1	Effect of ultrasound pretreatment and Maillard reaction on structure and antioxidant
2	properties of ultrafiltrated smooth hound viscera proteins-sucrose conjugates
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22	Ultrasound effect on ultrafiltrated peptides-sucrose properties
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25 Graphical abstract



46 Abstract

The effect of ultrasound (US) pre-treatment on the evolution of Maillard reaction (MR) induced between low molecular weight (LMW) peptides and sucrose was studied. LMW peptides (<1 kDa) were obtained by the ultrafiltration of smooth hound viscera protein hydrolysates, derived produced by Neutrase, Esperase and Purafect. MR was induced by heating the LMW peptides in the presence of sucrose for 2 h at 90 °C, without or with US pre-treatment. During reaction, a marked decrease in pH values, coupled to the increase in color of the Maillard reaction products (MRPs), were recorded. In addition, after sonication, the glycation degree was significantly enhanced in Esperase-derived peptides/sucrose conjugates (p<0.05). Moreover, results showed that US treatment reduced the bitter taste and enhanced the antioxidant capacities of the resulting conjugates. Hence, it could be concluded that US leads to efficient mixing of sugar-protein solution and efficient heat/mass transfer, contributing to increase the MR rate.

58 Keywords: Ultrasound; Maillard reaction; Low molecular weight peptides; Glycation degree; 59 Antioxidant activities.

69 **1. Introduction**

70 Among the numerous existing protein modification techniques, the glycation of proteins induced by sugar conjugation, via naturally-occurring Maillard reaction (MR), is commonly known 71 as the best process on improving functional properties of food proteins (Seo, Karboune, L'Hocine 72 73 & Yaylayan, 2013). Furthermore, the MR is one of the most complex reactions in food chemistry, 74 due to the variety of compounds generated throughout the reaction. In fact, the amine-carbonyl 75 condensation occurring during the early stages of the MR leads to form primordial intermediary products (Amadori products), which tended to be transformed during the advanced stages. These 76 77 products can undergo numerous reactions giving rise to protein-sugar conjugates and then the 78 formation of brown and polymeric products (Horvat & Jakas, 2004).

Different factors are involved in the MR evolution, the final glycation degree and MRPs 79 composition, including the protein/carbohydrate ratio, temperature, incubation time, pH, as well as 80 the interaction rate between the free amino groups of proteins and the carbonyl groups of sugars 81 (Lan et al., 2010). Besides conventional thermal treatment, the ultrasound (US) is a new processing 82 technology that has been recently reported to promote the interaction between proteins and 83 reducing sugars (Corzo-Martínez, Moreno, Megías-Pérez, Olano, Moreno & Villamiel, 2014). In 84 addition, it was reported that high-intensity sonication improve the propensity of proteins to 85 86 aggregate (Chandrapala, Zisu, Palmer, Kentish & Ashokkumar, 2011) and can modify the 87 secondary structure of proteins leading to increase their surface hydrophobicity (Stanic-Vucinic et al., 2012). On the other side, US treatment was found to modify the functional properties of 88 89 carbohydrates (Sun, Hayakawa & Izumori, 2004) and enhance the isomerization of glucose to fructose (Wang, Pan, Zhang, Sun, Fang & Yu, 2012). 90

Depending on the operating conditions, different MRPs with various biological and
functional properties and flavor sensory will be generated. Bioactive properties of MRPs include

93 the anti-cancer (Marko et al., 2003), anti-inflammatory (Chen & Kitts, 2015), antihypertensive (Rufia'n-Henares & Morales, 2007) and anti-oxidative (Kitts, Chen, & Jing, 2012) activities. 94 Matmaroh, Benjakul and Tanaka (2006) have shown that the capacity of MRPs to reduce reactive 95 oxygen species is influenced by various factors such as the concentration of reactants, duration of 96 97 reaction, temperature, starting pH, water activity and the presence of metal ions or salts. In addition, it has been previously reported that peptides below 1000 Da showed higher reactivity than high 98 molecular weight (HMW) peptides (Zhao, Zeng, Cui & Wang, 2007) and they are considered as a 99 key taste enhancer of flavor sensory, including umami, continuity and mouthfulness (Ogasawara, 100 101 Katsumata & Egi, 2006).

102 Proteins and peptides from animal origin are of great importance in food formulations such as sausages from poultry, bovine or porcine meats (Hathwar, Rai, Modi & Narayan, 2012; Nasri et 103 104 al., 2013). Particularly, proteins from fish meat or its by-products are considered as premium 105 quality proteins for their high content in essential amino acids and their numerous nutritional characteristics. In order to obtain new products from fish by-products, the major conventional 106 researches are still restricted to fishmeal production for animal feed (Péron, Mittaine & Le Gallic, 107 2010) and more recently for protein hydrolysates production (Abdelhedi et al., 2016; Zhou et al., 108 2012). Recently MR was described to modify odor in salmon byproducts hydrolysates (Christelle, 109 110 Jean-Pascal, Regis, Laurent, Carole & Mireille, 2014) and to enhance the physicochemical, antioxidant and hepato-protective properties of fish byproducts protein hydrolysate-ribose 111 conjugate (Yang, Lee, Pyo, Jeon, Kim & Lee, 2016). 112

Among byproducts, fish viscera represent a huge part of the total discards rejected in the local fish markets and fish processing industries. Smooth hound (*Mustelus mustelus*) is the most abundant hound shark in Tunisian coasts, with an interesting catch count of 192 tons (FAOSTAT, 2015). Regarding the huge quantities of visceral mass generated from freshwater fish marketing and the dearth of their scientific exploitation, the present investigation aims to obtain protein
hydrolysates from smooth hound viscera and their fractionation using ultrafiltration (UF) process
in order to produce low molecular weight peptides.

Because sucrose is the cheapest sugar and the most commonly used in food industries for the fabrication of sweet products, the fractionated peptides were thereafter conjugated to sucrose via the MR. On the other hand, the impact of US pre-treatment on the final products was also investigated by the study of the physicochemical, sensory, structural and antioxidant properties of the resulting MRPs.

125 **2. Materials and methods**

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2.1. Preparation of LMW peptides

Viscera were obtained following the processing of fresh filleted smooth hound (*M. mustelus*)
fish available in the local fish market of Sfax City, Tunisia. The biological material was brought to
the research laboratory in polyethylene bags, in iced conditions, within 30 minutes. Upon arrival,
they were immediately rinsed with tap water to remove contaminants, and then weighed and stored
in plastic bags at -20 °C until being used for protein hydrolysates elaboration.

132 Hydrolysis was carried out as previously described in our previou study (Abdelhedi et al., 2016). Smooth hound viscera were first cooked in distilled water at 95 °C for 15 min, with a 133 solid/solvent ratio of 1:1 (w/v), to inactivate endogenous enzymes. After being well homogenized, 134 135 the pH of the mixture was adjusted to the optimum value of the enzymatic activity by adding 4 N 136 NaOH solution. Thereafter, viscera treated mass was subjected to enzymatic hydrolysis, using three 137 different microbial enzymes, Neutrase® (pH 7.0), Esperase® (pH 9.0) and Purafect® (pH 10.0), 138 added at an enzyme/protein ratio of 6/1 (U/mg of protein). During the reaction (50 °C), the pH of 139 the mixture was maintained constant at the desired value by continuous addition of NaOH solution. 140 The degree of hydrolysis (DH) was calculated based on the volume of NaOH added during the141 reaction, as described by Adler-Nissen (1986) using the following formula:

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$$DH (\%) = \frac{h}{h_{tot}} \times 100 = \frac{B \times Nb}{MP} \times \frac{1}{\alpha} \times \frac{1}{h_{tot}} \times 100$$

where B is the amount of NaOH consumed (ml), *Nb* is the normality of the base, *MP* is the mass (g) of the protein (N = 6.25), and α represents the average degree of dissociation of the α -NH₂ groups in protein substrate expressed as:

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$$\alpha = \frac{10^{\text{pH-pK}}}{1+10^{\text{pH-pK}}}$$

where pH and pK are the values at which the proteolysis was conducted. The total number of peptide bonds (*h*tot) in protein substrate was assumed to be 8.6 meq/g (Adler-Nissen, 1986).

After the achievement of the digestion process, the reaction was stopped by heating the different solutions for 20 min at 95 °C to inactivate enzymes. Protein hydrolysates were then centrifuged at 9500 g for 20 min to separate soluble fractions (peptides) and insoluble fractions (non-hydrolyzed proteins). Hydrolysates obtained through the digestion of smooth hound viscera proteins with Neutrase, Esperase and Purafect were noted NH, EH and PH, respectively.

Hydrolysates were fractionated by ultrafiltration (UF), as illustrated in Fig. 1. Two successive 154 UF steps were applied on the liquid hydrolysates to obtain the LMW peptide fractions. First, a 155 tangential flow filtration (Millipore, Labscale[™] TFF System, USA) was performed using a 50 kDa 156 molecular weight cut-off (MWCO) membrane to remove HMW proteins and contaminants. Then, 157 permeates were separated by UF through 1 kDa MWCO membrane using a stirred cell (Amicon 158 159 Mini membrane system, 1-800-Millipore, USA). The inlet and the outlet pressure were 10 psi and 5 psi, respectively. UF experiments were carried out in batch mode, where the initial volume was 160 161 reduced to one-quart, *i.e.* to a volume concentration factor (VCF) = 4. After the UF processes, permeates (MW < 1 kDa) were freeze-dried (Bioblock scientific Christ ALPHA 1-2, IllKrich-
Cedex, France) and then stored at -20 °C until use.

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2.2. Preparation of Maillard reaction products

The conjugation between LMW peptides from smooth hound protein hydrolysates and 165 sucrose (S) was performed by suspending the freeze-dried peptide fractions and sucrose powder 166 (1:1; w/w) in distilled water (1:5; w/v). Then, the mixtures were homogenized at room temperature 167 168 $(25 \pm 1 \text{ °C})$ using a magnetic stirrer to form a uniform dispersion. After suspension, ultrasonic treatment (10 W/cm²) was performed at 40 °C for 30 min with a frequency of 25 kHz and a 169 maximum nominal power of 1,600 W, using SCIENTZ Electronics ultrasound device (JY98-3, 170 Ningbo, China). Thereafter, the mixtures were incubated at 90 °C during 2 h to induce MR. All 171 172 samples were then cooled immediately in iced water and then kept at 4 °C. The heated peptides-173 sucrose mixtures were termed as MRPs. A reference of lysine-sucrose conjugate (1:1, w/w) was prepared with the same manner and heated under the same conditions. 174

MRPs prepared from LMW-NH, LMW-EH and LMW-PH conjugated to sucrose without US
pre-treatment were named LMW-NH/S, LMW-EH/S and LMW-PH/S, respectively, while those
treated by US were referred to LMW-NH/S-Us, LMW-EH/S-Us and LMW-PH/S-Us, respectively.
The control groups consisted of the protein hydrolysate fractions (< 1 kDa) heated alone during 2
h at 90 °C without sugar addition or US pretreatment.

During heating, products were sampled every 30 min in order to evaluate physiochemical properties, color, free amino acid contents, taste score and structural changes produced owing to caramelization process. Antioxidant activities of the resulting MRPs were also investigated.

2.3. pH and browning intensity measurements of the MRPs

The pH values of MRPs, obtained at 0, 30, 60, 90 and 120 min of heating, were measured
using a 827-pH meter (Metrohm, Swiss made) calibrated with buffer solutions at pH 4.0 and 10.0.
The UV-absorbance and the browning intensity of MRPs were measured at 294 nm and 420
nm, respectively (Guan et al., 2011), using an UV-visible spectro-photometer (T70, UV/VIS
spectrometer, PG Instruments Ltd., China). Appropriate dilutions of the MRPs were made when
required. The analyses were carried out in duplicate.

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2.4. Color measurement of the MRPs

The color of the protein-sucrose conjugated samples was determined using a Color Flex spectrocolorimeter (Hunter Associates Laboratory Inc., Reston, VA, USA), readably calibrated using a white standard plate (L* = 92.84, a* = -1.25 and b* = 0.49). The results of the different samples were displayed as L*, a* and b* values which indicated lightness, redness and yellowness, respectively. L*, a* and b* values were determined in triplicate from three dependent samples. Whiteness (W*), differences in color (ΔE^*) and chroma (C*) were calculated using the following equations:

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$$W^* = 100 - \left[\left(100 - L^* \right)^2 + a^{*2} + b^{*2} \right]^{1/2}$$

200
$$C^* = (a^{*2} + b^{*2})^{1/2}$$

201
$$\Delta E^* = \left[\left(L^* - L_0 \right)^2 + \left(a^* - a_0 \right)^2 + \left(b^* - b_0 \right)^2 \right]^{1/2}$$

where L_0 , a_0 and b_0 are the color of the sample at t = 0.

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204 **2.5.** Free amino acids content and degree of glycation

205 Free amino acid contents of the freeze-dried LMW peptides and their MRPs (100 mg/ml) 206 were determined after being derivatizated with phenyl isothiocyanate (PITC) according to the method described by Aristoy and Toldrá (1991). The PITC derivates were quantified by reverse 207 phase HPLC with a 1200 Agilent liquid chromatography (Agilent Technologies, Palo Alto, CA, 208 209 USA) equipped with a diode array detector, using a PicoTagcolumn (300 mm \times 3.9 mm, Waters, USA). The temperature was set at 52 °C and the detection was carried out at 254 nm. The eluents 210 211 used were (A) 0.07 M sodium acetate adjusted to pH 6.55 and containing 2.5% acetonitrile and (B) 45:40:15 acetonitrile: water: methanol, with a flow rate of 1 ml/min. The following eluent 212 gradients: initially 0% B; several consecutive linear gradients of (B) as follows (3% at 13.5 min; 213 3.5% at 19 min; 4.5% at 21 min; 33% at 40 min; then a linear change to 40% at 50 min). Results 214 were expressed as mg of each amino acid per g of sample. For the glycation degree of the 215 conjugated protein hydrolysate fractions (*i.e.* after prolonged glycation reaction for 120 min), 216 results were expressed as the percentage of free amino acids content of the conjugates compared 217 with those of untreated protein hydrolysate fractions. 218

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2.6. Taste score of MRPs

The taste characteristics of the MRPs were assigned depending on the amino acid contents. Three categories of taste were selected (umami, bitter and sweet) using BIOPEP database (http://www.uwm.edu.pl/biochemia/index.php/pl/biopep) and bibliographic literature. Umami, bitter and sweet sensory attributes were evaluated based on Asp and Glu contents (Kabelová, Dvořáková, Čížková, Dostálek & Melzoc, 2009), Arg, Pro, Val, Ile, Leu, Lys, Phe and Trp contents (Yoshida & Saito, 1969) and Gly, Ala, Ser, Thr, Pro, Val and Lys contents (Yoshida & Saito, 1969), respectively.

227 **2.7. Fourier Transform Infrared (FTIR) analysis**

The infrared analysis was performed using the FTIR technique as previously reported by Liu, Kong, Han, Sun and Li (2014a) for MRPs. The spectroscopic measurements were performed using approximately 1 mg of each sample mixed with 100 mg of dried KBr The FTIR spectra at the mid IR region (500 and 4000 cm⁻¹) were recorded using a Thermo-Nicolet 5700 FTIR spectrometer equipped with an attenuated total reflection (ATR) accessory, with a resolution 4 cm⁻¹ and scan. The background noise was corrected using the data for pure KBr.

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2.8. Evaluation of the antioxidant activity

235 2.8.1. β-carotene bleaching assay

The prevention of β -carotene from bleaching was determined according to the method of 236 Koleva, van Beek, Linssen, de Groot and Evstatieva (2002). First, the emulsion of β -237 carotene/linoleic acid was freshly prepared by dissolving 0.5 mg of β-carotene, 25 µl of linoleic 238 239 acid and 200 µl of Tween 80 in 1 ml of chloroform. The chloroform was then completely evaporated under vacuum in a rotatory evaporator at 50 °C; then, 100 ml of distilled water were 240 added and the resulting mixture was vigorously stirred. Thereafter, 2.5 ml of the β-carotene/linoleic 241 242 acid emulsion were transferred to test tubes containing 0.5 ml of each sample. Control tube was prepared in the same conditions by adding 0.5 ml of water, instead of sample, to the emulsion. The 243 244 absorbance of every test tube was measured at 470 nm, before and after incubation for 1 to 2 h at 50 °C. The antioxidant activity was evaluated in terms of β -carotene bleaching and expressed as 245 μ M Trolox equivalent per mg of sample. The equation obtained for Trolox standard curve was y =246 0.0121 x + 0.4003 and the R² value was 0.99. The test was carried out in triplicate. 247

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249 2.8.2. Reducing power assay

The capacity of protein hydrolysate fractions and their MRPs to convert Fe³⁺ into Fe²⁺ was 250 251 determined as described by Yildirim, Mavi and Kara (2001). Samples (0.5 ml) were mixed with 252 1.25 ml of 0.2 M potassium phosphate buffer (pH 6.6) and 1.25 ml of 1% (w/v) potassium ferricyanide solution. After incubation for 30 min at 50 °C, 0.5 ml of 10% (w/v) trichloroacetic 253 254 acid was added and the reaction mixtures were then centrifuged at 2700 g for 10 min. Finally, 1.25 255 ml of the supernatant solution, from each sample mixture, was mixed with 1.25 ml of phosphate buffer (0.2 M; pH 6.6) and then 0.25 ml of 0.1% (w/v) ferric chloride was added. After a-10 min 256 reaction time, the absorbance of the resulting solutions was measured at 700 nm. The reducing 257 power was expressed as µM Trolox equivalent per mg of sample. The equation obtained for Trolox 258 standard curve was y = 0.0283 x + 0.0571 and the R² value was 0.99. The control was conducted 259 in the same manner, except that distilled water was used instead of sample. Values presented are 260 the mean of triplicate analysis. 261

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2.9. Statistical analysis

Data were expressed as mean \pm SD (Standard Deviation) and statistically analyzed using SPSS ver. 17.0, professional edition. A one-way analysis of variance (ANOVA) was then performed and means comparison was carried out by Duncan's multiple range test to estimate the significance among the main effects at the 5% probability level.

- **3. Results and discussion**
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3.1. Smooth hound viscera protein hydrolysates and MRPs preparation

Three protein hydrolysates from smooth hound viscera were elaborated using three different microbial enzymes, Neutrase, Esperase, and Purafect, with different final degrees of hydrolysis (DH) of 4%, 10% and 14%, respectively. The differences in the HD could be due to the difference in the specificity of enzymes used during hydrolysis. Hydrolysates were separated by UF using 1

273	kDa MWCO membrane and LMW peptides were recovered from NH, EH and PH with respective
274	yields of 13.25, 13.5 and 22.9 g per kg of fresh viscera. Thereafter, peptides (< 1 kDa) and sucrose
275	mixtures, with or without US-pretreatment, were heated at 90 °C for 2 h to induce MR.

3.2. pH change of the MRPs

The pH change of MRPs subjected or not to US pretreatment (25 kHz for 30 min at 40 °C) 277 was monitored to evaluate the MR evolution. As shown in Table 1, US pretreatment did not affect 278 the pH value of peptides-sucrose mixtures at t = 0, whereas the heat treatment induced its increase 279 during the first steps of the MR to be then gradually decreased (p<0.05) with advanced heating 280 time. MR is known to make changes in the final products (Nooshkam & Madadlou, 2016). 281 Meanwhile, for US pre-treated samples, the pH increased during the first 60 min of dry-heating 282 283 and then values were significantly decreased (p<0.05), which indicated that US induced the 284 acceleration of MR rate by enhancing the isomerization of glucose and fructose during the thermal hydrolysis of sucrose. In the same context, Wang, Pan, Zhang, Sun, Fang and Yu (2012) studied 285 286 the combination of US application and ionic liquid to enhance the enzymatic isomerization of glucose to fructose. Also, Corzo-Martínez, Moreno, Megías-Pérez, Olano, Moreno & Villamiel 287 (2014) investigated the effect of US on the isomerization of lactose, as a tool to accelerate the 288 289 formation of lactulose to enhance MR.

The initial heating stage resulted in the hydrolysis of intact peptides by thermal treatment coupled to the progressive exposure of hidden amino groups, leading to increasing pH values. However, in a later stage, the peptides-sugar conjugation took place and the pH decreased gradually, as it was expected during the MR evolution. This decrease might be the result of either, the release of organic acids or the consumption of free N-terminal amino groups of peptides during protein-sucrose conjugation (Jiang, Rai, O'Connor & Brodkorb, 2013). These findings are in agreement with those obtained by Nooshkam and Madadlou (2016), who assumed that the heating
process induced the pH increase, at the early stages of the reaction, by the denaturation of whey
protein hydrolysates.

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3.3. Color measurement of the MRPs

The color changes of the MRPs prepared without or with US pre-treatment and heated at 90 300 °C for 2 h are shown in Table 1. No significant differences were detected among all the color 301 parameters of the protein fractions heated alone (p>0.05) (Data not shown). Unlike control samples, 302 the protein-sucrose combination, preceded by ultra-sonication, induced marked changes of the 303 conjugated fractions color. In fact, whiteness values (W*) of the different samples decreased 304 significantly with the thermal processing time, and this decrease was more pronounced in the US 305 306 pre-treated samples rather than conventionally heated ones. Similarly, the color difference (ΔE^*) values markedly increased over time (p<0.05) to reach their maximal levels in the US treated 307 systems after 120 min of thermal treatment. Furthermore, increasing chroma (C*) values were 308 309 observed with the increase of time, indicating the formation of brown compounds. Similar findings were obtained by Małgorzata, Konrad and Zieliński (2016) after buckwheat groats roasting for 50 310 min at 160 °C. The observed chromacity and whiteness changes of the MRPs demonstrated the 311 312 occurring of the MR stages, characterized by the formation of brown colored pigments, which was relatively accelerated by US pretreatment. 313

According to the obtained results, the US treatment at relatively high temperature (40 °C) for 30 min was suitable to improve the MR rate in LMW peptides-sucrose model system. Corzo-Martínez, Moreno, Megías-Pérez, Olano, Moreno & Villamiel (2014) also observed an increase in the intermediate and advances stages of MR with US treatment at 40 °C with 50% of amplitude. In addition, Yu, Seow, Ong and Zhou (2017) showed that the ultrasonic MR process conducted at low temperatures (below 60 °C) leads to generate higher levels of intermediate MRPs and melanoidins,
than those produced at 65-75 °C.

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3.4. UV-absorbance and browning intensity measurement of the MRPs

The MR rate was monitored based on the evaluation of the color intensity of the intermediate 322 and final products, measured at 294 and 420 nm, respectively. In fact, carbonyl group-amino acid 323 conjugation is associated to the development of intermediate colorless compounds (Amadori), 324 which didn't absorb in the visible spectrum, prior to generation of brown pigments (melanoidins), 325 the final products of the MR. Fig. 2 (A, B) shows A₂₉₄ and A₄₂₀ evolution as a function of the 326 327 reaction time. As observed, the intermediate products and browning pigments formation began to increase in parallel after 60 min of dry-heating; particularly in LMW-NH/S and LMW-NH/S-Us, 328 329 which were the most suitable systems for the production of melanoidins ($A_{420 \text{ nm}}$). In addition, 330 products resulted from the US-assisted MR contained higher amount of final brown products within 331 the first hour of dry-heating compared to conventional treated ones. However, the content of these 332 advanced products still lower than that obtained in LMW-NH/S and LMW-NH/S-Us. Unlike A_{420nm} values, the US pre-treatment did not affect the formation of colorless intermediate 333 compounds. 334

It could be concluded that, in the present systems, the browning development occurs after an induction period, characterized by the production of fluorescent uncolored intermediates, and then the brown color appeared. Interestingly, US pre-treatment gave final products with darker color. Similar evolution results were reported by Kim and Lee (2009) and Nooshkam and Madadlou (2016).

340 **3.5. Free amino acid contents**

341 Free amino acid (FAA) contents of the initial peptide fractions and their MRPs, evaluated 342 each 60 min of thermal treatment, are summarized in Table 2. Data showed that there is a significant difference in the amounts of FAA between LMW peptides and the resulting MRPs. In addition, 343 FAA contents of all MRPs decreased with the incubation time progress, confirming the occurrence 344 345 of the MR during heating. The increase in the FAA amount in the heated peptides, without sucrose 346 addition, is mainly due to the thermal protein hydrolysis occurred during dry-heating for 2 h. 347 However, a significant decrease in total FAA, following the addition of sucrose, was observed, which resulted from the strong cross-linking between sucrose and free amino residues via the MR. 348 349 Similar results were reported by Eric et al. (2013) and Lan et al. (2010) who studied the MRPs 350 derived from xylose/cysteine-sunflower protein hydrolysate and xylose-soybean peptide model systems, respectively. In addition, Zhao, Zeng, Cui and Wang (2007) (2007) reported that in MR, 351 peptides below 1 kDa showed higher reactivity than peptides with higher MW and that the direct 352 peptide-sugar cross-linking via MR decreased the free amino acids content. 353

Moreover, results showed that sonication markedly promoted the MR between LMW 354 peptides and sucrose compared to the non-sonicated samples. In fact, the final total FAA level of 355 356 LMW-NH/S-Us, LMW-EH/S-Us and LMW-PH/S-Us, after 120 min of heating, were estimated at about 0.95, 1.55 and 0.97 mg/g, respectively, vs.1.08, 2.36 and 0.93 mg/g for LMW-NH/S, LMW-357 EH/S and LMW-PH/S, respectively. Similar data were found by Stanic-Vucinic, Prodic, 358 Apostolovic, Nikolic and Velickovic (2013) showing that high-intensity ultrasound in aqueous 359 solution promoted the glucose and fructose isomerization after the thermal hydrolysis of sucrose, 360 361 leading to the MR rate acceleration. Particularly, a fast rate of Lys free residue loss was observed, proving the successful protein-sugar linking during MR. The high reactivity of Lys was also 362 observed in the Lys/S reference system, where the total content of available Lys has been decreased 363 by 57.15, 69.26, and 84.69% after 30, 60, and 90 min of heat processing, respectively. In the same 364

365 context, Golon, Kropf, Vockenroth and Kuhnert (2014) have shown that, in the presence of sucrose
366 and after heating, lysine and tyrosine yielded in simple spectra of MRPs, followed by arginine and
367 aspartic acid, whereas cysteine and serine produced the highest number of compounds.

368 It has been reported that US induced significant loss of protein amino groups due to free 369 radical reactions generated by water sonolysis, and lysine residue is prone to be oxidized to lysine 370 aldehyde, resulting in loss of its amino group (Meltretter & Pischetsrieder, 2008). These aldehydes 371 are highly reactive and they can undergo spontaneous MR in the presence of sugars.

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3.6. Degree of glycation

The degrees of glycation in the final MRPs were determined and the results are shown in Fig. 373 374 2c. The ultrasound pre-treatment was found to enhance significantly only the LMW-EH/S system. 375 However, LMW-NH/S and LMW-PH/S systems, sonicated or not, had similar glycation degrees 376 (about 80%) (p>0.05), which were higher than those recorded in the LMW-EH/S-Us sample. Greater glycation degrees reflected the high reactivity of LMW peptides and their derived free 377 378 amino groups with sucrose hydrolysates. These data are in accordance with those reported by Nooshkam and Madadlou (2016) showing that the final MRPs, obtained from milk ultrafiltration 379 permeate and whey protein isolate heated at 90 °C for 45 min, exhibited a glycation degree of 46.73 380 381 $\pm 6.29\%$.

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3.7. Taste score of MRPs

Flavor and aroma development throughout the MR depends, particularly, on the type of sugars and amino acids involved in the reaction. In the present study, sweet, bitter and umami scores of the different MRPs were determined as a function of the reaction time. As shown in Fig. 3, the original peptides contained high amounts of sweet amino acids, which were gradually decreased from the first 30 min of heating, particularly in the US-pretreated samples. According to Table 2, the observed sweetness decrease is mainly due to the consumption of free Gly and Lys residues during MR, known as the most reactive amino acids. In fact, Guan et al. (2011) and Vhangani and Wyk (2013) have described the high reactivity of glycine and lysine in glucoseglycine, fructose–lysine and ribose–lysine model systems.

Similarly, hydrophobic FAA residues, responsible to the bitter taste, were significantly reduced by 41.18%, 6.74% and 19.31%, for LMW-NH/S, LMW-EH/S and LMW-PH/S, respectively, compared to the initial hydrolysates, after 60 min of heating. However, this reduction was more significant for US-derived MRPs and reached 54.55%, 19.69% and 51.45% in LMW-NH/S-Us, LMW-EH/S-Us and LMW-PH/S-Us, respectively, after the same period of heating. The reduction in bitter attribute reflects the decrease of bitter taste in the final products, which is an important criterion in food formulations.

For the umami amino acids, a similar behavior was observed, and a minimum content of umami residues was kept constant until the end of the reaction, which promises the sensorial quality of the final MRPs. In the same context, it has been reported that the Maillard peptides (MW = 1-5KDa) are good enhancers of flavor, including umami, continuity and mouthfulness, and then they could be considered as key taste enhancers in food formulations (Ogasawara, Katsumata & Egi, 2006). Hence, the obtained LMW peptides-sucrose derived MRPs could serve as potential flavor promoting agents in thermally processed foods.

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408 **3.8. Fourier transform infra-red analysis**

Fourier transform infra-red (FT-IR) spectroscopy is a useful technique to study protein– carbohydrate systems. In order to investigate the interaction between sugars and peptides before and after dry-heating, FT-IR analysis was performed (Fig. 3d).

As shown in FT-IR spectra, there are several typical regions in both, protein fractions and 412 413 their sucrose conjugates corresponding, respectively, to the N-H stretching at ~ 3301-3237 cm⁻¹ for the amide A and C–H stretching at ~ 2914-2939 cm⁻¹ for the amide B (Chang & Tanaka, 2002). 414 415 No significant changes were observed in the FT-IR spectra of the different protein hydrolysate fractions, before and after heat treatment, particularly in the amide A and B regions. Nevertheless, 416 the thermal treatment for 2 h affected the FT-IR spectra of the resulting conjugates, as compared 417 418 to the initial spectra of LMW peptide fractions. Particularly, US-pretreatment was found to affect significantly the amide A band wavenumbers, which decreased from 3274 to 2998 cm⁻¹ for LMW-419 NH/S and LMW-NH/S-Us, respectively. Furthermore, US-treated MRPs showed the lowest 420 421 intensities at these regions among all the other samples, suggesting that the free -OH groups of sucrose and amino acids $(3380-3340 \text{ cm}^{-1})$ were consumed during the heating process. 422

Moreover, after MR induction, important intensity reductions were noted in the C=O 423 424 stretching (1600-1700 cm⁻¹) and N-H deformation (1500-1550 cm⁻¹) of the amide I and II, respectively, especially in the NH and PH-derived MRPs. In addition, after 2 h of heating, marked 425 deformations of the bands attributable to C–N stretching and N–H bending of amide III vibrations 426 (1220-1400 cm⁻¹) were observed, particularly in the US-pretreated samples. The present changes 427 in the amide III band may be attributed to the complicated cross-linking occurred between peptides 428 429 and sucrose (Rhim, Mohanty, Singh & Ng, 2006). In this context, Stanic-Vucinic et al. (2013) reported that strong sheer forces generated during sonication led to efficient mixing of sugar-430 protein solution and efficient heat/mass transfer, contributing to increase the rate of intermediate 431 and advances stages of the MR. In addition, Corzo-Martinez, Moreno, Megías-Pérez, Olano, 432

433 Moreno and Villamiel (2014) found greater protein-sugar interactions in the ultrasonicated systems434 as compared to conventional treated samples under the same conditions.

Results of the FT-IR analysis reflected the interactions bonds between proteins and sugar
formed during dry-heating, which were typically caused by the MR and proved the effect of the
US-pretreatment on accelerating the reaction rate.

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3.9. Evaluation of the antioxidant activity of MRPs

The antioxidant activities of peptides/sucrose model systems, before and after thermal 439 treatment, were assayed in terms of their reducing power and β -carotene bleaching protection (Fig. 440 4). The three protein hydrolysates showed varying degrees of antioxidant activities, which is 441 mainly due to their initial amino acids composition and their sequences. Data showed that LMW-442 443 EH exhibited the highest reducing power, while LMW-PH displayed the highest β -carotene protection effect. Also, after 120 min of heating, the antioxidant activities of all peptides remained 444 constant, except for LMW-EH/S where the reducing power decreased by 20% after 120 min. On 445 446 the other hand, the addition of sucrose and US pretreatment, without heating process, did not affect significantly the antioxidant capacity of unheated peptides (p>0.05). In contrast, the protein-447 sucrose conjugation (2 h at 90 °C) enhanced the antioxidant potential of the resulting MRPs that 448 showed stronger reducing power and β -carotene bleaching protection effect, compared to the initial 449 peptides (p < 0.05). Interestingly, US treatment was found to enhance considerably the antioxidant 450 potential of MRPs. Among the three US-treated MRPs, the highest reducing power enhancement 451 was obtained with LMW-NH/S-Us, which increased by 200% compared to LMW-NH/S, while the 452 greater β-carotene bleaching protection improvement was found with LMW-EH/S-Us (+99% in 453 454 comparison with LMW-EH/S). Interestingly, US pre-treated MRPs exhibited higher antioxidant potential than those derived from conventional heating (without US). 455

456 This study revealed that the physicochemical and structural changes greatly correlated with the increase in the antioxidant activity. In addition, our results are in agreement with those obtained 457 by Huang et al. (2012) and Liu, Kong, Han, Sun and Li (2014a) who found that the glycation of 458 LMW peptides enhanced their reducing power and antiradical activity, compared to the peptides 459 460 alone. The increase of antioxidant properties might be attributed to generation of new products, resulting from peptides/sucrose cross-linking. Furthermore, Liu, Li, Kong, Jia and Li (2014b) 461 462 proposed that the MR could be a useful method to improve the free radical scavenging, iron chelating activity and reducing power of protein hydrolysates, and they explained this finding by 463 464 the caramelization reaction induced between an NH₂-protein and a sugar-reducing compound.

Moreover, Daglia, Papetti, Aceti, Sordelli, Gregotti and Gazzani (2008) suggested that advanced MRPs are considered as a particular complex mix that contained numerous compounds, which contributed mostly to the antioxidant activity of the product mixture. Hence, the present results suggested that the conjugates obtained from LMW peptides-sucrose system could be a potential food antioxidant ingredient in functional food products.

470 **4.** Conclusion

In the present study, LMW peptides-sucrose conjugates were elaborated through the MR 471 process, with or without ultrasonic pre-treatment. Neutrase-derived peptides were the most reactive 472 agents with sucrose, resulting in higher COOH-NH₂ interaction. Physical (pH, color and browning 473 intensity) and structural changes showed that, in the different studied systems, US pre-treatment 474 accelerated the rate of the MR evolution and give the higher glycation degrees as compared to 475 476 conventional heating process. In addition, MRPs showed better taste attributes (low amount of 477 bitter amino acids) and increased antioxidant capacities, evaluated based on their reducing power and β -carotene protection activities, compared to the un-conjugated proteins. As a result, the 478

479	proposed food model systems, formed based on LMW peptides (< 1 kDa) from smooth hound by-
480	products and sucrose, could serve as aromatic and bioactive ingredients for preparing foods.

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486 **References**

Abdelhedi, O., Jridi, M., Jemil, I., Mora, L., Toldrá, F., Aristoy M.-C., Boualga, A., Nasri,
M., & Nasri, R. (2016). Combined biocatalytic conversion of smooth hound viscera: Protein
hydrolysates elaboration and assessment of their antioxidant, anti-ACE and antibacterial activities. *Food Research International*, 86, 9-23.

Adler-Nissen, J. (1986). A review of food hydrolysis specific areas, in: J. Adler-Nissen
(Eds.), Enzymic hydrolysis of food proteins Elsevier Applied Science Publishers, Copenhagen,
Danemark, pp. 157-109.

Aristoy, M.-C., & Toldrá, F. (1991). Deproteinization techniques for HPLC amino acids
analysis in fresh pork muscle and dry cured ham. *Journal of Agricultural and Food Chemistry*, *39*,
1792-1795.

Chandrapala, J., Zisu, B., Palmer, M., Kentish, S., & Ashokkumar, M. (2011). Effects of
ultrasound on the thermal and structural characteristics of proteins in reconstituted whey protein
concentrate. *Ultrasonics Sonochemistry*, *18*, 951-957.

500	Chang, M. C., & Tanaka, J. (2002). FT-IR study for hydroxyapatite/collagen nanocomposite
501	cross-linked by glutaraldehyde. Biomaterials, 23, 4811-4818.

Chen, X., & Kitts, D. D. (2015). Evidence for inhibition of nitric oxide and inducible nitric
oxide synthase in Caco-2 and RAW 264.7 cells by a Maillard reaction product [5-(5, 6-dihydro4H-pyridin-3-ylidenemethyl) furan-2-yl]-methanol. *Molecular and Cellular Biochemistry*, 406,
205-215.

Christelle, K., Jean-Pascal, B., Regis, B., Laurent, L., Carole, P., & Mireille C. (2014). Odor
modification in Salmon hydrolysates using the Maillard reaction. *Journal of Aquatic Food Product Technology*, *23*, 453-467.

Corzo-Martínez, M., Moreno, A., Megías-Pérez, R., Olano, A., Moreno, F. J., & Villamiel,
M. (2014). Impact of high-intensity ultrasound on the formation of lactulose and Maillard reaction
glycoconjugates. *Food Chemistry*, *157*, 186-192.

Daglia, M., Papetti, A., Aceti, C., Sordelli, B., Gregotti, C., & Gazzani, G. (2008). Isolation
of high molecular weight components and contribution to the protective activity of coffee against
lipid peroxidation in a rat liver microsome system. *Journal of Agricultural and Food Chemistry*,
56, 11653-11660.

Eric, K., Raymond, L. V., Huang M., Cheserek M. J., Hayat K., Savio N. D., Amédée M., &
Zhang X. (2013). Sensory attributes and antioxidant capacity of Maillard reaction products derived
from xylose, cysteine and sunflower protein hydrolysate model system. *Food Research International*, *54*, 1437-1447.

520 FAOSTAT (2015). FishStat plus - Universal software for fishery statistical time series.

521	Golon, A., Kropf, C., Vockenroth, I., & Kuhnert, N. (2014). An investigation of the
522	complexity of Maillard reaction product profiles from the thermal reaction of amino acids with
523	sucrose using high resolution mass spectrometry. Foods, 3, 461-475.
524	Guan, YG., Zhang, BS., Yu, SJ., Wang, XR., Xu, XB, Wang, J., Han, Z., Zhang, P
525	J., & Lin, H. (2011). Effects of ultrasound on a Glycin-Glucose model system -A means of
526	promoting Maillard reaction. Food and Bioprocess Technology, 4, 1391-1398.
527	Hathwar, S. C., Rai, A. K., Modi, V. K., & Narayan, B. (2012). Characteristics and consumer
528	acceptance of healthier meat and meat product formulations - a review. Journal of Food Science
529	and Technology, 49, 653-664.
530	Horvat, S., & Jakas, A. (2004). Peptide and Amino Acid Glycation: New Insights into the
531	Maillard Reaction. Journal of Peptide Science, 10, 119-137.
532	Huang, X., Tu, Z., Xiao, H., Wang, H., Zhang, L., Hu, Y., Zhang, Q., & Niu, P. (2012).
533	Characteristics and antioxidant activities of ovalbumin glycated with different saccharides under
534	heat moisture treatment. Food Research International, 48, 866-872.
535	Jiang, Z., Rai, D. K., O'Connor, P. M., & Brodkorb, A. (2013). Heat-induced Maillard
536	reaction of the tripeptide IPP and ribose: Structural characterization and implication on bioactivity.
537	Food Research International, 50, 266-274.
538	Kabelová, I., Dvořáková, M., Čížková, H., Dostálek, P., & Melzoc, K. (2009). Determination
539	of free amino acids in cheeses from the Czech market. Czech Journal of Food Sciences, 27, 143-
540	150.

Kim, J. S., & Lee, Y. S. (2009). Antioxidant activity of Maillard reaction products derived
from aqueous glucose/glycine, diglycine, and triglycine model systems as a function of heating
time. *Food Chemistry*, *116*, 227-232.

Kitts, D.D., Chen, X., & Jing, H. (2012). Demonstration of antioxidant and anti-inflammatory
bioactivities from sugar–amino acid Maillard reaction products. *Journal of Agricultural and Food Chemistry*, 60, 6718–6727.

Koleva, I. I., van Beek, T. A., Linssen, J. P. H., de Groot, A., & Evstatieva, L. N. (2002).
Screening of plant extracts for antioxidant activity: a comparative study on three testing methods. *Phytochemical Analysis, 13*, 8-17.

Lan, X., Liu, P., Xia, S., Jia, C., Mukunzi, D., Zhang, X., Xia, W., Tian, H., & Xiao, Z.
(2010). Temperature effect on the non-volatile compounds of Maillard reaction products derived
from xylose–soybean peptide system: further insights into thermal degradation and cross-linking. *Food Chemistry*, *120*, 967-972.

Liu, Q., Kong, B., Han, J., Sun, C., & Li, P. (2014a). Structure and antioxidant activity of whey protein isolate conjugated with glucose via the Maillard reaction under dry-heating conditions. *Food Structure*, *1*, 145-154.

Liu, Q., Li, J., Kong, B., Jia, N., & Li, P. (2014b). Antioxidant capacity of Maillard reaction products formed by a porcine plasma protein hydrolysate-sugar model system as related to chemical characteristics. *Food Science and Biotechnology*, *23*, 33-41.

Małgorzata, W., Konrad, P. M., & Zieliński, H. (2016). Effect of roasting time of buckwheat
groats on the formation of Maillard reaction products and antioxidant capacity. *Food Chemistry*, *196*, 355-358.

Marko, D., Habermeyer, M., Kemény, M., Weyand, U., Niederberger, E., Frank, O., &
Hofmann, T. (2003). Maillard reaction products modulating the growth of human tumor cells *in vitro. Chemical Research in Toxicology*, *16*, 48-55.

Matmaroh, K., Benjakul, S., & Tanaka, M. (2006). Effect of reactant concentrations on Maillard reaction in a fructose-glycine model system and the inhibition of black tiger shrimp polyphenoloxidase. *Food Chemistry*, *98*, 1-8.

Meltretter, J., & Pischetsrieder, M. (2008). Application of mass spectrometry for the detection of glycation and oxidation products in milk proteins. *Annals of the New York Academy of Sciences, 1126*, 134-140.

Nasri, R., Younes, I., Jridi, M., Trigui, M., Bougatef, A., Nedjar-Arroume, N., Dhulster, P.,
Nasri, M., & Karra-Chaabouni, M. (2013). ACE inhibitory and antioxidative activities of Goby
(*Zosterissessor ophiocephalus*) fish protein hydrolysates: Effect on meat lipid oxidation. *Food Research International*, *54*, 552-561.

- Nooshkam, M., & Madadlou, A. (2016). Maillard conjugation of lactulose with potentially
 bioactive peptides. *Food Chemistry*, *192*, 831-836.
- 578 Ogasawara, M., Katsumata, T., & Egi, M. (2006). Taste properties of Maillard-reaction
 579 products prepared from 1000 to 5000 Da peptide. *Food Chemistry*, *99*, 600-604.
- Péron, G., Mittaine, J. F., Le Gallic, B. (2010). Where do fishmeal and fish oil products come
 from? An analysis of the conversion ratios in the global fishmeal industry. *Marine Policy*, *34*, 815820.

583	Rhim, J. W., Mohanty, K. A., Singh, S. P., Ng, P. K. W. (2006). Preparation and properties
584	of biodegradable multilayer films based on soy protein isolate and poly (lactide). Industrial &
585	Engineering Chemistry Research, 45, 3059-3066.

Rufia´n-Henares, J., & Morales, F. (2007). Functional properties of melanoidins: *In vitro*antioxidant antimicrobial and antihypertensive activities. *Food Research International*, 40, 9951002.

Seo, S., Karboune, S., L'Hocine, L., & Yaylayan, V. (2013). Characterization of glycated
lysozyme with galactose, galactooligosaccharides and galactan: Effect of glycation on structural
and functional properties of conjugates. *LWT – Food Science and Technology*, *52*, 44-53.

Stanic-Vucinic, D., Prodic, I., Apostolovic, D., Nikolic, M., & Velickovic, T. C. (2013).
Structure and antioxidant activity of β-lactoglobulin-glycoconjugates obtained by high-intensityultrasound-induced Maillard reaction in aqueous model systems under neutral conditions. *Food Chemistry*, *138*, 590-599.

Stanic-Vucinic, D., Stojadinovic, M., Atanaskovic-Markovic, M., Ognjenovic, J., Grönlund,
H., van Hage, M., Lantto, R., Sancho, A. I., & Velickovic, T. C. (2012). Structural changes and
allergenic properties of β-lactoglobulin upon exposure to high-intensity ultrasound. *Molecular Nutrition & Food Research, 56*, 1894-1905.

Sun, Y., Hayakawa, S., & Izumori, K. (2004). Modification of ovalbumin with a rare
ketohexose through the Maillard reaction: effect on protein structure and gel properties. *Journal of Agricultural and Food Chemistry*, *52*, 1293-1299.

603 Vhangani, L. N., & Wyk, J. V. (2013). Antioxidant activity of Maillard reaction products
604 (MRPs) derived from fructose–lysine and ribose–lysine model systems. *Food Chemistry*, *137*, 92605 98.

Wang, Y., Pan, Y., Zhang, Z., Sun, R., Fang, X., & Yu, D. (2012). Combination use of
ultrasound irradiation and ionic liquid in enzymatic isomerization of glucose to fructose. *Process Biochemistry*, 47, 976-982.

609 Yang, S.-Y., Lee, S., Pyo, M. C., Jeon, H., Kim, Y., & Lee K.-W. (2016). Physicochemical 610 and biological properties of fish protein hydrolysate-ribose conjugate by Maillard reaction. Journal 611 of International Society of **Antioxidants** in Nutrition & Health, DOI: 612 10.18143/JISANH_v2i1_1038.

Yildirim, A., Mavi, A., & Kara, A. A. (2001). Determination of antioxidant and antimicrobial
activities of *Rumex crispus* L. extracts. *Journal of Agricultural and Food Chemistry*, *49*, 40834089.

616 Yoshida, M., & Saito, S. (1969). Multidimensional scaling of the taste of amino acids.
617 *Japanese Psychological Research*, 11, 149-166.

Yu, H., Seow, Y.-X., Ong, P. K. C., & Zhou, W. (2017). Effects of high-intensity ultrasound
on Maillard reaction in a model system of D-xylose and L-lysine. *Ultrasonics Sonochemistry*, *34*,
154-163.

Zhao, M.M., Zeng, X.F., Cui, C., & Wang, J.S. (2007). Studies on the peptide degradation in
the Maillard reaction composed of chicken protein peptide and glucose. *Journal of Sichuan University* (Engineering Science Edition), *39*, 77-81.

624	Zhou, D.Y., Zhu, B.W., Qiao, L., Wu, H.T., Li, T.M., Yang, J.F., & Murata, Y. (2012). In
625	vitro antioxidant activity of enzymatic hydrolysates prepared from abalone (Haliotis discus hannai
626	Ino) viscera. Food and Bioproducts Processing, 90, 148-154.
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C A A	Table 1 all and color evolution of motoin budgebactor for them and their MDD, 1 is 1 with
644	Example 1. per and color evolution of protein hydrolysates fractions and their MIKP's during heating
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		Without	ultrasonic tre	With ultrasonic treatment								
Time (min) 0		30	60	90	120	0	30	60	90	120		
pH	5.42±0.11°	5.32±0.01°	5.53±0.02°	6.93±0.01 ^a	6.53 ± 0.02^{b}	5.38 ± 0.02^{b}	6.17±0.05 ^a	6.20±0.01ª	5.56±0.03 ^b	5.49 ± 0.09^{b}		
W*	92.55±0.01 ^a	90.90 ± 0.25^{b}	90.41 ± 1.2^{b}	$89.94{\pm}1.30^{bc}$	87.42±1.36 ^c	$91.26{\pm}0.00^a$	$80.15{\pm}0.43^{\text{b}}$	74.98±0.31°	$71.01{\pm}0.71^d$	$71.61{\pm}0.80^d$		
C*	$1.22\pm0.02^{\circ}$	1.48 ± 0.13^{b}	1.61 ± 0.14^{a}	1.48 ± 0.21^{ab}	1.77 ± 0.1^{a}	1.20±0.01°	2.13 ± 0.2^{a}	1.77 ± 0.1^{b}	1.78 ± 0.01^{b}	2.22±0.1ª		
ΔE^*	-	1.68 ± 0.06^{d}	$2.40\pm0.15^{\circ}$	$2.80{\pm}0.17^{b}$	5.17 ± 0.38^{a}	-	11.77 ± 0.29^{a}	16.48 ± 0.21^{b}	$20.33{\pm}0.50^{a}$	20.43 ± 0.43^{a}		
pH	6.30±0.1 ^{bc}	6.20±0.01°	6.37 ± 0.01^{b}	6.40±0.01 ^a	$5.62{\pm}0.01^d$	6.30±0.01 ^b	6.73 ± 0.03^{a}	6.81 ± 0.04^{a}	6.16±0.05 ^b	5.20±0.01°		
W*	89.22±1.03 ^a	84.02±2.31°	$87.15{\pm}0.63^a$	$87.35{\pm}1.09^{a}$	$87.87{\pm}0.68^{\text{b}}$	$90.22{\pm}1.00^{a}$	$79.83{\pm}1.32^{b}$	75.42±0.39°	$74.45 \pm 0.19^{\circ}$	71.74 ± 0.1^{d}		
C*	1.00 ± 0.12^{b}	1.26±0.10 ^{ab}	1.15±0.11 ^{ab}	1.28±0.13 ^{ab}	1.35±0.04 ^a	1.02 ± 0.21^{d}	1.79 ± 0.03^{b}	1.64±0.02°	1.63±0.01°	2.19 ± 0.07^{a}		
ΔE^*	-	1.93 ± 0.72^{b}	2.30 ± 0.29^{b}	$3.03{\pm}1.57^{b}$	5.47 ± 0.22^{a}	-	11.02 ± 0.05^{d}	15.22±0.04°	16.20 ± 0.01^{b}	19.96 ± 0.06^{a}		
pH	6.03±0.01 ^e	6.33 ± 0.02^d	6.38±0.01 ^c	7.08 ± 0.03^{a}	6.53±0.02 ^b	6.00 ± 0.00^{cd}	6.71±0.14 ^b	7.02 ± 0.10^{a}	6.62 ± 0.1^{b}	6.25 ± 0.3^{b}		
W*	85.95±2.69 ^a	83.81 ± 0.99^{a}	84.17±1.31 ^a	83.20±0.01 ^b	79.94±0.76°	$86.00{\pm}2.68^{a}$	$82.56{\pm}1.67^{a}$	84.22±2.61 ^a	76.58 ± 1.30^{b}	77.26±0.7 ^b		
C*	$1.48\pm0.04^{\circ}$	1.53±0.02 ^b	1.53 ± 0.05^{b}	1.62±0.13 ^{ab}	1.65 ± 0.01^{a}	1.36 ± 0.05^{b}	2.05±0.1ª	1.53±0.1 ^b	1.64±0.25 ^{ab}	1.77 ± 0.4^{ab}		
ΔE^*	-	1.91 ± 0.43^{d}	3.84±0.92°	5.57 ± 0.03^{b}	6.32 ± 0.32^{a}	-	3.80±0.05°	3.87±0.03°	9.44 ± 0.02^{b}	10.88±0.01ª		

647 LMW-NH/S, LMW-EH/S and LMW-PH/S indicate the MRPs prepared from LMW peptides derived from NH, EH

648 and PH and conjugated to sucrose (S), respectively.

649 Different letters in the same sample within different incubation time indicate significant differences at p < 0.05.

650	$W^* = 100 - \left[\left(100 - L^* \right)^2 + a^{*2} + b^{*2} \right]^{1/2}$
651	$C^* = (a^{*2} + b^{*2})^{1/2}$
652	$\Delta E^* = \left[\left(L^* - L_0 \right)^2 + \left(a^* - a_0 \right)^2 + \left(b^* - b_0 \right)^2 \right]^{1/2}$
653	where L_0 , a_0 and b_0 are the color of the sample at $t = 0$.
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Free amino acids (mg/g)		Asp	Glu	Нур	Ser	Gly	Tau	His	Thr	Ala	Arg	Pro	Tyr	Val	Met	Ile	Leu	Phe	Trp	Lys	Total
	0	0.2	0.24	0.02	0.16	0.83	0.1	0.07	0.22	0.87	0.05	0.32	0.07	0.42	0.11	0.19	0.34	0.16	0.03	0.36	4.76
	120	0.2	0.25	0.02	0.16	0.85	0.11	0.06	0.22	1.07	0.06	0.4	0.11	0.61	0.12	0.22	0.32	0.15	0.02	0.31	5.26
	60	0.07	0.13	0.01	0.06	0.35	0.03	0.03	0.1	0.35	0.02	0.17	0.06	0.36	0.06	0.13	0.22	0.11	0.02	0.07	2.35
	120	0.03	0.07	0.01	0.03	0.17	0	0.01	0.06	0.17	0.01	0.09	0.03	0.18	0.02	0.05	0.09	0.04	0	0	1.08
	60	0.03	0.09	0.01	0.05	0.29	0.02	0.02	0.08	0.31	0.01	0.16	0.05	0.34	0.06	0.12	0.22	0.11	0.02	0.04	2.04
	120	0.02	0.04	0.01	0.03	0.12	0	0.01	0.03	0.13	0	0.08	0.02	0.3	0.02	0.04	0.07	0.04	0	0	0.95
	0	0.1	0.27	0.02	0.06	0.41	0.07	0.03	0.14	0.49	0.04	0.35	0.08	0.28	0.1	0.25	0.45	0.12	0.04	0.4	3.69
	120	0.1	0.29	0.02	0.1	0.45	0.09	0.03	0.13	0.54	0.03	0.3	0.09	0.41	0.11	0.2	0.52	0.12	0.05	0.4	3.98
	60	0.16	0.2	0.03	0.12	0.06	0.11	0.08	0.1	0.49	0.09	0.2	0.16	0.23	0.16	0.22	0.4	0.18	0.04	0.44	3.47
	120	0.08	0.15	0.02	0.06	0.04	0.05	0.03	0.1	0.49	0.03	0.1	0.1	0.12	0.09	0.15	0.42	0.13	0.04	0.15	2.36
	60	0.16	0.12	0.02	0.1	0.03	0.1	0.05	0.19	0.49	0.06	0.25	0.11	0.15	0.12	0.15	0.44	0.11	0.03	0.36	3.05
	120	0.04	0.1	0.01	0.03	0.01	0.02	0.02	0.07	0.32	0.03	0.07	0.07	0.02	0.07	0.12	0.34	0.11	0.03	0.07	1.55
	0	0.11	0.17	0.02	0.12	0.48	0.06	0.03	0.1	0.61	0.1	0.19	0.09	0.34	0.08	0.14	0.35	0.13	0.03	0.17	3.32
	120	0.11	0.18	0.02	0.13	0.51	0.09	0.03	0.11	0.68	0.1	0.22	0.11	0.31	0.11	0.15	0.4	0.16	0.04	0.19	3.65
	60	0.2	0.17	0.03	0.1	0.3	0.12	0.08	0.26	0.5	0.08	0.11	0.09	0.25	0.12	0.21	0.21	0.1	0.09	0.12	3.15
	120	0.01	0.02	0.01	0.03	0.12	0	0.01	0.05	0.14	0.01	0.08	0.04	0.06	0.03	0.06	0.16	0.07	0.01	0.02	0.93
	60	0.05	0.07	0.01	0.05	0.23	0	0.01	0.07	0.21	0.02	0.11	0.04	0.1	0.04	0.08	0.2	0.08	0.01	0.07	1.46
	120	0.01	0.03	0.01	0.03	0.12	0	0.01	0.07	0.14	0.01	0.07	0.04	0.07	0.03	0.07	0.16	0.07	0.01	0.02	0.97

666 Figure captions:

Fig. 1. Experimental set-up of ultrafiltration process before Maillard reaction induction, where P:pressure and T: temperature points of measurement.

Fig. 2. Evolution of the absorbance at 294 nm (A) and 420 nm (B) as indicators of intermediate and advanced stages of MR, respectively, after different intervals of incubation time.

671 Glycation degree value of the final MRPs (C); different letters in the same system with different

treatments (without or with US) indicate significant differences (p < 0.05).

Fig. 3. Evolution of the taste scores of MRPs prepared without or with ultrasonic treatment, after
different intervals of incubation time: sweet (a), bitter (b) and umami (c) amino acid contents.
Effect of ultrasonic treatment on the infra-red spectra of the LMW peptides heated alone or in
conjugation to sucrose (d).

Fig. 4. Comparison of the antioxidant activities of LMW peptides and their sucrose conjugates, without or with US pre-treatment, before and after heat treatment at 90 °C for 2 h; Reducing power assay (a, b, c) and inhibition of β-carotene bleaching activity (d, e, f). Different letters in the same sample within different incubation times indicate significant differences (p < 0.05).

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