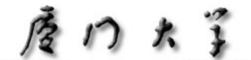
**University Code: 10384** 

Student ID No.: 20420131154525



## **MASTER DEGREE THESIS**

Lignin and lignin-acetoderivative oxidative degradation into

### valuable aromatic aldehydes

#### By

### **KISSA RAJIF ALUNGA**

(Uganda)

Supervisor: Professor Yunquan Liu Major: Chemical Engineering Date of Graduation: July, 2015

A THESIS SUBMITTED FOR THE DEGREE OF MASTER OF ENGINEERING DEPARTMENT OF CHEMICAL AND BIOCHEMICAL ENGINEERING COLLEGE OF CHEMISTRY AND CHEMICAL ENGINEERING XIAMEN UNIVERSITY

## CERTIFICATION

I, Professor....., hereby certify that I have read this manuscript and recommend for acceptance by the Xiamen University a dissertation entitled "*Lignin and lignin-acetoderivative oxidative degradation into valuable aromatic aldehydes*" in fulfillment of degree of Master of Engineering at Xiamen University, People's Republic of China.

Signed: Supervisor: Date:

Department of Chemical and Biochemical Engineering College of Chemistry and Chemical Engineering Xiamen University Xiamen, Fujian Province P.R. China

## **ORIGINAL STATEMENT**

The research described in this thesis for Master of Chemical Engineering was conducted under the supervision of Professor Yunquan Liu and Dr. Yue-Yuan Ye at the Institute for BioEnergy, School of Energy Research, Xiamen University. I hereby declare that the work submitted is my own and that appropriate credit has been given where reference has been made to the work of others. I also confirm that it has not been previously or concurrently submitted for any other degree, diploma or any other qualifications at Xiamen University, P.R China or other institutions.

## **COPYRIGHT DECLARATION**

All rights reserved. No part of this dissertation may be reproduced, stored in any retrieval system, or transmitted in any form by any means: electronic, mechanical, photographing, recording or otherwise without prior written permission of the author or Xiamen University.

NAME OF STUDENT: Date:

## ACKNOWLEDGEMENT

In all humility, I would like to acknowledge the Almighty God for giving me the opportunity to advance my academic and professional carrier in China. I am enormously indebted to my research supervisor, Professor Yunquan Liu, for all of his advice, support and guidance and also for providing me with the opportunity to join his research group in his laboratory. I am also very grateful to Dr. Yue-Yuan Ye, Dr. Shui-Rong Li, Dr. Duo Wang for their limitless help, orientation, advice and guidance throughout my study period, I am as well thankful to my fellow lab mates for their great help especially for teaching me the methodology of various experiments, their patience and friendly attitude towards me. I humbly thank all the lecturers in the college of Energy, most especially the college principal, special thanks also goes to my Lecturers in the college of Chemistry and Chemical Engineering; I believe they are the driving force behind my achievement now and in the future. My special thanks go to the Government of the People's Republic of China under the communist party who through the Chinese Scholarship Council provided me the scholarship to study at Xiamen University, China.

Finally, I would like to express my gratitude to my family members back home in Uganda for their encouragement, moral and other support in various ways, then my friends; Faisal Soyekwo, Musa Mubiru Kisiriko and William Brown, you have been always there for me when I needed you most. May the almighty God, in His infinite mercy spread His blessings on all those who in diverse ways contributed to my success.

#### Abstract

The main objective of this work was to study the transformation of lignin and its by-products (acetoderivatives) into valuable aromatic aldehydes as well as the kinetics of acetoderivative degradation during the alkaline catalytic oxidation. Bamboo lignin was transformed into valuable aromatic aldehydes and acetoderivatives. Vanillin was recovered from acetovanillone with the application of copper catalyst. Syringaldehyde recovery from acetosyringone was reported for the first time, thus adding value to the lignin oxidation process through transformations of its byproducts. A reliable and consistent relative response factor (RRF) method for the quantification of lignin and acetoderivatives products was successfully developed which was latter used in the GC-MS analysis. A total product yield of approximately 9.5% from bamboo lignin and a vanillin yield of 55% with greater than 90% selectivity from acetovanillone were achieved. The proposed reaction pathways suggest a two-step route towards the formation of value-added aromatic aldehydes from lignin via degradation of acetoderivatives. The kinetics study for the degradation of acetovanillone and the formation of vanillin were reported for the first time over the temperature range of 120-150 °C, with the activation energies of 85.29 kJ/mol and 120.7 kJ/mol respectively. Further kinetics studies on the oxygen concentration impacts indicated that the acetovanillone disappearance and the vanillin formation were of 0.5 and 1.6 reaction orders with respect to the oxygen concentration respectively. To date, catalytic wet air oxidation (CWAO) breaks down the lignin polymer and toxic phenolic compounds reasonably and effectively, producing value-added aldehydes, which could become a potential new route for the recovery of value-added products.

Key words: Lignin; Acetoderivatives; Catalytic oxidation; Vanillin; Syringaldehyde

Abstract

#### 摘要

本项工作研究了木质素的转化以及其副产品(乙酸基衍生物)转变为高附加值的芳香醛, 乙酸基衍生物在碱性条件下催化氧化降解的动力学方程等问题。竹材木质素被转化为高附 加值的芳香醛和乙酸基衍生物。在铜作为催化剂的条件下,香草乙酮被转化为香草醛。另 外,乙酰丁香酮首次转化为丁香醛,从而通过降解木质素氧化过程中的副产物来提高整个 过程的价值。本文用 GC-MS 分析得到了一种可靠的一致的相对影响因子法来定量木质素和 乙酸衍生物。竹材木质素的总产率接近 9.5%,且香草醛的产率为 55%,其中 90%以上来自 于香草乙酮。来源于木质素,通过降解乙酸衍生物而得到高附加值的芳香醛分为两步反 应。文章首次研究了 120-150℃条件下香草乙酮的降解动力学和香草醛的形成,其活化能 分别为 85.29 kJ/mol 和 120.7 kJ/mol。此外,对氧浓度的影响的研究表明香草乙酮的消失 和香草醛的形成分别为 0.5 和 1.6 级反应。迄今为止,催化湿式氧化有效地降解了木质素 聚合物和有毒酚类化合物,产生了高附加值的醛类,这将成为一种新的有潜力的回收高附 加值物质的手段。

关键词:木质素,乙酸基衍生物,催化氧化,香草醛,丁香醛

## Scope of the study

The extent and structural format of the thesis has been organized into the following five sections. A general introduction and a comprehensive review of the associated literature including the motivation resulting into this study is systematically presented in chapter one. In chapter two, the materials and experimental methods used in the study are introduced. Chapter three presents the detailed results and discussion on the depolymerization of lignin by catalytic oxidation in alkali aqueous solvent. More results are presented in chapter four where the catalytic acetoderivative degradation under alkali oxidation was investigated with the sole purpose of determining the fate of acetoderivatives during lignin oxidation as well as determining the possibility of recovery of value added products from them.

Finally in chapter five, the conclusions resulting from the study are noted and the prospects for future work are suggested.

## **Table of Contents**

Acknowledgement	i
Abstract	ii
摘要	,iv
Scope of the study	v
List of Symbols and Abbreviations	ix
List of Figures	
List of Tables	XV
Chapter 1: Literature Review and Background of the Study	1
1.1. Introduction	1
1.2. Motivation of the Study	2
1.3. Structure of Lignin and Its Occurrence in Wood	4
1.4. Production of Lignin in Industry	
1.4.1. The Kraft Process	
1.4.2. The Steam Explosion Process	10
1.5. Catalytic Oxidation of Lignin and Acetoderivatives in Alkaline Medium	11
Chapter 2: Materials and Methods Used in the Study	14
2.1. List of Chemicals and Materials Used in the Study	14
2.2. Procedures and Experimental Setup of the Study	15
2.2.1. Batch reactor for lignin and acetoderivative oxidation	15
2.2.2. Separation and Recovery of lignin Fractions	16
2.3. Characterization Experiments	
2.3.1. Alumina Supported Copper Catalyst	

#### **Table of Contents**

2.3.2. Lignin, Acetoderivatives and their Products	18
2.4. Development of a more reliable and consistent RRF method over ES method	20
2.4.1. Selection of the internal standard	20
2.4.2. Preparation of standard solutions and Calibration curves	20
2.4.3. Comparison of quantitative methods; ES and RRF for consistency	22
Chapter 3: The Oxidation of Lignin to Aromatic Monomers	
3.1. Spectroscopic Characteristics of Lignin and Acetovanillone	
3.2. Influence of Oxidation Parameters on Lignin Degradation	
3.2.1. Alkaline oxidation of lignin with and without oxygen	
3.2.2. Effect of Temperature	29
<ul><li>3.2.2. Effect of Lignin Concentration</li></ul>	30
3.2.4. Effect of Catalyst Types	31
Chapter 4: The Oxidation of Lignin-Acetoderivative to Aromatic Aldehydes	
4.1. Introduction	35
4.2. Recovery of Acetoderivatives from Lignin	35
4.3. Influence of Oxidation Parameters on Acetoderivative Degradation	39
4.3.1. Effect of Time on the Catalytic Oxidation of AV to V	39
4.3.2. Effect of Concentration of AV	41
4.3.3. Effect of Alkaline Conditions and Oxygen Partial Pressure	43
4.4. Reaction Mechanism of AV to V	44
4.5. Comparative Studies of Heterogeneous and Homogeneous Systems at Optimal Rea	ction
Conditions	46
4.6. The kinetics study	50

#### **Table of Contents**

4.6.1. Influence of Temperature on the oxidation of A	AV to V	50
4.6.2. Influence of oxygen concentration on AV deg	radation and V formation	54
Chapter 5: Conclusion and Recommendation		61
5.1. Conclusions		61
5.2. Suggested Future Prospects		63
References		65
Appendix		74
Publication		76

## List of Symbols and Abbreviations

**RRF-** Relative Response Factor **ES-External Standardisation** V- Vanillin AV- Acetovanillone Av.-Average STDEV-Standard Deviation **RSD-Relative Standard Deviation** Wt-Weight LOPs- Lignin oxidized products AS- Acetosyringone SY-Syringaldehyde CAV- Concentration of acetovanillone Cv- Concentration of vanillin Ea- Activation energy, kJ/mol k*i*- Empirical kinetic constant Form.-Formation Degrad.-Degradation A- Constant in the Arrhenius equation M-Concentration of NaOH, mol L<sup>-1</sup> *m* -Reaction order  $P^0$  -Water vapor pressure  $P_{\text{NaOH}}$  -Water vapour pressure above NaOH solution, bar Po2 -Oxygen partial pressure, bar Pt -Total pressure in the reactor, bar  $r_{\rm AV}$  -Rate of AV hydrolysis to vanillin, g min<sup>-1</sup> L<sup>-1</sup>

rv -Rate of V formation, g min<sup>-1</sup> L<sup>-1</sup> T-Temperature, °C or K as appropriate t -Time, min or s as appropriate  $y_{02}$  -Oxygen mole fraction in the gas CWAO-Catalytic Wet Air Oxidation Eq.-equation

# **List of Figures**

<b>Figure 1.1.</b> The chemical structures of the three building blocks of lignin, p-Coumaryl (A),
coniferyl (B), and sinapyl (C) alcohols, the precursors of the p-hydroxyphenyl, guaiacyl, and
syringyl units, respectively
<b>Figure 1.2.</b> The Chemical Structure of β-O-4-(β-Ether) Bond
Figure 1.3. Representative Chemical Structure of Linear Oligomer of Lignin
Figure 1.4. Schematic representation of the position of lignin and its linkages with other
carbohydrate polymers in a plant cell wall. Adapted from [6]7
Figure 1.5. Overview of a Kraft process in the paper industry, adapted from [25]
Figure 2.1: The high pressure batch wise setup for lignin/AV oxidation
Figure 2.2. Isolation and qualification/quantification of various fractions
Figure 3.1: FT-IR spectra of lignin and acetovanillone depicting their functional groups 25
Figure 3.2. FT-IR spectra for bamboo lignin prior to and after alkali oxidation
Figure 3.3. Effect of lignin oxidation with and without (CSL) oxygen on different lignin
materials under similar oxidative conditions, BML-Bamboo lignin, CSL-Cornstalk lignin, NaSL-
Sodium Sulfonate lignin, LOP-Lignin oxidized products, SY-Syringaldehyde, V-Vanillin 27
Figure 3.4. Influence of temperature on the bamboo lignin conversion and concentrations of
vanillin and Syringaldehyde
Figure 3.5. Solid residue and lignin conversion profile with lignin concentration
Figure 3.6. Effect of lignin Concentration on the lignin oxidation as applied on cornstalk lignin
material at Temp.= 170 °C, Run time=20 min.,Po2=3 bars (LOP-Lignin oxidised products) 31
Figure 3.7. Variation of lignin conversion and yields of (V %) vanillin and (SY %)
syringaldehyde with reaction time

## List of Figures

Figure 4.1. Comparison for LOPs in different lignin sources at 170 °C, 1.5 bars, 30 min (non
catalyst)
Figure 4.2. Yields of aromatic aldehydes and ketones in non-catatlyzed and catalyzed with
CuSO <sub>4</sub> oxidation of bamboo lignin with oxygen in NaOH solution (170 °C, 5 min, $P_{O2}$ =1.5 bar,
C <sub>NaOH</sub> =2 mol/L, load of CuSO <sub>4</sub> =20% w/w on lignin)
Figure 4.3. FT-IR spectra of bamboo lignin samples for purity test
Figure 4.4. Chromatograms of vanillin ions, m/z 151 and 152 from acetovanillone oxidation 39
Figure 4.5. The variation of acetovanillone conversion (a), vanillin yield (b), product loss (PL)
(c), selectivity (d) with reaction time, in comparison to the non-catalyzed (=) and CuSO4
catalyzed (0) processes at conditions; T=170 °C, CAv=0.7 g/L, Po2=1.5 bars, CNaOH=2.0 mol/L
and Catalyst weight=0.01 g
Figure 4.6. GC-MS spectrum of acetosyringone (AS) and its oxidation product; Syringaldehyde
(Sy) (0.05 g of AS, 0.01 g CuSO <sub>4</sub> , 2 mol/L NaOH solution, reaction temperature 170 °C,
reaction time 5 min, oxygen partial pressure 1.5 bars)
Figure 4.7. The variation of vanillin yield, product loss, selectivity of vanillin and conversion of
AV with AV concentration at conditions; T=170 °C, time=5 min, Po2=1.5 bars, CNaOH=2.0 mol/L
and Catalyst weight=0.01 g
Figure 4.8. Effects of alkaline concentration (a) and O <sub>2</sub> partial pressure (b) on the vanillin yield
and unreacted acetovanillone (U-AV)
Figure 4.9. Plausible reaction mechanism of acetovanillone oxidation to vanillin
Figure 4.10. XRD patterns of the prepared Cu/Al <sub>2</sub> O <sub>3</sub> catalysts with different loading
Figure 4.11. The SEM-EDS (a-b), SEM (c) characteristics of the prepared alumina supported
copper catalyst 5% Cu/Al <sub>2</sub> O <sub>3</sub>
Figure 4.12. The catalytic activity of copper salt and alumina supported copper with different
loading as compared to controlled experiment at 170 °C for 5 min
Figure 4.13. Concentration-time profiles of AV degradation (a) and V formation (b) during the
CWAO process at various temperatures; 120-150 °C

## List of Figures

Figure 4.14. Plots of first order kinetics with respect to AV concentration-plot of eq. (4-3) (a-	b),
and $\ell n \; (C_{V\infty} - C_V)$ against the reaction time at 135 °C and 150 °C (c)	. 53
Figure 4.15. Effect of oxygen concentration on the CWAO on AV degradation at 423K,	
5.72×10 <sup>-4</sup> mol/L Cu(ll), 2.1×10 <sup>-2</sup> mol/L AV at pH>13	. 57
Figure 4.16. Effect of oxygen concentration on the formation of vanillin and effect on the	
apparent rate constant	. 58

Degree papers are in the "Xiamen University Electronic Theses and

Dissertations Database".

Fulltexts are available in the following ways:

1. If your library is a CALIS member libraries, please log on

http://etd.calis.edu.cn/ and submit requests online, or consult the interlibrary

loan department in your library.

2. For users of non-CALIS member libraries, please mail to etd@xmu.edu.cn

for delivery details.