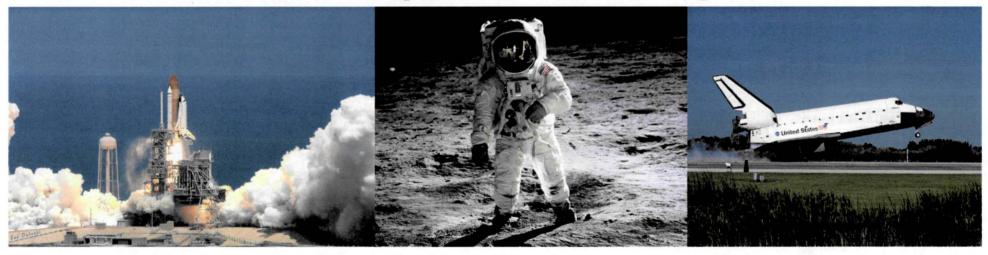


John F. Kennedy Space Center

# Advanced Active Materials for the Exploration of Space

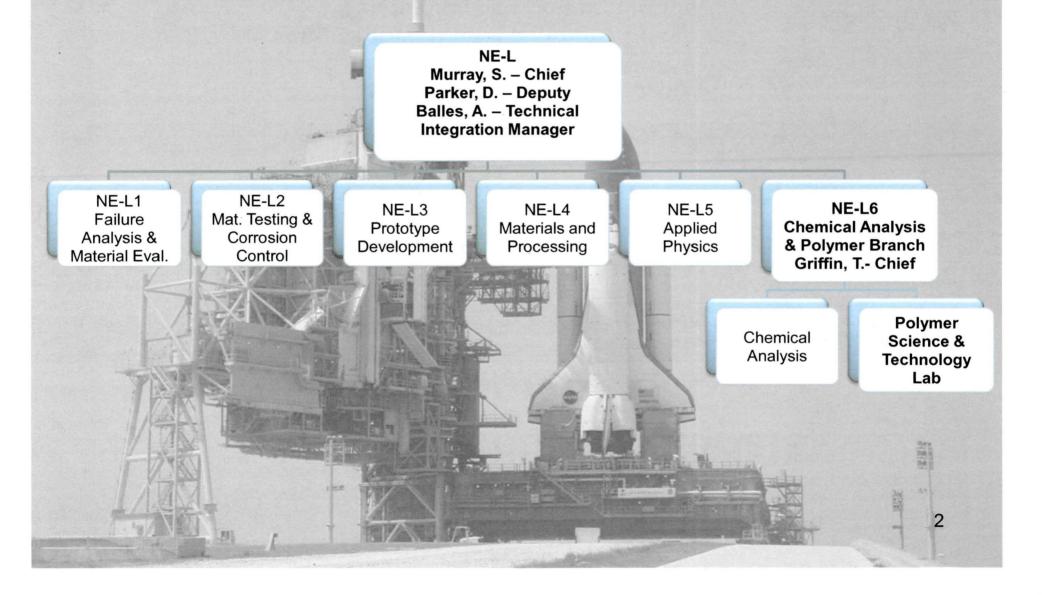


Materials Science Division Engineering and Technology Directorate Kennedy Space Center, Florida

> Luke Roberson, Ph.D. (<u>Luke.B.Roberson@nasa.gov</u>) 4/10/2012



### Materials Science Division Organizational Chart





### **Lab Overview**

#### Mission

To develop and apply new technologies in polymer and material chemistry that benefit NASA's programs and mission

#### Team

5 NASA scientists and 4 contractors

#### **Areas of Expertise**

Polymer Nanocomposites Next Generation Wire Materials Carbon Nanotube and Nanofiber Materials Conductive Polymers Polymer Processing Fire and Polymers Foam and Insulation Materials

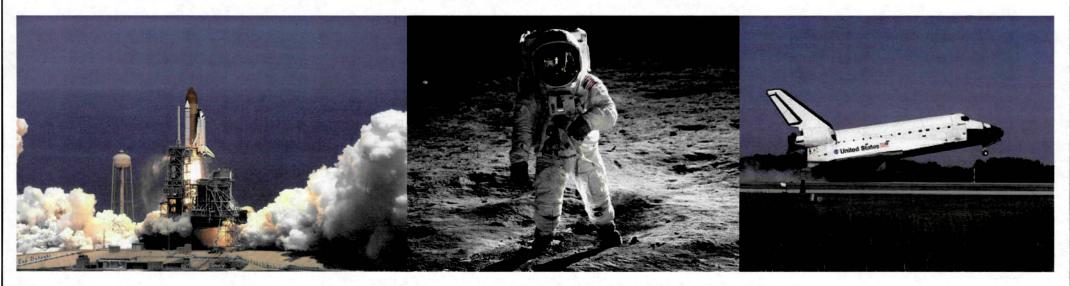
#### **Numerous Collaborative Efforts**

NASA Centers (JSC, LaRC, MSFC, GSFC, GRC) KSC Directorates (Shuttle, Ares, Orion, Ground support operations) Academia (Alberta, FIT, GT, Harding, Illinois-Urbana Champagne, UCF, UF, USF) Industry Space Act Agreements (Thermax, DeWAL, Sharklet, Crosslink, Sabic, Amalgam) Industry Contracts (ARCnano, Epner, Conductive Composites) 3



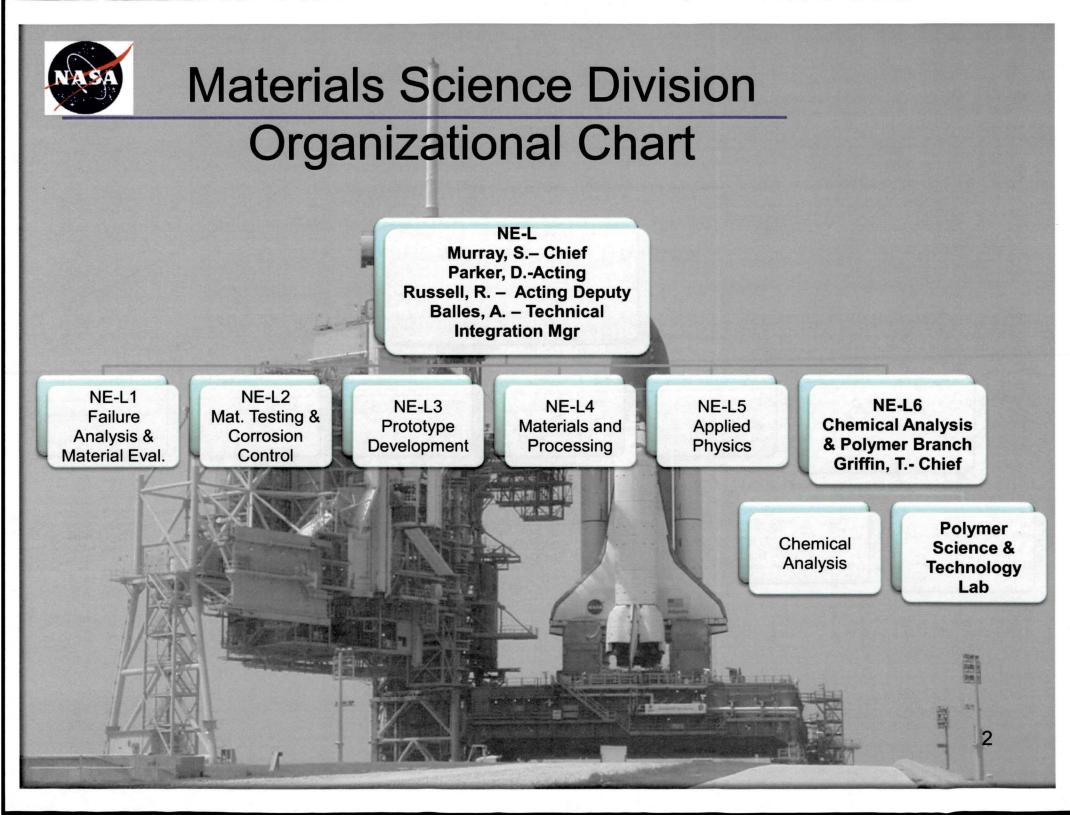
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# **Composite Materials for Space Exploration**



Materials Science Division Engineering and Technology Directorate Kennedy Space Center, Florida

> Luke Roberson, Ph.D. (<u>Luke.B.Roberson@nasa.gov</u>) 4/10/2012





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### Lab Overview

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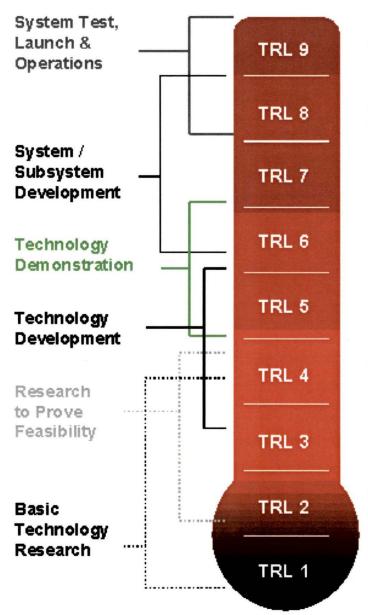
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# **Technology Readiness Levels**

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#### TECHNOLOGY READINESS LEVELS (TRL's)



Actual system proven through successful mission operations

Actual system completed and qualified through test and demonstration

System prototype demonstration in a relevant environment

System/subsystem model or prototype demonstration in a relevant environment

Component and/or breadboard validation in relevant environment

Component and/or breadboard validation in laboratory environment

Analytical and experimental critical function and/or characteristic proof-ofconcept

Technology concept and/or application formulated

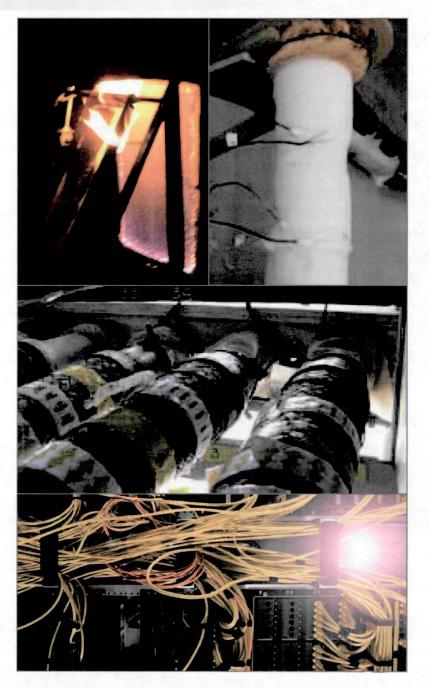
Basic principles observed and reported



### **Composites/Materials Development at KSC**

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- Smart Materials and Detection Systems
- Aerogel composites
- Aerogel for environmental remediation
- Chemochromic hazardous gas detectors
- Antimicrobial polymers
- CNTs and conductive polymer technologies





# Why Wiring?

### Aged Wire

- Cracks and frays over time
- Hard to detect damage
- Extensive maintenance related damage during ground processing work

### Space Shuttle Orbiter

- 183 miles of wiring buried deep within structure of vehicle
- Difficult to manually inspect





# **Next Generation Wiring Materials**

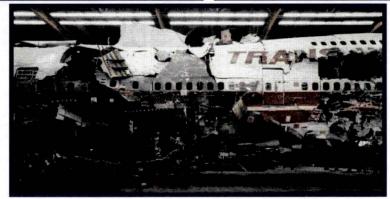
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#### **Wire System Failures**

- STS-93 (July 1999)
  - Short circuit in 14 AWG polyimide, Kapton<sup>®</sup> type insulated wire
- TWA 800 (July 1996)
  - -Frayed Kapton® wire in center tank area
- Swiss Air 111 (September 1998)
  - Damaged wire in plane's entertainment system







#### **Wiring Technology Solutions**

- Manual Repair Technologies for polyimide and fluorinated wires
- In-Situ Damage Detection Systems for Vehicle Health Monitoring
- Self-healing or self-repair insulation

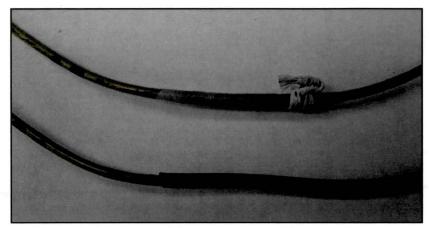


# **Wire System Materials**

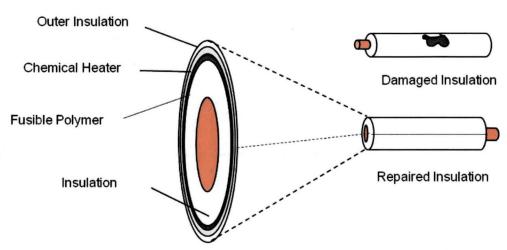
#### **Insulation and Repair Materials**

**Present Wiring Repairs** 

Manual Repair Concept

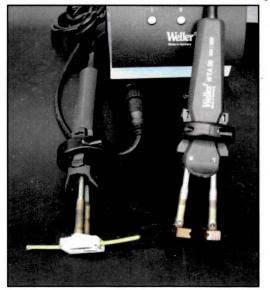


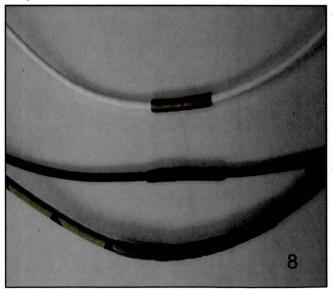
Casting of Wire Repair Films

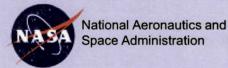




Laboratory Repair Process



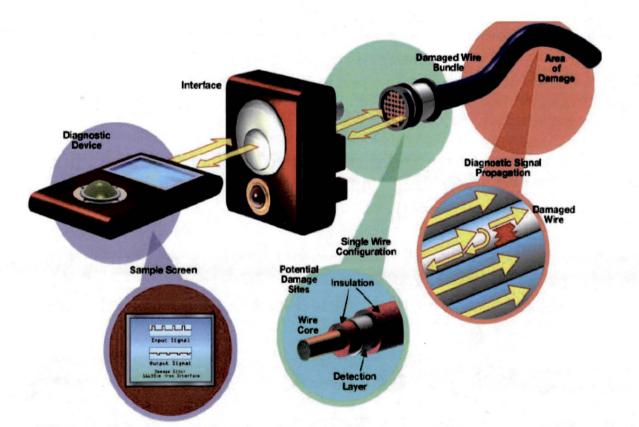


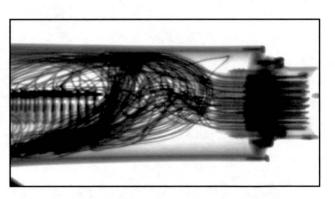


### **Wire Detection Systems & Integration**

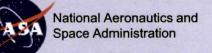
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- In-situ wire damage detection system
  - Capable of wire damage detection "on-the-fly"
- Smart Connectors
  - Small, lightweight, ultra reliable
- Integrated vehicle health monitoring (IVHM)
  - System-of-systems level, providing high level of reliability



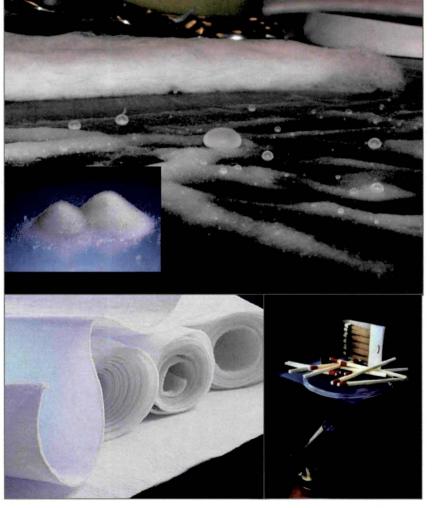


X-ray image of miniaturized TDR connector



# **Aerogel Technology**

- John F. Kennedy Space Center
- Aerogel materials are generally silica based, light weight materials, fully breathable, and treated to be super-hydrophobic.
- Aerogel granules are free flowing, fills small cavities, does not compact, no preconditioning required, and can be molded or formed using binders.
- Aerogel granules (Nanogel<sup>®</sup>) by Cabot Corp.:
  - 90% porous with a mean pore diameter of 20 nm.
  - Bead bulk density ≈ 80 kg/m<sup>3</sup> (5 lbs/ft<sup>3</sup>).
  - Individual beads are fragile; but have high elastic compression of over 50% with no damage.
  - k-value ≈ 18 mW/m-K @ 25 C and 760 torr.
- Aerogel Spaceloft<sup>®</sup> blanket manufactured by Aspen Aerogels:
  - Bulk density 6 to 8 lbs/ft<sup>3</sup>.
  - k-value ≈ 12 mW/m-K @ 38 C and 760 torr.
- Aerogel Pyrogel<sup>®</sup> blanket manufactured by Aspen Aerogels:
  - Flexible aerogel composite blanket designed for hightemperature applications (up to 650°C/1200°F).





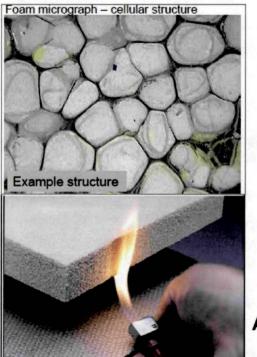
# **Aerogel Composites**

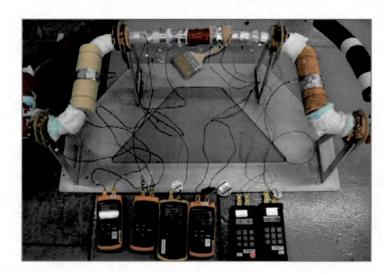
John F. Kennedy Space Center

AeroFoam<sup>™</sup> = polyimide foam + aerogel Enhanced thermal and vibration damping performance. Structural integrity to the aerogel and cryogen storage capabilities.

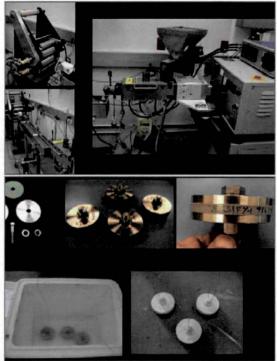
#### AeroPlastic<sup>™</sup> = thermoplastic + aerogel

Extruded process, composite reducing heat transfer by 40-60%. Cryogen storage and transfer applications such as piping and seal.





AeroPlastic demo testing on cryo-piping system





Fiber/Textile + aerogel structural composites



### **Aerogel for Oil Remediation**

John F. Kennedy Space Center

Lightest solid known (80 kg/m<sup>3</sup>) – floats on water High oil absorbency – 250 gallons/m<sup>3</sup> Super-hydrophobic material (repels water) Environmentally friendly – inert amorphous silica Stable – long consistent service life, no UV degradation Commercially manufactured in bulk quantities Aerogel incorporated into mesh bag, blanket, or filled boon for



### **Aerogels for Oil Remediation**

John F. Kennedy Space Center





# **KSC's Solution**

- Aerogel booms are 20% more effective than commercial PP/PE booms
- Reusable booms Oil recovered through distillation
- \$2800 per m<sup>3</sup> = 250 gallons oil
- Increase effectiveness through catalyst or bacterial infusion
- Cabot Nanogel and EnviroUSA: Commercial small business collaborations through existing SAA with NASA KSC

Domestic inventory	Europe inventory	Sustainable capacity per month
100 m <sup>3</sup>	2000 m <sup>3</sup>	600 m <sup>3</sup>
25,000 gallons equivalency	500,000 gallons equivalency	150,000 gallons equivalency



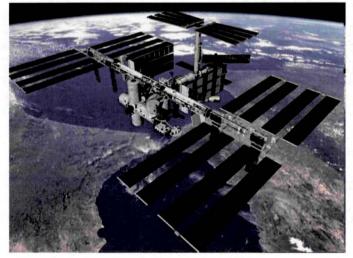
### **Antimicrobial Countermeasures**

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#### **Shuttle Potable Water**

Water generated on-orbit by fuel cells and stored in four 170-lb Inconel bellows tanks lodine (3-4 mg/L)





#### **ISS Potable Water**

Ground-supplied potable water (Shuttle, Progress, ATV, HTV, or commercial cargo) and reuse water recovered from humidity condensate and/or urine (SRV-K and WRS) Iodine, Silver Nitrate, Silver Fluoride

#### **Orion Potable Water**

Ground-supplied potable water stored in Five Inconel 718 Tanks (14.3 gal) Miles of Titanium water lines Silver Fluoride (0.4 mg/L)



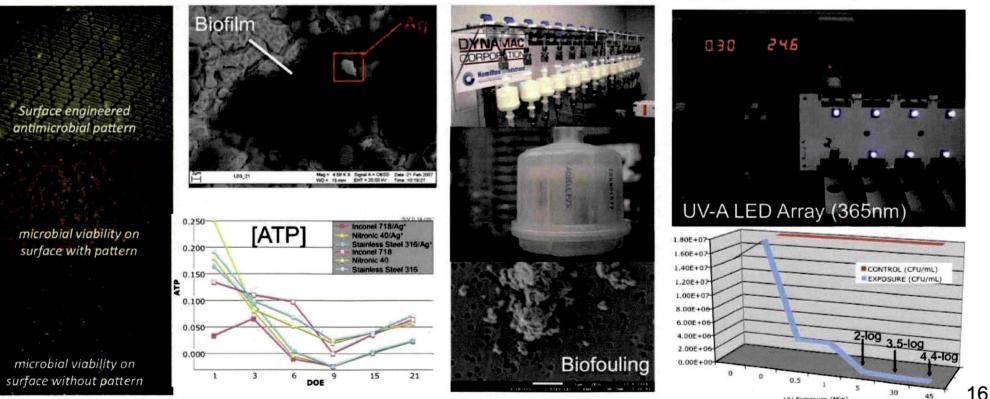


### **Antimicrobial Countermeasures**

John F. Kennedy Space Center

Multiple technologies are required for persistent microbial control in potable water systems

Antimicrobial materials Biocide delivery systems (ionic silver) Point-of-use sterilizing-grade filtration Solid state lighting systems (UV-A and UV-C LEDs)



UV Exposure (Min)

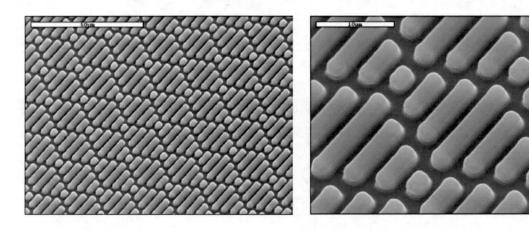


John F. Kennedy Space Center

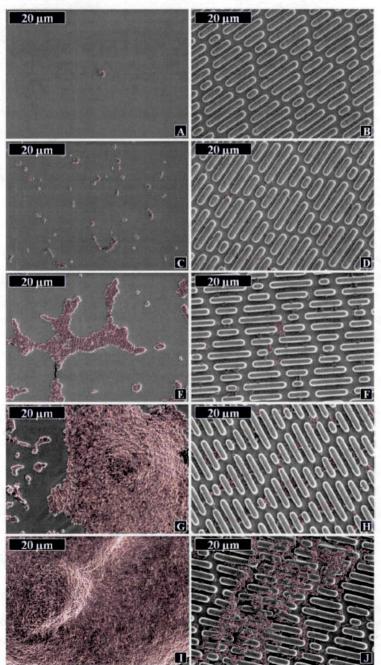
# **Antimicrobial Polymers**

In collaboration with Sharklet Technologies and UF

#### Surface Morphology and Surface Chemistry



- Efficacy studies after 21 days decreases biofilm formation
- Easy to imprint during manufacture of polymer articles through a coining process
- Can be used in conjunction with antimicrobial polymers



17

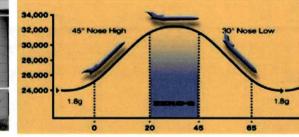


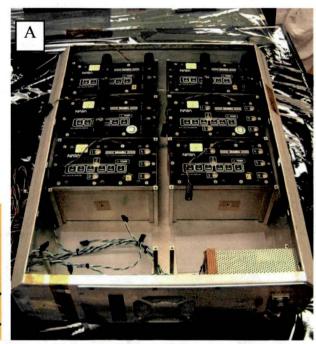
### **Antimicrobial Materials**

**Microgravity Flight Experiments** 

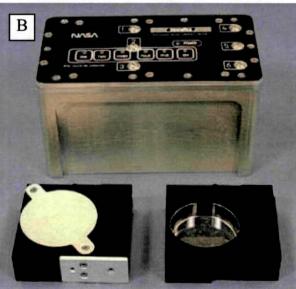
Measure ability of Sharklet<sup>®</sup> patterned coupons in combination with chemical surface treatments to inhibit biofilm formation by bacteria in reduced gravity













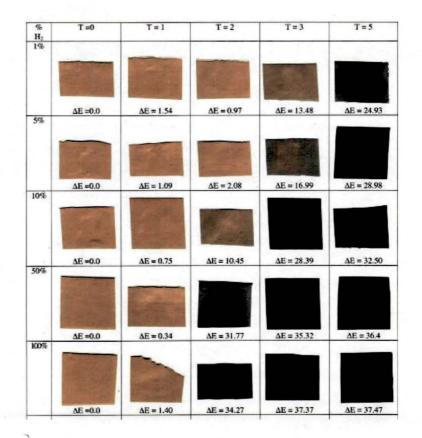
# **Chemochromic Hydrogen Sensors**

John F. Kennedy Space Center

In collaboration with FSEC/UCF

#### **Irreversible Sensor**

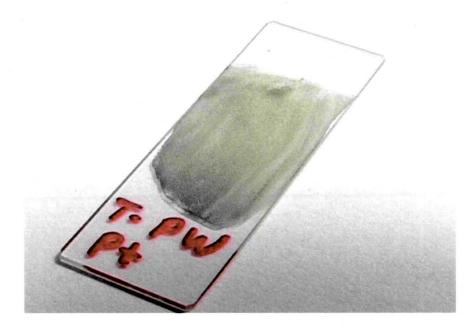
- A patent-pending irreversible color changing H<sub>2</sub> gas sensor was developed at KSC in partnership with UCF and ASRC.
- Changes color from a light tan to black in the presence of  $H_2$ .
- Can be manufactured into any polymer part, tape, fiber, or fabric material for unlimited potential uses.
  - Paint, Gloves, Coveralls, PPE
- Operates under ambient and cryogenic temperatures.

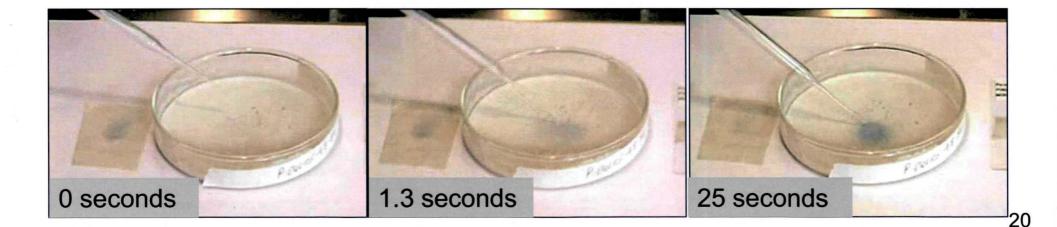




# Reversible Hydrogen Sensor

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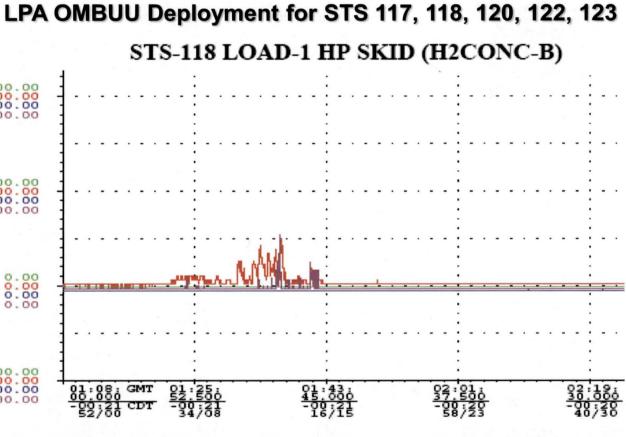






John F. Kennedy Space Center

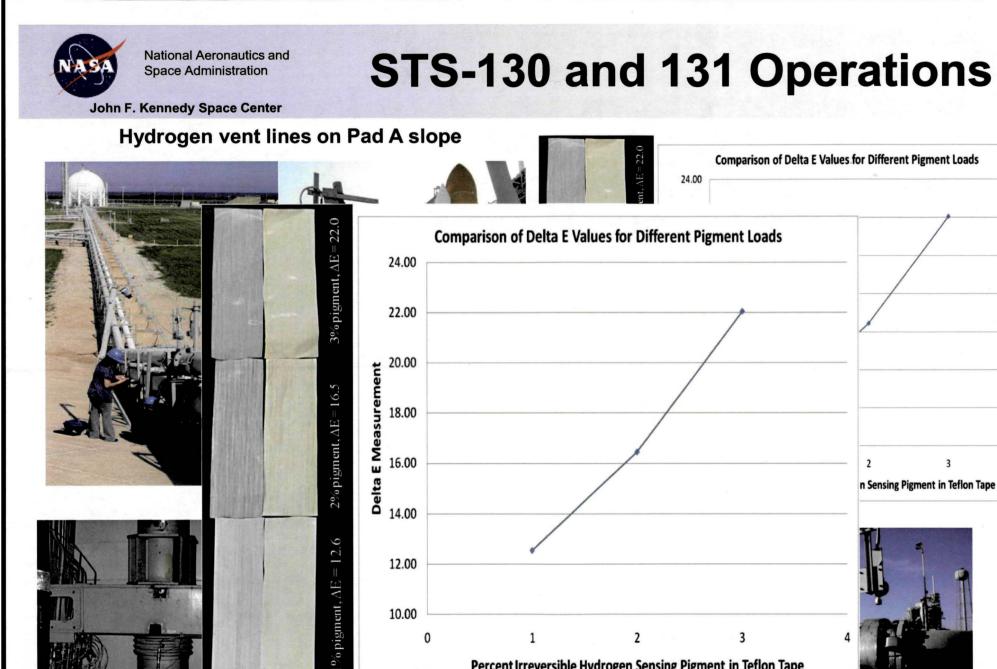
# **STS-129** Transfer Line 0000.00 20000.00 10000.00 10000.00 10000.00 10000.00 0.00 -10000.00 -10000.00 -10000.00 -10000.00











0

1

2

Percent Irreversible Hydrogen Sensing Pigment in Teflon Tape

3

3

STS-130 H2 Pressure Flange A3362

TSM for STS-131

22



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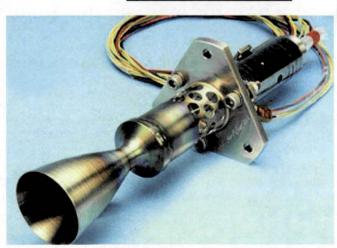
# **Hypergolic Fuels**

### **Direct Applications**

- Boiler Feed Water Treatment
- Monopropellant
- Bipropellant
- Fuel Cells
- Polymers
- Metallurgical

### **Derivative Applications**

- Solid Propellant
- Gun Propellant
- Explosives
- Pesticides
- Pharmaceutical









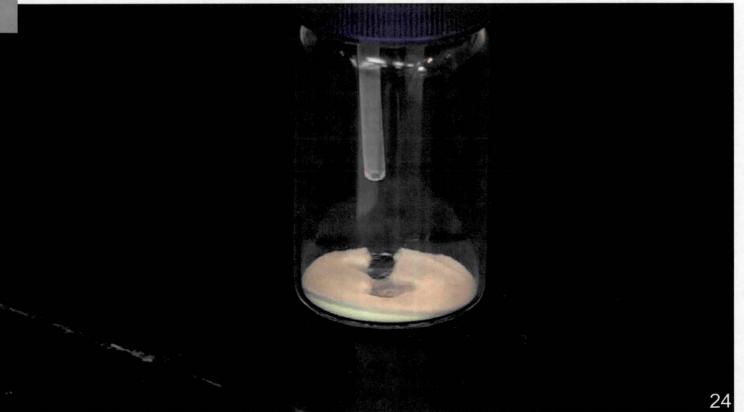
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### HyperPigment

Pigment shown is 1% by weight KAuCl<sub>4</sub> on silica

Concentrations were tested up to 5% to increase color response

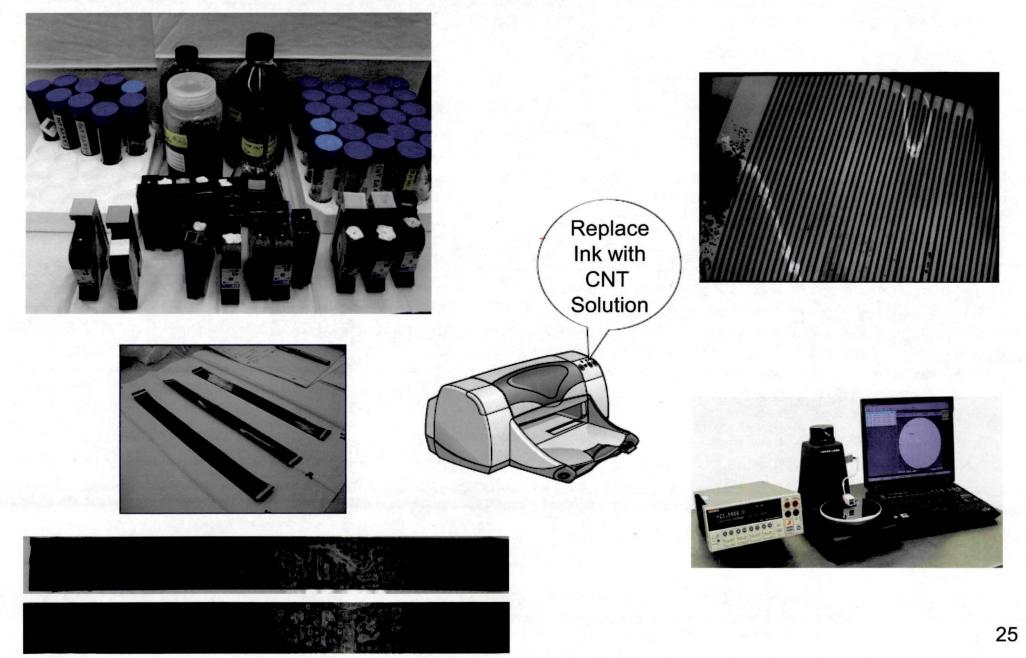
Pigment can be incorporated into most polymer materials - SCAPE suits





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### **Conductive Inks Formulations for Multiple Applications**

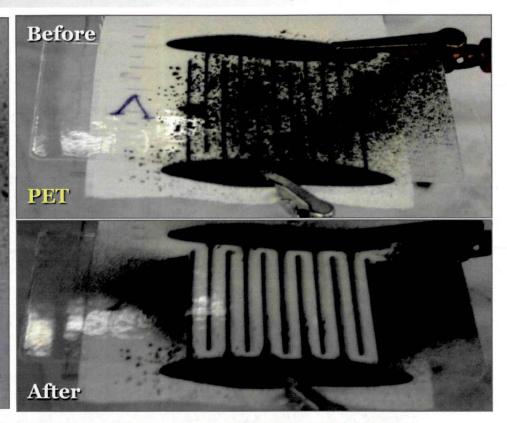


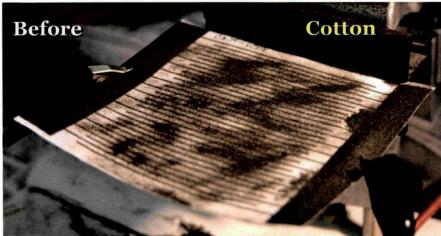


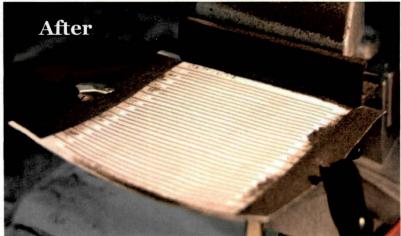
### **CNT Ink Dust Screens**

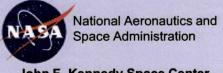
In collaboration with Electrostatics Laboratory

#### KSC Electrostatics and Surface Physics Laboratory









# **CNT Ink Printed Circuitry**

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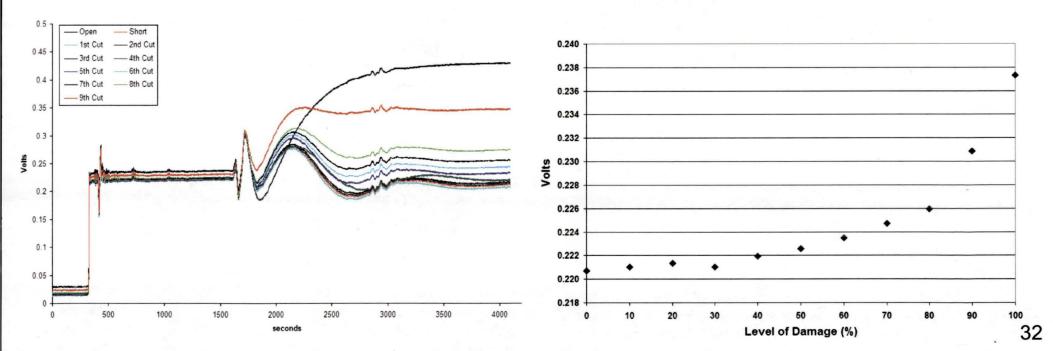
In collaboration with Crosslink

## Screen printed polymer-composite material

Line thickness and width increases conductivity

50 Ohm resistance able to measure damage to circuits

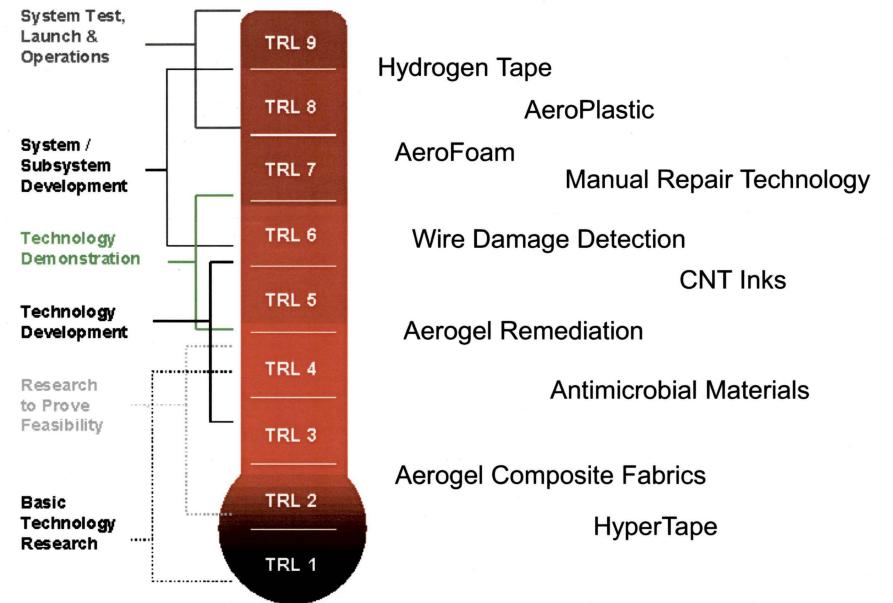


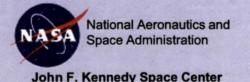




# **Technology Readiness Levels**

John F. Kennedy Space Center





### Acknowledgements

#### **Chemical Analysis and Polymer Lab**

Dr. Tim Griffin Anne Caraccio Dr. Kathy Loftin Jeremy O'Neal (2010 intern) Tyson Bevirt (2010 intern) Megan Morford (2010 intern)

#### **NASA Materials Science Division**

Dr. LaNetra Tate\* Gayle Krishingha Dean Lewis Rupert Lee

#### ASRC Applied Chemistry Lab

Dr. Pedro Medelius Dr. Tracy Gibson Dr. Mary Whitten Dr. Scott Jolley Dr. Robert Devor Dr. Steve Trigwell Sarah Snyder Lilly Fitzpatrick Rubie Vinje

#### NASA Technology Integration

Karen Thompson Nancy Zeitlin Orlando Melendez

#### **Environmental Life Support**

Dr. Ray Wheeler Dr. Mike Roberts Michelle Birmele David Smith LaShelle McCoy Dr. John Sager<sup>†</sup>

#### Surface Systems Office

Rob Mueller Dr. Carlos Calle David Smith\* Eddie Santiago-Maldonado Dr. Mike Hogue Dr. Charlie Buhler<sup>†</sup>

#### NASA/ASRC Fluids Division

Robert Johnson James Fesmire Jared Sass Angela Krenn Wes Johnson Brekke Coffman Stephen Huff\* Craig Fortier Dr. Barry Meneghelli Judy McFall

#### LaRC

Dr. Erik Weiser Bert Cano

#### **Applied Physics Lab**

Dr. Bob Youngquist Dr. Janine Captain Dr. Chris Immer<sup>+</sup>

#### **Technology Transfer Office**

David Makufka Alexis Hongaman Carol Dunn Jim Nichols Jeff Kohler Lewis Parrish Pasquale Ferrari

#### **Patent Counsel and Legal**

Randy Heald Ginger Arrington Amber Hufft Penny Chambers

#### **Shuttle Operations**

Trent Smith\* Norm Peters

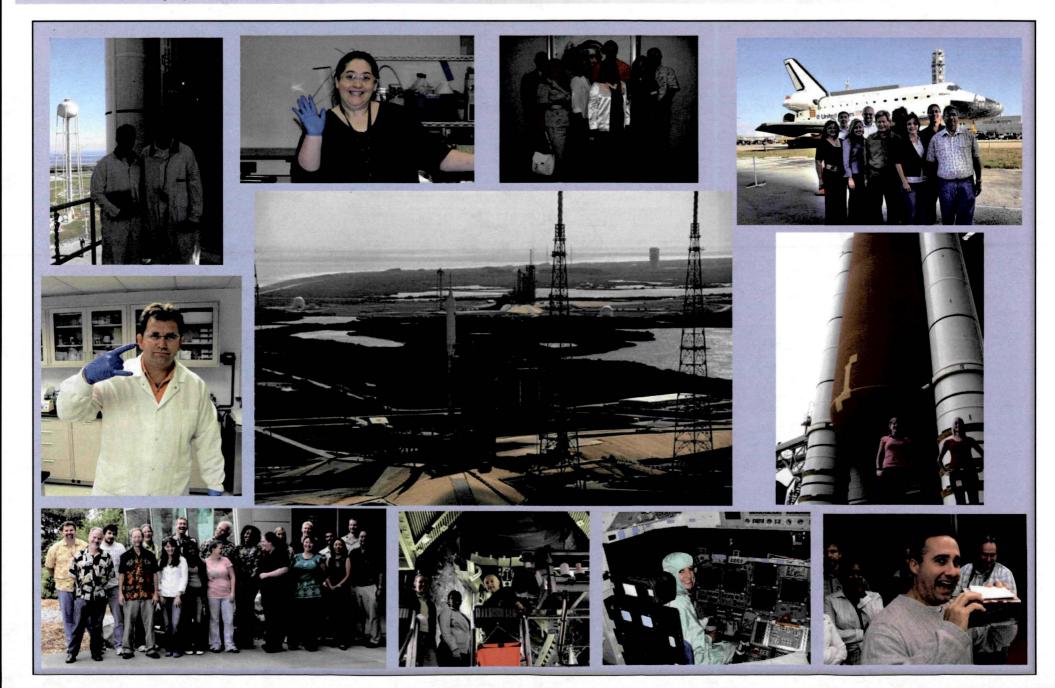
\* Former group members

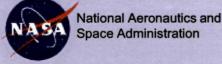
<sup>†</sup>No longer at KSC



### **Questions?**

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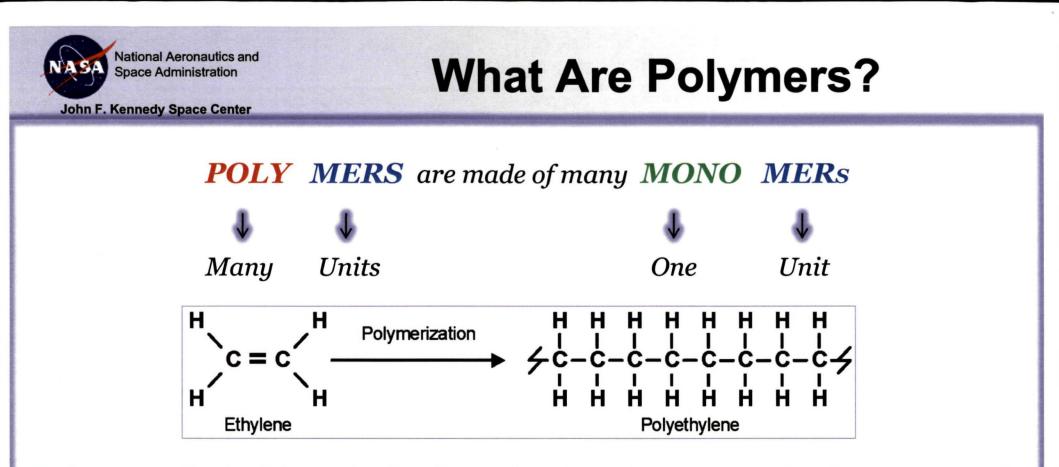
### Testing and Processing Equipment

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- Fire Testing
  - Cone Calorimeter
  - Oxygen Index\*\*
  - UL94 fire test
  - NASA Std 6001 fire test
  - Radiant Panel\*
  - NBS Smoke Chamber\*
  - Two foot tunnel\*
  - Glow wire ignition\*
- Cryogenic Materials Testing
  - Cryogenic moisture uptake (CMU)\*\*
  - Brittleness/Impact test \*\*
  - Liquid helium cold finger test\*\*
  - Single Pin-Socket Krytox Contamination Electrical Characterization under Cryogenic Conditions\*\*
- Specialty Test Equipment

\*in collaboration with Cryogenics Test Laboratory \*\*in collaboration with Florida Tech

- Cellular Solid Analysis
  - Pycnometer (closed/open cell)\*\*
  - Surface area measurement\*\*
- Thermal Analysis
  - Thermogravimetric analysis (TGA)
  - Differential Scanning Calorimetry (DSC)
  - Dynamic Mechanical Analysis (DMA)
- Physical Testing
  - Tensile Test
  - Compressive Test
  - Pull/Peel Test
- Electrical Testing
  - 4-point probe
  - Surface /Volume resistance
- •Polymer Processing capabilities
  - Extrusion
  - Injection molder
  - Fiber spinning equipment
  - Melt, ball, and high intensity mixers<sup>31</sup>



<u>Polymers</u>: Derived from the Greek words *poly* and *mers* meaning "*many parts*".
Large molecules composed of repeated chemical units

- •When you think of **POLYMER** most automatically think **> PLASTIC**. However, polymers are a wide range of *natural* and *synthetic* materials with a wide variety of properties.
- Molecular weight of the resulting synthesized polymer can range from the very lightest of molecules up to huge gels.