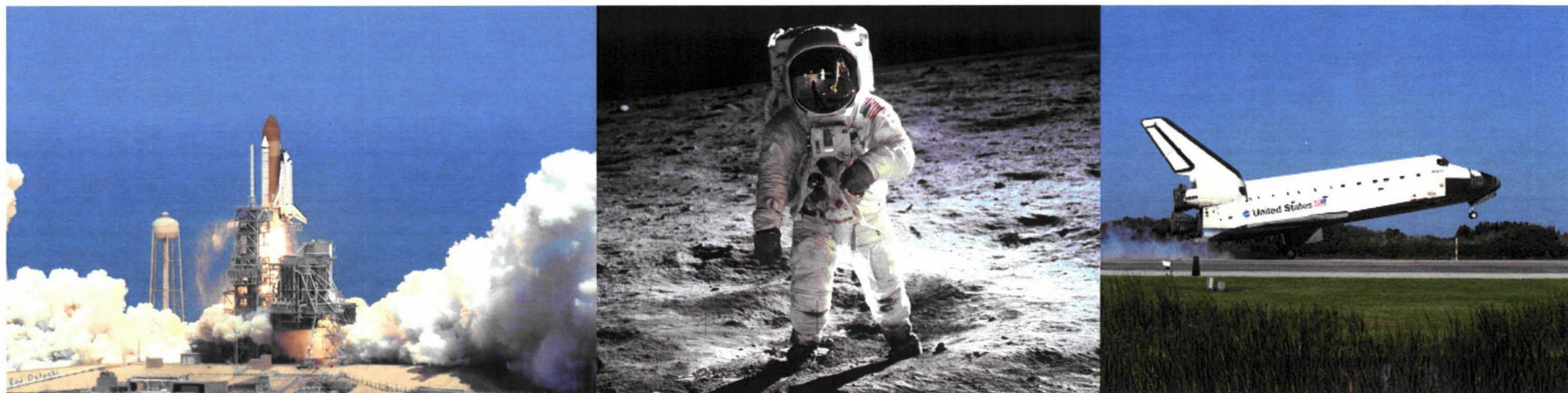




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# Polymer Chemistry



**Chemical Analysis and Polymer Branch  
Materials Science Division  
Engineering and Technology Directorate  
Kennedy Space Center, Florida**

Dr. Martha Williams (Martha.K.Williams@nasa.gov)

Dr. Luke Roberson (Luke.B.Roberson@nasa.gov)

Anne Caraccio (Anne.Caraccio@nasa.gov)

7/29/2010





# Materials Science Division Organizational Chart

NE-L  
Murray, S. – Chief  
~~Foster, A. – Deputy~~ *Delete*  
Balles, A. – Technical  
Integration Mgr

NE-L1  
Failure  
Analysis &  
Material Eval.

NE-L2  
Mat. Testing &  
Corrosion  
Control

NE-L3  
Prototype  
Development

NE-L4  
Materials and  
Processing

NE-L5  
Applied  
Physics

NE-L6  
Chemical  
Analysis &  
Polymer Branch

Chemical  
Analysis

Polymer  
Science &  
Technology  
Lab



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# Lab Overview

## Mission

*To develop and apply new technologies in polymer and material chemistry  
that benefit NASA's programs and mission*

## Team

*2 NASA scientists, 1 co-op, and 4 contractors*

## Areas of Expertise

*Polymer Nanocomposites*

*Next Generation Wire Materials*

*Carbon Nanotube and Nanofiber Materials*

*Conductive Polymers*

*Polymer Processing*

*Fire and Polymers*

*Foam and Insulation Materials*

## Numerous Collaborative Efforts

***NASA Centers*** (JSC, LaRC, MSFC, GSFC, GRC)

***KSC Directorates*** (Shuttle, Ares, Orion, Ground support operations)

***Academia*** (Alberta, FIT, GT, Harding, Illinois-Urbana Champagne, UCF, UF, USF)

***Industry Space Act Agreements*** (Thermax, DeWAL, Sharklet, Crosslink, Sabic, Amalgam)

***Industry Contracts*** (ARCnano, Epner)









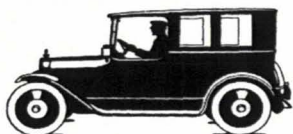
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# History of Polymer Chemistry

**1844**

Charles Goodyear  
patents  
**vulcanization**  
process.



**1907**

Leo Bakeland  
created the first  
completely synthetic  
polymer, Bakelite  
(Phenol/formaldehy  
de).



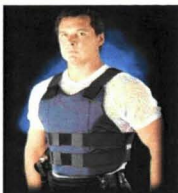
**1920**

**Vinyl Chloride**  
resin mass  
production begins  
(Invented in 1835).



**1960s-70s**

**Kevlar** and high  
performance polymer  
industries take off.



**1930s-40s**

**Polyethylene** mass  
production begins.



**1930**

**Polystyrene** mass  
production begins  
(Invented in 1839 by  
Eduard Simon).



We now use more **plastic** than steel, aluminum and copper combined.

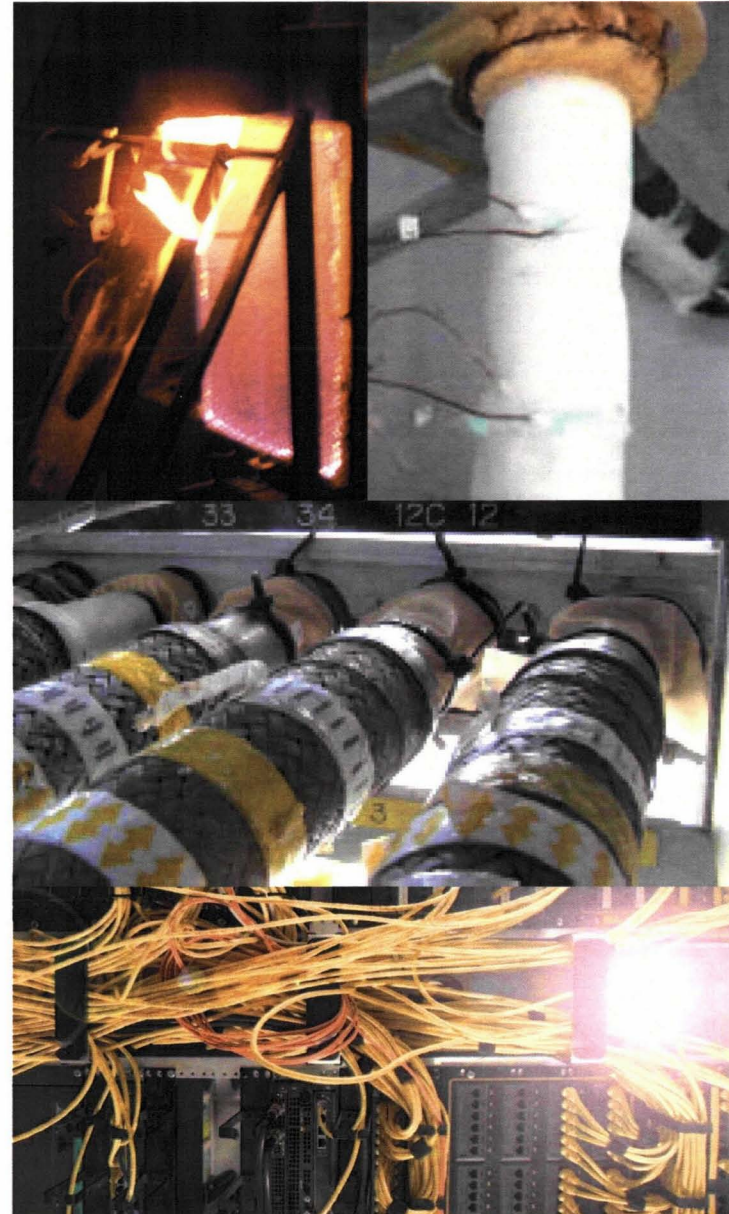


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# Composites/Materials Development at KSC

- Smart Materials and Detection Systems
- Self healing materials
- Flame retardant materials
- Aerogel composites
- Aerogel for environmental remediation
- Chemoschromic hazardous gas detectors
- Antimicrobial polymers
- CNTs and conductive polymer technologies







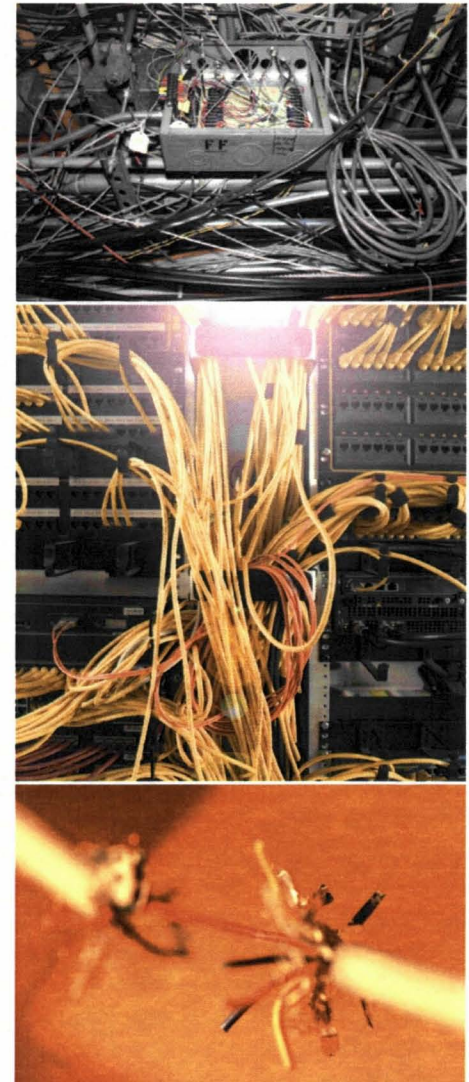
# Why Wiring?

- **Aged Wire**

- Cracks and frays over time
- Hard to detect damage
- Extensive maintenance related damage during ground processing work

- **Space Shuttle Orbiter**

- 183 miles of wiring buried deep within structure of vehicle
- Difficult to manually inspect







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# Next Generation Wiring Materials

## Wire System Failures

STS-93 (July 1999)

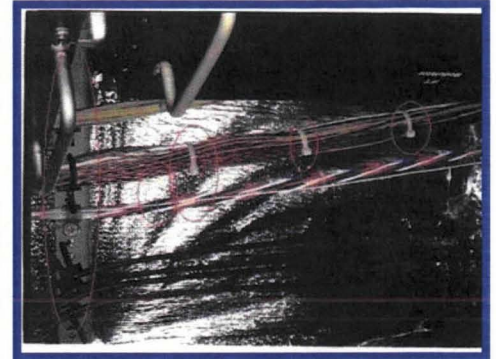
Short circuit in 14 AWG Kapton® insulated wire

TWA 800 (July 1996)

Frayed Kapton® wire in center tank area

SwissAir 111 (September 1998)

Damaged wire in plane's entertainment system



Manual Repair Technologies for Kapton and Teflon  
wires

In situ Damage Detection systems for vehicle health  
monitoring

Self-Healing insulation





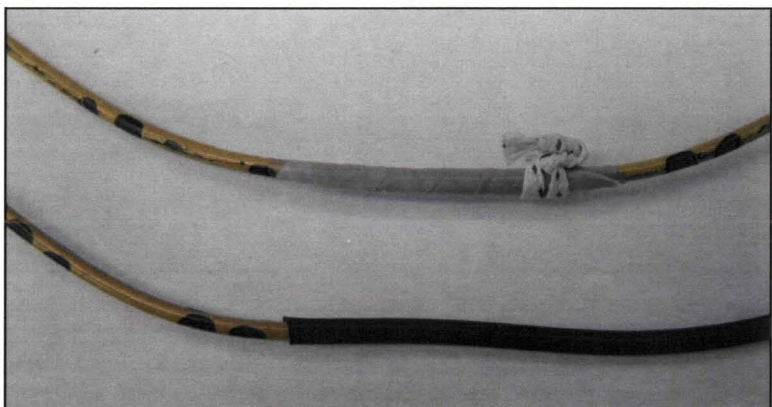
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# Wire System Materials

## Insulation and Repair Materials

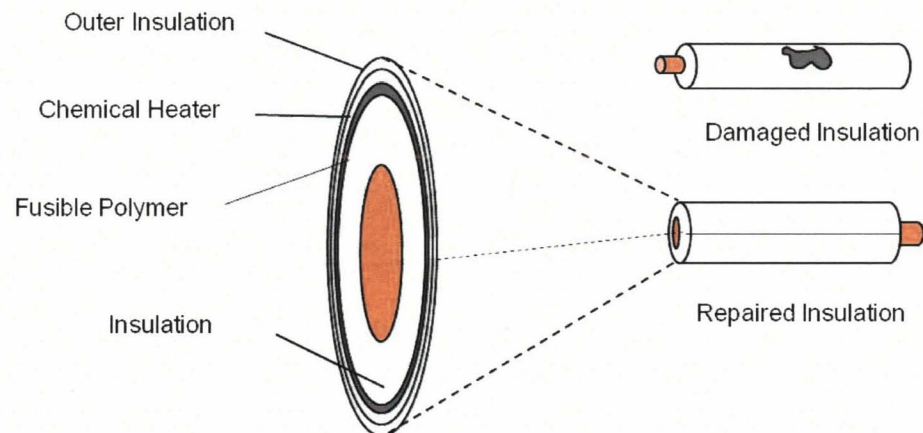
### Present Wiring Repairs



### Casting of wire repair materials



### Manual Repair Concept



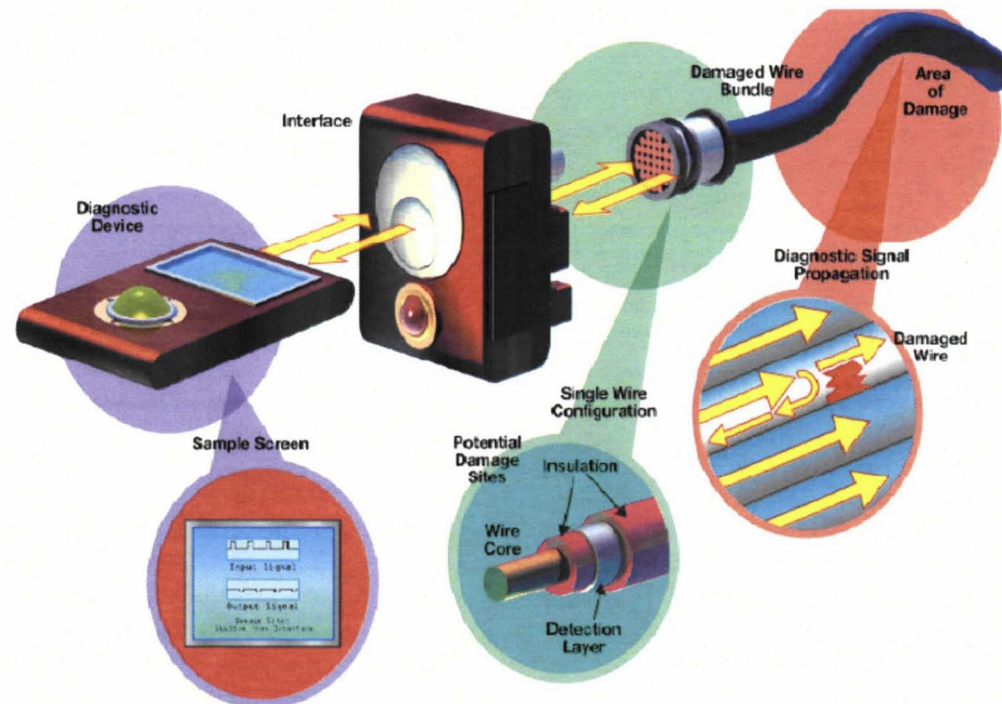
### Laboratory Repair Process





# Wire Systems Integration

- Smart Connectors
  - Small, lightweight, ultra reliable
- In-situ wire damage detection system
  - Capable of wire damage detection “on-the-fly”
- Integrated vehicle health monitoring (IVHM)
  - System-of-systems level, providing high level of reliability







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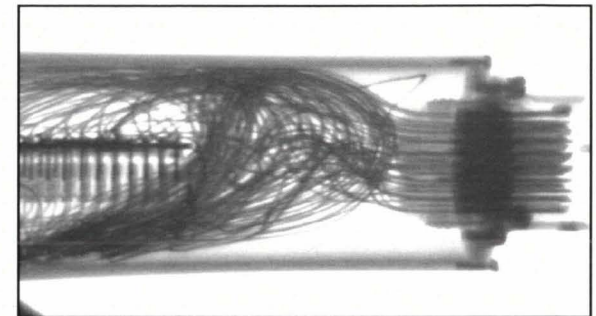
# Self-Healing Wire Repair

See Video



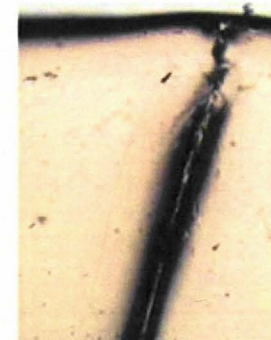
## “Smart Wiring” Summary

- Early wire fault detection and self-repair includes the development of an *in situ* detection system to detect and locate an electrical compromise on an energized “live” wire.
- The *in situ* detection monitoring system uses Time Domain Reflectometry (TDR) to locate failures such as opens, shorts, and intermittent faults in existing wiring systems.
- A new wire construction that contains a conductive composite detection layer for early detection to wire insulation damage and self-healing capabilities are also included in this “Smart Wiring” system.
- Applicable to Game Changing Technologies, Cross-cutting Capabilities, Flagship Inflatable Technology Demonstrations, and 21<sup>st</sup> Century Launch Complex.
- Currently partnering with wiring industry and NAVAIR.



*X-ray image of miniaturized TDR connector*

7 seconds



19 seconds



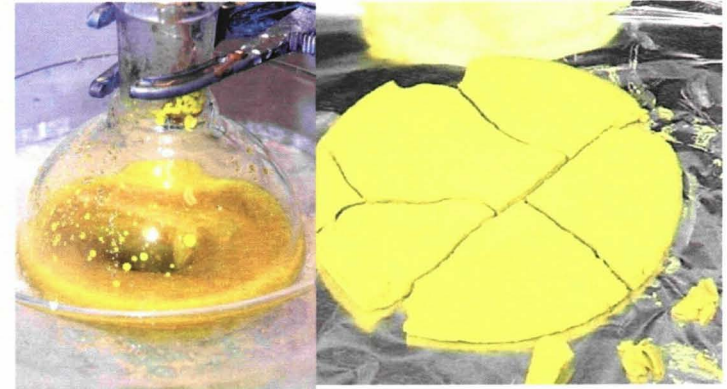
*Self healing occurring in repair film in seconds after damage from left to right*



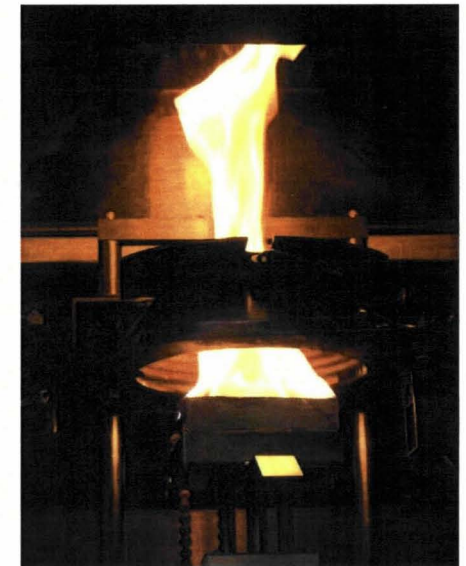


# Fire and Polymers

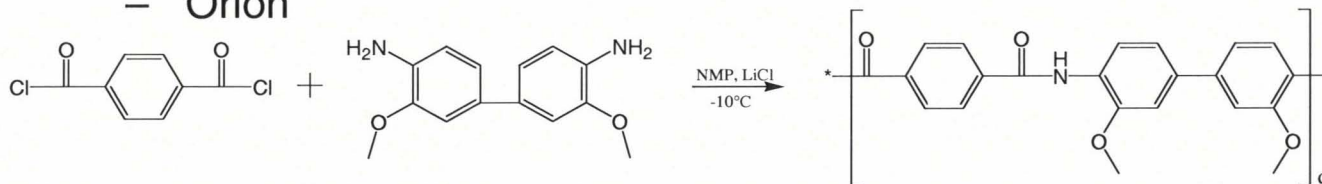
- Flame retardant strategies
  - Polymethoxyamide derivatives for high temperature engineering polymers (patent issued)
  - Carbon nanotube synergistic FR properties
  - Polyhedral Oligomeric Silsesquioxanes (POSS) FR properties



- Fire risk consultation
  - Wire insulation
  - Thermal insulation
  - Ablator



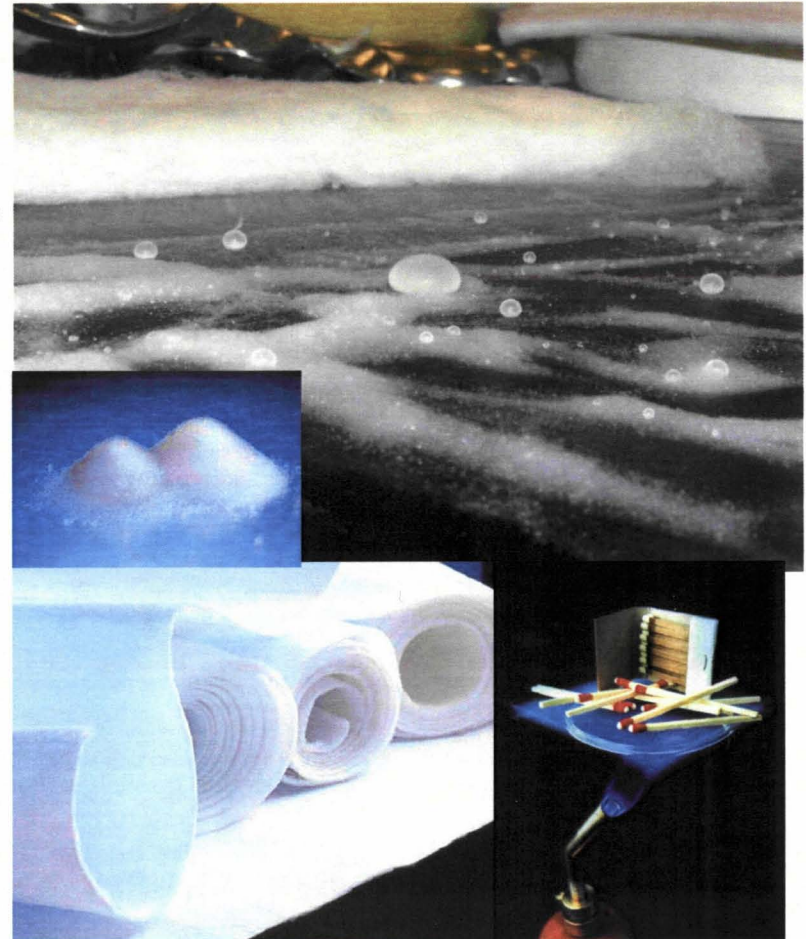
- Fire standards and risk
  - Ares I
  - Ares V
  - Orion





# Aerogel Technology

- Aerogel materials are generally silica based, light weight materials, fully breathable, and treated to be super-hydrophobic.
- Aerogel granules are free flowing, fills small cavities, does not compact, no preconditioning required, and can be molded or formed using binders.
- Aerogel granules (Nanogel®) by Cabot Corp.:
  - 90% porous with a mean pore diameter of 20 nm.
  - Bead bulk density  $\approx 80 \text{ kg/m}^3$  (5 lbs/ft<sup>3</sup>).
  - Individual beads are fragile (shear), but have high elastic compression of over 50% with no damage.
  - k-value  $\approx 18 \text{ mW/m-K}$  @ 25°C and 760 torr.
  - [www.cabot-corp.com/nanogel](http://www.cabot-corp.com/nanogel)
- Aerogel blanket (Spaceloft®) manufactured by Aspen Aerogels:
  - Bulk density 6 to 8 lbs/ft<sup>3</sup>.
  - k-value  $\approx 12 \text{ mW/m-K}$  @ 38°C and 760 torr.
  - Use temperature range -273°C to 650°C (-459°F to 1200°F).
  - <http://www.aerogel.com/>





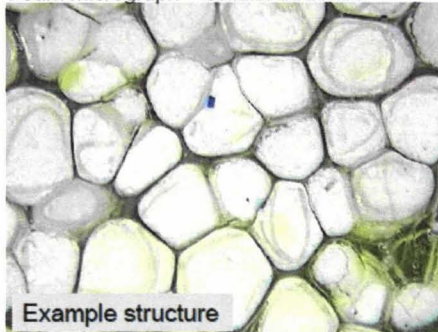


# Aerogel Composites

## AeroFoam - polyimide foam + aerogel

Enhanced thermal and vibration damping performance. Structural integrity to the aerogel and cryogen storage capabilities.

Foam micrograph – cellular structure

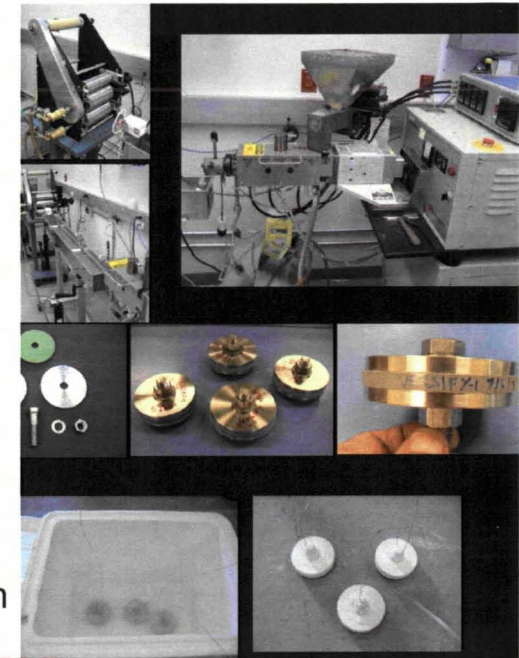


## AeroPlastic – thermoplastic + aerogel

Extruded process, composite reducing heat transfer by 40-60%. Cryogen storage and transfer applications such as piping and seal.



AeroPlastic demo testing on cryo-piping system



Fiber/Textile+ aerogel structural composites





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# Aerogels for Oil Remediation

Superdome Cap Structure - Anibal Karban (NE-M)

Aerogels for Environmental Remediation

## Why Aerogel?

Lightest solid known ( $80 \text{ kg/m}^3$ ) – floats on water

High oil absorbency – 250 gallons/ $\text{m}^3$

Super-hydrophobic material (repels water)

Environmentally friendly – inert amorphous silica

Stable – long consistent service life, no UV degradation

Commercially manufactured in bulk quantities

Aerogel incorporated into mesh bag, blanket, or filled boon for  
easy deployment





# KSC's Solution

- Cabot Nanogel: Commercial small business collaboration through existing SAA with NASA KSC
- \$2800 per  $m^3$  = 250 gallons oil
- 60,000 barrels of oil released per day
- 480,000 miles of booms to be deployed off Florida coast
- Inventory of aerogel for oil recovery

Domestic inventory	Europe inventory	Sustainable capacity per month
100 $m^3$	2000 $m^3$	600 $m^3$
25,000 gallons equivalency	500,000 gallons equivalency	150,000 gallons equivalency





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# Chromochromic Hydrogen Sensors

In collaboration with UCF

**A patent-pending irreversible color changing H<sub>2</sub> gas sensor was developed at KSC in partnership with UCF and ASRC.**















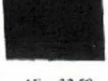



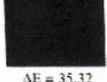
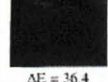
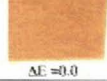
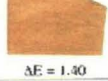
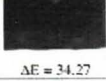
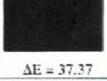
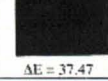
**Changes color from a light tan to black in the presence of H<sub>2</sub>.**

**Can be manufactured into any polymer part, tape, fiber, or fabric material for unlimited potential uses.**

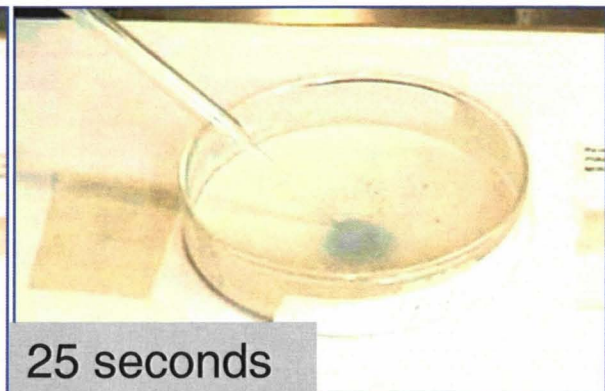
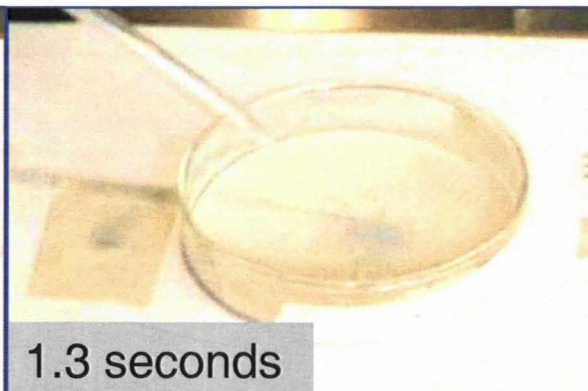
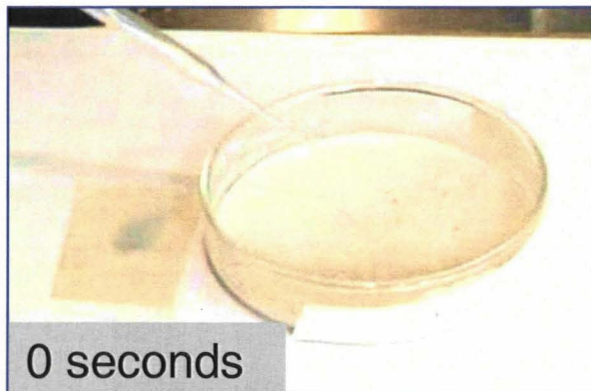
- Paint, Gloves, Coveralls, PPE

**Operates under ambient and cryogenic temperatures.**

## Irreversible Sensor

% H <sub>2</sub>	T = 0	T = 1	T = 2	T = 3	T = 5
1%	 ΔE = 0.0	 ΔE = 1.54	 ΔE = 0.97	 ΔE = 13.48	 ΔE = 24.93
5%	 ΔE = 0.0	 ΔE = 1.09	 ΔE = 2.08	 ΔE = 16.99	 ΔE = 28.98
10%	 ΔE = 0.0	 ΔE = 0.75	 ΔE = 10.45	 ΔE = 28.39	 ΔE = 32.50
50%	 ΔE = 0.0	 ΔE = 0.34	 ΔE = 31.77	 ΔE = 35.32	 ΔE = 36.4
100%	 ΔE = 0.0	 ΔE = 1.40	 ΔE = 34.27	 ΔE = 37.37	 ΔE = 37.47

## Reversible Sensor





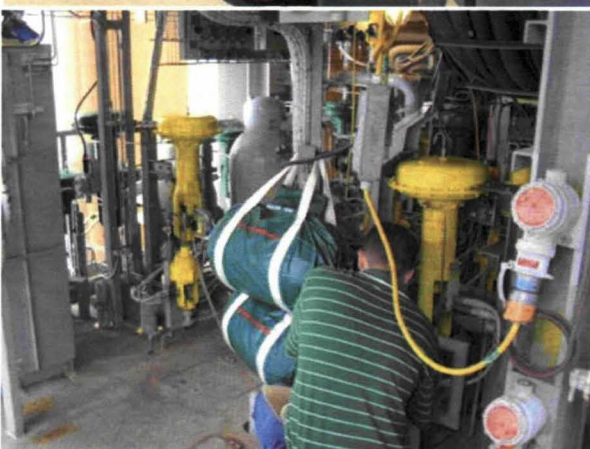


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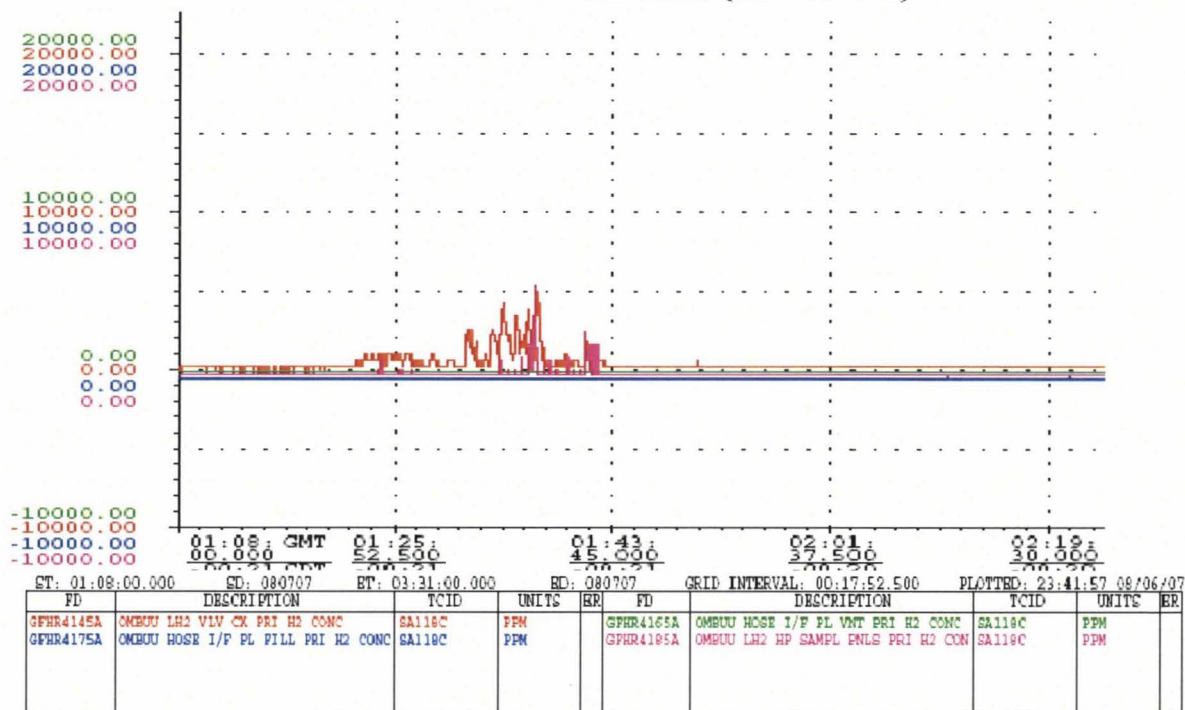
# Chromochromic Hydrogen Sensors

## STS-129 Transfer Line



## LPA OMBUU Deployment for STS 117, 118, 120, 122, 123

STS-118 LOAD-1 HP SKID (H2CONC-B)



TITLE: STS-118 LOAD-1 HP SKID (H2CONC-B)

OPT: -INDIV RANGE





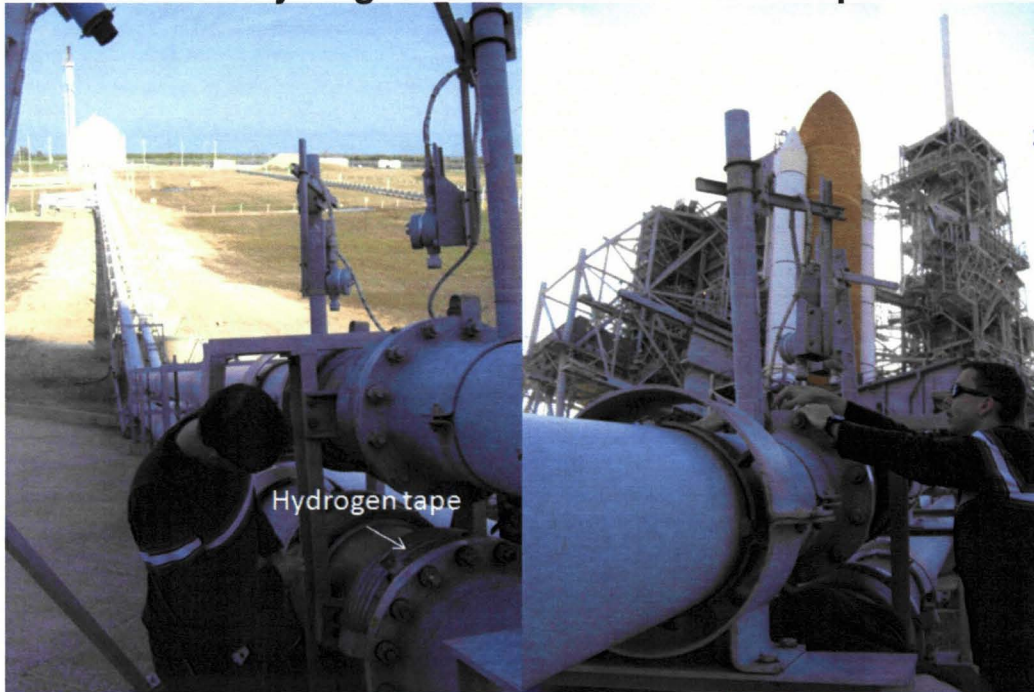


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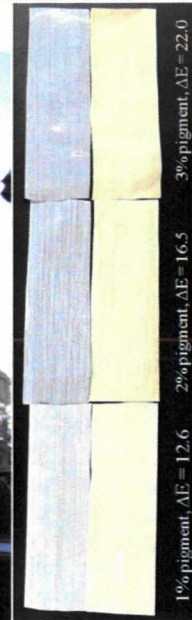
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# STS-130 and 131 Operations

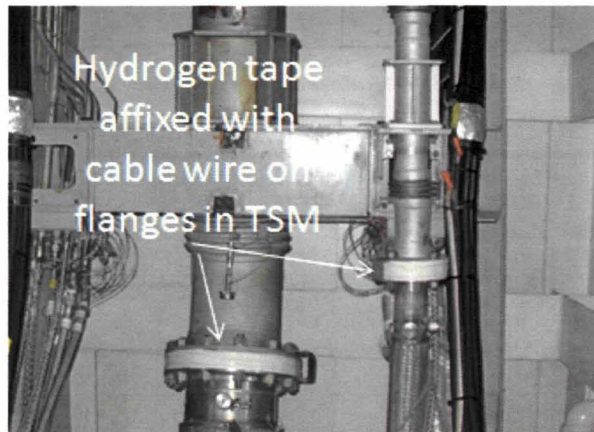
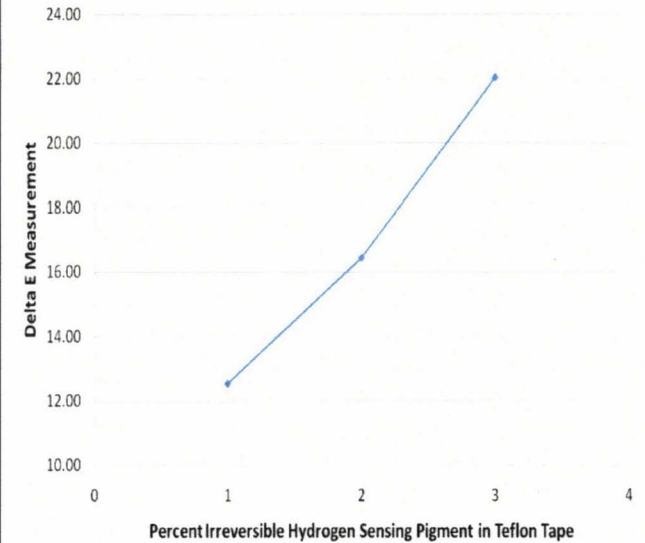
### Hydrogen vent lines on Pad A slope.



Hydrogen tape



### Comparison of Delta E Values for Different Pigment Loads



Hydrogen tape affixed with cable wire on flanges in TSM

### TSM for STS-131

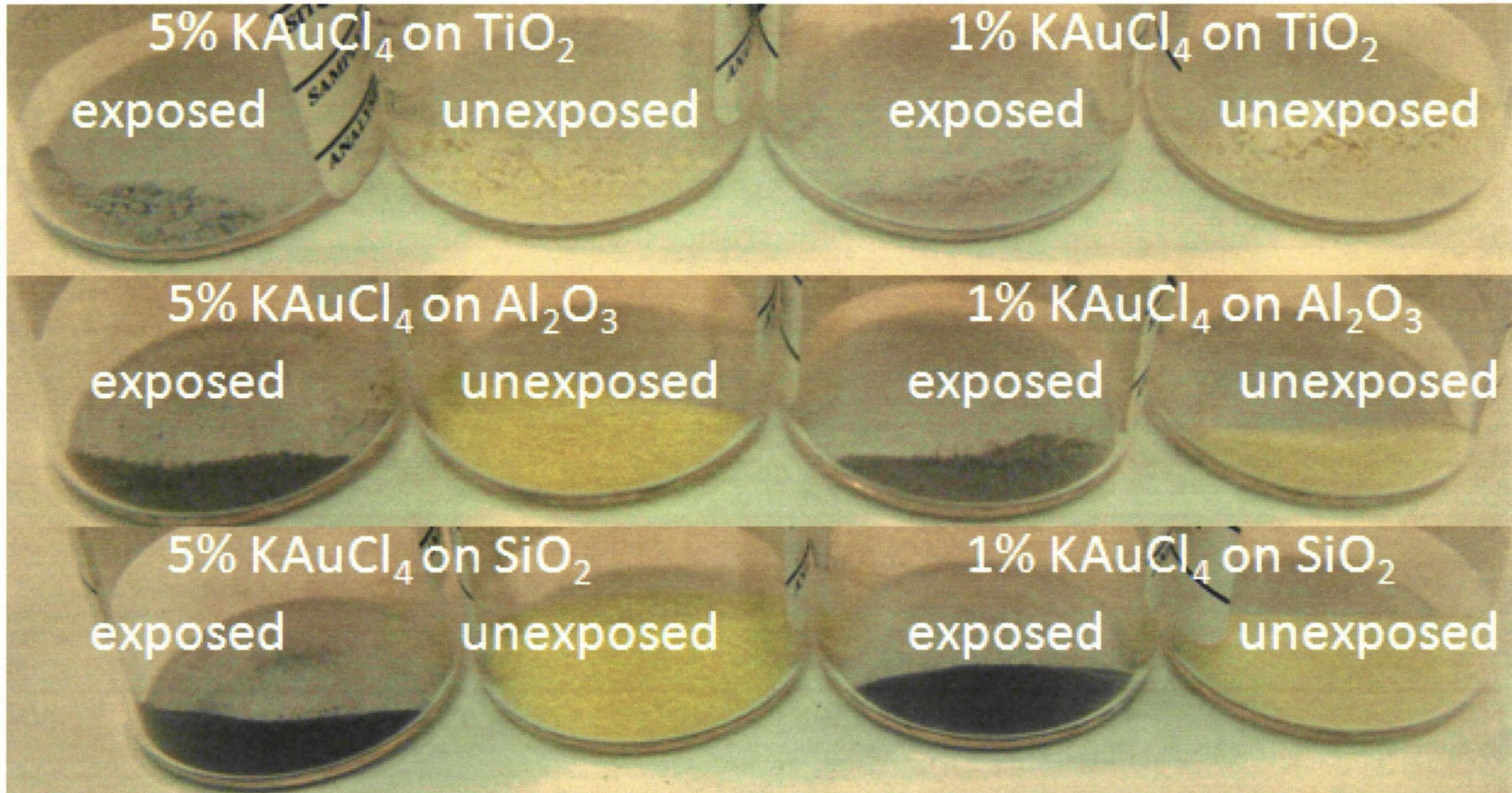


### STS-130 H2 Pressure Flange A3362





# HyperPigment



Comparison of 1% and 5% by weight KAuCl<sub>4</sub> on different substrates.

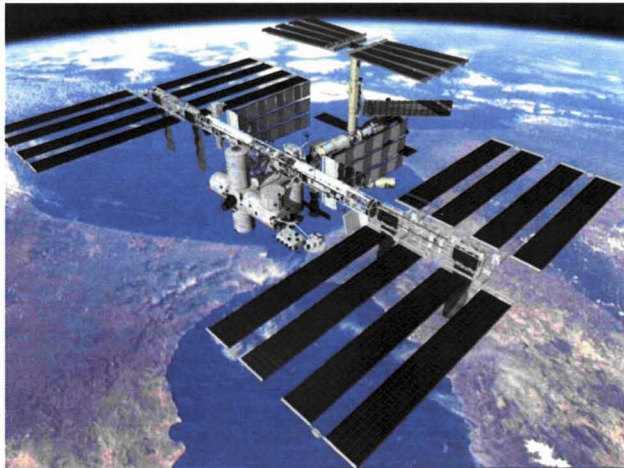




# Antimicrobial Materials

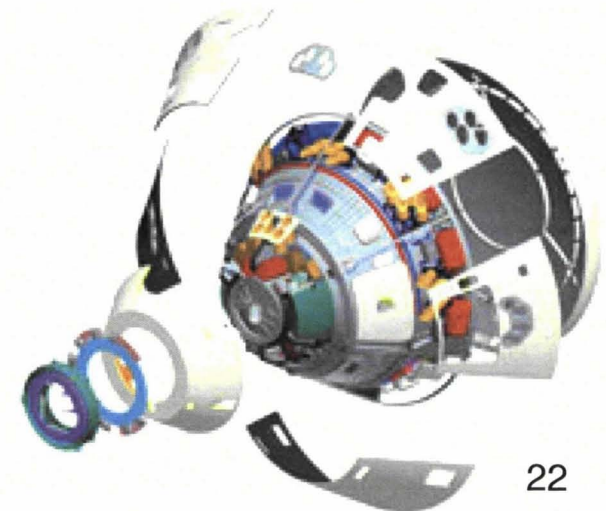
## Shuttle Potable Water

(4) 170 lb Inconel bellow tanks  
Iodine (3-4 mg/L)



## ISS Potable Water

Stainless tubing  
Ionic Silver Biocide



## Orion Potable Water

(5) Inconel 718 Tanks (14.3 gal)  
Miles of Titanium water lines

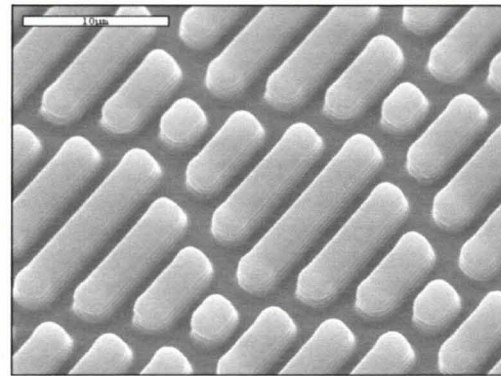
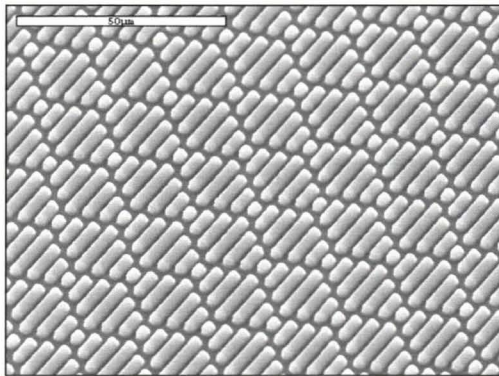




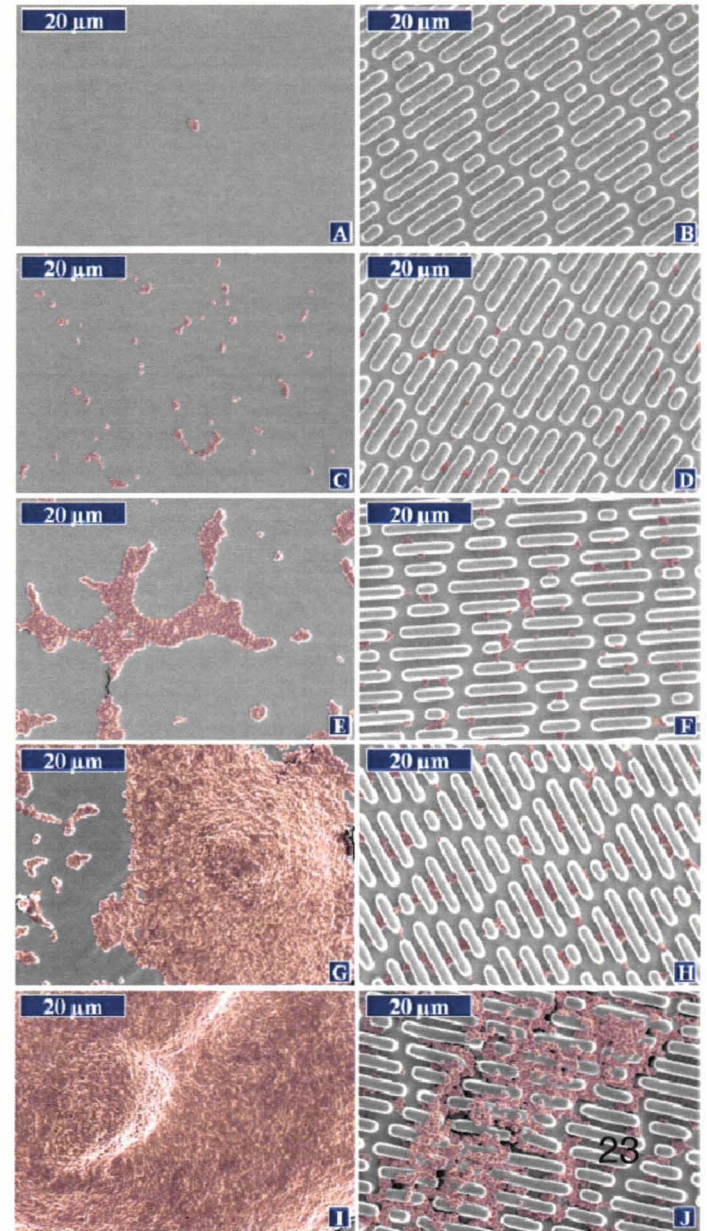
# Antimicrobial Materials

In collaboration with Sharklet Technologies and UF

## Surface Morphology and Surface Chemistry



- Efficacy studies after 21 days decreases biofilm formation
- Easy to imprint during manufacture of polymer articles through a coining process
- Can be used in conjunction with antimicrobial polymers







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# Antimicrobial Materials

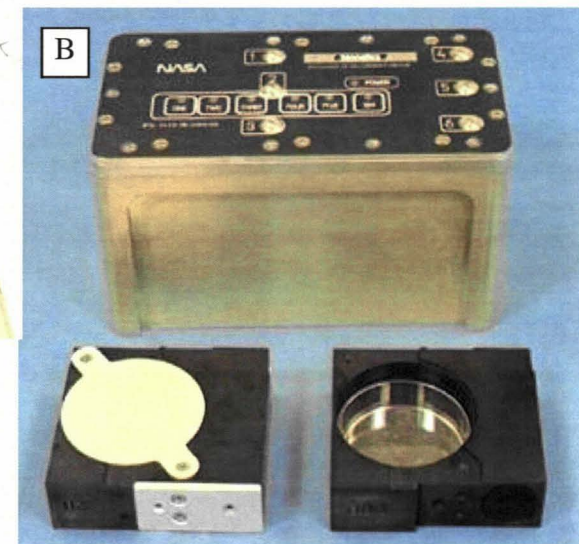
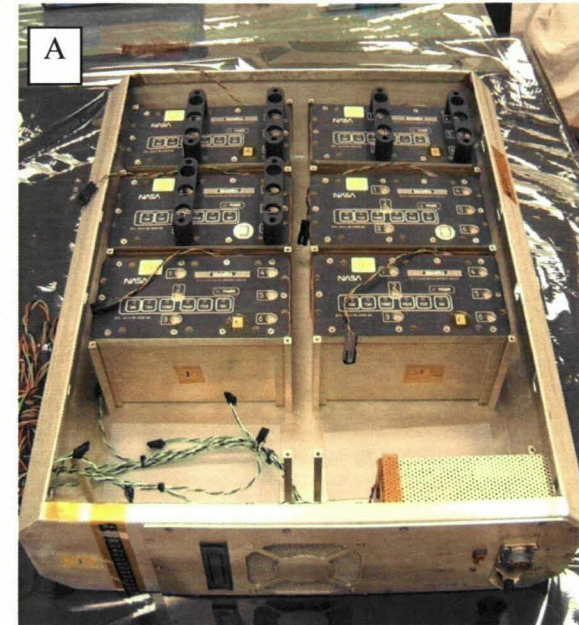
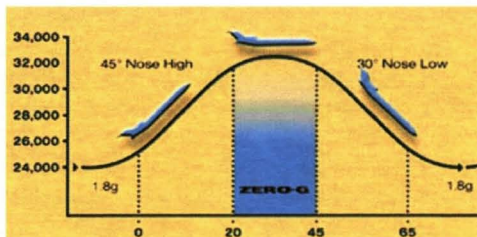
In collaboration with University of Alberta and Sharklet

## Microgravity Flight Experiments

### BIOLOGICAL ANALYSIS

Confirm efficacy of *Pseudomonas fluorescens* bacteria species with Sharklet® topography coupons and different surface treatments

- How well does it work in  $\mu$ G and lunar G compared to 1G?



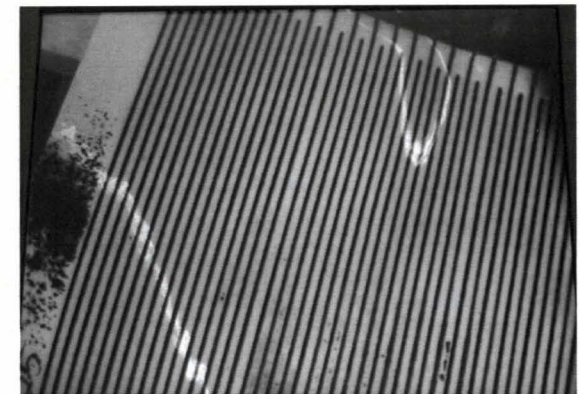
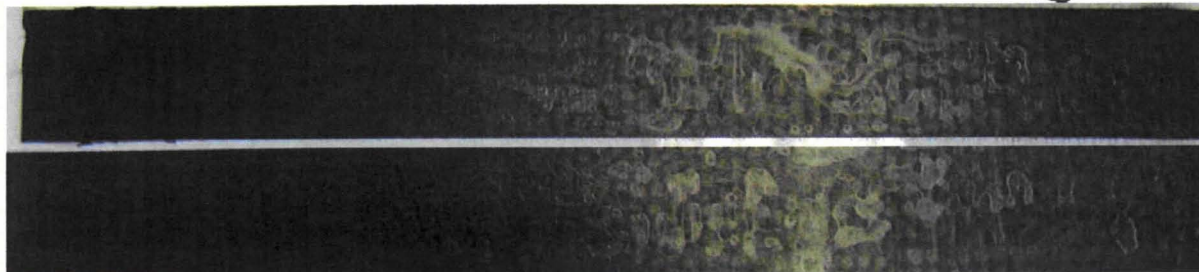
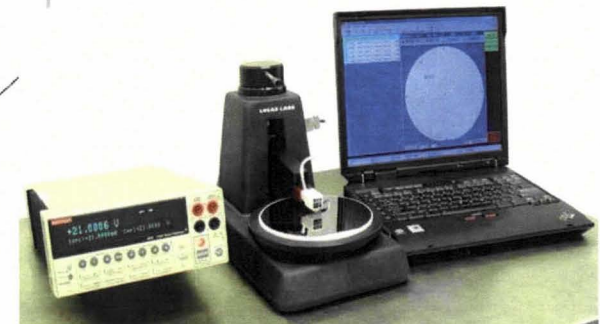
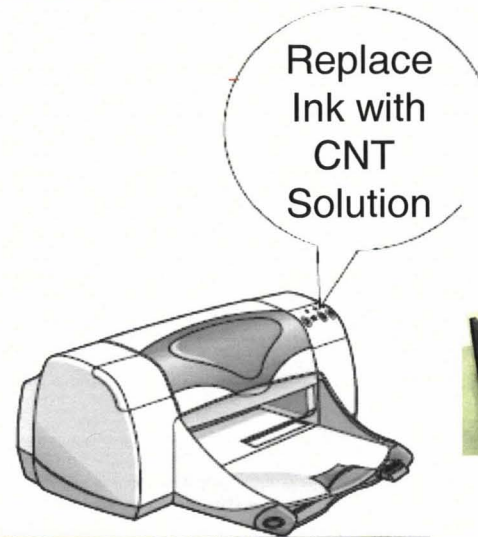
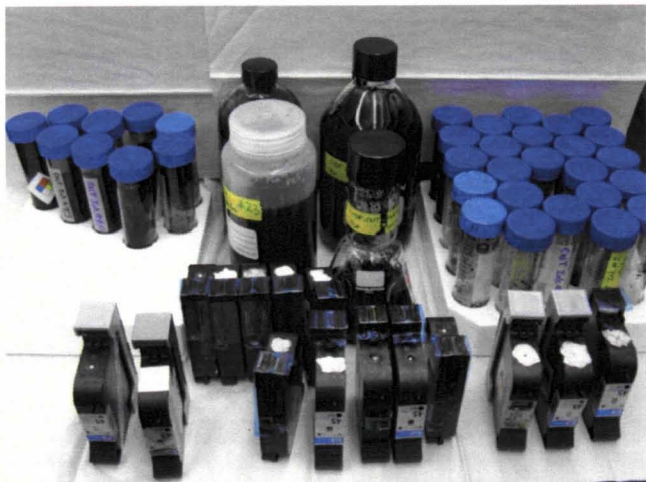




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# Conductive Inks Formulations for Multiple Applications



- Uses standard Inkjet printing technology
- 4pt probe used to measure resistivity or conductivity
- Formulations are solution blends; including carbon nanotubes, polymers and nanometallics. Patent application in work.
- Printing on multiple types of flat surfaces, including fabrics for dust screen technology
- Printing on curved surfaces for detection in process





# Testing and Processing Equipment

- Fire Testing
  - Cone Calorimeter
  - Oxygen Index\*\*
  - UL94 fire test
  - NASA Std 6001 fire test
  - Radiant Panel\*
  - NBS Smoke Chamber\*
  - Two foot tunnel\*
  - Glow wire ignition\*
- Cryogenic Materials Testing
  - Cryogenic moisture uptake (CMU)\*\*
  - Brittleness/Impact test \*\*
  - Liquid helium cold finger test\*\*
  - Single Pin-Socket Krytox Contamination Electrical Characterization under Cryogenic Conditions\*\*
- Specialty Test Equipment
- Cellular Solid Analysis
  - Pycnometer (closed/open cell)\*\*
  - Surface area measurement\*\*
- Thermal Analysis
  - Thermogravimetric analysis (TGA)
  - Differential Scanning Calorimetry (DSC)
  - Dynamic Mechanical Analysis (DMA)
- Physical Testing
  - Tensile Test
  - Compressive Test
  - Pull/Peel Test
- Electrical Testing
  - 4-point probe
  - Surface /Volume resistance
- Polymer Processing capabilities
  - Extrusion
  - Injection molder
  - Fiber spinning equipment
  - Melt, ball, and high intensity mixers

\*in collaboration with Cryogenics Test Laboratory

\*\*in collaboration with Florida Tech





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# Acknowledgements

## Polymer Science and Technology

*Dr. Martha Williams, Lead*  
*Dr. LaNetra Tate\**  
*Trent Smith\**  
*Anne Caraccio*

## ASRC Applied Chemistry Lab

*Dr. Pedro Medelius*  
*Dr. Tracy Gibson*  
*Dr. Mary Whitten*  
*Dr. Scott Jolley*  
*Dr. Robert Devor*  
*Dr. Steve Trigwell*  
*Sarah Snyder*  
*Lilly Fitzpatrick*

## Cryogenics Lab

*James Fesmire*  
*Jared Sass*  
*Wes Johnson*  
*Dr. Barry Meneghelli*

## Environmental Life Support

*Dr. Ray Wheeler*  
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# QUESTIONS?

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