Effects of a Nano-structured Surface Layer on Titanium Implants for Osteoblast Proliferation Activity

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Abstract: The goal of this study is to compare surface morphologies on dental implants created by a range of five different surface-modification technologies and, in addition, cell assays to assess the subsequent cell proliferation on each treated surface. In our research, we surface-treated 5 mm-diameter – 2 mm-thick discs, machined from Grade 1 titanium. We treated the surfaces of the discs with chemical etching, electro-polishing, Al₂O₃ sandblasting, and surface melting with 1 Joule or 3 Joule impulse-energy laser beam. We carried out quantitative as well as qualitative analyses with stereo and scanning electron microscopes (SEM), confocal and atomic force microscopes (AFM) and goniometer. We examined each surface with cell-testing, as a measure of osseointegration. In tests with fibroblasts, the highest cell proliferation occurred on the Al₂O₃-roughened surfaces. In the case of osteoblasts, we measured the greatest cell activity on the laser-melted samples with different energy levels.

Keywords: surface modification; surface analyses; surface morphology; cell proliferation

1 Introduction

Currently the examination and deliberate structuring of surface morphologies of dental implants, designing new processes for surface manipulation and the assessment of those results, are all at the cutting edge of implantology. Surface treatment of implants influences, to a major extent, the success of integration with bone [1], making these processes crucial to the functional value of implants. To ensure this integration, a suitable morphology must be created on the surface of a given implant [2].

1.1 Conditions for Osseointegration

Thirty years ago researchers considered how surface engineering, topography, and the morphology or lack of contamination of implant surface, might affect osseointegration. Since that time, investigators have made substantial progress in making issues of surface shape and contamination central to dental implantology research. In the early 1970s, development has sought to: understand the interaction between the bone and the implant surface, explain the process of osseointegration from the moment of implantation until the phase when secondary stability is attained, and define both the duration of, and the biological mechanisms involved in, each phase [3, 4, 5, 6].

Exact phases of the development of the interface zone are still unknown. The interface layer ensures a connection between the oxide layer on the implant surface and bone proteins. The layer around the osseointegrated implant has a thickness of 2 to 5 nm [7, 8]. By the beginning of the 1970s, Hulbert et al. had already established that porous surfaces enable bone regeneration to occur more quickly and allow osseous tissue to grow [9]. The consensus since the year 2000 has been that rough surfaces are better for osseointegration than smooth or polished surfaces [10-15].

Molecular biology has gradually gained significance in surface treatment. Researchers are developing increasingly sophisticated surface-engineering techniques that take into account biochemical signals and cues involved in how various materials bond with bone tissue. The possibility of applying organic materials (chemical modifications) and several proteins (BMP) on an implants surface, creates new areas of investigation [16]. The interaction of proteins at the nanometric level is emerging as a major decider in the integration of implants.

Up until the end of the 1990s, investigations mostly addressed surface topography and morphology. Since then, attention has increasingly focused on surfacechemistry research. Surfaces are modified with physical-chemical methods to promote faster and more comprehensive osseointegration. High surface energy optimally promotes interaction with the biological environment [17-19]. It is accepted that surface roughness influences alkaline phosphatase (ALP) and osteocalcin (OC) levels, thus indicating the osteoblast activity of the surrounding tissues [20, 21]. On a smooth surface, "osteoblast-like" cells show a sparse and flat distribution, while greater numbers and more dense distributions are found on rough surfaces [22, 23]. Rough surfaces have an impact on cytoskeleton functions as well [24]. Micro-scale surface structures influence cellular functions while nano-scale elements have an effect on sophisticated cascades involved in protein binding (Figure 1) [25-27].



Figure 1

The effect of surfaces with various degrees of structure on the binding of organic elements

According to the current status of the science for osseointegration, it is not a function of quantitative porosity but qualitative porosity [28]. In the case of dental implants the most commonly used and at least accepted parameters are the following: the average height differences of elements raised above the surface, usually represented as "Sa" (in the case of a two-dimensional examination the same value is signified by "Ra"), effectively the average of the wave-lengths (distances), which is named "Scx", and finally the sum of treated and untreated surfaces, which is called Sdr and is described as a "hybrid" value. According to Wennerber and colleagues as well as most of the research community, what can be described as ideal surface parameters are $Sa = 1.4 \mu m$, $Scx = 11.6 \mu m$, $Sdr = 1.5 \ \mu m$ [29-33]. On the basis of the literature a surface can be viewed as rough when the hybrid value is $>2 \mu m$. It is medium rough if that given value falls between 1 μ m and 2 μ m, and smooth, or at least mildly rough if <1 μ m [34]. Wennerberg and colleagues showed with animal experiments the 1.5 µm hybrid value is the most favorable and associated with cases of the greatest osseointegration [35]. Lazzara and colleagues showed this and Ivanov was doing human experiments supported with results involving human volunteer subjects [36, 37].

It's possible to characterize the quality of a surface with bone-implant contact (BIC) values, which show what percentage of an implant surface comes into direct contact with bone. After decalcification, with a histology-morphology examination of a tissue slice it is possible to tabulate the percentage ratio of bone-implant connection (BIC) [38]. For sufficient bone integration the lower jaw BIC average is 40.7%, while in the case of the upper jaw this average is only 37.2% [39].

The percentage ratio of bone-implant contact is not easy to calculate. Results recorded as obtained in various journals in the literature vary widely across a very broad scale. They depend on the species of experimental animal, the type of bone the implant was attached to (femur, tibia, jaw), the healing time, and the surface treatment applied to the implant.

Hansson devised a mathematical model describing interdependences between the measure of osseointegration and the surface roughness of an implant. Surfaces of commercially obtainable implants serve as the basis of the model. He established that hollows of depth 1.5 μ m and diameter of 3 to 5 μ m in the implant surface are advantageous for integration with bone [40].

1.2 Surface-Modification Methods

Current machining methods give dental implants unique geometries. Leading manufacturers most often create implants with diameters from 1 to 4.2 mm and lengths from 2 to 14 mm [41]. The geometric structure and surface-preparation of implants plays a central role in primary and secondary stabilization [42]. The most common methods for modifying surface morphology are chemical etching [43-45], sand-blasting [46], these two in combination [47], grit blasting [48], surface melting by laser [49], and anodic oxidation [50, 51].

A freshly machined surface is often used as a reference in experiments for mutual comparison. Literature discussing machined surfaces describes massive bone formation around the implants, which results in a stable connection between implant and bone tissue [52]. Anomalies on the machined surface go to a depth of 5 μ m in profile. On the surface at a spacing of 5-8 μ m, slightly unevenly arranged but roughly parallel grooves can be observed. The separation and depth of the grooves was equal. With the help of a profileometer the three most characteristic pieces of data were Sa = 0.836 μ m, Scx = 8.38 μ m and Sdr = 1.3 μ m [35]. In the literature there are reports of quite wide intervals (between grooves) on the machined surface with Ra, in the range 0.08 – 4.7 μ m [53].

The BIC ranges between 39 and 47% according to Ericsson and colleagues, as opposed to around 50% according to Wennerbeg et al. [54, 55]. In animal-experiment studies the value has been observed at 62% after 3 months [56]. One drawback is that in several places the surface is contaminated by noticeable machining fragments.

On the basis of the profilometer studies by Wennerberg et al (Sa = $1.9 \mu m$; Scx = $12.3 \mu m$; Sdr = $1.42 \mu m$) the parameters of the acid-treated surface closely approach the supposedly ideal values [35]. The results of animal studies support this, showing a BIC value of 88% after three months of healing [56]. According to Baker et al. animal experiments in the case of acid-treated surfaces bone adhesion starts earlier than with other treated surfaces [57]. Human trials carried out by Trisi and colleagues showed that in case of acid-treated surfaces the BIC

percentage value resembles that of freshly-machined surfaces [58]. They regard the payoff from this method to be the chance to load the implant early on. The other big benefit in each acid-treating process is a high level of surface cleanliness, in that the acid removes the outermost layer of the surface.

In the case of sand-blasting the surface can be blasted with particles of different sizes (25, 75, 250 μ m). With animal studies Wennerberg and colleagues showed that smaller particles resulted better osseointegration than larger, and suggested using 75 μ m sized granules. According to data obtained with the help of an optical profilometer, granules of the size of 25 μ m produced a surface characterized by Sa values = 1.13 μ m; Scx = 9.78 μ m; and Sdr = 1.39 μ m, while the surface produced by 75 μ m particle blasting has the following values Sa = 1.38 μ m; Scx = 11.62 μ m; Sdr = 1.47 μ m (although they approach the ideal parameters very closely). In the case of the 250 μ m particles the values were Sa = 2.15 μ m; Scx = 13.54 μ m; Sdr = 1.79 μ m. The BIC was 62/62% (the mandible/maxilla ratio). During animal-experiment studies this value rose after 3 months to 71% [14, 35, 56, 59]. The disadvantage of the process is that the material used for the sand-blasting can contaminate the surface, and because of this the chemical features of the implant can be unacceptable.

Buser et al examined surface treatments of the titanium used as dental-implant base material. The samples were treated with chemical etching, electro-polishing, or coated with hydroxyapatite. They established that surfaces treated with chemical etching and hydoxyapatite underwent more bone-integration than the samples with electro-polished surfaces [60].

The development of nanotechnology, analyzing the possibilities of implants with nanostructured surfaces, is at the center of numerous current research-and-development projects. Christenson and colleagues looked at organized nanostructures falling into the 1 to 100 nm size range. The "nano" expression can cover crystal structures of material, extracted or built out from the surface layer [61].

Carlos et al completed studies on surface treatments to dental implants made of Grade 4 titanium. They changed the surface morphology with sand-blasting, chemical etching, and anodic oxidation. They prepared SEM images of the surface and then established the level of surface roughness with confocal microscopy. They tested each surface for wettability with contact angle measurements. The surface-modified implants were inserted into live rat tibias and then after 12 weeks removed. They established that chemical etching created a homogeneous surface. Surfaces treated with anodic oxidation measured the lowest while at the same time those implants had the highest screw-out force values [62].

Luiz's research group demonstrated the surface modification and testing possibilities of titanium as a raw material for dental implants. In their work they compared surfaces modified by mechanical polishing, electro-polishing, nanohydroxyapatite, and sprinkling with titanium dioxide and fluoride granules. With an atomic force microscope, a SEM and an X-ray photoelectron spectroscope the surface structures of the samples were tested. Experiments carried out on rats (time period: 4 weeks) measured the differently surface-treated implants with pullout tests. It was observed that implants coated with hydroxyapatite and modified with fluoride had better osseointegration than polished-surface implants. They established that in the several weeks after insertion the nanostructure assisted osseointegration [63].

In the research work of Vinzenz usable surface-treatment techniques for titanium implants are demonstrated. Furthermore he discusses several coating techniques, giving special attention to anodic plasma-chemical treatment of surfaces. Using scanning electron microscopy, X-ray spectroscopy, Raman spectroscopy, and laser-surface-roughness measurements he examined surfaces created by coating. With the results of cell and animal-experimentation examinations he showed that from the point of view of osseointegration the anodic plasma-chemical treated surface performed better than when untreated [64].

Research by Göransson introduces several surface-treatment and examination possibilities for dental implants made of titanium. In the experimental work samples with surfaces modified in 12 different ways were examined. He tested measures of osseointegration using cell experiments, animal experiment models, and human experiments. His result was that the surface structure did not significantly influence osseointegration and further that bioactive coatings also failed to produce significant results aiding improved osseointegration [65].

1.3 Methods of Examining the Surface

1.3.1 In Vitro Methods

It's usual to categorize methods of examination on the basis of which properties of surface morphology are tested by which method. This creates the primary division into chemical methods of analysis (X-ray Photoelectron Spectroscopy, Auger Electron Spectroscopy, Secondary Ion Mass Spectroscopy) and physical methods of analysis (Atomic Force Microscopy, Scanning Electron Microscopy).

1.3.2 In Vivo Methods

An often-used method for quantitative evaluation of the bone-implant contact's load-bearing capacity and the effects of the implant's geometrical characteristics (disregarding some unusual biological and chemical consequences of bone integration) is the pull-out test [66, 67]. After the appropriate healing time has elapsed (in the literature 6 weeks, or else 3 or 6 months) with the help of a special instrument the already bone-embedded implant is removed from the bone and the torque necessary to do this is recorded [68, 69]. Values shown by the instrument can be checked against an already-derived table of standard results to obtain a Ncm value. Removal torque is a very widely used method for animal studies.

In addition to studying implant morphology and with the above progress in mind, we decided to investigate how the nanostructure surface influences the proliferation of osteoblasts. Micro- and nano-morphological analyses revealed some correlations, including the potential biological, biochemical and physiological effects of surface roughness on bone cells adjacent to the implants. In these experiments titanium Grade 1 material was used.

2 Materials and Methods

For these experiments titanium Grade 1 (ISO5832 Pt.2 Grade I.) supplied thanks to the PROTETIM[®] company was made use of, machined discs of diameter 5 mm and thickness 2 mm. The surfaces of the discs were first cleaned with ultrasonic equipment in pure ethanol. The chemical modification of the surfaces of the discs chemical treatment involved passivation in sodium-hydroxide + H₂O₂, oxalic acid, nitric acid solution, electropolishing, sandblasting was performed with Al₂O₃ particles of 250 μ m, along with laser modification (Pulsed Nd/Yag laser (Kvant 1) with a 1 J/pulse as the low-pulse energy laser, and the 3 J/pulse was used for surface treating as high-pulse energy laser). In these experiments, the surfacetreated samples were compared with the machined discs as reference standards.

The machined surfaces of the discs were examined and images taken before their various treatments, first with a stereomicroscope (type: Olympus SZX 16), then with a scanning electron microscope (type: Philips XL 30). The discs' surface roughness values were compared quantitatively and qualitatively using a confocal microscope (type: Alicona Infinite Focus). Microscopic measures of the discs' nanostructures were taken (type: Veeco® diInnovaTM). Comparisons were carried out to measure contact angles and therefore hydrophilic or hydrophobic characteristics of the surfaces of the discs (equipment type: Rame-hart). The measurements were taken at 23°C temperature and 50% humidity. The device released 5-5 droplets (5 μ l/droplet). The tests all used distilled water. Following the drop test (by 5 seconds) images were taken and the contact angles were measured using image-analysis software.

2.1 Experiments Done with Cells

Before the cell test the discs were placed for 20 minutes into tripsin solution, and for 20 minutes in alcohol, then sterilized in an autoclave. In twenty-four dishes, variously treated discs had NIH3T3 fibroblast and MC3T3 osteoblast cells added to their surfaces. For two days these cells were cultured on the disc surfaces in a cell-culturing medium (DMEM + 10% FBS). At the end of the incubation period discs with identical surface treatments are divided up into separate groups.

2.2 Cell-Counting

Cells dissolved from the disc surfaces in tripsin solution were counted using a Bürker camera. After cell-counting, cells were dissolved in a low-temperature 200 μ l Triton (30 mM HEPES, 100 mM NaCl, 1 mM EGTA, 20 mM NaF, 1% Triton X-100, 1 mM PMSF, 20 μ l/ml protease inhibitory cocktail, 1 mM Na₃VO₄) and then with the help of a spectrometer by the Bradford method, the concentration of dissolved protein (Bio-Rad Laboratories) was measured. Cell-counting was completed with a CyQuant DNA proliferation test. After culturing for two weeks CyQuant dye was added to the cells, coloring the DNA of each cell. After 5 minutes incubation the quantity of DNA was read off with a plate reader. Table 1 below lists the surface-treatment types and test methods used.

Type of surface treatment	Tests	Results
Machined	 stereomicroscope scanning electron microscope confocal microscope contact-angle measurement cell testing (MC3T3) 	 surface morphology surface roughness wettability cell-proliferation
Machined + chemical etched		
Machined + electro-polished		
Machined + sandblasted (Al ₂ O ₃)		
Machined + 1 Joule impulse-energy laser surface modified		
Machined + 3 Joule impulse-energy laser surface modified		

 Table 1

 Surface treatments and testing possibilities for Grade 1 titanium

3 Results

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In the illustrations below stereomicroscope images of samples are shown (Fig. 2).



Figure 2

Stereo-microscope images of titanium disks machined (a), chemical etched (b), electro_polished (c), Al₂O₃ sandblasted (d), 1 J laser surface modified (e), and 3 J laser surface modified

Machined: 3-10 μ m width concentric grooves, running parallel to each other, irregular microgrooves. The depth and spacing of the furrows mostly regular. At numerous places on the surface contamination and machine-turning residue observable.

Machined + chemical etched: 1-3 μ m width regularly arranged grooves. The primary leaf-form in the grooves run parallel to each other, but with grooves in the adjacent formations nearly perpendicular. This indicates that the groove patterning is not due to the original machining, but is caused by the subsequent surface modification. The leaf-shaped primary structure, about 20 to 25 μ m in diameter, is the basic unit of the surface.

Machined + electro-polished: smooth surface, although some traces of grooving from the machined surface are still visible.

Machined + sandblasted (Al₂O₃): irregularly-shaped, sharp-contoured raised areas, showing similar geometric depths.

Machined + surface modified with 1 Joule impulse energy laser: $50-70 \ \mu m$ width grooves, regular waves on the surface. The surface is regular and smooth, insofar as microgrooving is smoothed down.

Machined + surface modified with 3 Joule impulse energy laser: $20-50 \ \mu m$ width furrows and $10-20 \ \mu m$ diameter droplets. An orderly spaced, wave like primary structure can be observed with a distance of $30-60 \ \mu m$ from each other. The wave crest projections show 30 to 50 μm . The so-called secondary structure is pear shaped having a size of $10-15 \ \mu m$.

On the SEM images 3-10 μ m width grooves can be seen on the surface of the machined sample (Figure 3).



Figure 3

Scanning electron microscope images (magnification $1000\times$) machined (a), chemical etched (b), electropolished (c), Al₂O₃ sandblasted (d), 1 J laser surface modified (e), and 3 J laser surface modified

After chemical etching there were ordered structure on the surface of the disk, parallel – though oriented in the original direction of the surface machining – grooves separated by intervals of about 1-3 μ m. After electropolishing, a smooth uniform surface – absent the concentric grooving seen on the freshly machined samples – was apparent. The surface bombarded with Al₂O₃ sandblasting shows irregular roughly granular zones. The effect of 1 Joule impulse energy laser surface modification is 50-70 μ m wide, regular, wave-like grooving. Whereas 3 Joule impulse energy laser surface modification has the effect of creating 20-50 μ m width grooves on the surface, in which solidified droplets of diameter 10-20 μ m are visible.

With a confocal microscope of 3-3 measurements the average surface roughness (Ra) was established. Measurements were taken at 1500 μ m intervals. The different surface morphologies were compared in this case with the machined sample as a reference. The results of these measurements are brought together in Figure 4 below.



Figure 4 Measured values for the surface roughness (Ra) of the discs

The surface roughness of the sample treated with electropolishing was the lowest (Ra = $0.6 \pm 0.2 \ \mu$ m), while the highest value (Ra = $3.8 \pm 0.4 \ \mu$ m) was for the discs treated with 3 Joule impulse laser modification. The chemically etched discs' surface roughness (Ra = $1.0 \pm 0.1 \ \mu$ m) was slightly less than that of the freshly machined sample (Ra = $1.1 \pm 0.1 \ \mu$ m). The titanium discs sandblasted with Al₂O₃ particles had a roughness of Ra = $1.6 \pm 0.1 \ \mu$ m. The 1 Joule impulse energy laser modified surface had a roughness level (Ra = $3.2 \pm 0.2 \ \mu$ m) only slightly less than that created by the 3 Joule laser.

In the case of the measurements by the atomic-force microscope the measured surface was 400 μm^2 . The results of the measurements are collected together in Figure 5 below.



Figure 5

Titanium Grade 1 discs measurement results: machined (a), etched (b), electropolished (c), Al₂O₃ sandblasted (d)

The Ra roughness value for various treatments to the surface of the disc was as follows: machined 0.62 μ m; chemically etched 0.08 μ m; electropolished 0.04 μ m; Al₂O₃ sandblasting 0.27 μ m. Samples treated with laser surface-modification did not get subjected to AFM measurements. This is accounted for by physical limitations on the measurement method (extremely large Ra surface roughness, which goes beyond the measurement capacity of the instruments – NB: both the AFM and confocal microscope use methods which establish surface roughness values, but there are variations in the results which deviate from the surfaces examined). On the basis of the AFM measurements it was established that the value of surface roughness Ra was lowest for the electropolished discs and highest for the machined samples.

Contact angles were established, with the Al₂O₃ sandblasting $(60 \pm 10^{\circ})$ and 1 Joule impulse energy laser surface modification $(51 \pm 3^{\circ})$ results close together within margins of error. The electropolished sample surface had the lowest value $(27 \pm 4^{\circ})$, while the 3 Joule impulse energy laser treatment gave the highest contact-angle values $(102 \pm 4^{\circ})$.

The following quantitative and qualitative characteristics were subjected to cell testing.

3.1 Cell-Counting and Protein Concentration Measurement

Results of cell-counting and protein concentration in both the fibroblast and the osteoblast cases showed increased cell proliferation on the roughened surfaces when compared with the machined surface. Tables 2 and 3 present the measured results as averages \pm standard deviation. Numbers in the columns are relative values which show what the cell count or protein concentration was on the given surface to the value for the machined surface – used widely in the literature as the reference surface.

Table 2

Changes of cell numbers and protein concentration of NIH3T3 fibroblast cells measured on various modified surfaces compared to the group with a machined surface

Sample	Cell number	Prot. Conc.
Machined	1	1
Machined + chemical etched	1.85 ± 0.15	1.34 ± 0.28
Machined + electropolished	3.24 ± 0.12	2.53 ± 0.35
Machined + sandblasted (Al ₂ O ₃)	5.19 ± 0.59	3.18 ± 0.41
Machined + 1 Joule impulse energy laser surface modified	3.17 ± 0.33	2.66 ± 0.26
Machined + 3 Joule impulse energy laser surface modified	2.83 ± 0.36	2.31 ± 0.15

Table 3

Changes of cell numbers and protein concentration of MC3T3 osteoblast cells measured on various modified surfaces compared to the group with a machined surface

Sample	Cell number	Prot. Conc.
Machined	1	1
Machined + chemical etching	1.55 ± 0.16	1.40 ± 0.21
Machined + electropolished	2.15 ± 0.32	2.32 ± 0.41
Machined + sandblasted (Al ₂ O ₃)	3.25 ± 0.33	3.05 ± 0.45
Machined + 1 Joule impulse energy laser surface modification	3.25 ± 0.22	2.40 ± 0.37
Machined + 3 Joule impulse energy laser surface modification	3.83 ± 0.29	3.41 ± 0.41

Conclusions

In the experiments 5 mm diameter, 2 mm thickness discs machined from Grade 1 titanium were used. The chemical treatment to the surface of the discs involved passivation in sodium-hydroxide + H_2O_2 , oxalic acid, nitric acid solution, electropolishing, Al_2O_3 sandblasting, as well as laser pulse surface modification (Pulsed Nd/Yag laser (Kvant 1) with a 1 J/pulse is the low pulse energy laser, and

the 3 J/pulse was used for surface treating as high pulse energy laser). Images were taken of the discs using a stereomicroscope and a scanning electron microscope. On the scanning electron microscope images $3-10 \,\mu\text{m}$ width grooving can be seen on the freshly-machined sample surfaces.

After chemical etching, structures appear on the surface of the discs parallel to each other – but these follow the orientation of the structures on the machined surface, $1-3 \ \mu m$ grooves.

After electropolishing we obtain a smooth surface, without the concentric grooves visible on the machined samples.

The Al₂O₃ sandblasted surfaces showed irregular, roughly granular zones. The effect of the 1 Joule impulse energy laser surface modification was 50-70 μ m width, regular, wave-shaped surface grooving. Whereas the effect of 3 Joule impulse energy laser surface modification was to create 20-50 μ m width grooves on the surface, in which solidified droplets of diameter 10-20 μ m were visible.

During examination with a confocal microscope, it was established that the surface roughness value of the electropolished samples ($Ra = 0.6 \pm 0.2 \mu m$) was the lowest, while the highest roughness values ($Ra = 3.8 \pm 0.4 \mu m$) were produced on 3 Joule impulse energy laser modified discs. The roughness value for those discs treated with chemically etching ($Ra = 1.0 \pm 0.1 \mu m$) was slightly less than that for the freshly machined samples. The surface roughness of the titanium discs treated with Al₂O₃ sandblasting was $Ra = 1.6 \pm 0.1 \mu m$. Roughness values for discs surface-modified by the 1 Joule pulsed energy laser ($Ra = 3.2 \pm 0.2 \mu m$) deviated only slightly from the 3 Joule laser value.

AFM measurements of the samples were carried out. The machined discs had surface roughness values of 0.62 μ m, the chemically etched 0.08 μ m, the electropolished 0.04 μ m, the Al₂O₃ sandblasted samples 0.27 μ m. It was not possible to take accurate AFM measurements of the laser surface modified surfaces.

Contact angles were established for wetting of the Al₂O₃ sandblasted surfaces $(60 \pm 10^{\circ})$, the 1 Joule pulse energy laser modified surfaces $(51 \pm 3^{\circ})$, values that were very close together. On the electropolished surfaces these were measured with the lowest values $(27 \pm 4^{\circ})$, while the highest contact-angle values $(102 \pm 4^{\circ})$ appeared on the 3 Joule impulse-energy laser modified surfaces. Quantitative and qualitative cell-test characterizations of the surfaces were made. In the case of tests with fibroblasts the greatest degree of cell proliferation was the surfaces sandblasted with Al₂O₃ particles. With the osteoblast tests the highest levels of cell activity were for the two types of laser-modified surface.

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