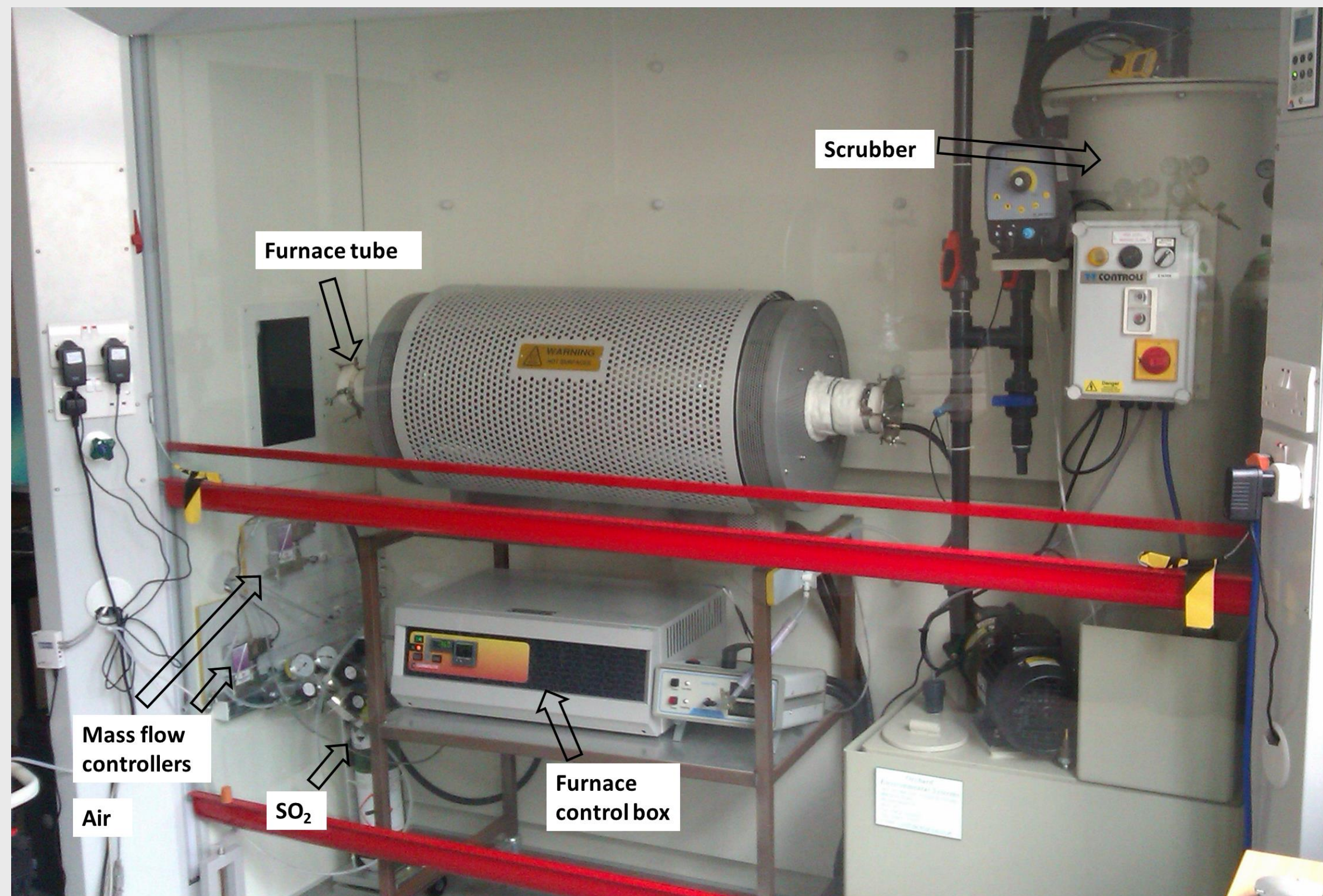


CALCIUM SULFOALUMINATE CEMENT CLINKERING

Galan I, Hanein T, Elhoweris A, Skalamprinos S, Jen G, Whittaker M, Bannerman MN, Imbabi MS, Glasser FP
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The aim of these studies was to gain understanding in the phase development of calcium sulfoaluminate cements, together with their resulting properties. Laboratory experiments, pilot plant trials and thermodynamic calculation were used for the studies. The results show that phase development is affected not only by temperature and composition but also by the partial pressures of SO₂ and O₂ gas in the kiln. The mineralogical contents of the sulfate- containing phases, in coexistence with other silicate and aluminate phases, can be controlled reproducibly.

LABORATORY EXPERIMENTS



Raw materials
SO₂+O₂
Lab grade Al₂O₃, SiO₂, CaCO₃, Fe₂O₃, CaSO₄
Bauxite, clay, limestone

Synthesis
C \bar{S} , C₄A₃ \bar{S} , C₅S₂ \bar{S} , clinkers (C₄A₃ \bar{S} -C₂S-C₄AF)
T 1200-1300 °C
Ratio SO₂:air (g/min) 1:2.5-1:100 (0.04:4)
Time 5-240 min
Powders/pellets

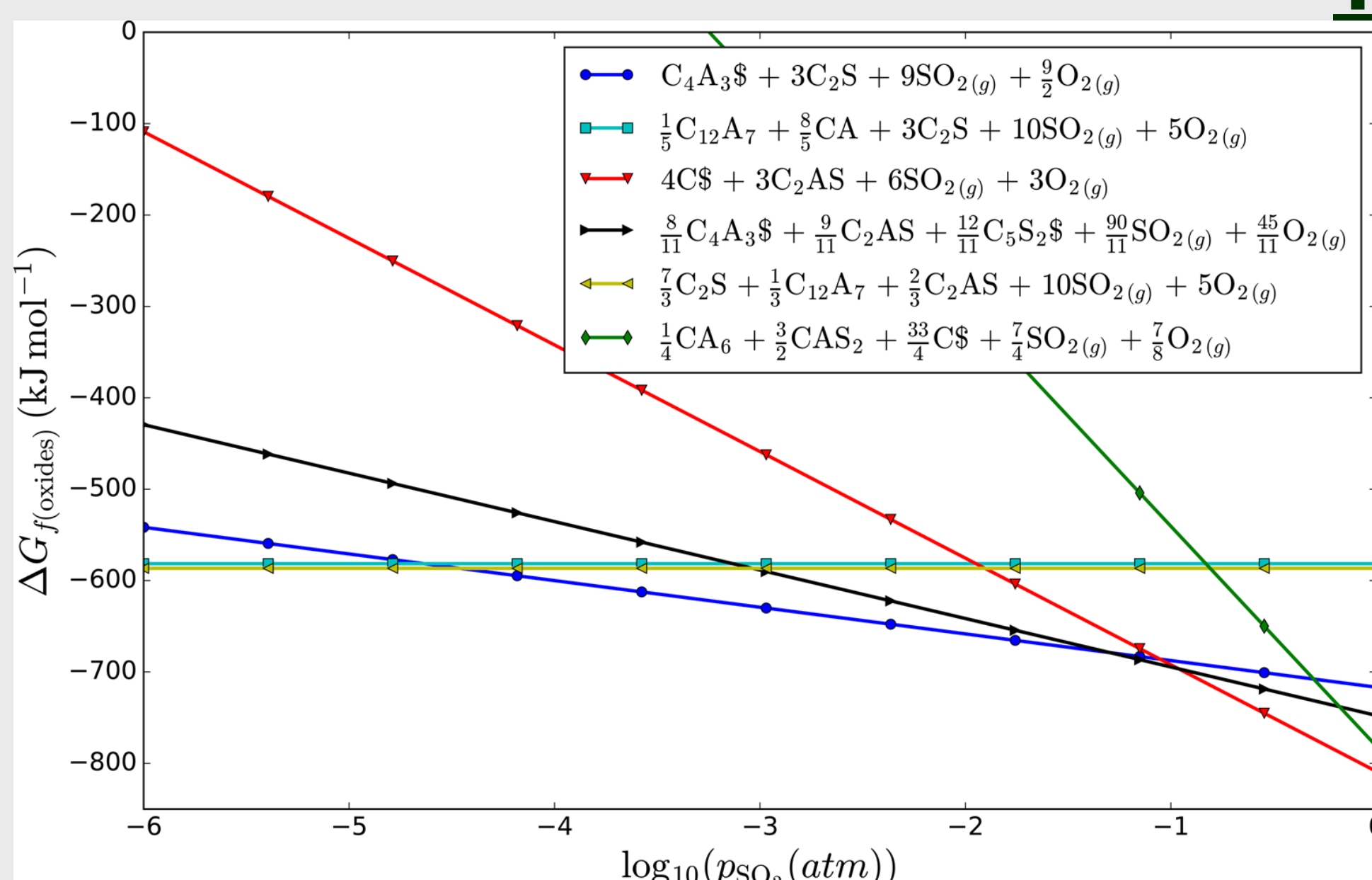
Characterisation
XRD-Rietveld

- Rapid transfer of SO₂ and O₂ to clinkering solids to form C \bar{S} A clinkers
- Temperature windows for clinkers with C₄A₃ \bar{S} and C₅S₂ \bar{S}
- SO₂ partial pressure threshold:
C \bar{S} +C₂AS <-> C₄A₃ \bar{S} +C₂S



Advances in clinkering technology of calcium sulfoaluminate cement, Galan, Elhoweris, Hanein, Bannerman, Glasser, Adv. Cem. Res. 2017, in press

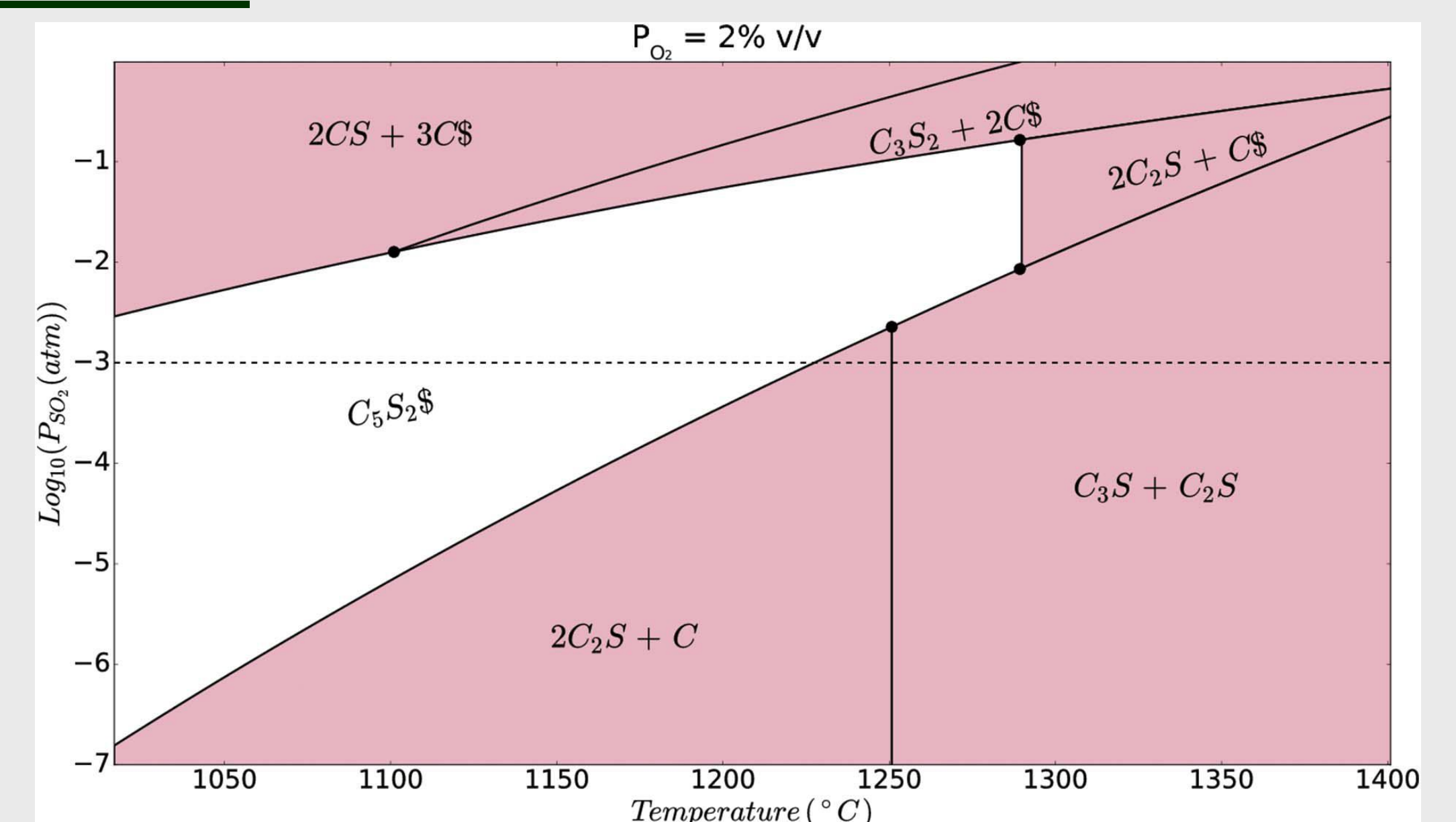
THERMODYNAMIC CALCULATIONS



Phase compatibility in the System CaO-SiO₂-Al₂O₃-SO₃-Fe₂O₃ and the Effect of Partial Pressure on the Phase Stability, Galan, Hanein, Elhoweris, Bannerman, Glasser, Ind. Eng. Chem. Res. 2017, 56

Thermodynamic stability of sulfate-containing phases

Ranges of SO₂, O₂ and temperature where the phases in the C-S-A- \bar{S} -F system form and coexist



Stability of ternesite and the production at scale of ternesite-based clinkers, Hanein, Galan, Glasser, Skalamprinos, Elhoweris, Imbabi, Bannerman, Cem. Concr. Res. 2017, 91

PILOT PLANT TRIALS



Kiln: 7.4 m long; 0.3 m diameter
Counter current flow
Natural gas + Sulfur powder (introduced with screw feeder to the burner)

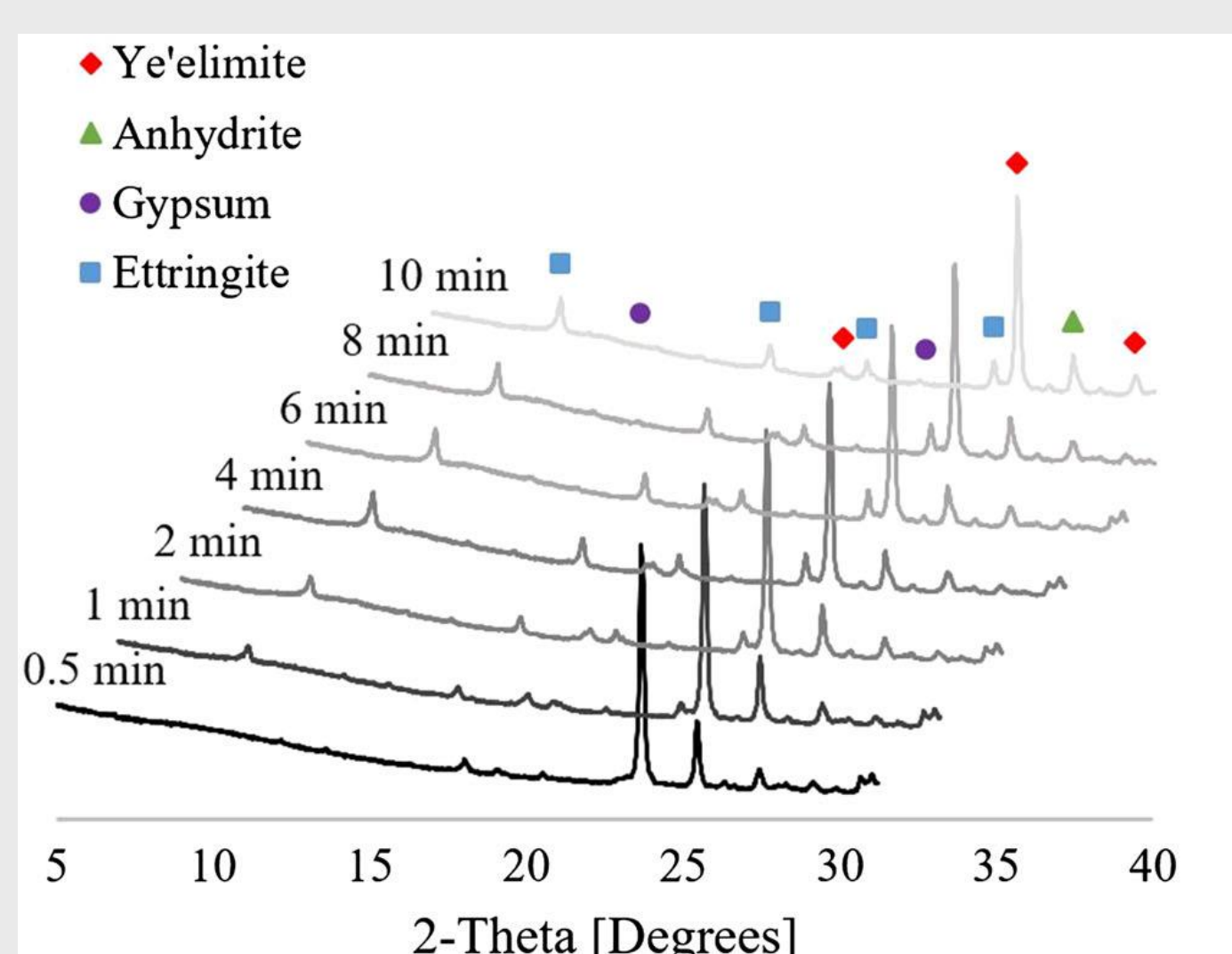
Limestone, bauxite and clay

Average operating conditions:
1260 °C; 7% O₂; 0.4% SO₂; rpm 2

Produced clinker
36% C₄A₃ \bar{S} 28% β-C₂S 15% α'-C₂S
10% C \bar{S} 4% C₂AS 2% C₄AF
2% CT, 2% C



Production of belite calcium sulfoaluminate cement using sulfur as a fuel and as a source of clinker sulfur trioxide: pilot kiln trial, Hanein, Galan, Elhoweris, Khare, Skalamprinos, Jen, Whittaker, Imbabi, Glasser, Bannerman, Adv. Cem. Res. 2016, 28



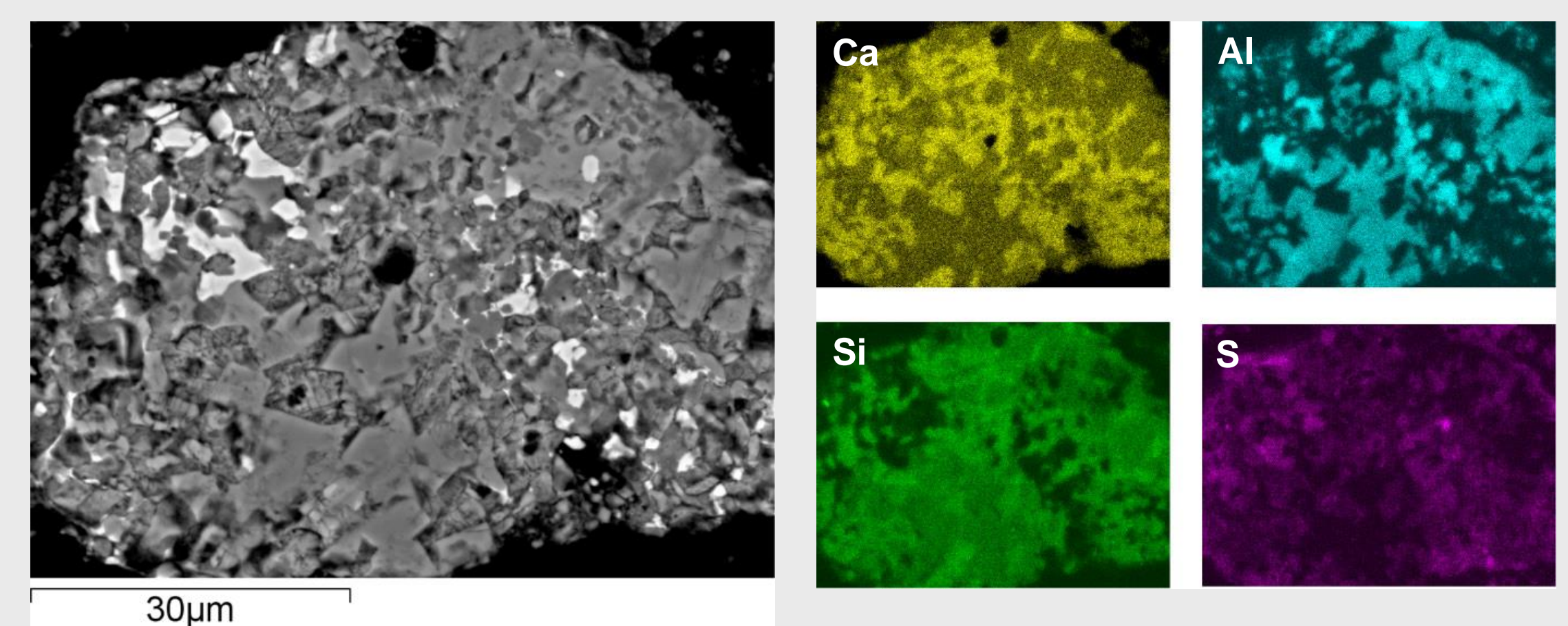
The impact of intrinsic anhydrite in an experimental calcium sulfoaluminate cement from a novel, carbon-minimized production process, Jen, Skalamprinos, Whittaker, Galan, Imbabi, Glasser, Mat. Struct. 2017, 50

HYDRATION

Experimental clinker: rapid hydration
Intrinsic anhydrite -> ettringite network

Setting time retardation: addition of citric acid + gypsum or additional mixing and higher w/c

Early strength higher without retarder
28d strength higher with retarder



CONCLUSIONS

The understanding of the influence of the SO₂ + O₂ partial pressure, together with the temperature, on the formation of phases in the system C-S-A- \bar{S} -F has enabled a new approach to clinker design and production.

A new generation of sulfoaluminate compositions are being developed which are readily clinkered using conventional processing. Advantages are: further reduction of CO₂ emissions, re-utilization of a waste material, and strength development improvement by controlling and optimising the mineralogy.