

*Candida rugosa* LIPASE SUPPORTED ON SILICA-COATED MAGNETITE  
NANOPARTICLES FOR HYDROLYSIS OF OLIVE OIL

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## **DEDICATION**

This dissertation is dedicated to my own sake, my husband, parents and family.

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## ABSTRACT

Oil palm leaves (OPL) has high content of silica ( $\text{SiO}_2$ ).  $\text{SiO}_2$  has a high surface area and large pore volume which could reduce the aggregation of magnetite ( $\text{Fe}_3\text{O}_4$ ). The coating of the superparamagnetic  $\text{Fe}_3\text{O}_4$  was to enable easy separation from the reaction mixture.  $\text{SiO}_2$  extracted from OPL was coated on  $\text{Fe}_3\text{O}_4$  followed by functionalization of 3-aminopropyltriethoxysilane (APTES) and activation of glutaraldehyde to prepare a nanosupport (G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$ ) for immobilisation of *Candida rugosa* lipase (CRL). The feasibility of the biocatalyst (CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$ ) has yet to be tested in aqueous environment. In this research, the CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  was used to determine the optimum condition for hydrolysis of olive oil. The kinetic and thermodynamic properties of the CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  was investigated for the hydrolysis of olive oil. The study first characterised the components and the treated OPL, whereby data of the thermal gravimetric analysis (TGA) indicated that the hemicellulose and lignin components in OPL were successfully reduced by acid treatment and calcination. The morphological and physiochemical facets of the extracted  $\text{SiO}_2$  were investigated by fourier transform infrared (FTIR) spectroscopy, X-ray diffraction (XRD) and thermogravimetry analysis-differential scanning calorimetry (TGA-DTG). The results revealed that  $\text{SiO}_2$  was successfully extracted from OPL and coated on the  $\text{Fe}_3\text{O}_4$ . Subsequently, it was activated by APTES and glutaraldehyde to yield CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$ . FTIR, XRD and TGA-DTG data showed that CRL was successfully immobilised on G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$ , as seen with the band arising at  $1639\text{ cm}^{-1}$  by C=O and C=N stretching in FTIR. Specifically, immobilisation of CRL onto the G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  yielded an enzyme loading and specific activity of  $14.7\text{ mg/g}$  and  $183\text{ U/g}$ . The CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  was then tested to establish the optimal conditions for catalysing hydrolysis of olive oil. It was found that the optimal conditions for the reaction that yielded the best activity were enzyme loading of  $1.00\text{ mg/mL}$ , incubation temperature of  $40\text{ }^\circ\text{C}$ , pH 8.0, ratio of olive oil: water of 2.5:1, and an agitation speed of  $200\text{ rpm}$ . Assessments of thermal stability showed that CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  was more resistant to thermally-induced denaturation, than free CRL. The CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  was kinetically shown to have higher affinity towards the substrate (Michaelis-Menten constant,  $K_m = 0.583\text{ g/mL}$ ) but catalysed at a lower maximum rate of reaction ( $V_{\text{max}} = 833.3\text{ }\mu\text{mol/ml.min}$ ) as compared to free CRL ( $K_m = 6.00\text{ g/mL}$ ,  $V_{\text{max}} = 3330\text{ }\mu\text{mol/ml.min}$ ), respectively. The thermodynamic parameters based on values of half-life ( $t_{1/2} = 38.94\text{ min}$ ), D-value ( $129.4\text{ min}$ ), thermal deactivation energy ( $E_d = 112.90\text{ kJ/mol}$ ), standard enthalpy of deactivation ( $\Delta H_d^\circ = 110.10\text{ kJ/mol}$ ) and Gibbs free energy ( $\Delta G_d^\circ = 11.32\text{ kJ/mol}$ ) for CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  conclusively showed that the lipase was appreciably more thermostable than free CRL ( $t_{1/2} = 23.89\text{ min}$ , D-value =  $79.67\text{ min}$ ,  $E_d = 93.3\text{ kJ/mol}$ ,  $\Delta H_d^\circ = 87.5\text{ kJ/mol}$  and  $\Delta G_d^\circ = 9.8111\text{ kJ/mol}$ ) at  $60^\circ\text{C}$ . The finding shows that  $\text{SiO}_2$  extracted from OPL could be coated on  $\text{Fe}_3\text{O}_4$  to be used as an inorganic support for enzyme immobilisation. The results thus demonstrated that the CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  biocatalyst was a potential candidate for catalysing hydrolytic reactions with good reaction rates, thus envisaging its prospective application as a commercially relevant biocatalyst.

## ABSTRAK

Daun kelapa sawit (OPL) mempunyai kandungan silika ( $\text{SiO}_2$ ) dan luas permukaan yang tinggi yang boleh mengurangkan pengagregatan magnetit ( $\text{Fe}_3\text{O}_4$ ). Penyalutan  $\text{Fe}_3\text{O}_4$  yang superparamagnetik ialah bagi membolehkan pemisahan lebih mudah daripada campuran reaksi.  $\text{SiO}_2$  yang diekstrak dari OPL dilapisi pada  $\text{Fe}_3\text{O}_4$  diikuti dengan fungsionalisasi 3-aminopropiltriethoxysilana (APTES) dan pengaktifan glutaraldehid untuk menyediakan-penyokong nano (G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$ ) bagi pemegungan lipase *Candida rugosa* (CRL). Kemampuan pemangkin lipase yang dipegunkan ini (CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$ ) masih belum diuji dalam persekitaran berair. Di dalam kajian ini, CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  digunakan untuk menentukan keadaan optimum untuk hidrolisis minyak zaitun. Ciri-ciri kinetik dan termodinamik CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  disiasat untuk hidrolisis minyak zaitun. Kajian pertama mencirikan komponen dan OPL yang dirawat, di mana data analisis TGA menunjukkan bahawa komponen hemiselulosa dan lignin dalam OPL telah berjaya dikurangkan dengan rawatan asid dan kalsinasi. Sifat morfologi dan fisiologi dari  $\text{SiO}_2$  yang diekstrak disiasat oleh FTIR, XRD dan TGA-DTG. Hasilnya menunjukkan bahawa  $\text{SiO}_2$  berjaya diekstrak dari OPL dan dilapisi pada  $\text{Fe}_3\text{O}_4$ . Selepas itu, ia diaktifkan oleh APTES dan glutaraldehyde untuk menghasilkan CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$ . Data FTIR, XRD dan TGA-DTG menunjukkan bahawa CRL berjaya dipegunkan pada G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$ , seperti yang dilihat dengan jalur yang timbul pada  $1639\text{ cm}^{-1}$  oleh C=O dan C=N yang ditunjukkan dalam FTIR. Khususnya, imobilisasi CRL ke G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  menghasilkan pemuatan enzim dan aktiviti khusus  $14.7\text{ mg/g}$  dan  $183\text{ U/g}$ . CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  kemudiannya diuji untuk menentukan keadaan optimum untuk menghidrolisis minyak zaitun. Didapati bahawa keadaan optimum untuk tindak balas yang menghasilkan aktiviti terbaik ialah penggunaan enzim  $1.00\text{ mg/mL}$ , suhu inkubasi  $40\text{ }^\circ\text{C}$ , pH  $8.0$ , nisbah minyak zaitun: air  $2.5: 1$ , dan kelajuan agitasi  $200\text{ rpm}$ . Penilaian kestabilan terma menunjukkan bahawa CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  lebih tahan terhadap denaturasi yang disebabkan oleh haba, daripada CRL bebas. CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  secara kinetika ditunjukkan mempunyai pertalian yang lebih tinggi terhadap substrat (Michaelis-Menten malar,  $K_m = 0.583\text{ g/mL}$ ) tetapi termangkin pada kadar tindak balas maksimum yang lebih rendah ( $V_{\text{max}} = 833.3\text{ }\mu\text{mol/ml. min}$ ) berbanding dengan CRL bebas ( $K_m = 6.00\text{ g / mL}$ ,  $V_{\text{max}} = 3330\text{ }\mu\text{mol/ml.min}$ ). Parameter termodinamik berdasarkan nilai-nilai separuh hayat ( $t_{1/2} = 38.94\text{ min}$ ), nilai D ( $129.4\text{ min}$ ),  $E_d = 112.90\text{ kJ/mol}$ ,  $\Delta H_d^\circ = 110.10\text{ kJ/mol}$ ) dan  $\Delta G_d^\circ = 11.32\text{ kJ/mol}$  untuk CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  secara konsisten menunjukkan bahawa lipase lebih tinggi kestabilan termanya daripada CRL bebas ( $t_{1/2} = 23.89\text{ min}$ , D-value =  $79.67\text{ min}$ ,  $E_d = 93.3\text{ kJ / mol}$ ,  $\Delta H_d^\circ = 87.5\text{ kJ/mol}$  dan  $\Delta G_d^\circ = 9.8111\text{ kJ/mol}$ ) pada  $60\text{ }^\circ\text{C}$ . Hasil kajian menunjukkan bahawa  $\text{SiO}_2$  yang diekstrak dari OPL boleh disalut pada  $\text{Fe}_3\text{O}_4$  untuk digunakan sebagai sokongan bukan organik untuk enzim terpegun. Hasilnya menunjukkan bahawa pemangkin CRL/G-AP- $\text{SiO}_2$ - $\text{Fe}_3\text{O}_4$  adalah berpotensi untuk memangkin tindak balas hidrolisis dengan kadar reaksi yang baik, dengan itu membayangkan penerapan prospektifnya sebagai pemangkin komersial yang berkaitan.

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## LIST OF ABBREVIATIONS

APTES	-	3-aminopropyltriethoxysilane
BSA	-	Bovine serum albumin
CRL	-	<i>Candida rugosa</i> lipase
$E_d$	-	Deactivation energy
$FeCl_2 \cdot 4H_2O$	-	Iron(II) chloride tetrahydrate
$FeCl_3 \cdot 6H_2O$	-	Iron(III) chloride hexahydrate
$FeSO_4 \cdot 7H_2O$	-	Iron(II) sulphate heptahydrate
FTIR	-	Fourier transform infrared spectroscopy
$Fe_3O_4$	-	Magnetic Nanoparticles
G	-	Glutaraldehyde
$H_2SO_4$	-	Sulphuric acid
HCl	-	Hydrochloric acid
$K_2HPO_4$	-	Dipotassium hydrogen phosphate
$KH_2PO_4$	-	Potassium dihydrogen phosphate
$K_d$	-	Deactivation constant
$K_m$	-	Michaelis-Menten constant
NaOH	-	Sodium hydroxide
-NH <sub>2</sub>	-	Amine group
NH <sub>4</sub> OH	-	Ammonium hydroxide
OPF	-	Oil palm fronds
OPFL	-	Oil palm frond leaves
OPL	-	Oil palm leaves
OPLA	-	Oil palm leaves ash
OPT	-	Oil palm trunks
OVAT	-	One-variable-at-a-time
$R^2$	-	Coefficient of determination
-SH	-	Thio group
$SiO_2$	-	Silica
TGA	-	Thermogravimetric analysis
$V_{max}$	-	Maximum rate of reaction

## LIST OF SYMBOLS

$^{\circ}\text{C}$	-	degree Celsius
$E_d$	-	denaturation activation energy
g	-	gram
h	-	hour
K	-	kelvin
kDa	-	kilo Dalton
$k_d$	-	deactivation rate constant
$\text{kJ/mol}$	-	kilojoules per mole
l	-	litre
mg	-	milligrams
min	-	minute
ml	-	millilitre
mM	-	millimolar
$\text{mg/g}$	-	milligram per gram
rpm	-	rotation per minutes
s	-	second
SF	-	stabilisation factor
$t_{1/2}$	-	half life
U	-	units
$\mu\text{mol}$	-	micro mole
v/v	-	volume per volume
w/v	-	weight per volume
w/w	-	weight per weight
%	-	percentage
$\Delta H_d^{\circ}$	-	standard energy of deactivation
$\Delta S_d^{\circ}$	-	standard entropy of deactivation
$\Delta G_d^{\circ}$	-	standard free energy of deactivation

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# CHAPTER 1

## INTRODUCTION

### 1.1 Background of Study

Oil palm or known as *Elaeis guineensis* is known as the most important plant species found in the *Elaeis* genus that belongs to the *Palmae* family. It is planted in large plantations in many tropical countries, for instance, Malaysia, Thailand and Indonesia (Nordin, Sulaiman, Hashim, & Mohamad Kassim, 2016). Palm oil extracted from pulp and kernel of the fruit are useful as edible oil to manufacture soap, food products, flavours and etc. Nonetheless, the high production rate of palm oil has taken its toll on the environment, as these large plantations leave behind large quantities of biomass, approximately 50-70 tonnes per hectare of the plantation (Shuit, Tan, Lee, & Kamaruddin, 2009). Oil palm fronds (OPF) including the oil palm leaves (OPL) and petioles constitute which is approximately 47% of the total oil palm waste (Nordin, Sulaiman, Hashim, & Mohamad Kassim, 2017). So far, these wastes are not fully utilised or recycled effectively, and are often eliminated by land-filling and open burning. Open burning in particular, pollutes the environment and damages the ecosystem (Sidik, Jalil, Triwahyono, Adam, Satar, & Hameed, 2012). Review of the literature conveyed that agricultural biomass i.e. OPL are potential raw materials for the manufacturing a myriad of value-added products such as animal food, fertiliser and absorbent. This is because of the abundance, readily available, and low-cost material of OPL. OPL in most part is used as a pelleting feed stock pulp and for the manufacturing of paper (Onoja, Attan, Chandren, Abdul Razak, Abdul Keyon, Mahat, & Wahab, 2017; Onoja, Chandren, Abdul Razak, Mahat, & Wahab, 2018a; Onoja, Chandren, Razak, & Wahab, 2018b).

According to a few reports, approximately 72.6% silica ( $\text{SiO}_2$ ) exists in palm oil fuel ash and, as much as 46.0 % is found in oil palm ash (Adam, Sulaiman, Baharuddin, Mokhtar, Busu, & Tengku Zainal Mulok, 2017; Faizul, Abdullah, &

Fazlul, 2012). In addition, another study reported that 95.2 % of SiO<sub>2</sub> found in acid treated OPL (Onoja *et al.*, 2017). These works highlighted the potential of OPL as a renewable SiO<sub>2</sub> source, aside to SiO<sub>2</sub> sources that are mined from the earth's crust (Faizul *et al.*, 2012). SiO<sub>2</sub> is prized for its multiple applications, largely because of its natural abundance of silanol groups (-SiOH), which contributes to compatibility for interaction with different types of proteins. SiO<sub>2</sub> is generally used for improving the stability, biocompatibility, hydrophilicity and surface functionality of the nanoparticles, i.e. magnetite (Fe<sub>3</sub>O<sub>4</sub>) (Abbas, Torati, Soo Lee, Rinaldi, & Kim, 2014). Onoja *et al.* (2018a) also reported the use of SiO<sub>2</sub> from OPL ash (OPLA) as a nanocoating material over nanoparticles of Fe<sub>3</sub>O<sub>4</sub> to covalently bind *Candida rugosa* lipase (CRL). The resultant biocatalyst was appreciably activated and stabilised by the SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> hybrid nanosupport (Onoja *et al.*, 2017; Onoja *et al.*, 2018a; Onoja *et al.*, 2018b).

Studies have shown that enzyme immobilisation onto solid supports can effectively distribute the enzyme molecules and prevent formation of inactive aggregates, in conjunction to stabilising their structures (Prlainovic, Bezbradica, Rogan, Uskokovic, Mijin, & Marinkovic, 2016). For such a purpose, inert polymers and inorganic materials are the preferred choice of carrier matrices. This has to do with their stability, inherent physical strength, regenerability, ability to increase catalytic activity, microbial resistance and reduce inhibition by the product during reactions (Datta, Christena, & Rajaram, 2012). In this milieu, the immobilisation of enzymes onto nanostructures such as nanoparticles, nanofibers, nanotubes and nanocomposites have been a topic of active research in enzyme technology. The high specific surface area of nanosupports is useful in enhancing the binding capacity of the enzymes, lowering transfer resistance with minimum diffusion limitation and lower operational cost (Singh & Mukhopadhyay, 2014). In fact, Fe<sub>3</sub>O<sub>4</sub> are quite popular supports for immobilising enzymes, as the nanoparticles facilitate easy separation of the biocatalyst from the reaction mixture using an external magnetic field. This approach permits the reuse of enzymes in continuous operations (Singh *et al.*, 2014). It was previously demonstrated that the coating of SiO<sub>2</sub> over Fe<sub>3</sub>O<sub>4</sub> (G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub>) before the covalent binding of CRL onto the surface of the support (Onoja *et al.*, 2018a) yielded a more activated and stabilised biocatalyst (CRL/G-AP-

SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub>) for catalysing an esterification production of butyl butyrate. This work proves that CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> was highly active in a water-free system. However, many questions remain unanswered in terms of stability and activity of the lipase when catalysing in a water-based system, i.e. a hydrolytic reaction (Onoja *et al.*, 2017; Onoja *et al.*, 2018a; Onoja *et al.*, 2018b).

## 1.2 Problem Statement

Although the G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> nanosupport was shown to be applicable for activation and stabilisation of CRL for an esterification reaction, the same cannot be assumed for its capability to improve lipase activity for a hydrolysis reaction. This is because conditions in an enzyme-catalysed esterification reaction are highly different to that in an aqueous reaction system. CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> is adversely affected by water molecules in an esterification reaction in which the produced ester is counterproductively hydrolysed in the presence of excess water (Abd Rahman, Abd Manan, Marzuki, Mahat, Attan, Abdul Keyon, Jamalis, Aboul-Enein, & Wahab, 2017; Elias, Chandren, Razak, Jamalis, Widodo, & Wahab, 2018; Manan, Attan, Zakaria, Keyon, & Wahab, 2018). Likewise, the high quantities of alcohol and acids of the starting materials can affect the conformation of CRL, causing the lipase to alter its activity and stability. In an aqueous system, all the above are absent and, the CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> was predicted to act differently, thereby influencing its stability, as well as the reaction kinetic and thermodynamic properties.

Herein, this study investigates the ability of CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> to catalyse a hydrolytic reaction, i.e. hydrolysis of olive oil. It is worth noting here that data on the ability of CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> to catalyse hydrolytic reactions remain unavailable. Thus, the findings of this study would greatly contribute to the body of knowledge in terms of the versatility of CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> as a biocatalyst. This study hypothesised that the catalytic properties CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> in a water-based system would be differ to that of the aqueous-free i.e. esterification. The presence of water would greatly change the stability of CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> in

catalysing the reaction, alongside changes in its kinetic and thermodynamic properties.

### 1.3 Research Objectives

The objectives of this study are:

- (i) To prepare and characterise the morphology of  $\text{SiO}_2$  and  $\text{Fe}_3\text{O}_4$ .
- (ii) To characterise the morphology and physiochemical properties of CRL immobilised onto oil-palm leaves ash (OPLA)-based magnetite-silica matrix ( $\text{CRL/G-AP-SiO}_2\text{-Fe}_3\text{O}_4$ ).
- (iii) To optimise and compare the free CRL and  $\text{CRL/G-AP-SiO}_2\text{-Fe}_3\text{O}_4$ -catalysed hydrolysis of olive oil.
- (iv) To assess the stability, as well as the kinetic and thermodynamic properties of the  $\text{CRL/G-AP-SiO}_2\text{-Fe}_3\text{O}_4$ -catalysed hydrolysis of olive oil.

### 1.4 Scope of Study

The study begins with the collection of OPL from Universiti Teknologi Malaysia (UTM). The OPL collected were washed and ground into powder for acid treatment. It was then calcined to obtain treated oil palm leaves ash (OPLA). Untreated OPL and treated OPLA were characterised using thermo gravimetric analysis (TGA) and fourier transform infrared (FTIR) spectroscopy. The  $\text{SiO}_2$  extracted from OPLA was coated onto  $\text{Fe}_3\text{O}_4$  to produce  $\text{SiO}_2\text{-Fe}_3\text{O}_4$  nanosupport. It was then activated with 3-aminopropyltriethoxysilane (APTES) and functionalised by glutaraldehyde to give  $\text{G-AP-SiO}_2\text{-Fe}_3\text{O}_4$ . Subsequently, CRL was immobilised onto  $\text{G-AP-SiO}_2\text{-Fe}_3\text{O}_4$  to produce the biocatalyst ( $\text{CRL/G-AP-SiO}_2\text{-Fe}_3\text{O}_4$ ). The morphology and physiochemical properties  $\text{SiO}_2\text{-Fe}_3\text{O}_4$ ,  $\text{G-AP-SiO}_2\text{-Fe}_3\text{O}_4$  and  $\text{CRL/G-AP-SiO}_2\text{-Fe}_3\text{O}_4$  were characterised by using FTIR, TGA and X-ray diffraction (XRD).

The next step involved the optimisation of CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> catalysed hydrolysis of olive oil for parameters temperature, pH and stirring rate. This part of the work attempts to establish and compare the best conditions to yield the highest percentage of the liberated free fatty acids. It was, in actual, to gauge which lipase was more efficient and activated to carry out the hydrolysis reaction.

Finally, the study assessed the stability of CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> by carrying out the reactions under different temperatures and leaching study. Subsequently, kinetic and thermodynamic parameters for the CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub>-catalysed hydrolysis of olive oil emulsion were assessed. The kinetic study assessed the values of Michaelis-Menten constant ( $K_m$ ) and maximum rate of reaction ( $V_{max}$ ) whereas the thermodynamic investigation were to estimate the values of thermal deactivation energy ( $E_d$ ), half-life ( $t_{1/2}$ ), standard enthalpy of denaturation ( $\Delta H_d^\circ$ ), standard entropy of denaturation ( $\Delta S_d^\circ$ ) and standard free energy of denaturation ( $\Delta G_d^\circ$ ).

## 1.5 Significance of Study

In this research work, the kinetic and thermodynamic parameters, as well as the ability of CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> to catalyse a hydrolytic reaction was established. The findings can further add to the body of knowledge with regards to the efficacy and versatility of CRL/G-AP-SiO<sub>2</sub>-Fe<sub>3</sub>O<sub>4</sub> to catalyse reactions from two different systems, *viz* the water-based or water-free (organic-solvent based) system.



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