

Strathprints Institutional Repository

Davidson, Christine M. and Urquhart, Graham J. and Ajmone-Marsan, Franco and Biasioli, Mattia and da Costa Duarte, Armando and Díaz-Barrientos, Encarnación and Grčman, Helena and Hossack, Iain and Hursthouse, Andrew S. and Madrid, Luis and Rodrigues, Sonia and Zupan, Marko (2006) *Fractionation of potentially toxic elements in urban soils from five European cities by means of a harmonised sequential extraction procedure.* Analytica Chimica Acta, 565 (1). pp. 63-72. ISSN 0003-2670

Strathprints is designed to allow users to access the research output of the University of Strathclyde. Copyright © and Moral Rights for the papers on this site are retained by the individual authors and/or other copyright owners. You may not engage in further distribution of the material for any profitmaking activities or any commercial gain. You may freely distribute both the url (http:// strathprints.strath.ac.uk/) and the content of this paper for research or study, educational, or not-for-profit purposes without prior permission or charge.

Any correspondence concerning this service should be sent to Strathprints administrator: mailto:strathprints@strath.ac.uk



Davidson, C.M. and Urquhart, G.J. and Ajmone-Marsan, F. and Biasioli, Mattia and da Costa Duarte, Armando and Díaz-Barrientos, Encarnación and Grčman, Helena and Hossack, Iain and Hursthouse, Andrew S. and Madrid, Luis and Rodrigues, Sonia and Zupan, Marko (2006) *Fractionation of potentially toxic elements in urban soils from five European cities by means of a harmonised sequential extraction procedure.* Analytica Chimica Acta. ISSN 0003-2670

http://strathprints.strath.ac.uk/528/

This is an author-produced version of a paper published in Analytica Chimica Acta (ISSN: 0003-2670). This version has been peer-reviewed, but does not include the final publisher proof corrections, published layout, or pagination.

Strathprints is designed to allow users to access the research output of the University of Strathclyde. Copyright © and Moral Rights for the papers on this site are retained by the individual authors and/or other copyright owners. Users may download and/or print one copy of any article(s) in Strathprints to facilitate their private study or for non-commercial research. You may not engage in further distribution of the material or use it for any profit-making activities or any commercial gain. You may freely distribute the url (<u>http://eprints.cdlr.strath.ac.uk</u>) of the Strathprints website.

Any correspondence concerning this service should be sent to The Strathprints Administrator: eprints@cis.strath.ac.uk

Fractionation of potentially toxic elements in urban soils from five European cities by means of a harmonised sequential extraction procedure

Christine. M. Davidson^{a*}, Graham J. Urquhart^a, Franco Ajmone-Marsan^f, Mattia Biasioli^f, Armando da Costa Duarte^b, Encarnación Díaz-Barrientos^e, Helena Grčman^c, Ian Hossack^d, Andrew S. Hursthouse^d, Luis Madrid^e, Sonia Rodrigues^b, Marko Zupan^c.

^a Department of Pure & Applied Chemistry, University of Strathclyde, 295 Cathedral St, Glasgow G1 1XL, UK.

^e Instituto de Recursos Naturales y Agrobiología de Sevilla (CSIC), Apartado 1052, 41080, Sevilla, Spain.

^f DI.VA.P.R.A., Chimica Agraria, Università di Torino, Via Leonardo da Vinci, 44, 10095 Grugliasco, Torino, Italy

* corresponding author: <u>c.m.davidson@strath.ac.uk</u>

Abstract

The revised (four-step) BCR sequential extraction procedure has been applied to fractionate the chromium, copper, iron, manganese, nickel, lead and zinc contents in urban soil samples from public-access areas in five European cities. A preliminary inter-laboratory comparison was conducted and showed that data obtained by different laboratories participating in the study were sufficiently harmonious for comparisons to be made between cities and land types (e.g. parks, roadside, riverbanks etc). Analyte recoveries by sequential extraction, with respect to direct *aqua regia* digestion, were generally acceptable ($100 \pm 15\%$). Iron, nickel and, at most sites, chromium were found mainly in association with the residual phase of the soil matrix. Copper was present in the reducible, oxidisable and residual fractions, whilst zinc was found in all four sequential extracts. Manganese was strongly associated with reducible material as, in some cities, was lead. This is of concern because high lead concentrations were generally highest in older, more heavily industrialised cities. Copper, lead and zinc showed marked (and often correlated) variations in concentrations between sites within the same city whereas manganese and, especially, iron, did not. No overall relationships were, however, found between analyte concentrations and land use, nor between analyte partitioning and land use.

Keywords

Potentially toxic elements, potentially toxic metals, urban soil, sequential extraction

^b Departamento de Química, Universidade de Aveiro 3810-193 Aveiro, Portugal.

^e Univerza v Ljubljani, Biotehniska fakulteta, 1000 Ljubljana, Slovenia.

^d School of Engineering & Science, University of Paisley, High Street, Paisley, PA1 2BE, UK.

Introduction

Urban soils are less studied than agricultural soil but are an increasing focus of attention [1]. Soils in urban environments are often highly variable in composition, due to a wide variety of human influences, and typically contain higher loadings of contaminants than those from rural settings due to the higher density of anthropogenic activity in urbanised areas. Only a small fraction of urban soil is used for food production (domestic gardens, allotments etc). However, since the majority of the population in developed countries live or work in urban areas, this is the type of soil with which most people have direct contact through, for example, use of public-access recreational areas such as parks and ornamental gardens.

Fractionation of the trace element content of environmental solids by application of a series of extractants provides (operationally-defined) information on reservoirs of metal that are likely to respond in different ways to changes in ambient conditions such as pH or redox potential [2]. The approach also gives an indication of the mineralogical phase(s) to which analytes are bound, although this information is limited by phenomena such as analyte re-adsorption, incomplete extraction, and non-specificity of reagents for "target" phases. The large variety of sequential extraction procedures now available was highlighted recently [3]. However, one of the most widely applied protocols remains that developed [4] and later revised [5] under the auspices of the European Commission, Community Bureau of Reference (BCR). Although originally intended for application to sediments, the BCR scheme has been applied successfully in the study of a wide variety of other substrates, including several different types of soils. It has the advantages of simplicity (only 4 steps are involved) and availability of certified reference materials (e.g. BCR CRM 701) for method validation.

Concentrations of potentially toxic elements (PTE) in urban soils have been reported for cities in diverse parts of the world. These include Aberdeen (UK) [6], Basel (Switzerland) [7], Bombay (India) [8], Damascus (Syria) [9], East St Louis (USA) [10], Falun (Sweden) [11], Gaborone (Botswana) [12], Glebe (Australia) [13], Hong Kong (China) [14], Honolulu (USA) [15], Jakobstad (Finland) [16], Central Jordan [17], Kyoto (Japan) [18], Madrid (Spain) [19], Montreal (Canada) [20], Naples (Italy) [21], New Orleans (USA) [22], Palermo (Italy) [23], Poznan (Poland) [24] and Sevilla (Spain) [25]. There has also been interest in the application of sequential extraction to urban soils, from the early work of Gibson and Farmer in Glasgow (UK) [26] to more recent studies in Bangkok (Thailand) [27], Danang-Hoian (Vietnam) [28], Hangzhou [29], Nanjing [30], Hong Kong [31, 32] (all in China) and Naples (Italy) [33]. However, a major limitation of previous research in the field

is the use of different sequential extraction protocols by different workers. Application of different reagents, extraction conditions, and number of extraction steps (between three and seven) makes it extremely difficult to compare results obtained in different cities.

This paper describes the application of a harmonised sequential extraction procedure, the revised BCR protocol [5], to fractionate the chromium, copper, iron, manganese, nickel, lead and zinc content in urban soils from public-access areas in five European cities, together with results of a preliminary exercise conducted to assess inter-laboratory comparability. The work was conducted as part of the EU URBSOIL project, the overall aims of which were to assess the role of soil as a source, or sink, for pollutants in the urban environment, and to promote the integration of soil quality information into the urban planning process.

Experimental

Sampling

Urban soil samples were collected, by means of a spade or trowel, from different types of public-access space (ornamental gardens (OG), parks and open spaces (PO), riverbanks (RB), roadsides (RD), school playing fields (SC) and urban agricultural land (AG)). Soils from five European cities were studied: Aveiro, Portugal (AVE); Glasgow, UK (GLA); Ljubljana, Slovenia (LJU); Sevilla, Spain (SEV) and Torino, Italy (TOR). Samples from twenty sites in each city, drawn from a larger set, were selected for sequential extraction. Sites were selected to give broad geographical coverage and to reflect the most important land use types within each city. A harmonised sampling regime was applied by all partners, although minor modifications were sometimes necessary because of the presence of paved pathways or other obstacles. Soil was collected from five individual sampling points at each site, to a depth of 10 cm, and bulked to give a composite sample. For PO sites, the sampling points were defined as the centre and corners of a square with a diagonal of length 20 m. For RD and RB sites the points were at 10 m intervals along a 40 m transect parallel to the road or river. OG, SC and AG sites were sampled similarly to PO, where feasible, but frequent alterations to the sampling pattern were necessary to take account of the locations and dimensions of the accessible soil areas. Soils were air dried and sieved to < 2 mm particle size in the city where they were collected before being re-distributed amongst project partners for analysis. Sequential extraction and measurement of PTE was carried out in three cities: soils from Aveiro, Sevilla and Torino were extracted and analysed in Ljubljana soils were extracted and analysed in Ljubljana; Glasgow soils were extracted at the University of Strathclyde then the extracts transported to the University of Paisley for analysis.

Apparatus and Reagents

The equipment and reagents used in extraction and determination of chromium, copper, iron, manganese, nickel, lead and zinc are summarised in Table 1. Quantification of cadmium was also attempted but levels proved to be too close to procedural detection limits (typically 0.5 mg kg⁻¹ for this element) in many of the extracts, for reliable results to be obtained. All plastic and glassware was acid-soaked overnight (5 % HNO₃) and rinsed with distilled water before use.

Procedures

Pseudo-total metal contents were determined by digesting air-dried urban soil samples in hot aqua regia (3 HCl: 1 HNO₃) as described in Table 1.

Sequential extraction was carried out according to the revised BCR protocol. The procedure is summarised below and described in full in reference [5].

Step 1 (water / acid soluble and exchangeable fraction): $40 \text{ ml of } 0.11 \text{ mol } l^{-1}$ acetic acid was added to 1 g soil and shaken for 16 h at room temperature. The extract was separated from the solid by centrifugation, decanted into a polyethylene bottle and stored at 4 °C. The residue was washed with distilled water and the washings discarded.

Step 2 (reducible fraction): 40 ml of $0.5 \text{ mol } l^{-1}$ hydroxylammonium chloride (adjusted to pH 1.5 by the addition of a prescribed volume of nitric acid) was added to the residue from step 1, and the extraction performed as described for step 1.

Step 3 (oxidisable fraction): The residue from step 3 was digested with hydrogen peroxide (for details see [5]), reduced in volume, then 50 ml of 1.0 mol l^{-1} ammonium acetate (adjusted to pH 2 with nitric acid) was added and the extraction performed as described for step 1.

Step 4 (residual fraction): The residue from step 3 was slurried in a small volume of *aqua regia*, transferred (as quantitatively as possible) to a suitable vessel and digested using the same procedure as described for the pseudo-total metal determination.

Results and Discussion

Analysis of CRMs

All three laboratories conducting the analysis extracted appropriate certified reference materials, in parallel with extraction of the urban soil samples, to assess the quality of their methodology. Results are presented in Table 2. Agreement between found and certified values was generally acceptable i.e. results were within two standards deviations of the certified concentration in most steps, and within three standard deviations in almost all steps. Exceptions occurred for the measurement of chromium in step 3 by the Glasgow laboratory (in BCR CRM 601); for copper, nickel and lead in step 2, copper in step 3, and nickel in step 4, by workers in Sevilla, and for copper in steps 1-3, lead in steps 2 and 4, and zinc in step 1, in Ljubljana. Some of the discrepancies may be due to underestimation of the uncertainty in the measured values, but there is a suggestion that a portion of the copper certified as being in the acid-exchangeable and reducible fractions of BCR CRM 701 is not being isolated until step 3.

Inter-laboratory comparison

A set of urban soil secondary reference materials had been prepared by participants in the URBSOIL project for use as internal quality control samples. One of these soils was used to compare the performance of the three laboratories conducting sequential extraction. The purpose was to assess whether minor deviations from the prescribed BCR protocol, such as use of shaking speeds outside the recommended range of 30 ± 10 rpm, would cause discontinuities between results generated in Glasgow, Ljubljana and Sevilla. Such an assessment is an important (but often neglected) preliminary stage of collaborative research projects, necessitated by the availability of different apparatus, and the use of different, preferred, analytical procedures, in different laboratories. The selected soil was from a park close to the centre of Glasgow. It was considered a more suitable basis for the inter-laboratory comparison than available certified reference materials because it is similar in matrix composition (and analyte concentrations) to the samples studied. In contrast, both BCR CRM 601 and BCR CRM 701 are freshwater lake sediments.

Statistical analysis (ANOVA at 95 % CI, except for Fe where a t-test assuming unequal variance was applied) indicated significant between-laboratory variability in the pseudototal metal concentrations reported by the Glasgow, Ljubljana and Sevilla laboratories for the Glasgow urban soil (Table 3). Results differed by up to 25% for some elements. This may be partly due to the inhomogeneity typical of urban soils. However, it is noticeable that, for many elements, lowest concentrations were those measured in Sevilla and highest were those from Ljubljana, suggesting that the microwave digestion procedures used were of unequal efficiency.

Despite the variability in pseudototal metal concentrations generally similar PTE fractionation patterns were obtained by the three laboratories (Figure 1). Also, metal recoveries by means of the BCR procedure, Σ (steps 1-4), were between 87 and 110 % of corresponding pseudototal values. There were a few exceptions. The Sevilla laboratory appeared to slightly under-extract iron and nickel (84 and 81%, recovery, respectively) and the Ljubljana laboratory obtained excess extractable zinc (134 % recovery). Inefficient attack on the solid phase, in steps 2 (reducible) and 4 (residual), may explain the poor iron and nickel results, but this is not supported by data obtained for other elements in Sevilla e.g. chromium and lead. The anomalous zinc recovery in Ljubljana arises because concentrations almost equivalent to the pseudototal value were found, in the step 4 extracts, in three of the four replicates analysed (the fourth replicate produced an acceptable recovery of 98 %). This suggests that gross contamination of some of the residual extracts had occurred – and highlights the necessity to cross-check sequential extraction results with those obtained by direct *aqua regia* digestion. Recoveries less than equivalent pseudototal values were reported for copper by all three participants. The worst performance for this element was that obtained in Glasgow (recovery 87 %). This appears to result from underestimation of the copper content in the oxidisable phase. The loss may represent copper associated with low-density, particulate, organic matter, which is difficult to isolate by centrifugation and may be inadvertently discarded with the washings at the end of steps 1 and 2; or loss of copper in the form of water-soluble complexes with dissolved organic matter. However, it is not clear why the effect should be more prevalent in one laboratory than in the others.

Between-laboratory discrepancies of the magnitude reported are typical in work of this nature (see, for example, results of the inter-comparisons conducted during certification of BCR CRM 601 [36]), Overall, there was sufficient similarity between results obtained in the three laboratories for comparisons to be made between analyte partitioning patterns in urban soil samples analysed in different laboratories.

PTE fractionation in urban soil samples

The fractionation of analytes in 20 urban soils from each of the five cities is shown in Figures 2-8. Results are arranged by analyte, city and land type. Mean recoveries with respect to pseudototal values for each analyte and city were in the range 87-110%, which is considered acceptable given the intrinsic heterogeneity of urban soils. Exceptions occurred for chromium in soils from Aveiro (142% recovery), copper in Glasgow soils (82%), iron in samples from Sevilla (77%) and nickel in soils from both Aveiro (66%) and Sevilla (70%). In each case, the effect (under- or over-extraction of a particular analyte relative to direct digestion with *aqua regia*) was observed for almost all soils from the city. This is probably due, in part, to inter-laboratory differences already seen in the analysis of the inter-comparison sample i.e.

low copper recovery in Glasgow, low iron and nickel recoveries in Sevilla (who also analysed the urban soils from Aveiro). However, the fact that recoveries closer to 100% were obtained when the Sevilla laboratory measured chromium, iron or nickel in other samples suggests that the soil type also has an influence on the results obtained by the BCR extraction.

Levels of extractable chromium (Figure 2) varied markedly between cities; from an average of 13.2 mg kg^{-1} in Aveiro to an average of 227 mg kg^{-1} in Torino. The high concentrations in the latter city are, at least partly, of geochemical origin [37]. Low chromium concentrations (< 60 mg kg⁻¹), relatively similar levels across all sites, and release of the analyte at all four stages of the sequential extraction, were characteristics of Aveiro and Sevilla soils. Soils from the other cities showed more noticeable variation between sites, although this did not appear to correlate with land use. Residual chromium species were predominant at most sites in Glasgow and Ljubljana. However, where a particularly high concentration was found, this often correlated with potentially labile (reducible and oxidisable) forms e.g. sites GLA PO.15, GLA PO.32 and LJU PO.15. The south-east of Glasgow is known to be contaminated with chromium because of the disposal of waste by chemical manufacturing industry in the 19th Century [38], and this may explain the high levels at GLA PO.15. However, GLA PO.32 is in a different area of the city and the reason for the elevated level in this soil is unknown. Site LJU PO.15 is relatively close (~ 200 m) to a chemical plant that manufactures products for various customers, including the leather industry, and so may constitute a point source for chromium. Soils from Torino contained mainly residual forms of chromium but, again, some of the sites with the highest analyte content also had higher than average contribution from reducible and oxidisable fractions. Higher chromium concentrations were generally observed in roadside than in park soils, from this city, with the highest overall concentration at sites TOR RD.161 and TOR RD.167, although there are no known anthropogenic point sources at these locations.

Aveiro soils again contained the lowest average concentration for copper (Figure 3) with the notable exception of site AVE OG/RD.17, which contained 10x more copper than found elsewhere in the study (474 mg kg⁻¹), and also an unusually high proportion of acid exchangeable species. The result is unlikely to be due to error during sequential extraction since the copper recovery with respect to the pseudototal concentration at this site was 103%. Rather, it appears to represent a "hot-spot" of copper contamination in the soil, perhaps arising from the nearby road or canal. Interestingly, a sample taken from a very similar location one year earlier contained only 17.4 mg kg⁻¹ Cu. This highlights the need to consider both spatial and temporal variability when assessing PTE content in urban systems. The other cities' soils showed the, well documented, importance of the oxidisable fraction for copper - probably representing associations with organic matter - together with mainly residual forms in Ljubljana and Sevilla, and reducible and residual forms in Glasgow and Torino. Marked concentration variations were evident in most cities, but no clear

relationship between copper content and type of land sampled. Recent work in Sevilla [39] has shown that copper, lead and zinc concentrations were slightly lower than average in riverbank samples and noticeably higher in soil from ornamental gardens, but this was based on a larger sample set than the current study.

Extractable concentrations of iron were higher in Glasgow and Torino soils than in soils from Aveiro and Sevilla (Figure 3) but the operational speciation was similarly dominated by residual species (as might be expected for an element whose minerals constitute a major structural components of soil) with small contributions from reducible forms. There was relatively little between-site variation within each city. Similar comments apply to manganese (Figure 5) although, here, the reducible fraction was generally the largest (as expected since manganese oxyhydroxides are a principal target of the hydroxylammonium chloride reagent) and variability between sites was somewhat larger than for iron. Ljubljana and Torino soils were, on average, the richest in extractable manganese (levels of 894 and 812 mg kg⁻¹, respectively), followed by Sevilla (424 mg kg⁻¹) and Glasgow (416 mg kg⁻¹). Lowest concentrations were found in the samples from Aveiro (mean = 97 mg kg⁻¹).

Nickel displayed some similar trends to chromium. Residual forms were dominant in all cities, highest concentrations were found in Torino (due, as for chromium, to the presence of serpentinites in the River Po catchment [37]), and sites TOR RD.161 and TOR RD.167 contained the highest levels of all those included in the study (490 and 824 mg kg⁻¹, respectively). In contrast, no sites from the other cities, except GLA PO.15 and GLA PO.24, contained > 50 mg kg⁻¹ extractable nickel.

Of the elements studied, lead showed the greatest between-site variation in concentration, especially in Sevilla and Torino (see Figure 7). Mean extractable lead concentrations were of a similar magnitude in Glasgow (212 mg kg⁻¹), Sevilla (183 mg kg⁻¹) and Torino (221 mg kg⁻¹) but contamination appeared to be more widespread in Glasgow, whereas the two other cities' soils were generally less polluted, except for specific sites with high lead levels (> 500 mg kg⁻¹). In terms of extraction patterns, lead was present mainly in reducible forms in Aveiro, Glasgow and Torino (in contrast to previous work involving the BCR extraction in Naples [33] which reported 77 % residual lead) but in oxidisable and residual forms in Sevilla. Results from Ljubljana were unusual in that some sites, within the same land-use category, were dominated by reducible lead species and some by oxidisable lead species. This effect may be related to carbonate content. Ljubljana soils containing > 30 % CaCO₃ (LJU.RD.02, LJU.RD.04, LJU.RD.05 and LJU.SC.12) released the most lead in step 3 whilst those with < 15 % CaCO₃ (except LJU.SC.08) released most lead in step 2. Soils with between 15 and 30 % showed variable behaviour.

In particularly carbonaceous samples, such as those obtained from some sites in Ljubljana and Sevilla, it may be the case that, not only is there insufficient acetic acid in step 1 of the extraction to dissolve all the CaCO₃ present, but also the residual carbonate is able to significantly raise the pH of the reagent used in step 2. This decreases the efficiency of the extraction, leading to underestimation of the amount of lead associated with the reducible phase. The analyte is thus recovered in step 3. The effect is clearest for lead but can be observed for other elements e.g. copper. This finding highlights the operational nature of sequential extraction and demonstrates clearly that such procedure cannot be relied upon to provide information on specific, mineralogical phase associations of PTEs.

Zinc was found in association with all four operationally-defined fractions in most soils (Figure 8) although the proportions varied between cities and sites. It is noticeable that many of the sites containing the highest extractable zinc levels were also the richest in copper and lead (GLA PO.15 and PO24; SEV OG.3, PO.25, PO.65 and RD.5; TOR PO.216 and RD.132). This suggests that the levels of these elements are influenced by common factors within the urban environment. Indeed, copper, lead and zinc are sometimes described as "urban" metals [see e.g. 25]. Fractionation patterns for the three elements were not necessarily similar, however, and differences were evident between analytes (compare copper, lead and zinc distributions in soil TOR PO.216) and between cities.

Conclusions

Application of a harmonised sequential extraction procedure has allowed comparison of the operationally defined partitioning of seven PTEs in urban soil samples from five European cities. Despite minor procedural differences, results obtained by different laboratories were sufficiently comparable, and similar to certified values, to allow measurements made in different countries to be included in the study.

Analyte recoveries by sequential extraction, with respect to direct *aqua regia* digestion, were generally acceptable $(100 \pm 15\%)$. In terms of overall extractable metal contents (sum of steps of sequential extraction) soils from Glasgow and Torino were generally more highly contaminated than those from Ljubljana and Sevilla. Lowest PTE concentrations were found in Aveiro soils. This is as expected i.e. the oldest cities, with historical legacies of heavy manufacturing industry, have the highest concentrations of potentially toxic elements in soil.

Urban metals (copper, lead and zinc) showed marked variations in concentrations between sites within the same city, with the highest values for all three elements often found together in the same samples. Some relations were also evident between concentrations of chromium and nickel, especially in Torino where levels were highest. In contrast, elements more likely to originate mainly from geochemical sources, such as iron, showed low variability in concentration across each city.

Iron, nickel and, at most sites, chromium were found mainly in association with the residual phase of the soil matrix. Copper was mainly in reducible, oxidisable and residual forms, whilst zinc was found in all four sequential extracts. Manganese, together with lead in Aveiro, Glasgow and Torino, was mainly associated with reducible material. This is of concern because lead concentrations at some sites are high and the potential exists for mobilisation under reducing conditions such as could result, for example, from water-logging. In contrast, lead in soils from Sevilla (and in some samples from Ljubljana) showed an apparent association with the oxidisable matrix component(s), although this may be partly an artefact caused by the differing carbonate content of the soil matrix.

No clear relationships were found between analyte concentrations and land use, nor between analyte partitioning and land use.

Acknowledgements

The URBSOIL project was supported by the EU FP5, Energy, Environment and Sustainable Development Program, under contract EVK4-CT-2001-00053.

References

- P Bullock and P J Gregory (Eds.), Soils in the Urban Environment, Blackwell, Oxford, 1991.
- 2. A M Ure and C M Davidson (Eds.), Chemical Speciation in the Environment (2nd Edn), Blackwell, Oxford, 2002.
- 3. A V Filgueiras, I Lavilla, C Bendicho, J. Environ. Monit., 4 (2002) 823.
- 4. A M Ure, Ph Quevauviller, H Muntau, B Griepink, Int. J. Environ. Anal. Chem., 51 (1993) 135.
- 5. G Rauret, J F Lopez-Sanchez, A Sahuquillo, R Rubio, C Davidson, A Ure, Ph Quevauviller, J. Environ. Monit., 1 (1999) 57.
- 6. E Paterson, M Sanka, L Clark, Appl. Geochem., 11 (1996) 129.
- 7. M Niederer, A Maschka-Selig, C Hohl, Environ. Sci. Pollut. Res., 2 (1995) 83.
- 8. D S Ratha, B K Sahu, Environ. Geol., 22 (1993) 276.
- 9. A Moller, H W Muller, A Abdullah, G Abdelgawad, J Utermann, Geoderma, 124 (2005) 63.
- 10. M D Kaminski, S Landsberger, J. Air Waste Manage. Assoc., 50 (2000) 1667.
- 11. Z X Lin, K Harsbo, M Ahlgren, U Qvarfort, Sci. Total Environ., 209 (1998) 47.
- 12. M Zhai, H A B Kampunzu, M P Modisi, O Totolo, Environ. Geol., 45 (2003) 171.
- 13. J A Markus, A B McBratney, Aust. J. Soil Res., 34 (1996) 453.
- 14. X D Li, S L Lee, S C Wong, W Z Shi, L Thornton, Environ. Pollut., 129 (2004) 113.
- 15. R A Sutherland, C A Tolosa, Water Air Soil Pollut., 127 (2001) 315.
- 16. P Peltola, M Astrom, Environ. Geochem. Health, 25 (2003) 397.
- 17. K M Banat, F M Howari, A A Al-Hamad, Environ. Res., 97 (2005) 258.
- 18. K Katahira, M Yoneda, A Morisawa, Environ. Technol., 21 (2000) 505.
- 19. E De Miguel, M J de Grado, L F Llamas, A Martin-Dorado, L F Mazadiego, Sci. Total Environ., 215 (1998) 113.
- 20. Y Ge, P Murray, W H Hendershot, Environ. Pollu., 107 (2000) 137.
- 21. M Angelone, G Armiento, D Cinti, R Somma, A Trocciola, Fresen. Environ. Bull., 11 (2002) 432.
- 22. H W Mielke, G Wang, C R Gonzales, B Le, V N Quach, P W Mielke, Sci. Total Environ., 281 (2001) 217.

- 23. D S Manta, M Angelone, A Bellanca, R Neri, M Sprovieri, Sci. Total Environ., 300 (2002), 229.
- 24. W Grzebisz, L Ciesla, J Komisarek, J Potarzycki, Pol. J. Environ. Stud., 11 (2002) 493.
- 25. L Madrid, E Diaz-Barrientos, R Reinoso, F Madrid, Eur. J. Soil Sci., 55 (2004) 209.
- 26. M J Gibson and J G Farmer, Environ. Pollut. B Chem. Phys., 11 (1986) 117.
- 27. W Wilcke, S Muller, N Kanchanakool, W Zech, Geoderma, 86 (1998) 211.
- 28. H T T Thuy, H J Tobschall, P V An, Environ. Geol., 39 (2000) 603.
- 29. M K Zhang, Z V Ke, Pedosphere, 14 (2004) 177.
- 30. Y Lu, Z T Gong, G L Zhang, W Burghardt, Geoderma, 115 (2003) 101.
- 31. C S C Wong, X D Li, Sci. Total Environ., 319 (2004) 185.
- 32. X D Li, C S Poon, P S Lui, Appl. Geochem., 16 (2001) 1361.
- 33. M Imperato, P Adamo, D Naimo, M Arienzo, D Stanzione, P Violante, Environ. Pollut., 124 (2003) 247.
- 34. Report EUR 19502 EN, G Rauret, J F Lopez-Sanchez, A Sahuquillo, H Muntau and Ph Quevauviller, European Commission, Brussels, 2000.
- 35. Report EUR 19775 EN, G Rauret, J F Lopez-Sanchez, D Luck, M Yli-Halla, H Muntau and Ph Quevauviller, European Commission, Brussels, 2001.
- 36. Report EUR 17554 EN, Ph. Quevauviller, G Rauret, J F Lopez-Sanchez, R Rubio, A M Ure and H Muntau, European Commission, Brussels, 1997.
- 37. M Biasioli, R Barberis and F Ajmone-Marsan, Sci. Total Environ., in press.
- 38. C Whalley, A Hursthouse, S Rowlatt, P Iqbal-Zahid, H Vaughan and R Durant, Water Air Soil Pollut., 112 (1999) 389.
- 39. E Ruiz-Cortes, R Reinoso, E Diaz-Barrientos and L Madrid, *Environ. Geochem. Health*, 27 (2005) 465.

Legends for Figures

Figure 1 Analyte fractionation patterns and recoveries with respect to equivalent pseudototal PTE concentrations obtained when three different laboratories sequentially extracted the same urban soil. Results presented are averages over replicate extractions: GLA (Glasgow, n = 7); LJU (Ljubljana, n = 4); SEV (Sevilla, n = 4). LJU did not measure iron concentration.





Figure 2. Fractionation of chromium in urban soils by means of the BCR sequential extraction.

Figure 3. Fractionation of copper in urban soils by means of the BCR sequential extraction (results for soil AVE OG/RD.17 plotted at 1/10th actual concentrations).





Figure 4. Fractionation of iron in urban soils by means of the BCR sequential extraction.



Figure 5. Fractionation of manganese in urban soils by means of the BCR sequential extraction.



Figure 6. Fractionation of nickel in urban soils by means of the BCR sequential extraction.



Figure 7. Fractionation of lead in urban soils by means of the BCR sequential extraction.



Figure 8. Fractionation of zinc in urban soils by means of the BCR sequential extraction

Table 1.	Details of apparatus	and reagents use	ed by project partne	ers involved in seque	ential extraction of	urban soil samples
	11	0		1		1

	Shaker (Type, model and speed)	Centrifuge (Model and conditions)	Microwave- assisted digestion (Model and conditions)	PTE measurement (Technique and instrument)	Calibration	Reagents (grade and supplier)
Glasgow	End-over-	MSE	CEM MDS-	ICPOES	Multi-element, reagent-	Nitric acid, hydrochloric acid and acetic acid (all
(Strathclyde/	end type	Mistral	2000,	(PerkinElmer	matched standards	AR grade) from Aldrich (Gillingham, UK)
Paisley)	(GFL	1000,	1 g soil : 20 ml	Optima 3000)	prepared from single-	Ammonium acetate and hydroxylammonium
	3040) 23	3000 g,	acid		element commercial	chloride (Analak) from Merck / BDH (Poole, UK)
	rpm.	10 min			solutions (Merck)	(Manahastar, LW)
T in h line a	Orbite1	II.44:-1	CEM MDC			Nitria and hadre alleria and a setting and
Ljubijana	Orbital	Hettich,	CEM, MDS-	FAAS	Single element standards	Nitric acid, nydrochloric acid, acetic acid,
	snaker (Vallass	Rotanta 96	2000,	(PerkinEimer	prepared from	ammonium acetate and hydrogen peroxide (all For
	(Yellow	S, 3000 g,	1 g soll : 20 ml	1100B)/	commercial solutions	Analysis grade) from Merck.
	Line OS 2	20 min	acid	EIAAS	(Merck)	Hydroxylammonium chloride (PA) from Fluka
	basic) 550			(PerkinElmer		Chimica
	rpm			Zeeman 5100)		
Sevilla	Orbital	Backman	Milestone	ICPOES	Multi-element reagent-	All reagent AR grade from Merck
	shaker	Coulter,	Ethos 900,	(Thermo Jarrell	matched standards	
	(New	3000 g,	0.5 g soil : 10	Ash IRIS	prepared from single-	
	Brunswick)	10 min	ml acid	Advantage)	element commercial	
	200 rpm				solutions (Merck)	

CRM			Replicates	Cr	Cu	Ni	Pb	Zn
BCR 601	Step 1	Indicative		0.35 ± 0.08	10.5 ± 0.8	7.82 ± 0.84	2.28 ± 0.44	261 ± 13
[34]		GLA	7	0.33 ± 0.10	10.8 ± 2.1	6.66 ± 0.98	1.57 ± 0.82	240 ± 34
	Step 2	Indicative		10.6 ± 0.9	72.8 ± 4.9	10.6 ± 1.2	205 ± 11	266 ± 17
		GLA	7	10.0 ± 0.6	68.9 ± 7.5	10.9 ± 0.4	208 ± 11	266 ± 16
	Step 3	Indicative		14.4 ± 2.6	78.6 ± 8.9	6.04 ± 1.27	19.7 ± 5.8	106 ± 11
		GLA	7	11.9 ± 0.2	65.3 ± 5.0	6.68 ± 0.24	16.9 ± 3.1	106 ± 5
	Step 4	Indicative		78.2 ± 6.5	60.4 ± 4.9	50.5 ± 4.3	38.0 ± 8.7	161 ± 14
		GLA	7	85.0 ± 2.7	59.7 ± 1.7	54.0 ± 2.9	40.0 ± 4.4	169 ± 4
BCR 701	Step 1	Certified		2.26 ± 0.28	49.3 ± 3.0	15.4 ± 1.6	3.18 ± 0.37	205 ± 10
[35]		LJU	3	2.10 ± 0.30	31.8 ± 1.5	15.9 ± 2.4	3.10 ± 0.20	154 ± 15
		SEV	6	2.32 ± 0.40	39.4 ± 7.0	11.9 ± 1.9	2.78 ± 0.63	162 ± 24
	Step 2	Certified		45.7 ± 3.1	124 ± 6	26.6 ± 2.0	126 ± 5	114 ± 8
		LJU	3	44.8 ± 1.4	96.9 ± 5.2	21.9 ± 1.9	108 ± 4	119 ± 6
		SEV	6	45.9 ± 8.8	103 ± 2	20.7 ± 0.7	102 ± 4	115 ± 9
	Step 3	Certified		143 ± 12	55.2 ± 6.1	15.3 ± 1.5	9.28 ± 3.03	45.7 ± 5.1
		LJU	3	150 ± 8	80.2 ± 7.5	14.4 ± 1.3	10.5 ± 1.3	44.9 ± 4.6
		SEV	6	148 ± 8	74.8 ± 3.7	17.1 ± 2.0	10.2 ± 1.7	61.4 ± 10.0
	Step 4	Indicative		62.5 ± 7.4	38.5 ± 11.2	41.4 ± 4.0	11.0 ± 5.2	95 ± 13
		LJU	3	64.5 ± 5.4	40.3 ± 3.6	45.8 ± 2.9	18.2 ± 2.1	104 ± 6
		SEV	6	49.2 ± 6.0	31.2 ± 8.0	30.0 ± 2.8	15.2 ± 2.3	80.3 ± 9.8

 Table 2.
 Results obtained for analysis of reference soils certified for extractable PTE contents (mg kg⁻¹)

Results presented are mean values ± one standard deviation for the number of replicates specified; GLA indicates results obtained by Team Glasgow (Strathclyde/Paisley); LJU results obtained in Ljubljana; SEV results obtained in Sevilla

	Replicates	Cr	Cu	Fe (%)	Mn	Ni	Pb	Zn
Glasgow	34	43.2 ± 3.0	111 ± 5	3.06 ± 0.12	442 ± 18	48.8 ± 7.0	389 ± 25	177 ± 11
Ljubljana	5	48.5 ± 0.6	133 ± 1	NA	487 ± 6	57.0 ± 0.8	439 ± 3	180 ± 7
Sevilla	6	37.2 ± 1.1	108 ± 2	2.76 ± 0.42	369 ± 6	50.5 ± 1.3	343 ± 7	151 ± 2

Table 3. Results obtained for analysis of Glasgow urban soil (pseudototal analyte concentration in mg kg^{-1}).

Results presented are mean values \pm one standard deviation for the number of replicates specified