pro

CRYSTAL STRUCTURE OF 1,2-CYCLO-PENTENOPHENANTHRENE

B. S. BASAK

PRESIDENCY COLLEGE, CALCUTTA

AND

M. G. BASAK

JHARGRAM RAJ COLLEGE, JHARGRAM, WEST BENGAL (Received, February 12, 1959)

ABSTRABT. The crystal structure of 1, 2-cyclo-pontenophenanthrene has been determined by the Fourier-synthesis method. A qualitative suggestion about the structure was made by Iball (1935) Following are the crystal data: Molecular formula C_{17} H_{14} ; Monoclinic system: a = 18.38 Å; b = 5.83 Å, c = 23 61 Å; $\beta = 114^{\circ}-18'$; space-group B_{21}/c (Real space-group $P_{21}/c-C_{24}$) · density = 1.23 gm/c.c. at 20°C; No. of molecules per unit cell ('b' face-centred cell) = 8.0.

Zero-layer and equi-inclination Weissenberg photographs were taken in a Unicam Weissenberg gomometer with copper radiation and the intensities estimated microphotometrically for stronger spots and visually for the weaker ones. The substance was kindly supplied by Dr. Devdas Mukherjee of Scottish Gas Board from Dr. Loudon of the Biochemistry Laboratory, Glasgow University.

In finding out the structure by trial, help was taken of the suggestion made by Iball (loc. cit). After a fairly satisfactory trial structure was arrived at, two projections were made about the a and b axes respectively. Because of the large values of a and c axial lengths, electron densities were calculated at intervals of 3° along them, using 3° strips of Lipson and Beevers. Values were calculated at 6° interval along b axis. Of the 17 carbon atoms, 13 were fairly well resolved in the b axis projection (figure 1). Only a very small number of atoms were resolved in the a axis projection, although the 'y' co-ordinates could be fixed to a fair degree of accuracy from the same. The considerations of the hydrogen atoms have been omitted for the present. Taking the new co-ordinates obtained from these projections, the values of structure factors were calculated when the reliability index R had the values 0.28 for both the $(o \ k \ l)$ and $(h \circ l)$ zones, where

$$R = \frac{\sum |F_{obs} - F_{calc}|}{\sum |F_{obs}|}$$



F1g. 1.

This high value of R is due most probably to the presence of a comparatively large number of very weak and absent reflections in these two zones. The coordinates of the carbon atoms of one asymmetric unit (one molecule) are given in Table I. The bond lengths and bond angles are all shown in figure 2. If χ_1, ψ_1, ω_1 be the angles made by the long axis of the molecule formed by joining atoms C_8 and C_{17} with the a, b and c axes respectively and χ_2, ψ_2 and ω_2 be the corresponding angles made by the short axis represented by the straight line joining C_{10} and C_{12} , then



F1g. 2.

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$\chi_1 = 109^{\circ}$	$\chi_2 = 121^\circ - 12'$
$\psi_1 = 90^{\circ}$	$\psi_2 = 32^{\circ} - 22'$
$\omega_1 = 5^{\circ} - 18'$	$\omega_2 = 98^{\circ} - 23'$

The minimum distance of approach between two molecules comes out to be 3.98 \AA .

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	C	Coordinates	of	\mathbf{the}	atoms	of	one	molecu	le
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Atom	x ā	y b	z
C_1	. 1204	.0616	.0656
C2	1629	1173	. 1060
\mathbf{C}_{a}	. 1948	- 2910	.0804
C4	.1827	2858	0174
C,	. 1656	2730	1063
C ₆	. 1565	2701	1675
C7	1130	0900	2060
C_8	. 0797	0851	1824
C _p	.0539	.2510	0991
C_{10}	.0638	.2410	0375
C ₁₁	. 1077	.0668	.0035
C_{12}	1403	1068	0212
C ₁₃	. 1303	1016	0832
C14	. 0895	.0773	1216
C15	.0884	.2177	.0993
C_{1G}	.1150	.1320	.1657
C17	.1596	0896	.1664

That the structure is substantially correct is confirmed from the following considerations; (i) The value of $\omega_1 = 5^{\circ} - 18'$ agrees very well with that of $\omega_1 =$ about 6° as predicted by Bernal (1935) from optical measurements, (ii) The 'b' axis projection of the molecule except the five-membered ring agrees very well with the corresponding electron density projection of phenanthrene crystal as obtained by Basak (1950). The structure obtained agrees with the suggestions put forward by Iball (loc.cit). The bond lengths, bond angles and intermolecular

distances all he within the limits of values generally obtained in other organic compounds of similar chemical formulae. The structure is now being refined with the help of latest techniques and when completed will give informations about the finer details.

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