altitude ranges for P-3B aircraft spirals (see Figure 4a for site locations). Boxes bracket the 336 interquartile range (IQR), lines within the boxes represent the median, and whiskers represent 1.5 times 337 the IQR from either end of the box.

In Figure 6, average TNO3 mixing ratios are shown for P-3B measurements at altitudes within the modeled PBL during 11-15 PST by model grid cell for cells with measurements on at least four days. Modeled TNO3 mixing ratios were generally biased high compared with the measurements, especially along Highway 99 between Fresno and Bakersfield. The relatively large TNO3 overpredictions between Bakersfield and Fresno resulted in weaker daytime gradients between the cities and surrounding areas for the model than were identified by *Pusede et al.* [2016]. Modeled TNO3 was biased low relative to the

ground site measurements in Fresno (see section 3.3).





351 In Figure 7, model predictions are compared with routine observations of 24-h average fine particle 352 NO<sub>3</sub><sup>-</sup> at four sites spanning SJV from north (Modesto) to south (Bakersfield) (see Figure 4b for site 353 locations). A peak in the NO<sub>3</sub><sup>-</sup> time series was observed at all sites on January 22<sup>nd</sup>. The model performed 354 well on this day for all sites except Bakersfield for which observations were underpredicted. On February 3, 355 high NO<sub>3</sub><sup>-</sup> concentrations (>20  $\mu$ g m<sup>-3</sup>) were observed in the south (Bakersfield and Visalia) and lower 356 concentrations (<10 µg m<sup>-3</sup>) were observed in the north (Modesto and Fresno). The model underpredicted 357 the NO<sub>3</sub><sup>-</sup> peaks on February 3 at Bakersfield and Visalia. The root-mean-square error (RMSE) for predictions 358 increased in magnitude from north (1  $\mu$ g m<sup>-3</sup>) to south (7  $\mu$ g m<sup>-3</sup>), whereas correlation coefficients were 359 high ( $r \ge 0.78$ ) at all sites.

360 The mean modeled PBL height at 15 PST was 320 m during the 18-22 January period when elevated 361 NO<sub>3</sub><sup>-</sup> was simulated in Modesto and Fresno and was 490 m during the 1-5 February period when the model 362 predicted lower NO<sub>3</sub><sup>-</sup> concentrations (Figure S8). Wind speeds were also lower during the January period (mean: 1.4 m/s) than the February period (mean: 1.9 m/s). Compared with profiler measurements at 363 364 Visalia, wind speeds were biased low in mid-January and were biased high near the surface in early 365 February (Table S1). Considering that meteorological stagnation is central to the conceptual model of NO<sub>3</sub><sup>-</sup> 366 formation and build-up in SJV, the relatively low NO<sub>3</sub><sup>-</sup> concentrations simulated during the early February 367 period are probably related to the greater transport and mixing in the model. The relatively large  $NO_3^{-1}$ underpredictions at Bakersfield, where meteorology is more influenced by the convergence of mountain 368 369 ranges to the south, may be attributed to challenges in simulating meteorology in complex terrain.

- 370 However, a full evaluation of the three-dimensional meteorological fields across the Valley and their impact
- 371 on air quality during these periods is not straightforward and is beyond the scope of this study.



![](_page_14_Figure_4.jpeg)

Figure 7. Comparison of 24-h average PM<sub>2.5</sub> NO<sub>3</sub><sup>-</sup> predictions of CMAQ with routine monitoring
 measurements at sites shown in Figure 4b.

376

## 377 3.3 Examining the NH<sub>4</sub>NO<sub>3</sub> system in Fresno

378 A relatively complete set of measurements for evaluating the  $NH_4NO_3$  system were made during 19 379 January – 10 February 2013 at the CARB Fresno-Garland site. In Figure 8, predictions of fine particle NO<sub>3</sub><sup>-</sup> 380 are compared with PILS-IC measurements at this site. Two major NO<sub>3</sub><sup>-</sup> episodes were identified in Fresno 381 during the campaign from about 14-23 January and 29 January to 5 February [Young et al., 2016]. During 382 the first episode, the model overpredicted the peak  $NO_3^-$  concentration on 22 January (Figure 8). The 383 modeled peak is due to overnight transport of NO<sub>3</sub><sup>-</sup> from the south (Figure S10), where modeled production of HNO<sub>3</sub> was particularly high around Visalia during this period (see section 3.4). Modeled wind speeds 384 385 were low in Visalia in reasonable agreement with observations (Figure S11–S13). However, observed winds 386 at sites in SJV were relatively disorganized overnight compared with model predictions and suggest that the 387 model overestimated transport of  $NO_3^-$  to Fresno on January 22. In early February, the model 388 underpredicted the elevated NO<sub>3</sub><sup>-</sup> concentrations in Fresno. As discussed above, modeled wind speeds and 389 PBL heights were relatively high across SJV during the February period, and modeled NO<sub>3</sub><sup>-</sup> concentrations

- 390 were relatively low. Comparisons of predictions of SO<sub>4</sub><sup>2-</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, and Cl<sup>-</sup> with PILS-IC measurements are
- 391 provided in Figure S14. Underpredictions of the generally modest measured Cl<sup>-</sup> concentrations are
- 392 consistent with findings of studies of other parts of the U.S. [Kelly et al., 2016; Kelly et al., 2014; Kelly et al.,
- 393 2010; Simon et al., 2010].
- 394

![](_page_15_Figure_5.jpeg)

395
396 Figure 8. Comparison of model predictions of fine particle NO<sub>3</sub><sup>-</sup> with PILS-IC measurements at the Fresno397 Garland ground site.

399 Distributions of hourly average modeled and measured NO<sub>3</sub><sup>-</sup> concentrations in Fresno are shown in 400 Figure 9 for the January and February episodes. Measured concentrations increase in the morning during 401 both periods in a pattern consistent with mixing of NH<sub>4</sub>NO<sub>3</sub> from the nocturnal residual layer to the surface during development of the daytime boundary layer [Parworth et al., 2017; Prabhakar et al., 2017; Young et 402 al., 2016]. The 75<sup>th</sup> percentiles of modeled concentrations increase in the morning during the 19-25 403 404 January episode, but median concentrations are relatively constant compared with the measurements. The 405 morning increase in NO<sub>3</sub><sup>-</sup> is also underpredicted during 30 January - 5 February. In the afternoon, measured 406 NO<sub>3</sub><sup>-</sup> concentrations reach a relatively constant level during the first period and decrease during the second period, whereas modeled concentrations decrease in the afternoon during both episodes (Figure 9). 407

![](_page_16_Figure_0.jpeg)

Figure 9. Hourly average modeled and measured NO<sub>3</sub><sup>-</sup> distributions at Fresno ground site during two
periods of interest. Boxes bracket the IQR, lines within the boxes represent the median, whiskers represent
1.5 times the IQR from either end of the box, and circles represent individual values less than and greater
than the range of the whiskers.

408

In Figure 10, concentrations of TNO3 and NHx and the percentage of the total concentrations in the gas phase are shown during 19-31 January when model performance for NO<sub>3</sub><sup>-</sup> was relatively good. The model is biased 27% low for TNO3 and 36% low for NHx during this period at the Fresno site. However, the model correctly predicts that most of NHx is in the gas phase and most of TNO3 is in the particle phase. This gas-particle partitioning behavior suggests that HNO<sub>3</sub> is the limiting precursor for NH<sub>4</sub>NO<sub>3</sub> formation in SJV in both the model and ambient. Sensitivity simulations with reductions in NH<sub>3</sub> and NOx emissions were conducted and confirmed that HNO<sub>3</sub> is the limiting precursor in the model.

421 Although gas-particle partitioning is generally predicted well, the fraction of TNO3 in the gas phase 422 is sometimes overestimated in the model (Figure 10b). The overestimates of partitioning to the gas phase 423 appear to be driven primarily by meteorology (i.e., RH and temperature, T) rather than issues with particle 424 composition predictions. The modeled gas-phase fraction of TNO3 is relatively high when RH is less than 425 50% and T is greater than 285 K (Figure S15a). The overpredictions of the gas-phase fraction of TNO3 under 426 these conditions could be due in part to challenges in representing the particle phase state under low RH 427 conditions. Recall that the model assumes crystallization does not occur and inorganic components exist as 428 ions in supersaturated solutions for low RH (e.g., RH<MDRH). Previous studies have found that this 429 assumption yields lower predicted NO<sub>3</sub><sup>-</sup> concentrations compared with the stable equilibrium assumption for RH < ~50% [Ansari and Pandis, 2000; Fountoukis et al., 2009]. To investigate the issue here, offline 430 431 simulations with ISORROPIA II were performed for cases of stable (i.e., including crystallization) and 432 metastable (i.e., no crystallization) equilibrium using T, RH, and concentration inputs based on CMAQ

433 output for hours where the sampling period average RH was <50%. These simulations confirmed that the 434 phase state assumption influences partitioning predictions under the low-RH conditions in Fresno. For 435 hours with RH between 37% and 54%, the average percentage of TNO3 in the gas phase was 50% for 436 simulations based on the metastable assumption and 24% for the stable assumption. Segregation of results 437 by time of day (Figure S15b) reveals that the overpredictions of partitioning of TNO3 to the gas phase occur 438 in the afternoon. The overestimate of the decreasing trend in  $NO_3^-$  concentration in the afternoon in the 439 top panel of Figure 9 could therefore be due in part to gas-particle partitioning prediction issues, which are 440 sensitive to particle phase state assumptions under low RH conditions. Deposition rates of TNO3 are relatively large in the afternoon due to the relatively low atmospheric resistance of the convective 441 boundary layer (Figure S16). The average modeled deposition velocity was 2.83 cm s<sup>-1</sup> for HNO<sub>3</sub> and 0.07 442 443 cm s<sup>-1</sup> for accumulation mode particles during 12-17 PST, 19-31 January. Given the relatively high deposition velocity of HNO<sub>3</sub> compared with that of fine particle NO<sub>3</sub>, excessive partitioning of TNO3 to the 444 445 gas phase could lead to excessive removal of TNO3 through HNO<sub>3</sub> dry deposition in the afternoon.

![](_page_17_Figure_1.jpeg)

![](_page_17_Figure_2.jpeg)

![](_page_17_Figure_3.jpeg)

Figure 10. Modeled and measured concentrations of (a) TNO3 and NHx and (b) percentage of total in the
gas phase during 19 – 31 January at the Fresno ground site.

450

451 3.4 Examining HNO<sub>3</sub> production

- 452 Previous studies and the current modeling indicate that the limiting precursor for NH<sub>4</sub>NO<sub>3</sub> formation in
- 453 SJV is HNO<sub>3</sub>. Understanding chemical production of HNO<sub>3</sub> is therefore important for understanding NH<sub>4</sub>NO<sub>3</sub>
- 454 formation. HNO<sub>3</sub> production during daytime when OH levels are high is typically dominated by R1:

455 
$$NO_2 + OH \rightarrow HNO_3.$$

- 456 At night, when OH mixing ratios are low and photolysis of NO<sub>3</sub> radical is negligible, heterogeneous
- 457 hydrolysis of gas-phase  $N_2O_5$  with particle-phase  $H_2O$  (R2) is important

(R1)

458 
$$N_2O_5 + H_2O(p) + Y Cl^{-}(p) \rightarrow Y(HNO_3 + CINO_2) + 2 (1-Y) HNO_3$$
 (R2)

459 where Y is the yield of ClNO<sub>2</sub> [*Bertram and Thornton*, 2009; *Roberts et al.*, 2009]. O<sub>3</sub> is an important oxidant 460 in the production of N<sub>2</sub>O<sub>5</sub> at night:

$$161 \qquad NO + O_3 \rightarrow NO_2 + O_2 \tag{R3}$$

$$NO_2 + O_3 \rightarrow NO_3 \tag{R4}$$

463 
$$NO_2 + NO_3 \leftrightarrow N_2O_5.$$
 (R5)

464 Hourly 75<sup>th</sup> percentile HNO<sub>3</sub> production rates for R1, R2, homogeneous hydrolysis of  $N_2O_5$  with water 465 vapor, and heterogeneous hydrolysis of organic nitrates over Fresno are shown in Figure 11a for model 466 layer 1, 5, and 7 during 17-22 January. Reaction of NO₂ with OH (R1) dominates HNO₃ production in all 467 layers during daytime. Overnight, heterogeneous  $N_2O_5$  hydrolysis (R2) dominates production in layer 5 and 468 7. This HNO<sub>3</sub> can condense to form fine particle NO<sub>3</sub><sup>-</sup> and increase surface NO<sub>3</sub><sup>-</sup> concentrations in the 469 morning as the daytime boundary layer develops (e.g., Figure 9). In the surface layer overnight, R1 and R2 470 contribute similarly to HNO<sub>3</sub> production over Fresno in the model. OH mixing ratios that drive R1 are 471 typically low at night because photolysis reactions important for OH production are negligible. The primary 472 source of OH in the model at night is the reaction NO + HO<sub>2</sub>  $\rightarrow$  OH + NO<sub>2</sub>. This reaction is important in the model surface layer over Fresno because of the substantial NO emissions and the limited vertical mixing at 473 474 night. HO<sub>2</sub> sources in the model that do not directly depend on sunlight include reactions of organics with 475  $O_3$  and NO. Measured increases in surface  $NO_3^-$  concentrations in Fresno in the morning suggest that 476 production in the ambient surface layer over Fresno is relatively small compared with production aloft. 477 Therefore, there is evidence that HNO<sub>3</sub> production in the nighttime surface layer over Fresno is too high. 478 Also, O<sub>3</sub> mixing ratios in the surface layer are overestimated at the Fresno site overnight during this period 479 (Figure S17a). Observations indicate that O<sub>3</sub> is almost entirely depleted at the site on most nights due to 480 the high NOx levels and reactions such as R3 and R4. NOx mixing ratios are lower in the model than the 481 ambient overnight and enable partial recovery of  $O_3$  mixing ratios following decreases during the evening 482 rush hour when NOx emissions are high.

The apparently excessive production of HNO<sub>3</sub> in the model surface layer over Fresno at night appears to be due to overpredictions of O<sub>3</sub> mixing ratios. The cause of high O<sub>3</sub> mixing ratios in the surface layer in the model is vertical transport from higher layers. To test the impact of vertical mixing at night on the production of HNO<sub>3</sub> over Fresno, a sensitivity simulation was conducted where CMAQ's parameterization for the minimum eddy diffusivity (K<sub>z,min</sub>) was replaced by a fixed K<sub>z,min</sub> of 0.01 m<sup>2</sup> s<sup>-1</sup> in all grid cells as is done

488 in ACM2 in the WRF model. The K<sub>z.min</sub> change reduced vertical mixing of species overnight over Fresno, 489 because ACM2 in CMAQ uses higher K<sub>z,min</sub> values in urban areas [USEPA, 2012a]. O<sub>3</sub> depletion in the surface 490 layer was nearly complete overnight during 17-22 January in the simulation with reduced vertical mixing 491 (Figure S17b), and HNO<sub>3</sub> production in the surface layer was significantly reduced compared with the base 492 simulation (Figure 11b). However, the underestimate of the morning increase in  $NO_3^-$  was not resolved by 493 reducing K<sub>z.min</sub>. Advection of the nocturnal residual layer from Fresno to the south likely contributed to the 494 underestimate of the morning increase in NO<sub>3</sub><sup>-</sup> in the model. In a simulation with increased CO emissions in 495 Fresno grid cells, the largest impacts on CO mixing ratios aloft at night were to the south of Fresno during this period (Figure S18). 496

![](_page_19_Figure_1.jpeg)

![](_page_19_Figure_2.jpeg)

Figure 11. Hourly 75<sup>th</sup> percentile HNO<sub>3</sub> production rates by chemical pathway over Fresno for the (a) base simulation and (b) simulation with  $K_{z,min}$ =0.01 m<sup>2</sup> s<sup>-1</sup> during January 17 – 22. N2O5homog: homogeneous gas phase reaction of N<sub>2</sub>O<sub>5</sub> + H<sub>2</sub>O; NTRhyd: heterogeneous hydrolysis of organic nitrates; N2O5hyd: heterogeneous hydrolysis of N<sub>2</sub>O<sub>5</sub>; and OH\_NO2: reaction of OH+NO<sub>2</sub>.

503

504 HNO<sub>3</sub> production integrated over model layers 1-20 is shown in Figure 12a for SJV grid cells during 17-22 January. R1 is the dominant production pathway in urban areas with large NOx emissions such as 505 506 Fresno and Bakersfield in the model. The R2 pathway is dominant in semi-urban and rural areas along Highway 99, particularly around Visalia and in northern SJV. HNO<sub>3</sub> production in SJV peaks in model layer 6 507 508 (160 - 240 m; Figure 12c). R1 is productive in the middle of the daytime boundary layer due to the 509 combination of relatively high OH and NO<sub>2</sub>, and R2 tends to be most productive in the nocturnal residual 510 layer due to the combination of high  $N_2O_5$  and aerosol surface area [*Riemer et al.*, 2003]. Overall, the 511 model estimates that R1 contributes 46% and R2 contributes 54% to total HNO<sub>3</sub> production for the 17-22

January period when the model predicted elevated NO<sub>3</sub><sup>-</sup>. This apportionment is similar to model estimates from previous episodes [*Ying and Kleeman*, 2009]. In early February, when the model under-predicted NO<sub>3</sub><sup>-</sup> concentrations, the modeled boundary layer was deeper during the day and production occurred over a wider range of altitudes (Figure 12d). The R2 pathway was relatively weak in the model in the area between Fresno and Bakersfield in early February (Figure 12b) compared with 17-22 January (Figure 12a).

517 Pusede et al. [2016] predicted that HNO<sub>3</sub> production from R1 would increase relative to R2 with decreasing NOx emissions. To explore the sensitivity of HNO<sub>3</sub> production to NOx levels in the model, a 518 519 sensitivity simulation was conducted with NOx emissions reduced by 40%. In this simulation, R1 520 contributed 49% to integrated HNO<sub>3</sub> production during 17-22 January (i.e., production from R1 was 521 enhanced relative to R2 compared with the base simulation). Decreases in NOx emissions lead to increases 522 in OH mixing ratios in urban areas and along major highways in the model and thereby increase the percent 523 contribution of R1 to total HNO<sub>3</sub> production relative to that of the base simulation. This behavior is 524 qualitatively consistent with predictions of *Pusede et al.* [2016], although that study focused on the entire 525 winter period rather than the multiday episode considered here. A wide range of  $N_2O_5$  heterogeneous 526 reaction probabilities (i.e., the fraction of gas-particle collisions that result in net removal of N<sub>2</sub>O<sub>5</sub> from the gas phase,  $\gamma$ ) have been used in previous studies of NO<sub>3</sub><sup>-</sup> formation in SJV [e.g., *Prabhakar et al.*, 2017; Ying 527 528 and Kleeman, 2009]. To explore the sensitivity of HNO<sub>3</sub> production to  $\gamma$  and the ClNO<sub>2</sub> yield (Y in R2), three 529 additional simulations were conducted with  $\gamma$  scaled by 0.5 and 1.5 and with Y=0. Total HNO<sub>3</sub> production 530 decreased by 11% relative to the base case when  $\gamma$  was reduced by 50% for the scenario in Figure 12a. A 24% reduction in HNO<sub>3</sub> production from R2 was partially compensated for by a 5% increase in production 531 532 from R1. Total HNO<sub>3</sub> production increased by 6% relative to the base case in the simulation with a 50% 533 increase in  $\gamma$ . Setting the yield of CINO<sub>2</sub> to zero had negligible impact on HNO<sub>3</sub> production consistent with the generally low concentrations of Cl<sup>-</sup> in SJV, although Cl<sup>-</sup> was underpredicted (Figure S14).  $\gamma$  values 534 535 predicted over the P-3B spiral sites during 17-22 January are shown in Figure S19.

![](_page_21_Figure_0.jpeg)

536

Figure 12. HNO<sub>3</sub> production integrated over layers 1-20 for SJV model grid cells during (a) 17-22 January and
(b) 29 January - 4 February and integrated over SJV grid cells by model layer during (c) 17-22 January and
(d) 29 January - 4 February.

# 541 4. Conclusions

This study demonstrates that regional photochemical grid models are capable of simulating NH<sub>4</sub>NO<sub>3</sub> formation and build-up during major recent PM<sub>2.5</sub> episodes in SJV. For example, routine measurements of NO<sub>3</sub><sup>-</sup> were generally predicted well at sites in SJV, including days where 24-h average NO<sub>3</sub><sup>-</sup> reached 20 µg m<sup>-</sup> <sup>3</sup>. Gas-particle partitioning predictions were in good agreement with measurements in Fresno and indicate that the model correctly predicts that NH<sub>4</sub>NO<sub>3</sub> formation is limited by HNO<sub>3</sub> availability. Modeled chemical production of HNO<sub>3</sub> via daytime and nighttime pathways was generally consistent with reports from previous studies and conceptual models of NO<sub>3</sub><sup>-</sup> formation in SJV. During a period of elevated NH<sub>4</sub>NO<sub>3</sub>, the 549 model predicted that the OH +  $NO_2$  pathway contributed 46% to total HNO<sub>3</sub> production in SJV and the  $N_2O_5$ 550 heterogeneous hydrolysis pathway contributed 54%.

551 Despite generally favorable model performance, the 2013 SJV DISCOVER-AQ dataset provided insights 552 on areas where additional work could improve NH<sub>4</sub>NO<sub>3</sub> modeling for SJV. First, additional study on 553 meteorological modeling of the major stagnation events that drive PM<sub>2.5</sub> episodes in the Valley would be 554 valuable, particularly for southern SJV where the terrain is more complex than in central and northern SJV. 555 Challenges in simulating meteorology in southern SJV could help explain the better NH<sub>4</sub>NO<sub>3</sub> model 556 performance for Fresno and Modesto than Bakersfield. Also, work toward improving the simulation of 557 diurnal patterns of vertical mixing would be valuable, because the coupling and decoupling of processes in 558 the surface layer from layers aloft influences HNO<sub>3</sub> production and the diurnal profiles of NH<sub>4</sub>NO<sub>3</sub> at the 559 surface. Additional evaluation of the degree to which urban-nonurban transport of NO<sub>3</sub><sup>-</sup> occurs in the 560 ambient would also be helpful because predictions suggest that this transport can be important. 561 Improvements in meteorological modeling are likely necessary to improve performance against the hourly 562 average NO<sub>3</sub><sup>-</sup> measurements in Fresno. Second, additional work on NH<sub>3</sub> modeling is warranted based on 563 underpredictions of NH<sub>3</sub> in emission source regions where very high mixing ratios were measured. 564 Although the NH<sub>3</sub> underpredictions do not appear to have a large impact on NO<sub>3</sub><sup>-</sup> predications (because 565  $NO_3^{-}$  is HNO<sub>3</sub>-limited), NH<sub>3</sub> levels are too low in the model in source regions and warrant further study. 566 Improvements in the spatial allocation of NH<sub>3</sub> emissions near Hanford and elsewhere are also warranted. 567 Third, there is evidence that gas-particle partitioning predictions under low-RH conditions could benefit 568 from additional study. Although the overall impact of gas-particle partitioning issues may be minor due to 569 the generally high RH during SJV PM<sub>2.5</sub> episodes, the potential for premature removal of TNO3 via rapid 570 deposition of HNO<sub>3</sub> when the gas-phase fraction is overestimated in afternoon makes this an area of 571 interest.

Another topic for future investigation is on HNO<sub>3</sub> production in the nocturnal residual layer over urban and surrounding areas. Although this pathway is central to the conceptual model of NO<sub>3</sub><sup>-</sup> formation in SJV, measurements that can directly constrain nighttime HNO<sub>3</sub> production aloft over SJV are extremely limited. Researchers have made progress by using indirect methods to infer characteristics of the nocturnal residual layer based on measurements over urban areas on the previous day and following morning, but direct measurements of the key species at night over urban and surrounding areas would be valuable.

578

- 580 The authors recognize contributions from Ellen Cooter, Rob Gilliam, Deborah Luecken, Limei Ran, Golam
- 581 Sarwar, Chris Allen, Allan Beidler, James Beidler, and Lara Reynolds.
- 582 Disclaimer
- 583 Although this work was reviewed by EPA and approved for publication, it may not necessarily reflect official
- 584 Agency policy.
- 585
- 586 References

Ansari, A. S., and S. N. Pandis (2000), The effect of metastable equilibrium states on the partitioning of nitrate between the gas and aerosol phases, *Atmospheric Environment*, *34*(1), 157-168,

589 doi:<u>http://dx.doi.org/10.1016/S1352-2310(99)00242-3</u>.

Appel, K. W., et al. (2017), Description and evaluation of the Community Multiscale Air Quality (CMAQ)
modeling system version 5.1, *Geoscientific Model Development*, *10*(4), 1703-1732, doi:10.5194/gmd-101703-2017.

Bash, J. O., K. R. Baker, and M. R. Beaver (2016), Evaluation of improved land use and canopy
representation in BEIS v3.61 with biogenic VOC measurements in California, *Geosci. Model Dev.*, 9(6), 21912207, doi:10.5194/gmd-9-2191-2016.

Bash, J. O., E. J. Cooter, R. L. Dennis, J. T. Walker, and J. E. Pleim (2013), Evaluation of a regional air-quality
model with bidirectional NH3 exchange coupled to an agroecosystem model, *Biogeosciences*, *10*(3), 16351645, doi:10.5194/bg-10-1635-2013.

Bertram, T. H., and J. A. Thornton (2009), Toward a general parameterization of N2O5 reactivity on aqueous
particles: the competing effects of particle liquid water, nitrate and chloride, *Atmospheric Chemistry and Physics*, 9(21), 8351-8363.

- Blanchard, C. L., P. M. Roth, S. J. Tanenbaum, S. D. Ziman, and J. H. Seinfeld (2000), The Use of Ambient
  Measurements To Identify which Precursor Species Limit Aerosol Nitrate Formation, *Journal of the Air & Waste Management Association*, *50*(12), 2073-2084, doi:10.1080/10473289.2000.10464239.
- 605 CDFA (2016a), California Department of Food and Agriculture, California Agriculural Statistics Review 2015-606 2016, *Available: <u>https://www.cdfa.ca.gov/statistics/PDFs/2016Report.pdf</u>.*
- 607 CDFA (2016b), California Department of Food and Agriculture, Dairy Statistics: 2016 Annual Data, Available:
   608 <u>https://www.cdfa.ca.gov/dairy/pdf/Annual/2016/2016\_Statistics\_Annual.pdf</u>.
- 609 CDOC (2015), California Department of Conservation, Division of Oil, Gas, & Geothermal Resources, 2015:
   610 Well Count and Oil and Gas Production by County, *Retrieved from <u>http://www.conservation.ca.gov/dog</u>.*
- 611 CDOF (2017), Calfornia Department of Finance, Press Release: Department of Finance Releases New State 612 Population Projections, *Available:*

613 http://www.dof.ca.gov/Forecasting/Demographics/projections/documents/P\_PressRelease.pdf

- 614 Chen, J., J. Lu, J. C. Avise, J. A. DaMassa, M. J. Kleeman, and A. P. Kaduwela (2014), Seasonal modeling of
- 615 PM2.5 in California's San Joaquin Valley, *Atmospheric Environment*, *92*, 182-190,
- 616 doi:<u>http://dx.doi.org/10.1016/j.atmosenv.2014.04.030</u>.
- 617 Chen, J., Q. Ying, and M. J. Kleeman (2009), Source apportionment of visual impairment during the
- 618 California regional PM10/PM2.5 air quality study, *Atmospheric Environment*, *43*(39), 6136-6144,
- 619 doi:<u>http://dx.doi.org/10.1016/j.atmosenv.2009.09.010</u>.
- 620 Chen, J., Q. Ying, and M. J. Kleeman (2010), Source apportionment of wintertime secondary organic aerosol
  621 during the California regional PM10/PM2.5 air quality study, *Atmospheric Environment*, 44(10), 1331-1340,
  622 doi:http://dx.doi.org/10.1016/j.atmosenv.2009.07.010.
- 623 Chen, L. W. A., J. G. Watson, J. C. Chow, and K. L. Magliano (2007), Quantifying PM2.5 source contributions
  624 for the San Joaquin Valley with multivariate receptor models, *Environmental Science & Technology*, *41*(8),
  625 2818-2826, doi:10.1021/es0225105.
- 626 Chow, J. C., L. W. A. Chen, J. G. Watson, D. H. Lowenthal, K. A. Magliano, K. Turkiewicz, and D. E. Lehrman
- 627 (2006), PM2.5 chemical composition and spatiotemporal variability during the California Regional
- PM10/PM2.5 Air Quality Study (CRPAQS), *Journal of Geophysical Research-Atmospheres*, *111*(D10),
  doi:10.1029/2005jd006457.
- Collett, J. L., K. J. Hoag, and X. Rao (1999a), Internal acid buffering in San Joaquin Valley fog drops and its
  influence on aerosol processing, *Atmospheric Environment*, *33*(29), 4833-4847, doi:10.1016/s13522310(99)00221-6.
- Collett, J. L., K. J. Hoag, D. E. Sherman, A. Bator, and L. W. Richards (1999b), Spatial and temporal variations
  in San Joaquin Valley fog chemistry, *Atmospheric Environment*, *33*(1), 129-140.
- Davis, J. M., P. V. Bhave, and K. M. Foley (2008), Parameterization of N2O5 reaction probabilities on the
  surface of particles containing ammonium, sulfate, and nitrate, *Atmos. Chem. Phys.*, *8*(17), 5295-5311,
  doi:10.5194/acp-8-5295-2008.
- Dennis, R. L., R. Mathur, J. E. Pleim, and J. T. Walker (2010), Fate of ammonia emissions at the local to
  regional scale as simulated by the Community Multiscale Air Quality model, *Atmospheric Pollution Research*, 1(4), 207-214, doi:<a href="http://dx.doi.org/10.5094/APR.2010.027">http://dx.doi.org/10.5094/APR.2010.027</a>.
- Fountoukis, C., and A. Nenes (2007), ISORROPIA II: a computationally efficient thermodynamic equilibrium
  model for K+-Ca2+-Mg2+-NH4+-Na+-SO42--NO3-Cl--H2O aerosols, *Atmos. Chem. Phys.*, 7(17), 4639-4659,
  doi:10.5194/acp-7-4639-2007.
- Fountoukis, C., A. Nenes, A. Sullivan, R. Weber, T. Van Reken, M. Fischer, E. Matías, M. Moya, D. Farmer,
  and R. C. Cohen (2009), Thermodynamic characterization of Mexico City aerosol during MILAGRO 2006, *Atmos. Chem. Phys.*, *9*(6), 2141-2156, doi:10.5194/acp-9-2141-2009.
- Ge, X. L., A. Setyan, Y. L. Sun, and Q. Zhang (2012a), Primary and secondary organic aerosols in Fresno,
  California during wintertime: Results from high resolution aerosol mass spectrometry, *Journal of Geophysical Research-Atmospheres*, *117*, 15, doi:10.1029/2012jd018026.

- 650 Ge, X. L., Q. Zhang, Y. Sun, C. R. Ruehl, and A. Setyan (2012b), Effect of aqueous-phase processing on
- aerosol chemistry and size distributions in Fresno, California, during wintertime, *Environmental Chemistry*,
   9(3), 221-235, doi:<u>https://doi.org/10.1071/EN11168</u>.
- 653 Gentner, D. R., et al. (2014), Emissions of organic carbon and methane from petroleum and dairy
- 654 operations in California's San Joaquin Valley, *Atmos. Chem. Phys.*, *14*(10), 4955-4978, doi:10.5194/acp-14-655 4955-2014.
- Herckes, P., A. R. Marcotte, Y. Wang, and J. L. Collett (2015), Fog composition in the Central Valley of
- 657 California over three decades, *Atmospheric Research*, *151*, 20-30,
  658 doi:http://dx.doi.org/10.1016/j.atmosres.2014.01.025.
- Herner, J. D., J. Aw, O. Gao, D. P. Chang, and M. J. Kleeman (2005), Size and Composition Distribution of
  Airborne Particulate Matter in Northern California: I—Particulate Mass, Carbon, and Water-Soluble Ions, *Journal of the Air & Waste Management Association*, *55*(1), 30-51, doi:10.1080/10473289.2005.10464600.
- 662 Herner, J. D., Q. Ying, J. Aw, O. Gao, D. P. Y. Chang, and M. J. Kleeman (2006), Dominant Mechanisms that
- 663 Shape the Airborne Particle Size and Composition Distribution in Central California, *Aerosol Science and*
- 664 *Technology*, *40*(10), 827-844, doi:10.1080/02786820600728668.
- Hixson, M., A. Mahmud, J. Hu, and M. J. Kleeman (2012), Resolving the interactions between population
  density and air pollution emissions controls in the San Joaquin Valley, USA, *Journal of the Air & Waste Management Association*, *62*(5), 566-575, doi:10.1080/10962247.2012.663325.
- Houyoux, M. R., J. M. Vukovich, C. J. Coats, N. J. M. Wheeler, and P. S. Kasibhatla (2000), Emission inventory
  development and processing for the Seasonal Model for Regional Air Quality (SMRAQ) project, *Journal of Geophysical Research: Atmospheres*, *105*(D7), 9079-9090, doi:10.1029/1999JD900975.
- Jacob, D. J., J. W. Munger, J. M. Waldman, and M. R. Hoffmann (1986), The H2SO4-HNO3-NH3 system at
- high humidities and in fogs: 1. Spatial and temporal patterns in the San Joaquin Valley of California, *Journal* of Geophysical Research: Atmospheres, 91(D1), 1073-1088, doi:10.1029/JD091iD01p01073.
- Jang, J. C. C., H. E. Jeffries, and S. Tonnesen (1995), Sensitivity of ozone to model grid resolution. 2. Detailed
  process analysis for ozone chemistry, *Atmospheric Environment*, *29*(21), 3101-3114, doi:10.1016/13522310(95)00119-j.
- Jeffries, H. E., and S. Tonnesen (1994), A comparison of 2 photochemical-reaction mechanisms using massbalance and process analysis, *Atmospheric Environment*, *28*(18), 2991-3003, doi:10.1016/13522310(94)90345-x.
- Kelly, J. T., J. Avise, C. Cai, and A. P. Kaduwela (2011), Simulating Particle Size Distributions over California
  and Impact on Lung Deposition Fraction, *Aerosol Science and Technology*, 45(2), 148-162,
- 682 doi:10.1080/02786826.2010.528078.
- Kelly, J. T., K. R. Baker, C. G. Nolte, S. L. Napelenok, W. C. Keene, and A. A. P. Pszenny (2016), Simulating the
  phase partitioning of NH3, HNO3, and HCl with size-resolved particles over northern Colorado in winter, *Atmospheric Environment*, 131, 67-77, doi:10.1016/j.atmosenv.2016.01.049.

- Kelly, J. T., et al. (2014), Fine-scale simulation of ammonium and nitrate over the South Coast Air Basin and
  San Joaquin Valley of California during CalNex-2010, *Journal of Geophysical Research: Atmospheres, 119*(6),
  3600-3614, doi:10.1002/2013JD021290.
- Kelly, J. T., P. V. Bhave, C. G. Nolte, U. Shankar, and K. M. Foley (2010), Simulating emission and chemical
  evolution of coarse sea-salt particles in the Community Multiscale Air Quality (CMAQ) model, *Geoscientific Model Development*, 3(1), 257-273.
- Kelly, J. T., A. S. Wexler, C. K. Chan, and M. N. Chan (2008), Aerosol thermodynamics of potassium salts,
  double salts, and water content near the eutectic, *Atmospheric Environment*, *42*(16), 3717-3728,
  dai:10.1016/j.stmassary.2009.01.001
- 694 doi:10.1016/j.atmosenv.2008.01.001.
- Kim, Y. J., S. N. Spak, G. R. Carmichael, N. Riemer, and C. O. Stanier (2014), Modeled aerosol nitrate
  formation pathways during wintertime in the Great Lakes region of North America, *Journal of Geophysical Research-Atmospheres*, *119*(21), 12420-12445, doi:10.1002/2014jd022320.
- Kleeman, M. J., Q. Ying, and A. Kaduwela (2005), Control strategies for the reduction of airborne particulate
  nitrate in California's San Joaquin Valley, *Atmospheric Environment*, *39*(29), 5325-5341,
  doi:http://dx.doi.org/10.1016/j.atmosenv.2005.05.044.
- Livingstone, P. L., et al. (2009), Simulating PM concentration during a winter episode in a subtropical valley:
  Sensitivity simulations and evaluation methods, *Atmospheric Environment*, *43*(37), 5971-5977,
  doi:10.1016/j.atmosenv.2009.07.033.
- Lonsdale, C. R., et al. (2017), Modeling the diurnal variability of agricultural ammonia in Bakersfield,
  California, during the CalNex campaign, *Atmos. Chem. Phys.*, *17*(4), 2721-2739, doi:10.5194/acp-17-27212017.
- McDonald, B. C., T. R. Dallmann, E. W. Martin, and R. A. Harley (2012), Long-term trends in nitrogen oxide
   emissions from motor vehicles at national, state, and air basin scales, *Journal of Geophysical Research- Atmospheres*, *117*, doi:10.1029/2012jd018304.
- Meng, Z., and J. H. Seinfeld (1996), Time scales to achieve atmospheric gas-aerosol equilibrium for volatile
   species, *Atmospheric Environment*, *30*(16), 2889-2900, doi:<a href="http://dx.doi.org/10.1016/1352-2310(95)00493-">http://dx.doi.org/10.1016/1352-2310(95)00493-</a>
   <u>9</u>.
- Miller, D. J., K. Sun, L. Tao, M. A. Khan, and M. A. Zondlo (2014), Open-path, quantum cascade-laser-based
  sensor for high-resolution atmospheric ammonia measurements, *Atmospheric Measurement Techniques*,
  7(1), 81-93, doi:10.5194/amt-7-81-2014.
- Miller, D. J., et al. (2015), Ammonia and methane dairy emission plumes in the San Joaquin Valley of
  California from individual feedlot to regional scales, *Journal of Geophysical Research: Atmospheres*, *120*(18),
  9718-9738, doi:10.1002/2015JD023241.
- Mlawer, E. J., S. J. Taubman, P. D. Brown, M. J. Iacono, and S. A. Clough (1997), Radiative transfer for
  inhomogeneous atmospheres: RRTM, a validated correlated-k model for the longwave, *Journal of Geophysical Research: Atmospheres, 102*(D14), 16663-16682, doi:10.1029/97JD00237.

- 722 Morrison, H., G. Thompson, and V. Tatarskii (2009), Impact of Cloud Microphysics on the Development of
- 723 Trailing Stratiform Precipitation in a Simulated Squall Line: Comparison of One- and Two-Moment Schemes,
- 724 *Monthly Weather Review*, *137*(3), 991-1007, doi:10.1175/2008mwr2556.1.
- NASA (2017), Airborne Science Data for Atmospheric Composition, <u>https://www-air.larc.nasa.gov/cgi-</u>
   <u>bin/ArcView/discover-ag.ca-2013</u>
- 727 Nenes, A., S. N. Pandis, and C. Pilinis (1998), ISORROPIA: A New Thermodynamic Equilibrium Model for
- 728 Multiphase Multicomponent Inorganic Aerosols, *Aquatic Geochemistry*, 4(1), 123-152,
- 729 doi:10.1023/a:1009604003981.
- 730 NOAA (2017), Earth System Research Laboratory Physical Sciences Division.
- 731 <u>ftp://ftp1.esrl.noaa.gov/psd2/data/realtime/Radar915/CnsWind/vis/2013/</u> Accessed June 2017.
- 732 Parworth, C. L., D. E. Young, H. Kim, X. Zhang, C. D. Cappa, S. Collier, and Q. Zhang (2017), Wintertime
- water-soluble aerosol composition and particle water content in Fresno, California, Journal of Geophysical
   *Research: Atmospheres*, 122(5), 3155-3170, doi:10.1002/2016JD026173.
- Pleim, J. E. (2007), A Combined Local and Nonlocal Closure Model for the Atmospheric Boundary Layer. Part
  I: Model Description and Testing, *Journal of Applied Meteorology and Climatology*, *46*(9), 1383-1395,
  doi:10.1175/jam2539.1.
- Pleim, J. E., J. O. Bash, J. T. Walker, and E. J. Cooter (2013), Development and evaluation of an ammonia
  bidirectional flux parameterization for air quality models, *Journal of Geophysical Research-Atmospheres*, *118*(9), 3794-3806, doi:10.1002/jgrd.50262.
- Pleim, J. E., and A. Xiu (2003), Development of a Land Surface Model. Part II: Data Assimilation, *Journal of Applied Meteorology*, *42*(12), 1811-1822, doi:10.1175/1520-0450(2003)042<1811:doalsm>2.0.co;2.
- Prabhakar, G., et al. (2017), Observational assessment of the role of nocturnal residual-layer chemistry in
  determining daytime surface particulate nitrate concentrations, *Atmos. Chem. Phys.*, *17*(23), 14747-14770,
  doi:10.5194/acp-17-14747-2017.
- Pun, B. K., R. T. F. Balmori, and C. Seigneur (2009), Modeling wintertime particulate matter formation in
  central California, *Atmospheric Environment*, *43*(2), 402-409,
  doi:http://dx.doi.org/10.1016/j.atmosenv.2008.08.040.
- Pun, B. K., and C. Seigneur (1999), Understanding particulate matter formation in the California San Joaquin
  Valley: conceptual model and data needs, *Atmospheric Environment*, *33*(29), 4865-4875,
  doi:http://dx.doi.org/10.1016/S1352-2310(99)00266-6.
- Pun, B. K., and C. Seigneur (2001), Sensitivity of Particulate Matter Nitrate Formation To Precursor
- Emissions in the California San Joaquin Valley, *Environmental Science & Technology*, *35*(14), 2979-2987,
   doi:10.1021/es0018973.
- Pusede, S. E., and R. C. Cohen (2012), On the observed response of ozone to NOx and VOC reactivity
  reductions in San Joaquin Valley California 1995-present, *Atmospheric Chemistry and Physics*, *12*(18), 83238339, doi:10.5194/acp-12-8323-2012.

- Pusede, S. E., et al. (2016), On the effectiveness of nitrogen oxide reductions as a control over ammonium
  nitrate aerosol, *Atmospheric Chemistry and Physics*, *16*(4), 2575-2596, doi:10.5194/acp-16-2575-2016.
- Pusede, S. E., et al. (2014), On the temperature dependence of organic reactivity, nitrogen oxides, ozone
   production, and the impact of emission controls in San Joaquin Valley, California, *Atmospheric Chemistry and Physics*, *14*(7), 3373-3395, doi:10.5194/acp-14-3373-2014.
- Riemer, N., H. Vogel, B. Vogel, B. Schell, I. Ackermann, C. Kessler, and H. Hass (2003), Impact of the
  heterogeneous hydrolysis of N2O5 on chemistry and nitrate aerosol formation in the lower troposphere
  under photosmog conditions, *Journal of Geophysical Research: Atmospheres*, *108*(D4), n/a-n/a,
  doi:10.1029/2002JD002436.
- Roberts, J. M., H. D. Osthoff, S. S. Brown, A. R. Ravishankara, D. Coffman, P. Quinn, and T. Bates (2009),
  Laboratory studies of products of N2O5 uptake on Cl- containing substrates, *Geophysical Research Letters*,
  36, doi:10.1029/2009gl040448.
- Russell, A. R., L. C. Valin, and R. C. Cohen (2012), Trends in OMI NO2 observations over the United States:
  effects of emission control technology and the economic recession, *Atmospheric Chemistry and Physics*,
  12(24), 12197-12209, doi:10.5194/acp-12-12197-2012.
- Sarwar, G., H. Simon, P. Bhave, and G. Yarwood (2012), Examining the impact of heterogeneous nitryl
  chloride production on air quality across the United States, *Atmospheric Chemistry and Physics*, *12*(14),
  6455-6473, doi:10.5194/acp-12-6455-2012.
- Shephard, M. W., and K. E. Cady-Pereira (2015), Cross-track Infrared Sounder (CrIS) satellite observations of
  tropospheric ammonia, *Atmospheric Measurement Techniques*, 8(3), 1323-1336, doi:10.5194/amt-8-13232015.
- Simon, H., et al. (2010), Modeling heterogeneous CINO2 formation, chloride availability, and chlorine
- 780 cycling in Southeast Texas, *Atmospheric Environment*, *44*(40), 5476-5488,
- 781 doi:10.1016/j.atmosenv.2009.09.006.
- SJVAPCD (2012), San Joaquin Valley Air Pollution Control District, 2012 PM2.5 Plan, Available:
   <u>http://www.valleyair.org/Air\_Quality\_Plans/PM25Plans2012.htm</u>.
- SJVAPCD (2016), San Joaquin Valley Air Pollution Control District, 2016 Moderate Area Plan for the 2012
   PM2.5 Standard, *Available: <u>http://www.valleyair.org/Air\_Quality\_Plans/PM25Plans2016.htm</u>.*
- Skamarock, W. C., J. B. Klemp, J. Dudhia, D. O. Gill, D. M. Barker, M. G. Duda, X. Huang, W. Wang, and J. G.
  Powers (2008), A description of the Advanced Reserch WRF version 3. NCAR Technical Note NCAR/TN475+STR.
- Stockwell, W. R., J. G. Watson, N. F. Robinson, W. Steiner, and W. W. Sylte (2000), The ammonium nitrate
   particle equivalent of NOx emissions for wintertime conditions in Central California's San Joaquin Valley,
   *Atmospheric Environment*, *34*(27), 4711-4717, doi:https://doi.org/10.1016/S1352-2310(00)00148-5.
- Sun, K., et al. (2015), Validation of TES ammonia observations at the single pixel scale in the San Joaquin
  Valley during DISCOVER-AQ, *Journal of Geophysical Research: Atmospheres*, *120*(10), 5140-5154,
  doi:10.1002/2014JD022846.

- Sun, K., L. Tao, D. J. Miller, M. A. Khan, and M. A. Zondlo (2014), On-Road Ammonia Emissions Characterized
- by Mobile, Open-Path Measurements, *Environmental Science & Technology*, 48(7), 3943-3950,
- 797 doi:10.1021/es4047704.
- 798 USEPA (2012a), CMAQ version 5.0 (February 2012 release) Technical Documentation. Available:
- https://www.airqualitymodeling.org/index.php/CMAQ\_version\_5.0\_(February\_2012\_release)\_Technical\_D
   ocumentation.
- 801 USEPA (2012b), Technical Support Document (TSD) Preparation of Emissions Inventories for the Version 5.0,
- 802 2007 Emissions Modeling Platform. Available: <u>https://www.epa.gov/sites/production/files/2015-</u>
- 803 <u>10/documents/2007v5\_2020base\_emismod\_tsd\_13dec2012.pdf</u>.
- USEPA (2016), 2011 National Emissions Inventory, version 2. Technical Support Document. Available:
   <u>https://www.epa.gov/sites/production/files/2015-10/documents/nei2011v2\_tsd\_14aug2015.pdf</u>.
- 806 USEPA (2017a), Air Quality System <u>https://www.epa.gov/aqs</u>.
- USEPA (2017b), Bayesian space-time downscaling fusion model (downscaler)-derived estimates of air
   quality for 2013.
- Watson, J. G., and J. C. Chow (2002), A wintertime PM2.5 episode at the Fresno, CA, supersite, *Atmospheric Environment*, *36*(3), 465-475, doi:<u>http://dx.doi.org/10.1016/S1352-2310(01)00309-0</u>.
- Watson, J. G., D. W. DuBois, R. DeMandel, A. Kaduwela, K. L. Magliano, C. McDade, P. K. Mueller, A. J.
- 812 Ranzieri, P. M. Roth, and S. Tanrikulu (1998), Field program plan for the California Regional PM2.5/PM10Air
- 813 Quality Study (CRPAQS), Desert Research Institute, Reno, NV, Available:
- 814 <u>http://www.arb.ca.gov/airways/crpaqs/publications.htm</u>.
- 815 Weibring, P., D. Richter, A. Fried, J. G. Walega, and C. Dyroff (2006), Ultra-high-precision mid-IR
- 816 spectrometer II: system description and spectroscopic performance, *Applied Physics B*, *85*(2), 207-218,
- 817 doi:10.1007/s00340-006-2300-4.
- 818 Wexler, A. S., and J. H. Seinfeld (1991), Second-generation inorganic aerosol model, Atmospheric
- Environment. Part A. General Topics, 25(12), 2731-2748, doi:<u>http://dx.doi.org/10.1016/0960-</u>
  1686(91)90203-J.
- Womack, C. C., et al. (2017), Evaluation of the accuracy of thermal dissociation CRDS and LIF techniques for
  atmospheric measurement of reactive nitrogen species, *Atmospheric Measurement Techniques*, *10*(5),
  1911-1926, doi:10.5194/amt-10-1911-2017.
- Ying, Q. (2011), Physical and chemical processes of wintertime secondary nitrate aerosol formation,
   *Frontiers of Environmental Science & Engineering in China*, 5(3), 348, doi:10.1007/s11783-011-0343-1.
- 826 Ying, Q., and M. Kleeman (2009), Regional contributions to airborne particulate matter in central California
- during a severe pollution episode, *Atmospheric Environment*, *43*(6), 1218-1228,
- 828 doi:<u>http://dx.doi.org/10.1016/j.atmosenv.2008.11.019</u>.
- Ying, Q., J. Lu, P. Allen, P. Livingstone, A. Kaduwela, and M. Kleeman (2008a), Modeling air quality during
- the California Regional PM10/PM2.5 Air Quality Study (CRPAQS) using the UCD/CIT source-oriented air

- quality model Part I. Base case model results, *Atmospheric Environment*, 42(39), 8954-8966,
  doi:<u>http://dx.doi.org/10.1016/j.atmosenv.2008.05.064</u>.
- Ying, Q., J. Lu, A. Kaduwela, and M. Kleeman (2008b), Modeling air quality during the California Regional
- 834 PM10/PM2.5 Air Quality Study (CPRAQS) using the UCD/CIT Source Oriented Air Quality Model Part II.
- 835 Regional source apportionment of primary airborne particulate matter, *Atmospheric Environment*, 42(39),
- 836 8967-8978, doi:<u>http://dx.doi.org/10.1016/j.atmosenv.2008.05.065</u>.
- 837 Ying, Q., J. Lu, and M. Kleeman (2009), Modeling air quality during the California Regional PM10/PM2.5 Air
- 838 Quality Study (CPRAQS) using the UCD/CIT source-oriented air quality model Part III. Regional source
- apportionment of secondary and total airborne particulate matter, *Atmospheric Environment*, *43*(2), 419430, doi:http://dx.doi.org/10.1016/j.atmosenv.2008.08.033.
- 841 Young, D. E., H. Kim, C. Parworth, S. Zhou, X. L. Zhang, C. D. Cappa, R. Seco, S. Kim, and Q. Zhang (2016),
- 842 Influences of emission sources and meteorology on aerosol chemistry in a polluted urban environment:
- results from DISCOVER-AQ California, *Atmospheric Chemistry and Physics*, 16(8), 5427-5451,
- 844 doi:10.5194/acp-16-5427-2016.
- Zhang, X. L., H. Kim, C. L. Parworth, D. E. Young, Q. Zhang, A. R. Metcalf, and C. D. Cappa (2016), Optical
- 846 Properties of Wintertime Aerosols from Residential Wood Burning in Fresno, CA: Results from DISCOVER-
- 847 AQ 2013, Environmental Science & Technology, 50(4), 1681-1690, doi:10.1021/acs.est.5b04134.
- Zhang, Y., P. Liu, X.-H. Liu, B. Pun, C. Seigneur, M. Z. Jacobson, and W.-X. Wang (2010), Fine scale modeling
  of wintertime aerosol mass, number, and size distributions in central California, *Journal of Geophysical Besearch: Atmospheres*, *115*(D15), p/a, p/a, doi:10.1020/2000/D012050.
- 850 *Research: Atmospheres, 115*(D15), n/a-n/a, doi:10.1029/2009JD012950.
- Zhu, L., D. Henze, J. Bash, G. R. Jeong, K. Cady-Pereira, M. Shephard, M. Luo, F. Paulot, and S. Capps (2015),
- 852 Global evaluation of ammonia bidirectional exchange and livestock diurnal variation schemes, *Atmos.*
- 853 *Chem. Phys.*, *15*(22), 12823-12843, doi:10.5194/acp-15-12823-2015.
- 854