



## Research paper

# In vitro evaluation of natural and methylated cyclodextrins as buccal permeation enhancing system for omeprazole delivery

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## ARTICLE INFO

## Article history:

Received 26 March 2008

Accepted in revised form 7 August 2008

Available online 29 August 2008

## Keywords:

Buccal mucosa

Cyclodextrins

L-Arginine

Omeprazole

Permeation

Stability

## ABSTRACT

In this work the enhancing effect of cyclodextrins on the buccal permeation of a hydrophobic model drug, omeprazole was studied. First, the influence of the complexation with cyclodextrins in the absence and in the presence of an alkali agent, L-arginine, on the drug stability was checked at neutral conditions since omeprazole alone is only stable in basic conditions. In vitro transbuccal permeation of omeprazole non-complexed and complexed with  $\beta$ - and methyl- $\beta$ -cyclodextrin and in presence of L-arginine was examined using freshly obtained porcine buccal mucosa. Tissue viability after incubation with sample solutions was assessed using a MTT (3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide) biochemical assay and histological evaluation. The toxicity of the sample solutions on buccal mucosa was evaluated by measuring lactate dehydrogenase activity. The present results show that complexation with cyclodextrins increases drug stability at neutral conditions; furthermore, L-arginine contributed to higher drug stability. Permeation studies indicate an increase on drug permeation in complexed form of 1.1- and 1.7-fold for  $\beta$ -cyclodextrin and methyl- $\beta$ -cyclodextrin, respectively. The presence of L-arginine increases drug permeation 1.4-fold in omeprazole complexed with  $\beta$ -cyclodextrin and 2.4-fold in the inclusion complex formed with methyl- $\beta$ -cyclodextrin. The cell viability of the buccal mucosa after a 3 h incubation period, with all sample solutions, remained around 70% and lactate dehydrogenase assay showed that studied cyclodextrins, even in the presence of an alkali agent are not cytotoxic for porcine buccal mucosa. Histological evaluation of the tissue demonstrated that the buccal epithelium remains viable after 3 h of incubation with sample solutions.

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## 1. Introduction

Buccal mucosa is a potential site for drug absorption in alternative to oral drug delivery. Active molecules administered through the buccal mucosa pass directly into the systemic circulation, thereby minimizing the first hepatic pass and adverse gastro-intestinal effects [1,2]. Other important advantages are the low enzymatic activity, suitability for drugs or excipients that mildly and reversibly damage or irritate the mucosa, painless administration, easy drug withdrawal, facility to include permeation enhancer/enzyme inhibitor or pH modifier in the formulation and versatility in designing as multidirectional or unidirectional release systems for local or systemic actions [3]. However, lower permeability of the buccal mucosa to large molecules can be problematic in order to achieve therapeutic levels of such molecules. Buccal permeation can be increased by using various penetration enhancers. Recently,

cyclodextrins have been classified as a new class of penetration enhancers [4–6]. These molecules are cyclic oligosaccharides with a hydrophilic outer surface and a hydrophobic central cavity. The hydrophilic exterior of the cyclodextrin molecules makes them water-soluble while the hydrophobic cavity provides a microenvironment for appropriate sized non-polar molecules [7]. Cyclodextrins are able to form dynamic molecular inclusion complexes with many drugs by incorporating the drug molecule, or commonly a lipophilic moiety of the molecule, into the central cavity [8]. These non-covalent complexes offer a variety of physicochemical advantages over the unmodified drugs such as the possibility to increase their water solubility and stability [9]. It is generally recognized that cyclodextrins act as true carriers by keeping the hydrophobic drug molecules in solution and deliver them to the surface of the biological membrane, where they partition into the membrane [10]. Cyclodextrins can enhance drug permeation by increasing drug availability and stability at the surface of the biological barriers [11]. However, derivative cyclodextrins, especially methylated cyclodextrins, act as absorption enhancers by different pathways. These hydrophobic cyclodextrins act as absorption enhancers, probably, by transiently changing membrane perme-

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ability, overcoming the aqueous diffusion barrier and opening tight junctions [12]. During cell differentiation process in buccal mucosa, small organelles called membrane coating granules (MCGs) composed of non-lamellar lipid sacks are formed in intercellular spaces of the non-keratinized regions [13,14]. These MCGs, first observed in epidermis, are not different from those observed in the skin and are believed to provide an intercellular permeability barrier to buccal mucosa [15]. Methyl- $\beta$ -cyclodextrin, a more hydrophobic cyclodextrin, can permeate the buccal mucosa and to form inclusion complexes with hydrophobic molecules, namely lipids from the cellular membrane, interacting strongly with these lipids; it could modify buccal mucosa permeability and could act as penetration enhancer for the buccal route.

Omeprazole (OME) a substituted benzimidazole, has been shown to effectively suppress gastric acid secretion by inhibiting the  $H^+K^+$ -ATPase (proton pump), in the parietal cells [16,17]. The bioavailability of OME following oral administration is usually very low, since it degrades quickly in the acidic environment of the stomach and undergoes hepatic first-pass metabolism. As an attempt to improve the oral bioavailability of OME various oral formulations have been developed over the years. However, these oral formulations revealed to have a large intra-individual variation in drug plasma concentration in human subjects [18,19].

The aim of this study is to evaluate the potential enhancement effect of the cyclodextrins on the buccal mucosa permeation of OME. Inclusion complexes between OME and both cyclodextrins in absence and in presence of an alkali agent, L-arginine (ARG), were prepared by a freeze-drying method [20]. The effect of  $\beta$ -cyclodextrin ( $\beta$ CD) and methyl- $\beta$ -cyclodextrin (M $\beta$ CD) in the absence and in the presence of ARG on the OME stability was assessed at neutral conditions by high performance liquid chromatography (HPLC). Permeation studies were performed using two different cyclodextrins, a natural cyclodextrin,  $\beta$ CD and a methylated derivative, M $\beta$ CD, in order to compare their permeation enhancing properties. Viability, integrity and cytotoxicity studies were carried out to investigate possible morphologic changes of the buccal mucosa, after contact with the complexed OME in the absence and presence of ARG. Structural alterations in the buccal mucosa were also considered and supported by histological studies.

## 2. Materials and methods

### 2.1. Materials

$\beta$ CD (Beta-cyclodextrin, KLEPTOSE<sup>®</sup>,  $M_w = 1135$ ) and M $\beta$ CD (Methyl- $\beta$ -cyclodextrin, CRYSMEB<sup>®</sup>,  $M_w \sim 1190$ , with an average degree of substitution of 0.5) were kindly donated by Roquette (Lestrem, France) and Omeprazole (OME,  $M_w = 345.42$ ) was kindly donated by Belmac Laboratory, S.A. (Madrid, Spain). L-Arginine (ARG) was purchased from Panreac (Santiago de Compostela, Spain). Thiazolyl blue tetrazolium bromide (MTT), fluorescein isothiocyanate-dextran 40 (FD40, average  $M_w$  40.000) were purchased from Sigma (St. Louis, MO, USA). The cytotoxicity detection kit (Dye solution + Catalyst) was purchased from Roche. All other reagents (chemicals and solvents) were of analytical grade.

### 2.2. Preparation of inclusion complexes

Solid inclusion complexes were prepared by freeze-drying method. Systems were prepared in a stoichiometry 1:1 (drug:cyclodextrin) according to previous phase solubility studies [21] and ARG was added in a molar proportion 6:1, relatively to OME. The same systems were prepared in the absence of ARG to observe the influence of the alkali agent on the drug stability and

permeation. All the clear solutions were frozen by immersion in an ethanol-bath at  $-50^\circ\text{C}$  (Shell Freezer, Labconco, Freezone<sup>®</sup> model 79490) and the frozen solutions were lyophilized in a freeze-dryer (Lyph-lock 6 apparatus, Labconco) for 72 h.

### 2.3. Stability studies

The stability of OME alone was checked at three different pH values: 7.0, 7.9 and 9.0. After, stability of OME alone, complexed with  $\beta$ CD or M $\beta$ CD in absence and in presence of ARG was examined at  $37^\circ\text{C}$  in 40 mM of Bis-Tris buffer (bis[2-hydroxyethyl]imino-tris[hydroxymethyl]methane) at  $\text{pH } 7.0 \pm 0.5$ . Solutions containing 100  $\mu\text{g/mL}$  of OME in free and complexed state were prepared. These solutions were stirred at 300 rpm during 8 h at  $37^\circ\text{C}$ . Samples were collected every hour and analyzed by HPLC.

### 2.4. High performance liquid chromatography (HPLC)

The official HPLC method described in the USP XXX [22] was used for quantification of the drug. An Elite Lachrom Liquid Chromatograph (Merck, Hitachi) system consisting of a quaternary pump (L2130), with a L2450 programmable multiple wavelength detector set at 300 nm and an autosampler L2200 was used. The separation was carried out at room temperature and the column used was a reverse-phase Purospher<sup>®</sup> RP-18 endcapped (5  $\mu\text{m}$ ), 125 cm  $\times$  4 mm. The mobile phase was a mixture of phosphate buffer ( $\text{pH } 7.6 \pm 0.5$ ) and acetonitrile (75:25, v/v), filtered through 0.20  $\mu\text{m}$  nylon filters, degassed and pumped at a constant flow rate of 1 mL  $\text{min}^{-1}$ . The chromatograms were recorded and the peak area response was measured using an automatic integrator. The injection volume was 20  $\mu\text{l}$  for all standards and samples.

#### 2.4.1. Validation of the HPLC method

The HPLC method was validated by using the following analytical parameters: linearity, precision, accuracy, repeatability and specificity. Linearity was evaluated by calculation of a regression line using the least squares method. Calibration curve was obtained from eight standard solutions, containing 0.7, 1.5, 3, 6, 12.5, 25, 50 and 100  $\mu\text{g/mL}$  of OME in Bis-Tris buffer injected three times. Precision was determined by injecting six times the standard solution containing 12.5  $\mu\text{g/mL}$  of OME. Accuracy was tested using three different standard solutions containing 6, 12.5 and 25  $\mu\text{g/mL}$  of OME injected three times. The repeatability was performed by six standard solutions containing 12.5  $\mu\text{g/mL}$  of OME prepared six times and injected. Specificity was determined by comparing the following solutions: 100  $\mu\text{g/mL}$  of OME alone, in the presence of  $\beta$ CD or M $\beta$ CD and in the presence of  $\beta$ CD or M $\beta$ CD and ARG injected three times each. At last, the interference of degradation products with OME peak retention time was checked by injection of two different solutions containing 100  $\mu\text{g/mL}$  of OME. In one solution, OME was submitted to degradation in acidic conditions and in the other solution, OME was degraded by high temperatures ( $60^\circ\text{C}$  during 1 h). The final pH of all solutions was adjusted to  $7.0 \pm 0.5$  and relative standard deviations (RSD) were calculated.

### 2.5. In vitro permeation studies in porcine buccal mucosa

#### 2.5.1. Tissue preparation

Porcine buccal mucosa was used as it resembles much better the human buccal mucosa regarding permeability, barrier lipid composition, histology and ultrastructural organization [23]. Buccal mucosa from pigs weighing 70–100 kg was obtained freshly from a local slaughterhouse and it was used at least within 3 h after animal slaughtering. Most of the underlying tissue was removed from the mucosa with surgical scissors. The buccal tissue

was dermatomed with a thickness of 500  $\mu\text{m}$  [24] using an electro-dermatome (Aesculap® Accu Dermatome, Germany).

### 2.5.2. Permeation studies

In vitro permeation studies were conducted in Ussing chambers with a diffusion area of 0.64  $\text{cm}^2$  and a compartment volume of 1 mL. All experiments were performed at 37 °C and protected against light. Dermatomed porcine buccal mucosa was mounted between the donor and the acceptor chambers, which were filled with 40 mM of Bis-Tris buffer at pH 7.0  $\pm$  0.5 and allowed to equilibrate for 15 min. The donor chamber was emptied after the equilibration period and replaced with 1 mL of a solution containing 300  $\mu\text{g}/\text{mL}$  of OME alone, complexed OME with cyclodextrins ( $\beta\text{CD}$  and  $\text{M}\beta\text{CD}$ ) and complexed OME with cyclodextrins in presence of ARG. Every 30 min, 100  $\mu\text{L}$  samples were withdrawn from the acceptor chamber and replaced by 100  $\mu\text{L}$  of Bis-Tris buffer at pH 7.0 and equilibrated at 37 °C. Samples were analyzed by HPLC in order to determine the amount of permeated OME. Cumulative corrections were made for the previously removed samples.

The apparent permeability coefficients ( $P_{\text{app}}$ ) for free and complexed OME were calculated according to the following equation:

$P_{\text{app}} = Q/(A \times c \times t)$ , where  $Q$  is the total amount permeated within the incubation time ( $\mu\text{g}$ );  $A$  is the diffusion area of the Ussing chamber ( $\text{cm}^2$ );  $c$  is the initial concentration of OME in the donor chamber ( $\mu\text{g}/\text{cm}^3$ );  $t$  is the total time of the experiment (s).

The cumulative amount of permeated drug was plotted versus time, and the steady state flux ( $J_{\text{ss}}$ ) was calculated using the formula:

$J_{\text{ss}} = \Delta M/(A \cdot \Delta t)$ , where  $\Delta M$  is the amount of drug transported across the membrane during the time  $\Delta t$  and  $A$  is the diffusional area.

### 2.6. Integrity studies

FD40 [impermeable fluorescein isothiocyanate (FITC)-labeled dextran] was added to the donor chamber after the permeability measurements (concentration 10  $\text{mg}/\text{mL}$ ), and quantified in the receptor chamber, after 1 h by fluorescence spectroscopy. Concentrations of FD40 were determined using a fluorescence spectrometer Infinite M200 TECAN, at an excitation wavelength of 495 nm and emission wavelength of 515 nm.

### 2.7. Viability studies

Samples of buccal mucosa of pig were cut, weighed and immersed in Bis-Tris buffer at pH 7.0  $\pm$  0.5 (negative control), in solutions containing inclusion complexes between OME and two cyclodextrins ( $\beta\text{CD}$  and  $\text{M}\beta\text{CD}$ ) in the absence and presence of ARG and in a solution with 2% (v/v) of Triton, used as positive control. Samples were incubated at 37 °C and 300 rpm during 3 h. MTT was dissolved (2  $\text{mg}/\text{mL}$ ) in fresh prepared buffer and filtered through a filter of 0.45  $\mu\text{m}$  to remove any dissolved crystals. After 3 h, 1 mL of MTT solution was added to each sample and the samples were placed on a rotating platform (300 rpm) at 37 °C for 2 h. After this time, the MTT solution was removed and the tissue was rinsed twice with 1 mL of buffer for 1 min and then minced with surgical scissors. To extract the water insoluble formazan, 2 mL of DMSO was added to each sample and stirred (300 rpm) for 80 min at 37 °C. The absorbance of formazan was measured at 540 nm (with DMSO as a blank) with the spectrophotometer Infinite M200 TECAN.

### 2.8. Cytotoxicity studies

The same procedure used in viability studies to prepare the samples was carried out. Samples were incubated at 37 °C while

stirred at 300 rpm during 3 h. Each hour a sample was removed and stored at 4 °C. All the samples were centrifuged at 5000 rpm during 2 min to remove possible interferences. Afterwards 50  $\mu\text{L}$  of each solution was mixed with 50  $\mu\text{L}$  of a specific reagent (Cytotoxicity Detection Kit, Dye solution + Catalyst). Final solutions were incubated at slow stirring for 30 min. The absorbance was measured at 496 nm with a spectrophotometer Infinite M200 TECAN.

### 2.9. Histological studies

Buccal tissues were cut with the electrical dermatome and incubated at 37 °C in vials containing Bis-Tris buffer pH 7.0  $\pm$  0.5, OME solution, complexed OME with cyclodextrins in the absence and presence of ARG and in a solution of Triton 2% (v/v). After 3 h, tissue samples were immersed in trypan blue solution during 20 min [25]. After that, samples were washed with buffer and fixed in Bovins solution (300 mL of picric acid, 100 mL of formaldehyde and 20 mL of acetic acid). A sample of buccal mucosa, extracted after slaughtering the animal, was used as control and transferred directly to the trypan blue solution and fixed in Bovins solution without any previous incubation. Tissue samples were then washed and dehydrated with a series of isopropanol grades ranging from 70% to absolute isopropanol, methylbenzoate, chloroform and finally samples were embedded in paraffin. Paraffin preparations were cut into slices and examined under an Olympus BH2 light microscope. The magnification of the microscope used was 10 $\times$ .

### 2.10. Data analysis

Statistical data analysis was performed using the *t*-test with  $p < 0.05$  as the minimal level of significance. The statistical analysis was done using GraphPad Prism® version 4.00 software. All experiments were run at least in triplicate.

## 3. Results and discussion

### 3.1. Stability studies

The OME stability in solution was studied in 40 mM of Bis-Tris buffer at different pH values (Fig. 1). After 3 h it was observed that 4% of the OME was degraded at pH 7.0 and 7.9. After 8 h, the remaining drug at pH 7.0, 7.9 and 9.0 was 78%, 87% and 100%, respectively. This study confirms that OME degrades at neutral

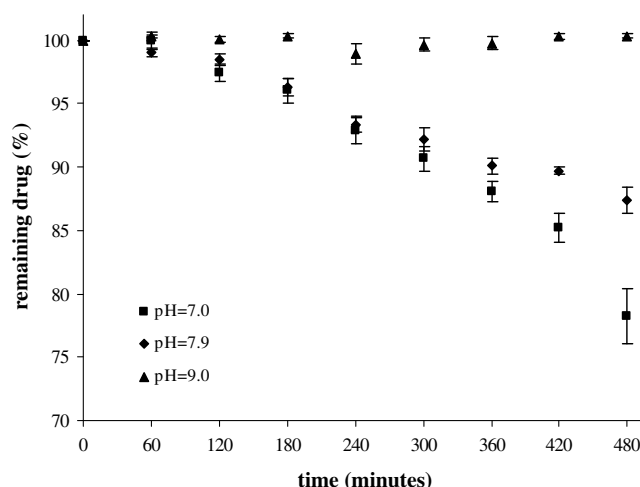
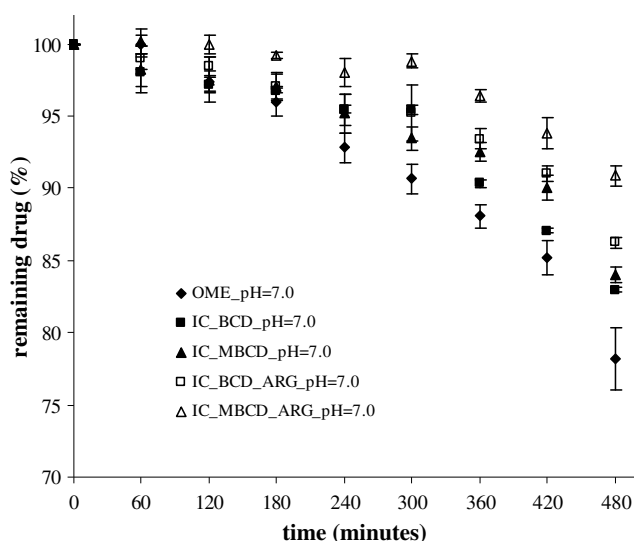


Fig. 1. Stability studies of OME in 40 mM of Bis-Tris buffer at different pH values. Each value is the mean of three independent assays ( $\pm$ SD,  $n = 3$ ).

conditions [26–28]. However, basic conditions are not suitable for drug delivery in physiological membranes. For that reason it is necessary to increase drug stability at neutral conditions in order to develop a pharmaceutical formulation with OME for administration in buccal mucosa.

The stability profiles of OME alone and complexed with  $\beta$ CD and M $\beta$ CD in the absence and presence of ARG are depicted in Fig. 2. Prior to start the stability studies, the pH of all solutions was adjusted to 7.0 in order to observe the real effect of the inclusion complexation in the absence and presence of ARG in OME stability. After 3 h the remaining drug calculated was 99% in the drug complexed with M $\beta$ CD in the presence of ARG and 97% in the OME complexed with  $\beta$ CD in presence of ARG. In the inclusion complex formed in the absence of ARG the remaining drug achieved was 97% for both cyclodextrins and 96% in the case of the drug alone. These results show that after 3 h the complexation with cyclodextrins cannot significantly increase OME stability. However, when the drug was complexed with M $\beta$ CD in the presence of ARG, it was not degraded at all during this period of time. The same result was not obtained for the inclusion complex formed with  $\beta$ CD in the presence of ARG. At the end of the study the remaining amount of drug in the solutions containing the drug alone, OME complexed with  $\beta$ CD and M $\beta$ CD and the inclusion complexes with ARG, was determined to be 78%, 83%, 84%, 86% and 91%, respectively. These values indicate that the inclusion of OME in the cyclodextrin cavity increases the OME stability. Nevertheless, in the presence of ARG the improvement on drug stability was more pronounced, especially when the drug was complexed with M $\beta$ CD. Previous studies have reported [21] that OME forms a more stable inclusion complex with the M $\beta$ CD than with the  $\beta$ CD due to the greater hydrophobic character of the former. Stability studies performed with OME in the presence of ARG at pH 7.0 (data not shown) could not increase OME stability suggesting that ARG alone acts as an alkali agent. These results propose that some interactions can occur between ARG and the inclusion complex. Consequently, ARG can stabilize the inclusion complex formed between OME and cyclodextrins, resulting in increased drug stability at neutral conditions.



**Fig. 2.** Stability studies of free and complexed OME with cyclodextrins ( $\beta$ CD and M $\beta$ CD) in absence and presence of ARG in 40 mM of Bis-Tris buffer at pH 7.0  $\pm$  0.5. Each value is the mean of three independent assays ( $\pm$ SD,  $n = 3$ ). Inclusion complex between OME and  $\beta$ CD (IC\_BCD); inclusion complex between OME and M $\beta$ CD (IC\_M $\beta$ CD); inclusion complex between OME and  $\beta$ CD in presence of ARG (IC\_BCD\_ARG) and inclusion complex between OME and M $\beta$ CD in presence of ARG (IC\_M $\beta$ CD\_ARG).

### 3.2. Validation of the HPLC method

The calibration curve and regression coefficient for the proposed method were:  $Y = 203742X - 140888$  and  $R^2 = 0.9998$  indicating a good linearity in the range of the study. The  $R^2$  obtained was higher than 0.999, as frequently recommended [29]. RSD values calculated are showed in Table 1. The results show a good precision, accuracy, repeatability and specificity of the analytical method. At last, results confirm that the presence of cyclodextrins ( $\beta$ CD and M $\beta$ CD), alkali agent (ARG) or degradation products do not interfere with the retention time corresponding to OME peak.

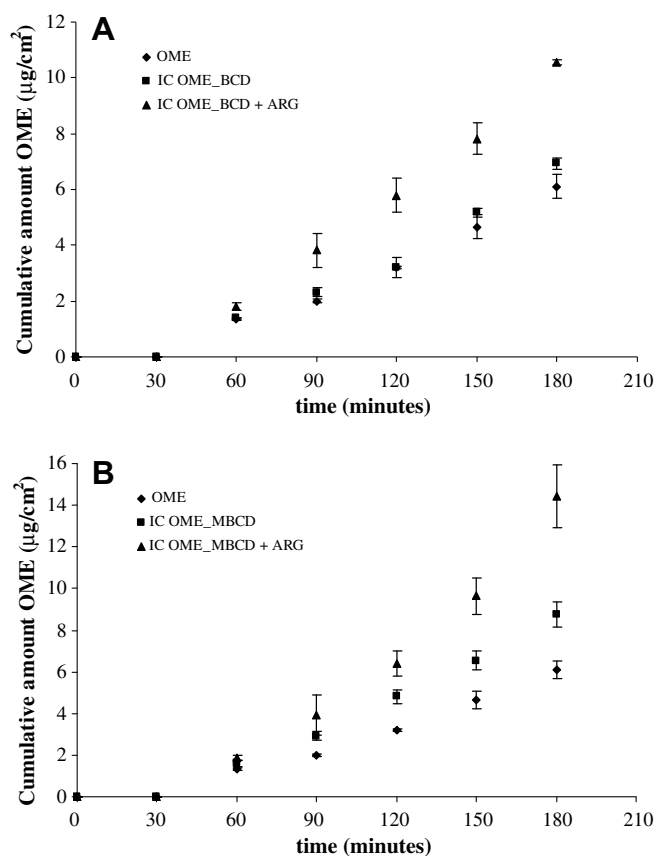
### 3.3. Permeation studies

In vitro permeability studies are an useful tool to assess the potential of a localized anatomical site as a route for drug delivery. However, in vitro conditions should simulate the in vivo situation as closest possible.

The permeability profiles of OME alone, complexed with  $\beta$ CD and M $\beta$ CD and complexed OME in presence of ARG, are shown in Fig. 3A and B. Table 2 contains apparent permeability coefficients ( $P_{app}$ ) and flux values of the different studied systems. The cumulative amount of complexed OME with  $\beta$ CD absorbed over 3 h through epithelium was only 1.1-fold greater than the amount of absorbed OME alone. In OME complexed with M $\beta$ CD the increase in drug absorbed was 1.4-fold compared to the OME alone. These results suggest that  $\beta$ CD acts as a carrier, keeping the hydrophobic OME molecules in solution and delivering them to the surface of the buccal mucosa. In the case of M $\beta$ CD, the enhancement in drug absorption indicates that this cyclodextrin acts as an enhancer of permeation by different mechanisms as reported in the literature [11,12]. Lipophilic cyclodextrins such as M $\beta$ CD, can permeate biomembranes, interact with the lipids and increase drug uptake through the buccal mucosa. Permeation studies with OME complexed with both cyclodextrins ( $\beta$ CD and M $\beta$ CD) in the presence of ARG show an enhancement of 1.7- and 2.4-fold, respectively, when compared with the amount of absorbed OME alone. This fact suggests that beyond the permeation enhancement effect of the cyclodextrins, the presence of ARG can stabilize the inclusion complex formed and, this stabilizing effect of the OME at the biomembrane surface increases the amount of OME available to permeate the buccal epithelium. On the other hand, a recent study [30] shows that the poly-L-arginine can act as an enhancer of the mucosal epithelia permeability by positively charged amino groups, interacting with negatively charged sites on the epithelial membrane and tight junctions. At neutral conditions, ARG is in the cationic form being able to interact with negatively charged molecules situated in the mucus layer [31] on the surface of the buccal

**Table 1**  
RSD values of the analytical parameters used to validate the HPLC method

Standard solution ( $\mu$ g/mL)	RSD %
<b>Precision (<math>n = 6</math>)</b>	
12.5	2.207
<b>Accuracy (<math>n = 9</math>)</b>	
6	2.515
12.5	0.715
25	0.599
<b>Repeatability (<math>n = 6</math>)</b>	
12.5	3.454
<b>Specificity (<math>n = 9</math>)</b>	
100	3.314
100 + $\beta$ CD	0.721
100 + M $\beta$ CD	0.247
100 + $\beta$ CD + ARG	0.308
100 + M $\beta$ CD + ARG	0.979



**Fig. 3.** In vitro permeation studies in porcine buccal mucosa of free and complexed OME with  $\beta$ CD (A) and M $\beta$ CD (B) in the absence and in presence of ARG solution. Each value is the mean of three assays ( $\pm$ SD,  $n = 3$ ).

**Table 2**

Apparent permeability coefficients ( $P_{app}$ ) and flux values of different studied systems

Systems	$P_{app}$ (cm/s)	Ratio <sup>a</sup>	Flux ( $\mu\text{g}/\text{cm}^2 \text{ h}$ )
OME	$1.883\text{E}-06 \pm 7.692\text{E}-08$	1.0	$2.382 \pm 0.185$
IC OME_ $\beta$ CD	$2.143\text{E}-06 \pm 3.638\text{E}-08$	1.1	$2.685 \pm 0.092$
IC OME_M $\beta$ CD	$2.702\text{E}-06 \pm 1.040\text{E}-07$	1.4	$3.455 \pm 0.217$
IC OME_ $\beta$ CD + Arg	$3.269\text{E}-06 \pm 1.900\text{E}-08$	1.7	$4.161 \pm 0.117$
IC OME_M $\beta$ CD + Arg	$4.445\text{E}-06 \pm 2.686\text{E}-07$	2.4	$5.588 \pm 0.564$

Indicated values are means ( $\pm$ SD,  $n = 3-4$ ).

<sup>a</sup> Enhancement ratio =  $P_{app}$  (sample)/ $P_{app}$  (control).

mucosa. This suggestion is supported by the permeation studies performed with OME in presence of ARG (data not shown) was observed a 1.2-fold increase in the amount of permeated drug in presence of ARG compared to the amount of OME alone permeated.

In the integrity studies, fluorescence values calculated for all the systems after 1 h of incubation with a solution containing 10 mg/mL of FD40, indicated a concentration of FD40 in the acceptor chamber below 0.01%. This data suggests that the tissue integrity of buccal mucosa is maintained during the permeation studies. The principle here is that FD40 being a very large molecule, cannot permeate the buccal mucosa and enter the acceptor chamber unless the mucosa is damaged.

### 3.4. Viability studies

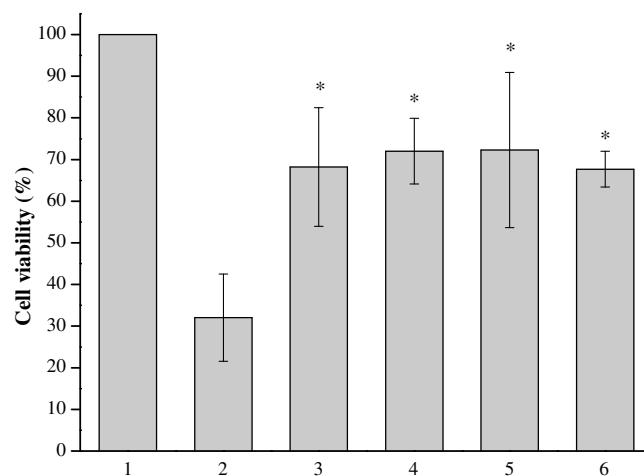
The MTT assay has been used for quantitative colorimetric measurements of mammalian cell survival and proliferation. The original assay has been modified to assess the viability of the tissue specimens [32].

MTT is converted in viable cells to formazan (a dark purple water insoluble compound) by enzymes in active mitochondria collectively known as tetrazolium reductase and the amount of formazan generated is directly proportional to the number of living cells [33]. The results of cell viability (%) after 3 h of incubation of porcine buccal mucosa with samples solutions, negative and positive control are shown in Fig. 4. To calculate the percentage of cell viability, sample in Bis-Tris buffer (negative control) was used as 100% of cellular viability. It was observed that cell viability of the buccal mucosa after 3 h of incubation remained around 70% in all samples as compared to Bis-Tris buffer (negative control). By contrast, Triton 2% (v/v) treated cells displayed a significant decrease ( $p < 0.05$ ) of viability after 3 h of incubation (around 30%) when compared with negative control and all sample solutions.

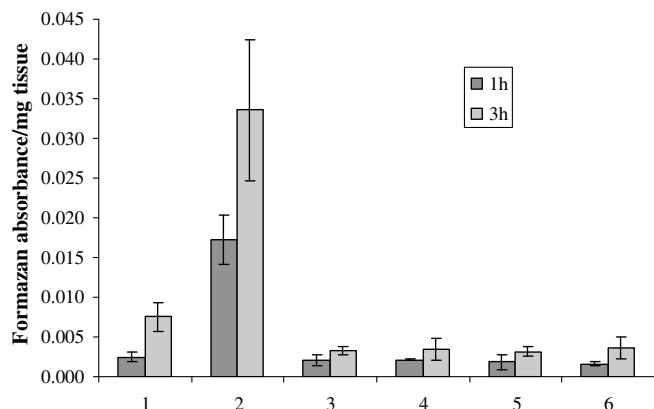
### 3.5. Cytotoxicity studies

Cell death is typically assayed by quantifying plasma membrane damage and in recent years, lactate dehydrogenase (LDH) activity has been used in toxicological studies in mucosal membranes [34,35]. LDH is a stable cytoplasmic enzyme which is present in all cells. When the plasmatic membrane is damaged, LDH is rapidly released into the supernatant medium and its activity can be determined by a colorimetric reaction. In the first step  $\text{NAD}^+$  is reduced to  $\text{NADH}/\text{H}^+$  by the LDH-catalyzed conversion of lactate to pyruvate. In the second step, a catalyst added to the supernatant medium (diaphorase) transfers  $\text{H}/\text{H}^+$  from  $\text{NADH}/\text{H}^+$  to the tetrazolium salt, which was reduced to a formazan dye. An increase in the number of dead or plasma membrane-damaged cells leads to an increased LDH activity in the supernatant medium, which directly correlates with the amount of formazan produced. Therefore, the amount of dye produced is proportional to the number of lysed (dead or plasmatic membrane damaged) cells [36].

Methylated  $\beta$ -cyclodextrins interact strongly with lipids [11], for that reason, cell membrane integrity was evaluated by quantification of the LDH released. Until 1 and 3 h, no significant differences in LDH activity (absorbance of formazan/mg of tissue) were observed in the samples incubated with inclusion complexes between OME and both cyclodextrins ( $\beta$ CD and M $\beta$ CD) in absence



**Fig. 4.** Cell viability in porcine buccal mucosa measured by MTT assay expressed as the mean values obtained from three inserts ( $\pm$ SD,  $n = 3$ ). (1) Buccal mucosa after 3 h exposed to Bis-Tris buffer 40 mM (positive control); (2) buccal mucosa after 3 h exposed to Triton 2% (v/v); (3) buccal mucosa after 3 h exposed to OME\_ $\beta$ CD inclusion complex in solution; (4) buccal mucosa after 3 h exposed to OME\_M $\beta$ CD inclusion complex in solution; (5) buccal mucosa after 3 h exposed to OME\_ $\beta$ CD inclusion complex in ARG solution; (6) buccal mucosa after 3 h exposed to OME\_M $\beta$ CD inclusion complex in ARG solution; all these samples are statistically different compared to the sample exposed to Triton 2% (v/v), ( $p < 0.05$ ).



**Fig. 5.** Formazan absorbance resultant of LDH activity released from porcine buccal mucosa after 1 and 3 h exposed to: (1) Bis-Tris buffer 40 mM; (2) Triton 2% (v/v) solution; (3) OME\_βCD inclusion complex in solution; (4) OME\_MβCD inclusion complex in solution; (5) OME\_βCD inclusion complex in ARG solution; (6) OME\_MβCD inclusion complex in ARG solution.

and in presence of ARG and the sample exposed to Bis-Tris buffer 40 mM (Fig. 5). At the same time, the sample exposed to Triton 2% (v/v) showed a significant increase in the absorbance of formazan between 1 and 3 h ( $p < 0.05$ ). These results demonstrate that cyclodextrins, namely βCD and MβCD even in presence of an alkali agent, ARG, do not present cytotoxic effects to the porcine buccal mucosa when compared with Triton solution.

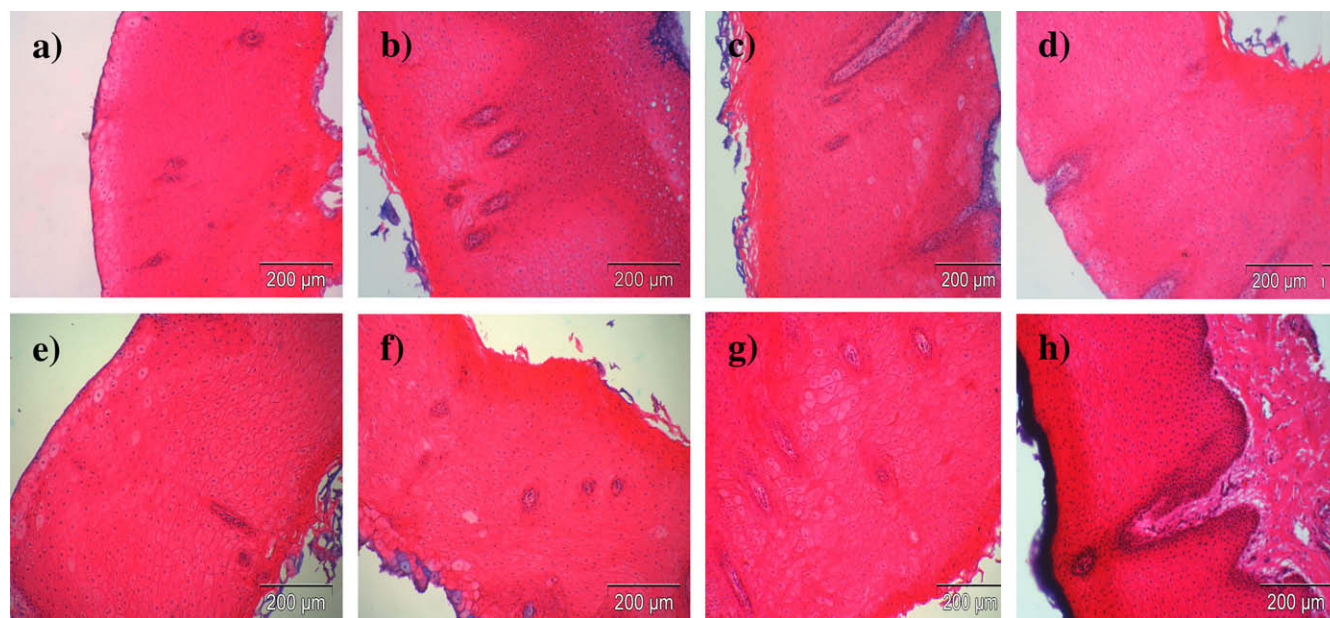
### 3.6. Histological studies

Fig. 6 shows the photomicrographs of the buccal epithelium after slaughtering the animal, 3 h after incubation at 37 °C with Bis-Tris buffer pH 7.0 ± 0.5, OME alone, complexed with cyclodextrins in the absence or presence of ARG and in a solution of Triton 2% (v/v), which is known to be extensively toxic [37]. In some photomicrographs it was possible to observe a separation of the epithelium from the connective tissue and only the buccal epithelium is present [38]. Histological studies showed that after 3 h of

incubation with sample solutions there were no cell leakage and no visible changes in the thickness of the superficial layer was observed, indicating that the buccal epithelium appeared viable when compared with the tissue incubated with buffer or the tissue removed immediately after death of the animal. Only few cells in the superficial layer of some preparations appeared to be dead, probably due to the mechanical stress during tissue preparation. However, the thickness of the buccal epithelium was greatly reduced after 3 h exposed to a Triton solution. Treatment of buccal epithelium with this solution resulted in cellular changes and tissue necrosis probably due the detergent effect of this compound. These histological results support data obtained in the viability and cytotoxicity studies.

### 4. Conclusion

Buccal drug delivery offers an alternative to conventional oral administration for drugs that show low stability at acidic conditions of the stomach and a strong first hepatic effect. However, buccal mucosa represents an effective absorption barrier and new strategies must be found to overcome it. The effect of cyclodextrins to increase OME stability and permeability through the buccal epithelium was studied and the integrity, viability and cytotoxicity effects were evaluated after the exposition of the buccal tissue with this kind of permeation enhancers. Stability studies performed at neutral conditions suggested that the complexation of OME increased drug stability and in the presence of ARG this effect was improved. The in vitro transbuccal permeation of OME was enhanced in the presence of cyclodextrins. This increase was highest with MβCD indicating that this lipophilic cyclodextrin can permeate buccal mucosa and consequently to enhance drug delivery through the biomembrane. The complexation of OME with MβCD in presence of ARG increases drug permeation 2.4-fold, suggesting that ARG favors drug permeation due to the establishment of ionic interactions with negative charges present in mucus layer on the surface of the mucosa. Viability studies showed that buccal mucosa remains viable after 3 h incubation period with OME in complexed form in absence and in presence of ARG. These results were supported by histological evaluations. No toxicity effects



**Fig. 6.** Light microscopic view of buccal mucosa extracted after slaughtering the animal (a) control; 3 h after incubation at 37 °C with: Bis-Tris buffer (b), OME solution (c), complexed OME with βCD and MβCD in solution (d and e); complexed OME with βCD and MβCD in presence of ARG in solution (f and g) and 2% (v/v) of Triton solution (h), (magnification, 10×).

were observed in buccal mucosa after incubation with OME complexed with cyclodextrins even in the presence of ARG. The results presented here demonstrate that complexation of OME with M $\beta$ CD in presence of ARG may be promising approach to increase drug stability and permeation through the buccal mucosa, that ultimately can result in improve drug bioavailability.

## Acknowledgements

This work was financially supported by a grant (Praxis SFRH/BD/19175/2004) from FCT (Fundação para a Ciência e a Tecnologia, Portugal). The authors thank the technical assistance of Prof. Klima and Rudolf Haring (Department of Histology and Embryology, University of Innsbruck) in the preparation and visualization of the histological samples and Dr. Christian Leiß (Department of Pharmaceutical Technology, Faculty of Pharmacy, University of Innsbruck) in the preparation of buccal tissue for the permeation studies. We also acknowledge Belmac Laboratory, S.A. (Madrid, Spain) for the kindly donating the OME and Roquette (Lestrem, France) for the  $\beta$ CD and M $\beta$ CD samples.

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