

## **AEC-NASA TECH BRIEF**



AEC-NASA Tech Briefs describe innovations resulting from the research and development program of the U.S. AEC or from AEC-NASA interagency efforts. They are issued to encourage commercial application. Tech Briefs are published by NASA and may be purchased, at 15 cents each, from the Clearinghouse for Federal Scientific and Technical Information, Springfield, Virginia 22151.

## Study of Fluoride Corrosion of Nickel and Nickel Alloys

In the fluoride volatility process for the recovery of uranium and plutonium from partially spent nuclear reactor fuels, volatile fluorides of uranium, plutonium, and selected fission products are formed by reaction with fluorine. Nickel or nickel alloys such as Monel are extensively used in construction of both laboratory-scale and pilot plant-scale equipment which is to be exposed to fluorine and to corrosive fluorides at elevated temperatures. To make estimates of the expected useful lifetimes of such equipment exposed to fluorine, uranium hexafluoride, and volatile fission product fluorides at high temperatures, a knowledge of the rates of corrosion by these fluorides is required.

A report has been published that contains the results of a survey program 1) to determine the resistance of several commonly used construction materials to volatile fluorides likely to be encountered in the fluoride volatility process and 2) to identify the more promising construction metals.

The metals investigated include: nickel-200, nickel-201, Monel, Inconel, Duranickel-301, HyMu-80, and INOR-8 coupons containing areas of nickel-200 weld, nickel-61 weld, or silver solder. Exposures of these metals for periods up to 30 hours at 500°C were carried out in elemental fluorine, UF<sub>6</sub>, SF<sub>6</sub> and some of the following volatile fission product fluorides, both individually and in the presence of fluorine: GeF<sub>4</sub>, AsF<sub>5</sub>, NbF<sub>5</sub>, SbF<sub>5</sub>, BrF<sub>5</sub>, SeF<sub>6</sub>, MoF<sub>6</sub>, and TeF<sub>6</sub>. Corrosion rates were calculated from both the weight gain of the coupons after exposure and the weight loss of the coupons after descaling in a KNO<sub>3</sub>-NaNO<sub>3</sub> bath at 500°C. All of the exposed coupons were examined microscopically for intergranular penetration of the base metal.

A survey of the unclassified literature on the rates of corrosion of nickel and nickel alloys by fluorine, uranium hexafluoride, and the volatile fission product fluorides is included in the report.

## Notes:

- 1. The findings have been published in "Laboratory Investigations in Support of Fluid-Bed Fluoride Volatility Processes—Part XIV. The Corrosion of Nickel and Nickel Alloys by Fluorine, Uranium Hexafluoride, and Selected Fission Product Fluorides at 500°C," by W. H. Gunther and M. J. Steindler of Argonne National Laboratory. ANL-7241, December 1966. The report is available from the Clearinghouse for Federal Scientific and Technical Information, Springfield, Va. 22151: Price: \$3.00 (microfiche \$0.65).
- 2. This information adds to the data currently available in literature concerning the corrosion of nickel and nickel alloys by fluorides.
- 3. This information may also have application to the chemical industries that handle fluorine products or in metal refining processes for semiconductor materials.
- 4. Inquiries concerning this report may be directed to:

Office of Industrial Cooperation Argonne National Laboratory 9700 South Cass Avenue Argonne, Illinois 60439 Reference: B69-10048

Source: W. H. Gunther, and M. J. Steindler Chemical Engineering Division Argonne National Laboratory (ARG-10224)

(continued overleaf)

This document was prepared under the sponsorship of the Atomic Energy Commission and/or the National Aeronautics and Space Administration. Neither the United States Government nor any person acting on behalf of the United States Government assumes any liability resulting from the use of the information contained in this document, or warrants that the use of any information, apparatus, method, or process disclosed in this document may not infringe privately owned rights.

## Patent status:

Inquiries about obtaining rights for commercial use of this innovation may be made to:

Mr. George H. Lee, Chief Chicago Patent Group U.S. Atomic Energy Commission Chicago Operations Office 9800 South Cass Avenue Argonne, Illinois 60439