Abstract

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Thermal Reactions between Sulfur Dioxide and H₂O₂ and their Relevance to the Jovian Icy Satellites and Other Small Bodies

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Laboratory experiments have demonstrated that magnetospheric radiation in the Jovian system drives reaction chemistry in ices at temperatures relevant to Europa and other icy satellites. Here we present new results on thermally-induced reactions occurring between 50 and 130 K in solid $H_2O + H_2O_2 + SO_2$ samples. In our studies, we find that warming our three component mixtures induces a thermal reaction that produces SO_4^{2-} , and this reaction appears to consume equal amounts of H_2O_2 and SO_2 . We suspect that the results may explain some of the observations related to the presence and distribution of H_2O_2 across Europa's surface as well as the lack of H_2O_2 on Ganymede and Callisto. If other molecules prove to be reactive with H_2O_2 at these or at even lower temperatures, then it may also explain why H_2O_2 has been absent from surfaces of many of the small icy bodies that are known to be exposed to ionizing radiation.

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