# EVALUATION OF SPECTROGRAMS OF HIGH SPEED STEELS FOR MINOR ELEMENTS. PLATE CALIBRATION METHOD

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**ABSTRACT.** Minor elements occurring in high speed steels do not exceed a total of two per cent. Such steels of our Works manufacture designated as T.H.S.<sub>2</sub> are of the conventional type 18-4-1. The time-scale method of plate calibration due to Smith has been adopted and applied to the ferrous analysis of the above samples; the spectra evaluated and compared with the usual method of log ratio of galvanometer deflections against composition. The elements receiving attention were manganese, silicon and vanadium. Conventional spark technique for exciting spectra was used. The standard deviations of results which have been computed, do not exceed 2 to 3 per cent of contents.

#### INTRODUCTION

The method of plate calibration for the evaluation of spectrograms can best be carried out in case of alloying elements occurring in small percentages. A method given by Smith (1945) of the British non-ferrous metals Research Association has been used in the analyses of non-ferrous alloys. The method is being adopted and applied to the analysis of ferrous samples such as high speed steels. The elements considered are manganese, silicon and vanadium, the total contents of which do not exceed two per cent in the given alloys.

The basis of method of plate calibration arises out of known relative intensities of manganese triplets 2949.2, 2939.3 and 2933.1. These values were derived from a consideration of "statistical weights" and given by D. M. Smith as 7:5:3. The ratios are independent of the source of excitation.

As applied to ferrous analysis, it was, however, not possible to include Mn 2949.2 owing to its own interference with an iron line having the same length (Fe II 2949.205). In what follows, therefore, two lines, Mn 2939.3 and Mn 2933.1 formed the calibration pair.

#### PROCESS

Technique: The apparatus used in exciting the spectra and in the photometric work and the processing of plates is described underneath

Volts	<u> </u>	230 A. C.
Condenser		0.005 µF
<ul> <li>For a star in the second se Second second sec</li></ul>	<i>.</i>	`15,000 V spark
	, ,	15,000 V spark 22,000 V wave peak.
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Trausformer Added inductance		1/4 KVA tapping from 15,000 V Nil
Spark gap	~	2 mm.
Electrodes	-	Samples, chill-cast, flat surface.
Counter electrode	-	H. S. brand carbon also flat surface,
Pre-sparking	-	3 mins.
Spectrograph	-	E 478 Littrow
Sp. phic slit	-	7 divs (=0.0175 mm)
Photometer slit	*	8 divs (=0.016 mm)

The apparatus are all of Hilger's specification and of Hilger manufacture.

		•
Photographic plates -	-	Ilford Process
Developer	-	ID 13 No. 1 and 2 in equal proportions.
Development –	-	4 mins.

Flat surface technique of spark excitation with a condensed spark of routine type was used.

Reproducibility : In order to test the reproducibility of spectral intensities a continuous spark of 25 secs. duration on a chosen sample against carbon was given with spectra distributed throughout the plate. The data are given in Table I.

	Mn 2939 Deflection	Density D 2939	D Av đ)		Mn 2933 Deflection	Density D 2933	D - Av (d)	
1	67	0.80	-0.01		120	0.54	0.00	1
2	69.5	0.78	-0.03		127.5	0.52	-0.02	
3	66	0.80	-0.01		121	0.54	0.00	
4	65.5	0.81	0.00		120	0.54	0.00	
5 6	70	0.78	-0.03	Σd <sup>2</sup> =.0100	126	0.52	-0.02	Zd <sup>3</sup> =.0047
6	69	0.78	-0.03	Std. deviation	125.5	0.52	-0.02	Std. deviation
7	66	0.80	-0.01	209	120	• 0.54	0,00	
8	63	0.82	0,01	$\sigma = \sqrt{\frac{2d^2}{n}}$	117.5	0.55	0.01	$\sigma = \sqrt{\frac{\mathbf{z}d^2}{n}}$
9	63	0.82	0.01	=0.023	117	0.35	0.01	<b>=0.016</b>
ó	66.5	0.80	-0.01	in %	127	0.52	-0.02	in %
1	65 58.5	0.81	0.00	2.8%	119.5	0.55	0.01	3%
2	58.5	0.86	0.05	,	113	0.57	0.03	370
3	62.5	0.83	0.02		117.5	0.55	0.01	
4	62.5	0.83	0.02		116.5	0.56	0.02	
	62	0.83	0.02		117	0.55	0.01	
5 6	58	0.86	0.05		113.5	0.57	0.03	
	62 58 62	0.83	0.02		116.5	0.56	0,02	
7 8	63	0.82	10.0		122	0,54	0.00	

TABLE I

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The deflections are the mean of two readings obtained by measuring the line each way as the plate is moved from one side to the other. The standard deviations of densities as obtained was 2.8% for Mn 2939.1 and 3% for Mn 2933.1.

### EXPERIMENTAL

A series of three plates A, B and C was used for making the blackening curves. As the distribution of spectra over the plate was of some consequence in sampling, spectra were distributed evenly and were noted. The sequence is shown in Table II

Plate	Alloy used	Exposure	Time of exposure
	,	I-3	Sec. 10
		4•6 7-0	,, 15
А.	(13)	7 <b>-9</b> 10-12	,, 20 ,, 30
	(-5)	13-15	,, 30 ,, 45
		16-18	,, 60
		19,20	<b>,,</b> 90
		1-4	,, 10
		5-8	,, 15
<b>D 0 0</b>	•   • • •	9-12	,, 20
в&С.	(13)	13-16	,, 25
		17-20 21-24	,, 30
		25-28	,, 45 ,, 60
	·		,,
·		1-10	<b>,,</b> <sup>2</sup> 5
		20-29	13 33
D.	(13)	11-13	,, то
		14-16	,, 20
		17-19	,, 30
	(13)	1,2,12,13,23,24	,, 10
		3,4,14,15,25,26	,, 20
		5,6,16,17,27,28	,, 30
E.	(3)	7,18	·· 25
	(4) (7)	8,19 9,20	11 11
	(30)	10,21	19 99
	(42)	11,22	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,
	(6)	1-4	,, 25
	(0)	13-16	
		25-28	1 11 11
	(r3)	5,22	,, 10
		6,23	yy 20
<b>1</b> 7.		7,24	,, 30
	(3)	8,17	,, 25
,	(4) (7)	9,18 10,19	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
	(30)	11,20	)) I) )) ))
	(42)	12,21	11 11

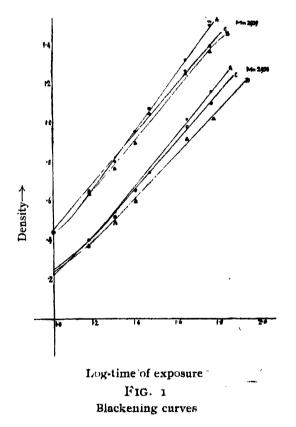
TABLE 11 Distribution of spectra

The table shows the distribution for all the plates used in the experiment. The spectra are numbered serially from top to bottom. Spectra with exposure time of 10, 15 etc. to 60 secs. were photographed. The spark used was continuous with pre sparking as described. The number of spectra of each duration in case of a plate is given in Table II. All such results are included in the table.

Blackening curves were drawn from data given in Table III, where mean values of densities and deflections for manganese lines from the three plates A, B and C are given, the density (Sug. Def. Glas., 1946) being defined as

$$D = \log d_0/d$$

where  $d_0$  is the deflection obtained from clear plate, *i.e.*, base plus unexposed processed emulsion and d, deflection due to the line. Such blackening curves are shown plotted in Fig. 1. The plot is of densities against log time of



exposure. The separation of curves along axes parallel to abscissa are measured giving  $\Delta \epsilon$  (log exposure time ratio) for different values of densities.. This is shown in Table III. The mean of log exposure time ratio ( $\Delta \epsilon$ ) thus measured gives, attributing a value 100 to the photographic intensity of Mn 2939, a value equal to 60.3 for Mn 2933; Baly's value on a consideration of statistical weights being 60.1. No correction for background was made but the agreement afforded was excellent.

Time of	l = 10 (Mn 2)		l=( (Mn	50.3 2933)	D (Density)	Plate A	Plate <b>B</b>	Piate C
exposure	Deflection	Density	Deflection	Density		<u>н</u>	2.58	<u>д</u> 2.25
	A 153	0.44			0.6			_
10″	B 152.5	0.44	238 437	0.23 0.25	1	2.40	2,50	2.20
10	C 144.4 Mean 150	0.44 0.44	222 232.3	0.26 0.25	0.7	2.25	2.40	2.25
	Log It	3.00			0.8	2.20	2.35	2.20
		3.00		2.78	0.9	2.10	2.35	2.20
	A 89	0.65	159	0.40	1.0	2.10	2.30	2.20
	B 79.5	0.63	177	0.37	1.1	3.15		{
15‴	C 90.9 Mean 86.46	<b>ს.64</b> ი.64	157 164.3	0.40 0.39		2.15	, }	2.15
	Log It	3.176			1.2	2.18		
م دند - و و است. بعرب		3.270		2.96	∆€	0.22	0.24	0.22
	A 62.7	0 <b>.8</b> 0	116.2	0.54	Меа	n ∆€	0.1	14
	B 72.4	0.76	137	0.48	Ant		0,	
20″	C 67.0 Mean 67.3	0.75	122 125.1	0.51 0.51	Pho	tograni	aic Intens	sitv
			123.1					
	LogIt	3.301		3.081		Mn : Mn :		00 0.3
	_		_		Dat	a accor	ding to B	Balv
	B 51.2	0.92	103.2	0.61		Mn	2939 1	00
25″	C 44.2	0.95	86	0.65		Mn	2933 6	0.1
	Mean 47.7	0 <b>.9</b> 4	94.6	0.63				
	Log It	3.398		3.178				
	A 36	1.04	73	0.74				
	B 35.5	1.07	74.5	0.75				
30″	C 34.1 Mean 35.2	1.07 1.06	66	<b>ა.</b> 78 ი.76				
			71.2					
	Log It	3.477		3.257		-		
	<b>61 A</b>	1.32	39	1.01				•
	B 24	1.25	51	0.91				
45″	C 22 Mean 21.7	1.26 1.28	42.4	0 <b>.97</b> 0.96				
				-				
	Log It	3.653		3-433				
	A 12.5	1.50	28	1.15	1	•		
	B 18.2	1,36	41.2	1.01				
60 <b>"</b>	C 16.0 Mean 15.6	1.39	32.5 33.9	1.09 1.08				
		1.42	3.3.9					
	Log It	3.778		3.558	4			

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Clear plate Av. 406 mm.

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In all subsequent experiments a calibration pattern as described above was included in the plate. Blackening curves were drawn in each case. Data for D, E and F plates are given in Table IV-V1.

Time of	Mn 2	939	Mu :	293,3	Densiter	
exposure	Deflection	Density	Deflection	Density	- Density	Plate D
	128	0.42	193	0.24	0.5	. 2.20
-	124	0.44	192	0.25		1
	114	0.47	186.5	0.26	0.6	2.45
Mean	123	0.44	190.5	0.25	0.7	2.70
20″	53	0 54	98.5	0.54		!
	53	0.53	99.5	0.53	Mean	2.45
	• 54	0.52	. 102	0.52	ƥ Antilog	0.245
Mean	53.3	0.53	100	0.53	-	
						phic In
30″	32 36	1.03	64.5	0 72	tensity	1
		0 <b>.97</b>	72	0.67	1 '	
ì	34.5	0.99	70	0.68	Mn 2939	
Mean	34.2	1.00	68.8	0.6 <b>9</b>	Mn2933	56.9

TABLE IV

Clear plate Av. 338 mm

Time of	Mn 2	939	M11 2	933	T)	
cxposure	Deflection	Density	Deflection	Density	Density	Plate F
10"	177.5 18 <b>3.</b> 0	0.35 0.34	251 261.5	0.20 0.19	0.4	2.55
	171.5	0.37	251.5	0.20	0.5	2.32
Mean	177.3	0.35	255.0	<u>0</u> .20	<b>U.</b> 6	2.15
20″	117.5	0.53	190	0.32	0.7	2.20
	91.5	0 <b>.64</b>	159.7	0.4 <b>0</b>	Mean	2.28
Mean	81.2 96.7	0.69 0.62	149 166.2	0.43 0.38	∆¢ Antilog	<b>0.2</b> 28 0.772
30″	57.7	0.84	110.2	0.56	Photogra	phic In-
	59.0	0.83 0.88	112.7	0.55	tensity	
Mean	52.5 56.4	0.85	103.2	0.59	Mn 2939	
	30.4	0.05	108.7	0.57	Mn 2933	59.2

TABLE V

Clear plate Av. 400 mm.

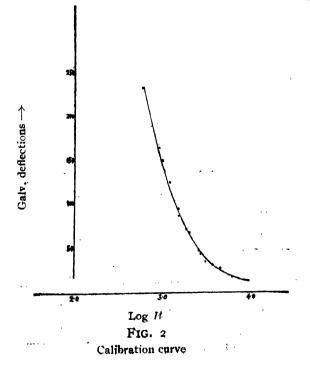
	Mn 2	939	Mn 3	939		·· · · ·
Time of Exposure	Deflection	Density	Deflection	Density	Density	Plate F
-	(Me	a11)	(Mea	1n)		
10" 20" 30"	172.8 81.7 48	0.36 0.69 0.92	258.7 152.7 100.2	0.44 0.42 0.60	0.4 0.5 0.6 0.7 0.8	2.40 2.13 2.00 2.00 2.03
-					Mean ∆€ Antilog Photogra	2.11 0.211 0.789

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Clear plate Av. 400 mm.

Giving the photographic intensity a value 100 for Mn 2939, the values were 56.9, 59.2 and 61.5 respectively for Mn 2933 for plates D, E and F.

Having constructed the blackening curves for plates A, B and C (Fig. 1) and obtained the mean  $\Delta e$ , a second plot was made with log *It* as one axis and deflection the other. This shows little scatter. The data are taken out of Table III. The calibration curve so obtained is shown in Fig. 2.



We now come to a stage of drawing intensity-calibration curves. As in all cases of high speed steel analysis exposures having been standardised at a fixed value of 25 secs., a parameter log  $I_{25}$  derived from log *It*-log 25 was used instead of log *It* in plotting intensity-calibration curves. Such a curve is shown plotted in Fig. 3, for plate D. The data are from Table VII.

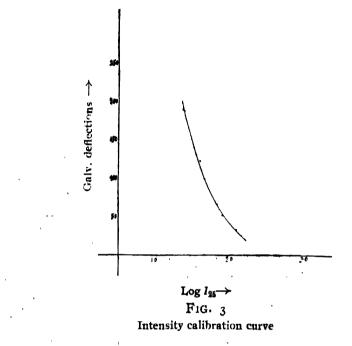


TABLE	VII
IANLE	

	Mn	2939	M11 2933		
Time of exposure	Deflections	Log 125 (log It-log 25)	Deflections	Log I <sub>25</sub> tlog It-log 25	
Plate D. 10" 20" 30"	123 53·3 34.2	1.602. 1.903 2.079	190.5 100 68.8	1.357 1.658 1.834	
······································	CI	ear plate Av. 338 mm	I		
Plate R. 10" 20" 30"	177-3 96.7 56.4	1.602 1.903 2.079	255 166.2 , 108.7	1.374 1.675 1.851	
	• CI	ear plate Av. 400 mm	· .		
Plate F. 10" 20" 30"	172.8 81.7 48.0	1.602 1.903 2.079	258.7 152.7 100.2	1.391 1.692 1.868	
,	C	ear plate Av. 400 mm			

Verification of results.-

Suggestion arising out of Smith's paper for verification of variation with time of the intensity of emission was tried. No systematic variation was evident and no generalisation was possible as with Smiths. The result for plate D is tabulated in Table VIII.

TABLE	VIII
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Exp. No.	Mn 2939		Mn 2933		Exp. No.	Mn 2930		Mn 2933	
	log I <sub>25</sub>	I <sub>25</sub>	log 1 <sub>25</sub>	1 <sub>25</sub>		log 1 <sub>25</sub>	I 25	log I <sub>25</sub>	l 25
1 2 3 4 5 6 7 8 9	1.93 1.90 1.92 1.99 1.96 2.01 1.99 1.99 1.99 1.98	85.1 70.4 83.2 97.7 91.2 102.3 97.7 97.7 91.2 95.5	1.68 1.63 1.66 1.72 1.71 1.75 1.71 1.71 1.71	47.9 42.7 52.5 51.3 56.2 51.3 51.3 51.3 50.1 51.3	20 21 22 23 24 25 26 27 28 29	2.00 1.98 1.99 1.98 2.02 1.98 1.93 1.93 1.94 1.93	100,0 95.5 97.7 95.5 104.7 95.5 85.1 85.1 87.1 85.1	1.73 1.72 1.72 1.72 1.74 1.70 1.65 1.63 1.66 1.66	53.7 52.5 52.5 52.5 55.0 50.1 44.7 42.7 45.7 43.7

Plate	D
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#### Av. on 100% basis.

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100:53.6

In a following table is shown the results of chemical analysis and the uses to which the alloys are put. Useful line pairs (Harrison, 1939) for purposes of calibration and for analysis and internal standards are shown in Tables IX and X.

#### TABLE IX

Chemical analysis

Sample No.	Use	-	Mı	n Si	v	Cr	W
(3)	Standard .		0.	22 0.17	1.17	4.27	18.64
(4)	Standard .	'-	- v.	20 0.22	1.22	4.42	1 <b>8</b> .84
(6)	Test alloy		<b>o.</b>	0.16	1.22	4-35	19.05
(7)	Standard			.20 0.14	1.26	4.40	19.83
(13)	Calibration		., 0.	24 0.20	1.44	4.70	20.79
(30)	Standard	·•• ;	q.	30 0.16	1.33	4.37	17.98
(42)	Standard		q.	.25 0.23	1.40	3,60	19.39

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## TABLE X

#### Line pairs

Calibration	Mn 2939.30	Mn 2933.06
Analysis	Mn 2433.06 Si I 2881 58 VII 3062.70	
Internal Standard	Fe I 2936.90 Fe II 2885.93 Fe II 3062.23	

#### Calibration curves.-

Two series of plates E and F are taken to include calibration patterns distributed evenly over the plate as also exposures from standards. Plate E includes exposures from a test alloy. Results from plates E and F are shown in Tables XI and XII. The results include the usual log R, ratio.

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TABLE >	٢	L
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#### Plate E

Sample No.	Mn 2933 deflec- tion	Fe 2937 Deflec- tion	$R = \frac{\mathrm{Mn}}{\mathrm{Fe}}$	log R	log / <sub>Mn</sub>	log I <sub>Fe</sub>	iog ratio of Ints. (+)
(3)	115.7	39.5	2.92	0.465	1.81	2.15	0.34
(4)	115.0	36.0	3.19	0.504	1 82	2.18	ი <b>.36</b>
(7)	115.0	35 5	3.24	0.511	1.82	2.18	0.36
(30)	80,0	37.0	2.16	0.335	1.9 <b>6</b>	2 17	0,21
(42)	83.0	37.7	2.20	0.342	1.95	2.16	0.21

Mn Chart

Sample No	V3063 D <b>eflec</b> - tion	Fe 3062 Deflec- tion	$R = \frac{V}{Fe}$	log R	$\log I_{ m V}$	Tog I Fe	log ratio of Ints, (+)
(3)	66.7	35.5	1 88	0 27	2 03	2.18	0 15
(4)	49.2	29.7 <sup>°</sup>	1.66	0.22	2,11	2.21	0.10
(7)	42.7	27.7	I 54	o <b>18</b>	2.14	2.22	0.0 <b>8</b>
(30)	48.7	28 2	1.73	0.24	2.11	2,21	0.10
(42)	42.5	29	1.46	0.16	2.14	2.21	0.07

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Sample No.	Si 2881 Deflee- tion	Fe 2886 Deflec- tion	R = Si Fe	log R	log I <sub>Si</sub>	log I <sub>Fe</sub>	log ratio of inlensities (+)
(3)	187	150	1.25	0.097	1.56	1.68	0.12
(4)	158.5	135.5	1.19	0.075	1.65	1.74	QULO
(7)	162.2	132.5	1.22	0 <b>.08</b> 5	1.64	1.75	0,11
(30)	142.7	136.5	1.04	0.017	1.71	1.73	0.02
(42)	131 5	133.7	0.98	0.01	1.75	1.74	-0.01

TABLE X1 (contd.)

# TABLE XII

### Plate F

# Mn Chart

Si Chart

Sample No.	Mn 2933 Deflec- tion	Fe 2937 Deflec- tion	$R = \frac{\mathrm{Mn}}{\mathrm{Fe}}$	log R	log I <sub>Mn</sub>	log I <sub>Fe</sub>	log ratio of intensities (+)
(3)	109.2	42.5	2.57	0.41	1.82	2 11	0.29
(4)	112.5	36.75	3.07	0.49	1.80	2.14	0.34
(7)	123.2	41 5	2 97	0.47	1.77	2 11	0 <b>.34</b>
(30)	85 7	46 2	1 85	0 27	1.91	2 00	81 O
(42)	79.2	42.5	1,86	0.27	1.94	2.10	0 16

# V Chart

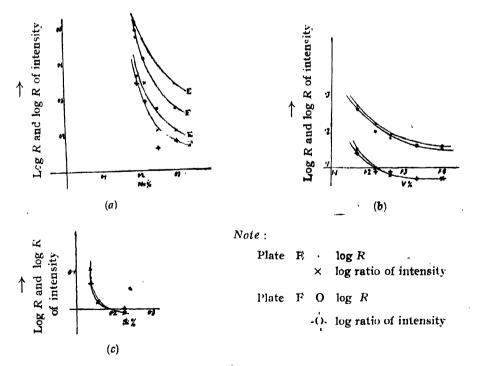
Sample No	V 3063 Deflec- tion	Fe 3062 Deflec- tion	$R = \frac{V}{Fe}$	log R	log I <sub>V</sub>	log I <sub>Fe</sub>	log ratio of intensities (+)
(3)	62	34.2	1 81	0.25	2.01	2.15	0.14
(4)	50.2	31.5	1.595	0.20	2.08	2.17	0.09
(7)	55.2	35 7	1.55	0.19	2.05	2.14	0.0 <b>9</b>
(30)	49.2	34 0	1.45	0,16	2,08	2.15	0.07
(42)	37-5	26.7	1.405	0.15	2.13	2.20	0.07

TABLE XII (contd.)

Si Chart

Sample No.	Si 2881 Deflec- tion	Fe 2886 Deflec- tion	$R = \frac{\text{Si}}{\text{Fe}}$	log R	log I <sub>Si</sub>	log I <sub>Fe</sub>	log Ratio of intensities (+)
(3)	169.5	130.5	1.30	0.11	1 63	1 74	0,11
(4)	145 0	120.7	1.20	0.08	1.69	1.78	0.09
(7)	152.7	130	1.17	0.07	1 67	1.74	0.0 <b>7</b>
(30)	132.2	136.2	0.971	0.01	1.73	1.72	0.01
(42)	123 ()	122.5	1.003	0.00	1.77	1.77	0.00

deflection due to analysis line to that due to internal standard and  $\log I_{25}$  values and their differences derived from appropriate calibration curves E and F (not shown) for elements manganese, silicon and vanadium. Calibration curves were plotted with log R and log ratio of intensities against composition. Such curves for elements manganese, silicon and vanadium are shown plotted in Fig. 4. Discussion of results follows. Plate F includes a test alloy. From the appropriate intensity calibration curve (F), compositions of Mn, Si, and V were derived. Values are tabulated in Tables X111, X1V and XV)



F1G. 4

$ \begin{array}{c c} b & Sq. of \\ b & Diffs. \\ (d) & \times_{10}^{-6} \end{array} $	ot f	03	۱ 	02 - 4		i  Q	1					υ4   1γ	<b>Z</b> 112
Diff. Av,	-0.004	- 0.002	0.000	1 0.00	- 0.002	000'0	0.000	0.000	0°00		- 0.004	400°U	
% Cal- culated (b)	<b>ν, 19</b> δ	<u>0.200</u>	0.202	0.200	0.200	0°-202	0.202	0.202	0.206	0.208	961.0	0,206 1	¥2015 0 1000
log ratio of Ints.	0.36	0.35	0.31	0.35	0.35	0.34	0.34	0.34	0.33	0.32	· 0.36	0.33	2
Mn	I.75	1.77	1.75	1.78	1.75	1.75	6ź-1	1.81	1.79	1.76	1.75	I.73	
<sup>I</sup> Fe I	2.11	2.12	5.00	2.13	2.10	2.09	2.13	2.15	2.12	2.08	2.11		
% (Cal- culated)	0.200	0.200	0.210	701.0	0.205	0.208	0.202	761.0	202.0	0.214	0.200	0.218	<b>2</b> 2.456
log R	0.49	0.48	0.44	0.50	0.46	0.45	14.0	0.50	0.46	0.42	0.48	0.41	
$R = \frac{Mn}{Fe}I$	3. IO	3-05	5.76	3.10	2.86	2.81	3.98	3.20	2.87	3.60	3.05	2-57	
Mn 2933 Deflection	178.5	122	128.5	120	128.5	128	114.5	112	115	тzб	126-5	135	
Fe I 2937 Deflection	41.5	04	46.5	Se	<u>.</u>	45.5	38.5	35		 	41.5	52.5	
				) <del>4</del>	13	I4	15	i vi	25	<b>3</b> 6	42	. 8g	-

TABLE XIII

Readings for Sample (6)

,	•	, 1	. \	<b>B</b> .	N.	Bi	had	uri				8.,	تي ١	
	Sq of Diffs. × 10 <sup>-6</sup>	I	1	I	I	-	I	I	I	Г	I	6	I	<b>N</b> 10
	Diff. b- Av, (d)	100.0-	100.0-	-0.001	100.0-	-0.001	100.0	100.0-	100.01	100.0 -	0.001	0.003	0.001	
	% (Cal- culated) b	0-140	0,140	0.140	0.140	0.140	0.140	0.140	0,140	0.140	- 2,142	0.144	0.142	<b>z</b> 1.688 M <b>ca</b> n o.141
	log ratio of Ints.	20°D	0.08	0.10	0.0	0.08	9.08	0.10	0.10	0.07	0.05	0.04	0.05	Ň
	<sup>I</sup> Si I	1.68	1.70	39. I	49.1	1.66	1.68	1.68	07.1	1.66	1.64	1.62	1.60	
	<sup>I</sup> Fe II	I.75	1.78	1.78	1.76	I.74	1.76	1.78	1.80	1.73	1.6g	у <b>.</b> г	1.65	.1660 .138 < 1% of content.
	% (Cal- culated)	0.139	0.137	0.136	0.137	0.139	0.137	981.0	0.135	0 <b>†1</b> .0	0.141	0.142	0.141	<b>2</b> 1.660 Mean 0.138 $\frac{ \mathbf{x}n ^3}{n} < 1\% \text{ of }$
	log R	0.08	0.0 <b>0</b>	0.10	0.0 <u>0</u>	0.08	0.09	0.10	0.11	0.07	90.0	0.05	0 <sup>.0</sup>	
	$R = \frac{Si}{Pe} \frac{\Gamma}{\Gamma}$	61.1	1.22	1.255	1.23	1.20	1.22	1.26	1.275	1.18	I.14	11.1	I.14	Standard deviation
	Si I 2881 (Deflection)	ISO	144.5	149-5	154	157-5	150.5	149	143.5	159	168	173.5	184	Ø
	Fe II 3886 Deflection	126	811	611	125	131	123.5	118.5	§.211	134.5	148	ISÓ.Ş	161.5 J	
	Kap. No	<b>⊢</b> 4	<b>61</b>	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	4	13	14	15	IŚ	28	36	52	28	

`, \*

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TABLE XIV Readings for Sample (6' Si (0.16)

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h **1**.... \*

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			_ <b>/</b>		051	4171	8 U		ugn	1.5	bee	d 2	tee		etc. I	361
		Sq. of Diffs. X10 <sup>-6</sup>	-1 <u>9</u> 6	625	961	961	0	961	3969	6	ıq6	- 6	961	96 *	<b>x</b> 4993	,
Readings for Sample (6) V (1.22)		Diff. b- Av. (d)	-0.0T4	0.025	-0.014	+10.0 <del>-</del>	0.003	-0.014	0.063	0.003	-0.014	0.003	-0.014	+10'0-		
		log ratio % Cal of Ints culated (b)	1.203	1.242	1.203	1.203	1.220	1.203	1.280	1.220	1.203	1.220	1.203	I.203	Z 14.603 Mean 1.217	
			0.11	<b>60</b> 0	0.11	0.11	0.10	0.11	0.08	0.10	0.11	0.10	0.11	0.11	We	
		I Fe II	2.07	2-14	2.18	2.18	2.17	2.15	3.16	2.19	2.17	2.16	2.15	2.14		
	.22)	IIA /	96.1	2.05	2.07	2.07	2.07	2.04	2.08	7.09	2.06	2.06	2.04	2.03		1.5% of content.
	I)	% Cal. culated	I.27!	1.271	I.200	061.1	I.220	1.210	1.271	1.190	I.200	I.238	I.200	1.254	Z 14.715 Mean 1.226	11
		log R	81.0	0.18	0.23	9.2 <b>4</b>	0.21	0.32	0.18	0.24	0.23	0.20	0.23	0.19	Ŵ	rd deviation $r = \sqrt{\frac{xd^3}{n}}$
		$R = \frac{VII}{\text{Fe II}}$	1-50	1.53	1.70	1.72	1.64	1.65	1.515	1-725	1.70	1.59	02.1	1.56		Standard devi
		Fe II 3062 Deflection	50 C	36	30	29	30.5	34.5	32	37	30.5	53	34	36		0.
		VII 3063 (Deflection)	75		£1	50	50	57	48.5	46.5	52	52.5	<b>3</b> 8	59		
		Rxp. No.	щ	п	e	*	£1 .	14	15	16	3E	<b>9</b> 2	11	8र		

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TABLE XV

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The standard deviations of 12 results are :

1.5 for Mn < 1.0 for Si 1.5 for V.

The accuracy obtained is comparable to usual  $\log R$  method.

#### DISCUSSION OF RESULTS.

In the calibration curves given by the two methods (Fig. 4) accuracy obtained was nearly constant. Calibration did not usually produce a single curve applicable to the two plates. This is the more remarkable in the case of manganese. For silicon, however, a single curve suffices. For vanadium two curves are given form sets of closely lying parallel curves.

On the application of background correction it can be said that no standard critique of spectral background evaluation exists; the emphasis is laid on having a clean plate with little background. Background calculation adds materially to time, a fact which militates against ordinary routine business. No attempt was thus made to calculate this factor. These facts point to a conclusion that plate calibration does not usually compensate for responses of plates. Nonetheless, the method described is simple and can be of general use. Further work might also give better corroborative results.

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