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Thallium isotopes as tracers of recycled materials in subduction zones: review and new data for lavas from Tonga-Kermadec and Central America

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18 | **Abstract** - Sediment is actively being subducted in every convergent margin worldwide.
19 | Yet, geochemical data for arc lavas from several subduction zones, such as Northern Tonga and
20 | Costa Rica have revealed either attenuated or limited evidence for sediment in their mantle
21 | source regions. Here we use thallium (Tl) isotopes to trace slab components in lavas from the
22 | Tonga-Kermadec and Central American arcs. In general, both arcs display Tl isotope data that
23 | are most compatible with addition of sediment to the sub-arc mantle from the subducting slab.
24 | This evidence is particular strong in the Tonga-Kermadec arc where pelagic clays dominate the
25 | Tl budget along the entire arc. Contributions from altered oceanic crust as well as the Louisville
26 | Seamount chain that subducts underneath Northern Tonga are not visible in Tl isotopes, which is
27 | likely due to the very high Tl concentrations found in pelagic sediments outboard of the Tonga-
28 | Kermadec arc. Lavas from Central America reveal variable and systematic Tl isotope
29 | compositions along-strike. In particular, lavas from Nicaragua are dominated by contributions
30 | from sediments, whereas Costa Rican samples reveal a significant altered oceanic crust
31 | component with little influence from sediments on thallium isotope composition. The absence of
32 | a sediment signature in Costa Rica corresponds with the Cocos Ridge and the seamount province
33 | subduction, which results in a thinner sediment cover. Furthermore, the subducted sediment is
34 | dominated by carbonates with very low Tl concentrations and, therefore, small amounts of
35 | carbonate sediment added to the mantle wedge does not contribute significantly to the overall Tl
36 | budget.

37 | A review of Tl isotope and concentration data from the Aleutians, Marianas, Tonga-Kermadec
38 | and Central American arcs demonstrate that pelagic sediments are detectable in most arcs,
39 | whereas altered oceanic crust components only become appreciable when sediment Tl
40 | concentrations are very low (e.g. carbonate) or if sediments are no longer a significant

41 component of the subducting slab (e.g. slab melting in Western Aleutians). As such, Tl isotopes
42 is a promising tool to trace sediment subduction although this requires at least some pelagic
43 sediment is present in the subducted sediment package.

44 We suggest that thallium partitioning between the slab and mantle wedge is most likely
45 controlled by retention in phengite or by partitioning into fluids. Residual phengite likely
46 produces high Cs/Tl ratios because Tl should be more compatible in phengite than is Cs,
47 however, this conclusion needs experimental verification. The stability of phengite is lower at
48 higher fluid contents, which results in hyperbolic relationships between Cs/Tl and possible
49 indicators of fluids such as Sr/Nd and Ba/Th. Thus, combined Tl isotopic and elemental
50 systematics not only provide detailed information about the specific slab components that
51 contribute to arc lavas, but also potentially shed light on the mineralogy and physical conditions
52 of subducting slabs.

53

54 **1. Introduction**

55 It is well established that material released from subducting slabs imparts distinct chemical
56 signatures to arc volcanism. Many lines of evidence suggest that both subducted sediments and
57 hydrothermally altered ocean crust (AOC) can play significant roles in the chemistry of arc lavas
58 (e.g. Elliott et al., 1997; Kay et al., 1978; Plank and Langmuir, 1993). However, even though
59 sediment appears to be almost ubiquitously part of the subducted package (Plank and Langmuir,
60 1998) there are several arcs (or segments of arcs) that display very weak or debatably absent
61 sediment signatures (Hawkesworth et al., 1997; Leeman et al., 2005; Morris et al., 1990;
62 Regelous et al., 1997; Tera et al., 1986). The reasons for the attenuated sediment signatures in
63 certain arcs are not clear and, in addition to the case of a very thin layer of initially subducted

64 sediments (e.g., Tonga), could be caused by effective dehydration/melting of the sediment
65 package at depths too shallow to be supplied to the arc itself, physical accretion of sediment in
66 the forearc, or because current techniques to detect the sediment component in arc lavas are
67 either not sufficiently sensitive or yield ambiguous results.

68 Several recent studies have shown that thallium (Tl) isotope compositions of lavas in the
69 Mariana, Aleutian and Ryukyu arcs provide a powerful and unusually sensitive tracer of
70 sediment involvement during arc lava genesis (Nielsen et al., 2016; Prytulak et al., 2013; Shu et
71 al., submitted). Thallium is a highly incompatible trace metal, whose chemical behavior is
72 classically considered to mirror large ion lithophile elements such as Rb, Cs, and K, due to
73 similarities in ionic radii (Shaw, 1952). Thallium abundances were first specifically investigated
74 in arc environments by Noll et al. (1996). They sought to determine the fluid mobility of
75 chalcophile and siderophile elements relative to boron in putative hydrothermal fluids associated
76 with arc magmatism by analyzing a suite of lavas from seven subduction zones. However, they
77 could not determine unambiguous co-variation of thallium with other notionally fluid mobile
78 elements. Whilst there are clear indications of thallium's high fluid mobility in ore-forming
79 fluids (Heinrichs et al., 1980), the extent of fluid mobility under subduction zone conditions
80 remains unconstrained. Complicating matters is the possibility of thallium retention in accessory
81 phases such as phengite in the residual slab (Nielsen et al., 2016; Prytulak et al., 2013). Whether
82 or not thallium behaves as a lithophile or chalcophile element is also setting specific, with clear
83 lithophile behavior demonstrated during magmatic processes, irrespective of tectonic setting
84 (Prytulak et al., 2016; Prytulak et al., 2013) and chalcophile affinities found in mantle conditions
85 and during incipient partial melting (Nielsen et al., 2014). Recent advances in the calibration and
86 determination of thallium abundances by laser ablation (Jenner and O'Neill, 2012; Nielsen and

87 Lee, 2013; Nielsen et al., 2014) are a promising means to better-constrain elemental Tl behavior.
88 In general, the power of thallium as a tracer lies in its vanishingly small concentrations in the
89 mantle (≤ 1 ppb; see review in (Nielsen et al., 2017)) versus potential inputs to a mantle source.

90 Thallium has two stable isotopes that display a wide of range of fractionation in terrestrial
91 environments (see review in (Nielsen et al., 2017)). Specifically, thallium isotopes can be used to
92 quantify sediment fluxes from subducted slabs because pelagic sediments are highly enriched in
93 Tl and display isotopic compositions that are heavier than the isotopically homogeneous upper
94 mantle (Nielsen et al., 2016; Prytulak et al., 2013; Rehkämper et al., 2004). The heavy Tl isotope
95 compositions of pelagic sediments are due to the large isotope fractionation that occurs when Tl
96 sorbs to Mn oxides that form a ubiquitous component in deep-sea sediments (Nielsen et al.,
97 2013; Rehkämper et al., 2004; Rehkämper et al., 2002). Most sections of oceanic crust altered by
98 hydrothermal fluids at low temperatures ($<100^{\circ}\text{C}$), on the other hand, display light Tl isotope
99 compositions coupled with strong Tl enrichment (Coggon et al., 2014; Nielsen et al., 2006b; Shu
100 et al., submitted). Given the disparate Tl isotopic reservoirs represented by pelagic sediments and
101 AOC, Tl isotopes should enable distinction between AOC and sediment components in arc lavas.
102 However, the Tl concentrations in pelagic sediments are, generally, one to two orders of
103 magnitude higher than AOC (1000-5000 ng/g and 10-100 ng/g, respectively) and, therefore, even
104 minor amounts of sediment might dominate the Tl budget of arc lava source regions. To date, arc
105 lavas investigated for Tl isotopes show almost exclusively signatures consistent with addition of
106 sediment to the arc mantle source region, which is expected to overwhelm the presence of any
107 thallium signature of AOC (Nielsen et al., 2016; Prytulak et al., 2013; Shu et al., submitted).
108 However, the arcs investigated for Tl isotopes had previously been shown based on other
109 geochemical data to contain significant sediment components (Elliott et al., 1997; Kay and Kay,

110 1988; Shinjo et al., 2000). Hence these studies provided confirmation that sediment cycling in
111 arcs can be traced with Tl isotopes. In addition, they also showed that the subduction process
112 itself does not appear to fractionate Tl isotopes because the major inputs outboard of the arcs
113 have very similar values to those found in the arc lavas. Thus, any stable isotope fractionation
114 between residual slab, melts and fluids must be smaller than the Tl isotope variations of the
115 subducting sediments that dominated these arcs. Such a conclusion is also consistent with other
116 studies of high-T magmatic systems in Hekla, Iceland (Prytulak et al., 2016) and the Colahuasi
117 deposit in Chile (Baker et al., 2010) where no systematic Tl isotope variations were observed
118 that could be related to fractional crystallization or high-T fluid transport. Thallium isotopes are,
119 therefore, likely to be an excellent source tracer in subduction zones.

120 Here we present new Tl isotope data for samples from the Tonga-Kermadec and Central
121 American arcs. Although sediment subduction appears almost continuous along-strike in both
122 arcs (Plank and Langmuir, 1998), sections have been hypothesized to have very minor or even
123 absent sediment components (Carr et al., 1990; Hawkesworth et al., 1997; Morris et al., 1990;
124 Patino et al., 2000; Regelous et al., 1997; Regelous et al., 2010). In addition, the nature of the
125 subducted sediment is very different in the Tonga-Kermadec and Central American arcs. A
126 relatively thin layer of pelagic clays accounts for the majority of the package subducted
127 underneath the Tonga-Kermadec arc, whereas thick and discrete pelagic and carbonate
128 sedimentary packages dominate outboard of the Central American arc (Patino et al., 2000; Plank
129 and Langmuir, 1988). Therefore, these two arcs present a unique opportunity to investigate the
130 ubiquity of sediment addition in subduction zones and if Tl isotopes are able to trace sediment
131 additions where other geochemical parameters might only return ambiguous results.

132

133 2. Samples and background

134 2.1 Tonga-Kermadec

135 The Tonga-Kermadec arc is located in the Southwest Pacific Ocean stretching over 3000km
136 from the Northern end of New Zealand to northwards to the Vitiaz Trench approximately 200km
137 south of Samoa (Fig. 1). The subducting Pacific plate drilled at Deep Sea Drilling Project
138 (DSDP) sites 595/596 consists of only 70 meters of mainly pelagic red and brown clays, rich in
139 ferromanganese oxide minerals overlying Cretaceous age oceanic crust (Menard et al., 1983;
140 Speeden, 1973). The thickness, and the proportion of continental clastic sediment increase
141 southward along the trench towards New Zealand (Gamble et al., 1996). The Louisville
142 Seamount chain subducts obliquely, intersecting the trench at the point where the Tonga and
143 Kermadec arcs meet (Fig. 1). Volcaniclastic material from Louisville Seamount Chain (LSMC)
144 is evidenced in the sediments found at DSDP Site 204 (Fig. 1), some of which bear geochemical
145 resemblance to Louisville Seamount rocks (Ewart et al., 1998).

146 Previous studies of lavas in the Tonga-Kermadec arc have found clear signatures of
147 sediment (for example, high $^{207}\text{Pb}/^{204}\text{Pb}$ for a given $^{206}\text{Pb}/^{204}\text{Pb}$) dominating most of the
148 Kermadec arc (Ewart et al., 1998; George et al., 2005; Haase et al., 2002; Hawkesworth et al.,
149 1997; Regelous et al., 1997; Turner and Hawkesworth, 1997), whereas the Tonga arc appears
150 more influenced by an AOC component (Regelous et al., 1997; Regelous et al., 2010; Wendt et
151 al., 1997) although sediment may also constitute a minor component (George et al., 2005). The
152 two northernmost islands in the Tonga arc, Tafahi and Niuatoputapu, display strong Pb isotopic
153 evidence suggesting that the LSMC is the primary source of Pb in these two islands (Regelous et
154 al., 1997; Regelous et al., 2010; Turner and Hawkesworth, 1997). As a whole, the Tonga-
155 Kermadec arc, therefore, displays along-strike chemical and isotopic variation that can be related

156 to changes in the composition of the down-going plate and, in particular, large variations in the
157 ratio between sediments and AOC supplied to the mantle wedge (Haase et al., 2002). With these
158 previous findings in mind, it would be expected that Tl isotopes display substantial and
159 systematic variation as a function of the composition of the subducted material along-strike in
160 the Tonga-Kermadec arc.

161 We have measured Tl isotope compositions for a set of 30 subaerial lavas from 12 islands
162 covering the entire length of the Tonga-Kermadec arc (Table 1). The majority of the samples are
163 basalts or basaltic andesites complemented by five dacites and three andesites where less evolved
164 samples were not available. All samples are likely less than a few 100ka in age. Most samples
165 have previously been investigated for major and trace elements and some radiogenic isotopes
166 (Ewart et al., 1998; Regelous et al., 1997; Regelous et al., 2010; Smith et al., 1988). Here, we
167 also present new major elements and Sr and Nd isotope data for samples that did not already
168 have published values (Table 1 and 2). We have also analyzed Tl isotopes for 12 representative
169 sediment samples (Table 3) from DSDP Sites 204 and 596 located outboard of the Tonga portion
170 of the arc (Fig. 1). These sediments have previously been the subject of detailed geochemical
171 analyses (Ewart et al., 1998). The AOC subducted beneath the Tonga-Kermadec arc is presently
172 not well sampled by any drill holes and we have not analyzed any samples that represent Tonga-
173 Kermadec AOC. A previous study analyzed composite samples of AOC from ODP Hole 801C
174 outboard of the Mariana arc that are similar in age to the oceanic crust subducted in the Tonga-
175 Kermadec arc (Prytulak et al., 2013). However, the Tl isotope and concentration data for ODP
176 801C revealed very different values to those that have been found in ODP 504B, IODP U1301
177 and DSDP 442B (Coggon et al., 2014; Nielsen et al., 2006b; Shu et al., submitted). This
178 discrepancy is possibly related to the unusual Top Alkali Basalt found in ODP Hole 801C, which

179 would could have severely changed the Tl concentration and isotopic profile of the oceanic crust
180 due to late hydrothermal systems associated with this magmatic activity. Hence, we assume that
181 AOC in the Tonga-Kermadec arc is most likely similar to the majority of AOC sections
182 investigated to date.

183

184 **2.2 Central America**

185 The Central American volcanic arc (CAVA), *sensu stricto*, results from the subduction of
186 the Cocos and Nazca plate beneath the Caribbean Plate and extends some 1200km from western
187 Guatemala to central Costa Rica (Fig. 2). The volcanic front terminates to the southeast where
188 the Cocos Ridge subducts beneath central Costa Rica. From Guatemala to northwestern Costa
189 Rica the Cocos plate consists of ~ 24 Ma oceanic crust formed at the East Pacific Rise. In
190 contrast, further southeast along the CAVA slightly younger oceanic crust (15-20 Ma) of the
191 Cocos-Nazca Ridge that is variably overprinted by Galapagos hotspot magmatism is subducting
192 (Werner et al., 2003). Some of the physical parameters of the subduction system vary
193 systematically along the strike of the arc. Most notably, variations in crustal thickness of the
194 overriding plate, and of slab dip, have been used to help explain the unique along-strike variation
195 in geochemistry along the CAVA (Abers et al., 2003; Carr, 1984; Carr et al., 1990; Feigenson
196 and Carr, 1993; Patino et al., 2000).

197 The active arc front has produced magmas since the Quaternary (Carr et al., 2007) and can
198 be structurally split into seven segments of aligned volcanic centers (Bolge et al., 2009; Stoiber
199 and Carr, 1973), spaced an average of 28km apart, allowing for higher resolution sampling of
200 volcanic products than most other convergent margins (Carr et al., 2003). Additional volcanic
201 centers lie behind the active arc front, enabling the investigation of both along-strike, and cross-

202 arc chemical variations (Geilert et al., 2012; Patino et al., 1997; Walker et al., 1995; Walker et
203 al., 2000). Furthermore, the study of magmatic sources in the CAVA is aided by the unusual
204 abundance of mafic lavas (Carr, 1984).

205 There are strong chemical constraints on the inputs to the system through a combination of
206 ocean dredging and drilling. For instance, the downgoing Cocos plate has been the focus of
207 numerous DSDP, Ocean Drilling (ODP) and International Ocean Drilling (IODP) expeditions
208 (Fig. 2). The sedimentary package covering the Cocos plate is thought to be fairly homogeneous
209 along the strike of the arc (Patino et al., 2000). The main lithologies are described from DSDP
210 Leg 67 Site 495, drilled 22km seaward of the middle America trench, outboard of the
211 Guatemala/El Salvador border (Aubouin et al., 1982; Plank and Langmuir, 1998). Site 495
212 consists of ~175m of hemipelagic sediments overlying ~250m of pelagic carbonate ooze
213 (Aubouin et al., 1982). Both of these sediment lithologies have been invoked in the source of
214 CAVA lavas (Patino et al., 2000). DSDP Hole 504B is also located in geographical proximity to
215 the arc on the Nazca plate south of the Cocos ridge. Site 504 has one of the best-studied sections
216 of in situ altered oceanic crust, and has been investigated by several drilling expeditions (Alt et
217 al., 1996). Furthermore, altered basalts from Hole 504B provided the first evidence for light
218 thallium isotopes in low temperature altered crust (Nielsen et al., 2006b).

219 The earliest investigations of mafic CAVA lavas revealed distinct and systematic along-
220 strike variations in chemistry (Carr, 1984), which has encouraged intense subsequent
221 investigation. In general, a symmetrical pattern with a maximum or minimum at Nicaragua is
222 apparent in a number of chemical signatures such as Ba/La, B/La and Be isotopes, which is
223 generally attributed to variations in sediment flux along the arc, with a maximum in Nicaragua
224 and a minimum in Costa Rica and western Panama (where the Cocos ridge is being subducted)

225 (Carr, 1984; Carr et al., 1990; Leeman et al., 1994; Patino et al., 2000). Although reaching an
226 apparent maximum in Nicaragua, the recycled sediment flux also shows its greatest variation in
227 Nicaragua, which may explain the eruption of contrasting basaltic magmas in this portion of the
228 CAVA (Walker et al., 1990; Walker et al., 2001).

229 More recent studies have confirmed and converged on the idea of an anomalous Nicaraguan
230 segment, including arguments for the involvement of serpentinite-derived fluids in the heavily
231 slab-influenced western Nicaraguan lavas (Eiler et al., 2005; Heydolph et al., 2012; Rupke et al.,
232 2002; Sadofsky et al., 2008). The Costa Rican segment is also atypical with variable subducted
233 inputs from the Galapagos plume (Gazel et al., 2009; Gazel et al., 2011; Hoernle et al., 2008;
234 Sadofsky et al., 2008). Finally, higher resolution seismic data has allowed the more detailed
235 examination of the tectonic relationship to volcanic location and chemical characteristics in the
236 southern half of the arc (Abers et al., 2003; Hoernle et al., 2008; Syracuse et al., 2008; Van
237 Avendonk et al., 2011). Thus the well-documented, strong and systematic chemical variations, in
238 particular the rich background of fluid mobile element analysis such as B, Be and Li in the
239 CAVA, make it an ideal place to test the ability of thallium isotopes to reflect individual slab
240 components in the source of the lavas.

241 We have measured 34 lavas from the CAVA system and two sediments from DSDP Site
242 495. All but three lava samples (from Guatemala) are from the active volcanic front. Most are
243 basaltic, with SiO₂ below 52 wt% and have been extensively studied by numerous investigations
244 for major element, trace element and both radiogenic and stable isotope variations. An extremely
245 useful chemical database has been compiled for CAVA lavas (Carr et al., 2014) and the chemical
246 compositions and references for the samples of this study are archived in that database.

247

248 **3. Methods**

249 **3.1 Sample preparation**

250 All sediment samples and lavas from Central America as well as the majority of lavas from
251 Tonga were received and processed as powders. Some of the lavas from Tonga-Kermadec were
252 received as rock fragments and these were carefully crushed into mm-sized chips and
253 handpicked under binocular microscope to avoid pieces with surficial alteration and
254 contamination. Subsequently the separated chips were ultrasonicated in ultra-pure water and any
255 particles in suspension were discarded. Separate aliquots of several of the rock chip samples
256 were also subjected to mild leaching in 1M HCl in order to investigate if contamination from any
257 Mn oxides or alteration minerals was present (Table 1). This procedure has been shown to
258 effectively remove these contaminants (Nielsen et al., 2016).

259 The rock and sediment powders as well as the handpicked and cleaned chips were dissolved
260 in a 1:1 mixture of concentrated distilled HNO₃ and HF on a hotplate at ~150°C for 24h. They
261 were then dried and fluxed several times using a 1:1 mixture of concentrated distilled HNO₃ and
262 HCl until the fluorides, which formed in the first step, were completely dissolved. Following
263 this, samples were dried on a hotplate and dissolved in 1M HCl in preparation for separation of
264 Tl from sample matrix. Isolation of Tl followed previously described anion exchange
265 chromatographic methods (Nielsen and Rehkämper, 2011; Nielsen et al., 2004). Total procedural
266 Tl blanks during this study were <3pg, which is insignificant compared to the indigenous Tl
267 processed for the samples of >3ng.

268

269 3.2 Thallium isotope and concentration measurements

270 Thallium isotope compositions were determined using a Thermo Finnigan Neptune multiple
271 collector inductively coupled plasma mass spectrometer (MC-ICPMS), located in the Plasma
272 Facility at Woods Hole Oceanographic Institution (WHOI). Following previous studies (Nielsen
273 et al., 2004), external correction for mass discrimination to NIST SRM 981 Pb and standard-
274 sample bracketing to the NIST SRM 997 Tl standard were applied for measurement of Tl
275 isotopic compositions and correction for instrumental mass bias. Thallium isotope compositions
276 are reported relative to the NIST SRM 997 standard in parts per 10,000:

$$277 \quad \epsilon^{205}\text{Tl} = 10,000 \times \left(\frac{{}^{205}\text{Tl}/{}^{203}\text{Tl}_{\text{sample}} - {}^{205}\text{Tl}/{}^{203}\text{Tl}_{\text{SRM 997}}}{{}^{205}\text{Tl}/{}^{203}\text{Tl}_{\text{SRM 997}}} \right) \quad (1)$$

278 The column chemistry procedure returns quantitative Tl yields (Nielsen et al., 2004;
279 Prytulak et al., 2013; Rehkämper and Halliday, 1999), thus Tl concentrations were estimated by
280 monitoring the ${}^{205}\text{Tl}$ intensity during the isotopic measurements. The measured ${}^{205}\text{Tl}/{}^{208}\text{Pb}$ ratios
281 were converted directly into Tl concentrations by adding a known quantity of NIST SRM 981
282 Pb. Recent studies in the NIRVANA ([Non-traditional Isotope Research on Various Advanced](#)
283 [Novel Applications](#)) lab at WHOI has documented that the long-term reproducibility of Tl
284 isotopes and concentrations in silicate samples are $\pm 0.4 \epsilon^{205}\text{Tl}$ -units (2sd) and $\pm 15\%$ (2sd),
285 respectively (Nielsen et al., 2015; Nielsen et al., 2016; Shu et al., submitted).

286

287 3.3 Measurements of Nd and Sr isotope compositions and major elements

288 For Sr and Nd isotope measurements, approximately 100 mg of sample was digested in
289 concentrated HNO_3 -HF, evaporated and treated with 15M HNO_3 until completely in solution.
290 The sample was then dissolved in 3M HNO_3 , and Sr and the REE were separated from the rock
291 matrix using Eichrom SrSpec and TRUSpec resins, respectively. The REE fraction was re-

292 dissolved in 0.25M HCl, and Nd separated from the other REE using ion exchange columns
293 containing 1.5 ml of Eichrom LNSpec resin. All reagents used were Teflon distilled, and total
294 procedural blanks were below 100 ng and 20 ng for Sr and Nd respectively. Isotope
295 measurements were carried out on a Thermo Triton thermal ionisation mass spectrometer in
296 static mode at the GeoZentrum Nordbayern. Measured $^{87}\text{Sr}/^{86}\text{Sr}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ ratios were
297 corrected for instrumental mass fractionation assuming $^{86}\text{Sr}/^{88}\text{Sr} = 0.1194$ and $^{146}\text{Nd}/^{144}\text{Nd} =$
298 0.7219 . Over the period of analysis, average values of the NBS987 Sr and La Jolla Nd standards
299 were $^{87}\text{Sr}/^{86}\text{Sr} = 0.710249$ and $^{143}\text{Nd}/^{144}\text{Nd} = 0.511850$, respectively, and the data in Table 1
300 were normalized to values of 0.710240 and 0.511857 , for direct comparison with the earlier data
301 from Ewart et al. (1998).

302 Major element analyses of Tonga lavas were carried out by XRF (AMETEK Spectro
303 XEPOS Plus) at the GeoZentrum Nordbayern, on fused glass discs prepared by using lithium
304 tetraborate as flux. Loss on ignition was determined on a 1 g aliquot heated at 1030°C in a
305 muffle furnace for 12 hours.

306

307 **4. Results**

308 **4.1 Tonga-Kermadec**

309 Thallium concentration in the Tonga-Kermadec lavas range from $\sim 5\text{ng/g}$ to 144ng/g , with
310 the highest concentrations found for the most evolved samples (Fig. 3a). Lavas in the Tonga arc
311 vary between $\epsilon^{205}\text{Tl} = -0.7$ to $+6.7$, with an average value of $\epsilon^{205}\text{Tl} = +1.2$, while lavas in the
312 Kermadec arc vary between $\epsilon^{205}\text{Tl} = -0.5$ to $+11.5$ (Table 1), with an average value of $\epsilon^{205}\text{Tl} =$
313 $+3.1$. Every lava analyzed in this study is significantly heavier than the average upper mantle, as
314 defined by MORB, which displays $\epsilon^{205}\text{Tl} = -2 \pm 0.5$ (Nielsen et al., 2006a; Nielsen et al., 2006b).

315 Sediments from DSDP Sites 204 and 596 contain primarily pelagic clays (Menard et al., 1983;
316 Speeden, 1973), but at DSDP Site 204 significant amounts of volcanoclastic sediments from the
317 LSMC are also found (Speeden, 1973). Thallium isotope compositions and concentrations are
318 substantially different for these two sediment types with pelagic clays exhibiting a concentration-
319 weighted average $\epsilon^{205}\text{Tl} = +3.6$ and $[\text{Tl}] = 1876 \text{ ng/g}$, whereas volcanoclastic sediments are
320 characterized by concentration-weighted average $\epsilon^{205}\text{Tl} = +0.2$ and $[\text{Tl}] = 214 \text{ ng/g}$ (Table 3).
321 We use these averages as representatives for the two different sediment components and do not
322 attempt to estimate the bulk average subducted sediment because the volcanoclastic sediment
323 component is highly variable along the arc to the extent that little or no LSMC volcanoclastics are
324 subducted beneath the Kermadec arc. The Tl compositional differences are consistent with
325 substantial enrichment of isotopically heavy Tl in pelagic clays from Mn oxides (Nielsen et al.,
326 2013; Rehkämper et al., 2004), whereas volcanoclastic sediments only contain minor Mn oxide
327 components.

328

329 **4.2 Central America**

330 Thallium concentrations in the CAVA lavas range from $<5 \text{ ng/g}$ in primitive lavas from
331 central Nicaragua to $\sim 150 \text{ ng/g}$ in more evolved lavas (Table 4). This behavior likely reflects the
332 near-perfectly incompatible behavior of Tl during differentiation processes (Prytulak et al.,
333 2016). Although the lavas span a large geographic area and encompass many volcanic centers,
334 the overall sense of increasing Tl concentration with decreasing MgO is apparent (Fig. 3b). A
335 clear exception is sample CR-IZ-D6 from Irazu volcano in Costa Rica, which has an
336 anomalously high thallium concentration (243 ng/g) for its degree of evolution (Fig. 3b). Note
337 that we did not estimate the Tl concentration of sample NE203. This sample was analyzed by

338 Noll et al. (1996) who reported a concentration of 1 ng/g. However, this concentration is
339 substantially lower than any other arc lava analyzed to date and also produces $Ce/Tl = 8400$,
340 which is far higher than what is realistic for an arc lava. We, therefore, only use the Tl isotope
341 composition for this sample.

342 Compared to other investigated arcs, thallium isotope signatures lighter than MORB are
343 more common in the CAVA (Table 4). At face value, the 34 lavas analyzed in this study also
344 display the largest range of isotope composition in mafic lavas documented to date, with $\epsilon^{205}Tl$
345 = -11.5 to +9.0 and does not correlate with classic recycled sediment indicators such as Ba/La,
346 U/La or ^{10}Be in the arc. The extremes of this range occur in the same volcanic system, the
347 Nejapa cinder cones of central Nicaragua, which are noteworthy for their overall chemical
348 variability (Walker et al., 1990; Walker et al., 2001). Significantly light isotope signatures of
349 $\epsilon^{205}Tl = -8.2$ are also found in Irazu volcano, located in central Costa Rica. The three behind the
350 volcanic front sample from Guatemala (GU-C201, GU-C303 and GU-C837) have average $\epsilon^{205}Tl$
351 = -1.1, which is not significantly anomalous when considered in the context of the overall
352 variation in CAVA lavas. The carbonate sediment yields a heavy isotope value of $\epsilon^{205}Tl = +5.8$,
353 albeit at a low Tl concentration of 33 ng/g. The hemipelagic sediment has a much higher
354 thallium concentration of 1077 ng/g and an isotope composition of $\epsilon^{205}Tl = +0.2$, similar to
355 sediments analyzed from outboard the Mariana, Aleutian, and Tongan subduction zones.

356

357 **5. Discussion**

358 **5.1 Effects of secondary processes on primary [Tl] and $\epsilon^{205}\text{Tl}$**

359 It has been documented that processes such as degassing, subaerial aqueous alteration,
360 assimilation of wall rock and fractional crystallization prior to eruption can alter the Tl budget of
361 subaerial lavas and potentially cause Tl isotope fractionation (Nielsen et al., 2016; Prytulak et al.,
362 2013). Before interpreting isotopic signatures in terms of source components we must first assess
363 if any of the above processes affected the investigated lavas from Tonga-Kermadec and Central
364 America.

365

366 ***5.1.1. Subaerial aqueous alteration***

367 Subaerial lavas are often observed to preferentially lose alkali metals during aqueous
368 alteration from meteoric water (Schiano et al., 1993). This effect is most likely due to the high
369 solubility of these metals in aqueous solution. Given the geochemical similarity between Tl and
370 the alkali metals (i.e. ionic charge and radius as well as high aqueous solubility) it has been
371 suggested that subaerial alteration may lead to significant Tl losses, reflected as high Ce/Tl ratios
372 coupled with high Th/Rb ratios (Nielsen et al., 2016). These losses, however, would likely not be
373 associated with Tl isotope fractionation because weathering has been shown to cause negligible
374 effect on the isotope system (Nielsen et al., 2005). Figure 4 illustrates the trajectory of alteration
375 in Ce/Tl vs Th/Rb space with a Hawaiian lava that had clear petrologic and geochemical
376 evidence for subaerial alteration although no Tl isotope fractionation appeared to have perturbed
377 this sample (Nielsen et al., 2006a). Also plotted is one sample from the Aleutians that likewise
378 was suspected of subaerial alteration (Nielsen et al., 2016). Compared to these two lavas, the
379 datasets from both Tonga-Kermadec and the CAVA do not show any clear indication of

380 subaerial alteration. High loss on ignition (LOI) measured for bulk rock powders can also be a
381 qualitative indicator of degree of weathering, but in our data set there are no correlations between
382 LOI and other indicators of weathering and generally LOI is very low (<0.5%) for the studied
383 lavas, which suggests that alteration is not significant.

384 Central American lavas span greater ranges of Ce/Tl and Th/Rb than unaltered lavas from
385 the Tonga-Kermadec and Aleutian arcs (Fig. 4). In principle, this variation could be interpreted
386 to reflect more extensive alteration of CAVA lavas. However, the vast majority of the samples
387 studied here are historical eruptions with essentially no or very minor alteration (Carr et al.,
388 2014), which suggests that the most elevated Th/Rb and Ce/Tl ratios of samples from Irazu
389 volcano in central Costa Rica are likely related to their unusual Galapagos-tainted source rather
390 than an alteration affect (Gazel et al., 2009; Gazel et al., 2011). This interpretation is also
391 supported by the high Th/Rb ratios observed in Galapagos lavas (Saal et al., 2007). Although no
392 Tl concentration data is available for Galapagos lavas, high Ce/Tl ratios are generally found in
393 other OIBs (Nielsen et al., 2014), which supports such an origin for the anomalous Ce/Tl and
394 Th/Rb ratios found in the Irazu samples. Hence, we conclude that none of the CAVA samples
395 were significantly affected by post-eruption aqueous alteration

396

397 ***5.1.2 Degassing***

398 Volcanic fumaroles contain elevated concentrations of thallium and display a range of Tl
399 isotope compositions that span both very light and heavy values, which has been interpreted to
400 reflect both evaporation and condensation processes (Baker et al., 2009). In principle, degassing
401 should be expressed as a loss of Tl associated with kinetic isotope fractionation whereby the light
402 Tl isotope would preferentially be lost. Given that Tl partitioning during mantle melting is

403 similar to Ce (Nielsen et al., 2014), it could be inferred that unusually high Ce/Tl ratios, not
404 otherwise explained by aqueous weathering, coupled with heavy Tl isotope compositions are
405 indicative of significant degassing - if lavas of the same degree of evolution and fractionating
406 assemblage are compared. Such effects have been reported for arc lavas (Nielsen et al., 2016),
407 but the lack of systematic Tl isotope fractionation for volcanic fumaroles also suggests that
408 degassing might not be ubiquitously associated with significant Tl isotope fractionation (Baker et
409 al., 2009). Within the present data set only one sample (Macaulay 10380) exhibits anomalously
410 high Ce/Tl relative to the other samples (at similar MgO) in this study (Fig. 4b). The sample is
411 also characterized by a heavy Tl isotope composition, as would be expected for a degassed lava.
412 However, a second sample from Macaulay (Table 1) has an almost identical Tl isotope
413 composition with a substantially lower Ce/Tl ratio, which would suggest that Tl loss due to
414 degassing in this sample might not have been accompanied by significant Tl isotope
415 fractionation and thereby consistent with published Tl isotope data for volcanic fumaroles (Baker
416 et al., 2009). In addition, the Ce/Tl ratio of Macaulay 10380 is lower than the upper mantle (Fig.
417 4b), which implies that any Tl loss due to degassing was relatively minor.

418 Higher Ce/Tl ratios are found in the CAVA than in Tonga-Kermadec, (Fig. 4b), the
419 Aleutians and Marianas (Prytulak et al., 2013). None of the high Ce/Tl ratios are associated with
420 systematically heavy Tl isotope compositions, as would be expected for kinetic isotope
421 fractionation associated with degassing. However, some of the highest Ce/Tl lavas in central
422 Nicaragua also have the most extreme positive ($\epsilon^{205}\text{Tl} = +9.0$) and negative ($\epsilon^{205}\text{Tl} = -11.5$)
423 thallium isotope compositions (samples NE-201 and NE-202; Fig. 4b), which mimics some of
424 the most fractionated Tl isotope effects observed for volcanic fumaroles (Baker et al., 2009). The
425 Tl isotope variations in volcanic fumaroles were even found to be large within individual

426 | volcanic centers (e.g. Kilauea volcano, Hawaii) (Baker et al., 2009), which suggests that lavas
427 | with very low Tl concentrations could more easily be perturbed to both heavier and lighter Tl
428 | isotope compositions by effects from degassing/condensation. Given that these two samples
429 | originate from neighboring Nejapa cinder cones and both have been classified as high-Ti basalts
430 | by Walker et al. (1990) with almost identical Sr and Nd isotopes (Carr et al., 2014), it seems
431 | improbable that the Tl isotope composition of these two samples reflect that of their mantle
432 | source. The lava sample GR5 also has similar major and trace element characteristics including
433 | anomalously low thallium concentration and heavy isotope composition ($\epsilon^{205}\text{Tl} = +3.5$; GR5),
434 | which we also suspect could be influenced by kinetic processes rather than a true reflection of
435 | mantle source.

436

437 ***5.1.3 Assimilation and fractional crystallization***

438 Thallium behaves as a near-perfect incompatible element during fractional crystallization of
439 anhydrous phases with no resolvable Tl isotope fractionation (Nielsen et al., 2016; Prytulak et
440 al., 2016). However, crystallization will affect key trace element ratios involving Tl as well as
441 other elements (e.g. Ce/Tl, Ce/Pb, Sr/Nd, U/Nb) due to systematic differences in the mineral-
442 melt partition coefficients (Blundy and Wood, 2003). In addition, any concomitant wall rock
443 assimilation is likely to both affect these trace element ratios and potentially alter the Tl isotope
444 composition of the original melt if the wall rock is isotopically different to the primitive melt.

445 Fractional crystallization affects all lavas, and our Tonga-Kermadec dataset contains
446 andesites and dacites that will have undergone more extensive fractionation. Therefore trace
447 element ratios should first be assessed for the effect of fractional crystallization. This is not a
448 straightforward task because the samples are not genetically related along a shared liquid line of

449 descent. However, the only trace element ratio that is systematically different in dacites relative
450 to less evolved samples is Sr/Nd, which is lower in dacites, likely due to crystallization of
451 plagioclase. No elemental ratios involving Tl are significantly different in dacites, suggesting
452 that Tl and most other trace elements (except for Sr) behaved highly incompatibly. Therefore
453 trace element ratios using thallium can be employed to investigate the sub-arc mantle and
454 contributions from the subducting slab in the Tonga-Kermadec system. In addition, there are no
455 differences between Tl isotope compositions measured for dacites and less evolved samples in
456 the Tonga-Kermadec system (Table 1). The dacite sample from Curtis Island is the isotopically
457 lightest sample from the Kermadec arc, which could be interpreted to reflect assimilation of
458 isotopically light wall rock material, although it is difficult to envision a process that created a
459 light Tl isotope signature in the first instance, since igneous ocean crust is isotopically similar
460 MORB (Nielsen et al., 2006b) whereas lower arc crust in the Tonga-Kermadec arc is most likely
461 heavier than MORB as evidenced by the overall heavy values observed in the lavas. However,
462 such an interpretation is consistent with the fact that this sample exhibits the most radiogenic Sr
463 isotope composition compared with other islands in the Tonga-Kermadec arc combined with Nd
464 isotope compositions similar to other Kermadec islands (Ewart et al., 1998), which might
465 indicate assimilation of older oceanic crust. However, given that the basaltic andesite (Table 1
466 and 2) displays Tl isotope and other basalts from Curtis Island display Pb, Sr and Nd isotope
467 compositions that are identical to dacites and rhyolites from the same island (Ewart et al., 1998;
468 Smith et al., 1988), we infer that assimilation processes are unlikely to have affected isotopic
469 compositions of Tl, Sr, Nd or Pb for these samples.

470 The nature of our sample set from the CAVA is generally a single sample per volcanic
471 edifice, which again makes evaluating the effects of fractional crystallization difficult.

472 Furthermore, the well-documented, high amplitude variation in the chemical characteristics of
473 the source regions combined with a variable sediment flux along the arc, mean that variation in
474 trace element ratios such as Ce/Tl, Ce/Pb, Sr/Nd, and U/Nb yield real information about source
475 rather than a reflection of secondary fractionation. The typical petrographic assemblage
476 documented in the CAVA are large, zoned plagioclase and pyroxene crystals in more evolved
477 magmas with ubiquitous olivine and magnetite in basaltic lavas, and only very rare reports of
478 amphibole in high Na lavas from Guatemala (Carr et al., 2003; Walker et al., 2000). Thus, the
479 assemblage is very similar to those investigated for the effects of differentiation on Tl isotopes
480 by Prytulak et al. (2016), where no correlation between thallium isotopes and indices of
481 differentiation were found, and thallium remained near-perfectly incompatible throughout
482 fractionation from basalt to rhyolite. Thus, we conclude that the dominant cause of trace element
483 variation in the presented CAVA lavas is mantle source differences rather than magmatic
484 processes. This conclusion is in agreement with previous studies that found silicic volcanic rocks
485 in Central America show the same regional variations in trace element ratios that the basaltic
486 rocks do (Vogel et al., 2004; Vogel et al., 2006).

487

488 **5.2 Slab components in Central America and Tonga-Kermadec lavas**

489 ***5.2.1. Pelagic sediment contributions throughout Tonga-Kermadec***

490 As outlined above, secondary processes are unlikely to account for much, if any, of the Tl
491 isotope variation observed in the Tonga-Kermadec lavas. Therefore, the mantle source region
492 along the entire length of the arc is characterized by Tl isotope compositions heavier than the
493 depleted upper mantle and AOC (Fig. 6). This provides clear evidence for a contribution of Tl
494 from subducted sediment in all Tonga-Kermadec lavas. In general, pelagic sediments display

495 substantial Tl isotope variation (e.g., Rehkamper et al., 2004; Nielsen et al., 2016). It is,
496 therefore, difficult to assess the exact Tl isotope composition of the sedimentary endmember in
497 the Tonga-Kermadec arc, which prevents a direct utilization of Tl isotopes to quantify sediment
498 contributions. Visually, it appears that Kermadec lavas are isotopically heavier than Tonga lavas
499 (Fig. 6), which might suggest a larger sediment component in the Kermadec lavas than in Tonga.

500 This observation is also generally supported by evidence from radiogenic isotopes (Regelous et
501 al., 1997). that suggest greater sediment contributions in Kermadec than in Tonga lavas (Fig. 7).

502 However, increased sediment contribution in the Kermadec arc versus Tonga should result in
503 higher Tl concentrations at a given MgO, which is not observed (Fig. 3a). The small difference
504 in Tl isotopes between Kermadec and Tonga is, therefore, more likely related to differences in
505 the average Tl isotope composition of subducted sediments.

506 Two of the lava samples investigated here extend to even heavier Tl isotope compositions
507 than observed for pelagic sediments (Fig. 6). These more extreme values could potentially be
508 explained by the occurrence of ferromanganese (Fe-Mn) nodules in the subducted sediment pile
509 (Menard et al., 1983; Speeden, 1973) that in other locations have been observed to exhibit Tl
510 concentrations as high as 100 $\mu\text{g/g}$ and $\epsilon^{205}\text{Tl} > +10$ (Nielsen et al., 2016; Rehkämper et al.,
511 2004; Rehkämper et al., 2002). Although the Sr, Nd and Pb concentrations of these nodules are
512 also high (Hein et al., 2000), their concentrations relative to Tl are sufficiently low that addition
513 of small amounts (<0.01% by weight) of pure Fe-Mn nodule material to the arc lava source
514 region would only have a minor effect on Sr, Nd and Pb isotopes whereas the contaminated
515 mantle source region would strongly inherit the Tl isotope signature of the Fe-Mn nodule (Fig.
516 7). The extreme Tl isotope variation in the investigated lavas, therefore, suggest that transfer of
517 material from slab to mantle wedge still contained sufficient heterogeneity to leave Tl isotope

518 values highly variable. Most islands in the Tonga-Kermadec arc display only little variation in Sr
519 and Nd isotopes (Ewart et al., 1998; Hergt and Woodhead, 2007; Regelous et al., 1997; Regelous
520 et al., 2010; Turner et al., 2012), which suggests that the mantle source for each island is fairly
521 homogenous. However, given the high Tl/Sr and Tl/Nd ratios for Fe-Mn nodules, heterogeneous
522 addition of sufficient Fe-Mn nodule material to generate the observed Tl isotope variation would
523 not register notably in Pb, Sr and Nd isotopes (Fig. 7).

524 Pelagic sediments clearly dominate the Tl budget of the subducted sediment package as
525 evidenced by the heavy Tl isotope composition of lavas across the entire Tonga-Kermadec arc
526 (Fig. 6), which is consistent with the recovered lithologies on the downgoing plate from DSDP
527 Sites 595/596. Based on Tl isotopes it is also evident that pelagic sediments are present even in
528 the Northern Tonga islands of Tafahi and Niuatoputapu where previous studies have found that
529 LSMC material is an important source of Pb (Ewart et al., 1998; Regelous et al., 1997; Regelous
530 et al., 2010; Turner and Hawkesworth, 1997). Although Pb isotopes in Tafahi and Niuatoputapu
531 clearly point towards an influence of LSMC material, the Sr isotope composition of lavas from
532 these two islands are significantly more radiogenic than LSMC material (Beier et al., 2011;
533 Vanderkluyzen et al., 2014), which requires an additional component with radiogenic Sr in the
534 Northern Tonga mantle source. This component could be sourced from the Samoan mantle
535 plume (Wendt et al., 1997), but given the heavy Tl isotope compositions also observed in
536 Northern Tonga (Fig. 7), at least some of the more radiogenic Sr must originate from small
537 amounts of pelagic sediment (<1% by weight). The small sediment component in northern Tonga
538 is in agreement with several previous studies that found evidence for sediment involvement
539 across the entire Tonga-Kermadec arc even in the Northern islands of Tafahi and Niuatoputapu
540 (George et al., 2005; Hergt and Woodhead, 2007).

541

542 **5.2.2. Central American isotope variations**

543 | Although the CAVA has large variations in thallium isotopes (Fig. 8a), without two
544 | anomalous regions (Central Nicaragua and Irazu volcano) and the three samples from behind the
545 | volcanic front, the average isotope composition is $\epsilon^{205}\text{Tl} = -1.6 \pm 3$ (n=23), which, although
546 | variable, is identical within error to the upper mantle as represented by MORB ($\epsilon^{205}\text{Tl} = -2 \pm 1$;
547 | Nielsen et al., 2006a). We have only measured two sediment samples from DSDP 495, but they
548 | represent the two major subducting lithologies (carbonate and hemipelagic sediment). The
549 | carbonate sample has a very low thallium concentration (33 ng/g) and a heavy isotope
550 | composition of +5.8. Investigation of corals, foraminifers, and biogenic oozes suggest that Tl
551 | concentrations of such materials is much lower than our carbonate sample (Nielsen and
552 | Rehkamper, 2011; Rehkamper et al., 2004). Furthermore, rivers draining carbonate lithologies
553 | have isotopically light thallium consistent with pure carbonate inheriting the seawater Tl isotope
554 | composition of $\epsilon^{205}\text{Tl} = -6.0$ (Nielsen et al., 2005). Therefore, the heavy Tl isotope signature is
555 | probably **inherited** from manganiferous clay that is reported to occur within this carbonate unit
556 | (Aubouin et al., 1982). Whatever the cause of the heavy isotope composition of our single
557 | carbonate sample, the total budget of thallium subducting in the carbonates is likely not
558 | sufficient to perturb the overall Tl isotope budget.

559

560 **5.2.3. Nicaraguan lavas: evidence for sediments?**

561 | Nicaragua has long been recognized as a location of chemically distinct lavas. For example,
562 | samples from Central Nicaragua are characterized by some of the highest $^{10}\text{Be}/^9\text{Be}$ ratios (Fig.
563 | **8b**) found in arc lavas globally, which provides unequivocal evidence that hemipelagic clays

564 found outboard of the CAVA contribute to magmatism in Nicaragua (Morris et al., 1990; Tera et
565 al., 1986). Furthermore, many studies have concluded that Nicaraguan volcanoes have the
566 strongest ‘slab signature’ in the CAVA, classically represented as elevated Ba/La ratios (Fig. 8c)
567 that are found to peak in Western Nicaragua approximately between the volcanoes Cosiguina
568 and Cerro Negro (e.g. Carr et al., 2003; Walker and Gazel, 2014). Ba/La ratios have commonly
569 been employed as a proxy for a slab fluid component (Sadofsky et al., 2008; Walker et al., 2000;
570 Woodhead et al., 1998; Woodhead and Johnson, 1993), which suggests that a high fluid flux
571 could be responsible for the elevated Ba/La in Nicaraguan lavas. This inference is supported by
572 several investigations that have invoked a higher fluid flux and significantly larger degree of
573 melting beneath Nicaragua compared to the rest of the CAVA, due to the steep dip ($\sim 75^\circ$) of the
574 Cocos slab (Abers et al., 2003; Carr et al., 1990; Patino et al., 2000; Sadofsky et al., 2008;
575 Syracuse et al., 2008). However, the regional variation in Ba/La actually reflects variation in La
576 concentrations in the lavas, not Ba, and thus it is difficult to relate to a variable fluid flux, but
577 could denote unusually high degrees of melting that vary as a function of fluid flux (Carr et al.,
578 1990; Carr et al., 2007). Alternatively, given that the subducting sediments are characterized by
579 extremely high Ba/La, it is also possible that the Nicaraguan peak in Ba/La is largely sediment
580 related (Patino et al., 2000).

581 Figure 8b shows the along-strike variation of thallium isotopes without the three samples
582 whose thallium isotope composition we suspect to be affected by kinetic processes (section
583 5.1.2). In Western Nicaraguan lavas, where Ba/La is at it highest, samples have average thallium
584 isotopes of $\epsilon^{205}\text{Tl} = -1.4 \pm 0.7$ (n=6), which is indistinguishable from the mantle value.
585 Therefore, arguably the greatest slab input does not correspond to heavier Tl isotopic
586 compositions as would be expected if this component was dominated by hemipelagic sediments

587 with high Tl concentrations. The very elevated Ba/La (and Ba/Th) ratios in Western Nicaragua
588 could, however, be sourced from carbonate sediments where Tl concentrations are low and,
589 therefore, do not significantly affect the total Tl budget. This offers an explanation for the lack of
590 co-variation between thallium isotopes and trace element indices of overall slab contributions in
591 Nicaragua.

592 Central Nicaragua is the location of the Nejapa and the Granada volcanic fields, from which
593 we have discounted three out of six samples. These fields are made up of cinder cones aligned
594 along fault traces rather than large composite volcanoes. Erupted lavas are generally very
595 primitive, high degree melts, with high MgO contents and very low concentrations of
596 incompatible elements, including thallium (Table 4). A subset of the Nejapa and Granada lavas
597 are characterized by an almost MORB-like absence of negative HFSE anomalies and have been
598 dubbed high-Ti basalts (Walker et al., 1990). These higher-Ti compositions could be, in part, a
599 product of re-melting mantle that had already generated LREE-enriched magmas, which might
600 explain the extremely low concentrations of many incompatible elements (Carr et al., 1990;
601 Feigenson and Carr, 1993; Reagan et al., 1994). Such a depleted mantle would also be more
602 susceptible to contamination with components characterized by anomalous Tl isotope
603 compositions.

604 The samples we include from this region are isotopically heavy with compositions of $\epsilon^{205}\text{Tl}$
605 = -1.2, +0.1, +1.0 and +4.7. These values are indicative of recycling of hemipelagic sediment in
606 the region, which is consistent with the generally elevated ^{10}Be values seen for this section of the
607 CAVA (Fig. 8b). However, only one of these three samples has been analyzed for ^{10}Be and was
608 found to be fairly low (Carr et al., 2014) and there is, therefore, no direct correlation between
609 ^{10}Be and Tl isotopes (Fig. 8). It is important to note that correlations between these two

610 parameters need not be apparent because ^{10}Be is high only in the youngest sediments, while Tl
611 isotopes are likely heavy throughout the hemipelagic clays.

612 From the perspective of Tl isotopes, an attractive feature of Central Nicaragua is that with a
613 locally trace element depleted mantle, the smallest addition of Tl from AOC, hemipelagic
614 sediments, and/or serpentinite-derived fluid (which might be isotopically heavy in Tl; (Nielsen et
615 al., 2015)) will dominate the Tl isotope signature, without being apparent in many other tracers.
616 Thus, Central Nicaragua could be a tremendously fruitful area for further systematic
617 examination, requiring more data on Cocos plate sediments and crust, oceanic serpentinites and
618 Nicaraguan volcanic rocks.

619

620 ***5.2.4. Irazu, Costa Rica: evidence for altered oceanic crust?***

621 Costa Rican lavas are notable in that not a single sample is isotopically heavy compared to
622 the mantle as represented by MORB, and the general tendency towards isotopically light
623 thallium (Fig. 8a) contrasts with all previously studied arcs. Irazu volcano in Costa Rica is
624 further unique in a number of respects. The volcano lies at the southeastern terminus of the arc as
625 well as in the path of subducting Galapagos-tinged crust. Two studies (Benjamin et al., 2007;
626 Sadofsky et al., 2008) have examined olivine-hosted melt inclusions for major, trace, and volatile
627 contents. Both note the high water contents of Irazu compared to many other volcanic centers in
628 the CAVA. In particular, Irazu was the exception in the study of Sadofsky et al. (2008) because,
629 unlike the rest of the investigated CAVA melt inclusions, Irazu did not show positive correlation
630 of Ba/La and B/La and water contents. It also has the highest F contents for a given olivine
631 composition coupled with the lowest B/La, Ba/La and highest La/Y, Nb/Y. Sadofsky et al.
632 (2008) explained these unusual features as resulting from a very fluid rich, but sediment poor

633 mantle, which is consistent with the subduction of sediment-poor seamounts at its present day
634 location. They also raised the possibility of melting the altered mafic crust to explain the high
635 La/Y and Nb/Y, which is consistent with regional isotopic and trace element studies (Gazel et al.,
636 2009; Gazel et al., 2011; Hoernle et al., 2008). Our study shows that sample CR-IZ-D6 has the
637 highest Tl concentration in the CAVA (243 ng/g), coupled with an extremely light isotope
638 signature of $\epsilon^{205}\text{Tl} = -8.2$. Thus the thallium concentration and isotope systematics are consistent
639 with previous interpretation of incorporation of altered oceanic crust. Irazu specifically, and
640 Costa Rica in general, may be one of the few regions where the downgoing oceanic crust is
641 sufficiently naked of Tl-rich sediments to allow detection of isotopically light Tl derived from
642 the AOC.

643

644

645 **6. Inter-arc comparisons and implications for Tl subduction cycling**

646 **6.1. Sediment control of arc lava Tl budgets**

647 As outlined in the introduction, there is overall consensus that both sediment and AOC
648 components are actively transported from the subducting slab to the mantle wedge and
649 participate in melt generation. However, even though sediment subduction is almost ubiquitous
650 some arcs display highly attenuated sediment signatures (Plank and Langmuir, 1998).

651 Compilation of all arc lava Tl isotope data published to date reveals that the vast majority of
652 samples plot towards heavier compositions than the DMM, which strongly implicates sediment
653 in these arcs (Fig. 9). The only exceptions are the Western Aleutians and Costa Rica. Magmatism
654 in the Western Aleutians likely reflects an unusual tectonic environment where orthogonal
655 subduction of the Pacific plate has stalled significantly, which may have heated up the oceanic

656 crust sufficiently to cause eclogite melting and eruption of adakitic magmas (Kay, 1978;
657 Yogodzinski et al., 2015; Yogodzinski et al., 1995; Yogodzinski et al., 2001). As such, this
658 section of the subduction zone represents a thermally anomalous environment where the
659 uppermost portions of the subducted slab (sediments and some AOC) could have been removed
660 by previous generations of magmatism, leaving the residue almost devoid of sediment (Kelemen
661 et al., 2003). In addition to trace element (high Sr/Y) and radiogenic isotope data (Sr, Nd, Pb)
662 that suggest slab melting is occurring there (Yogodzinski et al., 2015), the lack of heavy Tl
663 isotope compositions in the Western Aleutians also imply that sediment is not actively involved
664 in arc magma generation in this location (Nielsen et al., 2016). As discussed above, lavas from
665 Costa Rica also display a significant population of light Tl isotope compositions (Fig. 8a). These
666 could be related to the subduction of the Cocos Ridge and/or seamounts to its North, which
667 results in a thinner sediment cover for this portion of the subducted plate. In addition, sediments
668 subducted underneath Costa Rica are dominated by carbonates with very low Tl concentrations.
669 Hence, a minor sediment component in Costa Rica arc lavas might not dominate the overall Tl
670 budget.

671 The absence of light Tl isotope compositions in Central and Eastern Aleutians, Tonga-
672 Kermadec, Marianas and Central America north of Costa Rica is most easily explained if the Tl
673 budgets of arc lavas almost ubiquitously contain sediment, which is in agreement with the
674 subducted inputs in these arcs (Plank and Langmuir, 1998). Given that the Tl concentration of
675 most sediment types (excepting biogenic carbonate and opal) are more than an order of
676 magnitude larger than AOC (Coggon et al., 2014; Nielsen et al., 2006b; Nielsen et al., 2016;
677 Prytulak et al., 2013; Rehkämper et al., 2004) even small amounts of sediment would render this
678 component the dominant Tl contributor. However, Tl isotopes do not correlate with more

679 conventional indices of sediment or fluid contributions from the slab such as Th/La and Ba/Th
680 (Fig. 10). This lack of correlation could suggest that Tl fluxes from sediments are decoupled
681 from other sedimentary components, for example via the preferential mobilization of Mn oxides
682 where Tl is highly concentrated. Of course, both Tl isotope compositions and trace element ratios
683 of subducted sediments vary significantly (Nielsen et al., 2016; Plank and Langmuir, 1998;
684 Prytulak et al., 2013) such that mixing between mantle and different sediment components might
685 not be expected to generate globally significant correlations with Tl isotopes. Alternatively, it is
686 also possible that trace element signatures in arc lavas, in particular those that are fractionated in
687 subduction zones like Ba/Th, are decoupled from the process of mixing slab material with the
688 mantle wedge as inferred in models that invoke *mélange* melting (Marschall and Schumacher,
689 2012; Nielsen and Marschall, 2017).

690 Based on the currently available data for Tl isotopes in arc lavas, we conclude that there is
691 strong support for almost ubiquitous sediment involvement in arc magma genesis. Investigations
692 of Tl isotopes in additional volcanic arcs will further illuminate the effectiveness of Tl isotopes
693 to trace sediment recycling and whether this process is as widespread as current data suggests.

694

695 **6.2. Behaviour of Tl in slab material: fluids and accessory phases**

696 Although sediments, in most subduction zones, contain the bulk of Tl in subducted slabs, the
697 partitioning of Tl between slab and mantle wedge need not only be controlled by sediments.
698 Fluids and accessory phases that impart characteristic trace element fractionation observed in
699 subduction zones can originate from all portions of the subducted slab (Carter et al., 2015;
700 Hermann and Rubatto, 2009; Johnson and Plank, 1999; Kessel et al., 2005; Skora and Blundy,
701 2010). Here, we use new and previously published Tl concentration data for arc lavas to

702 investigate the controlling factors in determining Tl incorporation into arc magmas. When
703 plotting all arc lavas where fractional crystallization of phases such as clinopyroxene and
704 plagioclase are relatively minor (<55% SiO₂) a hyperbolic relationship between Cs/Tl and Sr/Nd
705 as well as Ba/Th ratios can be seen (Fig. 11). Thallium has previously been hypothesized as a
706 somewhat fluid mobile element (Noll et al., 1996), but the large ionic radius and univalent
707 charge of Tl likely also makes it compatible in minerals with large cation sites such as phengite
708 and phlogopite (Prytulak et al., 2013) that may be important phases in subducted slab material
709 (Schmidt and Poli, 1998). Natural data for previously subducted oceanic crust from Tian Shan
710 (van der Straaten et al., 2008) support this inference as Tl concentrations in these samples
711 correlate almost perfectly with K, Rb, Cs and Ba concentrations (Fig. 12) that are all controlled
712 by the abundance of phengite (van der Straaten et al., 2008). Based on these data, it follows that
713 Tl partitioning in subduction zones is likely strongly affected by the presence or absence of
714 phengite in the slab residue. The stability of phengite is itself strongly dependent on the presence
715 of fluids, with excess water lowering its stability or at least causing it to melt out at lower
716 pressures and temperatures (Hermann and Green, 2001). Therefore, high fluid abundances tend
717 to de-stabilize phengite at relatively lower temperatures and pressures. Thus, we interpret the
718 hyperbolic relationship between Cs/Tl and Ba/Th (Fig. 11) in terms of the stability of phengite
719 relative to the abundance of fluid present during melting. In this scenario, relatively lower Ba/Th
720 and higher Cs/Tl would indicate that phengite is stable in the arc lava residue. No known
721 subduction zone inputs display Cs/Tl as high as lavas in this portion of the diagram (Fig. 11),
722 which requires that Cs and Tl be fractionated during subduction processing. In both natural rocks
723 and experiments Cs has been shown to be the least compatible alkali metal in phengite (Busigny
724 et al., 2003; Hermann and Rubatto, 2009; Melzer and Wunder, 2000) and, given that the ionic

725 radius of Tl is most similar to that of Rb (Shannon, 1976), we predict that Tl is more compatible
726 in phengite than Cs. This difference in phengite partitioning can explain the unusually high Cs/Tl
727 ratios observed in many arc lavas (Figs. 5 and 11).

728 The absolute value of Ba/Th will be highly dependent on the initial Ba/Th ratio of
729 sediments, which can vary by more than an order of magnitude due to the presence of barite in
730 marine sediments (Plank and Langmuir, 1998). However, as the fluid abundance increases
731 phengite will de-stabilize and partitioning of elements like Ba, Cs and Tl will become entirely
732 dominated by the fluid, whereas Th will be controlled by partitioning into accessory phases like
733 monazite that can accommodate it (Hermann and Rubatto, 2009). Similarly, uniformly higher
734 Sr/Nd ratios relative to subduction zone inputs (Fig. 11b) is likely controlled by retention of Nd
735 in monazite or another REE-rich phase (Hermann and Rubatto, 2009; Skora and Blundy, 2012).
736 However, higher fluid fluxes could more effectively remove Sr from the slab (Kessel et al.,
737 2005) and thus further enhance the Sr/Nd fractionation. At high fluid fluxes, phengite
738 destabilizes and without a phase that can realistically retain Cs and Tl in the residue, the fluid
739 will most likely contain most of the Cs and Tl in the system and, therefore, obtain the Cs/Tl ratio
740 of the bulk slab material it was released from. This process would explain why Cs/Tl ratios are
741 generally in the range of subduction zone inputs for arc lavas that are characterized by high
742 Ba/Th and Sr/Nd. Although mechanistically somewhat different, this interpretation could be
743 compatible with models of arc trace element fractionation where fluid fluxes are relatively
744 uniform across arcs and variations in sediment abundance in the subducted slab controls ratios of
745 fluid mobile elements such as Sr/Nd and Ba/Th (Elliott, 2003). In this interpretation, high Sr/Nd
746 and Ba/Th would be observed for arc regions with low sediment abundances. However, low
747 sediment abundances would likely also produce less phengite that would probably melt out

748 earlier than in high sediment arc sections and thus produce low Cs/Tl. Hence, it is currently not
749 directly possible to distinguish between variations in sediment abundance in the slab (Elliott,
750 2003) and presence of excess fluid relative to phengite as implied here by Figure 11. In fact, the
751 two scenarios do not appear to be mutually exclusive.

752 Further support for the involvement of phengite in controlling the Tl concentration of arc
753 lavas is found in the slight tendency of heavy Tl isotope compositions in arc lavas to be
754 associated with higher Cs/Tl ratios (Fig. 5). Such a relationship is expected in arcs where
755 sediments contain high potassium concentrations (e.g. Aleutians, Tonga-Kermadec, Marianas),
756 which will favor phengite formation. Phengite is also stable in eclogitized oceanic crust (Carter
757 et al., 2015; van der Straaten et al., 2008) and metamorphosed continentally derived detrital
758 sediments (Hermann and Rubatto, 2009), which have $\epsilon^{205}\text{Tl} \leq -2$. Therefore, high Cs/Tl ratios
759 need not exclusively be associated with heavy Tl isotopes in order to be explained by residual
760 phengite.

761

762 **7. Conclusions and outlook**

763 We present Tl isotope data from the Tonga-Kermadec and Central American arcs. Lavas
764 from Tonga-Kermadec are offset in one direction from the mantle towards the heavy Tl isotope
765 compositions that are observed in pelagic clays in drill cores outboard of the arc. The Tl isotope
766 data show that sediments dominate the Tl budget of the subarc mantle in this arc.

767 The Tl isotope budgets of Central American lavas are influenced both by subducted
768 sediments and AOC. Specifically, we find evidence for involvement of hemipelagic sediments in
769 Central Nicaragua whereas Costa Rican lavas bear a strong influence from AOC. The lack of a
770 sediment signature in Costa Rica may be related to a thinner sediment where the Cocos ridge and

771 associated seamounts are subducted as well as the predominance of carbonates in the sediment
772 package that contain very low Tl concentrations.

773 Review of Tl isotope data from five arcs reveal that most lavas are displaced towards heavy
774 values that indicate involvement of pelagic sediments. Detection of AOC with Tl isotopes is very
775 limited, even for segments of arcs that have been hypothesized as dominated by AOC
776 components. The almost absent AOC signature for Tl isotopes is likely due to the much higher
777 Tl concentrations observed in pelagic sediments relative to AOC. Hence, Tl isotopic data for arc
778 lavas that imply involvement of sediments should not be seen as evidence against involvement of
779 AOC in their source regions. Rather, Tl isotopes appear to be an overall excellent tracer of
780 subducted sediment as long as these contain sufficient Mn oxides to generate Tl isotope
781 compositions heavier than the ambient mantle. Given that pelagic clays are very common within
782 the subducted sediment package in most subduction zones (Plank and Langmuir, 1998). Tl
783 isotopes promise to find utility in additional subduction zones. The almost ubiquitous sediment
784 signature for Tl isotopes in arcs suggests that most arc segments actively cycle sediments to
785 depth, which is consistent with observations of sediment subduction outboard of subduction
786 zones worldwide (Plank and Langmuir, 1998). Further studies of a wider range of arcs in which
787 sediment subduction might take place, will reveal if this conclusion holds true for subduction
788 zones in general.

789 Lastly, we suggest that Tl partitioning relative to Cs in arcs is controlled by residual
790 phengite during melting of slab components or, in cases where phengite is exhausted, fluids that
791 carry the bulk of Tl and the alkali metals with them resulting in little net fractionation between
792 Cs and Tl. This conclusion will benefit immensely from experimental verification involving
793 different slab components such as sediments, AOC, serpentinite and mélange.

794

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799

799 **FIGURE CAPTIONS**

800

801 Figure 1. Map of the Tonga-Kermadec arc. Islands from which we have analyzed samples
802 are marked with square symbols. Colours of symbols are the same as those used in Figures 3, [4](#),
803 [5](#), 6 and 7.

804

805 Figure 2. Map of the Central American arc. All volcanoes from which we have analyzed
806 samples are marked with coloured square symbol and the name of the volcano is indicated next
807 to the symbol. Symbol colours are also used in Figures 3, 4, 5 and 8.

808

809 Figure 3. Thallium concentrations plotted against MgO in lavas from (a) Tonga-Kermadec
810 and (b) Central America.

811

812 Figure 4. Ce/Tl plotted against Th/Rb ratios for lavas from Tonga-Kermadec and Central
813 America. In (a) is also shown two lavas, one from Hawaii (Nielsen et al., 2006a) and one from
814 the Aleutians (Nielsen et al., 2016), that were affected by subaerial alteration and loss of alkali
815 metals and Tl. (b) is a close-up of the area that contains all the arc lavas from [Central America](#).
816 [\(c\) close-up of the area that contains all the arc lavas from Tonga-Kermadec](#). Unaltered arc lavas
817 from the Aleutians plot inside the pink field for comparison. [DMM field is based on the average](#)
818 [composition of global MORBs](#) (Jenner and O'Neill, 2012). [It is noteworthy that the Th/Rb ratios](#)
819 [of Galapagos lavas \(Th/Rb = 0.11 to 0.22 \(Saal et al., 2007\)\) overlap with the values found in](#)
820 [Irazu volcano](#).

821

822 Figure 5. Thallium isotope compositions plotted against Cs/Tl ratios for (a) Central America
823 and (b) Tonga-Kermadec lavas. Also shown are fields for Aleutians and Mariana arcs.

824

825 Figure 6. Thallium isotope compositions of Tonga-Kermadec lavas along-strike in the arc.
826 Also shown is the field for the depleted mantle (grey bar), pelagic clays (brown bar) and
827 volcanoclastic sediments (pink bar). Volcanoclastic sediment subduction is only taking place in
828 the Tonga portion of the arc and is, therefore, not extended to the Kermadec portion of the arc.
829 The isotope compositions of the sediment components were calculated as concentration-
830 weighted averages of the individual pelagic and volcanoclastic sediments analyzed (Table 3).
831 Altered oceanic crust (AOC) is isotopically light and is indicated by an arrow.

832

833 Figure 7. Thallium isotope composition plotted against (a) $^{208}\text{Pb}/^{204}\text{Pb}$, (b) $^{87}\text{Sr}/^{86}\text{Sr}$ and (c)
834 $^{143}\text{Nd}/^{144}\text{Nd}$ isotope compositions for the Tonga-Kermadec arc. Symbols for arc lavas are the
835 same as in figure 1. Bulk mixing lines between the mantle and the pelagic and volcanoclastic
836 sediment components are shown in bold and dashed black lines. Tick marks indicate the amount
837 of sediment required to produce the relevant Tl, Pb, Sr and Nd isotope variations. The mantle
838 wedge $[\text{Pb}] = 0.034 \mu\text{g/g}$, $[\text{Sr}] = 9.8 \mu\text{g/g}$, and $[\text{Nd}] = 0.713 \mu\text{g/g}$ (Salters and Stracke, 2004) and
839 $^{208}\text{Pb}/^{204}\text{Pb} = 37.7$, $^{87}\text{Sr}/^{86}\text{Sr} = 0.7025$, $^{143}\text{Nd}/^{144}\text{Nd} = 0.51315$ is estimated from the most depleted
840 samples found in the Eastern Lau Spreading Center and the Valu Fa Ridge (Hergt and
841 Woodhead, 2007; Pearce et al., 2007). The Pb, Sr, Nd and Tl isotope compositions of the
842 sediment components were calculated as concentration-weighted averages of the individual
843 pelagic and volcanoclastic sediments analyzed (Table 3). The concentrations of the two sediment
844 components were calculated using the concentrations averages of the individual pelagic and

845 volcaniclastic sediments analyzed (Table 3) and the data in Ewart et al (1998). Also shown is
846 bulk mixing line between mantle and Fe-Mn nodules (light brown) and the Louisville Seamount
847 Chain (LSMC) in green. Concentration and isotope data for average LSMC rocks are averages of
848 published data (Beier et al., 2011; Vanderkluyesen et al., 2014) that yield values of [Pb] = 2.15
849 μg/g, [Sr] = 560 μg/g, and [Nd] = 36 μg/g and $^{208}\text{Pb}/^{204}\text{Pb} = 39.25$, $^{87}\text{Sr}/^{86}\text{Sr} = 0.7039$,
850 $^{143}\text{Nd}/^{144}\text{Nd} = 0.5129$. The Tl isotope composition for LSMC is estimated to be similar to AOC.
851 Tick marks indicate the amount of bulk sediment or LMSC material required to produce the
852 observed Pb, Sr, Tl and Nd isotope variations. Concentrations and isotope compositions for the
853 Fe-Mn nodules were assumed to be [Pb] = 1000 μg/g, $^{208}\text{Pb}/^{204}\text{Pb} = 38.7$, [Sr] = 1500 μg/g,
854 $^{87}\text{Sr}/^{86}\text{Sr} = 0.709$, [Nd] = 200μg/g, $^{143}\text{Nd}/^{144}\text{Nd} = 0.5124$, [Tl] = 100 μg/g, $\epsilon^{205}\text{Tl} = +12$.

856 Figure 8. Along-strike variation of (a) Tl isotopes, (b) ^{10}Be and (c) Ba/La ratios for the
857 CAVA. Tl isotopes and Ba/La ratios are for the same samples, whereas ^{10}Be data is a
858 compilation of all available data from CAVA (Carr et al., 2014). Grey field is the isotope
859 composition of the upper mantle as represented by MORB (Nielsen et al. 2006a). Most volcanic
860 centers have only a single sample. Circles are drawn around those that have two or more
861 samples. All lavas are from composite volcanoes of the active volcanic front except for 1) three
862 Guatemala lavas from behind the front, designated 'BVF' and with a small black circle and 2)
863 three samples from fault-aligned cinder cones of the Nejapa and Granada regions, between two
864 of the CAVAs linear volcanic segments. Samples where both ^{10}Be and Tl isotopes have been
865 measured are marked with a cross. There is no correlation between Tl isotopes and ^{10}Be even
866 though there is a tendency for Tl isotopes and ^{10}Be to be most enriched in the Central Nicaragua

867 region of the arc. Two-sigma standard deviation uncertainty on Tl isotope measurements is ± 0.4
868 epsilon units.

869

870 Figure 9. Probability distribution plotted for arc lavas from Aleutians, Marianas, Central
871 America and Tonga-Kermadec. Shown is also the field for the mantle (grey bar) and the general
872 direction that the isotope compositions of AOC and sediments plot in.

873

874 Figure 10. Thallium isotope compositions in arc lavas from Aleutians, Marianas, Central
875 America and Tonga-Kermadec plotted against (a) Ba/Th and (b) Th/La. These trace element
876 ratios have been suggested as indicators of fluid (Ba/Th) and sediment (Th/La) addition from the
877 slab to the mantle wedge (Elliott, 2003; Elliott et al., 1997; Plank, 2005; Woodhead et al., 1998).
878 Fields for average subducted sediment compositions are shown as boxes with same colors code
879 as the arc lavas. Sediment data are from (Nielsen et al., 2016; Plank and Langmuir, 1998;
880 Prytulak et al., 2013) and this study. Fields are also shown for DMM (Jenner and O'Neill, 2012;
881 Nielsen et al., 2014) and AOC (Coggon et al., 2014; Kelley et al., 2003; Nielsen et al., 2006b).

882

883 Figure 11. Ratio of Cs/Tl plotted against (a) Ba/Th and (b) Sr/Nd for lavas from Aleutians,
884 Marianas, Central America and Tonga-Kermadec. In (b) are also shown fields for the
885 compositions of the mantle (Jenner and O'Neill, 2012), detrital and pelagic sediments (Plank and
886 Langmuir, 1998) and AOC (Coggon et al., 2014; Kelley et al., 2003; Nielsen et al., 2006b).
887 Fields for the sediment endmembers are not shown for Ba/Th because sediment Ba
888 concentrations are extremely variable due to influence of barite deposition (Plank and Langmuir,

889 1998). Red arrows are also shown that indicate the general areas that arc lavas are expected to
890 occupy at different degrees of phengite and fluid involvement.

891

892 Figure 12. Thallium concentration plotted against (a) barium, (b) cesium, (c) potassium in
893 eclogites from Tian Shan (van der Straaten et al., 2008). The excellent covariations show that all
894 these elements are controlled by the only K-bearing phase in the rocks, which is phengite.

895

896

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1220 Table 1: Isotope data for Tonga-Kermadec lavas

Sample	Island	Rock type	$^{87}\text{Sr}/^{86}\text{Sr}$	$^{143}\text{Nd}/^{144}\text{Nd}$	Tl (ng/g)	$\epsilon^{205}\text{Tl}$	<u>n</u>
<i>Tonga</i>							
T068	Tafahi	BA	<i>0.70390</i>	0.51293	12.1	2.0	<u>1</u>
T071	Tafahi	B	0.70390	0.51295	5.4	2.2	<u>1</u>
T072	Tafahi	B	<i>0.70392</i>	0.51295	10.2	2.2	<u>1</u>
T073	Tafahi	BA	<i>0.70395</i>	0.51296	14.2	0.6	<u>1</u>
NT052A	Niuaotoputapu	BA	<i>0.70404</i>	0.51290	10.3	0.2	<u>1</u>
NT054	Niuaotoputapu	BA	<i>0.70404</i>	0.51290	23.4	1.4	<u>1</u>
NT059	Niuaotoputapu	BA	0.70404	0.51289	21.5	1.3	<u>1</u>
F8	Fonualei	D	<i>0.70383</i>	<i>0.51295</i>	117	1.1	<u>1</u>
F20	Fonualei	D	<i>0.70372</i>	0.51293	117	0.2	<u>1</u>
F30	Fonualei	A	<i>0.70351</i>	<i>0.51297</i>	100	0.8	<u>1</u>
F31*	Fonualei	A	<i>0.70392</i>	<i>0.51297</i>	17	-0.6	<u>2</u>
L3*	Late	BA	<i>0.70367</i>	<i>0.51298</i>	33	7.4	<u>2</u>
L13	Late	BA	<i>0.70364</i>	0.51297	144	0.6	<u>1</u>
L20	Late	BA	<i>0.70355</i>	0.51298	31.4	1.3	<u>1</u>
L21*	Late	BA	<i>0.70382</i>	<i>0.51297</i>	42.4	0.6	<u>2</u>
T101P	Kao	BA	<i>0.70334</i>	0.51303	32.1	-0.4	<u>1</u>
T102*	Kao	BA	<i>0.70329</i>	0.51303	33.8	0.8	<u>2</u>
TOF32*	Tofua	A	<i>0.70350</i>	<i>0.51302</i>	46.3	3.7	<u>2</u>
HHBF	Hunga Ha'apai	BA	<i>0.70376</i>	<i>0.51296</i>	106	-0.2	<u>1</u>
38983	Hunga Tonga	BA	<i>0.70365</i>	0.51299	47.7	0.2	<u>1</u>
<i>Kermadec</i>							
14782	Raoul Group	B	<i>0.70360</i>	<i>0.51300</i>	6.4	0.4	<u>1</u>
23374	Raoul Group	BA	<i>0.70347</i>	0.51305	21.3	0.7	<u>1</u>
23383*	Raoul Group	BA	<i>0.70376</i>	0.51302	12.5	2.0	<u>1</u>
23386*	Raoul Group	A	<i>0.70355</i>	0.51305	11.3	2.4	<u>2</u>
10379	Macauley	B	<i>0.70347</i>	<i>0.51301</i>	15.6	4.7	<u>2</u>
10380	Macauley	B	<i>0.70350</i>	0.51300	11.7	4.3	<u>2</u>
14849	Curtis	BA	<i>0.70410</i>	<i>0.51301</i>	39.2	-0.5	<u>1</u>
14864	Curtis	D	<i>0.70408</i>	<i>0.51301</i>	75.4	-0.1	<u>1</u>
14831	L'Esperance	BA		0.51297	22.4	5.3	<u>2</u>
14837*	L'Esperance	BA	<i>0.70402</i>	<i>0.51297</i>	21.3	12.3	<u>2</u>

1221 Numbers in italics denote data from (Ewart et al., 1998; Regelous et al., 1997; Regelous et al.,

1222 2010). B - basalt; BA - basaltic andesite; A - andesite; D - dacite

1223 * - samples that were received as chips (see text for description of sample processing)

1224 n - number of total procedural repeats (dissolution, chemical separation, mass spectrometry)

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1226 Table 2: Major element data for Tonga-Kermadec lavas in wt%

	SiO ₂	TiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MnO	MgO	CaO	Na ₂ O	K ₂ O	P ₂ O ₅	LOI	SUM
T068 Tafahi	52.52	0.360	16.83	11.16	0.195	6.62	11.52	0.358	0.185	0.117	<0.01	99.86
T071 Tafahi	51.53	0.355	17.52	10.85	0.189	6.65	12.21	0.312	0.163	0.091	<0.01	99.87
T072 Tafahi	51.85	0.347	17.00	10.70	0.191	6.93	12.26	0.312	0.172	0.092	<0.01	99.85
T073 Tafahi	52.51	0.435	16.96	11.57	0.198	5.81	11.46	0.594	0.186	0.107	<0.01	99.82
T052a Niuatoputapu	54.57	0.412	15.58	12.50	0.204	4.73	10.47	0.858	0.363	0.132	<0.01	99.82
T054 Niuatoputapu	54.18	0.305	15.39	10.46	0.192	7.10	11.30	0.316	0.321	0.105	0.17	99.84
T059 Niuatoputapu	54.13	0.303	15.22	10.83	0.192	7.23	11.10	0.347	0.321	0.109	0.05	99.83
FON-8 Fonualei	65.22	0.579	13.37	8.58	0.191	1.56	5.78	3.20	1.13	0.262	<0.01	99.87
FON-20 Fonualei	65.03	0.545	13.86	8.55	0.182	1.52	5.79	3.09	1.10	0.209	<0.01	99.87
FON-30 Fonualei	60.28	0.633	14.40	10.95	0.204	2.60	7.42	2.34	0.846	0.191	<0.01	99.86
FON-31 Fonualei	60.33	0.629	14.21	11.03	0.208	2.61	7.39	2.42	0.803	0.226	<0.01	99.86
Late 3	54.31	0.587	15.78	11.41	0.197	5.17	10.38	1.30	0.518	0.178	<0.01	99.83
Late 13	57.09	0.778	14.20	13.00	0.213	3.26	8.22	2.20	0.664	0.186	<0.01	99.81
Late 20	53.58	0.518	16.21	10.90	0.192	5.82	11.14	0.958	0.413	0.118	<0.01	99.84
Late 21	53.47	0.578	16.51	11.56	0.196	4.85	10.83	1.22	0.456	0.154	<0.01	99.82
T101P Kao	52.77	0.805	17.46	10.95	0.171	4.28	10.81	1.82	0.542	0.233	<0.01	99.84
T102 Kao	53.39	0.766	16.29	11.36	0.188	5.37	10.06	1.72	0.488	0.189	<0.01	99.83
TOF-32 Tofua	57.07	0.714	14.66	11.46	0.200	4.04	9.01	1.88	0.597	0.208	<0.01	99.84
HHBF Hunga Ha'apai	54.54	0.492	17.79	9.74	0.167	4.23	11.34	1.09	0.338	0.123	<0.01	99.85
38983 Hunga Tonga	55.70	0.578	14.87	11.28	0.196	5.05	10.26	1.29	0.435	0.160	<0.01	99.82
14782 Raoul	48.90	0.722	16.01	11.98	0.201	7.36	12.07	1.93	0.117	0.143	0.41	99.83
23374 Raoul	52.28	0.599	15.76	9.87	0.192	7.16	11.79	1.84	0.252	0.128	<0.01	99.87
23383 Raoul	53.03	0.646	17.29	10.40	0.178	5.14	10.72	2.12	0.200	0.105	0.05	99.87
23386 Raoul	56.38	0.703	17.51	9.05	0.179	3.34	9.79	2.54	0.260	0.143	<0.01	99.89
10379 Macauley	48.01	0.619	19.73	9.75	0.174	5.80	13.33	1.91	0.382	0.149	<0.01	99.85
10380 Macauley	48.56	0.861	14.89	13.40	0.229	7.05	12.07	2.11	0.454	0.170	<0.01	99.79
14849 Curtis	52.46	0.503	15.52	11.02	0.188	5.76	10.97	1.77	0.403	0.124	1.13	99.85
14864 Curtis	65.55	0.542	14.73	5.49	0.142	1.70	5.47	3.59	1.20	0.182	1.29	99.88
14831 L Esperance	52.86	1.02	16.56	13.54	0.258	4.62	9.59	0.341	0.220	0.547	0.24	99.79
14837 L Esperance	51.67	0.982	16.39	13.44	0.238	4.984	10.11	1.506	0.360	0.172	<0.01	99.85

1227 LOI - loss on ignition

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1230 Table 3: Thallium isotope compositions and concentrations for Tonga-Kermadec sediments

Site	Core	Section	Interval (cm)	Depth (m)	Tl (ng/g)	$\epsilon^{205}\text{Tl}$
<i>Pelagic sediments</i>						
204	1	1	30-33	0.3	834	1.0
204	1	3	52-55	3.52	681	1.5
204	5	5	60-63	100.6	2929	4.5
596	1	1	115-118	1.15	1556	3.6
596	2	3	85-88	9.35	1254	2.3
596	2	6	77-80	13.77	3067	6.6
596	3	6	84-87	23.44	3746	2.3
596	6	6	124-127	48.54	940	1.1
Average					1876	3.6
<i>Volcaniclastic sediments</i>						
204	1	4	103-106	5.53	328	0.0
204	3	2	103-106	50.53	151	-0.7
204	6	2	66-69	105.16	184	1.6
204	9	3	67-70	144.67	192	0.7
Average					214	0.4

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1233 Table 4: Thallium isotope compositions and concentrations for Central American lavas and
 1234 sediments

Sample	Volcano	Tl (ng/g)	$\epsilon^{205}\text{Tl}$	n
<i>lavas</i>				
CR-IZ-63A	Irazu	81	-1.4	2
CR-IZ-D6	Irazu	243	-8.2	2
CR-B2	Barba	39	-2.0	2
CR-PO10	Sabana Redonda	76	-3.8	1
CR-PP6	Platanar	91	-2.5	2
CR-124 (LA)	Arenal	15	-0.8	1
CR-123 (LA)	Arenal	16	-1.9	4
CR-TE9	Tenorio	36	-3.6	1
NIC-GR5	Granada	5.8	3.5	1
NIC-GR3	Granada	34	4.7	1
NIC-MS7	Masaya	29	-1.2	4
NIC-NE201	Nejapa	4.5	9.0	1
NIC-NE202	Nejapa	8.7	-11.5	3
NIC-NE5	Nejapa	14	1.0	1
NIC-NE203	Nejapa	nd	0.1	1
NIC-CN10	Cerro Negro	46	-0.1	2
NIC-TE1	Telica	92	-1.1	1
NIC-TE6	Telica	48	-1.8	1
NIC-TE123	Telica	75	-2.0	2
NIC-SC2	San Cristobal	34	-1.9	2
NIC-COS9A	Cosiguina	78	-1.8	2
SAL-CON-3	Conchagua	40	-3.1	2
SAL-SM-7	San Miguel	46	-1.6	2
SAL-AP-3	Apastapeque	134	-4.6	2
SAL-B-21	Boqueron	145	-1.4	2
SAL-IZ108	Izalco	59	2.0	2
SAL-CV1	Cerro Verde	60	-1.7	2
GU-T302	Tecuamburro	40	-1.3	2
GU-C201	C. Mongoy	51	-2.5	1
GU-C303	C. del Cemeterio	26	0.9	1
GU-E1	Pacaya	61	1.1	2
GU-C837	Cerro Las Olivas	71	-1.6	1
GU-74-27 (VF)	Fuego	35	-0.1	2
GU-SM126	Santa Maria	42	-1.3	2
<i>sediments, DSDP 495</i>				
41-43	hemipelagic	1077	0.2	1
111-113	carbonate	33	5.8	1

1235 n - number of isotopic measurements

1236 nd - not determined

1237 Major and trace elements and isotope ratios for these samples can be found in (Carr et al., 2014)























