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Mixed radical iodine charge-transfer salts of dithiadiazolyl diradicals. Structural characterization of the pyridine-bridged 2:1 salt 2,6-[(S₂N₂C)C₅H₃N(CN₂S₂)](2)[I]

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Table S1 Crystal and refinement data for [P-2,6-S]₂[I].

formula	<chem>IS8N10C14H6</chem>
fw	348.83
crystal size, mm	0.51 x 0.11 x 0.08
crystal color	black
crystal mount	on glass fiber by epoxy
<i>a</i> , Å	3.434(3)
<i>b</i> , Å	9.8914(9)
<i>c</i> , Å	30.803(4)
β , deg	91.52(3)
<i>V</i> , Å ³	1045.9(9)
cell detn, refls	25
cell detn, 2 <i>θ</i> range, deg	14-18
d(calcd), g cm ⁻³	2.22
space group	P2 ₁ /c
Z	2
F ₀₀₀	681.87
radiation	MoK _α , graphite monochromated
λ , Å	0.71073
temp, K	293
linear abs coeff, mm ⁻¹	2.31
diffractometer	Enraf-Nonius CAD-4
scan technique	θ -2 <i>θ</i>
scan speed, deg min ⁻¹	4-16 (in omega)
scan width, deg	1.0 + 0.35tan $θ$
2 <i>θ</i> range, deg	4-50
<i>h</i> , <i>k</i> , <i>l</i> ranges	0,4; 0,11; -36,36
exposure time, hrs	53.5
std refl indices	4,0,-8; 0,3,-7; 0,1,11
drift of stds, %	1.3
absorption correction	empirical psi scans

absorption, range	0.91-1.00
refl meas	2154
unique refls	1823
R for merge	0.066
data with $I > 3\sigma(I)$	1163
solution method	Direct Methods
parameters refined	154
$R(F), R_w(F)$	0.047, 0.081
GOF	1.36
$p, w^{-1} = [\sigma^2(I) + pI^2]/4F^2$	0.05
largest Δ/σ	0.000
extinction correction	none
final diff map, e Å ⁻³	-0.6(1), +0.5(1)
programs	NRC386 (PC version of NRCVAX)*
scattering factors	Internat. Tables for Crystallography Vol 4
H atom treatment	idealized positions (C-H = 0.95Å)

*PC version of NRCVAX, an interactive program system for structure analysis; see E.J. Gabe, Y. LePage, J.P. Charland, F.L. Lee, and P.S. White, *J. Appl. Cryst.* **22**, 383 (1989).

SPECIAL NOTE: During the refinement of the anisotropic temperature factors of I1 and I2, they were constrained to be equivalent due a large correlation. Iodine position occupancies were determined ($I1/I2 = 0.3974/0.3013$) during isotropic refinement and locked at those values prior to anisotropic refinement.

Table S2 Bond distances (\AA) and angles ($^\circ$) for $[\text{P}-2,6-\text{S}]_2[\text{I}]$. ESDs refer to the last digit printed.

	Distances	Angles	
S(1)-S(2)	2.065 (3)	S(2)-S(1)-N(1)	93.6 (3)
S(1)-N(1)	1.607 (9)	S(1)-S(2)-N(2)	95.2 (3)
S(2)-N(2)	1.613 (8)	S(4)-S(3)-N(3)	94.2 (3)
S(3)-S(4)	2.082 (4)	S(3)-S(4)-N(4)	94.6 (3)
S(3)-N(3)	1.629 (8)	S(1)-N(1)-C(1)	117.3 (6)
S(4)-N(4)	1.635 (8)	S(2)-N(2)-C(1)	114.7 (6)
N(1)-C(1)	1.325 (11)	S(3)-N(3)-C(7)	114.6 (6)
N(2)-C(1)	1.358 (10)	S(4)-N(4)-C(7)	113.4 (6)
N(3)-C(7)	1.313 (12)	C(2)-N(5)-C(6)	117.6 (7)
N(4)-C(7)	1.335 (12)	N(1)-C(1)-N(2)	119.1 (7)
N(5)-C(2)	1.351 (10)	N(1)-C(1)-C(2)	119.8 (7)
N(5)-C(6)	1.345 (11)	N(2)-C(1)-C(2)	121.0 (7)
C(1)-C(2)	1.443 (11)	N(5)-C(2)-C(1)	117.9 (7)
C(2)-C(3)	1.393 (12)	N(5)-C(2)-C(3)	121.9 (8)
C(3)-C(4)	1.375 (13)	C(1)-C(2)-C(3)	120.2 (7)
C(4)-C(5)	1.348 (15)	C(2)-C(3)-C(4)	119.7 (8)
C(5)-C(6)	1.382 (13)	C(3)-C(4)-C(5)	117.7 (8)
C(6)-C(7)	1.483 (13)	C(4)-C(5)-C(6)	121.6 (9)
		N(5)-C(6)-C(5)	121.4 (8)
		N(5)-C(6)-C(7)	117.5 (8)
		C(5)-C(6)-C(7)	121.1 (8)
		N(3)-C(7)-N(4)	123.3 (8)
		N(3)-C(7)-C(6)	116.5 (8)
		N(4)-C(7)-C(6)	120.1 (8)

Table S3 Intermolecular S---I contacts (\AA) for $[\text{P}-2,6-\text{S}]_2[\text{I}]$. ESDs refer to the last digit printed.

S3---I1	(at 1 - x, 0.5 + y, 0.5 - z)	3.479 (3)
S3---I2	(at 2 - x, 0.5 + y, 0.5 - z)	3.594 (7)
S3---I2	(at x - 1, 1.5 - y, 0.5 + z)	3.428 (7)
S4---I1	(at 1 - x, 0.5 + y, 0.5 - z)	3.748 (3)
S4---I2	(at 2 - x, 0.5 + y, 0.5 - z)	3.842 (6)
S4---I2	(at x - 1, 1.5 - y, 0.5 + z)	3.716 (6)

Table S4 Anisotropic temperature factors, $u(i,j) \times 100$, for $[P-2,6-S]_2[I]$. ESDs refer to the last digit printed.

	u11(U)	u22	u33	u12	u13	u23
I1	3.2 (3)	3.55(6)	3.02(6)	-0.22(24)	-0.34(21)	-0.22(4)
I2	3.2 (3)	3.55(6)	3.02(6)	-0.22(24)	-0.34(21)	-0.22(4)
S1	10.8 (3)	2.33(12)	3.32(13)	0.60(14)	-2.19(15)	-0.73(9)
S2	5.97(17)	3.16(12)	2.65(11)	0.30(12)	-0.79(11)	-0.17(9)
S3	8.78(23)	4.32(14)	2.79(12)	2.44(15)	0.11(14)	-0.50(10)
S4	6.47(19)	3.69(14)	3.74(14)	1.78(13)	-0.28(13)	-0.22(10)
N1	5.3 (6)	1.9 (3)	5.7 (5)	0.2 (4)	-1.7 (4)	-0.1 (3)
N2	3.0 (4)	2.3 (3)	3.6 (4)	0.2 (3)	-0.2 (3)	-0.2 (3)
N3	5.6 (5)	4.3 (4)	2.8 (4)	1.5 (4)	0.5 (4)	0.4 (3)
N4	6.3 (6)	3.9 (4)	2.1 (4)	1.2 (4)	0.8 (4)	0.2 (3)
N5	1.6 (4)	3.1 (4)	3.4 (4)	0.1 (3)	-0.7 (3)	0.3 (3)
C1	4.1 (5)	2.7 (4)	2.0 (4)	0.1 (4)	-1.1 (4)	0.2 (3)
C2	1.0 (4)	2.4 (4)	3.7 (5)	0.6 (3)	-0.2 (4)	0.3 (3)
C3	2.5 (5)	3.3 (5)	5.6 (6)	-0.1 (4)	0.2 (5)	0.8 (4)
C4	2.6 (5)	5.0 (6)	4.1 (5)	0.3 (5)	0.6 (4)	2.0 (4)
C5	2.8 (5)	5.1 (6)	3.6 (5)	0.5 (5)	-0.6 (4)	1.8 (4)
C6	2.1 (5)	3.6 (5)	3.2 (4)	1.1 (4)	-0.3 (4)	-0.1 (4)
C7	3.0 (5)	4.0 (5)	3.5 (5)	1.1 (4)	0.5 (4)	0.1 (4)

Anisotropic temperature factors are of the form:

$$\exp[-2\pi^2(h^2U_{11}a^{*2} + k^2U_{22}b^{*2} + l^2U_{33}c^{*2} + 2hkU_{12}a^*b^* + 2hlU_{13}a^*c^* + 2klU_{23}b^*c^*)].$$

Figure S1 ORTEP drawing of the asymmetric unit in $[P\text{-}2,6\text{-S}]_2[\text{I}]$.

