

Improvement of Thermoelectric Properties through Reduction of Thermal Conductivity by Nanoparticle Addition and Stoichiometric Change to Mg₂Si

W. Tanner Yorgason

Arden Barnes

N. A. Roberts

Mechanical and Aerospace Engineering

Utah State University

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https://flowcharts.llnl.gov/archive.html







SYSTEM





https://www.alphabetenergy.com/how-thermoelectrics-work/



Figure of Merit (*ZT*) is a measure of the thermoelectric material's efficiency in converting thermal energy to electrical current.

It is defined by the following equation: $ZT = S^2 \frac{\sigma T}{k}$









Why Use Mg₂Si?

- Mg is earth abundant in Utah and much of the US
- Mg₂Si is a relatively simple compound when compared to the other thermoelectrics
- Mg is inexpensive, resulting in the production of Mg₂Si being less expensive







https://mrdata.usgs.gov/geochem/doc/averages/mg/usa.html



Methodology

Molecular Dynamics





(Large-scale Atomic/Molecular Massively Parallel Simulator)

LAMMPS can only calculate k_p , which makes the follow up of experimental research to measure ZT critical



http://lammps.sandia.gov/#nogo

Experimental Setup





- Periodic Boundaries
- Extended modified embedded atom method (MEAM) potential

Zhang, Hengji, CMS, 2015





Experimental Setup







Experimental Setup

Nanostructure	300 K (Wm ⁻¹ K ⁻¹)	600 K (Wm ⁻¹ K ⁻¹)	900 K (Wm ⁻¹ K ⁻¹)
Pure Mg ₂ Si	k _p = ?	k _p = ?	k _p = ?
Mg ₂ Si with 1 Si NP	k _p = ?	$k_p = ?$	k _p = ?
Mg ₂ Si with 2 Si NP	k _p = ?	k _p = ?	k _p = ?
Mg ₂ Si with 4 Si NP	k _p = ?	$k_p = ?$	k _p = ?
Mg ₂ Si with 8 Si NP	k _p = ?	$k_p = ?$	k _p = ?
Mg ₂ Si with 16 Si NP	k _p = ?	$k_p = ?$	k _p = ?
Mg_xSi_x 34.29 % Si (matching stoichiometry of Mg_2Si with 1 Si NP)	k _p = ?	k _p = ?	k _p = ?
Mg_xSi_x 35.32 % Si (matching stoichiometry of Mg_2Si with 2 Si NP)	k _p = ?	$k_p = ?$	k _p = ?
Mg_xSi_x 37.29 % Si (matching stoichiometry of Mg_2Si with 4 Si NP)	k _p = ?	k _p = ?	k _p = ?
Mg_xSi_x 41.37 % Si (matching stoichiometry of Mg_2Si with 8 Si NP)	k _p = ?	$k_p = ?$	k _p = ?
Mg_xSi_x 49.55 % Si (matching stoichiometry of Mg_2Si with 16 Si NP)	k _p = ?	$k_p = ?$	k _p = ?





- Nonequilibrium Molecular Dynamics (NEMD)
- Applied Heat Flux
- Total simulated time of 15 ns



Methodology









Uncertainty Calculations

- Simulations were run at the 3 different equilibration temperatures as previously described, except that no heat flux was applied (we assumed stoichiometry did not significantly affect the uncertainty in temperature)
- The temperatures for each chunk were then averaged all together
- The absolute value of the difference between this value and the target equilibration temperature was taken as our uncertainty in temperature
- This value was then added to and subtracted from the ΔT in Fourier's Law to obtain the minimum and maximum k_p values, and therefore their associated k_p uncertainties





Equilibration Temperature (K)	Uncertainty (K)	
300	0.774	
600	0.721	
900	0.96	



Si NP in the wall and heat sink

Effect of NP Placement on k_p



- Some simulations had Si NPs in the walls and heat source/sink
- To ensure that this wasn't a problem, 2 simulations were run with 8 Si NPs each; one had an Si NP in the heat sink/wall area, the other did not.
- Their k_p values were 2.876 (Wm⁻¹K⁻¹) and 3.063 (Wm⁻¹K⁻¹), respectively, resulting in a percent change in k_p of 6.499%, which we considered negligible.





Results – Pure Mg₂Si

Work	k_p of Pure Mg ₂ Si at 300 K (Wm ⁻¹ K ⁻¹)	
LaBotz	7.8	
This Work	8.454 ± 1.094	

We determined that our calculated value for k_p above was sufficiently close that of Labotz, such that we could begin simulation of off-stoichiometry samples of Mg₂Si and calculation of their respective values for k_p .



LaBotz, JES, 1963



Results – Mg₂Si with Si NPs







Results – Mg₂Si with Si NPs



FIGURE 5: Lattice thermal conductivity (k_p) vs. temperature (T) for pure Mg₂Si (0 NPs) and Mg₂Si samples with 1, 2, 4, 8, and 16 Si NPs.



One Si NP Case



- Unexpectedly, the 1 NP cases resulted in lower in k_p values than their respective 2 NP cases
- We thought this might have to do with the NP spacing



 It appears that the closer the Si NP concentration is to the center of the sample, the lower the k_p, despite no change in stoichiometry





Results – Mg₂Si with Si Substitutionals





FIGURE 4: Lattice thermal conductivity (k_p) vs. percent Si present in Mg₂Si samples at 300, 600, and 900 K. The percent Si values correspond to the stoichiometry of the sample of pure Mg₂Si (33.33% Si), and respective samples of Mg₂Si having 1, 2, 4, 8, and 16 Si NPs.



Results – Mg₂Si with Si Substitutionals









Results – Table Summaries

Nanostructure	300 K (Wm ⁻¹ K ⁻¹)	600 K (Wm ⁻¹ K ⁻¹)	900 K (Wm ⁻¹ K ⁻¹)
Pure Mg ₂ Si	8.454 ±1.094	4.199 ±0.342	3.533 ±0.705
Mg ₂ Si with 1 Si NP	5.252 ±0.416	3.275 ±0.176	2.428 ±0.261
Mg ₂ Si with 2 Si NP	5.877 ±0.586	3.456 ±0.312	2.987 ±0.626
Mg ₂ Si with 4 Si NP	4.553 ±0.387	2.972 ±0.205	2.204 ±0.277
Mg ₂ Si with 8 Si NP	2.876 ±0.211	1.930 ±0.119	1.992 ±0.323
Mg ₂ Si with 16 Si NP	1.791 ±0.124	1.649 ±0.157	1.280 ±0.214
Mg_xSi_x 34.29 % Si (matching stoichiometry of Mg_2Si with 1 Si NP)	6.346 ±0.624	3.749 ±0.283	2.676 ±0.369
Mg_xSi_x 35.32 % Si (matching stoichiometry of Mg_2Si with 1 Si NP)	5.015 ±0.441	3.001 ±0.244	2.065 ±0.296
Mg_xSi_x 37.29 % Si (matching stoichiometry of Mg_2Si with 1 Si NP)	3.669 ±0.272	2.430 ±0.186	1.387 ±0.151
Mg_xSi_x 41.37 % Si (matching stoichiometry of Mg_2Si with 1 Si NP)	2.300 ±0.133	1.784 ±0.126	1.591 ±0.264
Mg_xSi_x 49.55 % Si (matching stoichiometry of Mg_2Si with 1 Si NP)	1.300 ±0.053	1.064 ±0.117	0.9347 ±0.292





Conclusions

- Increasing the atomic percent Si, either through substitutional atoms or Si NPs, decreases k_p
- Samples with substitutional Si atoms resulted in greater decreases in k_p when compared to the k_p of the samples with Si NPs
- Boundary resistance, rather than reduction in mean free path, seems have the greater influence in reducing k_p in the samples with Si NPs for the 1 and 2 NP cases





Future Work

- Experimental research will need to verify that these nanostructures actually result in an increased ZT for Mg_xSi_x as LAMMPS cannot account for changes in k_e
- Further work should be done to understand why substitutional Si atoms lower k_p more than the stoichiometric equivalent of Si NPs in Mg₂Si
- Further should be done to understand more fully how concentrations of stoichiometric changes at certain locations, such as Si NPs at the halfway point between a heat source and sink, can change k_p





References

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Questions?