

Contents lists available at [ScienceDirect](http://ScienceDirect.com)

Materials Letters

journal homepage: www.elsevier.com/locate/mlblue

Laser surface melting of stainless steel anodes for reduced hydrogen outgassing

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ARTICLE INFO

Article history:

Received 1 June 2016

Received in revised form 11 December 2016

Accepted 28 December 2016

Available online 29 December 2016

Keywords:

Grain boundaries

Laser processing

Metals and alloys

Vacuum electronics

Hydrogen

Outgassing

ABSTRACT

Anodes of 304 stainless steel have been processed by a continuous wave Yb fiber laser with a wavelength of 1.064 μm and subjected to 50 keV electron bombardment in order to determine the extent to which hydrogen outgassing is reduced by the laser surface melting treatment. The results show a reduction in outgassing, by approximately a factor of four compared to that from untreated stainless steel. This is attributed to a reduction in the number of grain boundaries which serve as trapping sites for hydrogen in stainless steel. Such laser treated anodes do not require post-processing to preserve the benefits of the treatment and are excellent candidates for use in high power source devices.

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1. Introduction

High Power Source (HPS) devices are used in numerous applications including vacuum electronics [1–3], particle acceleration [4,5], and microwave generation [2,6]. Stable, long term HPS operation is presently constrained by pulse shortening due to plasma formation in the anode-cathode gap region. Plasma is formed through the interaction of secondary electrons and gas molecules (primarily hydrogen), both of which are released by the impact of high energy electrons at the anode surface. Metals with relatively low hydrogen outgassing rates, such as austenitic stainless steel (SS) [7], are the most commonly used materials for vacuum applications [7]. To further reduce outgassing in such metals, several treatments have been proven effective including baking [8–12], vacuum baking [8,11,13,14], polishing [8,14], and surface treatments to create oxide or other protective surface films. Electropolishing has been the method of choice [8,11,15] since it both reduces the surface roughness of anode materials and creates an oxide layer that further reduces hydrogen outgassing. However, electropolishing may also introduce hydrogen and other contami-

nants into the surface layers in significant quantity [15] and may necessitate an additional bake to thoroughly degas the surface [8].

The purpose of the work described here was to determine the feasibility of reducing hydrogen outgassing by processing anodes of 304SS by laser surface melting (LSM). The LSM processing technique entails irradiating a sample with the output of a high energy continuous laser beam, thereby causing melting, flow and re-solidification of the material as the laser beam is scanned across the sample surface. This process has the potential of reducing outgassing by forming a more crystalline layer (with fewer grain boundaries), thereby reducing the number of potential sites that can trap hydrogen in the metal [16]. When compared to more conventional processing techniques, such as electropolishing, the LSM process introduces significantly fewer contaminants (especially hydrogen) into the anode surface. We show here that LSM processing of 304SS does indeed lead to reduced H_2 outgassing during electron bombardment.

2. Experimental

Samples of 304SS were irradiated at normal incidence by a non-polarized Continuous Wave (CW) SPI™ G3 Yb fiber laser ($M^2 = 2$, input beam diameter 4.3 mm), with a wavelength of 1.064 μm ,

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maximum output power of 20 W and nominal spot size of 39.4 μm . The treatment was carried out at atmospheric pressure under constant N_2 flow into the capped stage (O_2 levels $<0.2\%$), as shown in Fig. 1. The beam expander telescope (BET) supplied an input beam of 4.3 mm in diameter to the laser head.

The lens used in the laser setup was a Jenoptic™ fused silica lens with focal length of 125 mm. For patterning the sample, a bidirectional raster scan was applied with a line separation of 30 μm and irradiated with an average laser energy density (ED) of 13.54 kJ/cm^2 . Additional details of the LSM processing can be found elsewhere [17]. An Olympus BX51™ optical microscope with JENOPTIC™ ProgResC10+CCD camera was used to obtain images and depth measurements of the treated samples. For the depth measurements, the SS samples were cut along the laser-scanning track and mechanically polished using standard metallographic techniques. The samples were chemically etched in SS micro-etchant, chemical composition 10 g FeCl_3 , 30 ml HCl , 120 ml water, at room temperature to reveal the general microstructure. Scanning electron microscopy (SEM) images were acquired with a FEI™ Quanta 3D system equipped with a field emission gun (FEG). Microstructural characterization was conducted with the help of focused ion beam (FIB) microscopy and Philips XL30 SEM with FEG in secondary electron mode to obtain orientation maps. HKLTango™ software was used to quantify the grains and grain boundaries (GBs). GBs were categorized in two groups, special ($3 < \Sigma \leq 29$) and random ($29 < \Sigma \leq 49$), where Σ is the reciprocal of the fraction of the common lattice sites (CSL) from each grain at the boundary [18]. More restrictive Palumbo–Aust criterion [19] is used to determine the Σ number.

Outgassing characterization was carried out by bombarding at normal incidence the SS samples with the focused output of a 50 keV electron beam with a spot size of 1.6 mm in diameter (determined by measuring the size of a hole formed in a thin Ni foil under conditions identical to those used for the present work) with 60 s duration current pulse, and recording the time evolution during the pulse of the H_2 signal with a residual gas analyzer situated 45 degrees from the surface normal. The electron current density at the sample surface was approximately 16.4 mA/cm^2 , and the base pressure was 5×10^{-10} Torr.

3. Results & discussion

In order to form samples for depth characterization and outgassing evaluation, the SS samples were processed by raster scanning the laser beam across the surface in a uniform pattern. Shown in Fig. 2a is an SEM image of the raster scanned surface of the SS sample, tilted at 45 degrees to the electron beam column; the LSM-treated and untreated areas are visible. The individual laser tracks can be seen as can the ripples within each track that are ori-

ented in the direction of the laser beam scan; a phenomenon in CW laser melting commonly observed at high laser ED values [20]. Processing with lower ED tends to avoid ripple formation in the irradiated area but also reduces grain growth in the lattice of the sample. There are also small, bump-like formations seen within the treated area. Note that the features extend above the plane of the laser treated area and occur near the overlap between adjacent laser scan lines. These features represent areas of the SS surface morphology that were incompletely melted by LSM and most likely result from the specific choice of scan parameters (laser spot size and degree of overlap between adjacent scan lines) for the results shown here. These features can be minimized by a different choice of initial scan parameters or by subsequent laser passes over the sample. Shown in Fig. 2b is a longitudinal cross-section (optical) view of the treated sample, where the treated region is comprised of an LSM zone (LSMZ) and heat-affected zone (HAZ). The depth of the LSMZ is approximately 9.7 μm .

Shown in Fig. 3a and b are FIB images of the untreated and treated samples, respectively, demonstrating the microstructural changes induced by the laser radiation. It can be observed that the grains in the laser treated area, Fig. 3b, are elongated in the direction of the laser scan and have increased in size as a result of laser processing, as detailed in Table 1. The total number of grains was reduced from 1020 to 617 per 0.12 mm^2 unit volume. This transformation has the net effect of reducing the number of special and random GBs in the laser-treated volume and supports the outgassing data showing less H_2 released from such samples. Table 1 suggests that it is the grain boundary character distribution, i.e., the spectrum of misorientations and inclinations which is changed as a result of laser melting. Increasing further the grain size would entail extending the surface cooling time [21]. In fact, as shown in Fig. 2, during laser melting only the thin surface layer is melted which then rapidly solidifies due to the high heat outflow toward the solid substrate. Therefore, we believe that in our case the spectrum of misorientations and inclinations corresponds to a temperature close to the melting temperature T_m . However, it is well known that the equilibrium grain boundary character distribution changes significantly with temperature [22,23]. For instance, in conventional solidification, the temperature actually decreases slowly, and the spectrum of grain boundary misorientations and inclinations corresponds to a lower temperature of about 0.5 T_m . To optimize the reduction of hydrogen outgassing from stainless steel anodes using the laser surface treatment reported here, a more thorough analysis would require a study of the grain boundary character distribution as a function of the laser spot size and energy density [22,23].

Shown in Fig. 4 are the outgassing results, which are presented as the change in H_2 partial pressure (above baseline) with electron dose. Both curves exhibit an increase in H_2 signal with electron dose. Note that the H_2 signal for the untreated sample does not immediately rise with electron dose but instead shows an incubation period (extending from a dose of 0 to $\sim 3 \times 10^{18} \text{ cm}^{-2}$) in which the H_2 signal first increases gradually, after which there is a more rapid, nonlinear increase. Hydrogen is present within many metals, where it exists as atomic H at defects such as GBs [16]; Hydrogen outgassing from metals occurs [24] through a series of steps consisting of (a) diffusion of atomic H to the metal surface, (b) recombination of atomic H at the surface and (c) desorption of nascent, molecular hydrogen H_2 into the gas phase. The incubation period most likely is comprised of step (a) in which the majority of incident electron flux goes toward heating the sample and enhancing H atom diffusion to the surface. Only after the induction period is there sufficient atomic H at the surface that steps (b) and (c) proceed more efficiently and a larger desorbed H_2 signal is detected. The data from the LSM-treated sample exhibits a smaller initial slope, suggesting a lower rate of diffusion of atomic H to the

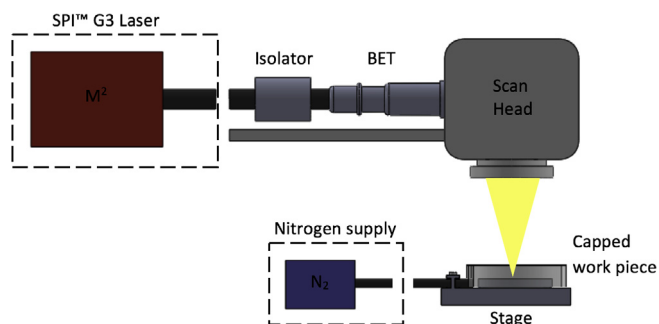


Fig. 1. Laser processing workstation showing the Yb fiber laser, Isolator for preventing unwanted feedback into the laser, BET, nitrogen supply and sample stage.

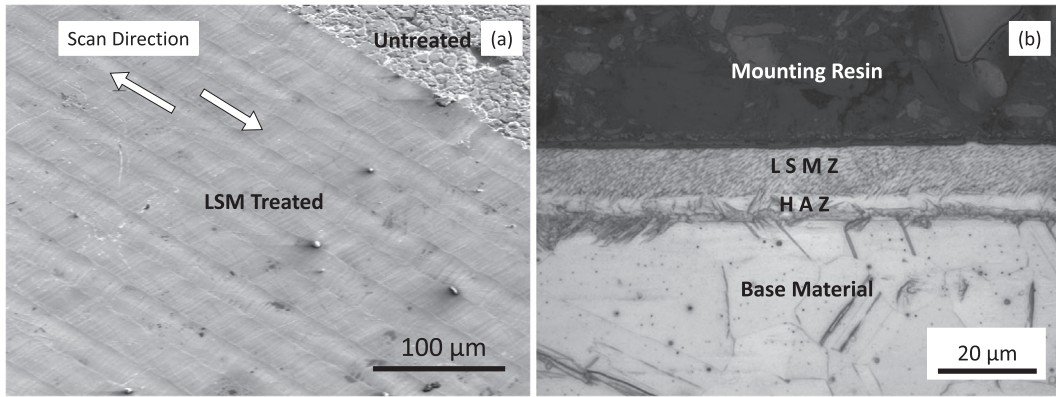


Fig. 2. (a) SEM image of the raster scanned SS sample with ED of 13.54 kJ/cm² tilted at 45 degrees to the electron beam column emphasizing the bump-like features and their occurrence at the overlap of laser scan lines, the arrows show the direction of the laser raster, and (b) optical micrograph of the section view showing the depth of the treated region.

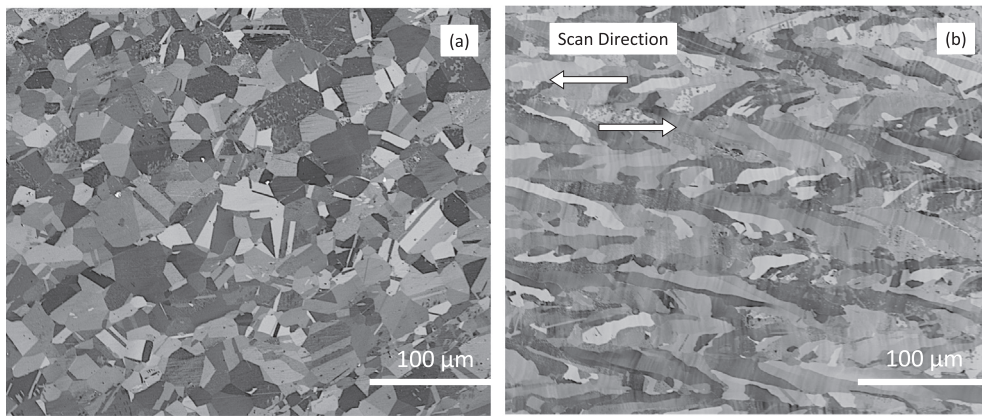


Fig. 3. FIB image showing the microstructure of the (a) untreated and (b) treated SS sample. The arrows show the direction of the laser raster.

Table 1
Grain boundary character distribution per 0.12 mm² surface area.

Position	$\Sigma 3$	$\Sigma 9$	$3 < \Sigma \leq 29$	$29 < \Sigma \leq 49$	Average grain size (μm)	Average grain area (μm^2)
Base material	33.81	1.43	35.11	0.07	10.72	99.88
LSMZ	15.39	0.82	16.66	0.05	14.12	201.59

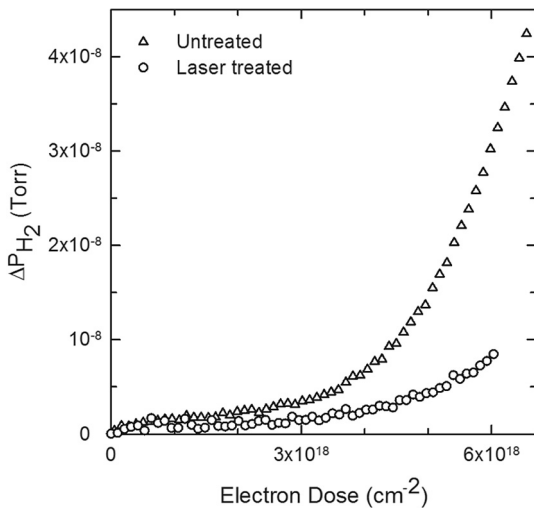


Fig. 4. Outgassing results showing the change in H₂ partial pressure with electron dose during 60 s electron irradiation.

surface because of the decreased number of GBs. The differences in these data sets are currently being investigated in more detail.

The data indicate that H₂ outgassing from the LSM-treated sample was approximately a factor four less than that from the untreated SS sample, for the conditions employed here.

4. Conclusions

304SS treated with CW, non-polarized Yb fiber laser radiation at an ED value of 13.54 kJ/cm² showed reduced hydrogen outgassing by a factor of ~4 at an electron dose of $6 \times 10^{18} \text{ cm}^{-2}$, indicating the feasibility of the LSM process for reducing hydrogen outgassing from SS anodes. Such laser treated anodes do not require post-processing to preserve the benefits of the treatment. The mechanism of suppression of hydrogen outgassing is caused by stimulating grain growth in the lattice of the specimen. Further studies are planned to maximize hydrogen outgassing reduction via grain boundary character distribution as a function of the laser spot size and energy density.

Acknowledgments

Work supported by US Air Force contract FA8650-11-D-5401 at the Materials & Manufacturing Directorate (AFRL/RXAP). The authors thank Lt Col Victor Putz of AFOSR/EOARD and Jason Marshall at AFOSR.D.G and M.S. wish to thank the EPSRC (EP/K503241/1).

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