# Fiber Optic Raman Sensor to Monitor Concentration Ratio of Nitrogen and Oxygen in a Cryogenic Mixture

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A spontaneous Raman scattering optical fiber sensor is developed for a specific need of NASA/SSC for long-term detection and monitoring of the quality of liquid oxygen (LOX) in the delivery line during ground testing of rocket engines. The sensor performance was tested in the laboratory and with different excitation light sources. To evaluate the sensor performance with different excitation light sources for the LOX quality application, we have used the various mixtures of liquid oxygen and liquid nitrogen as samples. The study of the sensor performance shows that this sensor offers a great deal of flexibility and provides a cost effective solution for the application. However, an improved system response time is needed for the real-time, quantitative monitoring of the quality of cryogenic fluids in harsh enviornonment.

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# 1. Introduction

The environment within a rocket engine is particularly severe with very high temperatures and pressures along with extremely rapid fluid and gas flows.[1] The escalating use of liquid phase compounds like oxygen, hydrogen, etc., as cryogenic fuels and their quality monitoring led to the development of a variety of sensor systems. Apart from quality control, sensors are required for quantitative determination of constituent species in a sample mixture, especially in aircraft fuel tanks where a disproportionate concentration of one sample against another can lead to an explosion.[2] Fuel vapors mixed with air containing a high level of oxygen are extremely flammable and raise serious questions on the survivability of aircraft. Implementation of sensors is not only restricted to realizing ignition susceptibility but also in monitoring the health of rocket engines, especially where an optimum level of cryogenic fuel purity is highly desirable. Several types of sensors like electronic sensors, capacitance based, electrochemical sensors, etc., are commercially available but lose their relevance in the supercritical environment of rockets and aircraft, where high temperature and pressure severely inhibit their optimal performance.[3-5] In the past, researchers have developed computational-analytical techniques for quality maintenance in cryogenic fuel like liquid methane at the entrance to the article tank.[6] Though interesting, this method requires sampling of fuel mixture and thus fails to provide real-time measurement of impurities in fuel. Application of fiber optic technology to rocket engine health monitoring offers an alternative approach towards realizing various parameters of interest and addresses several measurement issues associated with extremely harsh environments. Today, optical fiber sensing technology has evolved to the point where one can measure nearly all the physical parameters of interest and a very large number of chemical species as well.[7]

This paper reports an all-optical cryogenic fluid sensor based on Raman spectroscopy for concentration ratio monitoring of liquid nitrogen ($LN_2$) in liquid oxygen (LOX). The present work is motivated by a specific need of NASA Stennis Space Center (SSC) for long-term detection and monitoring of the quality of liquid oxygen in the delivery line during the testing of rocket engines.[8] An attractive design feature of a fiber optic sensor lies in real-time, *in-situ* qualitative as well as quantitative determination of LOX and/or $LN_2$ from their mixture sample on the basis of spectrum analysis for various sample compositions. Prior to this study, efforts have been made to develop a gas sensor for monitoring gaseous phase mixtures of nitrogen and oxygen.[9] Online characterization of cryogenic fluid mixtures like liquid nitrogen and liquid oxygen by optical diagnostics is an area that has not yet been explored to the best of our knowledge. Various issues involved in monitoring super critical fluid mixtures, like sample vaporization/condensation, system optimization, signal/noise ratio and other safety parameters, have been well addressed and provide a novel vision for encountering challenges involved in processing/controlling cryogenic fuels.

Certain distinct advantages of optical fiber sensors over conventional sensors, like remote sensing, multiplexing and distributed sensing, immunity towards EM signals, etc., make them more advantageous in real field applications. On the other hand, laser Raman spectroscopy (LRS) has been known for years as a relatively simple analytical method for identification of molecules in gases, liquids and solids by scattering of laser light. Raman spectroscopy is based on the Raman effect that results from energy exchange between incident photons and the scattered molecules.[10-21] The Raman scattered light occurs at frequencies that are shifted from the incident laser light by the change in vibrational, rotational or electronic energies of a molecule. By measuring the frequency and intensity of inelastically scattered light from the

3

sample, the molecules in the sample can be qualitatively and quantitatively measured. In order to explore various possibilities of employing a low cost, portable optical sensor based on LRS in the harsh environment of a rocket engine, we evaluated three excitation light sources for the sensor. Two frequency doubled 532-nm continuous wavelength (CW) Nd:YAG lasers (TEM$_{00}$ and Multimode) were first used as excitation light source with a fiber optic state-of-the-art miniaturized Raman probe. Then a diode laser operating at 670-nm was used as the excitation source in an attempt towards configuring a cost effective optical sensor, which caters to the needs of various national laboratories and cryogenic industries. The sensor performances with different excitation sources in the measurements of liquid nitrogen in liquid oxygen are compared and the results are reported in this paper.

## 2. Experimental Details

The main focus of this research effort was to develop a real-time optical fiber sensor that employs a CW 532-nm laser as an excitation light source for monitoring the concentration of LN$_2$ in LOX/LN$_2$ liquid mixture. In the process of developing a miniaturized and cost effective optical fiber Raman sensor, we have also evaluated a diode laser as an excitation light source. The details of the excitation sources used in this study are given in Table 1.

In the initial experiment, an optical fiber Raman sensor is set up using a frequency doubled 532-nm continuous wavelength (CW) Nd:YAG laser (Coherent DPSS 532) as the excitation light source, with maximum output power at 330 mW. The laser output energy was attenuated through proper neutral density filters in order to protect the detector from potential damage by the high intensity light. The experimental set-up is shown in Fig. 1(a). The sensor configuration employed a modified In-Photonics fiber optic state-of-the-art miniaturized Raman

4

probe. This probe consisted of just two parallel optical fibers; one (90-µm core diameter) was guiding the launched light to the liquid sample mixture enclosed within the dewar, whereas the other (200-µm core diameter) was collecting the emitted Raman signal and feeding it to a USB spectrometer (USB 2000, Ocean Optics Inc.). This spectrometer had 1200 1/mm grating along with a 2048-element charge-coupled device (CCD) attached to the exit of the spectrograph, and was interfaced with a computer via a USB port. Probe performance in terms of Raman signal response and background noise level was improved by incorporating a miniaturized lens at the tip of the probe with a working distance of about 30-mm. Experiments were carried out for the multiple concentration (weight) ratios of the liquid $N_2/O_2$ and mixed to ensure homogeneity within the mixture. For each sample, a 3-s sampling time was used to minimize errors due to rapid phase transforming, i.e., liquid to gaseous phase nature of the sample constituents. Owing to the continuous vaporization of the supercritical liquid mixture, the greatest challenge of the experiment was to minimize the vaporization loss, and also the accuracy with which the weight ratio of the sample constituents were determined during the preparation of mixtures. In the initial experiment, a cap with a hole was mounted on the dewar. Though it allowed a clear passage for the laser light to be focused onto the liquid $N_2/O_2$ mixture from the probe, it could not prevent the vapors from escaping and condensing onto the tip of the fiber probe. As a result, attenuation in the laser power was observed that severely hampered the liquid $N_2/O_2$ spectrum, thereby reducing the signal-to-noise ratio of the Raman signal. To overcome this problem, two identical quartz windows (1/4") were introduced into the dewar cap hole to provide optical access as well as to prevent vapors from reaching the tip of the probe. A dryer was also installed in the vicinity of the fiber probe to eliminate the remote possibility of vapor deposition onto the tip of the probe and thus to achieve long-term stable operation of the Raman sensor.

To study the effects of laser property to Raman signal, the Raman data from a narrow band 532-nm laser (Coherent DPSS 532) and a high-power and inexpensive broadband 532-nm laser (Milles Griot GHS 309) were compared. The experimental configuration was kept the same as we recorded these data. A neutral density filter was applied to the broadband 532-nm laser output to attenuate the laser power so that the laser energy on the sample mixture was about ~ 140 mW from both lasers.

Later we designed a prototype Raman sensor based on a 700 mW power, CW diode laser (Power Technology Inc.), operating at 670-nm, as an excitation light source and a Y-shaped reflection/backscattering probe (R200-REF, Ocean Optics Inc.). Figure 1(b) shows the schematic diagram of this system. This Y-shaped probe consists of seven optical fibers, each having 200 μm core diameters and 0.22 as numerical aperture with one launching fiber and six surrounding collecting fibers. Uniqueness of the prototype sensor lies in its compact design configuration that included carefully aligned optical components, viz., laser diode, filter holder, cut-off filter (725-nm) and a compact OOI spectrometer (USB 2000). Raman Spectrums of different concentration ratio mixtures of LOX and LN$_2$ were recorded with the prototype Raman sensor in the same manner as described earlier.

### 3. Results and Discussions

The present study was primarily focused on quantitative as well as qualitative analysis of component ratio of liquid N$_2$/O$_2$ mixtures at varied levels of their weight ratio. The mixture sample was prepared by adding liquid O$_2$ in liquid N$_2$, owing to the fact that the molecular weight of oxygen is higher than nitrogen, and therefore suppressing the rate of vaporization of liquid nitrogen.[22] Raman Spectrum recorded for 40% of LOX and 60% of LN$_2$ mixture is shown in the Fig.2 (a). Raman bands of LOX (~ 580-nm) and LN$_2$ (~ 607-nm), corresponding Raman shift for

6

$O_2$ at 1556.4 cm$^{-1}$ and $N_2$ at 2330.7cm$^{-1}$, were marked on the spectrum.[23] The concentration of $LN_2$ was varied from 1% to 100% against LOX at normal atmospheric pressure and temperature, while maintaining a total mixture weight of 100 gram. The Raman band intensity corresponding to $LN_2$ was increasing as the concentration of $LN_2$ was increased against the LOX in the sample mixture as shown in Fig. 2(b). A narrow band Coherent CW laser, operating at 532-nm, was used in this experiment.

The observed Raman bands of LOX and $LN_2$ were critically analyzed in terms of spectral band shape, S/N ratio, peak intensity, etc. Peak intensity corresponding to $LN_2$/LOX, at various concentrations of their weight ratio, was estimated for obtaining a calibration curve as shown in Fig. 3. Sensor response exhibits a linear trend within the concentration limit of 1 % to 60 % of $LN_2$ and observed to be saturating in the concentration range of 60% to 80% $LN_2$ in a sample mixture. Slower response of the sensor towards a higher concentration ratio of $LN_2$ attributes to the fact that the vaporization rate of liquid nitrogen increases manifolds resulting in the formation of a cloud of semi-liquid gaseous phase mixture; and thereby affecting the sensor performance. We further calculated the spectral band area ratios of Raman $LN_2$ and LOX peaks. This calculation was performed for all the sets of spectra corresponding to a particular concentration ratio of LOX/$LN_2$, and averaged to obtain a single data point. As a result, error due to the fluctuation in the peak intensity was minimized. Finally, a calibration curve was drawn between the averaged spectral band area ratios of Raman $LN_2$ and LOX peaks against the corresponding weight ratios as shown in Fig. 4. The following conclusions have been drawn from the experimental results obtained above.

First, the sensor system is capable of successfully monitoring the $LN_2$/LOX level in a $LN_2$/LOX liquid mixture. Second, the Raman peak area follows the same characteristics as the

intensity ratio curve. For example, both intensity and area ratio curves exhibit a linear trend within a low concentration range of $LN_2$ which tend to saturate above 60% $LN_2$ concentration. This implies that the sensor has high detection sensitivity in low concentration range and can measure $LN_2$ percentages as low as 1% in a sample mixture. For a typical sample mixture of 60% $LN_2$ and 40% LOX, Limit of Detection (L.O.D.) of sensor was calculated as ~1%. Also, intensity/area ratio data points are least fluctuating as reflected by the error bars on the calibration curves which are barely visible in Fig. 3 and Fig. 4. This signifies not only the long-term stable operation of the sensor but also the reproducibility in the sensor response towards analyte concentration. Although the calibration curve obtained by the area ratio method involved statistical averaging of multiple sets of spectra, it did not provide better sensor response. The linear calibration data for $LN_2$/LOX ratio indicates that the fiber optic Raman sensor has a great potential for qualitative as well as quantitative monitoring of sample constituent in supercritical $LN_2$/LOX liquid mixtures.

To develop a less expensive sensor for this application, the feasibility of using an inexpensive, high power, broadband laser (Milles Griot) operating at 532-nm as an excitation light source was tested. Using the same laser energy as with Coherent laser (~140 mW), and same experimental setup, spectra were recorded for various concentration ratios of LOX and $LN_2$ and compared with the 532-nm Coherent laser used in the previous case. Comparative analysis of spectra recorded with two lasers suggest that in both the cases, spectra share common features and depending upon the requirement, either of these could be employed as excitation light sources. The laser line-width of a Milles Griot laser(~0.18nm) is higher than that of a Coherent laser(~0.00047nm) and showed a slightly higher spectral background in comparison with a

Coherent laser. Also, a Milles Griot laser could be operated in the high range of power which is suitable for trace $LN_2$/LOX monitoring.

To develop a prototype sensor using a 670-nm diode laser, we first estimated the performance of the sensor by considering the laser energy and detector response near 532 and 670 nm. Since the Raman cross-section depends inversely on the fourth power of the laser wavelength $(\lambda)$,[21] the Raman signal is also affected by the excitation wavelength. The various parameters that can affect Raman signal are summarized in Table 2. Estimation based on these parameters (the effects of optical response from other optics at these two wavelengths are not included in this calculation) shows that the prototype sensor should give about 4.4 times less signal than with the system with more expensive 532-nm excitation source. This signal level should be good enough for this type of application. A typical spectrum of a 60% $LN_2$ and 40% LOX sample mixture, recorded with the prototype sensor that included a 670nm laser as an excitation light source, is shown in Fig. 5(a), where the $LN_2$ Raman peak corresponds to ~ 795-nm and the LOX peak corresponds to ~ 749-nm, respectively. The Limit of Detection (L.O.D.) of the prototype sensor for 60%$LN_2$ and 40% LOX was calculated as ~20% which is much higher in comparison to the L.O.D. of the sensor with a 532-nm laser (~1%) for the same concentration ratio mixture. The spectrum recorded with the 670-nm laser diode also shows a strong background that severely affects the Raman signal of LOX/$LN_2$. The overall spectral quality (signal-to-noise) of the prototype sensor using a 670-nm diode laser is not as good as that observed in the case of a 532-nm laser (see Fig. 2a). The performance of the sensor systems with different lasers is actually determined by many factors such as the spectral characteristics of the laser systems, the optics, and also the scattered light. Due to the broadband spectral line width of the 670-nm laser diode, a cut-off filter, centered on 725-nm, is unable to suppress the scattered

incident laser light. The scattered light has contributed to part of the strong background we observed.  The optical fiber can also give rise to its Raman spectral feature and fluorescence background signal that will superimpose on the LOX and $LN_2$ spectrum.  Our current prototype sensor is very compact, and we are unable to effectively remove these background features.  To suppress the background and to improve the signal/noise ratio of the Raman signal, appropriate combinations of various filters will need to be tested and added to the collection optics of the miniaturized prototype sensor. However, we can apply some data processing technique to improve the quality of the data. Figure 5(b) shows the processed Raman LOX and $LN_2$ spectrum of the prototype sensor.  This spectrum was obtained by subtracting the raw data from the background spectrum (the background spectrum was obtained by fitting with some polynomial functions).  The spectral averaging can also be applied to further improve the signal-to-noise ratio.  Intensity ratios for a $LN_2$ and LOX weight ratios of 1.5 were calculated to be ~1.4 from background subtracted spectrum obtained from the prototype sensor.  The Intensity ratios obtained with a 532-nm laser for the same weight ratios of $LN_2$/LOX is about 1.3.

## 4. Conclusions

This paper presents a consistent optical fiber Raman sensor for cryogenic applications and the evaluation of sensor performance with different excitation light sources.  The sensor performance with two different 532-nm lasers and an Inphotonic probe was employed.  Efficient light collection optics and good signal-to-noise ratio of Raman $LN_2$/LOX signal make these laser-based sensor systems an effective tool for rapid as well as sensitive monitoring of $LN_2$/LOX during the testing of a rocket engine.  To design a low-cost, reliable sensor for this type of application, we have also tested with a compact and an inexpensive diode laser.  Although this miniaturized prototype sensor system yielded a low signal-to-noise ratio of Raman

$LN_2$/LOX signal, we have shown that with an appropriate data processing technique, the extracted intensity ratio of Raman signal of $LN_2$ and LOX are close to that obtained from the more-expensive 532-nm excitation systems. The miniaturized sensor system offers a great deal of flexibility and provides a cost effective solution for measuring the quality as well as quantity of cryogenic fluids. However, due to the relatively poor signal-to-noise ratio and background problem, the performance of the miniaturized sensor system is not suitable for real time monitoring. On the other hand, the system with a high-power 532-nm laser can offer good quality LOX/$LN_2$ spectra with reasonable response time, and therefore it is more suitable for real-time application. Future research will be focussed on improving the performance of the sensor for high cryogenic liquids pressure.

## Acknowledgments

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| Excitation Source | Wavelength (nm) | Laser Line Width (nm) | Max. Power (mW) | Beam Diameter* (mm) | Transverse Mode |
|---|---|---|---|---|---|
| **A**. 532-nm laser | | | | | |
| ❖ Coherent | 532 | 0.00047 | 330 | 0.70 | TEM$_{00}$ |
| ❖ Milles Griot | 532 | 0.18 | 3000 | 0.24 | Multimode |
| **B**.670-nm diode laser Power Technology Inc. | 670 | 0.70 | 700 | 1.5 | Multimode |

**Table 1. The excitation sources used in this study**

*$1/e^2$ at waist

| | 532-nm Laser | 670-nm Diode Laser |
|---|---|---|
| Maximum Laser Power on Sample | 140 mW | 380 mW |
| CCD Detector Efficiency | 2.3 | 0.4 |
| Grating Efficiency | 60% | 72% |
| Relative Raman Cross Section ($\lambda^{-4}$) | 12.48 | 4.96 |
| Overall Relative Raman signal | 4.44 | 1 |

**Table 2**. **Estimation of the Raman Signal from 532-nm and 670-nm laser**

**List of Figure Captions**

Fig. 1. Experimental set-up of optical fiber Raman sensor. (a) InPhotonic Raman Probe based system (b) Reflection probe based prototype Raman system.

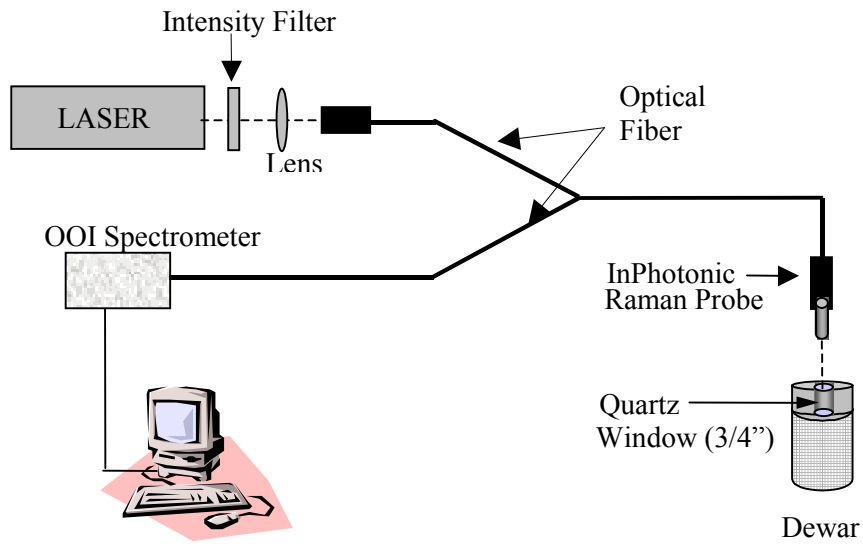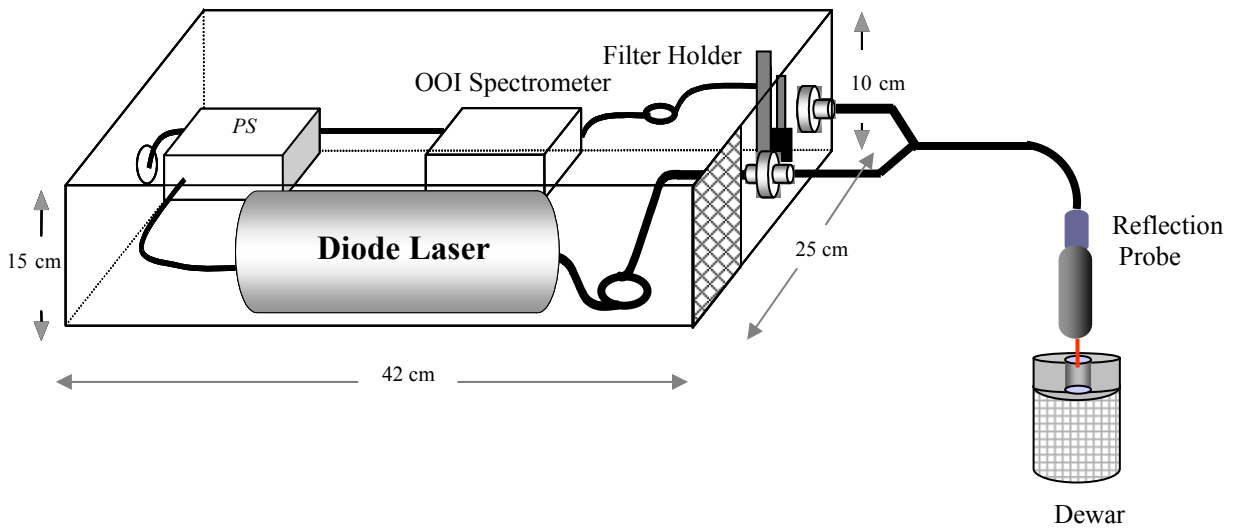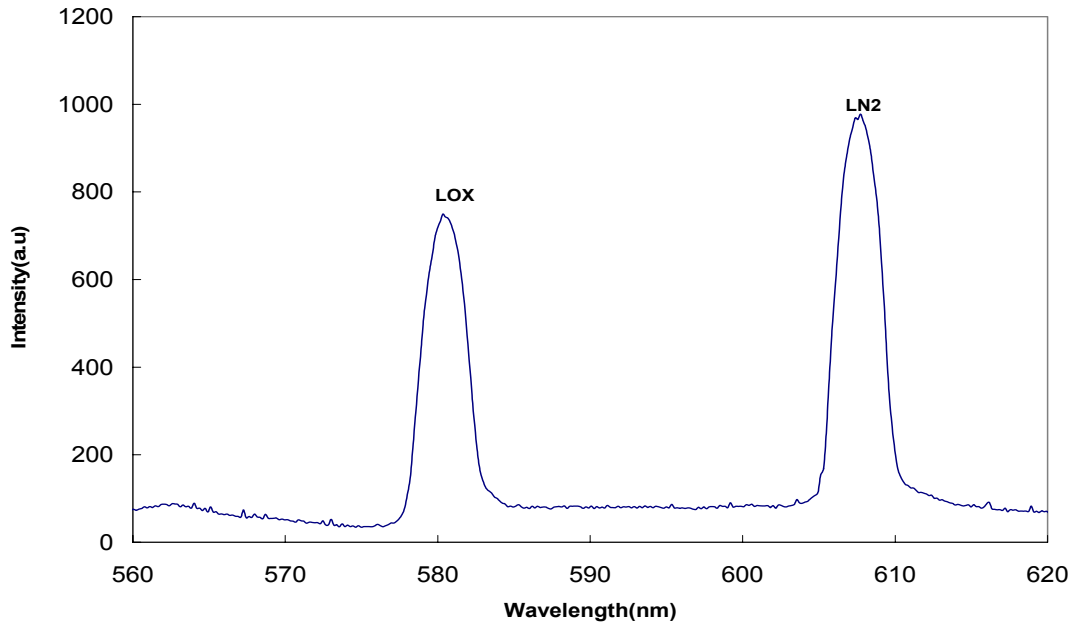Fig. 2. (a) Raman Spectrum of 40% LOX and 60% $LN_2$ liquid mixture. Laser wavelength: 532nm
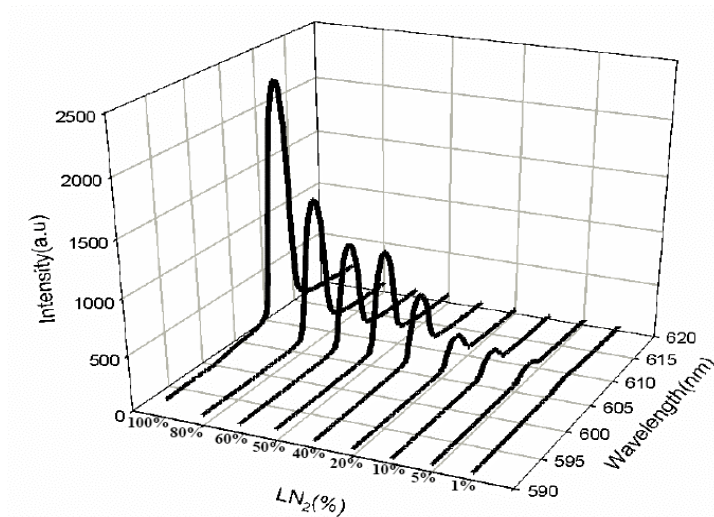
Fig. 2 (b) Sensor response to the liquid $N_2$ /$O_2$ mixture as a function of $LN_2$ concentration.

Fig. 3 Sensor response to the liquid $N_2$ /$O_2$ mixture as a function of $LN_2$/LOX weight ratio.

Fig. 4. The area ratio variation of $LN_2$/$O_2$ against their weight ratios.

Fig. 5 Raman spectrum for 60% $LN_2$ & 40% LOX with 670nm diode laser (a) actual spectrum (b) background subtracted spectrum

(a)



(b)



Fig. 1. Experimental set-up of optical fiber Raman sensor. (a) InPhotonic Raman Probe based system (b) Reflection probe based prototype Raman system.

17

Fig. 2. (a) Raman Spectrum of 40% LOX and 60% $LN_2$ liquid mixture. Laser wavelength: 532nm



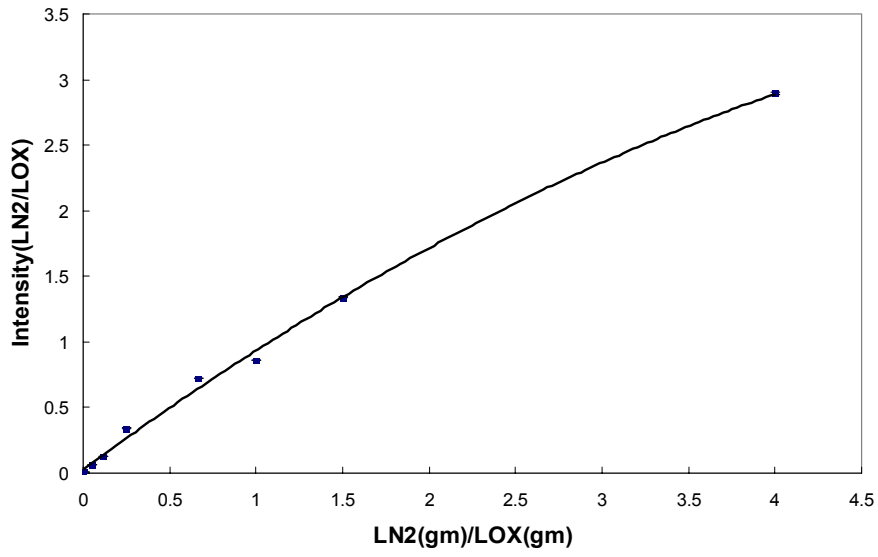Fig. 2 (b) Sensor response to the liquid $N_2$ /$O_2$ mixture as a function of $LN_2$ concentration.

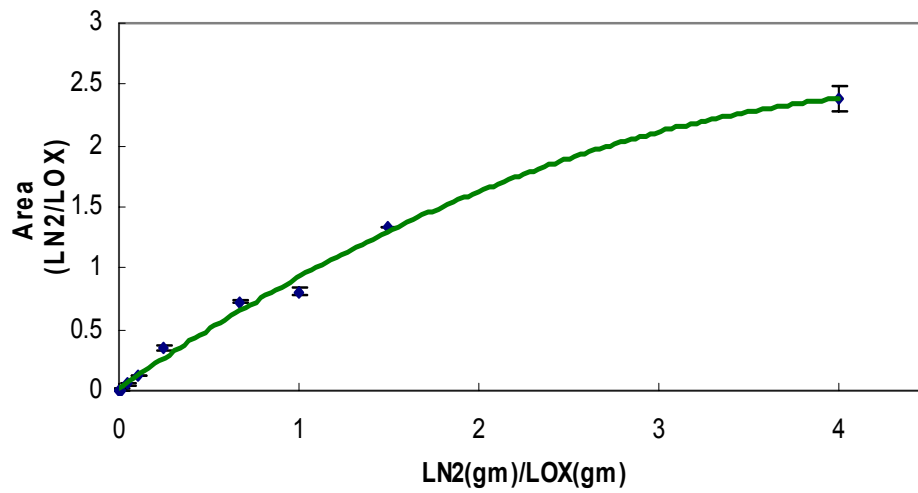Fig.3 Sensor response to the liquid $N_2$ /$O_2$ mixture as a function of $LN_2$/LOX weight ratio.

Fig. 4. The area ratio variation of $LN_2/O_2$ against their weight ratios.
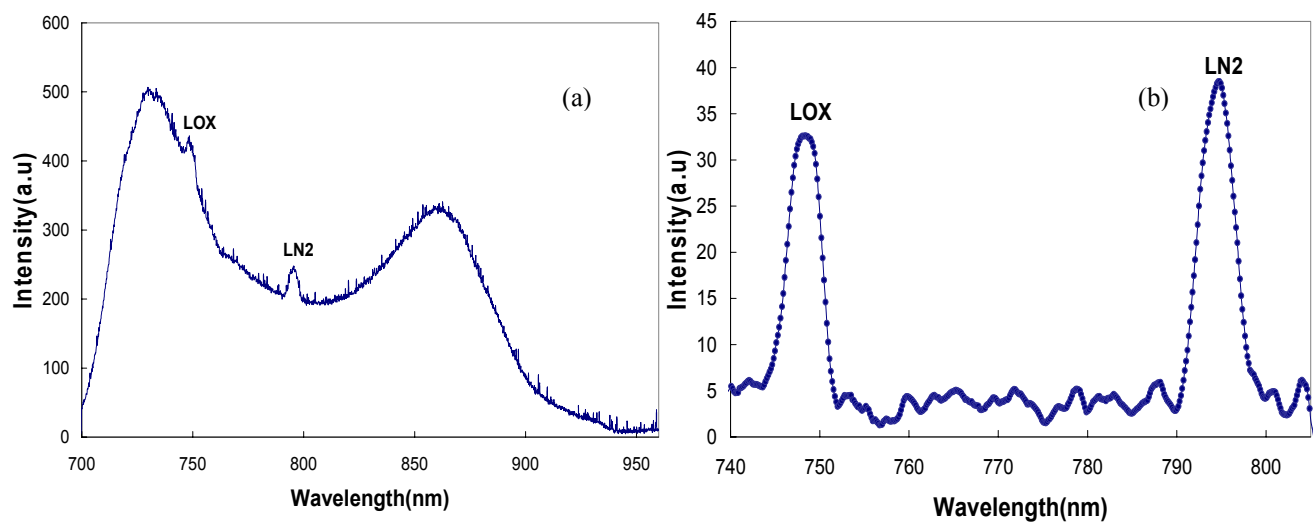
Fig. 5 Raman spectrum for 60% $LN_2$ & 40% LOX with 670nm diode laser (a) actual spectrum

(b) background subtracted spectrum