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Mechanically and Electrically Robust Self-Assembled Monolayers for Large-Area Tunneling Junctions

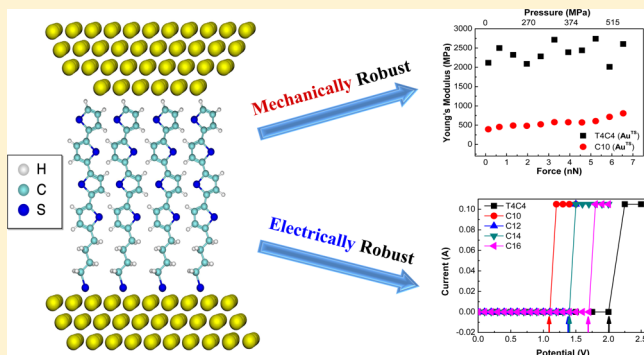
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S Supporting Information

ABSTRACT: This paper examines the relationship between mechanical deformation and the electronic properties of self-assembled monolayers (SAMs) of the oligothiophene 4-([2,2':5',2'':5'',2'''-quaterthiophen]-5-yl)butane-1-thiol (T4C4) in tunneling junctions using conductive probe atomic force microscopy (CP-AFM) and eutectic Ga–In (EGaIn). We compared shifts in conductivity, transition voltages of T4C4 with increasing AFM tip loading force to alkanethiolates. While these shifts result from an increasing tilt angle from penetration of the SAM by the AFM tip for the latter, we ascribe them to distortions of the π system present in T4C4, which is more mechanically robust than alkanethiolates of comparable length; SAMs comprising T4C4 shows about five times higher Young's modulus than alkanethiolates. Density functional theory calculations confirm that mechanical deformations shift the barrier height due to changes in the frontier orbitals caused by small rearrangements to the conformation of the quaterthiophene moiety. The mechanical robustness of T4C4 manifests as an increased tolerance to high bias in large-area EGaIn junctions suggesting that electrostatic pressure plays a significant role in the shorting of molecular junctions at high bias.



INTRODUCTION

Emerging challenges in information technology are driving research into new computer architectures and circuit designs¹ that require new materials and concepts in nanoelectronics. Molecular electronics, specifically tunneling junctions comprising discrete molecules, are well suited to address these challenges^{2,3} because they control charge-transport directly at the quantum level, however, it remains impractical to integrate single-molecule junctions^{4,5} into devices. Bottom-up junctions comprising self-assembled monolayers^{6–8} (SAMs,) on the other hand, can already be incorporated into wafer-scale fabrication processes⁹ and diode logic circuits.¹⁰ When molecules pack into a SAM, collective effects, such as the overlap of interacting electronic states and charges, give rise to new properties that affect tunneling charge-transport significantly as compared to single-molecule junctions.^{11–14} In addition to electronic and electrostatic effects, SAMs exhibit mechanical properties derived from the interactions between individual molecules, which play a critical role in the tolerance of SAMs toward particular top-contacts and, ultimately, technological applications. Small changes in the conformation of a molecule or ensemble of molecules (e.g., in a SAM) between two electrodes can have dramatic effects on conductance by altering electronic states in the metal/molecule/metal junction.^{12,15} Large-area junctions are typically constructed using SAMs of molecules with anchoring groups

such as thiols that drive self-assembly into ordered structures that impose fixed conformations. The effects of these conformations and their relationship to the bulk mechanical properties of the SAM can not be ignored, particularly for π -conjugated molecules, since intermolecular interactions affect charge transport via electrostatic effects and because both hopping and tunneling charge-transport are sensitive to electronic delocalization, which is maximized in coplanar conformations. Establishing a structure–function relationship between mechanical deformation and electrostatics in SAMs of π -conjugated molecules is, therefore, important fundamentally and for the potential for exploitation in molecular-scale devices that are sensitive to force/pressure/deformation.

Conductive probe atomic force microscopy (CP-AFM) is capable of characterizing the electrical properties of SAMs while varying the loading force applied to a probe tip that doubles as a top-contact. Changes in the resistances of SAMs of alkanethiolates with applied force have been ascribed to changes in the tilt angle of the alkane chains.^{16–20} Transition voltage spectroscopy (TVS) indicates that the transition voltage V_{trans} shifts to a lower bias with increasing force (i.e., as the tilt angle increases).^{19,20} This observation implies a decrease in

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barrier height of molecular junctions because V_{trans} is proportional to the energy offset between the Fermi level E_f and the highest-occupied molecular state.²¹ If there are more subtle influences to the electrostatics of the junctions from bond distortions they are masked by the larger effect of the tilt angle increasing as the tip penetrates the SAM, which is stabilized only by relatively weak intermolecular dispersion forces. The mechanical properties of SAMs of π -conjugated molecules have not been similarly investigated. Thus, we synthesized 4-([2,2':5',2'':5'',2'''-quaterthiophen]-5-yl)butane-1-thiol (T4C4), a molecule containing both a flexible butanethiol chain to facilitate the formation of a densely packed monolayer and a rigid quarterthiophene to impart mechanical stability through relatively strong π - π interactions. The molecular structure and the geometry of a CP-AFM junction are shown in Figure 1. We studied the mechanical and

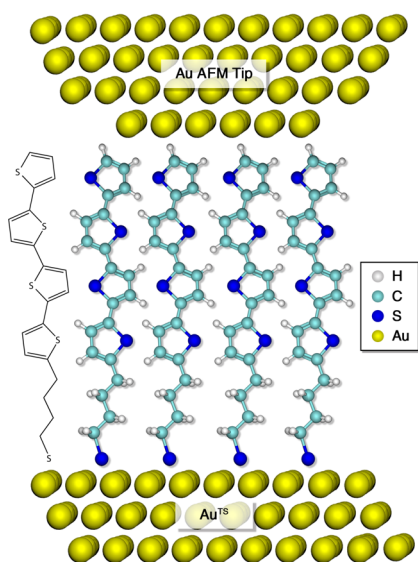


Figure 1. Representative schematic of molecular junction comprising T4C4 with a Au coated CP-AFM tip as top electrode and template-stripped Au as bottom electrode.

electrical properties of SAMs of T4C4 using CP-AFM and density functional theory (DFT). They are quantitatively more robust than SAMs of alkanethiolates, but the electrostatics of the junction respond to small distortions of the π -system. This robustness translates into junctions that are capable of withstanding larger bias windows than alkanethiolates in large-area junctions using eutectic Ga–In (EGaIn) top-contacts.²²

METHODS

Self-Assembled Monolayers. The formation of SAMs follows the reported methods.²³ The Ag and Au substrates were prepared by template stripping (TS) described in details elsewhere.²⁴ 200 nm of Ag (99.99%) and 100 nm of Au (99.99%) were deposited by thermal deposition at 10^{-7} mbar onto a 3" Silicon wafer (without adhesion layer). Glass substrates (1 cm \times 1 cm) were glued onto deposited metal by using UV-curable optical adhesive (Norland 61) with 300 s exposure of UV. All samples were made by incubation of freshly cleaved silver and gold substrates into either 3 mM solution of the corresponding n -alkanethiols ($n = 10, 12, 14, 16$) in ethanol or 0.5 mM solution of T4C4 in toluene at room temperature

and kept inside a nitrogen flow box (in which the O_2 was below 3% and the humidity was below 10%) for 3 h. Then the substrates of alkanethiols and T4C4 were rinsed by ethanol and toluene respectively and dried gently by N_2 . Prior to forming the SAMs, the solution was degassed by bubbling N_2 for at least 20 min and all solution were kept under N_2 atmosphere to prevent oxidation of thiol and Ag substrates. Characterization of the SAMs by water contact angles, ellipsometry, and ultraviolet photoelectron spectroscopy (UPS) is described in [Supporting Information](#).

Current–Voltage Measurements. CP-AFM. I – V measurements were performed on a Bruker AFM Multimode MMAFM-2 equipped with a Peak Force TUNA Application Module (Bruker). The SAMs were contacted with an Au-coated silicon nitride tip with a nominal radius of 30 nm (NPG-10, Bruker; tip A, resonant frequency = 65 kHz, spring constant = 0.35 N/m; tip B, resonant frequency = 23 kHz, spring constant = 0.12 N/m; tip D, resonant frequency = 18 kHz, spring constant = 0.06 N/m; tip A was chosen in this work) in TUNA mode. The AFM tip was grounded and for all loading forces, T4C4 on Au^{TS} were biased from -1.0 to $+1.0$ V and from $+1.0$ to -1.0 V while C10 on were biased from -1.5 to $+1.5$ V and from $+1.5$ to -1.5 V on Au^{TS} to record the I – V curves: a max of 10 trace/retrace cycles per junction were performed and the top electrode was removed from SAMs between junctions. Between different samples, a new tip was used. The total number of I – V traces recorded by CP-AFM is summarized in the [Supporting Information](#), Table S3. It is difficult to determine V_{trans} for an individual I – V trace due to the inherent noise in the raw data. The peaks of Gaussian fits of histograms of I for each value of V at different loading forces obtained by CP-AFM were plotted and transformed into axes of $\ln(I/V^2)$ versus $1/V$. The position of the V_{trans} was determined manually by the center of the dips in the plots.

EGaIn. The electrical measurement with the EGaIn, as well as sample preparation and handling, was performed in the nitrogen flow box in which O_2 was maintained below 3% and the humidity was kept below 10%. At least two samples were examined for each molecule. The potential windows include the following: (1) 0 V \rightarrow 1 V \rightarrow -1 V \rightarrow 0 V, steps of 0.05 V; (2) 0 V \rightarrow 1.5 V \rightarrow -1.5 V \rightarrow 0 V, steps of 0.1 V; (3) 0 V \rightarrow 2 V \rightarrow -2 V \rightarrow 0 V, steps of 0.1 V; (4) 0 V \rightarrow 2.5 V \rightarrow -2.5 V \rightarrow 0 V, steps of 0.25 V. A total of 20 trace/retrace cycles were measured for each junction, and shorts occurring during the cycles are counted for the failure of junction. The statistics of the stability test on EGaIn junctions are summarized in the [Supporting Information](#), Table S5.

PeakForce QNM. Measurements of Young's modulus were performed on a Bruker Multimode MMAFM-2 in PeakForce QNM mode. The tip used in the measurement was ScanAsyst-Air from Bruker (resonant frequency, 70 kHz; spring constant, 0.4 N/m; tip radius, 2 nm). Deflection sensitivity was calibrated by measuring 5 force curves on fused silica sample provided by Bruker and taking the average of the results. Spring constant was calibrated by thermal tune before and after the measurements. Tip radius was calibrated before and after the measurements using scanning electron microscope (SEM). Deformation under each force load was measured from five spots of the sample and averaged. Young's modulus was calculated by the DMT model from the averaged deformation of each force load.

DFT Calculations. All calculations were performed using Gaussian 09 software. Structures were first minimized by

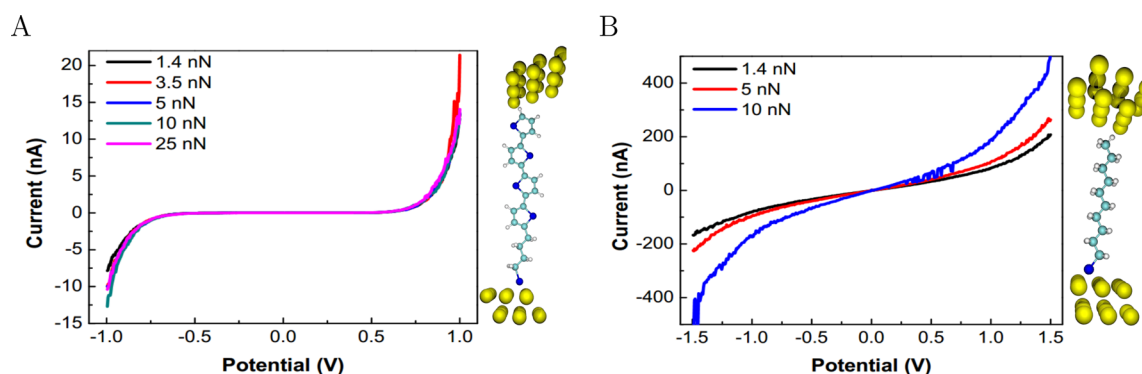


Figure 2. (A) I - V plots of T4C4 with different forces: black for 1.4 nN, red for 3.5 nN, blue for 5 nN, dark cyan for 10 nN, and pink for 25 nN. (B) I - V plots of C10 with different forces: black for 1.4 nN, red for 5 nN, and blue for 10 nN. Both SAMs were measured on Au^{TS} substrates by CP-AFM.

B3LYP/6-311G*; then, the single-point energies were computed by B3LYP/LANL2DZ. Single-molecule junctions were constructed by attaching the minimized structures to 17- or 18-atom Au(111) clusters via the terminal sulfur atoms at hexagonal close-pack hollow sites at a distance of 1.75 Å from the center of the hollow site. To calculate the energy difference $E_f - E_{HOPS}$ and $E_f - E_{HOSS}$, the value of E_f for Au electrodes was taken to be -4.2 eV from the UPS measurements.

RESULTS AND DISCUSSION

We synthesized T4C4, a “ σ - π ” hybrid molecular structure containing both alkyl (σ) and thiophene (π) moieties according to the protocol described in the Supporting Information. We chose T4C4 because it is known to form densely packed SAMs.^{25–27} For comparison to previously reported mechanical studies, we used decanethiol (C10). We chose C10 specifically because the properties of SAMs of C10 have been studied extensively by CP-AFM.^{16,28,29} We prepared both template-stripped gold (Au^{TS}) and silver (Ag^{TS}) ultrasmooth electrodes following literature procedures template-stripping.²⁴ These substrates are particularly well suited to large-area junctions³⁰ and are compatible with CP-AFM.¹²

CP-AFM Measurements. We formed metal-molecule-metal junctions by placing the gold coated CP-AFM tip (denoted Au^{AFM}) with spring constant of 0.35 N m⁻¹ and radius of 30 nm in a stationary point contact with the SAM under a controlled tip loading force, which translates into an applied pressure that depends on the radius of the tip; CP-AFM tips are larger than ordinary Si₃N₄ tips due to the additional metallic layers. We refer to the molecular junctions as Au^{TS}/SAM//Au^{AFM}, where “/” and “//” denote a covalent interface and a van der Waals contact, respectively. We measured the I - V characteristics of Au^{TS}/C10//Au^{AFM} and Au^{TS}/T4C4//Au^{AFM} junctions at low applied forces, which we define as 25 nN or less. Characteristic data are shown in Figure 2 for C10 and T4C4. (The I - V curves for C10 at and above 25 nN shorted when bias was applied and are, therefore, omitted from the figure.) The I - V characteristics of C10 were sufficiently similar to published data to validate our measurement technique.^{16–20} The I - V curves of T4C4 did not change at low forces (Figure 2A), passing approximately 10 nA at 1 V. The I - V curves of C10, however, varied by about a factor of 2, passing approximately 200 nA at 1 V with a force of 10 nN and 100 nA at 1.4 nN. We were only able to measure T4C4 to ± 1 V without saturating the current amplifier, while we were able to measure C10 to ± 1.5 V using the low-gain amplifier because

the absolute current in the intermediate-bias regime (i.e., where the I - V dependence becomes exponential) increases more slowly for C10 than for T4C4. To verify the reproducibility of the data in Figure 2, we performed a statistical analysis by constructing histograms of I at ± 1 V (Figures S6 and S7) and fitting them to Gaussian distributions. The data points with error bars reported throughout the main text and the Supporting Information are the peak and standard deviations, respectively, derived from such histograms.

The I - V curves of T4C4 are sigmoidal, passing nearly invariant, low current in the linear, low-bias regime (below 0.5 V) and increasing dramatically in the exponential, intermediate-bias regime, which is consistent for π -conjugated (or σ - π) “molecular wire” molecules.³¹ The I - V curves of C10 are sigmoidal, but increase throughout the low-bias regime, which is consistent for alkanethiols.^{16,32} The evolution of the I - V curves with increasing loading force can be caused by any combination of three factors: (1) the molecular tilt increases; (2) molecules in the SAM are deformed; and (3) the contact area increases. As mentioned above, the response of C10 is attributed mainly to the tilt angle, but T4C4 showed no change at forces up to 30 nN as can be seen in Figure 3. (Note that the dependence of pressure on loading force is nonlinear due to the dependence of contact-area on force, thus, the values across the top X-axis are only meant to show the range of pressures experienced by the SAM; see Supporting Information for

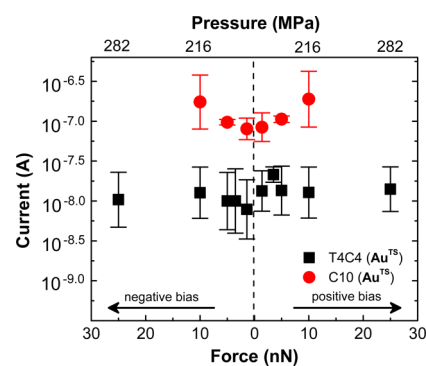


Figure 3. Current at 1.0 V versus loading force on Au^{TS} plots: dark squares for T4C4, red circles for C10. Each data point is the peak of Gaussian fit to a histogram of I at that value of V . The error bars are standard deviations. The values listed on the top X-axis are the pressures calculated explicitly for the corresponding values of force on the bottom X-axis.

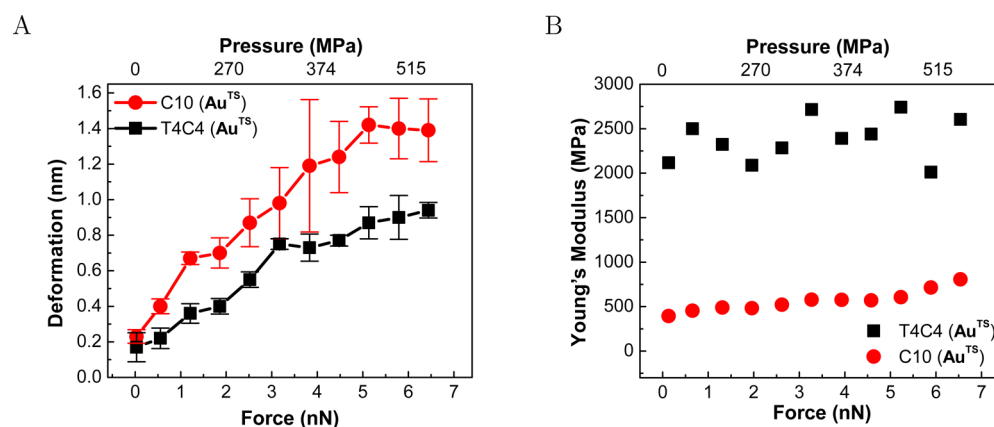


Figure 4. (A) Deformation of SAMs on Au^{TS} : black squares for T4C4; red circles of C10. The error bars are standard deviations. (B) Young's modulus of SAMs: black squares for T4C4; red circles of C10. The values listed on the top X-axis are the pressures calculated explicitly for the corresponding values of force on the bottom X-axis.

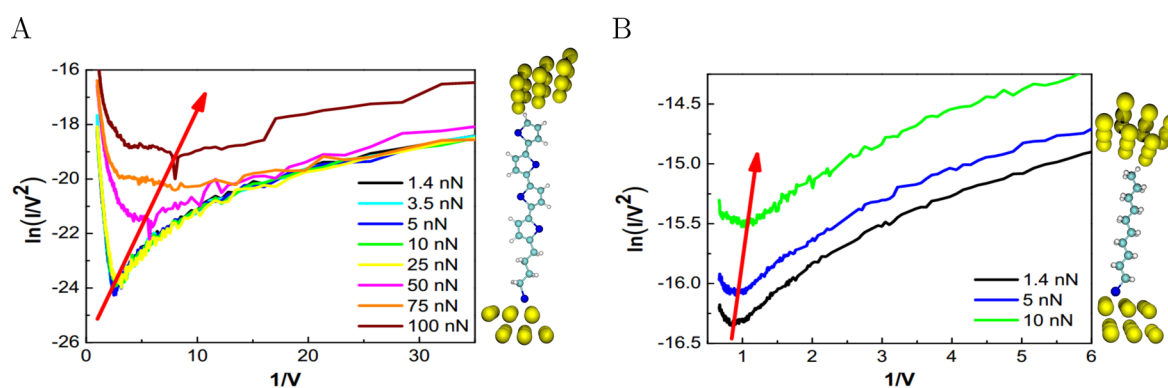


Figure 5. Transition voltage spectroscopy of Au^{TS} /SAM//Au junctions versus loading force from the peaks of Gaussian fits obtained by CP-AFM at each loading force. (A) T4C4, 200 traces at each loading force. (B) C10, 30 traces at each loading force. The equivalent pressures for each force are shown in Table 1.

details.) This semilog scale plot compresses the data somewhat, but there is still a clear, increasing trend for C10 that is absent for T4C4 even up to 30 nN (i.e., three times the loading force). At high force (30–150 nN) the conductivity of T4C4 begins to increase, but C10 either shorts or saturates the current amplifier (both manifest as hitting the compliance limit) above 10 nN (Figure S5). Thus, we measured SAMs of dodecanethiol (C12) in Ag^{TS} /C12// Au^{AFM} and Ag^{TS} /C12//EGaIn junctions in order to compare the effects at high forces. We switched to Ag^{TS} substrates for C12 to facilitate comparisons to literature reports for EGaIn, for which Ag^{TS} is the preferred substrate as described below.^{23,33,34} We also measured Ag^{TS} /T4C4// Au^{AFM} and Ag^{TS} /T4C4//EGaIn junctions for comparison. The increase in current of Ag^{TS} /C12// Au^{AFM} as a function of loading force is even more dramatic than Au^{TS} /C10// Au^{AFM} and Ag^{TS} /C10// Au^{AFM} , spanning 3 orders of magnitude up to 150 nN; see the Supporting Information, Figures S8 and S9, for details.

Mechanical Properties. In addition to I – V measurements via CP-AFM, we measured the mechanical properties of SAMs of T4C4 and C10 on Au^{TS} using PeakForce QNM AFM. Figure 4A shows the deformation as a function of loading force up to 7 nN. To enable a comparison between these data and CP-AFM data, we estimated the pressure applied to the SAM by considering the loading force and the radius of the tip (see the Supporting Information, Table S4). At loading forces below 3 nN, the magnitudes and slopes are similar, but above 3 nN the

displacement of T4C4 begins to level off at approximately 0.8 nm while C10 continues to increase. We hypothesize that the inflection point in the T4C4 curve is caused by compression/deformation of the butyl tail, which deforms at lower loading force than the quarterthiophene unit (but similar to C10). Figure 4B shows the Derjaguin–Muller–Toporov (DMT) Young's modulus (stiffness in the elastic region) over the same range of loading force.³⁵ (There are no error bars because the Young's modulus was calculated from the average deformation of each force load using the DMT model.) The difference is unambiguous; the modulus of T4C4 is five times higher than C10, indicating that SAMs of T4C4 are considerably stiffer than SAMs of C10. Our measured values for C10 are also in good agreement with the moduli for SAMs of alkanethiolates reported previously; 280 MPa for octanethiol (C8) and 860 MPa for C12.³⁶ From the electrical and mechanical measurements, we conclude that SAMs of T4C4 are more mechanically robust than C10, which translates into more stable conductance across a wider range of loading force; however, conductance alone does not provide much insight into the electrostatics of the junctions or address the question of why the I – V characteristics of T4C4 are stable despite deforming considerably at low loading forces.

Transition Voltage Spectroscopy. Transition voltage spectroscopy (TVS) is a useful tool to gain insights into the electrostatics of molecular junctions by providing an indirect measure of ϕ , the offset between E_f and the frontier orbital that

participates most strongly in tunneling transport (the highest-occupied state for both C10 and T4C4, i.e., hole transport). The transition voltage V_{trans} corresponds to the transition from ohmic, low-bias conduction to exponential/hyperlinear conduction at intermediate bias, which can be estimated by replotting the I - V curves and looking for minimums. The value of V_{trans} is proportional to the height of the tunneling barrier imposed by ϕ . Shifts in V_{trans} , therefore, reveal changes to ϕ , which is a function of the electrostatics (i.e., level-alignment) near E_f . These shifts can occur independently of changes in conductance, either because they are below the threshold for detection or are offset by other changes, for example, the barrier width, which is related to the distance between the electrodes and, therefore, decreases as the SAM deforms. To compute V_{trans} , we plotted $\ln(I V^{-2})$ vs V^{-1} using the peaks of Gaussian fits of histograms of I for each value of V at different loading forces (200 traces for T4C4 and 30 for C10) and recorded the center of the dips in the plots. These data are plotted in Figure 5 and summarized in Table 1. At loading forces above 75 nN the dips were not very pronounced, but they were well-resolved at all other forces, revealing clear differences between T4C4 and C10.

Table 1. Measured Values of V_{trans}^+ at Different Loading Forces

pressure (MPa)	force (nN)	V_{trans}^+ (V)	
		T4C4	C10
163.12	1.4	0.4	1.20
177.35	3.5	0.4	–
186.94	5.0	0.4	1.10
215.93	10	0.38	0.95
282.28	25	0.34	–
352.43	50	0.18	–
396.49	75	0.14	–
426.72	100	0.13	–

The trend for C10 shown in Figure 5B and Table 1 is in excellent agreement with literature values; V_{trans}^+ (V_{trans} at positive bias) decreases from a maximum of 1.20 to 0.95 V, a change of approximately 20%. Table 2 compares literature

Table 2. Comparison of V_{trans}^+ of Alkanethiols on Au Substrate at Low Loading Forces to Literature Values

	V_{trans}^+ (V)		
	C8	C10	C12
this work	–	1.20	–
ref 29	1.28	1.27	1.20
ref 37	1.21	–	1.33

values of V_{trans}^+ for C10, C12 and C8 at low loading force to our value for C10; these values, which are typically 1.10 to 1.40 V for alkanethiols, are also in excellent agreement.^{29,37} Because V_{trans} is proportional to ϕ and E_f is invariant (i.e., the value for Au^{TS}), V_{trans} is almost always smaller for π -conjugated molecules than for alkanethiols by virtue of the fact that the HOMO tends to lie closer to E_f .^{38,39} Indeed, V_{trans}^+ for T4C4 is about one-third the value of C10. Moreover, it decreases from 0.4 to 0.13 V, a change of approximately 70% over a range of 1.4 nN to 100 nN. From 1.4 nN to 10 nN, the range over which C10 could be measured, T4C4 only changes by approximately 5%, compared to 20% for C10. Thus, the

changes in conductance in both SAMs correspond to a lowering of the barrier height, but it requires about 1 order of magnitude more loading force to induce a change in T4C4 as compared to C10. Given the substantial differences in chemical structure and mechanical properties, it is unlikely that the cause of the shifts in V_{trans} are the same for T4C4 as they are for C10 (i.e., increased tilt angle).

DFT Calculations. For insights into the electrostatics of SAMs of T4C4 under deformation, we constructed model junctions and computed their properties using DFT. The model junctions consist of single molecules spanning two clusters of Au atoms; these are not meant as direct simulations of $\text{Au}^{\text{TS}}/\text{SAM}/\text{Au}$ junctions, rather, they are computationally accessible models from which we can compute trends to compare to experimental data. First, we optimized the geometry of the molecule in the gas-phase using B3LYP/6-311G*. Given the coplanar geometry of the quarterthiophene moiety and the tendency for alkanes to adopt a trans-extended conformation in SAMs, this geometry is a reasonable approximation for T4C4 in a SAM. Second, we attached a cluster of Au at a hollow site via the thiol anchor on one end and positioned an identical cluster above the terminal thiophene ring/methyl group at the other end. (The Au–S and Au–thiophene distances do have a small effect on the computed electrostatics, but they are kept constant across all calculations such that the effect is constant.)

Finally, we computed point energies using B3LYP/LANL2DZ for the molecule before and after attaching the metal electrodes to compare the orbital energies and isoplots of the molecule in gas phase and in the model junctions, respectively. To model the deformation of the SAM, we distorted the T4C4 molecules in the model junctions systematically either by hand or by using displacements predicted from frequency calculations. The figure of merit of these calculations is the offset between the metal Fermi level and the highest-occupied π -state (HOPS) of T4C4 ($E_f - E_{\text{HOPS}}$), which is a direct approximation of ϕ and, therefore, will vary accordingly with V_{trans} . Because these are Gaussian (i.e., discrete, aperiodic) calculations the “HOMO” corresponds to E_f , thus we locate the HOPS by comparing the model junction to the gas-phase calculation. Figure S11 shows isoplots of the HOPS for T4C4 and the highest-occupied σ -state (HOSS) for C10 obtained using this method.

We estimated $E_f - E_{\text{HOPS}}$ of SAMs of T4C4 on Au^{TS} and Ag^{TS} experimentally from ultraviolet photoelectron spectroscopy (UPS) data according to ref 39 (Table 3). To relate the

Table 3. Energy Levels Determined by UPS

	HOPS (eV)		$E_f - E_{\text{HOPS}}$ (eV)	
	center	onset	center	onset
T4C4 on Au	–5.42	–4.88	1.23	0.68
T4C4 on Ag	–5.25	–4.71	1.31	0.77

DFT calculations to experimental data, we computed $E_f - E_{\text{HOPS}}$ using the value of E_f from UPS and the value of E_{HOPS} from DFT of the minimized geometry of T4C4 in a model Au/T4C4/Au junction. This method produced excellent agreement for $E_f - E_{\text{HOPS}}$ between UPS and DFT.

Figure 6 shows $E_f - E_{\text{HOPS}}$ of model junction as a function of in-plane bending. Unsurprisingly, there is hardly any effect on C10; however, the response of T4C4 is nonlinear, increasing at first and then rapidly decreasing. The initial increase is due to

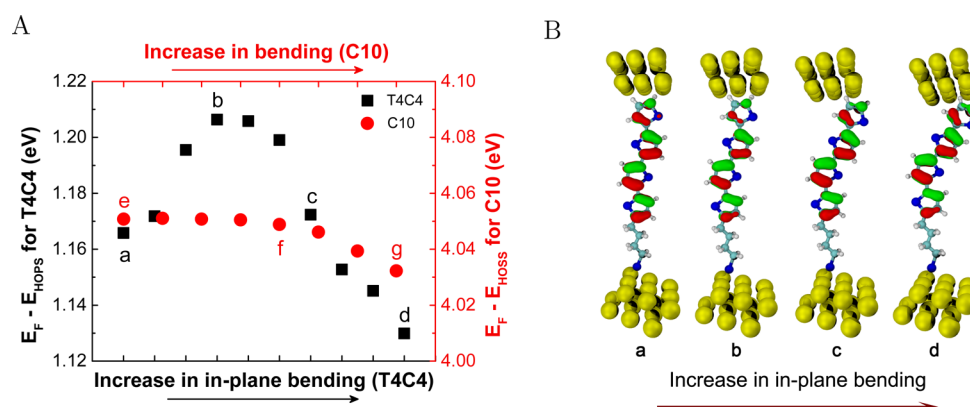


Figure 6. (A) Shift in the energy of $E_f - E_{\text{HOPS}}$ of $\text{Au}^{\text{TS}}/\text{SAM}/\text{Au}$ model junctions with the increased in-plane bending of the T4C4 molecules (black squares), and $E_f - E_{\text{HOSS}}$ with the increased bending of C10 alkanethiol molecules (red dots), relative to their equilibrium geometries. The labeled data points (a, b, c, and d) correspond to the energies of the T4C4 geometries showed in the bottom schematic (B). The first points a and e correspond to optimized geometries of T4C4 and C10 molecules, respectively. The geometries corresponding to the data points e, f, and g and further details are given in the [Supporting Information](#). The E_f of Au electrodes was set to -4.20 eV for these plots from the UPS measurements.

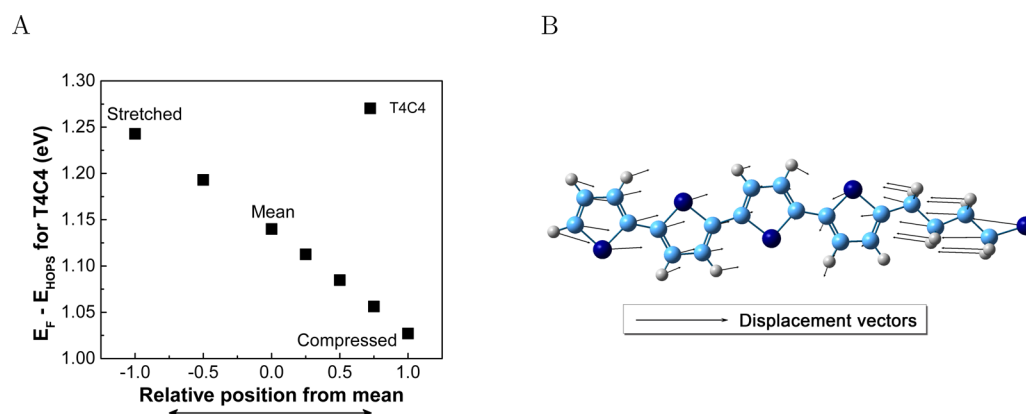


Figure 7. (A) Shift in the energy of $E_f - E_{\text{HOPS}}$ of $\text{Au}^{\text{TS}}/\text{T4C4}/\text{Au}$ model junctions as a function of the displacement of atoms along vibrational vectors from frequency calculations. The points on X-axis span from -1 (fully stretched geometry) to $+1$ (fully compressed geometry), where 0 corresponds to the equilibrium geometry. (B) Arrows represent the displacement vectors of individual atoms as they vibrate. The E_f of Au electrodes was set to -4.20 eV for these plots from the UPS measurements.

the decrease in orbital overlap in the π -system, which lowers the energy of the HOPS (the total energy still increases). It is not clear why $E_f - E_{\text{HOPS}}$ then decreases, but since we did not observe any increase in V_{trans} experimentally, we conclude that in-plane bending (a relatively high-energy deformation, particularly in a SAM) does not play a large role in the mechanical deformation of SAMs of T4C4; we cannot exclude its contribution to C10, however. The exact bond angles of each conformation are shown in [Figure S12](#) and [Tables S6](#) and [S7](#).

In-plane bending is a relatively high-energy process. Deformations in which atoms are allowed to displace along all vibrational vectors are generally lower-energy processes, but are more difficult to rationalize because it translates a compressive force (from the AFM tip) into motion in all directions within a SAM. Nonetheless, molecules of T4C4 stretched and compressed along these vectors show a linear response of $E_f - E_{\text{HOPS}}$ as a function of relative displacement as is shown in [Figure 7](#). This response (as we go from “stretched” to “compressed” forms) also correctly predicts the direction of change in V_{trans} . Given the high Young’s modulus and relatively small tip displacement, we hypothesize that the shifts in V_{trans} for $\text{Au}^{\text{TS}}/\text{T4C4}/\text{Au}$ junctions are, therefore, the result of compressing molecules of T4C4 along displacement vectors

corresponding to vibrational modes that are allowed by the constraints of the SAM. This is a very different mechanism from that of C10 and provides a coherent explanation for the change in conductance that occurs at high loading forces. Other bending and twisting modes yielded either no change or an increase in V_{trans} ; see [Figures S13](#) and [S14](#) and the [Supporting Information](#) for further details.

Stability of Large-Area Junctions. The studies enumerated above probe areas on the order of tens of nm^2 to give insight into the bulk mechanical properties of a SAM (e.g., stiffness,) however, the electrical properties that SAMs exhibit in large-area junctions include the influence of defects (e.g., local disorder) driven by inhomogeneities in the substrate, chemical impurities and grain boundaries.^{24,40–42} Shorter alkanethiols exhibit more resilience to defects because they are more liquid-like;⁴³ however, when a voltage is applied to a large-area junction, a substantial electrostatic pressure develops that can deform and induce the reorganization⁴⁴ in which case the stiffness of longer alkyl chains is advantageous. We hypothesize that there is, therefore, a relationship between the mechanical stability of a SAM and its breakdown voltage; SAMs that can withstand higher electrostatic pressures should form large-area tunneling junctions that resist shorting at high bias. There is no consensus on the mechanism of failure of

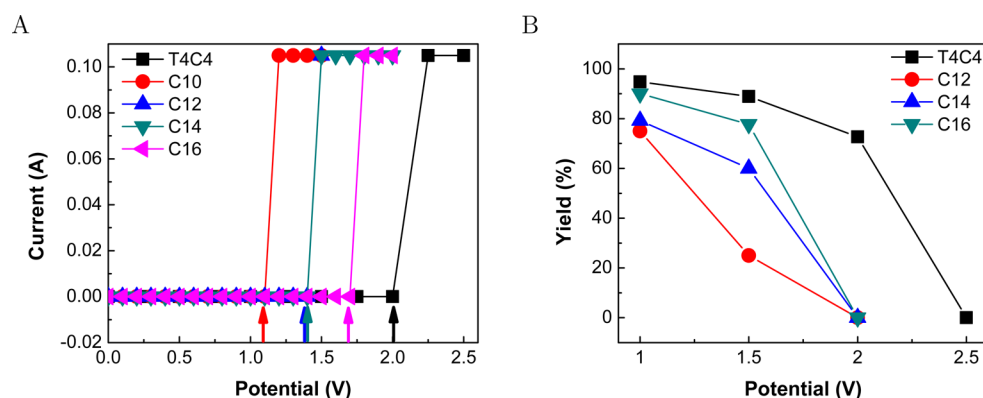


Figure 8. (A) Representative I – V plots show the breakdown voltage of $\text{Ag}^{\text{TS}}/\text{SAM}/\text{EGaIn}$ junctions comprising T4C4, C10, C12, C14, and C16. (B) Yield of nonshorting junctions as a function of potential window.

large-area junctions at high bias, which could be (1) entirely a function of the ability of a SAM to resist penetration by the top-contact, (2) the migration of metal atoms from the bottom-contact, e.g., the formation of filaments of Au, (3) electrochemical instability, or (4) some combination of the three. A clear correlation between breakdown voltage and the mechanical robustness of T4C4 would imply that mechanism 1 is dominant because the electrochemical window of T4C4 is much smaller than that of an alkanethiol. Extending the potential window in which a SAM can operate in a large-area junction is particularly relevant to molecule diodes⁴⁵ such as SAMs incorporating ferrocenyl,^{46,47} bipyridyl,^{48,49} pyrimidyl,¹⁴ and fullerene⁵⁰ moieties because the degree of rectification tends to scale with bias and they function under bias in integrated circuits.¹⁰

To investigate the influence of mechanical stability on breakdown voltages in large-area molecular junctions, we formed $\text{Ag}^{\text{TS}}/\text{SAM}/\text{EGaIn}$ junctions²² of T4C4, C12, tetradecanethiol (C14), and hexadecanethiol (C16). As mentioned above, we chose Ag^{TS} because it is the most commonly reported substrate for EGaIn top-contacts. We swept junctions of each SAM through increasing bias windows and recorded the frequency of shorts, defined by the sudden increase in current to the compliance limit of the instrument. Figure 8A shows representative I – V plots (on a linear scale) revealing a clear trend of increasing breakdown potential: T4C4 > C16 > C14 \approx C12 > C10. Figure 8B shows the percent-yield of nonshorting junctions scanned from ± 1 , ± 1.5 , ± 2 , and ± 2.5 V (see Table S5 for details). All SAMs shorted 100% of the time at ± 2.5 V, but only 20% of junctions comprising T4C4 shorted at ± 2 V, whereas 100% of junctions comprising C12, C14 and C16 shorted. At ± 1 and ± 1.5 V there is a clear trend of increasing percentage of shorts: C12 > C14 > C16 > T4C4. This trend supports the hypothesis that the primary mode of failure of these $\text{Ag}^{\text{TS}}/\text{SAM}/\text{EGaIn}$ junctions is mechanical failure due to electrostatic pressure from the applied bias; the mechanical robustness of SAMs of alkanethiolates scales with chain-length, but T4C4 is considerably more robust than a SAM of alkanethiolates of any number of carbons up to at least C16.

CONCLUSION

Technological applications of molecular electronics in the medium-term will almost certainly utilize SAMs; they simplify fabrication and large-area junctions, in particular, can already be incorporated into integrated circuits and wafer-scale manufac-

turing processes. The usefulness of molecular tunneling junctions derives from the nonlinear dependence of I – V characteristics on the conformation and electronic structure of the molecules. However, the I – V properties of bottom-up junctions comprising SAMs are affected by mechanical force. For SAMs of alkanethiolates, mechanical forces disturb the packing of the SAM, causing tilt angles to increase. We have shown that the electronic structure of π -conjugated molecules (i.e., the electrostatics of the junction) is also directly affected by mechanical force. Thus, it is important to develop an understanding of this relationship and relate it to molecular structure such that the mechanical properties of a SAM and how a tunneling junction responds to forces can be tailored synthetically both to increase the robustness and stability of junctions and to develop devices that respond to mechanical inputs.

We have shown that SAMs of a σ – π molecule designed to maximize intermolecular interactions, T4C4, are significantly more mechanically robust than SAMs of alkanethiolates. Self-assembled monolayers of T4C4 undergo less deformation as a function of loading force by AFM and Young's modulus is approximately five times higher. At relatively low loading forces, tunneling junctions comprising SAMs of T4C4 show no changes in conductance or values of V_{trans} ; SAMs of C10 show significant changes. At higher loading forces than SAMs of C10 are capable of withstanding, junctions comprising T4C4 begin to show differences. Our DFT calculations suggest that these changes are due to force-induced distortions of the π -system and not, as is the case for C10, a change in tilt angle; the AFM tip does not penetrate SAMs of T4C4 as it does C10. The mechanical stability of T4C4 translates into a higher breakdown potential in large-area tunneling junctions with EGaIn top-contacts. This observation suggests that electrostatic pressure plays a significant role in the shorting of molecular tunneling junctions at high bias; SAMs of T4C4 are more mechanically robust than SAMs of alkanethiolates, but T4C4 is significantly less electrochemically stable than an alkanethiol. Our results demonstrate that it is possible to design molecules that maximize mechanical properties and breakdown voltages in large-area tunneling junctions comprising SAMs.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jpcc.7b03853.

Details of synthesis and characterization of molecules and SAMs, details on EGaIn and CP-AFM measurements, and details on DFT calculations and parameters used, including more deformation simulations (PDF)

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Notes

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