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1 Effects of dissolved organic matter and nitrification on biodegradation of

2 pharmaceuticals in aerobic enrichment cultures

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12 Abstract

13 Natural dissolved organic matter (DOM) and nitrification can play an important role in 14 biodegradation of pharmaceutically active compounds (PhACs) in aerobic zones of constructed 15 wetlands (CWs). This study used an enrichment culture originating from CW sediment to study 16 the effect of DOM and nitrification on aerobic biodegradation of seven PhACs. The enriched 17 culture degraded caffeine (CAF), metoprolol (MET), naproxen (NAP), and ibuprofen (IBP) with a 18 consistent biodegradability order of CAF > MET > NAP > IBP. Biodegradation of propranolol, 19 carbamazepine, and diclofenac was insignificant (<15%). CAF biodegradation was inhibited by 20 the easily biodegradable DOM. Conversely, DOM enhanced biodegradation of MET, NAP, and 21 IBP, potentially by contributing more biomass capable of degrading PhACs. Nitrification 22 enhanced biodegradation of NAP and IBP and mineralization of the PhAC mixture as well as less 23 biodegradable DOM, which may result from co-metabolism of ammonia oxidizing bacteria or enhanced heterotrophic microbial activity under nitrification. MET biodegradation was inhibited
in the presence of nitrification. DOM and nitrification effects on PhAC biodegradation in CWs
gained from this study can be used in strategies to improve CW operation, namely: designing
hydraulic retention times based on the biodegradability order of specific PhACs; applying DOM
amendment; and introducing consistent ammonium streams to increase removal of PhACs of
interest.

30 Keywords

Pharmaceuticals; Aerobic biodegradation; DOM; Nitrification; Constructed wetland; Microbial
 degradation preference

33 **1. Introduction**

34 The occurrence and accumulation of PhACs become a growing concern with the extensive 35 detection of pharmaceutically active compounds (PhACs) in wastewater effluent, surface water, 36 groundwater and sediment (Hernando et al., 2006). Some PhACs are beginning to be associated 37 with adverse effects in aquatic organisms at environmentally relevant concentrations, i.e. ng/l to 38 μ g/l (Fatta-Kassinos et al., 2011). The conventional wastewater treatment plants (WWTPs) could 39 be a key barrier to eliminate PhAC discharge into the surface water, but are at this moment not 40 designed to treat PhACs sufficiently (Li et al., 2014). Therefore, development of post-treatment 41 processes in addition to conventional WWTPs are important ways to further remove PhACs from 42 domestic effluents.

An alternative technique that could be coupled to WWTP is a constructed wetland (CW), which
is sustainable, inexpensive in construction and easy to operate and maintain (Meng et al., 2014).
In recent years CWs have shown promising functionality for PhAC removal (Zhang et al., 2014).

46 PhAC removal in CWs involves various mechanisms such as photodegradation, sorption, 47 biodegradation, and phytoremediation (Hijosa-Valsero et al., 2016). Of those mechanisms, 48 biodegradation plays a main role in the transformation and mineralization of PhACs (Li et al., 49 2014). PhAC biodegradation from wastewater is favored by aerobic conditions (Zhang et al., 50 2014). Oxygen supporting aerobic biodegradation of PhACs can be present in CWs throughout 51 the water column in free water surface flow CWs, in the upper sediment layer, and in 52 rhizosphere area of CWs (Eger, 1994). Therefore, it is of importance to study the aerobic 53 biodegradation of PhACs in CWs functioning as contaminant remediation systems. Compared to 54 related studies with soil and river sediment (Koba et al., 2016; Lin and Gan, 2011; Zhang et al., 55 2013), aerobic biodegradation of PhACs in CW sediment or using enrichment cultures 56 originating from CW sediment is hardly investigated. A rare study of aerobic biodegradation of 57 carbamazepine (CBZ) and ibuprofen (IBP) in CW sediment focused on the biodegradation 58 kinetics of these PhACs (Conkle et al., 2012).

59 CW sediments contain a large and potentially reactive pool of dissolved organic matter (DOM), 60 and DOM plays an important role in the bioavailability and fate of contaminants in CWs (Chin et 61 al., 1998). DOM is a complex mixture of organic compounds varying widely in molecular weight, 62 from small molecules like hydrophilic organic acids to intermediate polymers like hemicellulose, 63 and large humic substances such as humic acids (HA) and fulvic acids (FA) (Khadka et al., 2016). 64 DOM is believed to interact with aerobic microbial communities influencing the biodegradation 65 of organic contaminants through various mechanisms, including catabolic processes, redox 66 cycling, and nutrient bioavailability (Van Trump et al., 2006). The effect of DOM on the 67 biodegradation of pesticides, herbicides, and polycyclic aromatic hydrocarbon are well 68 investigated (Van Trump et al., 2006). However, studies on the DOM impact on aerobic 69 biodegradation of PhACs as emerging contaminants are limited.

70 Nitrification is an important redox reaction under aerobic conditions. Nitrification is known to 71 enhance PhAC removal in various systems, e.g. nitrifying activated sludge (NAS), and membrane 72 bioreactors (Batt et al., 2006; De Gusseme et al., 2009; Tran et al., 2009). It is believed that the 73 ammonia monooxygenase (AMO) enzyme in ammonia oxidizing bacteria (AOB) can co-74 metabolize non-target PhACs in biochemical processes (Benner et al., 2013; Yi and Harper, 2007). 75 Nevertheless, the co-metabolism occurs for some PhACs, but is ineffective for others. For 76 example, the biodegradation rate of 17α -ethinylestradiol (EE2) significantly decreased after 77 AOBs were inhibited, whereas trimethoprim biodegradation was not influenced by AOB 78 inhibition in the same NAS systems (Khunjar et al., 2011). Moreover, contradicting findings of 79 nitrification on PhAC biodegradation were reported in different treatment systems. Fernandez-80 Fontaina et al. (2016) showed that AOB enhanced biodegradation of IBP in NAS, whereas AOB 81 was ineffective for IBP biodegradation in biofiltration columns packed with anthracite (Rattier et 82 al., 2014). To date, there is no information available on the role of nitrification in biodegradation 83 of PhACs in CW sediment microbial culture, especially for a group of PhACs with different 84 physiochemical properties.

This study applied a microbial enrichment culture originating from CWs to investigate the effect of DOM and nitrification on the aerobic biodegradation of a mixture of seven PhACs with different physiochemical properties. This research aims to elucidate the role of DOM and nitrification in CWs on PhAC biodegradation under aerobic conditions.

89 2. Materials and methods

90 2.1. Chemicals and reagents

PhACs were purchased commercially: (\pm)-metoprolol (+)-tartrate salt (MET, \geq 98%), caffeine (CAF, \geq 99%), (\pm)-propranolol hydrochloride (PRO, \geq 99%), CBZ (\geq 99%), naproxen (NAP, \geq 93 98%), IBP (\geq 98%), and fenoprofen calcium salt (FEN, \geq 97%) from Sigma Aldrich Chemie B.V (the Netherlands). PhAC mixture stock (20 mg/L) was prepared with MilliQ water (Millipore, USA). Acetonitrile and water with 0.1% formic acid (Biosolve B.V., the Netherlands) were used for ultra-performance liquid chromatography (UPLC) analysis. All other chemicals used were of analytical grade.

98 2.2. Batch experiments

99 Batch experiments were performed to study the effect of DOM and nitrification on the aerobic 100 biodegradation of a mixture of 7 PhACs in a microbial enrichment culture. The enriched culture 101 is originally from CW sediment microbial culture, which was acclimated aerobically with PhAC 102 mixture (1 mg/l each) for 22 months in growth medium (Table S1 of the Supplementary Materials). The spiked PhAC concentration is above the environmental relevant concentrations 103 104 of PhACs to enrich PhAC-degrading cultures with PhACs as the sole carbon source. The mixed 105 sediment was collected from two CWs that are applied as post-treatment processes for two 106 Dutch WWTPs.

107 15 g sediment inoculum was mixed with 120 ml growth medium for the incubation. Batch 108 bottles sealed with rubber stoppers (250 ml) were incubated on a shaker (120 rpm) at 20 °C and 109 covered with aluminum foil to avoid photodegradation of PhACs. The gas phase was filled with 110 over-pressured air (1.3 bar) and was exchanged to maintain the O₂ supply (>10% of the gas 111 phase) during the incubation. During 22 months of incubation, the inoculum was amended with 112 PhAC mixture repeatedly once easily biodegradable PhACs (CAF, MET, NAP, IBP) were almost 113 depleted to enrich a PhAC-degrading microbial community. Afterwards, 10% of the enrichment (11 ml) was transferred to fresh aerobic medium (110 ml) spiked with PhAC mixture, and incubated for 2 weeks to obtain sediment free enrichment culture. After repeating the transfer twice, 10% of the active PhAC degrading enrichment was used as inoculum for our batch experiments. TOC analysis confirmed that no residual organic matter (< 0.3 mg/l, detection limit) was transferred from the original sediment inoculum to our batch experiments.

Table 1 Designed experimental conditions to investigate the effect of DOM and nitrification onremoval of PhACs.

Conditions	PhACs	DOM	Nitrification
	(1 mg/l each)	(22.8 mg/l in carbon)	(ammonium, 41.7 mg/l)
А	V	×	×
В	V	\checkmark	×
С	V	×	V
D	V	\checkmark	V
E	×	V	x

121 For the experiments, five batch conditions were designed as shown in Table 1. Nitrification was 122 inhibited by spiking with allylthiourea (ATU, 5 mg/l) in both conditions A and B. Condition E was 123 only amended with DOM to determine its biodegradability in the microbial enrichment culture. 124 For each condition, biotic batches were conducted in triplicate and abiotic controls in duplicate. 125 Abiotic bottles were prepared by autoclaving for 20 min at 120 °C and adding chemical inhibitors 126 (0.3 g/L HgCl₂ and 1.3 g/L NaN₃) after cooling down. Milli-Q water was added to batches without 127 DOM, ATU or chemical inhibitor addition to maintain the same liquid volume (131.9 ml) and 128 PhAC concentration in all batches.

The batch experiments lasted for 35 days and liquid samples were withdrawn at day 0, 1, 2, 4, 7,
10, 14, 22, 28, and 35. Samples were pre-treated by centrifugation at 10000 rpm for 5 min prior
to UPLC and ion chromatography (IC) analyses. DOM samples were taken at day 0 and 35 from

triplicate batches and mixed to compare the composition variation during the biodegradation.
50 ml liquid samples were filtered using 0.2 µm Isopore[™] membrane filters (Milipore, Ireland)
and stored in centrifuge tubes at -20°C for DNA extraction. Based on pre-test of ATU efficiency,
at day 23 ATU was re-spiked to maintain nitrification inhibition.

136 2.3. DOM extraction, fractionation and characterization

137 DOM was extracted from the sediment (together with litters) in the CW to obtain natural 138 organic matter. The sediment was mixed with Milli Q water (1:1, wet w/w) by shaking at 180 139 rpm for 1 h. The mixture was centrifuged at 3750 rpm for 15 min and the supernatant obtained 140 was further centrifuged at 10000 rpm for 20 min. Afterwards the supernatant was filtered 141 through 0.45 μm mixed cellulose ester filters (ME25/21, Whatman, USA) to collect the DOM. For 142 the 35 days batch experiments, DOM samples were collected at day 0 and 35 and fractionated 143 to get the fractions including HA, FA, hydrophilic compounds (Hy), and hydrophobic neutrals 144 (HON), according to the procedure reported by Van Zomeren and Comans (2007). Work-flow of 145 the procedure is summarized in Fig. S1. Recovery of the fractionation procedure was 84-96% in 146 this work. For characterization, the DOM samples were first freeze-dried to solid material by an 147 alpha 2-4 LD+ freeze dryer (Martin Christ GmbH, Germany) and then characterized by using a 148 Bruker Tensor 27 Fourier transform infrared spectrometer (FTIR) equipped with a Platinum ATR 149 accessory (Bruker, Germany).

150 2.4. Chemical analysis

151 Concentrations of PhACs were measured by a UPLC (Ultimate 3000, Thermo, USA) equipped 152 with diode array detection, as previously described (He et al., 2016). Detection limits of the 153 PhACs analyses were 50 μ g/l. Quantification of PhACs was performed based on the internal 154 standard (FEN) and external calibration standards. Concentrations of NO₂⁻ and NO₃⁻ were analyzed by an IC (Dionex ICS-2100, Thermo, USA) as described in the work of Wilt et al. (2018).
Concentrations of O₂ and CO₂ were followed by gas chromatography (GC-2010, Shimadzu,
Japan) according to the method described previously (Wilt et al., 2018). A Sievers[™] 900 Series
TOC Analyzer (GE Analytical Instrument, USA) was used to quantify the concentrations of DOM
and its fractions.

160 2.5. Molecular analysis

161 DNA was extracted from the filtered samples using PowerSoil® DNA Isolation Kits (MoBio 162 Laboratories, USA) according to the manufacturer's protocol. Concentration and purity of the 163 extracted DNA were estimated by a Nanodrop spectrophotometer (Denovix, DS-11, USA). The 164 DNA extracts were stored at -20 °C until further analysis. Quantitative PCR (qPCR) was used to 165 quantify the abundances of bacterial 16S rRNA gene and AOB amoA genes. The qPCR 166 measurements were performed in triplicate by a CFX384 touch real-time PCR detection system 167 (Bio-Rad, Veenendaal, the Netherlands). In the reaction mixture, $2 \mu l$ diluted sample (10 times 168 dilution) was mixed with 2 µl master mix comprising of iQ[™] SYBR[®] Green (Bio-Rad), primers 169 (forward and reverse), and DNase-free water (50:2:2:26). Standards and blanks were prepared 170 similarly to the samples. Standard curves were obtained by using serial dilutions of a known 171 amount of plasmid DNA containing a fragment of the respective genes. Detailed information of 172 the primers and cycling conditions used for qPCR analyses are summarized in Table S2. 173 Abundance of amoA was represented in absolute concentration and 16S rRNA gene-normalized 174 values.

175 3. Results and discussion

176 3.1. Biodegradation of PhACs

177 Biodegradation of the seven PhACs tested showed remarkable differences. For example, in 178 condition C with nitrification present, half of MET was removed and CAF, NAP, and IBP were 179 completely removed within 35 days (Fig. 1c). This can be ascribed to biodegradation as no 180 obvious removal was found in abiotic controls (Fig. S2). Our study obtained comparable or 181 higher removal for the biodegradation of CAF, MET, and NAP than biodegradation observed in 182 microbial cultures of soil and river sediments (Koba et al., 2016; Lin and Gan, 2011; Zhang et al., 183 2013). A CW sediment-based research presented that 50% of IBP can be removed within nine 184 days resulting from sorption onto sediment and biodegradation by natural microbial population 185 (Conkle et al., 2012). In comparison, our sediment-free enrichment cultures gave a 50% 186 biodegradation approximately within 22 days. Similar with our findings (Fig. S3), PRO, CBZ, and 187 DFC were reported to be persistent during biodegradation tests (Tixier et al., 2003; Xu et al., 188 2016; Zhang et al., 2008). The same conclusions can be drawn for the biodegradability of the 189 seven PhACs in conditions A (nitrification inhibition), B (nitrification inhibition and DOM 190 addition), and D (with both DOM and nitrification present), as indicated in Fig. 1. Especially, in 191 conditions B and D with DOM addition, no obvious sorption of PhACs onto the extracted DOM 192 was found in the abiotic controls (Fig. S2). In Fig. 1 the notable increase of IBP concentrations in 193 between may be attributed to detection deviations. PRO, CBZ, and DFC will not be discussed in 194 the following content considering their insignificant biodegradation (<15%, Fig. S3).



Fig. 1. Biodegradation of a mixture of seven PhACs in the microbial enrichment culture under different conditions: conditions A (nitrification inhibition), B (nitrification inhibition and DOM addition), C (with nitrification present), and D (with both DOM and nitrification present). Results are mean values ± standard deviation (n = 3).

201 Microorganisms seem to have a preferential order when biodegrading PhACs, i.e. less 202 biodegradable PhACs were only biodegraded once the readily degradable PhACs were removed. 203 As shown in Fig. 1c, CAF was first biodegraded during day 0-2 followed by MET during day 2-14, 204 and finally NAP and IBP during day 14-28 in condition C. The notable standard deviation of MET 205 biodegradation was due to different biodegradation activity in our triplicate batches, but the 206 biodegradation tendency in the triplicates were consistent where MET was not further 207 biodegraded after day 14 (Fig. S4a). The microbial biodegradation preference was also observed 208 for the other conditions: CAF > MET > NAP > IBP (Fig. 1). Therefore, our study demonstrates a 209 clear microbial preference in the order of PhACs biodegraded regardless of conditions with or





Fig. 2. Microbial respiration in the five conditions: a) consumption of O_2 and b) production of O_2 . Results are mean values ± standard deviation (n = 3).

214 PhACs could be a sole carbon substrate to support microbial growth, as seen in condition A. 215 Looking at the microbial respiration, O₂ was consumed and CO₂ was produced from PhACs in 216 condition A (Fig. 2). Moreover, the amount of total bacteria in this condition increased from 217 1×10^4 to 4.7×10^4 copies/ml after 35 days incubation (Fig. S5a). Both the microbial respiration 218 and growth supported by PhACs indicate that PhACs could be used as a sole carbon substrate, 219 and results in a selective pressure to enrich microorganisms capable of biodegrading PhACs. 220 Similarly, ketoprofen was reported to act as a sole source of carbon and energy for 221 microorganisms in a membrane bioreactor (Quintana et al., 2005). These findings demonstrate 222 the possibility to enrich microorganisms capable of biodegrading PhACs for the use of enhanced 223 biodegradation technologies such as bioaugmentation. lasur-Kruh et al. (2011) successfully enriched and isolated a pure culture of *Novosphingobium JEM-1* from CW sediment and bioaugmented the culture to CW sediment film to achieve 40% removal of estron (E1) and 17estradiol (E2) in batch experiments.

3.2. Effect of DOM on PhAC biodegradation

228 First, the characteristics of the DOM extracted from CW sediment were investigated. The DOM 229 fractions comprised 5% HA, 35% FA, 27% Hy, and 33% HON (Fig. 3, condition E with only DOM 230 present). As presented in the FTIR spectra (Fig. S6), the main absorption bands of the extracted 231 DOM corresponded to the following functional groups: 1) 3390 cm⁻¹ (O-H stretching of inter- and 232 intra- molecular hydrogen bonds); 2) 1639 cm⁻¹ (aromatic C=C skeletal vibrations, asymmetric 233 stretching of C=O, symmetric stretching of COO-); 3) 1359 cm⁻¹ (C-O asymmetric stretching of 234 COOH conditions); 4) 1105 cm⁻¹ (C-O stretching of alcohols and carbohydrate) (Chen et al., 2002; 235 Traversa et al., 2014).

236 Second, the biodegradability of the DOM was evaluated by microbial respiration and 237 fractionation studies. DOM was rapidly mineralized to CO_2 in the first two days followed by a 238 relatively slow mineralization rate in the following days (Fig. 2b, condition E). A similar rapid CO_2 239 production was also found in conditions B and D with DOM present. Furthermore, we confirmed 240 that the significant CO₂ production in the first two days was related to DOM mineralization 241 rather than DOM breakdown due to abiotic process (results not shown). The observed fast 242 production of CO₂ may indicate that the microbial enrichment culture first rapidly consumed 243 relatively easily biodegradable DOM, followed by utilization of the less biodegradable DOM with 244 a slower rate. This microbial preference towards carbon substrate was reported in previous 245 work, where the authors found that Gram-negative bacteria in the soil prefer to use fresh plant-246 derived carbon while Gram-positive bacteria use more older soil organic matter-derived carbon

independent on the soil depth (Kramer and Gleixner, 2008). The microbial biodegradation
preference for DOM compositions is similar to the microbial preference found among the PhACs.



Fig. 3. Composition of DOM fractions at day 0 and 35 in conditions with DOM present (HA = humic acid, FA= fulvic acid, Hy = hydrophilic compounds, HON = hydrophobic neutrals).

252 In terms of the microbial respiration, the easily biodegradable DOM contributes to half of the 253 CO₂ production and the less biodegradable DOM is responsible for the other half. In total 22% 254 DOM was consumed during 35 days incubation, with HA and FA being the main metabolized 255 DOM fractions (Fig. 3, respectively 46% and 49% consumed). This is in line with a previous 256 modelling study, where HA and FA are mentioned as significant DOM fractions for microbial respiration (Straathof, 2015). The functional groups of 3390 and 1639 cm⁻¹ in DOM disappeared 257 258 after 35 days biodegradation (Fig. S6), which indicates that the microbial enrichment culture is 259 capable to transform DOM with those structures.

260 3.2.1. DOM inhibits biodegradation of CAF

249

261 The effect of DOM on PhAC biodegradation was identified by comparing the biodegradation 262 efficiencies of PhACs between conditions A and B as well as between conditions C and D. In the 263 presence of DOM, CAF biodegradation was delayed (Fig. 1). CAF in conditions A and C without 264 DOM addition was completely biodegraded within 2 days. This coincides with the time to 265 consume the easily biodegradable DOM (Fig. 2). Furthermore, CAF contains functional groups of 266 asymmetric stretching of C=O (1639 cm⁻¹) (Gunasekaran et al., 2005). These functional groups 267 are also present in our extracted DOM and they disappeared during 35 days of incubation due to 268 biodegradation (Fig. S6). Therefore, we assume that the easily biodegradable DOM competed 269 with the readily biodegradable CAF resulting in a delay in CAF biodegradation till day 4 in the 270 presence of DOM (Fig. 1b and 1d). The competition between the easily biodegradable DOM and 271 CAF also indicates a microbial biodegradation preference, which is similar to the preferences 272 found among PhACs and among DOM compositions.

273 3.2.2. DOM enhances biodegradation of MET, NAP, and IBP

As opposed to CAF, biodegradation of MET, NAP, and IBP was enhanced by DOM addition. After the easily biodegradable DOM and the readily biodegradable CAF was utilized, microorganisms started degrading the other PhACs. A faster biodegradation of MET, NAP, and IBP was found in conditions B and D compared to conditions A and C in general (Fig. 1). As with MET in condition C, the standard deviation of IBP biodegradation at day 22 was caused by different biodegradation activity in triplicate batches (Fig. S4b), but in general the IBP biodegradation in this condition was slower than in condition D (Fig. 1c and 1d).

More biomass was found in the conditions with DOM addition (Fig. S5a, B > A, D > C). Furthermore, the extracted DOM can act as a substrate or a co-substrate together with PhACs for the microbial enrichment culture as indicated by the increased CO₂ production in conditions

B and D compared to conditions A and C (Fig. 2b). The observed higher biomass production and higher microbial respiration with DOM addition indicate a more active microbial community that may have increased PhAC biodegradation capacity. Rauch-Williams et al. (2010) hypothesized that removal of organic contaminants including NAP and IBP was caused by a specialized microbial community growing on hydrophobic acids, i.e. HA and FA, as in our case.

289 There appears to be other hypotheses to explain this DOM enhanced PhAC biodegradation. First, 290 DOM may act as structural analogues to enhance PhAC biodegradation. By biodegrading DOM 291 with similar molecular structures to PhACs, microorganisms may activate more enzymes that 292 can readily biodegrade PhACs with a similar structure. This structural analogue theory has been 293 applied for the biodegradation of organic compounds such as pesticides, or cyclic ethers (Foght 294 et al., 2001; Kim et al., 2008). Second, DOM can act as electron shuttles to stimulate 295 microorganisms to degrade PhACs. DOM is a complex mixture of organics with many functional 296 groups that could be potentially functional in redox reactions (Van Trump et al., 2006). For 297 example, quinones in DOM (especially in HA) can act as redox-mediating functional groups to 298 stimulate the electron transfer between electron donors and electron acceptors and thus 299 enhance redox reactions (Martinez et al., 2013).

It has been confirmed that biodegradation of certain PhACs are dependent on the availability
and concentrations of DOM (Lim et al., 2008; Maeng et al., 2011). However, there are no studies
that explicitly describe the mechanism of DOM in the enhancement of PhAC biodegradation.
Further research efforts need be dedicated to understand the underlying mechanisms of these
processes.

305 3.3. Effect of nitrification on PhAC biodegradation

306 Nitrification occurred in conditions C and D whereas it was inhibited in conditions A and B by the 307 addition of ATU. 70 % of the ammonium in batches was converted to NO_3^{-1} in conditions C and D 308 along with biodegradation of PhACs. NO₃⁻ was not formed in the other two conditions (Fig. S7). 309 In the q-PCR tests, amoA abundance was higher in conditions C and D as compared to the other 310 two conditions, indicating an enrichment of AOBs (Fig. S5b). The detection efficiency of amoA 311 absolute abundance in conditions B and D might be influenced by the presence of humic acids 312 due to inhibition of enzymatic reaction (Fig. S5b), as reported by Schriewer et al. (2011). 313 However, this effect in those two conditions was absent after normalization for abundance 314 relative to the 16S rRNA gene abundance (Fig. S5b). In our nitrification conditions, nitrification plays an important role in microbial respiration as it stoichiometrically accounts for 39% and 38% 315 316 of the O_2 consumption for NO_2 and NO_3 production in conditions C and D, respectively.

317 3.3.1. Nitrification enhances biodegradation of NAP and IBP

The effect of nitrification on PhAC biodegradation is seen by comparing the biodegradation efficiencies between conditions A and C and between conditions B and D. Nitrification enhanced the biodegradation of NAP and IBP. Significantly higher biodegradation of NAP and IBP was found in condition C than in condition A (Fig. 1). No difference was observed between conditions B and D, which were both amended with DOM.

The enhancement by nitrification is likely a result of AOB driven co-metabolism. As demonstrated by Fernandez-Fontaina et al. (2016), biodegradation of NAP and IBP was enhanced by AOB instead of contribution from heterotrophic microorganisms. AOB containing AMO can easily hydroxylate linear alkyl carbons especially for the secondary and tertiary carbons in the chain, due to its low specificity and broad substrate spectrum (Fernandez-Fontaina et al., 2016). Such structures can be found in NAP and IBP (Fig. 5, circles). Co-metabolic

function of AOB was reported for many other PhACs in nitrifying bioreactors and membrane bioreactors (Roh et al., 2009; Yi and Harper, 2007). It has been demonstrated that sludge-based wastewater treatment processes with significant nitrification showed higher removal efficiencies than those without nitrification for 90% of the investigated PhACs (Xu et al., 2016). Comparable with sludge-based microbial cultures, our findings demonstrate a positive effect of nitrification on PhAC biodegradation also in the CW sediment derived microbial cultures.

335 In our case, nitrification not only enhanced biodegradation of NAP and IBP but also enhanced 336 mineralization of PhACs, as illustrated in the increased CO₂ production in condition C compared 337 to condition A (Fig. 2b). In addition to increase PhAC mineralization, nitrification also stimulated mineralization of less biodegradable DOM. DOM fractionation reveals that condition D with 338 339 nitrification showed more biodegradation of HA, Hy, and HON, compared to conditions B (Fig. 340 3). The more biodegraded DOM in condition D was mainly mineralized to CO₂, as illustrated by 341 comparing the carbon balances between conditions D and B. In condition D, more DOM was 342 consumed than in condition B (equal to 0.85 mmol CO_2/I medium, Fig. 3) and more CO_2 was 343 produced (0.70 ± 0.04 mmol/l medium, Fig. 2b). The enhanced biodegradation of less 344 biodegradable DOM might be due to co-metabolism via AOB and/or enhanced heterotrophic 345 microbial activity. Microbial CO₂ production was higher in conditions C and D with nitrification 346 than in conditions A and B, respectively (Fig. 2b), which indicates higher heterotrophic microbial activity under nitrifying conditions. The higher heterotrophic microbial activity might contribute 347 348 to the higher mineralization of PhACs and less biodegradable DOM under nitrifying conditions.

349 3.3.2. Nitrification inhibits biodegradation of MET

350 MET biodegradation was inhibited in the presence of nitrification, as indicated in Fig 4. To date, 351 there is no literature referring to inhibition of PhAC degradation in nitrifying cultures; only

enhanced PhAC biodegradation has been reported, as described at 3.3.1, or no effect on PhAC biodegradation has been observed (Khunjar et al., 2011). To confirm the unexpected finding, we repeated our experiment with MET and the same conclusion was obtained (Table S3). The inhibitory effect of nitrification on MET is not yet well understood. Further research is needed to elucidate the observed discrepancies of nitrification effect on NAP and IBP, and MET.

Compared to the other three biodegradable PhACs, biodegradation of CAF was not obviously
influenced by nitrification. The reason might be that CAF was readily biodegradable and thus an
effect of nitrification could not be observed.

360 3.4. Implications for theory and practice

361 In summary, the effect of DOM and nitrification on the biodegradation of PhACs is shown in Fig. 362 5. Translating our findings to practice, first the clear biodegradability order among PhACs can 363 provide an operational strategy for designing hydraulic retention time (HRT) for removing 364 specific PhACs. Second, DOM addition could be a strategy to enhance the aerobic biodegradation of less biodegradable PhACs in CWs. Instead of harvesting dead plants 365 seasonally and transporting them for special disposal, dead plant biomass can be (partially) 366 367 disposed directly into a CW to obtain more natural DOM. Third, nitrification is beneficial for 368 biodegradation of PhACs with certain structures. However, ammonium concentration is 369 normally low in the influent of CWs used as post-treatment, resulting in limited nitrification. A 370 sub-stream of wastewater influent of WWTPs rich in ammonium could be introduced to CWs to 371 strengthen the biodegradation of PhACs of interests.



Fig. 5. Proposed mechanisms of the effect of DOM and nitrification on biodegradation of caffeine, metoprolol, naproxen, and ibuprofen. DOM pathway represented by solid lines; nitrification pathways represented by dashed lines. Directions of arrows show enhancement (up), inhibition (down), and no effect (horizontal). The red line represents the biodegradability order of the microbial enrichment culture. Red circles on the structure of NAP and IBP label the secondary and tertiary carbons in linear alkyl chains for hydroxylation by ammonia oxidizing bacteria (AOB).

380 4. Conclusions

372

This study contributes to the understanding of the impact of DOM on aerobic biodegradation of PhACs and the role of nitrification in PhAC biodegradation in enrichment cultures originating from CW sediment.

Our enrichment culture showed different biodegradation capacities for the seven PhACs
 that all had different physiochemical properties. Aerobic microbial communities appear

to have a biodegradation preference for the biodegradable PhAC in the order of CAF > MET > NAP > IBP. It is feasible to use PhACs as a sole carbon substrate as a selective pressure to enrich microorganisms capable of biodegrading PhACs for the use of enhanced biodegradation technologies.

With DOM present, the easily biodegradable DOM was rapidly consumed followed by
 less biodegradable DOM. Biodegradation of CAF competed with easily biodegradable
 DOM during the first two days. In contrast, DOM enhanced the biodegradation of MET,
 NAP, and IBP, potentially by either contributing more biomass capable of degrading
 PhACs, acting as structural analogues, or acting as electron shuttles to stimulate
 microorganisms to metabolize or co-metabolize PhACs. The exact mechanisms need to
 be further identified.

Nitrification enhanced the biodegradation of NAP and IBP which may result from AOB
 driven co-metabolism. MET biodegradation was inhibited in the presence of nitrification.
 In comparison, no difference was observed for the readily biodegradable CAF in the
 presence or absence of nitrification.

For application, the clear biodegradability preference found among PhACs can offer a
 strategy to design different HRTs to remove corresponding PhACs. From the operational
 perspective, DOM additions and introducing consistent ammonium streams may be
 used to increase the potential of CWs to achieve higher PhAC biodegradation as post treatment processes.

406

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