

Advancement of benthic indicators and biomarker-based tools for biomonitoring and risk assessment in the Barents Sea region



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# Advancement of benthic indicators and biomarker-based tools for biomonitoring and risk assessment in the Barents Sea region

Ariadna S. Szczybelski Ciordia

#### **Thesis**

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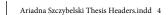


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But that in the simple appreciation of a world not our own to define, that poised arctic landscape, we might find some solace by discovering the ki-lin hidden within ourselves, like a shaft of light.

- Barry Lopez -







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# 1 General introduction







The Arctic Ocean is subject to a substantial climate change [1], causing Arctic marine ecosystems to exhibit particularly fast changes in the coming decades [2,3]. Sea ice loss and an extended open water period within Arctic marine areas are expected to contribute to increased human activities such as oil and gas (O&G) production, commercial fisheries and tourism development [4,5]. This raises a global concern about Arctic marine oil pollution [6,7] and its long-term impacts [8], based on experiences and environmental impacts from previous oil spills in the Arctic or elsewhere [9,10]. Awareness is also reinforced by the vast amounts of bunker fuel on board ships currently transiting International Maritime Organization Arctic waters being operated on heavy fuel oil [11], salient difficulties during oil spill response such as remoteness, seasonal darkness, stronger winds and currents, more fragmented sea ice and a lack of effective oil spill response methods for icy conditions [12,13]. Consequently, sustained and adequate long-term monitoring of the impacts of oil spills on Arctic ecosystems and development of plans for mitigation before and after Arctic oil spills should be implemented [14-16].

Current environmental risk assessment procedures contain uncertainties on the potential long-term effects of oil spills and operational discharges of produced water in the Arctic [17]. Environmental risk assessment combines exposure assessment and chemical-specific hazard data. Oil spills may pose an acute risk to the environment, with ecological consequences being highly dependent on the type of oil, size of spill, environmental characteristics of the habitat, selection and effectiveness of emergency response systems [16]. Risk assessment of O&G and shipping activities also needs to include long-term effects of continuous exposure to drilling waste and produced water [18].

Environmental impact assessment procedures should delineate oil pollution sources, transport mechanisms, routes and duration of exposure to Arctic species or habitats prior to the implementation of forecasting methods to predict O&G environmental impacts. Arctic characteristics affecting the environmental fate and effects of oil related chemicals may result in different exposure scenarios between Arctic and temperate species. Therefore, environmental monitoring tools should rely on the identification of factors that are critical for the exposure to oil related chemicals and development of sublethal effects in Arctic biota. Such factors may be used to define acceptable environmental assessment criteria and monitoring methods for the early detection of O&G environmental impacts in the Arctic.







For any species or group of species to be an effective indicator of O&G related chemicals, the abovementioned factors should meet the following criteria:

- 1. Its response is sensitive to changes in the chemical concentration;
- 2. Its response is specific and causally related to the chemical exposure;
- 3. Its response is adequate given the anticipated concentration of the chemical [19];
- 4. The species' uptake/elimination of the chemical can be quantified;
- 5. The toxic effects of the chemical can be measured/interpreted.

For the specific purpose of the thesis, biological indicators (bioindicators) are defined as biological species which may be affected by the exposure to a given O&G target chemical at different levels of biological organization and under realistic O&G exposure scenarios. In addition, bioindicators may present different functional traits and habitat choices so that their combined use may indicate different levels of stress for different ranges of environmental conditions and/or exposure to an O&G chemical group. Since adaptations of Arctic marine individuals to their environment underlie their sensitivity and vulnerability to O&G chemicals, the genetic diversity and phenotypic plasticity of species traits influencing the combined effect of natural stressors (e.g., ocean acidification) and O&G waste exposure should be reflected by regional impact monitoring. This ensures that bioindicators of potential pressures arising from O&G activities are identified, and that our understanding of how certain ecological attributes increase resilience of Arctic marine ecosystems is improved and used to develop oil spill restoration plans.

### Chemical characterization and environmental fate of offshore oil and gas waste in the Arctic with special emphasis on the Barents Sea region

Waste materials produced by offshore O&G activities that may be discharged into the ocean are water-based drilling muds (WBM) and cuttings, synthetic-based mud (SBM) cuttings, treated produced water (PW), treated sanitary and domestic waste, deck drainage, once-through fire water and non-contact cooling water (Figure 1). WBM are formulated mixtures of clays, natural and synthetic organic polymers, mineral weighting agents, and other additives, all dissolved or suspended in freshwater, saltwater, or in brine (Table 1). The composition of WBM may vary during drilling of a single well because different additives may be required to drill different well sections through varying geological formations [20]. WBM cuttings that are discharged to the ocean tend to accumulate at seafloor level downstream from the discharge source at distances of about 100 m to > 1 km. SBM cuttings do not







disperse on contact with sea water and settle rapidly when discharged, mainly exposing the (local) benthic ecosystem [21].

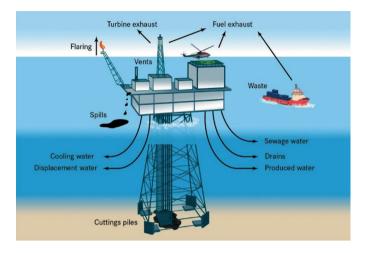


Figure 1. Offshore O&G waste discharges and air emissions depiction.

Sewage water includes industrial household and sanitary sewage in compliance with MARPOL 73/78 Annex V. <a href="Drains">Drains</a> include ballast water, storm/rain water and other liquid waste in compliance with MARPOL 73/78 Annex I, regulation 39. <a href="Produced water">Produced water</a> (PW), <a href="cutting-piles">cutting-piles</a> (WBM attached to WBM cuttings and SBM cuttings, if allowed) and <a href="displacement water">displacement water</a> discharges are in compliance with OSPAR Decision 2000/3. Emissions to air shall be in compliance with the provisions of MARPOL 73/78 Annex VI. Shipping discharges to sea shall be in compliance with the provisions of MARPOL 73/78 Annex I-V. Source: <a href="http://www.ospar.org/work-areas/oic/chemicals">http://www.ospar.org/work-areas/oic/chemicals</a>

**Table 1.** Functional categories of additives used in WBM to improve drilling performance and examples of chemical products as provided by the International Association of Oil and Gas Producers [20].

Functional Category of	Examples
Additives	
Weighting materials	barite, calcium carbonate, ilmenite or hematite
Thinners	lignite, lignosulfonates, polymers
Filtrate reducers	clay, lignite, polymers, starch
Lost circulation	inert soluble solids (e.g., calcium carbonate, ground nut shells, graphite,
	mica and cellulose fibres)
Shale control	soluble salts (e.g., KCl), amines, glycols
Bactericides	glutaraldehyde, triazine disinfectants
Pipe-freeing agents	water-based lubricants, enzymes, surfactants
Corrosion inhibitors	amines, phosphates
Viscosifiers	clay, organic polymers
Flocculants	inorganic salts, acrylamide polymers
pH control	inorganic acids and bases (caustic soda)
Lubricants	water-based lubricants, glycols and beads
Emulsifiers, surfactants	detergents, soaps, organic fatty acids
Defoamers	alcohols, silicones, aluminium stearate, alkyl phosphates
Calcium reducers	sodium carbonate, bicarbonate, polyphosphates
Temperature stability	acrylic or sulfonated polymers, lignite, lignosulfonate







PW is a complex mixture of dissolved and particulate organic and inorganic chemicals (Tables 2 and 3). Naturally occurring compounds that are dissolved or dispersed include inorganic salts, metals, radioisotopes, and a wide variety of organic chemicals, mainly polycyclic aromatic hydrocarbons (PAHs). Although these compounds may occur naturally, local concentrations after a spill may exceed natural concentrations by several orders of magnitude, creating a non-natural situation. Much of the petroleum hydrocarbons discharged to the ocean in properly treated PW are dissolved low-molecular-weight (LMW) PAHs and smaller amounts of saturated hydrocarbons. Because PW treatment is not always optimally effective, treated PW usually still contains some dispersed oil which includes high-molecular-weight (HMW), less soluble saturated and aromatic hydrocarbons. If PW is discharged to shallow waters, some metals and HMW aromatic and saturated hydrocarbons may accumulate in sediments near the PW source, with an increasing trend observed for compounds with a relatively high octanol-water partitioning coefficient (Kow). In well-mixed waters, high concentrations of saturated hydrocarbons and PAHs in sediments sometimes can be observed as far as hundreds of meters from the PW source [14,22,23].

**Table 2.** Concentration of naturally occurring organic chemicals in produced water worldwide as provided by Neff [24], and heavy metals and radioisotopes in produced water from the Norwegian Continental Shelf as provided by Neff [24] and the Norwegian Radiation Protection Authority [25].

Natural chemical	Treatment conc. (mg/L)
Total organic carbon	≤0.1 −>11,000
Total organic acids	≤0.001 <b>-</b> >10,000
Total saturated hydrocarbons	17 - 30
Total benzene, toluene, ethylbenzene and xylenes (BTEX)	0.068 - 578
Total polycyclic aromatic hydrocarbons (PAH)	0.04 - 3.0
Total steranes/triterpanes	0.14 - 0.175
Ketones	1.0 - 2.0
Total phenols (primarily C <sub>0</sub> -C <sub>5</sub> - phenols)	0.4 - 23
Metals (mainly Barium)	$107,000 - 228,000 (\mu g/L)$
Radioisotopes (mainly <sup>226</sup> Ra and <sup>228</sup> Ra)	n.d. — 567 ( <u>pCi/L</u> )

n.d.: not detected.







**Table 3.** Summary of additives in offshore produced water provided by Zheng et al. [26].

Function	Function Offshore chemical class			
	Amine imidazolines Amine salts	25 <b>–</b> 100		
Corrosion inhibitor (acid)	Quaternary ammonium salts			
	Nitrogen heterocyclics			
Corrosion inhibitor (oxygen)	Ammonium bisulfite	5 – 15		
Bactericide	Quaternary amine salt			
	Amine acetate	10 - 200		
	Gluteraldehyde			
Hydrate inhibitor	Methanol	n.a.		
	Ethylene glycol	n.a.		
Dehydration	Triethylene glycol	n.a.		
Scale inhibitor	Phosphate esters	3-10		
Scale IIIII01101	Phosphonates	3 – 10		
Emulsion breaker	Oxyalkylated resins			
	Polyglycol esters	10 - 200		
	Sulfonates			
Solid removal	Polyamine	< 3		
Solid Tellioval	Quaternary polyamine			

n.a.: not available.

Among all pollutants released during O&G drilling and production, alkylphenols (APs) and PAHs are seen as potential causes of long-term biological effects [23,27-29] although there is high variability in, for instance, PW composition between different fields and over drilling time (Table 2). Prolonged O&G waste discharges could have a cumulative effect on chronically exposed Arctic biota and their impacts could be particularly relevant for ecosystem components vulnerable to oil spills, such as benthic systems [30,31], since they mainly accumulate hydrocarbons from sediments to which persistent fractions of oil remain adsorbed. Despite their relevance, long-term AP and PAH effects on benthic populations or communities at concentrations found in PW effluents are basically unknown [32] and there are scarce means to evaluate their long-term impacts on the marine environment [18,33].

Besides, the effects of long-term exposure of Arctic benthos to oil or O&G waste components could be enhanced by increased primary productivity in seasonal ice regions where PAHs may be absorbed by phytoplankton, which in turn serves as major food source for benthic species [34-37]. Furthermore, exposure to oil or O&G waste may be aggravated by limited protein function (e.g., enzymatic detoxification) of benthos due to a relatively low tolerance of (sub-)Arctic temperatures (i.e., > 5 °C) in polar ectotherms [38,39]. Chronic impacts of O&G waste discharge on Arctic benthic systems could eventually result in a reduction of the productivity of the Arctic marine ecosystem since benthic systems are mainly responsible for







carbon and nutrient recycling [30] and they form an important food source for many top predators.

Because of the important role of benthic communities within the Arctic marine ecosystem and their susceptibility to O&G pollution, the scientific goals of the current thesis focused on the Barents Sea region which is characterized by the highest benthic species diversity among the different Arctic seas [30]. Major gaps on the physical and biological conditions of the Barents Sea seabed currently prevent the integration of baseline knowledge about the benthic ecosystem in O&G monitoring programmes and environmental impact assessment [40-42].

In the following sections, I will justify and underpin the selection of PAHs and model Arctic benthic invertebrate species from the Svalbard inshore region as indicators for chemical and biological pressures arising from O&G activities in the Barents Sea benthic system.

#### Polycyclic aromatic hydrocarbons as oil and gas target chemicals

PAHs are one of the main components of crude oil and can be classified into (1) LMW PAHs which are highly toxic to aquatic organisms and highly volatile typically remaining in the environment for hours to days following a spill, and (2) HMW PAHs which are less volatile and degradable, and are associated with chronic effects in biota due to their persistence in the environment [43]. PAHs have both natural and anthropogenic sources which overlap, particularly for combustion, and processes such as diagenesis (i.e., alteration of sediments into sedimentary rock) can modify sediment PAH distributions over time. In the Arctic Ocean, relatively high atmospheric concentrations and increasing deposition of PAHs with increasing latitude have been attributed to combustion of biomass and/or coal [44]. Therefore, the environmental fate and effects of anthropogenic PAHs must be evaluated against a site-specific background of natural PAHs and a broad range of hydrocarbon indicators needs to be considered in order to define sources, transport and sedimentary persistence of PAHs [45].

Once released into the ocean, PAHs are subject to partitioning processes including adsorption, where the hydrocarbon attaches to the surface of a solid or other interface, and absorption, where the chemical partitions into an (a)biotic compartment [46]. Understanding the distribution or partitioning of PAHs between the dissolved phase, aquatic particles and sediment is crucial for the determination of fate and bioavailability of PAHs to marine biota, as this will strongly affect the environmental fate of PAHs and mechanisms and magnitude of exposure of aquatic organisms to PAHs [47,48]. Similarly, PAH accumulation in organisms







will be affected by PAH desorption rates from particles in biota digestive tissues and metabolic rates that degrade or eliminate PAHs.

#### Environmental fate of PAHs: sorption to sediment and bioavailability

The fate and persistence of petroleum in marine systems are controlled by physical, chemical and biological properties and processes that subsequently modify petroleum impact. All these properties and processes are driving the weathering of oil, which however also depends on factors such as temperature, light conditions and water mixing by, for instance, wind. A key factor of oil weathering is based on the viscosity of the oil mixture, which largely affects its tendency to spread out on surfaces [49]. Since cold temperatures can increase the viscosity of oil, the resulting spreading of oil under Arctic conditions may be lower than expected and so are the evaporation of volatile components from oil [49,50] and from the water-accommodated fraction of some fuel oils [51]. This may delay oil weathering and biodegradation [52] and increase PAH sorption affinity [46], particularly near river runoff sources where alluviums of finely dispersed materials would sequestrate the suspended fractions of oil.

Desorption rate constants of PAHs from sediments contaminated with petroleum are inversely correlated with the logarithm of K<sub>ow</sub> of the PAH compound and directly correlated with its aqueous solubility [53,54]. PAHs associated with dissolved organic matter (colloidal; DOC) behave like dissolved PAHs, whereas PAHs associated with suspended organic matter will usually settle. They readily desorb from DOC, striving at an equilibrium between water, DOC, suspended and settled particles, defined by organic carbon normalised partition coefficients (K<sub>oc</sub>). However, when PAHs remain adsorbed to sediment for long periods, their desorption rate will decrease since prolonged contact time of PAHs with organic materials may lead to increased K<sub>oc</sub> values over time [55]. This 'ageing' phenomenon effectively decreases the bioavailability of adsorbed PAHs [48].

Assessment based on bioavailability is considered a useful tool for the risk evaluation of contaminated sites. PAH bioavailability can be estimated from PAH bioaccumulation which results from PAH uptake by an organism and subsequent chemical excretion after digestion and metabolism. Uptake of hydrocarbons in target tissues (bioavailability) occurs from sediments, suspended particulate matter, the water column, or through the diet, depending on the trophic level and ecological lifestyle of the organism [56]. In general, bioaccumulation is measured in order to (1) determine contaminant-specific bioavailability (for non-metabolised







contaminants), (2) identify possible causative agent(s) of toxicity, (3) relate body residues to predator-prey accumulation values (biomagnification factors) and (4) assess/predict effects of chronic, low-level exposures [57].

Under environmentally realistic PAH exposure scenarios, dealing with the fourth objective may seem compelling. Since PAHs exert toxic effects following metabolism, PAH body residues in organisms with low metabolic rate such as cold stenothermal bivalves will mainly reflect PAH bioavailability [58]. For organisms with a higher metabolic rate, measurement of parent PAH compounds is difficult due to low concentrations, hence PAH metabolic products need to be analysed to determine a relationship between bioaccumulation and toxicity. However, toxicity from bioaccumulation will also depend on the prior history of organism exposure and the differences in organism sensitivity. Thus, both bioaccumulation of PAHs (and their derivatives) and toxic effects should be simultaneously reported in benthic organisms exposed to PAHs in order to identify any likely tolerance mechanisms or sensitivity under low PAH exposure level [59].

#### Arctic benthic invertebrates as oil and gas bioindicators

In coastal environments, where oil is most likely to strand and accumulate, local biota will be subject to long-term impacts at source point or in surrounding areas after oil remobilization. In these coastal areas, or hydrodynamic systems, a causal link between an increase of PAHs in suspended solids and higher PAH bioavailability is suggested due to frequent sediment resuspension and ageing [60,61]. This process may be enhanced by the physical disturbance of sediment and increased riverine input of terrestrial PAH sources within inshore distance to land (< 200 m). Freshwater drainage with terrigenous particles may, in addition, create steep gradients of sedimentation and salinity [62] which generally underlie a shift from filter- to (sub-)surface-deposit feeders within the benthic community. Allowing for the coexistence of different functional groups along such small-scale gradients, a comprehensive understanding of the influence of species traits on PAH uptake and selection of bioindicators may be promoted.

Many polar benthic organisms typically display *K* strategies [63], such as reduced development and lower resource thresholds. In low water energy zones, absence or temporary drifting of larval stages and selective feeding mode would cause benthic invertebrates to be constantly exposed to local PAH sources and their possible PAH uptake routes could be more easily defined. Additionally, among true residents, ectotherms may show a high vulnerability







to increasing temperatures with greater fluctuations in their abundances on an inter-annual scale, depending on their recent recruitment [64]. Such ectotherms may be far more vulnerable to cumulative, but relatively minor impacts causing local extinctions such as the combined effect of rising temperatures, food shortage and oil spill incidents.

### Validation of long-term PAH exposure indicators in Arctic benthic invertebrates

In the Arctic, development of offshore O&G has undergone extensive technological and engineering advances that have enabled the industrial sector to operate safe and efficiently from a human and economic perspective [65]. One important aspect derived from such advances is the prioritization of early detection of environmental effects of anthropogenic pollution sources, notably effects derived from release of compounds included in the List of Substances/Preparations Used and Discharged Offshore that are Considered to Pose Little or No Risk to the Environment (PLONOR) [66]. The PLONOR list encompasses compounds for which an array of ecotoxicological properties has been evaluated according to the Harmonized Offshore Chemical Notification Format (HOCNF) [67] and set against a suite of validity criteria in order to ensure environmental risks are minimized during operational discharges. In the Lofoten-Barents Sea region, the development of O&G resources is permitted under a strict regulatory policy of zero discharge of oil-based mud (OBM) and SBM cuttings [68,69], whereas most of the additives used in WBM are either PLONOR substances or if non-PLONOR their use is restricted to emergency situations.

For the scientific community, there is one key question about current ecotoxicological data validation in the Arctic, which is whether the actual parameters used in HOCNF classification criteria need to be adjusted for this region. This question is motivated by the fact that low temperatures and regional physiological adaptations in Arctic species may give a difference in sensitivity to oil components of Arctic ecosystems when compared to non-Arctic systems. It has been shown that in general, differences in sensitivity to oil or PW components between Arctic and non-Arctic test species are small [70-72], the former ones including mainly boreal species with either a planktonic or nektonic adult stage. However, factors such as O&G waste discharge conditions, ecological seasonal variations or a delayed manifestation of toxicity in polar environments as recently suggested [73], could play an important role in establishing differences between the susceptibilities of both regional groups in the long term.

Parameters such as bioaccumulation and biotransformation are currently included in national or regional biomonitoring programmes of the marine environment [74], and they have been







compared across a wide range of temperate aquatic systems and benthic species, enabling us to make a read across between Arctic and temperate species. In this thesis, both parameters are used as part of an evaluating approach for the identification of long-term PAH exposure indicators in Arctic benthic species. In addition, the thesis evaluates the sensitivity of behavioural indicators in benthic model species to reflect on their potential use for the prediction of sublethal effects of oil from the perspective of Arctic benthic biota.

# PAH bioaccumulation from sediment: biota-sediment accumulation factors (BSAFs) for Arctic benthos

Bioconcentration of PAHs from water is directly proportional to their respective K<sub>ow</sub> value, whereas bioaccumulation from sediments and food involves an intermediate step in which PAHs desorb from the solid matrix and partition into the lipid tissues of the organism via the aqueous gut content. Bioaccumulation models generally assume that (1) both uptake and elimination routes are regulated by passive partitioning and will finally result in steady-state conditions and that (2) there is no biological metabolism once the PAH is absorbed. Bioaccumulation from ingestion has been neglected by traditional models such as the Equilibrium Partitioning Theory (EPT) [75]. The EPT method assumes that organic chemicals such as PAHs partition between the sediment organic carbon and biota lipid phase and that there is an equilibrium. However, the use of EPT methodology and the validity of its extrapolation to sediment-dwelling organisms have been questioned because of the large variability in tissue PAH concentrations across species and environmental conditions [56,76,77]. Hence, bioaccumulation processes are not solely explained by passive partitioning.

In both filter- and deposit-feeding organisms, PAHs are mainly accumulated via food intake as these chemicals are generally adsorbed on dissolved and particulate organic matter (OM). Because of the high affinity of PAHs for (sediment) condensed carbon (e.g., black carbon; BC) [78], there is an inverse relationship between the concentration of BC in sediments and the bioavailability of PAHs in sediments to marine organisms [79]. Additionally, the relative contribution of water, food and sediment PAH sources to the total body burden greatly varies from one species to another [56,76,77,80]. Within the EPT context, the estimation of a PAH compound concentration in a certain species may be obtained by multiplying its sediment concentration by an empirically derived biota-sediment accumulation factor (BSAF) for that same PAH, species and sediment quality/composition. In practical terms, the observed







variability in PAH concentrations requires a broad interpretation since the bioavailability of PAHs will be modified by the species behaviour and the composition of their sorbent.

The BSAF has been used as a metric to evaluate the bioaccumulation of contaminants from sediment and has been compared across a wide range of aquatic ecosystems [56,80-83]. The use of the BSAF concept assumes that the concentration of chemicals in organisms is a linear, non-threshold function of the concentration in sediment. However, assumptions of steady-state conditions and absence of metabolism may not always be suitable, particularly if food sources are predominantly waterborne or if PAHs are effectively metabolised after summer periods of intense foraging. In the Arctic, BSAFs for organic pollutants have been barely reported for benthic biota under field or laboratory conditions [84,85], and have never been reported for PAHs in benthic biota.

# A PAH biotransformation approach: biomarkers of PAH exposure and pyrene metabolites in Arctic benthos

Biomarkers of chemical exposure may be useful as estimators of chemical stress to monitor trends in ecological impacts of O&G activities [86,87]. For instance, aromatic xenobiotics may enhance the production of intracellular reactive oxygen species (ROS) through induction of the cytochrome P450 system (CYP). CYP is a multigene family of haem-containing enzymes, which catalyse a variety of oxidative reactions, including hydroxylation, epoxidation, dealkylation, deamination, sulfoxidation, and desulfuration (i.e., Phase I). During these reactions, the lipophilicity of the xenobiotic is lowered by the addition of a polar functional moiety, which often makes the metabolite more hydrophilic and, in most cases, ready for further conjugation and/or excretion. Metabolites produced in Phase I reactions can be eliminated or covalently conjugated to various endogenous compounds (e.g., reduced glutathione, glucuronic acid, sulfate) to further decrease their lipophilic properties. Such reactions, commonly known as Phase II, are catalysed either by glutathione transferases (GSTs), uridine-diphosphate-glucuronosyl-transferases (UDPGTs), or sulfotransferases (STs), usually yielding water-soluble products that can be readily excreted. Suites of biomarkers have been successfully applied at low levels of biological organization in combination with chemical body residues in fish and invertebrates [74,88,89]. However, the use of biomarkers also has been criticized as a result of the lack of a clear ecological relevance [86] and of difficulty in data interpretation due to the influence of confounding factors [90] or lack of baseline values [91].







The major trait that could enhance oxidative stress sensitivity in polar species is the elevated unsaturation level in membrane lipids [92]. A higher percentage of the polyunsatured fatty acid (PUFA) represents an important mechanism by which polar ectotherms maintain biological membrane structural and functional properties against low temperature. Since PUFAs are highly susceptible to ROS attack, and high unsaturation levels enhance the velocity and propagation of lipid radical chain reactions, the difference in fatty acid composition increases the vulnerability of polar organisms to oxidative stress. PAH metabolites may, in addition, exert pro-oxidant effects through the redox cycle, a well-known source of chemically-mediated ROS generation [93], while metabolites containing an epoxide may be mutagenic and/or carcinogenic or even inhibit some CYP components [94]. From a practical point of view, the analysis of PAH metabolites in polar benthos may facilitate the definition of acceptable oxidative stress thresholds since metabolic responses may not rely on a large variety of homeostatic controls and feedback mechanisms as in the case of changes in enzymatic induction (i.e., gene expression and protein production), which ensures their meaningful interpretation [95].

# Behavioural response approach: avoidance of Distillate Marine grade A (DMA) oil-spiked sediment

Active spatial avoidance ('escape') of contaminated sediment relates to the ability of an organism to detect toxicants via olfaction or taste and to move to a lesser contaminated area [96]. Such behaviour can be up to 1000 times more sensitive than conventional lethal endpoints (e.g., LC50), and it may also potentially reflect changes at a population level [97]. Among benthic invertebrates, amphipods have been extensively used in avoidance assays [98-101] since they are equipped with a sensory system and they form an important food source for benthic fishes and other organisms. To date only few studies have addressed avoidance behaviour of oil-contaminated sediment by benthic amphipods, all of which were restricted to temperate marine or estuarine species such as *Corophium volutator* or *Melita plumulosa*.

In the Arctic, exposure to the water-soluble fraction (WSF) of oil revealed little mortality in sea-ice amphipods (i.e., *Gammarus wilkitzkii*), although biomarker results indicated sublethal effects after exposure to WSF [102]. However, oil toxicity in pelagic amphipods may be quickly dampened due to considerable evaporation and dissolution of oil in the water column. This may not be the case in the water-sediment interface where microbial degradation of oil under Arctic conditions [103] may result in persistent and localised sediment oil sources.







These may trigger an escape response in benthic amphipods, while their response to lethal concentrations may be slower than avoidance as previously observed in amphipods exposed to the water-accommodated fraction of oil [104,105].

Since heavy fuel oils are being gradually replaced with marine distillate fuel oils (e.g., DMA) where increased marine traffic is expected [106] and the organic fraction of distillate fuels might be able to persist sufficiently long to adsorb to sediment and cause toxicity on a similar scale than heavier fuels [51,107], escape by Arctic benthic amphipods from oil-spiked sediment could become a sensitive tool for the early detection of oil sublethal effects in benthic communities [108,109]. So far, studies have been performed on the toxicity of diesel in sediment to polar benthic species such as the ophiuroid *Ophiura crassa* [110] and the amphipod *Gammarus setosus* [111], but their ability to avoid oil-contaminated sediment has never been addressed under field or laboratory conditions. Integrating escape responses in oil risk assessment for the Arctic may not only give an indication on benthic amphipods' spatial distribution, which is inversely related to sediment toxicity, but also on oil related adverse effects from chronic exposure.

#### Aim and scope of the thesis

As described in the foregoing sections there are four main knowledge gaps presently hampering our ability to monitor long-term impacts of O&G activities on Arctic benthic ecosystems:

- BSAFs for organic pollutants have been barely reported for Arctic benthic biota, neither evaluated as possible indicators of sediment PAH bioavailability.
- Biomarkers of exposure to organic pollution and their responsiveness to sediment pollution sources have not been evaluated in Arctic benthos chronically exposed to PAHs.
- Biotransformation metabolites of PAHs and their persistence along sediment exposure have not been evaluated in Arctic benthos chronically exposed to PAHs.
- Avoidance of DMA-spiked sediment and its potential as an oil impact assessment tool
  have never been studied from the perspective of Arctic benthic biota.

#### Aim of thesis

The aim of this thesis is to develop candidate methods for the early identification of Arctic bioindicators of O&G and shipping related chemical stress in Arctic coastal benthic systems. The bioindicator potential of some Arctic benthic invertebrate species is evaluated by







specifically measuring the species traits influence on their PAH uptake and PAH biotransformation induction during PAH exposure. Species with either sessile or with reduced mobility, different feeding mode, food selectivity and low dispersal capacity are used to define a sensitivity range to PAHs and other stressors within our chosen taxa. From that perspective, two of the most representative Barents Sea and Chukchi Sea bivalve species, with differing feeding and reproduction traits, the filter feeder *Astarte borealis* (Schumacher, 1817) and the filter/deposit feeder *Macoma calcarea* (Gmelin, 1791) were selected as target species in combination with a predator polychaete, *Nephtys ciliata* (O.F. Müller, 1776) [112-115].

#### Research questions:

**Q1:** Can PAH body residues or BSAFs be used as bioavailability indicators in Arctic coastal benthic systems and which target species is most suitable to detect trends?

**Q2:** Can species traits explain differences in PAH bioaccumulation between Arctic benthic invertebrates?

**Q3:** Does bioaccumulation of PAHs differ between Arctic and temperate benthic invertebrate species with different traits, and can modelling assist in reading across between species?

**Q4:** What is the feasibility of using biomarkers of exposure to monitor PAH pollution in the Arctic? Is the identification of biotransformation metabolites a better alternative?

**Q5:** Can behavioural tests like avoidance behaviour become a suitable monitoring tool for oil impact assessment in the Arctic?

**Q6:** What are prospects and advantages of using Arctic benthic bioindicator species for monitoring long-term impacts of O&G and other maritime activities in the Arctic?

The above research questions are answered through a combination of field and laboratory research, and modelling.

To address Q1, PAHs, PCBs and HCB body residues were measured in the field in *A. borealis*, *M. calcarea* and *N. ciliata*. Differences in body residues among two locations in Kongsfjorden ecosystem (Svalbard, Norway), chemicals and species were assessed and BSAF values were obtained. Whether the BSAF metric is useful as an indicator of PAH availability in Arctic benthic invertebrates under field conditions is investigated in **Chapter 2**.







To address Q2 and Q3, PAH body residues were measured and BSAF values were obtained for the aforementioned Arctic invertebrates and two Atlantic counterpart species (i.e., *Limecola balthica* [Linnaeus, 1758] and *Alitta virens* [Sars, 1835]) under laboratory conditions. PAH bioaccumulation and bioavailability were compared between both regional groups and bioaccumulation was modelled in order to make BSAF reading across between groups possible. Results from laboratory experiments and modelling exercises are discussed in Chapter 3 and set against field bioaccumulation conclusions in Chapter 2.

To address Q4, the responsiveness of some biochemical biomarkers of exposure, namely acyl-CoA oxidase (AOX) and glutathione S-transferase (GST), or acetylcholinesterase (AChE) inhibition in the Arctic bivalve *A. borealis* exposed to a PAH-contaminated sediment in laboratory conditions were studied. Additionally, two biotransformation metabolites of the model PAH pyrene (e.g., 1-hydroxypyrene [OHPyr] and pyrene-1-glucuronide [GluPyr]) were measured in Arctic (*M. calcarea*, *N. ciliata*) and temperate (*L. balthica*, *A. virens*) bivalves and polychaetes exposed to the same sediment and differing (Arctic *vs* temperate) laboratory conditions. Both groups' PAH biotransformation capacity and estimated chronic toxicity derived from PAH metabolites bioaccumulation are discussed in Chapter 4.

To address Q5, short-term avoidance by the temperate freshwater amphipod *Gammarus pulex* (Linnaeus, 1758) and temperate marine benthic amphipod *Gammarus locusta* (Linnaeus, 1758) to DMA-spiked sediment was studied and set against results from the available literature in **Chapter 5**. How and to what extent avoidance may affect the long-term performance of the amphipods and how the response of the Arctic benthic amphipods may differ from their temperate counterparts are discussed in **Chapter 6**, based on available literature on oil acute toxicity in Arctic sea-ice and benthic amphipods.

Lastly, motivated by **Q6**, a synthesis is provided in **Chapter 6** which discusses how physiological and ecological traits may render Arctic coastal benthic species particularly vulnerable to chronic exposure to O&G related chemicals and how we can identify those at an early stage during O&G production as part of existing biomonitoring programmes. Incorporating sublethal effects and long-term impacts of O&G waste on Arctic benthic systems into biomonitoring and risk assessment procedures will help to determine the odds of oil spill and O&G impacts on Arctic marine ecosystems and thus, improve environmental assessment and restoration planning.







2 Bioaccumulation of polycyclic aromatic hydrocarbons, polychlorinated biphenyls and hexachlorobenzene by three Arctic benthic species from Kongsfjorden (Svalbard, Norway)

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#### Based on:

Szczybelski, A.S., M.J. van den Heuvel-Greve, T. Kampen, C. Wang, N.W. van den Brink & A.A. Koelmans (2016). Bioaccumulation of polycyclic aromatic hydrocarbons, polychlorinated biphenyls and hexachlorobenzene by three Arctic benthic species from Kongsfjorden (Svalbard, Norway). *Marine Pollution Bulletin* 112(1-2), 65-74.







#### **Abstract**

The predicted expansion of oil and gas (O&G) activities in the Arctic urges for a better understanding of impacts of these activities in this region. Here we investigated the influence of location, feeding strategy and animal size on the bioaccumulation of polycyclic aromatic hydrocarbons (PAHs), polychlorinated biphenyls (PCBs) and hexachlorobenzene (HCB) by three Arctic benthic species in Kongsfjorden (Svalbard, Norway). No toxicity was expected based on biota PAH critical body residues. Biota PCB levels were mainly below limit of detection, whereas samples were moderately polluted by HCB. PAH concentrations in biota and biota-sediment accumulation factors (BSAFs) were generally higher in Blomstrandhalvøya than in Ny-Ålesund, which was explained by a higher abundance of black carbon in Ny-Ålesund harbour. BSAFs differed significantly among species and stations. We conclude that contaminant body residues are a less variable and more straightforward monitoring parameter than sediment concentrations or BSAFs in Arctic benthos.







#### 1. Introduction

The Arctic region is undergoing an unprecedented change, with global rising temperatures causing an annual summer retreat of sea ice and changes to for instance, seasonal weather patterns and even ecosystems [116,117]. The retreat of sea ice will allow expansion of oil and gas (O&G) activities in the next decades, posing possible impacts on the Arctic ecosystem [14]. Besides O&G activities, changes in climatic parameters as such may affect contaminant transport to and cycling in the Arctic. Furthermore, primary productivity and food web energetics may be affected and thus the trophic transfer of contaminants. Besides allowing possible expansion of O&G activities, retreat of ice also opens alternative commercial shipping routes, with associated anthropogenic activities and risk of maritime accidents.

It has been claimed that baseline information on effects of O&G activities on the Arctic ecosystem or ecosystem components is still inadequate or unavailable [72]. Further research on linking effects in organisms to exposure to O&G related contaminants is therefore essential. In this context, identifying chemical (e.g., petroleum marker compounds like petrogenic PAHs), biological indicators for cumulative effects of O&G activities that can be applied across the Arctic is highly relevant. Petroleum hydrocarbons are generally considered to be one of the main pollutants related to O&G activities. Although anthropogenic inputs of PAHs are only a small proportion of total hydrocarbon burden in the Arctic environment [118], they can create substantial local pollution [119]. Several studies suggest that atmospheric transport of polycyclic aromatic hydrocarbons (PAHs) has only a minor influence on sedimentary PAH distributions in the Arctic Ocean [45,120]. However, petrogenic and pyrogenic PAHs have been identified as an emerging concern in the Arctic [121] and there is little information on PAH bioaccumulation in Arctic benthic species [122]. Since the hydrophobicity of PAHs may promote their adsorption to settling organic matter (OM), marine sediments and benthic systems may act first as a sink and ultimately as a secondary source of PAHs.

Besides PAHs, legacy persistent organic pollutants (POPs) like polychlorinated biphenyls (PCBs), are still ubiquitous in Arctic regions. Local sources are mainly found in the form of secondary sources after, for instance, breakdown and decommission of equipment and demolishing of buildings, although long range atmospheric transport also greatly contributes to the general supply. Contaminants deposited in polar regions typically become mobile during the summer months as glaciers, sea ice and snow melting may introduce POPs into the





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marine system [123,124]. Input of these contaminants may directly impact the pelagic system when taken up by algae and zooplankton [125,126], and indirectly the benthic communities due to a strong pelagic-benthic coupling in the Arctic [127]. PCBs are persistent and hydrophobic organochlorine contaminants with a wide range of biological effects in chronic exposed biota. This makes them a suitable chemical group for environmental monitoring in general.

Hexachlorobenzene (HCB) is another chemical that is known for its persistence and ability to being transported to the cold polar regions [128]. Potential sources of HCB to the marine ecosystem may comprise by-products and waste such as solid and liquid residues, solid waste, oils and wastewater.

Coal, crude oil and atmospheric dust have been identified as possible PAH sources in sediment samples of Svalbard [129]. PAH patterns in sediment samples showed predominance of petrogenic PAHs (20-60%  $\Sigma$ PAH) and Svalbard coal was identified in all sediment sampling stations [129]. Kongsfjorden is a relatively pristine fjord, as direct anthropogenic impact on the marine system at local scales is generally much lower in Svalbard than along the Norwegian coast and in European waters [130]. Strict environmental policy has resulted in a limited contamination of the Kongsfjorden environment [131], although accidental spills or chronic impacts from local anthropogenic activities still may occur.

Arctic benthic species can be applied as relevant and sensitive bioindicators for the impact assessment of O&G activities in the Arctic [122,132]. Markert et al. [133] defined a bioindicator as an "organism (or a part of an organism or a community of organisms) that contains information on the quality of the environment (or a part of the environment)". Arctic benthic systems are characterised by a high trophic diversity, relatively long life-span, and sedentary lifestyles of the species [134]. This makes them well suited for studying spatial but also temporal variability of O&G related pollution. Omnivore strategies are common among many benthic species from Arctic shallow benthic communities such as scavenging amphipods [135] making them resilient to changes in seasonal conditions which results in little change in food web structure through the year [136]. Benthic invertebrates are an important food source for higher trophic levels in the Arctic such as benthic fishes, bottom-feeding seals (e.g., bearded seals, walruses) and birds (such as the eider duck) [137,138]. The accumulation and effects of contaminants in benthic species depends on both biological traits







such as diet, habitat preference and longevity [56], and sediment characteristics such as grain size and organic carbon content [80]. This results in differences in bioaccumulation potential among benthic invertebrate species, which may be further enhanced by differences in their biotransformation capacity of contaminants [139]. Some information is available on contaminant levels in Arctic benthic species [140,141], however effects of contaminants in Arctic invertebrates are still poorly understood [142].

The biota-sediment accumulation factor (BSAF) has been used as a metric to assess the bioaccumulation of contaminants in sediment and has been compared across a wide range of aquatic ecosystems [80-82]. Only a few examples of BSAFs for Arctic species are available. BSAFs are reported for PCBs in (sub-)Arctic shorthorn sculpins [84] and for volatile siloxanes in Svalbard Atlantic cods and shorthorn sculpins [85]. We are aware of only one study reporting BSAFs for Arctic benthic invertebrates. PCB BSAFs have been published for *Astarte* sp. and *Nephtys incisa* from an Atlantic temperate region [143].

Aim of the present study was to provide a more systematic evaluation of bioaccumulation of PAHs, PCBs and HCB by three Arctic benthic species. This includes assessing differences between species, size classes within species, contaminant groups and sampling stations. In this study we explore for the first time to what extent the BSAF metric is useful as an indicator for PAH, PCB and HCB accumulation in Arctic benthic invertebrate species. The results will be interpreted in light of the usefulness of such metric for assessment of potential impact of future O&G activities in the Arctic.

Biota was collected in Kongsfjorden (Svalbard, Norway). Three benthic species (*Astarte borealis*, *Macoma calcarea* and *Nephtys ciliata*) were selected based on their feeding habits, sessility and relative abundance or Valuable Ecosystem Component (VEC) condition (i.e., species that are specifically abundant at ecologically relevant habitats) [144].

Results of this study show the bioaccumulation potential of PAHs, PCBs and HCB in Arctic benthic species in the field and are used to select the most relevant benthic species and tools for monitoring organic contaminants in Arctic sediment.







#### 2. Materials and Methods

#### 2.1. Study sites

Sediment was collected in an (sub-)Arctic fjord (Kongsfjorden Bay, Svalbard) and an offshore region (Barents Sea) (Figures 1A-1B). The Svalbard inshore and Barents Sea study areas were located between 71° and 79° latitude (Table S3) along which the waters are influenced by the West Spitsbergen Current (WSC). Barents Sea sampling stations depth ranged 110 to 320 m. For a general description of PAH levels in the W Barents Sea the reader is referred to Boitsov et al. [118].

Kongsfjorden ecosystem is a high-latitude (sub-)Arctic fjord (79°N, 12°E), but the waters are influenced by both the Atlantic water masses of the WSC as well as the Arctic-type coastal waters, and additionally glacial input of melt water [145]. Depths in the outer basin average 200 to 300 m, whereas the inner basin is considerably shallower (average depth 50 - 60 m). Muds dominate the subtidal sediments throughout the fjord. Kongsfjorden inner and outer basins differ with respect to differences in their deposition rates, which are much higher in the inner basin, and differences in bioturbatory activity by infaunal organisms distribution [131].

Kongsfjorden shallow benthic macrofauna distribution and abundance are influenced by sediment characteristics such as grain size, water currents, as well as by different carbon sources such as fresh settling OM, reworked settling OM, microphytobenthos and terrestrial OM [127,136,141,146]. Two faunal communities can be distinguished in the Kongsfjorden ecosystem according to the frequency of species occurrence, and to several indices of community fidelity of each species in each community [147]. The Glacial Bay community (inner basin) is dominated by small mobile bivalves (e.g., *Yoldiella solidula*) while the Outer Basin community is dominated by larger, often less mobile, mostly tubiculous polychaetes (e.g., *Pectinaria hyperborea*).

#### 2.2. Collection of field samples

Sampling of sediment and biota in Kongsfjorden was performed at a locally impacted harbour station (Ny-Ålesund) [148,149] and two reference stations (Blomstrandhalvøya and Kapp Guissez) (Figure 1C). Sampling took place outside the protected areas of Kongsfjorden on board of research vessel MS Teisten (Kings Bay AS, Ny-Ålesund) between 15 and 19 July 2013 and within the Outer Basin community distribution area as defined by Włodarska-







Kowalczuk and Pearson [147]. Additionally, sediment samples were collected at five stations in the West Barents Sea on board of the R/V Lance (Norwegian Polar Institute, Tromsø) between 3 and 10 October 2013 (Table S3). Sediment samples consisted of soft bottom substrates and were collected using a Van Veen grab (0.1 m<sup>2</sup>). Biota samples were collected at Ny-Ålesund and Blomstrandhalvøya using either a Van Veen grab, or a small dredge. Three Arctic marine benthic invertebrate species with different feeding behaviours were collected: A. borealis (Schumacher, 1817) (mollusc; suspension feeder), M. calcarea (Gmelin, 1791) (mollusc; deposit and suspension feeder), and *N. ciliata* (O.F. Müller, 1776) (polychaete; omnivore). Sediment samples were preserved in 250 mL-glass jars at -20 °C until analysis. The organisms were allowed to depurate for 24 h in clean sea water from the fjord before dissection. Animals were weighed, measured, dissected and pooled by size class (Table S2). Size classes were defined such that each class had sufficient numbers of individuals to allow statistical testing. The soft tissue of bivalves was removed from the shell and samples were pooled in groups of 5 to 12 individuals before freezing at -20 °C. Sediment and biota samples were all shipped frozen in dry-ice to Wageningen University (Wageningen, the Netherlands) for chemical analysis.

#### 2.3. Chemical analysis

Chemical analysis was described before [150]. The following PAHs and POPs were analysed: Phenanthrene (PHE), Anthracene (ANT), Fluoranthene (FLT), Pyrene (PYR), Benz(a)anthracene (BaA), Chrysene (CHR), Benzo(e)pyrene (BeP), Benzo(b)fluoranthene (BbF), Benzo(k)fluoranthene (BkF), Benzo(a)pyrene (BaP), Indeno(1,2,3-cd)pyrene (Ind123P), Dibenz(a,h)anthracene (dBahA), Benzo(ghi)perylene (BghiP), hexachlorobenzene (HCB), PCB-18, -20, -28, -29, -31, -44, -52, -101, -105, -118, -138, -153, -170, -180, -194, -204 and -209.

For details on extraction, clean-up and instrumental analysis see the Supporting Information (SI). In short: sediment and biota samples were Soxhlet-extracted with hexane/acetone. Extracts were cleaned over Al<sub>2</sub>O<sub>3</sub> (PAH), or Al<sub>2</sub>O<sub>3</sub>/silica (HCB and PCBs) and analysed using HPLC-UV and GC-ECD, respectively. Recoveries were 53 to 77% (PAHs in biota), 66 to 79% (HCB and PCBs in biota), 57 to 67% (PAHs in sediment) and 79 to 103% (HCB and PCBs in sediment). Besides recoveries, numerous blanks were used. All data were corrected for the resulting values.







Subsamples of each sediment sample were analysed for weight percentages (wt. %) of OM content, measured as loss on ignition (550 °C, 3 h). Biota lipids were extracted with chloroform:methanol:water and quantified gravimetrically.

#### 2.4. Data analyses

Lipid-normalised biota concentrations were calculated. BSAFs were calculated as  $(C_{org}/f_{lip})/(C_{sed}/f_{SOM})$  with  $C_{org}$  being the chemical concentration in the organism in wet weight (wet wt.;  $\mu g/kg$ ),  $C_{sed}$  the chemical concentration in sediment in dry weight (dry wt.;  $\mu g/kg$ ),  $f_{lip}$  the fraction of lipids in the organism based on wet wt., and  $f_{SOM}$  the fraction of sediment organic matter (SOM) based on dry wt. BSAFs of a specific compound were calculated provided that there were at least three detects (i.e., >LOD) for the same compound in sediment samples. Only those compounds for which  $\geq 90\%$  BSAF values were above zero in both stations were included in the statistical analysis. Data were checked for normality with Q-Q plots and Shapiro-Wilk test and equality of variances with Levene's test. Lipid-normalised biota concentrations and BSAFs were tested for location and size effect with the Mann-Whitney U test and Kruskal-Wallis test, respectively for each PAH and PCB compound as well as for HCB and the sum of PAHs ( $\Sigma_{13}$ PAH) and 7 PCBs ( $\Sigma_{7}$ PCB). Significance level of an overall statistical comparison was set at  $p \leq 0.05$ , while for pairwise comparisons a Bonferroni correction was applied. All statistical calculations were performed using SPSS version 22.

PAH diagnostic ratios were applied to assess emission sources of PAHs in sediment. Sources can be pyrogenic (originated from combustion), petrogenic (originated from oil) or a combination of these. It has been shown that PAH isomers with masses 202 (e.g., FLT, PYR) and 276 (e.g., BghiP, Ind123P) have the greatest range in stability of PAHs, thus they are good indicators of petroleum *vs* combustion sources [120,151]. PAH ratios FLT/(FLT+PYR) and Ind123P/(Ind123P+BghiP) were calculated. A FLT/(FLT+PYR) ratio of < 0.4, 0.4 – 0.5 and > 0.5 points out to petrogenic sources, fuel combustion or grass/wood/coal combustion, respectively. A Ind123P/(Ind123P+BghiP) ratio of < 0.5 or > 0.5 points out to fuel combustion or grass/wood/coal combustion, respectively. However, this relationship is not universal and interpretation of this measure should only be used indicatively.

Sediment (based on dry wt.) and biota (based on lipid wt.) PAH and PCB concentrations were evaluated based on Environmental Quality Standards (EQSs) established by the Norwegian







Environment Agency [152], Environmental Assessment Criteria (EACs) established by the OSPAR Commission [153] and Critical Body Residues (CBRs) established by the USEPA [154].

Concentrations below the EACs are considered to present no significant risk to the environment, and to that extent may be considered as being related to the EQSs applied to concentrations of contaminants in water [155]. CBRs established by the USEPA are based on the target lipid model suggested by Di Toro et al. [156].

#### 3. Results and Discussion

#### 3.1. Concentrations of PAHs, PCBs and HCB in sediment

*PAHs in sediment.* For Kongsfjorden samples, concentrations of  $\Sigma_{13}$ PAH varied between 12 and 2315 μg/kg dry wt. (geometric mean 139 μg/kg dry wt.), whereas for Barents Sea samples, this range was between 39 and 4270 μg/kg dry wt. (geometric mean 823 μg/kg dry wt.) (Table S4). Ny-Ålesund sediments were contaminated quite uniformly with  $\Sigma_{13}$ PAH concentrations ranging from 1815 to 2315 μg/kg dry wt. (geometric mean 2011 μg/kg dry wt.). Towards the northern fjord shelf (Blomstrandhalvøya) and the inlet from Kongsfjorden to Krossfjorden (Kapp Guissez), PAH levels decreased. In the Barents Sea, significant differences were found for PHE and BeP among distant stations (e.g., SE Edgeøya vs Snøhvit, Figures 1A-1B). This may indicate differences in deposition, the presence of local sources such as natural seepages of oil or other organic fossils, or differences in SOM content. No significant differences were found among stations relatively close to each other (e.g., SE Edgeøya S2-S4) or among distant stations located within the NW Barents Sea study area (e.g., E Kong Karls Land vs SE Edgeøya).

Sediment PAH diagnostic ratios. Based on PAH ratios, sediment PAHs seemed to originate from combustion of grass, wood, coal or petroleum (Barents Sea, Ny-Ålesund), from combustion of grass, wood and coal (Kapp Guissez), and from combustion of grass, wood and coal or from petroleum (Blomstrandhalvøya) (Table 1). In Ny-Ålesund sediments, uncombusted coal also is relevant. After all, PAHs from coal are ubiquitous in the Svalbard environment where extensive coal mining was carried out by Kings Bay in Ny-Ålesund until 1962 [157]. In a study of surface sediment collected between the Barents and Kara Sea shelves, the highest pyrogenic PAH levels (sum of 4- to 6-ring hydrocarbons, perylene







excluded) were found in sediments from Kola, the Pechenga Bays and Svalbard inshore [129]. Highest pyrogenic PAH levels in Barents Sea and Kongsfjorden were reached in sediments from station 2 (SE Edgeøya) and 8 (Kapp Guissez) (Table 1).

**Table 1.** Sediment parent PAH diagnostic ratios.

			Barents Sea			Kongsfjorden			
			(October 2013)			(July 2013)			
PAH ratio		S1	S2	S3	S4	S5	S6	S7	S8
FLT/(FLT+PYR)	Min	0.43	0.53	0.45	0.44	0.00	0.63	0.31	0.48
		(B)	(C)	(B)	(B)	(A)	(C)	(A)	(B)
	Max	0.47	0.57	0.48	0.49	0.52	0.65	0.35	0.55
		(B)	(C)	(B)	(B)	(C)	(C)	(A)	(C)
Ind123P/(Ind123P+BghiP)	Min	0.80	0.82	0.77	0.79	0.81	0.34	0.21	0.83
		(C)	(C)	(C)	(C)	(C)	(B)	(B)	(C)
	Max	0.83	0.85	0.79	0.80	0.87	0.43	0.79	0.86
		(C)	(C)	(C)	(C)	(C)	(B)	(C)	(C)

<sup>(</sup>A) Petrogenic sources; (B) Fuel combustion sources; (C) Grass/wood/coal combustion sources.

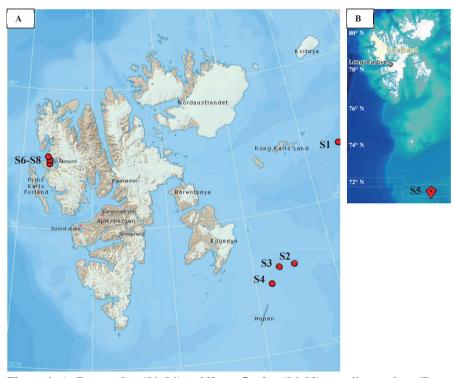
Sediment Quality Assessment based on PAH concentrations. Effect Range (ER) values were established as sediment quality guidelines to be used to predict adverse biological effects on marine organisms [158]. ERL (ER lower tenth percentile) values are available for 9 individual PAHs (PHE, ANT, FLT, PYR, BaA, CHR, BaP, Ind123P and BghiP) below which adverse effects on organisms are rarely observed. An ERL in sediment for the sum of PAHs (2990 μg/kg dry wt.) was calculated as the sum of the 9 PAH ERLs. The geometric means of the sum of 9 PAHs in sediment samples from our study (Barents Sea and Kongsfjorden, respectively) were 32 to 39 times lower than the 9 PAH ERL. However, in almost all Barents Sea (S2-S4) and all Ny-Ålesund (S6) samples PHE concentrations were above its individual ERL value (i.e., 240 μg/kg dry wt.). Additionally, FLT, BbF, dBahA and Ind123P concentrations in some Barents Sea samples (stations S3-S4, Figure 1A) were within the interval of acute effects for short term exposure (Class IV) established by the Norwegian Environment Agency [152]. In some cases (station S4), PYR concentrations were even within the interval of severe toxic effects (> 140 μg/kg dry wt., Class V) established by this same institution.





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**Figure 1. A:** Barents Sea (S1-S4) and Kongsfjorden (S6-S8) sampling stations (Basemap © Norwegian Polar Institute). **B:** Snøhvit sampling station (S5) (MAREANO mapping © Institute of Marine Research).

*PCBs in sediment*. For Kongsfjorden samples,  $\Sigma_7$ PCB concentrations varied between 0.00002 and 0.13 μg/kg dry wt. (geometric mean 0.001 μg/kg dry wt.), whereas for Barents Sea samples this range was between below limit of detection and 0.04 μg/kg dry wt. (geometric mean below limit of detection) (Table S5). Highest concentrations of  $\Sigma_7$ PCB were detected in Ny-Ålesund sediments where concentrations varied between 0.002 and 0.13 μg/kg dry wt. (geometric mean 0.01 μg/kg dry wt.). Significant differences were found between Ny-Ålesund and Blomstrandhalvøya station for PCB-18, -105, -118, -138 and -155 (Mann-Whitney U test, p < 0.05). PCB concentrations in Kongsfjorden sediment were within the range of earlier reported PCB concentrations in Kongsfjorden sediment [149]. In other Svalbard fjords,  $\Sigma_7$ PCB sediment concentrations of 0.74 to 5.41 μg/kg dry wt. in Barentsburg (Grønfjorden) and 1.8 to 20.2 μg/kg dry wt. in Pyramiden (Billefjorden) were reported in 2008, showing 6 to 14 orders of magnitude higher concentrations than in Ny-Ålesund samples included in our study [159,160]. PCB-138 and PCB-153 made up 60% of  $\Sigma_7$ PCB in some Ny-







Ålesund sediment samples and 53% of  $\Sigma_7$ PCB in Barentsburg sediment samples in 2008 [159] which may point to different local sources [140,161,162].



**Figure 1. C:** Kongsfjorden sampling stations (S6-S8). From north to south: Kapp Guissez (S8), Blomstrandhalvøya (S7) and Ny-Ålesund (S6) harbour (Basemap © Norwegian Polar Institute).

*HCB in sediment*. HCB concentrations in Kongsfjorden samples were between below limit of detection and 0.01 μg/kg dry wt., whereas concentrations of the Barents Sea sediment samples varied between below limit of detection and 0.37 μg/kg dry wt. (Table S5). For Kongsfjorden and Barents Sea samples, significant differences were found between Ny-Ålesund and Blomstrandhalvøya station (Kruskal-Wallis test, p = 0.037) as well as between SE Edgeøya (S3) sediments where HCB concentrations ranged 0.17 to 0.37 μg/kg dry wt. and Snøhvit (S5) sediments where HCB concentrations were  $\leq 0.00003$  μg/kg dry wt. (Kruskal-Wallis test, p = 0.037).





### 3.2. Sediment organic matter content

For Kongsfjorden, SOM percentages varied between 3.7% in Kapp Guissez and 6.3% in Ny-Ålesund, and for the Barents Sea samples between 4.9% in E Kong Karls Land and 9.1% in SE Edgeøya (S2) (Table S4).

# 3.3. Concentrations of PAHs, PCBs and HCB in biota

# 3.3.1. Polycyclic aromatic hydrocarbons

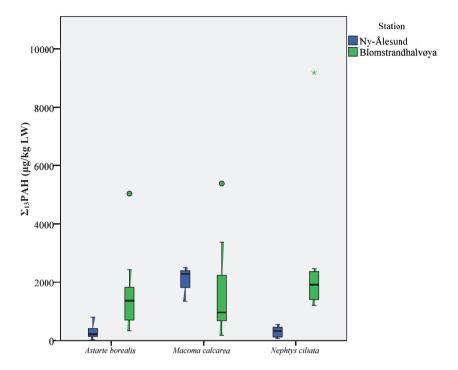
*PAH concentrations in biota.* Among species and stations,  $\Sigma_{13}$ PAH concentrations in benthic invertebrates of Kongsfjorden varied between 27 and 9185 μg/kg lipid wt. (geometric mean 729 μg/kg lipid wt.) (Table S9). Lowest  $\Sigma_{13}$ PAH concentrations were found in *A. borealis* at the Ny-Ålesund sampling station, whereas highest  $\Sigma_{13}$ PAH concentrations were observed in *N. ciliata* samples at Blomstrandhalvøya (Figure 2). In general the highest concentrations in *A. borealis* and *N. ciliata* were found in Blomstrandhalvøya samples, whereas concentrations in *M. calcarea* samples did not differ much between stations. In our study, *A. borealis* specimens (Ny-Ålesund) were mainly exposed to both petroleum and coal combustion-derived PAH sources in the field based on the agreement between mean PAH ratios (FLT/[FLT+PYR] and Ind123P/[Ind123P+BghiP]) calculated for both Ny-Ålesund bivalves (0.49/0.52) and sediment (0.64/0.38) samples [163,164].





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**Figure 2.**  $\Sigma_{13}$ PAH concentrations (µg/kg lipid wt.) in benthic invertebrates of Kongsfjorden.

*PAH Critical Body Residues*. To assess potential effects resulting from PAH bioaccumulation, the measured concentrations can be compared to USEPA CBR levels. Hansen et al. [154] calculated a  $\Sigma_{13}$ PAH total lipid LC50 body burden (96 h) for *N. arenaceodentata* (polychaeta) and for *L. stagnalis* (gastropoda) as 280,000 and 330,000 μg/g lipid wt. (using a  $\Sigma_{13}$ PAH W<sub>m</sub> (molar weight) = 3058 μg/μmol). This is a factor of 30,500 to 65,500 times higher than the mean  $\Sigma_{13}$ PAH lipid-normalised concentrations found in any of our species (Blomstrandhalvøya).

Hwang et al. [165] found a CBR of  $\Sigma_{24}$ PAH for lysosomal destabilization (at least 50% of destabilized cells) of 2100 µg/kg dry wt. in eastern oysters (*C. virginica*). Assuming a 20% and 12% dry wt. (based on soft tissue wet wt.) in *A. borealis* [166] and in *M. calcarea* [167], respectively, the maximum  $\Sigma_{13}$ PAH values detected in Blomstrandhalvøya *A. borealis* and *M. calcarea* specimens are 3 to 8 times lower (258 and 620 µg/kg dry wt., respectively) than the  $\Sigma_{24}$ PAH CBR calculated by Hwang et al. [165].







Based on EACs for PAHs established in mussels and oysters [153], and assuming 20% dry wt. in *A. borealis*, mean concentrations of PHE, FLT and PYR in *A. borealis* Blomstrandhalvøya specimens were 22, 6 and 2 times lower than the corresponding EAC.

Liu et al. [168] observed the induction of aryl hydrocarbon hydroxylase (AHH) and 7-ethoxyresorufin O-deethylase (EROD) enzymatic activities in R. philippinarum individuals with internal BaP concentrations of approximately 2.02  $\mu$ g/g dry wt. in the digestive gland. Assuming 20% and 12% dry wt. (based on soft tissue wet wt.) in A. borealis and M. calcarea, respectively, and accepting that a great proportion of BaP should be accumulated in the digestive gland of bivalves [169,170], then the maximum BaP concentrations detected in A. borealis (Blomstrandhalvøya) and M. calcarea (Ny-Ålesund) would be 108 to 465 times lower (0.02 and 0.004  $\mu$ g/g dry wt., respectively) than the same concentration observed by Liu et al. [168].

In summary, no potential toxicity of  $\Sigma_{13}$ PAH to *A. borealis*, *M. calcarea* or *N. ciliata* Blomstrandhalvøya populations was indicated according to three CBRs mentioned. Biotransformation enzymatic activity might be also negligible in the case of bivalves, according to the results obtained by Liu et al. [168] for *R. philippinarum*.

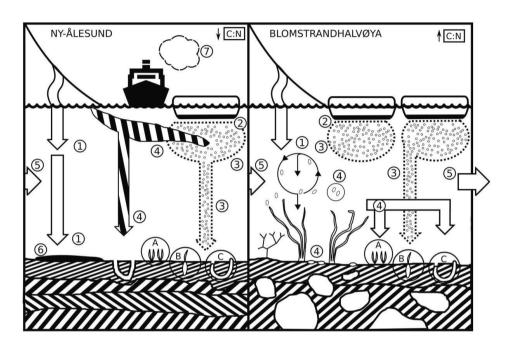
Differences in PAH concentrations among stations. Several factors were identified that can explain differences in PAH bioaccumulation between Ny-Ålesund (NY) and Blomstrandhalvøya (BL). A summary of factors in relation to ecological processes is provided as Figure 3, and in the discussion below we refer to these factors and processes as NY-x and BL-y, where x and y correspond to the numbers as indicated in Figure 3. Nearly all lipid-normalised individual PAH concentrations in the polychaete N. ciliata were significantly higher in Blomstrandhalvøya than in Ny-Ålesund specimens (Table S6). The lower  $\Sigma_{13}$ PAH content in Ny-Ålesund specimens (Table S9) can be explained by different factors. The most important factor may be differences in bioavailability related to differences in OM quality between sites. For instance, sediment in Ny-Ålesund harbour contained clearly visible black carbon (BC) particles that are known to strongly bind organic contaminants such as PAHs making them less available for uptake [78] (Figure 3, factor NY-7). BC particles (≥10 μm) predominate in surface sediments from the middle part of the Kongsfjorden compared to the outer part [171]. BC ≥10 µm are usually associated with local sources such as fossil-fuel fired power stations located in different Svalbard human settlements. Secondly, transport of coal particles from former coal mines within the Bayelva river catchment area, west of Ny-







Ålesund, might also influence the final bioavailability of sediment-based PAH sources, particularly in Ny-Ålesund *A. borealis* and *N. ciliata* specimens [148,172]. Thirdly, ongoing degradation of weathered oil in Ny-Ålesund [148,172] could have a highly significant impact on Ny-Ålesund individuals (Figure 3, factor NY-6) as this would minimize their in depth and deposit feeding in the weathered oil immediately surrounding area.



**Figure 3.** Possible factors contributing to observed differences between sampling stations. **Ny-Ålesund (NY):** (1) terrestrial particulate organic matter (POM), (2) ice algae, (3) fresh pelagic/settling organic matter (OM), (4) clayish river runoff, (5) advected and reworked detritus, (6) weathered oil, (7) volatile organic compounds (VOCs). **Blomstrandhalvøya (BL):** (1) terrestrial POM, (2) ice algae, (3) pelagic/settling OM, (4) brown macroalgal material, (5) advected and reworked detritus. A: *Astarte borealis*, B: *Macoma calcarea*, C: *Nephtys ciliata*.

Additionally, differences in OM quality related to differences in species and biomass composition of phytoplankton blooms and in macroalgae biomass would partly explain different benthic PAH bioaccumulation patterns between stations. A higher predominance of ice diatoms (Figure 3, factor BL-2) promoted by the resuspension of nutrients as a direct consequence of the continuous down-fjord advection of deep freshwater [145] (Figure 3, factor BL-5), as well as a higher input of macroalgal detritus [173] (Figure 3, factor BL-4),





are expected to increase the PAHs partitioning from the water phase to fresh settling OM (Figure 3, factor BL-3) as both algal materials are rich in polyunsaturated fatty acids (PUFAs) [174]. A higher relative intake of PUFA-rich material may increase the bioavailability of PAHs in Blomstrandhalvøya, particularly in the case of filter feeders (e.g., *A. borealis*). Finally, streams leaving Blomstrandhalvøya carried a much higher terrestrial settling OM (Figure 3, factor BL-1) concentration per both water volume and dry wt. of suspended matter than the Bayelva river (Ny-Ålesund) [146] (Figure 3, factor NY-1). A higher input of terrestrial OM in Blomstrandhalvøya can contribute to the mineralization of settling OM during sedimentation which eventually would increase the OC-normalised partition coefficients in settling particles [175-177].

A short response time to fresh settling OM may also increase PAH bioaccumulation in M. calcarea as concentrations of ANT, BaA, BaP and dBahA in M. calcarea Ny-Ålesund specimens were significantly higher than in Blomstrandhalvøya specimens. Bayelva river outflow usually contains a high proportion of clay materials (Figure 3, factor NY-4) that could also enhance phytoplankton mortality while zooplankton may suffer from direct mortality [131]. The immediate decrease of phytoplankton grazing by zooplankton due to the oxygen depletion of surface waters, would ultimately favour the sedimentation of more fresh phytoplankton (Figure 3, factors NY-2 and NY-3), which M. calcarea basically feeds on and to which PAHs may adsorb. M. calcarea is also one of the few benthic species in the Kongsfjorden of which a low trophic level and  $\delta^{13}$ C may suggest that this species is mainly reliant on fresh phytoplankton or fresh phytodetritus during summer [136,178]. Additionally, a lower PAH excretion capacity in M. calcarea [179,180] compared to the other two species may contribute to the observed differences between Ny-Ålesund and Blomstrandhalvøya M. calcarea individuals.

In summary, lipid-normalised PAH concentrations in all species seemed to be affected by some station related variation. This may be explained by either a difference in SOM content, origin and sedimentation rate of settling OM and BC strength source between stations.

Differences in PAH concentrations among species. BaP, dBahA and Ind123P burden in N. ciliata was significantly higher than in both bivalve species at Blomstrandhalvøya (Table S7). This may be caused by a more opportunistic feeding pattern of N. ciliata when compared to the bivalve species. Regarding Ny-Ålesund samples, nearly all lipid-normalised individual PAHs concentrations were significantly higher in M. calcarea compared with the other two







species (Table S7). Since steady-state PAH body burdens in polychaetes can also be greatly influenced by biotransformation processes [139,181], chronic PAH exposure is suggested to increase the inducibility of PAH biotransformation enzymes (e.g., CYP) in Ny-Ålesund polychaetes, and thus the probability of finding lower PAH body burdens in Ny-Ålesund individuals [182]. Additionally, specific CYP enzymes involved in xenobiotic biotransformation often are induced by substrates upon which they act, so that the mere presence of some PAHs may trigger the expression of CYP enzymes (e.g., PYR and PYR hydroxylase) [183]. However, long-term PAH detoxification might be slower, if not negligible as explained in the PAH critical body residues section, in bivalves since the major site of uptake and accumulation in these organisms is the digestive gland [169] whereas CYP genes transcription has been described as a tissue specific process reflecting the importance of gills in PAH detoxification [184].

In general, ingestion of SOM has been described as the major uptake pathway of organic contaminants in temperate benthic species (*A. virens*, *L. balthica*) [56]. Additionally, a higher ingestion rate, more opportunistic diet and higher dermal exposure of *N. ciliata* could lead to an extended bioaccumulation of sediment-bound PAHs [185]. *A. borealis* (filter-feeder) and *M. calcarea* (deposit-feeder) feed mainly on fresh phytoplankton or fresh phytodetritus, as shown by an abundance in algal-derived fatty acids in the same species [186,187]. In a Chukchi Sea study performed by Neff et al. [188], ΣPAH (parent and alkylated compounds) concentrations were also higher in *Macoma* sp. clams and maldanid worms than in *Astarte* sp. This may be due to the fact that, as a deposit-feeder, *M. calcarea* may also actively ingest SOM.

In general, differences in species-specific PAH accumulation patterns were seen between the three species, which can be explained by differences in uptake and elimination rates between these species due to underlying mechanisms that however could not be assessed unambiguously from the data.

Differences in PAH concentrations among species size-classes. For the Kruskal-Wallis test, a Bonferroni correction was applied and all effects are reported at a 0.0167 level of significance. We conclude that there is no significant size-class effect on A. borealis lipid-normalised PAH concentrations (Table S10), neither on M. calcarea (Blomstrandhalvøya) PAH concentrations. Similarly, no significant differences were found in  $\Sigma_{13}$ PAH lipid-normalised concentrations between both bivalves size-classes.







# 3.3.2. Polychlorinated biphenyls

Concentrations of PCBs in biota were below limit of detection in  $\geq$  90% of the biota samples analysed for every species and from both Ny-Ålesund and Blomstrandhalvøya station. PCBs were therefore excluded from further analyses.

### 3.3.3. Hexachlorobenzene

Among species and stations, concentrations of HCB varied between < 0.01 and 14.16  $\mu$ g/kg lipid wt. The minimum and maximum values were detected in *A. borealis* (Ny-Ålesund) and *M. calcarea* (Blomstrandhalvøya), respectively. In general the highest concentrations in biota tissues were found in Blomstrandhalvøya specimens. *M. calcarea* Blomstrandhalvøya individuals were assessed as markedly polluted (Class I to Class III for HCB, between 0.1 and 1  $\mu$ g/kg wet wt.) according to the EQSs for blue mussels defined by the Norwegian Environment Agency [189]. HCB lipid-normalised concentrations were found to be significantly higher in *A. borealis* (Mann-Whitney *U* test, p < 0.001) and *M. calcarea* (Mann-Whitney *U* test, p = 0.005) Blomstrandhalvøya specimens than in their Ny-Ålesund counterparts. An interspecies comparison within Blomstrandhalvøya station samples showed significantly higher (Kruskal-Wallis test, p = 0.027) HCB lipid-normalised concentrations in *A. borealis* specimens. No size effect on HCB lipid-normalised concentrations was reported after applying a Bonferroni correction.

### 3.4. Biota-sediment accumulation factors

For organisms that are closely linked to the sediment and are exposed mainly to sediment-bound contaminants, the BSAF has been shown to be a useful metric of bioaccumulation [56,82]. Here we explore to what extent BSAF is useful to assess bioaccumulation in the Arctic species studied, hypothesizing that BSAF has the lowest value for suspension feeder *A. borealis*. Data were sufficient to calculate BSAF values only for PAHs. Mean BSAF values are presented in Tables S11, S13 and S15 for both stations.

For Blomstrandhalvøya, the 10-90th percentile range of  $\Sigma_{13}$ PAH BSAF values per species was 0.59 to 3.50 for *A. borealis*, 0.35 to 4.76 for *M. calcarea* and 1.88 to 5.50 for *N. ciliata*, which thus shows a somewhat higher accumulation than the value of BSAF = 1-2 as predicted by Equilibrium Partitioning Theory (EPT) for hydrophobic contaminants [190]. Similar BSAF values have been reported earlier in the literature [56,83,191-193]. One explanation is

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that EPT neglects the ingestion pathway, whereas this pathway is relevant in nature. Another potential explanation is that uptake was not from the sediment compartment but merely from freshly deposited or settling suspended solids and their ambient water [136,194], having higher chemical concentrations than the sampled sediment, due to atmospheric or melt-water based sources of PAHs [121,177,195]. In Ny-Ålesund, the 10-90th percentile range of  $\Sigma_{13}$ PAH BSAF values per species was 0.00 to 0.02 for *A. borealis*, 0.05 to 0.07 for *M. calcarea* and 0.00 to 0.02 for *N. ciliata*. These general ranges of the  $\Sigma_{13}$ PAH BSAF values in Ny-Ålesund comply very well with the BSAF values found for BC impacted sites [196,197]. Sediments in Ny-Ålesund contain BC [171,198]. This is consistent with information from the mean PAH diagnostic ratios FLT/(FLT + PYR) and Ind123P/(Ind123P + BghiP), which were > 0.5 and 0.2 - 0.5, respectively (Table 1), and reflect a pyrogenic origin of these chemicals [120].

PAH BSAF values were significantly higher in Blomstrandhalvøya specimens compared to Ny-Ålesund specimens (Tables S12, S14 and S16). In the case of *A. borealis*, all considered BSAFs were significantly higher (Mann-Whitney U test, p < 0.001, Table S12) in Blomstrandhalvøya specimens. As for *M. calcarea* and *N. ciliata* Blomstrandhalvøya samples, significant differences between the stations were found in FLT, PYR, BaA, CHR and BghiP (Mann-Whitney U test, p < 0.05, Table S14) BSAFs and FLT, BaA, CHR, BbF, BghiP and Ind123P (Mann-Whitney U test, p < 0.05, Table S16) BSAFs in the first and second species, respectively.

An interspecies comparison within Blomstrandhalvøya station samples showed significantly higher BbF BSAFs in *A. borealis* (Kruskal-Wallis test, p = 0.006) whereas the same comparison within Ny-Ålesund showed significantly higher FLT, BaA and CHR BSAFs in *M. calcarea* (Kruskal-Wallis test, p < 0.05). No size class effect was observed in either of the species PAH BSAFs.

# 4. Implications and Conclusions

Our data analysis provides the first systematic evaluation of BSAF patterns across chemicals, locations and species in the Arctic. PAH concentrations and BSAFs were higher in Blomstrandhalvøya specimens which was explained by stronger sorption to BC and lesser uptake from ingested sediment in Ny-Ålesund which in turn implies that the use of the BSAF concept is not straightforward in these ecosystems. We conclude that PAH body residues can







be applied as a less variable and more straightforward pollution monitoring parameter than sediment concentrations or BSAFs as biota body concentrations in low trophic level species give a good estimate of the bioavailable fraction of organic contaminants present in the system. Based on this field study, the filter/deposit-feeding bivalve *M. calcarea* is proposed as a (sub-)Arctic indicator of O&G-derived environmental impacts due to its ability to accumulate PAHs in low O&G impacted areas (e.g., Ny-Ålesund) regardless of its body size class and to its selective feeding mode which decreases uncertainty on the species bioavailable PAH sources.

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# **Supporting Information**

## Extraction, clean-up and instrumental analysis

Frozen sediment samples (Barents Sea) and freeze-dried sediment samples (Kongsfjorden) were thawed at room temperature. Frozen pooled biota samples were thawed at room temperature and subdivided into replicates of equal weight before drying. Sediment subsamples and biota replicates were dried and grinded with diatomaceous earth and Soxhlet-extracted for 40 min with hexane/acetone (3:1) (DionexASE 350 System, Thermo Scientific). All extracts were split in two. One part was cleaned-up over an Al<sub>2</sub>O<sub>3</sub> column and used for PAH analysis, which was carried out on an HP 1100 HPLC equipped with a 4.6 mm Vydac guard and analytical reverse phase C18 column (201GD54T and 201TP54, respectively) which were kept at 22.00 °C. Detection was performed by an HP 1100 multi-wavelength fluorescence detector operating in the multi-emission wavelength mode. The mobile phase consisted of methanol/water (mixture and flow gradient). The injection volume was 20 µL.

The other part was cleaned-up using an  $Al_2O_3$ /silica column, desulfurized with Cu-powder, and used for PCB and HCB analysis. PCBs were measured by splitless injection of 1  $\mu$ L of sample in an upgraded HP 5890 series II gas chromatograph equipped with an HP 7673A autosampler system, two fused silica capillary columns, CP Sil-8 CB and CP Sil-5 CB (both 50 m.; i.d. 0.15 mm; d.f. 0.20  $\mu$ m), and two <sup>63</sup>Ni electron capture detectors. The injector and detector temperatures were 250 and 325 °C, respectively. Carrier gas was N<sub>2</sub> (1 mL/min).

Besides recoveries 3 blanks per each batch of 30 samples were used, and values were corrected for blanks.







Table S1. HCB and PCBs limit of detection (LOD, μg/L) and limit of quantification (LOQ, μg/L).

Compound	LOD	LOQ
HCB	0.08	0.27
PCB 18	0.25	0.85
PCB 20	0.13	0.43
PCB 28	0.13	0.42
PCB 29	0.47	1.58
PCB 31	0.11	0.37
PCB 44	0.34	1.14
PCB 52	0.76	2.53
PCB 101	0.12	0.40
PCB 105	0.09	0.30
PCB 118	0.09	0.32
PCB 138	0.08	0.28
PCB 153	0.13	0.42
PCB 155	0.09	0.29
PCB 170	0.08	0.28
PCB 180	0.07	0.23
PCB 194	0.08	0.27
PCB 204	0.03	0.09
PCB 209	0.10	0.32

Table S2. Biota samples size classification.

Species	Size class	Shell/body length (mm)	Blomstrandhalvøya specimens	Ny-Ålesund specimens
	1	≤ 35.9	39	
1-tt- 11:-	2	36.0 – 37.9	20	
Astarte borealis	3	38.0 – 39.9	15	25
,	4	≥ 40.0	12	45
17	1	≤ 25.5	30	
Macoma calcarea	2	25.6 – 26.5	15	10
caicarea	3	≥ 26.6	30	
	1	≤ 84.9	20	
N7 1, 11 ,	2	85.0 – 93.9	25	
Nephtys ciliata	3	94.0 - 102.9		10
,	4	≥ 103.0		15







Table S3. Sampling stations in Barents Sea and Kongsfjorden Bay (Svalbard, Norway).

Location	Station	Latitude	Longitude	Water depth (m)	Sediment samples (0.1 m <sup>2</sup> )	Biota spp.
E Kong Karls Land	S1	78° 45.8'	32° 53.6'	250	4	n.a.
SE Edgeøya	S2	77° 14.0'	27° 37.0'	230	3	n.a.
SE Edgeøya	S3	77° 13.8'	26° 39.6'	130	3	n.a.
SE Edgeøya	S4	77° 00.0'	26° 00.0'	110	3	n.a.
Snøhvit	S5	71° 37.1'	21° 04.3'	320	5	n.a.
Ny-Ålesund	S6	78° 55.0'	n.a.	15-20	4	Astarte borealis, Macoma calcarea, Nephtys ciliata
Blomstrandhalvøya	S7	78° 59.0'	n.a.	20-25	4	Astarte borealis, Macoma calcarea, Nephtys ciliata
Kapp Guissez	S8	79° 02.0'	n.a.	45-50	4	Astarte borealis, Nephtys ciliata

n.a.: not available.







 $\textbf{Table S4.} \ PAH \ (\mu g/kg \ dry \ wt.) \ and \ OM \ (\% \ dry \ wt.) \ concentrations \ (mean \pm SD) \ in \ sediment \ samples \ collected \ in \ Kongstjorden \ Bay \ and \ the \ Barents \ Sea.$ 

		Bare	Barents Sea (October 2013)	3)		Kongs	Kongsfjorden (July 2013)	(3)
Station	S1	S2	S3	S4	S5	9S	S7	88
OM	$4.97 \pm 0.16$	$9.04 \pm 0.10$	$6.97 \pm 0.56$	$6.15 \pm 0.26$	$8.22 \pm 0.46$	$6.10 \pm 0.11$	$4.25 \pm 0.23$	$4.17 \pm 0.29$
PHE	$86.17 \pm 102.61$	$371.44 \pm 284.13$	$657.79 \pm 261.61$	$1015.60 \pm 300.92$	$17.38 \pm 17.38$	$707.05 \pm 84.88$	$7.52 \pm 6.54$	$16.86 \pm 6.02$
ANT	$0.00\pm0.00$	$4.07 \pm 5.25$	$4.76 \pm 2.37$	$10.08 \pm 3.32$	$0.04 \pm 0.07$	$52.14 \pm 4.29$	$0.22 \pm 0.25$	$0.00 \pm 0.00$
FLT	$23.83 \pm 25.90$	$143.81 \pm 104.12$	$118.27 \pm 46.70$	$184.89 \pm 57.64$	$4.02 \pm 3.99$	$197.72 \pm 18.81$	$2.02 \pm 1.46$	$6.88 \pm 2.25$
PYR	$27.59 \pm 28.79$	$113.00 \pm 79.28$	$133.24 \pm 54.59$	$203.51 \pm 47.39$	$5.14 \pm 4.03$	$113.76 \pm 13.65$	$4.03 \pm 2.65$	$6.63 \pm 2.89$
BaA	$8.22 \pm 7.35$	$33.46 \pm 22.38$	$26.60\pm10.39$	$32.53 \pm 9.84$	$1.80 \pm 0.69$	$45.80 \pm 5.65$	$0.34\pm0.07$	$1.07 \pm 0.14$
CHR	$19.94 \pm 20.71$	$106.93 \pm 79.03$	$149.77 \pm 78.80$	$244.57 \pm 64.87$	$5.22 \pm 2.60$	$63.54 \pm 6.77$	$5.51 \pm 2.45$	$6.25 \pm 1.43$
BeP	$36.93 \pm 52.23$	$217.45 \pm 172.38$	$494.50 \pm 207.24$	$698.89 \pm 220.75$	$0.00 \pm 0.00$	$566.51 \pm 96.80$	$0.00 \pm 0.00$	$2.33 \pm 2.21$
BbF	$52.43 \pm 49.31$	$208.06 \pm 141.43$	$202.79 \pm 78.64$	$248.90 \pm 74.94$	$24.39 \pm 16.73$	$34.16 \pm 5.24$	$3.43 \pm 0.60$	$8.95 \pm 1.43$
BkF	$13.28 \pm 11.66$	$57.62 \pm 39.02$	$45.50 \pm 17.86$	$57.07 \pm 17.05$	$6.08 \pm 4.67$	$27.19 \pm 2.26$	$0.41 \pm 0.36$	$2.32 \pm 0.36$
BaP	$0.00\pm0.00$	$71.41 \pm 63.65$	$41.51\pm40.88$	$84.99 \pm 52.65$	$0.00 \pm 0.00$	$67.82 \pm 16.41$	$0.00\pm0.00$	$1.84 \pm 3.68$
BghiP	$9.26 \pm 6.66$	$26.81 \pm 16.97$	$24.01 \pm 9.18$	$27.23 \pm 8.54$	$4.22 \pm 2.41$	$15.75 \pm 1.55$	$1.11 \pm 0.42$	$1.07 \pm 0.18$
dBahA	$102.77 \pm 87.86$	$226.68 \pm 130.41$	$314.44 \pm 144.50$	$437.48 \pm 112.55$	$17.68 \pm 8.56$	$120.17 \pm 20.19$	$2.05 \pm 1.79$	$9.12 \pm 1.80$
Ind123P	$43.13 \pm 34.78$	$138.96 \pm 91.91$	$87.96 \pm 32.30$	$106.86 \pm 30.94$	$25.80 \pm 19.64$	$9.92 \pm 2.08$	$1.45 \pm 1.01$	$5.68 \pm 0.51$
$\Sigma_{13}$ PAH	$423.54 \pm 427.85$	5 $1719.69 \pm 1221.84$ $2301.14 \pm 963.26$	$2301.14 \pm 963.26$		$111.75 \pm 76.70$	$3352.61 \pm 991.97$ $111.75 \pm 76.70$ $2021.53 \pm 260.85$	$28.09 \pm 14.24$	$68.99 \pm 16.48$





**Table S5.** HCB ( $\mu g/kg$  dry wt.), CB ( $\mu g/kg$  dry wt.) and OM (% dry wt.) concentrations (mean  $\pm$  SD) in sediment samples collected in Kongstjorden Bay and the Barents Sea. LOD ( $\mu g/L$ ) values are provided on Table S1.

		Bare	Barents Sea (October 2013)	013)		Kor	Kongsfjorden (July 2013)	113)
Station	S1	S2	S3	S4	SS	9S	S7	88
	$4.97 \pm 0.16$	$9.04 \pm 0.10$	$6.97 \pm 0.56$	$6.15 \pm 0.26$	$8.22 \pm 0.46$	$6.10 \pm 0.11$	$4.25 \pm 0.23$	$4.17 \pm 0.29$
HCB	<0.01	$0.21 \pm 0.05$	$0.28 \pm 0.10$	$0.27 \pm 0.02$	<0.01	<0.01	<0.01	<0.01
PCB 18	$0.07 \pm 0.05$	$0.03 \pm 0.00$	$0.06 \pm 0.03$	$0.11 \pm 0.17$	<0.01	$0.01 \pm 0.01$	<0.01	<0.01
PCB 20	<0.01	<0.01	$0.01 \pm 0.00$	$0.02 \pm 0.00$	<0.01	<0.01	<0.01	<0.01
PCB 28	<0.01	$0.01 \pm 0.00$	<0.01	$0.01 \pm 0.00$	<0.01	<0.01	<0.01	<0.01
PCB 29	$0.02 \pm 0.02$	$0.08 \pm 0.02$	$0.01 \pm 0.02$	<0.01	$0.08 \pm 0.13$	<0.01	$0.01 \pm 0.01$	<0.01
PCB 31	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
PCB 44	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
PCB 52	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
PCB 101	<0.01	<0.01	<0.01	$0.01 \pm 0.01$	<0.01	<0.01	<0.01	<0.01
PCB 105	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
PCB 118	<0.01	$0.01 \pm 0.00$	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
PCB 138	<0.01	$0.01 \pm 0.00$	<0.01	<0.01	<0.01	$0.01 \pm 0.02$	<0.01	<0.01
PCB 153	<0.01	<0.01	<0.01	<0.01	<0.01	$0.02 \pm 0.03$	<0.01	<0.01
PCB 155	<0.01	$0.01 \pm 0.00$	$0.11 \pm 0.01$	$0.09 \pm 0.12$	<0.01	<0.01	<0.01	<0.01
PCB 170	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	$0.01 \pm 0.02$
PCB 180	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
PCB 194	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	$0.02 \pm 0.01$
PCB 204	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	$0.01 \pm 0.00$
PCB 209	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	$0.02 \pm 0.02$
$\Sigma_7$ PCB	<0.01	$0.03 \pm 0.01$	$0.01 \pm 0.01$	$0.02 \pm 0.01$	<0.01	$0.05 \pm 0.07$	$0.01 \pm 0.01$	<0.01





Table S6. Single species lipid-normalised PAH concentrations statistical comparison (Mann-Whitney U test p-value) between stations (BL: Blomstrandhalvøya; NY: Ny-Ålesund; AB: Astarte borealis; MC: Macoma calcarea; NC: Nephtys ciliata). Only compounds for which statistical significance was found are included.

Station – PHE PHE	0.001 0.026 0.000	8	C	C 0.015	BL - NC 0.003 0.020	۵
PYR	0.000 0.0			0.0	0.016 0.0	
СНК	0.033 0.001			0.030	003 0.003	
Beb	0.000				0.004	
BPF	0.000				0.009	
BKF	0.001					
$\mathrm{B}^{\mathrm{g}}\mathrm{b}$	0.024			0.007	0.020	
qidga	0.049				0.020	
Adsab				0.043	0.004	
4821bal	0.002				0.004	
$\mathrm{HAq}_{\epsilon_1} \mathfrak{Z}$	0.000				0.003	

**Table S7.** Single station lipid-normalised PAH concentrations statistical comparison (Kruskal-Wallis test *p*-value) between species (BL: Blomstrandhalvøya; NY: Ny-Ålesund; AB: Astarte borealis; MC: Macoma calcarea; NC: Nephtys ciliata). Only compounds for which statistical significance was found are included.

$\Sigma^{13}$ byH			6 8		0.023	
4621bnl			0.032		0.018	
AdsBb			0.032		0.028	
$\mathrm{B}^{\mathrm{g}}\mathrm{b}$			0.002		0.005	
BPE	900.0					
В¢Ь					0.037	
CHK					0.048	
$ m B_aA$					0.023	
РҮК					0.037	
TLT					0.013	
TNA					0.003	
ЬНЕ					0.011	
Station - Spp.	BL-AB	BL-MC	BL-NC	NY - AB	NY - MC	NY - NC







Table S8. PAH (µg/kg wet wt.) concentrations in biota samples collected in Kongstforden Bay (Svalbard, Norway).

Station			Blomstrandhalvøya	ıdhalvøya					Ny-Ålesund	esnud		
Spp.	Astarte	Astarte borealis	Macoma calcarea	calcarea	Nephtys ciliata	ciliata	Astarte borealis	orealis	Macoma calcarea	calcarea	Nephtys ciliata	ciliate
u	~	98	7	75	45	5	75		15	5	25	
Mean lipid wt. (% wet wt.)	-	1.17	2.67	57	1.24	24	3.17	7	3.23	23	3.55	5
РАН	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
PHE	3.84	3.95	7.36	6.65	9.61	9.19	1.06	1.23	15.42	2.53	2.31	1.65
ANT	0.62	1.15	0.26	69.0	0.64	0.75	0.02	0.07	1.98	1.66	0.24	0.45
FLT	1.40	0.93	4.29	2.70	1.76	1.85	0.78	09.0	6.56	1.90	1.09	0.11
PYR	3.97	2.37	6.71	4.65	5.76	6.49	1.95	2.16	14.23	5.21	2.33	2.39
BaA	0.29	0.29	0.65	0.46	0.63	0.48	0.22	0.16	3.00	1.22	0.18	0.00
CHR	1.45	1.14	4.96	3.65	3.38	3.64	0.91	0.87	5.81	2.79	0.88	99.0
ВеР	1.58	1.13	3.46	2.51	3.12	2.33	0.74	0.51	3.28	0.98	1.25	1.30
BbF	0.95	0.63	0.65	0.36	0.95	0.85	99.0	0.26	1.49	0.85	0.37	0.40
BkF	0.46	0.35	0.61	0.47	0.48	0.59	0.41	0.33	1.47	1.09	0.27	0.38
BaP	0.44	0.95	0.17	0.19	0.58	0.58	0.16	0.10	1.42	0.33	0.08	90.0
BghiP	0.38	0.40	1.13	0.84	1.07	1.54	0.28	0.18	1.38	1.07	0.35	0.45
dBahA	0.37	0.59	0.25	0.29	0.99	1.46	0.07	0.08	1.47	0.84	0.17	0.22
Ind123P	1.48	1.59	2.85	3.30	4.40	4.64	69.0	1.04	7.45	1.63	1.22	0.99
$\Sigma_{13}$ PAH	17.23	12.45	33.36	21.48	33.35	32.58	7.96	5.00	64.98	15.52	10.73	7.26





Table S9. PAH (µg/kg lipid wt.) concentrations in biota samples collected in Kongsfjorden Bay (Svalbard, Norway).

Station			Blomstrandhalvøya	ıdhalvøya					Ny-Ålesund	puns		
Spp.	Astarte	Astarte borealis	Масота	Macoma calcarea	Nephtys ciliata	ciliata	Astarte i	Astarte borealis	Macoma calcarea	alcarea	Nephtys ciliata	ciliata
и	8	98	75	5	45	5	75	5	15		25	
PAH	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
PHE	342.85	382.59	379.76	420.04	763.11	711.10	32.31	31.51	483.54	110.56	64.79	47.00
ANT	60.28	111.00	11.80	28.72	51.81	60.71	0.42	1.52	66.09	52.88	7.06	13.77
FLT	123.91	88.75	185.52	130.43	138.90	143.57	26.33	22.92	205.59	67.23	30.75	3.85
PYR	355.52	230.85	324.16	302.83	449.05	497.74	87.11	117.54	449.77	190.63	63.73	62.50
BaA	24.25	24.51	30.71	30.45	50.22	36.67	8.93	12.07	94.10	41.53	4.95	2.48
CHR	128.34	103.90	241.95	263.73	270.09	280.77	38.98	52.72	184.03	98.54	24.85	18.74
BeP	142.19	104.87	180.21	191.05	253.40	184.58	25.21	15.89	103.23	36.88	36.52	40.24
BbF	82.17	53.70	29.27	23.05	74.90	64.75	23.49	12.82	47.60	28.78	10.14	10.73
BkF	38.32	23.70	30.28	35.54	37.57	45.96	15.99	15.85	47.38	35.85	7.81	11.53
BaP	39.39	91.91	5.62	98.9	45.24	44.24	6.04	4.55	44.05	10.22	2.32	1.54
BghiP	29.71	27.58	46.58	31.25	84.41	118.67	10.90	6.67	44.06	34.81	10.10	13.61
dBahA	31.73	49.66	14.17	21.17	77.03	112.76	3.56	5.48	47.02	28.16	4.61	99.5
Ind123P	130.84		147.84	217.94	345.74	358.12	24.48	31.13	233.28	63.41	34.59	28.62
$\Sigma_{13}$ PAH	1529.50	1184.37	1627.87	1526.95	2641.48	2497.47	303.74	244.98	2044.63	613.36	302.22	207.35







Table S10. Astarte borealis PAH (µg/kg lipid wt.) concentrations in samples collected in Kongstjorden Bay (Svalbard, Norway) for different size classes.

Station				Blomstrandhalvøya	dhalvøya					Ny-Å	Ny-Ålesund	
Size class		1	2					4	(,,			4
и	6	39	20		-	5	1	2	2	25	4	45
PAH	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
PHE	641.50	529.63	148.12	105.09	124.57	144.14	322.99	235.38	51.03	38.14	22.96	26.11
ANT	90.34	147.84	1.40	2.80	6.84	11.84	142.14	137.33	1.18	2.63	0.04	0.12
FLT	179.58	107.34	94.51	64.29	73.00	17.58	121.22	110.47	29.66	22.51	24.51	25.59
PYR	471.00	270.22	315.55	194.01	212.59	171.25	359.27	272.05	135.60	99.53	67.79	129.4
BaA	35.19	28.43	20.27	23.43	28.35	30.99	7.23	4.09	6.38	4.55	10.13	15.4
CHR	171.20	147.48	135.82	62.77	75.84	38.26	99.40	120.80	56.02	34.82	33.85	62.1
BeP	232.67	125.92	99.04	52.16	83.09	15.30	108.03	98.90	36.93	19.09	19.83	11.3
BbF	129.30	59.77	62.96	34.34	68.93	44.86	42.49	18.59	30.25	16.87	20.59	10.00
BkF	45.81	20.35	35.62	25.81	42.57	37.98	25.17	14.84	22.84	13.49	13.05	17.3
BaP	90.68	155.17	22.56	29.82	9.65	11.10	8.80	7.46	6.49	3.67	6.03	5.37
BghiP	41.83	17.82	24.87	28.97	37.80	46.74	7.86	4.07	14.15	9.11	9.10	10.58
dBahA	00.69	70.62	21.88	31.89	2.80	3.11	11.68	15.06	4.53	4.88	3.42	6.20
Ind123P	241.14	187.24	111.67	90.53	43.03	18.85	60.40	62.18	50.60	31.30	12.54	23.5
$\Sigma_{13}$ PAH	2437.63	1520.01	1094.29	611.32	809.06	512.61	1316.67	1043.02	445.66	195.41	243.83	256.27







0.09

0.05

0.12

0.20

2.21

69.0

4.09

98.9

8.84

Table S11. Astarte borealis PAH BSAFs. BSAFs of a specific compound were calculated provided that there were at least three detects in sediment

Station				Blomstra	Blomstrandhalvøya					Ny-Alesund	puns	
Size class			2		3		7	4	3		4	
и	3	39	2	20	15	2		12	25		45	
Mean lipid wt. (% wet wt.)		1.12		1.13	1.31	11	-:	1.18	3.09	6	3.20	0.
PAH	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
PHE	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.00	0.00	0.00	0.00
ANT	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.00	0.00	0.00	0.00
FLT	3.61	2.16	1.90	1.29	1.47	0.35	2.44	2.22	0.01	0.01	0.01	0.01
PYR	4.74	2.72	3.18	1.95	2.14	1.73	3.62	2.74	0.07	0.05	0.04	0.07
BaA	4.17	3.37	2.40	2.78	3.36	3.67	98.0	0.49	0.01	0.01	0.01	0.02
CHR	1.26	1.09	1.00	0.46	0.56	0.28	0.73	0.89	0.05	0.03	0.03	90.0
ВеР	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.00	0.00	0.00	0.00
BbF	1.53	0.71	0.74	0.41	0.82	0.53	0.50	0.22	0.05	0.03	0.04	0.02
BkF	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.05	0.03	0.03	0.04
BaP	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.01	0.00	0.01	0.00
BghiP	1.17	0.50	0.70	0.81	1.06	1.31	0.22	0.11	60.0	90.0	90.0	0.07
dBahA	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.00	0.00	0.00	0.00

BghiP
dBahA
Ind123P
n.a.: not available.





**Table S12.** Astarte borealis PAH BSAFs: station effect. Only compounds for which ≥ 90% BSAF values were higher than zero in both stations were included in the statistical analysis. Maximum mean ranks for all significant results were found in Blomstrandhalvøya specimens.

			FLT	BaA	CHR	BbF	BghiP
Mann-Whitney $U$			0.000	0.000	2.000	0.000	8.000
Wilcoxon W			120.000	120.000	122.000	120.000	128.000
Z			-4.667	-4.666	-4.585	-4.666	-4.335
Asymp. sig. (2-tailed)			0.000	0.000	0.000	0.000	0.000
Exact sig. [2*(1-tailed sig.)]			0.000 <sup>b</sup>				
	Sig.		$0.000^{\circ}$	$0.000^{\circ}$	$0.000^{\circ}$	$0.000^{\circ}$	$0.000^{\circ}$
Monte Carlo sig. (2-tailed)	000	Lower bound	0.000	0.000	0.000	0.000	0.000
	99% confidence interval	Upper bound	0.000	0.000	0.000	0.000	0.000
	Sig.		$0.000^{\circ}$	$0.000^{\circ}$	$0.000^{\circ}$	$0.000^{\circ}$	$0.000^{\circ}$
Monte Carlo sig. (1-tailed)	000/ 2006 5 000	Lower bound	0.000	0.000	0.000	0.000	0.000
	99% confidence interval	Upper bound	0.000	0.000	0.000	0.000	0.000

<sup>a</sup> Grouping variable: Station

<sup>b</sup>Not corrected for ties.

<sup>c</sup> Based on 10000 sampled tables with starting seed 2000000.





**Table S13.** Macoma calcarea PAH BSAFs. BSAFs of a specific compound were calculated provided that there were at least three detects in sediment samples.

1   2   2     30   15     15   15     16   16   15     17   2.84   2.72     10   10   1.8   1.72     10   1.72   1.72   1.85     10   2.40   3.21   3.48   0.67     10   3.21   3.48   0.67     10   3.21   3.48   0.67     10   3.21   3.48   0.67     10   3.21   3.48   0.67     10   3.21   3.48   0.67     10   3.21   3.48   0.67     10   3.21   3.48   0.67     10   3.21   3.48   0.67     10   3.21   3.48   0.76     10   3.21   3.48   0.76     10   3.21   3.48   0.76     10   3.21   3.48   0.76     10   3.21   3.48   0.76     10   3.21   3.48   0.16     10   3.21   3.48   0.16     10   3.21   3.48   0.16     10   3.21   3.48   0.16     10   3.21   3.48   0.18     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10   3.21   3.48     10	Station			Blomstrandhalvøya	ıdhalvøya			Ny-Ålesund	puns
30         15           sid wt. (% wet wt.)         2.84         2.72           Mean         SD         Mean         SD         Mean           n.a.         n.a.         n.a.         n.a.         n.a.         n.a.           n.a.         n.a.         n.a.         n.a.         n.a.         n.a.           2.07         1.72         5.60         2.57         3.63           2.73         3.17         2.35         0.83         3.99           2.40         3.21         3.48         0.76         4.34           1.03         1.12         1.43         0.67         2.34           n.a.         n.a.         n.a.         n.a.         n.a.           n.a.         n.a.         n.a.         n.a.           n.a.         n.a.	Size class	1		2			3	2	
ipid wt. (% wet wt.)         2.84         2.72           Mean         SD         Mean         Mean           n.a.         n.a.         n.a.         n.a.         n.a.           n.a.         n.a.         n.a.         n.a.         n.a.           2.07         1.72         5.60         2.57         3.63           2.73         3.17         2.35         0.83         3.99           2.40         3.21         3.48         0.76         4.34           1.03         1.12         1.43         0.67         2.34           n.a.         n.a.         n.a.         n.a.         n.a.           n.a.		3(	0	I	5	3	0	10	
Mean         SD         Mean         SD           n.a.         n.a.         n.a.         n.a.           n.a.         n.a.         n.a.         n.a.           2.07         1.72         5.60         2.57           2.73         3.17         2.35         0.83           2.40         3.21         3.48         0.76           1.03         1.12         1.43         0.67           n.a.         n.a.         n.a.         n.a.	ean lipid wt. (% wet wt.)	2.8	84	2.7	72	2.	57	3.05	6
n.a.     n.a.     n.a.     n.a.       n.a.     n.a.     n.a.     n.a.       2.07     1.72     5.60     2.57       2.73     3.17     2.35     0.83       2.40     3.21     3.48     0.76       1.03     1.12     1.43     0.67       n.a.     n.a.     n.a.     n.a.       n.a.     n.a.	Н	Mean	SD	Mean	SD	Mean	SD	Mean	SD
n.a.     n.a.     n.a.     n.a.       2.07     1.72     5.60     2.57       2.73     3.17     2.35     0.83       2.40     3.21     3.48     0.76       1.03     1.12     1.43     0.67       n.a.     n.a.     n.a.     n.a.	IE	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.05	0.00
2.07 1.72 5.60 2.57 2.73 3.17 2.35 0.83 2.40 3.21 3.48 0.76 1.03 1.12 1.43 0.67 n.a. n.a. n.a. n.a. n.a. 0.22 0.22 0.28 0.16 n.a. n.a. n.a. n.a. n.a. n.a. n.a. n.a.	IN	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.07	0.09
2.73 3.17 2.35 0.83 2.40 3.21 3.48 0.76 1.03 1.12 1.43 0.67 n.a. n.a. n.a. n.a. n.a. 0.22 0.22 0.28 0.16 n.a. n.a. n.a. n.a. n.a. n.a. n.a. n.a.	T	2.07	1.72	5.60	2.57	3.63	2.75	0.07	0.02
2.40 3.21 3.48 0.76 1.03 1.12 1.43 0.67 1.03 1.12 1.43 0.67 1.03 0.22 0.28 0.16 1.04 0.22 0.28 0.16 1.05 0.29 0.28 0.16 1.05 0.20 0.28 0.16 1.05 0.20 0.28 0.16 1.05 0.20 0.28 0.16 1.05 0.20 0.20 0.20 1.05 0.35 1.05 0.35 1.05 0.35	/R	2.73	3.17	2.35	0.83	3.99	3.86	0.29	0.08
1.03 1.12 1.43 0.67  n.a. n.a. n.a. n.a. n.a.  0.22 0.22 0.28 0.16  n.a. n.a. n.a. n.a. n.a.  n.a. n.a. n.	aA	2.40	3.21	3.48	0.76	4.34	4.77	0.15	90:0
n.a.     n.a.     n.a.     n.a.     n.a.       0.22     0.22     0.28     0.16       n.a.     n.a.     n.a.     n.a.       n.a.     n.a.     n.a.     n.a.       0.96     0.81     0.79     0.35       n.a.     n.a.     n.a.     n.a.       s. 34     6.00     1.77     1.73	HR	1.03	1.12	1.43	0.67	2.34	2.62	0.22	0.09
0.22 0.28 0.16  n.a. n.a. n.a. n.a. n.a.  n.a. n.a. n.	e <b>P</b>	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.01	0.00
n.a.         n.a.         n.a.         n.a.           n.a.         n.a.         n.a.         n.a.           0.96         0.81         0.79         0.35           n.a.         n.a.         n.a.         n.a.           p         5 34         6 06         1.77         1.72	bF	0.22	0.22	0.28	0.16	0.44	0.33	0.11	0.01
n.a. n.a. n.a. n.a. n.a. n.a. n.a. n.a.	KF.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.15	0.00
0.96 0.81 0.79 0.35 n.a. n.a. n.a. n.a. n.a. n.a.	aP	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.04	0.01
n.a. n.a. n.a. n.a. n.a. n.a. p. s 3.4 6.00 1.27 1.72	ghiP	96.0	0.81	0.79	0.35	1.74	96:0	0.37	0.17
5.34 6.09 1.27 1.72	SahA	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.03	0.00
7/11 /21 60:0 +5:0	Ind123P	5.34	60.9	1.27	1.72	7.53	10.42	0.99	0.28

n.a.: not available.





Table S14. Macoma calcarea PAH BSAFs: station effect. Only compounds for which ≥ 90% BSAF values were higher than zero in both stations were included in the statistical analysis. Maximum mean ranks for all significant results were found in Blomstrandhalvøya specimens.

		Test statistics <sup>a</sup>	icsa						
			FLT	PYR	BaA	CHR	BbF	BghiP	Ind123P
Mann-Whitney U			0.000	1.000	1.000	3.000	5.000	3.000	10.000
Wilcoxon W			000.9	7.000	7.000	000.6	11.000	000.6	16.000
Z			-2.598	-2.454	-2.454	-2.165	-1.876	-2.165	-1.155
Asymp. sig. (2-tailed)			0.009	0.014	0.014	0.030	0.061	0.030	0.248
Exact sig. [2*(1-tailed sig.)]			$0.004^{b}$	0.009 <sup>b</sup>	0.009 <sup>b</sup>	$0.031^{b}$	$0.070^{b}$	$0.031^{b}$	$0.295^{b}$
	Sig.		$0.005^{\circ}$	$0.010^{\circ}$	$0.010^{\circ}$	$0.032^{\circ}$	$0.072^{\circ}$	$0.032^{\circ}$	$0.298^{\circ}$
Monte Carlo sig. (2-tailed)	Lorse Combine	Lower bound	0.003	0.007	0.007	0.027	0.066	0.027	0.287
		Upper bound	0.007	0.012	0.012	0.036	0.079	0.036	0.310
	Sig.		$0.003^{\circ}$	$0.006^{\circ}$	$0.006^{\circ}$	$0.017^{c}$	$0.037^{\circ}$	$0.017^{c}$	$0.147^{c}$
Monte Carlo sig. (1-tailed)	Longing Santamon	Lower bound		0.004	0.004	0.014	0.032	0.014	0.138
		Upper bound	0.004		0.007	0.020	0.042	0.020	0.156

a Grouping variable: Station



 $<sup>^{\</sup>rm b}$  Not corrected for ties.  $^{\rm c}$  Based on 10000 sampled tables with starting seed 2000000.



Table S15. Nephys ciliata PAH BSAFs. BSAFs of a specific compound were calculated provided that there were at least three detects in sediment

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Station		Blomstra	Blomstrandhalvøya			Ny-Alesund	esund	
Size class		1		2	3		4	
u u		20	2	25	10		15	5
Mean lipid wt. (% wet wt.)	T	1.24	1.	24	3.6	0	3.51	51
PAH	Mean	SD	Mean	SD	Mean	SD	Mean	SD
PHE	n.a.	n.a.	n.a.	n.a.	0.00	0.01	0.01	0.00
ANT	n.a.	n.a.	n.a.	n.a.	0.00	0.00	0.01	0.02
FLT	1.86	1.35	3.54	3.70	0.01	0.00	0.01	0.00
PYR	2.57	1.95	80.9	6.37	0.04	90.0	0.03	0.02
BaA	4.61	0.95	7.02	5.82	0.00	0.01	0.01	0.00
CHR	1.43	0.63	2.43	2.78	0.02	0.01	0.03	0.02
BeP	n.a.	n.a.	n.a.	n.a.	0.00	0.00	0.01	0.01
BbF	0.95	0.54	0.84	0.97	0.02	0.02	0.02	0.02
BkF	n.a.	n.a.	n.a.	n.a.	0.01	0.01	0.02	0.03
BaP	n.a.	n.a.	n.a.	n.a.	0.00	0.00	0.00	0.00
BghiP	2.34	1.58	3.70	5.91	0.04	0.04	0.08	0.11
dBahA	n.a.	n.a.	n.a.	n.a.	0.00	0.00	0.00	0.00
Ind123P	7.61	4.18	11.37	13.45	0.12	0.10	0.14	0.14
							J	

n.a.: not available.





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# Chapter 2

Table S16. Nephtys ciliata PAH BSAFs: station effect. Only compounds for which  $\geq 90\%$  BSAF values were higher than zero in both stations were included in the statistical analysis. Maximum mean ranks for all significant results were found in Blomstrandhalvøya specimens.

			est statistics.					
			FLT	BaA	CHR	BbF	BghiP	Ind123P
Mann-Whitney U			5.000	0.000	0.000	0.000	0.000	0.000
Wilcoxon W			20.000	15.000	15.000	15.000	15.000	15.000
Z			-2.333	-3.000	-3.000	-3.000	-3.000	-3.000
Asymp. sig. (2-tailed)	H)		0.020	0.003	0.003	0.003	0.003	0.003
Exact sig. [2*(1-tailed sig.)]	ed sig.)]		$0.019^{b}$	$0.001^{b}$	$0.001^{b}$	$0.001^{b}$	$0.001^{b}$	$0.001^{b}$
M	Sig.		$0.020^{\circ}$	$0.001^{\circ}$	$0.001^{\circ}$	$0.001^{\circ}$	$0.001^{\circ}$	$0.001^{\circ}$
Monte Carlo sig.	0000 / 0000 f. Jan 60 / 000	Lower bound	0.016	0.000	0.000	0.000	0.000	0.000
(z-talleu)	9970 confidence interval	Upper bound	0.023	0.002	0.002	0.002	0.002	0.002
Mante Carlo air	Sig.		$0.010^{\circ}$	$0.001^{\circ}$	$0.001^{\circ}$	$0.001^{\circ}$	$0.001^{\circ}$	$0.001^{\circ}$
Monte Carlo sig.	000% sometimes and another some	Lower bound	0.008	0.000	0.000	0.000	0.000	0.000
(1-talled)	9970 confidence interval	Upper bound	0.013	0.001	0.001	0.001	0.001	0.001

<sup>&</sup>lt;sup>a</sup> Grouping variable: Station





<sup>&</sup>lt;sup>b</sup> Not corrected for ties.

<sup>&</sup>lt;sup>c</sup> Based on 10000 sampled tables with starting seed 2000000.



# 3 Bioaccumulation of polycyclic aromatic hydrocarbons by Arctic and temperate benthic species

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Environmental Toxicology and Chemistry, under revision







### Abstract

Increasing oil and gas activities may substantially increase chemical stress to benthic ecosystems in the Arctic. Polycyclic aromatic hydrocarbons (PAHs) can be used as indicator compounds for this increase. For temperate benthos, more information is available on the potential accumulation of PAHs, which may be useful to address Arctic species. Here we compare for the first time, the bioaccumulation of PAHs by Arctic benthic invertebrate species, with that of temperate invertebrate species. PAH biota-sediment accumulation factors (BSAFs) ranged 0.01 to 2.48 and were generally higher in the temperate bivalve (*Limecola* balthica) than in the Arctic bivalve (Macoma calcarea, ranging 0.00 – 0.21). Arctic polychaetes (Nephtys ciliata) showed higher BSAFs (0.00 – 1.79) than temperate polychaetes (Alitta virens, 0.00 – 0.34). Differences in bioaccumulation were found among Arctic and temperate species, reflecting species-specific feeding modes and traits. Bioaccumulation modelling revealed that steady state was not likely to be reached in the 28-d tests for all chemicals and organisms. Due to low numbers of individuals, most species-specific parameters carried too much uncertainty to reveal differences between Arctic and temperate species. No preference for ingestion of sediment organic matter (OM) over that of suspended OM was detected, indicating a high PAH affinity for OM. The results from the present study did not detect crucial differences in bioaccumulation between temperate and Arctic species. This means that there is no indication that data from temperate species cannot be used as a proxy for Arctic species in risk assessment.







### 1. Introduction

The Arctic has a high sensitivity to oil spill impacts and has limited capacity for natural recovery due to a very pronounced seasonality, mainly dictated by the reduction in sea ice. Sensitivity may be increased by expanding shipping and oil and gas (O&G) activities and the lack of appropriate oil spill response methods for this area [199]. Therefore, biological targets (i.e., bioindicators) for priority monitoring during all phases of O&G activities should be used, to assess, minimize and mitigate adverse effects [199]. Polycyclic aromatic hydrocarbons (PAHs) constitute a large group of hydrophobic contaminants and have been the focus of previous environmental assessments because of their potential toxicity and bioaccumulation [23,200]. Such chemical properties may be used to select appropriate bioindicators of acute and chronic effects of O&G production.

Besides knowledge about chemical concentrations in the abiotic environment, bioindicators are considered particularly useful to monitor trends in O&G related pollution, as they integrate chemical and non-chemical stress and account for in situ ecological conditions [74,201]. Arctic benthic systems are characterised by a high trophic diversity, relatively long life-span and sedentary lifestyles of species [134], which makes them suitable for monitoring purposes, particularly in areas of O&G production. A particularly useful exposure metric to define the bioindicator potential of O&G related bioaccumulative chemicals in benthic invertebrates is the biota-sediment accumulation factor (BSAF) of PAHs, which can adequately mirror the pollution state of the local Arctic ecosystem near O&G activities [202]. Compared to bioaccumulation (i.e., net result of influx (uptake) and efflux of contaminants) by temperate species, little is known on bioaccumulation of PAHs from sediment by Arctic benthic species. Cross-chemical extrapolation techniques such as read-across may help to predict PAH bioaccumulation by Arctic species in the absence of experimental data [203]. If bioaccumulation would be comparable between Arctic and temperate species with similar traits, risk assessment for the Arctic might be simplified by using temperate species data as a surrogate for Arctic species. Therefore, it is useful to explore (a) the potential for reading across Arctic and temperate species and (b) the potential for reading across species with different feeding traits and from the same region, with respect to bioaccumulation. Similarity between Arctic and temperate species however may not be self-evident. For instance, exposure times to contaminants may be longer in Arctic systems due to, for instance, Arctic species having a longer life-span or biological reaction times being generally slower in polar







biota than in temperate biota [125]. Clearance of PAHs and their metabolites may also be slower in Arctic invertebrates, since chemical elimination rates are temperature-dependent and affected by seasonal variability in lipid content [14]. Finally, seasonality in the environmental conditions in the Arctic may have a major impact on the physiology of local species [36].

The BSAF has traditionally been used as a metric to assess the bioaccumulation of contaminants from sediment and has been compared across a wide range of aquatic ecosystems [80-82]. If Equilibrium Partitioning Theory (EPT) applies, BSAF values can be expected to range between 1 and 2 [204]. This is based on the assumption that chemicals partition between biota lipids and sediment organic carbon. However, EPT does not consider the possibility of feeding as a route of uptake leading to higher than equilibrium steady-state concentrations. In order to accommodate such situations where EPT does not apply, a kinetic BSAF model was used [56,83]. Potentially, the BSAF may be a useful metric to compare effects of species traits on bioaccumulation, because it can correct for differences in chemical concentration, sediment organic matter (OM) as well as organism lipid content between the sites or species that are compared. For benthic invertebrates, of which the geographic distribution can span to (sub-)Arctic areas, we are aware of only two studies reporting BSAFs [143,Chapter 2].

The objective of the present study is to compare bioaccumulation between four Arctic and two temperate species, with different species traits, using 28-d whole-sediment tests [56,83]. An additional objective was to assess whether the results can be generalized by using a time-dependent BSAF model, previously used for describing bioaccumulation in temperate marine and freshwater benthic invertebrates [56,83]. For the first time, we explore to what extent BSAF is useful as a metric for bioaccumulation in Arctic benthic invertebrate species and for conducting a comparison between Arctic and temperate species for bioaccumulation.

### 2. Materials and Methods

Based on their feeding habits, sessility and relative abundance [144], four Arctic benthic species were selected; three bivalve species: *Astarte borealis* (Schumacher, 1817), *Macoma calcarea* (Gmelin, 1791), *Nuculana pernula* (O.F. Müller, 1771) and a polychaete species: *Nephtys ciliata* (O.F. Müller, 1776). These species are primarily abundant in Arctic climate zones, although they are not restricted to this area (Table S1). Additionally, two temperate benthic species (a bivalve: *Limecola balthica* [Linnaeus, 1758] and a polychaete: *Alitta virens* 







[Sars, 1835], formerly known as *Macoma balthica* and *Nereis virens*) were selected based on their comparable feeding habits, sessility, availability of chronic test protocols and thus the potential for a comparison with the Arctic species. We limited the present study to two temperate species because we prioritized comparing bioaccumulation among Arctic and temperate deposit feeding bivalves and polychaetes. Due to scarcity of *M. calcarea* individuals at the sampling area, a second Arctic deposit feeder (*N. pernula*) was included. Since *M. calcarea* may feed both on suspended and sediment OM, an obligate suspension feeder (*A. borealis*) was also included to estimate the contribution of suspended OM to the bioaccumulation of PAHs by *M. calcarea*.

# 2.1. Test organisms

We performed a 28-d sediment bioaccumulation experiment, with Arctic species and conditions at Kings Bay AS (Ny-Ålesund, Svalbard) between September and October 2014, here onwards named as "AE" (Arctic Experiment); and one with temperate species and conditions at Wageningen Marine Research (Yerseke, the Netherlands) between July and August 2014, here onwards named as "TE" (Temperate Experiment). Sediments in the Arctic (AE) and temperate (TE) experiments were the same for the same phylum. Emphasis was on simulating environmentally realistic PAH exposure levels, which implied use of representative species of the macrobenthic community of each climate region and field-contaminated sediments. This also implied that sometimes numbers of individuals were too low to allow for testing all sediment treatments. This is a limitation inherent to Arctic marine ecological research.

Permission for sampling of Arctic and temperate species was issued by The Governor of Svalbard and the Province of Zeeland, respectively. Sampling of Arctic species was performed along a transect from Tønsneset (79°0'22"N, 11°57'10"E) to Blomstrandhalvøya (78°59'14"N, 11°57'28"E), Kongsfjorden Bay (Svalbard, Norway), whereas *L. balthica* was collected at low tide at the Oesterdam (The Netherlands, 51°26'24"N, 4°13'16"E). *A. virens* was obtained from a professional bait farm, Topsy baits (Wilhelminadorp, the Netherlands). Arctic and temperate species were acclimatized under test conditions. Arctic species were kept in filtered natural sea water (20 μm) during 2 to 6 d without food supply, whereas in the TE experiment animals were kept in filtered natural sea water (0.2 μm) during 5 d and were fed once at the start of the acclimatization. For further details of the acclimatization of test species and sediment collection, see the Supporting Information (SI).







### 2.2. Sediment treatments preparation

Since both *A. borealis* and *M. calcarea* usually reside in sandy sediment and *N. ciliata* is generally found in sandy mud [40], two different batches of sediment were prepared after sediment collection to ensure an optimal habitat for the test species. Each contained different ratios of muddy and coarse sediment, respectively defined as < 0.5 and 0.5 – 1 mm grain size. Bivalves sediment (BS) contained two-thirds of coarse and one-third of muddy sediment, whereas polychaetes sediment (PS) contained two-thirds of muddy and one-third of coarse sediment, based on volume. Sediment was collected in the Oosterschelde estuary (The Netherlands, 51°36′13.5″N, 3°47′49.3″E) on 24 to 25 April 2014 and 2 May 2014 and used to create the sediment treatments for both the AE and TE experiments. This was done to ensure homogeneity of exposure of treatments among experiments. These sediment mixtures contain background PAH levels and therefore are referred to as the 'low' (L) treatment hereafter (BSL, PSL). PAH concentrations differed between BSL and PSL treatments (Tables S2 and S3) due to differences in their mud content and other variables creating sediment heterogeneity, and were one order of magnitude higher than in sediment previously collected at the Arctic species collection site (Blomstrandhalvøya) (Chapter 2).

From each of these two initial batches with low chemical concentration, another two sediment treatments were prepared with a higher chemical concentration, referred to as the 'medium' (M) and 'high' (H) concentration. The 'medium' treatment contained 5% (vol.) of harbour sediment (Rotterdam, the Netherlands), while the 'high' treatment contained 10% (vol.) of harbour sediment. These percentages were used (a) on a precautionary basis since neither the harbour sediment chemical concentrations nor the effect threshold for each of the species were known and (b) to be able to link non-lethal effects of chemical concentration to endpoints studied (Chapter 4). These preparations thus resulted in six sediments: three for bivalves (BSL, BSM, BSH) and three for polychaetes (PSL, PSM, PSH) (Figure S1). All sediment treatments were thoroughly mixed with a turbine mixer for approximately 15 min before storage/transport from the Netherlands to Svalbard and again right before use in the exposure experiments. Of all sediment treatments, chemical concentrations and other characteristics (i.e., dry wt. and OM content) were assessed (Tables S2 and S3). Due to logistics, storage time  $(3-7 \, ^{\circ}\text{C})$  for sediment treatments was 9 weeks longer in the AE than in the TE experiment. Prior to start of exposure, sediment treatments were allowed to stay in contact with filtered sea water in a 1:6 sediment-to-water volume ratio without aeration for 3 d and







with aeration for the following 4 d. Some experimental units in the AE experiment were aerated for 12 to 17 d because biota field sampling took place for a longer period than initially expected (the first 3 weeks since the aeration started).

# 2.3. Experimental design

Experimental designs were similar for AE and TE except for ambient temperature and photoperiod. The AE experiment was a 28-d test in a temperature controlled room at 3 °C under a photoperiod of 12 h light: 12 h dark. All planned sediment treatments were run with N. ciliata (PSL, PSM and PSH) (n = 1 - 3; Table S4). However, due to low numbers of available organisms, only two treatments (BSL and BSH) were tested with M. calcarea (n = 1 - 2) and only one treatment (BSH) was tested with M. borealis and M. pernula (n = 2 - 3; Table S4).

The TE experiment was a 28-d test in a temperature controlled room at 18 °C under a photoperiod of 16 h light: 8 h dark. All sediment treatments were tested with *A. virens* (PSL, PSM and PSH) (n = 2 - 3). Due to low availability of test organisms only two treatments were tested with *L. balthica* (BSL and BSH) (n = 1 - 3).

For statistical purposes, only measured data for BSH and PSH treatments were used due to either a low level of replication of BSL and PSM treatments (Table S4) or the heterogeneity among PSL replicates in their OM content (Tables S2 and S3). Data for all treatments were used for bioaccumulation modelling.

Individuals were randomly divided in groups of 16 to 50 individuals per experimental unit (Table S4). Animals were not fed during exposure, although sediment OM content (2-6%) and in some cases food supply during acclimatization (i.e., temperate species) prevented body [56] and lipid weight loss (Tables S5 and S6) during the experiment.

# 2.4. Endpoints

Prior to the start and at end of the experiments, organisms were allowed to depurate their guts for 24 h in filtered sea water. At the end of each experiment, animals were weighed, measured for body or shell length, dissected (e.g., bivalves) and pooled by species. Samples were stored at -20 °C for later determination of wet wt. and lipid fraction, expressed as % wet wt., and chemical concentrations (µg/kg wet wt.). Experimental units were checked daily for mortality (i.e., animals at surface and immobile after poked), and dead organisms were removed daily.





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At the start of the AE experiment only subsamples of A. borealis and N. pernula were archived, to assess the initial conditions ('Background'). This was however, not feasible for M. calcarea and N. ciliata due to the low number of individuals. At the end of the experiments, samples of each sediment treatment were stored at -20 °C for later determination of dry wt., OM content and chemical concentrations (µg/kg dry wt.). Sediment samples were also taken at the beginning of the AE experiment (Table S2). Biota and sediment samples from both experiments were shipped frozen in dry-ice to Wageningen University (Wageningen, the Netherlands) for chemical analysis.

### 2.5. Chemical analysis

Chemical analysis was conducted according to methods used by Kupryianchyk et al. [150]. The following PAHs were analysed: Phenanthrene (PHE), Anthracene (ANT), Fluoranthene (FLT), Pyrene (PYR), Benz(a)anthracene (BaA), Chrysene (CHR), Benzo(e)pyrene (BeP), Benzo(b)fluoranthene (BbF), Benzo(k)fluoranthene (BkF), Benzo(a)pyrene (BaP), Benzo(ghi)perylene (BghiP), Dibenz(a,h)anthracene (dBahA) and Indeno(1,2,3-cd)pyrene (Ind123P). PAH recoveries were 85 to 94% and 64 to 75% for biota and sediment samples, respectively. Three blanks per each batch of 30 samples were used, and values were corrected for blanks. Benzo(k)fluoranthene and benzo(a)pyrene could not be accurately determined due to overlapped retention times with an unknown compound. For further details on extraction, clean-up and instrumental analysis see the SI.

### 2.6. Data analyses

Lipid-normalised concentrations after 28 d were calculated in biota samples for all available sediment treatments. Biota-sediment accumulation factors were calculated as  $(C_{org}/f_{lip})/(C_{sed}/f_{OC})$  with  $C_{org}$  being the chemical concentration in the organism ( $\mu g/kg$  wet wt.), C<sub>sed</sub> the chemical concentration in sediment (μg/kg dry wt.), f<sub>lip</sub> the fraction of lipids in the organism based on wet weight and foc the fraction of sediment organic carbon (OC) based on loss of ignition and an OC/OM conversion ratio of 0.4 [56,83]. Lipid-normalised biota concentrations and BSAFs of BSH exposed bivalves and PSH exposed polychaetes were checked for normality with Q-Q plots and Shapiro-Wilk test and equality of variances with Levene's test. If data were normally distributed, lipid-normalised biota concentrations and BSAFs were tested for species and climate region effect with a one-way ANOVA or an independent t test, respectively, for each PAH compound. If data were non-normally







distributed, these were log-transformed, and in case a normal distribution still was not reached, lipid-normalised biota concentrations and BSAFs were tested for species and climate region effect with the Kruskal-Wallis test and Mann-Whitney U test in combination with Kolmogorov-Smirnov Z test, respectively, for each PAH compound. Significance level of an overall statistical comparison was set at  $p \le 0.05$ , while for pairwise comparisons among species or chemicals the Bonferroni and Holm's sequential Bonferroni corrections were applied, respectively. All statistical calculations were performed using SPSS version 22.

## 2.7. Bioaccumulation modelling

The usefulness of modelling bioaccumulation in invertebrate lipids was explored according to methods described in Diepens et al. [56], following previously published models (e.g., [205]). In short:

$$BSAF_t = \frac{c_{L,t=0}}{c_{OC,t=0}^{SED}} \times e^{-(k_e + k_g)t} + \frac{\frac{k_e \times K_{lip}}{\kappa_{OC}} + \alpha \times I \times [\beta + (1-\beta) \times \gamma]}{(k_e + k_g)} \times \left(1 - e^{-(k_e + k_g) \times t}\right)$$
(1)

in which  $C_{L,t=0}$  (µg × kg<sup>-1</sup>) is the measured concentration in the biota at time zero, normalised to lipid content,  $C_{OC,t=0}^{SED}$  (µg × kg<sup>-1</sup>) is the measured concentration in sediment at time zero, normalised to OC content, t (d) is time,  $k_e$  and  $k_g$  (d<sup>-1</sup>) are the rate constants for overall elimination and growth dilution,  $K_{lip}$  is the ratio  $k_w$  /  $k_e$  which equates to an apparent lipid-water partition coefficient, and  $k_w$  the rate constant for dermal uptake (L × kg<sup>-1</sup> × d<sup>-1</sup>),  $\alpha$  (-) is the chemical assimilation efficiency (assumed to be independent of food source) and  $I \geq 0$ , kgoc × kgLipids<sup>-1</sup> × d<sup>-1</sup>) represents the mass of ingested OC per unit of time and organism lipid weight,  $\beta$  (0 <  $\beta$  < 1) is the fraction of ingested OC originating from the sediment, whereas 1 –  $\beta$  is the fraction of ingested OC originating from the suspended and freshly deposited (sediment top layer) solids, and  $\gamma$  is a constant ratio between the sorption affinities for suspended OC and sediment OC ( $K_{OC}^{SS} = \gamma K_{OC}^{SED}$ ). Note that Eq. (1) models an OC based BSAF, whereas Diepens et al. [56] used a similar version yet normalised on OM. The ingested OC is thus allowed to originate partly from suspended solids (SS) from the overlying water and partly from the sediment (SED). Ingestion of multiple food items by benthic invertebrates has been modelled before in a similar manner [56,80,83].

The parametrization process and calculations of the percentage of uptake through water and of the fraction of steady state reached in the bioaccumulation test have been described before





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[56,83] and are provided in the SI. For further explanation of modelling and calculation of 90% Confidence Intervals (CIs) the reader is referred to the SI.

## 3. Results and Discussion

### 3.1. PAH background concentrations

In the Arctic bivalve *A. borealis* average  $\Sigma_{13}$ PAH concentrations when the animals were collected were 9.4 mg/kg lipid wt. This was 9 times higher than in *A. borealis* individuals collected by Szczybelski et al. (**Chapter 2**) at the same sampling area (Blomstrandhalvøya) in July 2013 (Figure S4; Table S10). Concentrations of PHE and CHR were highest (46% of the  $\Sigma_{13}$ PAH) in *A. borealis* collected in September, whereas individuals collected in July accumulated PHE and PYR (46% of the  $\Sigma_{13}$ PAH) which was similar to concentrations analysed in local sediments by Szczybelski et al. (**Chapter 2**). Changes of an order of magnitude in PAH concentrations between both years were observed for PHE, ANT, CHR and BghiP. A significant increase in CHR *A. borealis* concentration would comply with generally high 4-ring PAHs concentration in sediment surface layers from the inner areas of Kongsfjorden [206], whereas an increase in BghiP might be linked to some disturbance of deeper layers [206] and/or deposition of terrestrial petrogenic PAHs after bedrock erosion and sediment transport by Blomstrandhalvøya rivers and tributary streams during late summer [207].

In the temperate bivalve L. balthica average  $\Sigma_{13}PAH$  concentrations in individuals collected at Oesterdam (The Oosterschelde National Park, the Netherlands) were 35.5 mg/kg lipid wt. (Table S6). Concentrations of PYR and BaA were below concentrations found in L. balthica collected in the Westerschelde estuary (The Netherlands) [208]. In the temperate polychaete A. virens,  $\Sigma_{13}PAH$  concentrations were lower than in L. balthica (10.4 mg/kg lipid wt.) (Table S6). This may be expected as A. virens were obtained from an aquaculture farm and L. balthica were collected in the field.

# 3.2. Effects of Arctic species traits on PAH bioaccumulation

Lipid-normalised  $\Sigma_{13}$ PAH concentrations in Arctic invertebrates in the high treatment (BSH or PSH) were 13.1 mg/kg in *M. calcarea*, 20.2 mg/kg in *N. ciliata*, 37.7 mg/kg in *A. borealis* and 144.8 mg/kg in *N. pernula* (Table S5). The 3- to 4-ring PAHs were generally accumulated by the filter feeder *A. borealis* which agreed with a higher fraction of these compounds in







BSH sediment. The 5- to 6-ring PAHs were strongly accumulated (i.e., up to one order of magnitude higher than background concentrations) by the deposit feeder *N. pernula* which did not show this same agreement with BSH sediment composition. This disagreement might be explained by a higher PAH accumulation by *N. pernula* than the other two Arctic bivalves prior to the experiment start [209], slow PAH excretion [210] and non-equilibrium between *N. pernula* lipid tissue and sediment during the AE experiment.

The 3- to 4-ring PAHs (i.e., PHE, ANT, FLT and PYR) concentrations were on average 2.5 to 4 times higher in the BSH exposed filter feeder *A. borealis* compared to the deposit feeding bivalves (*M. calcarea* and *N. pernula*). An increase in FLT and PYR concentrations can be explained by the fact that the concentrations of these PAHs in BSH sediment were also high (Table S2). Concentrations of CHR were on average 42 times higher in BSH exposed *N. pernula* than *A. borealis* (Table S5), although CHR concentration in BSH sediment was very low (Table S2). In this case, considerably high CHR concentrations in *N. pernula* might be explained by the species ability to ingest sediment to a larger extent than the other Arctic bivalves [209] and to retain PAHs as observed by Neff et al. [210]. In the polychaete *N. ciliata*, differences in the concentration of PHE, FLT, PYR, BaA, BbF and BghiP between the low (PSL) and high (PSH) exposed individuals reflected the difference in concentrations measured in the PSL and PSH sediments (Tables S2 and S5).

No statistical differences in PAH concentrations were found between the deposit feeding bivalves *M. calcarea* and *N. pernula* in the BSH exposures. However, this conclusion should be made with caution as the number of samples available per species was low and the two species show different feeding behaviours (Table S1; [136,209,211]). Variability in metabolism of PAHs by some of our target species (*A. virens*, *L. balthica*) is well described in the literature [139], with generally higher metabolism rates in temperate polychaetes than bivalves. Although not evaluated in the present study, an increase in general metabolic activity [212,213] and a moderate production of reactive oxygen in Arctic species at the end of summer [214,215] may promote the efficiency and prevent the inhibition of PAH biotransformation, respectively. Metabolism of PAHs in Arctic species is expected to be lower than in temperate species due to low temperature and generally low food availability [73,104].

Polycyclic aromatic hydrocarbons BSAFs were generally low (i.e., BSAF <1) and higher in BSH exposed *A. borealis* than PSH exposed *N. ciliata*. This may be linked to a higher black





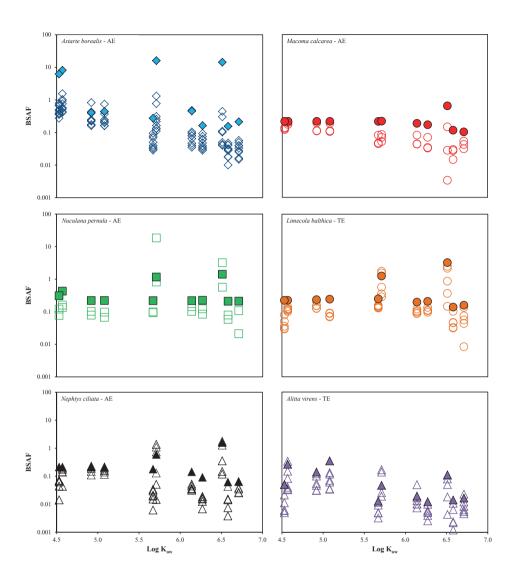


carbon (BC) and amorphous OM content in PSH sediment, which would decrease PAH bioavailability in polychaetes. Harbour sediments are known to contain more BC [53,216] than sediments from the collection site (The Oosterschelde National Park, the Netherlands) and the Rotterdam harbour area has been subject to considerable BC deposition [217]. The BSAFs for all PAHs ranged from < 0.0001 to 18, but they were often smaller than one which is considered to be caused by the strong sorption of the planar PAHs to BC [196,218]. The 10-90th percentiles of the BSAF for all PAHs in the high treatment (BSH or PSH) exposed biota ranged 0.02 to 1.13 for *A. borealis*, 0.01 to 0.21 for *M. calcarea*, 0.03 to 16.64 for *N. pernula* and 0.01 to 1.70 for *N. ciliata* (Figure 1; Table S11). The BSAFs for PHE, ANT, FLT and PYR were on average 2 to 5 times higher in *A. borealis* compared to the deposit feeding bivalves, whereas CHR BSAFs were up to 58 times higher in *N. pernula* than in the other two Arctic bivalves.









**Figure 1.** Modelled biota-sediment accumulation factors (BSAFs) (closed symbols) based on the minimum sum of squares and actual measured BSAFs (open symbols) for different polycyclic aromatic hydrocarbons in Arctic and temperate species exposed to the high treatment (BSH/PSH). AE: Arctic experiment; TE: temperate experiment.

Significantly higher BSAFs were found for ANT in BSH exposed *A. borealis* than in PSH exposed *N. ciliata* (Tables S11 and S15). Higher PAH BSAFs in bivalves compared to polychaetes may be explained by a higher BC content in PSH than in BSH sediment, which would decrease PAH bioavailability in the polychaete experiment, and by a higher OM content in PSH than in BSH sediment, which may result in a higher nutritional value of PSH

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and reduction in polychaetes feeding. Both BSH and PSH treatments presented a similar pyrogenic PAH component, which may be indicative of BC presence in both sediment treatments [151].

### 3.3. Effects of temperate species traits on PAH bioaccumulation

Lipid-normalised  $\Sigma_{13}$ PAH concentrations in the high treatment (BSH or PSH) exposed invertebrates were 32.9 mg/kg in *L. balthica* and 9.4 mg/kg in *A. virens* (Table S6). *L. balthica* accumulated PAHs to a larger extent or excreted less than the polychaete *A. virens*, yet increasing PAH concentrations between the low and high treatment were reflected better by *A. virens* body residues. Increasing PAH concentrations in clams and sediment did not correlate between the low and high treatments. Such lack of correlation might be explained by other PAH uptake routes being dominant in *L. balthica* such as dermal absorbance or suspended solids ingestion.

Concentrations of PAHs were significantly higher in the BSH exposed deposit feeder *L. balthica* than in the PSH exposed polychaete *A. virens* (Tables S6 and S17). This may be related to several factors, like higher PAH background concentrations, higher bioavailability, lower elimination rates, longer exposure times and high food selectivity in *L. balthica* [56,179].

In *L. balthica*, PAH concentrations were 2-fold (FLT, BbF, dBahA, Ind123P), 3-fold (PYR, BeP) or 4-fold (BaA) higher in BSH than BSL individuals, whereas the concentrations were 4 to 10 times higher in BSH than in BSL sediment (Tables S3 and S6). An explanation to this disproportionality between organisms and sediment may be either a low PAH uptake from sediment compared to dermal uptake [56,219] or differences in congener-specific absorption efficiencies from sediment in *L. balthica* [220].

In *A. virens*, a 2-fold increase in FLT, PYR, BeP and dBahA body residue was observed between the low and high treatment exposed individuals which was expected from the congeners increase between PSL and PSH sediment concentrations. Fluoranthene (FLT), pyrene (PYR) and benzo(e)pyrene (BeP) were also correlated between *A. virens* body residue (wet wt. basis) and sediment concentrations after 28 d of exposure to polluted sediments from Oslo (Norway) [221].







Polycyclic aromatic hydrocarbons BSAFs were higher in BSH exposed *L. balthica* than in PSH exposed *A. virens* which is likely due to differences in biota PAH background concentrations and bioavailability between sediment treatments. The 10-90th percentiles for BSAFs for all PAHs in high treatment exposed biota ranged from 0.02 to 2.36 for *L. balthica* and 0.00 to 0.25 for *A. virens* (Figure 1; Table S12). Biota-sediment accumulation factors were on average 1.5 to 24 times higher in BSH exposed *L. balthica* than in PSH exposed *A. virens*. This may be explained by a combination of different factors such as lower PAH bioavailability in *A. virens* due to a probably higher BC and other organic material content in PSH than BSH sediment [222], lower sediment ingestion by *A. virens* compared to *L. balthica* [56] and generally higher PAH biotransformation capacity in polychaetes [181].

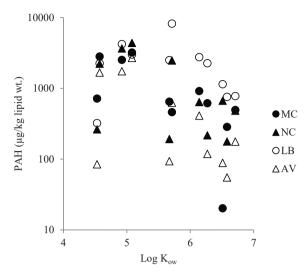
### 3.4. PAH bioaccumulation among Arctic and temperate species

Lipid-normalised PAH concentrations were generally higher in the temperate deposit feeding bivalve (*L. balthica*) than the Arctic one (*M. calcarea*), when exposed to the same sediment treatment (e.g., BSH). Only ANT concentrations were found to be 2 times higher in *M. calcarea* when compared to *L. balthica* (Figure 2; Tables S5 and S6). This may be explained by the fact that temperature affects OM-water and lipid-water partition coefficients, and thus bioaccumulation and BSAF values if other conditions are the same. Lower temperature has been found to increase PAH affinity to OM [223], and also to decrease lipid partitioning [224]. In the case of *M. calcarea*, these processes probably play a role but it was not possible to unambiguously identify the main reason for the apparent difference between the species.









**Figure 2.** Geomean polycyclic aromatic hydrocarbons (PAHs) concentrations (μg/kg lipid wt.) in Arctic and temperate species exposed to the high treatment (BSH/PSH). MC: *Macoma calcarea*; NC: *Nephtys ciliata*; LB: *Limecola balthica*; AV: *Alitta virens*.

Additionally, differences between field-sampled L. balthica and M. calcarea and differences in food selectivity may have led to differences in their general performance during the TE and AE experiments. For instance, in the case of M. calcarea, strong temperature gradients (0 – 4  $^{\circ}$ C) described at sampling depth (20 m) within a 2 month period in late summer [145,225] could have increased energy allocation to physiological maintenance in the Arctic bivalves decreasing their energy budget [38,226], whereas the absence of phytoplankton input in the experiment can cause a rapid onset of lowered metabolic rate in M. calcarea adults [136,227].

The Arctic polychaete (*N. ciliata*) showed on average 2 to 3 times higher ANT, PYR, BaA and BbF concentrations than the temperate polychaete (*A. virens*) (Figure 2; Tables S5 and S6). This may be related to the fact that food conversion can be more effective in *N. ciliata* under low temperature compared to *A. virens* [228], thus lowering energy loss and improving the Arctic polychaete's fitness. Reproductive and morphological differences between the species can also affect their feeding activity rates and PAH bioaccumulation. A <12 h photophase is known to trigger onset of sexual maturity in polychaetes and to reduce their feeding rate [229,230]. It is possible that the feeding rate for the Arctic species *N. ciliata* was reduced during the period of sampling (8 – 26 September), when the natural photophase rapidly reaches <12 h [229,231]. In the case of the temperate species *A. virens*, both farm







growing and TE conditions (photophase >12 h, 18 °C) ensured gametogenic development arrest in mature females [232].

Following this last assumption, stable feeding activity would have made a PAH assimilation peak possible, which in this case was assumed to have been reached at an earlier exposure time in the temperate polychaete *A. virens* than in the Arctic polychaete *N. ciliata*. This can be due to continuous feeding by *A. virens*, which may not only increase the contact time of the species intestinal epithelium and coelomic fluids with PAHs [233], and thus increase PAH solubilisation, but may also lead to oxidative stress and ultimately to biotransformation enzymes induction [181,234]. *Nephtyidae* species such as *N. ciliata*, usually have a much smaller gut volume than other deposit feeding polychaetes and take discrete meals as part of their carnivore diet [235]. This may to some extent restrict the species PAH absorption [236], although stable pre-oogenesis conditions (12 h light: 12 h dark) in the absence of animal food (i.e., AE experimental conditions), would make it necessary for the animals to increase their deposit feeding activity over time in order to maintain PAH tolerance mechanisms. Although we have no conclusive evidence for this explanation, this could have possibly created a difference between both species so that *A. virens* energy budget at exposure start and also its PAH bioaccumulation were higher than in *N. ciliata*.

Biota-sediment accumulation factors for 2- to 3-ring PAHs were higher in BSH exposed M. calcarea than in L. balthica which, similarly to the differences between PAH lipid-normalised concentrations, could be explained by higher waterborne PAH uptake in M. calcarea. Biota-sediment accumulation factors for PAHs were generally higher in PSH exposed N. ciliata than in A. virens. Ranges of BSAFs for all PAHs were < 0.0001 to 18, in the order M.  $calcarea \approx A$ . virens < A. vivens <







#### 3.5. Modelling PAH bioaccumulation by Arctic and temperate species

The modelled BSAFs matched well with the measured BSAF data. Confidence Intervals (CIs) for modelled BSAFs however were generally wide for most parameters and species, reflecting the variability in the biological data (Figure 1; Table S21). Complete data sets, with all treatments, were available for *N. ciliata* and *A. virens* and only data for the high treatment were present for all species. This also defines the cases for which parameters were estimated (Table S21).

The intercept b in the relation  $LogK_{OC}^{SED} = LogK_{OW} + b$ , determining the affinity of chemical partitioning to OC, was optimized to a value of 1.07 (0.81 – 1.33; 90% CI). This is higher than the well-established value of -0.21 for natural sediment [237], and can be explained from PAHs being efficiently bound to BC present in the sediment [78,197]. This is consistent with the aforementioned PAH diagnostic ratios indicating pyrogenic PAH sources, and with many measured BSAF values being smaller than one. The sorption affinity ratio  $\gamma$  was fitted and appeared indistinguishable from one, whereas the fractions of ingested sediment (parameter  $\beta$ ) had 90% CIs extending beyond parameter constraints and were overlapping among species. This implies that the present experiments did not identify a difference in sorption to suspended OM and sediment OM (i.e.,  $K_{OC}^{SS} = K_{OC}^{SED}$ ), and thus that it does not matter what type of OM is ingested. Accordingly, the fraction of ingested sediment (parameter  $\beta$ ), was set to one. This reduced the number of parameters fitted (p = 19; Table S21), which yielded narrower CIs for the remaining parameters. Hence, the present model analysis was less rigorous as those provided by Diepens et al. [56] and Sidney et al. [83] who detected significant values for  $\beta$  based on larger data sets.

In general, BSAFs decreased or remained at a constant value with increasing log K<sub>ow</sub> (Figures 1 and S5), which agrees with earlier findings [56,83]. One explanation for this, as discussed by Diepens et al. [56], may be that steady state was only reached for PHE, ANT, FLT and PYR in *M. calcarea* and *L. balthica* during the 28-d experiment (Table S22).

However, for two chemicals (CHR and BghiP) high BSAF values were measured and predicted, which was not in line with the other values and expected trends with log K<sub>ow</sub>. This can be explained by the high CHR and BghiP background concentrations in the organisms, at start of the experiment (Tables S5 and S6), and low clearance rates.







Since particle ingestion may be a dominant PAH uptake route in benthic organisms [80], ingestion rates (I) were fitted and expressed on an OM basis. In general, ingestion rates were low or even zero (Figure 3; Table S21). For Arctic species a low ingestion rate would be expected as metabolic rates are reduced and growth rates are constrained in cold-adapted stenotherms [38]. However, comparison of the temperate species L. balthica and A. virens to previously published data shows that these values are below the lower 90% CI boundary and below the range of 0.13 to 0.62 kg<sub>OM</sub> × kg<sub>Lipids</sub><sup>-1</sup> × d<sup>-1</sup> reported by Thomann et al. [238]. We speculate that the low apparent ingestion rates may be caused by a high nutritional value of the sediment, resulting in the dominance of dermal uptake.

For M. calcarea and A. borealis, the values for proportionality parameter 'a' in 'log  $k_e = -\log K_{ow} + a$ ', required to calculate the elimination rate ( $k_e$ ), were either overlapping or above previously published CIs for L. balthica, respectively (Table S21). For N. ciliata it was higher than previously published CIs for A. virens [56]. The magnitudes of 'a' for the temperate species L. balthica were higher than previously published CIs for PCBs [56,239]. For A. virens, magnitudes of 'a' were lower than previously published CIs for PCBs [56], whereas PAHs often are considered to be metabolised easier than PCBs [139,181].

The relative importance of chemical uptake pathways depends on the species, the chemical, treatment and estimated value of the ingestion parameter (Table S21). In all cases where the ingestion parameter fitted rates were zero, consequently the contribution of the OM ingestion pathway is modelled as 0%. In the high treatment (BSH or PSH), *N. pernula* and *A. virens* show 100% uptake from OM ingestion, whereas for *A. borealis*, *M. calcarea* and *N. ciliata* the model suggests 100% uptake from water (Table S23). For *L. balthica*, the chemical uptake from sediment increases with increasing log K<sub>ow</sub>. These patterns and values agree with earlier published data [56,239,240]. Although lower elimination rates and ingestion rates for Arctic species compared to their temperate counterparts may be expected, this cannot be seen from the present data and parameters. Because the present bioaccumulation data set was not large enough for model validation purposes as in previous modelling studies [56,83], we ascribe this to the higher variability and uncertainty in the present bioaccumulation data.

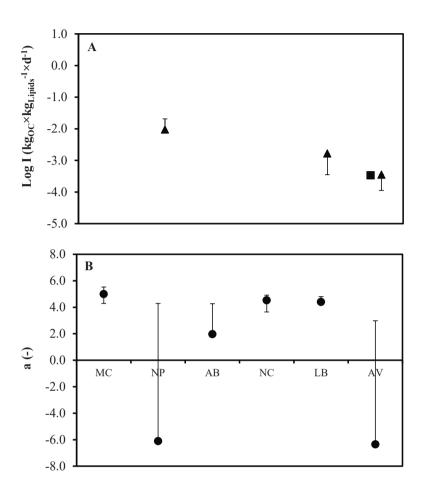






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## Chapter 3



**Figure 3.** Optimized model parameters and 90% confidence limits (CLs) for ingestion rate (I;  $kg_{OC} \times kg_{Lipids}^{-1} \times d^{-1}$ ) and intercept for elimination rate constant ( $k_e$ ) (a; -). **A:** 'I'; **B:** 'a' for species exposed to the high treatment (BSH/PSH). 'I' was fitted separately for medium (squares) and high (triangles) treatment. Parameter values are only included if a 90% CL could be assessed in at least one direction. AB: *Astarte borealis*; MC: *Macoma calcarea*; NP: *Nuculana pernula*; NC: *Nephtys ciliata*; LB: *Limecola balthica*; AV: *Alitta virens*.

### 4. Conclusions and Implications

We showed significant differences in PAH bioaccumulation among temperate species with different feeding traits, and only for ANT among Arctic species. Differences between Arctic and temperate species with similar feeding traits, were generally not significant.







Bioaccumulation of PAHs from sediment was generally higher in the Artic polychaete *N. ciliata* than in the temperate *A. virens*. On the contrary, the temperate deposit feeding bivalve *L. balthica* accumulated PAHs to a greater extent from sediment than its Arctic counterpart *M. calcarea*. Consequently, bioaccumulation metrics experimentally determined in temperate species might become too conservative in the risk assessment for similar Arctic species and in other cases too moderate.

The time-dependent BSAF model shows that model parameters did not significantly differ among species and climate regions. This is supported by the generally non-significant differences in PAH bioaccumulation between Arctic and temperate species. This implies that although modelling has been shown to be a valid tool in earlier work [56,83], insufficient data for an appropriate parametrization limited the insight provided by modelling in the present study. Field validation of bioaccumulation models, of which data assumptions are built on temperate species data, will however require Arctic standard single-species tests to be able to further characterize sediment bioaccumulation mechanisms, such as contaminant uptake and elimination routes, under local conditions.

The present study used the same sediments and conditions mimicking Arctic and temperate environmental conditions and used Arctic and temperate species with pairwise matching functional groups. Differences in bioaccumulation in a direct comparison of the two climate regions characteristics were observed, which however were not statistically significant due to considerable variability in the biological control data, and similarity in exposure routes. This means that a cautionary note should be sounded on the use of temperate benthic species as a surrogate for Arctic benthic species in bioaccumulation assessment.

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### **Supporting Information**

#### Field sampling

Sampling of Arctic species. Sampling in Kongsfjorden was mainly performed near Blomstrandhalvøya and to a lesser extent at Gluudneset and Tyskerhytta, Kongsfjorden Bay (Svalbard, Norway). Sampling took place outside the protected areas of Kongsfjorden on board of research vessel MS Teisten (Kings Bay AS, Ny-Ålesund) between 8 and 26 September 2014 and within the Outer Basin community distribution area as defined by Włodarska-Kowalczuk and Pearson [147]. Biota samples were collected using a Van Veen grab (0.1 m²). Four Arctic marine benthic invertebrate species with different feeding behaviors were collected: Astarte borealis (mollusc; suspension feeder [Schumacher, 1817]), Macoma calcarea (mollusc; surface deposit and suspension feeder [Gmelin, 1791]), Nuculana pernula (mollusc; sub-surface deposit feeder [O.F. Müller, 1771]) and Nephtys ciliata (polychaete; predator [O.F. Müller, 1776]) (Table S1).

Sampling of temperate species. Limecola balthica (mollusc; surface deposit and suspension feeder [Linnaeus, 1758]) was collected at low tide at the Oesterdam, Zeeland, the Netherlands, between 16 and 17 July 2014. *Alitta virens* (polychaete; predator and deposit feeder [Sars, 1835]) were obtained from a professional bait farm, Topsy baits, Wilhelminadorp, the Netherlands (Table S1).

#### **Acclimatization**

Arctic species were kept in glass aquaria with aerated, filtered (20 µm) and UV-treated sea water from the fjord (3 °C) for 2 to 6 d, before start of the experiment. In the case of *Nephtys ciliata*, a thin layer of sieved (1 mm Ø) sediment from the original sampling area was also included. Animals were not fed during acclimatization.

Temperate species were kept in glass aquaria with aerated filtered (0.2  $\mu$ m) sea water for 5 d and were fed once with spiked ground fish food (TetraMin) suspended in deionised water at the start of the acclimatization. In order to have sufficient food supply, 10 mg of dry food for *A. virens* and 3 mg (shell length  $\geq$  15 mm) or 1.5 mg (shell length < 15 mm) for *L. balthica* was added per individual.







Water quality variables such as pH, dissolved oxygen, salinity and temperature were measured every two days during exposure time (Tables S7 and S8).

#### **Sediment collection**

A moderately contaminated sub-toxic marine sediment was prepared by mixing a non-contaminated 'clean' sediment with a naturally contaminated harbour sediment. Batches of muddy sediment (132 L) and coarse sediment (176 L) were collected near Wissenkerke (The Oosterchelde Natural Park, the Netherlands) at low tide between 24 and 25 April 2014 and 2 May 2014. Coarse sediment was sieved in the field with a 1-mm sieve whereas muddy sediment was sieved at laboratory inside a 1-mm sieve with some filtered sea water (0.2  $\mu$ m). Once sieved both samples were stored at 3 °C.

A batch of harbour sediment (30 L) was collected near Rotterdam Nesserdijk by Nautisch Service Centrum (Rotterdam, the Netherlands) in mid-April 2014 and sieved with a 1-mm sieve on 1 May 2014. A subsample of the sieved wet harbour sediment (800 mL) was sent to Wageningen Marine Research laboratories (IJmuiden, the Netherlands) for further chemical analyses.

### **Test species**

Arctic experiment. A. borealis, M. calcarea, N. pernula and N. ciliata were obtained as described in the field sampling section. A. borealis randomly selected healthy individuals with an average and standard deviation (SD) shell length of 31.30 (4.63) mm and wet wt. (flesh) of 7.47 (4.16) g were used. N. pernula randomly selected healthy individuals with an average (SD) shell length of 14.60 (3.66) mm and accumulated (n = 16 - 17) wet wt. (flesh) of 7.38 (1.65) g were used. As for M. calcarea and N. ciliata, we did not measure any morphological parameters due to the scarcity of individuals. Animals were not fed during the acclimatization, neither during the exposure experiments.

Temperate experiment. L. balthica and A. virens were obtained as described in the field sampling section. L. balthica randomly selected healthy individuals with an average (SD) shell length of 13.49 (2.46) mm and wet wt. (shell + flesh) of 0.47 (0.31) g were used. A. virens randomly selected healthy individuals with an average (SD) length of 10.83 (1.70) cm and wet wt. of 2.75 (0.98) g were used. Animals were fed once at the beginning of the acclimatization.





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#### Chemical analysis

Subsamples of each sediment sample were analyzed for weight percentages of organic matter (OM) content, measured as loss on ignition (550 °C, 3 h). Biota lipids were extracted with chloroform:methanol:water and quantified gravimetrically.

Frozen sediment samples were thawed at room temperature. Frozen pooled biota samples were thawed at room temperature and subdivided into replicates of equal weight before drying. Sediment subsamples and biota replicates were dried and grinded with diatomaceous earth and Soxhlet-extracted for 40 min with hexane:acetone (3:1, v/v) (DionexASE 350 System, Thermo Scientific). Extracts were cleaned-up over an Al<sub>2</sub>O<sub>3</sub> column and used for polycyclic aromatic hydrocarbons (PAHs) analysis, which was carried out on an HP 1100 HPLC equipped with a 4.6 mm Vydac guard and analytical reverse phase C18 column (201GD54T and 201TP54, respectively) which were kept at 22.00 °C. Detection was performed by an HP 1100 multi-wavelength fluorescence detector operating in the multi-emission wavelength mode. The mobile phase consisted of methanol/water (mixture and flow gradient). The injection volume was 20 μL.

### **Bioaccumulation modelling**

Bioaccumulation modelling was used to link observed bioaccumulation with species-specific traits. Bioaccumulation in invertebrate lipids was modelled according to Diepens et al. [56] and is briefly described here:

$$BSAF_{t} = \frac{c_{L,t=0}}{c_{OC,t=0}^{SED}} \times e^{-(k_{e}+k_{g})\times t} + \frac{\frac{k_{e}\times K_{lip}}{K_{OC}^{SED}} + \alpha \times I \times [\beta + (1-\beta)\times \gamma]}{(k_{e}+k_{g})} \times \left(1 - e^{-(k_{e}+k_{g})\times t}\right)$$
(S1)

in which  $C_{L,t=0}$  (µg × kg<sup>-1</sup>) is the measured concentration in the biota at time zero, normalised to lipid content,  $C_{OC,t=0}^{SED}$  (µg × kg<sup>-1</sup>) is the measured concentration in sediment at time zero, normalised to OC content, t (d) is time,  $k_e$  and  $k_g$  (d<sup>-1</sup>) are the rate constants for overall elimination and growth dilution,  $K_{lip}$  is the ratio  $k_w/k_e$  which equates to an apparent lipid-water partition coefficient, and  $k_w$  the rate constant for dermal uptake (L × kg<sup>-1</sup> × d<sup>-1</sup>),  $\alpha$  (-) is the chemical assimilation efficiency (assumed to be independent of food source) and I ( $\geq$  0, kg<sub>OC</sub> × kg<sub>LIP</sub> × d<sup>-1</sup>) represents the mass of OC ingested per unit of time and organism lipid weight,  $\beta$  (0 <  $\beta$  < 1) is the fraction of ingested OC originating from the sediment whereas 1 –  $\beta$  is the fraction of ingested OC originating from the suspended and freshly deposited (sediment top







layer) solids and  $\gamma$  is a constant ratio between the sorption affinities for suspended OC and sediment OC ( $K_{OC}^{SS} = \gamma K_{OC}^{SED}$ ). The ingested OC thus is assumed to originate partly from suspended solids (SS) from the overlying water and partly from the sediment (SED). Ingestion of multiple food items by benthic invertebrates has been modelled in a similar manner [56.80,83]. A detailed explanation of Eq. S1 is provided below.

The percentage of uptake through water is calculated based on Eq. S1 as:

$$\%WaterUptake = \frac{k_e}{k_e + \alpha I [\beta + (1 - \beta)\gamma] K_{OC}^{SED} / K_{lip}}$$
(S2)

The fraction of steady state reached ( $F_{SS}$ ,  $0 < F_{SS} < 1$ ) in the bioaccumulation test (t = 28 d) was calculated as:

$$F_{SS} = 1 - e^{-(k_e + k_g)t} ag{S3}$$

### Model parametrization

Eq. S1 was implemented in Microsoft Excel and the model was fitted to the log-transformed experimental BSAF data using the Excel Solver tool with scaling of parameters and a relative least-squares criterion. The first term in Eq. S1 ( $C_{L,t=0}$ ) is omitted when concentrations in organism lipids at time zero are below limit of detection. The sediment OC-water partition coefficient  $K_{OC}^{SED}$  was assumed proportional to  $LogK_{OW}$ :  $LogK_{OC}^{SED} = LogK_{OW} + b$  [241,242]. The parameter  $k_g$  could be obtained from either soft tissue weight differences between t = 28 d and t = 0 d for A. borealis (0.032), N. pernula (0.001), L. balthica (-0.001) and A. virens (-0.019), or estimated from previously calculated yearly growth rates (mm/year) [56,243] for M. calcarea (0.002) and N. ciliata (-0.001). The parameters  $k_e$ , I,  $\beta$  and  $\gamma$  were estimated using a two-stage iterative approach [83]. First, the chemical sorption parameters b and  $\gamma$  were set at default literature values [175,241,242], and the species-specific parameters a, I (constrained: I  $\geq 0$ ) and  $\beta$  (constrained:  $0 < \beta < 1$ ) were optimized for each species separately by minimizing their individual sum of squares. Subsequently, the parameters b and  $\gamma$  were optimized by minimizing the total sum of squares, after which the parameters a, I and  $\beta$  were fitted again for each of the species. This procedure was repeated until all minimum sums of squares had been reached.

Confidence intervals (90% CIs) were calculated according to Draper and Smith [244]:





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$$SS_{90} = SS_{min} \left( 1 + \frac{p}{(n-p)} F(p, n-p, 90\%) \right)$$
 (S4)

in which  $SS_{90}$  is the sum of squares at the 90% confidence contour,  $SS_{min}$  is the minimum sum of squares, n is the number of BSAF measurements (n = 479), p is the number of estimated parameters (p = 19) and F (p, n-p, 90%) is the F-distribution according to Fisher. Confidence intervals were estimated using n, p and F either for the whole data set for the general parameters or the species-specific data set. Negative confidence limits for I were set to zero. Since the present experiments did not identify a difference in sorption to suspended OC and sediment OC (i.e.,  $K_{OC}^{SS} = K_{OC}^{SED}$ ), the fraction of ingested sediment ( $\beta$ ), was set at one. An overview of all model parameters is provided as Table S9.







Table S1. Overview of target species and their ecological traits.

Experiment	Species	Geographical distribution	Feeding mode	Picture
	Astarte borealis (Schumacher, 1817)	Chukchi Sea <sup>a</sup> Barents Sea <sup>b</sup> Baltic Sea <sup>c,d</sup> Temperature: -2 - 16 °C	Suspension feeder °	
Arctic	Macoma calcarea (Gmelin, 1791)	Bering Sea <sup>f</sup> Chukchi Sea <sup>a</sup> Barents Sea <sup>b</sup> Kara Sea <sup>g</sup> Baltic Sea <sup>d</sup> Temperature: -2 – 12 °C	Surface deposit and suspension feeder °	
	Nuculana pernula (O.F. Müller, 1771)	Bering Sea <sup>f</sup> Barents Sea <sup>b</sup> Temperature: -2 – 9 °C	Sub-surface deposit feeder <sup>h</sup>	
	Nephtys ciliata (O.F. Müller, 1776)	Bering Sea <sup>1</sup> Barents Sea <sup>b</sup> Baltic Sea <sup>c,d,j</sup> Temperature: -1 – 14 °C	Predator °	
Temnerate	Limecola balthica (Linnaeus, 1758)	Extensive distribution in the northern hemisphere $^k$ Temperature: $0-25^{\circ}\mathrm{C}$	Surface deposit and suspension feeder <sup>k,l</sup>	
	Alitta virens (Sars, 1835)	Cosmpolitan, often found in areas of chronic oil pollution $^{\rm m}$ Temperature: +3 - 12 $^{\circ}$ C	Predator and deposit feeder no	

 ${}^{a}[114], {}^{b}[245], {}^{c}[166], {}^{d}[246], {}^{c}[136], {}^{f}[247], {}^{g}[248], {}^{h}[249], {}^{i}[250], {}^{i}[251], {}^{k}[240], {}^{l}[252], {}^{m}[253], {}^{n}[253], {}^{n}[254], {}^{c}[255], {}^{h}[251], {}^$ 







Table S2. Polycyclic aromatic hydrocarbons concentrations (mean ± SD; µg/kg dry wt.) in sediment treatments at the start of the Arctic experiment (AE). Concentrations at AE end (t = 28 d) were assumed to be close to the experiment start values.

Treatment	BSL	BSH	PSL	PSM	PSH
OM (% dry wt.)	$2.05 \pm n.a.$	$2.52 \pm 0.03$	$6.02 \pm n.a.$	3.89 ± n.a.	3.76 ± n.a.
PHE	$48.11 \pm 8.46$	$164.69 \pm 48.81$	$155.03 \pm 26.35$	$159.71 \pm 15.59$	$250.26 \pm 29.11$
ANT	$8.52 \pm 1.25$	$61.62 \pm 14.90$	$34.09 \pm 3.75$	$49.27 \pm 9.77$	$81.30 \pm 8.37$
FLT	$58.83 \pm 11.28$	$306.13 \pm 128.02$	$249.20 \pm 35.11$	$288.08 \pm 58.13$	$406.93 \pm 52.78$
PYR	$42.94 \pm 5.80$	$240.91 \pm 109.12$	$180.19 \pm 24.54$	238.76 ± 75.43	$313.86 \pm 38.50$
BaA	$15.45 \pm 3.14$	$148.82 \pm 70.13$	$83.17 \pm 20.31$	$121.87 \pm 54.17$	$156.96 \pm 18.12$
CHR	n.d.	$92.95 \pm 59.11$	$19.64 \pm 11.43$	$80.57 \pm 41.76$	$170.90 \pm 110.17$
BeP	$25.42 \pm 7.53$	$209.65 \pm 88.39$	$128.04 \pm 19.40$	$216.94 \pm 117.22$	$251.10 \pm 31.37$
BbF	$30.44 \pm 6.45$	$174.13 \pm 69.34$	$145.95 \pm 16.18$	$166.15 \pm 32.69$	$234.81 \pm 30.82$
BghiP	$0.80 \pm 0.65$	$11.45 \pm 7.23$	$8.17 \pm 2.22$	$13.46 \pm 5.70$	$18.84 \pm 6.94$
dBahA	$32.09 \pm 12.89$	$146.19 \pm 58.01$	$129.82 \pm 17.92$	$140.35 \pm 27.26$	$203.35 \pm 25.71$
Ind123P	$24.12 \pm 3.57$	$152.92 \pm 60.80$	$150.68 \pm 20.01$	$154.70 \pm 32.43$	$218.81 \pm 39.02$
Σ <sub>13</sub> PAH	$286.72 \pm 58.50$	$1709.46 \pm 709.16$	$1283.99 \pm 184.63$	$1629.86 \pm 470.15$	$2307.13 \pm 162.56$

BS: bivalves sediment; PS: polychaetes sediment; L: low treatment; M: medium treatment; H: high treatment; n.d.: not detected; n.a.: not available.

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**Table S3.** Polycyclic aromatic hydrocarbons concentrations (mean  $\pm$  SD;  $\mu$ g/kg dry wt.) in sediment treatments at the end of the temperate experiment (t = 28 d).

Treatment	BSL	BSH	TSd	PSM	PSH
OM (% dry wt.)	$2.18 \pm 0.06$	$2.73 \pm 0.76$	$3.52 \pm 0.04$	$4.01 \pm 0.04$	$4.00 \pm 0.04$
PHE	$53.71 \pm 4.21$	$229.66 \pm 42.90$	$159.69 \pm 82.88$	$278.72 \pm 98.45$	$381.60 \pm 258.88$
ANT	$14.43 \pm 2.23$	$98.66 \pm 48.61$	$39.03 \pm 21.19$	$80.80 \pm 35.04$	$128.92 \pm 85.59$
FLT	$80.04 \pm 9.79$	$460.54 \pm 196.09$	$316.06 \pm 156.73$	$430.07 \pm 125.39$	$667.36 \pm 280.22$
PYR	$60.09 \pm 7.37$	$349.31 \pm 149.85$	$254.04 \pm 146.96$	$321.96 \pm 101.17$	$509.52 \pm 215.43$
BaA	$21.79 \pm 3.26$	$227.37 \pm 130.50$	$140.35 \pm 122.63$	$158.02 \pm 32.66$	$256.14 \pm 70.35$
CHR	n.d.	$158.98 \pm 125.12$	$220.13 \pm 340.36$	$113.51 \pm 38.92$	$207.44 \pm 127.47$
BeP	$34.62 \pm 6.50$	$352.06 \pm 195.45$	$205.57 \pm 141.23$	$257.50 \pm 74.64$	$479.18 \pm 250.18$
BbF	$42.21 \pm 5.79$	$255.53 \pm 116.72$	$168.62 \pm 85.51$	$243.55 \pm 63.64$	$387.37 \pm 169.03$
BghiP	$4.26 \pm 3.91$	$19.16 \pm 10.47$	$19.88 \pm 23.15$	$16.31 \pm 4.89$	$42.54 \pm 25.68$
dBahA	$49.43 \pm 11.12$	$212.64 \pm 82.18$	$127.75 \pm 53.39$	$232.47 \pm 82.85$	$352.82 \pm 194.89$
Ind123P	$39.37 \pm 9.81$	$205.33 \pm 88.24$	$136.06 \pm 67.36$	$204.31 \pm 54.36$	$374.39 \pm 202.59$
Σ <sub>13</sub> PAH	$400.01 \pm 52.03$	$2569.23 \pm 1182.50$	$1787.18 \pm 1103.86$	$2337.23 \pm 698.72$	$3787.26 \pm 1880.29$
		;			

BS: bivalves sediment; PS: polychaetes sediment; L: low treatment; M: medium treatment; H: high treatment. n.d.: not detected.





**Table S4.** Overview of the number of experimental units (n) per species and sediment treatment.

	PSH					PC	
	PS				6.1		(,,
	PSL PSM				1		2
	PSL				1		2
Treatment $(n)^a$	BSH	3	2	2		3	
T	BSL	n.t.	1	n.t.		1	
	Background	3	n.t.	1	n.t.	3	3
	$Ind./EU^b$	25	40-50	40-50	20	50	16
	Species	Astarte borealis	Macoma calcarea	Nuculana pernula	Nephtys ciliata	Limecola balthica	Alitta virens
	Experiment		A	Arcac		H	ı emperate

<sup>a</sup> Sediment treatments with different levels of PAH contamination are referred as BSL: Low Bivalve Sediment; BSH: High Bivalve Sediment; PSL: Low Polychaete Sediment; PSM: Medium Polychaete Sediment; PSH: High Polychaete Sediment. 'Background' refers to pooled field samples in which individuals were allowed to depurate for 24 h in filtered sea water before storage.

<sup>b</sup> Experimental unit' (EU) refers to a sediment treatment replicate or aquarium in which a specific number of individuals of a single species are exposed during 28 d to the corresponding sediment treatment, allowed to depurate for 24 h in filtered sea water and pooled for analytical treatment. n.t.: not tested.

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Table S5. Polycyclic aromatic hydrocarbons (mean ± SD; µg/kg lipid wt.) in test species from the Arctic experiment.

Spp.	Масота	Macoma calcarea	$Ne_{l}$	Nephtys ciliata	iata	Astarte borealis	borealis	Nucuh	Nuculana pernula
Treatment	BSL	BSH	PSL	PSM	PSH	Background	BSH	Background	BSH
и	1	2	1	-	3	3	3	1	2
Lipid wt. (% wet wt.)	0.82	0.84	0.39	0.41	0.45	0.42	0.46	1.71	1.70
PHE	$3566 \pm 393$	$2821 \pm 287$	$1402 \pm 681$	729	$2462 \pm 941$	9697 ± 1482	$11746 \pm 4998$	2457	$2792 \pm 324$
ANT	$68 \pm 068$	718 ± 48	$336 \pm 143$	149	$297 \pm 120$	$2384 \pm 442$	2977 ± 957	359	$686 \pm 216$
FLT	$1612 \pm 178$	$3329 \pm 1174$	$2353 \pm 788$	1771	$4468 \pm 969$	$1376 \pm 280$	$8504 \pm 4206$	374	$3192 \pm 837$
PYR	$1314 \pm 87$	$2623 \pm 930$	$1713 \pm 530$	1384	$3758 \pm 920$	$1119 \pm 120$	$6926 \pm 3575$	346	$2828 \pm 531$
BaA	n.d.	$673 \pm 240$	$113 \pm 70$	77	$218 \pm 104$	n.d.	.p.u	129	1847 ± 70
CHR	$7434 \pm 6614$	470 ± 113	$1327 \pm 1709$	0.15	$6425 \pm 6647$	$6164 \pm 1276$	$2926 \pm 2658$	0986	$122550 \pm 158586^*$
BeP	$316 \pm 0$	957 ± 374	$427 \pm 178$	298	$650 \pm 129$	$884 \pm 169$	$2066 \pm 2300$	183	$3222 \pm 761$
BbF	$175 \pm 30$	$658 \pm 305$	117 ± 78	100	$228 \pm 69$	223 ± 56	$845 \pm 269$	218	$2200 \pm 586$
BghiP	$1162 \pm 745$	50 ± 65	$344 \pm 454$	n.d.	$1106 \pm 955$	$1409 \pm 2154$	.p.u	1192	$2953 \pm 2925$
dBahA	275 ± 2	$506 \pm 133$	$166 \pm 76$	40	n.d.	n.d.	$479 \pm 233$	296	$1173 \pm 1112$
Ind123P	$69 \pm 40$	$298 \pm 102$	$34 \pm 37$	n.d.	n.d.	$205 \pm 144$	$421 \pm 187$	113	$1307 \pm 283$
S <sub>13</sub> PAH	16933 ± 6714	16933 ± 6714 13104 ± 3552	792 ± 283	4550	20242 ± 9762	8333 ± 267 4550 20242 ± 9762 34245 ± 20336 37668 ± 17371	37668 ± 17371	15526	15526 144750 ± 156791

BS: bivalves sediment; PS: polychaetes sediment; L: low reatment; M: medium treatment; H: high treatment. \* Individual data are 10413 and 234688 µg/kg lipid wt. n.d.: at least one sample shows a concentration below the limit of detection.

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 $\textbf{Table S6. Polycyclic aromatic hydrocarbons (mean} \pm SD; \ \mu g/kg \ lipid \ wt.) \ in \ test \ species \ from \ the \ temperate \ experiment.$ 

Spp.	Lime	Limecola balthica	hica		Alitta virens	irens	
Treatment	Background	BSL	BSH	Background	PSL	PSM	PSH
n	33		3	ю	2	2	33
Lipid wt. (% wet wt.)	0.85	0.84	0.78	1.30	0.97	1.71	1.39
PHE	$2499 \pm 571$	2849	$2257 \pm 165$	$2497 \pm 798$	$2267 \pm 1376$	$1324 \pm 496$	$2278 \pm 2000$
ANT	$225 \pm 60$	262	$334 \pm 102$	$136 \pm 64$	$108 \pm 70$	$100 \pm 27$	$113 \pm 96$
FLT	$1431 \pm 394$	2853	$3061 \pm 685$	$5421 \pm 1780$	$1596 \pm 598$	$1824 \pm 664$	$3151 \pm 1903$
PYR	$1484 \pm 325$	3513	$4340 \pm 1153$	$1611 \pm 530$	$893 \pm 379$	$1157 \pm 419$	$2034 \pm 1254$
BaA	$680 \pm 177$	1101	$2547 \pm 484$	n.d.	$83 \pm 62$	$134 \pm 19$	$104 \pm 51$
CHR	86 ± 88	964	$11585 \pm 9242$	$268 \pm 278$	n.d.	$887 \pm 1061$	n.d.
BeP	$937 \pm 247$	1618	$2833 \pm 702$	$125 \pm 50$	$249 \pm 220$	$279 \pm 102$	$475 \pm 286$
BbF	$938 \pm 259$	1442	$2295 \pm 432$	$51 \pm 26$	$100 \pm 88$	$151 \pm 26$	$125 \pm 45$
BghiP	$2216 \pm 4746$	47	$1754 \pm 1454$	99 <del>=</del> 9 <i>L</i>	n.d.	$134 \pm 134$	n.d.
dBahA	717 ± 491	702	$1083 \pm 818$	$83 \pm 29$	$115 \pm 79$	$137 \pm 36$	$239 \pm 238$
Ind123P	$452 \pm 206$	448	$839 \pm 473$	n.d.	62 ± 89	96 ± 56	$75 \pm 62$
V.,PAH	35512 + 56664	15799	35517 + 56664 15700 32028 + 13246 10360 + 3000 8005 + 5818 6223 + 2220 0371 + 5430	10360 + 3000	8005 + 5818	0000 + 2009	0371 + 5430

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Table S7. Mean water quality parameters in the Arctic experiment H treatment aquaria.

			Astarte borealis - BSH	s - BSH			W	Macoma calcarea - BSH	a - BSH			~	Nuculana pernula - BSH	a - BSH	
Day	Hd	D.O. (%)	D.O. (mg/L)	Salinity (g/L)	T(°C)	Hd	D.O. (%)	D.O. (mg/L)	Salinity (g/L)	T (°C)	Hd	D.O. (%)	D.O. (%) D.O. (mg/L)	Salinity (g/L)	T (°C)
_	8.1	82.7	9.0	34.7	2.7	7.9	81.0	8.8	35.6	2.9	7.9	80.5	8.8	35.9	3.1
7	8.1	81.3	8.8	34.7	3.0	7.9	82.5	8.9	35.7	2.9	7.9	82.0	8.8	36.0	3.1
3	8.0	82.3	8.9	34.8	2.7	8.0	84.0	9.0	35.8	2.9	8.0	84.5	0.6	36.2	3.1
4	7.9	83.1	9.0	34.9	2.7	7.9	85.0	9.2	35.8	2.9	7.9	82.0	8.8	36.3	3.1
w	7.9	83.5	0.6	34.9	2.6	7.9	85.5	9.3	35.9	2.9	7.9	82.0	8.9	36.5	3.1
9	7.8	83.7	9.1	35.0	2.6	7.8	81.5	8.7	35.7	2.9	7.8	83.0	8.9	36.1	3.1
r-	7.9	85.0	9.2	35.1	5.6	7.9	83.5	9.1	36.0	2.9	7.9	81.5	8.8	36.5	3.0
<b>∞</b>	7.8	84.4	9.1	35.2	2.7	8.0	86.0	9.4	35.8	2.8	8.0	88.5	7.6	36.0	3.0
6	7.8	83.8	9.0	35.2	2.7	n.a.	82.0	9.1	36.0	2.9	n.a.	83.5	9.1	36.3	3.0

			Nephtys ciliata - PSH	SH	
Day	Hd	D.O. (%)	D.O. (mg/L)	Salinity (g/L)	(°C)
1	8.0	81.2	8.7	35.1	2.9
2	8.0	81.5	8.8	35.2	2.8
3	8.0	82.5	8.9	35.2	2.8
4	8.0	82.8	8.9	35.3	2.8
w	7.9	82.3	8.9	35.5	2.8
9	7.9	82.7	8.9	35.5	2.8
7	8.0	81.7	8.8	35.7	2.9
<b>∞</b>	8.0	84.8	9.2	35.8	2.8
6	8.0	85.3	9.0	35.8	2.8

PSH: High Polychaetes Sediment containing 10% (vol.) of harbour sediment; D.O.: dissolved oxygen.





Table S8. Mean water quality parameters in the temperate experiment H treatment aquaria.

		1	imecola balthic	a - BSH				Alitta virens -	PSH	
Day	pН	D.O. (%)	D.O. (mg/L)	Salinity (g/L)	T (°C)	pН	D.O. (%)	D.O. (mg/L)	Salinity (g/L)	T (°C)
1	8.2	83.8	6.6	32.8	17.8	8.1	79.4	6.2	33.8	18.1
2	n.a.	80.3	6.3	n.a.	17.8	n.a.	77.9	6.1	n.a.	18.1
3	n.a.	79.3	6.2	n.a.	17.7	n.a.	75.3	5.9	n.a.	17.9
4	8.2	80.8	6.2	33.3	18.0	8.1	76.7	5.9	34.4	18.2
5	8.2	82.0	6.4	33.5	18.1	8.2	78.9	6.1	34.0	18.2
6	8.2	82.2	6.4	33.6	17.9	8.2	80.7	6.2	34.5	18.0
7	8.3	84.2	6.5	34.1	17.8	8.2	80.9	6.2	35.2	18.0
8	8.2	84.2	6.5	34.5	17.8	8.2	81.6	6.3	35.1	18.0
9	8.3	86.0	6.6	34.7	18.1	8.2	79.7	6.0	35.5	18.3
10	8.3	85.2	6.6	35.0	17.9	8.2	82.1	6.3	34.8	18.0
11	8.3	85.3	6.6	35.4	17.9	8.2	82.4	6.3	35.2	18.0
12	8.3	84.5	6.5	35.7	17.9	8.2	81.4	6.2	35.7	17.9

BSH: High Bivalves Sediment containing 10% (vol.) of harbour sediment; PSH: High Polychaetes Sediment containing 10% (vol.) of harbour sediment; D.O.: dissolved oxygen; n.a.: not available.







**Table S9.** Parameters for the biota-sediment accumulation factor model.

Parameter	Symbol	Unit
Affinity of chemical partitioning to peat in the relation $LogK_{OC}^{SED} = LogK_{OW} + b$	b	-
Biota-sediment accumulation factor	$BSAF_t$	-
Chemical assimilation efficiency	α	-
Concentration in invertebrate lipids	$C_L$	$\mu g \times kg^{\text{-}1} \ lipids$
Concentration in water	$C_W$	$\mu g \times L^{-1}$
Concentration in sediment	$C_{OC}^{SED}$	$\mu g \times kg^{-1} OC$
Concentration in suspended solids	$C_{OC}^{SS}$	$\mu g \times kg^{-1} OC$
Constant ratio between sorption affinities for suspended OC and sediment OC in the relation $K_{OC}^{SS} = \gamma K_{OC}^{SED}$	γ	-
Dermal uptake rate constant	$k_w$	$L\times kg^{\text{-}1}\times d^{\text{-}1}$
Elimination rate constant	$k_e$	d-1
Food ingestion rate	I	$kg_{OC} \times kg_{Lipids}^{-1} \times d^{-1}$
Fraction of ingested OC originating from the sediment	β (0<β<1)	-
Growth rate constant	$k_g$	d <sup>-1</sup>
Lipid water partition coefficient $(k_w/k_e)$	$K_{lip}$	$L \times kg^{-1}$
Sediment water partition coefficient	$K_{OC}^{SED}$	L × kg <sup>-1</sup>
Species-specific elimination parameter in the relation $LogK_e = -LogK_{ow} + a$	a	-
Suspended solids water partition coefficient	$K_{OC}^{SS}$	$L \times kg^{-1}$
Time	t	d







Table S10. Astarte borealis background polycyclic aromatic hydrocarbons (PAHs) concentrations (mean  $\pm$  SD;  $\mu$ g/kg lipid wt.).

PAH	2013 <sup>a</sup>	2014
PHE	$193.52 \pm 169.41$	$2708.86 \pm 341.85$
ANT	$45.25 \pm 93.27$	$664.10 \pm 91.68$
FLT	$96.07 \pm 67.43$	$385.65 \pm 80.61$
PYR	$297.78 \pm 198.32$	$313.13 \pm 30.09$
BaA	$18.78 \pm 21.82$	n.a.
CHR	$106.90 \pm 74.79$	$4614.76 \pm 5290.49$
BeP	$96.95 \pm 56.92$	$245.85 \pm 32.57$
BbF	$58.61 \pm 32.37$	$62.18 \pm 14.44$
BkF	$34.57 \pm 25.36$	n.a.
BaP	$14.56 \pm 19.59$	n.a.
BghiP	$23.65 \pm 30.32$	$393.05 \pm 616.28$
dBahA	$13.09 \pm 21.49$	n.a.
Ind123P	$75.70 \pm 68.40$	$56.89 \pm 39.89$
$\sum_{13}$ PAH	$1075.43 \pm 684.05$	$9444.47 \pm 6537.91$

a Individuals collected in July 2013 by Szczybelski et al. (Chapter 2) belong to size classes 2 to 4 as described by the authors (36 to ≥ 40 mm shell length), whereas individuals from 2014 collected by the present study averaged  $31.3 \pm 4.63$  mm (shell length). Both groups were collected in different sampling years near Blomstrandhalvøya, Kongsfjorden Bay (Svalbard, Norway).

n.a.: not available.

Table S11. High treatment (BSH) polycyclic aromatic hydrocarbons biota-sediment accumulation factors (BSAFs) percentile ranges for Arctic (Astarte borealis, Macoma calcarea and Nuculana pernula) and temperate (Limecola balthica) bivalves.

Spp.	Astarte b	orealis	Масота с	alcarea	Limecola	balthica	Nuculana	pernula
BSAF percentile	10th	90th	10th	90th	10th	90th	10th	90th
PHE	0.43	1.13	0.18	0.21	0.10	0.13	0.14	0.16
ANT	0.34	0.72	0.12	0.14	0.03	0.07	0.08	0.12
FLT	0.16	0.50	0.11	0.17	0.07	0.09	0.07	0.09
PYR	0.17	0.51	0.11	0.18	0.13	0.19	0.08	0.10
BaA	0.03	0.08	0.04	0.07	0.13	0.15	0.09	0.10
CHR	0.13	0.88	0.06	0.08	0.31	1.67	2.57	16.64
BeP	0.04	0.18	0.04	0.07	0.09	0.11	0.11	0.14
BbF	0.03	0.08	0.03	0.06	0.10	0.13	0.09	0.12
BghiP	0.04	0.36	0.01	0.12	0.23	2.36	0.83	2.94
dBahA	0.02	0.05	0.03	0.05	0.02	0.11	0.03	0.10
Ind123P	0.02	0.04	0.02	0.03	0.03	0.08	0.06	0.08







**Table S12.** High treatment (PSH) polycyclic aromatic hydrocarbons biota-sediment accumulation factors (BSAFs) percentile ranges for Arctic (*Nephtys ciliata*) and temperate (*Alitta virens*) polychaetes.

Spp.	Nephtys	ciliata	Alitta	virens
BSAF percentile	10th	90th	10th	90th
PHE	0.09	0.19	0.03	0.25
ANT	0.03	0.07	0.01	0.04
FLT	0.13	0.20	0.04	0.13
PYR	0.13	0.22	0.03	0.11
BaA	0.01	0.03	0.00	0.01
CHR	0.03	1.23	0.02	0.16
BeP	0.03	0.05	0.01	0.03
BbF	0.01	0.02	0.00	0.01
BghiP	0.13	1.70	0.01	0.10
dBahA	0.03	0.05	0.01	0.02
Ind123P	0.01	0.03	0.00	0.01

**Table S13.** Statistical test for differences in lipid-normalised polycyclic aromatic hydrocarbons concentrations between Arctic species exposed to the high treatment (Kruskal Wallis test *p*-value).

Spp.	PHE	ANT	FLT	PYR	CHR	BeP	BbF
Macoma calcarea							
Nephtys ciliata							
Astarte borealis	0.110	0.046	0.088	0.080			
Nuculana pernula					0.091	0.075	0.054







**Table S14.** Statistical test for differences in lipid-normalised polycyclic aromatic hydrocarbons concentrations between Arctic bivalve species exposed to the high treatment (Kruskal Wallis test *p*-value).

Spp.	PHE	ANT	FLT	PYR	CHR	Ind123P
Macoma calcarea						
Astarte borealis	0.095	0.105	0.095	0.105		
Nuculana pernula					0.069	0.069

**Table S15.** Statistical test for differences in biota-sediment accumulation factors of polycyclic aromatic hydrocarbons between Arctic species exposed to the high treatment (Kruskal Wallis test *p*-value).

Spp.	PHE	ANT	FLT	PYR	CHR	BbF
Macoma calcarea						
Nephtys ciliata						
Astarte borealis	0.068	0.037	0.103	0.103		
Nuculana pernula					0.103	0.054

**Table S16.** Statistical test for differences in biota-sediment accumulation factors of polycyclic aromatic hydrocarbons between Arctic bivalve species exposed to the high treatment (Kruskal Wallis test *p*-value).

Spp.	PHE	ANT	FLT	PYR	CHR	Ind123P
Macoma calcarea						
Astarte borealis	0.069	0.069	0.107	0.107		
Nuculana pernula					0.069	0.107







Table S17. Statistical test for differences in lipid-normalised polycyclic aromatic hydrocarbons (PAHs) concentrations between temperate species exposed to the high treatment after log transformation (independent t test).

					equality of	equality of variances				T test fo	T test for equality of means	ans	
	Group statistics	p sta	tistics			5		9	Sig.	Mean	SE	95% confidence interval of the difference	interval of the nce
PAH		z	Mean	SE	Τ,	<u>හ</u> පු	t	đ	(2- tailed)	difference	difference	Lower	Upper
TINA	AV	3	1.99	0.18			00	-	2000	330	0	-	
ANI	ΓB	33	2.54	0.07	3.91	0.12	-7.00	4	0.043	-0.33	0.19	-1.0/	-0.02
	AV	ж	3.49	0.07	700	70 0	300	-	040	000	0	200	
FLI	ΓB	3	3.45	0.07	0.0	0.00	0.53	4	0./40	0.03	0.10	-0.24	0.31
a ya	AV	3	3.30	0.07	000	1	00 0	-	0.00	100	0	03.0	
FIK	LB	ж	3.60	0.07	0.00	1.00	-3.00	4	0.040	-0.31	0.10	4C.U-	-0.02
4.0	AV	3	2.01	0.07	0.03	90 0	14 43	-	0000	137	000	1 63	-
DaA	ΓB	3	3.38	90.0	0.0	0.00	-14.43	4	0.000	1.3/	60.0	-1.03	-1.11
Den	AV	33	2.67	90.0	73.0	0.40	7 23	-	0000	25.0	0	1 03	20 0
Der	ΓB	3	3.41	0.08	0.5	64:0	-7.73	†	0.007	+/.0-	0.10	-1.03	1.0-
DFL	AV	3	2.09	0.03	2 43	21.0	10 27	-	0000	201	100	1	1 04
BOF	ΓB	33	3.33	90.0	5.45	0.14	-10.7	4	0.000	-1.24	0.0	-1.42	-1.03
JD5LA	AV	3	2.30	0.18	000	070	32 2	-	0.030	090	000	30 1	0.10
UDAIIA	ΓB	3	2.99	0.09	0.00	0.4.0	-5.30	†	0.030	60:0-	0.50	27:1-	-0.12
T. 4132B	AV	3	1.81	0.18	-	000	00.4	-	0000	1 05	100	1 64	0
1C71DUI	LB	3	2.86	0.11	7+.1	67:0	4.30	4	0.00	CU:1-	0.21	-1.04	-0.40

<sup>a</sup> Significant *p*-value after Holm's sequential Bonferroni correction based on the number of possible paired comparisons among chemicals is applied ( $p \le 0.0014$ ). AV: Alitta virens; LB: Limecola balthica.



**Table S18.** Statistical test for differences in biota-sediment accumulation factors of polycyclic aromatic hydrocarbons between temperate species exposed to the high treatment (Mann Whitney U test p-value).

Spp.	ANT	PYR	BaA	BeP	BbF	dBahA	Ind123P
Limecola balthica	0.050	0.050	0.050	0.050	0.050	0.050	0.050
Alitta virens							

**Table S19.** Statistical test for differences in lipid-normalised polycyclic aromatic hydrocarbons concentrations between Arctic and temperate species (Mann Whitney U test p-value) based on input data from Tables S5 and S6.

Spp.	PHE	ANT	BaA	CHR	BeP	BbF	BghiP	dBahA	Ind123P
Macoma calcarea	0.083	0.083							
Limecola balthica			0.083	0.083	0.083	0.083	0.083	0.083	0.083
Nephtys ciliata	0.827	0.127	0.127	n t	0.127	0.050	n t	n t	n t
Alitta virens				n.t.			n.t.	n.t.	n.t.

n.t.: not tested.

**Table S20.** Statistical test for differences in biota-sediment accumulation factors of polycyclic aromatic hydrocarbons between Arctic and temperate species exposed to the high treatment (Mann Whitney U test p-value) based on input data from Tables S11 and S12.

Spp.	PHE	ANT	FLT	PYR	BaA	CHR	BeP	BbF	BghiP	dBahA	Ind123P
Macoma calcarea	0.083	0.083	0.083								
Limecola balthica				1.000	0.083	0.083	0.083	0.083	0.083	0.248	0.083
Nephtys ciliata	0.827	0.127	0.050	0.050	0.050	n t	0.050	0.050	n t		t
Alitta virens						n.t.			n.t.	n.t.	n.t.

n.t.: not tested.







Table S21. Parameters and their 90% confidence intervals (CIs) (Eq. S4) obtained from fitting with data for Astarte borealis, Macoma calcarea, Nuculana pernula, Limecola balthica, Nephtys ciliata and Alitta virens using the full model.

Parameters	C	Astarte borealis		Macoma calcarea Nuculana pernula Limecola balthica Nephys ciliata Alitta virens Limecola balthica <sup>a</sup> Alitta virens <sup>a</sup>	Limecola balthica	Nephtys ciliata	Alitta virens	Limecola balthica <sup>a</sup>	Alitta virensª
	T90	*	4.29	*	3.15	3.64	*	2.8	
a (-)		1.97	5.00	-6.11	4.40	4.53	-6.35	3.9	2.7
	H90	4.27	5.53	4.29	4.81	4.91	2.98	4.3	3.4
	T60	n.a.	*#	n.a.	*#	*#	*#	0.07-0.16 <sup>b</sup>	0.11-0.20 <sup>b</sup>
I (kgocxkg <sup>-1</sup> xd <sup>-1</sup> ) - L		n.a.	0.00	n.a.	0.00	0.00	0.00	0.12-0.24 <sup>b</sup>	0.14-0.25 <sup>b</sup>
I	H90	n.a.	0.02	n.a.	90.0	0.01	0.00	0.16-0.32 <sup>b</sup>	0.17-0.31 <sup>b</sup>
	T90	n.a.	n.a.	n.a.	n.a.	*	*		
I (kgocxkg <sup>-1</sup> xd <sup>-1</sup> ) - M		n.a.	n.a.	n.a.	n.a.	00.00	0.00		
I	06H	n.a.	n.a.	n.a.	n.a.	0.01	0.00		
	T90	*	*	0.00	*	*	*		
I (kgocxkg <sup>-1</sup> xd <sup>-1</sup> ) - H		0.00	00:00	0.01	0.00	00:00	0.00		
	H90	0.00	0.01	0.02	0.01	0.00	0.00		
	T90			0.41					
p (-)				0.67					
	H90			0.93					
N of experimental data points		96	52	22	65	95	149		
Parameters (p)		3	4	3	4	5	5		
F-ratio value	Per species	2.14	2.07	2.40	2.04	1.91	1.89		
SSmin		59.13	11.35	3.72	11.43	34.57	40.45		
N of experimental data points				479					
Parameters (p)	T.	s 5		19					
F-ratio value	1 0121			1.45					C
SSmin				303.16					ha
									p





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Table S21. Continued.

estimated value, \* = parameter set to zero or one because fit was out of constrain boundary, L90 = lower boundary of the 90% CI, H90 = higher boundary of the <sup>a</sup> Diepens et al. [56], <sup>b</sup> Ranges for all treatments as described in Diepens et al. [56]; # = confidence limit not within two orders of magnitude above or below 90% CI. Ingestion I was fitted separate for low (L), medium (M) and high (H) treatments. n.a.: not available.

Table S22. Fraction of steady state reached (Fss) (Eq. S3) in 28-d bioaccumulation main tests with Astarte borealis, Macoma calcarea, Nuculana pernula, Limecola balthica, Nephtys ciliata and Alitta virens for low (L), medium (M) and high (H) treatments for the full model.

					Ŧ	action (	Fraction of steady state reached	y state 1	eached			
Treatment	Species	PHE	ANT	FLT	PYR	BaA	CHR	BeP	BbF	BghiP	dBahA	Ind123P
	A. borealis	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
	M. calcarea	1.00	1.00	1.00	1.00	1.00	1.00	0.87	0.79	09.0	0.45	0.54
-	N. pernula	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
7	L. balthica	1.00	1.00	0.99	1.00	0.46	0.38	-0.45	-0.65	-0.94	-1.10	-1.00
	N. ciliata	1.00	1.00	1.00	1.00	0.91	0.89	99.0	09.0	0.50	0.45	0.48
	A. virens	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25
	A. borealis	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
	M. calcarea	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
7	N. pernula	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
Z.	L. balthica	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
	N. ciliata	1.00	1.00	1.00	1.00	0.94	0.92	0.75	0.71	0.64	09.0	0.62
	A. virens	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32	0.32
	A. borealis	-8.02	-7.96	-8.48	-8.39	-8.63	-8.64	-8.67	-8.67	-8.68	-8.68	-8.68
	M. calcarea	1.00	1.00	1.00	1.00	1.00	1.00	0.87	0.78	0.59	0.43	0.53
I	N. pernula	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04
=	L. balthica	1.00	1.00	1.00	1.00	0.63	0.57	-0.01	-0.15	-0.34	-0.46	-0.39
	N. ciliata	1.00	1.00	1.00	1.00	0.90	0.88	0.62	0.55	0.44	0.38	0.42
	A. virens	0.31	0.31	0.31	0.31	0.31	0.31	0.31	0.31	0.31	0.31	0.31
aldelieve ton . e n	ohlo											







Table S23. Relative importance of polycyclic aromatic hydrocarbons (PAHs) uptake by sediment ingestion in 28-d bioaccumulation main tests with Astarte borealis, Macoma calcarea, Nuculana pernula, Limecola balthica, Nephys ciliata and Alitta virens for low (L), medium (M) and high (H) treatments for the full model.

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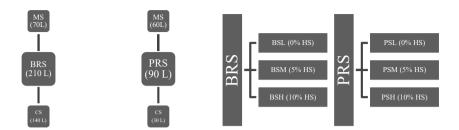
					% P.	% PAH uptake by sediment ingestion	te by sedi	ment ing	estion			
Treatment	Species	PHE	ANT	FLT	PYR	BaA	CHR	BeP	BbF	BghiP	dBahA	Ind123P
	A. borealis	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
	M. calcarea	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
-	N. pernula	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
J	L. balthica	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
	N. ciliata	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
	A. virens	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
	A. borealis	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
	M. calcarea	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
Y	N. pernula	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
K	L. balthica	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
	N. ciliata	0.00	0.00	0.00	0.00	0.00	0.00	0.00	00.00	0.00	0.00	0.00
	A. virens	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
	A. borealis	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
	M. calcarea	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
=	N. pernula	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Ξ	L. balthica	06.0	0.82	2.86	2.00	10.29	11.17	25.29	31.35	44.13	53.14	47.30
	N. ciliata	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
	A. virens	100.00		100.00 100.00	100.00	100.00	100.00		100.00 100.00 100.00	100.00	100.00	100.00

n.a.: not available



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**Figure S1.** Preparation of Bivalves Reference Sediment (BRS) and Polychaetes Reference Sediment (PRS) from both muddy (MS) and coarse (PS) sediment.

BRS: Bivalves Reference Sediment; BSL: Low Bivalves Sediment containing background levels of contaminants; BSM: Medium Bivalves Sediment containing 5% (vol.) of harbour sediment (HS); BSH: High Bivalves Sediment containing 10% (vol.) of harbour sediment (HS); PSL: Low Polychaetes Sediment containing background levels of contaminants; PSM: Medium Polychaetes Sediment containing 5% (vol.) of harbour sediment (HS); PSH: High Polychaetes Sediment containing 10% (vol.) of harbour sediment (HS).

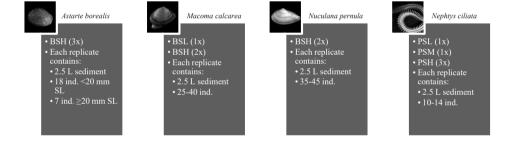


Figure S2. Arctic experiment (AE) replicates per species and treatment.

BSL: Low Bivalves Sediment containing background levels of contaminants; BSH: High Bivalves Sediment containing 10% (vol.) of harbour sediment; PSL: Low Polychaetes Sediment containing background levels of contaminants; PSM: Medium Polychaetes Sediment containing 5% (vol.) of harbour sediment; PSH: High Polychaetes Sediment containing 10% (vol.) of harbour sediment; SL: shell length; ind.: number of individuals.







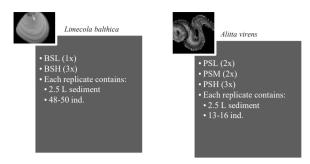


Figure S3. Temperate experiment (TE) replicates per species and treatment.

BSL: Low Bivalves Sediment containing background levels of contaminants; BSH: High Bivalves Sediment containing 10% (vol.) of harbour sediment; PSL: Low Polychaetes Sediment containing background levels of contaminants; PSM: Medium Polychaetes Sediment containing 5% (vol.) of harbour sediment; PSH: High Polychaetes Sediment containing 10% (vol.) of harbour sediment; ind.: number of individuals.

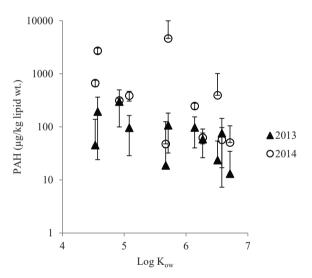


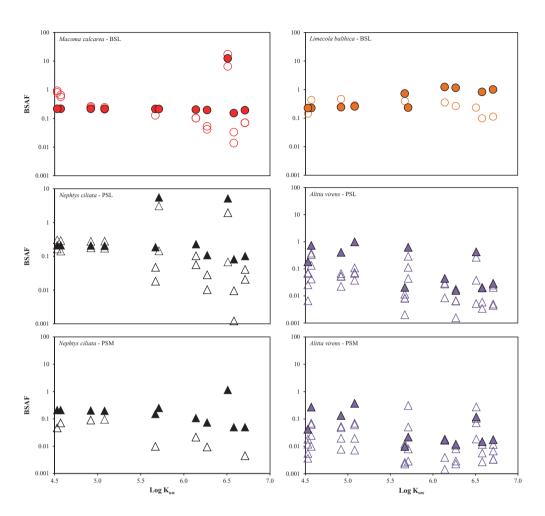
Figure S4. Astarte borealis background polycyclic aromatic hydrocarbons (PAHs) concentrations (mean  $\pm$  SD;  $\mu$ g/kg lipid wt.).

Individuals collected in July 2013 by Szczybelski et al. (**Chapter 2**) belong to size classes 2 to 4 as described by the authors. Individuals were collected in different sampling years near Blomstrandhalvøya, Kongsfjorden Bay (Svalbard, Norway).









**Figure S5.** Modelled polycyclic aromatic hydrocarbons biota-sediment accumulation factors (BSAFs) (closed symbols) and actual measured BSAFs (open symbols) for Arctic (*Macoma calcarea* and *Nephtys ciliata*) and temperate (*Limecola balthica* and *Alitta virens*) species exposed to the low (BSL/PSL) and medium (PSM) treatments.







4 Biomarker responses and biotransformation capacity in Arctic and temperate benthic species exposed to polycyclic aromatic hydrocarbons

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Science of the Total Environment, submitted







#### Abstract

Monitoring parameters for the assessment of oil and gas related contaminants and their biological effects need validation before application in the Arctic. For such monitoring purposes, we evaluated the potential use of three biomarkers (acetylcholinesterase, acyl-CoA oxidase and glutathione S-transferase) for application to an Arctic bivalve (Astarte borealis) and determined the body residue of pyrene and two pyrene metabolites (1-hydroxypyrene and pyrene-1-glucuronide) in Arctic benthic species (bivalve: Macoma calcarea; polychaete: Nephtys ciliata) and temperate benthic species (bivalve: Limecola balthica; polychaete: Alitta virens) in order to establish the potential of polycyclic aromatic hydrocarbons (PAHs) metabolite profiles as biomarkers of exposure in such species. Experimental PAH exposure levels were probably too low (0.2 - 1.7 mg/kg dry wt. in sediment) to induce or inhibit biomarker responses in A. borealis. Concentrations of pyrene and pyrene metabolites varied between species, although no consistent patterns could be established among taxonomic groups and locations. Metabolites made up to 79% of the total pyrene concentrations indicating that basal metabolic activity is affecting pyrene kinetics even at low concentrations in all species. This indicates that Arctic and temperate species could show similar metabolism patterns of PAHs, although more insight into the effects of confounding factors is needed.







#### 1. Introduction

Current baseline information on potential effects of oil and gas (O&G) activities on the Arctic ecosystem or ecosystem components is still inadequate or unavailable [72], particularly on the cumulative effects of O&G related toxic compounds. More information is needed because the retreat of sea ice in the Arctic will cause an increase in O&G related activities [14]. Hence, further studies on assessing the vulnerability of Arctic biota to O&G related chemicals are essential, in order to inform policy and to ensure that O&G risk assessment procedures are targeted to the Arctic environment.

Polycyclic aromatic hydrocarbons (PAHs) are the main organic pollutants related to O&G activities, which have been identified as an emerging concern in the Arctic [121]. Once released into the marine environment, partitioning of PAHs to settling particulate organic matter generally results in contamination of sediments and chronic exposure of benthic organisms. Accumulation of PAHs by benthic organisms depends on several factors such as the species' feeding behavior, physiological responses to PAH exposure like detoxification and excretion of PAHs [256], as well as sediment characteristics [80]. Biological responses (biomarkers) to exposure to PAHs may be used as indicators of exposure to or sublethal ecological effects of O&G related activities, and the use of such biomarkers may provide an early warning for hazard [257] and ecological risk assessment [86,258].

Studies have documented effects of oil at the cellular, individual and community level in Arctic benthic communities [104,122,259,260], but they provide little information on the specific chemicals in the complex oil matrix causing the effects. Although PAHs are the major group of organic contaminants in oil, there is a large knowledge gap on their specific role in affecting the responsiveness of biochemical biomarkers in Arctic benthos chronically exposed to O&G derived chemicals [261]. Furthermore, almost no information on PAH metabolites in benthic invertebrates is available that may be used to assess oil-derived hazards and risks for Arctic benthic organisms [213]. In order to address these knowledge gaps the aims of the present study were to (1) identify the responsiveness of biochemical biomarkers of exposure and effects in Arctic benthic species, and to (2) assess the use of profiles of metabolites of a model PAH (i.e., pyrene) for monitoring of exposure and effects in Arctic benthic invertebrates. This was performed experimentally, using representative species of the Arctic macrobenthic community (the suspension feeder *Astarte borealis* [Schumacher, 1817], the suspension/deposit feeder *Macoma calcarea* [Gmelin, 1791] and the predator/deposit feeder









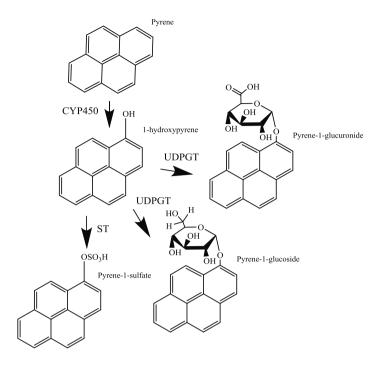
Nephtys ciliata [O.F. Müller, 1776]), which are compared to temperate counterpart species (the suspension/deposit feeder *Limecola balthica* [Linnaeus, 1758] and the omnivore/deposit feeder *Alitta virens* [Sars, 1835], formerly known as *Macoma balthica* and *Nereis virens*, respectively).

Selected biomarkers of exposure were the peroxisomal  $\beta$ -oxidation enzyme acyl-CoA oxidase (AOX, E.C.1.3.3.6) and Phase II biotransformation enzyme glutathione S-transferase (GST, E.C.2.5.1.18). The biomarker of neurotoxicity consisted of acetylcholinesterase (AChE, E.C.3.1.1.7). These were determined in A. borealis, upon exposure to PAH-contaminated sediment under laboratory conditions. In a second experiment, selected pyrene metabolites (1-hydroxypyrene [OHPyr, Phase I metabolite, Figure 1] and pyrene-1-glucuronide [GluPyr, Phase II metabolite, Figure 1]) were analyzed in Arctic (M. calcarea, N. ciliata) and temperate (L. balthica, A. virens) bivalves and polychaetes exposed to PAH-contaminated sediment under laboratory conditions. Due to limited availability of test animals it was not feasible to analyze biochemical biomarkers and metabolite profiles in the same experiments. In the first experiment, an Arctic species was selected for which ample information was available on biomarker responses in similar temperate species. As for metabolite profiles, much less information is available, so we conducted a broader experimental approach, comparing species with different feeding modes and geographical origin.









**Figure 1.** Biotransformation pathway for pyrene in *Alitta virens* (adapted from Jørgensen et al. [183]). CYP450: cytochrome P450 enzymes; ST: sulfotransferase enzymes; UDPGT: glucuronosyltransferase enzymes. CYP450 forms Phase I metabolites, whereas ST and UDPGT form Phase II metabolites.

#### 2. Materials and Methods

# 2.1. Test organisms

Based on their feeding habits, sessility and relative abundance in Kongsfjorden Bay (Svalbard, Norway), three Arctic benthic species were selected: the suspension feeder *A. borealis*, the suspension/deposit feeder *M. calcarea*, and the carnivore/omnivore *N. ciliata*. Additionally, two counterpart temperate species (the suspension/deposit feeder *L. balthica* and the deposit feeder/omnivore *A. virens*) were selected based on their comparable habitat and feeding traits with Arctic species. Permission for sampling of Arctic and temperate (*L. balthica*) species was issued by The Governor of Svalbard and the Province of Zeeland, respectively. *A. virens* was obtained from a professional bait farm, Topsy baits







(Wilhelminadorp, the Netherlands). For details of the sampling and acclimatization of test species and sediment collection, the reader is referred to the Supporting Information (SI).

#### 2.2. Sediment treatments preparation

Reference sediment was collected in the Oosterschelde estuary (The Netherlands, 51°36'13"N, 3°47'49"E) on 24 to 25 April 2014 and 2 May 2014, and naturally contaminated sediment was collected near Nesserdijk (Rotterdam, the Netherlands, 51°54'14"N, 4°31'17"E) by Nautisch Service Centrum in April 2014. Two different batches of reference sediment were prepared to ensure an optimal habitat for the test species. Since the bivalve species (*A. borealis*, *M. calcarea* and *L. balthica*) usually reside in sandy sediment, their sediment batch consisted of two thirds of coarse and one third of muddy sediment, based on volume. *N. ciliata* and *A. virens* are generally found in sandy mud, so a sediment batch consisting of two thirds of muddy and one third of coarse sediment was prepared for these species.

The reference sediments are referred to as the 'low' (L) treatment: BSL (Bivalve Sediment Low for bivalves) and PSL (Polychaete Sediment Low for polychaetes). Two additional sediment treatments were prepared with a higher amount of harbour sediment increasing the chemical concentrations, referred to as 'medium' (BSM and PSM with 5% harbour sediment) and 'high' (BSH and PSH with 10% harbour sediment) treatments. These preparations thus resulted in six sediments: three for bivalves (BSL, BSM, BSH) and three for polychaetes (PSL, PSM, PSH) (Tables S5 and S6).

All sediment treatments were thoroughly mixed before storage and transport from the Netherlands to Svalbard, before use in the exposure experiments. Due to logistics, storage time (at 3-7 °C) for sediment treatments was 8 and 17 weeks in the temperate and Arctic experiment, respectively. Prior to the start of exposure, sediment was allowed to settle in contact with filtered sea water in a 1:6 sediment-to-water volume ratio without aeration during 3 d and with aeration during the following 4 d. In the Arctic experiment some aquaria were aerated for 7 to 14 d because biota field sampling took longer than initially expected.

 $\Sigma_{13}$ PAH concentrations in the high (H) treatment were selected to be above or within average sediment  $\Sigma_{16}$ PAH concentration detected in Arctic oil-impacted areas like Ny-Ålesund harbour (Svalbard, Norway) [149], and Gulf of Alaska (USA) after 4 and 13 years of the *Exxon Valdez* oil spill [262] (Table 1).







**Table 1.** Total polycyclic aromatic hydrocarbons (PAHs) concentrations (min-max; μg/kg dry wt.) in sediment and biota (soft tissue) available in the literature.

			Total	PAH
Present study	y <sup>a</sup>	Analytes (No. of compounds)	Sediment	Biota
	BSL		219 - 322	n.a.
	BSM		n.a.	n.a.
Treatment	BSH	3- to 6-ring PAHs (13)	839 - 2780	446 - 1878
	PSL		1130 - 1489	249 - 264
	PSH		2125 - 2437	147 - 1183
Payne et al.	[262] <sup>b</sup>			
	Gold Creek	2 4 6 : PAH 1 1 4 1	~50	~600
1993	Alyeska Marine Terminal	2- to 6-ring PAHs and selected alkylated homologues (44)	20 - 300	300 - 400
	Disk Island	alkylated homologues (44)	1830°	~200
Payne et al.	[262] <sup>b</sup>			
1993-2002	Gold Creek	2- to 6-ring PAHs and selected	40 – 125	100 - 800
		alkylated homologues (44)		
Van den Hei	uvel-Greve et al. [149]			
2012-2013	Kongsfjorden	3- to 6-ring PAHs (16)	1 - 26	n.a.
2012-2013	Ny-Ålesund	3- 10 0-Filig PARS (10)	476 - 2550	n.a.

<sup>&</sup>lt;sup>a</sup> Sediment and A. borealis samples from the Arctic experiment.

## 2.3. Experimental design

Two 28-d exposure experiments were conducted: a first with Arctic species and conditions at Kings Bay AS (Ny-Ålesund, Svalbard) (September to October 2014); and another with temperate species and conditions at Wageningen Marine Research (Yerseke, the Netherlands) (July to August 2014). The Arctic experiment was performed in a temperature controlled room (3±1 °C) with a photoperiod of 12 h light: 12 h dark, whereas the temperate experiment was performed in a temperature controlled room (18±1 °C) with a photoperiod of 18 h light: 6 h dark.

Exposure treatments were achieved in quadruplicate or more for *A. borealis* (BSL, BSM and BSH), whereas for the rest of the test species, these were achieved in duplicate or more due to low availability of test organisms (Table S1). For *A. borealis*, each aquarium contained 25 individuals of which 18 individuals were classified as 'small' (< 2 cm) and 7 individuals were classified as 'large' (2 – 4 cm), based on shell length. For *M. calcarea*, *L. balthica*, *N. ciliata* and *A. virens*, each aquarium contained 40 to 50, 50, 20 and 16 individuals, respectively (Table S1). Individuals were not fed during exposure.





<sup>&</sup>lt;sup>b</sup> Sediment and *Mytilus trossulus* samples.

<sup>&</sup>lt;sup>c</sup> Sediment samples from Boehm et al. [263].

BSL: Low Bivalve Sediment; BSH: High Bivalve Sediment; PSL: Low Polychaete Sediment; PSM: Medium Polychaete Sediment; PSH: High Polychaete Sediment; n.a.: not available.



No mortality was observed for any species in any treatment. At the end of the experiments, organisms were allowed to depurate their guts for 24 h in filtered sea water. Animals were weighed, measured for body or shell length, dissected (e.g., bivalves), snap-frozen and stored at -80 °C for the determination of enzymatic activities in *A. borealis* samples or at -20 °C for the analysis of pyrene and pyrene metabolites in Arctic (*M. calcarea*, *N. ciliata*) and temperate (*L. balthica*, *A. virens*) species.

A. borealis samples were shipped in a dry-shipper in liquid nitrogen (Arctic Express 20, Thermo Fisher Scientific) to Plentzia Marine Station (University of the Basque Country) at Plentzia (Biscay, Spain), while the remaining samples were shipped in dry-ice to the Department of Analytical Chemistry (University of the Basque Country) at Leioa (Biscay, Spain).

#### 2.4. Chemicals

2',7'-dichlorofluorescin diacetate (DCF), N,N-dimethylformamide (DMF), palmitoyl coenzyme A lithium salt, peroxidase from horseradish, sodium azide, Triton<sup>TM</sup> X-100 and ethylenediaminetetraacetic acid disodium salt dihydrate (EDTA-N<sub>2</sub>) (Sigma-Aldrich) were used for the determination of AOX activity. Butylated hydroxytoluene (BHT), acetylthiocholine iodide (ATC), 5,5'-dithiobis(2-nitrobenzoic acid) (DTNB) (Sigma-Aldrich) and sodium bicarbonate (VWR) were used for the determination of AChE activity. 1-chloro-2,4-dinitrobenzene (CDNB) and L-glutathione reduced (GSH) (Sigma-Aldrich) were used for the determination of GST activity. Quick Start<sup>TM</sup> Bradford Kit 4 (Bio-Rad) was used for protein determination.

Pyrene (98%), 1-hydroxypyrene (98%) (Sigma-Aldrich), pyrenyl-1-O-glucuronide (98.2%) (Isostandards Material), acetone, methanol, acetonitrile (HPLC grade) (Macrom Fine Chemicals), 0.45 μm polyamide filters (Macherey-Nagel) and Oasis HLB (200 mg) cartridges (Waters) were used for the determination of pyrene metabolites.

# 2.5. Enzymatic activities

Pooled digestive glands of either 9 'small' or 3 'large' *A. borealis* individuals were homogenized in 1:5 volumes of 100 mM potassium phosphate buffer (pH 7.4) in a tissue homogenizer (Precellys®24, Bertin Technologies) at 6,000  $rpm \times 30$  s (5 °C). Homogenates for AOX determination were obtained after a centrifugation of the homogenate at 500  $g \times 15$ 







min (4 °C) (Allegra® 25R Benchtop Centrifuge, Beckman Coulter). 100  $\mu$ L of supernatant was used for AOX determination whereas the remaining volume was centrifuged at 13,280 g × 20 min (4 °C) (Microfuge® 22R Microcentrifuge, Beckman Coulter) to obtain the postmitochondrial fraction (S12) in the supernatant for further biochemical determinations of AChE and GST activities. Samples were stored at -80 °C if not directly used. All assays were carried out in quadruplicate per sample at 22 °C (AChE, GST) or in duplicate at 25 °C (AOX).

AOX activity was analyzed in *A. borealis* 1:4 500 g homogenates following Small et al. [264]. AChE activity was analyzed in *A. borealis* S12 homogenates following Guilhermino et al. [265]. GST activity was analyzed in S12 homogenates following Habig et al. [266] with some modifications (i.e., reaction medium is 100 mM potassium phosphate buffer pH 7.4). AChE and GST activity rates were recorded in 96-well Sterilin<sup>TM</sup> Clear Microtiter<sup>TM</sup> plates (Thermo Fisher Scientific) at 412 and 340 nm every 60 s during 5 and 3 min, respectively (PowerWave HT Microplate Spectrophotometer, BioTek). Total cytosolic protein was measured in the same homogenate fraction where enzymatic activities were analyzed following Lowry et al. [267].

# 2.6. Pyrene metabolites

Samples were frozen and freeze-dried prior to the extraction step. Each sample was weighed and extracted in 10 mL of acetone in a 40 mL polypropylene vessel. Focused ultrasound solid-liquid extraction (FUSLE) was performed in the pulsed mode for 2 min, with a pulsed time 'on' of 0.8 s and pulsed time 'off' of 0.2 s, at 20% of irradiation power [268]. Extractions were performed at 0 °C in an ice-water bath [269]. After the extraction step, the supernatant was filtered through 0.45  $\mu$ m poliamide filters and evaporated to 500  $\mu$ L under a nitrogen stream at 35 °C using a Turbovap LV evaporator (Zymark).

200-mg Oasis HLB cartridges were conditioned with 10 mL of Milli-Q:acetonitrile (1:9, v/v). Then, the sample was loaded and 1 mL of Milli-Q water was added and cartridges were dried for an hour under vacuum. Then, the analytes were eluted using 10 mL of acetonitrile and collected in a test tube. The eluate was evaporated until dryness under a gentle stream of nitrogen at 35 °C and reconstituted in 150  $\mu$ L of methanol. Finally, the analyses of pyrene metabolites were performed by high-performance liquid chromatograph with fluorescence detector (HPLC-FLD). For further details on the extraction, clean-up and instrumental analysis see the SI.







#### 2.7. Data analyses

Data were checked for normality with Q-Q plots and the Shapiro-Wilk test and for equality of variances with Levene's test. Differences among treatment groups were assessed either by one-way ANOVA and Tukey HSD as a post-hoc test in case of normally distributed data, or by the Kruskal-Wallis non-parametric rank test for each of the biomarkers in *A. borealis*. For each taxonomic group (i.e., bivalves and polychaetes), differences in the content of GluPyr among L and H treatment groups and among climatic groups were analyzed by either ANCOVA, or multiple linear regression, respectively. For the ANCOVA test the content of pyrene (Pyr) was used as the covariate and for multiple linear regression the treatment group and source region of the species were used as predictors. A simple linear regression between Pyr and GluPyr concentration was also calculated for each of the species. The simple linear regression was used to analyze any likely difference in GluPyr/Pyr ratios among species. Significance level of an overall statistical comparison was set at  $p \le 0.05$ , while for pairwise comparisons a Bonferroni correction was applied. All statistical calculations were performed using SPSS version 22.

#### 3. Results and Discussion

## 3.1. Enzymatic activities

Exposure to the selected  $\Sigma_{13}$ PAH concentration range (i.e.,  $287 - 1710 \,\mu\text{g/kg}$  dry wt. sediment; Table S5) had no significant effect on any biomarker response in *A. borealis* digestive gland (Tables 2, S11-S13). Nevertheless, AChE activity in BSH exposed *A. borealis* was relatively low and within the range as detected in mussels from either historically polluted areas [270] or transplanted to harbour areas [271], indicating a potential neurotoxic response. Examples of classical AChE inhibitors are organophosphate and carbamate pesticides, although PAHs may also show AChE inhibition capacity [270,272]. However, low AChE activity could also be due to either a low AChE substrate specificity in digestive gland with respect to gills as observed in scallops [273] or a stressed physiological status in *A. borealis* as observed in caged mussels [274].

The absence of AChE inhibition in *A. borealis* is also in line with the lack of GST and AOX induction (Table 2). GST activity in *A. borealis* was below baseline levels as detected in digestive gland of scallops (*Chlamys islandica*) [215,275] and slightly lower than those in







mussels (*Mytilus galloprovincialis*) experimentally exposed to similar PAH concentrations (Table 3; [276]). GST activity is mainly correlated to the concentration of 5- to 6-ring PAHs in mussels [277]. Yet, higher concentrations of high-molecular-weight PAHs in BSH exposed *A. borealis* compared to sediment PAH exposed *M. galloprovincialis* yielded lower GST activity values in *A. borealis* than in mussels [276]. Furthermore, a negative correlation was observed between GST activity rates in *M. galloprovincialis* and sediment PAH exposure (Table 3; [276]). This indicates that even if lower-molecular-weight PAHs such as phenanthrene, anthracene, fluoranthene and pyrene were mainly accumulated by *A. borealis* during BSH exposure (**Chapter 3**), pointing to an absence of GST induction, likely confounding factors such as the nutritional and reproductive state of *A. borealis* could have masked the bivalves' response to organic pollution [278].

**Table 2.** Acyl-CoA oxidase, acetylcholinesterase and glutathione S-transferase activities (mean  $\pm$  SD) in *Astarte borealis* digestive gland.

Treatment	$\mathrm{EA}\left( n\right)$	AOX	AChE	GST
BSL	4	$2.1\pm0.4$	$27.9 \pm 8.4$	$49.8 \pm 13.4$
BSM	4	$2.0 \pm 0.4$	$35.5 \pm 4.2$	$47.8\pm14.2$
BSH	6	$2.3 \pm 0.6$	$30.7 \pm 14.4$	$46.7\pm11.2$

EA: experimental aquarium; AOX: Acyl-CoA oxidase (mU AOX/mg prot.); AChE: Acetylcholinesterase (nmol/min/mg prot.); GST: Glutathione S-transferase (nmol/min/mg prot.). Additional abbreviations are defined in Table 1.

AOX activity in all *A. borealis* pooled samples was above levels as detected in digestive gland of mussels from or transplanted to harbour areas [279] and at O&G produced water discharge points [280]. This indicates that AOX levels in the present study were high, even in the controls (i.e., BSL exposed *A. borealis*). Relatively high AOX activity in *A. borealis* may be related to non-toxic factors like low food availability or changing metabolic status under post bloom conditions [134]. Although AOX induction has been described as a rapid and reversible response to PAH and PCB exposure [279,281], this process can also be affected by seasonal changes [281]. AOX activity in mussels (*M. galloprovincialis*) is usually induced during late winter and spring while AOX levels are negatively correlated with lipid content in digestive tubules of mussels during phytoplankton blooms [282]. AOX may be further induced in *A. borealis* due to the increased synthesis of prostaglandins during the species spawning in early autumn [282,283]. Hence, nutritional and reproductive status impacts on peroxisome proliferation should be considered as likely confounding factors when interpreting AOX as a biomarker for exposure to environmental contaminants [281].







**Table 3.** Total polycyclic aromatic hydrocarbon (PAH) concentrations in whole body soft tissue and glutathione S-transferase activity in digestive gland of exposed bivalve species.

		Total	PAH	
Source	Species (type of exposure)	Analytes (No. of compounds)	Whole body soft tissue (weight units)	GST
Present study	Astarte borealis <sup>a</sup> (sediment)	3- to 6-ring PAHs (13)	170 (wet wt.) 34830 (lipid wt.)	34 – 62
Baussant et al.	Chlamys islandica (dispersed oil)	2- to 6-ring PAHs	0 (wet wt.) 6 (wet wt.) 8 (wet wt.)	475 355 425
[275]	Mytilus edulis (dispersed oil)	(16)	0 (wet wt.) 3 (wet wt.) 10 (wet wt.)	71 62 85
Nahrgang et al.	Chlamys islandica b (bakground)	2- to 6-ring PAHs	< 5 – 5.7 (wet wt.)	500 - 800
[215]	Mytilus edulis <sup>b</sup> (background)	(16)	5.7 – 6.3 (wet wt.)	0 – 100
Giuliani et al. [276]	Mytilus galloprovincialis (sediment)	2- to 6-ring PAHs (15)	140 (dry wt.) 300 (dry wt.) 600 (dry wt.)	120 110 90

<sup>&</sup>lt;sup>a</sup> A. borealis exposed to BSH treatment.

In summary, we observed a general lack of dose related responses of the selected biomarkers in digestive gland of *A. borealis*. This may indicate that exposure levels were too low to induce such changes. However, within treatment variation in catalytic activity was rather large, which prevented a sensitive analysis of dose response relationships, but for which knowledge of possible confounding factors was lacking. Therefore, in order to interpret variation in biomarker responses in Arctic marine invertebrates, relevant confounding parameters should be identified such as time scales for enzymatic induction, conditions of the assays, and maturation and nutritional status for field individuals.

#### 3.2. Pyrene metabolites

Biotransformation and excretion of hydrophobic organic pollutants is mostly mediated by CYP enzymes, although biotransformation capacity may be low in marine invertebrates [284]. Only a few studies showed the existence of the aryl hydrocarbon receptor (AHR) transcription factor in marine invertebrates and suggested that the expression of Phase I enzymes through AHR in response to hydrocarbon exposure might be tissue-specific in bivalves [285,286]. Additionally, relatively large differences in dominating Phase II biotransformation pathways between invertebrates could yield different PAH metabolite patterns. In the present study, GluPyr was selected as the major Phase II metabolite because glucuronosyltransferases







<sup>&</sup>lt;sup>b</sup> Biota samples collected from the field in September 2010.

Whole body soft tissue (µg/kg); GST: Glutathione S-transferase (nmol/min/mg prot.).



(UDPGTs) have a higher activity rate than sulfotransferases (STs), and also glucuronidation is the main conjugation pathway in A. virens (Figure 1). In the following subsections, pyrene metabolites results are discussed based on the possibilities and limitations for an interspecies comparison inherent in our selection of metabolites.

### 3.2.1. Pyrene hydroxylation

OHPyr concentrations were often below the limit of detection among all analyzed individuals (Tables 4, S9 and S10). This agrees with typically low OHPyr/Pyr ratios found in other freshwater [287] and marine worms [183,233], as well as in marine clams [288] due to generally high Phase II biotransformation levels and/or low induction capacity of Phase I biotransformation pathways in polychaetes and bivalves. Based on the species-specific sediment threshold for A. virens (i.e., 10 µg pyrene/g dry wt. [183]), an absence of pyrene hydroxylase induction and therefore low OHPyr concentrations, would be expected in A. virens exposed to PSH (i.e., 0.5 μg pyrene/g dry wt.). This is also expected in L. balthica since its PAH biotransformation capacity is generally lower than A. virens [139].

### 3.2.2. 1-hydroxypyrene glucuronidation

A significant effect of the high treatment (PSH) compared to the low sediment treatment (PSL) was observed on the concentration of GluPyr in polychaetes (p = 0.005, Table S14). This was not detected between BSH and BSL exposed bivalves (Table S15). Similarly, a multiple linear regression on the concentration of GluPyr did generate a significant model for polychaetes when both treatment and climatic groups were included as predictors ( $R^2 = 0.466$ , p < 0.001, Table S16). This was not the case for bivalves (Table S17).

# 3.2.3. Biotransformation of pyrene among climatic groups

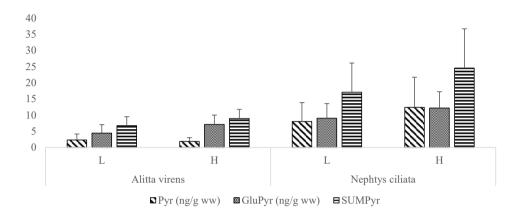
For polychaetes and bivalves different GluPyr/Pyr ratios were observed when comparing Arctic and temperate species (Figures 2 and 3). GluPyr concentrations in the Arctic N. ciliata were on average twice as high as in the temperate A. virens in both L and H treatments. This agrees with higher pyrene concentrations in the former species (Figure 2) and a positive correlation between pyrene and GluPyr concentrations in N. ciliata ( $R^2 = 0.252$ , p = 0.012), but not in A. virens (Tables S18 and S19). In H exposed A. virens, low GluPyr concentrations may be due to too low pyrene concentrations in A. virens to lead to significant induction of Phase I (e.g., pyrene hydroxylase) and thus, consecutive Phase II (UDPGT) biotransformation







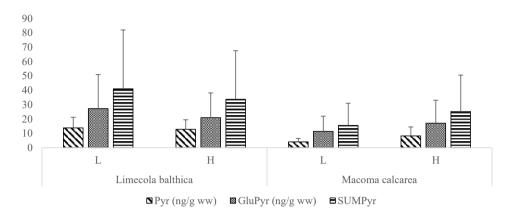
[233]. However, GluPyr/Pyr ratios in H exposed *A. virens* were on average 3.5 times higher than in *N. ciliata*. This shows that biotransformation rates were limited among polychaete species according to the low level of exposure (Table 4), although a higher induction of CYP450 enzymes may be expected in *A. virens* than *N. ciliata* possibly due to a stronger bioactivity of PAHs and additional inducer chemicals, such as PBDEs (Table S4; [289]), in the broad omnivore *A. virens* than in *N. ciliata* [136,235].



**Figure 2.** Concentrations of pyrene (Pyr; ng/g wet wt.), pyrene-1-glucuronide (GluPyr; ng/g wet wt.) and the sum of both compounds (SUMPyr) in whole body tissue of temperate (*Alitta virens*) and Arctic (*Nephtys ciliata*) polychaete species exposed to the low (L) and high (H) sediment treatments.







**Figure 3.** Concentrations of pyrene (Pyr; ng/g wet wt.), pyrene-1-glucuronide (GluPyr; ng/g wet wt.) and the sum of both compounds (SUMPyr) in soft body tissue of temperate (*Limecola balthica*) and Arctic (*Macoma calcarea*) bivalve species exposed to the low (L) and high (H) sediment treatments.

**Table 4.** Pyrene, 1-hydroxypyrene and pyrene-1-glucuronide concentrations (geomean [min-max]; ng/g wet wt.) in Arctic (*Macoma calcarea*, *Nephtys ciliata*) and temperate species (*Limecola balthica*, *Alitta virens*).

Species	Treatment	Pyr	OHPyr	GluPyr
Limecola balthica	BSL	11.6 ( <loq 36.4)<="" td="" –=""><td>0.4 (<lod 2.98)<="" td="" –=""><td>17.4 (<lod 74.0)<="" td="" –=""></lod></td></lod></td></loq>	0.4 ( <lod 2.98)<="" td="" –=""><td>17.4 (<lod 74.0)<="" td="" –=""></lod></td></lod>	17.4 ( <lod 74.0)<="" td="" –=""></lod>
Limecola balinica	BSH	11.8 ( <loq 26.7)<="" td="" –=""><td>0.4 (<lod 2.22)<="" td="" –=""><td>12.1 (<loq -="" 50.7)<="" td=""></loq></td></lod></td></loq>	0.4 ( <lod 2.22)<="" td="" –=""><td>12.1 (<loq -="" 50.7)<="" td=""></loq></td></lod>	12.1 ( <loq -="" 50.7)<="" td=""></loq>
Macoma calcarea	BSL	3.6 (0.7 – 12.3)	0.5 ( <lod -="" 1.5)<="" td=""><td>7.8 (<lod 34.7)<="" td="" –=""></lod></td></lod>	7.8 ( <lod 34.7)<="" td="" –=""></lod>
Macoma carcarea	BSH	5.8 (1.1 – 21.3)	0.6 ( <lod 3.1)<="" td="" –=""><td>7.1 (<loq 41.6)<="" td="" –=""></loq></td></lod>	7.1 ( <loq 41.6)<="" td="" –=""></loq>
Alitta virens	PSL	1.7(0.5 - 5.4)	0.1 ( <lod -="" 1.5)<="" td=""><td>3.9 (2.2 – 10.0)</td></lod>	3.9 (2.2 – 10.0)
Attita virens	PSH	1.5(0.5-4.3)	<lod< td=""><td>6.7(4.5 - 13.7)</td></lod<>	6.7(4.5 - 13.7)
Nephtys ciliata	PSL	5.9 (2.0 – 15.5)	0.2 ( <lod -="" 0.9)<="" td=""><td>8.2 (4.1 – 20.9)</td></lod>	8.2 (4.1 – 20.9)
Nepniys Ciliaia	PSH	9.2(3.6 - 28.5)	0.3 ( <lod 3.7)<="" td="" –=""><td>11.4 (6.4 – 22.2)</td></lod>	11.4 (6.4 – 22.2)

Pyr: Pyrene; OHPyr: 1-hydroxypyrene; GluPyr: pyrene-1-glucuronide; LOD: limit of detection; LOQ: limit of quantification. Additional abbreviations are defined in Table 1.

Mean GluPyr concentrations in the temperate L. balthica were higher than in the Arctic M. calcarea. This complies with higher pyrene concentrations in L. balthica than in M. calcarea (Figure 3). In H exposed bivalves, pyrene concentrations were within the same range as observed concentrations of another 4-ring PAH, chrysene, in scallops ( $Chlamys\ farreri$ ) after exposure to  $0.1\ \mu g/L$  (i.e., 35-45 ng chrysene/g dry wt. [290]). Within such range, concentrations of chrysene in scallops did not reveal a significant correlation with 7-ethoxyresorufin O-deethylase activity, which is the catalytic measurement of CYP450







induction. Absence of CYP induction would explain similar biotransformation rates between *L. balthica* and *M. calcarea* (Figure 3), but it would not account for a seemingly decreasing trend in GluPyr concentration with PAH exposure (Table 4). This stresses the importance of considering different Phase II biotransformation pathways for the characterization and comparison of pyrene metabolite patterns among bivalves [288,291].

In summary, pyrene metabolite patterns appeared to be similar among all target species after exposure to PAHs in sediment at the different sediment treatment levels. However, GluPyr yielded different concentration profiles between Arctic (*N. ciliata*) and temperate polychaetes (*A. virens*) exposed to the high treatment, possibly related to a higher induction of CYP450 enzymes in *A. virens*. Besides, the suitability of this metabolite as biomarker of exposure could be confirmed for *N. ciliata* according to its positive correlation with pyrene exposure. In bivalves, both temperate (*L. balthica*) and Arctic species (*M. calcarea*) showed similar GluPyr concentration profiles, although lower GluPyr concentrations with PAH exposure pointed towards the activation of alternative Phase II biotransformation pathways in *L. balthica*.

# 4. Conclusions and Implications

No effects of environmentally relevant PAH concentrations in sediment were found on biomarkers AChE, GST and AOX activities in digestive gland of *A. borealis*. Overall exposure levels were probably too low to induce detectable biomarker responses in *A. borealis* digestive gland, while relatively low AChE levels and high baseline AOX levels potentially indicated a stressed physiological status of *A. borealis*. A further identification of crucial confounding parameters in both enzymatic induction and inhibition, as well as optimizing the biomarker assays for Arctic species is therefore needed for an appropriate evaluation.

In Arctic and temperate benthic invertebrates similarly exposed to PAHs, the concentrations of Phase I and Phase II biotransformation metabolites of pyrene did not increase with pyrene exposure. Biotransformation rates appeared to be limited by low exposure to sediment-bound chemicals among bivalve and polychaete species, although glucuronide conjugates represented up to 74% and 79% of the total pyrene concentrations in Arctic bivalves and temperate polychaetes, respectively. This shows that in order to characterize and quantify







PAH exposure among invertebrate species, the metabolic products of Phase I and Phase II biotransformation should be equally considered.

# Acknowledgements

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# **Supporting Information**

### Field sampling

Sampling of Arctic species. Sampling in Kongsfjorden Bay (Svalbard, Norway) was mainly performed along a transect from Tønsneset (79°0'22"N, 11°57'10"E) to Blomstrandhalvøya (78°59'14"N, 11°57'28"E), and to a lesser extent at Gluudneset and Tyskerhytta, on board of research vessel MS Teisten (Kings Bay AS, Ny-Ålesund) on 8 to 26 September 2014. Biota samples were collected using a Van Veen grab (0.1 m²). Three Arctic marine benthic invertebrate species with different feeding behaviours were collected: Astarte borealis (mollusc; suspension feeder [Schumacher, 1817]), Macoma calcarea (mollusc; surface deposit and suspension feeder [Gmelin, 1791]) and Nephtys ciliata (polychaete; predator and deposit feeder [O.F. Müller, 1776]).

Sampling of temperate species. Limecola balthica (mollusc; surface deposit and suspension feeder [Linnaeus, 1758]) were collected at low tide at the Oesterdam (The Netherlands, 51°26'24"N, 4°13'16"E) on 16 to 17 July 2014. Alitta virens (polychaete; predator and deposit feeder [Sars, 1835]) were obtained from a professional bait farm, Topsy baits (Wilhelminadorp, the Netherlands).

#### Acclimatization

Arctic species were kept in glass aquaria with aerated, filtered (20  $\mu$ m) and UV-treated sea water from the fjord (3 °C) for 2 to 6 d, before start of the experiment. In the case of *N. ciliata*, a thin layer of sieved (1 mm  $\emptyset$ ) sediment from the original sampling area was also included. Animals were not fed during acclimatization.

Temperate species were kept in glass aquaria with aerated, filtered (0.2  $\mu$ m) sea water for 5 d and were fed once with spiked ground fish food (TetraMin) suspended in deionised water at the start of the acclimatization. In order to have sufficient food supply, 10 mg of dry food for *A. virens* and 3 mg (shell length  $\geq$  15 mm) or 1.5 mg (shell length  $\leq$  15 mm) for *L. balthica* was added per individual.

Water quality variables such as pH, dissolved oxygen, salinity and temperature were measured every two days during exposure time (Tables S2 and S3).







#### **Sediment collection**

A moderately contaminated sub-toxic marine sediment was prepared by mixing a non-contaminated 'clean' sediment with a naturally contaminated harbour sediment. Batches of muddy sediment (132 L) and coarse sediment (176 L) were collected in the Oosterchelde estuary (The Netherlands, 51°36'13"N, 3°47'49"E) at low tide on 24 to 25 April 2014 and 2 May 2014. Coarse sediment (0.5 – 1 mm grain size) was sieved in the field with a 1-mm sieve whereas muddy sediment (< 0.5 mm grain size) was sieved at laboratory inside a 1-mm sieve with some filtered sea water (0.2 μm). Once sieved both samples were stored at 3 °C.

A batch of harbour sediment (30 L) was collected near Nesserdijk (Rotterdam, the Netherlands, 51°54'14"N, 4°31'17"E) by Nautisch Service Centrum in April 2014 and sieved with a 1-mm sieve on 1 May 2014. A subsample of the sieved wet harbour sediment (800 mL) was sent to Wageningen Marine Research laboratories (IJmuiden, the Netherlands) for further chemical analyses (Table S4).

## **Test species**

Arctic experiment. A. borealis, M. calcarea and N. ciliata were obtained as described in the field sampling section. A. borealis randomly selected healthy individuals with an average and standard deviation (SD) shell length of 31.30 (4.63) mm and wet wt. (flesh) of 7.47 (4.16) g were used. As for M. calcarea and N. ciliata, we did not measure any morphological parameters due to the scarcity of individuals. Animals were not fed during the acclimatization, neither during the exposure experiments.

Temperate experiment. L. balthica and A. virens were obtained as described in the field sampling section. L. balthica randomly selected healthy individuals with an average (SD) shell length of 13.49 (2.46) mm and wet wt. (shell + flesh) of 0.47 (0.31) g were used. A. virens randomly selected healthy individuals with an average (SD) length of 10.83 (1.70) cm and wet wt. of 2.75 (0.98) g were used. Animals were fed once at the beginning of the acclimatization.

# Pyrene metabolites analysis

Chemicals. Pyrene (98%) and 1-hydroxypyrene (OHPyr) (98%) were purchased from Sigma Aldrich. Pyrenyl-1-O-glucuronide (GluPyr) (98.2%) was supplied by Isostandards Material.







The dilutions at lower concentrations were prepared daily, depending on the experiment. All chemical standards were stored at 4  $^{\circ}$ C and stock solutions at -20  $^{\circ}$ C. Acetone, methanol and acetonitrile (all HPLC grade) were purchased from Macrom Fine Chemicals. For filtration 0.45  $\mu$ m polyamide filters (Macherey Nagel) were used. Oasis HLB (200 mg) cartridges were purchased from Waters in order to perform clean-up step.

High-performance liquid chromatograph with a fluorescence detector (HPLC-FLD) analysis. A high-performance liquid chromatograph (Agilent Technologies, Series 1100) with a fluorescence detector (FLD, Agilent Technologies, Series 1100) was used for all the quantitative measurements. The chromatographic method was modified from Beach et al. [292]. A Supelcolsil LC-PAH HPLC column (10 cm  $\times$  4.6 mm, 3  $\mu$ m) was used for the separation. Mobile phases were prepared: of A: only Milli-Q water, B: methanol:acetonitrile:Milli-Q water (38:57:5, v/v/v), both with 10 mM ammonium acetate. The solvent gradient was 15 min, starting with 20% B with a linear gradual increase to 100% B from 0 to 2.5 min with a 4 min hold at 100% B and continued with a linear decrease to 80% B from 6.5 to 8.5 min with a 6.5 min hold at 80% B. An injection volume of 5  $\mu$ L and a flow rate of 800  $\mu$ L/min were used throughout. Fluorescence detection was performed at an excitation/emission wavelength pair of 235/388 nm for pyrene, OHPyr and GluPyr.

Limits of detection (LODs) were 0.2, 0.35 and 0.20 ng/mL for pyrene, OHPyr and GluPyr, respectively. LODs were expressed in biomass units taking 0.4, 0.8, 0.05 and 1.7 g as the average wet wt. per sample of *M. calcarea*, *N. ciliata*, *L. balthica*, *A. virens*, respectively. Therefore, pyrene, OHPyr and GluPyr LODs were recalculated as 0.5, 0.88 and 0.5 ng/g for *M. calcarea*; 0.25, 0.44 and 0.25 ng/g for *N. ciliata*; 4, 7 and 4 ng/g for *L. balthica*; and 0.12, 0.21 and 0.12 ng/g for *A. virens*. Samples with pyrene or GluPyr below the limit of quantification (LOQ) were assigned a "LOQ/2" value.







Table S1. Overview of the number of experimental aquaria per species and sediment treatment.

				Trea	atment (	$(n)^a$	
Experiment	Species	Ind./aq.b	BSL	BSM	BSH	PSL	PSH
	Macoma calcarea	40-50	2	n.a.	2		
Arctic	Astarte borealis	25	4	4	6		
	Nephtys ciliata	20				2	3
Tomanomoto	Limecola balthica	50	2	n.a.	3		
Temperate	Alitta virens	16				3	4

<sup>&</sup>lt;sup>a</sup> n: number of replicates per sediment treatment. <sup>b</sup> Ind./aq.: number of organisms per aquarium.

BSL: Low Bivalve Sediment; BSM: Medium Bivalve Sediment; BSH: High Bivalve Sediment; PSL: Low Polychaete Sediment; PSH: High Polychaete Sediment; n.a.: not available







Table S2. Mean water quality parameters in the Arctic experiment H treatment aquaria.

	(°C)	2.9	2.9	2.9	2.9	2.9	2.9	2.9	2.8	2.9
a - BSH	Salinity (g/L)	35.6	35.7	35.8	35.8	35.9	35.7	36.0	35.8	36.0
Macoma calcarea - BSH	D.O. (mg/L)	8.8	8.9	9.0	9.2	9.3	8.7	9.1	9.4	9.1
V	D.O. (%)	81.0	82.5	84.0	85.0	85.5	81.5	83.5	86.0	82.0
	Hd	7.9	7.9	8.0	7.9	7.9	7.8	7.9	8.0	n.a.
	(°C)	2.7	3.0	2.7	2.7	2.6	2.6	2.6	2.7	2.7
s - BSH	Salinity (g/L)	34.7	34.7	34.8	34.9	34.9	35.0	35.1	35.2	35.2
Astarte borealis - BSH	D.O. (mg/L)	0.6	8.8	6.8	9.6	0.6	9.1	9.2	9.1	9.0
	D.O. (%)	82.7	81.3	82.3	83.1	83.5	83.7	85.0	84.4	83.8
	Ηd	8.1	8.1	8.0	7.9	7.9	7.8	7.9	7.8	7.8
	Day	1	2	Э	4	5	9	7	∞	6

PSH
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(C) <sub>0</sub> ()	2.9	2.8	2.8	2.8	2.8	2.8	2.9	2.8	2.8
Salinity (g/L) T								35.8	
D.O. (mg/L)	8.7	8.8	6.8	6.8	8.9	8.9	8.8	9.2	9.0
D.O. (%)	81.2	81.5	82.5	82.8	82.3	82.7	81.7	84.8	85.3
μd	8.0	8.0	8.0	8.0	7.9	7.9	8.0	8.0	8.0
	-								

D.O.: dissolved oxygen; n.a.: not available. Additional abbreviations are defined in Table S1.







Table S3. Mean water quality parameters in the temperate experiment H treatment aquaria.

Limecola balthica - BSH

Day	рН	D.O. (%)	D.O. (mg/L)	Salinity (g/L)	T (°C)
1	8.2	83.8	6.6	32.8	17.8
2	n.a.	80.3	6.3	n.a.	17.8
3	n.a.	79.3	6.2	n.a.	17.7
4	8.2	80.8	6.2	33.3	18.0
5	8.2	82.0	6.4	33.5	18.1
6	8.2	82.2	6.4	33.6	17.9
7	8.3	84.2	6.5	34.1	17.8
8	8.2	84.2	6.5	34.5	17.8
9	8.3	86.0	6.6	34.7	18.1
10	8.3	85.2	6.6	35.0	17.9
11	8.3	85.3	6.6	35.4	17.9
12	8.3	84.5	6.5	35.7	17.9

Alitta virens - PSH

Day	рΗ	D.O. (%)	D.O. (mg/L)	Salinity (g/L)	T (°C)
1	8.1	79.4	6.2	33.8	18.1
2	n.a.	77.9	6.1	n.a.	18.1
3	n.a.	75.3	5.9	n.a.	17.9
4	8.1	76.7	5.9	34.4	18.2
5	8.2	78.9	6.1	34.0	18.2
6	8.2	80.7	6.2	34.5	18.0
7	8.2	80.9	6.2	35.2	18.0
8	8.2	81.6	6.3	35.1	18.0
9	8.2	79.7	6.0	35.5	18.3
10	8.2	82.1	6.3	34.8	18.0
11	8.2	82.4	6.3	35.2	18.0
12	8.2	81.4	6.2	35.7	17.9

D.O.: dissolved oxygen; n.a.: not available. Additional abbreviations are defined in Table S1.







**Table S4.** Polycyclic aromatic hydrocarbons, polychlorinated biphenyls and polybrominated diphenyl ethers concentrations in Rotterdam harbour sediment (Nesserdijk, Rotterdam).

Compound	Mean (μg/kg dry wt.)
PHE	1378
ANT	459
FLT	2551
PYR	1811
BaA	1174
CHR	1352
BeP	510
BbF	1403
BkF	536
BaP	1148
BghiP	612
dBahA	135
Ind123P	1148
CB-28	< 0.20
CB-52	0.64
CB-101	0.97
CB-118	0.89
CB-138	0.97
CB-153	0.64
CB-180	< 0.50
BDE-28	0.23
BDE-47	2.04
BDE-99	6.12
BDE-100	0.51
BDE-153	10.71
BDE-154+BDE-155	0.26

PHE: phenanthrene; ANT: anthracene; FLT: fluoranthene; PYR: pyrene; BaA: benz(a)anthracene; CHR: chrysene; BeP: benzo(e)pyrene; BbF: benzo(b)fluoranthene; BkF: benzo(k)fluoranthene; BaP: benzo(a)pyrene; BghiP: benzo(ghi)perylene; dBahA: dibenz(a,h)anthracene; Ind123P: indeno(1,2,3-cd)pyrene.







Table S5. Polycyclic aromatic hydrocarbons concentrations in sediment treatments at the start of the Arctic experiment. Source: Chapter 3.

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Compaind (110/kg dry wt)	BSI	ر	BSH	ж	PSI	Γ	PSM	$\mathbb{Z}$	PSH	T
Compound (pg/ng and we.)	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
OM (% dry wt.)	2.05	n.a.	2.52	0.03	6.02	n.a.	3.89	n.a.	3.76	n.a.
PHE	48.11	8.46	164.69	48.81	155.03	26.35	5 159.71 15.59 250.26 29.11	15.59	250.26	29.11
	8.52	1.25	61.62	14.90	34.09	3.75	49.27	9.77	81.30	8.37
FLT	58.83	11.28	306.13	128.02	249.20	35.11	288.08	58.13	406.93	52.78
	42.94	5.80	240.91	109.12	180.19	24.54	238.76	75.43	313.86	38.50
	15.45	3.14	148.82	70.13	83.17	20.31	121.87	54.17	156.96	18.12
	n.d.	n.d.	92.95	59.11	19.64	11.43	80.57	41.76	170.90	110.17
	25.42	7.53	209.65	88.39	128.04	19.40	216.94	117.22	251.10	31.37
	30.44	6.45	174.13	69.34	145.95	16.18	166.15	32.69	234.81	30.82
	0.80	0.65	11.45	7.23	8.17	2.22	13.46	5.70	18.84	6.94
	32.09	12.89	146.19	58.01	129.82	17.92	140.35	27.26	203.35	25.71
Ind123P	24.12	3.57	152.92	08.09	150.68	20.01	154.70	32.43	218.81	39.02
$\Sigma_{13}$ PAH	286.72	58.50	1709.46	709.16	1283.99	184.63	1629.86	470.15	2307.13	162.56

OM: organic matter; n.a.: not available; n.d.: not detected. Additional abbreviations are defined in Tables S1 and S4.

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**Table S6.** Polycyclic aromatic hydrocarbons concentrations in sediment treatments at the end of the temperate experiment (t = 28 d). Source:

Treatment	BS	J	BS	H	PS	3T	PSI	M	PS	Н
Compound (µg/kg dry wt.)	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
OM (% dry wt.)	2.18	90.0	2.73	0.76	3.52	0.04	4.01	0.04	4.00	0.04
PHE	53.71	4.21	229.66	42.90	159.69			98.45	381.60	258.88
ANT	14.43	2.23	99.86	48.61	39.03			35.04	128.92	85.59
FLT	80.04	62.6	460.54	196.09	316.06			125.39	98.799	280.22
PYR	60.09	7.37	349.31	149.85	254.04			101.17	509.52	215.43
BaA	21.79	3.26	227.37	130.50	140.35			32.66	256.14	70.35
CHR	n.d.	n.d.	158.98	125.12	220.13	340.36	113.51	38.92	207.44	207.44 127.47
BeP	34.62	6.50	352.06	195.45	205.57			74.64	479.18	250.18
BbF	42.21	5.79	255.53	116.72	168.62			63.64	387.37	169.03
BghiP	4.26	3.91	19.16	10.47	19.88			4.89	42.54	25.68
dBahA	49.43	11.12	212.64	82.18	127.75			82.85	352.82	194.89
Ind123P	39.37		205.33	88.24	136.06			54.36	374.39	202.59
$\Sigma_{13}$ PAH	400.01	$\overline{}$	2569.23	1182.50	1787.18	1103.86		698.72	3787.26	1880.29

AUU.UI 32.U3 | L209.L3 | L102.30 | L767.18 | 1103.80 | L531. OM: organic matter; n.d.: not detected. Additional abbreviations are defined in Tables S1 and S4







Table S7. Polycyclic aromatic hydrocarbons concentrations (mean ± SD; μg/kg lipid wt.) in biota samples collected from the Arctic experiment. Source: Chapter 3.

Species	Масота	Macoma calcarea	$Ne_{p}$	Nephtys ciliata	iata	Astarte borealis	orealis
Treatment	BSL	BSH	PSL	PSM	PSH	Background	BSH
PHE	$3566 \pm 393$	$2821 \pm 287$	$1402 \pm 681$	729	$2462 \pm 941$	$9697 \pm 1482$	$11746 \pm 4998$
ANT	$68 \mp 068$	$718 \pm 48$	$336 \pm 143$	149	$297 \pm 120$	$2384 \pm 442$	$2977 \pm 957$
FLT	$1612 \pm 178$	$3329 \pm 1174$	$2353 \pm 788$	1771	$4468 \pm 969$	$1376 \pm 280$	$8504 \pm 4206$
PYR	$1314 \pm 87$	$2623 \pm 930$	$1713 \pm 530$	1384	$3758 \pm 920$	$1119 \pm 120$	$6926 \pm 3575$
BaA	n.d.	$673 \pm 240$	$113 \pm 70$	77	$218 \pm 104$	n.d.	n.d.
CHR	$7434 \pm 6614$	$470 \pm 113$	$1327 \pm 1709$	0.15	$6425 \pm 6647$	$6164 \pm 1276$	$2926 \pm 2658$
BeP	$316 \pm 0$	$957 \pm 374$	$427 \pm 178$	298	$650 \pm 129$	$884 \pm 169$	$2066 \pm 2300$
BbF	$175 \pm 30$	$658 \pm 305$	$117 \pm 78$	100	$228 \pm 69$	$223 \pm 56$	$845 \pm 269$
BghiP	$1162 \pm 745$	$50 \pm 65$	$344 \pm 454$	n.d.	$1106 \pm 955$	$1409 \pm 2154$	n.d.
dBahA	$275 \pm 2$	$506 \pm 133$	$166 \pm 76$	40	n.d.	n.d.	$479 \pm 233$
Ind123P	$69 \pm 40$	$298 \pm 102$	$34 \pm 37$	n.d.	n.d.	$205 \pm 144$	$421 \pm 187$
$\Sigma_{13}$ PAH	$16933 \pm 6714$	$13104 \pm 3552$	$8333 \pm 267$	4550	$20242 \pm 9762$	$34245 \pm 20336$	$37668 \pm 17371$
n.d.: not dete	n.d.: not detected. Additional abbreviations are defined in Tables S1 and S4.	abbreviations are	defined in Table	es S1 and	d S4.		





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Table S8. Polycyclic aromatic hydrocarbons concentrations (mean ± SD; μg/kg lipid wt.) in biota samples collected from the temperate experiment. Source: Chapter 3.

Species	Lime	Limecola balthica	hica		Alitta 1	irens	
Treatment	Background	BSL	BSH	Background	PSL	PSM	PSH
PHE	$2499 \pm 571$	2849	$2257 \pm 165$	$2497 \pm 798$	$2267 \pm 1376$	$1324 \pm 496$	$2278 \pm 2000$
ANT	$225 \pm 60$	262	$334 \pm 102$	$136 \pm 64$	$108 \pm 70$	$100 \pm 27$	$113 \pm 96$
FLT	$1431 \pm 394$	2853	$3061 \pm 685$	$5421 \pm 1780$	$1596 \pm 598$	$1824 \pm 664$	$3151 \pm 1903$
PYR	$1484 \pm 325$	3513	$4340 \pm 1153$	$1611 \pm 530$	$893 \pm 379$	$1157 \pm 419$	$2034 \pm 1254$
BaA	$680 \pm 177$	1101	$2547 \pm 484$	n.d.	$83 \pm 62$	$134 \pm 19$	$104 \pm 51$
CHR	$863 \pm 98$	964	$11585 \pm 9242$	$268 \pm 278$	n.d.	$887 \pm 1061$	n.d.
BeP	$937 \pm 247$	1618	$2833 \pm 702$	$125 \pm 50$	$249 \pm 220$	$279 \pm 102$	$475 \pm 286$
3bF	$938 \pm 259$	1442	$2295 \pm 432$	$51 \pm 26$	$100 \pm 88$	$151 \pm 26$	$125 \pm 45$
3ghiP	$2216 \pm 4746$	47	$1754 \pm 1454$	99 <del>+</del> 9 <i>L</i>	n.d.	$134 \pm 134$	n.d.
iBahA	$717 \pm 491$	702	$1083 \pm 818$	$83 \pm 29$	$115 \pm 79$	$137 \pm 36$	$239 \pm 238$
Ind123P	$452 \pm 206$	448	$839 \pm 473$	n.d.	$62 \pm 89$	$96 \pm 56$	$75 \pm 62$
Σι3PAH	$35512 \pm 56664$	15799	$32928 \pm 13246$	$10360 \pm 3099$	$8005 \pm 5818$	$6223 \pm 2229$	$9371 \pm 5439$

n.d.: not detected. Additional abbreviations are defined in Tables S1 and S4







**Table S9.** Pyrene (Pyr), 1-hydroxypyrene (OHPyr) and pyrene-1-glucuronide (GluPyr) concentrations in polychaete samples (ng/g wet wt.).

Species	Treatment	Aquarium	Pyr	OHPyr	GluPyr
A. virens	L	PSL-1	1.5	1.5	10
A. virens	L	PSL-1	4.7	<lod< td=""><td>2.4</td></lod<>	2.4
A. virens	L	PSL-1	5.2	<lod< td=""><td>2.9</td></lod<>	2.9
A. virens	L	PSL-1	0.5	<lod< td=""><td>2.2</td></lod<>	2.2
A. virens	Ĺ	PSL-2	5.4	<lod< td=""><td>3.5</td></lod<>	3.5
A. virens	Ĺ	PSL-2	1	<lod< td=""><td>2.7</td></lod<>	2.7
A. virens	Ĺ	PSL-2	3.4	<lod< td=""><td>3.2</td></lod<>	3.2
A. virens	L	PSL-2	1.3	<lod< td=""><td>9.2</td></lod<>	9.2
A. virens	L	PSL-3	1.5	<lod< td=""><td>3.2</td></lod<>	3.2
A. virens	L	PSL-3	0.6	<lod< td=""><td>3.6</td></lod<>	3.6
A. virens	L	PSL-3	1.8	<lod< td=""><td>6.1</td></lod<>	6.1
A. virens	L	PSL-3	1.0	<lod< td=""><td>4.3</td></lod<>	4.3
A. virens	H	PSH-1	1.4	<lod< td=""><td>8.1</td></lod<>	8.1
	Н		0.9		
A. virens		PSH-1		<lod< td=""><td>7.6</td></lod<>	7.6
A. virens	H	PSH-1	2	<lod< td=""><td>5.7</td></lod<>	5.7
A. virens	H	PSH-2	1.6	<lod< td=""><td>13.7</td></lod<>	13.7
A. virens	H	PSH-2	0.5	<lod< td=""><td>11.8</td></lod<>	11.8
A. virens	H	PSH-3	3.6	<lod< td=""><td>5.1</td></lod<>	5.1
A. virens	H	PSH-3	1.4	<lod< td=""><td>4.6</td></lod<>	4.6
A. virens	H	PSH-3	1.1	<lod< td=""><td>7</td></lod<>	7
A. virens	Н	PSH-3	1.7	<lod< td=""><td>4.8</td></lod<>	4.8
A. virens	H	PSH-4	4.3	<lod< td=""><td>6.2</td></lod<>	6.2
A. virens	H	PSH-4	3.1	<lod< td=""><td>6.4</td></lod<>	6.4
A. virens	H	PSH-4	0.6	<lod< td=""><td>4.5</td></lod<>	4.5
N. ciliata	L	PSL-1	3	<lod< td=""><td>4.1</td></lod<>	4.1
N. ciliata	L	PSL-1	2.8	0.6	10
N. ciliata	L	PSL-1	11.8	<lod< td=""><td>6.9</td></lod<>	6.9
N. ciliata	L	PSL-1	15.5	<lod< td=""><td>7.5</td></lod<>	7.5
N. ciliata	L	PSL-1	2.6	<loq< td=""><td>9.1</td></loq<>	9.1
N. ciliata	L	PSL-1	14.3	<lod< td=""><td>8.7</td></lod<>	8.7
N. ciliata	L	PSL-2	2.7	0.5	8.3
N. ciliata	L	PSL-2	2	0.8	5.6
N. ciliata	L	PSL-2	15.3	< LOD	13.8
N. ciliata	L	PSL-2	2.7	<loq< td=""><td>4.5</td></loq<>	4.5
N. ciliata	L	PSL-2	10.1	<lod< td=""><td>8.9</td></lod<>	8.9
N. ciliata	L	PSL-2	13.8	< LOD	20.9
N. ciliata	H	PSH-2	28.5	< LOD	22.2
N. ciliata	H	PSH-2	19.1	<lod< td=""><td>15.7</td></lod<>	15.7
N. ciliata	Н	PSH-2	5.2	2.5	8.2
N. ciliata	H	PSH-2	27	<lod< td=""><td>10.9</td></lod<>	10.9
N. ciliata	H	PSH-2	4.2	1.5	7.8
N. ciliata	H	PSH-2	4.3	<lod< td=""><td>10</td></lod<>	10
N. ciliata	H	PSH-5	3.6	3.7	6.4
N. ciliata	H	PSH-5	4.6	2.6	20.6
N. ciliata	H	PSH-5	15.8	<lod< td=""><td>10.7</td></lod<>	10.7
N. ciliata	H	PSH-6	15.8	<lod< td=""><td>9.3</td></lod<>	9.3
N. ciliata	H	PSH-6	3.8	<lod< td=""><td>14.5</td></lod<>	14.5
N. ciliata	Н	PSH-6	16.9	<lod< td=""><td>10</td></lod<>	10

L: low treatment; H: high treatment; LOD: limit of detection; LOQ: limit of quantification.

LOQs: 0.39 ng/mL (Pyr), 0.9 ng/mL (OHPyr) and 0.38 ng/mL (GluPyr).







**Table S10.** Pyrene (Pyr), 1-hydroxypyrene (OHPyr) and pyrene-1-glucuronide (GluPyr) concentrations in bivalve samples (ng/g wet wt.).

Species	Treatment	Aquarium	Pyr	OHPyr	GluPyr
L. balthica	L	BSL-1	17.90	2.18	47.51
L. balthica	L	BSL-1	16.50	2.04	40.94
L. balthica	L	BSL-1	10.50	1.87	48.87
L. balthica	L	BSL-1	11.60	<lod< td=""><td>5.40</td></lod<>	5.40
L. balthica	L	BSL-1	22.70	<lod< td=""><td>5.90</td></lod<>	5.90
L. balthica	L	BSL-1	19.69	2.20	56.60
L. balthica	L	BSL-1	17.90	<lod< td=""><td>7.40</td></lod<>	7.40
L. balthica	L	BSL-1	13.70	<lod< td=""><td>6.70</td></lod<>	6.70
L. balthica	L	BSL-1	22.30	<lod< td=""><td>12.50</td></lod<>	12.50
L. balthica	L	BSL-1	11.30	<lod< td=""><td>6.00</td></lod<>	6.00
L. balthica	L	BSL-1	36.38	2.98	73.97
L. balthica	L	BSL-1	10.00	1.19	56.36
L. balthica	L	BSL-2	1.30	<lod< td=""><td>1.30</td></lod<>	1.30
L. balthica	L	BSL-2	2.40	<lod< td=""><td><loq< td=""></loq<></td></lod<>	<loq< td=""></loq<>
L. balthica	L	BSL-2	14.12	1.83	50.06
L. balthica	L	BSL-2	9.23	1.40	46.43
L. balthica	L	BSL-2	9.09	1.56	34.64
L. balthica	L	BSL-2	13.50	<lod< td=""><td>11.70</td></lod<>	11.70
L. balthica	L	BSL-2	6.45	1.06	45.73
L. balthica	L	BSL-2	11.00	<lod< td=""><td>6.50</td></lod<>	6.50
L. balthica	L	BSL-2	13.60	<lod< td=""><td>4.80</td></lod<>	4.80
L. balthica	L	BSL-2	13.20	<lod< td=""><td>15.10</td></lod<>	15.10
L. balthica	L	BSL-2	24.07	2.66	67.66
L. balthica	L	BSL-2	7.40	<lod< td=""><td>22.80</td></lod<>	22.80
L. balthica	L	BSL-2	8.90	<lod< td=""><td>5.50</td></lod<>	5.50
L. balthica	H	BSH-1	8.04	0.88	16.43
L. balthica	Н	BSH-1	14.49	1.26	44.81
L. balthica	H	BSH-1	11.40	1.27	37.22
L. balthica	H	BSH-1	18.00	<lod< td=""><td>1.80</td></lod<>	1.80
L. balthica	H	BSH-1	13.00	<lod< td=""><td>1.40</td></lod<>	1.40
L. balthica	H	BSH-1	11.90	<lod< td=""><td>5.00</td></lod<>	5.00
L. balthica	H	BSH-1	13.10	<lod< td=""><td>2.20</td></lod<>	2.20
L. balthica	H	BSH-2	3.30	<lod< td=""><td>2.10</td></lod<>	2.10
L. balthica	H	BSH-2	6.34	0.96	26.35
L. balthica	H	BSH-2	6.89	2.03	24.31
L. balthica	H	BSH-2	16.72	2.22	48.66
L. balthica	H	BSH-2	2.90	<lod< td=""><td>3.90</td></lod<>	3.90
L. balthica	H	BSH-2	10.88	1.47	43.56
L. balthica	H	BSH-2	19.60	<lod< td=""><td>5.00</td></lod<>	5.00
L. balthica	H	BSH-2	14.31	1.66	36.31
L. balthica	H	BSH-2	<loq< td=""><td><lod< td=""><td>1.50</td></lod<></td></loq<>	<lod< td=""><td>1.50</td></lod<>	1.50
L. balthica	H	BSH-2	12.10	<lod< td=""><td>4.50</td></lod<>	4.50
L. balthica	H	BSH-3	16.25	0.99	50.65
L. balthica	H	BSH-3	12.55	1.75	47.08
L. balthica	H	BSH-3	6.03	1.01	22.13
L. balthica	H	BSH-3	11.85	1.30	18.75
L. balthica	H	BSH-3	22.60	<lod< td=""><td>11.10</td></lod<>	11.10
L. balthica	H	BSH-3	21.50	<lod< td=""><td>22.70</td></lod<>	22.70
L. balthica	H	BSH-3	21.00	<lod< td=""><td>14.20</td></lod<>	14.20
L. balthica	H	BSH-3	26.70	<lod< td=""><td>31.30</td></lod<>	31.30
M. calcarea	L	BSL-1	2.06	0.68	4.24
M. calcarea	L	BSL-1	0.70	1.10	0.70
M. calcarea	L	BSL-1	2.49	0.54	11.09
M. calcarea	L	BSL-1	5.11	0.88	23.48
M. calcarea	L	BSL-1	0.70	0.90	0.30
M. calcarea	L	BSL-1	2.16	0.69	8.37







Table S10. Continued.

g :	TD		D	OIID	CI D
Species	Treatment	Aquarium	Pyr	OHPyr	GluPyr
M. calcarea	L	BSL-1	3.95	0.74	18.65
M. calcarea	L	BSL-1	4.10	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
M. calcarea	L	BSL-1	2.80	1.10	<loq< td=""></loq<>
M. calcarea	L	BSL-1	5.50	<lod< td=""><td>5.30</td></lod<>	5.30
M. calcarea	L	BSL-1	6.52	0.79	24.78
M. calcarea	L L	BSL-1	3.50	<lod< td=""><td><loq< td=""></loq<></td></lod<>	<loq< td=""></loq<>
M. calcarea		BSL-1	5.41	0.89	<loq< td=""></loq<>
M. calcarea	L L	BSL-1	5.20	1.50	<lod< td=""></lod<>
M. calcarea	_	BSL-1	4.76	0.55	15.50
M. calcarea	L	BSL-1	2.92	0.33	9.40
M. calcarea	L	BSL-1	12.25	0.93	25.12
M. calcarea	L	BSL-1	5.54	1.00	21.15
M. calcarea	L	BSL-1	3.99	0.70	26.40
M. calcarea	L	BSL-1	3.10	<lod< td=""><td>3.20</td></lod<>	3.20
M. calcarea	L	BSL-1	5.30	0.95	34.72
M. calcarea	L	BSL-1	4.70	<lod< td=""><td>2.90</td></lod<>	2.90
M. calcarea	L	BSL-1	5.64	0.79	17.74
M. calcarea	Ĺ	BSL-2	2.34	0.51	6.80
M. calcarea	L	BSL-2	3.30	<lod< td=""><td>1.70</td></lod<>	1.70
M. calcarea	H	BSH-6	2.30	<lod< td=""><td><lod< td=""></lod<></td></lod<>	<lod< td=""></lod<>
M. calcarea	H	BSH-6	1.20	0.60	<loq< td=""></loq<>
M. calcarea	H	BSH-6	5.90	<lod< td=""><td>0.50</td></lod<>	0.50
M. calcarea	H	BSH-6	11.40	<lod< td=""><td>1.70</td></lod<>	1.70
M. calcarea	H	BSH-6	1.20	<lod< td=""><td>0.30</td></lod<>	0.30
M. calcarea	H	BSH-6	1.10	1.10	0.80
M. calcarea	H	BSH-6	4.30	<lod< td=""><td>0.40</td></lod<>	0.40
M. calcarea	Н	BSH-6	16.67	1.17	33.63
M. calcarea	Н	BSH-6	8.10	<loq< td=""><td>2.00</td></loq<>	2.00
M. calcarea	Н	BSH-6	2.90	<lod< td=""><td>3.00</td></lod<>	3.00
M. calcarea	Н	BSH-6	6.30	<lod< td=""><td>3.90</td></lod<>	3.90
M. calcarea	Н	BSH-6	8.30	<loq< td=""><td>0.80</td></loq<>	0.80
M. calcarea	Н	BSH-6	13.07	1.23	15.24
M. calcarea	Н	BSH-6	21.30	<loq< td=""><td>2.50</td></loq<>	2.50
M. calcarea	Н	BSH-6	17.34	1.30	24.78
M. calcarea	Н	BSH-6	17.39	1.66	40.16
M. calcarea	Н	BSH-6	21.18	1.50	30.23
M. calcarea	Н	BSH-6	3.54	1.28	32.48
M. calcarea	Н	BSH-6	6.11	0.89	34.12
M. calcarea	Н	BSH-6	5.25	0.86	38.24
M. calcarea	H	BSH-7	6.04	1.67	41.63
M. calcarea	H	BSH-7	5.72	1.37	15.23
M. calcarea	H	BSH-7	3.66	1.25	23.47
M. calcarea	H	BSH-7	3.78	1.63	29.22
M. calcarea	Н	BSH-7	4.83	1.84	36.21

L: low treatment; H: high treatment; LOD: limit of detection; LOQ: limit of quantification. LOQs: 0.39 ng/mL (Pyr), 0.9 ng/mL (OHPyr) and 0.38 ng/mL (GluPyr).







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**Table S11.** Mean acetylcholinesterase activity statistical comparison between high (BSH), medium (BSM) and low (BSL) treatment exposed *Astarte borealis* individuals.

					Descriptives			
Mean AChE	AChE	[47]						
	Z	Moon	Ç	20	95% confidence interval for mean	nterval for mean	Minimum	Maximo
	Ξ.	Mean	OS.	SE.	Lower bound Upper bound	Upper bound	IVIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	Maxillulli
BSL		0.0284	7 0.0284 0.00824 0.0031	0.00311	0.0207	0.0360	0.01	0.04
BSM 8	∞	0.0355	0.01005 0.00355	0.00355	0.0271	0.0439	0.02	0.05
BSH	∞	0.0353	0.01614	0.00571	0.0218	0.0488	0.00	0.05
Total	23	Total 23 0.0333	0.01202 0.00251	0.00251	0.0281	0.0385	0.00	0.05

Fest of homogeneity of variances	hE	atistic dfl df2 Sig.	2 20 0.333
Test of ho	Mean AChE	Levene statistic	1.163

		A	ANOVA				
Mean AChE							
			Sum of squares df Mean square	дþ	Mean square	H	Sig.
	(Combined)	ined)	0.000	7	0.000	0.827	0.827 0.452
		Unweighted	0.000	1	0.000	1.234	0.280
D	Linear term	Weighted	0.000	_	0.000	1.176	1.176 0.291
Detween groups		Deviation	0.000	1	0.000	0.478 0	0.497
		Unweighted	0.000	_	0.000	0.478	0.478 0.497
	Quadranc term	Weighted	0.000	_	0.000	0.478 0	0.497
Within groups			0.003	20	0.000		
Total			0.003	22			





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			Multiple comparisons	sons			
Mean AChE							
						95% confidence interval	nce interval
(I) Treatment	ent		Mean difference (I-J)	SE	Sig.	Lower bound	Upper bound
	19.0	BSM	-0.00715	0.00715 0.00627 0.501	0.501	-0.0230	0.0087
	BSL	BSH	-0.00697	0.00627	0.519	-0.0228	0.0089
Golf Light	7,00	BSL	0.00715	0.00627	0.501	-0.0087	0.0230
т икеу пэр	BSM	BSH	0.00019	0.00606	0.999	-0.0151	0.0155
	1120	BSL	0.00697	0.00627	0.519	-0.0089	0.0228
	рэп	BSM	-0.00019	0.00606	0.999	-0.0155	0.0151
	ISG	BSM	-0.00715	0.00472	0.317	-0.0196	0.0053
	BSL	BSH	-0.00697	0.00650	0.550	-0.0246	0.0107
	7,00	BSL	0.00715	0.00472	0.317	-0.0053	0.0196
Games-ноwеп	BSM	BSH	0.00019	0.00672	1.000	-0.0178	0.0182
	IISG	BSL	0.00697	0.00650	0.550	-0.0107	0.0246
	БЭП	BSM	-0.00019	0.00672	1.000	-0.0182	0.0178
	BSM	BSL	0.00715	0.00627	0.215	-0.0055	
Dunnett t (>control)"	BSH	BSL	0.00697	0.00627	0.223	-0.0057	

<sup>&</sup>lt;sup>a</sup> Dunnett t-tests treat one group as a control, and compare all other groups against it.







Table S12. Mean glutathione S-transferase activity statistical comparison between high (BSH), medium (BSM) and low (BSL) treatment exposed Astarte borealis individuals.

					Descriptives			
Mean GST	GST							
	7	Maga	Ę	Ç	95% confidence interval for mean	nterval for mean		
	Z	Mean	SD	oe.	Lower bound	Upper bound	MINITINI	Maximum
BSL		49.7688	8 49.7688 26.64573 9.42069	9.42069	27.4924	72.0452	18.26	95.48
BSM	$\infty$	47.8317	47.8317 20.47659 7.23957	7.23957	30.7128	64.9505	25.99	84.19
BSH	11	46.5008	BSH 11 46.5008 15.16479 4.57236	4.57236	36.3130	56.6887	27.73	79.98
Total	27	47.8634	Total 27 47.8634 19.85929 3.82192	3.82192	40.0074	55.7195	18.26	95.48

ances		Sig.	0.170
of vari		df2	24
neity o		dfl	2
Test of homogeneity of variances	Mean GST	Levene statistic dfl df2	1.909

		AN	ANOVA				
Mean GST							
			Sum of squares df Mean square	дþ	Mean square	Н	Sig.
	(Combined)	ned)	49.475	2	24.737	0.058	0.058 0.944
		Unweighted	49.463	-	49.463	0.116	0.116 0.736
0	Linear term	Weighted	48.961	_	48.961	0.115	0.115 0.737
Between groups		Deviation	0.513	_	0.513	0.001	0.973
	or of support	Unweighted	0.513	1	0.513	0.001	0.001 0.973
	Quadranc term	Weighted	0.513	1	0.513	0.001	0.001 0.973
Within groups			10204.707	24	425.196		
Total			10254.181	26			





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			Multiple comparisons	risons			
Mean GST						95% confidence interval	nce interval
(I) Treatment	ent		Mean difference (I-J)	SE	Sig.	Lower bound Upper bound	Upper bound
	Inci	BSM	1.93708	1.93708 10.31014 0.981	0.981	-23.8103	27.6845
	BSL	BSH	3.26796	9.58143 0.938	0.938	-20.6596	27.1955
Tulon	Deng	BSL	-1.93708	-1.93708 10.31014 0.981	0.981	-27.6845	23.8103
Tukey H3D	DSIM	BSH	1.33088	9.58143 0.989	0.989	-22.5967	25.2584
	וואם	BSL	-3.26796	-3.26796 9.58143 0.938	0.938	-27.1955	20.6596
	руп	BSM	-1.33088	9.58143 0.989	0.989	-25.2584	22.5967
B(1	BSM	BSL	-1.93708	-1.93708 10.31014 0.730	0.730	-22.5711	
Dunnett (>control)"	BSH	BSL	-3.26796	-3.26796 9.58143 0.782	0.782	-22.4436	

<sup>&</sup>lt;sup>a</sup> Dunnett t-tests treat one group as a control, and compare all other groups against it.

Table S13. Mean acyl-CoA oxidase activity statistical comparison between high (BSH), medium (BSM) and low (BSL) treatment exposed Astarte borealis individuals.

	Test statistics <sup>a,b</sup>		
		N	Mean AOX
Chi-Square			0.050
Df			2
Asymp. sig.			0.976
	Sig.		0.977°
Monte Carlo sig.	000% comply down	Lower bound	0.973
	99 70 collidence interval	Upper bound	0.981

<sup>&</sup>lt;sup>a</sup> Kruskal Wallis test





b Grouping variable: Treatment

<sup>°</sup> Based on 10000 sampled tables with starting seed 2000000.



**Table S14.** ANCOVA analysis of pyrene-1-glucuronide (GluPyr) concentration for polychaete species (*Alitta virens* and *Nephtys ciliata*). L: low treatment (PSL); H: high treatment (PSH).

Levene's t	est of equal	ity of error	variancesa
Dependent	variable: Lo	og_GluPyr	
F	df1	df2	Sig.
1.594	1	46	0.213

Tests the null hypothesis that the error variance of the dependent variable is equal across groups.

<sup>&</sup>lt;sup>a</sup> Design: Intercept + Log\_Pyr + Treatment

	Tests of between-su	ıbject	s effects		
Dependent variab	le: Log_GluPyr				
Source	Type III sum of squares	df	Mean square	F	Sig.
Corrected model	1.030 <sup>a</sup>	2	0.515	12.657	0.000
Intercept	10.735	1	10.735	263.876	0.000
Log_Pyr	0.607	1	0.607	14.933	0.000
Treatment	0.350	1	0.350	8.603	0.005
Error	1.831	45	0.041		
Total	37.255	48			
Corrected total	2.860	47			•

<sup>&</sup>lt;sup>a</sup> R squared = 0.360 (adjusted R squared = 0.332)

		Paramet	er estimates			
Dependent variab	le: Log_Glu	Pyr				
Parameter	r B SE t		_	G:-	95% confidence interval	
Parameter	В	SE	ι	Sig	Lower bound	Upper bound
Intercept	0.804	0.054	14.804	0.000	0.694	0.913
Log_Pyr	0.239	0.062	3.864	0.000	0.114	0.363
[Treatment=L]	-0.171	0.058	-2.933	0.005	-0.289	-0.054
[Treatment=H]	0 <sup>a</sup>	•	•		•	•

<sup>&</sup>lt;sup>a</sup> This parameter is set to zero because it is redundant.

	Ε	Estimates		
Dependent v	ariable: Log	_GluPyr		
Treatment	Mean	SE -	95% con inter	
Treatment	Mean	SE .	Lower	Upper
			bound	bound
L	0.761 <sup>a</sup>	0.041	0.678	0.844
Н	0.932a	0.041	0.849	1.015

 $<sup>^{\</sup>rm a}$  Covariates appearing in the model are evaluated at the following values: Log\_Pyr = 0.5385.







		P	airwise o	comparis	sons	
Depe	endent var	iable: Log_GluPyr				
(I) T.		Maam difference (L.D.	SE	Sig.b	95% confidence inter	val for difference <sup>b</sup>
(1) I reatment		Mean difference (I-J)	SE	Sig.	Lower bound	Upper bound
L	Н	-0.171*	0.058	0.005	-0.289	-0.054
Н	L	0.171*	0.058	0.005	0.054	0.289
Base	d on estin	nated marginal means				

<sup>\*</sup> The mean difference is significant at the 0.05 level.

**Table S15.** ANCOVA analysis of pyrene-1-glucuronide (GluPyr) concentration for bivalve species (*Limecola balthica* and *Macoma calcarea*). L: low treatment (BSL); H: high treatment (BSH).

Levene's test of eq	uality of err	or variance	s <sup>a</sup>		
Dependent variable: Log_GluPyr					
F	dfl	df2	Sig.		
0.108	1	95	0.743		
Tests the null hypothesis th	at the error v	ariance of	the		

dependent variable is equal across groups.

<sup>a</sup> Design: Intercept + Log Pyr + Treatment

	Tests of betwee	n-subjects	effects	·	
Dependent variable: Lo	g_GluPyr				
Source	Type III sum of squares	df	Mean square	F	Sig.
Corrected model	18.669 <sup>a</sup>	2	9.334	10.950	0.000
Intercept	0.012	1	0.012	0.014	0.906
Log_Pyr	18.393	1	18.393	21.576	0.000
Treatment	0.063	1	0.063	0.074	0.786
Error	80.130	94	0.852		
Total	166.421	97			
Corrected total	98.799	96			

<sup>&</sup>lt;sup>a</sup> R squared = 0.189 (adjusted R squared = 0.172)

Parameter estimates  Dependent variable: Log_GluPyr						
Lower bound	Upper bound					
Intercept	0.049	0.224	0.219	0.827	-0.395	0.493
Log_Pyr	0.964	0.208	4.645	0.000	0.552	1.377
[Treatment=L]	-0.051	0.188	-0.272	0.786	-0.424	0.322
[Treatment=H]	$0^{a}$					

<sup>&</sup>lt;sup>a</sup> This parameter is set to zero because it is redundant.





<sup>&</sup>lt;sup>b</sup> Adjustment for multiple comparisons: Bonferroni.



	Estimates			
Dependent variable: Log	g_GluPyr			
Treatment	Mean	SE.	95% con inter	
Treatment	Mean	SE -	Lower bound	Upper bound
L	$0.809^{a}$	0.133	0.544	1.074
Н	0.860a	0.132	0.598	1.122

<sup>&</sup>lt;sup>a</sup> Covariates appearing in the model are evaluated at the following values: Log\_Pyr = 0.8412.

		Pairwise	compari	sons		
Depe	ndent variable: Log_GluPyr					
	(I) Treatment	Mean difference	SE	Sig.a	95% confidence differe	
		(I-J)			Lower bound	Upper bound
L	Н	-0.051	0.188	0.786	-0.424	0.322
Н	L	0.051	0.188	0.786	-0.322	0.424
Basec	d on estimated marginal means					

<sup>&</sup>lt;sup>a</sup> Adjustment for multiple comparisons: Bonferroni.







Table S16. Multiple linear regression of pyrene-1-glucuronide (GluPyr) concentration for polychaete species (Alitta virens and Nephtys ciliata).

	Durhin	Watson	1.853
		Sig. F change	0.000
		df2	45
	hange statistics	lfb	2
yo.	Cha	F change	19.610
Aodel summary		R square change	0.466
M	SE of	the estimate	0.18429
	Adineted	R square	0.442
		R square	0.466
		2	$0.682^{a}$
		Model	1

<sup>a</sup> Predictors: (Constant), Treatment, Source

<sup>b</sup> Dependent variable: Log\_GluPyr

		ANOVAª			
Model	Jo mnS	J+C	Mean	Ĺ	6:5
INDOC	squares	∄	square	. 1	518.
Regression	1.332	2	999.0	19.610	$0.000^{b}$
Residual	1.528	45	0.034		
Total	2.860	47			

<sup>a</sup> Dependent variable: Log\_GluPyr

<sup>b</sup> Predictors: (Constant), Treatment, Source

					Coeffi	Coefficients <sup>a</sup>						
Model	Unstandar coefficie	ardized ients	Standardized coefficients			95% confidence interval for B	ifidence I for B		Correlations		Collinearity statistics	statistics
Model	В	SE	Beta	_	Z Sp	Lower	Upper bound	Zero- order	Partial	Part	Tolerance	VIF
(Constant)	0.152	0.116		1.312	0.196	-0.081	0.386					
1 Source	0.275	0.053	0.564	5.175	0.000	0.168	0.382	0.564	0.611	0.564	1.000	1.000
Treatment	0.188	0.053	0.384	3.526	0.001	0.080	0.295	0.384	0.465	0.384	1.000	1.000

<sup>a</sup> Dependent variable: Log\_GluPyr







Table S17. Multiple linear regression of pyrene-1-glucuronide (GluPyr) concentration for bivalve species (Limecola balthica and Macoma calcarea).

ary <sup>b</sup>	Change statistics	F change dfl df2 Sig. F Duroni- change dfl df2 change	3.764 2 94	
Model summary		R square F	0.074	
Mo	SE of	the estimate	0.98647	
	L 04011110 A	Aujusteu R square	0.054	(
		R square	0.074	E
		R	0.272ª	
		Model		E

<sup>a</sup> Predictors: (Constant), Treatment, Source
<sup>b</sup> Dependent variable: Log\_GluPyr

- 1			ANOVAª			
	Model	Sum of	J+C	Mean	Ĺ	Si.
	MODE	squares	ī	square	1	51g.
	Regression	7.326	2	3.663	3.764	0.027 <sup>b</sup>
_	Residual	91.473	94	0.973		
	Total	98.799	96			
ľ						

<sup>a</sup> Dependent variable: Log\_GluPyr

<sup>b</sup> Predictors: (Constant), Treatment, Source

**(** 

					Coeffi	Coefficients <sup>a</sup>						
Model	Unstandardized coefficients	rdized ients	Standardized coefficients	t	Sig.	95% confidence interval for B	fidence for B	J	Correlations		Collinearity statistics	statistics
	В	SE	Beta			Lower	Upper	Zero- order	Partial	Part	Tolerance	VIF
(Constant)	1.467	0.433		3.386	0.001	0.607	2.327					
Source	-0.539	0.200	-0.267	-2.692	0.008	-0.937	-0.142	-0.267	-0.268	-0.267	1.000	1.000
Treatment	0.112	0.200	0.056		0.561 0.576	-0.285	0.510	0.053	0.058	0.056	1.000	1.000
Dependent variable: I og Glul	Je I og Glul	P <sub>v/r</sub>										

<sup>a</sup> Dependent variable: Log\_GluPyr





Table S18. Linear regression of pyrene (Pyr) and pyrene-1-glucuronide (GluPyr) concentration for the polychaete species Alitta virens.

**(** 

				M	Model summary	ry <sup>b</sup>				
			A 4:::40 A	SE of		Che	Change statistics	s		.:
Model	×	R square	Aujusieu R square	the estimate	R square change	F change	IJP	df2	Sig. F change	Watson
_	0.187 <sup>a</sup>	0.035	0.000	0.21989	0.035	0.796	-	22	0.382	1.407
g 1 (7 + 5)										

<sup>a</sup> Predictors: (Constant), Log\_Pyr <sup>b</sup> Dependent variable: Log\_GluPyr

			$ANOVA^a$			
	Model	Sum of	J.	Mean	Ĺ	S.S.
	Model	squares	Ħ	square	1	51 <u>8</u> .
	Regression	0.038	1	0.038	962.0	$0.382^{b}$
_	Residual	1.064	22	0.048		
	Total	1.102	23			

<sup>a</sup> Dependent variable: Log\_GluPyr
<sup>b</sup> Predictors: (Constant), Log\_Pyr

					Coeffi	Coefficients						
Model	Unstandar coefficie	lardized cients	Standardized coefficients	t	Sig.	95% confidence interval for B	fidence I for B	J	Correlations		Collinearity statistics	statistics
	Q	25	Doto			Lower	Upper	Zero-	Dominal	Down	Tolomoro	1/115
	D	or.	Deta			punoq	punoq	order	raittai	rait	1 OICI AIICC	. T
(Constant)	0.736	0.054		13.602	0.000	0.624	0.848					
Log_Pyr	-0.128	0.144	-0.187	-0.187 -0.892	0.382	-0.427	0.170	-0.187	-0.187 -0.187 -0.187	-0.187	1.000 1.000	1.000

<sup>a</sup> Dependent variable: Log\_GluPyr

**(** 



Table S19. Linear regression of pyrene (Pyr) and pyrene-1-glucuronide (GluPyr) concentration for the polychaete species Nephtys ciliata.

				Mc	Model summary	ryb				
			A dissorted	SE of		Cha	Change statistics			D.1.4b.;
Model	×	R square	Adjusted R square	the estimate	R square change	F change	dfl	q£2	Sig. F change	Duroin- Watson
1	$0.502^{a}$	0.502 <sup>a</sup> 0.252	0.218	0.218 0.16988	0.252	7.400	-	22	0.012	2.215
a Drodiotor	a Dradiators: (Constant) I ag Dra	J I or Drm								

<sup>a</sup> Predictors: (Constant), Log\_Pyr
<sup>b</sup> Dependent variable: Log\_GluPyr

		$ANOVA^a$			
Model	Sum of	Jł.	Mean	Ĺ	S.S.
INIONCI	squares	<b>5</b>	square	1	31g.
Regression	0.214	1	0.214	7.400	$0.012^{b}$
Residual	0.635	22	0.029		
Total	0.848	23			

<sup>a</sup> Dependent variable: Log\_GluPyr

<sup>b</sup> Predictors: (Constant), Log\_Pyr

**(** 

	statistics	VIF
	Collinearity statistics	Tolerance
		Part
	Correlations	Partial
		Zero- order
	95% confidence interval for B	Upper Bound
Coefficientsa	95% con interval	Lower
Coefi	Sig.	
	t	
	Standardized coefficients	Beta
	<sup>J</sup> nstandardized coefficients	SE
	Unstand	В

1.000

1.000

0.502

0.502

0.502

0.945 0.455

0.575 0.061

0.000 0.012

8.506 2.720

0.089 0.095

0.760 0.258

1 (Constant) Log\_Pyr

Model

0.502

<sup>a</sup> Dependent variable: Log\_GluPyr





Table \$20. Linear regression of pyrene (Pyr) and pyrene-1-glucuronide (GluPyr) concentration for the bivalve species Limecola balthica.

**(** 

				Ň	Model summary	ryb				
			Lotoric A	SE of		Cha	Change statistics			- iq
Model	~	R square	Aujusieu R square	the estimate	R square change	F change	IJp	q£2	Sig. F change	Watson
	0.423a	0.179	0.162	0.162 0.62617	0.179	0.179 10.471	1	48	48 0.002	1.789
diotomo	a Dundintonia, (Comptent) I on Dru	I Lac Dem								

<sup>a</sup> Predictors: (Constant), Log\_Pyr

<sup>b</sup> Dependent variable: Log\_GluPyr

			$ANOVA^a$			
	Model	Sum of	J+C	Mean	Ĺī	S.S.
	Model	squares	<b>5</b>	square	4	J. G.
	Regression	4.106	-	4.106	10.471	$0.002^{b}$
_	Residual	18.820	48	0.392		
	Total	22.926	49			

<sup>a</sup> Dependent variable: Log\_GluPyr <sup>b</sup> Predictors: (Constant), Log\_Pyr

Coefficients<sup>a</sup>

statistics	VIE	A TIL		1.000
Collinearity statistics	Tolerance	1 Olei allee		1.000
	Dart	ו מונ		0.423
Correlations	Dortial	ı aıtıaı		0.423
	Zero-	order		0.423
fidence for B	Upper	ponnoq	0.904	0.981
95% confidence interval for B	Lower	punoq	0.065	0.229
Sig.	ļ		0.025	0.002
ţ			2.320	3.236
Standardized coefficients	Beta	Deta		0.423
rdized ents	SE	35	0.209	0.187
Unstandardized coefficients	В	a	0.484	0.605

<sup>a</sup> Dependent variable: Log\_GluPyr

1 (Constant) Log\_Pyr

Model

**(** 



Table \$21. Linear regression of pyrene (Pyr) and pyrene-1-glucuronide (GluPyr) concentration for the bivalve species Macoma calcarea.

				Me	Model summary	ry <sup>b</sup>				
			A 41:3240.4	SE of		Cha	Change statistics		8	- : <del> </del>
Model	М	R square	Aujusteu R square	the estimate	R square change	F change	dfl	q£2	Sig. F change	Watson
-	0.398ª	0.159	0.140	0.140 1.13464	0.159	8.482	-	45	0.006	1.444
1: 4	n - 1 (7 - 7 - 0)	- L								

<sup>a</sup> Predictors: (Constant), Log\_Pyr <sup>b</sup> Dependent variable: Log\_GluPyr

			$ANOVA^a$			
	Model	Sum of	Эf	Mean	П	Sig
	Model	squares	Ħ	square	-	51g.
	Regression	10.919	-	10.919	8.482	900°0
_	Residual	57.934	45	1.287		
	Total	68.853	46			

<sup>a</sup> Dependent variable: Log\_GluPyr <sup>b</sup> Predictors: (Constant), Log\_Pyr

$^{1}$ ts $^{a}$	ĺ
tricien	
Coe	

**(** 

statistics	VIE	ΛTL		1.000
Collinearity statistics	Tolomoron	1 Olei allee		1.000
	Dowt	ган		0.398
Correlations	Dostiol	r ai tiai		0.398
Ŭ	Zero-	order		0.398
fidence for B	Upper	punoq	0.357	2.346
95% confidence interval for B	Lower	punoq	-1.075	0.428
Sig.			0.318	0.006
t			-1.010	2.912
Standardized coefficients	Doto	Deta		0.398
ndardized fficients	15	31.	0.356	0.476
Unstandar	Q	р	-0.359	1.387
Model			(Constant)	Log_Pyr

a Dependent variable: Log\_GluPyr





# 5 Avoidance tests as a tool to detect sublethal effects of oilimpacted sediments

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### Based on:

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#### **Abstract**

Currently, risk assessment for oil contamination does not consider behavioral responses of benthos to oil toxicity. Avoidance of oil-contaminated sediment by benthic amphipods, however, may be a highly sensitive endpoint for sublethal effects of commonly used distillate fuels. In the present study, the avoidance behavior of temperate freshwater ( $Gammarus\ pulex$ ) and marine ( $Gammarus\ locusta$ ) amphipods is tested, allowing them to choose between a reference sediment and a Distillate Marine grade A (DMA) oil-spiked sediment. Avoidance of DMA-spiked sediment at 1000 mg/kg (dry weight) was significant within total exposure time (96 h) in G. pulex and within the first 72 h in G. locusta in one of two tests. Absence of DMA avoidance at lower concentrations ( $\leq$  250 mg/kg dry weight) indicates that test species can only detect DMA above these concentrations. However, sensitivity to oil may vary according to the phenology and physiological conditions of the populations involved, such as the species temperature tolerance and reproductive stage. The results suggest that avoidance tests may be used as an alternative to traditional chronic toxicity tests provided that a causal link between avoidance and long-term effects can be established.







#### 1. Introduction

Increasing global offshore oil and gas (O&G) production, oil transportation and shipping entail higher risks of accidental oil spills which are known to cause severe environmental impacts. The assessment of potential environmental consequences should be based on the relationship between the extent of oil exposure and the sensitivity of marine species and their habitats. Following a spill, the composition of oil and its toxicity will mostly vary according to the timing and magnitude of oil spreading and evaporation on sea surface [293]. In this process, the more water soluble and bioavailable fraction of oil volatilizes or degrades in the first several days, while the more hydrophobic oil components will be deposited to the seafloor where degradation is much slower [294]. Exceptions may apply to polar regions where the presence of ice can greatly reduce oil spreading and low temperatures may lower oil hydrocarbons partitioning and depletion rates [51]. Thus, the identification of biological features affecting oil exposure in local benthic organisms may help to better characterize the sensitivity of environmental receptors such as shoreline habitats [295,296].

Monitoring of environmental stress resulting from O&G activities can be based on assays that use behavioral endpoints, like movement ability [105,297]. Animal behavior integrates biochemical and physiological processes in response to chemical stressors. As such it is expected to be more sensitive than a lethal response [298], provided that an organism is able to sense a toxicant via olfaction or taste [299], and it may potentially reflect changes at a population level [97]. In a natural environment, mobile species may be able to move away from oil-contaminated sites before the oil can exert its toxic effects. In some cases, such responses may be absent. For example, in case of chronic toxicity [300] or when the individuals' energy allocation to locomotion, and thus avoidance, is attenuated [301-303].

Previous research addressed avoidance behavior in laboratory assays, studying spatial avoidance of either water or sediment containing different chemicals (summarised by Araújo et al. [96]). Amphipods have been used in avoidance assays since they are equipped with a sensory system [98-101]. Studies on the avoidance behavior of oil-contaminated sediment by benthic organisms are limited and restricted to the marine environment [99,304-306]. In these studies, generally a bi-compartmented system container was used to test the ability of amphipods to discern between a clean and a contaminated sediment. However, a preference for the clean sediment or avoidance of the contaminated sediment is not always evident, because for instance exposure to sublethal concentrations of oil may lead to contrasting







sensitivities among different groups of amphipods or 'attractive' odours may be masked by oil [108,109]. The effect of time is generally not considered or limited to a single time point per exposure day, which eliminates the possibility of identifying likely experimental biases such as a (pre-)adaptation to the chemical cue [100].

Spatial avoidance can be divided into active and passive avoidance [96], each occurring under the changing physical state of chemicals over exposure time. Active avoidance or 'escape' response relates to the ability of an organism to detect a chemical and to move to a lesser contaminated area, whereas passive avoidance or 'drift' response will occur when water flow is responsible for the organism displacement. In a bi-compartmented static system, amphipods will display active spatial avoidance provided that no other physical disturbances are present (e.g., sand, seaweed). The effect of time on the amphipods distribution may help to predict possible interactions between avoidance and their physiological susceptibility to oil [307] as well as their long-term implications [308].

The objective of the present study is to evaluate the potential of sediment avoidance tests using benthic amphipods as a monitoring tool for oil spill impact assessment. Prior to testing, oil was mixed with the sediment. We tested the avoidance behavior of a freshwater amphipod *Gammarus pulex* and a marine benthic amphipod *Gammarus locusta* as model invertebrate species, to sediment spiked with a commonly used distillate fuel oil (i.e., Distillate Marine grade A [DMA]) during a 96-h test period. DMA oil is widely used as fuel in midsized to larger ships such as large inland vessels, large cutters and coasters [309], and is often used in oil toxicity studies [43,46]. These species were deployed as a first step to develop ecologically relevant tests for oil impact assessment.

#### 2. Materials and Methods

### 2.1. Test organisms

Freshwater experiment (FW). Following De Lange et al. [100], G. pulex (Linnaeus, 1758) adult individuals (9.7±1.8 mm) were collected from a non-polluted freshwater stream near Heelsum (The Netherlands, 51°58'40.3"N, 5°45'27.8"E) on 13 March 2014. Animals were transported to the laboratory in plastic buckets with water from the collection stream and acclimatized in a water bath during 10 d (15±1 °C). Since animals were collected in winter and seasonal effects may influence tolerance to toxicants exposure [310], the temperature and







duration of acclimatization were chosen according to the temperature range used by similar avoidance studies [99,100] and studies on effects of temperature on *G. pulex* physiology [311,312]. During acclimatization, oxygenated copper-free tap water was used and animals were fed unconditioned poplar leaves ad libitum. Prior to their use, poplar leaves were leached in tap water to remove active leachable substances, and dried.

Water quality variables such as temperature, dissolved oxygen, electric conductivity and pH were measured regularly during acclimatization (Table S1).

Marine water experiments (MW1 and MW2). G. locusta (Linnaeus, 1758) individuals were collected over 4 d in the Oosterschelde estuary (The Netherlands, 51°32'41.3"N, 3°55'30.0"E) on 24 to 25 June 2016 and 25 to 26 July 2016, hereafter MW1 and MW2 experiments. Since G. locusta life-span and maturity age are known to be greatly reduced by water temperature [313], individuals of the species were collected at two different times during summer in order to assess the consistency of experimental results. Animals were transported to the laboratory in 15-L jerry cans with water from the collection site and acclimatized during 5 d (15±1 °C). Sea water temperature from the collection site was 17±1 °C in June and 21±1 °C in July. During acclimatization, animals were kept in 18-L glass aquaria each of them holding approximately 200 to 500 individuals in 15 L of oxygenated artificial sea water (ASW) (40 g/L, Pro Reef Salt, Colombo B.V.). A few stones from the collection site were also placed inside each aquarium for comfort of the test animals and animals were fed fish food tabs (Tetra Wafermix) ad libitum.

#### 2.2. Sediment spiking

For the freshwater experiment, sediment was collected from a pond at Wageningen University campus. Sediment was sieved with a 2-mm sieve (83% dry wt., 0.88% organic matter) and spiked with DMA to obtain four different nominal concentrations: 0 (Control), 50 (Low), 250 (Medium) and 1000 (High) mg DMA/kg sediment (dry wt.). DMA oil was supplied by Gulf Oil (Nigtevecht, the Netherlands) and was the same as used in our earlier studies, where also chemical characteristics are provided [43,46]. Nominal DMA concentrations were chosen to match field-relevant concentrations [314,315] and to match the range of oil concentrations for which PAH sediment-water partition coefficients (K<sub>d</sub>) are generally independent of oil concentration (i.e., 100 – 1000 mg/kg dry wt. [46]). This was observed after a 2 year weathering process of oil-spiked sediment, although in some cases (i.e., 3-ring PAHs) K<sub>d</sub>







values were increased after weathering [46]. A relative stability of 2- to 3-ring PAHs in weathered surface oil [316] and in slick oil spiked to sediment [317], as well as their sublethal effects on benthic biota exposed at 1.7 mg/kg (dry wt.) ( $\sum_{50}$ PAH = 2.6 mg/kg dry wt.) or more, have been reported in literature [317,318]. Therefore, concentrations of 3-ring PAHs (e.g., phenanthrene, anthracene) in the high DMA treatment were considered sublethal (~1 mg/kg dry wt.) assuming a fair similarity in weathering conditions between Jonker et al. [43] and the present study.

For each sediment treatment containing DMA-spiked sediment (50-1000 mg/kg dry wt.), the required volume of DMA was divided into three batches to minimize the adsorption of DMA onto the barrel walls and its evaporation during stirring. Sediment was continuously stirred while each of the DMA batches was added drop by drop from a glass pipette. Once each oil batch was added, sediment stirring continued for about 5 min and allowed to rest for another 5 min before addition of following batch under the same procedure. After spiking, each sediment treatment was mixed on a roller bank during approximately 22 h and stored ( $7\pm1$  °C) for 4 weeks before use. For the control treatment, sediment was stirred and mixed in the same way without the addition of oil, and stored.

For the marine *G. locusta* experiments, sediment was collected at the animals collection site during low tide. Sediment was sieved (86% dry wt., 0.82% organic carbon) and spiked with DMA following the same procedure as in the freshwater experiment. After spiking, each sediment treatment was mixed on a roller bank during approximately 24 h and stored (4±1 °C) for 2 to 6 weeks before use. The control sediment treatment was prepared as mentioned above.

#### 2.3. Experimental design

The avoidance experiments were performed in glass aquaria ( $20 \times 20$  cm). A glass barrier was placed in the middle of each aquarium dividing the bottom into two equal parts. One side was filled with a 2-cm layer of control sediment and the other side with a 2-cm layer of control or spiked sediment. In each experiment, 800 mL of either copper-free tap water or ASW was gently added to each aquarium in order to avoid sediment disturbance, after which the systems were allowed to stabilise for 24 h. The freshwater *G. pulex* experiment was performed at Wageningen University laboratories, whereas the marine experiments with *G. locusta* were performed at the marine research laboratories of Wageningen Marine Research (Yerseke, the Netherlands). In the freshwater experiment, the aquaria were randomly distributed within a





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water bath (15±1 °C), whereas for marine water experiments the aquaria were randomly placed on a table located inside a climate room (15±1 °C). Once the systems were stabilised, another 800 mL of either copper-free tap water or ASW was carefully added, after which the systems were allowed to stabilise for 3 d. Five replicates (aquaria; n = 5) were used for each of the four sediment treatments (control, low, medium and high DMA concentration) during each experiment.

Prior to the addition of the animals, the water layer was partially renewed. A volume of 800 mL of the system water was pumped out of each aquarium and renewed with 800 mL of clean copper-free tap water or ASW, respectively. This was done to minimize the concentration of water soluble DMA toxicants in the water column with respect to the concentration in the spiked sediment layer at start. The glass barriers were carefully removed and 20 individuals were added in the middle of each aquarium to ensure that the starting position did not influence the experiment outcome [100]. Animals were kept under a 16 h light: 8 h dark red light cycle since *Gammarus* spp. usually become photonegative when they reach an adult size [319]. However, dim light was needed for maintenance and to facilitate the visual recording of the animal distribution.

## 2.4. Endpoints

During the removal of barriers, some mixing of control and spiked sediment could not be prevented at surface level in the middle area. Therefore, the middle area was excluded from the statistical analyses. Each system sediment surface was divided into three areas: control (37.5% of the surface), middle (25%) and spiked (37.5%). The number of individuals staying for at least 5 to 6 consecutive seconds within the control or spiked area during an observation bout was recorded as 'control' or 'spiked', respectively. Recordings took place during a 96-h period allowing an estimation of the effect of DMA concentration (treatment) on Gammarus spp. avoidance of the DMA-spiked area over time (day). Only the results of the 'control' scores of each replicate were used to test whether their relative distribution at the control section was higher than 37.5%, in which case the null hypothesis was rejected and avoidance of the spiked area was considered occurred. Water quality variables were measured at the beginning of each recording day in three experimental control replicate aquaria without animals during the first marine experiment (Table S2) and in each experimental replicate at the end (Table S3) of both marine experiments.







#### 2.5. Data analyses

Logistic regression within the framework of Generalized Linear Mixed Models (GLMM) was used to analyse the relationship between the number of individuals in the control section treated as repeated measures for each aquarium (experimental unit) and the effect of treatment, day and their respective interaction as effect builders. GLMM statistical tests provide a more flexible approach than other non-parametric tests for analysing non-normal data when random effects are present. Only observations obtained from the second to the fourth experimental day were used since they include an equal number of recordings per day and handling of the animals could also have induced stress on the first day. Observations were collected 3 times per day for each aquarium and treatment in the freshwater experiment, that is 180 observations in total; and 7 times per day for each aquarium and treatment in the marine experiments, making a total of 420 observations for each experiment. Observations were assumed to be Poisson distributed following the repeated recording of a single event (i.e., 'control' score takes integer values between 0 and 20) and a pairwise contrast among treatments and recording days was chosen in order to calculate the estimated means. The significance level of an overall statistical comparison was set at  $p \le 0.05$ , while for pairwise comparisons the least significant difference correction was applied. All statistical calculations were performed using SPSS version 22.

### 3. Results and Discussion

## 3.1. Freshwater Gammarus pulex avoidance

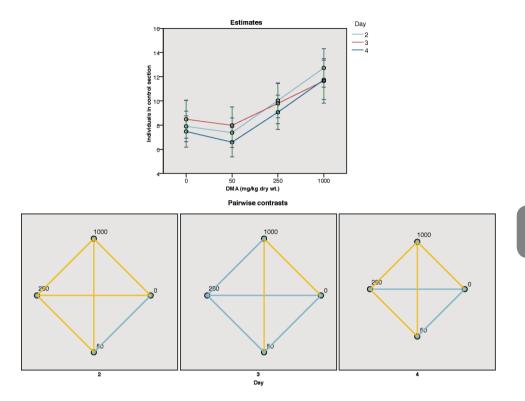
 $G. \ pulex$  avoidance was significantly and consistently higher in the high DMA treatment than in the control and low treatment (day 2 to 4) (Figure 1; Tables S4-S6).  $G. \ pulex$  avoidance behavior in the medium DMA treatment was significantly different from that in the control, low and high treatment on day 2, after which an apparent attenuation of avoidance was observed (Figure 1). No mortality was observed within the total test duration (4 d) throughout the range of nominal DMA concentrations used  $(0-1000 \ mg/kg \ dry \ wt.)$ .





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**Figure 1.** Generalized linear mixed models avoidance estimated means with confidence intervals and 'treatment  $\times$  day' pairwise contrasts for *Gammarus pulex* exposed to Distillate Marine grade A (DMA) oil (FW experiment). Avoidance is expressed as number of individuals staying for at least 5 to 6 consecutive seconds within the control section per DMA treatment (0-1000 mg/kg dry wt.) and day (2-4). Significant contrasts  $(p \le 0.05)$  are shaded gold.

A consistent avoidance response of *G. pulex* exposed to the high DMA treatment from day 2 to 4 could be explained by a combined result of exposure to a high DMA concentration and to the species upper thermal tolerance limit (i.e., 15 °C) in northern populations of *G. pulex* [311]. According to the 'oxygen- and capacity-limited thermal tolerance' model, failure of oxygen transport systems to match body oxygen demand dictates thermal tolerance [320]. An increase in the oxygen (energy) demand of *G. pulex* winter populations (~5 °C at which the organisms were sampled) at the exposure temperature (15 °C) is expected to be met by an increase in their ventilation rate [321,322]. An increase in ventilation may facilitate the detection and transmission of chemical cues [323] and possibly triggered avoidance in *G. pulex* exposed to the high DMA treatment from the first day of exposure. Another





consequence of the maintenance of *G. pulex* within its optimal thermal window can be that glycogen reserves, which is the main energy fuel for oxygen deficiency and sustained activity, are relatively stable provided that food is not limiting [324]. The mobilization of these energy stores would allow *G. pulex* to actively avoid DMA-spiked sediment and to counterbalance the stress produced by DMA exposure at the experiment start.

DMA exposure in the low and medium DMA treatments was most probably too low to elicit a continuous avoidance by *G. pulex*. The observed differences in DMA avoidance response between the low and medium treatment exposed *G. pulex* on day 2 and 4 of exposure were most likely a reflection of within treatment variability (Figure 1; Table S4). Variability in the actual exposure to DMA and avoidance response of *G. pulex* exposed to the medium DMA treatment may be underlain by intraspecific differences in *G. pulex* molecular response to starvation or reproduction [311,325] among other factors.

In summary, *G. pulex* active avoidance of DMA-spiked sediment was significant and continuous over exposure time (day 2 to 4) at a DMA concentration of 1000 mg/kg (dry wt.), and sometimes at 250 mg/kg (dry wt.). Since no mortality was observed within the total test duration (4 d) in none of the treatments, it was assumed that no deleterious effects were caused by DMA on the sensory system of individuals.

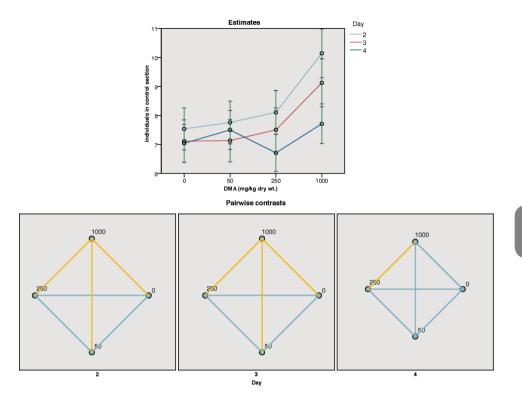
#### 3.2. Marine Gammarus locusta avoidance

 $G.\ locusta$  avoidance during the first marine experiment was significantly higher in the high DMA treatment compared to the control, the low and medium DMA treatments from day 2 to 3 (Figure 2), whereas in the second marine experiment no effect of treatment or day on  $G.\ locusta$  avoidance was found (Figure 3).  $G.\ locusta$  avoidance was also significantly higher in individuals exposed to the high DMA treatment in the first compared to the second marine experiment from day 2 to 3 (Tables S13 and S14). No mortality was observed within the total test duration (4 d) throughout the range of nominal DMA concentrations used  $(0-1000\ mg/kg\ dry\ wt.)$ .



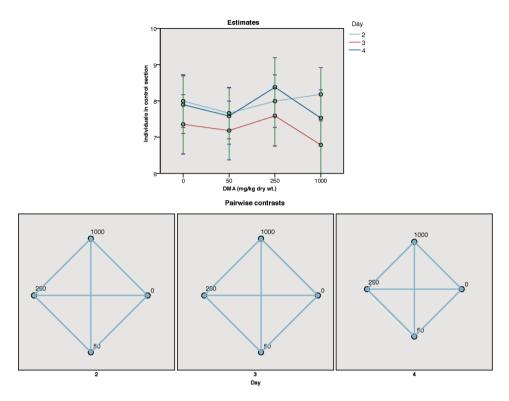






**Figure 2.** Generalized linear mixed models avoidance estimated means with confidence intervals and 'treatment  $\times$  day' pairwise contrasts for *Gammarus locusta* exposed to Distillate Marine grade A (DMA) oil (MW1 experiment). Avoidance is expressed as number of individuals staying for at least 5 to 6 consecutive seconds within the control section per DMA treatment (0-1000 mg/kg dry wt.) and day (2-4). Significant contrasts  $(p \le 0.05)$  are shaded gold.





**Figure 3.** Generalized linear mixed models avoidance estimated means with confidence intervals and 'treatment  $\times$  day' pairwise contrasts for *Gammarus locusta* exposed to Distillate Marine grade A (DMA) oil (MW2 experiment). Avoidance is expressed as number of individuals staying for at least 5 to 6 consecutive seconds within the control section per DMA treatment (0-1000 mg/kg dry wt.) and day (2-4). Significant contrasts  $(p \le 0.05)$  are shaded gold.

In an experiment performed by Hellou et al. [99], the preference for reference sediment could also not be clearly elucidated in the marine amphipod *Corophium volutator* exposed to freshly diesel oil-spiked sediment (> 0.01%). As explained by the authors, a high variation in the response between exposure replicates and high mortality of *C. volutator* were caused by a narcotic effect of diesel oil. Besides avoidance behavior, re-emergence and failure to burrow did not follow a concentration-related response in *C. volutator* exposed to sediment spiked with artificially weathered crude oil at 220 to 880 mg/kg (dry wt.) [326]. In our experiments, the highest two DMA concentrations were 0.02% and 0.1%, respectively, but neither mortality nor narcotization were observed in *G. locusta*. As in *C. volutator* example [326], *G.* 







*locusta* avoidance behavior did not correlate with DMA concentration (50 – 1000 mg/kg dry wt.) over exposure time (i.e., day 2 to 4).

Although the cause of the difference in *G. locusta* avoidance behavior between the first and second experiments is not known, this may be linked to differences in life stage-related factors such as body size [327], lipid content or maturity stage [99,313] between the two experiments. Since body size and structural (membrane) lipid content may influence the chemical exchange at body surface, changes in these variables may be reflected in differences in the apparent sensitivity to DMA and avoidance between summer cohorts. Variability in body size and structural lipid content [328] may be further increased by a reduction in the life-span of *G. locusta* above 15 °C [313], the maximum expression of which can be found among summer cohorts. A *G. locusta* generational succession was observed between June and July by Kolding and Fenchel [329] in the Limfjorden strait (Denmark) and by Costa and Costa [327] in the Sado estuary (Portugal), according to which a higher proportion of larger-sized gravid females would be expected in *G. locusta* collected in June compared to July.

Lack of avoidance by *G. locusta* exposed at 50 and 250 mg DMA/kg (dry wt.) (respectively low and medium concentration) in both marine experiments could be related to a low exposure of sediment treatments as also observed in the freshwater experiment. Both age- and sex-related differences in their oxidative stress tolerance [327,330] might account for the lack of a DMA concentration- nor time-dependent avoidance response by *G. locusta* [331].

## 3.3. Comparison of avoidance between Gammarus pulex and Gammarus locusta

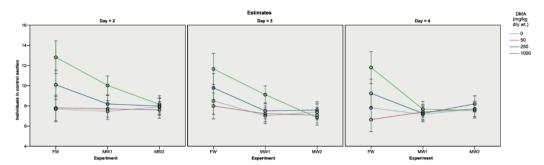
Avoidance behavior of DMA-spiked sediment (50 – 1000 mg/kg dry wt.) could not be clearly observed in *G. locusta*, whereas in *G. pulex* avoidance was significant and continuous over exposure time (i.e., day 2 to 4) at the highest DMA concentration (Figures 4 and 5; Tables S16 and S17). Interpretation of avoidance results may be complicated by differences in behavior of DMA in freshwater and marine systems as the aqueous solubility of PAHs tends to decrease with increasing salinity [24]. Therefore, oil may have been easier detectable by *Gammarus* sp. in the freshwater experiment. PAH degradation will be generally higher in freshwater than in marine systems, although degradation will be strongly influenced by oxygen availability in the sediment and the molecular size of PAHs [43]. In the present study, oxygen concentration was not measured in the freshwater experiment, but lower oxygen levels were detected at the end of exposure in the second compared to the first marine experiment







(Table S3). This might point to a higher microbial degradation of PAHs [43] and therefore, lower DMA toxicity, in the second marine experiment than in the first marine experiment.



**Figure 4.** Generalized linear mixed models avoidance estimated means with confidence intervals for *Gammarus pulex* (FW experiment) and *Gammarus locusta* (MW1 and MW2 experiments) exposed to Distillate Marine grade (DMA) oil. Avoidance is expressed as number of individuals staying for at least 5 to 6 consecutive seconds within the control section per DMA treatment (0 - 1000 mg/kg dry wt.) and day (2 - 4).

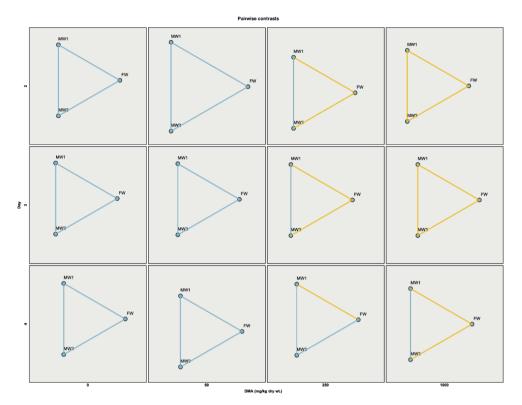
As indicated by previous studies [99,306], lack of reproducibility of avoidance results among test organisms may be caused by differences in their growth rate and reproductive stage. Factors such as the individual size of *G. pulex* (~10 mm) and temperature at collection time (~5 °C) may have affected the experimental results. *G. pulex* has a natural life-span of 1 to 2 years and possibly experienced a low reproductive effort during the freshwater experiment [332]. A longer life-span of *G. pulex* (11 – 15 months at 15 °C [333]) compared to *G. locusta* (5 – 6 months at 15 °C [313]) and/or lower reproductive effort of female *G. pulex* during the freshwater experiment may have caused an increase in the energy budget in *G. pulex* compared to that of *G. locusta*, and thus their capacity of avoidance during DMA exposure.

In addition, it is not known whether slightly hypersaline conditions (40 – 42 mS/cm, Table S3) could have resulted in intraspecific differences in *G. locusta* growth [334] or if a likely osmotic imbalance in *G. locusta* [335] may have been enhanced by oil exposure. Osmoregulation is an energy-consuming regulatory function in *Gammarus* spp. in a way that its alteration may reduce the energy allocated to locomotion and ventilation. A combined effect of osmoregulation- and reproduction-related energy consumption may have reduced the ability of *G. locusta* to 'escape' from DMA, whereas *G. pulex* energy allocation to hyperventilation during acclimatization, may be compensated by an early DMA detection, via ventilation, and long-term stability of glycogen reserves.









**Figure 5.** Generalized linear mixed models 'experiment × treatment × day' pairwise contrasts for *Gammarus pulex* (FW experiment) and *Gammarus locusta* (MW1 and MW2 experiments) exposed to Distillate Marine grade A (DMA) oil. Avoidance is expressed as number of individuals staying for at least 5 to 6 consecutive seconds within the control section per DMA treatment (0-1000 mg/kg dry wt.) and day (2-4). Significant contrasts  $(p \le 0.05)$  are shaded gold.

### 3.4. Implications for the use of avoidance tests with Gammarus spp.

The present study showed that spatial avoidance may be integrated in the current risk assessment framework due to its higher sensitivity than mortality [326]. However, the interpretation of spatial avoidance results required some assumptions to be made related to inter-individual variability. Such variability may be greater in summer generations because their reproductive effort should be maximized when their life-span is shortened by high temperatures and this would restrict their capacity to cope with metabolic stress. Therefore, a supporting analysis of energy consumption in amphipods may help to establish if their energy budget is affected by the energy allocation to avoidance of oil and if such trade-off follows a seasonal pattern according to the species reproductive strategy. Avoidance results obtained





under such combined analysis could also be translated into a recognizable pattern of effects on growth and reproduction in short life-span species.

In the present study, nominal DMA concentrations were chosen according to the range of DMA oil concentrations for which a K<sub>d</sub> decrease had been observed for most PAHs by Jonker et al. [46]. However, variability in PAH partitioning due to differences in the storage time of sediment treatments might have affected the comparability between the present avoidance tests. This stresses the importance of including a chemical characterization of sediment treatments prior to and after avoidance tests, as well as of the overlying water during exposure for each replicate [83]. The interpretation of avoidance results may also be improved after the inclusion of a statistically sufficient number of recordings between the time to achieve a uniform distribution in control treatment replicates [101] and the end of recording on the first day of exposure. Such additional recordings would serve to identify the possibility of a reversible narcotic effect, reducing the risk of bias.

The interpretation of results may be further improved using a more gradual acclimation to standard temperature and salinity conditions in field-collected organisms. Acclimation can be energy demanding in amphipods and may influence their energy allocation to avoidance. This was hypothesized for *G. pulex* as the acclimation to standard temperature regimes could have affected its energy metabolism and thus, sensitivity to temperature and other stressors [336]. Similarly, it is not known if the variation in salinity could have contributed to the observed variability in *G. locusta* experiments. In both cases, the observed variability could have been associated to different rates of energy consumption among life-span stages during acclimatization. In addition, factors such as the reproductive or moulting stage could influence the chemical exchange at body surface and sensitivity to oil of amphipods.

Consequently, sampling efforts should be addressed to specific periods at which contributing factors to the sensitivity to oil in amphipods are more uniform among gammarid populations. One needs to consider as well how seasonality may influence the energy reserves and activity of test organisms as this would facilitate the extrapolation of avoidance results between studies from different climate regions. Avoidance of oil can be a useful tool to relate energy consumption to growth reduction and delated maturation in adult amphipods, helping to predict the environmental risks derived from their behavioral reaction. A combined monitoring of spatial avoidance and metabolic responses and its integration in risk analysis







may therefore help to anticipate in which exposure scenarios gammarid populations can recover from an oil spill following habitat fragmentation [16,295].

### 4. Conclusions

In our experiments, it was observed that the freshwater amphipod *G. pulex* avoided DMA-spiked sediment at 1000 mg/kg (dry wt.). For the marine amphipod *G. locusta*, avoidance was either not consistent with time or below effect threshold. Oil behavior may have differed in the freshwater experiment compared to the marine experiments, despite a similar spiking protocol. Season-related differences in the physiology of amphipod species may restrict the applicability of avoidance tests to non-reproductive and/or winter individuals. Besides the inclusion of scoring on the first day of exposure, we recommend the analysis of energy reserves dynamics in amphipods in order to ascertain whether any experimental factors other than the exposure duration can significantly alter the energy allocation to locomotion. The predictive power of avoidance tests using amphipod species could be increased after the identification of biological factors that may influence sensitivity to oil exposure in sediment.

### Acknowledgements

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## **Supporting Information**

Table S1. Gammarus pulex average water quality variables during acclimatization.

Day	Temperature (°C)	Oxygen (mg/L)	Electric conductivity (µS/cm)	pН
1	$15.6 \pm 0.1$	$9.3 \pm 0.1$	$401.8 \pm 0.5$	$7.3 \pm 0.0$
2	$15.6 \pm 0.0$	$9.2 \pm 0.2$	$404.5 \pm 1.7$	$7.2 \pm 0.0$
3	$15.6 \pm 0.0$	$9.9 \pm 0.1$	$378.8 \pm 4.3$	$7.5 \pm 0.0$
4	$15.7 \pm 0.1$	$9.5 \pm 0.2$	$357.5 \pm 5.1$	$7.5 \pm 0.0$
5	$15.8 \pm 0.0$	$9.5 \pm 0.2$	$342.3 \pm 7.3$	$7.6 \pm 0.0$
6	$15.7 \pm 0.1$	$9.4 \pm 0.2$	$327.3 \pm 9.0$	$7.7 \pm 0.0$
7	$16.4 \pm 0.0$	$10.1 \pm 0.1$	$198.6 \pm 4.2$	$7.9 \pm 0.0$
8	$16.4 \pm 0.0$	$9.8 \pm 0.2$	$204.6 \pm 7.1$	$7.8 \pm 0.0$
9	$16.4\pm0.0$	$9.4 \pm 0.1$	$208.8 \pm 8.2$	$7.8 \pm 0.0$

**Table S2.** First marine experiment (MW1) average water quality variables based on three experimental replicates without *Gammarus locusta* individuals.

Exposure day	Salinity (mS/cm)	Temperature (°C)	Oxygen (mg/L)	рН
1	40.1	16.1	7.1	7.8
2	40.7	16.8	6.5	8.0
3	41.3	16.6	7.0	8.2
4	41.5	16.3	7.4	8.2

**Table S3.** End of exposure (day 4) water quality variables per Distillate Marine grade A (DMA) oil sediment treatment (average  $\pm$  SD) in *Gammarus pulex* (FW) and *Gammarus locusta* (MW1 and MW2) experiments.

Experiment	DMA (mg/kg dry wt.)	Salinity (mS/cm)	Temperature (°C)	Oxygen (mg/L)	рН
	0	-	n.a.	n.a.	n.a.
FW	50	-	n.a.	n.a.	n.a.
I' W	250	-	n.a.	n.a.	n.a.
	1000	-	n.a.	n.a.	n.a.
	0	$42.2 \pm 0.2$	$16.3 \pm 0.1$	$6.3 \pm 0.3$	$8.2 \pm 0.0$
MW1	50	$42.1 \pm 0.1$	$16.4 \pm 0.2$	$6.0 \pm 0.3$	$8.2 \pm 0.0$
IVI VV I	250	$42.2 \pm 0.3$	$16.3 \pm 0.1$	$6.3 \pm 0.4$	$8.2 \pm 0.0$
	1000	$42.2 \pm 0.1$	$16.4 \pm 0.2$	$6.1 \pm 0.2$	$8.2 \pm 0.0$
	0	$41.4 \pm 0.9$	$15.4 \pm 0.1$	$5.2 \pm 0.3$	$8.1 \pm 0.1$
MW2	50	$40.5 \pm 0.9$	$15.4 \pm 0.1$	$5.4 \pm 0.8$	$8.0 \pm 0.2$
IVI VV Z	250	$40.7 \pm 1.9$	$15.4 \pm 0.1$	$5.2 \pm 0.3$	$8.0 \pm 0.0$
	1000	$40.2 \pm 0.6$	$15.4 \pm 0.0$	$5.3 \pm 0.4$	$8.0\pm0.0$

n.a.: not available.





**Table S4.** Estimated means of target variable (Individuals in control section) for "Treatment  $\times$  Day" effect on *Gammarus pulex* avoidance.

DMA treatment	Day	Mean	SE	95% confider	nce interval
(mg/kg dry wt.)	Day	Mean	SE	Lower	Upper
	2	7.896	0.636	6.639	9.152
0	3	8.485	0.794	6.919	10.052
	4	7.474	0.652	6.186	8.762
	2	7.378	0.615	6.164	8.593
50	3	7.982	0.770	6.462	9.501
	4	6.588	0.613	5.379	7.797
	2	10.034	0.717	8.618	11.450
250	3	9.798	0.853	8.114	11.482
	4	9.066	0.719	7.647	10.485
	2	12.727	0.808	11.131	14.322
1000	3	11.648	0.930	9.812	13.483
	4	11.746	0.818	10.131	13.360





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**Table S5.** Pairwise contrasts for "Treatment × Day" effect on *Gammarus pulex* avoidance.

D	DMA treatment	Contrast	CE	_	16	A 4: _:_	95% confide	nce interval
Day	pairwise contrast (mg/kg dry wt.)	estimate	SE	t	df	Adj. sig.	Lower	Upper
	0 - 50	0.517	0.885	0.584	168	0.560	-1.230	2.265
	0 - 250	-2.138	0.959	-2.230	168	0.027	-4.032	-0.245
	0 - 1000	-4.831	1.029	-4.697	168	5.461E-06	-6.862	-2.801
	50 - 0	-0.517	0.885	-0.584	168	0.560	-2.265	1.230
	50 - 250	-2.656	0.945	-2.810	168	0.006	-4.521	-0.790
	50 - 1000	-5.348	1.016	-5.266	168	4.212E-07	-7.353	-3.343
2	250 - 0	2.138	0.959	2.230	168	0.027	0.245	4.032
	250 - 50	2.656	0.945	2.810	168	0.006	0.790	4.521
	250 - 1000	-2.693	1.081	-2.492	168	0.014	-4.826	-0.560
	1000 - 0	4.831	1.029	4.697	168	5.461E-06	2.801	6.862
	1000 - 50	5.348	1.016	5.266	168	4.212E-07	3.343	7.353
	1000 - 250	2.693	1.081	2.492	168	0.014	0.560	4.826
	0 - 50	0.504	1.106	0.456	168	0.649	-1.679	2.686
	0 - 250	-1.313	1.165	-1.127	168	0.262	-3.612	0.987
	0 - 1000	-3.162	1.223	-2.587	168	0.011	-5.576	-0.749
	50 - 0	-0.504	1.106	-0.456	168	0.649	-2.686	1.679
	50 - 250	-1.816	1.149	-1.581	168	0.116	-4.084	0.452
	50 - 1000	-3.666	1.207	-3.037	168	0.003	-6.049	-1.283
3	250 - 0	1.313	1.165	1.127	168	0.262	-0.987	3.612
	250 - 50	1.816	1.149	1.581	168	0.116	-0.452	4.084
	250 - 1000	-1.850	1.262	-1.466	168	0.145	-4.341	0.641
	1000 - 0	3.162	1.223	2.587	168	0.011	0.749	5.576
	1000 - 50	3.666	1.207	3.037	168	0.003	1.283	6.049
	1000 - 250	1.850	1.262	1.466	168	0.145	-0.641	4.341







Table S5. Continued.

	DMA treatment	Contrast	ar.		10		95% confider	nce interval
Day	pairwise contrast (mg/kg dry wt.)	estimate	SE	t	df	Adj. sig.	Lower	Upper
	0 - 50	0.886	0.895	0.991	168	0.323	-0.880	2.653
	0 - 250	-1.592	0.971	-1.640	168	0.103	-3.508	0.324
	0 - 1000	-4.271	1.046	-4.083	168	6.875E-05	-6.337	-2.206
	50 - 0	-0.886	0.895	-0.991	168	0.323	-2.653	0.880
	50 - 250	-2.478	0.944	-2.625	168	0.009	-4.342	-0.614
	50 - 1000	-5.158	1.022	-5.048	168	1.154E-06	-7.175	-3.141
4	250 - 0	1.592	0.971	1.640	168	0.103	-0.324	3.508
	250 - 50	2.478	0.944	2.625	168	0.009	0.614	4.342
	250 - 1000	-2.680	1.089	-2.461	168	0.015	-4.829	-0.530
	1000 - 0	4.271	1.046	4.083	168	6.875E-05	2.206	6.337
	1000 - 50	5.158	1.022	5.048	168	1.154E-06	3.141	7.175
	1000 - 250	2.680	1.089	2.461	168	0.015	0.530	4.829

**Table S6.** Overall generalized linear mixed models test results for "Treatment  $\times$  Day" effect on *Gammarus pulex* avoidance.

Day	F	dfl	df2	Sig.
2	11.084	3	168	1.122E-06
3	3.567	3	168	0.015
4	9.433	3	168	8.541E-06







**Table S7.** Estimated means of target variable (Individuals in control section) for "Treatment  $\times$  Day" effect on *Gammarus locusta* avoidance in the first marine experiment (MW1).

DMA treatment	D	Mean	CE	95% confider	nce interval
(mg/kg dry wt.)	Day	Mean	SE	Lower	Upper
	2	7.536	0.369	6.812	8.261
0	3	7.113	0.372	6.382	7.844
	4	7.041	0.332	6.389	7.692
	2	7.759	0.374	7.024	8.495
50	3	7.133	0.372	6.401	7.865
	4	7.502	0.342	6.829	8.174
	2	8.106	0.382	7.354	8.857
250	3	7.505	0.382	6.754	8.256
	4	6.711	0.324	6.075	7.347
	2	10.140	0.428	9.300	10.981
1000	3	9.123	0.421	8.295	9.951
	4	7.712	0.347	7.030	8.394







Table S8. Pairwise contrasts for "Treatment × Day" effect on Gammarus locusta avoidance (MW1 experiment).

į	DMA treatment	Contrast	Ç	•	JF		95% confidence interval	ice interval
Day	(mg/kg dry wt.)	estimate	SE.	1	αI	Adj. sig.	Lower	Upper
	0 - 50	-0.223	0.525	-0.424	408	0.672	-1.256	0.810
	0 - 250	-0.569	0.531	-1.072	408	0.284	-1.614	0.475
	0 - 1000	-2.604	0.565	-4.612	408	5.357E-06	-3.714	-1.494
	50 - 0	0.223	0.525	0.424	408	0.672	-0.810	1.256
	50 - 250	-0.346	0.535	-0.648	408	0.518	-1.398	0.705
c	50 - 1000	-2.381	0.568	-4.190	408	3.415E-05	-3.498	-1.264
1	250 - 0	0.569	0.531	1.072	408	0.284	-0.475	1.614
	250 - 50	0.346	0.535	0.648	408	0.518	-0.705	1.398
	250 - 1000	-2.035	0.574	-3.547	408	0.000	-3.163	-0.907
	1000 - 0	2.604	0.565	4.612	408	5.357E-06	1.494	3.714
	1000 - 50	2.381	0.568	4.190	408	3.415E-05	1.264	3.498
	1000 - 250	2.035	0.574	3.547	408	0.000	0.907	3.163
	0 - 50	-0.021	0.526	-0.039	408	696.0	-1.055	1.014
	0 - 250	-0.392	0.533	-0.736	408	0.462	-1.440	0.656
	0 - 1000	-2.010	0.562	-3.578	408	0.000	-3.115	-0.906
	50 - 0	0.021	0.526	0.039	408	0.969	-1.014	1.055
	50 - 250	-0.372	0.533	-0.697	408	0.486	-1.420	0.677
,	50 - 1000	-1.990	0.562	-3.539	408	0.000	-3.095	-0.885
n	250 - 0	0.392	0.533	0.736	408	0.462	-0.656	1.440
	250 - 50	0.372	0.533	0.697	408	0.486	-0.677	1.420
	250 - 1000	-1.618	0.569	-2.846	408	0.005	-2.736	-0.500
	1000 - 0	2.010	0.562	3.578	408	0.000	0.906	3.115
	1000 - 50	1.990	0.562	3.539	408	0.000	0.885	3.095
	1000 - 250	1.618	0.569	2.846	408	0.005	0.500	2 736





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Table S8. Continued.

∑ . O	OMA treatment	Contrast	Ę	,	9		95% confidence interval	ice interval
pairwis (mg/kg	pairwise contrast (mg/kg dry wt.)	estimate	SE	+	đ	Adj. sig.	Lower	Upper
0 - 50	•	-0.461	0.476	-0.967	408	0.334	-1.397	0.476
0 - 250	0	0.330	0.463	0.712	408	0.477	-0.581	1.241
0 - 1000	00	-0.671	0.480	-1.398	408	0.163	-1.614	0.272
50 - 0		0.461	0.476	0.967	408	0.334	-0.476	1.397
50 - 250	50	0.791	0.471	1.679	408	0.094	-0.135	1.717
50 - 1000	000	-0.210	0.487	-0.431	408	0.667	-1.168	0.748
250 - 0	0	-0.330	0.463	-0.712	408	0.477	-1.241	0.581
250 - 50	50	-0.791	0.471	-1.679	408	0.094	-1.717	0.135
250 - 1000	1000	-1.001	0.474	-2.110	408	0.036	-1.934	-0.068
1000 - 0	0 -	0.671	0.480	1.398	408	0.163	-0.272	1.614
1000 - 50	- 50	0.210	0.487	0.431	408	0.667	-0.748	1.168
1000	1000 - 250	1.001	0.474	2.110	408	0.036	0.068	1.934

Table S9. Overall generalized linear mixed models test results for "Treatment × Day" effect on Gammarus locusta avoidance (MW1 experiment).

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Sig.	2.042E-05	0.001	0.143	
q£2	408	408	408	
lJp	3	3	8	
ഥ	8.378	5.463	1.818	
Day	2	3	4	





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**Table S10.** Estimated means of target variable (Individuals in control section) for "Treatment  $\times$  Day" effect on *Gammarus locusta* avoidance in the second marine experiment (MW2).

DMA treatment				95% confiden	ce interval
(mg/kg dry wt.)	Day	Mean	SE	Lower	Upper
	2	7.995	0.369	7.270	8.720
0	3	7.355	0.416	6.537	8.174
	4	7.897	0.403	7.104	8.689
	2	7.662	0.361	6.952	8.371
50	3	7.184	0.412	6.375	7.993
	4	7.583	0.395	6.807	8.360
	2	7.994	0.369	7.269	8.718
250	3	7.589	0.423	6.757	8.420
	4	8.378	0.415	7.561	9.194
	2	8.182	0.373	7.448	8.915
1000	3	6.787	0.400	6.001	7.574
	4	7.527	0.394	6.754	8.301









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# Chapter 5

Table S11. Pairwise contrasts for "Treatment × Day" effect on Gammarus locusta avoidance (MW2 experiment).

2	DMA treatment	Contrast	S T	+	4	7 6:5	95% confidence interval	ice interval
Jay	(mg/kg dry wt.)	estimate	3E	1	T	Auj. sig.	Lower	Upper
	0 - 50	0.334	0.516	0.646	408	0.518	-0.681	1.348
	0 - 250	0.001	0.521	0.003	408	0.998	-1.024	1.027
	0 - 1000	-0.186	0.525	-0.356	408	0.722	-1.218	0.845
	50 - 0	-0.334	0.516	-0.646	408	0.518	-1.348	0.681
	50 - 250	-0.332	0.516	-0.644	408	0.520	-1.346	0.682
c	50 - 1000	-0.520	0.519	-1.002	408	0.317	-1.540	0.500
4	250 - 0	-0.001	0.521	-0.003	408	0.998	-1.027	1.024
	250 - 50	0.332	0.516	0.644	408	0.520	-0.682	1.346
	250 - 1000	-0.188	0.525	-0.358	408	0.720	-1.219	0.843
	1000 - 0	0.186	0.525	0.356	408	0.722	-0.845	1.218
	1000 - 50	0.520	0.519	1.002	408	0.317	-0.500	1.540
	1000 - 250	0.188	0.525	0.358	408	0.720	-0.843	1.219
	0 - 50	0.171	0.585	0.293	408	0.770	-0.979	1.322
	0 - 250	-0.233	0.594	-0.393	408	0.694	-1.400	0.933
	0 - 1000	0.568	0.577	0.984	408	0.326	-0.567	1.703
	50 - 0	-0.171	0.585	-0.293	408	0.770	-1.322	0.979
	50 - 250	-0.405	0.590	-0.686	408	0.493	-1.565	0.755
3	50 - 1000	0.397	0.574	0.691	408	0.490	-0.731	1.525
	250 - 0	0.233	0.594	0.393	408	0.694	-0.933	1.400
	250 - 50	0.405	0.590	989.0	408	0.493	-0.755	1.565
	250 - 1000	0.802	0.582	1.377	408	0.169	-0.343	1.946
	1000 - 0	-0.568	0.577	-0.984	408	0.326	-1.703	0.567
	1000 - 50	-0.397	0.574	-0.691	408	0.490	-1.525	0.731







Table S11. Continued.

غ ا	DMA treatment	Contrast	2	+	<b>\$</b>	.; .; V	95% confidence interval	ice interval
Cay	(mg/kg dry wt.)	estimate	SE	1	īn	Auj. sig.	Lower	Upper
3	1000 - 250	-0.802	0.582	-1.377	408	0.169	-1.946	0.343
	0 - 50	0.314	0.564	0.556	408	0.579	-0.796	1.423
	0 - 250	-0.481	0.579	-0.831	408	0.407	-1.618	0.657
	0 - 1000	0.369	0.563	0.656	408	0.512	-0.738	1.477
	50 - 0	-0.314	0.564	-0.556	408	0.579	-1.423	0.796
	50 - 250	-0.795	0.573	-1.387	408	0.166	-1.921	0.332
-	50 - 1000	0.056	0.558	0.100	408	0.921	-1.040	1.152
4	250 - 0	0.481	0.579	0.831	408	0.407	-0.657	1.618
	250 - 50	0.795	0.573	1.387	408	0.166	-0.332	1.921
	250 - 1000	0.850	0.572	1.486	408	0.138	-0.274	1.975
	1000 - 0	-0.369	0.563	-0.656	408	0.512	-1.477	0.738
	1000 - 50	-0.056	0.558	-0.100	408	0.921	-1.152	1.040
	1000 - 250	-0.850	0.572	-1.486	408	0.138	-1.975	0.274

Table S12. Overall generalized linear mixed models test results for "Treatment × Day" effect on Gammarus locusta avoidance (MW2 experiment).

Sig.	0.789	0.564	0.433
df2	408	408	408
lJp	3	3	3
H	0.351	0.682	0.916
Day	2	3	4





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**Table S13.** Estimated means of target variable (Individuals in control section) for "Experiment  $\times$  Treatment  $\times$  Day" effect on *Gammarus locusta* avoidance (MW1 and MW2 experiments).

г	DMA treatment	Ъ		C.E.	95% confide	nce interval
Experiment	(mg/kg dry wt.)	Day	Mean	SE	Lower	Upper
		2	7.443	0.381	6.695	8.192
	0	3	7.027	0.394	6.253	7.801
		4	7.159	0.379	6.414	7.903
		2	7.682	0.388	6.921	8.443
	50	3	7.197	0.399	6.414	7.981
MW1		4	7.396	0.386	6.639	8.153
IVI W I		2	8.206	0.401	7.420	8.992
	250	3	7.495	0.407	6.696	8.295
		4	7.257	0.382	6.507	8.007
		2	9.918	0.440	9.053	10.782
	1000	3	9.099	0.449	8.218	9.980
		4	7.674	0.393	6.903	8.444
		2	7.952	0.394	7.178	8.726
	0	3	7.422	0.405	6.626	8.218
		4	7.701	0.393	6.928	8.473
		2	7.644	0.387	6.885	8.403
	50	3	7.058	0.395	6.282	7.834
) (IV)		4	7.684	0.393	6.913	8.456
MW2		2	8.068	0.397	7.288	8.847
	250	3	7.584	0.410	6.779	8.388
		4	8.194	0.406	7.398	8.991
		2	8.085	0.398	7.305	8.866
	1000	3	6.872	0.390	6.106	7.637
		4	7.555	0.390	6.790	8.320



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Table S14. Pairwise contrasts for "Experiment × Treatment × Day" effect on Gammarus locusta avoidance (MW1 and MW2 experiments).

DMA treatment	4	Experiment	Contrast	Ç	,	31	:	95% confidence interval	ice interval
(mg/kg dry wt.)	Day	pairwise contrast	estimate	SE	1	aı	Adj. sig.	Lower	Upper
	c	MW1 - MW2	-0.508	0.549	-0.927	816	0.354	-1.585	0.568
	4	MW2 - MW1	0.508	0.549	0.927	816	0.354	-0.568	1.585
c	,	MW1 - MW2	-0.395	0.566	869.0-	816	0.485	-1.505	0.715
o .	n	MW2 - MW1	0.395	0.566	0.698	816	0.485	-0.715	1.505
	_	MW1 - MW2	-0.542	0.547	-0.992	816	0.322	-1.615	0.531
	4	MW2 - MW1	0.542	0.547	0.992	816	0.322	-0.531	1.615
	ر	MW1 - MW2	0.038	0.547	0.069	816	0.945	-1.037	1.112
	4	MW2 - MW1	-0.038	0.547	-0.069	816	0.945	-1.112	1.037
Ċ.	,	MW1 - MW2	0.139	0.562	0.247	816	0.805	-0.964	1.242
00	n	MW2 - MW1	-0.139	0.562	-0.247	816	0.805	-1.242	0.964
	-	MW1 - MW2	-0.288	0.551	-0.524	816	0.601	-1.369	0.792
	4	MW2 - MW1	0.288	0.551	0.524	816	0.601	-0.792	1.369
	ر	MW1 - MW2	0.138	0.564	0.245	816	0.806	-0.969	1.246
	1	MW2 - MW1	-0.138	0.564	-0.245	816	0.806	-1.246	696.0
030	,	MW1 - MW2	-0.089	0.578	-0.153	816	0.878	-1.223	1.046
730	n	MW2 - MW1	0.089	0.578	0.153	816	0.878	-1.046	1.223
	_	MW1 - MW2	-0.937	0.557	-1.681	816	0.093	-2.031	0.157
	†	MW2 - MW1	0.937	0.557	1.681	816	0.093	-0.157	2.031
	c	MW1 - MW2	1.832	0.593	3.088	816	0.002	0.668	2.997
	7	MW2 - MW1	-1.832	0.593	-3.088	816	0.002	-2.997	-0.668
0001	,	MW1 - MW2	2.227	0.595	3.745	816	0.000	1.060	3.395
1000	ر ا	MW2 - MW1	-2.227	0.595	-3.745	816	0.000	-3.395	-1.060
	-	MW1 - MW2	0.118	0.553	0.214	816	0.831	-0.968	1.204
	+	MW2 - MW1	-0.118	0.553	-0.214	816	0.831	-1.204	896.0





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**Table S15.** Overall generalized linear mixed models test results for "Experiment  $\times$  Treatment  $\times$  Day" effect on *Gammarus locusta* avoidance (MW1 and MW2 experiments).

DMA treatment (mg/kg dry wt.)	Day	F	dfl	df2	Sig.
	2	0.859	1	816	0.354
0	3	0.488	1	816	0.485
	4	0.984	1	816	0.322
	2	0.005	1	816	0.945
50	3	0.061	1	816	0.805
	4	0.274	1	816	0.601
	2	0.060	1	816	0.806
250	3	0.023	1	816	0.878
	4	2.827	1	816	0.093
	2	9.538	1	816	0.002
1000	3	14.026	1	816	0.000
	4	0.046	1	816	0.831







**Table S16.** Estimated means of target variable (Individuals in control section) for "Experiment  $\times$  Treatment  $\times$  Day" effect on *Gammarus pulex* (FW experiment) and *Gammarus locusta* avoidance (MW1 and MW2 experiments).

Evenovimont	DMA treatment	Davi	Mean	SE	95% confider	nce interval
Experiment	(mg/kg dry wt.)	Day	Mean	SE	Lower	Upper
		2	7.688	0.639	6.434	8.942
	0	3	8.498	0.674	7.174	9.821
		4	7.810	0.658	6.520	9.101
		2	7.800	0.644	6.537	9.063
	50	3	7.991	0.654	6.707	9.274
FW		4	6.644	0.607	5.454	7.835
ΓW		2	10.093	0.732	8.656	11.530
	250	3	9.784	0.724	8.364	11.204
		4	9.249	0.716	7.844	10.653
		2	12.819	0.825	11.199	14.438
	1000	3	11.663	0.790	10.112	13.214
		4	11.802	0.808	10.215	13.388
		2	7.467	0.412	6.659	8.276
MW1	0	3	7.024	0.392	6.255	7.793
		4	7.159	0.379	6.414	7.903
		2	7.709	0.418	6.888	8.530
	50	3	7.222	0.397	6.443	8.002
		4	7.396	0.386	6.639	8.153
		2	8.190	0.431	7.344	9.036
	250	3	7.506	0.405	6.711	8.300
		4	7.257	0.382	6.508	8.007
		2	10.026	0.477	9.090	10.963
	1000	3	9.111	0.446	8.236	9.987
		4	7.674	0.393	6.903	8.444
		2	7.905	0.424	7.074	8.737
	0	3	7.402	0.402	6.612	8.191
		4	7.701	0.393	6.929	8.473
		2	7.601	0.415	6.785	8.416
	50	3	7.079	0.393	6.307	7.851
MW2	4		7.684	0.393	6.913	8.456
1V1 VV Z			7.985	0.426	7.149	8.820
	250	3	7.598	0.407	6.798	8.397
		4	8.194	0.406	7.398	8.991
		2	8.154	0.430	7.309	8.998
	1000	3	6.845	0.387	6.086	7.604
		4	7.555	0.390	6.790	8.320







Table S17. Pairwise contrasts for "Experiment × Treatment × Day" effect on Gammarus pulex (FW experiment) and Gammarus locusta avoidance (MW1 and MW2 experiments).

	2	Experiment	Contrast	Ç	,	31	-	95% confidence interval	ce interval
DMA treatment (mg/kg dry wt.)	Day	pairwise contrast	estimate	OE.	1	Œ.	Adj. sig.	Lower	Upper
		FW - MW1	0.221	092.0	0.290	984	0.772	-1.271	1.712
		FW - MW2	-0.217	0.767	-0.283	984	0.777	-1.722	1.287
	c	MW1 - FW	-0.221	0.760	-0.290	984	0.772	-1.712	1.271
	7	MW1 - MW2	-0.438	0.591	-0.741	984	0.459	-1.597	0.722
		MW2 - FW	0.217	0.767	0.283	984	0.777	-1.287	1.722
		MW2 - MW1	0.438	0.591	0.741	984	0.459	-0.722	1.597
		FW - MW1	1.473	0.780	1.889	984	0.059	-0.057	3.004
		FW - MW2	1.096	0.785	1.396	984	0.163	-0.445	2.637
c	ŗ	MW1 - FW	-1.473	0.780	-1.889	984	0.059	-3.004	0.057
D)	n	MW1 - MW2	-0.378	0.561	-0.672	984	0.501	-1.479	0.724
		MW2 - FW	-1.096	0.785	-1.396	984	0.163	-2.637	0.445
		MW2 - MW1	0.378	0.561	0.672	984	0.501	-0.724	1.479
		FW - MW1	0.652	0.759	0.859	984	0.391	-0.838	2.142
		FW - MW2	0.110	0.766	0.143	984	0.886	-1.394	1.614
	-	MW1 - FW	-0.652	0.759	-0.859	984	0.391	-2.142	0.838
	4	MW1 - MW2	-0.542	0.547	-0.992	984	0.322	-1.615	0.530
		MW2 - FW	-0.110	0.766	-0.143	984	0.886	-1.614	1.394
		MW2 - MW1	0.542	0.547	0.992	984	0.322	-0.530	1.615
		FW - MW1	0.092	0.768	0.119	984	0.905	-1.415	1.598
		FW - MW2	0.200	0.766	0.260	984	0.795	-1.304	1.703
CV	c	MW1 - FW	-0.092	0.768	-0.119	984	0.905	-1.598	1.415
20	4	MW1 - MW2	0.108	0.590	0.183	984	0.855	-1.049	1.265
		MW2 - FW	-0.200	0.766	-0.260	984	0.795	-1.703	1.304
		MW2 - MW1	-0.108	0.590	-0.183	984	0.855	-1.265	1.049

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Table S17. Continued.

DMA A LOCAL COLUMN COLU	2	Experiment	Contrast	ç	•	31	1 1	95% confidence interval	ice interval
DIMA treatment (mg/kg dry wt.) Day	Day	pairwise contrast	estimate	NE.	-	đ	Adj. sig.	Lower	Upper
		FW - MW1	0.768	0.765	1.004	984	0.316	-0.733	2.270
		FW - MW2	0.911	0.763	1.194	984	0.233	-0.586	2.409
	ŗ	MW1 - FW	-0.768	0.765	-1.004	984	0.316	-2.270	0.733
	n	MW1 - MW2	0.143	0.559	0.256	984	0.798	-0.954	1.240
		MW2 - FW	-0.911	0.763	-1.194	984	0.233	-2.409	0.586
Q.		MW2 - MW1	-0.143	0.559	-0.256	984	0.798	-1.240	0.954
00		FW - MW1	-0.752	0.719	-1.046	984	0.296	-2.162	0.659
		FW - MW2	-1.040	0.723	-1.439	984	0.150	-2.459	0.378
	-	MW1 - FW	0.752	0.719	1.046	984	0.296	-0.659	2.162
	4	MW1 - MW2	-0.288	0.551	-0.524	984	0.601	-1.369	0.792
		MW2 - FW	1.040	0.723	1.439	984	0.150	-0.378	2.459
		MW2 - MW1	0.288	0.551	0.524	984	0.601	-0.792	1.369
		FW - MW1	1.903	0.850	2.240	984	0.025	0.235	3.570
		FW - MW2	2.108	0.847	2.489	984	0.013	0.446	3.770
	c	MW1 - FW	-1.903	0.850	-2.240	984	0.025	-3.570	-0.235
	4	MW1 - MW2	0.205	909.0	0.339	984	0.735	-0.984	1.395
		MW2 - FW	-2.108	0.847	-2.489	984	0.013	-3.770	-0.446
030		MW2 - MW1	-0.205	909.0	-0.339	984	0.735	-1.395	0.984
067		FW - MW1	2.279	0.829	2.748	984	90000	0.651	3.906
		FW - MW2	2.187	0.831	2.633	984	0.009	0.557	3.816
	ŗ	MW1 - FW	-2.279	0.829	-2.748	984	0.006	-3.906	-0.651
	3	MW1 - MW2	-0.092	0.574	-0.160	984	0.873	-1.219	1.035
		MW2 - FW	-2.187	0.831	-2.633	984	0.009	-3.816	-0.557
		MW2 - MW1	0.092	0.574	0.160	984	0.873	-1.035	1.219

**(** 



Table S17. Continued.

	2	Experiment	Contrast	20	+	J+7	: T	95% confidence interval	nce interval
DIMA treatment (mg/kg dry wt.)	Day	pairwise contrast	estimate	SE	1	Ħ	Auj. sig.	Lower	Upper
		FW - MW1	1.992	0.811	2.455	984	0.014	0.400	3.584
		FW - MW2	1.054	0.823	1.282	984	0.200	-0.560	2.669
Cac	-	MW1 - FW	-1.992	0.811	-2.455	984	0.014	-3.584	-0.400
720	4	MW1 - MW2	-0.937	0.557	-1.681	984	0.093	-2.031	0.157
		MW2 - FW	-1.054	0.823	-1.282	984	0.200	-2.669	0.560
		MW2 - MW1	0.937	0.557	1.681	984	0.093	-0.157	2.031
		FW - MW1	2.792	0.953	2.930	984	0.003	0.922	4.663
		FW - MW2	4.665	0.931	5.013	984	6.356E-07	2.839	6.491
	c	MW1 - FW	-2.792	0.953	-2.930	984	0.003	-4.663	-0.922
	7	MW1 - MW2	1.872	0.643	2.914	984	0.004	0.611	3.133
		MW2 - FW	-4.665	0.931	-5.013	984	6.356E-07	-6.491	-2.839
		MW2 - MW1	-1.872	0.643	-2.914	984	0.004	-3.133	-0.611
		FW - MW1	2.552	0.907	2.812	984	0.005	0.771	4.333
		FW - MW2	4.818	0.880	5.477	984	5.504E-08	3.092	6.545
0001	Ç	MW1 - FW	-2.552	0.907	-2.812	984	0.005	-4.333	-0.771
1000	n	MW1 - MW2	2.266	0.590	3.838	984	0.000	1.107	3.425
		MW2 - FW	-4.818	0.880	-5.477	984	5.504E-08	-6.545	-3.092
		MW2 - MW1	-2.266	0.590	-3.838	984	0.000	-3.425	-1.107
•		FW - MW1	4.128	0.899	4.593	984	4.939E-06	2.364	5.892
		FW - MW2	4.247	0.898	4.732	984	2.553E-06	2.485	800.9
	-	MW1 - FW	-4.128	0.899	-4.593	984	4.939E-06	-5.892	-2.364
	4	MW1 - MW2	0.118	0.553	0.214	984	0.831	-0.967	1.204
		MW2 - FW	-4.247	0.898	-4.732	984	2.553E-06	-6.008	-2.485
		MW2 - MW1	-0.118	0.553	-0.214	984	0.831	-1.204	0.967





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**Table S18.** Overall generalized linear mixed models test results for "Experiment × Treatment × Day" effect on *Gammarus pulex* (FW experiment) and *Gammarus locusta* avoidance (MW1 and MW2 experiments).

DMA treatment (mg/kg dry wt.)	Day	F	df1	df2	Sig.
	2	0.275	2	984	0.760
0	3	1.784	2	984	0.168
	4	0.642	2	984	0.526
	2	0.038	2	984	0.963
50	3	0.731	2	984	0.482
	4	1.036	2	984	0.355
	2	3.265	2	984	0.039
250	3	4.124	2	984	0.016
	4	3.455	2	984	0.032
	2	13.590	2	984	1.507E-06
1000	3	17.783	2	984	2.588E-08
	4	12.033	2	984	6.873E-06













# 6 Synthesis





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# Introduction

In the Arctic, combined effects of climate change and globalization will increase oil and gas (O&G) exploration prospects. This will lead to more O&G and shipping activities [337] causing higher risks of pollution (Figure 1). In Norway and Russia, oil production from existing offshore wells will likely increase in the Barents Sea and Pechora Sea, and Arctic shipping routes such as the Northern Sea Route along the Russian coast are expected to be used more frequently. Such activities will be subjected to sea ice, strong winds and currents, which may enhance the risk of accidents during O&G transport [338] (Chapter 1).

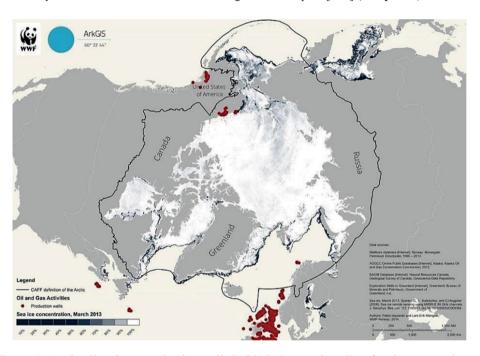


Figure 1. Arctic oil and gas production wells in 2013. Source: http://wwfarcticmaps.org/

Until now, environmental risks of O&G activities in the Arctic have mainly been evaluated using a conventional environmental risk assessment approach (Figure 2). For instance, threshold values for effects have been established based on species sensitivity distribution (SSD) curves for sub-cellular biomarker responses in water exposed species and for whole organism responses [339,340]. On the Norwegian Continental Shelf, environmental risk assessment of produced water (PW) involves the traditional steps in risk assessment (Figure 2). These are derived from the European Commission [341], and commonly using oil dispersions as proxies for PW and other crude oil discharges. Such conventional approaches





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mainly apply to the water column. So far, specific approaches to assess risks for benthic systems in the Arctic have not been defined.

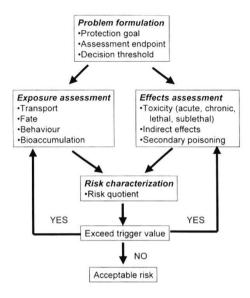


Figure 2. Environmental risk assessment process. Source: Maltby [342].

Current programmes monitoring the impacts of O&G activities on the Arctic parts of the Norwegian Continental Shelf include the water column and benthic habitats [343]. According to the Norwegian Environment Agency, both need to be surveyed every three years in both regional and field-specific stations of the same geographical region. A minimum of three wild-caught fish species and caged mussels are usually recommended for monitoring of exposure and effects in the water column [343]. As for the monitoring of benthic habitats, only the taxonomic identification and species density are required. In the Barents Sea, monitoring of benthic habitats in the Barents Sea South region is planned to start in 2019.

Polycyclic aromatic hydrocarbons (PAHs) are often used as indicators of oil related pollution [14], as petrogenic PAHs are enriched in weathered oil and reside in sediment with oilspecific chemical signatures [23,43,120]. Oil does not behave identically to PAHs in the marine environment, especially when it concerns fate processes. However, PAHs can be considered a major fraction contributing to the toxicity of oil. Therefore, in this thesis, several chapters are focused on the behaviour and effects of PAHs (**Chapters 2-4**), whereas one chapter studies a widely used gas oil; Distillate Marine grade A (DMA; **Chapter 5**).







As specified before, there is a high need for Arctic specific approaches and monitoring. To achieve this, the evaluation of environmental risks of O&G activities in the Arctic should take into account the specific physiology of Arctic species and environmental characteristics. Furthermore, the uncertainty of exposure pathways of oil related compounds should be addressed. In this thesis, four potential tools for the monitoring and risk assessment of O&G impacts on Arctic benthic species were addressed: (1) the use of biota-sediment accumulation factors (BSAFs) as indicators for sediment PAH bioavailability (Chapters 2,3), (2) the applicability of biomarkers of exposure (Chapter 4) and (3) the suitability of biotransformation metabolites in Arctic benthos chronically exposed to PAHs (Chapter 4), and (4) the relevance of avoidance tests as a tool for oil impact assessment (Chapter 5). These four methods were used to identify potential bioindicators for O&G and shipping related chemical stress in Arctic coastal benthic systems. For these tools six research questions were described (Chapter 1), which are discussed below.

# Q1: Can PAH body residues or BSAFs be used as bioavailability indicators in Arctic coastal benthic systems and which target species is most suitable to detect trends?

In this thesis, bioaccumulation from sediment was evaluated under field (**Chapter 2**) and laboratory conditions (**Chapter 3**). The results showed that PAH partitioning deviated from what can be expected from Equilibrium Partitioning Theory (e.g., BSAF having a value of 1-2 [82]). This was explained by the presence of black carbon in the sediment, a bioaccumulation non-equilibrium, experimental variability or by any combination of these factors. This led to both lower and higher BSAF values than the range predicted by the Equilibrium Partitioning Theory.

In the field study (**Chapter 2**), significantly higher BSAFs were found in individuals from all species collected at a reference area (Blomstrandhalvøya) compared to those collected at a relatively impacted area (Ny-Ålesund). This was mostly ascribed to a higher abundance of black carbon in Ny-Ålesund sediment. For the majority of PAH compounds, body residues differed significantly between species. In Ny-Ålesund, body residues were highest in *M. calcarea* compared to body residues in *A. borealis* and *N. ciliata*.

In the laboratory study (**Chapter 3**), significantly higher BSAFs (i.e., for 3-ring PAHs) and body residues (i.e., for 3- and 4-ring PAHs) were observed in *A. borealis* than in *N. ciliata*. This was explained by a difference in feeding activity between these species and by a higher content in black carbon and amorphous organic matter in the polychaetes than in the bivalves







sediment treatment that may have resulted in a lower bioavailability and thus lower PAH uptake in *N. ciliata*.

In the literature, many examples apply body residues and/or BSAF values as indicators for bioavailability and bioaccumulation (e.g., [56,80,82]). However, deviations from BSAF values of 1-2 have also been reported frequently, sometimes limiting the applicability of BSAF as an unambiguous and reliable monitoring metric. This is also seen in the present research, where BSAFs were observed to vary widely. This means that the present work confirms the limited applicability of BSAF values as a relevant metric for biomonitoring in the Arctic as well.

As for body residues, also many literature reports exist where biomonitoring is based on concentrations in organisms over time and space [344,345]. When lipid-normalised, these concentrations are a more pure descriptor of bioavailability and bioaccumulation as they directly reflect exposure on the individual level. Furthermore, body residues can be compared directly with lipid based toxicity thresholds (e.g., the critical body burden – CBR – concept, see **Chapter 2**), which is a well-established approach in environmental risk assessment [154].

Based on the above, I conclude that PAH body residues provide a more realistic estimate of the bioavailable PAH fraction than BSAFs. This also implies that body residues are a better estimator of exposure in the context of effect- and risk assessment in Arctic benthic communities than BSAFs. Therefore, body residues are more suitable to derive relevant environmental quality standards for Arctic species. However, BSAFs are still useful indicators to help detect possible mechanisms affecting body residues, like for instance, variability in organic matter composition (e.g., black carbon), biomagnification and/or non-equilibrium as mentioned above.

Of all evaluated species I recommend *Macoma calcarea* as target species being most suitable to detect trends. This is based on its ability to accumulate PAHs in low impacted areas (e.g., Ny-Ålesund), regardless of their body size, and on their selective feeding mode, which decreases the variability of body residues measured in the species (**Chapter 2**). Recently, *M. calcarea* also was suggested as an ecological indicator in the context of developing trait-based approaches to explore changes in rapidly changing ecosystems [346]. However, in the absence of *M. calcarea* a similar method can be applied to select the next suitable indicator organism, using the criteria from **Chapter 1**. For instance, *Nuculana pernula* would be the closest species based on its traits (**Chapter 3**).







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Recommendations. Because of the extensive variability in BSAFs and body residues it is recommended to minimize experimental variability. This can be achieved by enhancing sediment homogeneity, representativeness and uniformity of test animals taken from the field, quality and homogeneity of food supply, in laboratory tests with Arctic species. Standardized sediment toxicity testing [347] can be used to further develop protocols for Arctic benthic testing. A limiting practical factor is the presence of often less elaborated laboratory facilities for Arctic research that has to be dealt with.

A second recommendation is to further explore the relation between bioaccumulation (body residues) and environmental factors (e.g., temperature, particulate organic carbon, contaminant concentration) through multivariate statistical techniques [315,348-350] and based on this develop bioaccumulation models and scenarios for exposure testing.

A final recommendation is to apply the suggested biomonitoring tool, that is, use *M. calcarea* to detect trends and spatial differences in Arctic areas with O&G related pollution gradients.

Together, this will provide necessary information on spatial and temporal scales of gradients in bioaccumulation in the Arctic. When sediments need to be sampled for laboratory exposure experiments, such information on spatial heterogeneity in chemical exposure can be used to select sampling locations for these sediments.

# Q2: Can species traits explain differences in PAH bioaccumulation between Arctic benthic invertebrates?

In this thesis, bioaccumulation was explored for species with different traits under both field (Chapter 2) and laboratory conditions (Chapter 3). The uncertainty in bioaccumulation metrics (BSAFs, body residues) was large, especially in the laboratory data. In the field study (Chapter 2), a higher bioaccumulation of PAHs was observed for *M. calcarea* in Ny-Ålesund compared to *A. borealis* and *N. ciliata*. This was explained by the high preference for freshly deposited organic matter, higher ingestion rate and lower PAH excretion capacity of *M. calcarea*. In the laboratory study (Chapter 3), a lower bioaccumulation of PAHs was observed for *M. calcarea* compared to *A. borealis* and *N. ciliata*. This was explained by a general decrease of PAH uptake and metabolic rate in *M. calcarea* under laboratory conditions. This decrease of uptake can be explained from the high diet selection of *M. calcarea* [211]. The decrease in metabolic rate can possibly be explained from the metabolic rate depression experienced by Arctic bivalves during starvation periods [351]. This illustrates





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how laboratory settings can yield quite different results when compared to observations from field data.

According to the literature, bioaccumulation is strongly influenced by exposure routes, which in turn are determined by species traits, such as the feeding mode, ingestion – egestion rates, capability to metabolise chemicals, and diet composition [56,83,274,352,353]. Validated models are available to quantify the link between these traits and bioaccumulation [56,80,354]. For Arctic ecosystems, it has been reported that different benthic groups respond differently to the seasonality in the abundance of organic matter [127]. However, according to Włodarska-Kowalczuk et al. [355], a relatively high abundance of macroalgal detritus in the Kongsfjorden system makes benthic organisms less sensitive to the seasonality in pelagic production. Under such conditions of abundant food supply, differences in bioaccumulation will be mostly explained by other factors, such as the species' ability to metabolize and excrete PAHs.

Based on the aforementioned literature data and data from this thesis, I conclude that bioaccumulation is also dependent on species traits in Arctic species. The link between species-specific feeding mode and bioaccumulation was mechanistically explained using models (**Chapter 3**). However, the model's statistical significance remained limited as compared to earlier model evaluations, because of the considerable residual error between observations and model predictions.

*Recommendations*. Because of the high variability in BSAFs and body residues, it is recommended to reduce experimental variability. This will increase the rigour of tests that aim to detect differences in bioaccumulation among species with different traits, as well as the significance of models aimed at mechanistically explaining the observations.

To be able to explain differences in bioaccumulation among species from the field, the characteristics relating to quality and quantity of organic matter in all different diet components (e.g., sediment and freshly deposited organic matter) need to be determined. This includes quantifying organic matter, black carbon, oil and other possible solid components involved in binding hydrophobic chemicals.

Finally, it is recommended to assess bioaccumulation for a wider range of species in the field, together with community composition and sediment characteristics at the sites of interest. This will allow to study bioaccumulation patterns over time and space, for a much wider







range of species traits, including improved opportunities for mechanistic modelling. Furthermore, this allows for establishing causal relationships between bioaccumulation and community composition *in situ*.

Q3: Does bioaccumulation of PAHs differ between Arctic and temperate benthic invertebrate species with different traits, and can modelling assist in reading across between species?

In my research, no significant differences were observed in PAH body residues and BSAFs between Arctic and temperate benthic species under laboratory conditions (**Chapter 3**). This was mainly due to the substantial biological variability in the data.

Little is known about bioaccumulation of PAHs from sediment by Arctic benthic species, compared to bioaccumulation by temperate species. To my understanding, no studies have been published on the direct comparison of bioaccumulation by Arctic versus temperate benthic species. Two studies report PAH body residues in Arctic benthic species [121,356]. One of these studies included the modelling of bioaccumulation factors (BAFs) in blue mussels obtained from literature to infer sea water concentrations [121]. In this sense, this study (Chapter 3) is the first attempt to compare and to model bioaccumulation in similar Arctic and temperate benthic species.

Hypotheses were provided based on published differences in species traits with relevance for bioaccumulation. For instance, feeding rate, clearance rate and digestion rates often are lower in Arctic species [213,357,358], that may affect both bioaccumulation and risks. These differences were not observed in our experiments, due to the fact that they were probably smaller than the current variability in the data (**Chapter 3**).

For temperate species, the use of biodynamic modelling has helped to provide a mechanistic understanding of bioaccumulation processes [56,83]. The models can also be used to predict PAH bioaccumulation in the absence of experimental data [203], and to read across species, compartments (freshwater *vs* marine) and potentially climate regions. Examples of modelling of contaminants across temperate and Arctic pelagic species are provided by some studies [359,360]. These studies concluded that bioaccumulation in the Arctic may be described in the same way as bioaccumulation in temperate regions. In this thesis, however, dynamic simulations among species with similar traits and exposed to PAHs under laboratory conditions did not reveal any clear differences in the model parameters between species or







tests. This was due to the substantial variability in the biological control data and the similarity in exposure routes (Chapter 3).

Despite the lack of significant differences in PAH bioaccumulation between Arctic and temperate species in our study, I argue that differences between Arctic and temperate species can be detected if more rigorous testing methods are applied in these tests, and/or if other selections of species, individuals or sediments would be used in these tests.

*Recommendations*. As specified above, there is a need to improve the accuracy of bioaccumulation testing. This can be achieved by a series of measures, such as the use of non-exposed test animals, artificial sediment, a higher number of replicates and time points, and/or a more constant food addition.

These improvements will also increase the applicability of bioaccumulation models and lead to more certain parameter values. Additionally, several model parameters need to be measured in an experimental set up, in order to increase the reliability of bioaccumulation models for Arctic species (such as growth and feeding rates of species). To get a better understanding when comparing Arctic and temperate species, specific tests can be done where sediments either are identical for both Arctic and temperate species, or a suitable representative of the region's sediment type. The tests can be done with Arctic and temperate species at the same temperature, or at temperatures specific to Arctic and temperate conditions. This way the relative importance of both temperature and sediment properties can be identified which helps to further elucidate potential differences among species and regions.

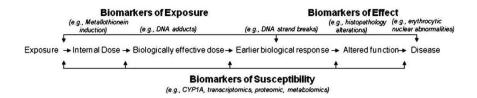
# Q4: What is the feasibility of using biomarkers of exposure to monitor PAH pollution in the Arctic? Is the identification of biotransformation metabolites a better alternative?

For an integrated monitoring and assessment programme for the Arctic, biomarkers and bioassays need to be tested and validated in Arctic organisms that are representative for the water and sediment compartment [91,361]. For the benthic compartment, benthic invertebrates may fulfil the role of bioindicators of chemical pollution, just as fish species do for the water compartment [343,362]. In this respect it is essential to causally link chemical body residues and biomarkers measured at different suborganismal levels (Figure 3), in order to increase the interpretative value of individual measurements [345]. Therefore, in my study, I analysed both PAH body residues and biomarkers of exposure/effect in target species.









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**Figure 3.** Continuum from chemical exposure to organism level effects. Source: Chapman et al. [363].

For an integrated exposure an effect assessment (Figure 2), two biomarkers of PAH exposure (i.e., acyl-CoA oxidase [AOX] and glutathione S-transferase [GST]) and one biomarker of neurotoxicity (i.e., acetylcholinesterase [AChE]) in *A. borealis* were addressed (**Chapter 4**). GST and AChE have been validated in temperate bivalves before. They are commonly used in the assessment of oil impact [364] and pollution monitoring [365], whereas AOX generally shows a rapid response to PAH exposure in temperate bivalves [279,281]. In the present work, interpretation of enzymatic biomarkers data in *A. borealis* was limited. This is because either baseline studies are lacking for Arctic benthic species [215], or these biomarkers were assessed in water exposed rather than sediment exposed Arctic benthic organisms [366,367].

Pyrene metabolites, and particularly 1-hydroxypyrene, have been used before as a suitable biomarker of PAH exposure in temperate species [368,369]. Our study made the first multispecies comparison of pyrene biotransformation between Arctic and temperate benthic invertebrates (**Chapter 4**). This was done to estimate differences in their biotransformation capacity as well as to validate the use of 1-hydroxypyrene and pyrene-1-glucuronide [183] as biomarkers of PAH exposure in Arctic species.

Unlike pyrene body residues, a significant increase in the concentration of pyrene-1-glucuronide was observed in polychaetes exposed to the high sediment treatment compared to the low sediment treatment (**Chapter 4**). This increase was not detected in bivalve species. In polychaetes, biotransformation rates did not vary between treatments in our study. This was illustrated by the similar relative concentrations of pyrene-1-glucuronide in polychaetes exposed to the different sediment treatments, implying biotransformation also at low concentrations. As for enzymatic biomarkers, there was no correlation between the activity of biomarkers in *A. borealis* and PAH contamination levels in sediment treatments (**Chapter 4**). This could mainly be explained by the low PAH contamination levels in *A. borealis*, although the influence of confounding factors [134,283] could not be excluded. There is a lack of



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knowledge on pyrene metabolites in Arctic invertebrates and the use of biomarkers in Arctic invertebrates in general [213,370].

Based on the available literature data and data from this thesis, I conclude that the analysis of pyrene metabolites provides a feasible approach to assess PAH exposure in Arctic benthic populations. However, it would be advisable to review the sensitivity of additional biomarkers of exposure and biomarkers of effect in order to develop a further weight-of-evidence approach.

Recommendations. In order to provide a more robust and integrated assessment of PAH exposure in Arctic species, the ability of different markers of exposure needs to be evaluated to provide an early warning of risks on PAH mediated biological effects. This can be achieved by exploring the relative effect of environmental and biological factors on the activity of biomarkers with respect to PAH exposure [275]. The natural metabolic activity and seasonal dynamics of biomarkers in Arctic species are needed to be able to determine relevant biomarker responses in Arctic benthic invertebrates to PAH exposure in the field. Finally, our findings illustrate that Arctic benthic macroinvertebrates have a biotransformation capacity, which should not be overlooked when evaluating accumulation and risks of PAHs in the Arctic [93,94].

# Q5: Can behavioural tests like avoidance behaviour become a suitable monitoring tool for oil impact assessment in the Arctic?

In my research, a significant avoidance of DMA oil-spiked sediment was observed at 1000 mg/kg (dry weight). Furthermore, it was shown that for the freshwater species *G. pulex* this occurred within an exposure time of 96 h, whereas for the marine species *G. locusta* in one of two tests, this occurred within an exposure time of 72 h (**Chapter 5**). Hence, this shows that the tests are able to detect behavioural effects of gammarids to DMA oil, as well as a time dependency of these effects.

Differences in time dependency of avoidance behaviour between the two species, were explained by the experimental conditions (i.e., water parameters, feeding) on the species' sensitivities to DMA oil and to their ability to maintain active over time. The sensitivity to detect DMA oil was explained by a higher ventilation rate, caused by the exposure of *G. pulex* at its upper thermal tolerance limit (i.e., 15 °C). As for *G. locusta*, the ability to detect and







avoid DMA may have been restricted by a higher salinity in the marine experiments and low feeding potential.

Within-species differences in the avoidance response of *G. locusta* among tests were considerable, thereby limiting the rigour of the test (**Chapter 5**). This variability may have been related to the variability among individuals used in the tests. This originates from differences in structural growth and reproductive status of the tested individuals, as observed more often among summer *G. locusta* generations [327,329].

In the literature, behavioural endpoints such as the avoidance response to oil of benthic organisms have been proposed before as a tool to detect sublethal effects of oil in sediment [99,304-306]. Studies on the avoidance behaviour of oil-contaminated sediment by temperate marine amphipods [298] and of PAH-spiked sediment by temperate freshwater invertebrates [100] showed that avoidance behaviour was not affected by the background exposure history of the organisms to either oil or PAHs, and that avoidance tests could be used as an alternative to chronic toxicity testing. These studies also concluded that homogeneity in exposure conditions (e.g., sediment characteristics, feeding) or in test species individuals is crucial.

Variability among individuals may be less for Arctic species than for temperate species. Variability in energy levels of organisms is likely to affect avoidance behaviour [322,371]. Energy levels are affected by environmental factors (e.g., temperature and salinity [335,372]) and biological variables (e.g., structural growth rate [313,333,372] and reproductive status [327,329,372]). Some Arctic benthic marine amphipods (e.g., *Gammarus setosus*) show less variability than *G. locusta* with regards to a longer life-span and fewer reproductive events [373]. Based on this, *G. setosus* may be a suitable candidate species for effective avoidance tests for Arctic sediments exposed to oil.

Given the above, I conclude that avoidance tests are suitable for monitoring oil impact in temperate regions, whereas their applicability in the Arctic is likely but needs more investigation.

*Recommendations*. The avoidance tests need to be validated with Arctic species under Arctic conditions prior to implementation.

I suggest the use of immature adult *G. setosus* collected during early spring for the avoidance tests (i.e., March-April) [373]. During this period, *G. setosus* synchronizes its activity with the







spring bloom [374,375] and water parameters are stable [62]. Immature adults will limit its energy allocation to acclimation and growth, and start building up energy reserves.

Finally, several aspects of the test conditions can be further improved. For instance, sediment oxygen demand may be used to evaluate the physiological state of individuals during tests, and the applicability of these tests at low PAH exposure level [104,259]. Similarly, the homogeneity of sediment characteristics and feeding can be considered for an optimised test.

O6: What are prospects and advantages of using Arctic benthic bioindicator species for monitoring long-term impacts of O&G and other maritime activities in the Arctic? Here the prospects and advantages of using Arctic benthic bioindicator species are discussed based on the criteria provided in **Chapter 1**:

- 1. Its response is sensitive to changes in the chemical concentration;
- 2. Its response is specific and causally related to the chemical exposure;
- 3. Its response is adequate given the anticipated concentration of the chemical;
- 4. The species' uptake/elimination of the chemical can be quantified;
- 5. The toxic effects of the chemical can be measured/interpreted.

These criteria, based on Goodsell et al. [19], do not necessarily need to apply to every single aspect measured on a bioindicator species. For instance, measuring a biomarker of exposure can be meaningful in the context of biomonitoring, even without measuring the response of a biomarker of effect (criterion 3).

As for the first two criteria, the correlation between bioaccumulation of PAHs and PAH exposure levels was tested for species with different traits under field conditions (Chapter 2). It was suggested that exposure was lower in the presence of black carbon, thus implying a positive correlation between exposure and uptake. However, the causality of bioaccumulation changes could also be explained by the species' feeding traits as was shown under experimental conditions (Chapter 3). In that respect, M. calcarea can be used to detect spatial and temporal gradients in bioaccumulation in the Arctic due to its selective feeding mode (Q1).

As for the third criterion, confounding non-target chemicals or environmental conditions have to be taken into account. The causality of the interrelationships between biomarkers at different suborganismal levels needs to be established for the bioindicator candidates (Figure 3; [376]). This is needed for the validation of the bioindicator species as these relationships will be used to select appropriate biomarkers which can provide early warning/are indicative







of PAH biological effects. In my research, a direct causality between the concentration of pyrene body residue and the rate of pyrene biotransformation was not found in *M. calcarea* (Chapter 4).

In addition, for the actual validation of *M. calcarea* as a biomonitoring species, future studies should address the combined effects of relevant oil chemical components and environmental factors on biomarker responses [361]. Although a long-term strategy is essential for such studies [377], the use of Arctic region specific environmental quality standards in an integrated monitoring and assessment programme is essential for the understanding of the dynamics of toxicants and their effects on Arctic marine ecosystems [345]. This implies that much work is still needed (e.g., see **Q4**) and that the present study is only a first start for what is needed to fully develop a validated system for biomonitoring in the Arctic.

The adaptation of existing methodology to Arctic species and conditions may form a suitable base for the environmental risk assessment of O&G and shipping activities in the Arctic, and the use of Arctic benthic bioindicator species is feasible and highly necessary for the prediction of long-term impacts of such activities on Arctic benthic habitats.

#### Conclusion

For the selection of relevant bioindicators for O&G impact assessment in the Arctic, sediment tests can be used in combination with chemical data and biomarkers measured in the organism collected *in situ* as part of an integrated weight-of-evidence approach. Bioaccumulation testing requires the implementation of more standardized methods for sediment preparation, exposure conditions and acclimatization of field organisms (Q3). A better understanding of the natural variability in biomarker responses in Arctic species is also needed in combination with the influence of environmental confounding factors (Q4).

With a few exceptions, it is not common to find all properties of a good bioindicator in one single species. Thus, different species may be used in the evaluation of different sublethal endpoints. This also accounts for the selection of biomarkers. A single biomarker response is generally not adequate, but a suite of biomarkers can be used to indicate early effects of exposure (as in this thesis) and/or effects of a toxicant. Biomarkers present early warning information that biodiversity indicators, such as biotic indexes, do not offer. Biomarkers of exposure are closely related to chemical stress, while biodiversity indicators are not specific







and may indicate all kinds of different stressors. Hence, biomarkers may enable mitigation measures before irreversible impacts occur.

In this thesis, *M. calcarea* was selected as a potential bioindicator species for O&G impact assessment in the Arctic (Q2). In short, *M. calcarea* was able to accumulate PAHs at a low exposure level under field conditions, although the species showed a variable level of pyrene metabolization under laboratory conditions. This was mostly related to experimental conditions, but potentially also to the likely variability in the physiological state of *M. calcarea* individuals. Its capacity for the excretion of PAH metabolites was low. Although the actual metabolization and excretion of PAHs still need to be further tested, the retention of PAH metabolites at a low exposure level in *M. calcarea* makes it worthwhile to study the implementation of several biomarkers with temporal differences in their response and differences in their effective dose.

A bioindicator like *M. calcarea* could be incorporated into a signalling system. Signalling systems can be seen as a weight-of-evidence approach, being an integrative index based on biomarkers that allow us to classify sampling sites with respect to their pollution status. For instance, the Integrated Biomarker Response (IBR) [378] is a mathematical tool based on biomarker data standardisation and their aggregation into a single value. Other approaches like the Integrated assessment of hazardous substances in the Baltic Sea (CHASE tool) combine chemical and biological measurements. Such approaches may reduce the uncertainty in interpretation of biomarker responses and can be used for regular and impact-specific biomonitoring.

A crucial part of such a signalling system is a solid knowledge on the baseline levels. Such baseline studies need to characterize natural levels in PAH body residues and biomarker responses commonly recommended for biomonitoring and to evaluate the sensitivity to oil pollution in these Arctic species. Furthermore, baseline studies help to identify the strengths and limitations for application of biomarkers to (sub-)Arctic regions as well as to determine their seasonal variability in potential bioindicator species. Seasonality may result in particular physiological characteristics of Arctic species that affect their toxic response to oil and PAHs (e.g., seasonality in lipid reserves). In addition, environmental fluctuations in the light regime and temperature may affect the bioavailability and exposure to oil-related chemicals. Baseline studies using bioindicator species will help to better characterize the general mechanisms and patterns underlying negative effects of O&G and shipping activities.







The overall utility of this work consists of a first evaluation of promising candidate species for Arctic biomonitoring and research programmes, and of various alternative approaches for the early identification of negative ecological impacts of O&G and shipping in Arctic benthic bioindicators.

Particularly, this thesis proposed one potential benthic bioindicator species (*M. calcarea*) and one potential tool (avoidance tests) for the assessment of oil impacts on sediment for its further validation under Arctic conditions. It also underlined the need for standard sediment test protocols and baseline studies as an integral part of environmental risk assessment in the Arctic, and provided several recommendations based on existing guidelines and a comparative study of bioaccumulation data from field and laboratory studies with Arctic species.







- 1. Larsen JN, Anisimov OA, Constable A, Hollowed AB, Maynard N, Prestrud P, et al. Polar regions. Climate change 2014: Impacts, adaptation, and vulnerability Part B: Regional aspects contribution of working group II to the fifth assessment report of the Intergovernmental Panel on Climate Change. Cambridge (United Kingdom): Cambridge University Press. 2014; p. 1567-612.
- 2. Grebmeier JM. Shifting patterns of life in the Pacific Arctic and sub-Arctic seas. Annual Review of Marine Science. 2012;4:63-78.
- 3. Krause-Jensen D, Duarte CM. Expansion of vegetated coastal ecosystems in the future Arctic. Frontiers in Marine Science. 2014 December 17, 2014;1(77).
- 4. ICES. Final Report of the Working Group on the Integrated Assessments of the Barents Sea (WGIBAR). Murmansk (Russia): International Council for the Exploration of the Sea. 2016 February 22-26, 2016.
- 5. Lien T. The world needs a stable oil market. Norwegian Ministry of Petroleum and Energy. 2016.
- 6. Gelcich S, Buckley P, Pinnegar JK, Chilvers J, Lorenzoni I, Terry G, et al. Public awareness, concerns, and priorities about anthropogenic impacts on marine environments. Proceedings of the National Academy of Sciences. 2014 October 21, 2014;111(42):15042-7.
- 7. Gulas S, Downton M, D'Souza K, Hayden K, Walker TR. Declining Arctic Ocean oil and gas developments: Opportunities to improve governance and environmental pollution control. Marine Policy. 2017;75:53-61.
- 8. Peterson CH, Rice SD, Short JW, Esler D, Bodkin JL, Ballachey BE, et al. Long-term ecosystem response to the Exxon Valdez oil spill. Science. 2003;302(5653):2082-6.
- 9. Atlas RM, Hazen TC. Oil biodegradation and bioremediation: A tale of the two worst spills in U.S. history. Environmental Science & Technology. 2011;45(16):6709-15.
- 10. Schmidt CW. Arctic oil drilling plans raise environmental health concerns. Environmental Health Perspectives. 2011;119(3):A116-A7.
- 11. Comer B, Olmer N, Mao X. Heavy fuel oil use in Arctic shipping in 2015. The International Council on Clean Transportation. 2016.
- 12. Sørstrøm SE, Brandvik PJ, Buist I, Daling P, Dickins D, Faksness L-G, et al. Joint industry program on oil spill contingency for Arctic and ice-covered waters. 2010; 32 Contract No.: 978-82-14-04759-2.
- 13. Thorsell DE, Leschine TM. An evaluation of oil pollution prevention strategies in the Arctic: A comparison of Canadian and U.S. approaches. Marine Policy. 2016;72:255-62.
- 14. AMAP. Assessment 2007: Oil and Gas Activities in the Arctic Effects and Potential Effects. Arctic Monitoring and Assessment Programme. 2010.
- 15. Kristoffersen B, O'Brien K, Selboe E. Opportunistic adaptation. The adaptive challenge of climate change. Cambridge (United Kingdom): Cambridge University Press. 2015; p. 118-39.
- 16. Nevalainen M, Helle I, Vanhatalo J. Preparing for the unprecedented Towards quantitative oil risk assessment in the Arctic marine areas. Marine Pollution Bulletin. 2017;114(1):90-101.
- 17. Hauge KH, Blanchard A, Andersen G, Boland R, Grøsvik BE, Howell D, et al. Inadequate risk assessments A study on worst-case scenarios related to petroleum exploitation in the Lofoten area. Marine Policy. 2014;44(Supplement C):82-9.







- 18. Blanchard A, Hauge KH, Andersen G, Fosså JH, Grøsvik BE, Handegard NO, et al. Harmful routines? Uncertainty in science and conflicting views on routine petroleum operations in Norway. Marine Policy. 2014:43:313-20.
- Goodsell PJ, Underwood AJ, Chapman MG. Evidence necessary for taxa to be reliable indicators of environmental conditions or impacts. Marine Pollution Bulletin. 2009;58(3):323-31.
- IOGP. Environmental fates and effects of ocean discharge of drill cuttings and associated drilling fluids from offshore oil and gas operations. International Association of Oil and Gas Producers. 2016; Report No. 543.
- U.S.E.P.A. Final Ocean Discharge Criteria Evaluation of the Arctic NPDES General Permit for Oil and Gas Exploration (Permit No. AKG280000). Seattle, WA: U.S. Environmental Protection Agency, Region 10. 2006.
- 22 Neff JM, Lee K, DeBlois EM. Produced Water: Overview of Composition, Fates, and Effects. In: Lee K. Neff JM, editors, Produced Water: Environmental Risks and Advances in Mitigation Technologies, 1 ed: Springer-Verlag New York. 2011; p. 3-54.
- Bakke T, Klungsøyr J, Sanni S. Environmental impacts of produced water and drilling waste discharges from the Norwegian offshore petroleum industry. Marine Environmental Research. 2013;92:154-69.
- 24. Neff JM. Bioaccumulation in Marine Organisms: Effects of Contaminants from Oil Well Produced Water. Amsterdam (The Netherlands): Elsevier B.V. 2002.
- NRPA. Natural Radioactivity in Produced Water from the Norwegian Oil and Gas Industry in 2003. Østerås (Norway): Norwegian Radiation Protection Authority. 2004; Strålevern Rapport 2005:2.
- Zheng J, Chen B, Thanyamanta W, Hawboldt K, Zhang B, Liu B. Offshore produced water 26. management: A review of current practice and challenges in harsh/Arctic environments. Marine Pollution Bulletin. 2016;104(1-2):7-19.
- Holth TF, Beckius J, Zorita I, Cajaraville MP, Hylland K. Assessment of lysosomal membrane stability 27. and peroxisome proliferation in the head kidney of Atlantic cod (Gadus morhua) following long-term exposure to produced water components. Marine Environmental Research. 2011;72(3):127-34.
- 28. Geraudie P, Nahrgang J, Forget-Leray J, Minier C, Camus L. In vivo effects of environmental concentrations of produced water on the reproductive function of polar cod (Boreogadus saida). Journal of Toxicology and Environmental Health - Part A: Current Issues. 2014;77(9-11):557-73.
- Jensen LK, Halvorsen E, Song Y, Hallanger IG, Hansen EL, Brooks SJ, et al. Individual and molecular level effects of produced water contaminants on nauplii and adult females of Calanus finmarchicus. Journal of Toxicology and Environmental Health, Part A. 2016;79(13-15):585-601.
- 30. Wassmann P, Reigstad M, Haug T, Rudels B, Carroll ML, Hop H, et al. Food webs and carbon flux in the Barents Sea. Progress in Oceanography. 2006;71(2-4):232-87.
- Montagna PA, Baguley JG, Cooksey C, Hyland JL. Persistent impacts to the deep soft-bottom benthos one year after the Deepwater Horizon event. Integrated Environmental Assessment and Management. 2016.
- Harman C, Thomas KV, Tollefsen KE, Meier S, Bøyum O, Grung M. Monitoring the freely dissolved concentrations of polycyclic aromatic hydrocarbons (PAH) and alkylphenols (AP) around a Norwegian oil platform by holistic passive sampling. Marine Pollution Bulletin. 2009;58(11):1671-9.
- Beyer J, Myhre LP, Sundt RC, Meier S, Tollefsen K-E, Vabø R, et al. Environmental risk assessment of alkylphenols from offshore produced water on fish reproduction. Marine Environmental Research. 2012;75:2-9.
- 34. Wassmann P, Reigstad M. Future Arctic Ocean seasonal ice zones and implications for pelagic-benthic coupling. Oceanography. 2011;24(3):220-31.







- 35. Søreide JE, Carroll ML, Hop H, Ambrose WG, Hegseth EN, Falk-Petersen S. Sympagic-pelagic-benthic coupling in Arctic and Atlantic waters around Svalbard revealed by stable isotopic and fatty acid tracers. Marine Biology Research. 2013;9(9):831-50.
- 36. Kedra M, Moritz C, Choy ES, David C, Degen R, Duerksen S, et al. Status and trends in the structure of Arctic benthic food webs. Polar Research. 2015;34(23775).
- 37. Frey KE, Comiso JC, Cooper LW, Gradinger RR, Grebmeier JM, Tremblay J-É. Arctic Ocean Primary Productivity. Arctic Essays [Internet]. 2016. Available from: <a href="http://www.arctic.noaa.gov/Report-Card/Report-Card-2016/ArtMID/5022/ArticleID/284/Arctic-Ocean-Primary-Productivity">http://www.arctic.noaa.gov/Report-Card/Report-Card-2016/ArtMID/5022/ArticleID/284/Arctic-Ocean-Primary-Productivity</a>.
- 38. Pörtner HO, Peck L, Somero G. Thermal limits and adaptation in marine Antarctic ectotherms: an integrative view. Philosophical Transactions of the Royal Society B: Biological Sciences. 2007;362(1488):2233-58.
- 39. Richard J, Morley SA, Deloffre J, Peck LS. Thermal acclimation capacity for four Arctic marine benthic species. Journal of Experimental Marine Biology and Ecology. 2012;424–425:38-43.
- 40. Jørgensen LL, Ljubin P, Skjoldal HR, Ingvaldsen RB, Anisimova N, Manushin I. Distribution of benthic megafauna in the Barents Sea: baseline for an ecosystem approach to management. 2014 July 2, 2014.
- 41. CAFF. Arctic Marine Biodiversity Monitoring Plan Annual Report 2016: Annual report on the implementation of the CBMP's Arctic Marine Biodiversity Monitoring Plan. 2016.
- 42. Olsen E, Holen S, Hoel AH, Buhl-Mortensen L, Røttingen I. How integrated ocean governance in the Barents Sea was created by a drive for increased oil production. Marine Policy. 2016;71:293-300.
- 43. Jonker MTO, Brils JM, Sinke AJC, Murk AJ, Koelmans AA. Weathering and toxicity of marine sediments contaminated with oils and polycyclic aromatic hydrocarbons. Environmental Toxicology and Chemistry. 2006;25(5):1345-53.
- 44. Ma Y, Xie Z, Yang H, Möller A, Halsall C, Cai M, et al. Deposition of polycyclic aromatic hydrocarbons in the North Pacific and the Arctic. Journal of Geophysical Research: Atmospheres. 2013;118(11):5822-9.
- 45. Yunker MB, Macdonald RW. Composition and origins of polycyclic aromatic hydrocarbons in the Mackenzie river and on the Beaufort Sea shelf. Arctic. 1995;48(2):118-29.
- 46. Jonker MTO, Sinke AJC, Brils JM, Koelmans AA. Sorption of polycyclic aromatic hydrocarbons to oil contaminated sediment: Unresolved complex? Environmental Science & Technology. 2003;37(22):5197-203.
- 47. NRC. Oil in the Sea III: Inputs, Fates, and Effects. Washington, DC: National Research Council. 2003; 280 p.
- 48. Moermond CTA, Roessink I, Jonker MTO, Meijer T, Koelmans AA. Impact of polychlorinated biphenyl and polycyclic aromatic hydrocarbon sequestration in sediment on bioaccumulation in aquatic food webs. Environmental Toxicology and Chemistry. 2007;26(4):607-15.
- 49. Payne JR, McNabb GD, Clayton JR. Oil-weathering behavior in Arctic environments. Polar Research. 1991;10(2):631-62.
- 50. Brandvik PJ, Faksness L-G. Weathering processes in Arctic oil spills: Meso-scale experiments with different ice conditions. Cold Regions Science and Technology. 2009;55(1):160-6.
- 51. Brown KE, King CK, Kotzakoulakis K, George SC, Harrison PL. Assessing fuel spill risks in polar waters: Temporal dynamics and behaviour of hydrocarbons from Antarctic diesel, marine gas oil and residual fuel oil. Marine Pollution Bulletin. 2016;110(1):343-53.







- 52. Crisafi F, Giuliano L, Yakimov MM, Azzaro M, Denaro R. Isolation and degradation potential of a cold-adapted oil/PAH-degrading marine bacterial consortium from Kongsfjorden (Arctic region). Rendiconti Lincei. 2016;27(1):261-70.
- 53. Koelmans AA, Poot A, Lange HJD, Velzeboer I, Harmsen J, Noort PCMv. Estimation of in situ sediment-to-water fluxes of polycyclic aromatic hydrocarbons, polychlorobiphenyls and polybrominated diphenylethers. Environmental Science & Technology. 2010;44(8):3014-20.
- 54. Rakowska MI, Smit MPJ, Kupryianchyk D, Qin J, Koelmans AA, Rijnaarts HHM, et al. Turbulent mixing accelerates PAH desorption due to fragmentation of sediment particle aggregates. Journal of Soils and Sediments. 2017;17(1):277-85.
- 55. Jonker MTO, Koelmans AA. Sorption of polycyclic aromatic hydrocarbons and polychlorinated biphenyls to soot and soot-like materials in the aqueous environment: Mechanistic considerations. Environmental Science & Technology. 2002;36(17):3725-34.
- 56. Diepens NJ, Van den Heuvel-Greve MJ, Koelmans AA. Modeling of bioaccumulation in marine benthic invertebrates using a multispecies experimental approach. Environmental Science & Technology. 2015;49(22):13575-85.
- 57. Neff JM. Chapter 15 Polycyclic Aromatic Hydrocarbons in the Ocean. Bioaccumulation in Marine Organisms: Effect of Contaminants from Oil Well Produced Water. Oxford: Elsevier. 2002; p. 241-318.
- 58. Varanasi U. Metabolism of Polycyclic Aromatic Hydrocarbons in the Aquatic Environment: CRC Press. 1989; 352 p.
- 59. Benedetti M, Martuccio G, Fattorini D, Canapa A, Barucca M, Nigro M, et al. Oxidative and modulatory effects of trace metals on metabolism of polycyclic aromatic hydrocarbons in the Antarctic fish Trematomus bernacchii. Aquatic Toxicology. 2007;85(3):167-75.
- 60. Yang Z, Feng J, Niu J, Shen Z. Release of polycyclic aromatic hydrocarbons from Yangtze river sediment cores during periods of simulated resuspension. Environmental Pollution. 2008;155(2):366-74.
- 61. Oh S, Wang Q, Shin WS, Song D-I. Sorption and desorption kinetics of PAHs in coastal sediment. Korean Journal of Chemical Engineering. 2013;30(1):145-53.
- 62. Beuchel F, Gulliksen B. Temporal patterns of benthic community development in an Arctic fjord (Kongsfjorden, Svalbard): results of a 24-year manipulation study. Polar Biology. 2008;31(8):913-24.
- 63. MacArthur RH, Wilson EO. The Theory of Island Biogeography: Princeton University Press. 2001.
- 64. Moe SJ, De Schamphelaere K, Clements WH, Sorensen MT, Van den Brink PJ, Liess M. Combined and interactive effects of global climate change and toxicants on populations and communities. Environmental Toxicology and Chemistry / Setac. 2013;32(1):49-61.
- 65. Neff JM. Fates and Effects of Water Based Drilling Muds and Cuttings in Cold-Water Environments. Duxbury, MA: Neff & Associates LLC. 2010.
- OSPAR. OSPAR List of Substances Used and Discharged Offshore which Are Considered to Pose Little or No Risk to the Environment (PLONOR). OSPAR Agreement 2013-06 ed2004.
- 67. OSPAR Decision 2000/2 on a Harmonised Mandatory Control System for the Use and Reduction of the Discharge of Offshore Chemicals (as amended by OSPAR Decision 2005/1). 2000.
- 68. NME. Integrated Management of the Marine Environment of the Barents Sea and the Sea Areas off the Lofoten Islands. Oslo (Norway): Norwegian Ministry of the Environment. 2006.
- 69. Renaud PE, Jensen T, Wassbotten I, Mannvik HP, Botnen H. Offshore sediment monitoring on the Norwegian shelf: A regional approach 1996-2006. Tromsø (Norway): Akvaplan-niva AS. 2008.









- 70. De Hoop L, Schipper AM, Leuven RSEW, Huijbregts MAJ, Olsen GH, Smit MGD, et al. Sensitivity of polar and temperate marine organisms to oil components. Environmental Science and Technology. 2011;45(20):9017-23.
- 71. Gardiner WW, Word JQ, Word JD, Perkins RA, McFarlin KM, Hester BW, et al. The acute toxicity of chemically and physically dispersed crude oil to key arctic species under arctic conditions during the open water season. Environmental Toxicology and Chemistry. 2013;32(10):2284-300.
- 72. Camus L, Brooks S, Geraudie P, Hjorth M, Nahrgang J, Olsen GH, et al. Comparison of produced water toxicity to Arctic and temperate species. Ecotoxicology and Environmental Safety. 2015;113:248-58.
- 73. Chapman PM. Toxicity delayed in cold freshwaters? Journal of Great Lakes Research. 2016;42(2):286-9.
- 74. Martínez-Gómez C, Vethaak AD, Hylland K, Burgeot T, Köhler A, Lyons BP, et al. A guide to toxicity assessment and monitoring effects at lower levels of biological organization following marine oil spills in European waters. ICES Journal of Marine Science: Journal du Conseil. 2010 September 1, 2010;67(6):1105-18.
- 75. Di Toro DM, Zarba CS, Hansen DJ, Berry WJ, Swartz RC, Cowan CE, et al. Technical basis for establishing sediment quality criteria for nonionic organic chemicals using equilibrium partitioning. Environmental Toxicology and Chemistry. 1991;10(12):1541-83.
- 76. Morrison HA, Gobas FAPC, Lazar R, Haffner GD. Development and verification of a bioaccumulation model for organic contaminants in benthic invertebrates. Environmental Science & Technology. 1996;30(11):3377-84.
- 77. Kaag NHBM, Foekema EM, Scholten MCT, van Straalen NM. Comparison of contaminant accumulation in three species of marine invertebrates with different feeding habits. Environmental Toxicology and Chemistry. 1997;16(5):837-42.
- 78. Poot A, Jonker MTO, Gillissen F, Koelmans AA. Explaining PAH desorption from sediments using Rock Eval analysis. Environmental Pollution. 2014;193:247-53.
- 79. Koelmans AA, Jonker MTO, Cornelissen G, Bucheli TD, Van Noort PCM, Gustafsson Ö. Black carbon: The reverse of its dark side. Chemosphere. 2006;63(3):365-77.
- 80. Selck H, Drouillard K, Eisenreich K, Koelmans AA, Palmqvist A, Ruus A, et al. Explaining differences between bioaccumulation measurements in laboratory and field data through use of a probabilistic modeling approach. Integrared Environmental Assessment and Management. 2012;8(1):42-63.
- 81. Burkhard LP, Cook PM, Lukasewycz MT. Comparison of biota-sediment accumulation factors across ecosystems. Environmental Science & Technology. 2005;39(15):5716-21.
- 82. Weisbrod AV, Woodburn KB, Koelmans AA, Parkerton TF, McElroy AE, Borgå K. Evaluation of bioaccumulation using in vivo laboratory and field studies. Integrated Environmental Assessment and Management. 2009;5(4):598-623.
- 83. Sidney LA, Diepens NJ, Guo X, Koelmans AA. Trait-based modelling of bioaccumulation by freshwater benthic invertebrates. Aquatic Toxicology. 2016;176:88-96.
- 84. Kuzyk ZZA. Bioaccumulation of PCBs from contaminated sediments in a coastal marine ecosystem of Northern Labrador. Kingston, Ontario (Canada): Queen's University. 2000.
- 85. Warner NA, Evenset A, Christensen G, Gabrielsen GW, Borgå K, Leknes H. Volatile siloxanes in the European Arctic: Assessment of sources and spatial distribution. Environmental Science & Technology. 2010;44(19):7705-10.
- 86. Hagger JA, Jones MB, Leonard DRP, Owen R, Galloway TS. Biomarkers and integrated environmental risk assessment: Are there more questions than answers? Integrated Environmental Assessment and Management. 2006;2(4):312-29.







- 87. Regoli F, Giuliani ME. Oxidative pathways of chemical toxicity and oxidative stress biomarkers in marine organisms. Marine Environmental Research. 2014;93:106-17.
- 88. Regoli F, Winston GW, Gorbi S, Frenzilli G, Nigro M, Corsi I, et al. Integrating enzymatic responses to organic chemical exposure with total oxyradical absorbing capacity and DNA damage in the european eel Anguilla anguilla. Environmental Toxicology and Chemistry. 2003;22(9):2120-9.
- 89. Nahrgang J, Camus L, Carls MG, Gonzalez P, Jönsson M, Taban IC, et al. Biomarker responses in polar cod (Boreogadus saida) exposed to the water soluble fraction of crude oil. Aquatic Toxicology. 2010;97(3):234-42.
- 90. Thain JE, Vethaak AD, Hylland K. Contaminants in marine ecosystems: developing an integrated indicator framework using biological-effect techniques. ICES Journal of Marine Science: Journal du Conseil. 2008 November 1, 2008;65(8):1508-14.
- 91. Davies IM, Vethaak AD. Integrated marine environmental monitoring of chemicals and their effects. ICES Cooperative Research Report. Copenhagen (Denmark): International Council for the Exploration of the Sea. 2012; Report No. 315.
- 92. Abele D, Puntarulo S. Formation of reactive species and induction of antioxidant defence systems in polar and temperate marine invertebrates and fish. Comparative Biochemistry and Physiology Part A: Molecular & Integrative Physiology. 2004;138(4):405-15.
- 93. Livingstone DR. Contaminant-stimulated reactive oxygen species production and oxidative damage in aquatic organisms. Marine Pollution Bulletin. 2001;42(8):656-66.
- 94. Hylland K. Polycyclic aromatic hydrocarbon (PAH) ecotoxicology in marine ecosystems. Journal of Toxicology and Environmental Health, Part A. 2006;69(1-2):109-23.
- 95. Hampel M, Blasco J, Martín Díaz ML. Biomarkers and Effects. Marine Ecotoxicology: Current Knowledge and Future Issues: Elsevier. 2016; p. 45.
- 96. Araújo CVM, Moreira-Santos M, Ribeiro R. Active and passive spatial avoidance by aquatic organisms from environmental stressors: A complementary perspective and a critical review. Environment International. 2016;92–93:405-15.
- 97. Basu N. Applications and implications of neurochemical biomarkers in environmental toxicology. Environmental Toxicology and Chemistry. 2015;34(1):22-9.
- 98. Kravitz MJ, Lamberson JO, Ferraro SP, Swartz RC, Boese BL, Specht DT. Avoidance response of the estuarine amphipod Eohaustorius estuarius to polycyclic aromatic hydrocarbon-contaminated, field-collected sediments. Environmental Toxicology and Chemistry. 1999;18(6):1232-5.
- 99. Hellou J, Cheeseman K, Jouvenelle M-L, Robertson S. Behavioral response of Corophium volutator relative to experimental conditions, physical and chemical disturbances. Environmental Toxicology and Chemistry. 2005;24(12):3061-8.
- 100. De Lange HJ, Sperber V, Peeters ETHM. Avoidance of polycyclic aromatic hydrocarbon–contaminated sediments by the freshwater invertebrates Gammarus pulex and Asellus aquaticus. Environmental Toxicology and Chemistry. 2006;25(2):452-7.
- 101. Ward DJ, Simpson SL, Jolley DF. Avoidance of contaminated sediments by an amphipod (Melita plumulosa), a harpacticoid copepod (Nitocra spinipes), and a snail (Phallomedusa solida). Environmental Toxicology and Chemistry. 2013;32(3):644-52.
- 102. Hatlen K, Camus L, Berge J, Olsen GH, Baussant T. Biological effects of water soluble fraction of crude oil on the Arctic sea ice amphipod Gammarus wilkitzkii. Chemistry and Ecology. 2009;25(3):151-62.







- 103. Sharma P, Schiewer S. Assessment of crude oil biodegradation in arctic seashore sediments: effects of temperature, salinity, and crude oil concentration. Environmental Science and Pollution Research. 2016;23(15):14881-8.
- 104. Olsen GH, Carroll ML, Renaud PE, Ambrose WG, Olssøn R, Carroll J. Benthic community response to petroleum-associated components in Arctic versus temperate marine sediments. Marine Biology. 2007;151(6):2167-76.
- 105. Brown KE, King CK, Harrison PL. Lethal and behavioral impacts of diesel and fuel oil on the Antarctic amphipod Paramoera walkeri. Environmental Toxicology and Chemistry. 2017.
- 106. Security NMoJaP. Svalbard: Report No. 22 (2008–2009) to the Storting. Oslo (Norway): Ministry of Justice and Public Security. 2009.
- 107. Woolfenden ENM, Hince G, Powell SM, Stark SC, Snape I, Stark JS, et al. The rate of removal and the compositional changes of diesel in Antarctic marine sediment. Science of the Total Environment. 2011;410–411:205-16.
- 108. Gesteira JLG, Dauvin JC. Amphipods are good bioindicators of the impact of oil spills on soft-bottom macrobenthic communities. Marine Pollution Bulletin. 2000;40(11):1017-27.
- 109. Washburn T, Rhodes ACE, Montagna PA. Benthic taxa as potential indicators of a deep-sea oil spill. Ecological Indicators. 2016;71:587-97.
- 110. Lane A, Riddle MJ. Toxicity of dispersed and undispersed diesel in sediments to the Antarctic ophiuroid Ophiura crassa. Australasian Journal of Ecotoxicology. 2004;10(2):115-21.
- 111. Vázquez M. Bioaccumulation and effects of parent and alkylated PAHs in an Arctic marine amphipod (Gammarus setosus) originating from pristine and historically contaminated sites in Svalbard: implications for sensitivity and adaptation. Tromsø (Norway): University of Tromsø. 2016.
- 112. Thorson G. The larval development, growth and metabolism of arctic marine bottom invertebrates compared with those of other seas. Copenhague: C.A. Reitzel. 1936.
- 113. Sirenko BI, Gagaev SY. Unusual abundance of macrobenthos and biological invasions in the Chukchi Sea. Russian Journal of Marine Biology. 2007;33(6):355-64.
- 114. Blanchard AL, Parris CL, Knowlton AL, Wade NR. Benthic ecology of the northeastern Chukchi Sea. Part I. Environmental characteristics and macrofaunal community structure, 2008–2010. Continental Shelf Research. 2013;67:52-66.
- 115. Grebmeier JM, Bluhm BA, Cooper LW, Danielson SL, Arrigo KR, Blanchard AL, et al. Ecosystem characteristics and processes facilitating persistent macrobenthic biomass hotspots and associated benthivory in the Pacific Arctic. Progress in Oceanography. 2015;136:92-114.
- 116. C. Emmerson GL. Arctic opening: opportunity and risk in the High North. London (United Kingdom): Chatham House. 2012.
- 117. Miller GH, Lehman SJ, Refsnider KA, Southon JR, Zhong YCGL. Unprecedented recent summer warmth in Arctic Canada. Geophysical Research Letters. 2013;40(21):5745-51.
- 118. Boitsov S, Jensen HKB, Klungsøyr J. Natural background and anthropogenic inputs of polycyclic aromatic hydrocarbons (PAH) in sediments of south-western Barents Sea. Marine Environmental Research. 2009;68(5):236-45.
- 119. Dahle S, Savinov VM, Matishov GG, Evenset A, Næs K. Polycyclic aromatic hydrocarbons (PAHs) in bottom sediments of the Kara Sea shelf, Gulf of Ob and Yenisei Bay. Science of the Total Environment. 2003;306(1-3):57-71.







- 120. Yunker MB, Macdonald RW, Vingarzan R, Mitchell RH, Goyette D, Sylvestre S. PAHs in the Fraser River basin: a critical appraisal of PAH ratios as indicators of PAH source and composition. Organic Geochemistry. 2002;33(4):489-515.
- 121. De Laender F, Hammer J, Hendriks AJ, Soetaert K, Janssen CR. Combining monitoring data and modeling identifies PAHs as emerging contaminants in the Arctic. Environmental Science & Technology. 2011;45(20):9024-9.
- 122. Camus L, Birkely SR, Jones MB, Børseth JF, Grøsvik BE, Gulliksen B, et al. Biomarker responses and PAH uptake in Mya truncata following exposure to oil-contaminated sediment in an Arctic fjord (Svalbard). Science of the Total Environment. 2003;308(1–3):221-34.
- 123. Hung H, Kallenborn R, Breivik K, Su Y, Brorström-Lundén E, Olafsdottir K, et al. Atmospheric monitoring of organic pollutants in the Arctic under the Arctic Monitoring and Assessment Programme (AMAP): 1993–2006. Science of the Total Environment. 2010;408(15):2854-73.
- 124. Ma J, Hung H, Tian C, Kallenborn R. Revolatilization of persistent organic pollutants in the Arctic induced by climate change. Nature Climate Change. 2011;1(5):255-60.
- 125. Chapman PM, Riddle MJ. Toxic effects of contaminants in polar marine environments. Environmental Science & Technology. 2005;39(9):200A-6A.
- 126. Hallanger IG, Ruus A, Herzke D, Warner NA, Evenset A, Heimstad ES, et al. Influence of season, location, and feeding strategy on bioaccumulation of halogenated organic contaminants in Arctic marine zooplankton. Environmental Toxicology and Chemistry. 2011;30(1):77-87.
- 127. Renaud P, Morata N, Carroll ML, Denisenko SG, Reigstad M. Pelagic-benthic coupling in the western Barents Sea: Processes and time scales. Deep-Sea Research Part II: Topical Studies in Oceanography. 2008;55(20-21):2372-80.
- 128. van den Brink NW, Riddle MJ, van den Heuvel-Greve M, van Francker JA. Contrasting time trends of organic contaminants in Antarctic pelagic and benthic food webs. Marine Pollution Bulletin. 2011;62(1):128-32.
- 129. Dahle S, Savinov, V., Petrova, V., Klungsøyr, J., Savinova, T., Batova, G. & Kursheva, A. Polycyclic aromatic hydrocarbons (PAHs) in Norwegian and Russian Arctic marine sediments: concentrations, geographical distribution and sources. Norwegian Journal of Geology. 2006;86(1):41-50.
- 130. Kędra M, Włodarska-Kowalczuk M, Wesławski JM. Decadal change in macrobenthic soft-bottom community structure in a high Arctic fjord (Kongsfjorden, Svalbard). Polar Biology. 2010;33(1):1-11.
- 131. Hop H, Pearson T, Hegseth EN, Kovacs KM, Wiencke C, Kwasniewski S, et al. The marine ecosystem of Kongsfjorden, Svalbard. Polar Research. 2002;21(1):167-208.
- 132. Olsen GH, Smit MGD, Carroll J, Jæger I, Smith T, Camus L. Arctic versus temperate comparison of risk assessment metrics for 2-methyl-naphthalene. Marine Environmental Research. 2011;72(4):179-87.
- 133. Markert B, Wappelhorst O, Weckert V, Herpin U, Siewers U, Friese K, et al. The use of bioindicators for monitoring the heavy-metal status of the environment. Journal of Radioanalytical and Nuclear Chemistry. 1999;240(2):425-9.
- 134. Renaud PE, Tessmann M, Evenset A, Christensen GN. Benthic food-web structure of an Arctic fjord (Kongsfjorden, Svalbard). Marine Biology Research. 2011;7(1):13-26.
- 135. Nygård H, Berge J, Søreide JE, Vihtakari M, Falk-Petersen S. The amphipod scavenging guild in two Arctic fjords: seasonal variations, abundance and trophic interactions. Aquatic Biology. 2012;14(3):247-64.
- 136. Kędra M, Kuliński K, Walkusz W, Legeżyńska J. The shallow benthic food web structure in the high Arctic does not follow seasonal changes in the surrounding environment. Estuarine, Coastal and Shelf Science. 2012;114:183-91.









- 137. Richman SE, Lovvorn JR. Effects of clam species dominance on nutrient and energy acquisition by spectacled eiders in the Bering Sea. Marine Ecology Progress Series. 2003 October 17, 2003;261:283-97.
- 138. Dunton KH. Chukchi Sea Offshore Monitoring in Drilling Area (COMIDA): Chemical and Benthos (CAB). Anchorage, AK: University of Texas Marine Science Institute. 2012; OCS Study BOEM 2012-012.
- 139. Rust AJ, Burgess RM, Brownawell BJ, McElroy AE. Relationship between metabolism and bioaccumulation of benzo[a]pyrene in benthic invertebrates. Environmental Toxicology and Chemistry. 2004;23(11):2587-93.
- 140. Tessmann M. Trophic transfer of persistent organic pollutants in the benthic food web of Kongsfjorden, Spitzbergen. Christian-Albrechts-Universität zu Kiel. 2008.
- 141. Evenset A, Hallanger IG, Tessmann M, Warner N, Ruus A, Borgå K, et al. Seasonal variation in accumulation of persistent organic pollutants in an Arctic marine benthic food web. Science of the Total Environment. 2016;542, Part A:108-20.
- 142. Chapman PM, Riddle MJ. Missing and needed: polar marine ecotoxicology. Marine Pollution Bulletin. 2003;46(8):927-8.
- 143. Lake JL, Rubinstein NI, Lee H, Lake CA, Heltshe J, Pavignano S. Equilibrium partitioning and bioaccumulation of sediment-associated contaminants by infaunal organisms. Environmental Toxicology and Chemistry. 1990;9(8):1095-106.
- 144. Word JQ, Gardiner W, Melbye A, Merlin F, Brakstad O, Camus L, et al. Environmental effects of spilled oil and response technologies in the Arctic: Literature review and recommendations. International Association of Oil and Gas Producers (OGP). Arctic Oil Spill Response Technology Joint Industry Programme (JIP). 2014.
- 145. Svendsen H, Beszczynska-Møller A, Hagen JO, Lefauconnier B, Tverberg V, Gerland S, et al. The physical environment of Kongsfjorden-Krossfjorden, and Arctic fjord system in Svalbard. Polar Research. 2002;21(1):133-66.
- 146. Kuliński K, Kędra M, Legeżyńska J, Gluchowska M, Zaborska A. Particulate organic matter sinks and sources in high Arctic fjord. Journal of Marine Systems. 2014;139:27-37.
- 147. Włodarska-Kowalczuk M, Pearson TH. Soft-bottom macrobenthic faunal associations and factors affecting species distributions in an Arctic glacial fjord (Kongsfjord, Spitsbergen). Polar Biology. 2004;27(3):155-67.
- 148. Sander G. Limits of acceptable change caused by local activities in Ny-Ålesund: Report from a Preproject, Containing a Proposal for a Main Project. Norwegian Polar Institute. 2014.
- 149. van den Heuvel-Greve MJ, Szczybelski AS, van den Brink NW, Kotterman MJJ, Kwadijk CJAF, Evenset A, et al. Low organotin contamination of harbour sediment in Svalbard. Polar Biology. 2016;39(10):1699-709.
- 150. Kupryianchyk D, Reichman EP, Rakowska MI, Peeters ETHM, Grotenhuis JTC, Koelmans AA. Ecotoxicological effects of activated carbon amendments on macroinvertebrates in nonpolluted and polluted sediments. Environmental Science & Technology. 2011;45(19):8567-74.
- 151. Tobiszewski M, Namieśnik J. PAH diagnostic ratios for the identification of pollution emission sources. Environmental Pollution. 2012;162:110-9.
- 152. Arp HP, Ruus A, Macken A, Lillicrap A. Quality assurance of environmental quality standards. Oslo (Norway): NIVA and NGI. 2014; Contract No.: M-241.
- 153. Webster L, Fryer R, Davies I, Roose P, Moffat C. Background Document on CEMP assessment criteria for the QSR 2010. London (United Kingdom): OSPAR Commission. 2009; 461/2009 Contract No.: 978-1-907390-08-1.







- 154. Hansen DJ, DiToro DM, McGrath JA, Swartz RC, Mount DR, Burgess RM, et al. Procedures for the derivation of equilibrium partitioning sediment benchmarks (ESBs) for the protection of benthic organisms: PAH mixtures. Washington, DC.: United States Environmental Protection Agency. 2003; 600R02013.
- Directive 2013/39/EU of the European Parliament and of the Council of 12 August 2013 amending Directives 2000/60/EC and 2008/105/EC as regards priority substances in the field of water policy Text with EEA relevance. 2013.
- Di Toro DM, McGrath JA, Hansen DJ. Technical basis for narcotic chemicals and polycyclic aromatic hydrocarbon criteria. I. Water and tissue. Environmental Toxicology and Chemistry. 2000;19(8):1951-70.
- Auen JE. Kings Bay-ulykken og borgerlig samarbeid. 2016 [cited 2016 31 July]; Available from: http://ndla.no/nb/node/161573.
- U.S.E.P.A. Mid-Atlantic Integrated Assessment (MAIA) 1997-98 Summary Report. 2002; Contract No.: EPA/620/R-02/003.
- Evenset A. Norwegian and Russian monitoring of PCB contamination at settlements in Svalbard: Comparison of field and analysis methods and results. Tromsø (Norway): Akvaplan-niva AS. 2009.
- Evenset A. Pollution in settlements in Svalbard. The need for follow-up surveys and measures. Tromsø (Norway): Akvaplan-niva AS. 2010.
- Holte B, Dahle S, Gulliksen B, Næs K, Some macrofaunal effects of local pollution and glacier-induced sedimentation, with indicative chemical analyses, in the sediments of two Arctic fjords. Polar Biology. 1996;16(8):549-57.
- 162. Zhang P, Ge L, Gao H, Yao T, Fang X, Zhou C, et al. Distribution and transfer pattern of polychlorinated biphenyls (PCBs) among the selected environmental media of Ny-Ålesund, the Arctic: As a case study. Marine Pollution Bulletin. 2014;89(1-2):267-75.
- Evenset A, Christensen GN, Palerud R. Miljøgifter i marine sedimenter, Isfjorden, Svalbard 2005. 163. Tromsø (Norway): Akvaplan-niva AS, 2006; Contract No.: APN-414.3341.
- Jörundsdóttir HÓ, Jensen S, Hylland K, Holth TF, Gunnlaugsdóttir H, Svavarsson J, et al. Pristine Arctic: Background mapping of PAHs, PAH metabolites and inorganic trace elements in the North-Atlantic Arctic and sub-Arctic coastal environment. Science of the Total Environment. 2014;493:719-28.
- Hwang H-M, Wade TL, Sericano JL, Residue-response relationship between PAH body burdens and lysosomal membrane destabilization in eastern oysters (Crassostrea virginica) and toxicokinetics of PAHs. Journal of Environmental Science and Health, Part A. 2008;43(12):1373-80.
- Gusev AA, Rudinskaya LV. Shell form, growth, and production of Astarte borealis (Schumacher, 1817) (Astartidae, Bivalvia) in the southeastern Baltic Sea. Oceanology. 2014;54(4):458-64.
- Sokołowski A, Wolowicz M, Hummel H. Free amino acids in the clam Macoma balthica L. (Bivalvia, Mollusca) from brackish waters of the southern Baltic Sea. Comparative Biochemistry and Physiology Part A: Molecular & Integrative Physiology. 2003;134(3):579-92.
- Liu D, Pan L, Li Z, Cai Y, Miao J. Metabolites analysis, metabolic enzyme activities and bioaccumulation in the clam Ruditapes philippinarum exposed to benzo[a]pyrene. Ecotoxicology and Environmental Safety. 2014;107:251-9.
- Livingstone DR. Organic Xenobiotic Metabolism in Marine Invertebrates. Advances in Comparative and Environmental Physiology. Berlin, Heidelberg: Springer Berlin Heidelberg. 1991; p. 45-185.
- 170. Widdows J, Bayne BL, Donkin P, Livingstone DR, Lowe DM, Moore MN, et al. Measurement of the responses of mussels to environmental stress and pollution in Sullom Voe: a base-line study. Proceedings of the Royal Society of Edinburgh Section B: Biological Sciences. 1981;80:323-38.









- 171. Singh N, Singh V. Spheroidal carbonaceous particles (SCPs): Indicator of anthropogenic pollutants in Kongsfjorden, Ny-Ålesund, Arctic. ARCTIC FRONTIERS; Tromsø (Norway). 2016.
- 172. Hop H, Sagerup K, Schlabach M, Gabrielsen GW. Persistent organic pollutants in marine macrobenthos near urban settlements in Svalbard; Longyearbyen, Pyramiden, Barentsburg and Ny Ålesund. Tromsø (Norway): Norwegian Polar Institute. 2001.
- 173. Renaud PE, Løkken TS, Jørgensen LL, Berge J, Johnson BJ. Macroalgal detritus and food-web subsidies along an Arctic fjord depth-gradient. Frontiers in Marine Science. 2015;2.
- 174. Mayzaud P, Boutoute M, Noyon M, Narcy F, Gasparini S. Lipid and fatty acids in naturally occurring particulate matter during spring and summer in a high arctic fjord (Kongsfjorden, Svalbard). Marine Biology. 2012;160(2):383-98.
- 175. Koelmans AA, Gillissen F, Makatita W, Van Den Berg M. Organic carbon normalisation of PCB, PAH and pesticide concentrations in suspended solids. Water Research. 1997;31(3):461-70.
- 176. Koelmans AA, Lijklema L, Jiménez CS. Sorption of chlorobenzenes to mineralizing phytoplankton. Environmental Toxicology and Chemistry. 1993;12(8):1425-39.
- 177. Galbán-Malagón C, Berrojalbiz N, Ojeda M-J, Dachs J. The oceanic biological pump modulates the atmospheric transport of persistent organic pollutants to the Arctic. 2012;3:862.
- 178. Kelton WM, William GA, Jr., Beverly JJ, Ming-Yi S, Glenn RL, Lisa MC, et al. Benthic community response to ice algae and phytoplankton in Ny-Ålesund, Svalbard. Marine Ecology Progress Series. 2006;310:1-14.
- 179. Rust AJ, Burgess RM, McElroy AE, Cantwell MG, Brownawell BJ. Influence of soot carbon on the bioaccumulation of sediment-bound polycyclic aromatic hydrocarbons by marine benthic invertebrates: An interspecies comparison. Environmental Toxicology and Chemistry. 2004;23(11):2594-603.
- 180. McElroy AE, Barron MG, Beckvar N, Kane Driscoll SB, Meador JP, Parkerton TF, et al. A review of the tissue residue approach for organic and organometallic compounds in aquatic organisms. Integrated Environmental Assessment and Management. 2011;7(1):50-74.
- 181. Jørgensen A, Giessing AMB, Rasmussen LJ, Andersen O. Biotransformation of polycyclic aromatic hydrocarbons in marine polychaetes. Marine Environmental Research. 2008;65(2):171-86.
- 182. Zheng S, Chen B, Qiu X, Lin K, Yu X. Three novel cytochrome P450 genes identified in the marine polychaete Perinereis nuntia and their transcriptional response to xenobiotics. Aquatic Toxicology. 2013;134–135:11-22.
- 183. Jørgensen A, Giessing AMB, Rasmussen LJ, Andersen O. Biotransformation of the polycyclic aromatic hydrocarbon pyrene in the marine polychaete Nereis virens. Environmental Toxicology and Chemistry. 2005;24(11):2796-805.
- 184. Lüchmann KH, Dafre AL, Trevisan R, Craft JA, Meng X, Mattos JJ, et al. A light in the darkness: New biotransformation genes, antioxidant parameters and tissue-specific responses in oysters exposed to phenanthrene. Aquatic Toxicology. 2014;152:324-34.
- 185. Granberg ME, Selck H. Effects of sediment organic matter quality on bioaccumulation, degradation, and distribution of pyrene in two macrofaunal species and their surrounding sediment. Marine Environmental Research. 2007;64(3):313-35.
- 186. North CA, Lovvorn JR, Kolts JM, Brooks ML, Cooper LW, Grebmeier JM. Deposit-feeder diets in the Bering Sea: potential effects of climatic loss of sea ice-related microalgal blooms. Ecological Applications. 2014;24(6):1525-42.
- 187. Legeżyńska J, Kędra M, Walkusz W. Identifying trophic relationships within the high Arctic benthic community: how much can fatty acids tell? Marine Biology. 2014;161(4):821-36.







- 188. Neff JM, Durell GS, Trefry JH, Brown JS, Environmental studies in the Chukchi Sea 2008: Chemical characterization. Duxbury, MA: Battelle Memorial Institute. Exponent, Inc. Florida Institute of Technology. Neff & Associates. 2010; Service Contract No. 68393.0-SA-AKR.
- Molvær J, Knutzen J, Magnusson J, Rygg B, Skei J, Sørensen J. Classification of environmental quality in fjords and coastal waters. A guide. Statens forurensningstilsyn. 1997; TA-1467/1997 Contract No.: 82-7655-367-2.
- 190. Weisbrod AV, Sahi J, Segner H, James MO, Nichols J, Schultz I, et al. The state of in vitro science for use in bioaccumulation assessments for fish. Environmental Toxicology and Chemistry. 2009;28(1):86-96.
- Menone ML, Miglioranza KSB, Iribarne O, Aizpún de Moreno JE, Moreno VcJ. The role of burrowing beds and burrows of the SW Atlantic intertidal crab Chasmagnathus granulata in trapping organochlorine pesticides. Marine Pollution Bulletin. 2004;48(3-4):240-7.
- Jantunen APK, Tuikka A, Akkanen J, Kukkonen JVK. Bioaccumulation of atrazine and chlorpyrifos to Lumbriculus variegatus from lake sediments. Ecotoxicology and Environmental Safety. 2008;71(3):860-8.
- Besseling E, Wegner A, Foekema EM, van den Heuvel-Greve MJ, Koelmans AA. Effects of microplastic on fitness and PCB bioaccumulation by the lugworm Arenicola marina (L.). Environmental Science & Technology. 2013;47(1):593-600.
- Han YM, Bandowe BAM, Wei C, Cao JJ, Wilcke W, Wang GH, et al. Stronger association of polycyclic aromatic hydrocarbons with soot than with char in soils and sediments. Chemosphere. 2015;119:1335-45.
- Nizzetto PB, Aas, W., Krogseth, I.S. Monitoring of environmental contaminants in air and precipitation, annual report 2013 Kjeller (Norway): NILU - Norwegian Institute for Air Research. 2014; Report No.: M-202/2014 Contract No.: 978-82-425-2690-8.
- Moermond CTA, Zwolsman JJG, Koelmans AA. Black Carbon and ecological factors affect in situ biota to sediment accumulation factors for hydrophobic organic compounds in flood plain lakes. Environmental Science & Technology. 2005;39(9):3101-9.
- Ruus A, Bøyum O, Grung M, Næs K. Bioavailability of PAHs in aluminum smelter affected sediments: Evaluation through assessment of pore water concentrations and in vivo bioaccumulation. Environmental Science & Technology. 2010;44(24):9291-7.
- Ma Y, Halsall CJ, Crosse JD, Graf C, Cai M, He J, et al. Persistent organic pollutants in ocean sediments from the North Pacific to the Arctic Ocean. Journal of Geophysical Research C: Oceans. 2015;120(4):2723-35.
- PAME. Arctic Offshore Oil and Gas Guidelines. Arctic Council. 2009.
- OSPAR. Background Document concerning Techniques for the Management of Produced Water from Offshore Installations. London (United Kingdom): OSPAR Commission. 2002.
- Montagna PA, Baguley JG, Cooksey C, Hartwell I, Hyde LJ, Hyland JL, et al. Deep-sea benthic footprint of the Deepwater Horizon Blowout. PLOS ONE. 2013;8(8):e70540.
- DeBlois EM, Kiceniuk JW, Paine MD, Kilgour BW, Tracy E, Crowley RD, et al. Examination of body burden and taint for Iceland scallop (Chlamys islandica) and American plaice (Hippoglossoides platessoides) near the Terra Nova offshore oil development over ten years of drilling on the Grand Banks of Newfoundland, Canada. Deep Sea Research Part II: Topical Studies in Oceanography. 2014;110:65-83.
- ECETOC. High information content technologies in support of read-across in chemical risk assessment. Brussels: European Centre for Ecotoxicology and Toxicology of Chemicals. 2010; Report No.: 109.
- 204. Boese BL, Lee Ii H, Specht DT, Randall R, Pelletier J. Evaluation of PCB and hexachlorobenzene biota-sediment accumulation factors based on ingested sediment in a deposit-feeding clam. Environmental Toxicology and Chemistry. 1996;15(9):1584-9.









- 205. Thomann RV, Komlos J. Model of biota-sediment accumulation factor for polycyclic aromatic hydrocarbons. Environmental Toxicology and Chemistry. 1999;18(5):1060-8.
- 206. Pouch A, Zaborska A, Pazdro K. Concentrations and origin of polychlorinated biphenyls (PCBs) and polycyclic aromatic hydrocarbons (PAHs) in sediments of western Spitsbergen fjords (Kongsfjorden, Hornsund, and Adventfjorden). Environmental Monitoring and Assessment. 2017;189(4):175.
- 207. Bogen J, Bønsnes TE. Erosion and sediment transport in High Arctic rivers, Svalbard. Polar Research. 2003;22(2):175-89.
- 208. Steur C, Seys J, Eppinga J. Ecologisch profiel van het Nonnetje (Macoma balthica). 1996.
- 209. Weems J, Iken K, Gradinger R, Wooller MJ. Carbon and nitrogen assimilation in the Bering Sea clams Nuculana radiata and Macoma moesta. Journal of Experimental Marine Biology and Ecology. 2012;430–431:32-42.
- 210. Neff JM, Hillman RE, Scott Carr R, Buhl RL, Lahey JI. Histopathologic and biochemical responses in Arctic marine bivalve molluses exposed to experimentally spilled oil. Arctic. 1987;40(Suppl 1):220-9.
- 211. Sun M-Y, Clough LM, Carroll ML, Dai J, Ambrose Jr WG, Lopez GR. Different responses of two common Arctic macrobenthic species (Macoma balthica and Monoporeia affinis) to phytoplankton and ice algae: Will climate change impacts be species specific? Journal of Experimental Marine Biology and Ecology. 2009;376(2):110-21.
- 212. Jensen LK, Honkanen JO, Jæger I, Carroll J. Bioaccumulation of phenanthrene and benzo[a]pyrene in Calanus finmarchicus. Ecotoxicology and Environmental Safety. 2012;78:225-31.
- 213. Carrasco-Navarro V, Jæger I, Honkanen JO, Kukkonen JVK, Carroll J, Camus L. Bioconcentration, biotransformation and elimination of pyrene in the arctic crustacean Gammarus setosus (Amphipoda) at two temperatures. Marine Environmental Research. 2015;110:101-9.
- 214. Tschischka K, Abele D, Portner HO. Mitochondrial oxyconformity and cold adaptation in the polychaete Nereis pelagica and the bivalve Arctica islandica from the Baltic and White Seas. Journal of Experimental Biology. 2000;203(21):3355.
- 215. Nahrgang J, Brooks SJ, Evenset A, Camus L, Jonsson M, Smith TJ, et al. Seasonal variation in biomarkers in blue mussel (Mytilus edulis), Icelandic scallop (Chlamys islandica) and Atlantic cod (Gadus morhua)—Implications for environmental monitoring in the Barents Sea. Aquatic Toxicology. 2013;127:21-35.
- 216. Cornelissen G, Gustafsson Ö, Bucheli TD, Jonker MTO, Koelmans AA, Van Noort PCM. Extensive sorption of organic compounds to black carbon, coal, and kerogen in sediments and soils: Mechanisms and consequences for distribution, bioaccumulation, and biodegradation. Environmental Science & Technology. 2005;39(18):6881–95.
- 217. Keuken MP, Moerman M, Zandveld P, Henzing JS. Total and size-resolved particle number and black carbon concentrations near an industrial area. Atmospheric Environment. 2015;122:196-205.
- 218. Cornelissen G, Breedveld GD, Næs K, Oen AMP, Ruus A. Bioaccumulation of native polycyclic aromatic hydrocarbons from sediment by a polychaete and a gastropod: Freely dissolved concentrations and activated carbon amendment. Environmental Toxicology and Chemistry. 2006;25(9):2349-55.
- 219. Beukema JJ, Cadée GC, Dekker R, Philippart CJM. Annual and spatial variability in gains of body weight in Macoma balthica (L.): Relationships with food supply and water temperature. Journal of Experimental Marine Biology and Ecology. 2014;457:105-12.
- 220. McLeod PB, van den Heuvel-Greve MJ, Allen-King RM, Luoma SN, Luthy RG. Effects of particulate carbonaceous matter on the bioavailability of benzo[a]pyrene and 2,2<sup>c</sup>,5,5<sup>c</sup>-tetrachlorobiphenyl to the clam, Macoma balthica. Environmental Science & Technology. 2004;38(17):4549-56.







- 221. Ruus A, Allan IJ, Øxnevad S, Schaanning MT, Borgå K, Bakke T, et al. In vivo bioaccumulation of contaminants from historically polluted sediments Relation to bioavailability estimates. Science of the Total Environment. 2013;442:336-43.
- 222. Yates K, Pollard P, Davies IM, Webster L, Moffat CF. Application of silicone rubber passive samplers to investigate the bioaccumulation of PAHs by Nereis virens from marine sediments. Environmental Pollution. 2011;159(12):3351-6.
- 223. Tremblay L, Kohl SD, Rice JA, Gagné J-P. Effects of temperature, salinity, and dissolved humic substances on the sorption of polycyclic aromatic hydrocarbons to estuarine particles. Marine Chemistry. 2005;96(1–2):21-34.
- 224. Koelmans AA, Jiménez CS. Temperature dependency of chlorobenzene bioaccumulation in phytoplankton. Chemosphere. 1994;28(12):2041-8.
- 225. Hanelt D, Tüg H, Bischof K, Groß C, Lippert H, Sawall T, et al. Light regime in an Arctic fjord: a study related to stratospheric ozone depletion as a basis for determination of UV effects on algal growth. Marine Biology. 2001;138(3):649-58.
- 226. Abele D, Burlando B, Viarengo A, Pörtner H-O. Exposure to elevated temperatures and hydrogen peroxide elicits oxidative stress and antioxidant response in the Antarctic intertidal limpet Nacella concinna. Comparative Biochemistry and Physiology Part B: Biochemistry and Molecular Biology. 1998;120(2):425-35.
- 227. Prevodnik A, Gardeström J, Lilja K, Elfwing T, McDonagh B, Petrović N, et al. Oxidative stress in response to xenobiotics in the blue mussel Mytilus edulis L.: Evidence for variation along a natural salinity gradient of the Baltic Sea. Aquatic Toxicology. 2007;82(1):63-71.
- 228. Neuhoff H-G. Influence of temperature and salinity on food conversion and growth of different Nereis Species (Polychaeta, Annelida). Marine Ecology Progress Series. 1979;1:255-62.
- 229. Olive PJW. Reproduction and annual gametogenic cycle in Nephtys hombergii and N. caeca (Polychaeta: Nephtyidae). Marine Biology. 1978;46(1):83-90.
- 230. Lawrence AJ, Soame JM. The endocrine control of reproduction in Nereidae: a new multi-hormonal model with implications for their functional role in a changing environment. Philosophical Transactions of the Royal Society B: Biological Sciences. 2009;364(1534):3363-76.
- 231. Graf G, Bengtsson W, Diesner U, Schulz R, Theede H. Benthic response to sedimentation of a spring phytoplankton bloom: Process and budget. Marine Biology. 1982;67(2):201-8.
- 232. Peter JWO, Simon WR, Ali D. Influence of photoperiod and temperature on oocyte growth in the semelparous polychaete Nereis (Neanthes) virens. Marine Ecology Progress Series. 1998;172:169-83.
- 233. Christensen M, Andersen O, Banta GT. Metabolism of pyrene by the polychaetes Nereis diversicolor and Arenicola marina. Aquatic Toxicology. 2002;58(1–2):15-25.
- 234. Catalano B, Moltedo G, Martuccio G, Gastaldi L, Virno-Lamberti C, Lauria A, et al. Can Hediste diversicolor (Nereidae, Polychaete) be considered a good candidate in evaluating PAH contamination? A multimarker approach. Chemosphere. 2012;86(9):875-82.
- 235. Jumars PA, Dorgan KM, Lindsay SM. Diet of worms emended: An update of polychaete feeding guilds. Annual Review of Marine Science. 2015;7(1):497-520.
- 236. Ahrens MJ, Hertz J, Lamoureux EM, Lopez GR, McElroy AE, Brownawell BJ. The effect of body size on digestive chemistry and absorption efficiencies of food and sediment-bound organic contaminants in Nereis succinea (Polychaeta). Journal of Experimental Marine Biology and Ecology. 2001;263(2):185-209.
- 237. Karickhoff SW, Brown DS, Scott TA. Sorption of hydrophobic pollutants on natural sediments. Water Research. 1979;13(3):241-8.







- 238. Thomann RV, Connolly JP, Parkerton TF. An equilibrium model of organic chemical accumulation in aquatic food webs with sediment interaction. Environmental Toxicology and Chemistry, 1992;11(5):615-29.
- McLeod PB, Luoma SN, Luthy RG. Biodynamic modeling of PCB uptake by Macoma balthica and Corbicula fluminea from sediment amended with activated carbon. Environmental Science & Technology. 2008;42(2):484-90.
- McLeod PB, van den Heuvel-Greve MJ, Luoma SN, Luthy RG. Biological uptake of polychlorinated biphenyls by Macoma balthica from sediment amended with activated carbon. Environmental Toxicology and Chemistry. 2007;26(5):980-7.
- 241. Seth R, Mackay D, Muncke J. Estimating the organic carbon partition coefficient and its variability for hydrophobic chemicals. Environmental Science & Technology. 1999;33(14):2390-4.
- 242. Schwarzenbach RP, Gschwend PM, Imboden DM. Environmental Organic Chemistry: John Wiley & Sons. 2005.
- Høpner Petersen G. Life cycles and population dynamics of marine benthic bivalves from the Disko Bugt area of West Greenland. Ophelia. 1978;17(1):95-120.
- 244. Draper NR, Smith H. Applied Regression Analysis: Wiley. 1981.
- Anisimova NA, Jørgensen LL, Lyubin PA, Manushin IE. Mapping and monitoring of benthos in the Barents Sea and Svalbard waters: Results from the joint Russian - Norwegian benthic programme 2006-2008. Institute of Marine Research - IMR. Polar Research Institute of Marine Fisheries and Oceanography - PINRO. 2010; Contract No.: 1502-8828.
- Gogina M, Nygård H, Blomqvist M, Daunys D, Josefson AB, Kotta J, et al. The Baltic Sea scale inventory of benthic faunal communities. ICES Journal of Marine Science: Journal du Conseil. 2016 January 26, 2016.
- 247. Lovvorn JR, Cooper LW, Brooks ML, De Ruyck CC, Bump JK, Grebmeier JM. Organic matter pathways to zooplankton and benthos under pack ice in late winter and open water in late summer in the northcentral Bering Sea. Marine Ecology Progress Series. 2005;291:135-50.
- Denisenko NV, Rachor E, Denisenko SG. Benthic Fauna of the Southern Kara Sea. In: Stein R, Fahl K, Fütterer DK, Galimov EM, Stepanets OV, editors. Siberian river run-off in the Kara Sea: Elsevier Science B.V. 2003.
- Strand J, Jacobsen JA, Pedersen B, Granmo Å. Butyltin compounds in sediment and molluscs from the shipping strait between Denmark and Sweden. Environmental Pollution. 2003;124(1):7-15.
- Oxtoby LE, Budge SM, Iken K, O'Brien DM, Wooller MJ. Feeding ecologies of key bivalve and polychaete species in the Bering Sea as elucidated by fatty acid and compound-specific stable isotope analyses. Marine Ecology Progress Series. 2016;557:161-75.
- Pedersen TF. Metabolic adaptations to hypoxia of two species of polychaeta, Nephtys ciliata and Nephtys hombergii. Journal of Comparative Physiology B. 1991;161(2):213-5.
- Decho AW, Luoma SN. Time-courses in the retention of food material in the bivalves Potamocorbula amurensis and Macoma balthica significance to the absorption of carbon and chromium. Marine Ecology Progress Series. 1991;78:303-14.
- Lewis C, Galloway T. Genotoxic damage in polychaetes: A study of species and cell-type sensitivities. Mutation Research/Genetic Toxicology and Environmental Mutagenesis. 2008;654(1):69-75.
- McElroy AE, Farrington JW, Teal JM. Influence of mode of exposure and the presence of a tubiculous polychaete on the fate of benz[a]anthracene in the benthos. Environmental Science & Technology. 1990;24(11):1648-55.









- 255. Oliver M, Desrosiers G, Caron A, Retière C. Juvenile growth of the polychaete Nereis virens feeding on a range of marine vascular and macroalgal plant sources. Marine Biology. 1996;125(4):693-9.
- 256. Amiard-Triquet C, Berthet B. Individual Biomarkers. In: Amiard-Triquet C, Amiard J-C, Mouneyrac C, editors. Aquatic Ecotoxicology. 2015.
- 257. Esler D, Bowman TD, Trust KA, Ballachey BE, Dean TA, Jewett SC, et al. Harlequin duck population recovery following the 'Exxon Valdez' oil spill: Progress, process and constraints. Marine Ecology Progress Series. 2002;241:271-86.
- 258. Eason C, O'Halloran K. Biomarkers in toxicology versus ecological risk assessment. Toxicology. 2002;181–182:517-21.
- 259. Olsen GH, Sva E, Carroll J, Camus L, De Coen W, Smolders R, et al. Alterations in the energy budget of Arctic benthic species exposed to oil-related compounds. Aquatic Toxicology. 2007;83(2):85-92.
- 260. Camus L, Jones MB, Børseth JF, Regoli F, Depledge MH. Heart rate, respiration and total oxyradical scavenging capacity of the Arctic spider crab, Hyas araneus, following exposure to polycyclic aromatic compounds via sediment and injection. Aquatic Toxicology. 2002;61(1–2):1-13.
- 261. Jewett SC, Dean TA, Woodin BR, Hoberg MK, Stegeman JJ. Exposure to hydrocarbons 10 years after the Exxon Valdez oil spill: evidence from cytochrome P4501A expression and biliary FACs in nearshore demersal fishes. Marine Environmental Research. 2002;54(1):21-48.
- 262. Payne JR, Driskell WB, Short JW, Larsen ML. Long term monitoring for oil in the Exxon Valdez spill region. Marine Pollution Bulletin. 2008;56(12):2067-81.
- 263. Boehm PD, Mankiewicz PJ, Reilly JEO, Hartung R, Neff JM, Page DS, et al. Characterization of mussel beds with residual oil and the risk to foraging wildlife 4 years after the Exxon valdez oil spill. Environmental Toxicology and Chemistry. 1996;15(8):1289-303.
- 264. Small GM, Burdett K, Connock MJ. A sensitive spectrophotometric assay for peroxisomal acyl-CoA oxidase. Biochemical Journal. 1985;227(1):205-10.
- 265. Guilhermino L, Lopes MC, Carvalho AP, Soared AMVM. Inhibition of acetylcholinesterase activity as effect criterion in acute tests with juvenile Daphnia Magna. Chemosphere. 1996;32(4):727-38.
- 266. Habig WH, Pabst MJ, Jakoby WB. Glutathione S-transferases. The first enzymatic step in mercapturic acid formation. Journal of Biological Chemistry. 1974;249(22):7130-9.
- 267. Lowry OH, Rosebrough NJ, Lewis Farr A, Randall RJ. Protein measurement with the Folin phenol reagent. Journal of Biological Chemistry. 1951;193(1):265-75.
- 268. Zabaleta I, Bizkarguenaga E, Prieto A, Ortiz-Zarragoitia M, Fernández LA, Zuloaga O. Simultaneous determination of perfluorinated compounds and their potential precursors in mussel tissue and fish muscle tissue and liver samples by liquid chromatography—electrospray-tandem mass spectrometry. Journal of Chromatography A. 2015;1387:13-23.
- 269. Errekatxo A, Prieto A, Zuloaga O, Usobiaga A, Etxebarria N, Fernández LA. Simultaneous extraction of several persistent organic pollutants in sediment using focused ultrasonic solid-liquid extraction. Analytical and Bioanalytical Chemistry. 2008;392(7):1471-8.
- 270. Leiniö S, Lehtonen KK. Seasonal variability in biomarkers in the bivalves Mytilus edulis and Macoma balthica from the northern Baltic Sea. Comparative Biochemistry and Physiology Part C: Toxicology & Pharmacology. 2005;140(3–4):408-21.
- 271. Vidal-Liñán L, Bellas J, Etxebarria N, Nieto O, Beiras R. Glutathione S-transferase, glutathione peroxidase and acetylcholinesterase activities in mussels transplanted to harbour areas. Science of the Total Environment. 2014;470–471:107-16.







- 272. Froment J, Langford K, Tollefsen KE, Bråte ILN, Brooks SJ, Thomas KV. Identification of petrogenic produced water components as acetylcholine esterase inhibitors. Environmental Pollution. 2016;215:18-26.
- 273. Bonacci S, Corsi I, Focardi S. Cholinesterases in the Antarctic scallop Adamussium colbecki: Characterization and sensitivity to pollutants. Ecotoxicology and Environmental Safety. 2009;72(5):1481-8.
- 274. Lehtonen KK, Turja R, Budzinski H, Devier M-H. An integrated chemical-biological study using caged mussels (Mytilus trossulus) along a pollution gradient in the Archipelago Sea (SW Finland, Baltic Sea). Marine Environmental Research. 2016;119(Supplement C):207-21.
- 275. Baussant T, Bechmann RK, Taban IC, Larsen BK, Tandberg AH, Bjørnstad A, et al. Enzymatic and cellular responses in relation to body burden of PAHs in bivalve molluscs: A case study with chronic levels of North Sea and Barents Sea dispersed oil. Marine Pollution Bulletin. 2009;58(12):1796-807.
- 276. Giuliani ME, Benedetti M, Arukwe A, Regoli F. Transcriptional and catalytic responses of antioxidant and biotransformation pathways in mussels, Mytilus galloprovincialis, exposed to chemical mixtures. Aquatic Toxicology. 2013;134–135:120-7.
- 277. Gowland BTG, McIntosh AD, Davies IM, Moffat CF, Webster L. Implications from a field study regarding the relationship between polycyclic aromatic hydrocarbons and glutathione S-transferase activity in mussels. Marine Environmental Research. 2002;54(3–5):231-5.
- 278. González-Fernández C, Albentosa M, Sokolova I. Interactive effects of nutrition, reproductive state and pollution on molecular stress responses of mussels, Mytilus galloprovincialis Lamarck, 1819. Marine Environmental Research. 2017;131:103-15.
- 279. Orbea A, Cajaraville MP. Peroxisome proliferation and antioxidant enzymes in transplanted mussels of four basque estuaries with different levels of polycyclic aromatic hydrocarbon and polychlorinated biphenyl pollution. Environmental Toxicology and Chemistry. 2006;25(6):1616-26.
- 280. Brooks S, Harman C, Soto M, Cancio I, Glette T, Marigómez I. Integrated coastal monitoring of a gas processing plant using native and caged mussels. Science of the Total Environment. 2012;426:375-86.
- 281. Cajaraville MP, Cancio I, Ibabe A, Orbea A. Peroxisome proliferation as a biomarker in environmental pollution assessment. Microscopy Research and Technique. 2003;61(2):191-202.
- 282. Cancio I, Ibabe A, P. Cajaraville M. Seasonal variation of peroxisomal enzyme activities and peroxisomal structure in mussels Mytilus galloprovincialis and its relationship with the lipid content. Comparative Biochemistry and Physiology Part C: Pharmacology, Toxicology and Endocrinology. 1999;123(2):135-44.
- 283. Von Oertzen J-A. Cycles and rates of reproduction of six Baltic Sea bivalves of different zoogeographical origin. Marine Biology. 1972;14(2):143-9.
- 284. Livingstone DR. Biotechnology and pollution monitoring: Use of molecular biomarkers in the aquatic environment. Journal of Chemical Technology & Biotechnology. 1993;57(3):195-211.
- 285. Skarphéðinsdóttir H, Ericson G, Dalla Zuanna L, Gilek M. Tissue differences, dose–response relationship and persistence of DNA adducts in blue mussels (Mytilus edulis L.) exposed to benzo[a]pyrene. Aquatic Toxicology. 2003;62(2):165-77.
- 286. Jenny MJ, Walton WC, Payton SL, Powers JM, Findlay RH, O'Shields B, et al. Transcriptomic evaluation of the American oyster, Crassostrea virginica, deployed during the Deepwater Horizon oil spill: Evidence of an active hydrocarbon response pathway. Marine Environmental Research. 2016;120:166-81.
- 287. Carrasco Navarro V, Brozinski J-M, Leppänen MT, Honkanen JO, Kronberg L, Kukkonen JVK. Inhibition of pyrene biotransformation by piperonyl butoxide and identification of two pyrene derivatives in Lumbriculus variegatus (Oligochaeta). Environmental Toxicology and Chemistry. 2011;30(5):1069-78.







- 288. Simpson CD, Cullen WR, He TYT, Ikonomou M, Reimer KJ. Metabolism of pyrene by two clam species, Mya arenaria and Protothaca staminea. Chemosphere. 2002;49(3):315-22.
- 289. Díaz-Jaramillo M, Miglioranza KSB, Gonzalez M, Barón E, Monserrat JM, Eljarrat E, et al. Uptake, metabolism and sub-lethal effects of BDE-47 in two estuarine invertebrates with different trophic positions. Environmental Pollution. 2016;213:608-17.
- 290. Guo R, Pan L, Lin P, Zheng L. The detoxification responses, damage effects and bioaccumulation in the scallop Chlamys farreri exposed to single and mixtures of benzo[a]pyrene and chrysene. Comparative Biochemistry and Physiology Part C: Toxicology & Pharmacology. 2017;191:36-51.
- 291. Beach DG, Quilliam MA, Rouleau C, Croll RP, Hellou J. Bioaccumulation and biotransformation of pyrene and 1-hydroxypyrene by the marine whelk Buccinum undatum. Environmental Toxicology and Chemistry. 2010;29(4):779-88.
- 292. Beach DG, Quilliam MA, Hellou J. Analysis of pyrene metabolites in marine snails by liquid chromatography using fluorescence and mass spectrometry detection. Journal of Chromatography B. 2009;877(22):2142-52.
- 293. Neff JM, Ostazeski S, Gardiner W, Stejskal I. Effects of weathering on the toxicity of three offshore australian crude oils and a diesel fuel to marine animals. Environmental Toxicology and Chemistry. 2000;19(7):1809-21.
- 294. Sammarco PW, Kolian SR, Warby RAF, Bouldin JL, Subra WA, Porter SA. Distribution and concentrations of petroleum hydrocarbons associated with the BP/Deepwater Horizon Oil Spill, Gulf of Mexico. Marine Pollution Bulletin. 2013;73(1):129-43.
- 295. Helle I, Jolma A, Venesjärvi R. Species and habitats in danger: estimating the relative risk posed by oil spills in the northern Baltic Sea. Ecosphere. 2016;7(5):e01344-n/a.
- 296. Fernández-Macho J. Risk assessment for marine spills along European coastlines. Marine Pollution Bulletin. 2016;113(1):200-10.
- 297. Laramore S, Krebs W, Garr A. Effects of exposure of pink shrimp, Farfantepenaeus duorarum, larvae to Macondo Canyon 252 Crude Oil and the Corexit dispersant. Journal of Marine Science and Engineering. 2016;4(1):24.
- 298. Hellou J. Behavioural ecotoxicology, an "early warning" signal to assess environmental quality. Environmental Science and Pollution Research International. 2011;18(1):1-11.
- 299. Zimmer-Faust RK. The relationship between chemoreception and foraging behavior in crustaceans. Limnology and Oceanography. 1989;34(7):1367-74.
- 300. Gray RH. Fish behavior and environmental assessment. Environmental Toxicology and Chemistry. 1990;9(1):53-67.
- 301. De Coen WM, Janssen CR. The missing biomarker link: Relationships between effects on the cellular energy allocation biomarker of toxicant-stressed Daphnia magna and corresponding population characteristics. Environmental Toxicology and Chemistry. 2003;22(7):1632-41.
- 302. Luís LG, Guilhermino L. Short-term toxic effects of naphthalene and pyrene on the common prawn (Palaemon serratus) assessed by a multi-parameter laboratorial approach: mechanisms of toxicity and impairment of individual fitness. Biomarkers. 2012;17(3):275-85.
- 303. Tierney KB. Chemical avoidance responses of fishes. Aquatic Toxicology. 2016;174:228-41.
- 304. Lotufo GR. Toxicity of sediment-associated PAHs to an estuarine copepod: Effects on survival, feeding, reproduction and behavior. Marine Environmental Research. 1997;44(2):149-66.







- 305. Hellou J, Cheeseman K, Desnoyers E, Johnston D, Jouvenelle M-L, Leonard J, et al. A non-lethal chemically based approach to investigate the quality of harbour sediments. Science of the Total Environment. 2008;389(1):178-87.
- 306. Scarlett A, Rowland SJ, Canty M, Smith EL, Galloway TS. Method for assessing the chronic toxicity of marine and estuarine sediment-associated contaminants using the amphipod Corophium volutator. Marine Environmental Research. 2007;63(5):457-70.
- 307. Scott GR, Sloman KA. The effects of environmental pollutants on complex fish behaviour: integrating behavioural and physiological indicators of toxicity. Aquatic Toxicology, 2004;68(4):369-92.
- 308. Blocker TD, Ophir AG. Cryptic confounding compounds: A brief consideration of the influences of anthropogenic contaminants on courtship and mating behavior. Acta ethologica. 2013;16(2):10.1007/s10211-012-0137-x.
- 309. Brils JM, Huwer SL, Kater BJ, Schout PG, Harmsen J, Delvigne GAL, et al. Oil effect in freshly spiked marine sediment on Vibrio fischeri, Corophium volutator, and Echinocardium cordatum. Environmental Toxicology and Chemistry. 2002;21(10):2242-51.
- 310. Dehedin A, Piscart C, Marmonier P. Seasonal variations of the effect of temperature on lethal and sublethal toxicities of ammonia for three common freshwater shredders. Chemosphere. 2013;90(3):1016-22.
- 311. Foucreau N, Cottin D, Piscart C, Hervant F. Physiological and metabolic responses to rising temperature in Gammarus pulex (Crustacea) populations living under continental or Mediterranean climates. Comparative Biochemistry and Physiology Part A: Molecular & Integrative Physiology. 2014;168:69-75.
- 312. Henry Y, Piscart C, Charles S, Colinet H. Combined effect of temperature and ammonia on molecular response and survival of the freshwater crustacean Gammarus pulex. Ecotoxicology and Environmental Safety. 2017;137:42-8.
- 313. Neuparth T, Costa FO, Costa MH. Effects of temperature and salinity on life history of the marine amphipod Gammarus locusta. Implications for ecotoxicological testing. Ecotoxicology. 2002;11(1):61-73.
- 314. Stronkhorst J, Hattum Bv. Contaminants of concern in Dutch marine harbor sediments. Archives of Environmental Contamination and Toxicology. 2003;45(3):306-16.
- 315. Koelmans AA, Kaag K, Sneekes A, Peeters ETHM. Triple domain in situ sorption modeling of organochlorine pesticides, polychlorobiphenyls, polyaromatic hydrocarbons, polychlorinated dibenzo-p-dioxins, and polychlorinated dibenzofurans in aquatic sediments. Environmental Science & Technology. 2009;43(23):8847-53.
- 316. Faksness L-G, Altin D, Nordtug T, Daling PS, Hansen BH. Chemical comparison and acute toxicity of water accommodated fraction (WAF) of source and field collected Macondo oils from the Deepwater Horizon spill. Marine Pollution Bulletin. 2015;91(1):222-9.
- 317. Brown-Peterson NJ, Krasnec M, Takeshita R, Ryan CN, Griffitt KJ, Lay C, et al. A multiple endpoint analysis of the effects of chronic exposure to sediment contaminated with Deepwater Horizon oil on juvenile Southern flounder and their associated microbiomes. Aquatic Toxicology. 2015;165(Supplement C):197-209.
- 318. Lotufo GR, Farrar JD, Biedenbach JM, Laird JG, Krasnec MO, Lay C, et al. Effects of sediment amended with Deepwater Horizon incident slick oil on the infaunal amphipod Leptocheirus plumulosus. Marine Pollution Bulletin. 2016;109(1):253-8.
- 319. Hunte W, Myers RA. Phototaxis and cannibalism in gammaridean amphipods. Marine Biology. 1984;81(1):75-9.
- 320. Verberk WCEP, Overgaard J, Ern R, Bayley M, Wang T, Boardman L, et al. Does oxygen limit thermal tolerance in arthropods? A critical review of current evidence. Comparative Biochemistry and Physiology Part A, Molecular & Integrative Physiology. 2016;192:64-78.







- 321. Sutcliffe DW. Quantitative aspects of oxygen uptake by Gammarus (Crustacea, Amphipoda): a critical review. Freshwater Biology. 1984;14(5):443-89.
- Vellinger C, Gismondi E, Felten V, Rousselle P, Mehennaoui K, Parant M, et al. Single and combined effects of cadmium and arsenate in Gammarus pulex (Crustacea, Amphipoda): Understanding the links between physiological and behavioural responses. Aquatic Toxicology. 2013;140-141:106-16.
- Kaufmann RS. Structure and function of chemoreceptors in scavenging lysianassoid amphipods. Journal of Crustacean Biology. 1994;14(1):54-71.
- Maazouzi C, Piscart C, Legier F, Hervant F. Ecophysiological responses to temperature of the "killer shrimp" Dikerogammarus villosus: Is the invader really stronger than the native Gammarus pulex? Comparative Biochemistry and Physiology Part A: Molecular & Integrative Physiology. 2011;159(3):268-74.
- Charron L, Geffard O, Chaumot A, Coulaud R, Jaffal A, Gaillet V, et al. Influence of molting and starvation on digestive enzyme activities and energy storage in Gammarus fossarum. PLOS ONE. 2014;9(4).
- Scarlett A, Canty MN, Smith EL, Rowland SJ, Galloway TS. Can amphipod behavior help to predict chronic toxicity of sediments? Human and Ecological Risk Assessment: An International Journal. 2007;13(3):506-18.
- 327. Costa FO, Costa MH. Life history of the amphipod Gammarus locusta in the Sado estuary (Portugal). Acta Oecologica. 1999;20(4):305-14.
- Jager T, Altin D, Miljeteig C, Hansen BH. Stage-dependent and sex-dependent sensitivity to watersoluble fractions of fresh and weathered oil in the marine copepod Calanus finmarchicus. Environmental Toxicology and Chemistry. 2016;35(3):728-35.
- 329. Kolding S, Fenchel TM. Coexistence and life cycle characteristics of five species of the amphipod Genus Gammarus. Oikos. 1979;33(2):323-7.
- 330. Correia AD, Costa MH, Luis OJ, Livingstone DR. Age-related changes in antioxidant enzyme activities, fatty acid composition and lipid peroxidation in whole body Gammarus locusta (Crustacea: Amphipoda), Journal of Experimental Marine Biology and Ecology. 2003;289(1):83-101.
- 331. Gismondi E, Cossu-Leguille C, Beisel JN. Do male and female gammarids defend themselves differently during chemical stress? Aquatic Toxicology. 2013;140-141:432-8.
- Welton JS, Clarke RT. Laboratory studies on the reproduction and growth of the amphipod, Gammarus pulex (L.). Journal of Animal Ecology. 1980;49(2):581-92.
- Sutcliffe DW, Carrick TR, Willoughby LG. Effects of diet, body size, age and temperature on growth rates in the amphipod Gammarus pulex. Freshwater Biology. 1981;11(2):183-214.
- Delgado L, Guerao G, Ribera C. Effects of different salinities on juvenile growth of Gammarus aequicauda (Malacostraca: Amphipoda). International Journal of Zoology. 2011;2011:6.
- Dobrzycka A, Szaniawska A. The effect of salinity on osmoregulation in Corophium volutator (Pallas) and Saduria entomon (linnaeus) from the gulf of Gdańsk. Oceanologia. 1995;37(1):111-22.
- Vereshchagina KP, Lubyaga YA, Shatilina Z, Bedulina D, Gurkov A, Axenov-Gribanov DV, et al. Salinity modulates thermotolerance, energy metabolism and stress response in amphipods Gammarus lacustris. PeerJ. 2016;4:e2657.
- 337. LeMiere C, Mazo J. Arctic opening: Insecurity and opportunity. Series A, editor: Routledge. 2014.
- 338. Aksenov Y, Popova EE, Yool A, Nurser AJG, Williams TD, Bertino L, et al. On the future navigability of Arctic sea routes: High-resolution projections of the Arctic Ocean and sea ice. Marine Policy. 2017;75:300-17.









- 339. Forskningsrådet. Long-term effects of discharges to sea from petroleum-related activities: The results of ten years' research. Oslo (Norway): Research Council of Norway. 2012; Contract No.: 978-82-12-03027-5.
- Sanni S, Lyng E, Pampanin DM, Smit MGD. II. Species sensitivity distributions based on biomarkers and whole organism responses for integrated impact and risk assessment criteria. Marine Environmental Research. 2017;127:11-23.
- De Bruijn J, Hansen BG, Johansson S, Luotamo M, Munn SJ, Musset C, et al. Technical Guidance Document on Risk Assessment (Part 2). European Commission. 2003; EUR 20418 EN.
- Maltby L. Environmental Risk Assessment. In: Harrison RM, Hester RE, editors. Chemicals in the Environment: Assessing and Managing Risk, London (United Kingdom): Royal Society of Chemistry, 2006.
- Iversen PE, Lind MJ, Ersvik M, Rønning I, Skaare BB, Green AMV, et al. Guidelines for environmental monitoring of petroleum activities on the Norwegian continental shelf (Retningslinjer for miljøovervåking av petrolemsvirksomheten til havs). Norwegian Environment Agency. 2015; M-408.
- Hylland K, Tollefsen K-E, Ruus A, Jonsson G, Sundt RC, Sanni S, et al. Water column monitoring near oil installations in the North Sea 2001-2004. Marine Pollution Bulletin. 2008;56(3):414-29.
- Vethaak AD, Davies IM, Thain JE, Gubbins MJ, Martínez-Gómez C, Robinson CD, et al. Integrated indicator framework and methodology for monitoring and assessment of hazardous substances and their effects in the marine environment. Marine Environmental Research. 2017;124:11-20.
- Degen R, Aune M, Bluhm BA, Cassidy C, Kedra M, Kraan C, et al. Trait-based approaches in rapidly changing ecosystems: A roadmap to the future polar oceans. Ecological Indicators. 2018;91:722-36.
- 347. Diepens NJ, Arts GHP, Brock TCM, Smidt H, Van Den Brink PJ, Van Den Heuvel-Greve MJ, et al. Sediment toxicity testing of organic chemicals in the context of prospective risk assessment: A review. Critical Reviews in Environmental Science and Technology. 2014;44(3):255-302.
- 348. Peeters ETHM, Gardeniers JJP, Koelmans AA. Contribution of trace metals in structuring in situ macroinvertebrate community composition along a salinity gradient. Environmental Toxicology and Chemistry. 2000;19(4):1002-10.
- Peeters ETHM, Dewitte A, Koelmans AA, Velden JAvd, Besten PJd. Evaluation of bioassays versus contaminant concentrations in explaining the macroinvertebrate community structure in the Rhine-Meuse delta, the Netherlands. Environmental Toxicology and Chemistry. 2001;20(12):2883-91.
- Washburn TW, Reuscher MG, Montagna PA, Cooksey C, Hyland JL. Macrobenthic community structure in the deep Gulf of Mexico one year after the Deepwater Horizon blowout. Deep Sea Research Part I: Oceanographic Research Papers. 2017;127:21-30.
- 351. Souster TA, Morley SA, Peck LS. Seasonality of oxygen consumption in five common Antarctic benthic marine invertebrates. Polar Biology. 2018.
- Fernández B, Campillo JA, Martínez-Gómez C, Benedicto J. Assessment of the mechanisms of detoxification of chemical compounds and antioxidant enzymes in the digestive gland of mussels, Mytilus galloprovincialis, from Mediterranean coastal sites. Chemosphere. 2012;87(11):1235-45.
- Ademollo N, Patrolecco L, Matozzo V, Marin MG, Valsecchi S, Polesello S. Clam bioaccumulation of alkylphenols and polyciclic aromatic hydrocarbons in the Venice lagoon under different pressures. Marine Pollution Bulletin. 2017;124(1):121-9.
- McLeod AM, Paterson G, Drouillard KG, Haffner GD, Ecological implications of steady state and nonsteady state bioaccumulation models. Environmental Science & Technology. 2016;50(20):11103-11.
- Włodarska-Kowalczuk M, Górska B, Deja K, Morata N. Do benthic meiofaunal and macrofaunal communities respond to seasonality in pelagial processes in an Arctic fjord (Kongsfjorden, Spitsbergen)? Polar Biology. 2016;39(11):2115-29.





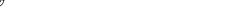




- 356. Svendsen TCC, Lionel, Hargrave B, Fisk A, Muir D, C. G., Borgå K. Polyaromatic hydrocarbons, chlorinated and brominated organic contaminants as tracers of feeding ecology in polar benthic amphipods. Marine Ecology Progress Series. 2007;337:155-64.
- 357. Pernet F, Tremblay R, Comeau L, Guderley H. Temperature adaptation in two bivalve species from different thermal habitats: energetics and remodelling of membrane lipids. Journal of Experimental Biology. 2007;210(17):2999.
- 358. Petersen JK, Sejr MK, Larsen JEN. Clearance rates in the Arctic bivalves Hiatella arctica and Mya sp. Polar Biology. 2003;26(5):334-41.
- 359. Gewurtz SB, Laposa R, Gandhi N, Christensen GN, Evenset A, Gregor D, et al. A comparison of contaminant dynamics in arctic and temperate fish: A modeling approach. Chemosphere. 2006;63(8):1328-41.
- 360. Sobek A, McLachlan MS, Borgå K, Asplund L, Lundstedt-Enkel K, Polder A, et al. A comparison of PCB bioaccumulation factors between an arctic and a temperate marine food web. Science of the Total Environment. 2010;408(13):2753-60.
- 361. Lehtonen KK, Sundelin B, Lang T, Strand J. Development of tools for integrated monitoring and assessment of hazardous substances and their biological effects in the Baltic Sea. Ambio. 2014;43(1):69-81.
- 362. Hanson N, Halling M, Norin H. Biomarkers for Environmental Monitoring Suggestions for Norwegian monitoring programmes. Miljødirektoratet. 2013; Contract No.: M-88.
- 363. Chapman PM, Wang F, Caeiro SS. Assessing and managing sediment contamination in transitional waters. Environment International. 2013;55:71-91.
- 364. Bocquené G, Chantereau S, Clérendeau C, Beausir E, Ménard D, Raffin B, et al. Biological effects of the "Erika" oil spill on the common mussel (Mytilus edulis). Aquatic Living Resources. 2004;17(3):309-16.
- 365. Broeg K, Lehtonen KK. Indices for the assessment of environmental pollution of the Baltic Sea coasts: Integrated assessment of a multi-biomarker approach. Marine Pollution Bulletin. 2006;53(8):508-22.
- 366. Frantzen M, Regoli F, Ambrose Jr WG, Nahrgang J, Geraudie P, Benedetti M, et al. Biological effects of mechanically and chemically dispersed oil on the Icelandic scallop (Chlamys islandica). Ecotoxicology and Environmental Safety. 2016;127:95-107.
- 367. Geraudie P, Bakkemo R, Milinkovitch T, Thomas-Guyon H. First evidence of marine diesel effects on biomarker responses in the Icelandic scallops, Chlamys islandica. Environmental Science and Pollution Research. 2016;23(16):16504-12.
- 368. Carrasco Navarro V, Leppänen MT, Kukkonen JVK, Godoy Olmos S. Trophic transfer of pyrene metabolites between aquatic invertebrates. Environmental Pollution. 2013;173:61-7.
- 369. Vethaak AD, Baggelaar PK, van Lieverloo JHM, Ariese F. Decadal trends in polycyclic aromatic hydrocarbon (PAH) contamination assessed by 1-hydroxypyrene in fish bile fluid in the Netherlands: Declining in marine waters but still a concern in estuaries. Frontiers in Marine Science. 2016 November 11, 2016;3(215).
- 370. Nørregaard RD, Nielsen TG, Møller EF, Strand J, Espersen L, Møll M. Evaluating pyrene toxicity on Arctic key copepod species Calanus hyperboreus. Ecotoxicology. 2014;23(2):163-74.
- 371. Sornom P, Felten V, Médoc V, Sroda S, Rousselle P, Beisel J-N. Effect of gender on physiological and behavioural responses of Gammarus roeseli (Crustacea Amphipoda) to salinity and temperature. Environmental Pollution. 2010;158(5):1288-95.
- 372. Jager T, Salaberria I, Altin D, Nordtug T, Hansen BH. Modelling the dynamics of growth, development and lipid storage in the marine copepod Calanus finmarchicus. Marine Biology. 2017;164(1):1.
- 373. Węsławski JM, Legeżyńska J. Life cycles of some Arctic amphipods. Polish Polar Research. 2002;23(3-4):253-64.







- 374. Legeżyńska J, Kędra M, Walkusz W. When season does not matter: summer and winter trophic ecology of Arctic amphipods. Hydrobiologia. 2012;684(1):189-214.
- 375. Wiedmann I, Reigstad M, Marquardt M, Vader A, Gabrielsen TM. Seasonality of vertical flux and sinking particle characteristics in an ice-free high arctic fjord—Different from subarctic fjords? Journal of Marine Systems. 2016;154(Part B):192-205.
- 376. Galloway TS, Brown RJ, Browne MA, Dissanayake A, Lowe D, Jones MB, et al. Ecosystem management bioindicators: the ECOMAN project a multi-biomarker approach to ecosystem management. Marine Environmental Research. 2004;58(2):233-7.
- 377. Melwani AR, Gregorio D, Jin Y, Stephenson M, Ichikawa G, Siegel E, et al. Mussel watch update: Long-term trends in selected contaminants from coastal California, 1977–2010. Marine Pollution Bulletin. 2014;81(2):291-302.
- 378. Beliaeff B, Burgeot T. Integrated biomarker response: A useful tool for ecological risk assessment. Environmental Toxicology and Chemistry. 2002;21(6):1316-22.













## **Summary**

Expanding oil and gas (O&G) and shipping activities in the absence of sufficient baseline data on the sensitivity of Arctic marine biota to oil related compounds makes environmental risk assessment for the Arctic challenging. Delayed oil weathering and biodegradation, ecological seasonal variations or a delayed manifestation of toxicity in Arctic marine biota could create differences between the susceptibilities of Arctic and temperate monitoring species in the long term. Such differences may be better defined in estuarine or shallow areas, where the frequent resuspension of sediment and strong affinity of oil for sediment organic matter suggest that a large proportion of oil compounds will be transferred to the seafloor, increasing exposure of benthic organisms. Arctic benthos, and in particular sediment-dwelling organisms, can be applied as relevant and sensitive bioindicators of chemical stress and are important food sources for higher trophic levels. In addition, Arctic benthic organisms may benefit from the projected increase in phytoplankton primary production under climate change, strengthening the pelagic-benthic coupling and trophic transfer efficiency within Arctic marine ecosystems. Their intrinsic ecological value calls for the evaluation and validation of oil toxicokinetic parameters in Arctic benthos as part of current oil risk assessment procedures and monitoring programmes. Such knowledge may form an important basis for the selection of relevant Arctic benthic bioindicators of long-term environmental impacts of O&G and shipping activities.

The aim of this thesis is to develop a suite of candidate exposure methods, including bioaccumulation and biochemical markers, for the identification of bioindicators of chemical stress derived from O&G and shipping in Arctic coastal benthic systems. The second aim is to propose, based on experiments with temperate species, the avoidance of oil contaminated sediment by Arctic benthic amphipods as an oil risk assessment tool for the Arctic.

In **Chapter 2**, the applicability of biota-sediment accumulation factors (BSAFs) as chemical exposure metric was evaluated in three representative species of an Arctic coastal benthic system under field conditions. In a preliminary round, target species were selected based on their feeding mode, reduced motility and relative abundance. Two bivalves, the suspension feeder (*Astarte borealis*) and the surface deposit and suspension feeder (*Macoma calcarea*), and one polychaete, the predator and deposit feeder (*Nephtys ciliata*), were selected. These species and corresponding sediment samples were collected at a reference area (Blomstrandhalvøya) and a relatively oil impacted area (Ny-Ålesund) to study the influence of







## **Summary**

local sediment components on the bioaccumulation of polycyclic aromatic hydrocarbons (PAHs), polychlorinated biphenyls (PCBs) and hexachlorobenzene (HCB).

Bioaccumulation differences were detected between both areas for PAHs, with higher 10-90th  $\sum_{13}$ PAH BSAF percentiles (i.e., 0.35-5.50) in Blomstrandhalvøya than in Ny-Ålesund organisms (i.e., 0.00-0.07). Differences in the quality of organic matter in sediment and other food items probably affected the sediment ingestion and thus, PAH uptake by organisms from different areas. However, similar patterns across sampling areas were observed in M. calcarea based on  $\sum_{13}$ PAH body residues, with higher lipid-normalised concentrations in Ny-Ålesund specimens. This pointed at possible differences in toxicokinetic parameters such as PAH uptake and elimination, and higher preference for biogenic organic matter sources for M. calcarea as compared to the other species. It was concluded that PAH body residues provide a more realistic estimate of the bioaccessible PAH fraction since their variability among individuals and environmental conditions was lower than for BSAFs. The study also reflected on the necessity to include black carbon as a major influencing factor of sediment exposure to persistent organic contaminants for Arctic benthos.

In Chapter 3, the relative importance of feeding traits and sediment composition for the bioaccumulation of PAHs by the aforementioned species was evaluated using a dynamic modelling approach. Modelling input data were obtained from a laboratory exposure experiment performed with specimens collected at Blomstrandhalvøya. The experiment followed the design of an earlier exposure experiment developed for temperate counterpart species for comparison purposes. The latter included the surface deposit and suspension feeding bivalve (*Limecola balthica*) and the deposit feeding polychaete (*Alitta virens*) frequently used as test species in bioaccumulation studies. In both experiments, species were exposed to sediment treatments made from the combination of different ratios of a reference and a naturally contaminated sediment collected in a temperate estuary. In this laboratory study, PAH body residues were again selected as the most suitable exposure metric due to considerable BSAF variability in both the control and exposed groups.

In the Arctic experiment, differences in the bioavailability of PAHs between sediment treatments for each taxonomy group possibly explained significantly higher lipid-normalised concentrations observed for 3- and 4-ring PAHs in *A. borealis* compared to *N. ciliata*. However, PAH uptake was limited among bivalve species which was likely caused by the presence of pyrogenic PAH sources in the contaminated sediment, commonly associated with







black carbon emissions from shipping, and low nutritional value of the bivalves reference sediment. The experimental approach used did not reduce the uncertainty in the relative contribution of different PAH sources to body residues observed in **Chapter 2**. It was concluded that future studies should look at the influence of the quality and quantity of organic matter in different sediment PAH sources to help to interpret chemical exposure in Arctic species.

A comparison with the temperate experiment did not yield any significant differences in PAH body residues nor measured BSAFs between the Arctic and temperate counterpart species. Similar exposure routes and the considerable biological variability in the control groups could explain the similarity in the model output between experiments, also reflected by the absence of differences in the model parameters between test species or experiments. Because of limitations on the availability of Arctic species and the variability observed during exposure, the modelling approach did not provide any insightful information on the bioaccumulation mechanisms. Nevertheless, it is expected that such constraints may be overcome after further characterization of PAH uptake and elimination routes in Arctic species under local conditions.

In **Chapter 4**, a suite of biomarkers of PAH exposure (acyl-CoA oxidase, AOX; glutathione S-transferase, GST; pyrene metabolites) and neurotoxicity (acetylcholinesterase, AChE) were analysed in the aforementioned species in order to evaluate the species-specific sensitivity of Arctic species to PAHs in comparison to their temperate counterparts and the general applicability of standard biomarker protocols for predicting sub-lethal ecological effects in Arctic species. Due to limited sample availability and general lack of knowledge of PAH biotransformation pathways among marine invertebrates, enzymatic biomarkers (i.e., AOX, AChE, GST) were analysed in *A. borealis* samples collected from experiments included in **Chapter 3**, whereas pyrene metabolites were analysed in samples from Arctic (*M. calcarea*, *N. ciliata*) and temperate species (*L. balthica*, *A. virens*).

The results of the study showed neither induction or correlation between the activity of AOX or GST in *A. borealis* and PAH contamination levels in sediment treatments, nor evidence of PAH neurotoxicity according to AChE activity. Lack of enzymatic induction may be related to either too low PAH concentrations in sediment or to the influence of confounding factors such as low food availability or changing metabolic status of *A.borealis* during sampling. Pyrene biotransformation rates were limited among Arctic and temperate species as the ratio







## **Summary**

of pyrene-1-glucuronide (Phase II metabolite) to pyrene content did not correlate with pyrene exposure in test organisms. It was concluded that PAH metabolites may be used to estimate PAH exposure as a flux balance between PAH parental and metabolite compounds, although Phase II biotransformation pathways need further characterization in test species.

In Chapter 5, the avoidance response to sediment contaminated with Distillate Marine grade A (DMA) oil of a freshwater (*Gammarus pulex*) and marine (*Gammarus locusta*) model benthic species was evaluated in a 96-h test period and used as the basis for future pilot testing in Arctic species. *G. pulex* avoided contaminated sediment at the highest concentration (1000 mg DMA/kg dry wt.), while *G. locusta* avoidance was either not consistent with exposure time or below effect threshold. Lack of consistency in *G. locusta* avoidance was possibly caused by differences in the energy allocation to growth and reproduction between *G. locusta* life-span stages, affecting their ability to escape from DMA. In contrast, a lower reproductive effort in *G. pulex* winter individuals may have increased their energy storage capacity and thus, energy allocation to avoidance. In addition, differences in DMA behaviour between freshwater and marine tests and PAH partitioning between marine tests likely affected the comparability between tests.

The results of this study highlighted the need for baseline assessments of inter-individual variability in the avoidance response of benthic amphipods as well as the chemical characterization of sediment treatments and overlying water in exposure systems. It was concluded that the combined analysis of energy consumption and spatial avoidance could help to elucidate whether the avoidance ability of amphipods may be affected by their reproductive strategy and how could avoidance be translated into a recognizable pattern of effects on life-history traits of *Gammarus* spp. Future studies should consider as well the influence of reproductive and moulting stage on the sensitivity to oil of benthic amphipods.

The final **Chapter (6)** summarizes and discusses the outcomes of Chapters 2 to 5 in order to select the most appropriate approach to risk identification of O&G and shipping activities in Arctic benthos and to identify research priorities for the validation of candidate methods. Recommendations for the selection and validation of bioindicators and biomarkers of long-term oil spill and O&G environmental effects in Arctic coastal benthic systems included (1) the implementation of more standardized methods for sediment preparation, exposure conditions and acclimatization of field organisms, (2) the collection of baseline data on PAH body residues and biomarker responses in Arctic bioindicator species, (3) the characterization







of quality and quantity of organic matter in all different diet components of exposure treatments, (4) the evaluation of the relative effect of environmental and biological factors on the activity of biomarkers, (5) the design of community assessments to explore the relationship between PAH body residues, sediment characteristics and environmental factors and (6) the inclusion of oil avoidance tests using Arctic benthic amphipods as a candidate tool for oil spill risk assessment for the Arctic. This work contributes to the development of Arctic benthic bioindicators of the cumulative effects of O&G and shipping activities through the

identification of robust PAH exposure metrics in representative benthic species of the Barents



Sea region.









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Y a mi pequeño pero también gran hermano, Kordian, con quien espero que podamos seguir siempre teniendo nuestras conversaciones en familia. ¡Nunca pierdas tu esencia!

How do people imagine the landscapes they find themselves in?

How does the land shape the imaginations of the people who dwell in it?

How does desire itself, the desire to comprehend, shape knowledge?

- Barry Lopez -







## Curriculum vitae

Ariadna S. Szczybelski Ciordia (Zaragoza, Spain, April 2nd 1985) obtained her BSc degree in Biology from the University of Alicante, Spain. During her studies, she was acquainted with several marine biology aspects and completed an Erasmus exchange at the University of Liège (Belgium) where she became interested in the subject of marine ecotoxicology and related research performed therein. In 2010, she decided to enrol on an MSc programme in Environmental Contamination and Toxicology (University of Valencia, Spain). She focused on the use of biochemical biomarkers in freshwater and marine fish species to monitor effects of organic pollution and did an internship at the Institute of Marine Sciences (ICM-CSIC) in Barcelona, Spain. After finishing her MSc degree, she was awarded a Leonardo Da Vinci fund to complete a 6 month work-related training at the Ecotoxicology laboratory of IFREMER (Nantes, France). During her stay she was involved in the analysis of marine fish and shellfish samples from two projects funded by the French National Agency for Research (CONPHYPOP and GIMEPEC) and in the sampling of flatfish species in the Seine estuary (France) as part of the BIOCO 2012 campaign. After finishing her internship, she started a PhD in the Aquatic Ecology and Water Quality Management Group in Wageningen University (Wageningen, the Netherlands). The PhD revised the applicability of exposure assessment methods to oil related compounds in Arctic and temperate benthic macroinvertebrates for the selection of relevant Arctic benthic bioindicators to be used in current environmental risk assessment procedures in the Arctic. During her PhD she cosupervised one BSc student and three MSc students, presented at two international conferences (ARCTIC FRONTIERS and SETAC Europe) and was member of the Dutch National Committee of the Association of Polar Early Career Scientists (APECS) in 2016-2017. She is currently taking her second MSc programme in Marine Biotechnology at the University of Tromsø (Tromsø, Norway) with an eye to studying different strategies for optimizing oil bioremediation in Arctic marine sediments.





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## List of publications

**Szczybelski A.S.**, M.J. van den Heuvel-Greve, A.A. Koelmans, N.W. van den Brink. Submitted. Biomarker responses and biotransformation capacity in Arctic and temperate benthic species exposed to polycyclic aromatic hydrocarbons. Science of the Total Environment.

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Szczybelski, A.S., T. Kampen, J. Vromans, E.T.H.M. Peeters, M.J. van den Heuvel-Greve, N.W. van den Brink, A.A. Koelmans (2018). Avoidance tests as a tool to detect sublethal effects of oil-impacted sediments. Environmental Toxicology and Chemistry 37(6), 1757-1766.

Szczybelski, A.S., M.J. van den Heuvel-Greve, T. Kampen, C. Wang, N.W. van den Brink, A.A. Koelmans (2016). Bioaccumulation of polycyclic aromatic hydrocarbons, polychlorinated biphenyls and hexachlorobenzene by three Arctic benthic species from Kongsfjorden (Svalbard, Norway). Marine Pollution Bulletin 112(1-2), 65-74.

Van den Heuvel-Greve, M.J., **A.S. Szczybelski**, N.W. van den Brink, M.J.J. Kotterman, C.J.A.F. Kwadijk, A. Evenset, A.J. Murk (2016). Low organotin contamination of harbour sediment in Svalbard. Polar Biology 39(10), 1699-1709.

Akcha, F., A. Barranger, E. Bachère, C. Heude Berthelin, D. Piquemal, P. Alonso, R. Rondon Sallan, G. Dimastrogiovanni, C. Porte, D. Menard, A. Szczybelski, A. Benabdelmouna, M. Auffret, J. Rouxel, T. Burgeot (2016). Effects of an environmentally relevant concentration of diuron on oyster genitors during gametogenesis: responses of early molecular and cellular markers and physiological impacts. Environmental Science and Pollution Research 23(8), 8008-8020.







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Wageningen, 5 November 2018

On behalf of the SENSE board

the SENSE Director of Education

Prof. dr. Huub Rijnaarts

Dr. Ad van Dommelen

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The SENSE Research School declares that Ariadna Sabina Szczybelski Ciordia has successfully fulfilled all requirements of the Educational PhD Programme of SENSE with a work load of 34.9 EC, including the following activities:

#### **SENSE PhD Courses**

- Environmental research in context (2013)
- Research in context activity: 'Preparing and organizing of Netherlands / Belgium Symposium for Association of Polar Early Career Scientists, APECS (2 November 2016)'

### Selection of Other PhD and Advanced MSc Courses

- Basic statistics, Wageningen university (2013)
- Introduction to R for statistical analysis, Wageningen University (2014)
- Proficiency in English, Radboud in'to Languages (2014)
- o Molecular toxicology, Postgraduate Education in Toxicology (2016)
- Writing grant proposals, Wageningen University (2016)
- Career assessment, Wageningen University (2016)
- Marine sustainability, University of Tromsø, Norway (2017)
- Toxicogenomics, Postgraduate Education in Toxicology (2017)

## External training at a foreign research institute

Enzymatic determination of a suite of exposure biomarkers, PIE Research Centre, University of the Basque Country, Spain (2015)

## Management and Didactic Skills Training

- Supervising two MSc students with theses entitled 'Avoidance of DMA contaminated sediment by Gammarus pulex' (2014) and 'Bioaccumulation of polycyclic aromatic hydrocarbons (PAHs) and polychlorinated biphenyls (PCBs) in the Arctic Ocean' (2015)
- Teaching in the BSc course 'Environmental Toxicology' (2014-2015)

#### **Oral Presentations**

- ARCIND Development of ARCtic biological INDicators for the impact assessment of (new) human activities in the Arctic ecosystem. NWO Symposium 2014, 5 November 2014, The Hague, The Netherlands
- o Biological indicators for the impact assessment of (new) human activities in the Arctic. SETAC Europe, 7 May 2015, Barcelona, Spain
- o Biological indicators for the impact assessment of (new) human activities in the Arctic. APECS Netherlands Symposium, 6 November 2015, The Hague, The Netherlands

SENSE Coordinator PhD Education

Dr. Peter Vermeulen











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## **Propositions**

- Mechanistic bioaccumulation models for Arctic coastal benthic systems should account for the relative proportion of terrestrial dissolved organic matter and marine dissolved organic matter and its change over time. (this thesis)
- The variation in bioaccumulation by Arctic benthic macroinvertebrates can largely be explained by site-specific differences in diet choice and food composition. (this thesis)
- Community initiatives to promote physical activity and strength training among pre-frail older adults can significantly reduce health care and social assistance spending.
  - J. Sacha, M. Sacha, J. Soboń, Z. Borysiuk and P. Feusette (2017). "Is It Time to Begin a Public Campaign Concerning Frailty and Pre-frailty? A Review Article." Frontiers in Physiology 8(484)
- Geospatial data visualization should be used in decision making about climate change risks to define and communicate extreme events.
- 5. In the era of political micro-targeting and mass surveillance, our digital footprint can obstruct a functioning democracy.
- 6. Closing the gender pay gap begins by introducing a universal basic income.

Propositions belonging to the thesis, entitled

'Advancement of benthic indicators and biomarker-based tools for biomonitoring and risk assessment in the Barents Sea region'

Ariadna S. Szczybelski Ciordia Wageningen, 5 November 2018