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From UTP to AR-C118925, the Discovery of a Potent non Nucleotide Antagonist of the P2Y2 Receptor

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ABSTRACT

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Purinergic receptors are divided into P1 (adenosine) and P2 (ATP, ADP) receptors,¹ with P2 subdivided into P2X (trimeric ion channels) and P2Y (metabotropic G-coupled receptors).² The eight members of the P2Y family of receptors, so far characterized, have been further subdivided based on their primary signaling through specific coupled G-proteins. The first subgroup P2Y_{1,2,4,6,11} act through Gq and the second group, P2Y_{12,13,14}, through Gi.³ The P2Y₂ receptor (P2Y₂R) has been found in a variety of different tissues and cell types. Cell types include: epithelial cells, endothelial cells, smooth muscles cells and leukocytes. A study using P2Y₂ knockout mice revealed that the receptor mediates 85-

using P2Y₂ knockout mice revealed that the receptor mediates 85-95% of nucleotide-stimulated chloride secretion in the trachea. This suggests that P2Y₂R agonists have therapeutic potential as a treatment for cystic fibrosis (CF), as activation of this chloride secretion channel could compensate for the defective chloride secretion in the respiratory epithelium of CF patients.^{4, 5} Indeed, a P2Y₂ agonist, diquafosol **2**, mediating chloride secretion, has been approved in Japan for the topical treatment of dry eye disease.⁶

Agonism of the P2Y₂R can also lead to keratinocyte proliferation and neutrophil migration, indicating that P2Y₂ antagonists have therapeutic potential as a treatment for psoriasis.^{5, 7} It has also been reported that ATP released from tumor-cell activated platelets, acting through the P2Y₂R, induce opening of the endothelial barrier, leading to migration of tumor cells and hence cancer proliferation. P2Y₂R antagonists, therefore have therapeutic potential as anti-metastatic agents.⁸

The G protein-coupled P2Y₂ receptor, activated by ATP and UTP has been reported as a potential drug target for a wide range of important clinical conditions, such as tumor metastasis, kidney disorders, and in the treatment of inflammatory conditions. However, pharmacological studies on this receptor have been impeded by the limited reported availability of stable, potent and selective P2Y₂R antagonists. This article describes the design and synthesis of AR-C118925, a potent and selective non-nucleotide antagonist of the P2Y₂ receptor discovered using the endogenous P2Y₂R agonist UTP as the chemical starting point.

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Despite the appeal of the P2Y₂R as an important drug target, limited reports on P2Y₂R antagonists have appeared to date^{9–11} and indeed the only reported discovery of drug-like P2Y₂R antagonists¹² is from an industrial research group within AstraZeneca, disclosed within a series of chemical patents.^{13,14} One of the most potent and selective of these antagonists, AR-C118925, has been used as a tool for pharmacological studies on the P2Y₂ receptor.^{15,16} More recently, its selectivity profile against a range of P2 receptors has been published. AR-C118925 was at least 50-fold selective against P2Y₄, P2Y₆, P2Y₁₁, P2Y₁₂, P2Y₁₄, P2X₂, P2X₄ and P2X₇, whilst ~ 40-fold against P2X₁ and ~15-fold against P2X₃.¹⁷



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Herein we report the design and synthesis of $P2Y_2R$ antagonists, which led to the discovery of AR-C118925.

In the absence of a high throughput screening (HTS), the endogenous agonists, adenosine triphosphate (ATP) 3 and uridine triphosphate (UTP) 4 were considered as starting points for the research program. The greater selectivity of UTP for P2Y₂R over the other P2 purinoceptors led to its selection as the chemical starting point. Replacing the $\beta\gamma$ -oxygen of the triphosphate moiety of UTP with a dichloromethylene unit and substitution of the oxygen in the 4-position of the uracil with sulfur, gave enhanced metabolic stability, whilst maintaining agonist activity.¹⁸⁻²⁰ Interestingly, it was discovered that that P2Y₂R antagonism could be achieved through substitution of the 5-position of the uracil ring with lipophilic substituents (Table 1). The introduction of a benzhydryl group into this position (5) gave a compound with a pA2 for the P2Y2R of 6. P2Y2R activity was substantially increased by either symmetrically adding substituents to the benzhydryl group, compounds 6, 7, 8, and 9, or by linking the two phenyl rings of the benzhydryl to form a tricycle, compounds 10, 11, 12, and 13, with the highest activity achieved with the dibenzosuberenyl group, compound 13 (pA₂ 8.5).

Table 1: Exploration of the 5-position of the uracil ring



Compound	$P2Y_2 pA_2^{a, b}$	
5	6.0 ± 0.2	
6	7.9 ± 0.2	
7	7.8 ± 0.2	
8	7.1 ± 0.2	
9	7.7 ± 0.2	
10	7.2 ± 0.2	
11	7.3 ± 0.2	
12	8.0 ± 0.2	
13	8.5 ± 0.2	

a) The assay used a human P2Y₂R clone which was isolated from HL60 cells cDNA and then stably transfected into a Jurkat cell line. The cloned receptor mediates an increase in intracellular calcium in the cell line, which possesses no endogenous nucleotide receptor of its own. Inhibition of UTP mediated calcium responses were measured using 17 ktM fluo-3AM dye on a SPEX Fluomax using 508 nm excitation and 525 nm emission wavelengths at room temperature.^b n= \geq 2 replicates

The P2Y₂R antagonist program ultimately required the development of an oral drug-like compound.²¹ This necessarily meant moving away from the high molecular weight, highly charged substituted nucleotides. Removing the triphosphate group gave the nucleoside analogue **14** (Table 2). Whilst this compound was substantially less potent than its parent triphosphate **13**, it did retain some antagonist activity (pA₂ 4.7). Furthermore, it was discovered that the ribose ring could be replaced with the structurally less complex 3-methylenebenzoic acid to give **15**, with a slight gain in activity. The rationale for having the carboxylic

acid was to mimic any binding interactions the alpha phosphate of UTP might have with the $P2Y_2R$.

Table 2: Replacement of the ribose-triphosphate group

HN N	R=		но	HOHO
R			14	15
HO	HOHO		HO	но
16	17		18	19
HONN	HO		HO	но
20, 21	22		23	24
Compou	nd	Х		P2Y ₂ pA2 ^a
14		S		4.7 ± 0.2
15				
15		S		5.4 ± 0.2
16		S S		5.4 ± 0.2 5.7 ± 0.2
16 17		S S S		5.4 ± 0.2 5.7 ± 0.2 5.6 ± 0.2
10 16 17 18		S S S		$5.4 \pm 0.2 \\ 5.7 \pm 0.2 \\ 5.6 \pm 0.2 \\ 5.4 \pm 0.2$
16 17 18 19		S S S S		$5.4 \pm 0.2 \\ 5.7 \pm 0.2 \\ 5.6 \pm 0.2 \\ 5.4 \pm 0.2 \\ 5.9 \pm 0.2 \\$
16 17 18 19 20		S S S S S		$5.4 \pm 0.2 5.7 \pm 0.2 5.6 \pm 0.2 5.4 \pm 0.2 5.9 \pm 0.2 5.7 \pm 0.2 $
16 17 18 19 20 21		S S S S S O		5.4 ± 0.2 5.7 ± 0.2 5.6 ± 0.2 5.4 ± 0.2 5.9 ± 0.2 5.7 ± 0.2 4.4 ± 0.2
16 17 18 19 20 21 22		S S S S S O S		5.4 ± 0.2 5.7 ± 0.2 5.6 ± 0.2 5.4 ± 0.2 5.9 ± 0.2 5.7 ± 0.2 4.4 ± 0.2 5.8 ± 0.2
16 17 18 19 20 21 22 23		S S S S O S S		5.4 ± 0.2 5.7 ± 0.2 5.6 ± 0.2 5.4 ± 0.2 5.9 ± 0.2 5.7 ± 0.2 4.4 ± 0.2 5.8 ± 0.2 5.8 ± 0.2 5.2 ± 0.2

^a $n \ge 2$ replicates

Changing the linker between the carboxylic acid and phenyl ring from a direct bond to a one or two atom linker, potentially placing the carboxylic acid group further into the triphosphate binding region, gave a small increase in activity (**16** and **17**). In addition, the P2Y₂R appeared tolerant of a range of hetero aromatic replacements for the phenyl ring. For example, the furanyl analogue **19** had a further slight gain in activity (pA₂ 5.9). The continuing importance of the 4-thiouracil relative to uracil was demonstrated with thiazole replacement, where the uracil analogue was > 10-fold less potent (compare compound **20** with **21**).

Having been able to replace the ribose triphosphate with a structurally less complicated and synthetically less challenging group, $P2Y_2R$ antagonism could be increased through symmetrical substitution on the 5-suberenyl substituent **25**, **26** and **27** (Table 3). This combines the two features of change to the 5-benzhydryl group in the ribose triphosphate series, which independently gave an increase in receptor antagonism (Table 1).

Table 3: Further changes to the substituent on the 1-position of the uracil ring



In a similar manner to compounds discussed in Table 2, moving the acidic group further into the triphosphate binding region gave an increase in receptor antagonism. Although the dicarboxylic acids 28 and 29 had similar activity to the mono carboxylic acid **30**, indicating perhaps that only one of the carboxylic acids is involved in binding to the receptor. Finally, bioisosteric replacement of the carboxylic acid was employed giving the most potent compound, AR-C118925 (1).



Scheme 1: Representative synthesis of nucleotide analogues (5-13). (a) (i) TMEDA (2.2 eq), secBuLi (2.1 eq), THF, -78°C (ii) 5-dibenzosuberenone (1.5 eq), THF, 55% (b) (i) Et₃SiH (2 eq), BF₃.Et₂O (2 eq), CH₂Cl₂, 0°C (ii) ^tBuMe₂SiCl (1.5 eq), imidazole (1.5 eq), DMF, 70% (c) P₄S₁₀, pyridine, reflux, 68% (d) TBAF, THF, r.t., 85% (e) (i) 1.3 eq POCl₃, (1.3 eq) proton sponge, PO(OEt)₃ (ii) clodronic acid (3 eq), ⁿBu₃N (9 eq), PO(OEt)₃, 16%.

Scheme 1, which shows the synthesis of compound 13, illustrates the general procedure for synthesising the compounds in Table 1. Selective lithiation of the 5-position of the uracil ring (32) using the method of Miyasaka²² followed by addition of 5dibenzosuberenone gave the 5-dibenzosuberenol (33). The alcohol was then reduced using triethylsilane and borontrifluoride etherate. During this reaction, there was some loss of the silyl

protecting groups, giving a mixture of mono, di and tri silylated products. Reprotection using standard procedure gave the trisilylated product (33). Thiation of the 4-position of the uracil was achieved using P_4S_{10} in pyridine under reflux and removal of the silyl protecting groups using tetrabutylammonium fluoride gave the 5-substituted 4-thiouridine (14). Phosphorylation of the 5'-position of the ribose was achieved using phosphorus oxychloride followed by treatment with clodronic acid. The crude product was purified using DEAE Sephadex® ion exchange chromatography eluting with a 0 to 1 molar gradient of triethylammonium bicarbonate followed by elution through a Na+ form Dowex® ion exchange column to give the sodium salt of the product (13).



Scheme 2: Synthesis of AR-C118925. (a) (i) "BuLi, THF. -78°C (ii) 2,8dimethyldibenzosuberanone, THF (b) AcOH, reflux, 96% over two steps (c) (i) BSTFA (2.2 eq), 1,2-dichloroethane, reflux (ii) 5-bromomethyl-2furancarboxylic acid methyl ester, MeCN, reflux, 53% (d) Lawesson's reagent (1 eq.), 1,4-dioxane reflux, 65% (e) LiOH (5 eq.), 3:1 MeOH:H₂O, r.t., 95% (f) aminotetrazole, (1.5 eq) pybrop (1.5 eq), NEt₃(3 eq), DMF, 60%.

The synthesis of AR-C118925 is shown in scheme 2. Bromo lithium exchange of 5-bromo-2,4-di-tert-butoxypyrimidine²³ followed by treatment with 2,8-dimethyldibenzosuberanone gave the corresponding dibenzosuberanol product (34). Treatment with acetic acid under reflux gave the 5-substituted uracil (35) in very good yield over the two steps. Bis-silvlation of uracil (35) followed by treatment with 5-bromomethyl-2-furancarboxylic acid methyl ester, selectively alkylated in the 1-position of the uracil. Treatment with Lawesson's reagent then effected the conversion to the 4-thio uracil. Hydrolysis of the methyl ester gave the carboxylic acid (25) in good yield. Activation of the carboxylic acid bromo-tris(pyrrolidino)phosphonium with hexafluorophosphate and coupling with 1-aminotetrazole gave AR-C118925 (1).

In conclusion, a potent and selective non nucleotide P2Y2R antagonist, AR-C118925, was discovered using the endogenous agonist UTP as the chemical starting point for the program. AR-C118925 has reasonable physicochemical properties for a tool compound (MWt 537, LogD_{7.4} of 2.26) and was shown to have a reasonable rat in vivo pharmacokinetics [i.v. clearance (75 mL/min/Kg), Vss 4.34 L/Kg, T_{1/2} 2.12 h)] but was not bioavailable (F 0%). This approach of starting with the endogenous ligand has also been used in the $P2Y_{12}$ program ²⁴ and complimented the alternative, HTS - Active to Hit, Hit to Lead approach to initiating receptor agonist/antagonist projects at AstraZeneca.25-27 Utilization of AR-C118925 in the chemical literature supporting future potential clinical applications for P2Y2R antagonists has necessitated this present disclosure and we will be reporting further studies in due course.

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