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27 **Abstract**

28

29 The influence of the characteristics of soluble algal organic matter (AOM) on the fouling of a
30 7-channel tubular ceramic microfiltration membrane ($\text{ZrO}_2\text{-TiO}_2$, 0.1 μm) was investigated at
31 lab scale. The AOM (~3 mg DOC/L) extracted from a *Microcystis aeruginosa* culture at three
32 phases of growth (10, 20 and 35 days) all caused severe flux decline, and its fouling
33 potential increased with increasing growth time. Size exclusion chromatography,
34 fluorescence excitation-emission matrix spectra and organic matter fractionation showed that
35 the high MW biopolymers were the major component determining the severity of the AOM
36 fouling of the ceramic membrane. For the AOM at stationary phase (35 days), 0.45 and 1 μm
37 pre-filtration gave greater flux decline and hydraulically irreversible fouling than 5 μm pre-
38 filtration due to the denser foulant layer formed and greater amounts of small organic
39 molecules entering membrane pores. However, the non-pre-filtered AOM (with algal cells)
40 caused the greatest flux decline which was likely due to the presence of the high fouling
41 potential cell surface organic matter. The addition of calcium to the feed solutions led to a
42 marked improvement in flux and reduction in membrane irreversible fouling due to the lower
43 fouling potential of the AOM-calcium complexes formed.

44

45 *Keywords:* Algal organic matter; Ceramic membrane; Characterisation; Fouling;
46 Microfiltration

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53 **1. Introduction**

54

55 Low pressure membrane processes such as microfiltration (MF) and ultrafiltration (UF) are
56 being widely used for the purification of drinking water and wastewater treatment due to their
57 high cost-effectiveness [1]. The use of ceramic MF and UF membranes for water treatment
58 has become popular in recent years as the ceramic membranes possess many advantages
59 over conventional polymeric membranes, such as higher selectivity, higher mechanical and
60 chemical stability, and higher hydrophilicity [2]. However, membrane fouling remains a major
61 drawback for most of the membrane-mediated water treatment processes, since it can lead
62 to substantial losses of product water flux over time and the consequent great reduction in
63 the efficiency of the treatment systems [3].

64

65 Blooms of cyanobacteria (also termed blue green algae) such as *Microcystis aeruginosa* in
66 increasingly eutrophic aquatic systems have become a serious environmental issue
67 worldwide. The blooms in natural surface water and treated wastewater can result in a large
68 amount of soluble algal organic matter (AOM) entering downstream water treatment systems
69 [4]. The algal organic compounds are commonly dominated by hydrophobic proteins and
70 hydrophilic polysaccharides which have been widely regarded as responsible for the
71 significant fouling issues in membrane filtration processes [5]. It has been demonstrated that
72 the presence of AOM associated with natural organic matter in surface water or effluent
73 organic matter in wastewater can further reduce the flux of polymeric MF/UF membranes [6-
74 8]. Some efforts have been made since to characterise the AOM fouling of the polymeric
75 MF/UF membranes, with a view to understanding the fouling mechanisms. Qu et al. [9]
76 investigated the influence of the interfacial characteristics of AOM extracted from *M.*
77 *aeruginosa* including surface charge, molecular size and hydrophilicity on the fouling of UF
78 polymeric membranes. They also studied the impact of the AOM and algal cells on
79 membrane fouling, and reported that the AOM caused greater flux decline than algal cells
80 due to greater pore plugging and less porous cake layer formed by the AOM [10]. It was

81 found in a further study by the research group that the dissolved AOM could cause greater
82 flux decline but less irreversible membrane fouling compared with cell surface AOM. They
83 suggested that this was because the cell surface AOM contained more large and
84 hydrophobic molecules, which could result in the foulant layer being more porous but having
85 a higher affinity to the membrane surface than dissolved AOM [11]. In another study,
86 Huang et al. [12] observed that different AOM compositions due to different nutrient
87 conditions had different impacts on the fouling of the polymeric MF membranes. The high
88 fouling potential of AOM was attributed to the high molecular weight polysaccharide-like and
89 proteinaceous substances.

90

91 Although great attention has been drawn to the application of ceramic membranes in water
92 and wastewater treatment, the information regarding AOM fouling of these membranes is
93 very limited to date. A better understanding of AOM fouling of ceramic membranes (which
94 are significantly different from polymeric membranes in terms of physical, chemical and
95 mechanical properties) is essential for the effective design and operation of the processes.
96 As such, the aim of this study was to investigate the impact of the characteristics of soluble
97 AOM on the fouling of a commercially available ceramic MF membrane at lab scale. The
98 influence of the AOM from different phases of *M. aeruginosa* growth, feed solution pre-
99 filtration, and the presence of calcium ions on the fouling was investigated. Advanced
100 organic matter characterisation techniques including size exclusion chromatography (SEC)
101 using liquid chromatography with organic carbon detection (LC-OCD), fluorescence
102 excitation-emission matrix (EEM) spectra and fractionation using resin adsorption
103 chromatography were employed to gain a better insight into the characteristics of the organic
104 compounds involved.

105

106

107

108 2. Experimental

109

110 2.1. Cultivation of algae, AOM extraction and preparation of feed solutions

111 *M. aeruginosa* (CS 566/01-A01) was purchased from the CSIRO Microalgae Research
112 Centre (Tasmania, Australia). The algal cultures were grown in 5 L Schott bottles at 22 °C
113 using MLA medium [13] under humidified aeration. A 16/8 hour light/dark cycle was used to
114 simulate natural light conditions. According to several reports, the algae have high
115 absorbance at 684 nm [14-16]. Optical density (OD) of the algal cell suspension was
116 therefore used to measure algal cell concentration. The correlation between OD₆₈₄ and cell
117 count (5×10^3 - 5×10^6 cells mL⁻¹) was verified as indicated by their strong linear relationship
118 ($R^2 > 0.99$) (data not shown).

119

120 Algal cultures were harvested at the 10th (early exponential phase), 20th (late exponential
121 phase) and 35th day (stationary phase) of growth. Centrifugation ($3270 \times g$ for 30 mins) of
122 the algal cell suspensions and the subsequent filtration of the supernatant (using 1 µm
123 membranes unless otherwise stated) were conducted to extract the dissolved extracellular
124 AOM [8, 9]. In the preparation of the MF feed solutions, the extracted AOM was diluted to
125 approximately the same dissolved organic carbon (DOC) concentration (5.0 ± 0.2 mg/L) with
126 tap water (1.9 ± 0.05 mg DOC/L). The pH of the MF feed solution was adjusted to 8.0 ± 0.2
127 using 1 M HCl or 1 M NaOH prior to each filtration run.

128

129 2.2. Ceramic membrane filtration rig

130 A 7-channel tubular ceramic ZrO₂-TiO₂ MF membrane with a nominal pore size of 0.1 µm
131 (CeRAM™ INSIDE, TAMI Industries) was used in the filtration experiments. The active layer
132 of this membrane is made of a mixture of ZrO₂ and TiO₂, and the support layer is made of
133 TiO₂. These materials give the membrane a highly hydrophilic nature which can reduce the
134 fouling potential to some extent. According to the manufacturer, the membrane can be
135 operated at high temperature (up to 350 °C) and is insensitive to bases and acids. A

136 schematic diagram of the lab-scale ceramic membrane system is presented in Fig. 1. The rig
137 can be operated in either dead-end or cross-flow mode by closing or opening the
138 downstream valve (Valve 3). All filtration runs were carried out in inside-out and dead-end
139 modes at a constant transmembrane pressure (TMP) of 70 ± 1 kPa and under room
140 temperature (22 ± 2 °C). Membrane backwashing was carried out by filtering tap water in
141 outside-in operation mode (i.e., closing Valves 1 and 4, opening Valves 2 and 3) at the same
142 TMP as the filtration runs.

143

144 **Fig. 1.** Schematic diagram of the ceramic membrane filtration system, P1, P2, P3 are
145 manometers.

146

147 2.3. Microfiltration test

148 Prior to each MF run, the clean water flux of the clean membrane (J_0) was obtained by
149 filtering tap water for 2 minutes. The AOM solution was then filtered for 90 minutes under the
150 defined conditions. Membrane permeate flow rate was recorded continuously, and the
151 permeate was sampled after 15, 30, 60 and 90 minutes filtration for chemical analyses. After
152 AOM solution filtration, the clean water flux of the fouled membrane (J_a) was determined by
153 filtering tap water for 2 minutes. The membrane was then backwashed for 2 minutes, and
154 the clean water flux of the backwashed membrane (J_b) was measured by filtering tap water
155 for 2 minutes. Reversible fouling (RF), an indicator of the affinity of foulant for the membrane,
156 was estimated using the following equation (Eq. (1)) [17]. The series resistances including
157 reversible (R_r) and irreversible filtration resistance (R_i) were also calculated using the
158 method described elsewhere [18].

159

$$160 \quad RF = \frac{J_b - J_a}{J_0 - J_a} \times 100\% \quad (1)$$

161

162

163 The same membrane was used for all MF runs, and after each run the membrane was
164 restored by Cleaning in Place (CIP) until the permeate flux reached 138-148 LMH. CIP was
165 carried out through the following steps: 1) 0.1M NaOH solution (65 °C) for 30 minutes; 2)
166 0.1M HNO₃ solution (65 °C) for 20 minutes; 3) tap water (18 - 20 °C) for 2 minutes. All
167 filtration tests were run in duplicate. As the final flux of the duplicate tests typically agreed
168 within 5% and the trend was found to be consistent between the duplicate runs, only one set
169 of flux data was reported. Reversible fouling results were reported using average values.

170

171 2.4. Analytical methods

172 DOC was determined using a Sievers 820 TOC analyser. UVA₂₅₄ and OD₆₈₄ were measured
173 using a UV/vis spectrophotometer (UV2, Unicam). The pH was measured using a Hach
174 Sension 156 pH meter. The concentration of calcium was measured with an atomic
175 absorption spectrometer (AA240FS, Varian). Apparent molecular weight distribution of the
176 AOM was determined by SEC with LC-OCD at the Water Research Centre of the University
177 of New South Wales, Sydney, Australia. The LC-OCD system (LC-OCD Model 8, DOC-
178 Labor Dr. Huber, Germany) utilised a SEC column (Toyopearl TSK HW-50S, diameter 2 cm,
179 length 25 cm) and the chromatograms were processed using the Labview based program
180 Fiffikus (DOC-Labor Dr. Huber, Germany). Fluorescence excitation-emission matrix (EEM)
181 spectra were obtained using a fluorescence spectrometer (LS 55, PerkinElmer) at an
182 excitation and emission wavelength range of 200–550 nm. In order to remove the Raman
183 scatter and other background noise, the fluorescence spectra of Milli-Q water were
184 subtracted from all EEM spectra using Origin software. Fractionation using XAD-4 and DAX-
185 8 resins was used to separate the organics into hydrophobic (HPO), transphilic (TPI), and
186 hydrophilic (HPI) fractions [19].

187

188

189

190

191 **3. Results and discussion**

192

193 3.1. Influence of AOM from different phases of *M. aeruginosa* growth

194 3.1.1. Flux decline and reversibility of AOM fouling

195 Rapid flux decline was observed during the MF of the solutions containing the AOM
196 extracted at 10, 20 and 35 days of *M. aeruginosa* growth, with the majority of flux decline
197 occurring before the specific permeate volume reached 30 L/m² (Fig. 2). In the initial stage of
198 the filtration (< 30 L/m²), the solution containing Day 35 AOM gave a much more rapid and
199 greater flux reduction compared with Day 10 and Day 20 AOM. The maximum flux decline
200 was reached at about 40 L/m² for the Day 35 AOM, whereas Day 10 and Day 20 AOM
201 solutions led to continued flux reduction until the runs were terminated. On reaching 80 L/m²,
202 Day 20 AOM exhibited a similar flux to Day 35 AOM, which was about 5% greater than Day
203 10 AOM. Control filtration tests with tap water and MLA solution (the concentration of MLA
204 in tap water was the same as that in the Day 10 AOM solution) showed the flux decline was
205 relatively insignificant compared with the AOM solutions. The impact of the organic matter in
206 the tap water and MLA on the membrane performance was therefore considered negligible
207 in this study.

208

209 The extent of reversible fouling by the AOM decreased with increasing *M. aeruginosa* growth
210 time, with 35% for Day 10 (R_r 39.4 × 10¹⁰ m⁻¹, R_i 48.9 × 10¹⁰ m⁻¹), 17% for Day 20 (R_r 27.7 ×
211 10¹⁰ m⁻¹, R_i 82.8 × 10¹⁰ m⁻¹) and 10% for Day 35 (R_r 18.3 × 10¹⁰ m⁻¹, R_i 103.9 × 10¹⁰ m⁻¹).
212 Hence the AOM obtained from a later phase of algal growth had higher affinity for the
213 ceramic membrane compared with the AOM from an earlier growth phase, and consequently
214 led to more severe irreversible membrane fouling.

215

216 **Fig. 2.** Normalized flux vs. specific volume for the MF of tap water, MLA solution and the
217 solutions containing AOM from different phases of *M. aeruginosa* growth.

218

219 3.1.2. AOM rejection by the ceramic MF membrane

220 DOC rejection for the AOM from the three phases of *M. aeruginosa* growth was similar (31-
221 35%), and the DOC rejection for each AOM sample was fairly consistent (variation 2-4%)
222 over the filtration period (Fig. 3). A similar trend was observed for UVA₂₅₄ rejection (16-20%),
223 however the UVA rejection was markedly lower than DOC rejection. This indicated that the
224 organic matter retained by the membrane contained less UV-absorbing organic materials. As
225 suggested by Zheng et al. [20], the retained organic matter could contain a large portion of
226 biopolymer substances in the AOM, such as proteins and polysaccharides.

227

228 **Fig. 3.** DOC and UVA₂₅₄ rejection during MF of the three AOM solutions.

229

230 3.1.3. Characterisation of the AOM by LC-OCD

231 In order to interpret the diverse fouling behaviour of the AOM from the different phases of
232 algal growth, the molecular weight distribution of the AOM was examined using SEC with
233 LC-OCD (Fig. 4). Day 10 AOM contained significantly less high MW organic components
234 such as biopolymers (> 20,000 Da) and humic-like substances (500-20,000 Da), but more
235 medium-MW components (i.e., building blocks, 350-500 Da), low MW acids and low MW
236 humic substances (< 350 Da) compared with Day 20 and Day 35 AOM. Although Day 35
237 AOM had a similar content of biopolymers and low-MW compounds to the Day 20 AOM, it
238 contained significantly more humic-like substances.

239

240 Biopolymers such as polysaccharides and proteins have been proven to result in severe
241 fouling of polymeric MF membranes [1, 21]. On the other hand, the deposition of humic
242 substances via hydrophobic interaction with the surface of membranes can also lead to
243 significant membrane fouling [22]. The flux and reversible fouling results presented in section
244 3.1.2 were therefore related to the relative content of the high MW organic compounds
245 (biopolymers and humic-like substances) in the AOM derived from the different phases of
246 algal growth. The lower fouling potential for the AOM from the early exponential phase (Day

247 10) was due to the lower content of biopolymers and high MW humic-like substances. The
248 greater fouling potential of Day 35 AOM compared with Day 20 AOM was associated with a
249 greater amount of high MW humic-like substances, which could result in an enhanced
250 fouling effect on the membrane as reported by Yuan and Zydney [22]. However, it may also
251 be possible that the different fouling behaviour of the Day 20 and Day 35 AOM was due to
252 the different chemical compositions of their biopolymers.

253

254 **Fig. 4.** LC-OCD chromatograms of the AOM from different phases of *M. aeruginosa* growth.

255 HS = humic substances.

256 3.1.4. Characterisation of AOM by fluorescence EEM spectra

257 Fluorescence EEM spectra are widely used for the characterisation of fluorescent organic
258 components in natural organic matter or wastewater effluent organic matter. According to
259 Chen et al. [23], EEM spectra can be divided into 5 regions (Fig. 5a). Regions I (Ex/Em:
260 220-270 nm/280-330 nm) and II (Ex/Em: 220-270 nm/330-380 nm) correspond to aromatic
261 proteins, and region III (Ex/Em: 220-270 nm/380-540 nm) is associated with fulvic acid (FA)-
262 like substances. Regions IV (Ex/Em: 270-440 nm/280-380 nm) and V (Ex/Em: 270-440
263 nm/380-540 nm) represent soluble microbial products (SMPs, e.g., proteins and
264 polysaccharide-like materials) and humic acid (HA)-like materials, respectively. The AOM
265 extracts for the different phases of *M. aeruginosa* growth exhibited different EEM spectral
266 features (Figs. 5a, b & c). The fluorophores increased with increasing *M. aeruginosa* growth
267 time in all EEM regions, which was likely due to the changes in molecular weight/size
268 distribution and/or chemical composition of the AOM over the algal growth phases, i.e., from
269 lower MW to higher MW as shown in Fig. 4. There was a marked increase in fluorescence
270 for all regions between the late exponential (20 days) and the stationary phase (35 days)
271 (Figs. 5b & c).

272

273 The fluorescence regional integration (FRI) method [23] was used to quantify the changes in
274 the fluorescent organic species before and after the MF runs in terms of EEM spectra (EEMs)
275 volume in each region (Fig. 6). The EEMs volumes of the tap water before and after MF are
276 provided as a reference. The EEMs volumes for the HA-like and FA-like regions of the
277 organic matter in Day 10 AOM solution was mainly contributed by the organic matter in the
278 tap water, whereas almost all of the aromatic proteins (AP) and SMPs in the solution were
279 contributed by the AOM. After MF, the reductions in EEMs volume in all five regions for Day
280 10 AOM were markedly lower than for Day 20 and Day 35 AOM. This suggested that less
281 organic matter associated with these regions for the Day 10 AOM was retained by the
282 membrane, and hence led to less fouling of the membrane. The lower rejection of these
283 fluorescent organic components was attributed to the relative abundance of low MW
284 compounds in the early phase of *M. aeruginosa* growth (Fig. 4). There were considerably
285 greater reductions in EEMs volumes in both the AP and SMPs regions for Day 35 AOM and
286 Day 20 AOM (i.e., 66% and 39% for AP, 38% and 24% for SMPs, respectively) compared
287 with the reductions in the FA-like and HA-like regions (i.e., 34% and 18% for FA-like, 18%
288 and 5% for HA, respectively). This indicated that aromatic proteins and SMP-like substances
289 were the major fouling organic components of the ceramic MF membrane. It should be noted
290 that a considerably greater amount of humic substances in the Day 35 AOM was retained by
291 the membrane compared with Day 20 AOM, this could also play a role in the enhanced
292 membrane fouling for the Day 35 AOM.

293

294 **Fig. 5.** Fluorescence EEM spectra of (a) Day 10, (b) Day 20 and (c) Day 35 AOM.
295 Region I and II: aromatic proteins (AP); Region III: fulvic acid-like (FA); Region IV: soluble
296 microbial products (SMPs); Region V: humic acid-like (HA)

297

298 **Fig. 6.** EEM spectra volumes for the AOM before and after MF.

299

300

301 3.1.5. AOM fractionation

302 The AOM before and after microfiltration was fractionated into different organic groups using
303 resin adsorption chromatography, and the results are presented in Fig. 7 in terms of DOC
304 concentrations. Day 10 AOM contained significantly more hydrophobic (HPO) but less
305 transphilic compounds (TPI), and its hydrophilic fraction (HPI) was only slightly lower than
306 Day 20 and 35 AOM. Although the rejection of bulk DOC was similar for the three AOM
307 samples (Fig. 3), the rejection of their fractions varied significantly. There were greater
308 reductions in HPO (38%) and TPI (34%) fractions for the Day 10 AOM after MF compared
309 with Day 20 (29% HPO and 19% TPI) and Day 35 AOM (33% HPO and 23% TPI). However,
310 there was significantly less reduction in HPI (29%) for Day 10 AOM compared with Day 20
311 and Day 35 AOM which had 46 and 63% reduction, respectively. The results suggested that
312 HPI had greater fouling potential than the HPO and TPI fractions, as the greater retention of
313 HPO and TPI for Day 10 AOM did not result in poorer membrane performance compared
314 with Day 20 or Day 35 AOM. This was consistent with some previous studies where the
315 hydrophilic organic fraction that contained a greater proportion of high MW compounds such
316 as biopolymers was found to have higher fouling potential compared with hydrophobic (such
317 as humic substances) and transphilic compounds [1, 18].

318

319 Although the HPI content was only slightly higher for Day 35 AOM than Day 10 and Day 20
320 AOM, there was a markedly greater retention of this fraction for the Day 35 AOM (63% cf.
321 46% Day 20 and 29% Day 10). This indicated that the composition/physico-chemical
322 properties of the HPI fractions in the three AOM samples were markedly different, and hence
323 they exhibited different fouling behaviour. It has been reported that the AOM derived from the
324 stationary phase of *M. aeruginosa* contained more hydrophilic biopolymer substances (i.e.,
325 hydrophilic proteins and carbohydrates) compared with the AOM in the exponential growth
326 phase [24]. Therefore, the increased fouling by the AOM derived from a later algal growth
327 phase in our study was attributed to the hydrophilic biopolymers, which had greater fouling
328 potential compared with the other fractional components in the AOM.

329

330 **Fig. 7.** Fractional components of AOM in MF feed and permeate. HPO: hydrophobic fraction;
331 TPI: transphilic fraction; HPI: hydrophilic fraction.

332

333 3.2. Influence of AOM pre-filtration

334 The impact of pre-filtration of the AOM on the fouling of the ceramic MF membrane was
335 studied by comparing the flux decline and reversible fouling for 0.45, 1 and 5 μm pre-filtered
336 AOM and non-pre-filtered AOM (with *M. aeruginosa* cells) (Fig. 8). The AOM after 5 μm pre-
337 filtration gave significantly less flux reduction during the whole filtration period compared with
338 the other feed solutions. It was observed that around 70% of the algal cells were removed by
339 5 μm pre-filtration (data not shown), and the lower flux reduction for the 5 μm pre-filtered
340 AOM indicated that the remaining particulates (including the smaller algal cells) formed a
341 fouling layer with lower filtration resistance. The 0.45 and 1 μm filtered AOM caused a
342 similar flux decline over the filtration period, which was likely due to these two pre-filtration
343 membranes being relatively similar in pore size and hence their filtrates would have similar
344 physico-chemical properties. However, the non-pre-filtered AOM produced the greatest flux
345 decline which suggested that the algal cell-AOM and/or cell-membrane interactions could
346 have played an influential role in the fouling of the ceramic MF membrane.

347

348 The above results imply that both dissolved AOM (< 0.45 or 1 μm) and the particulates in the
349 AOM solutions can affect the filtration process. The dissolved AOM can cause much more
350 rapid and greater flux decline due to the resultant denser fouling layer, and the presence of
351 particulates can alleviate the initial rapid flux decline due to the formation of a more porous
352 layer of lower resistance. However, the particulates can build up on the membrane surface
353 and make the fouling layer thicker as the filtration proceeds, and hence increase the filtration
354 resistance, leading to greater further reduction in flux at the later stage of filtration (e.g., after
355 40 L/m^2 for the 5 μm pre-filtered AOM). In addition, the AOM attached to the algal cells

356 (also termed cell surface AOM or bound extracellular organic matter) could cause linkages
357 between the cells, leading to a more compact cake layer under the system pressure and
358 hence greater reduction in flux [25].

359

360 **Fig. 8.** Normalized flux vs. specific volume for the MF of the solutions of: 1) AOM with cells;
361 2) 0.45 μm pre-filtered AOM; 3) 1.0 μm pre-filtered AOM; 4) 5.0 μm pre-filtered AOM.
362 (All feed solutions contained AOM extracted from stationary phase)
363

364 It was observed that the 5 μm pre-filtered AOM gave the highest reversible fouling (around
365 21%, R_r $43.9 \times 10^{10} \text{ m}^{-1}$, R_i $97.8 \times 10^{10} \text{ m}^{-1}$). This was likely due to the loosely bonded “pre-
366 layer” formed by the particulates which can prevent smaller particulates from entering the
367 membrane pores, and hence lead to reduced irreversible fouling. As a comparison, only 8-
368 10% reversible fouling (R_r $24.6 \times 10^{10} \text{ m}^{-1}$, R_i $119.9 \times 10^{10} \text{ m}^{-1}$) was obtained for the AOM
369 pre-filtered by the 0.45 or 1 μm filters. The non-pre-filtered AOM solution (i.e., with cells) also
370 produced lower reversible fouling (10%, R_r $25.8 \times 10^{10} \text{ m}^{-1}$, R_i $128.9 \times 10^{10} \text{ m}^{-1}$). This was
371 likely due to the presence of the algal cell surface AOM which was reported to have higher
372 potential for irreversible membrane fouling compared with the dissolved AOM [11].

373

374 3.3. Influence of calcium ion

375 In order to get further insights into the interfacial characteristics of the AOM, calcium (i.e.,
376 CaCl_2) of different concentrations was added to feed solutions containing stationary phase
377 AOM. Addition of calcium reduced the flux decline markedly (Fig. 9), with 2.5 mM of calcium
378 giving slightly greater flux improvement compared with other dosages. As noted by Qu et al.
379 [9], the addition of calcium increased the AOM molecular sizes due to complexation. A
380 significant amount of calcium was retained by the membrane at all dosages, i.e., 17.5 -
381 49.7% (Table 1), which indicated the formation of large complexes of calcium ions and the
382 AOM (i.e., $> 0.1\mu\text{m}$). The large AOM complexes would then form a more porous pre-layer
383 and so result in a higher filtration flux. It was observed that the addition of calcium also led to
384 increased reversible fouling at all calcium dosages (from 11% to 20-25%, R_r from 18.3×10^{10}

385 m^{-1} to $\sim 43.7 \times 10^{10} \text{ m}^{-1}$, R_f from $103.9 \times 10^{10} \text{ m}^{-1}$ to 77.7×10^{10} – $97.8 \times 10^{10} \text{ m}^{-1}$). At 10 mM
386 calcium, there was less flux improvement. This was most probably due to the increased
387 amount of AOM-calcium complexes (as indicated by the increased calcium retention at 10
388 mM), which would result in a thicker cake layer and hence higher resistance to the filtration.

389

390 The DOC retained by the ceramic membrane during MF of the feed solution without addition
391 of calcium was slightly higher than the feed with the calcium addition at 1.0, 2.5 or 5.0 mM
392 (Table 1). This was attributed to the denser fouling layer formed by the AOM solution (no
393 calcium addition), which led to the greater retention of some AOM molecules compared with
394 the more porous layer formed by the AOM-calcium complexes. However, the calcium
395 dosage at 10 mM resulted in a greater DOC retention compared with the feed without
396 calcium addition. This could be due to the trapping of some AOM molecules by the resultant
397 thicker fouling layer and/or the increased complexation of the AOM at the high calcium
398 dosage. It is known that at high calcium concentration aquatic organic matter becomes
399 insoluble when maximum complexation is attained [26].

400

401 **Fig. 9.** Normalized flux vs. specific volume for the MF of AOM (stationary phase) solutions
402 with and without addition of calcium.

403

404 **Table 1.** Retention of calcium and DOC by the ceramic MF membrane at different calcium
405 dosages.

406

407

408 **4. Conclusions**

409

410 The influence of the soluble AOM extracted from three different phases of *M. aeruginosa*
411 growth, AOM pre-filtration and the presence of calcium ion on the fouling of a $0.1 \mu\text{m}$ ZrO_2 -
412 TiO_2 ceramic MF membrane was studied. AOM from the different algal growth phases all

413 caused rapid and great flux decline, but exhibited different fouling potentials, with fouling for
414 the stationary phase > late exponential phase > early exponential phase. Characterisation of
415 the AOM using SEC with LC-OCD, fluorescence EEMs and organic matter fractionation
416 indicated that the biopolymers (containing mainly proteinaceous materials and
417 polysaccharides) were the major organic component that determined the severity of the
418 AOM fouling of the ceramic MF membrane. Since the amount of biopolymer in the late
419 exponential and the stationary phase AOM was fairly similar, it is suggested that a difference
420 in the properties of the biopolymers led to the higher fouling potential of the stationary phase
421 AOM.

422

423 For the stationary phase, the soluble AOM (i.e., 0.45 or 1 µm pre-filtered) caused more rapid
424 flux decline compared with the 5 µm pre-filtered AOM. The relatively lower flux decline for
425 the 5 µm pre-filtered AOM was attributed to a more porous foulant layer due to the presence
426 of particulates in the feed solution. However, the non-pre-filtered AOM (with algal cells)
427 produced the greatest flux reduction, which was likely due to the presence of the high fouling
428 potential cell surface organics. The addition of calcium to the AOM solutions led to reduced
429 flux decline and increased reversible fouling due to complexation of the calcium ions with the
430 AOM molecules to form large complexes and consequently a more porous foulant layer on
431 the membrane surface.

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433 The results indicate that monitoring algal growth can be important for the effective prediction
434 of fouling and implementation of maintenance measures for ceramic membrane systems
435 during cyanobacterial bloom events. Removal of cyanobacterial cells by a loose MF pre-filter
436 (e.g., 5 µm) may mitigate membrane fouling due to the reduction of the cell surface organic
437 matter. Furthermore, chemical coagulation may be an effective pre-treatment of AOM
438 containing water for improving the filtration performance of the ceramic MF membranes
439 since the fouling potential of AOM can be greatly reduced by its complexation with metal
440 ions.

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442 **References**

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- 444 [1] N. Lee, G. Amy, J.-P. Croué, H. Buisson, Identification and understanding of fouling in
445 low-pressure membrane (MF/UF) filtration by natural organic matter (NOM), *Water Res.*
446 38 (2004) 4511-4523.
- 447 [2] B. Hofs, J. Ogier, D. Vries, E.F. Beerendonk, E.R. Cornelissen, Comparison of ceramic
448 and polymeric membrane permeability and fouling using surface water, *Sep. Purif.*
449 *Technol.* 79 (2011) 365-374.
- 450 [3] P. Bacchin, P. Aimar, R.W. Field, Critical and sustainable fluxes: Theory, experiments
451 and applications, *J. Membr. Sci.* 281 (2006) 42-69.
- 452 [4] J. Fang, X. Yang, J. Ma, C. Shang, Q. Zhao, Characterization of algal organic matter and
453 formation of DBPs from chlor(am)ination, *Water Res.* 44 (2010) 5897-5906.
- 454 [5] Y.-T. Chiou, M.-L. Hsieh, H.-H. Yeh, Effect of algal extracellular polymer substances on
455 UF membrane fouling, *Desalination* 250 (2010) 648-652.
- 456 [6] N. Lee, G. Amy, J.-P. Croué, Low-pressure membrane (MF/UF) fouling associated with
457 allochthonous versus autochthonous natural organic matter, *Water Res.* 40 (2006) 2357-
458 2368.
- 459 [7] Y. Goh, J. Harris, F. Roddick, Impact of *Microcystis aeruginosa* on membrane fouling in a
460 biologically treated effluent, *Water Sci. Technol.* 63 (2011) 2853-2859.
- 461 [8] Y. Goh, J. Harris, F. Roddick, Reducing the effect of cyanobacteria in the microfiltration of
462 secondary effluent, *Water Sci. Technol.* 62 (2010) 1682.
- 463 [9] F. Qu, H. Liang, Z. Wang, H. Wang, H. Yu, G. Li, Ultrafiltration membrane fouling by
464 extracellular organic matters (EOM) of *Microcystis aeruginosa* in stationary phase:
465 Influences of interfacial characteristics of foulants and fouling mechanisms, *Water Res.*
466 46 (2012) 1490-1500.

- 467 [10] F. Qu, H. Liang, J. Tian, H. Yu, Z. Chen, G. Li, Ultrafiltration (UF) membrane fouling
468 caused by cyanobacteria: Fouling effects of cells and extracellular organics matter (EOM),
469 Desalination 293 (2012) 30-37.
- 470 [11] F. Qu, H. Liang, J. He, J. Ma, Z. Wang, H. Yu, G. Li, Characterization of dissolved
471 extracellular organic matter (dEOM) and bound extracellular organic matter (bEOM) of
472 *Microcystis aeruginosa* and their impacts on UF membrane fouling, Water Res. 46 (2012)
473 2881-2890.
- 474 [12] W. Huang, H. Chu, B. Dong, Characteristics of algogenic organic matter generated
475 under different nutrient conditions and subsequent impact on microfiltration membrane
476 fouling, Desalination 293 (2012) 104-111.
- 477 [13] C. Bolch, S. Blackburn, Isolation and purification of Australian isolates of the toxic
478 cyanobacterium *Microcystis aeruginosa* Kütz, J. Appl. Phycol. 8 (1996) 5-13.
- 479 [14] P. Rajasekhar, L. Fan, T. Nguyen, F.A. Roddick, Impact of sonication at 20 kHz on
480 *Microcystis aeruginosa*, *Anabaena circinalis* and *Chlorella* sp, Water Res. 46 (2012)
481 1473-1481.
- 482 [15] G. Zhang, P. Zhang, H. Liu, B. Wang, Ultrasonic damages on cyanobacterial
483 photosynthesis, Ultrason. Sonochem. 13 (2006) 501-505.
- 484 [16] G. Zhang, P. Zhang, B. Wang, H. Liu, Ultrasonic frequency effects on the removal of
485 *Microcystis aeruginosa*, Ultrason. Sonochem. 13 (2006) 446-450.
- 486 [17] M. Hashino, K. Hiram, T. Katagiri, N. Kubota, Y. Ohmukai, T. Ishigami, T. Maruyama, H.
487 Matsuyama, Effects of three natural organic matter types on cellulose acetate butyrate
488 microfiltration membrane fouling, J. Membr. Sci. 379 (2011) 233-238.
- 489 [18] L. Fan, J.L. Harris, F.A. Roddick, N.A. Booker, Influence of the characteristics of natural
490 organic matter on the fouling of microfiltration membranes, Water Res. 35 (2001) 4455-
491 4463.
- 492 [19] G.R. Aiken, D.M. McKnight, K.A. Thorn, E.M. Thurman, Isolation of hydrophilic organic
493 acids from water using nonionic macroporous resins, Org. Geochem. 18 (1992) 567-573.

- 494 [20] X. Zheng, M. Ernst, M. Jekel, Identification and quantification of major organic foulants
495 in treated domestic wastewater affecting filterability in dead-end ultrafiltration, *Water Res.*
496 43 (2009) 238-244.
- 497 [21] C.N. Laabs, G.L. Amy, M. Jekel, Understanding the Size and Character of Fouling-
498 Causing Substances from Effluent Organic Matter (EfOM) in Low-Pressure Membrane
499 Filtration, *Environ. Sci. Technol.* 40 (2006) 4495-4499.
- 500 [22] W. Yuan, A.L. Zydney, Humic acid fouling during microfiltration, *J. Membr. Sci.* 157
501 (1999) 1-12.
- 502 [23] W. Chen, P. Westerhoff, J.A. Leenheer, K. Booksh, Fluorescence Excitation–Emission
503 Matrix Regional Integration to Quantify Spectra for Dissolved Organic Matter, *Environ.*
504 *Sci. Technol.* 37 (2003) 5701-5710.
- 505 [24] R.K. Henderson, A. Baker, S.A. Parsons, B. Jefferson, Characterisation of algogenic
506 organic matter extracted from cyanobacteria, green algae and diatoms, *Water Res.* 42
507 (2008) 3435-3445.
- 508 [25] S. Babel, S. Takizawa, Microfiltration membrane fouling and cake behavior during algal
509 filtration, *Desalination* 261 (2010) 46-51.
- 510 [26] S. Hong, M. Elimelech, Chemical and physical aspects of natural organic matter (NOM)
511 fouling of nanofiltration membranes, *J. Membr. Sci.* 132 (1997) 159-181.

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523 **Table 1.** Retention of calcium and DOC by the ceramic MF membrane at different calcium
524 dosages.

Calcium dosage (mM)	Calcium retention (%)	DOC retention (%)
0	-	36.5
1.0	17.5	33.4
2.5	37.0	31.8
5.0	42.5	32.3
10.0	49.7	41.1

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543 **Figure Captions**

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545 **Fig. 8.** Schematic diagram of the ceramic membrane filtration system, P1, P2, P3 are
546 manometers.

547 **Fig. 9.** Normalized flux vs. specific volume for the MF of tap water, MLA solution and the
548 solutions containing AOM from different phases of *M. aeruginosa* growth.

549 **Fig. 10.** DOC and UVA₂₅₄ rejection during MF of the three AOM solutions.

550 **Fig. 11.** LC-OCD chromatograms of the AOM from different phases of *M. aeruginosa*
551 growth. HS = humic substances.

552 **Fig. 12.** Fluorescence EEM spectra of (a) Day 10, (b) Day 20 and (c) Day 35 AOM. Regions
553 I and II: aromatic proteins (AP); Region III: fulvic acid-like (FA); Region IV: soluble
554 microbial products (SMPs); Region V: humic acid-like (HA)

555 **Fig. 13.** EEM spectra volumes for the AOM and tap water before and after MF.

556 **Fig. 14.** Fractional components of AOM in MF feed and permeate. HPO: hydrophobic
557 fraction; TPI: transphilic fraction; HPI: hydrophilic fraction.

558 **Fig. 15.** Normalized flux vs. specific volume for the MF of the solutions of: 1) AOM with cells;
559 2) 0.45 µm pre-filtered AOM; 3) 1.0 µm pre-filtered AOM; 4) 5.0 µm pre-filtered AOM.
560 (All feed solutions contained AOM extracted from stationary phase)

561 **Fig. 16.** Normalized flux vs. specific volume for the MF of AOM (stationary phase) solutions
562 with and without addition of calcium.

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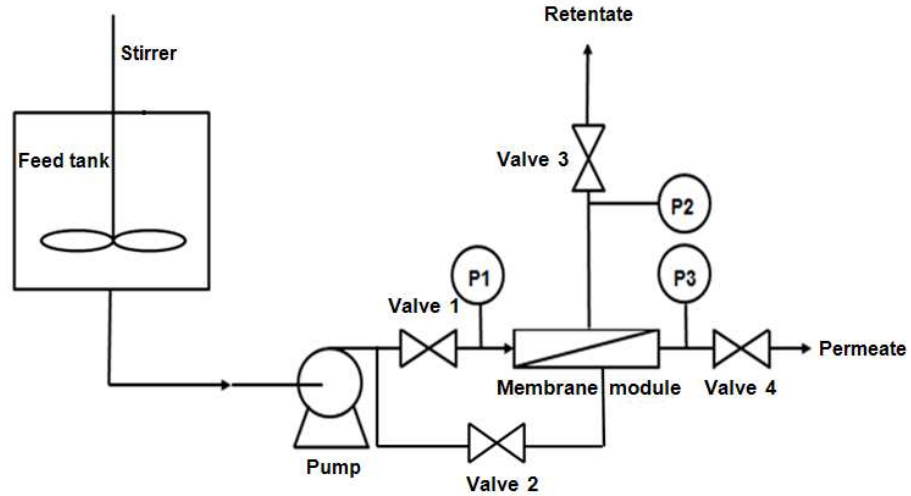
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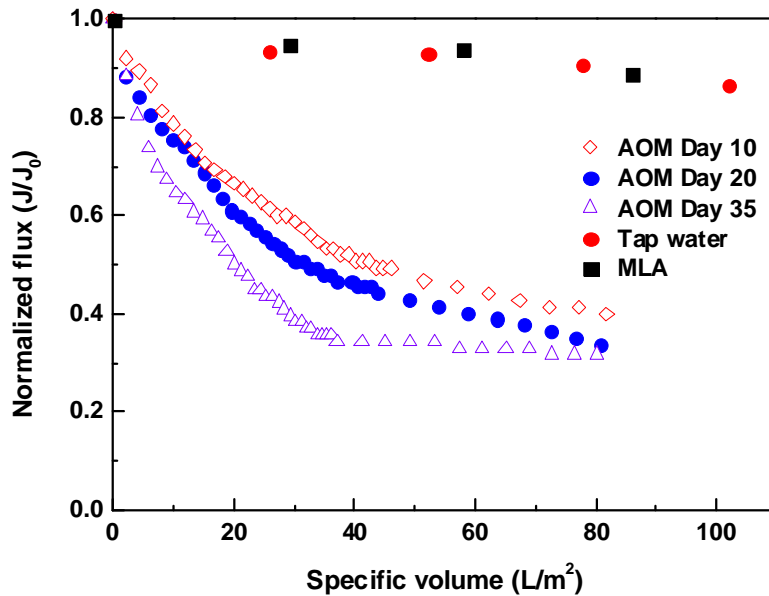
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575 **Fig. 1.** Schematic diagram of the ceramic membrane filtration system, P1, P2, P3 are
576 manometers.

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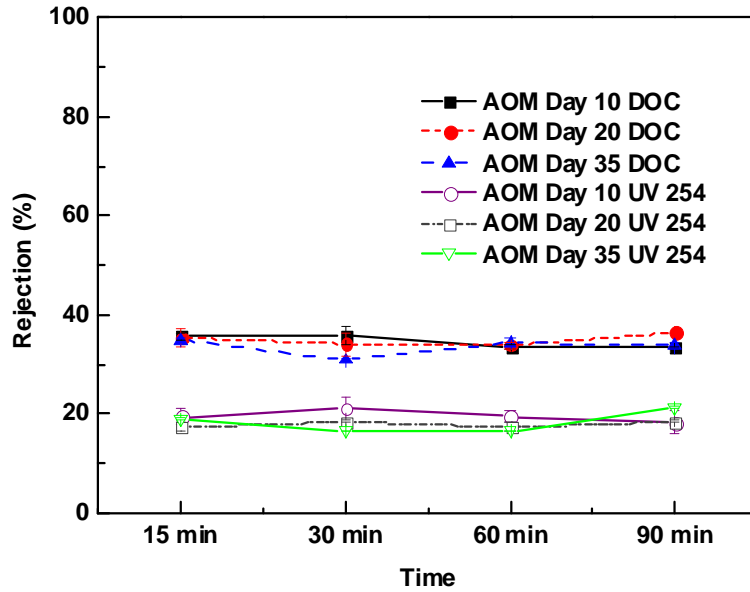


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582 **Fig. 2.** Normalized flux vs. specific volume for the MF of tap water, MLA solution and the
583 solutions containing AOM from different phases of *M. aeruginosa* growth.

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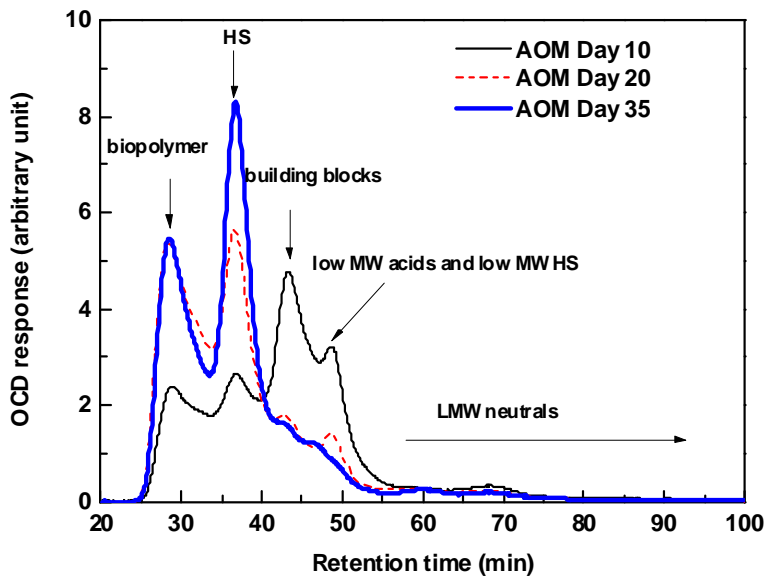
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Fig. 3. DOC and UVA₂₅₄ rejection during MF of the three AOM solutions.

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Fig. 4. LC-OCD chromatograms of the AOM from different phases of *M. aeruginosa* growth.

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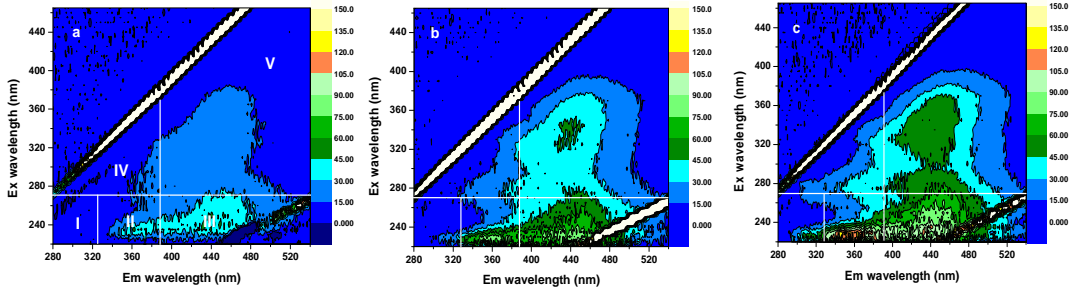
HS = humic substances.

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600 **Fig. 5.** Fluorescence EEM spectra of (a) Day 10, (b) Day 20 and (c) Day 35 AOM. Regions I

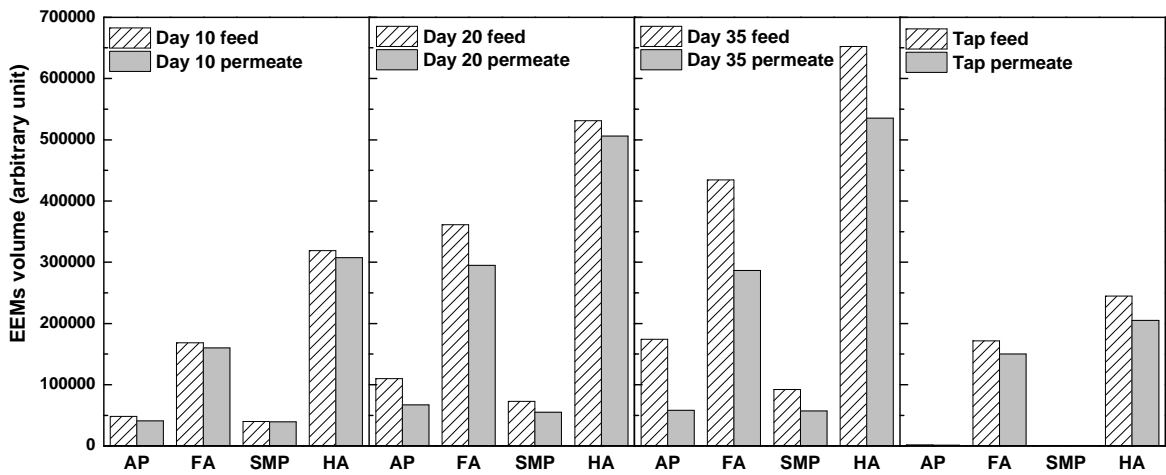
601 and II: aromatic proteins (AP); Region III: fulvic acid-like (FA); Region IV: soluble

602 microbial products (SMPs); Region V: humic acid-like (HA).

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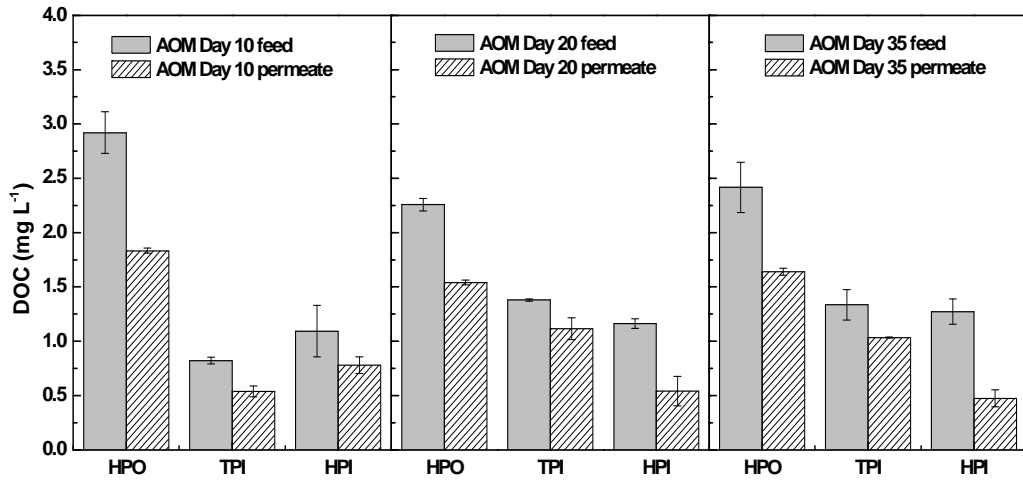
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Fig. 6. EEM spectra volumes for the AOM and tap water before and after MF.

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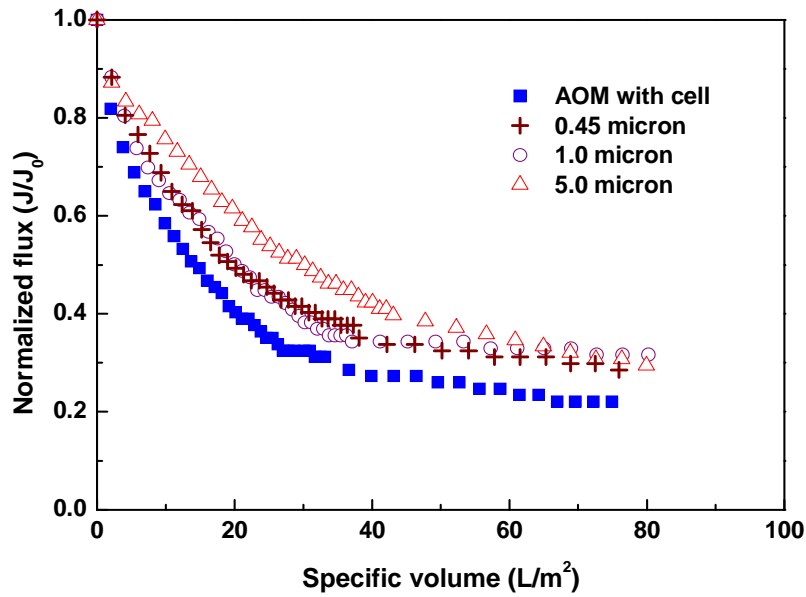
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613 **Fig. 7.** Fractional components of AOM in MF feed and permeate. HPO: hydrophobic fraction;
614 TPI: transphilic fraction; HPI: hydrophilic fraction.

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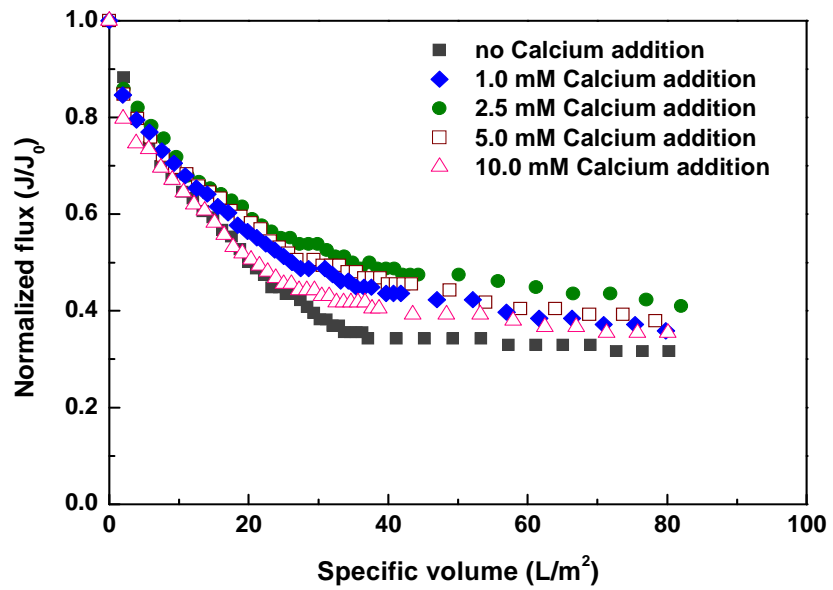
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620 **Fig. 8.** Normalized flux vs. specific volume for the MF of the solutions of: 1) AOM with cells;
621 2) 0.45 μm pre-filtered AOM; 3) 1.0 μm pre-filtered AOM; 4) 5.0 μm pre-filtered AOM.
622 (All feed solutions contained AOM extracted from stationary phase)
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628 **Fig. 9.** Normalized flux vs. specific volume for the MF of AOM (stationary phase) solutions
629 with and without addition of calcium.

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