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1	Limits to Mercury's magnesium exosphere from MESSENGER second flyby
2	observations
3 4	Menelaos Sarantos ^{a,b,*} , Rosemary M. Killen ^c , William E. McClintock ^d , E. Todd
5	Bradley ^e , Ronald J. Vervack, Jr. ^f , Mehdi Benna ^{b.g} , and James A. Slavin ^a
6	
7	^a Heliophysics Science Division, NASA Goddard Space Flight Center, Greenbelt, MD
8	20771, USA
9	⁶ Goddard Earth Sciences and Technology Center, University of Maryland, Baltimore
10	County, Baltimore, MD 21228, USA
11	° Planetary Magnetospheres Branch, NASA Goddard Space Flight Center, Greenbelt, MD
12	20771, USA
13	^d Laboratory for Atmospheric and Space Physics, University of Colorado, Boulder, CO
14	80303, USA
15	^e Department of Physics, University of Central Florida, Orlando, FL 32816, USA
16	^f The Johns Hopkins University Applied Physics Laboratory, Laurel, MD 20723, USA
17	^g Solar System Exploration Division, NASA Goddard Space Flight Center, Greenbelt,
18	MD 20771, USA
19	
20	*Corresponding author. Fax: +1 301 286 1648. Phone: +1 301 286 2945
21	E-mail address: menelaos.sarantos-1@nasa.gov

23 Abstract

24 The discovery measurements of Mercury's exospheric magnesium, obtained by the 25 MErcury Surface, Space ENvironment, GEochemistry, and Ranging (MESSENGER) 26 probe during its second Mercury flyby, are modeled to constrain the source and loss 27 processes for this neutral species. Fits to a Chamberlain exosphere reveal that at least two 28 source temperatures are required to reconcile the distribution of magnesium measured far 29 from and near the planet: a hot ejection process at the equivalent temperature of several 30 tens of thousands of degrees K, and a competing, cooler source at temperatures as low as 31 400 K. For the energetic component, our models indicate that the column abundance that 32 can be attributed to sputtering under constant southward interplanetary magnetic field 33 (IMF) conditions is at least a factor of five less than the rate dictated by the 34 measurements. Although highly uncertain, this result suggests that another energetic 35 process, such as the rapid dissociation of exospheric MgO, may be the main source of the 36 distant neutral component. If meteoroid and micrometeoroid impacts eject mainly 37 molecules, the total amount of magnesium at altitudes exceeding ~100 km is found to be 38 consistent with predictions by impact vaporization models for molecule lifetimes of no 39 more than two minutes. Though a sharp increase in emission observed near the dawn 40 terminator region can be reproduced if a single meteoroid enhanced the impact vapor at 41 equatorial dawn, it is much more likely that observations in this region, which probe 42 heights increasingly near the surface, indicate a reservoir of volatile Mg being acted upon 43 by lower-energy source processes.

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- 45 Keywords:
- 46 Mercury
- 47 Mercury atmosphere

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- 48 Mercury surface
- 49 Atmospheric structure
- 50 Mercury magnetosphere
- 51 MESSENGER

53 1. Introduction

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55 Mercury is enveloped by a rarefied collisionless exosphere whose source is largely the surface of the planet itself, which is continuously being bombarded by a flux of 56 57 micrometeoroids, solar-wind ions, and ultraviolet (UV) photons. Known exospheric constituents include H, He, Na, K, O, and Ca (e.g., Killen et al., 2007), and Mg, which 58 59 was recently discovered by the MErcury Surface, Space ENvironment, GEochemistry, 60 and Ranging (MESSENGER) spacecraft (McClintock et al., 2009). Because the exosphere reflects a combination of surface composition, the composition of vaporizing 61 micrometeoroids, and the escape rates of atoms from the various source processes, we 62 expect the detection of other atomic and molecular constituents of Mercury's exosphere 63 64 by MESSENGER (Solomon et al., 2001) and by future missions such as BepiColombo 65 (Milillo et al., 2010).

66 On its way to insertion into orbit around Mercury in 2011, MESSENGER flew by the planet three times, on January 14, 2008, October 6, 2008, and September 29, 2009. 67 68 During the first of these encounters (M1), the Ultraviolet and Visible Spectrometer 69 (UVVS) channel on the Mercury Atmospheric and Surface Composition Spectrometer 70 (MASCS) instrument on MESSENGER observed the planet's exosphere at the resonant 71 emission lines of Na (589.0 and 589.5 nm), Ca (422.7 nm), and H (121.6 nm) 72 (McClintock et al., 2008). During the second (M2) and third (M3) flybys, UVVS also 73 observed the Mg (285.2 nm) line (McClintock et al., 2009; Vervack et al., 2010). 74 MESSENGER probed the nightside (tail) and terminator exospheric regions with

unprecedented spatial coverage but did not observe the dayside exosphere (except for H).
These observations revealed substantial differences even for chemically related species.
Calcium appeared to peak near the equatorial dawn flank during all three flybys, sodium
was enhanced at high latitudes, and magnesium appeared to be uniformly distributed
during M2 but enhanced above the northern pole during M3.

80 The differences in the distributions of individual species point to differing ejection 81 mechanisms and relate, perhaps, to different regions of surface composition and/or 82 mechanisms of global circulation and surface replenishment. Sodium has been the most 83 extensively studied species because its favorable spectroscopic properties enable its longterm monitoring from Earth, yet the relative importance of several proposed source 84 85 processes is still under debate (e.g., Killen et al., 2001; Killen et al., 2004, Mura et al., 86 2009) and their interplay may exhibit an annual trend (Leblanc and Johnson, 2010). At 87 the time of the MESSENGER flybys, the main source of low-latitude sodium production 88 was found to be photon-stimulated desorption (PSD); the efficiency of this source was 89 enhanced at high latitudes by solar wind precipitation (Burger et al., 2010); finally, the 90 dayside content was consistent with a moderate degree of thermal accommodation upon 91 interaction with the surface (Mouawad et al., 2011). Although the calcium distribution 92 obtained by MESSENGER remains unexplained, its high temperature in the tail is 93 consistent with the possible photodissociation of an exospheric molecule such as CaO as 94 suggested by ground-based observations (Killen et al., 2005).

The measurements of Mg by MESSENGER were first analyzed by Killen et al. (2010). They showed that the observed magnesium tail was hotter than can be expected by direct impact vaporization, and they attributed these high temperatures (>10,000 K) to

98 vaporization of ~30% of the magnesium vapor in molecular form (MgO) followed by its 99 dissociation. They noted that the exospheric temperature was poorly constrained, even 1 00 more so near the planet, where no single temperature could be made to fit. In this paper 101 we elaborate upon the Killen et al. (2010) analysis using a large number of simulations 102 that seek to constrain many unknown physical constants for this species, such as the 103 lifetime of the proposed molecule, the inferred exospheric source rates, and the relation 104between the exosphere and the surface content. In Sections 2 and 3 we describe the 105 observational sequence, we analyze the measurements using a Chamberlain (1963) model, 106 and we provide evidence of multiple source components and rates. In Section 4 we detail 107 our exosphere model assumptions and place limits on the possible contributions from 108 several proposed source processes of exospheric magnesium: (1) solar-wind ion 109 sputtering in polar areas that are only partially shielded by the planetary magnetic field, (2) production of magnesium atoms due to micrometeoroid impact vaporization, and (3) 110 111 dissociation of a Mg-bearing molecule such as MgO with a finite lifetime. In Section 5 112 we demonstrate that low-altitude data obtained near the dawn terminator strongly suggest 113 the presence of either an additional local source or, more likely, a source that is colder 114 than impacts or sputtering. The paper concludes with a summary of the major findings in 115 Section 6.

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117	2. Observations of magnesium during the second MESSENGER flyby	 Sean Solomon 3/4/11/9/16 AM	
		Comment:	

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119 During the second MESSENGER flyby, the UVVS spectrometer scanned Mercury's

120 exosphere at the Mg 285.2-nm line with a $0.1^{\circ} \times 1^{\circ}$ field of view. The measurements,

detailed by McClintock et al. [2009], were organized spatially and temporally in thefollowing sequence:

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• As the spacecraft approached Mercury and at downstream distances between 14 125 $R_{\rm M}$ and 2 $R_{\rm M}$ (where $R_{\rm M}$ is Mercury's radius, or 2,440 km), both the equatorial and 126 the high-latitude neutral tail were observed.

While inside Mercury's shadow (and near closest approach), the spacecraft
 executed a ~180° roll, initially pointing toward dawn, then rotating through north
 toward the dusk direction, while moving toward the planet from about 3,000 km
 (1.2 R_M) to within 450 km (0.2 R_M) above the planetary surface.

As the spacecraft exited the shadow past closest approach and at distances ≤ 1,000
 km, the lines of sight pointed toward the planetary shadow and intersected the
 surface in the equatorial near-nightside and dawn terminator regions.

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In what follows we refer to the set of observations obtained first upon initial approach to Mercury as "tail measurements," to the set of observations obtained second during the spacecraft roll as "fantail measurements," and to the third and final set as "nearterminator measurements." Observations of magnesium on the dayside were not conducted during the flybys. The excitation mechanism is presumed to be resonant scattering of sunlight. Hence, the instrument sees that portion of the exosphere that is illuminated, i.e., outside Mercury's shadow.

Excepting the shadow, the tail and fantail lines of sight probed exospheric material from the spacecraft to infinity. During these two sequences the closest tangent point outside the shadow was ~1100 km above the surface. Because the minimum ejection speed for a magnesium atom to reach this altitude is 2.3 km/s, or ~50% of the escape velocity, these lines of sight survey energetic atoms. In marked contrast, the

147 measurements near the dawn terminator probed material much closer to the surface, with 148 lines intercepting the shadow at progressively diminishing heights, starting at 162 km 149 altitude and eventually reaching the surface. Obviously, a wider spectrum of ejection 150 speeds can populate these altitudes compared to altitudes scanned by the tail and fantail 151 observations. Another difference is that lines of sight near the terminator are only 200– 152 800 km long (i.e., from the spacecraft to the shadow or surface).

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154 3. Fits to a Chamberlain exosphere

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156 In order to remove geometrical effects, the exospheric measurements were first fit to a 157 Chamberlain (1963) model. The instrument measures intensity, 158 $4\pi I = gN_{LOS} = \int_{R_{SP}}^{\infty} g(\mathbf{r}) n(\mathbf{r}) ds$, (1)

where g is the emission probability (photons s^{-1} atom⁻¹) at the relevant Doppler-shifted 159 160 resonance line, N_{LOS} is the column abundance, r is the radius vector from the planet 161 center, s is the distance along the line of sight (LOS) from the spacecraft location R_{SP} to 162 infinity (or to the shadow in the case of the measurements near the terminator), and the 163 neutral density n is a function of altitude and source temperature. Two model free 164 parameters must be determined by least squares fitting of the model to the data weighted 165 by the measurement uncertainty: the temperature, T, and the surface density, n_0 , of the 166 source. We have run models having with exobase temperatures between 100 K and 167 50,000 K and have excluded from the model intensities, I, the portion lying inside the 168 planetary shadow.

Comparisons of modeled and observed magnesium in the tail regions are presented inFig. 1, followed, in Fig. 2, by fits of the fantail and near-terminator measurements. The

171	modeled column abundances were converted to line intensity using a g -factor for Mg at
172	285.2 nm of $g_{Mg} = 0.317$ photons s ⁻¹ atom ⁻¹ given Mercury's radial velocity of $v_r = -9.2$
173	km/s during the flyby (Killen et al., 2009, 2010). The tail data (Fig. 1) are ordered by the
174	distance -x downstream of Mercury where each line of sight intercepts the noon-midnight
175	meridian $(x-z)$ plane, and by the distance z of this intercept above the equatorial plane.
176	The fantail data (Fig. 2a) are shown versus the boresight angle with respect to dawn
177	(where dawn = 0° and north = 90°), and the near-terminator lines (Fug. 2b) are plotted
178	against the spacecraft altitude from the planetary shadow. The best-fit parameters, n_0 and
179	T, for each model, along with the reduced chi-squared error, χ^2/ν (where ν is the number
180	of measurements minus two) and the corresponding production rate of magnesium atoms,
181	$S = \frac{n_0}{2\sqrt{\pi m/2K_BT}}$, for an assumed Maxwellian velocity distribution (where <i>m</i> is the
182	mass of a magnesium atom and $K_{\rm B}$ is the Boltzmann constant), were computed using all
182 183	mass of a magnesium atom and $K_{\rm B}$ is the Boltzmann constant), were computed using all ~200 lines of sight.
183	~200 lines of sight.
183 184	~200 lines of sight. [Insert Fig.1; Fig.2]
183 184 185	~200 lines of sight. [Insert Fig.1; Fig.2] Inspection of Figs. 1 and 2 leads to the following conclusions:
183 184 185 186	 ~200 lines of sight. [Insert Fig.1; Fig.2] Inspection of Figs. 1 and 2 leads to the following conclusions: 1) A single "warm" (T ≤ 10,000 K) source underestimates the tail measurements
183 184 185 186 187	 ~200 lines of sight. [Insert Fig.1; Fig.2] Inspection of Figs. 1 and 2 leads to the following conclusions: A single "warm" (T ≤ 10,000 K) source underestimates the tail measurements and overestimates data near the planet.
183 184 185 186 187 188	 ~200 lines of sight. [Insert Fig.1; Fig.2] Inspection of Figs. 1 and 2 leads to the following conclusions: A single "warm" (T ≤ 10,000 K) source underestimates the tail measurements and overestimates data near the planet. A single hot source can describe measurements obtained far from the planet
183 184 185 186 187 188 189	 ~200 lines of sight. [Insert Fig.1; Fig.2] Inspection of Figs. 1 and 2 leads to the following conclusions: A single "warm" (T ≤ 10,000 K) source underestimates the tail measurements and overestimates data near the planet. A single hot source can describe measurements obtained far from the planet with production rate ~(8-15) × 10⁵ atoms cm⁻² s⁻¹. Models having

192	3) No single temperature can describe the data near and far from the planet in an
193	internally consistent manner. The rates and temperatures needed to fit the tail
194	measurements underestimate the magnesium abundance near the surface.
195	These remarks effectively summarize the conclusions of the analysis by Killen et al.
196	(2010). The improvement up to this point is that we have fit all measurements together
197	here, a treatment that accentuates the conclusion of several distinct components.
198	In Fig. 3 we have fit the measurements in the weighted least-squares sense to a two-
199	component Chamberlain model, $I(\mathbf{r}) = n_{0,LT} I(\mathbf{r}, T_{COOL}) + n_{0,HT} I(\mathbf{r}, T_{HOT})$, where $n_{0,LT}$
200	$(n_{0,\text{HT}})$ and $T_{\text{COOL}}(T_{\text{HOT}})$ are the surface density and temperature, correspondingly, of the
201	cool (hot) components of the exosphere. In this case the added constraint of non-negative
202	surface densities must be enforced in minimizing the squared residuals in order to obtain
203	physically meaningful solutions. We have found that the superposition of a hot source
204	$(T_{HOT} \ge 20,000 \text{ K})$ with a cooler source $(T_{COOL} \le 5000 \text{ K})$ improves the description near
205	dawn compared to residuals obtained with a single-component model (e.g., compare Figs.
206	2 and 3a-b and the related χ^2/v error shown in Table 1 when $T_{COOL} = 3000$ K is assumed).
207	In particular, Figs. 3c-d demonstrate that the exospheric signature at low altitudes is
208	reproduced best if T_{COOL} = 400 K is assumed, which describes thermalized ejecta (also
209	see Table 1). At $T = 400$ K the mean velocity of a magnesium atom leaving the surface is
210	1 km/s, so the altitude reached is \sim 152 km. Such a low-energy source is undetectable
211	given only tail and fantail measurements, and lies below the scale of Fig. 3c.
212	[Insert Figure 3; Insert Table 1]

In conclusion, Chamberlain models indicate that at least two distinct temperatures and source rates of a few times 10^6 atoms cm⁻² s⁻¹ are required to fit the data. In the next

215 section we investigate the kinds of physical processes that might be consistent with these

216 requirements. 217 218 4. Relative roles of sources populating the magnesium tail 219 220 4.1. Model formulation 221 222 Mercury's neutrals are subject to losses via ballistic escape, radiation acceleration 223 (strong for Na, weak for Ca, and negligible for Mg), adsorption upon surface impact, and 224 photoionization. Thus, they are continuously resupplied by the surface via a number of 225 processes, including photon-stimulated desorption and thermal desorption that act upon 226 volatiles, as well as micrometeoroid impact vaporization and sputtering by the solar wind

and recycled magnetospheric ions that act upon both volatile and refractory species (e.g.,

228 Killen et al., 2007).

Near Mercury the magnesium neutrals are subjected only to gravitational forces, because for this species the radiation pressure is a very small fraction of Mercury's gravity and the photoionization lifetime is ~ 57 h (McClintock et al., 2009). Hence, the magnesium production and its distribution are modeled with an analytical model of particle transport in a collision-free exosphere (Hartle, 1971). The model uses Liouville's theorem to compute the density of neutrals,

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$$n(\mathbf{r}) = \int f(\mathbf{r}, \mathbf{v}) d^3 \mathbf{v},$$
 (2)

at an exospheric location **r**, where f is a truncated distribution function relating the constants of motion in a gravitational field to $f_0(\mathbf{r_0}, \mathbf{v_0})$, the velocity distribution function of released particles from the surface. Prior to applying it to the study of

magnesium transport at Mercury, we modified this model to simulate the velocity distribution function of sputtered particles, we used it to describe the source rates of the lunar sodium exosphere, and we validated it against ground-based observations (Sarantos et al., 2010).

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244 4.1.1. Exospheric sources

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246 The sources we initially considered are the typical processes that have been 247 hypothesized to produce exospheric refractory gases: impact vaporization, sputtering, and 248 molecular dissociation. However, we later allowed for the possibility of less energetic 249 sources acting upon a limited reservoir of volatile particles because we found that low-250 altitude data could not be reproduced by high-energy sources. Source processes can be 251 co-added because of the collisionless nature of the exosphere, but they compete for the 252 same surface reservoir. Thermal processes act upon the top monolayer; sputtering affects 253 the top ~ 10 nm, and impacts tap the top 1 μ m or more of the surface.

254 Sputtering by solar wind ions is a potential source of energetic atoms. Originally 255 introduced to explain the rapid temporal variability and the high-latitude enhancements 256 seen in Mercury's sodium exosphere (e.g., Killen et al., 2001), this source is regulated by 257 the interaction of the solar wind with the magnetosphere formed by Mercury's planetary 258 magnetic field. The fraction of the surface that can be exposed to solar wind ions varies 259 in response mainly to changes in the interplanetary magnetic field (IMF) (e.g., Sarantos et 260 al., 2001; Sarantos et al., 2007). The rate for this source is proportional to three uncertain 261 parameters: the abundance of magnesium in the regolith, the sputtering vield per incident 262 ion, and the influx onto Mercury's surface and composition of the solar wind.

Comment: Menelaos, is this OK?

263 Impact-driven vaporization caused by micrometeoroids is expected to be a source of 264 exospheric atoms (e.g., Morgan et al. 1988; Cintala 1992; Morgan and Killen, 1997, 265 Killen et al., 2001). The amount of vapor produced in this way is proportional to the 266 influx of micrometeoroids and depends on their velocity distribution. The mean impact 267 velocity may exceed 20 km/s at Mercury's orbit, and is even higher than that for larger 268 meteoroids or during meteoroid streams. Possible temperatures for impact vaporization 269 ejecta of 2500-5000 K, depending on the impact energy and the thermophysical 270 properties of the target material, are obtained in hypervelocity impact experiments 271 (Eichhorn, 1978). Uncertainties in the physical properties of the regolith and the 272 impactors, the assumed micrometeoroid mass flux and velocity, and the method used to 273 calculate the vapor yields each contribute to roughly a factor of five uncertainty in the 274 estimated vapor production rates (e.g., Cintala, 1992; Killen et al., 2005).

275 Molecules can be produced in the vapor + liquid + solid phase that follows 276 micrometeoroid impact. According to quenching theory, chemical reactions during the 277 collisional phase of the cloud $(10^{-7} - 10^{-5} \text{ s})$ lead to the formation of metallic oxides and 278 hydroxides (e.g., MgO, CaO, CaOH) (Berezhnoy and Klumov, 2008; Berezhnoy, 2010). 279 Such molecules may photolyze or dissociate due to their high internal energy and 280 produce high-energy atoms of Mg, Ca, O, as well as other species. We adopt the premise 281 that a dissociating molecule is a major source of energetic atomic Mg (Killen et al., 2010). 282 The same mechanism has been suggested by Killen et al. (2005) to be the main reason 283 why Mercury's neutral Ca tail is extremely hot. Besides the uncertainty in total impact 284 vapor, the unknowns for this source include the abundance ratio of atoms versus 285 molecules of the same species in the vapor cloud, and the molecular dissociation lifetime 286 and temperature.

287 Although less energetic sources such as photon-stimulated desorption or thermal 288 desorption cannot act on intrinsic magnesium that is bound in silicate phases, such 289 processes could act on recycled, gravitationally returning atoms that were originally 290 vaporized from silicates by impacts and sputtering. The main uncertainty here is whether 291 reabsorbed Mg physisorbs, or weakly adsorbs to the surface following return, so that 292 lower energy processes can re-eject them to the gas phase. The rate for low-energy 293 sources cannot exceed the return rate of atoms delivered by energetic processes, which 294 can be constrained from high-altitude measurements.

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296 4.1.2. Exospheric sinks

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Our model explicitly accounts for losses due to ballistic escape and sticking. Each returning atom is assumed to stick with unit efficiency. For those atoms that are generated by a dissociating molecule we also include losses due to the formation of a stable molecule with a finite lifetime (Section 4.4). However, the possible role of the surface as a sink requires some discussion.

303 Previous investigations have demonstrated that the efficiency of exospheric Na 304 ejection is related to a surface reservoir whose content varies with distance from the Sun. 305 A complex balance between the exospheric and the surface supply arises because (1) the 306 dayside reservoir for the very efficient processes (e.g., thermal vaporization and PSD) 307 can become depleted in local time if the resupply rate is not sufficiently large (Leblanc 308 and Johnson, 2010); (2) the resupply rate is diffusion-limited (Killen et al., 2004; Burger 309 et al., 2010); and (3) solar wind precipitation may enrich the surface by enhancing 310 diffusion through creation of vacancies (Mura et al., 2009; Burger et al., 2010). In

311 sodium simulations by Leblanc and Johnson (2010) and Mura et al. (2009), each test

312 particle is tracked even when trapped at the surface.

313 Unlike Na, which is a trace species of the regolith, observations of Mg by 314 MESSENGER allow us to infer that the regolith turnover rate ("gardening rate") can 315 most likely provide sufficient fresh targets to populate the exosphere with this species. 316 We demonstrate that only if the impact vapor consists mainly of molecules with half-317 lives in excess of ~2 min do the required source rates approach the gardening rate 318 (Section 4.4). Otherwise, our results suggest that, to first order, the Mg surface supply for 319 hot processes can be treated as infinite. However, what is limited is a possible reservoir 320 of volatiles which may be needed to explain the measurements at low altitudes (Sections 321 3 and 5.2). Therefore, we assume in our simulations two populations of Mg at the 322 surface: one that is strongly bound in silicate phases and can be released only by impacts 323 and sputtering, and one that originates from atoms and molecules that have returned to 324 the surface following emission from the silicate population. The latter component should 325 physisorb with lower binding energy than Mg in silicates so that it can be vaporized by 326 lower-energy processes.

Comment: Menelaos, I can't understand this sentence. What is a "thermal particle"? What is your "evidence"?

Comment: Sean, "volatiles" has substituted "thermal particles", "Elsewhere, "thermal processes" are now referred to "low-energy" or "less energetic" processes. Five added the Section numbers where this "evidence" is presented.

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328 4.2. Ion sputtering

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Simulations of sputtered magnesium are initialized with the flux of precipitating solar wind ions predicted by a magnetohydrodynamic (MHD) model of Mercury's magnetosphere (Benna et al., 2010). The model predictions under the southward IMF **B** field that prevailed during the second MESSENGER flyby are shown in Fig. 4a. The assumed solar-wind conditions were: $n_{sw} = 20 \text{ cm}^{-3}$, $V_{sw} = 400 \text{ km/s}$, and $[B_s, B_y, B_z] = [-$

8, 4, -10] nT, where n_{sw} and V_{sw} are the solar wind density and velocity, respectively, and the IMF is given in Mercury solar orbital coordinates, where x is directed from the center of the planet toward the Sun, z is normal to Mercury's orbital plane and toward the north celestial pole, and y is in the direction opposite to orbital motion.

339 The production rate due to ion sputtering is obtained by assuming a sputtering yield, 340 Y, weighted by protons and alpha particles, of 0.1 per ion impact (Wurz et al., 2007), an 341 upper limit for the magnesium abundance in the regolith of c = 0.17 by number (Wurz et al., 2010), and a precipitating flux of $F_{SW} = 2 \times 10^8$ solar wind ions cm⁻² s⁻¹ poleward of 342 343 $\pm 50^{\circ}$ dayside latitude. These parameters yield a sputtered flux, $F_{SPUTT} = cYF_{SW}$, of 3.6 x 10^6 Mg atoms cm⁻² s⁻¹ and a modeled near-surface density, n_0 , of ~6 atoms cm⁻³ at polar 344 345 cusp latitudes. The mean ejection energy for this process is approximately half the 346 binding energy, which is assumed to be 3.6 eV in this simulation, and the energy, E, and 347 directionality of the ejecta are described by the Sigmund-Thompson function (e.g., Wurz 348 et al., 2007):

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$$f(E,\delta) = \frac{6E_b}{3 - 8\sqrt{E_b}/E_{Max}} \frac{E}{(E + E_b)^3} \{1 - \sqrt{(E + E_b)/E_{Max}}\} \cos\delta$$
(3)

where $E_b = 3.6 \text{ eV}$ the magnesium binding energy to the regolith grains, $E_{\text{Max}} = 450 \text{ eV}$ the maximum energy that can be imparted to the ejected magnesium atoms by 3 keV protons, and δ the angle from local vertical. Due to the effects of soil porosity, the atoms are sputtered primarily perpendicular to the surface, with the yield assumed to lessen as cos δ .

Subject to these assumptions, our simulations demonstrate that the contribution by sputtering to the fantail measurements varies from about 10% for observations near the equator to about 50% for those over the northern pole (Fig. 4b). On average, about 20%

358 of the column abundance observed during these measurements can be attributed to 359 sputtering.

360 [Insert Fig. 4]

361 These predictions are uncertain for two reasons. First, the predicted location and flux 362 of the plasma reaching the surface are not only model-dependent but are also sensitive to 363 the assumed solar wind conditions. These conditions can be inferred only indirectly 364 because MESSENGER observes neither the solar wind nor the IMF while inside the 365 magnetosphere. More importantly, the magnetosphere during M2 was extremely dynamic 366 (Slavin et al., 2009), yet the exospheric consequences of short-term (~1 min) 367 magnetospheric variability due to magnetic reconnection have not been evaluated by any 368 model to date. Bearing these uncertainties in mind, we conclude that, although sizeable, 369 the contributions by sputtering alone cannot provide the entire energetic component and 370 so the MESSENGER measurements indicate that some other energetic source process is 371 at play.

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373 4.3. Impact vaporization

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To first order the influx of micrometeoroids to Mercury's surface is assumed to be isotropic. We model the production of atomic magnesium during impacts as following a Maxwellian distribution at temperatures 3,000–5,000 K. Under the assumption that meteoroid impacts produce both atoms and molecules, we include in our simulations the dissociation products of MgO \rightarrow Mg + O. Three unknowns are the dissociation cross sections, the resulting velocity distribution of the dissociated ejecta, and the molecule lifetime.

382 Lacking experimental results, we assume that the molecules dissociate due to their 383 high internal energy (they are hot, hence unstable, if they are produced during impacts). 384 In this case, a wide spectrum of ejecta energies can be expected, so our assumed 385 energetic Mg distributions from dissociation are Maxwellian at equivalent temperatures 386 of at least 5,000 K (> 0.5 eV). The dissociation lifetime τ of MgO may be very short, 4 s 387 according to scalings from other diatomic molecules (Berezhnoy, 2010). However, 388 metallic oxides in Earth's atmosphere are fairly stable, with a known lifetime for NaO of 389 42 s (Self and Plane, 2002). In this Section we present models in which the molecule is 390 assumed to break up immediately upon production ($\tau = 0$) and the exobase is Mercury's surface; in the following section we present models with finite τ , where the exobase is 391 392 extended.

Comment: Menelaos, you did not include this reference in the References. I have guessed the paper you intended. Comment: Sean, you guessed right—thank you.

393 If $\tau = 0$ and sputtering contributions are small, then we effectively recover the two-394 component Chamberlain fits (see Table 1) for the partitioning of atoms and molecules in 395 the vapor. A match between the model results and the magnesium measurements in the 396 fantail and near-terminator regions, fit together to describe the global distribution of 397 magnesium self-consistently, is shown in Fig. 5, where "Model Sum" refers to the total 398 contribution by sputtered magnesium, impact-driven atomic magnesium, and magnesium 399 from dissociation. We also ran models where the molecules were allowed to dissociate 400 only on the dayside and beyond the shadow (photolytic scenario), and found, in that case, 401 that the molecule production rates must be about a factor of three higher than the rates 402 quoted in Table 1 to put the same number of atoms in the tail. As in Section 3, the wide 403 range of possible values for these sources can be traced to measurement uncertainties 404 because pairwise combinations of all models differ by statistically insignificant amounts. 405 In summary, we conclude that the high-altitude measurements can be explained by the 406 processes studied here if approximately one-third to one-half of the total amount of

407 exospheric magnesium due to impacts comes directly in atomic form, and the rest results 408 from the dissociation of a molecule, with the total rate summing to $(2-4) \times 10^6$ Mg 409 atoms cm⁻² s⁻¹ if the molecule photolyzes, or to about $(1-2) \times 10^6$ Mg atoms cm⁻² s⁻¹ if 410 the molecule vibrationally dissociates.

411 [Insert Fig. 5]

412 Can impact vaporization rates produce the observed tail abundances? The total vapor rate produced by micrometeoroid impacts at 0.342 AU was estimated by Killen et al. 413 (2010) to be $M_{\rm vap} = 2.7 \times 10^7$ atoms cm⁻² s⁻¹ for all species. The model is based on the 414 assumption that the delivery of meteoroid material at Earth's vicinity is at a rate of 3 × 10 415 ⁻¹⁶ g cm⁻³ for particles smaller than 1 cm (Love and Brownlee, 1993); this rate can be 416 417 scaled to Mercury's distance from the Sun, 0.342 astronomical units (AU) at the time of 418 the second MESSENGER flyby, following Cintala (1992); and the thermodynamic 419 parameters of aluminum onto enstatite and a regolith porosity of 0.5 (Morgan and Killen, 420 1997; Killen et al., 2005) may be adopted. On assuming a magnesium abundance in the 421 regolith between the lunar values and those recently estimated for Mercury's regolith by 422 Wurz et al. (2010), c = 0.05 - 0.17 by number, we estimate the magnesium production rate due to a uniform micrometeoroid influx to be $(1.3-4.7) \times 10^6$ Mg atoms cm⁻² s⁻¹. 423 This vaporization rate, which includes the products of MgO if formed in the cloud 424 425 expansion, is consistent with the rate inferred from the tail measurements, $\leq (2-4) \times 10^6$ Mg atoms $\text{cm}^{-2} \text{ s}^{-1}$, if the molecule lifetime is very short. A point that will be studied next, 426 and which was not addressed by Killen et al. (2010), regards the effect of a finite 427 428 molecule lifetime on impact-driven arguments.

Secan Solomon 24(11.1.16 AM **Comment:** Menclaos, is 'rate" the correct word here, given that you are citing a flux. (See my. question at this point in the previous draft.)

429

430 4.4. Extended exosphere models

433 because a fraction of the molecule population returns to the surface before it is 434 dissociated. We tested a source of Mg by dissociation of MgO at different altitudes from 435 the surface. We computed the loss flux to dissociation of MgO molecules as a function of 436 altitude, r, having corrected the distribution function f(r) for the survival probability, 437 $e^{-t(r,v)/\tau}$, where t(r,v) is the time elapsed since ejection from an altitude r_0 and τ the 438 molecule lifetime (e.g., Cui et al., 2008). This loss flux, which is the production rate of 439 new atoms, was used to initialize the atom redistribution model of Eq. 2 where the limits 440of integration now relate to jumps both from higher and from lower altitudes. The 441 molecule lifetime was treated as a free parameter with an assumed range between 1-1000 442 s.

If the molecule has a finite lifetime, the required rates will exceed those quoted above

443 In the right panel of Fig. 6 we present profiles of the dissociating flux from an MgO molecule produced from the surface at a rate of 10⁶ molecules cm⁻² s⁻¹. On the left the 444 445 resulting profiles for energetic Mg atoms are shown, along with the $\tau = 0$ profile at the 446 same rate and dissociation temperature. It can be seen that: (1) the profiles from an 447 extended exosphere are less steep, meaning that particles can escape easier from a given 448 altitude at the same assumed dissociation temperature; (2) at $\tau = 1$ s the model approaches 449 the idealized $\tau = 0$ model of the previous section, and (3) if the molecules are stable, 450 higher ejection rates are necessary to populate the tail with the same number of atoms. 451 With the exception of the production rate, the profiles are so similar that we cannot 452 separate the effects of a different temperature of the original molecule (T = 3000 - 5000453 K), the different dissociation temperatures (5000 - 20000 K tested), and the different 454 lifetime τ .

Two questions that we can now answer are: can the surface limit the delivery rate of magnesium to the exosphere? And what is the possible lifetime of the putative Mg**Comment:** Menelaos, it is not a good idea to have boldface r denote radius and italicized r denote altitude. I suggest that the pick a different symbol for altitude, say, z or h.

. Inenelaos sarantos 3/4/11.9 16 AM Comment: Sean, is this a clear answer to your question?

457	bearing molecule? A limit to the rate that the surface can provide is the gardening rate.
458	For a turnover rate at Mercury that is ten times that of the Moon, the top μm is turned
459	over once every 125 years; this rate brings $\sim 6 \times 10^7$ new Mg atoms cm ⁻² s ⁻¹ over such a
460	time to the upper μ m, which is the depth from which impact vaporization draws its
461	supply. This rate exceeds the exospheric rates by more than an order of magnitude if the
462	molecule has an infinitesimal lifetime (see previous Section). If τ is longer, our
463	simulations indicate that:
464	• For $\tau = 1000$ s the required source rates to populate the tail exceed the
465	gardening rate.
466	• For $\tau = 100$ s the data necessitate a rate of (2–5) ×10 ⁷ mol cm ⁻² s ⁻¹ , i.e., less
467	than the gardening rate but up to 10 times the impact vaporization rate given
468	by Cintala (1992).
469	• For $\tau = 10$ s a best fit requires (3-7) ×10 ⁶ mol cm ⁻² s ⁻¹ , no more than three
470	times the rate quoted in the previous Section.
471	Our prediction of short dissociation lifetimes for MgO can be tested with laboratory
472	experiments.
473	
474	5. Magnesium near the surface
475 476	We can approximately reproduce the magnesium tail under the assumption that the
477	source processes are sputtering and uniform impact-driven release, but we cannot
478	reproduce the distribution near the terminator where half the exospheric content is not
479	predicted by these processes (compare Fig. 5b to Fig. 3). Because of observational
480	constraints, we cannot ascertain what causes the enhancements observed in this region.

481 We suggest two possibilities that may explain the magnesium distribution near the

482 planetary surface: one relating to a local source of impacts (Section 5.1), and one relating

483 to a reservoir of adsorbed particles (Section 5.2).

484

- 485 5.1. A single ~0.5 m impactor at equatorial dawn?
- 486

We can visualize what is inherently different about the measurements obtained near the dawn terminator by mapping with our particle transport model the "footprint" of atoms that scatter light into the instrument for different boresights and tangent heights. Some examples are shown in Fig. 7 for a uniform source having temperature T = 5,000 K.

491 [Insert Fig. 7]

492 For lines of sight pointing far from the surface (in the tail and fantail observations), 493 MESSENGER UVVS observed primarily the escaping component; hence the 494 instrument's effective field of view is large. Fig. 7a shows a typical measurement 495 obtained in the tail prior to the spacecraft entering the shadow, and Figs. 7b to 7d show 496 the beginning, middle, and end of the fantail sequence. As expected, the main 497 contributions to the observed column abundance drift from being approximately equally 498 weighted between dawn and dusk in the tail to being primarily dawn, then north, then 499 dusk during the fantail sequence. Particles mapping outside the shadow originate mainly 500 from the nightside surface and contribute approximately two-thirds of the Chamberlain 501 column abundance; ~one-third of the total column abundance is contributed by particles 502 originating on the dayside. (This result implies that the molecule production rates quoted 503 in the paper should be increased by a factor of three if the molecule is destroyed by 504 photons rather than being vibrationally dissociated.). As a wide area of the surface

505 contributes to the tail and fantail measurements, localized "disturbances" such as a 506 surface density enhancement or a meteoroid impinging on Mercury's near-equatorial, 507 morning sector would go undetected until the spacecraft pointed near the surface.

In contrast, for lines of sight nearly intercepting the dawn terminator surface, the ejecta mapping into the instrument's field of view are very localized. As the spacecraft comes out of the shadow, the population sampled is slowly drifting towards the equatorial dawn and morning sectors in successive lines of sight (Fig. 7e – f). We conclude that the measurements near the terminator are sensitive to localized sources. This point is further studied in Fig. 8.

514 Having evaluated the impact-driven column abundances element by element as in Fig. 515 7, we then combine them by different amounts to investigate possible asymmetries in a 516 least-squares manner. First, we determine the location of the possible meteoroid impact by iteratively enhancing the weight of "pixels" starting at equatorial dawn, then 517 518 progressively expanding the "size" of the assumed impact area. The shape of the near-519 terminator curve is fit best if we enhance only one pixel, that sitting at the dawn 520 terminator with an extent of $\pm 10^{\circ}$ in longitude and latitude. Then, the production rate and 521 its partitioning into the two impact-driven populations is determined by least-squares 522 regression. This outcome is shown in Fig. 8 for assumed local enhancements of vapor 523 rates over the uniform model by factors of four and eight. As seen in the left column, 524 these models are indistinguishable from fantail measurements, but the models having 525 enhancement factors of 4-8 represent the near-terminator observations increasingly well. 526 [Insert Fig. 8]

527 A meteoroid stream is not a suitable explanation because streams would be expected 528 to have a hemispherical dependence and not be so localized (J. Vaubaillon, personal 529 communication, 2010). A single impactor in a small region around dawn is possible. 530 According to our estimates, enhancements of the local impact-driven release rate by 531 factors of 4-8 would require the meteoroid to have a diameter of ~ 0.3 m; impacts of 532 objects of such a size occur once every 3-4 days at Mercury. The likelihood that this 533 impact happened around dawn within ~1000 s prior to this observation so that the cloud 534 remained localized is low.

535

536 5.2. A source of volatile Mg?

537

538 An alternative, more likely way to interpret the near-terminator measurements is to 539 assume that returning Mg atoms from an energetic process do not strongly bind with the 540 surface and can be re-emitted to the gas phase when exposed to UV photons or to the 541 high temperatures of the dayside surface. Such a hypothesis is suggested by the two-542 component Chamberlain fits (Section 3). As illustrated by Fig. 3, a low-energy source 543 produces a sharp increase in column abundance as the spacecraft probes areas near the 544 planet, i.e., within 100 km of the surface. Note, however, that the population consistent 545 with a T = 400 K Chamberlain model would be effectively concentrated on the nightside 546 during this sequence (e.g., see Fig. 7).

A source concentrated on the dayside would produce the same sharp exospheric profile over the terminator but would require much higher source rates than those of a uniform model (Table 1). To investigate the possibility that the terminator measurements

550 for Mg might be consistent with photon-stimulated desorption (PSD) or thermal 551 desorption of adsorbed ejecta, we ran models under the premise that, in addition to 552 uniform impact vaporization of atoms and molecules, sources included a dayside source having a cos (χ) or cos^{1/4} (χ) dependence on the solar zenith angle, χ a varying 553 554 temperature of 100-1500 K, and no emission from the nightside. We found excellent fits 555 to the observed profile for assumed source temperatures of ~700-1200 K. However, the 556 required rates to match the measurements with these dayside source models were too high, 5×10^8 Mg atoms cm⁻² s⁻¹, to be provided either by the return rate or by the 557 558 gardening rate; such rates can be justified only if returning particles do not stick and 559 rebound multiple times, which is probably a poor assumption for refractory elements 560 such as Mg.

561 In conclusion, the illuminated limb-scan profiles can be matched at acceptable 562 source rates only if the putative cold source extends to the nightside. That is, some 563 process other than PSD and thermal desorption would be required to eject low-energy Mg 564 neutrals. Perhaps electron-stimulated desorption (ESD) is responsible for T = 400 K 565 neutrals since magnetospheric electrons have access to the nightside; no ESD 566 measurements are available for this species, however. Note that the inferred rate from a T = 400 K source corresponds to a few times 10^6 atoms cm⁻² s⁻¹ (Table 1). Although our 567 568 simulations have not carefully treated the reservoir in the top monolayer of the surface. 569 flux of this magnitude could approximately be provided by the return rate of impact-570 driven atomic Mg and by surface-directed Mg from dissociation of a molecule with a 571 finite lifetime. All but 1-2 % of magnesium atoms from a 3000-5000 K source are

572 gravitationally bound, and approximately half the dissociating flux from a molecule will

573 be directed towards the surface.

574

575 6. Conclusions

576

577 Comparison of the measurements of Mg in Mercury's exosphere with a large number of 578 simulations have suggested the following possibilities: (1) scale-height arguments imply the 579 presence of at least two distinct temperatures; (2) energetic processes, such as 580 micrometeoroid impact vaporization, ion sputtering, and dissociation of a Mg-bearing oxide, 581 can supply the exospheric population at high altitudes, but no single process dominates; (3) 582 the lifetime of the putative molecule may not exceed ~ 100 s in order to supply the needed 583 rates given the replenishment rates of the surface by gardening; and (4) at low altitudes, low-584 energy processes appear to act on a limited reservoir of volatiles that may be provided by 585 recycled atoms from hot processes. A visual summary of the paper's results appears in Fig. 9. 586 Preliminary analysis of third flyby data (not discussed at length in this paper) 587 indicates similar temperatures and source rates in the tail and polar regions to those 588 presented here. We cannot confirm the volatile source because observations of 589 magnesium on the equatorial dawn region were not conducted by MESSENGER during 590 its third flyby as a result of a spacecraft safe-hold event (Vervack et al., 2010). 591 Improvements to the model will require laboratory measurements of appropriate physical 592 constants (e.g., molecule lifetime and dissociation cross-sections, and temperature-593 programmed desorption yields for adsorbed Mg). The measurements expected during the 594 orbital phase of the MESSENGER mission promise to further constrain these results. For 595 instance, if impacts populate the high-altitude Mg exosphere as inferred here, there

596 should be a correlation of exospheric content with spacecraft passages through the 597 interplanetary dust plane (Kameda et al., 2009). A source of volatiles can be verified or 598 refuted during repeated illuminated limb-scanning opportunities.

599

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- 709
- 710

- Table 1. Two-component Chamberlain models obtained by superposing a cool source,
- 100-5000 K, and a hot source, $T \ge 10000$ K.

	Mo	del: n _{0,LT} I(T=3	000 K) + $n_{0,\rm HI}$	I(T)	
$n_{0,LT}$ (atoms cm ⁻³)	$T (\times 10^3 \text{ K})$	$n_{0,\mathrm{HT}}$ (atoms cm ⁻³)	$S_{LT} (\times 10^5 \text{ cm}^2 \text{ s}^{-1})$	$S_{\rm HT} (\times 10^6 {\rm cm}^2 {\rm s}^{-1})$	χ^2/v
6.1	10	12.7	2.4	0.9	6.23
14.8	20	8.7	6.0	1.0	5.48
16.8	30	8.5	6.8	1.1	5.29
17.7	40	8.7	7.2	1.3	5.20
18.1	50	9.0	7.4	1.5	5.15
	Mo	del: $n_{0,LT} I(T =$	400 K) + n _{0,H1}	I(T)	
$n_{0,LT}$ (atoms cm ⁻³)	$\frac{T}{(\times 10^3 \text{ K})}$	$n_{0,\mathrm{HT}}$ (atoms cm ⁻³)	$\frac{S_{\rm LT} (\times 10^6}{\rm cm^{-2} s^{-1}})$	$\frac{S_{\rm HT} ~(\times 10^6}{\rm cm^{-2} ~s^{-1}})$	χ^2/ν
298	10	12.1	4.4	0.9	5.87
397	20	9.0	5.9	0.9	5.03
418	30	8.9	6.2	1.1	4.84
426	40	9.2	6.3	1.4	4.75
430	50	9.5	6.4	1.6	4.71

Notes: Listed are the corresponding surface density, n_0 , production rate, S, and reduced chi squared error, χ^2/ν , for different assumed values of the exobase temperature, T. The models are fit to all measurements. (LT=Low Temperature; HT = High Temperature) 716 717

*

720

721	Fig. 1. All Mg observations obtained during MESSENGER's second flyby were fit to
722	Chamberlain models. Shown here are lines of sight through the tail that intercept the
723	noon-midnight meridian plane (a) 2 to 4 R_M from the equatorial plane; (b) 1 to 2 R_M from
724	the equator; and (c) inside Mercury's shadow. The measurements and their uncertainties
725	are shown in red. Three models shown, which have temperatures of 3,000 (dotted black
726	line), 20,000 (blue line) and 50,000 K (magenta line), indicate that atoms in the tail are
727	very energetic.

728

Fig. 2. Weighted least-squared fits of (a) the fantail and (b) near-terminator measurements to Chamberlain models subject also to the tail data. Only a poor description of the low-altitude measurements near dawn can be achieved with a single temperature.

733

Fig. 3. Two-component models are improvements over single-component models at low altitudes, especially if a low-energy source (T = 400 K) is assumed (panels c and d). The fantail observations, where the line of sight drifts to infinity, constrain the energetic process, whereas the source having the smaller temperature dominates the nearterminator measurements, which probe altitudes near the surface.

739

740 Fig. 4. (a) The adopted location and flux of protons bombarding the surface of Mercury,

741 obtained from a MHD model of Mercury's magnetosphere (Benna et al., 2010); (b)

742 amount of magnesium that can be attributed to sputtering under these conditions.

743

Fig. 5. A possible model of Mercury's magnesium exosphere consisting of three assumed source processes: sputtering (in green), impact vaporization of atomic Mg (in black), and dissociation of MgO (in blue). In contrast to the model shown in the next figure, here the molecule is assumed to break up immediately upon production ($\tau = 0$). In this limiting case the inferred rates can be provided by impact vaporization of atoms and molecules for magnesium abundances in the regolith of five percent or higher. The sputtering contribution to the near-terminator measurements is below the scale shown.

751

752 Fig. 6. (a) Flux of dissociating MgO as a function of radial distance and an assumed 753 molecule lifetime, τ , as the source rate for an extended exosphere model; (b) profiles of 754 energetic Mg resulting from the dissociation of MgO with arbitrary τ . A production rate 755 of 10⁶ molecules cm⁻² s⁻¹ and a dissociation temperature of 20,000 K are assumed; $\tau = 0$ 756 is the profile used in the simplified case of Fig. 5 at the same temperature. Ejection rates 757 of a stable molecule must evidently be higher in order to produce the same column 758 abundance in the tail. Based on a comparison to the gardening rate, the molecule must be 759 short-lived, $\tau \leq 2$ min, to explain the measurements.

760

Fig. 7. Relative contribution by planetary surface elements to the modeled column abundance for an isotropic impact vaporization source (T = 5000 K) and for selected lines

763 of sight: (a) typical line through the equatorial tail when the spacecraft lies outside the 764 shadow; (b)-(d) the beginning, mid-point, and end of the fantail sequence; (e)-(f) first and 765 last lines of sight for the near-terminator sequence. In these plots, the integrated column 766 abundance from a model of uniform ejection has been divided to extract the originating 767 locations of particles that map outside the planetary shadow (and hence contribute to the 768 measurements). The main point illustrated is that particles that affect the pre-dawn 769 observations originate from a limited region nearby. The subsolar point is indicated by 770 the white dot.

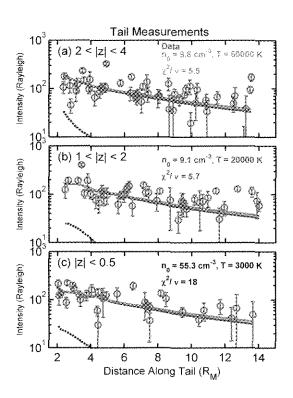
771

772 Fig. 8. In addition to the assumption of a low-energy source (Fig. 3), models of the near-773 terminator measurements markedly improve if a single meteoroid impacted Mercury 774 within $\pm 10^{\circ}$ of equatorial dawn around the time of these observations: (upper panel) 775 model with no equatorial enhancement at dawn; (middle panel) model with a factor of 776 four, and (lower panel) model with a factor of eight enhancement over the uniform 777 impact vaporization rate. In each case the "background" surface density, n_0 , needed to match the data is shown in black (T = 3000 K) and blue (T = 20,000 K). It is surmised 778 779 that a single meteoroid should have produced the equivalent flux of $\sim (6-8) \times 10^6$ atoms/molecules cm⁻² s⁻¹ within a brief period prior to the last few measurements for this 780 781 scenario to work.

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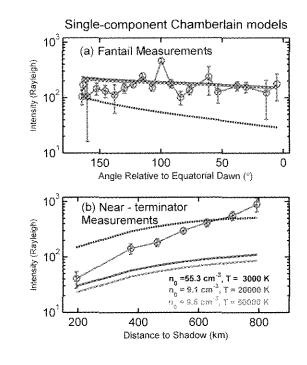
Fig. 9. Schematic depiction of potential processes promoting Mg to the exosphere. Thegardening rate in the top micron suffices to provide the exospheric rates, shown here for

- 785 brief molecule lifetimes. The recycling rate of bound ejecta could replenish an
- 786 unexpected reservoir of volatiles at rates of $\sim 10^6$ atoms cm⁻² s⁻¹.

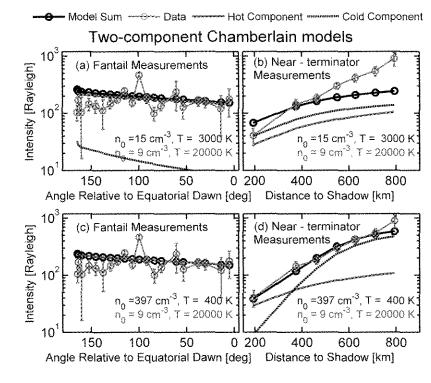




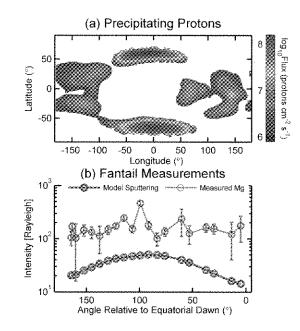




791 Figure 2

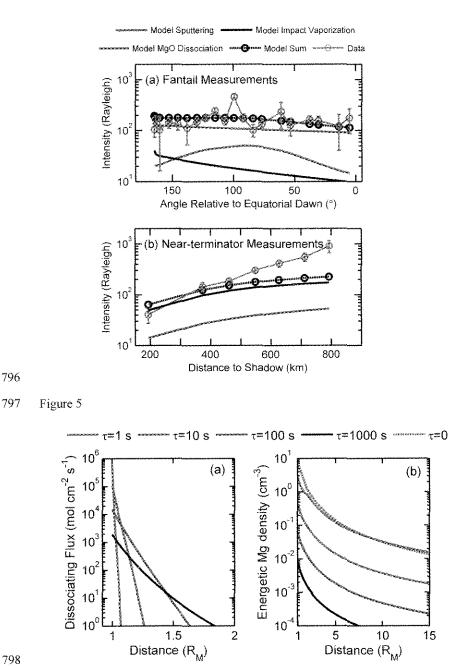


793 Figure 3



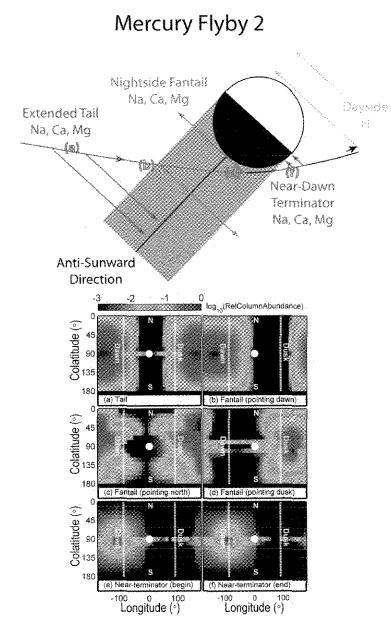






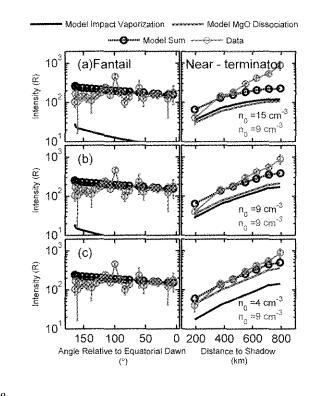




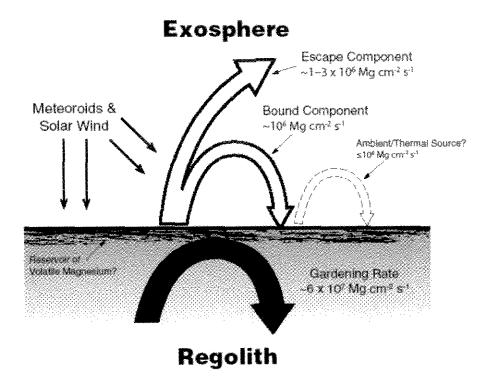












805 Figure 9