

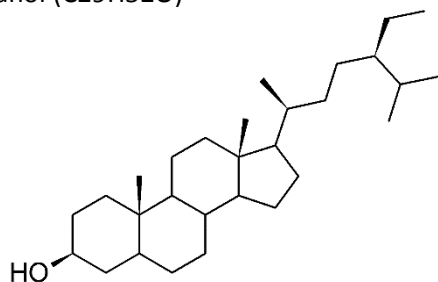
## Supplementary Material

### Rapid measurement of phytosterols in fortified food using gas chromatography with flame ionization detection

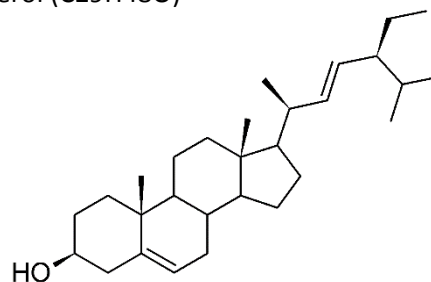
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**Figure S1.** Structures of common phytosterols (Moreau et al., 2002).

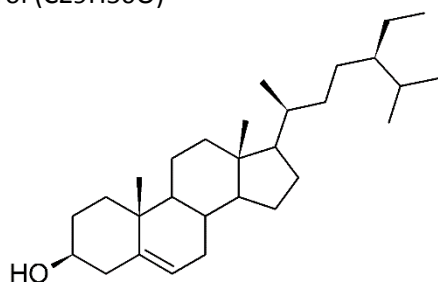
Stigmastanol (C<sub>29</sub>H<sub>52</sub>O)



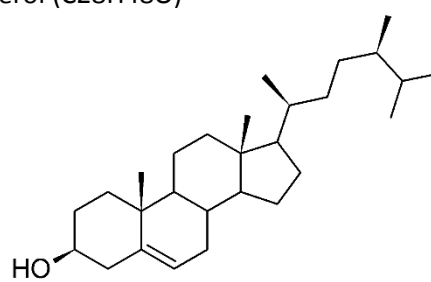
Stigmasterol (C<sub>29</sub>H<sub>48</sub>O)



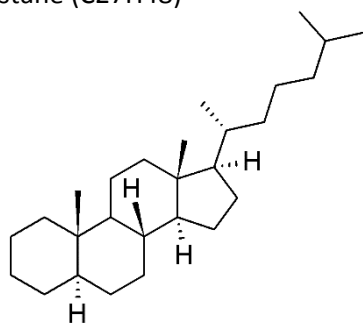
$\beta$ -sitosterol (C<sub>29</sub>H<sub>50</sub>O)



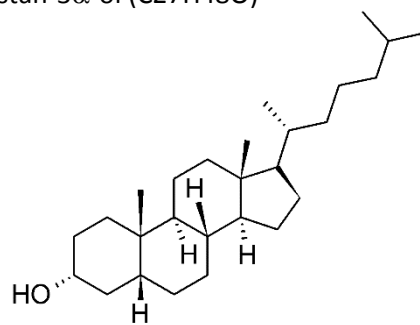
Campesterol (C<sub>28</sub>H<sub>48</sub>O)



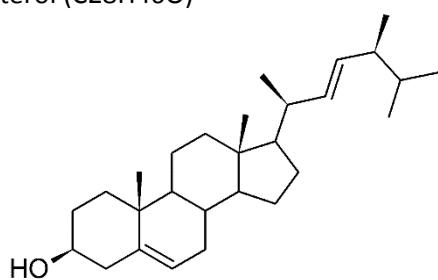
5 $\alpha$ -cholestane (C<sub>27</sub>H<sub>48</sub>)



5 $\beta$ -cholestan-3 $\alpha$ -ol (C<sub>27</sub>H<sub>48</sub>O)

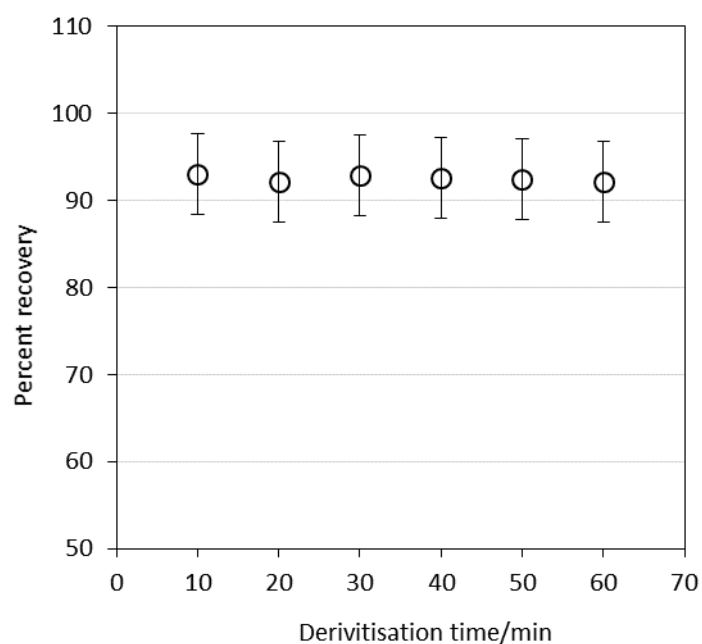


Brassicasterol (C<sub>28</sub>H<sub>46</sub>O)



**Table S1.** Plackett Burnman experimental design to investigate critical method parameters.

Experiment	Extraction solvent	Incubation temperature /°C	Saponification mixture	Saturated NaCl	Aqueous HCl	Water
A	Toluene	100	2.3 M NaOH in methanol	Yes	Yes	
B	Toluene	80	2.3 M KOH in methanol	Yes	Yes	
C	Toluene	100	2.3 M NaOH in methanol	Yes	Yes	
D	Toluene	100	2.3 M NaOH in methanol	Yes	No	
E	Heptane	100	2.3 M NaOH in methanol	Yes	Yes	
F	Toluene	100	2.3 M NaOH in methanol	No	Yes	
G	Heptane	80	5M ethanolic KOH	No		Yes
H	Heptane	80	5M ethanolic KOH	Yes		Yes
I	Heptane	80	5M ethanolic KOH	Yes		No

**Figure S2.** Effect of BSTFA/TCMS derivatisation time on sterol recovery.**Equation S1.** Recovery calculation.

$$\text{Recovery \%} = \frac{\text{sample value}}{\text{theoretical certified value}} \times 100$$