



University of Groningen

Charge and Sodium Ordering in β-Na0.33V2O3

Loosdrecht, P.H.M. van; Presura, C.N.; Popinciuc, M.; Marel, D. van der; Maris, G.; Palstra, Thomas; Bentum, P.J.M. van; Yamada, H.; Yamauchi, T.; Ueda, Y.

Published in: Journal of Superconductivity%3A Incorporating Novel Magnetism

DOI: 10.1023/A:1021255410333

IMPORTANT NOTE: You are advised to consult the publisher's version (publisher's PDF) if you wish to cite from it. Please check the document version below.

Document Version Publisher's PDF, also known as Version of record

Publication date: 2002

Link to publication in University of Groningen/UMCG research database

Citation for published version (APA): Loosdrecht, P. H. M. V., Presura, C. N., Popinciuc, M., Marel, D. V. D., Maris, G., Palstra, T. T. M., ... Ueda, Y. (2002). Charge and Sodium Ordering in β-Na0.33V2O3. Journal of Superconductivity%3A Incorporating Novel Magnetism, 15(6). DOI: 10.1023/A:1021255410333

Copyright Other than for strictly personal use, it is not permitted to download or to forward/distribute the text or part of it without the consent of the author(s) and/or copyright holder(s), unless the work is under an open content license (like Creative Commons).

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

Downloaded from the University of Groningen/UMCG research database (Pure): http://www.rug.nl/research/portal. For technical reasons the number of authors shown on this cover page is limited to 10 maximum.

Charge and Sodium Ordering in β -Na_{0.33}V₂O₃

P. H. M. van Loosdrecht,¹ C. N. Presura,¹ M. Popinciuc,¹ D. van der Marel,¹ G. Maris,¹ T. T. M. Palstra,¹ P. J. M. van Bentum,² H. Yamada,³ T. Yamauchi,³ and Y. Ueda³

Polarized Raman and optical spectra for the quasi one-dimensional metallic vanadate β -Na_{0.33}V₂O₃ are reported for various temperatures. The spectra are discussed in the light of the sodium and charge ordering transitions occurring in this material, and demonstrate the presence of strong electron–phonon coupling.

KEY WORDS: charge ordering; metal-insulator transition; vanadates; phase transitions; Raman and optical spectroscopy.

1. INTRODUCTION

The recent discovery [1] of a clear metalinsulator transition (MIT) in the vanadium bronze β -Na_{0.33}V₂O₃ has sparked a revival of interest in this quasi one-dimensional(1D) metallic system. In addition to the MIT, β -Na_{0.33}V₂O₃ undergoes a structural sodium ordering transition at higher temperatures, a magnetic transition at low temperatures, and a transition into a superconducting state under high pressure [2]. Several important aspects of this 1D material have remained unclear, including the nature of the spin and charge excitations, and the relation between the Na ordering and the MIT. This contribution presents a Raman and optical study of β -Na_{0.33}V₂O₃ focusing on the various phase transitions and electron-phonon coupling in this compound.

2. STRUCTURE AND PHASE TRANSITIONS

At room temperature β -Na_{0.33}V₂O₃ has a monoclinic structure (spacegroup C_{3h}^2 , a = 10.08 Å, b = 3.61 Å, c = 15.44 Å, $\beta = 109.6^{\circ}$) [3,4]. The structure consists of zigzag double chains of VO₆ octahedra,

forming sheets by joining corners. These sheets are separated by additional chains of double VO₅ trigonal bipyramids, giving rise to unidirectional tunnels along **b** (see Fig. 1) in which the Na ions are located. At a sodium stoichiometry of 0.33, there is exactly one sodium atom per primitive cell. The sodium ion can occupy two closely spaced positions, although simultaneous occupation of the two sites is prohibited (the distance between the sites is 1.95 Å). NMR experiments [5] suggest that, at room temperature, the occupation of these two sites is random, giving rise to a disordered structure. At $T \simeq 240$ K a second-order phase transition occurs in which the sodium atoms order in a zigzag fashion along the unique axis accompanied by a doubling of the *b*-axis [1]. Two additional transitions occur at lower temperatures. At $T \simeq 136$ K an MIT has been observed, in which the unit cell undergoes an additional tripling along the b-axis [6]. It has been suggested that the charge ordering occurs because of a localization of the charge on chains along the b-direction formed by the V_1 ions. Finally, at $T \simeq 22$ K there is a transition from the paramagnetic high-temperature state to a canted antiferromagnetic state [7].

3. RAMAN SPECTROSCOPY

The samples used in this study have been prepared as described in [1]. Typical sizes are 5-6 mm along the *b*-direction and about 0.3–1 mm in the other two directions. For the Raman experiments, samples

¹Material Science Center, University of Groningen, Nijenborgh 4, 9747 AG Groningen, The Netherlands.

²University of Nijmegen, Toernooiveld, 6525 ED Nijmegen, The Netherlands.

³Institute for Solid State Physics, University of Tokyo, Tokyo, Japan.



Fig. 1. Room temperature structure of β -Na_{0.33}V₂O₃ showing chains of V₁ (dark octahedra), V₂ (light octahedra), and V₃ ions (pyramids).

were mounted in a flow cryostat (stabilization better than 1 K). Polarized spectra have been recorded in a back reflection geometry using an Ar⁺ ion laser (514 nm) for excitation (power <5 mW, spot size 100 μ m) and a state of the art triple pass Raman spectrometer with diode array detection. Typical spectra are shown in Fig. 2 (left part). Because of the low symmetry and the large number of atoms in the unit cell, the spectra show a large number of active phonon modes in all symmetries. What immediately draws attention is the large width of most of the observed phonon modes. Particularly in the 400–800 cm⁻¹ region, the line widths are 10– 50 cm^{-1} . The active phonons here are expected to be



Fig. 2. Left: Polarized Raman spectra of β -Na_{0.33}V₂O₃ for T = 15 K and 300 K. Upper panel, $A_g(\mathbf{bb})$; middle panel, $A_g(\mathbf{\perp b \perp b})$; lower panel, $B_g(\mathbf{b \perp b})$ symmetry. Right: Detailed temperature dependence of the 400–800 cm⁻¹ region of the $A_g(\mathbf{\perp b \perp b})$ Raman spectrum. In all panels the subsequent curves have been given an offset for clarity.

Charge and Sodium Ordering in β -Na_{0.33}V₂O₃

vanadium–oxygen bending modes. This strong broadening is believed to be due to a strong electron– phonon coupling, consistent with a polaronic picture of the electronic properties of this material [8,9]. The coupling is then a result of the modulation of the hopping parameters for O(2p)-V(3d) hopping by the phonons.

A group theoretical analysis shows that the room temperature k = 0 optical phonons can be classified as $\Gamma = 11A_u + 22B_u + 20A_g + 10B_g$, in which the gerade modes are Roman active $(A_g \text{ in } (aa), (bb), (cc))$, and (ac), and B_g in (ab) and (bc) polarization) and the ungerade IR active $(A_u \text{ for } b \text{ polarization}, B_u \text{ for } a \text{ and} c$ polarizations). Consistent with the crystal symmetry 16 modes are observed in A_g symmetry, and 9 in B_g symmetry. The missing modes might be too weak to be observed, but might also escape detection because of near degeneracy.

Below the sodium ordering transition the crystal structure adheres to a C_{2h}^5 symmetry [6,10]. The symmetry analysis leads now to a decomposition $\Gamma = 65A_u + 64B_u + 66A_g + 66B_g$ for the optical modes. Experimentally 55 modes are observed in A_g symmetry, and 20 in B_g symmetry. Below the charge ordering transition the unit cell triples, leading to about 190 active phonons for each irreducible representation. Although a few phonons seem to be activated below the metal insulator transition, the spectra do not show this tripling. Again, this might be due to a lack of scattering strength or near degeneracies.

The strongest changes in the spectra are observed below the MIT at T = 136 K. This is exemplified in Fig. 5 (right panel). Apparently, the localization of the electrons leads to a decrease of the coupling of the phonons to electronic excitations.

4. OPTICAL SPECTROSCOPY

Temperature-dependent optical spectra of β -Na_{0.33}V₂O₃ in the NIR–UV range (6000–35000 cm⁻¹) have been measured using ellipsometry on a surface containing the *b*-axis. In addition, the reflectivity spectra have been measured as a function of temperature and polarization in the IR–MIR range (20–6000 cm⁻¹). The optical conductivity of β -Na_{0.33}V₂O₃ has been determined by combining the reflectivity data with the ellipsometric data and performing a Kramers–Kronig analysis [9]. The room temperature results are shown in Fig. 3.

The strong absorption band observed above $10,000 \text{ cm}^{-1}$ in both spectra is due to charge transfer excitations from the oxygen 2p to the vanadium 3d



Fig. 3. Optical conductivity of β -Na_{0.33}V₂O₃ at room temperature calculated from reflectivity spectra (FIR–MIR) and ellipsometric data (MIR–UV) [9].

levels, similar as has been observed in α -Na_{0.33}V₂O₃ [11]. The optical conductivity perpendicular to the *b*-axis shows an absorption band around 8000 cm⁻¹, again similar to what is observed in α -NaV₂O₃. This band is probably due to a bonding-antibonding transition on the V₁-O-V₃ bonds [9].

At lower energy the 1D metallic nature of β -Na_{0.33}V₂O₃ is clearly demonstrated by the finite spectral weight for $E \rightarrow 0$ for a polarization along the *b*-axis, and the tendency to zero weight in the perpendicular polarization. Indeed the data parallel to *b* may be fitted by a Drude model, yielding an unscreened plasma frequency of $\simeq 3200 \text{ cm}^{-1}$, and a scattering time $\tau \simeq 40$ ps. This low energy response has been attributed to mobile small polarons. From the optical data the estimated DC conductivity is $200 \ \Omega^{-1} \text{ cm}^{-1}$, which compares well to the published DC conductivity ity of $100 \ \Omega^{-1} \text{ cm}^{-1}$ [1]. Another manifestation of the polaronic nature of β -Na_{0.33}V₂O₃ is found in the characteristic [12] absorption peak around 3000 cm⁻¹.

At room temperature one expects a total of 32 active phonon modes, 11 in A_u symmetry (polarization along b), and 22 in B_u symmetry (\perp b). Along the insulating direction (Fig. 3, lower curve) we indeed observe 22 active phonon modes. In contrast, the metallic direction (figure 5, upper curve) exhibits only two strong phonon modes, all other modes appear as strongly broadened weak features in the spectrum, indicating again the importance of electron–phonon coupling in this material. Already below the sodium ordering transition one expects many new phonon modes, though only a few are actually observed (for an example see Fig. 4 lower right panel). In contrast, the MIT leads to a spectacular gradual appearance



Fig. 4. *Upper panel*: Optical conductivity of β -Na_{0.33}V₂O₃ along the metallic direction above and below the charge ordering transition. *Lower panels*: Phonon modes appearing below the sodium (c) and charge ordering (b) transitions. The inset shows the temperature dependence of the phonons shown in panels (b) and (c).

of a large number of phonon modes (see Fig. 4, upper and lower left panels). The observed behavior is reminiscent of the observation of phase or charged phonons in several charge density wave materials, resulting from electron–phonon interactions [13].

5. CONCLUSIONS

The Raman and optical spectra presented in this contribution demonstrate in many ways the importance of electron–phonon coupling in β -Na_{0.33}V₂O₃.

The observed phonon modes and symmetry are found to be in overall agreement with the phase transitions in this material, although the appearance of the many modes in the IR spectra along the metallic direction probably does not result from symmetry breaking, but rather gain strength from electron–phonon coupling in the charge ordered phase. Finally, one interesting undiscussed aspect of the optical data is the appearance of a continuum and Fano distortions of phonon modes (see Fig. 4 upper panel). The origin of this lowenergy continuum is presently unknown, but likely to be of electronic nature.

ACKNOWLEDGMENTS

This work was partially supported by the Dutch Foundation for Fundamental Research on Matter (FOM) and by INTAS (99-155).

REFERENCES

- 1. H. Yamada and Y. Ueda, J. Phys. Soc. Jpn. 68, 2735 (1999).
- 2. T. Yamauchi, Y. Ueda, and N. Môri, *Phys. Rev. Lett.* **89**,057002-1 (2002).
- 3. A. D. Wadsley, Acta Cryst. 8, 695 (1955).
- 4. H. Kobayashi, Bull. Chem. Soc. Jpn. 52, 1315 (1979).
- K. Maruyama and H. Nagasawa, J. Phys. Soc. Jpn. 48, 2159 (1980).
- J.-I. Yamaura, M. Isobe, H. Yamada, T. Yamauchi, and Y. Ueda, J. Phys. Chem. Solids 63, 957 (2002).
- A. N. Vasil'ev, I. Marchenko, A. I. Smirnov, S. S. Sosin, P. L. Kapitza, H. Yamada, and Y. Ueda, *Phys. Rev. B* 64, 174403 (2001).
- N. F. Mott, *Metal–Insulator Transitions* (Taylor and Francis, London, 1974).
- C. Presura, M. Popinciuc, P. H. M. van Loosdrecht, D. van der Marel, M. Mostovoy, G. Maris, T. T. M. Palstra, H. Yamada, and Y. Ueda, 2002; unpublished manuscript (2002).
- 10. G. Maris et al., private communication (2002).
- C. Pressura, D. van der Marel, M. Dischner, C. Geibel, and R. K. Kremer, *Phys. Rev. B* 62, 16522 (2000).
- C. A. Kuntscher, M. Dressel, F. Lichtenberg, J. Mannhart, D. van der Marel, Y. Kanai, S. Kagoshima, and H. Nagasawa (2002); *Preprint* cond-mat/0205049; D. Emin, *Phys. Rev. B* 48, 13691 (1993); D. Kaplan and A. Zylbersztejn, *J. de Physique* 37, L-123 (1976).
- M. J. Rice, *Phys. Rev. Lett.* **37**, 36 (1976); L. Degiorgi, P. Wachter, and C. Schlenker, *Physica B* **164**, 305 (1990), and references therein.