

## 1 Designing Polar and Magnetic Oxides: $Zn_2FeTaO_6$ - in Search of 2 Multiferroics

3 Man-Rong Li,<sup>†</sup> Peter W. Stephens,<sup>‡</sup> Maria Retuerto,<sup>†</sup> Tapati Sarkar,<sup>†</sup> Christoph P. Grams,<sup>§</sup>  
4 Joachim Hemberger,<sup>§</sup> Mark C. Croft,<sup>||</sup> David Walker,<sup>⊥</sup> and Martha Greenblatt<sup>\*,†</sup>

5 <sup>†</sup>Department of Chemistry and Chemical Biology, Rutgers, the State University of New Jersey, 610 Taylor Road, Piscataway, New  
6 Jersey 08854, United States

7 <sup>‡</sup>Department of Physics & Astronomy, State University of New York, Stony Brook, New York 11794, United States

8 <sup>§</sup>II. Physikalisches Institut, Universität zu Köln, D 50937 Köln, Germany

9 <sup>||</sup>Department of Physic and Astronomy, Rutgers, the State University of New Jersey, 136 Frelinghusen Road, Piscataway, New Jersey  
10 08854, United States

11 <sup>⊥</sup>Lamont Doherty Earth Observatory, Columbia University, 61 Route 9W, PO Box 1000, Palisades, New York 10964, United States

### 12 **S** Supporting Information

13 **ABSTRACT:** Polar oxides are technically of great interest  
14 but difficult to prepare. Our recent discoveries predicted  
15 that polar oxides can be synthesized in the corundum-  
16 derivative  $A_2BB'O_6$  family with unusually small cations at  
17 the A-site and a  $d^0$  electron configuration ion at B'-site.  
18 When magnetic transition-metal ions are incorporated  
19 more interesting polar magnetic oxides can form. In this  
20 work we experimentally verified this prediction and  
21 prepared  $LiNbO_3$  (LN)-type polar magnetic  $Zn_2FeTaO_6$   
22 via high pressure and temperature synthesis. The crystal  
23 structure analysis indicates highly distorted  $ZnO_6$  and  $(Fe/$   
24  $Ta)O_6$  octahedra, and an estimated spontaneous polar-  
25 ization ( $P_S$ ) of  $\sim 50 \mu C/cm^2$  along the  $c$ -axis was obtained  
26 from point charge model calculations.  $Zn_2Fe^{3+}Ta^{5+}O_6$  has  
27 a lower magnetic transition temperature ( $T_N \sim 22$  K) than  
28 the  $Mn_2FeTaO_6$  analogue but is less conductive. The  
29 dielectric and polarization measurements indicate a  
30 potentially switchable component.

31 **T**he study of polar materials has become an important topic  
32 of materials science due to the fundamental and  
33 technological opportunities arising from their interesting and  
34 useful physical properties,<sup>1,2</sup> including pyroelectric, piezo-  
35 electric, ferroelectric, multiferroic behavior, and second  
36 harmonic generation (SHG) effect. When magnetic ions are  
37 incorporated into polar materials and, in the rare cases, when  
38 strong coupling between magnetization and polarization occurs,  
39 the so-called magnetoelectric effect allows magnetic control of  
40 the electrical polarization and *vice versa* for multiple-state  
41 memory applications. However, materials with both magnetic  
42 and ferroelectric polarization are difficult to prepare because of  
43 the conflicting electronic requirements of these properties,<sup>3,4</sup>  
44 and therefore only a few multiferroic materials are known to  
45 date. So far, perovskites ( $ABO_3$ ) and related oxides are the  
46 most promising and widely studied multiferroic systems. One  
47 strategy to produce ferroelectric polarization in perovskites is to  
48 locate a lone pair electron active cation at the A-site, such as  
49  $Pb^{2+}$  or  $Bi^{3+}$  where the stereochemical effect of the nonbonding

60 lone pair electron drives the electric polarization, as  
61 observed in  $BiFeO_3$ ,<sup>5,6</sup> and  $PbVO_3$ .<sup>7</sup> There are only a few  
62 elements with a lone electron pair, and the presence of such  
63 element is not a guarantee for a polar structure, e.g.,  $BiScO_3$ ,<sup>8</sup>  
64 and  $BiGaO_3$ .<sup>6</sup> A second approach to obtain ferroelectricity is via  
65 structural distortion. Examples are the polar  $LiNbO_3$  (LN)-type  
66 compounds, which contain unusually small A-site cations and  
67 may be considered as highly distorted perovskite-based  
68 structures with small tolerance factors ( $t$ ) and octahedral  
69 coordination for both the A- and B-site cations, such as  
70  $ScFeO_3$ ,<sup>9</sup>  $(In_{1-x}□_x)MO_{3-\delta}$  ( $M = Mn/Fe$ ,  $□ =$  vacancy),<sup>10</sup>  
71  $FeTiO_3$ ,<sup>11</sup> and  $MnMO_3$  ( $M = Ti, Sn$ ).<sup>12</sup> In these compounds  
72 the polarization arises from the structural distortions due to  
73 atomic displacements and octahedral distortions (Section 1,  
74 Supporting Information (SI)). This strategy does not require  
75 the presence of lone pair electron cations and thus provides  
76 more opportunities in the search and design of new polar  
77 materials.

78 We recently expanded the LN-type structure to the  $A_2BB'O_6$   
79 system.<sup>13</sup> For the high pressure and temperature (HPT)  
80 synthesized  $Mn_2^{2+}Fe^{3+}M^{5+}O_6$  ( $M = Nb$ , and  $Ta$ ) the calculated  
81 spontaneous polarizations are 32 and  $23 \mu C \cdot cm^{-1}$  for  $M = Nb$   
82 and  $Ta$ , respectively; magnetic ordering ( $T_N$ ) is observed above  
83 200 K, due to magnetic interactions between the high spin  
84 (HS)  $d^5$  electrons of  $Mn^{2+}$  and  $Fe^{3+}$ . Moreover, significant  
85 pyroelectric response at low temperature and SHG effect was  
86 also observed. First-principle theoretical calculations revealed  
87 that the polarization in  $Mn_2FeMO_6$  is due to the second-order  
88 Jahn–Teller (SOJT) effect of  $d^0$  ions  $Nb^{5+}$  and  $Ta^{5+}$ . These  
89 findings suggested many new LN-type materials in the  
90  $A_2BB'O_6$  system, because of the large variety of ions that the  
91 structure can accommodate and the structural versatility of the  
92 cation arrangements. With these considerations we predicted  
93 possible novel polar compounds in the  $A_2^{2+}B^{3+}B'^{5+}O_6$  and  
94  $A_2^{2+}B^{2+}B'^{6+}O_6$  series, where A is an unusually small cation and  
95 B' is a  $d^0$  ion. In this paper, we present the synthesis at HPT  
96 and characterization of LN-type  $Zn_2FeTaO_6$  and confirm the 86

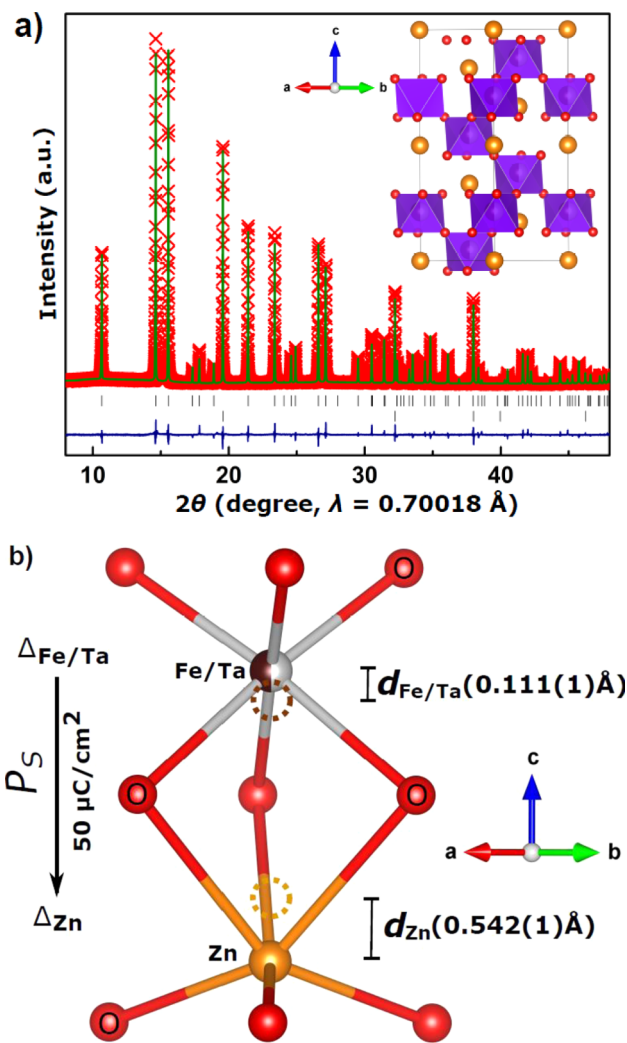
Received: March 18, 2014

87 validity of our prediction. High-resolution synchrotron powder  
88 X-ray diffraction (SPXD) analysis and dielectric and polar-  
89 ization studies were also performed to understand the  
90 experimental data at the microscopic level.

91 The small tolerance factor of  $\text{Zn}_2\text{FeTaO}_6$  (0.836)<sup>14</sup> predicts  
92 unstable perovskite structure under ambient pressure, which  
93 was confirmed by experiments (Figure S2). Subsequent HPT  
94 synthesis at 1623 K under 9 GPa of pure polycrystalline  
95  $\text{Zn}_2\text{FeTaO}_6$  was demonstrated by powder X-ray diffraction  
96 (PXD) analysis (Figure S2; experimental details and the  
97 crystallographic information file (CIF) are provided in SI). The  
98 phase purity and rhombohedral cell dimensions ( $a = 5.1709(2)$   
99 Å,  $c = 13.9353(4)$  Å) of the as-prepared sample were further  
100 confirmed by high-resolution SPXD, which suggested possible  
101 space groups: the centrosymmetric  $R\bar{3}c$  (no. 167, corundum-  
102 type structure) or the noncentrosymmetric  $R3c$  (no. 161, LN-  
103 type structure). Although these space groups have the same  
104 reflection conditions, Rietveld refinements using the SPXD data  
105 yielded poor fit of the corundum structure ( $R_p/R_{wp} = 7.55/$   
106  $10.66\%$ ,  $\chi^2 = 2.52$ ) and confirmed unambiguously the  
107 noncentrosymmetric LN structure ( $R_p/R_{wp} = 5.82/7.72\%$ ,  $\chi^2$   
108  $= 1.32$ , Figure 1). The final refined structural parameters are  
109 listed in Table 1.

110 The LN-type crystal structure of  $\text{Zn}_2\text{FeTaO}_6$  (inset of Figure  
111 1a) contains corner-sharing  $\text{ZnO}_6$  and  $(\text{Fe/Ta})\text{O}_6$  octahedral  
112 sublattices interconnected via face-sharing along the  $c$ -axis and  
113 edge-sharing in the  $ab$ -plane to form face- and edge-sharing  
114 octahedral dimers with heterometal sites (Figure S1). To  
115 overcome the electrostatic repulsions, Zn and disordered Fe/Ta  
116 in the face-sharing octahedral pairs displace away from the  
117 octahedral centroids (dashed circles in Figure 1b) by  $0.542(1)$   
118 ( $d_{\text{Zn}}$ ) and  $0.111(1)$  Å ( $d_{\text{Fe/Ta}}$ ), respectively, in opposite  
119 directions along the  $c$ -axis. These displacements ( $d_{\text{Zn}}$  and  
120  $d_{\text{Fe/Ta}}$ ) generate three short and three long metal–oxygen  
121 bonds in  $(\text{Fe/Ta})\text{O}_6$  (1.951(7) and 2.077(8) Å) and  $\text{ZnO}_6$   
122 (1.997(7) and 2.365(7) Å) octahedra. The octahedral  
123 distortions are reflected by the distortion parameters ( $\Delta$ )<sup>15</sup>  
124 (Table S1). The short and long Fe/Ta–O bond lengths are  
125 comparable with a difference of 0.126 Å, giving  $\Delta_{\text{Fe/Ta}}$  of  $9.8 \times$   
126  $10^{-4}$ , which is comparable with that of the B-site Sn in the LN-  
127 analogue  $\text{ZnSnO}_3$  ( $\Delta_{\text{Sn}} = 5 \times 10^{-4}$ ),<sup>16,17</sup> but smaller than  $\Delta_{\text{Nb}}$   
128 in  $\text{LiNbO}_3$  ( $40 \times 10^{-4}$ ).<sup>18</sup> In contrast, the differences between  
129 the long and short Zn–O bonds are nearly three times larger  
130 (0.368 Å) than those of Fe/Ta–O, which suggest that the  
131  $\text{ZnO}_6$  octahedra are more distorted.  $\Delta_{\text{Zn}}$  is calculated to be  $71.2$   
132  $\times 10^{-4}$ , almost three times that of  $\Delta_{\text{Li}}$  ( $18 \times 10^{-4}$ )<sup>18</sup> in  $\text{LiNbO}_3$   
133 and double that of  $\Delta_{\text{Zn}}$  in  $\text{ZnSnO}_3$ <sup>16</sup> and  $\text{ZnTiO}_3$  ( $38 \times$   
134  $10^{-4}$ ).<sup>19</sup> Accordingly, the structural distortion in  $\text{Zn}_2\text{FeTaO}_6$   
135 results in spontaneous polarization ( $P_s$ ) of  $\sim 50 \mu\text{C}/\text{cm}^2$  along  
136 the  $c$ -axis calculated from the point charge model ( $P_s =$   
137  $\sum_i q(i)\delta d(i)/V$ , where  $q(i)$  is the nominal charge on the  $i^{\text{th}}$   
138 atom,  $\delta d(i)$  is the displacement along the  $c$ -axis of the  $i^{\text{th}}$  atom  
139 from its position in the pseudocentrosymmetric structure, and  
140  $V$  is the unit cell volume).<sup>20</sup> This large polarization is  
141 comparable with the  $P_s$  of  $\text{ZnSnO}_3$  ( $\sim 58 \mu\text{C}/\text{cm}^2$ ) observed  
142 on thin film<sup>21</sup> and somewhat smaller than that of  $\text{LiNbO}_3$  ( $67$   
143  $\mu\text{C}/\text{cm}^2$ ).<sup>22</sup> The X-ray absorption spectra (XAS) of Fe and Ta  
144 indicated formal cation oxidation states of  $\text{Zn}^{2+}\text{Fe}^{3+}\text{Ta}^{5+}\text{O}_6$   
145 (Figure S3), which is consistent with the bond valence sums  
146 calculations (Table S1).

147 The thermal evolution of the magnetic susceptibility ( $\chi_{\text{dc}}$ ) of  
148  $\text{Zn}_2\text{FeTaO}_6$  is shown in Figure 2a. The susceptibility keeps on  
149 increasing as the temperature is decreased but lacks any sharp



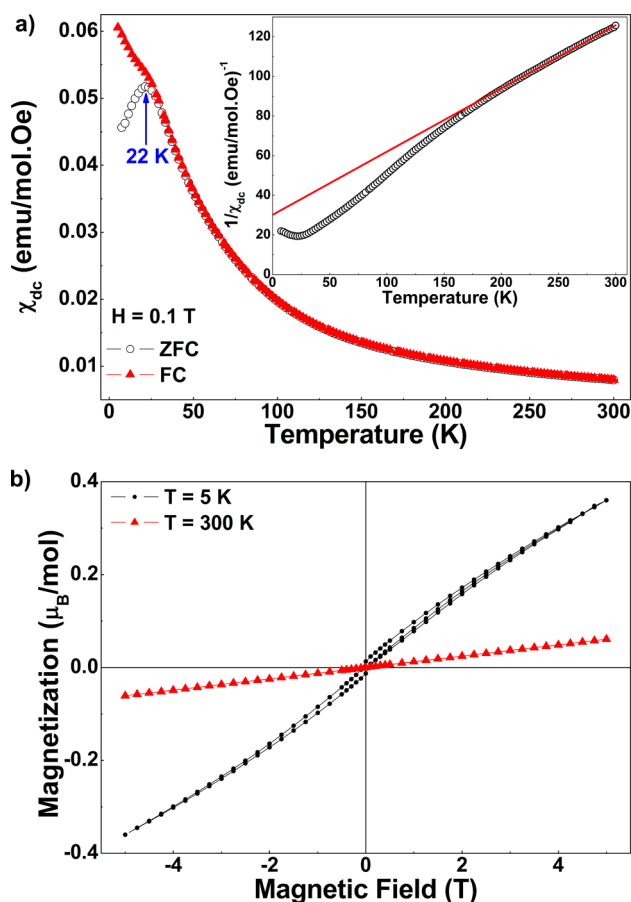
**Figure 1.** (a) Rietveld refinement of the SPXD data for  $\text{Zn}_2\text{FeTaO}_6$ . Red cross represents the observed data, the green line the calculated fit, the deep blue line the difference, upper and lower black tick marks the peak positions of  $\text{Zn}_2\text{FeTaO}_6$  and diamond (internal standard), respectively. Inset shows the crystal structure viewed along  $[110]$  direction. Zn, orange spheres;  $(\text{Fe/Ta})\text{O}_6$  octahedra, purple; O, red spheres. (b) Crystal structure of the face-sharing  $\text{ZnO}_6/(\text{Fe/Ta})\text{O}_6$  octahedral pair. The atomic displacements ( $d_s$ ) away from the  $\text{ZnO}_6$  and  $(\text{Fe/Ta})\text{O}_6$  octahedral site centroids (highlighted by dashed circles) are indicated as  $d_{\text{Zn}}$  (0.542(1) Å) and  $d_{\text{Fe/Ta}}$  (0.111(1) Å), respectively.  $P_s$  is for the spontaneous polarization.

**Table 1. Refined Structural Parameters in  $\text{Zn}_2\text{FeTaO}_6$  from SPXD Data Collected at Room Temperature<sup>a</sup>**

atom	site	$x$	$y$	$z$	$B, \text{Å}^2$ <sup>b</sup>
Zn	6a	0	0	0	0.39(1)
Fe/Ta <sup>c</sup>	6a	0	0	0.2135(1)	0.39(1)
O	18b	0.034(1)	0.334(2)	0.2889(6)	0.39(1)

<sup>a</sup>Rhombohedral, space group  $R3c$  (no. 161),  $a = 5.1709(2)$  Å,  $c = 13.9353(4)$  Å,  $V = 322.68(3)$  Å<sup>3</sup>,  $Z = 3$ ,  $R_{wp} = 7.72\%$ ,  $R_p = 5.82\%$ ,  $\chi^2 = 1.32$ . <sup>b</sup>Isotropic atomic displacement parameters ( $B$ ) were constrained to be the same value. <sup>c</sup>Occupancy of Fe/Ta site slightly deviated from half/half and thus fixed to 0.5/0.5 during the refinements.

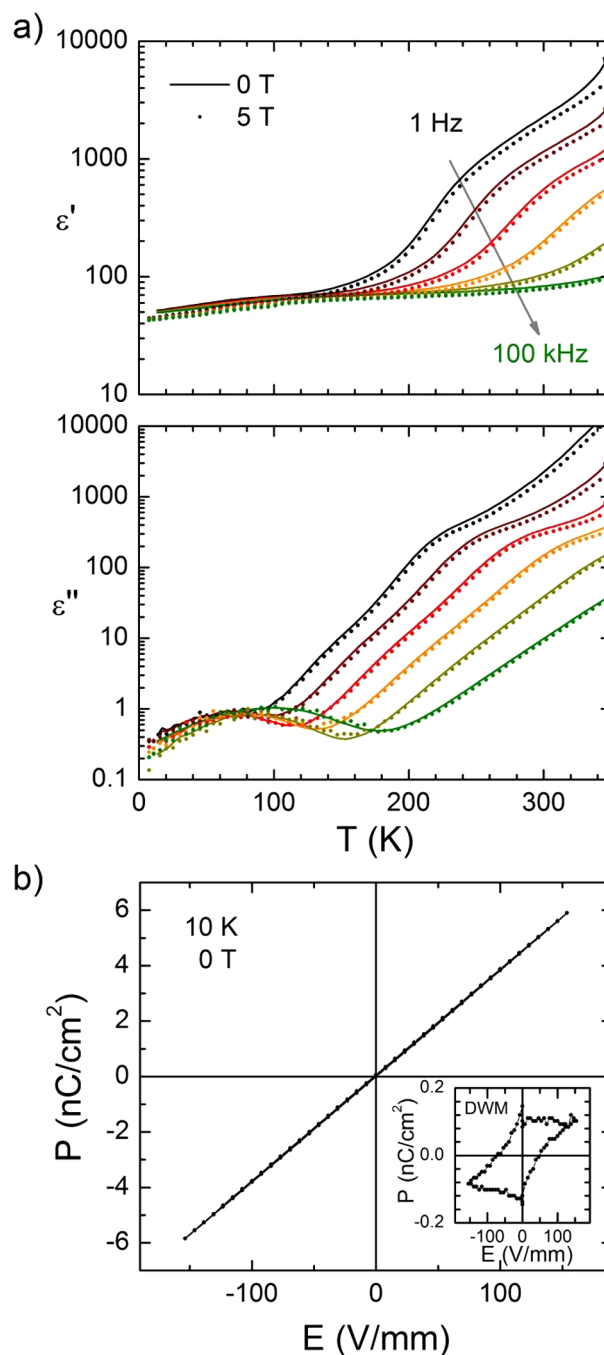
transition, which indicates the absence of proper long-range 150  
order. At low temperature ( $T \sim 22$  K), the zero field cooled 151  
(ZFC) curve shows a cusp-like feature (marked by the blue 152



**Figure 2.** Magnetic behavior of  $\text{Zn}_2\text{FeTaO}_6$ . (a) Temperature dependence of the ZFC and FC dc susceptibility ( $\chi_{\text{dc}}$ ) at 1000 Oe, inset shows the susceptibility inverse ( $1/\chi_{\text{dc}}$ ) vs temperature plot. (b) Isothermal magnetization measured at 5 and 300 K, respectively.

153 arrow in Figure 2a). This cusp-like feature, along with the fact  
 154 that the ZFC and field cooled (FC) curves diverge below 22 K,  
 155 is reminiscent of systems undergoing spin glass-like transitions  
 156 and may well suggest the presence of magnetic frustration in  
 157 this system, with competing short-range ferromagnetic (FM)  
 158 and antiferromagnetic (AFM) interactions. Although the  
 159 sample exhibits no clear long-range magnetic transition, the  
 160 inverse susceptibility data ( $1/\chi_{\text{dc}}$ ) reveal a deviation from strict  
 161 linearity at temperatures below 200 K (inset of Figure 2a).  
 162 Hence we fit the data above 200 K to the Curie–Weiss law  $\chi =$   
 163  $C/T - \theta_{\text{CW}}$  (red line in the inset of Figure 2a). The fitting  
 164 allowed us to extract the value of the effective magnetic  
 165 moment  $\mu_{\text{eff}} = 4.99 \mu_{\text{B}}$ . This is slightly less than the calculated  
 166 spin only moment per formula unit ( $\mu_{\text{cal}} = 5.92 \mu_{\text{B}}$ ) of  $\text{Fe}^{3+}$ ,  
 167 probably due to the disordering of the Fe and Ta ions. The  
 168 Curie–Weiss constant ( $\theta_{\text{CW}} = -94 \text{ K}$ ) indicates the presence of  
 169 dominant AFM interactions. Figure 2b presents the isothermal  
 170 magnetization ( $M$ ) vs  $H$  curves of  $\text{Zn}_2\text{FeTaO}_6$  recorded at 5  
 171 and 300 K, respectively. At 300 K, the sample is in the  
 172 paramagnetic state, while at 5 K, it shows a small hysteresis  
 173 with a coercive field  $H_{\text{C}} \sim 0.07 \text{ T}$ . This might be indicative of the  
 174 presence of some FM interactions in the system at low  
 175 temperature. However, even at this temperature the  $M$ – $H$   
 176 curve shows no sign of saturation, which indicates the presence  
 177 of competing AFM interactions that might lie at the root of the  
 178 magnetic frustration.

179 Compared with the isostructural  $\text{Mn}_2\text{FeTaO}_6$ ,<sup>13</sup> the replace-  
 180 ment of the magnetic  $\text{Mn}^{2+}$  by the diamagnetic  $\text{Zn}^{2+}$  dilutes the  
 181 magnetic interactions in the system and increases the electric  
 182 resistance as reflected by the dielectric measurements (Figure  
 183 3a,b). The dielectric behavior of  $\text{Zn}_2\text{FeTaO}_6$  is similar to that of  
 184  $\text{Mn}_2\text{FeTaO}_6$ <sup>13</sup> and other analogues,<sup>23</sup> showing a high- $\epsilon$  contact  
 185 feature near room temperature and no sign of (switchable)



**Figure 3.** Dielectric properties of  $\text{Zn}_2\text{FeTaO}_6$ . (a) Temperature dependence of the dielectric function ( $\epsilon$ ) is dominated by a contact feature that masks the intrinsic value of  $\epsilon$  for temperatures higher than 100 K. Only at low temperatures the intrinsic value  $\epsilon \approx 50$  can be measured. (b) Electric-field-dependent polarization measurements at 10 K in zero magnetic field show a nearly flat “loop” and also the result of the DWM<sup>24</sup> gives only very small values for the switchable polarization as shown in the inset.

186 ferroelectricity. The polarization-electric field ( $P(E)$ ) “loop” at  
187 low temperature is nearly completely flat (as in paraelectrics in  
188 Figure 3b), the inset of Figure 3b denotes the difference  
189 between the first and second half-cycle of the double wave  
190 method (DWM)<sup>24</sup> and gives very small values, which can be  
191 understood as experimental resolution giving a lower boundary  
192 for a possible switchable component in the multidomain  
193 polycrystalline sample. The small differences in the temper-  
194 ature-dependent permittivity for zero field and 5 T data are  
195 probably due to a small magnetoresistive contribution altering  
196 the effective RC-element of the contacts/grain boundaries.  
197 Further exploration of ferroelectric transition in monodomain  
198 thin films or single crystal samples of  $Zn_2FeTaO_6$  is planned  
199 similar to those of  $ZnSnO_3$ , where not the bulk, only the film  
200 sample showed switchable polarization.<sup>21</sup>  
201 In conclusion, the successful synthesis and study of the novel  
202 polar and magnetic  $LiNbO_3$ -type  $Zn_2FeTaO_6$  in this work  
203 further confirmed the theoretical prediction for polar structures  
204 in the  $A^{2+}B^{3+}B'^{5+}O_6$  corundum-based family with a  $d^0$  electron  
205 configuration ion at the  $B'$ -site. The designed polar structure  
206 and multifunctional properties in  $Zn_2FeTaO_6$  suggest strong  
207 potential in search of multifunctional materials in  $A_2BB'O_6$   
208 phases, especially ferroelectric materials with  $Al^{3+}$ ,  $Ga^{3+}$ , or  $Sc^{3+}$   
209 at the B-site and multiferroic materials with transition metals at  
210 both the B- and  $B'$ -sites and  $Zn^{2+}$  or  $Mg^{2+}$  at the A-site.

## 211 ■ ASSOCIATED CONTENT

### 212 ● Supporting Information

213 Comparison of the perovskite and  $LiNbO_3$  type crystal  
214 structures, details of high and ambient pressure syntheses,  
215 powder synchrotron X-ray diffraction studies and crystallo-  
216 graphic data, X-ray absorption spectroscopy analysis, magnetic,  
217 dielectric, and ferroelectric measurements. This material is  
218 available free of charge via the Internet at <http://pubs.acs.org>.

## 219 ■ AUTHOR INFORMATION

### 220 Corresponding Author

221 [martha@rutchem.rutgers.edu](mailto:martha@rutchem.rutgers.edu)

### 222 Notes

223 The authors declare no competing financial interest.

## 224 ■ ACKNOWLEDGMENTS

225 This work was supported by the DOD-VV911NF-12-1-0172  
226 (ARO-434603) grant and Rutgers University (Board of  
227 Governor Professor Grant). Use of the National Synchrotron  
228 Light Source, Brookhaven National Laboratory was supported  
229 by the DOE BES (DE-AC02-98CH10886). The authors would  
230 like to thank Ms. Jean Hanley at Lamont-Doherty Earth  
231 Observatory in Columbia University for making the high-  
232 pressure assemblies.

## 233 ■ REFERENCES

- 234 (1) Ok, K. M.; Chi, E. O.; Halasyamani, P. S. *Chem. Soc. Rev.* **2006**,  
235 35, 710.  
236 (2) Rao, C. N. R.; Sundaresan, A.; Saha, R. *J. Phys. Chem. Lett.* **2012**,  
237 3, 2237.  
238 (3) Hill, N. A. *J. Phys. Chem. B* **2000**, 104, 6694.  
239 (4) Donakowski, M. D.; Gautier, R.; Yeon, J.; Moore, D. T.; Nino, J.  
240 C.; Halasyamani, P. S.; Poeppelmeier, K. R. *J. Am. Chem. Soc.* **2012**,  
241 134, 7679.  
242 (5) Wang, J.; Neaton, J. B.; Zheng, H.; Nagarajan, V.; Ogale, S. B.;  
243 Liu, B.; Viehland, D.; Vaithyanathan, V.; Schlom, D. G.; Waghmare, U.

- V.; Spaldin, N. A.; Rabe, K. M.; Wuttig, M.; Ramesh, R. *Science* **2003**,  
299, 1719. 244  
(6) Belik, A. A. *J. Solid State Chem.* **2012**, 195, 32. 245  
(7) Singh, D. J. *Phys. Rev. B* **2006**, 73, 094102. 246  
(8) Belik, A. A.; Iikubo, S.; Kodama, K.; Igawa, N.; Shamoto, S.-i.; 247  
Maie, M.; Nagai, T.; Matsui, Y.; Stefanovich, S. Y.; Lazoryak, B. I.; 248  
Takayama-Muromachi, E. *J. Am. Chem. Soc.* **2006**, 128, 706. 249  
(9) Li, M.-R.; Adem, U.; McMitchell, S. R. C.; Xu, Z.; Thomas, C. I.; 250  
Warren, J. E.; Giap, D. V.; Niu, H.; Wan, X.; Palgrave, R. G.; 251  
Schiffmann, F.; Cora, F.; Slater, B.; Burnett, T. L.; Cain, M. G.; 252  
Abakumov, A. M.; van Tendeloo, G.; Thomas, M. F.; Rosseinsky, M. 253  
J.; Claridge, J. B. *J. Am. Chem. Soc.* **2012**, 134, 3737. 254  
(10) Belik, A.; Furubayashi, T.; Matsushita, Y.; Tanaka, M.; Hishita, 255  
S.; Takayama-Muromachi, E. *Angew. Chem., Int. Ed.* **2009**, 48, 6117. 256  
(11) Varga, T.; Kumar, A.; Vlahos, E.; Denev, S.; Park, M.; Hong, S.; 257  
Sanehira, T.; Wang, Y.; Fennie, C. J.; Streiffer, S. K.; Ke, X.; Schiffer, 258  
P.; Gopalan, V.; Mitchell, J. F. *Phys. Rev. Lett.* **2009**, 103, 047601. 259  
(12) Aimi, A.; Katsumata, T.; Mori, D.; Fu, D.; Itoh, M.; Kyômen, T.; 260  
Hiraki, K.-i.; Takahashi, T.; Inaguma, Y. *Inorg. Chem.* **2011**, 50, 6392. 261  
(13) Li, M.-R.; Walker, D.; Retuerto, M.; Sarkar, T.; Hadermann, J.; 262  
Stephens, P. W.; Croft, M.; Ignatov, A.; Grams, C. P.; Hemberger, J.; 263  
Nowik, I.; Halasyamani, P. S.; Tran, T. T.; Mukherjee, S.; Dasgupta, T. 264  
S.; Greenblatt, M. *Angew. Chem., Int. Ed.* **2013**, 52, 8406. 265  
(14) Lufaso, M. W.; Woodward, P. M. *Acta Crystallogr., Sect. B* **2001**, 266  
57, 725. 267  
(15) Brown, I. D.; Shannon, R. D. *Acta Crystallogr., Sect. A* **1973**, 29, 268  
266. 269  
(16) Inaguma, Y.; Yoshida, M.; Katsumata, T. *J. Am. Chem. Soc.* **2008**, 270  
130, 6704. 271  
(17) Hoel, C. A.; Amores, J. M. G.; Morán, E.; Álvaro-Franco, M. A.; 272  
Gaillard, J.-F.; Poeppelmeier, K. R. *J. Am. Chem. Soc.* **2010**, 132, 16479. 273  
(18) Hsu, R.; Maslen, E. N.; Boulay, D. d.; Ishizawa, N. *Acta* 274  
*Crystallogr., Sect. B* **1997**, 53, 420. 275  
(19) Inaguma, Y.; Aimi, A.; Shirako, Y.; Sakurai, D.; Mori, D.; 276  
Kojitani, H.; Akaogi, M.; Nakayama, M. *J. Am. Chem. Soc.* **2014**, 136, 277  
2748. 278  
(20) Resta, R.; Vanderbilt, D. *Theory of Polarization: A Modern* 280  
*Approach*; Springer: Berlin, Heidelberg, 2007, 105, 31. 281  
(21) Son, J. Y.; Lee, G.; Jo, M.-H.; Kim, H.; Jang, H. M.; Shin, Y.-H. *J.* 282  
*Am. Chem. Soc.* **2009**, 131, 8386. 283  
(22) Hsu, R.; Maslen, E. N.; du Boulay, D.; Ishizawa, N. *Acta* 284  
*Crystallogr., Sect. B* **1997**, 53, 420. 285  
(23) Inaguma, Y.; Sakurai, D.; Aimi, A.; Yoshida, M.; Katsumata, T.; 286  
Mori, D.; Yeon, J.; Halasyamani, P. S. *J. Solid State Chem.* **2012**, 195, 287  
115. 288  
(24) Fukunaga, M.; Noda, Y. *J. Phys. Soc. Jpn.* **2008**, 77, 064706. 289