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# Study on the Migration Rules of Sb in Antimony Ore Soil based on HYDRUS-1D

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**Abstract:** since waste residues of antimony ores are piled disorderly for a long term, heavy metal elements of the residues are dissolved and precipitated under the eluviation effect of snow and rain, causing serious pollution to the surrounding soil. This paper takes the antimony ore of Hunan stannary as the research area, and the surrounding soil of the storage yard of antimony ores as the research object, and carries out a research on the migration rules of Sb in the leachate of waste residues of antimony ores in soil by means of spot sampling, lab analysis and test and simulation, and simulates the migration of Sb in the surrounding soil of mining areas. The results show that through the dynamic penetration experiment of indoor soil columns, the hydrodynamic dispersion coefficient($D=2.485cm^2/h$), adsorption distribution coefficient ($K_d=48.826cm^3/g$), retardation factor ($R_d=78.50$) and other parameters of Sb migrating in soil are obtained; the study makes use of the HYDRUS-1D model to conduct dynamic simulation of Sb in the soil nearby the antimony mine areas on the stannaries of Hunan Province, and the results indicate that the measured value is quite close to the fitted value, and in the correlation analysis of double variables, R is equal to 0.986, indicating that the simulation effect is fairly good. This study aims to provide theoretical foundation and scientific basis for evaluating, controlling, curbing

and repairing the surrounding ecological environment of antimony mine areas.

## 1. Introduction

In the surrounding soil of the areas of antimony ores mining, mineral separation and smelting, the content of antimony increases markedly, and the reason lies in that, on one hand, in the mining and separation process of antimony ores, native antimony ore residual particles formed by antimony-bearing rocks and the formation of secondary minerals result in that antimony is obviously enriched in the surrounding soil of mining areas[1,2].On the other hand, the "three wastes" materials produced in the mining, separation and smelting process of antimony ores become an important source of soil antimony pollution around the mining areas through the wet and dry deposition of atmosphere, rain leaching of waste residues and waste water irrigation[3,4].

HYDRUS-1D is developed and advances on the basis of SUMATRA, WORM, SWMI and other models, and this model can simulate the migration and reaction of one-dimensional flow, nitrogen-phosphorus element, heat and carbon dioxide in the saturated and unsaturated media on a micro or macro scale, and besides, users can make a precise subdivision of grid based on the actual situation of the medium structure, and simulate the transport of a variety of water flows and solutes according to different reactants and boundary conditions. Currently, it has been widely used in agriculture, water conservancy, microbiology, environmental science and other fields[5-7].

## 2. Materials and methods
### 2.1 Profile of the research area

The antimony ore of Hunan stannary is located in Lengshuijiang City, the county-level city in the middle of Hunan Province, and its geographical coordinates are N27°30′49″~N27°50′38″, E111°18′57″~E111°36′40″ respectively. The terrain of

the city is high in the north and south and low in central areas, and takes on an asymmetrical saddle shape, and it belongs to subtropical continental monsoon moist climate. Affected by continental cold-high pressure, northerly winds prevail in winter, and the prevailing wind directions are NE and NNE, and due to subtropical high pressure, southerly winds prevail in summer, and the prevailing wind directions are S and SSE. Lengshuijiang City has four distinctive seasons and is abundant in rainfall, especially, there is much rainfall at the end of spring and the beginning of summer and less rain in the height of summer and early autumn. Its average annual rainfall is 1,444.0 millimeters.

## 2.2 Sample collection and analysis

### 2.2.1 Sample collection and preprocessing of waste residues of antimony ores.

This study collects the samples (waste stone, smelting slag, milltailings) of waste residues of antimony ores by means of spatial simple random distributed points (as shown in Figure 1), and each point respectively takes three layers of samples (0-0.4m, 0.4-0.8m, 0.8-1.2m) with 0.5kg or so for each, after mixing the samples of three layers of each point, put it into a special sample bag, in order to prevent the change of water content in slag sample results in the change of sample properties, eliminate air in the sample bag after bagging and seal and sample bag. Take the three samples of waste residues of antimony ores back to the laboratory and make them dry naturally, respectively mix the three kinds of waste residues evenly on the pulverizer, filter them through nylon mesh(respectively through 8mm, 0.83mm, 0.38mm and 0.15mm mesh), and keep them for stand-by application.

### 2.2.2 Collection of soil samples and physical and chemical analysis

The soil samples collected in this study are the unpolluted undisturbed soils in the close region of antimony ores of the stannary. When collecting the soil samples, strip off the dead twigs and withered leaves on the surface, adopt stratified collection and try not to damage the structure of soil. In addition, collect samples with a soil

engineering cutting ring (diameter: 6.18cm, and height: 4cm) in the same position, and after sealing the collected samples (including the samples of cutting ring) with polyethylene plastic bags, quickly take them back to the laboratory for processing, the measured results are shown in Table 1.

## 2.3 Experiment and results

### 2.3.1 Semi-dynamic leaching experiment

Semi-dynamic leaching experiment[8,9]： take 500g waste stone, smelting slag, milltailings of antimony ores respectively, place them in the leaching column and gently vibrate the leaching column so that the waste residues will be dense and thick, then lay a piece of qualitative filter paper on the surface of waste dues to ensure that the water distribution is uniform when simulating the leaching of acid rain, and add 500mL hyperpure water to wet the waste residues in the leaching column and dewater them after 24 hours. Every day, add it with the pH of 5.0 (annual weighted average of antimony mine areas) representing simulated acid rain amount from January to December, and conduct a semi-dynamic leaching with a peristaltic pump keeping the velocity 30mL/h. The experiment is a total of 12 cycles (parallel and double samples), and based on the different kinds of waste residue (waste rock, smelting slag and milltailings), the precipitation amount of Sb in each waste residue in each month of the year was simulated.

In Figure 2, the same condition of the experiment: the semi-dynamic leaching with the solution with a PH of 5. The different conditions of the experiment: the different kinds of waste residue (waste rock, smelting slag and milltailings). The experiment result: the precipitation amount of Sb in the each waste residue. Thereinto 12 days of the experiment respectively represent the 12 months of the year. See Figure 2 for the variation of Sb with time under the conditions of simulating a semi-dynamic leaching of different waste residues of antimony ores:

## 2.3.2 Dynamic penetration experiment of soil column

Soil column experiment[10'11]：lay a piece of qualitative filter paper, non-woven fabrics and 5cm-thick quartz sand rinsed thoroughly at the bottom before putting it into the soil column, then put the soil with the aggregate thickness of 10cm into it in a stratified manner (soil of each layer is 2cm thick, and make the soil density close to that of the site with a wooden stick, namely $\rho$=1.344g/cm$^3$). After soil filling is completed, lay a layer of 3cm clean quartz sand and a piece of qualitative filter paper on the surface of soil in turn to make the water distribution evenly in the test, and control the leaching rate with a peristaltic pump (as shown in Figure 3).

In the penetration test of Cl$^-$ (the solution concentration of standard Cl$^-$ is 0.01mol/L), and the change rule of the concentration of chloride ions with time is shown in Figure 4.

In the penetration test of Sb in the tested soil, take the same test devices used in the penetration test of chloride ions (the concentration of standard Sb solution is 10 mg/L and the pH of the standard antimony solution is 8.0), and the change rule of the concentration of Sb at the sample exit at the bottom of the soil column along with time is shown in Figure 5(when the antimony concentration reached 9.979mg/L after the 1780-hours experiment, the pH of leachate at the bottom of the soil column was 7.9).

## 3. Model building and operation

### 3.1 Equation of flow and solute transport

In some saturated porous media, the equation of motion of one-dimensional uniform flow is commonly represented by the corrected form of Richards[12], and in the equation, the effects of thermal gradient and vapour-phase period in the process of liquid phase flow are ignored:

$$\frac{\partial \theta(h)}{\partial t} = \frac{\partial}{\partial z}\left[K(h)\frac{\partial h}{\partial z} + K(h)\right] - S(h)$$

(1)

In the equation, h is the pressure head (L), which is positive when saturated and negative when unsaturated; $\theta$ is volume water content ($L^3/L^3$); z is the vertical coordinate(L), which is possible when it is upwards. S is the source sink term ($T^{-1}$); t is time (T); K is unsaturated hydraulic conductivity (L/T). The two hydraulic conductivity parameters $\theta(h)$ and $K(h)$ in the equation are the nonlinear functions of the pressure head H. HYDRUS-1D provides 6 different analytical models of hydraulic characteristics. This study adopts the hydraulic conductivity function of Van Genuchten model[13,14], and the curve function of hydraulic characteristics of the soil is derived based on a large quantity of previous experimental research, and the expression is as follows:

$$\theta(h) = \theta r + \frac{\theta s - \theta r}{\left[1 + |\alpha h|^n\right]^m}$$

(2)

$$K(h) = Ks \cdot Se^l \left[1 - (1 - Se^{1/m})^m\right]^2$$
$$Se = \frac{\theta - \theta r}{\theta s - \theta r}$$

(3)

In the expression, m=1-1/n, n >1; $\theta$ is the volume moisture content of the soil; $\theta_s$ is the saturated water content of the soil; $\theta_r$ is the residual water content of the soil; $K_s$ is the saturated hydraulic conductivity; $S_e$ is the relative saturation; $\alpha$, n and l are the empirical parameters of hydraulic characteristic curve.

The model of solute transport of HUDRUS-1D follows the convection-dispersion equation[15,16]. Considering the effect of soil on solute adsorption, the equation of solute control can be expressed as follows:

$$\frac{\partial \theta c}{\partial t} + \frac{\partial \rho s}{\partial t} = \frac{\partial}{\partial z}(\theta D \cdot \frac{\partial c}{\partial z}) - \frac{\partial qc}{\partial z} - \mu_w \theta c - \mu_s \rho s + \gamma_w \theta + \gamma_s \rho - SC_s$$

(4)

In the equation, c is the concentration of solute in the solution; D is the longitudinal dispersion coefficient; q is the velocity of Darcy single width; $\mu_w$ is the first order natural attenuation coefficient (liquid phase); $\mu_s$ is the first order natural

attenuation coefficient (solid phase); $\gamma_w$ is the zero order reaction rate constant (liquid phase); $\gamma_s$ is the zero order order reaction rate constant (solid phase). $\rho$ is the density of the soil; S is the root adsorption item; $C_s$ is the concentration of absorption solute in the root system. In this study, ignoring the absorption item of plants and not considering the attenuation of heavy metals, the equation can be simplified as:

$$\frac{\partial \theta c}{\partial t} + \frac{\partial \rho s}{\partial t} = \frac{\partial}{\partial z}(\theta D \cdot \frac{\partial c}{\partial z}) - \frac{\partial qc}{\partial z}$$

(5)

Thereinto,

$$D = \varepsilon \frac{|q|}{\theta} + D^o \tau, \tau = \theta^{7/3} / \theta s^2$$

(6)

In the expression, $D_0$ is the free diffusion coefficient of the solute in the still water; $\varepsilon$ is the degree of dispersion of the soil; $\tau$ is the tortuous factor of the soil pores, which is related to the soil structure.

## 3.2 Selection of initial conditions and boundary conditions

For independent systems, the equations of flow and solute transport must be solved by specifying the following conditions[17], soil surface (z=l), and infiltration surface of soil profile (z=l):

$$h(z,t) = h_0(z), z = 0 \quad or \quad z = l$$

$$-K(\frac{\partial h}{\partial z} + \cos\alpha) = q_0(t), z = 0 \quad or \quad z = l$$

$$\frac{\partial h}{\partial z} = 0, z = 0$$

In the expression, $h_0$ and $q_0$ are the initial values of the boundary pressure head and the water flux of soil.

$$c(z,t)\big|_{z=a} = c_a(t), c(z,t)\big|_{z=b} = c_b(t)$$

In the expression, $c_a(t)$ and $c_b(t)$ are the given concentrations of the two endpoints a and b in the research area.

$$n(D\frac{\partial c}{\partial z}-u_z c)\cos(n,z)\Big|_{z=a}= g(t),t\geq 0$$

In the expression, g(t) is the given pollutant convection diffusion flux function on the Cauchy conditions of r, take a positive value for the research area of inflow and a negative value for the research area of outflow.

**3.3 Determination of hydrodynamics and migration parameters**

**3.3.1 Hydrodynamics parameters and dispersion coefficient of soil medium**

According to the analysis and measurement results of the soil bulk density and particles after collecting soil on site and dry it naturally in the lab, it is able to obtain all hydrodynamic parameters by inputting Neural Network Prediction in HYDRUS-1D, and the results are shown in Table 2.

This study makes use of the experimental data of the chloride ion (tracer) penetration curve to calculate the hydrodynamic diffusion coefficient (D). Not considering the migration and degradation of chloride ions in soil migration, and process the hydrodynamic diffusion coefficient according to one-dimensional (vertical) issue[18˙19], and the approximate solution is calculated with the following formula:

$$D=\frac{v^2}{8t_{0.5}}(t_{0.84}-t_{0.16})^2$$

(7)

In the expression, $t_{0.16}$ is the time when the relative concentration $C_t/C_0$ of the sampling point is up to 0.16; $t_{0.5}$ and $t_{0.84}$ is in a similar way; v is the actual flow velocity of flow in soil column, which is determined by $v=L/t_{0.5}$; L is the actual length of soil column. In this study, L value is 10cm. The values of $t_{0.16}$, $t_{0.5}$ and $t_{0.84}$ are calculated by way of interpolation, and the calculated value of D is shown in Table 3.

**3.3.2 Sb absorption distribution coefficient and retardation factor**

Adsorption can affect the migration of heavy metals in soil and the environmental

remediation of the site polluted by heavy metals. At present, it is difficult to solve the adsorbed distribution coefficient ($K_d$) and retardation factor ($R_d$) with a nonlinear equation. Therefore, the linear adsorption equation is adopted to solve $K_d$ and $R_d$[20.21], and the formula is as follows:

$$K_d = \frac{\eta_e}{\rho_b}(\frac{vt_{0.5}}{L} - 1) \tag{8}$$

$$R_d = 1 + \frac{\rho_b}{\eta_e} \cdot K_d \tag{9}$$

In the expression, $\rho_b$ is the dry bulk density of the soil, g/cm$^3$; $\eta_e$ is the effective porosity of the soil medium, %; $t_{0.5}$ is the time when the relative concentration $C/C_0$ at the sampling point reaches 0.5 ( it can be obtained from the penetration curve of Sb in the figure that $t_{0.5}$=228.33h), h; v is the the actual flow rate of the solution in the soil column, which can be determined by the formula $v=L/t_{0.5}$; L is the distance from the bottom sampling port to the starting point of the soil column (when L is 10cm).

Put $\rho_b$, $\eta_e$, $t_{0.5}$, L and v into (8), then: $K_d$=48.826×10$^{-3}$ (m$^3$/kg);

Put $\rho_b$, $\eta_e$ and $K_d$ into (9), then:$R_d$=78.50.

## 4. Simulation results and analysis

## 4.1 Migration simulation of antimony

The parameters required for simulating the migration of Sb in the soil column by making use of HYDRUS-1D model are shown in Table 4, and when defining the model boundary, define it as the boundary of constant water head (5cm water head), and set the lower boundary to be free drainage boundary. The simulation results are shown in Figure 6 and Figure 7.

In Figure 6, it can be drawn that the concentration of Sb at the 10cm point of the soil column reaches 9.1mg/L after 1000-hour migration; In Figure 7, it can be drawn that the measured values (Figure 5) and the simulated value are highly close to each other, and the bivariate correlation analysis is conducted via SPSS19.0: the correlation coefficient between the measured value and the simulated value is 0.986 (P=0.001), which indicates that the model can well simulate the migration of Sb in the local soil and the simulation results are relatively reliable.

## 4.2 Application and prediction of antimony migration model

According to the release rule of Sb under the condition of simulating acid rain

leaching of waste residues of antimony ores, it can be concluded that the concentration of Sb in acid rain leaching of smelting slags of antimony ores is the highest at 35.20mg/L. This study takes the highest concentration of Sb in leaching solution as the concentration of input liquid, builds the model with HYDRUS-1D, and makes use of the parameters of Table 4 to simulate the migration of Sb in the storage yard of waste residues of antimony ores in the surrounding soil. According to the actual investigation on groundwater level of the surrounding soil in the storage yard of waste residues of antimony ores, it is found that the groundwater level is buried shallow, and it is generally 70-90cm from the earth surface. Assume that the soil is 80cm thick, simulate the migration of Sb into the soil, and analyze the possible groundwater pollution. See Figure 8 for the simulation results.

It can be seen from Figure 8 that it takes over 500 days for the Sb in storage yard of waste residues of antimony ores to penetrate 80cm thick soil, indicating that the penetration ability of Sb is relatively weak. It requires 597.3d to completely penetrate the 80cm soil, but only needs 160.3d to produce the leachate exceeding the limit value (0.005mg/L) of Sanitary Standard for Drinking Water which is injected into the groundwater. With the increase in the deposit volume of waste residues of antimony ores, the concentration of Sb in the leaching solution of waste residues of antimony ores is also enhanced.

## 5. Conclusions

This study carries out an analytical test of hydrodynamic and migration parameters of Sb in soil by means of an indoor soil column test. Besides, it takes a mathematical model to simulate the migration of Sb in the soil column, and on the basis of Sb migration model and Sb precipitation rule under the condition of simulating the acid rain leaching of waste residues of antimony ores, it conduct a simulation and prediction of the migration of Sb in the leaching solution of waste residues of antimony ores in the waste residue disposal site of mine areas as well as its pollution to the groundwater, the conclusions are as follows:

(1) The study takes advantage of the indoor soil column dynamic simulation test to obtain the hydrodynamic dispersion coefficient ($D=2.485cm^2/h$), adsorption distribution coefficient ($K_d=48.826cm^3/g$) and retardation factor ($R_d=78.50$) and the like of Sb in soil migration, the time required for Sb to completely penetrate the soil column is 100 times of $Cl^-$, indicating that Sb has obvious tailing phenomena in soil migration.

(2) It takes advantage of the HYDRUS-1D model to conduct a dynamic simulation of the migration of Sb in the surrounding soil of antimony mine areas of Hunan stannary,

and the results show that the measured value and the predicted value are rather close to each other, and the analyzed bivariate correlation is R=0.986, indicating that the simulation parameters are selected appropriately.

(3) The study applies the HYDRUS-1D model in the migration simulation of Sb in the surrounding soil of the storage yard of waste residues of antimony ores, and it takes 597.3d for Sb ($C_0$=35.2mg/L) in the leaching solution of waste residues of antimony ores to completely penetrate the 80cm soil in the surrounding storage yard of waste residues. Therefore, the leaching solution of waste residues in the storage yard of waste residues of antimony ores cause serious Sb pollution to the surrounding environment.

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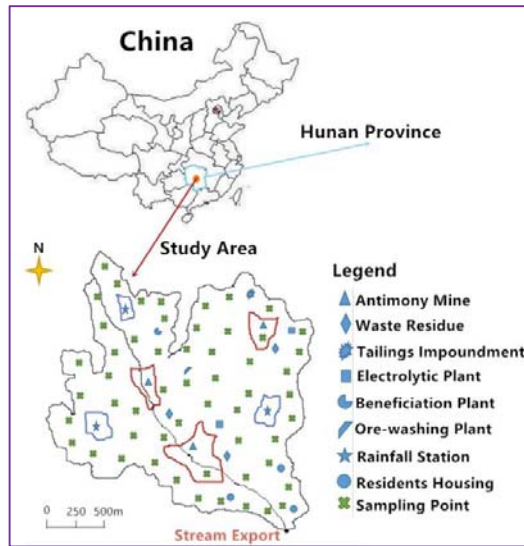
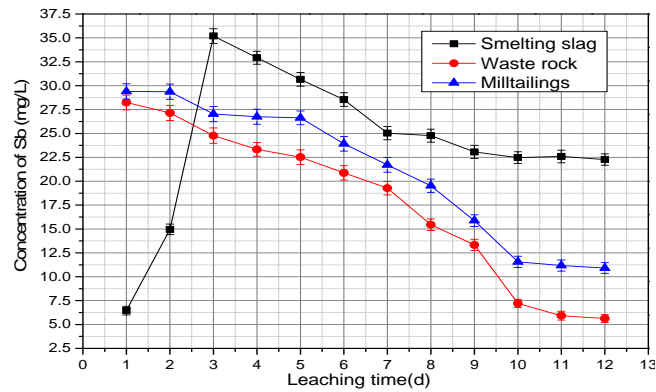Figure 1    Profile of the research area



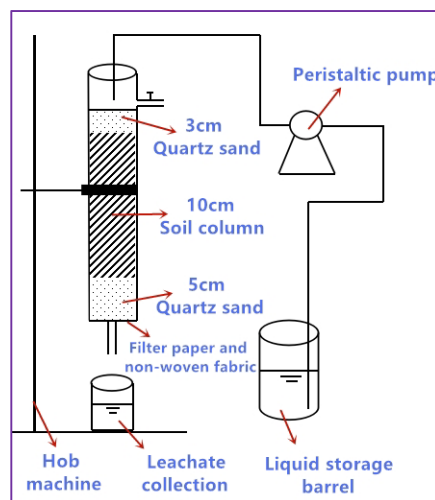Figure 2    Leaching and precipitation of antimony in different categories of wastes
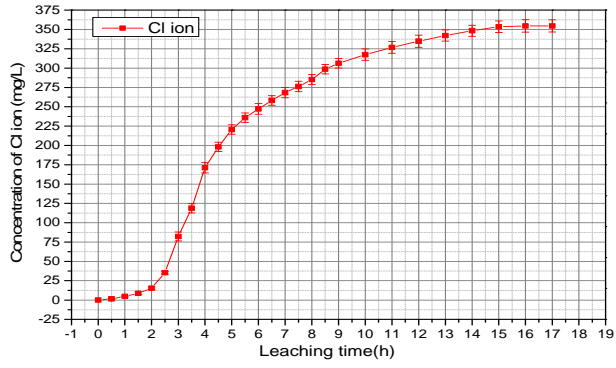


Figure 3 Device of the soil column experiment

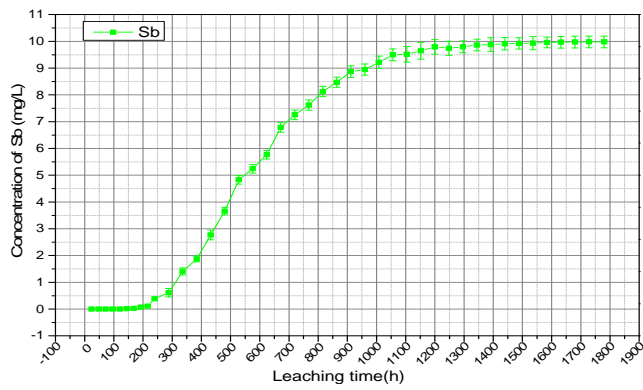Figure 4    Breakthrough curve of Cl⁻ in soil



Figure 5    Breakthrough curve of the concentration of Sb
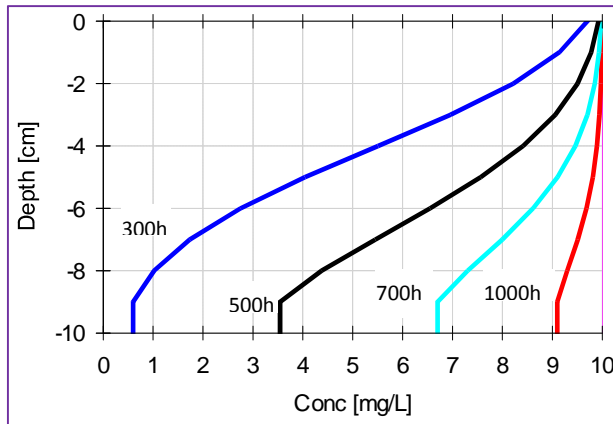


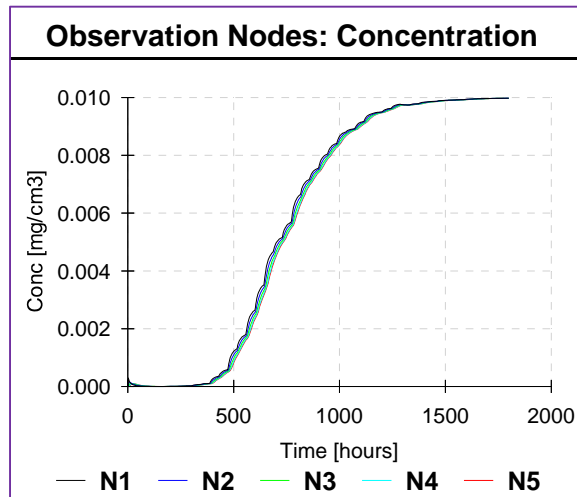Figure 6 Distribution of Sb content of soil profile at different times

Figure 7   Predicted value of simulation of Sb content at observation points

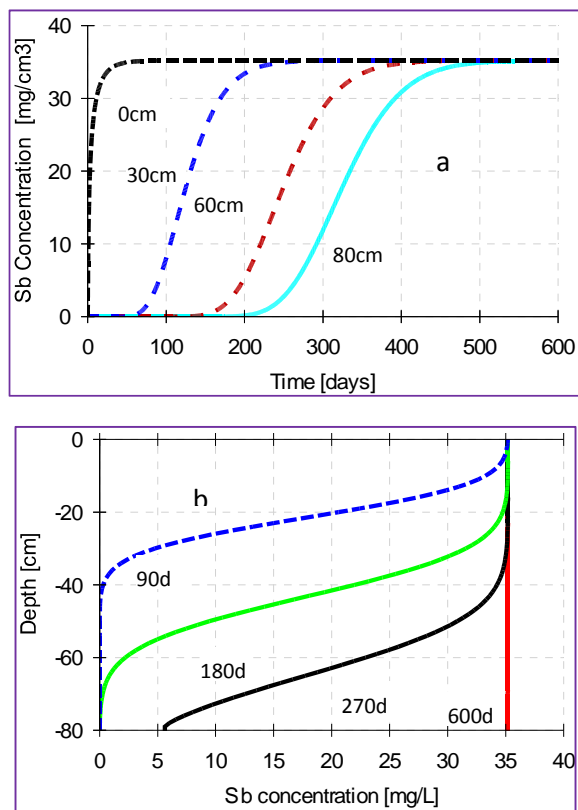(Figure 7 is prepared in colour)



Figure 8   Predicting results of the migration of Sb in soil for 600 days

(a: different profile depths; b: different times/d)

Table 1 Analysis of physical and chemical properties of soil

| Item | Index content and measured value | | | |
|---|---|---|---|---|
| Particle size content (%) | Sand particle (2~0.05mm) | Silt particle (0.05~0.002mm) | Clay particle (<0.002mm) | |
| | 33 | 27 | 40 | |
| Physical properties | Soil category | Soil density (g/cm$^3$) | Soil moisture content (%) | Porosity (%) |
| | Red soil | 1.344 | 20.25 | 49.28 |
| Chemical properties | pH | | Organic matter | Sb content |
| | Water | KCl | (%) | (mg/kg) |
| | 6.75 | 5.32 | 0.45 | 1.012 |

Table 2 Motion parameters of soil moisture

| $\theta_r$ | $\theta_s$ | a | n | $K_s$ | l |
|---|---|---|---|---|---|
| 0.0911 | 0.4683 | 0.0155 | 1.3691 | 15.96 | 0.5 |

Table 3 Calculation of the hydrodynamic dispersion coefficient D

| v(cm/h) | $t_{0.16}$(h) | $t_{0.5}$(h) | $t_{0.84}$(h) | D(cm$^2$/h) |
|---|---|---|---|---|
| 2.448 | 2.728 | 4.085 | 8.488 | 2.485 |

Table 4 Parameters required for migration simulation of Sb

| Parameter | Value | Parameter | Value |
|---|---|---|---|
| $\theta_r$ (cm$^3$/cm$^3$) | 0.0911 | L(cm) | 10 |
| $\theta_s$ (cm$^3$/cm$^3$) | 0.4683 | $C_0$(mg/L) | 10 |
| a (1/cm) | 0.0155 | $\rho_b$(g/cm$^3$) | 1.344 |
| n | 1.3691 | $K_d$(cm$^3$/g) | 48.826 |
| $K_s$ (cm/h) | 0.665 | D(cm$^2$/h) | 2.485 |

| 1 | 0.5 | T(h) | 2400 |
| --- | --- | --- | --- |