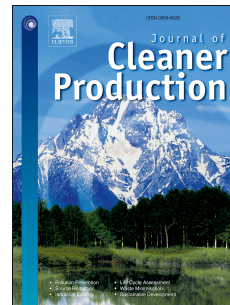


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Sustainable Aluminium Recycling of End-of-Life Products: A Joining Techniques Perspective

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Abstract

The sustainable management of aluminium has become crucial due to the exponential growth in global demand. The transition to a sustainable society with lightweight electric vehicles has led to the increasing use of aluminium in the transportation sector. This has consequently led to the importance of aluminium recycling to prevent the valuable material stream going to landfill. In addition, the extraction of primary aluminium has high environmental impact due to the high energy consumption and waste generation in comparison to secondary aluminium processing. Despite being one of the most recycled metals, ongoing trends of multi-material designs and the associated joining choices have caused increasing difficulty of separating aluminium with high purity.

This paper evaluates the types of joining techniques causing impurities in the aluminium streams, and the relationship between particle size reduction and the presence of impurities due to joints particularly for end-of-life vehicles. An empirical experiment in a leading European recycling facility was conducted and demonstrated that mechanical fasteners, such as machine screws, socket screws, bolt screws and rivets, are the major types of joining technique causing impurities. Based on the observations from this case study, the characteristics of imperfectly liberated joints are examined. A Life Cycle Assessment (LCA) is also performed to evaluate the environmental impact of recycling different aluminium scrap qualities with varying impurity levels. The outcomes are then used to

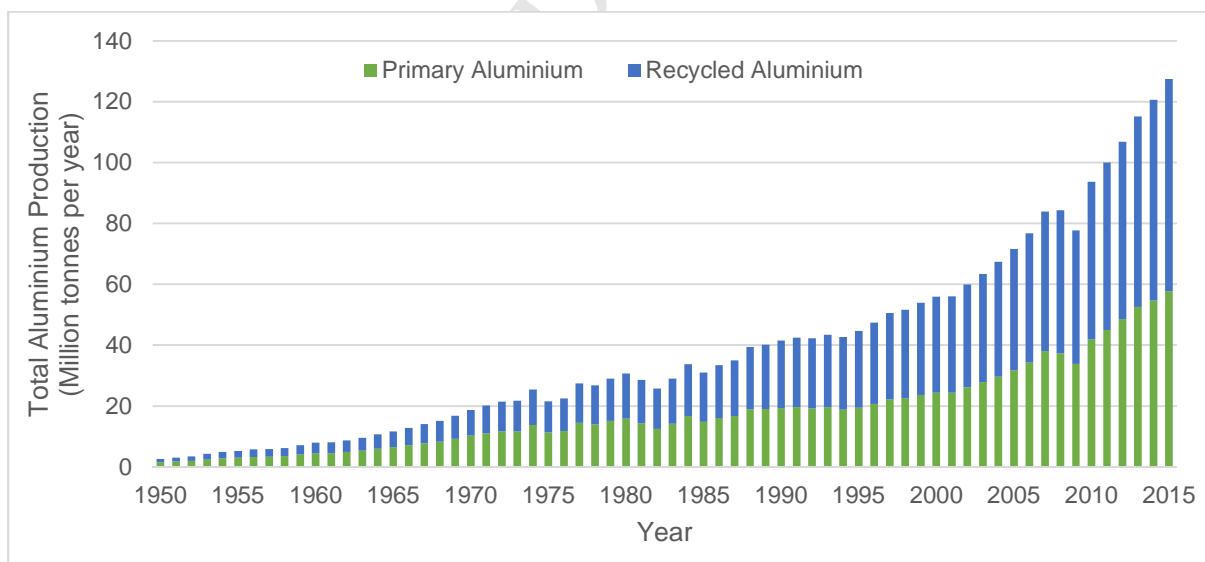
27 provide ecodesign guidelines aimed at improving the quality and increase the quantity of recycled
 28 aluminium.

29 **Keywords**

30 Joining technologies; AI recycling; End-of-Life; recycling efficiency; ecodesign; Life Cycle Assessment

31 **1 Introduction**

32 Aluminium (Al) is used in a variety of products due to its high strength-to-weight ratio, good
 33 formability, and high corrosion resistance. The global demand for Al has seen significant growth,
 34 leading to the importance of sustainable metal management. The amount of Al used globally has
 35 been increasing since 1950, as can be seen in Figure 1, and this trend is projected to continue
 36 (Cullen and Allwood, 2013; Martchek, 2006). One of the major concerns is the continuous energy-
 37 intensive extraction of primary Al to supply for the growing demand worldwide. This activity has
 38 contributed significantly to the global carbon dioxide emissions (Norgate et al., 2007). Although Al is
 39 one of the highly recycled metals, offering significant energy saving during secondary production, the
 40 benefits of Al recycling are influenced by the purity level of scrap sources (Liu and Müller, 2012).



41

42 Figure 1: Amount of primary and recycled Al used globally (International Aluminium Institute, 2017).

43

44 The transportation industry is one of the major consumers of Al worldwide and is responsible for
45 35-40% of the overall Al consumption (Nappi, 2013). In recent years, the focus on producing
46 lightweight vehicles has led to the increasing use of high purity Al in vehicle design to replace
47 conventional steels (Goede et al., 2008). Multi-material design concepts have been progressively
48 adopted by vehicle manufacturers due to the emphasis on reducing vehicle mass, thereby lowering
49 the vehicle carbon footprint (Cui et al., 2011; Miller et al., 2000). Al or Al alloys are among the most
50 suitable material category candidates for the manufacture of multi-material car bodies for automotive
51 applications such as Body-in-White (BIW), chassis components, doors closure and outer panels
52 (Carle and Blount, 1999; Hirsch, 2011; Volkswagen Group, 2009).

53 **1.1 Joining Trends in Multi-Material Products – An Example using Vehicle Design**

54 The increasing use of multi-material products focussing on mass reduction has led to changing
55 joining trends (Groche et al., 2014; Mori et al., 2013; Soo et al., 2015). For example, the material
56 composition in newer vehicle designs has undergone significant transformation particularly with the
57 use of light metals, such as Al (Barnes and Pashby, 2000a; Carle and Blount, 1999; Miller et al.,
58 2000), and lightweight materials, such as plastics and composites (U.S Department of Energy, 2013).
59 As a result, the choice of joining techniques that are feasible to combine these multi-material
60 combinations is limited (Meschut et al., 2014). Although there are several ongoing developments in
61 more advanced joining technologies, their application in large-scale production is still restricted due to
62 the proven design requirements in this risk averse sector (Barnes and Pashby, 2000a, 2000b), and
63 the additional manufacturing costs of installing new equipment (Davies, 2012). Table 1 shows that the
64 Al-intensive vehicle spaceframe structures (Audi A6 and A8) have increased the use of some joining
65 techniques, while reducing others. The large amount of wrought Al used for these luxury vehicles is
66 expected to be adopted also in electric vehicle and mass-optimised vehicle designs. Hatayama et al.
67 (2012) have predicted the increasing demand of wrought Al to produce the power-supply box in
68 electric vehicles. The growing Al composition in vehicle designs has led to the increasing use of
69 mechanical joining techniques, such as screws and rivets, and adhesive bonding. In contrast,
70 traditional welding techniques (e.g. spot welding and MIG welding) are showing a decreasing trend.
71 The observed joining trends are also supported by the vehicle manufacturers' viewpoint on the

72 development of joining processes used for future large scale vehicle production (Grote and
73 Antonsson, 2009).

74 Table 1: Joining trends of different vehicle models' spaceframe structure (Adapted from European Aluminium Association,
75 2013; Mirdamadi and Korchnak, n.d.).

Joint type	Audi A6		Trend	Audi A8		Trend
	2001-2004	2005-2008		1994-2002	2009-present	
<i>Share of point joints (%)</i>						
Spot welding	91.5	81.0	↓	28.1	7.5	↓
Stud welding	3.3	6.5	↑	0	0	
Clinching	0.9	1.3		10.0	0	↓
Screw joints	0	0		0	23.6	↑
Rivets	0	5.8	↑	61.9	68.9	↑
<i>Share of linear joints (%)</i>						
Laser welding	8.3	3.3	↓	0	8	↑
MIG welding	6	4.3	↓	100	33.3	↓
Laser brazing	0	3.1	↑	0	0	
Adhesive bonding	85.7	89.3	↑	0	58.7	↑

76

77 Perfect material separation during the end-of-life (EoL) phase is not possible in the shredder-
78 based recycling practices due to the complex product designs and the difficulty in separating different
79 material types from their associated joining techniques (Van Schaik and Reuter, 2007). Castro et al.
80 (2005) and Van Schaik and Reuter (2007) have shown that the increasing complexity in vehicle
81 designs further hinders perfect liberation of dissimilar materials. As a result, lower grades and
82 qualities of recyclates will be retrieved due to the presence of impurities that lead to cascade recycling
83 (Paraskevas et al., 2015) and the loss of valuable material streams. This is particularly the case for
84 recycling Al scrap that has more limitations during metallurgical recycling in comparison to other
85 metals such as iron and copper (Nakajima et al., 2010). One of the main reasons is the relatively low
86 melting point of Al, which makes it difficult to remove impurities or tramp elements—contaminants
87 affecting the quality of metals that are not added on purpose—during the secondary Al smelting and
88 refining processes. The most common strategies used to address this challenge are either dilution
89 using primary Al or down-cycling to lower grade Al alloys that are associated to additional

90 environmental burden (Castro et al., 2004; Paraskevas et al., 2015). The ability to retrieve high quality
91 Al with low impurities increases the scrap value for recyclers; however, the extra recycling costs need
92 to be justified by the volume of different scrap qualities.

93 **1.2 Generic Design Guidelines for Joining Choices**

94 Most of the design rules specific to joining choices facilitate Design for Disassembly (DfD) for part
95 repair or material reuse that may not cater well for destructive recycling (Ferrão and Amaral, 2006;
96 Güngör, 2006; Kriwet et al., 1995; Reuter, 2011). Screwing, for example, is preferred over adhesive
97 bonding to ease the maintenance and repair of highly complex products, as well as part reuse.
98 However, most EoL products are put through a shredding process in industrialised regions.
99 Consequently, screws may not be separated well from the base materials, particularly when they are
100 used to combine different material types (Castro et al., 2005; Van Schaik and Reuter, 2007), whereas
101 weak adhesive bonding might be easily released in a shredding process. With the increasing
102 complexity of multi-material structures in products, there is a need to consider the implications of both
103 product design and the choice of joining elements on the quality of recycled materials (Reuter, 2011).

104 There are a few ecodesign guidelines that provide advice on joint selection for destructive
105 recycling, such as the “Ten Golden Rules” (Luttrupp and Lagerstedt, 2006) and VDI 2243 design
106 guidelines (VDI 2243, 1993). In spite of that, the details on joint selection are very generic. According
107 to the ‘golden rule number 10’, joining elements should be minimal, and the use of screws, adhesives,
108 welding, snap fits and geometric locking should be appropriate for different life cycle scenarios. In
109 contrast, VDI 2243 and recycling guidelines by Bras (2005) differentiate the guidelines between
110 disassembly and destructive recycling to assist designers in choosing the most appropriate product
111 design based on the fastening principles.

112 The choice of joining techniques during the design phase is gaining prominence for the
113 sustainability of high purity metal recycling due to the current recycling practices (Castro et al., 2005;
114 Van Schaik and Reuter, 2007; Worrell and Reuter, 2014). Studies on the limitations of Al recycling are
115 mostly focussed on the efficiency of current sorting and separation processes (Froelich et al., 2007;
116 Gaustad et al., 2012) and the challenges during the metallurgical recycling phase (Nakajima et al.,
117 2010; Paraskevas et al., 2015; Reck and Graedel, 2012). There is a lack of understanding of the

118 influence of joining choices during initial product design on the quality of recycled Al that is retrieved
119 from highly complex products.

120 This work investigates the types of joining techniques causing impurities in the Al streams when
121 advanced recycling technologies are applied. The study is based on an industrial trial carried out in a
122 leading recycler located in Belgium. Impurities due to joints are identified to understand to what extent
123 they are affecting the collected Al streams. The observations are then expanded to assess the
124 correlation between the presence of impurities due to specific joint types, and the different particle
125 liberation sizes. Based on the case study data, a Life Cycle Assessment (LCA) is performed to
126 evaluate the environmental impact of recycling different Al scrap qualities. This study assists
127 manufacturers and designers to promote closed-loop recycling by mitigating the source of impurities
128 through effective joining technology selection during the initial design stage that caters for current
129 recycling practices. In addition, recyclers and policy-makers can target effective recycling processes
130 and standards to ensure perfectly liberated joints for high purity Al to minimise the loss of valuable
131 material streams.

132 **2 Materials and Methods**

133 This study analyses the cause of impurities present in the different Al output fractions sampled
134 from a Belgium recycling facility. Section 2.1-2.3 describe the Al recycling processes and sampling
135 procedures. To assess to what extent the impurities are affecting the environmental impact of
136 recycling different Al scrap qualities, the LCA method is used, as detailed in Section 2.4.

137 **2.1 Recovery of Aluminium in Belgium**

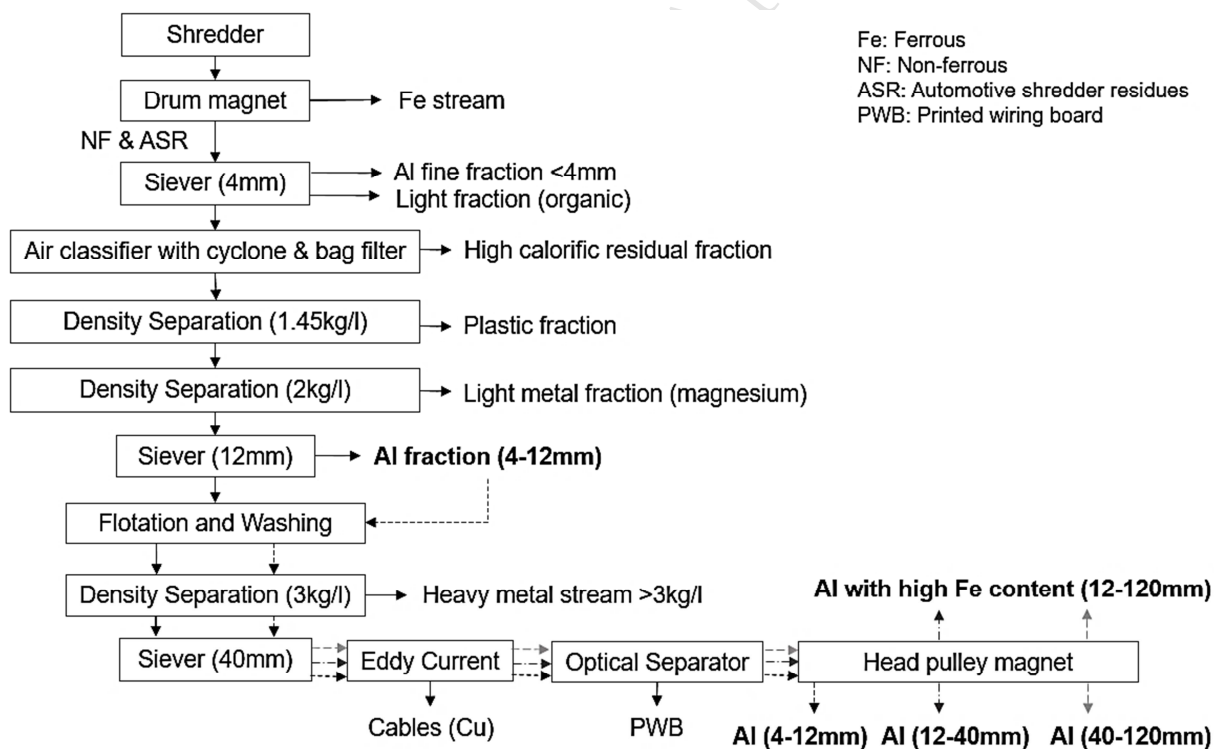
138 The types of scrap sources as considered in the studied Belgian recycling facility, one of the
139 leading recyclers in Europe, are shown in Table 2. The high content of Al in different scrap sources
140 has made it one of the most intensely recycled metals besides steel. Most of the Al scrap is
141 contributed by the end-of-life vehicle (ELV) and household waste streams. The Al content in the ELV
142 and household waste accounts for 4.9wt.% (Muchová and Eder, 2010; RDC Environment, 2015) and
143 4.7wt.% (Muchová and Eder, 2010) respectively. The Al content in demolition and building scrap is
144 relatively low, less than 1wt.% (Muchová and Eder, 2010).

145 Table 2: The sources of the Al containing scrap as recycled in the Belgian recycling facility.

Scrap sources	Relative share of total scrap stream (%)
Depolluted vehicle hulks (ELV)	30
Demolition scrap	30
Household waste	20
Building scrap	20

146

147 The material process flow specific to Al is shown in Figure 2. The processes involved in Al
 148 recycling can be categorised into three main clusters: Al sorting, refinement of sorted aluminium, and
 149 particle size sorting. For Al sorting, density separation is the first step to retrieve Al from the mixture of
 150 scrap. Subsequently, other major processes, such as eddy current separator, optical separator, and
 151 head pulley magnet, are used to further separate Al from other material types.



152

153 Figure 2: Al material flow in the Belgian recycling facility.

154

155 The sorting process targeting Al begins with the density separation after the shredding and
156 magnetic separation. Density separation sorts different materials based on their material densities. It
157 typically starts with separating lighter material fractions (e.g. plastics, foam, rubber, etc.) and is
158 followed by the sorting of materials with higher density. Through density separation at 3kg/l, Al alloys
159 float and can be separated from materials with higher density, such as copper, zinc, and other heavy
160 metals, that sink to the bottom. In some other recycling facilities, Al retrieval through eddy current
161 separation is carried out (Gaustad et al., 2012).

162 An air classifier is used to remove fine shredder residues targeted for energy recovery before the
163 density separation. This allows the fine mixture of dust, metals, glass, and polymers to be removed
164 before the first density separation for lighter material fraction. Other separation techniques, such as
165 sievers, are also used. The material flow is sorted to different particle sizes based on the siever sizes
166 used at various screening stages. This is a common practice in the recycling industry in Europe to
167 segregate different material grades based on the particle sizes (Cui and Forsberg, 2003).

168 An eddy current separator, optical separator and head pulley magnet are used to further sort
169 unwanted materials that are still present in the Al flow. The remaining cable wires that did not sink
170 during earlier density separation are further sorted using the eddy current separator. Through this
171 process, an electrical current is induced within the conductive metal flow, and all metals are repelled
172 through the rotor that produces an external magnetic field. Since Al and copper have a different
173 conductivity, and thus produce varying eddy currents, they are ejected to different distances from the
174 rotor. An optical separator is then utilised to further sort the commonly green coloured printed wiring
175 board (PWB) from the grey coloured aluminium. To further remove small particles with ferrous content
176 from the Al flow, a very strong head pulley magnet with a deeper magnetic field is used.

177 **2.2 Sampling Method**

178 The different Al fraction categories recovered from the facility are shown in Table 3. These
179 categories were chosen for sampling to understand the effect of particle sizes on the purity level of
180 various Al fractions, and the extent of impurities due to joints in the different particle sizes. Sampling
181 was also carried out for Al with high Fe content. The collection of a minimum of 10 samples from each
182 Al fraction was performed in accordance with the field sampling guidance for shredded scrap by the

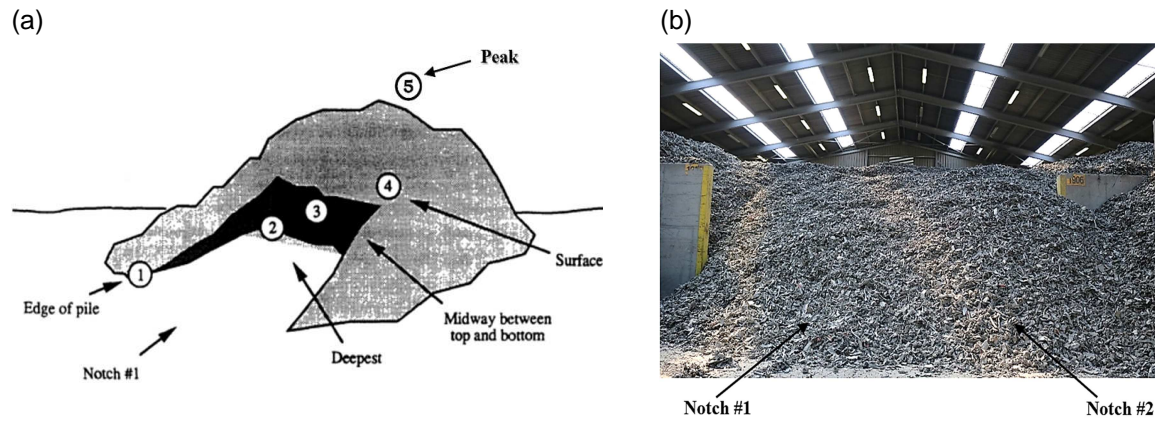
183 United States Environmental Protection Agency (USEPA, 1993). The field sampling guidance
 184 provides information on different sampling methods, estimated sample size, and the statistical
 185 analysis methods to accurately approximate the impurity level of different Al fractions. These
 186 guidelines were based on previous case studies carried out at different shredder sites.

187 Table 3: Amount of Al samples taken from each category, and the generated annual amount in the Belgian recycling facility.

Category	Particle size class (mm)	Number of samples	Mass range of each sample (kg)	Overall sample mass (kg)	Annual amount (ton)
Al with high Fe content	12-120	10	2.685-3.737	32.689	644
Al fraction	40-120	20	2.290-3.896	61.363	6132
Al fraction	12-40	10	1.506-2.408	19.210	4147
Al fraction	4-12	10	1.494-1.947	16.662	1114

188

189 There are different field sampling methods for shredded metal scraps on-site based on the
 190 guidelines by USEPA. Stockpile sampling, as explained in Figure 3, was chosen in this case study to
 191 obtain a more representative sample of the normal shredder output (USEPA, 1993). Al samples were
 192 taken from the Al stockpile warehouse where different qualities and particle sizes were stored
 193 separately. The bucket used to collect the samples has a diameter of 27.5cm with a height of 22.5cm.
 194 Each sample taken only filled up half the bucket. First, Al samples were collected at the edge of pile
 195 (location 1) at notch 1 and notch 2. The two notches were then dug to equal depth with the help of a
 196 front-loader truck. Finally, samples were gathered at locations 2 to 5 for notch 1 and 2. In total, there
 197 were 10 buckets of samples collected for each targeted Al output stream. 20 samples were taken only
 198 for the Al fraction 40-120mm to ensure a good representation of the stockpile, since it is the largest
 199 fraction produced in the facility.



200 Figure 3: Stockpile sampling of different Al fractions (a) Sampling location for each Al stockpile; (b) Location of notches made
 201 for each Al stockpile to carry out sampling.

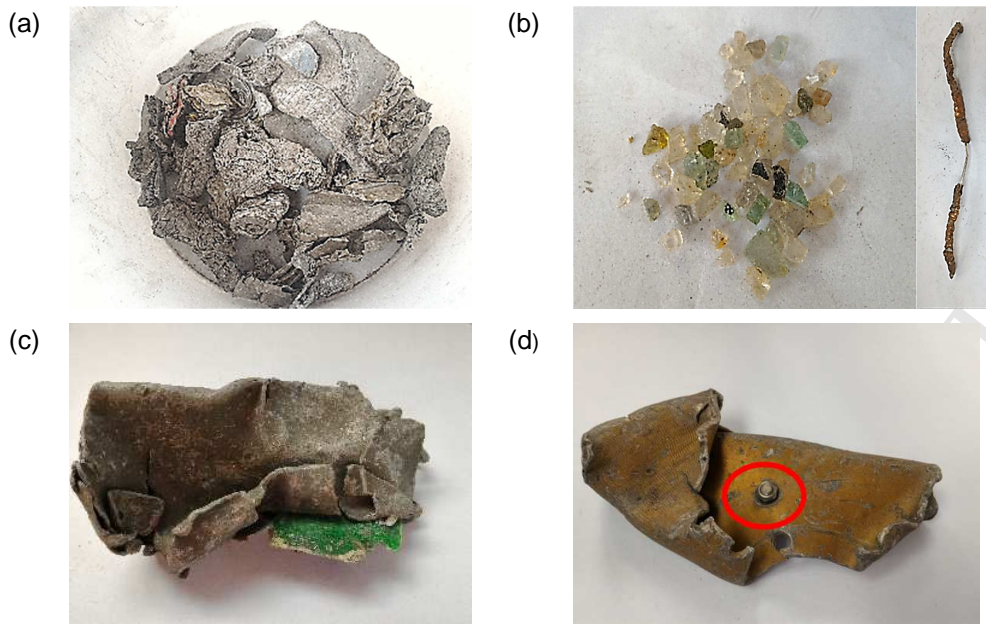
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203 2.3 Sample Analysis Procedures

204 The Al with high Fe content fraction was sieved through a 40mm mesh sieve to separate
 205 particles to two particle size categories: 12-40mm and 40-120mm. This step was carried out to allow
 206 comparability with the observations made for the Al fractions of similar particle size classes.

207 Each particle was weighed and hand-sorted according to the different liberation classifications, as
 208 shown in Figure 4 and as follows.

- 209
- Liberated Al samples consisting of Al only (Figure 4a).
 - 210 • Liberated impurities were particles consisting of a single material type other than Al
 211 (Figure 4b).
 - 212 • Unliberated impurities were particles consisting of material combinations other than Al
 213 (Figure 4b).
 - 214 • Unliberated Al samples were particles consisting of Al that was still attached to other
 215 material types without the presence of a joint (Figure 4c).
 - 216 • Unliberated Al samples due to joint were particles consisting of Al that was still attached
 217 to other material types with the presence of a joint (Figure 4d).



218 Figure 4: Examples of liberation classification for particles end up in the Al stream. (a) Liberated Al samples (Al particles only);
 219 (b) Liberated/unliberated impurities (liberated glass and unliberated Cu-Fe particles); (c) Unliberated Al sample not due to joint
 220 (PWB inserted in Al particle); (d) Unliberated Al sample due to joint (screw and bolt attached to Al particle).

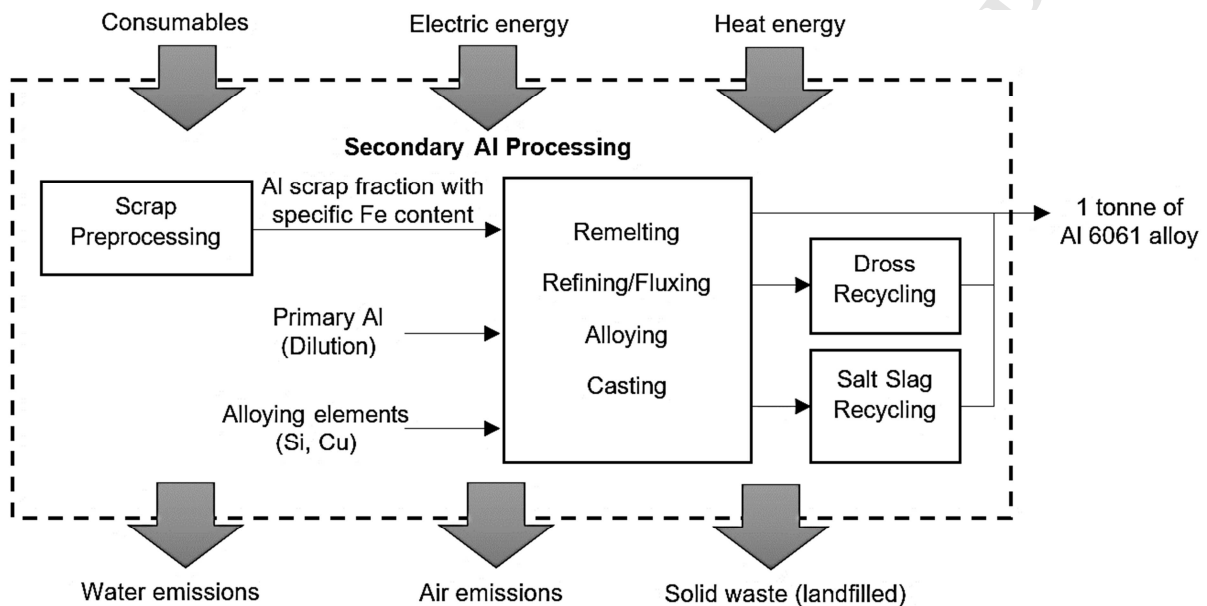
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222 Unliberated particles were further separated into their individual materials. The mass of each
 223 material was recorded. For cases where further material separation was not possible due to
 224 entanglement or rust, the mass of individual materials was calculated using their volumes and
 225 average material densities: Al (2745kg/m^3), steel (7825kg/m^3), copper (8595kg/m^3), plastic
 226 (1253kg/m^3), rubber (1270kg/m^3), and foam (255kg/m^3) (Callister and Rethwisch, 2013; Gaustad et
 227 al., 2012). The types and characteristics of joints causing impurities were observed, and the range of
 228 joint sizes, joint material liberation, and the number of rusty joints were recorded quantitatively.

229 2.4 Environmental Impact Assessment

230 To evaluate the environmental impacts associated with the quality of different Al scrap fractions
 231 collected from the case study, LCA was carried out to assess the dilution and quality losses in
 232 remelting the scrap to be reused as Al 6061 alloy (AA6061). During remelting, dilution losses occur
 233 due to the need to dilute the residual element concentration (e.g. Fe) with primary Al. To avoid quality
 234 losses, alloying elements (e.g. Si and Cu) are added (Paraskevas et al., 2015). The environmental
 235 impact assessment only takes into consideration the secondary Al processing of the defined system
 236 boundary shown in Figure 5. The wrought AA6061 was chosen as the target secondary alloy since it

237 is widely used in automotive applications and thus, likely to be close to the average composition of the
 238 Al scrap retrieved from ELV. To compare the environmental impact of smelting different Al scrap, the
 239 functional unit is defined as Al recycling to produce 1 tonne of AA6061. The calculations for the
 240 required primary Al for dilution purposes and the additional alloying elements are attached in
 241 Appendix A. The credits for subsequent recycling of by-products, such as dross and salt slag, were
 242 also taken into consideration.



243

244 Figure 5: System boundary and functional unit of secondary Al processing for different Al scrap fractions.

245

246 GaBi software was used to model all the processes and resources involved during the secondary
 247 Al processing. Electricity generation was modelled based on the average electricity consumption mix
 248 in Europe. The life cycle inventories were obtained from GaBi Professional database v6.115 and a
 249 previous comprehensive report from the Aluminium Association (The Aluminium Association, 2013),
 250 as detailed in Table 4. The environmental performance was calculated based on the midpoint
 251 categories of the International Reference Life Cycle Data System (ILCD recommendations v1.09).
 252 These recommendations were based on the ILCD handbook in accordance with the ISO 14040 series
 253 (European Commission et al., 2010; ISO, 2006). Following this method, the midpoint results were
 254 normalised to person-equivalent (PE) units—the environmental impact caused by an average
 255 European annually. An equal weighting was applied for the midpoint impact categories, and the

256 normalised PE scores for all midpoint categories were added to allow comparison of the overall
 257 environmental performance for different Al scrap fractions to achieve 1 tonne of AA6061. The single
 258 impact score for climate change (include biogenic carbon) and resource depletion, mineral, fossils and
 259 renewables impact are provided in Table 4 due to their significant contribution to the environmental
 260 performance.

261 Table 4: The life cycle inventory data sources for materials and recycling processes, and the respective impact score for climate
 262 change (include biogenic carbon) and resource depletion, mineral, fossils and renewables.

Process	Climate change (PE/ton)	Mineral and fossil depletion (PE/ton)	Source	Description
Al scrap preprocessing	6.01E-03	3.62E-04	(The Aluminium Association, 2013)	The dataset includes scrap collection, separation, cleaning, and preprocessing.
Al scrap remelting	2.59E-01	1.15E-02	(The Aluminium Association, 2013)	The dataset includes remelting, refining, alloying, and casting of secondary Al. Dross and salt slag recycling are included.
Primary Al ingot	1.08E+00	2.84E-01	GaBi Professional Database v6.115	The dataset includes cradle to gate inventory for primary Al ingot production in Europe.
Primary Cu	4.04E-01	3.56E+02	GaBi Professional Database v6.115	The dataset includes cradle to gate inventory for primary Cu (99.999%) in Germany.
Primary Si	5.45E-01	4.77E-02	GaBi Professional Database v6.115	The dataset includes cradle to gate inventory for primary Si (99%) in global context. The chemical composition is approximated based on Si-2202 (BAIDAO, 2007; SINO GU, 2016).

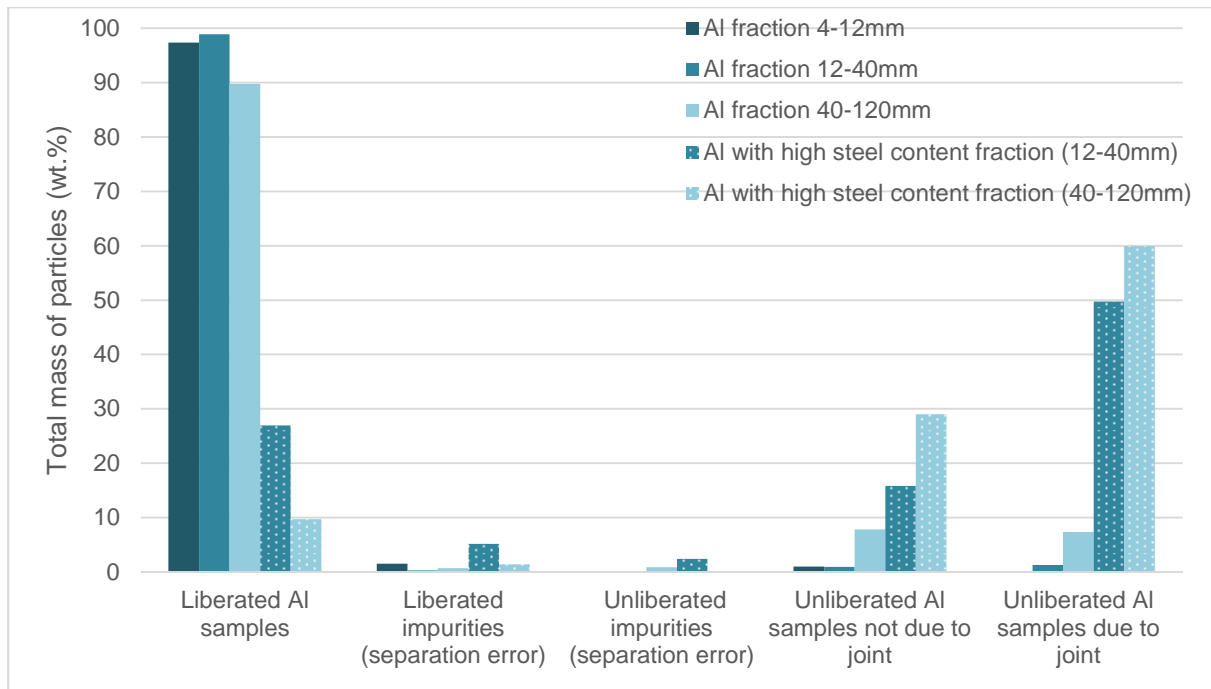
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264 3 Results

265 The liberation categories of the collected Al samples from different fractions were studied. The
266 average Al purity of each fraction was determined. The presence of impurities due to joints was
267 further analysed, and the types of joining techniques causing impurities were characterised.

268 3.1 Al Sample Analysis

269 The mass distribution of particles in the different liberation categories is shown in Figure 6.
270 Liberated and unliberated impurities were mainly caused by separation errors during the recycling
271 processes, and can be characterised as fine particles (<4mm); materials with similar density range to
272 Al; small and longitudinal heavy metal particles; and materials with density less than Al (<2kg/l). The
273 types of impurities consisted of ferrosilicon fines, glass, PWB, Cu, Fe, wires, plastics, and other light
274 fraction of non-metals. Ferrosilicon fines are an example of fine particles easily trapped in Al samples
275 during the density separation. Glass and PWB have a density range of 2.47-2.54kg/l (Malone and
276 Dolter, 2008) and 1.5-2.89kg/l (Bizzo et al., 2014; Zhang and Forssberg, 1997) respectively that can
277 be similar to Al density. Small heavy metal particles, such as Cu, Fe and wires, with thin and long
278 shapes caused them to be entangled between Al particles during the density separation. Plastics,
279 rubber, fabric, fibrous materials and foam are examples of impurities that were not well separated
280 through density separation at earlier stages. Unliberated Al samples both with or without the presence
281 of joint have higher Al content in the particles by mass and therefore, they were more likely to end up
282 in the Al streams.



283

284 Figure 6: Liberation categories for particles in different Al fractions.

285

286 From Figure 6, the total mass of unliberated Al samples both with and without joint is showing an
 287 increasing trend when the particle sizes are larger. This observation is valid for both the Al fraction
 288 and Al with high steel content fraction. To understand the purity level of Al samples for different
 289 particle sizes, the mass fraction of impurities was calculated and the result is shown in Table 5.

290 Table 5: Al purity for different Al fractions with 95% confidence interval.

Category	Particle size class (mm)	Al purity (wt.%)
Al with high Fe content fraction	40-120	82.07 ± 3.86
Al with high Fe content fraction	12-40	80.75 ± 3.38
Al fraction	40-120	98.66 ± 0.58
Al fraction	12-40	99.57 ± 0.29
Al fraction	4-12	98.11 ± 0.58

291

292 In general, the quality of recycled Al can be separated into two classes: Al purity more than 98%,
 293 and Al purity less than 83%. Al purity less than 83% consisted of Al with high steel content fractions

294 that were separated through a strong head pulley magnet as the final separation process in the
295 recycling facility.

296 Based on the analysis of the shredded samples, smaller particle sizes do not indicate higher Al
297 purity. Al with high Fe content fraction (40-120mm), and Al fraction (12-40mm) have higher Al purity
298 values in their respective categories. The geometry, joint size, and material types of the combined
299 parts also affect the purity levels of Al fractions in different particle sizes. For instance, when a large
300 number of small steel screw fasteners (e.g. steel screw with diameter and length of 2mm and 4mm
301 respectively) are used, the likelihood of Fe impurities due to screw fasteners present in the Al fraction
302 in smaller particle sizes is quite high with respect to mass.

303 The material types of impurities were identified to understand the extent of contamination in the Al
304 samples. Some of the impurity types can be removed easily during the secondary Al production
305 whereas others, such as Fe, require a dilution process using primary Al. As seen in Table 6, the types
306 of impurities are Fe, Cu, organic and inorganic. It can be observed that the smaller particle size
307 fraction, 4-12mm has a higher impurity level than the 12-40mm fraction due to the material types and
308 the physical characteristics of impurities. These impurities are largely contributed by ferrosilicon fines
309 (consisting of ferrous and silicon), thin and long-shaped wires (consisting of Cu and plastics), small
310 pieces of shattered glass (silicon) and plastics that typically have small dimensions or high brittleness.
311 One of the most undesired tramp elements during Al recycling are Fe impurities (Cho et al., 2015;
312 Paraskevas et al., 2015) due to their detrimental effect on the mechanical properties of Al alloys
313 (Belov et al., 2002). Therefore, this case study focussed on the source of Fe impurities in unliberated
314 samples due to joints to understand the impact of joining choices on the purity level of recycled Al.

315 Table 6: Types of impurities present in the Al output streams in the Belgian recycling facility.

Category	Particle size class (mm)	Average mass percentage (wt.%)			
		Fe impurities	Cu impurities	Organic impurities	Inorganic impurities
Al with high Fe fraction	40-120	11.32	0.27	5.82	0.42
Al with high Fe fraction	12-40	9.82	1.38	6.40	1.56
Al fraction	40-120	0.36	0.25	0.71	0.05
Al fraction	12-40	0.03	0.13	0.23	0.06
Al fraction	4-12	0.14	0.26	0.96	0.43

3.2 Observations on the Joint Type Causing Impurities

From the collected Al samples, it was observed that mechanical fastening and adhesive bonding were the two main types of joining techniques causing impurities. The amount of unliberated Al samples due to adhesive bonding was extremely small. They were mostly combinations of Al and lower density materials, such as Al-plastic and Al-foam particles, using lap joint. Lower density materials assisted in breakage during the shredding process due to centrifugal force, and hence, were less likely to cause impurities in the Al samples.

In contrast, mechanical fasteners were the major type of joining method contributing to the presence of Fe impurities in the Al stream, since they are typically made of steel. They were further classified to understand the different types of mechanical fasteners, and how their characteristics contributed to the presence of impurities. Figure 7 shows the various types of mechanical fasteners that were observed in the unliberated Al samples due to joints.

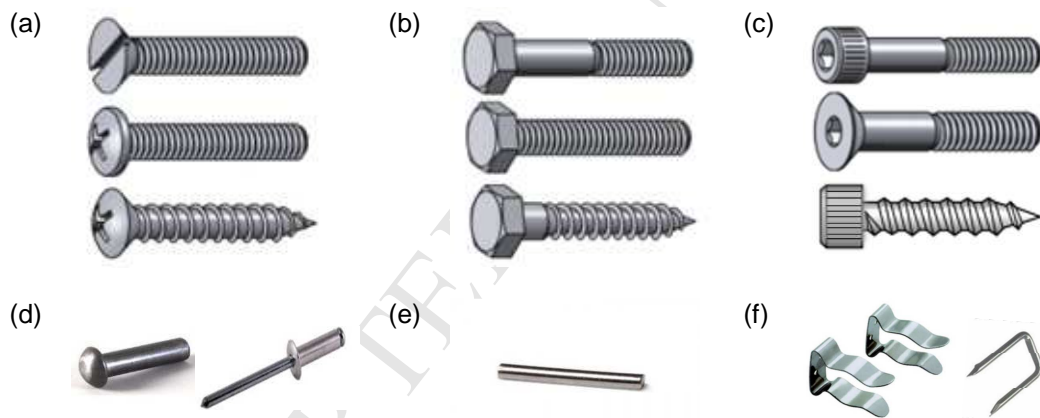


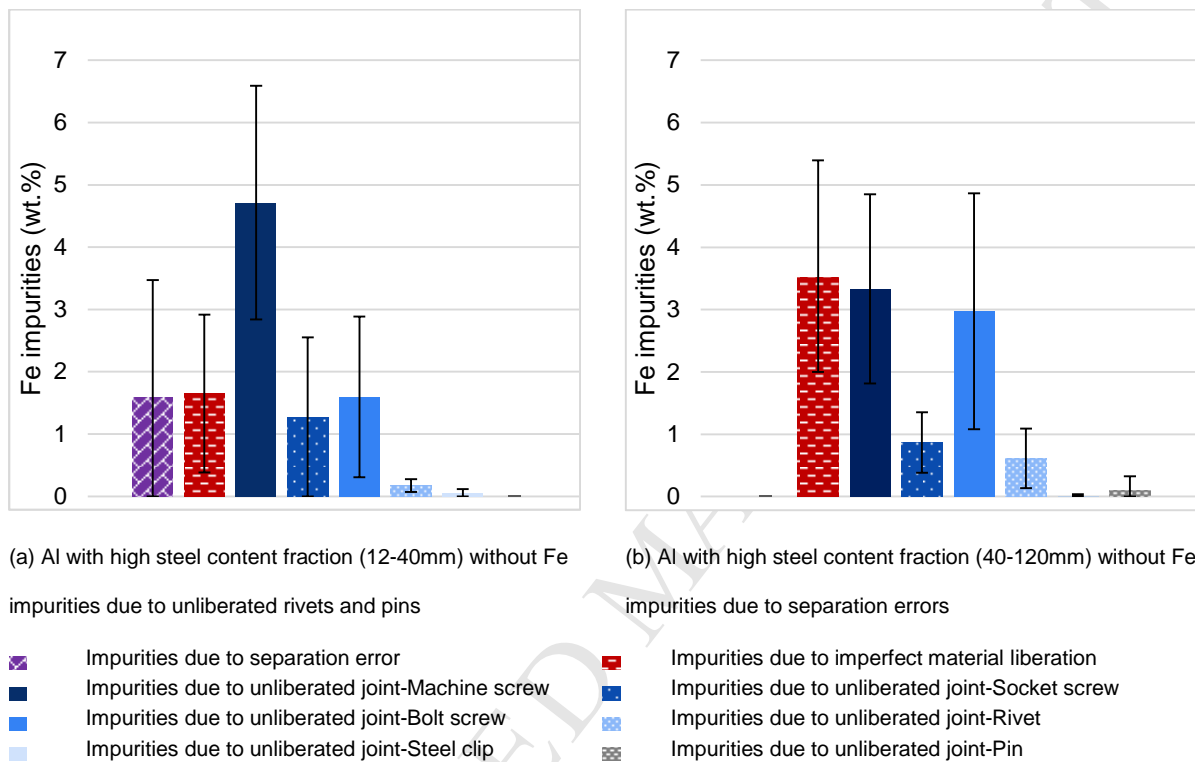
Figure 7: Classification of different mechanical fastening joining methods (Bolt Depot, 2013). (a) Machine screw; (b) Bolt screw; (c) Socket screw; (d) Rivet; (e) Pin; (f) Steel clip.

330

3.2.1 Al with High Steel Content Fraction (12-120mm)

In Figure 8, it can be observed that the likelihood of Fe impurities due to separation errors decreases for larger particle sizes in Al fraction. The Fe impurities observed in the Al with high Fe content fraction (12-40mm) were mostly small in size due to the inability of the head pulley magnet to pick up small Fe content. There were no Fe impurities due to separation errors observed in the larger particle sizes. Fe impurities that were larger in size have higher likelihood of being separated by the

337 magnetic separator after the shredding process. On the other hand, impurities due to imperfect
 338 material liberation were largely caused by structural design, such as enclosures (parts surrounded by
 339 different material types) and entanglement (parts that were twisted together or caught in), after the
 340 shredding process. Therefore, the likelihood of Fe impurities due to imperfect material liberation
 341 increases for larger particle sizes in the Al fraction.



342 Figure 8: Fe impurities present in the Al with high steel content fraction with 95% confidence intervals.

343

344 There were a variety of mechanical fastener types causing Fe impurities in the Al with high Fe
 345 content fractions. Fe impurities observed in smaller particle sizes were caused by unliberated
 346 machine screws, socket screws, bolt screws, rivets, and steel clips. No pins were observed for this
 347 fraction due to the smoother joining surface that allowed them to be well liberated when shredded to
 348 smaller particle sizes. The types of mechanical fasteners causing Fe Impurities in the larger particle
 349 sizes were machine screws, socket screws, bolt screws, rivets, steel clips and pin. For both fractions,
 350 machine screws were more likely to cause impurities when compared to other mechanical fastener
 351 types.

352 The types of mechanical fasteners causing impurities were further characterised through
 353 observation of their physical attributes, as shown in Table 7. The percentages are with respect to the
 354 total number of units for each joint type. It is observed that the number of mechanical fasteners in the
 355 larger particle sizes were higher compared to the smaller particle sizes except for machine screws
 356 and steel clips. Moreover, the fraction with larger particle sizes has a wider range of joint sizes when
 357 compared to smaller particle sizes. In spite of that, the number of joint sizes with diameter and length
 358 more than 6mm and 10mm respectively (large joint sizes) is similar for both particle size classes.
 359 Partial liberation joints, those with more than 50 wt.% of the joint material liberated, were more likely
 360 for threaded fasteners such as machine screws and bolt screws. In most cases, the fasteners' head
 361 was liberated due to protrusion. Rusty threaded fasteners were also more likely to cause impurities in
 362 the Al samples.

363 Table 7: Characteristics of joint causing Fe impurities in Al with high Fe content fractions.

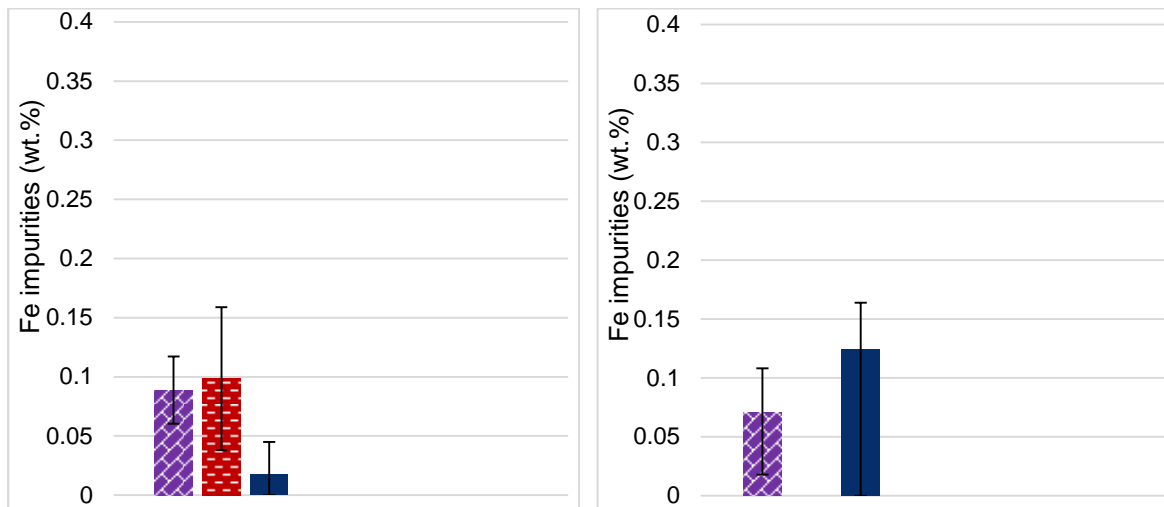
Joint types	Total (unit)	Joint size (mm)		Large joint size (%)	Partial liberation (%)	Rust (%)
		Diameter	Length			
12-40mm						
Machine screw	101	2-10	3-30	12	9	86
Socket screw	11	4-7	9-36	27	0	64
Bolt screw	16	4-10	8-50	56	0	94
Rivet	13	4-5	3-13	0	0	46
Steel clip	19	2-3	10	0	0	0
40-120mm						
Machine screw	94	2-12	2-30	12	4	76
Socket screw	20	3-9	10-60	30	0	85
Bolt screw	39	3-14	7-125	52	8	76
Rivet	48	5-6	3-50	1	0	48
Steel clip	2	2-3	12	0	0	0
Pin	1		10	0	0	0

364

365 3.2.2 Al Fraction (4-120mm)

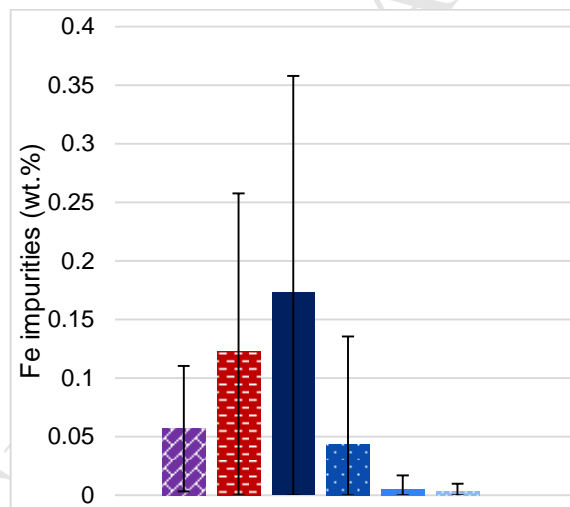
366 Similar to Al samples with high Fe content, the likelihood of Fe impurities due to separation errors
 367 decreases for larger particles sizes in the Al fraction, as seen in Figure 9, since they can be easily
 368 sorted through magnetic separation. In contrast, impurities due to imperfect material liberation could

369 potentially be higher for larger particle sizes, although they were not observed in the Al fraction (12-
 370 40mm).



(a) Al fraction (4-12mm) with Fe impurities due to separation errors, imperfect material liberation, and unliberated machine screws

(b) Al fraction (12-40mm) with Fe impurities due to separation errors and unliberated machine screws



(c) Al fraction (40-120mm) with Fe impurities due to a variety of unliberated joint types

- Impurities due to separation error
- Impurities due to unliberated joint-Machine screw
- Impurities due to unliberated joint-Bolt screw
- Impurities due to imperfect material liberation
- Impurities due to unliberated joint-Socket screw
- Impurities due to unliberated joint-Rivet

371 Figure 9: Fe impurities present in the Al fraction with 95% confidence intervals.

372

373 The likelihood of Fe impurities due to mechanically fastened joints in the Al fraction is higher for
 374 larger particle sizes. There was more variety of mechanical fastener types that contribute to the Fe

375 impurities in the Al fraction (40-120mm). Machine screws were the only type of joint causing impurities
 376 in the smaller particle sizes, whereas machine screws, socket screws, bolt screws and rivets were
 377 observed in Al fraction (40-120mm). Despite the use of a strong head pulley magnet to remove small
 378 Fe content, machine screws contaminating the different Al fractions were still present.

379 Table 8 shows the attributes of mechanical fasteners causing Fe impurities in the different Al
 380 fractions. The number of machine screws observed in Al fraction (40-120mm) was larger compared to
 381 the fraction containing the smaller particle sizes. However, there was still a small number of machine
 382 screws present in this smaller particle size fraction. This was due to the lower magnetic force
 383 experienced by small screws located at enclosed spots despite the use of a strong head pulley
 384 magnet. In contrast, the presence of mechanical fasteners other than machine screws (socket screws,
 385 bolt screws, and rivets) was only seen in Al fraction (40-120mm). Socket screws and bolt screws have
 386 a more protruded head compared to machine screws that facilitate liberation during the shredding
 387 process. On the other hand, rivets have a smooth surface that allows them to be easily set free when
 388 shredded into smaller particle sizes. The likelihood of impurities due to larger joint sizes or of partial
 389 liberation is higher for larger particle sizes particularly for the machine screw fastener type.

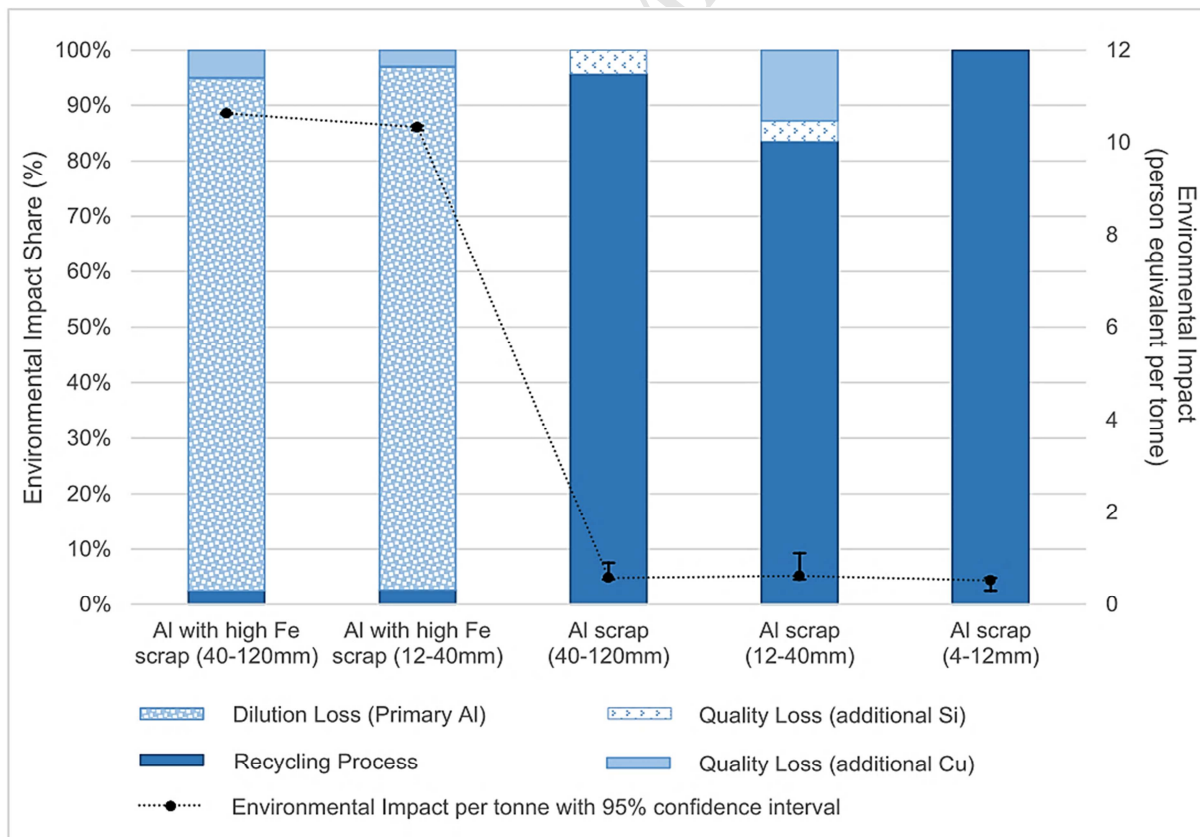
390 Table 8: Characteristics of joint causing Fe impurities in Al fractions.

Joint types	Total (unit)	Joint size (mm)		Large joint size (%)	Partial liberation (%)	Rust (%)
		Diameter	Length			
4-12mm						
Machine screw	2	3-4	8	0	0	100
12-40mm						
Machine screw	1	5	20	0	0	100
40-120mm						
Machine screw	17	3-8	5-25	18	12	50
Socket screw	1	4	18	0	0	0
Bolt screw	2	4	11-12	0	0	50
Rivet	2	5	7	0	0	0

391

392 4 Environmental Impact Assessment Results

393 From the LCA results presented in Figure 10, the total environmental impact for Al with high Fe
 394 scrap fractions (both particle sizes) has increased by at least 28 times in comparison to the Al scrap
 395 fractions (4-12mm, 12-40mm, and 40-120mm). This is caused by the higher concentration of Fe, that
 396 can be considered as impurities rather than useful alloying elements, and the addition of Si and Cu
 397 alloying elements to produce AA6061. The contribution of different midpoint impact categories for the
 398 different Al scrap fractions to produce 1 tonne of AA6061 is provided in Appendix A. The use of
 399 primary Al as dilution agent is the major contributor to the environmental impact for Al scrap with high
 400 Fe content with an impact share of at least 92%, as supported in other studies (Amini et al., 2007;
 401 Paraskevas et al., 2015). To achieve higher purity wrought Al alloy, a substantial amount of primary Al
 402 is required for the dilution of these streams, and alloying elements are added to meet the
 403 compositional limits. This results in scrap underutilisation since only 3-6wt.% of the produced AA6061
 404 consists of recycled Al scrap. The use of primary Al for dilution can be minimised by using other high
 405 purity scrap streams and optimised Al scrap blending (Paraskevas et al., 2015).



406

407 Figure 10: Total environmental impact and the share of recycling, dilution and quality losses for different Al scrap fractions to
 408 achieve 1 tonne of AA6061.

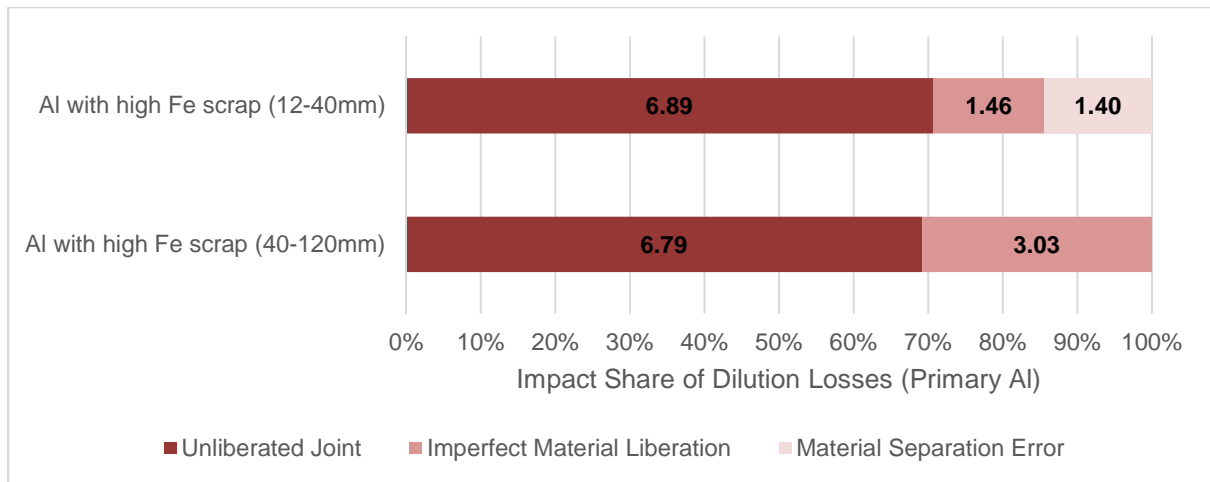
409 A sensitivity analysis was carried out to assess the influence of varying element values for
 410 different Al scrap fractions. The range of values for Fe, Si and Cu elements are shown in Table 9. As
 411 can be seen from Figure 10, the total environmental impact is sensitive to the range of values for Fe,
 412 Si and Cu in different Al scrap fractions. It is shown that the margin of error for the total environmental
 413 impact can be up to ± 0.5 person equivalent per tonne. However, the trend of the total environmental
 414 impact for the range of element values is largely unaffected. The total environmental impact for Al with
 415 high Fe scrap fractions is greatly influenced by the dilution losses. In contrast, the total environmental
 416 impact for Al scrap fractions is largely contributed by recycling process and quality losses since no
 417 dilution losses occur. Despite the sensitivity to the range of Fe, Si and Cu values, the negative impact
 418 of the Al scrap fractions is insignificant compared to Al scrap with high Fe content.

419 Table 9: The range of values for Fe, Si and Cu present in the different Al scrap fractions with 95% confidence interval.

Category	Particle sizes (mm)	Fe (wt.%)		Si (wt.%)		Cu (wt.%)	
		min	max	min	max	min	max
Al with high Fe fraction	40-120	9.95	12.69	0.00	0.94	0.00	0.46
Al with high Fe fraction	12-40	7.12	12.53	0.00	3.51	0.51	2.24
Al fraction	40-120	0.03	0.68	0.00	0.11	0.06	0.46
Al fraction	12-40	0.00	0.06	0.03	0.09	0.00	0.28
Al fraction	4-12	0.07	0.20	0.27	0.54	0.14	0.38

420

421 As can be seen from Figure 11, about 70% of the total impact share of dilution losses for Al scrap
 422 with high Fe content is caused by unliberated joints. Dilution losses due to material separation errors
 423 can only be observed for Al with high Fe scrap in smaller particle sizes due to the presence of silicon
 424 from the shattered glass. The environmental evaluation based on the case study data shows that the
 425 dilution and quality loss impacts are tightly-linked to the quality or purity level of the recovered Al
 426 streams resulting from material liberation. The high Fe content due to improper material liberation has
 427 become a limiting factor for the recyclability of the Al streams. It is worth noting that the environmental
 428 performance may vary according to the efficiency of recycling technologies used in different countries.



429

430 Figure 11: The impact share of dilution losses (primary Al addition) and the environmental impact (person equivalent per tonne)
 431 due to liberation categories.

432 5 Discussion

433 The annual Al output streams in the Belgian recycling facility are shown in Table 10. The material
 434 composition in each Al stream is estimated from the performed sampling, and subjected to the
 435 variation based on the 95% confidence interval. Despite the large variance for certain tramp elements
 436 in different Al output streams, the impact on the environmental performance is insignificant. Al
 437 fractions (12-40mm) and (40-120mm) are the two largest fractions with high Al purity levels of 99.57%
 438 and 98.66% respectively, whereas the Al with high Fe content fraction has the lowest annual amount
 439 with low Al purity level of 81.28% (combination of both particle sizes). With the increasing complexity
 440 of multi-material designs, particularly in the automotive sector which is one of the largest consumers
 441 of Al, it is projected that the Al with high Fe content fraction will be growing and thus, lead to the
 442 reduction of the Al fraction with higher purity (Soo et al., 2015, 2016).

443 Table 10: Estimated material composition for the Belgian recycling facility's annual Al output streams based on the
 444 extrapolation of sampling results.

Material type	Al with high Fe content fraction (12-120mm)		Al fraction (40-120mm)		Al fraction (12-40mm)		Al fraction (4-12mm)	
	ton	wt. %	ton	wt. %	ton	wt. %	ton	wt. %
Al	523.46	81.28	6048.50	98.64	4128.60	99.56	1093.00	98.11
PWB	2.09	0.32	3.14	0.05	1.45	0.03	6.04	0.54
Wire	4.21	0.65	18.80	0.31	2.41	0.06	2.72	0.24
Cu	0.62	0.10	7.60	0.12	4.24	0.10	0.61	0.06
Plastic/composite	12.13	1.88	19.44	0.32	3.87	0.09	3.57	0.32
Rubber	18.26	2.84	6.71	0.11	2.67	0.06	0.20	0.02
Steel	75.82	11.77	21.39	0.35	0.73	0.02	1.29	0.12
Foam	0.57	0.09	0.46	0.01	0.21	0.01	0	0
Fabric	2.19	0.34	2.16	0.04	0	0	0	0
Synthetic leather	0.28	0.04	0	0	0	0	0	0
Glass	0	0	0	0	0	0	5.06	0.45
Fibrous material	2.66	0.41	0	0.41	0	0	0.40	0.04
Ferrosilicon fine	1.70	0.26	3.86	0.06	2.80	0.07	1.16	0.10
TOTAL:	644	100	6132	100	4147	100	1114	100

445

446 5.1 Observational Study Outcomes

447 From the analysed samples, most of the Fe impurities were due to unliberated joints particularly
 448 for Al with high Fe fractions, and Al particles of larger sizes, as seen in Table 11. Particles with
 449 unliberated joints in the Al with high Fe fractions have contributed at least by 69% to the total Fe
 450 impurities. When the particle sizes for different Al fractions decrease, the total Fe impurities due to
 451 unliberated joints decrease by at least 33%. Therefore, smaller particle sizes can assist in reducing Fe
 452 impurities due to unliberated joints. However, the proportion of Fe impurities due to separation errors
 453 or imperfect material liberation is higher for Al fractions with smaller particle sizes. The presence of
 454 these impurities is strongly influenced by the material structural design, joint size used, and the
 455 efficiency of the recycling processes in sorting small to fine particles. Thus, additional loops in Fe
 456 impurity removal or adjustment of the installation with strong magnets could assist in reducing
 457 material separation errors for smaller particle sizes.

458 Table 11: The proportion of Fe impurities due to separation error, imperfect material liberation and unliberated joint.

Category	Particle size class (mm)	Total Fe impurities (wt.%)		
		Separation error	Imperfect material liberation	Unliberated joint
Al with high Fe fraction	40-120	0	3.52	7.88
Al with high Fe fraction	12-40	1.41	1.47	6.94
Al fraction	40-120	0.04	0.08	0.24
Al fraction	12-40	0.03	0	0.01
Al fraction	4-12	0.06	0.07	0.01

459

460 **5.2 Recommendations to Improve Al Scrap Quality**

461 Based on the case study observations, the suggestions to improve the quality of Al recycling from
 462 ecodesign (materials and connections) and recycling process perspectives are as follows.

463 **Ecodesign**

- 464 • Encourage the use of low cost disassembly embedded design, such as the use of active
 465 fasteners (connections that can be unfastened simultaneously through a specific trigger
 466 or a combination of triggers), to maximise the material/part reuse without compromising
 467 the product use phase (Duflo et al., 2008; Peeters et al., 2015; Peeters et al., 2017).
- 468 • Encourage the use of mechanical fasteners with greater protrusion and smoother joining
 469 surface, such as socket screws, rivets and pins, to ease particle liberation.
- 470 • Avoid the use of machine screws. Otherwise, minimise the total number and sizes of
 471 joints particularly for machine screws to reduce the mass of Fe impurities.

472 **Recycling processes**

- 473 • Encourage shredding of particles into smaller sizes to decrease the presence of
 474 impurities due to mechanical fastening joints and imperfect liberation.
- 475 • Encourage the use of strong head pulley magnets to further sort small Fe content
 476 particularly for Al particles that still contain smaller size mechanical fastening joints. This
 477 process is not commonly used in recycling facilities.

- 478 • Encourage the use of more efficient sorting processes for PWB, Cu and wires to refine Al
479 scrap after density separation, such as wet shaker tables (Jordão et al., 2016), kinetic
480 gravity separators (Rem, 2009), and nail roll separators (Fabrizi et al., 2003).

481 The feasibility of the proposed recommendations can be influenced by other factors, such as the
482 economic aspects and legislative boundaries (Soo et al., 2017). Recycling of high purity materials can
483 be affected by the additional recycling costs, the profit of the end products, or the generated mass or
484 volume of different quality scrap fractions. In addition, a governmental role can also be of importance
485 through the imposing of a minimum purity level required for outputs from recycling activities.

486 **6 Conclusions and Future Work**

487 Despite the rigorous recycling processes used in the Belgian recycling facility, unliberated joints
488 are one of the major contributors to the presence of Fe impurities in the Al output streams, particularly
489 for Al with high steel content fractions. The Fe content level highly influences the recyclability of Al
490 scrap; thus, the environmental impact of dilution and quality losses during Al recycling needs to be
491 integrated into LCA for better-informed decisions towards closed-loop recycling.

492 The main type of joining techniques causing impurities in the Al streams are mechanical
493 fasteners, such as machine screws, socket screws, bolt screws and rivets, which are commonly used
494 for assembling Al with other materials. Although adhesive bonding was also observed to cause
495 impurities in the Al particles, these were relatively small and almost negligible when compared to the
496 effects of mechanical fastening joints.

497 Based on the observations of the collected samples, machine screws were the major type of
498 mechanical fasteners causing Fe impurities in different Al fractions due to their joint characteristics.
499 This was consistently observed for various particle sizes. Machine screws are normally less protruded
500 compared to other mechanical fasteners, such as bolt screw, and socket screw. A higher level of
501 protrusion eases liberation during the shredding process. In addition, machine screws that were
502 smaller in size, and corroded due to moisture will make particle liberation more challenging. There
503 were also cases of partial liberation due to the threaded structure that further hindered full material
504 liberation.

505 Unliberated Al samples due to the presence of joints are less likely for smaller shredder output
506 fractions with respect to the total mass of particles. It was found that smaller particle sizes ease
507 liberation of Fe impurities from the joints. However, when considering the Al purity level for different
508 particle sizes, they do not indicate a higher purity level for smaller particle sizes. This was largely
509 caused by the increasing proportion of Fe impurities due to separation errors and imperfect material
510 liberation. Although sorting of Al scrap into different fractions is proven to be effective in obtaining high
511 quality Al in most European countries, it is important to understand the quality of recycled Al scrap in
512 high consumption countries, such as in China (RBC Capital Markets, 2015), from a global
513 perspective.

514 It can be concluded that the choice of joining techniques during the design phase has a significant
515 impact on the environmental performance during the EoL phase. The share of unliberated joints
516 causing the environmental impact due to dilution losses in the Al scrap with high Fe content is the
517 highest compared to material separation errors and imperfect material liberation. Dilution losses
518 cause a significant environmental impact and reduce the avoided environmental impact during Al
519 recycling.

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524 **Appendix A. Supplementary Data**

525 The supplementary data related to this article is as attached.

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Highlights

- Identified the types of joining techniques causing impurities in the Al streams.
- Characterised the joints causing Fe impurities in the Al streams.
- Recycling efficiency of Al was measured through an industrial trial in Europe.
- Identified the linkage between particle sizes and impurities due to joints.
- Assessed the environmental impact of dilution and quality losses due to joints.