

Preparation and characterization of metal supported solid oxide fuel cells with screen-printed electrodes and thin-film electrolyte

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Content

Introduction of SOFCs

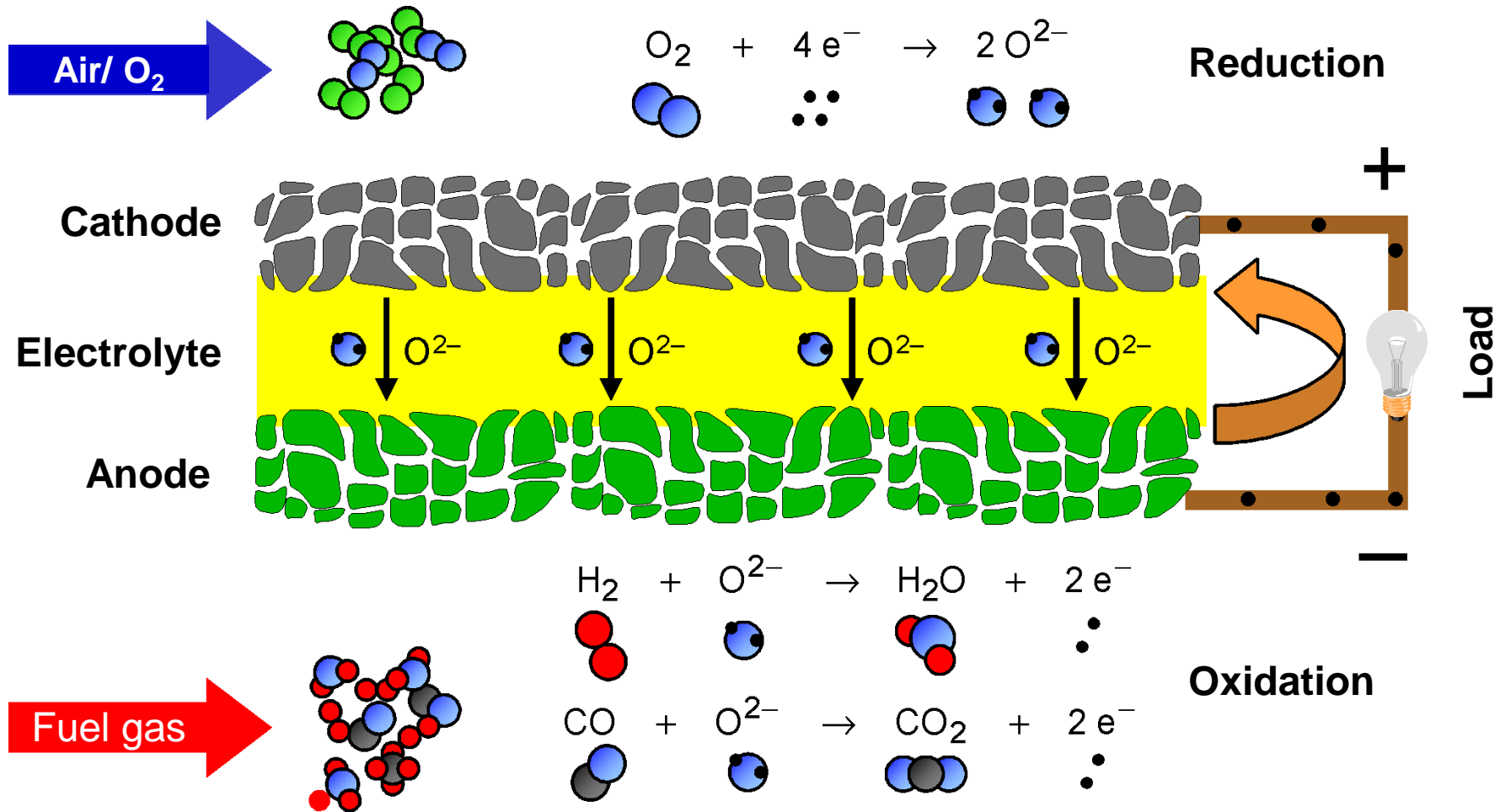
Processing of planar SOFCs

MSC with screen-printed electrodes and thin-film electrolyte

Summary & Perspectives



Repetition of Basic Principle



Source : L.G.J. de Haart, IEF-3, FZJ



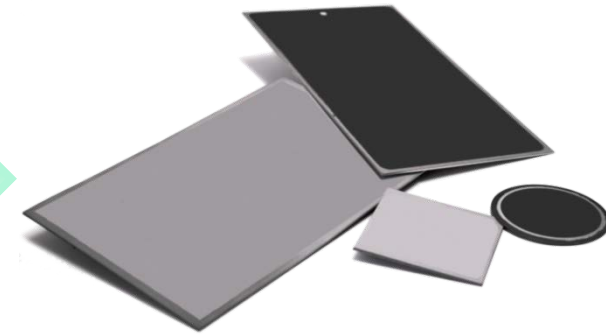
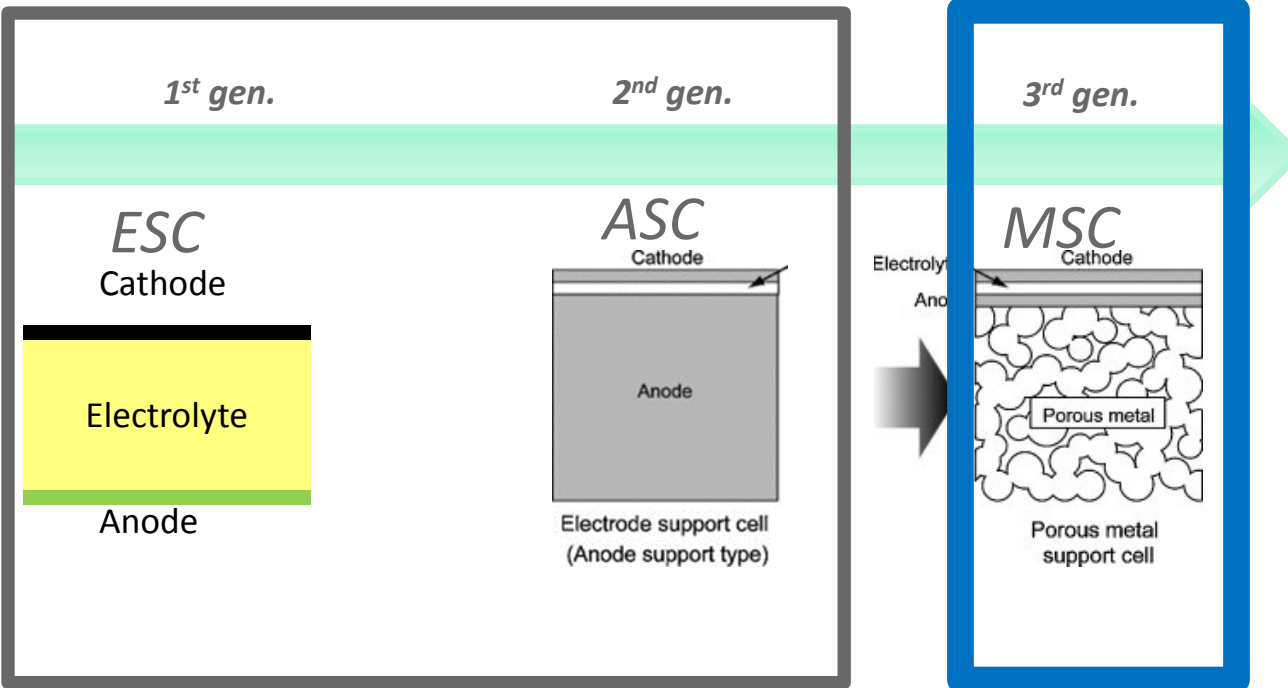
Motivation: towards the next generation SOC



Flexible architecture for multiple applications
Which materials for the next generation of SOCs?



Potential Advantages of Metal Supported Cells



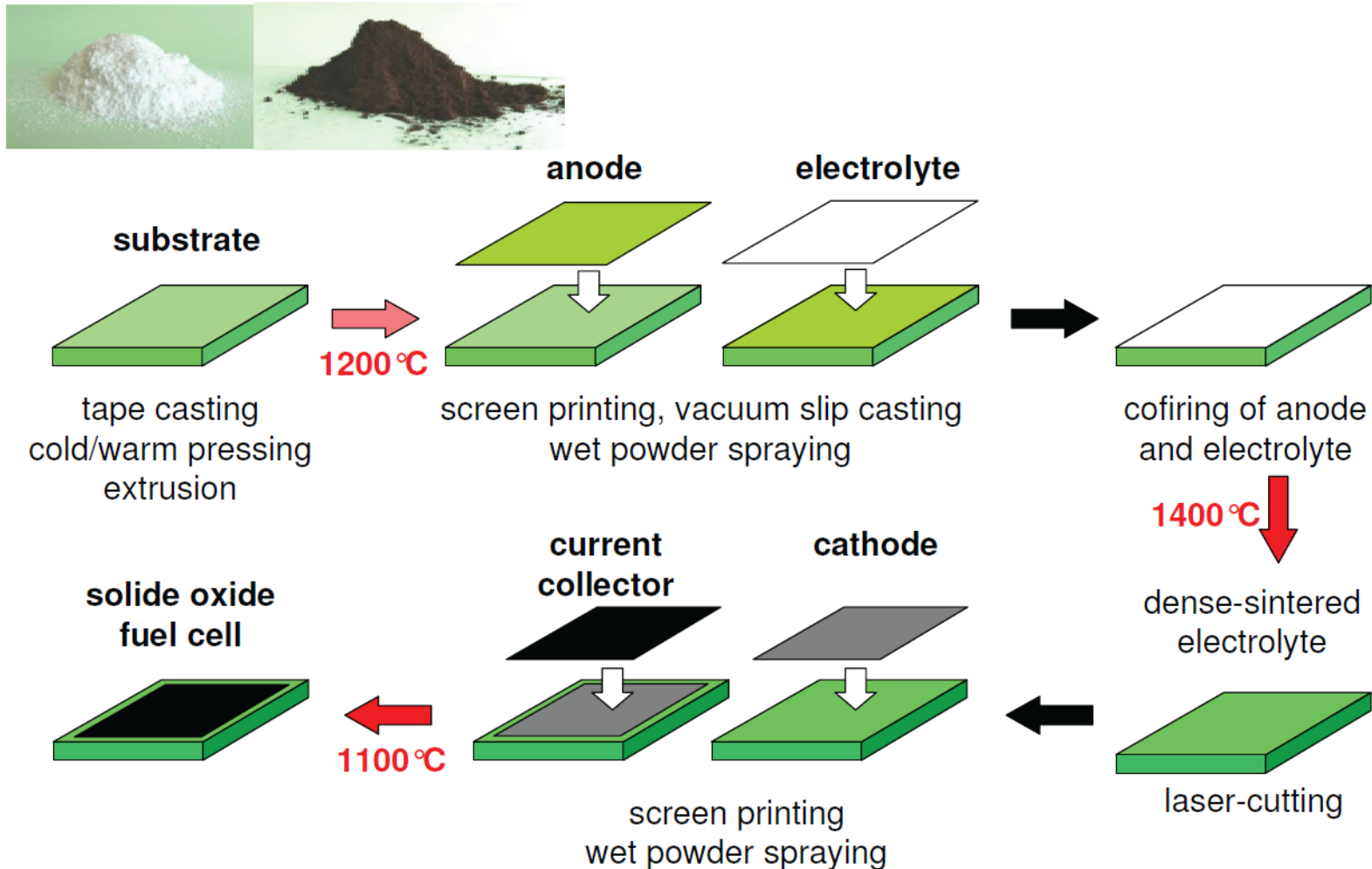
- To Replace ceramic components by metals
- Operating temperature > 600 °C
- Atmosphere: Hydrogen / or Synthetic Gas, Air
- Reversible operation

Metal supported Cell (MSC):

- High robustness
- High resistance against thermal and redox cycling
- Good integration into interconnects (bipolar plates) via brazing or welding
- Low cost of metal support and cell materials (thin layers)
- High electronic and thermal conductivity



Manufacturing Route (conventional ASC)

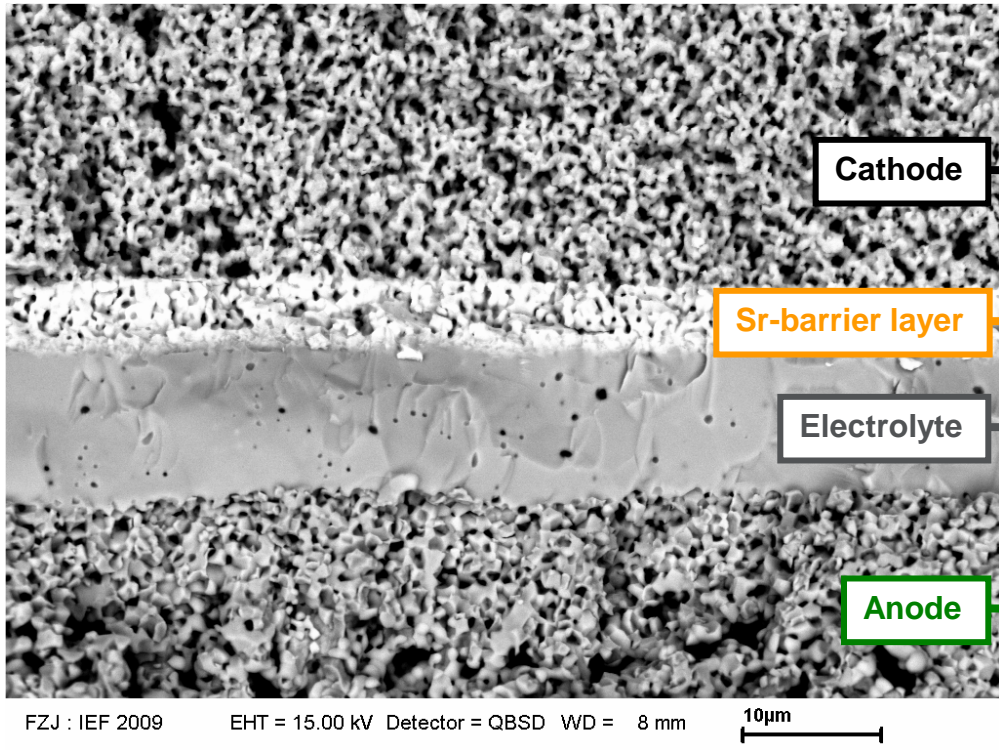


Alternatives without sintering: plasma spraying, PVD

M. Bram, IEF-1, FZJ



SEM fracture surface of anode-supported SOFC



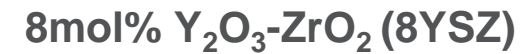
Cathode



Sr-barrier layer



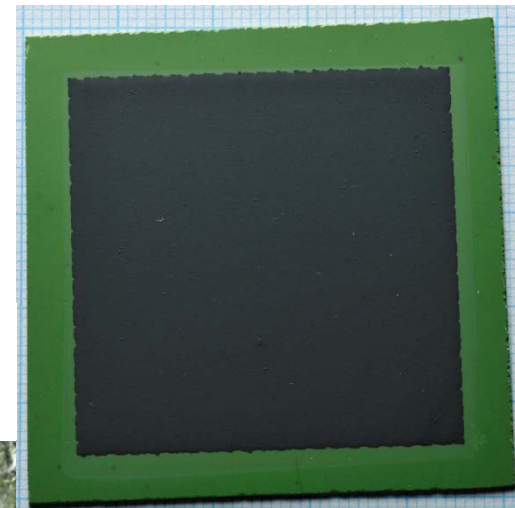
Electrolyte



Anode



Sintered single cell
50mmx50mm

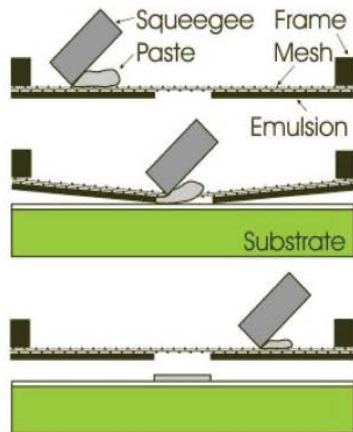


Screen Printing

origin: printing industry (non-paper prints, e.g. CD etc.)
to print letters and graphics

ceramic applications: large screen opening
(50-70% open area)
easy to create contoured coatings

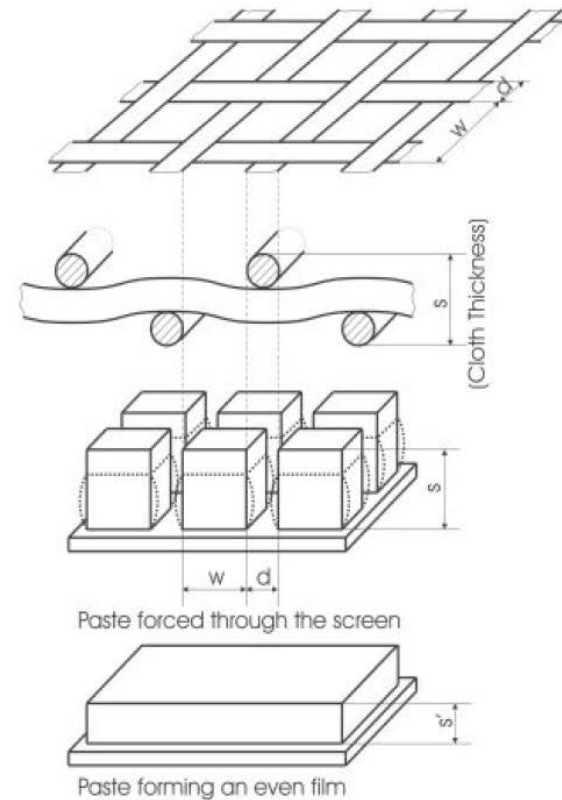
thickness of dried layer $\sim 1/4..1/2$ of wet layer
(spreading and drying)
 $\Rightarrow 5 \dots 100 \mu\text{m}$ feasible



printing scheme



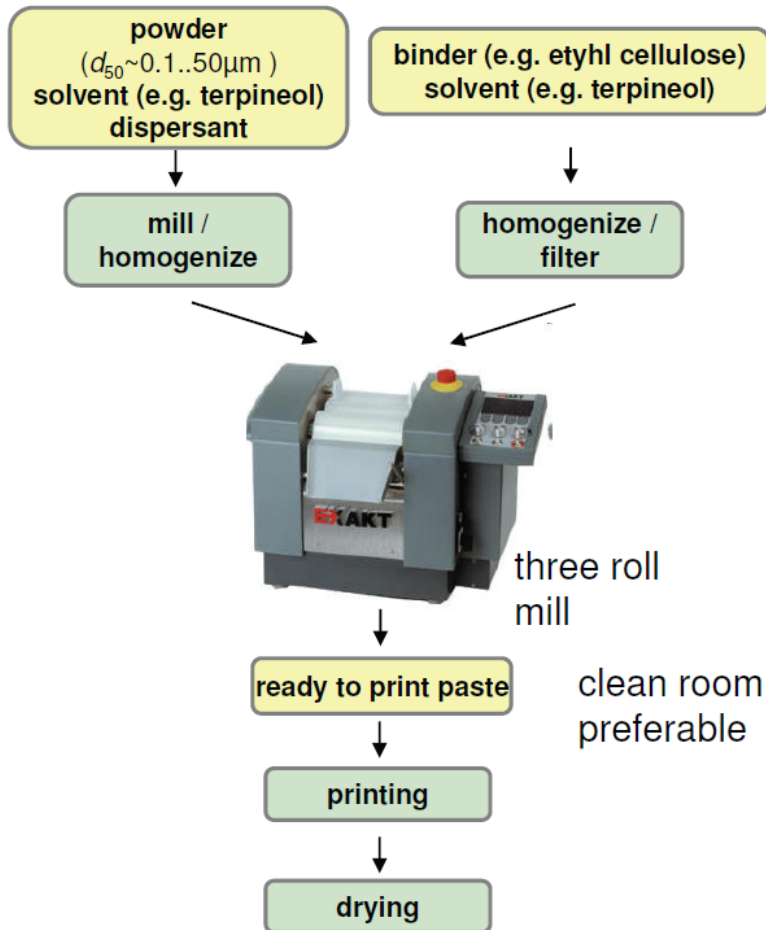
threads by polymer, steel,
or liquid crystal polyarylates



screen mesh and wet film



Screen Printing Paste Preparation

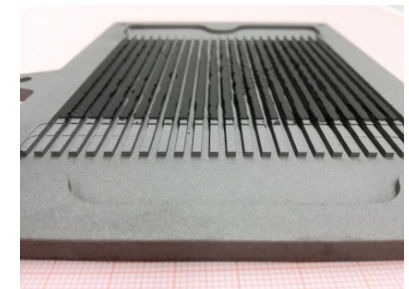


Paste for Energy Devices at DLR

1. SOFC/SOEC
2. PEMFC/PEMEC
3. Battery
4. Thermoelectric Generator
5. Gas Separation Membrane

Paste for SOFC functional layers

1. Anode: NiO-YSZ, NiO-GDC, NiO-BCZY, LST-NiO
2. Cathode: LSM-YSZ, LSCF, LSC
3. Electrolyte: YSZ, GDC, BCZY
4. Contact layer: LSCF
5. Sealing: glass sealant



Contact coating for Bi-polar plate



Objectives in this work

Demonstration of the feasibility of the cell concept and design

Development of metal supported SOFCs without nickel as structural components, improving redox cycling stability and sulfur tolerance.

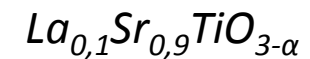
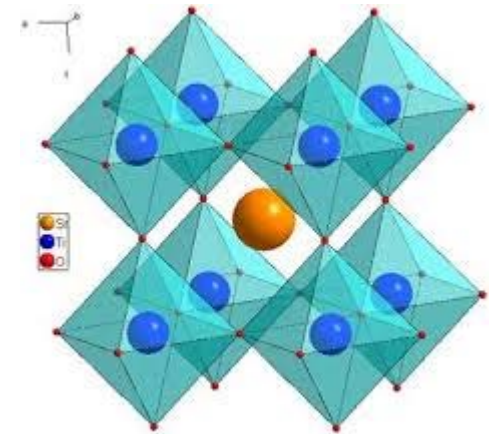
Deposition of gas-tight thin-film electrolyte ($\sim 3 \mu\text{m}$ thick) layers

Apply perovskite based materials for anode functional layer in MSC



Material Selection

- *Improving durability of the metallic substrate*
 - **Implementing alumina forming alloys (NiCrAl)**
- *Enhancing sulfur tolerance and redox stability at the anode*
 - **Perovskite based anode materials**
- *Ensuring the electrolyte with good gas tightness and electrical property*
 - **Multi-layered thin film**
- *Avoiding High T sintering and reducing atmosphere*
 - **low T (max. 1000 °C) processing in air**



screen printing

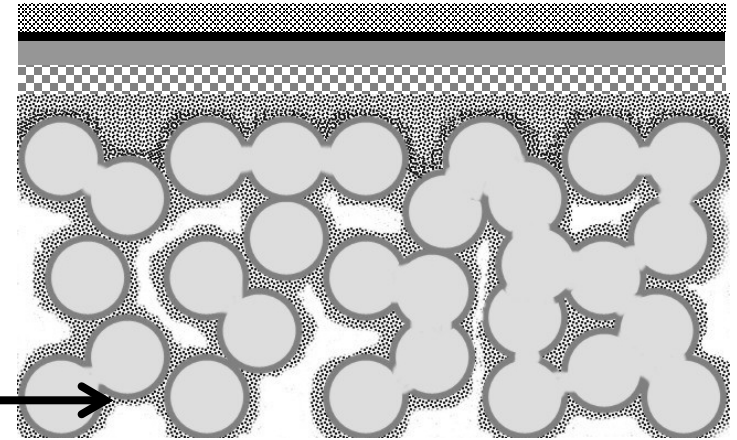


Materials and Architecture

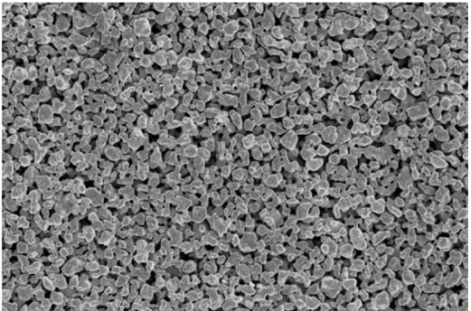
Cathode : $La_{0,4}Sr_{0,6}Co_{0,2}Fe_{0,8}O_{3-\alpha}$ (LSCF)

Electrolyte: 8YSZ / GDC10

Composition of the anode: GDC10-LST (w/o 5-10%Ni)


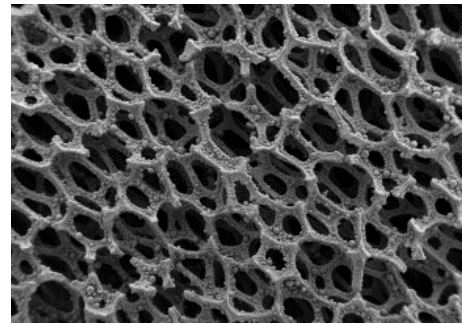




Metallic substrate at the fuel side



*Porous steel
pore size <math><50\mu\text{m}</math>*

Type 1



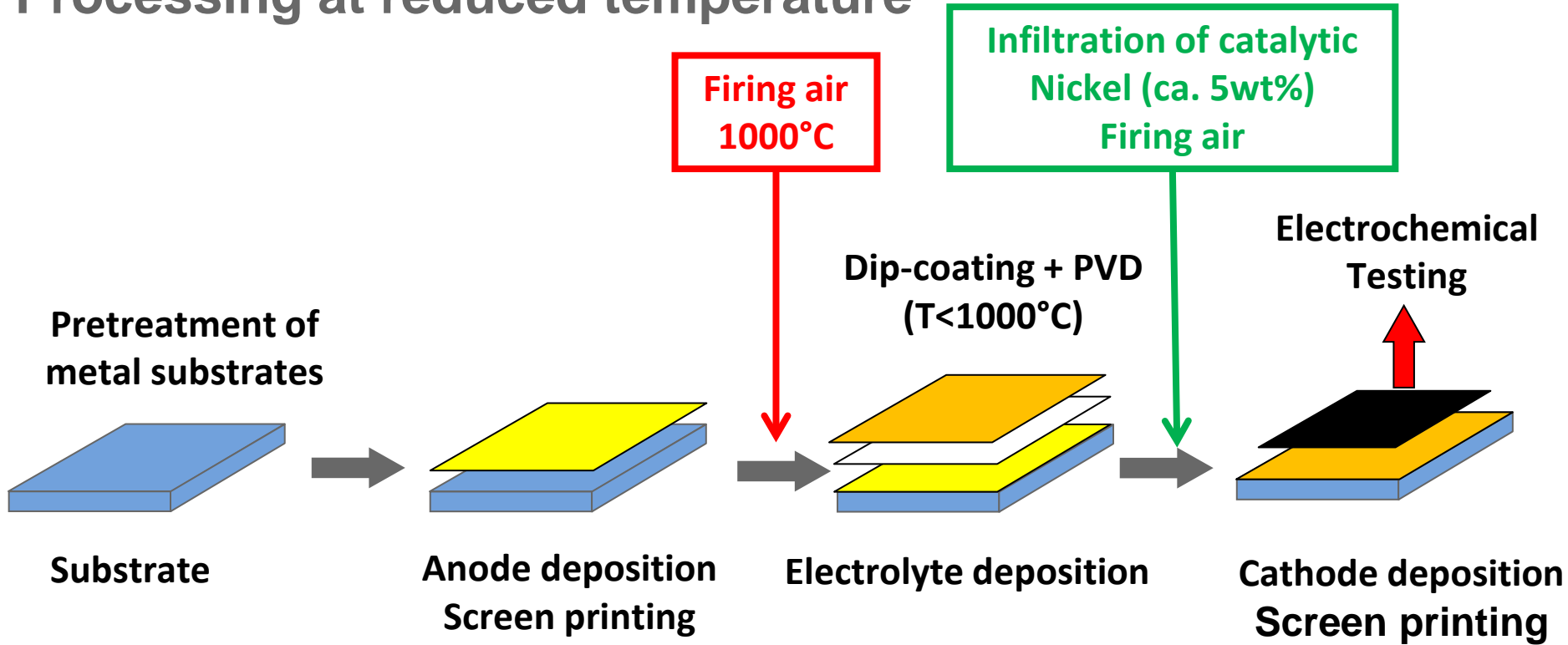
*Metal foam (NiCrAl)
pore size $450\mu\text{m}$*

*$La_{0,1}Sr_{0,9}TiO_{3-\alpha}$
 $NiO + La_{0,1}Sr_{0,9}TiO_{3-\alpha}$ (50:50)*

Type 2

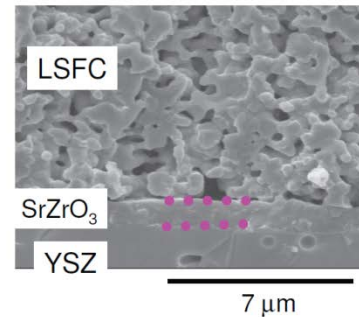


Processing at reduced temperature



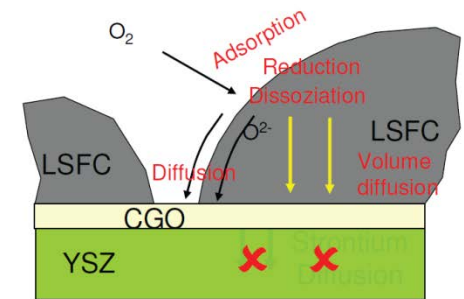
Fabrication options for electrolyte:

1. PVD
2. Plasma-spray (VPS)
3. Sintering
4. Sol-gel



disadvantage

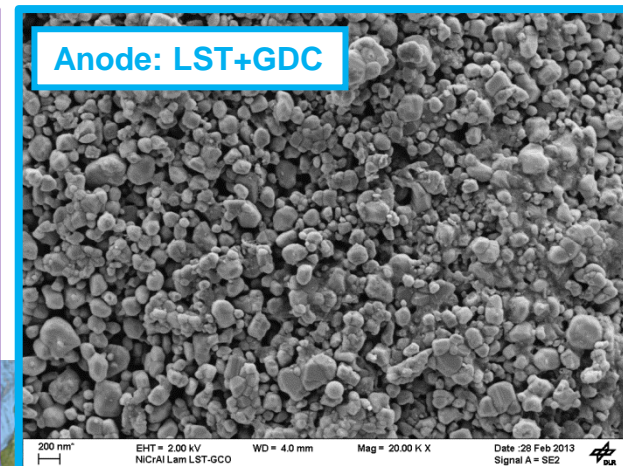
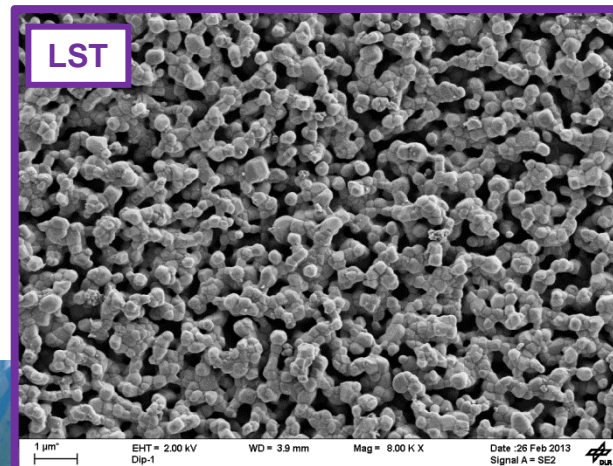
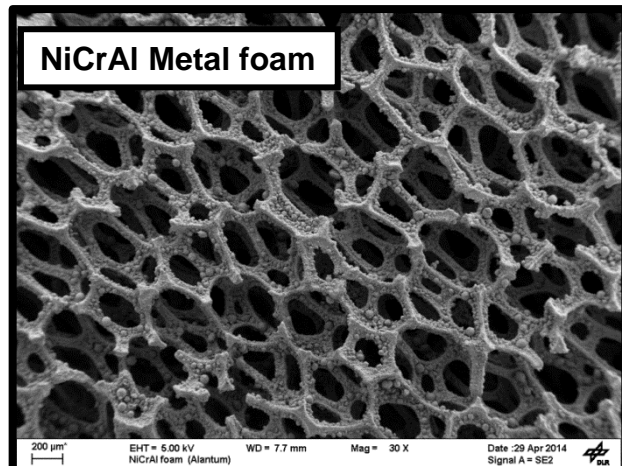
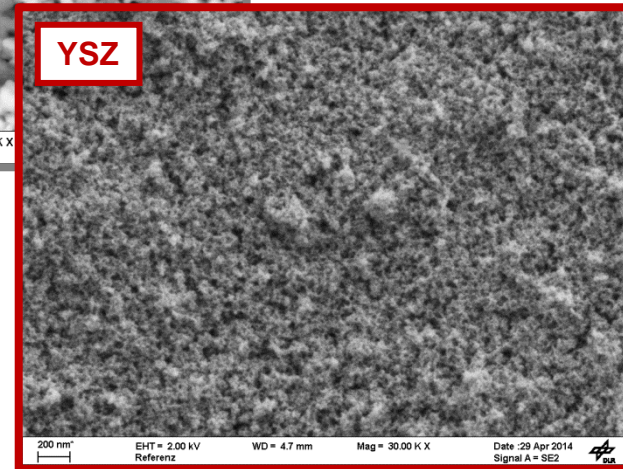
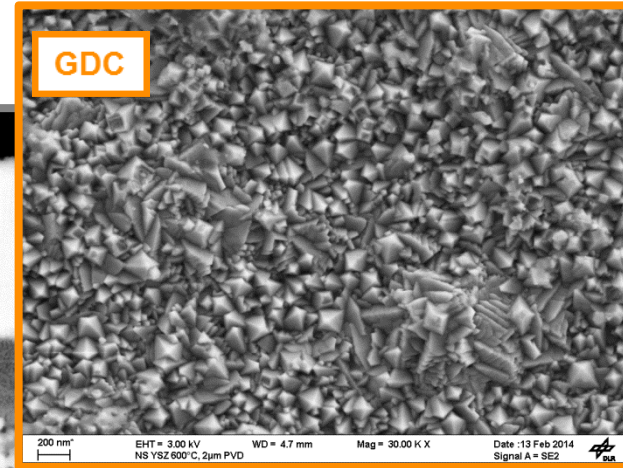
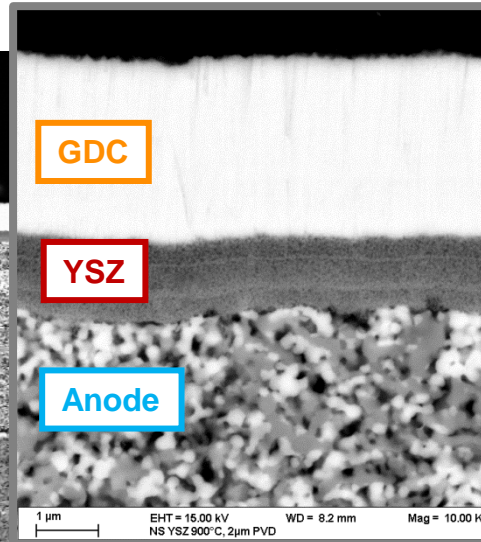
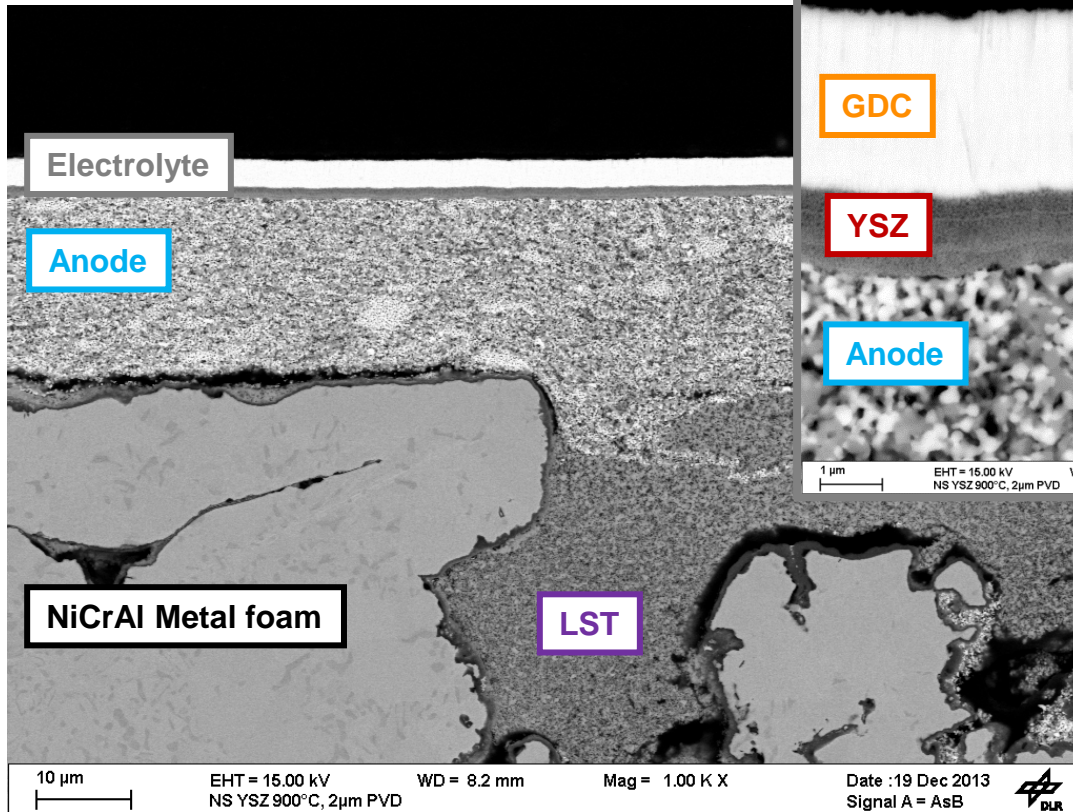
Reaction with YSZ
forming SrZrO₃



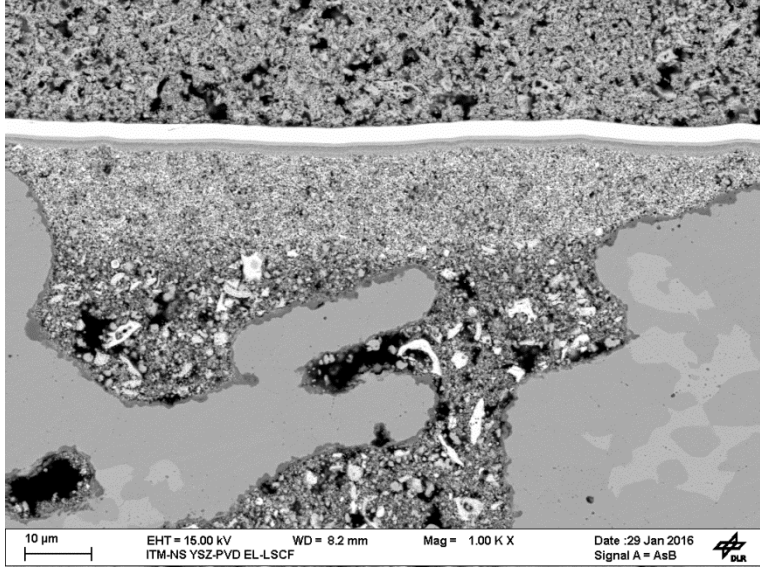
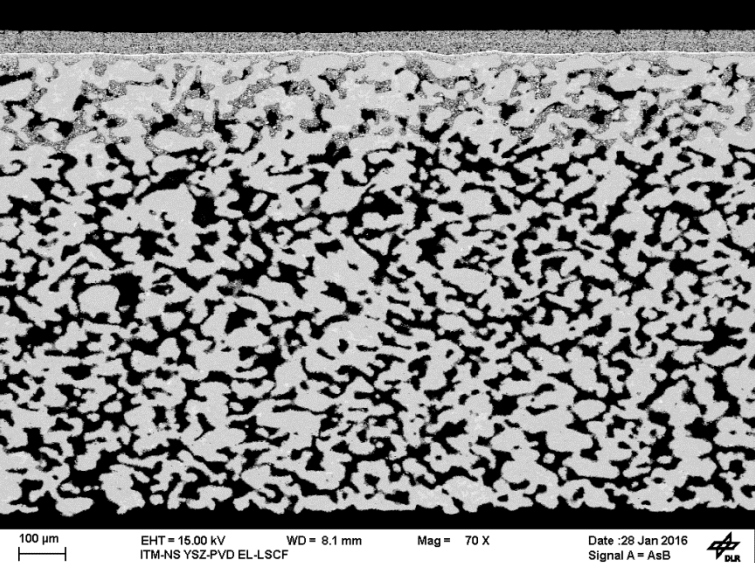
solution

Interlayer Ce_{0.8}Gd_{0.2}O_{1.9}

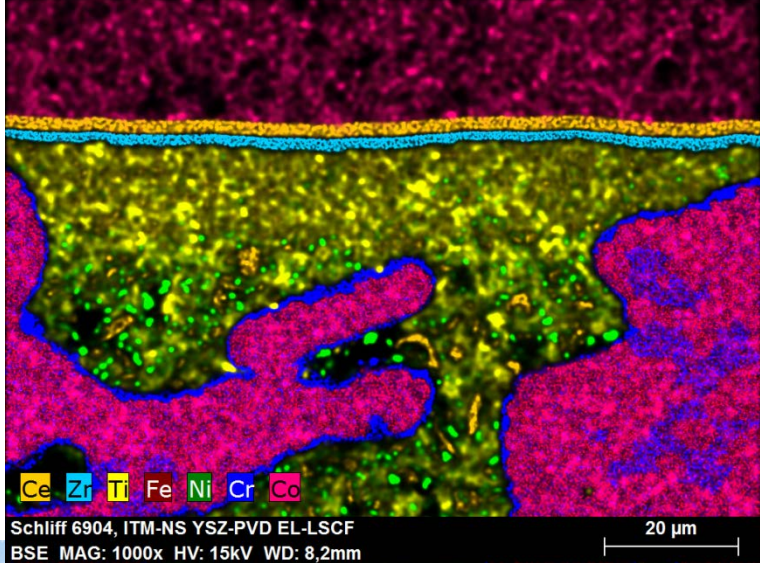
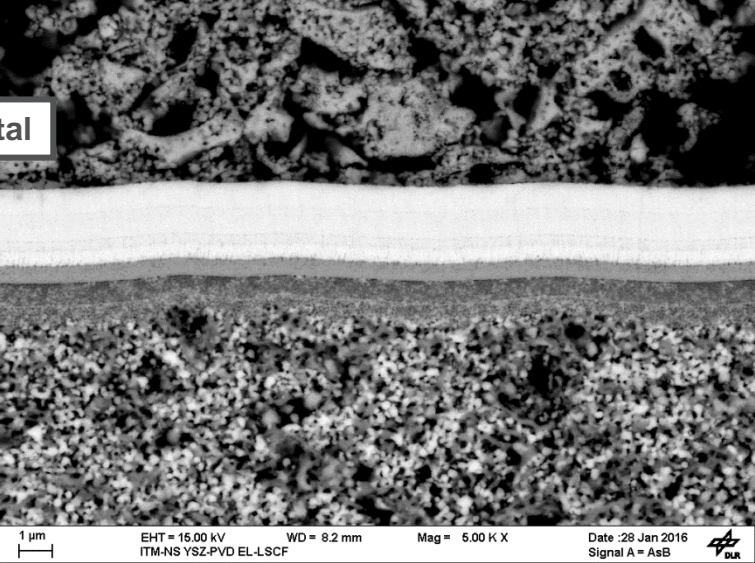
A Way to Thin Electrolyte



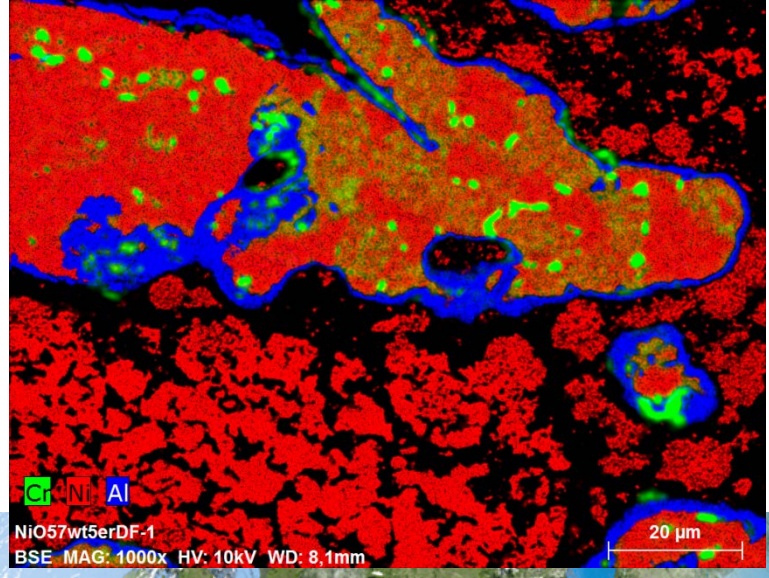
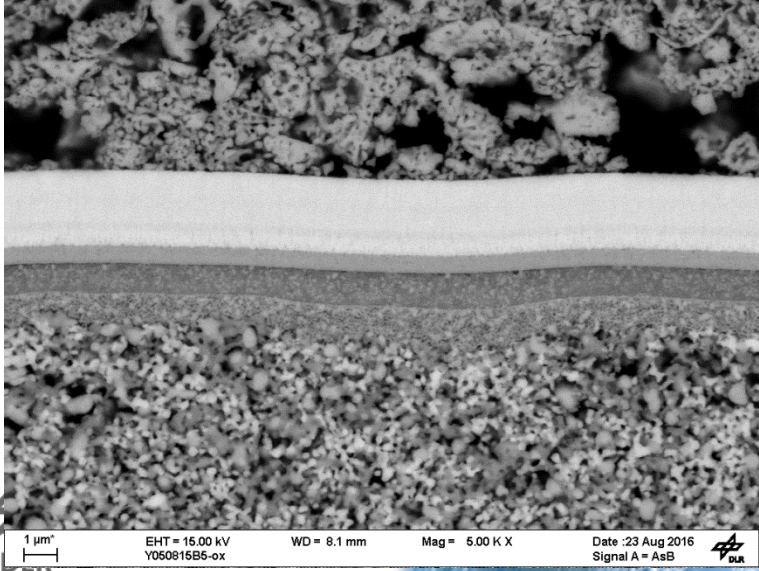
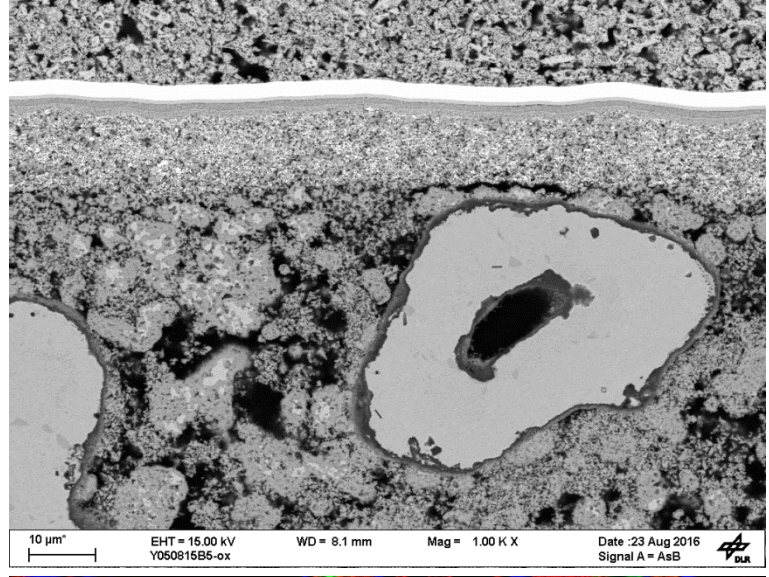
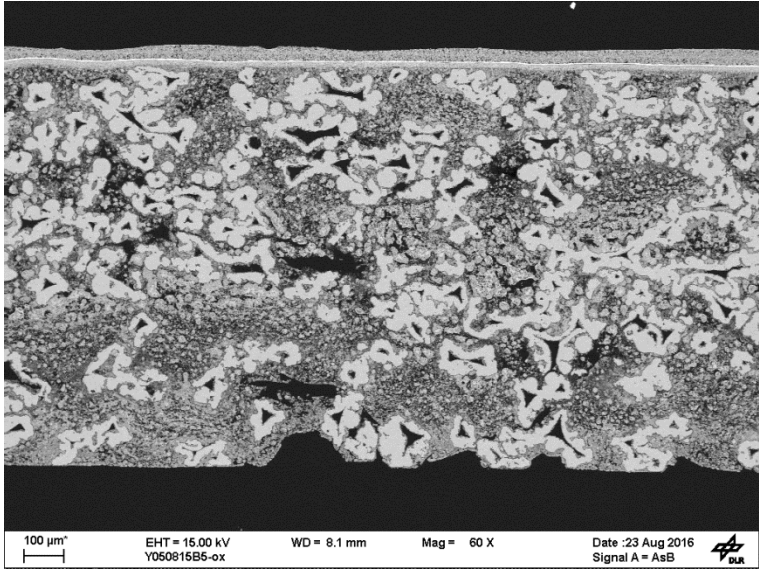
MSC with ITM Substrate (Type 1)



Porous metal



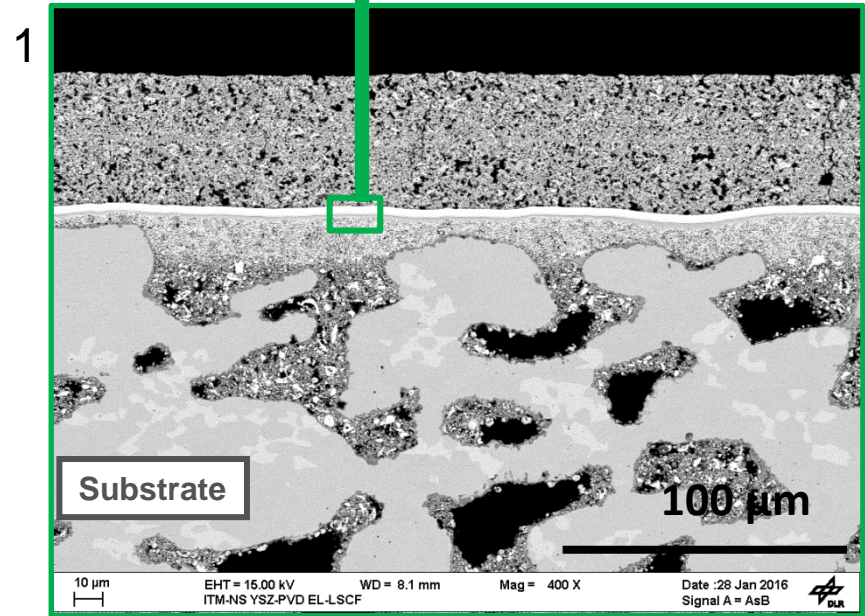
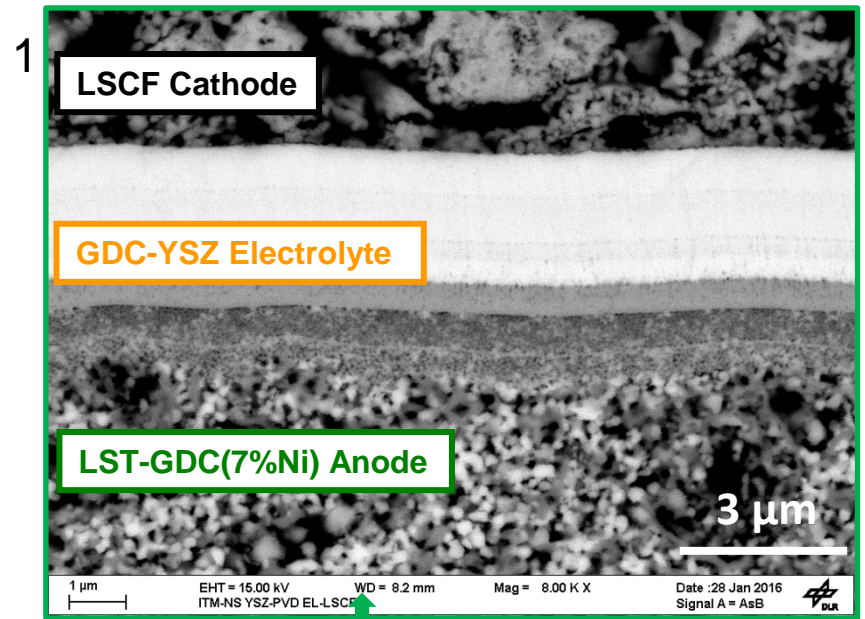
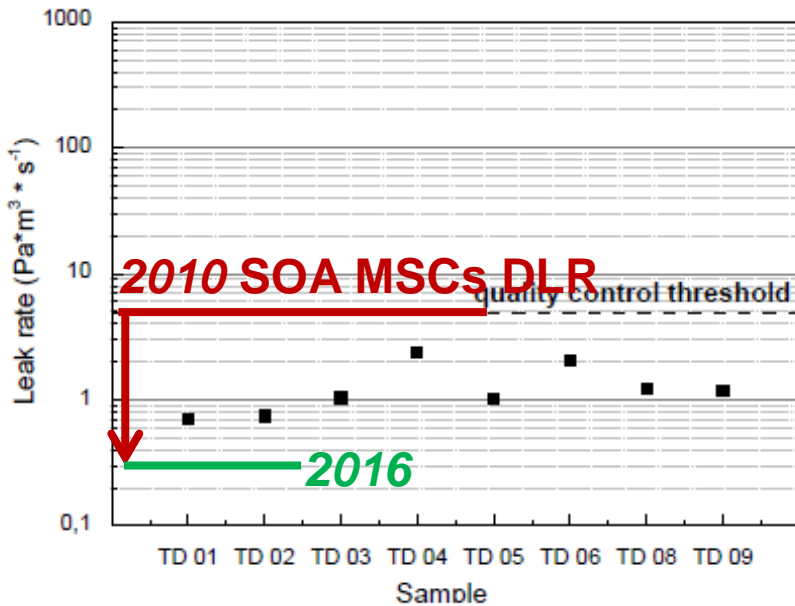
MSC with Metal Foam Substrate (Type 2)



Hermiticity of the electrolyte

Compared to DLR Plasma Sprayed MSCs:
Gas tightness improved by 1 order of magnitude
Material consumption reduced

- PVD: 1,2mg/cm² of YSZ + 1,5mg/cm² of CGO
- PS MSCs: 20-30 mg/cm² of YSZ

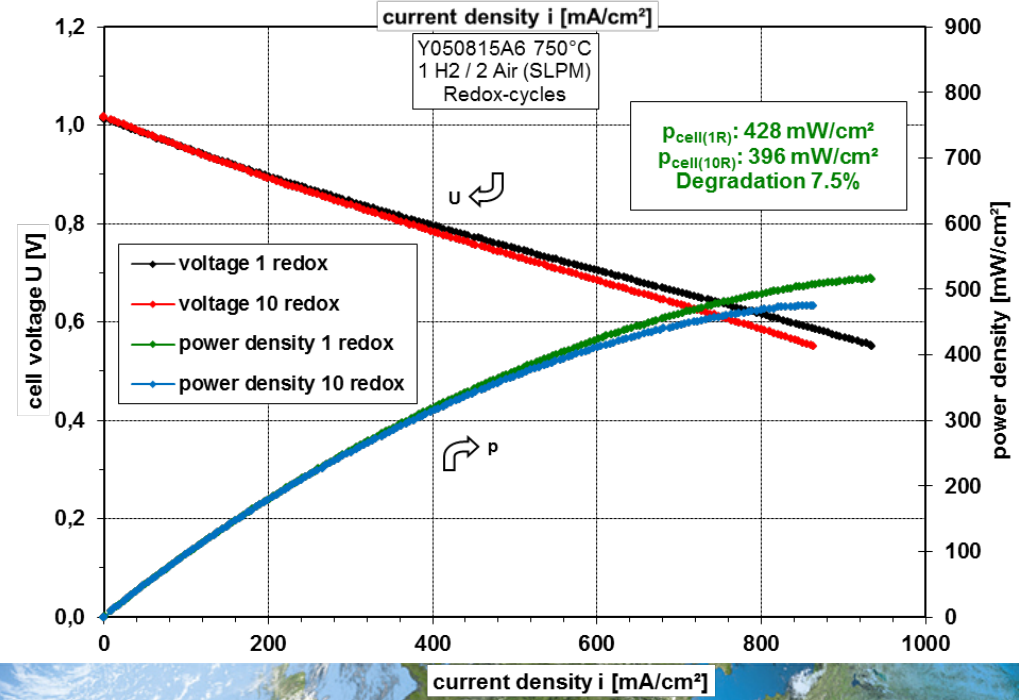
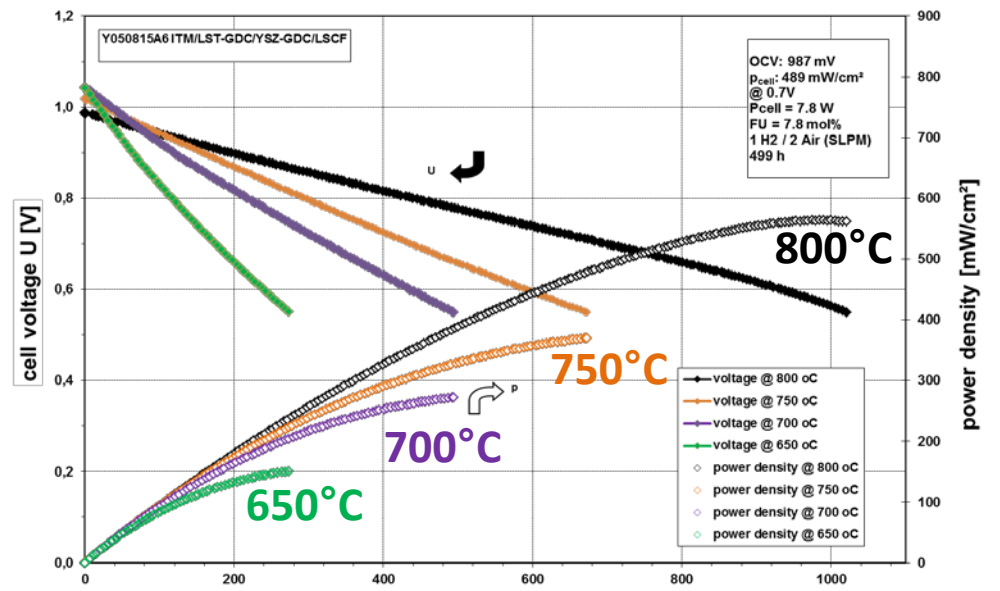


Performance Cell type 1 ITM (w 5-10wt%Ni) – 16cm²

@ 750°C 1slpm H₂ / 2slpm air

OCV: 1,03V
Power density at 0,7 V ca. 430 mW/cm²
(520 mW/cm² @ 800°C)

No OCV drop after 10 redox cycles



Performance Cell (Type 2)

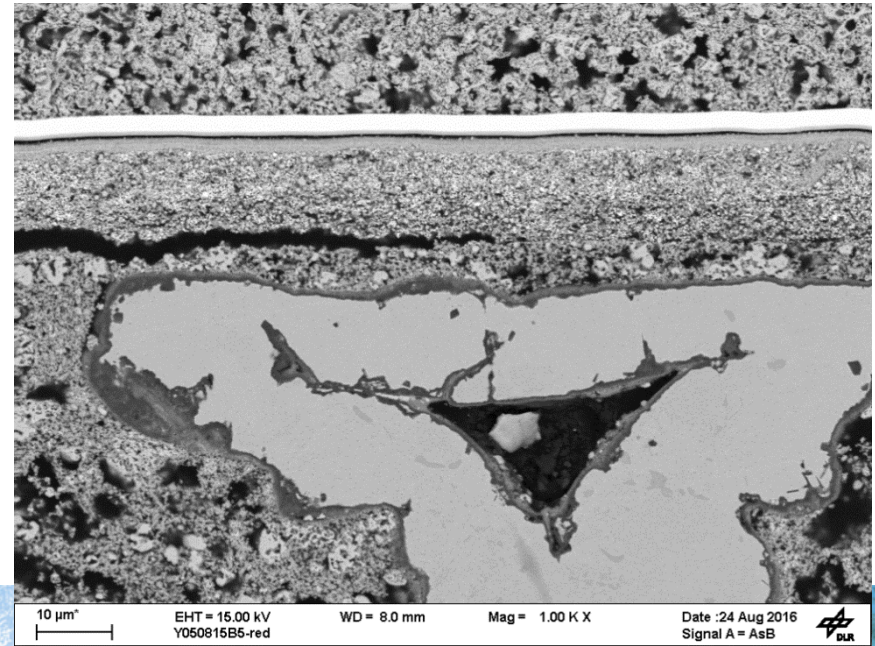
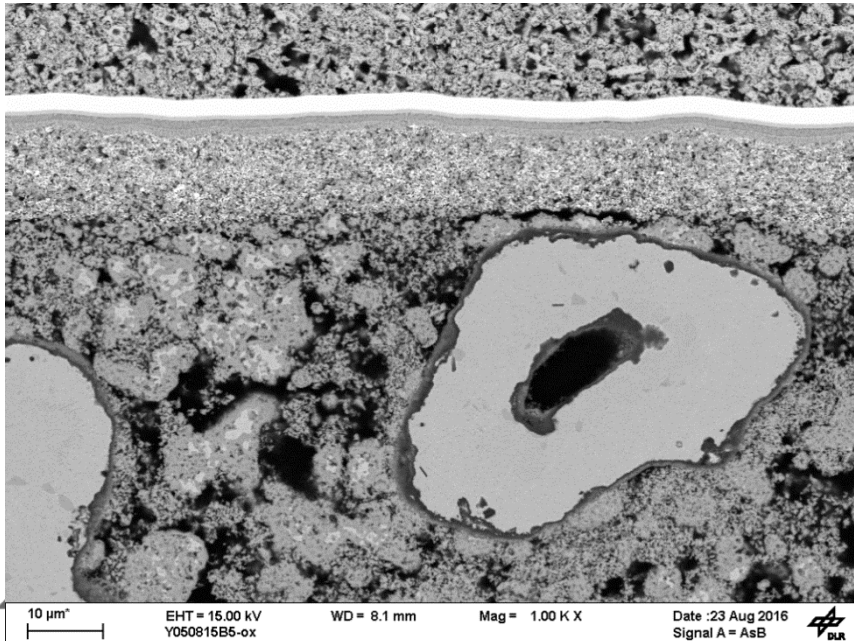
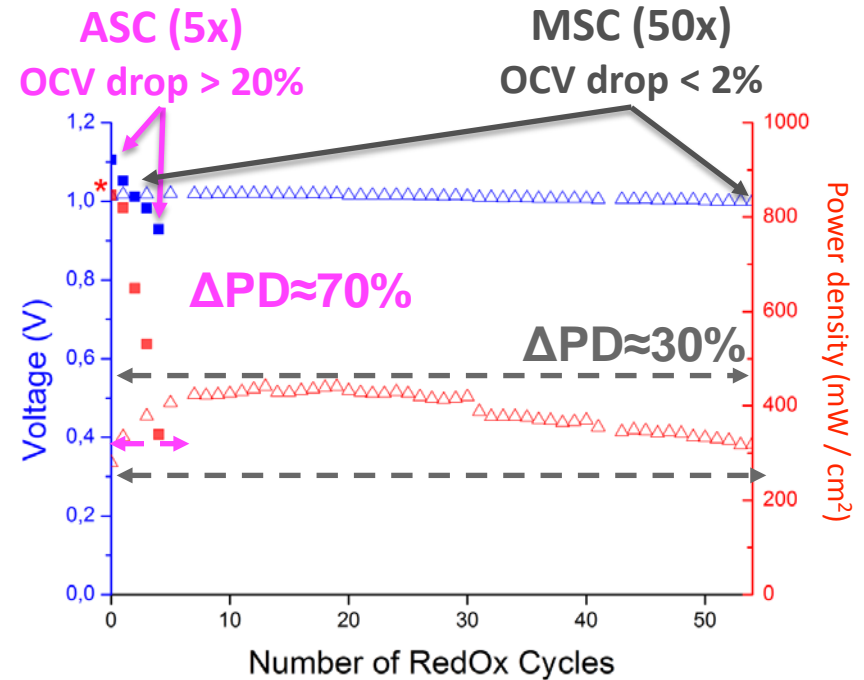
NiCrAl + LST + NiO – (w 5-10wt%Ni)

@ 750°C 1slpm H₂ + 3% water / 1 slpm air
OCV: 1,03V

Power density at 0,7 V ca. 430 mW/cm²

OCV drop of less than 2% (50cycles)

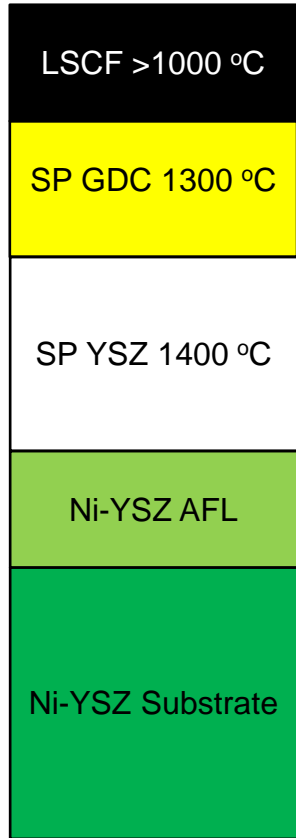
- good tolerance toward redox cycles
- Performance enhancement with addition of catalytic nickel



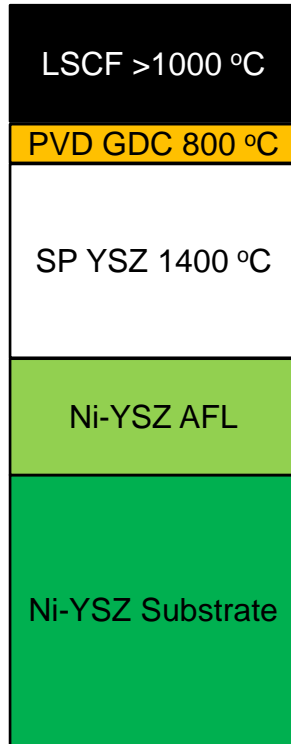
MSC with thin-film electrolyte vs. ASC with sintered YSZ

Operated@750°C & 0.7V

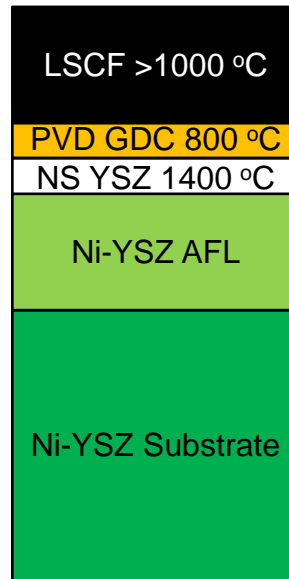
0,9 W/cm²



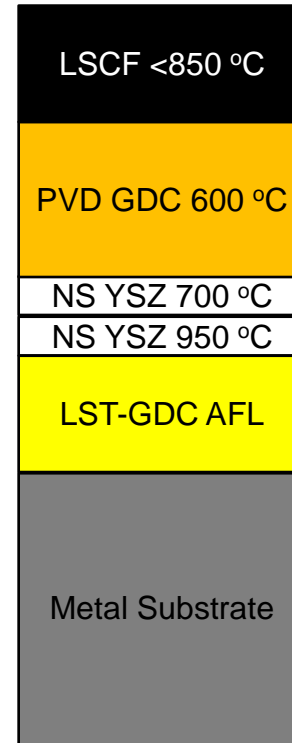
1,2 W/cm²



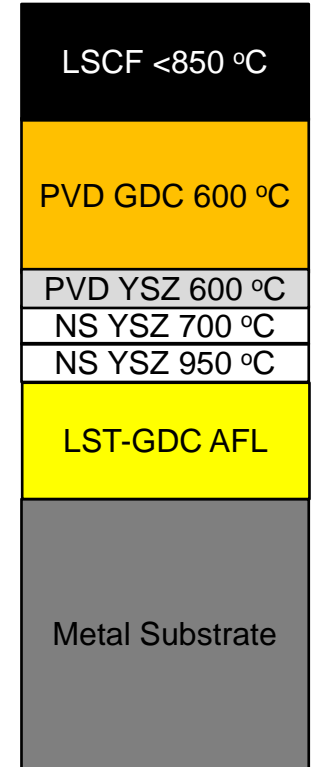
1,6 W/cm²



0,32 W/cm²



0,43 W/cm²

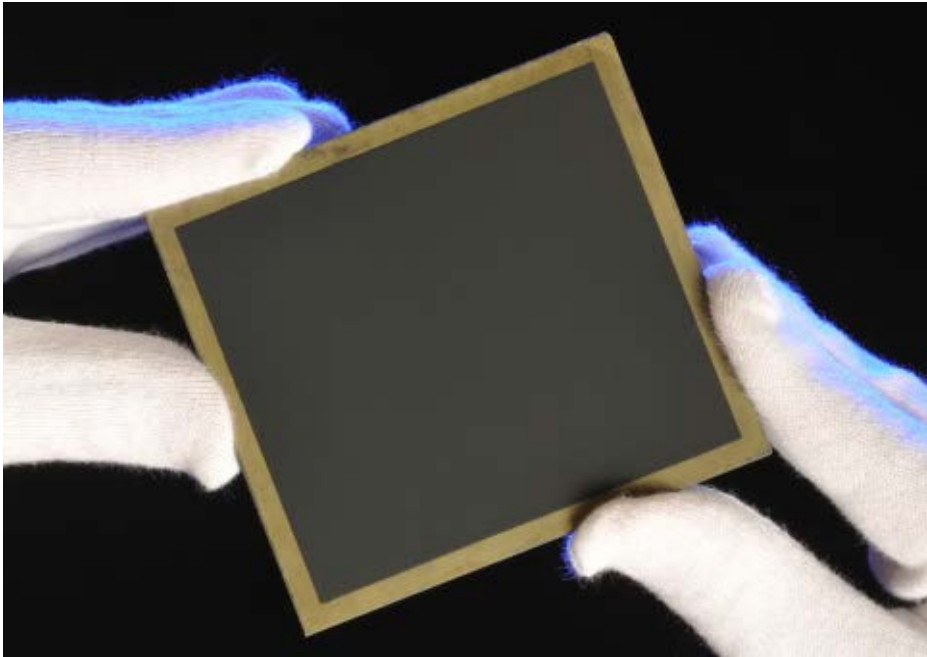


Poor redox stability

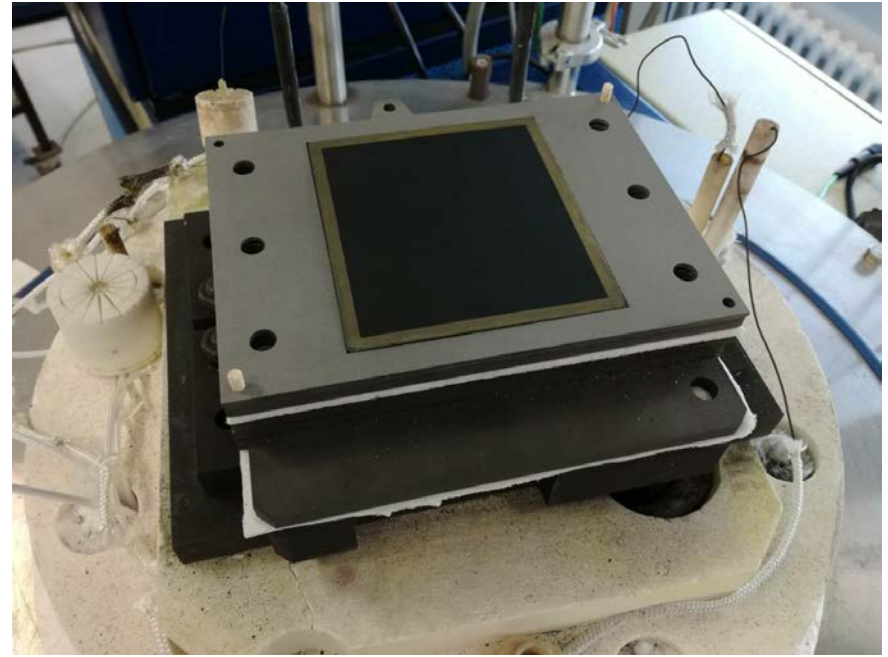
Good redox stability



Up-scaled cells for stacks



- Size up to 90 mm x 100 mm
- Laser cut substrate
- Infiltrated Ni catalyst



- Leak rate moderate or poor
- Stackable cells 8 pcs
- Needs for LT sealing solution



Conclusion & Perspectives

- *MSCs cell with various substrates (Stainless Steel & NiCrAl) delivered fair Power density despite low nickel content (< 10wt%)*
- *OCV drop of less than 2% for 50 forced redox cycles (30min in Oxygen) at 750°C*
- *Thin film electrolyte technology developed and demonstrated*
- *Estimated cost reduction: 20 Euro for a 10cm²x10 cm² cell (10 MW per year)*
- **LST based anode materials can operate without pre-reduction at high temperature (>1000°C)**
- *Without addition of catalysts in LST based anode is performance limited*
- *Degradation issue at the cathode*
- *Operando migration of cobalt from the cathode size in the electrolyte layer*
- *Improvement of the cathode performance*
- *Development of appropriate low T sealing (750°C) solutions*
- *Implementation in Single Repeat Unit and Test for completing Assessment*
- *Assessment against Sulfur Poisoning*



Acknowledgement

This work was funded by the European Union's Seventh Framework Programme (FP7/2007-2013) for the Fuel Cells and Hydrogen Joint Technology Initiative under grant agreement n°303429.



Thanks for your attention!

