# Stable isotope compositions of speleothems from the last interglacial – spatial patterns of climate fluctuations in Europe

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#### **Highlights**

- Stable isotope compositions of a last interglacial (LIG) stalagmite, Central Europe
- A sequence of climate change events during the LIG for European speleothems
- Hydrogen isotope data of inclusion water reveal a major event at about 125 ± 2 ka
- Temperature and winter precipitation changes around the Mediterranean at  $125 \pm 2$  ka

ABSTRACT

Studies on the last interglacial (LIG) can provide information on how our environment behaved in a period of slightly higher global temperatures at about  $125 \pm 4$  ka, even if it is not the best analogue for the Holocene. The available LIG climate proxy records are usually

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better preserved and can be studied at a higher resolution than those of the preceding interglacials, allowing detailed comparisons. This paper presents complex stable hydrogen, carbon and oxygen isotope records obtained for carbonate ( $\delta^{13}$ C and  $\delta^{18}$ O<sub>carb</sub>) and fluid inclusion hosted water ( $\delta D$  and  $\delta^{18}O_w$ ) of a stalagmite from the Baradla Cave system in Central Europe that covers most of the LIG, as proven by U-Th dates. Comparing its C and O isotope data with records reported for other speleothem (cave-hosted carbonate) deposits from Europe revealed the complex behavior of these climate proxies, with a concerted relative increase in <sup>18</sup>O of carbonates from 128 to 120 ka and synchronized shifts in the opposite direction after 119 ka. The hydrogen isotope analyses of inclusion-hosted water extracted from the BAR-II stalagmite also correspond to the regional climate proxy records, with meaningful deviations from global temperature trends. Beside following the general paleotemperature pattern from the climate optimum (high δD values up to -64‰ around 120 ka) to the subsequent cooling starting at about 119 ka (low  $\delta D$  values down to -90% at about 109 ka), a period between 126.5 and 123 ka with low  $\delta D$  values (down to -81%) is detected in the BAR-II stalagmite. Although the isotope shifts are muted in the C-O isotope data of carbonate due to competitive fractionation processes, the  $\delta^{13}$ C data show a positive relationship with the  $\delta D$  pattern, indicating humidity – and possibly temperature – variations. The periods with low  $\delta D$  values fit well to temperature and humidity changes inferred from proxy records from western Europe to the eastern Mediterranean. Spatial distributions of these variables show, that at about 125 ± 2 ka the Mediterranean region was characterized by warm, humid conditions and enhanced seasonality with elevated winter precipitation. The combined interpretation of stable isotope data revealed that the Alpine and Mediterranean regions behaved differently during Greenland Stadial 26 (GS26, ~119 to 116.2 ka). While the Alpine records fluctuated in close agreement with the Central

Greenland ice core  $\delta^{18}O$  data, the BAR-II stalagmite show a positive  $\delta^{18}O_{carb}$  anomaly. The Baradla data indicate enhanced aridity and seasonality for a part of GS26, with the relative dominance of summer precipitation and Mediterranean moisture contribution. Following the GS26 event, the effect of long-term global cooling becomes dominant in the Baradla isotope records and leads to glacial inception at about 109 ka.

**Keywords**: last interglacial, speleothems, Europe, stable isotope compositions, fluid inclusions

#### 1. Introduction

In the last 5 million years Earth's climate has experienced substantial cooling, leading to an ice age characterized by the repeated expansions of extensive ice sheets (Lisiecki and Raymo, 2005). Stable isotope compositions of deep sea drilling cores and ice cores from the Antarctic continent convincingly indicate that the period of the last ~1 million years was dominated by cold (glacial) periods cyclically interrupted by warm periods, called interglacials (EPICA community members, 2004; Lisiecki and Raymo, 2005). At present we are living in an interglacial that provides a comfortably warm and humid environment for the human society. The future of our society depends largely on the evolution of the natural environment, hence the understanding of how the present interglacial will change and how natural processes and anthropogenic influences can contribute to the temperature and humidity changes is increasingly important. Beside global climate models that forecast meteorological changes from the present to the near future, paleoproxy records of sedimentary formations (marine sediments, lake deposits, speleothems, etc.) formed during past interglacial periods can provide important data on how temperature and precipitation

conditions evolved on decadal to millennial time scales. The best natural analogue for the present interglacial would be the warm period around 400 ka before present, as the orbital parameters that determine insolation were similar to the present day situation (Loutre and Berger, 2003). Unfortunately, as we go back in time, suitable paleoproxy records become less and less frequent due to their decreasing availability and subsequent alterations. This is especially true for speleothems, as the formations of earlier interglacials are generally covered by younger deposits and the large stalagmites are usually strictly protected, prohibiting appropriate sampling (i.e., microdrilling along the growth axis on a polished cross section). A rare exception is the compiled database of Chinese speleothems that cover the last 640 ka and whose composite stable oxygen isotope record could be compared with solar insolation values and orbital parameters (Cheng et al., 2016).

Speleothems are valuable in paleoclimate studies for a number of reasons (see the comprehensive review by Fairchild and Baker, 2012): i) they can be precisely dated using the U-Th series method providing a solid base for investigations of temporal evolution of paleoclimate proxies; ii) speleothems are formed in protected and relatively stable cave environments and hence are not exposed to late-stage alterations (like in case of sedimentary diagenesis that is generally induced by significant temperature changes coupled with influx of exotic fluids); iii) speleothems can be studied by geochemical methods that provide valuable environmental proxies such as stable isotope and trace element compositions. Compared to earlier periods, stalagmites from the last interglacial (LIG) are rather frequent and can be studied in sufficient numbers (Govin et al., 2015). Stable C and O isotope records, for instance, from European caves' carbonate deposits are available from Belgium (Vansteenberge et al., 2016), the Alps (Austria; Meyer et al. 2008, Boch et al., 2011; Moseley et al. 2015), the Massif Central (France; Couchoud et al. 2009), and the Apuane Alps

(Italy; Drysdale et al., 2005; 2007). These records have been dated by the U-Th series dating method providing numerical ages, making them valuable for paleoclimate studies on the LIG, although their geochemical compositions have not been evaluated in detail as yet (Govin et al., 2015). Besides conventional carbonate C and O isotope analyses, H and O isotope values of water trapped in the fluid inclusions of stalagmites have gained increasing importance in recent years as technical developments improved sampling resolution, analytical precision and throughput (Vonhof et al., 2006; van Breukelen et al., 2008; Dublyansky and Spötl, 2009; Griffiths et al., 2010; Wainer et al., 2011; Rowe et al., 2012; Arienzo et al., 2013; Ayalon et al., 2013; Affolter et al., 2014). The main advantage of fluid inclusion analyses is that - in contrast to C and O isotope ratios of the precipitating carbonate that depend on a large number of factors (see Fairchild and Baker, 2012) - H and O isotope compositions of the water are not fractionated during entrapment, and hence can directly reflect the composition of dripwater (Harmon and Schwarz, 1981; Yonge, 1982). Although, following deposition, the oxygen isotope composition of the trapped water can change in the course of diagenetic alteration (Demény et al., 2016a), the hydrogen isotope composition remains intact during and after the entrapment. In this study we present the carbonate C and O isotope compositions as well as fluid inclusion H and O isotope values of a stalagmite from the Baradla Cave (NE Hungary) that were used to detect paleohydrological changes in the last interglacial. These records were compared with paleoproxy data obtained from Western and Central Europe and the Mediterranean region in order to reveal spatial temperature and humidity changes in the course of the LIG.

#### 2. Cave site and stalagmites

The Baradla Cave is situated in north-east Hungary (N48°28′ E20°30′) at the Hungarian-Slovakian border, beneath a forested area with a mean elevation of 300 to 450 m a.s.l. The cave system was formed in lagoon facies of the Wetterstein Limestone of the Triassic (Ladinian) age and is characterized by almost horizontal passages. The area is under the influence of European cyclone tracks with the most important cyclogenesis centers in the Icelandic and the Western Mediterranean regions (Bartholy et al., 2006, 2009). Cyclones starting-up near these regions transport marine moisture to Hungary (Bottyán et al., 2013) by the induced frontal systems. Most Icelandic cyclones arrive in winter, but approximately 30 cyclones derive from the Mediterranean area, most frequently in spring (Kelemen et al. 2015). Significant weather influences could also come from the East (Bartholy at al., 2006). However, this weather pattern brings cold/dry air into the region, so precipitation is typically not accompanied with these macrocirculation situations. The most characteristic season and situation when such influences can be prevailing is winter, when cold polar/continental air masses intrude to the region. Monthly averages of precipitation amount, temperature and potential evapotranspiration data were extracted from the Carpatclim dataset (Szalai et al., 2013) that indicate that infiltration may occur in the period of November to April, while evapotranspiration exceeds precipitation in the warm months (Demény et al., 2016b).

On the basis of combined evaluation of cave monitoring data, as well as textural, stable isotope and trace element analyses of four actively growing stalagmites and their comparison with meteorological data, Demény et al. (2016b) determined that carbon isotope compositions of compact, slowly growing stalagmites of the Baradla Cave reflect variations in winter precipitation amount, while their oxygen isotope composition can be considered as a proxy for winter temperature. Although these relationships were calibrated

for the current interglacial conditions, they can provide a basis for the interpretation of stable isotope compositions of the stalagmite formed during the last interglacial.

The collection site of the stalagmite studied in this paper is about 1 km away from the nearest entrance (Fig. 1) and under about 150 m of host rock cover. Cave conditions were characterized by spot measurements conducted during water sampling (Demény et al., 2016b). Cave temperature was  $9.7 \pm 0.5$  °C,  $CO_2$  content was  $3500 \pm 1100$  ppm, pH was  $7.9 \pm$ 0.3 (n=14). Dripwater compositions were fairly constant in the entire cave ( $\delta D = -66 \pm 1.5\%$ ,  $\delta^{18}O = -9.5 \pm 0.2\%$ , n= 27). The stalagmite studied in this paper (BAR-II) is actually a double one, with two columns, 34 and 36 cm long (BAR-II#L and BARII#B, respectively). The internal textures of the cut and polished surfaces of the two separate columns are very similar, with clear hiata (Fig. 1). The stalagmites are composed of dense, well-crystallized calcite. Four major textural types characterize both stalagmite columns: the formation starts with a finely laminated texture in which the lamination is expressed by whitish, inclusion-rich layers (#1). This is followed by white, cloudy, inclusion-rich texture (#2), then translucent calcite with fine lamination (#3) and texture #2 again. The white, cloudy texture becomes gradually pale and lamination appears (#4) that turns into translucent, finely laminated calcite (#3). Hiatus #1 (see Fig. 1) closes this sequence as a ~ 1 mm thick white (yellowish at the sides) layer. Stalagmite formation continues with translucent calcite, but this part is not investigated in this paper. Both columns were dated by U-Th series method. High-resolution carbon and oxygen isotope analyses were conducted on the BAR-II#B column, BAR-II#L was used for a replication test. Both columns were analysed by means of hydrogen and oxygen isotope measurements of fluid inclusion-hosted water.

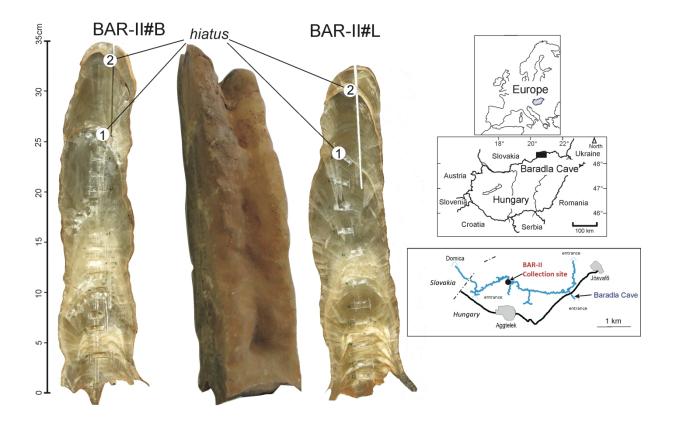


Fig. 1

# 3. Analytical methods

Altogether 23 U-Th dates were obtained for the two columns of the BAR-II double stalagmite (Fig. 1), placed either at hiata or distributed evenly along the stalagmite's axis. The U-Th age determinations were conducted at the Department of Geosciences, National Taiwan University using standard procedure (Shen et al., 2002; 2003). The BAR-IIB stalagmite column was analysed in two analytical runs in 2007 and 2015, whereas the analyses of the BAR-IIL column was conducted only during the 2015 run. Sample chips (0.1–0.2 g) were dissolved in HNO<sub>3</sub>, spiked with a mixed <sup>229</sup>Th-<sup>233</sup>U-<sup>236</sup>U tracer, refluxed and evaporated. U and Th were co-precipitated with an Fe-carrier, then separated to form a clean U fraction and a clean Th fraction using an anion exchange resin (Shen et al., 2003). The isotopic compositions of U and Th in the solutions were measured by a Thermo Electron NEPTUNE

multi-collector inductively coupled plasma mass spectrometer (MC-ICP-MS) (Shen et al., 2012). The obtained ages are given as year BP ("Before Present", i.e., before the analyses' dates). The results of the two analytical runs in 2007 and 2015 showed excellent agreement (see Supplementary Table 1). The estimated uncertainity of individual dates range from 683 to 1288 years (with an average of 864 years). Age depth models for the stalagmite columns BAR-II#B and BAR-II#L (Fig. 2) were established using the StalAge algorithm (Scholz and Hoffmann, 2011). The Stalage model fit all the age dates of the BAR-II#B column well, while it deviates from the youngest age date in case of the BAR-II#L column. This calculation problem has also been detected by Mudelsee et al. (2012) for a case when the youngest date deviates significantly from the age-depth relationship for other samples from the same stalagmite. Since only the youngest part and only one age date point is influenced in the BAR-II#L column that served as a replication test for the BAR-II#B column analysed at higher resolution, StalAge estimates have been replaced by a simple linear interpolated age-depth relationship above the last two age dates of BAR-II#L.

Stable carbon and oxygen isotope compositions were determined on calcite powder drilled at a spatial resolution of about 5 mm for a preliminary investigation, then at <1 mm resolution for a detailed study using the continuous flow isotope ratio mass spectrometric method (Spötl and Vennemann, 2003). The low-resolution analyses of the BAR-II#L column were conducted at the University of Lausanne, the higher resolution measurements on the BAR-II#B column were made at the Institute for Geological and Geochemical Research (IGGR), Budapest. Isotopic compositions are expressed as  $\delta^{13}$ C and  $\delta^{18}$ O in % relative to V-PDB, with a precision better than  $\pm$  0.1% (based on replicate analyses of samples and standards). Stable hydrogen and oxygen isotope compositions were determined in the water extracted from fluid inclusions at the IGGR, Budapest, using the laser spectrometric method

described by Czuppon et al. (2014) and Demény et al. (2016a). The reproducibilities of the  $\delta D$  and  $\delta^{18}O$  values are better than  $\pm$  2 and 0.5%, respectively. As hydrogen and carbon isotope compositions are reported only for water and carbonate, respectively,  $\delta D$  and  $\delta^{13}C$  will not be distinguished for water or carbonate, whereas  $\delta^{18}O$  will be expressed as  $\delta^{18}O_{carb}$  or  $\delta^{18}O_{w}$  when carbonate or water is not explicitly stated.

# 4. Results

The two columns of the BAR-II stalagmite were dated by U-Th method from the bases to a major hiatus (hiatus 1 marked in Fig. 1), starting from  $127.5 \pm 1.3$  ka (BAR-II#B) and  $128.7 \pm 0.9$  ka (BAR-II#L) and ending at  $110.1 \pm 1.3$  ka (BAR-II#B) and  $109.6 \pm 0.9$  ka (BAR-II#L) (see Supplementary Table 1). The separate age-depth model curves are quite similar (Fig. 2) and show that the stalagmite fulfils the suitability criteria of sufficiently fast and stable growth rate to achieve appropriate resolution (Genty et al., 2013; Govin et al., 2015).

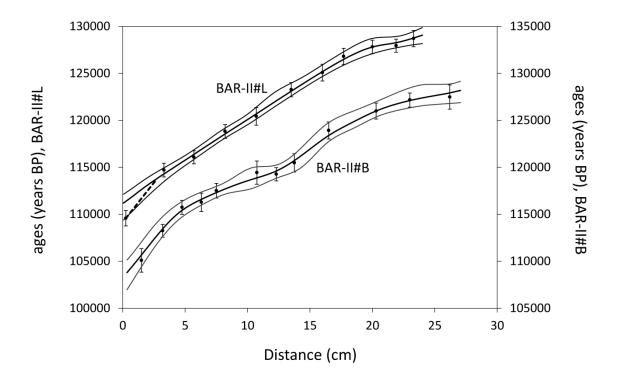


Fig. 2

Hendy test analyses (Hendy, 1971) were conducted for six laminae of the BAR-II#B stalagmite (Supplementary Table 2, Fig. 3), although it should be noted that sampling of annual laminae at a growth rate of ~0.01 mm/year was not possible, and the sampling of collected layers spans 50-100 years. This spatial resolution issue was one of the reasons why Dorale and Liu (2009) criticized the use of the Hendy test and suggested the Replication Test instead. Nevertheless, we should note that the sampled layers showed no positive  $\delta^{18}$ O- $\delta^{13}$ C correlation, only the  $\delta^{13}$ C values were shifted in the positive direction by ~0.7% (average shift for the six layers).

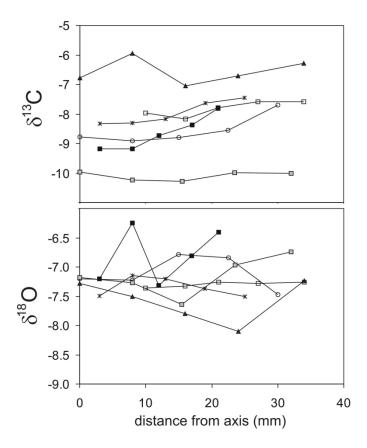


Fig. 3.

Stable carbon and oxygen isotope compositions obtained for the stalagmite columns are listed in Supplementary Table 3. The high-resolution  $\delta^{13}C$  and  $\delta^{18}O_{carb}$  values gathered for the BAR-II#B column at the IGGR, Budapest match the isotopic compositions obtained for the BAR-II#L column at the University of Lausanne (Fig. 4) that gives credit to the compositions by means of laboratory-related data precision and stalagmite record replication. The youngest part of the BAR-II#L column was resampled at a spatial resolution of 0.5 mm and measured at the IGGR, where the BAR-II#B could be analysed only at low resolution due to earlier sampling for U-Th dating. Both the low and the high-resolution data of the BAR-II#L column fit the isotopic pattern of the BAR-II#B, although there seems to be a systematic ~500 years shift in the two records. It should be noted, however, that this

difference is smaller than the analytical error of the U-Th dating. The  $\delta^{13}$ C values show a larger difference between the two records than the  $\delta^{18}$ O<sub>carb</sub> values, but this is in agreement with the stronger kinetic effect on C isotopes caused by CO<sub>2</sub> degassing during carbonate precipitation (Hendy, 1971; Mickler et al., 2004).

The textural types (see section 2) show a correspondence to the isotope data as the inclusion-rich, white layers are associated with low  $\delta^{13}$ C values (Fig. 4). Additionally, the highest water contents are found in the period of 122.1 to 118.6 ka (see Supplementary Table 3), overlapping with the part with the lowest  $\delta^{13}$ C values.

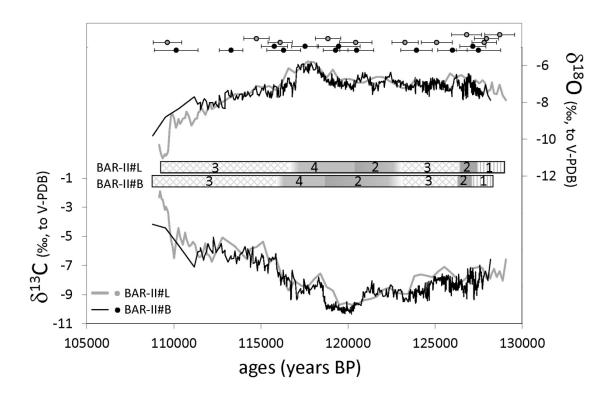


Fig. 4 The results of the high-resolution analyses on the BAR-II#B column are shown in Fig. 4 and listed in Supplementary Table 3. The  $\delta^{13}$ C values range from -10.3 to -4.2%. The

compositions start at a relatively high  $\delta^{13}$ C values of -6.6% at 128.2 ka and decrease to about -10% with fluctuations. The  $\delta^{13}$ C record has a minimum of around -10% in the period of 120.7 to 118.8 ka, followed by a sudden increase by about 2% over 1000 years. After about 118.5 ka the  $\delta^{13}$ C values gradually increase, with a maximum of -4.2% at 108.8 ka. The  $\delta^{18}$ O<sub>carb</sub> values start at about -8% at 128.2 ka and show a strong fluctuation of  $\sim$ 1% and an average  $\delta^{18}$ O<sub>carb</sub> value of about -7% until 120 ka. The  $\delta^{18}$ O<sub>carb</sub> values start increasing at 120 ka with a steeper increase from 119 ka (contemporaneously with the positive  $\delta^{13}$ C shift), and reach a maximum between 118.4 and 117.1 ka, ended with a decrease by 1.2% at 117 ka. Finally the  $\delta^{18}$ O<sub>carb</sub> values show a steep decline to -9.8% until 108.8 ka.

The H and O isotope compositions of fluid inclusion water (Supplementary Table 3) are positively correlated ( $R^2$ = 0.73), but some of the  $\delta^{18}O_w$  values are shifted to irrealistically negative values down to -16.4‰. Plotting the H and O isotope compositions as a function of age (Fig. 5), similar patterns are obtained for both stalagmite columns and for both isotope compositions. Low  $\delta D$  and  $\delta^{18}O_w$  values characterize the period of 129 to 123 ka, high and relatively constant compositions appear between 122 and 119 ka, then both the  $\delta D$  and  $\delta^{18}O_w$  values decrease. Evaporation as a cause of the positive  $\delta D - \delta^{18}O$  correlation is not regarded as a viable process, as in that case the starting compositions (least affected by evaporation) should correspond to the very low  $\delta^{18}O_w$  values that – together with the  $\delta D$ values - would mean an irrealistic shift (with a calculated d-excess value of +41) from the Global Meteoric Water Line (Craig, 1961). The  $\delta^{18}O_w$  shift to very low values (down to – 16.4‰) rather indicates that the post-depositional oxygen isotope exchange processes detected by Demény et al. (2016a) affected the BAR-II stalagmite. Demény et al. (2016a) observed coupled changes in  $\delta^{18}O_w$  and calcite crystallinity parameters and related the isotope shift to recrystallization of the stalagmite carbonate and oxygen isotopic re-

equilibration between the surrounding calcite and the inclusion-hosted water. As a reason for the re-equilibration they proposed that initially the carbonate was precipitated partly as metastable amorphous calcium carbonate (ACC) that crystallized to calcite after the deposition. On the basis of literature data, they assumed that the ACC-water oxygen isotope fractionation is smaller than the calcite-water fractionation value, hence the  $\delta^{18}O_w$  of the inclusion-hosted water trapped in the ACC-calcite mixture would be shifted to more negative values upon the ACC-calcite transformation. In order to approach the larger calcite-water fractionation, the water composition should change without any appreciable  $\delta^{18}O_{carb}$  shift due to the mass dominance of the host carbonate. Additionally, Demény et al. (2016c) have found direct mineralogical evidences for ACC precipitation at the cave site and determined the ACC-water oxygen isotope fractionation that turned out to be at least  $2.4 \pm 0.8\%$  smaller than the calcite-water value, giving support to the assumptions described above. It is interesting to note that the strongest  $\delta^{18}O_w$  shift occurs in the youngest part (<115 ka) that belongs to the cooling period of the last interglacial (Govin et al., 2015). Demény et al. (2016a,c) suggested that as ACC is stabilized at low temperature (<10 °C), cooling may induce ACC precipitation at higher proportion, and consequently stronger  $\delta^{18}O_w$  changes, in accordance with the present observation. Hence the  $\delta^{18}O_w$  pattern can be partly related to formation temperatures and may support the primary, climate-related nature of the  $\delta D$ values. However, as the  $\delta^{18}O_w$  values are shifted by the post-depositional processes, only the δD values will be discussed in the followings. As both stalagmite columns provided reliable and comparable age-depth relationships, the  $\delta D$  values will be plotted as a stacked record containing both columns' data.

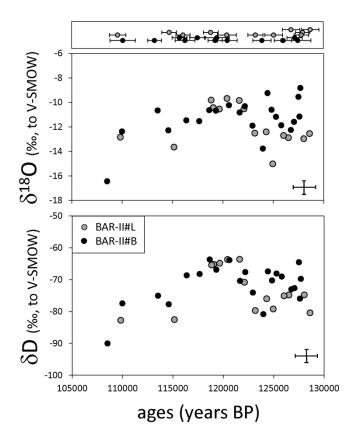


Fig. 5

# 5. Discussion

# 5.1. Suitability assessment of the BAR-II stalagmite

As was demonstrated by Demény et al. (2016b), the carbon and oxygen isotope compositions of the stalagmites grown under interglacial conditions of the Baradla Cave can reflect environmental parameters, but the suitability for paleoclimate analysis of the BAR-II stalagmite should also be investigated, *i.e.* if the stalagmite's calcite was precipitated under equilibrium conditions. There are two classical tests used to assess equilibrium precipitation: i) the Hendy test (see above), and ii) the comparison of compositions directly measured and calculated from temperature and water composition data.

As shown in Fig. 3, none of the laminae analysed for Hendy test (Hendy, 1971) showed coupled positive  $\delta^{13}$ C and  $\delta^{18}$ O<sub>carb</sub> shifts, hence degassing-related kinetic fractionation was negligible during the formation of the stalagmite. It is important to note that Hendy test analyses on recently forming stalagmites collected close to the sampling location of the BAR-II stalagmite showed no significant  $\delta^{18}$ O- $\delta^{13}$ C correlations, hence the site can be considered to be free of ventillation-related kinetic fractionation (Demény et al., 2016c).

Another important test for equilibrium is the comparison of measured and calculated compositions. Under conditions of equilibrium the oxygen isotope composition of the calcite is determined by temperature and the water oxygen isotope composition (O'Neil et al., 1969). The main problem is that many equations have been proposed for the calcite-water oxygen isotope fractionation (O'Neil et al., 1969; Friedman and O'Neil, 1977; Kim and O'Neil, 1997; Chacko and Deines, 2008; Coplen, 2007; Horita and Clayton, 2007; Dietzel et al., 2009; Demény et al., 2010; Zheng, 2011; Day and Henderson, 2011; Tremaine et al., 2011; Gabitov et al., 2012); these, however, provided different fractionation values for the same temperature. Two equations were selected as the Devils Hole location fulfils most of the theoretical requirements of equilibrium calcite precipitation (slow precipitation rate, stable environment, Coplen, 2007), while the equation established by Tremaine et al. (2011) is based on a collection of natural cave deposit data selected by avoiding kinetic fractionation bias. The present day temperature of the cave site (9.7  $\pm$  0.5 °C) and the dripwater composition (see section 2) yield calculated calcite compositions of -7.1 and -7.6% for the Coplen (2007) and Tremaine et al. (2011) equations, respectively (with standard deviation of 0.2‰ as a result of temperature and water composition variations). The BAR-II  $\delta^{18}O_{carb}$ values range from -7.7 to -6.4% in the fully developed interglacial period (124 to 120 ka),

which range partly overlaps the calculated compositions, suggesting that the BAR-II stalagmite was grown close to equilibrium conditions.

5.2. Interpretation of stable isotope compositions and variations in climate conditions on the basis of the BAR-II record

The carbon isotope signal of the Baradla Cave stalagmites can be basically interpreted as a proxy of soil activity depending on temperature and precipitation amount, while the oxygen isotope composition of the carbonate reflects mainly the combined effect of temperature, moisture source and seasonality changes (Demény et al., 2016b). The hydrogen isotope compositions of inclusion-hosted water depend on the same factors that determine the drip water compositions and the  $\delta^{18}O_{carb}$  values, only the effect of carbonate precipitation temperature can be excluded. Hence, the  $\delta D$  and  $\delta^{18}O_{carb}$  records are expected to have some shared patterns if the environmental parameters change. Temporal variations of  $\delta^{13}C$ ,  $\delta^{18}O_{carb}$  and  $\delta D$  records of the BAR-II stalagmite revealed interesting correspondences (Fig. 6).

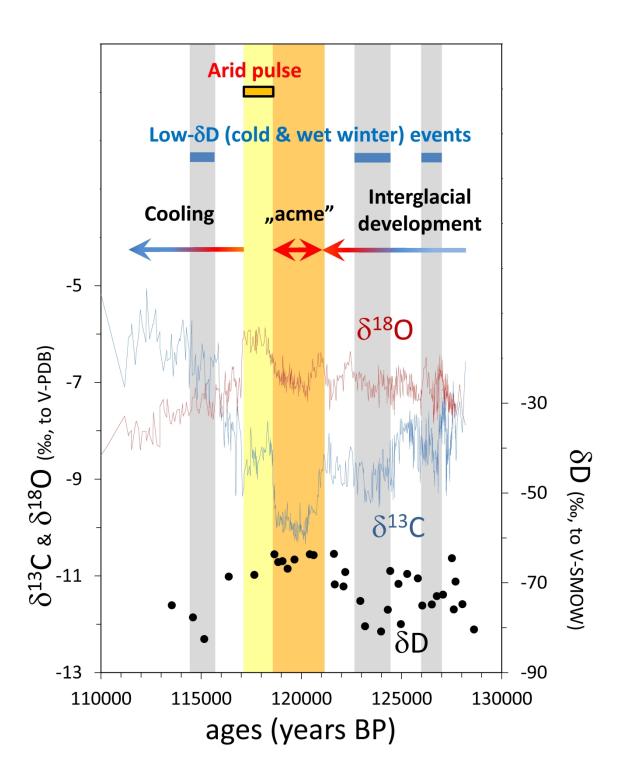


Fig. 6

The BAR-II stalagmite's formation started at around 129 ka, indicating that the climate ameliorated at that time to the temperature and humidity conditions appropriate for stalagmite formation at this karstic region. This date corresponds to the starting of maximum interglacial conditions and highest growth rates recorded by stalagmites in Europe (Drysdale et al., 2009; Genty et al., 2013). In the first millennium of the BAR-II stalagmite growth the  $\delta^{13}$ C and  $\delta^{18}$ O<sub>carb</sub> values decreased and increased by ~2.5 and 1.5%, respectively, then they continued the decrease/increase with fluctuations. This suggests that onset of the interglacial conditions was achieved around 128 ka, then the interglacial conditions was gradually developing. Climate optimum would mean warm conditions with elevated yearround precipitation amount that is most favourable for soil biogenic activity. Such conditions can be assumed for the period with the lowest  $\delta^{13}$ C (increased soil activity) and elevated  $\delta^{18}O_{carb}$  and  $\delta D$  values (warm atmospheric temperature). Such compositions are reached in the Baradla Cave's region around 121 ka, while the period between 127 and 121 ka can be regarded as the developing phase (Fig. 6). The δD values reached their maximum between 121.6 and 118.6 ka overlapping with the period of lowest  $\delta^{13}$ C values (120.7 to 118.7 ka). Following the nomenclature suggested by Govin et al. (2015), this period can be regarded as the "acme", meaning climate optimum within the interglacial. In the developing phase the δD values are strongly fluctuating from -65 down to -81‰. The low-δD events are associated with pronounced negative  $\delta^{13}$ C shifts and less expressed negative  $\delta^{18}$ O<sub>carb</sub> shifts. These data collectively suggest that these low- $\delta D$  periods were wet (hence the low  $\delta^{13}C$ ) and cold. The reason why the low  $\delta D$  values are not associated with similarly low  $\delta^{18}O_{carb}$  shifts can be found in the competition of dripwater composition change (driving  $\delta^{18}O_{carb}$  in negative direction) and formation temperature decrease (driving  $\delta^{18}O_{carb}$  in positive direction) that effects could partially cancel each other providing a great challenge to

environmental interpretation based solely on  $\delta^{18}O_{carb}$  despite it can be read rather frequently in the speleological literature. The negative  $\delta D$  shift of about 15% would correspond to about 5 °C decrease (see temperature calculations below), which is not realistic under full interglacial conditions. Seasonality change such as higher contribution of winter precipitation more depleted in heavy water isotopologues to the infiltrating dripwater (see also Meyer et al., 2008) can explain the observed  $\delta D$  shifts. This is elaborated in the paleotemperature calculations described below.

After the developing phase that was punctuated with cold, wet and winterdominated events, followed by a calm optimum period ("acme", Govin et al., 2015), the  $\delta^{13}$ C values suddenly rose by 2% at 119 ka, indicating a major climate deterioration event. The strong positive  $\delta^{13}$ C shift was coeval with  $\delta^{18}$ O<sub>carb</sub> rise, indicating either sudden warming and drying, or seasonality change to summer precipitation dominance. Global cooling had already started before (around 120 ka, Govin et al., 2015), hence warming is not expected (although cannot be excluded) in the global cooling phase. Increased aridity during cooling, however, can explain the elevation in  $\delta^{13}C$  and  $\delta^{18}O_{carb}$  values as it would lead to decreased soil activity and <sup>18</sup>O-enrichment in water due to evaporation. Reduced humidity can also result in drying of the fractures in the overlying rocks, where the migrating solution may experience carbonate precipitation ("Prior Calcite Precipitation", PCP). Evaporation in the water conduit system and associated PCP can also result in elevated  $\delta^{13}$ C and  $\delta^{18}$ O values in the solution and hence in the precipitating speleothem carbonate (Fairchild and Baker, 2012), contributing to the observed isotope shifts. The δD values remained still close to the "acme" compositions contrary to the effect of global cooling, supporting the assumption of increased aridity as evaporation could also cause D-enrichment in the infiltrating water. It is important to note, that a strong aridity pulse was detected at about 118.5 ka on the basis of

greyscale analyses of Eifel maar lake sediments (Sirocko et al. 2005), overlapping the short period of the sudden positive  $\delta^{13}$ C shift. Finally, the isotopic compositions follow the global cooling trend from 117 ka with increasing  $\delta^{13}$ C and decreasing  $\delta$ D and  $\delta^{18}$ O<sub>carb</sub> values.

Opposite to the carbonate's carbon and oxygen isotope compositions, the advantage of the  $\delta D$  data is that they can be used to infer quantitative formation temperatures, provided that the quantified temperature-composition relation is known for the region's precipitation, and it did not change over time. An approximately three year long sampling campaign (2013-2015) has been conducted at the cave site to determine the H and O isotope compositions of precipitation (rain and snow) and their relationships with meteorological conditions (Czuppon et al., in prep) that yielded a 2.1% °C<sup>-1</sup> gradient for the δD values and surface air temperature (stable H and O isotope compositions of the precipitation waters and their relationships with the local air temperature are shown in Supplementary Fig. 1). This gradient can be translated to a 0.488 °C shift for every ‰ change in the  $\delta D$  value. Hence, higher  $\delta D$  values compared to the present day dripwater composition would correspond to elevated mean annual temperature at the surface during the last interglacial, and vice versa. The present day gradient may be used for the "acme" (Govin et al., 2015) period, but cooler periods may have been characterized by higher isotope-temperature gradients. The climate conditions of the high-altitude regions of the Alps or the Carpathians north of the Baradla Cave can provide present day analogues where elevated composition-temperature gradients have been reported (Holko et al., 2012; Salamalikis et al., 2016). The hydrogen isotope compositions of precipitation waters can reach a 3.5% °C<sup>-1</sup> gradient in the Carpathians (Holko et al., 2012) that can be chosen as a cold-climate related gradient value. The present day dripwater composition obtained for the Baradla Cave ( $\delta D = -66 \pm 1.5\%$ ) can be used as a reference value. First the difference of the

measured  $\delta D$  values from this reference composition was calculated, then the  $\delta D$  difference was multiplied by the temperature effect value of 0.488 °C‰<sup>-1</sup> to obtain the deviation from the present day annual mean temperature. Finally, the temperature deviation value was added to the present day cave site temperature of 9.7 °C to obtain the annual mean temperature that corresponds to the  $\delta D$  value measured on inclusion-hosted water. The main sources of bias in this calculation are the analytical error of the  $\delta D$  analysis, the uncertainty in temperature-composition gradient and the possible changes in the Local Meteoric Water Line (i.e., the verification of the use of GMWL equation). In order to estimate the uncertainty of temperature values, the calculations were conducted for  $\delta D$ temperature gradients of 2.1 and 3.5%  $^{\circ}$ C<sup>-1</sup> and a 2% error in the  $\delta$ D values. The results are shown in Fig. 7 along with sea surface temperature (SST) data reported by Sánchez-Goñi et al. (2012) for the Bay of Biscay. In order to reduce the noise caused by high-frequency δD variations and sample inhomogeneity effects, the  $\delta D$  values were averaged for every millennium and the average  $\delta D$  values were used in the calculations. The highest temperature obtained in this calculation is 11.2 ± 1.4 °C for 120.5 ± 0.5 ka, whereas the lowest temperature value found for 108.5 ka is 3.2 ± 0.6 °C (temperatures below zero were excluded as freezing would stop stalagmite formation). The inferred ~8 °C drop from peak interglacial conditions to the beginning of glacial times is in very good agreement with the independent estimate of mean annual air temperature (MAAT) difference of  $10 \pm 2$  °C between the last glacial maximum and present day conditions based on a multiproxy evaluation of regional paleotemperatures in the Pannonian lowlands (Ruszkiczay-Rüdiger and Kern, 2016). Additionally, the calculated paleotemperatures closely follow the pattern of the SST variations of Sánchez-Goñi et al. (2012), indicating that the  $\delta D$  values record meaningful paleoclimate information.

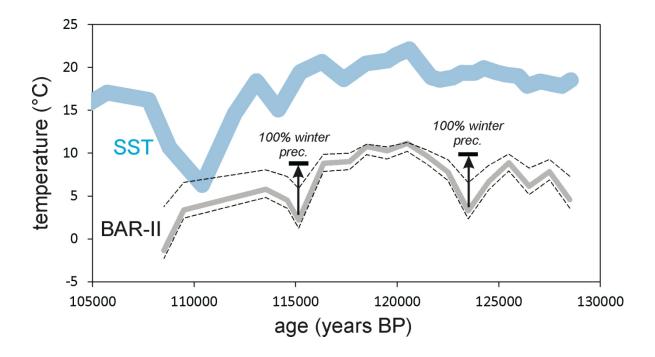


Fig. 7

This calculation procedure would yield an approximately 5 °C decrease for the low- $\delta D$  event at 124 ka, which is unrealistic. However, seasonality change can explain the observed D-depletion. Raising the winter precipitation contribution to the infiltrating water would result in decreased dripwater composition. Using -80% as a Holocene reference composition that corresponds to the average winter precipitation composition (Czuppon et al., 2016, in prep.), the temperature decrease inferred for the 124 ka peak would totally disappear. This situation is shown by the arrows in Fig. 7 that point to temperature values obtained using the  $\delta D_w$ = -80% reference value in the calculation. Both the 5 °C temperature decrease and the 100% winter precipitation contribution represent extreme cases that provide the boundary conditions, a realistic assumption can be characterized by intermediate parameters. These considerations imply that a slight cooling coupled with a rise in winter precipitation contribution to the karstic water can provide an explanation for the low  $\delta D$  values . This speculation can be strengthened if other records showing similar

seasonality changes are found. Such paleoclimate records were gathered and compared with the BAR-II record as shown in the following sections.

#### 5.3. Carbon and oxygen isotope variations – comparisons with LIG speleothems

Fewer than a dozen speleothem C and O isotope records represent at least a part of the LIG in Europe (Linge et al., 2001; Holzkämper et al., 2004; Drysdale et al., 2005; Muñoz-García et al., 2007; Meyer et al., 2008; Couchoud et al 2009; Boch et al., 2011; Genty et al., 2013; Regattieri et al., 2014; Moseley et al., 2015), and only a few of them cover more than 10 ka including peak interglacial conditions as well as the glacial inception (Drysdale et al., 2005; Meyer et al., 2008; Drysdale et al., 2009; Moseley et al., 2015; Vansteenberge et al., 2016). In this section the BAR-II data will be compared with eleven stalagmite records from Europe and the Levant (Fig. 8).

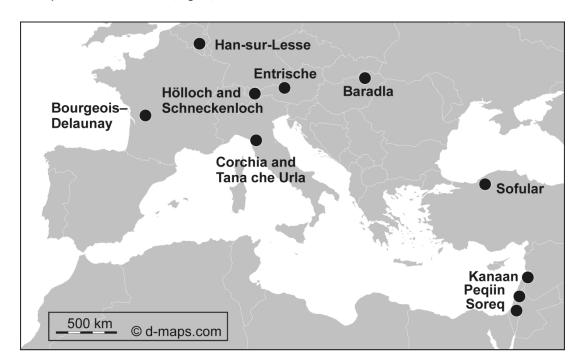


Fig. 8

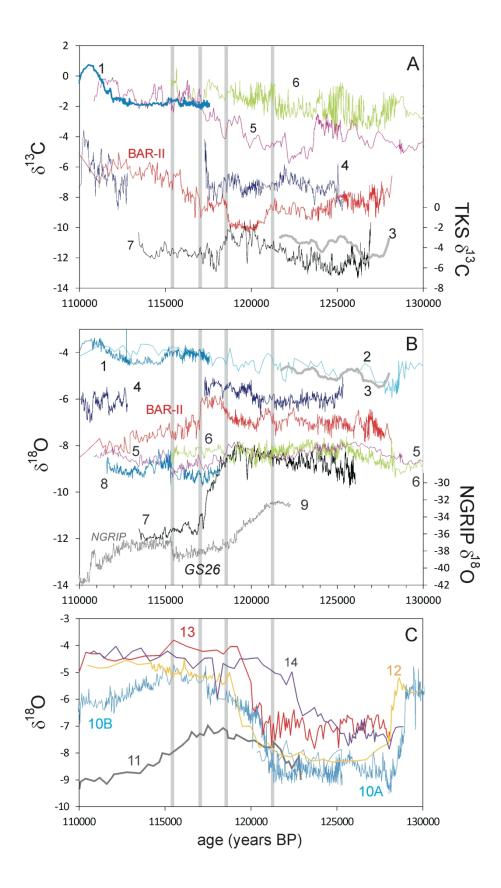


Fig. 9.

Since the formation of the BAR-II stalagmite started after about 129 ka, records extending before 130 ka were truncated to keep the focus on their relation to the Baradla records. The paleoclimatic interpretation of isotope records and their comparison depends on the precisions of age determinations of the individual speleothems. On the basis of the speleothem dating precisions (Supplementary Table 4) the chronology of climate variations will be discussed in a ka scale.

The  $\delta^{13}$ C fluctuations (Fig. 9A) show different overall patterns in the stalagmite records, and this is in accordance with earlier conclusions (Meyer et al., 2008; Moseley et al., 2015) concerning the difficulties with the interpretation of the carbon isotope compositions. From 128-129 ka, the  $\delta^{18}$ O<sub>carb</sub> records show a continuous increase, with similar slopes until about 119 ka (Fig. 9B). The stalagmite from the Han-sur-Lesse cave (Belgium, Vansteenberge et al., 2016) represents an exception as its  $\delta^{18}$ O values slightly decrease from ~125 ka until about 121 and then rise abruptly at 120 ka. The coupled  $\delta^{13}$ C- $\delta^{18}$ O<sub>carb</sub> patterns suggest that the climate optimum (warm and humid conditions) was attained in the period of 129-119 ka, when  $\delta^{18}$ O<sub>carb</sub> values were increasing and  $\delta^{13}$ C values were relatively low, indicating a warm and humid climate. This is also valid for the HÖL-10 (Moseley et al., 2015) Alpine stalagmite record, whereas the TKS  $\delta^{13}$ C data (Meyer et al., 2008) display an opposite pattern. The increase in  $\delta^{13}$ C values in the TKS record during the period of  $\delta^{13}$ C minimum in the BAR-II stalagmite indicates humidity differences affecting soil activity and/or kinetic isotope fractionation processes at the two locations.

A marked change in the caves' behavior appeared around 119.5  $\pm$  0.5 ka. The Hansur-Lesse stalagmite is characterized by a  $\delta^{18}O_{carb}$  rise of ~0.6‰ at about 120 ka, which was related to a shift in the moisture source due to the starting growth of polar ice sheets (Vansteenberge et al., 2016). The Entrische cave's TKS record (Meyer et al., 2008) shows a

decrease in  $\delta^{18} O_{\text{carb}}$  values, which is not reproduced with a comparable amplitude in other Alpine records, although the  $\delta^{18}O_{carb}$  values slightly decrease in the Schneckenloch and the Hölloch caves' stalagmites (Boch et al., 2011; Moseley et al., 2015). The Corchia  $\delta^{18}O_{carb}$ records (Drysdale et al., 2005; 2007) show a slight  $\delta^{18}O_{carb}$  increase between 118 and 115 ka on the top of the continuous <sup>18</sup>O enrichment, accompanied by small positive  $\delta^{13}$ C peaks at the same time. With regard to oxygen isotope compositions, the BAR-II stalagmite displays opposite shifts to the Alpine stalagmites after 120 ky (within dating uncertainties). The 119-115 ka changes correlate with the Greenland Stadial GS26 in the NGRIP  $\delta^{18}$ O record (North Greenland Ice Core Project members, 2004; using the GICC05 age scale of Wolff et al., 2010) (Fig. 9B). Almost all of the stalagmite isotope records (except the TKS data, Meyer et al., 2008) show positive  $\delta^{13}$ C shifts to different degrees that, taken together with the global cooling reflected by polar ice core data (Figs. 9A and B) indicate soil degradation due to deteriorating environmental conditions. As the North Atlantic Current is suppressed from the north (Müller and Kukla, 2004), less heat and moisture might reach the Alpine cave sites, leading to colder and more arid conditions, hence to lower  $\delta^{18}$ O and higher  $\delta^{13}$ C values in the precipitating carbonate. The strong shift in the Entrische cave's  $\delta^{18}O_{carb}$  record has been interpreted as a result of changes in seasonality (stronger winter precipitation contribution, Meyer et al., 2008) and/or the fact that the catchment was already affected by glaciation with an increased glacier melt water contribution (Moseley et al., 2015). The GS26 stadial seems to appear in various speleothem records in the territory of the Asian monsoon (Kelly et al., 2006), in the Levant (Bar-Matthews et al., 2003) and on the Black Sea coast (Zumbühl, 2010) (Fig. 9C). These records indicate lower rainfall amount during the GS26 stadial. It is interesting to note that the BAR-II  $\delta^{18}O_{carb}$  record indicates a strongly arid pulse in the same period that points again to the decisive connection between the Mediterranean moisture

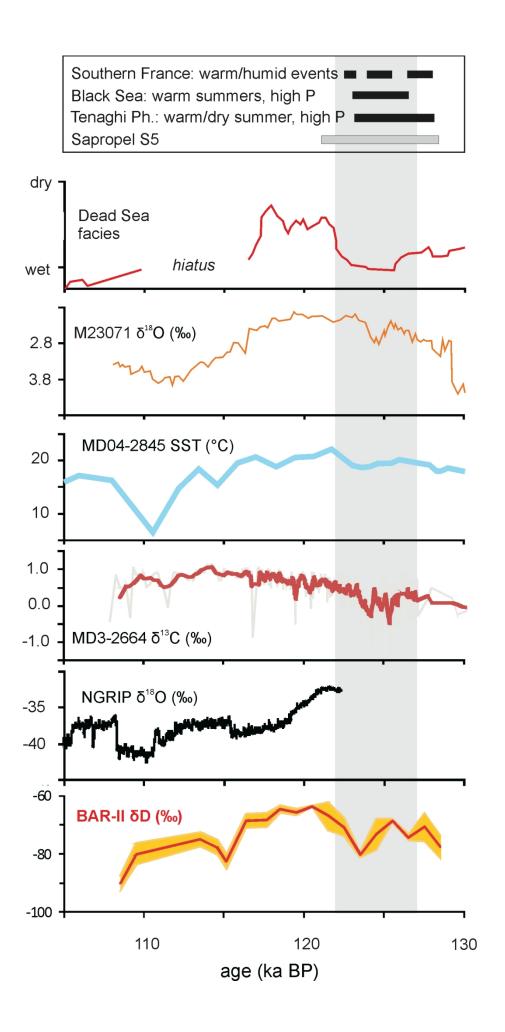
availability and the Carpathian Regions's hydroclimate. The period of elevated  $\delta^{18}$ O values ends with a decrease at about 117 ka that occurred contemporaneously with the growth stop of the Han-sur-Lesse stalagmite (Vansteenberge et al., 2016) and the steep  $\delta^{18}$ O decrease in the TKS record (Meyer et al., 2008), suggesting different responses in the regional temperature and humidity changes in Western and Central Europe to the large-scale event triggering the GS26 cooling at Greenland.

With the exception of the TKS stalagmite, all the records display a concerted change at the end of the GS26 stadial, at 116 ka (Fig. 9A-C). This abrupt change may be related to an approximately 5 °C warming in Greenland (Kindler et al., 2014), which can be detected in small positive  $\delta^{18}O_{carb}$  peaks in the BAR-II and the Schneckenloch and the Hölloch caves' records (Boch et al., 2011; Moseley et al., 2015). This peak is, however, negligible compared to the sudden  $\delta^{18}O_{carb}$  decrease in the BAR-II stalagmite at 117 ka coupled with a rise in  $\delta^{13}C$ . This may reflect the appearance of Atlantic moisture in the region of the Baradla Cave, and hence the BAR-II  $\delta^{18}O_{carb}$  pattern returns to the long-term global trend. Finally, the BAR-II record follows the global cooling that lead to glacial inception and stalagmite formation ceased at about 108 ka. Due to the more elevated position, the Alpine sites would be more sensitive to glaciation, thus stalagmite formation ceased about 1 ka earlier (Fig. 9B), although it might result from dating uncertainty, too.

# 5.4 The hydrogen isotope record - global and regional relationships

The pattern of hydrogen isotope compositions of fluid inclusion hosted water is similar to the records representing regional and global climate conditions (Fig. 10) such as the Greenland ice core  $\delta^{18}$ O data (North Greenland Ice Core Project members, 2004; using the GICC05 age scale of Wolff et al., 2010) and selected stable isotope compositions and sea

surface temperature reconstructions derived from North Atlantic deep sea sediment cores (Bauch et al., 2011; Sánchez-Goñi et al., 2012; Galaasen et al., 2014). It should be noted here that dating of deep sea records is a challenging task (e.g., Drysdale et al., 2009), and the North Atlantic region may behave heterogeneously with respect to temperature anomalies during climate change events (Capron et al., 2014) that must be kept in mind during the evaluation. The selected records demonstrate that the period of low  $\delta D$  values in the BAR-II record was associated with climate changes in the wider region. Nevertheless, the good match with these records and the rather complex relationships in the C-O isotope data of speleothems may suggest that the  $\delta D$  record is less influenced by local, cave-related effects compared to the C-O isotope values.



One of the most prominent features in the  $\delta D$  pattern is the low  $\delta D$  peaks detected between 127 and 123 ka. The very large  $\delta D$  changes (up to 17% within 2.5 ka, see Fig. 5) during an interglacial period are unusual and require additional evidence that prove their primary nature (not related to the alteration processes or analytical procedure). First it should be noted that the low  $\delta D$  values were obtained in different analytical runs and from two stalagmite columns that reduces the chance of analytical bias. Second, the low δD values are associated with decreased  $\delta^{13}$ C values in the BAR-II stalagmite (Fig. 6) that excludes the possibility of procedure-related isotope shifts. As discussed above, low  $\delta^{13}$ C values indicate increased humidity, and the same (with relative increase of winter precipitation) is assumed for the low  $\delta D$  values (see above). Regional humidity proxies are provided for the period by the carbon isotope compositions of the Han-9 stalagmite (Hansur-Lesse Cave, Vansteenberge et al., 2016) and the BDinf stalagmite (Bourgeois-Delaunay Cave, Couchoud et al., 2009) that can serve as independent test records. Fig. 11 shows a comparison of these records, indicating that the low  $\delta D$  values in the BAR-II stalagmite are indeed associated with low  $\delta^{13}$ C values and higher humidity in the period of 127-123 ka. The magnitude of the  $\delta D$  fluctuation is related to local fractionation processes, but the timing and directions of isotope shifts can be correlated in a regional scale.

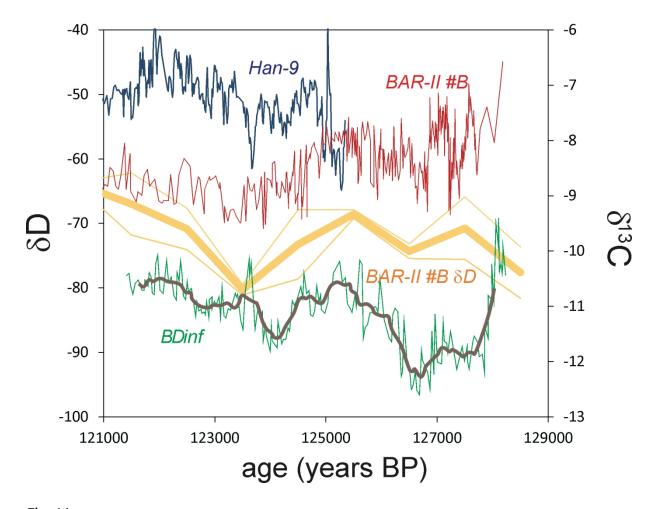


Fig. 11

The low  $\delta D$  period fits several paleoclimate records from the Mediterranean to the North Atlantic regions (Fig. 10). At the same time the Dead Sea had elevated water levels (Torfstein et al., 2015), in agreement with the humid climate inferred for the eastern Mediterranean and increased Nile floods, leading to the formation of sapropel S5 (Grant et al., 2012). For the period of 128 to 123 ka Milner et al. (2012) have suggested pronounced seasonality with increased summer aridity and winter precipitation on the basis of acicular aragonite formation (indicating enhanced evaporation) in an otherwise aquatic phase of the Tenaghi Philippon peatland (NE Greece). The inferred precipitation seasonality is in agreement with the assumptions of increased winter precipitation around the Baradla Cave based on paleotemperature calculations (see section 5.2.). On the basis of pollen data from a

Black Sea sediment core, Shumilovskikh et al. (2013) have also come to the same conclusion, assuming warm summers and mild winters with increased annual precipitation for the period of 126.4 to 122.9 ka. In a sediment core from the Western Mediterranean Sea, nalkane based temperature estimates showed a maximum at about 125 ka (Kandiano et al., 2014), suggesting that the humid period was accompanied with a warm event in this area. In the C and O isotope data from a stalagmite from Southern France, Couchoud et al. (2009) detected fluctuations of warm/wet and cold/dry periods, with increased rainfall amount at 127.9–126.3, 125.3–123.8 and 123.1–122.4 ka, as shown also in Fig. 11. In summary, South European proxy data indicate that the 128-123 ka period represented by these records was characterized by warm conditions (especially during summer) and elevated annual precipitation.

Beside the Mediterranean data, several North Atlantic deep-sea isotope records from further afield display interesting similarities to the BAR-II δD pattern. As also shown in Fig. 10, stable isotope peaks obtained for planctonic foraminifers at 125-124 ka appear in several North Atlantic deep sea cores around Greenland (Bauch et al., 2011, Van Nieuwenhove et al., 2011; Irvali et al., 2012; Galaasen et al., 2014), and may be interpreted as a result of a Younger Dryas type event (Bauch, 2013). The event is generally explained as resulting from a freshwater influx from melting ice sheets that induced a reduction in the Atlantic Meridional Overturning Circulation (AMOC) and associated cooling in the northernmost parts of the Atlantic realm. However, the event was not homogeneous in the entire North Atlantic as shown by the temperature anomaly distributions compiled by Capron et al. (2014).

In contrast to the Mediterranean data, certain European paleoclimate proxies indicate cooling and precipitation changes during the early-mid LIG. As shown in Fig. 11 in detail, the precisely (better than  $\pm$  700 years) dated  $\delta^{13}$ C record of a stalagmite from the

Han-sur-Lesse cave (Belgium) indicates humidity fluctuations during the period of 128 to 121 ka (Vansteenberge et al., 2016). Slight temperature decrease at about 126 and 122 ka has been detected in a sea surface temperature record at the Bay of Biscay by Sánchez-Goñi et al. (2012) that fit the low-δD peaks (Fig. 10) within dating uncertainties.

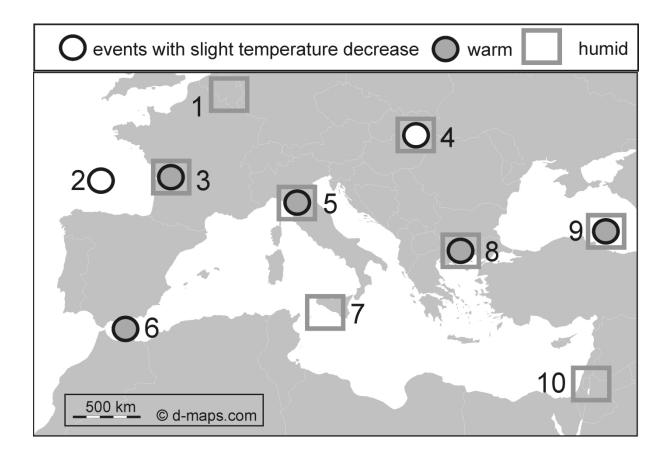


Fig. 12

In order to investigate the spatial differences in climate conditions, proxy information gathered from those records that detected a climate change event at about 126-122 ka at a sufficient resolution (~200 yr/sample) to resolve this event, were plotted on a map (Fig. 12).

The spatial distribution confirms the impression that most of the European continent was characterized by elevated humidity, while the temperature conditions were different

with significant warming in the Mediterranean. A new climate model simulation for the last interglacial period of the Atlantic-Mediterranean region was conducted including freshwater flux to the northernmost Atlantic Ocean (Stone et al., 2016). Interestingly, the simulation experiment resulted in a similar picture (see Fig. 7 of Stone et al., 2016) to that shown in Fig. 12, suggesting that freshwater flux to the Northern Atlantic changed moisture and heat transport trajectories.

#### 6. Synthesis and conclusions

Stable carbon and oxygen isotope compositions of carbonate, and hydrogen and oxygen isotope compositions of inclusion-hosted water were obtained for a stalagmite covering a large part of the last interglacial (129 to 109 ka) from the Baradla Cave (NE Hungary). The stable isotope records were correlated with various coeval regional proxy and paleoclimate records, including stable isotope data of speleothems, ice cores, deep sea cores, and sea surface temperature values. The comparison revealed a series of events that characterized the complex climate evolution of the European continent and the eastern Mediterranean.

The climate reached full interglacial conditions at the Baradla Cave at about 128 ka, when the BAR-II stalagmite formation commenced. The C and O isotope compositions indicate an approach to the climate optimum that was attained around 124 ka and lasted until about 120 ka. However, the BAR-II  $\delta D$  pattern displays negative  $\delta D$  peaks between 127 and 123 ka, which is associated with negative  $\delta^{13}C$  shifts, but not reflected in the  $\delta^{18}O_{carb}$  values, probably due to competing fractionation effects. The low  $\delta D$  period can be correlated with climate change events detected in paleoclimate records ranging from  $\delta^{18}O$  and SST records from deep sea cores from the North Atlantic to paleobiological and

sedimentological records from marine and continental sediments from Western Europe to the Eastern Mediterranean. Paleotemperature and -moisture information was gathered from proxy records of Western and Central Europe and the Mediterranean region that cover the period of interest with appropriate age constraints and resolution. A systematic spatial pattern was revealed with overall increased humidity from NW to SE, and significant warming in the Mediterranean. Stable isotope records of several North Atlantic deep sea cores and climate simulations suggest that the humidity and temperature changes are related to freshwater influx from melting ice sheets at about 125 ka (e.g., Bauch, 2013, Stone et al., 2016), with this perturbation being responsible for shifting heat and moisture transport towards the Mediterranean. The event is also associated with increased humidity and seasonality (elevated winter/summer precipitation ratio) in East-Central Europe as documented at the Baradla Cave in accordance with observations from the eastern Mediterranean and the Black Sea region.

After the climate optimum between 124 and 120 ka (the Baradla Cave "acme", Govin et al., 2015), global cooling processes induced changes in the Western-Central European and Mediterranean regions. Strong cooling started in the Alps at about 119 ka, while the Baradla  $\delta^{13}$ C values are the lowest and the  $\delta D$  record shows a maximum, suggesting increased summer precipitation in still favorable temperature conditions for soil activity. These mild conditions ended at 118 ka, when the Alpine stalagmite records indicate a strong cooling in concert with the onset of the Greenland stadial GS26. In the course of the Greenland stadial GS26 the North Atlantic Current was shifted to the south, resulting in cooler and drier conditions, while the Baradla Cave experienced stronger evaporation effects. These conditions changed as reflected by sharp  $\delta^{13}$ C and  $\delta^{18}$ O shifts in the BAR-II record at 117 ka, still within the GS26 event. The sudden negative  $\delta^{18}$ O shift at 117 ka in the

BAR-II stalagmite likely indicates the appearance of Atlantic moisture and the effect of global cooling. The GS26 event ended with a sudden warming, that appears as slight  $\delta^{18}$ O shifts in the Alpine stalagmites and in the Corchia and the Baradla caves. Finally, global cooling was reflected by continuous negative shifts in the BAR-II  $\delta^{18}$ O<sub>carb</sub> and  $\delta$ D values until the glacial inception at 109 ka.

The complex set of climate change events can be related to the north-south movements of the North Atlantic Current combined with global temperature trends. Although C and O isotope compositions reflected most of the events and proved their synchronicity, the competing fractionation effects may partially attenuate the climate signal in speleothem proxy records. The BAR-II  $\delta D$  record not only helped interpret the seasonality signal in the  $\delta^{13}C-\delta^{18}O_{carb}$  record, but was also successfully used to detect a significant midinterglacial climate event at 125 ka.

#### **Acknowledgements**

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### Figure captions

- Fig. 1. Cave site locations and pictures of stalagmites collected from the Baradla Cave, NE Hungary. Numbers in circles within the BAR-II#B and L stalagmites denote hiatuses.
- Fig. 2. Age-depth relationships for the BAR-II#L and BAR-II#B stalagmite columns. U-Th ages and their errors are listed in Supplementary Table 1. Model curves and their uncertainties were established using the StalAge algorithm of Scholz and Hoffmann (2011), except the youngest part of the BAR-II#L stalagmite column, where a linear age-depth model was applied (marked by dashed line).
- Fig. 3. Stable carbon and oxygen isotope compositions (in‰, relative to V-PDB) of the BAR-II#B stalagmite along six laminae (Hendy test, Hendy, 1971).
- Fig. 4. Stable carbon and oxygen isotope compositions of the two columns (BAR-II#L and BAR-II#B) of the BAR-II stalagmite as a function of age. Positions of age dates and their errors are shown in the uppermost panel. Textural types are plotted as a function of age in the middle panel. Texture: #1: finely laminated texture in which the lamination is expressed by whitish, inclusion-rich layers, #2: white, cloudy, inclusion-rich texture, #3: translucent calcite with fine lamination, #4: pale whitish, cloudy texture with weak lamination.
- Fig. 5. Stable hydrogen and oxygen isotope compositions of inclusion-hosted water as a function of age. Positions of age dates and their errors are shown in the uppermost panel.
- Fig. 6. Combined plot of stable carbon and oxygen isotope compositions of stalagmite carbonate (BAR-II#B stalagmite column) and hydrogen isotope compositions of inclusion-

hosted water. Arrows indicate the periods of interglacial development, "acme" ( $\delta^{13}$ C minimum and  $\delta^{18}O_{carb}$  maximum) and global cooling following Govin et al. (2015). Low  $\delta D$  periods are marked as cold and wet events, while the positive  $\delta^{18}O_{carb}$  peak is interpreted as an arid period.

Fig. 7. Paleotemperatures calculated from stable hydrogen isotope compositions (averaged for millennium periods) of fluid inclusion hosted water of the BAR-II stalagmite. Dashed lines indicate the accompanied uncertainty range obtained by error propagation using different isotope-temperature gradients and analytical scatter (see text). Arrows indicate the shift of calculated temperature (marked by horizontal bars) if a reference water hydrogen isotope composition of –80 ‰ is used in the calculation. Thick blue line shows sea surface temperature (SST) variations reported by Sánchez-Goñi et al. (2012) for the Bay of Biscay.

Fig. 8. Locations of caves where stalagmite  $\delta^{13}C$  and  $\delta^{18}O_{carb}$  records were gathered for the last interglacial. Map is from d-maps.com.

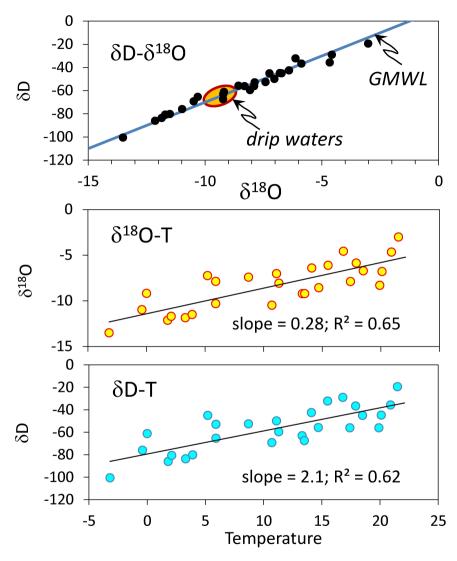
Fig. 9. Stable carbon and oxygen isotope compositions (in‰ relative to V-PDB) of stalagmite and ice core records. BAR-II: this study. 1: Corchia cave, CC28, Drysdale et al. (2007); 2: Corchia Cave, CC5, Drysdale et al. (2005); 3: Bourgeois–Delaunay Cave, Bdinf, Couchoud et al. (2009); 4: Han-sur-Lesse Cave, Han-9, Vansteenberge et al. (2016); 5: Hölloch Cave, HÖL-10, Moseley et al. (2015); 6: Schneckenloch Cave, SCH-5, Moseley et al. (2015); 7: Entrische Cave, TKS, Meyer et al. (2008), 8: Schneckenloch Cave, Boch et al. (2011), 9: North Greenland Ice Core Project members (2004) using the GICC05 age scale (Wolff et al., 2010), 10A and 10B: Dongge Cave, Kelly et al. (2006), 11: Sofular Cave, Zumbühl (2010), 12 and 13: Soreq

and Peqiin Caves, Bar-Matthews et al. (2003), 14: Kanaan Cave, Nehme et al. (2015). Vertical grey bars mark the onsets and ends of  $\delta^{13}$ C minimum and  $\delta^{18}$ O maximum in the BAR-II stalagmite, and the end of the GS26 event.

Fig. 10. Stable hydrogen isotope compositions of inclusion-hosted water from the BAR-II stalagmite (lowermost panel) and various proxy records from the Atlantic and Mediterranean regions. The BAR-II  $\delta$ D values are millennial averages. NGRIP: North Greenland Ice Core Project members (2004), using the GICC05 age scale (Wolff et al., 2010). MD3-2664  $\delta$ <sup>13</sup>C: North Atlantic drilling site, south of Greenland, Galaasen et al. (2014). MD04-2845 SST: Biscay Bay, Sánchez-Goñi et al. (2012). M23071  $\delta$ <sup>18</sup>O: Norwegian Sea, Bauch et al. (2011). Dead Sea sedimentary facies distribution: Torfstein et al. (2015). Sapropel S5: Grant et al. (2012). Tenaghi Philippon: Milner et al. (2012). Black Sea: Shumilovskikh et al. (2013). Southern France: Couchoud et al. (2009).

Fig. 11. Millennial averages of fluid inclusion  $\delta D$  values (in % relative to V-SMOW, with  $1\sigma$  standard deviations shown by thin lines) and  $\delta^{13}C$  records (in % relative to V-PDB) of the Han-9 (Vansteenberge et al., 2016), the BDinf (Couchoud et al., 2009) and the BAR-II (this study) stalagmites.

Fig. 12. Temperature and humidity information in last interglacial proxy records between 126 and 122 ka. 1: Vansteenberge et al. (2016). 2: Sánchez-Goñi et al. (2012). 3: Couchoud et al. (2009). 4: BAR-II (this study). 5: Drysdale et al. (2007) and Regattieri et al. (2014). 6: Kandiano et al. (2014). 7: Censi et al. (2010). 8: Milner et al. (2012). 9: Shumilovskikh et al. (2013). 10: Torfstein et al. (2015).



**Fig. S1.** Stable H and O isotope compositions (in ‰ relative to V-SMOW) of precipitation waters and drip waters, and surface air temperature (°C) at the Baradla Cave. GMWL: Global Meteoric Water Line (Craig, 1961).

Supplementary Table 1. U-Th isotopic compositions and  $^{230}$ Th ages for stalagmites Distances are below hiatus #1 within the stalagmite columns BAR-II#B and BAR-II#L

chemistry	sample	distance	<sup>238</sup> U		232
date	number	cm	ppb		pŗ
BAR-II#B stalagmite	column				
June, 2007	BAR-II#B-1	26, 2	202, 79	$\pm$ 0, 22	7091, 9
June, 2007	BAR-II#B-2	23, 0	295, 07	$\pm$ 0,43	394, 3
June, 2007	BAR-II#B-3	20, 3	180, 93	$\pm$ 0,31	95, 8
June, 2007	BAR-II#B-4	16, 5	114, 82	$\pm$ 0, 17	107, 4
December, 2015	BAR-II#B2015-5	13, 8	61, 27	$\pm$ 0, 11	137, 2
June, 2007	BAR-II#B-5	12, 3	61, 77	$\pm$ 0,07	465, 1
December, 2015	BAR-II#B2015-4	10, 8	50, 56	$\pm$ 0, 11	1677, 8
June, 2007	BAR-II#B-6	7, 5	61, 72	$\pm$ 0,06	681, 7
December, 2015	BAR-II#B2015-3	6, 3	42, 10	$\pm$ 0, 10	205, 8
December, 2015	BAR-II#B2015-2	4,8	40, 83	$\pm$ 0,07	162, 4
December, 2015	BAR-II#B2015-1	3, 2	40, 43	$\pm$ 0,04	225, 8
June, 2007	BAR-II#B-7	1, 5	23, 87	$\pm$ 0,04	1062, 5
BAR-II#L stalagmite	column				
December, 2015	BAR-II#L2015-16	23, 3	101, 30	$\pm$ 0, 15	374, 1
December, 2015	BAR-II#L2015-15	21, 9	225, 26	$\pm$ 0,31	439, 4
December, 2015	BAR-II#L2015-14	20, 0	306, 16	$\pm$ 0,43	397, 8
December, 2015	BAR-II#L2015-13	17, 7	196, 11	$\pm$ 0,36	685, 3
December, 2015	BAR-II#L2015-12	16, 0	150, 74	$\pm$ 0, 24	655, 2
December, 2015	BAR-II#L2015-11	13, 5	101, 12	$\pm$ 0, 15	620, 5
December, 2015	BAR-II#L2015-10	10, 7	62, 25	$\pm$ 0,09	379, 8
December, 2015	BAR-II#L2015-9	8, 2	64, 09	$\pm$ 0,09	415, 3
December, 2015	BAR-II#L2015-8	5, 7	42,67	$\pm$ 0,05	353, 7
December, 2015	BAR-II#L2015-7	3, 3	41, 25	$\pm$ 0,05	293, 1
December, 2015	BAR-II#L2015-6	0, 3	30, 94	$\pm$ 0,05	521, 9

Chemistry was performed in two analytical runs (Shen et al., 2003), instrumental analysis was done on MC-ICP-MS (Shen Analytical errors are  $2\sigma$  of the mean.

 $<sup>{}^{</sup>a}[{}^{238}U] = [{}^{235}U] \times 137.818 \ (\pm 0.65\%) \ (Hiess \ et \ al., \ 2012); \\ \delta^{234}U = ([{}^{234}U/{}^{238}U]_{activity} - 1) \times 1000.$ 

 $<sup>^</sup>b\delta^{234}$ U<sub>initial</sub> corrected was calculated based on  $^{230}$ Th age (T), i.e.,  $\delta^{234}$ U<sub>initial</sub> =  $\delta^{234}$ U<sub>measured</sub> X e  $^{\lambda_{234}*T}$ , and T is corrected age.  $^c$ [ $^{230}$ Th/ $^{238}$ U]<sub>activity</sub> = 1 - e  $^{-\lambda_{230}T}$  + ( $\delta^{234}$ U<sub>measured</sub>/ $^{1000}$ )[ $\lambda_{230}$ /( $\lambda_{230}$  -  $\lambda_{234}$ )](1 - e  $^{-(\lambda_{230} - \lambda_{234})T}$ ), where T is the age.

Decay constants are  $9.1705 \times 10^{-6} \text{ yr}^{-1}$  for  $^{230}\text{Th}$ ,  $2.8221 \times 10^{-6} \text{ yr}^{-1}$  for  $^{234}\text{U}$  (Cheng et al., 2013), and  $1.55125 \times 10^{-10} \text{ yr}^{-1}$  for  $^{d}$ The degree of detrital  $^{230}\text{Th}$  contamination is indicated by the [ $^{230}\text{Th}/^{232}\text{Th}$ ] atomic ratio instead of the activity ratio.

 $<sup>^{</sup>e}$ Age, relative to chemistry date, corrections for samples were calculated using an estimated atomic  $^{230}$ Th/ $^{232}$ Th ratio of 4: Those are the values for a material at secular equilibrium, with the crustal  $^{232}$ Th/ $^{238}$ U value of 3.8. The errors are arbitraril

of Hungarian Academy for Sciences, Hungary, on MC-ICP-MS at the Dept Geoscien , (see Fig. 1) .

Th	δ <sup>234</sup> U		[ <sup>230</sup> Th/	<sup>/238</sup> U]	[ <sup>230</sup> Th/ <sup>23</sup>	<sup>32</sup> Th]	Ag	
pt	measure	ed <sup>a</sup>	activ	ity <sup>c</sup>	$\mathbf{ppm}^{d}$		uncorr	
	,	,				,		
$\pm 24,8$	270, 7	$\pm$ 1,9	0,9070	$\pm$ 0,0047	428, 2	$\pm$ 2,6	127 815	
$\pm$ 4,0	248, 2	$\pm$ 2, 3	0,8865	$\pm$ 0,0023	10953, 8	$\pm$ 113	127 175	
$\pm$ 3, 4	249, 1	$\pm$ 2, 5	0,8826	$\pm$ 0,0027	27532, 6	$\pm$ 970	125 993	
$\pm$ 3, 3	250, 8	$\pm$ 2,4	0,8758	$\pm$ 0,0031	15451, 5	$\pm$ 477	123 927	
$\pm$ 3,0	273, 3	$\pm$ 4, 1	0,8800	$\pm$ 0,0027	6482	$\pm$ 143, 8	120 517	
$\pm$ 4,6	256, 7	$\pm$ 1,9	0,8616	$\pm$ 0,0026	1889, 3	$\pm$ 19	119 334	
$\pm$ 4, 7	261, 2	$\pm$ 3, 0	0,8688	$\pm$ 0,0044	431, 6	$\pm 2,3$	120 081	
$\pm$ 4, 1	284, 5	$\pm$ 1,8	0,8755	$\pm$ 0,0031	1308, 8	$\pm$ 9,0	117 613	
$\pm$ 3, 4	284, 3	$\pm$ 3, 5	0,8705	$\pm$ 0,0034	2936	$\pm$ 49	116 363	
± 1,8	309, 4	$\pm$ 2, 4	0,8869	$\pm$ 0,0028	3676	$\pm$ 42	115 808	
± 2,8	335, 5	$\pm$ 1,6	0,8950	$\pm$ 0,0030	2643	$\pm$ 34	113 345	
$\pm$ 3,6	386, 4	$\pm$ 4, 1	0,9176	$\pm$ 0,0053	340, 3	$\pm$ 2, 2	110 476	
± 2,8	271,6	$\pm$ 2,4	0,9122	$\pm$ 0,0028	4073	$\pm$ 32	128 778	
$\pm$ 2, 5	266, 5	$\pm$ 2, 2	0,9049	$\pm$ 0,0021	7649	$\pm$ 46	127 985	
$\pm$ 2, 2	242, 7	$\pm$ 2, 3	0,8853	$\pm$ 0,0020	11234	$\pm$ 66	127 839	
$\pm$ 3,6	241, 5	$\pm$ 2,6	0,8806	$\pm$ 0,0026	4155	$\pm$ 24	126 877	
$\pm$ 3,0	254, 7	$\pm$ 2, 5	0,8844	$\pm$ 0,0029	3355	± 18	125 164	
$\pm$ 2, 7	251, 1	$\pm$ 2,0	0,8745	$\pm$ 0,0026	2349	$\pm$ 12	123 379	
$\pm$ 2, 7	251, 5	$\pm$ 2, 1	0,8632	$\pm$ 0,0035	2333	$\pm$ 19	120 541	
$\pm$ 3, 3	277, 4	$\pm$ 2, 1	0,8765	$\pm$ 0,0027	2230	$\pm$ 19	118 946	
± 2,9	281, 7	$\pm$ 2, 1	0,8679	$\pm$ 0,0026	1727	$\pm$ 15	116 230	
$\pm$ 3, 3	322, 3	$\pm$ 1,9	0,8922	$\pm$ 0,0031	2070	$\pm$ 24	114 835	
<u>±</u> 3, 7	420, 7	± 2,7	0, 9401	± 0,0038	918, 9	± 7,3_	109 862	

n et al., 2012).

or <sup>238</sup>U (Jaffey et al., 1971).

 $\pm$  2 ppm.

y assumed to be 50%.

nces, NTU.

		-	C23411		
зе	Age		$\delta^{234}$ Uinitial		
rected	CO:	rrected <sup>c, e</sup>	correct	ed	
$\pm 1248$	127 481	$\pm$ 1288	388, 1	$\pm$ 3, 1	
$\pm$ 754	127 162	$\pm$ 754	355, 6	$\pm$ 3, 4	
$\pm$ 849	125 988	$\pm$ 849	355, 7	$\pm$ 3, 7	
$\pm$ 909	123 918	$\pm$ 909	356, 0	$\pm$ 3,6	
$\pm$ 978	120 474	$\pm$ 978	384, 0	$\pm$ 5,8	
$\pm$ 706	119 261	$\pm$ 709	359, 6	$\pm$ 2, 7	
$\pm$ 1198	119 436	$\pm 1234$	365, 8	$\pm$ 4,4	
$\pm$ 775	117 508	$\pm$ 782	396, 6	$\pm$ 2, 7	
$\pm$ 978	116 270	$\pm$ 978	394, 7	$\pm$ 5,0	
$\pm$ 747	115 735	$\pm$ 747	428, 9	$\pm$ 3, 5	
$\pm$ 681	113 244	$\pm$ 683	461, 9	$\pm$ 2, 4	
$\pm$ 1196	110 090	$\pm$ 1252	527, 4	$\pm$ 5, 9	
$\pm$ 865	128 708	$\pm$ 865	390, 6	$\pm$ 3,6	
$\pm$ 698	127 947	$\pm$ 698	382, 3	$\pm$ 3, 3	
$\pm$ 711	127 814	$\pm$ 711	348, 1	$\pm$ 3, 4	
$\pm$ 862	126 809	$\pm$ 862	345, 4	$\pm$ 3, 7	
± 888	125 079	± 889	362, 5	$\pm$ 3, 7	
$\pm$ 757	123 260	$\pm$ 758	355, 6	$\pm$ 2,9	
$\pm$ 938	120 422	$\pm$ 939	353, 2	$\pm$ 3, 1	
$\pm$ 729	118 822	$\pm$ 730	387, 9	$\pm$ 3,0	
$\pm$ 695	116 073	$\pm$ 698	390, 8	$\pm$ 3,0	
$\pm$ 733	114 705	$\pm$ 735	445, 6	± 2,8	
± 806	109 578	± 816	573, 2	± 3,9	

Supplementary Table 2. Stable carbon and oxygen isotope compositions (in % calcite samples drilled along 6 growth layers ("Hendy test", see text).

1 _	1: £		
sample	distance from	s13a	c18o
	the axis (mm)	δ <sup>13</sup> C	δ <sup>18</sup> 0
BAR-2BH 1/1	0	-6, 8	-7, 3
BAR-2BH 1/2	8	-5, 9	-7, 5
BAR-2BH $1/3$	16	-7, 0	-7, 8
BAR-2BH $1/4$	24	-6, 7	-8, 1
BAR-2BH 1/5	34	-6, 3	-7, 2
BAR-2BH $2/1$	0	-8, 8	-7, 2
BAR-2BH $2/2$	8	-8, 9	-7, 2
BAR-2BH $2/3$	15	-8, 8	-6, 8
BAR-2BH $2/4$	23	-8, 5	-6, 8
BAR-2BH $2/5$	30	-7, 7	-7, 5
BAR-2BH 3/1	0	-10, 0	-7, 2
BAR-2BH $3/2$	8	<b>−10,</b> 2	-7, 3
BAR-2BH $3/3$	16	<b>−10,</b> 3	-7, 6
BAR-2BH $3/4$	24	-10, 0	-7, 0
BAR-2BH $3/5$	32	-10, 0	-6, 7
BAR-2BH $4/1$	10	-8, 0	-7, 4
BAR-2BH $4/2$	16	-8, 2	-7, 3
BAR-2BH $4/3$	21	-7, 8	<b>−7,</b> 3
BAR-2BH $4/4$	27	-7, 6	<b>−7,</b> 3
BAR-2BH 4/5	34	-7, 6	-7, 3
BAR-2BH 5/1	3	-9, 2	-7, 2
BAR-2BH 5/2	8	-9, 2	-6, 2
BAR-2BH 5/3	12	-8, 7	-7, 3
BAR-2BH 5/4	17	-8, 4	-6, 8
BAR-2BH 5/5	21	-7, 8	-6, 4
BAR-2BH 6/1	3	-8, 3	-7, 5
BAR-2BH 6/2	8	-8, 3	-7, 1
BAR-2BH 6/3	13	-8, 2	-7, 2
BAR-2BH 6/4	19	-7, 6	-7, 4
BAR-2BH 6/5	25	-7, 5	-7, 5

Supplementary Table 3. Stable carbon and oxygen isotope compositions (relative to (relative to V-SMOW) of water extracted from chips collected from stalagmites. Dis

BAR-II#B column				BAR-II#B col	umn, cont.	
distance	Age	$\delta^{13}$ C	$\delta^{18}0$	distance	Age	$\delta^{13}$ C
from top	J			from top	J	
(mm)	(years BP)	(‰)	(%)	(mm)	(years BP)	(‰)
270, 5	128183	-6, 6	-7, 9	144, 7	121407	-8, 9
265, 7	128029	-8, 0	-7, 6	144, 4	121377	-8, 0
261,0	127895	-7, 4	-7, 0	143, 8	121317	-8, 6
256, 3	127775	-7, 8	-7, 6	143, 3	121268	-8, 9
253, 9	127718	-8, 6	-7, 7	142, 8	121219	-8, 8
253, 4	127706	-8, 5	-7, 4	142, 3	121171	-8, 6
252, 5	127685	-7, 9	-7, 5	141, 7	121113	-8, 5
252, 3	127680	-8, 2	-7, 6	141, 3	121074	-8, 8
251, 3	127659	-7, 5	-7, 4	140, 8	121026	-8, 8
251, 2	127655	-7, 7	-7, 6	140, 2	120969	-8, 9
250, 6	127641	-7, 7	-7, 4	139, 9	120951	-9, 2
249, 8	127623	-7, 4	-7, 6	139, 2	120876	-8, 8
249, 3	127613	-7, 4	-7, 5	138, 8	120839	-9, 4
248, 8	127599	-7, 2	-7, 4	138, 4	120802	-9, 3
248, 3	127590	-7, 5	-7, 4	137, 8	120747	-9, 7
247, 6	127572	-7, 0	-7, 2	137, 4	120710	-9, 7
247, 0	127558	-7, 8	-7, 6	136, 7	120647	-9, 7
246, 7	127551	-8, 0	-7, 7	136, 4	120620	-9, 6
246, 1	127536	-7, 5	-7, 3	135, 9	120576	-9, 7
245, 6	127525	-7, 6	-7, 8	135, 8	120567	-9, 6
245, 1	127513	-8, 2	-7, 2	135, 0	120497	-9, 6
244, 6	127501	-7, 9	-7, 4	134, 6	120463	-9, 7
244, 0	127486	-8, 3	-7, 6	133, 9	120403	-9, 7
243, 7	127479	-8, 4	-7, 7	133, 8	120395	-9 <b>,</b> 5
243, 1	127464	-8 <b>,</b> 0	-7, 2	133, 4	120361	-9 <b>,</b> 9
242, 8	127457	-8 <b>,</b> 5	-7, 5	133, 4	120361	-9, 5
242, 2	127442	-8 <b>,</b> 1	-7, 5	132, 9	120319	-9, 8
241, 7	127430	-8, 5		132, 8	120311	-10, 1
241, 0	127412			132, 4	120278	-9, 8
240, 5	127399	-9, 1 $-8, 7$		132, 2	120262	-10, 0
239, 8	127381	-8, 7 -8, 7		132, 1 131, 7	120254 120221	-10, 1 $-9, 8$
239, 6 238, 7	127376 127352	-8, <i>t</i> -8, 4	−7, 6 −7, 3	131, 7	120221	-9, 8 -9, 9
238, 6	127332		-7, 5 -7, 6	131, 4	120205	
237, 8	127343	−8, 0		131, 4	120203	-10, 3 -10, 1
237, 7	127325	-7, 4	-7, 0	130, 6	120134	-10, 1 -10, 2
236, 8	127323	-8, 7	-7, 3	130, 6	120134	-9, 9
236, 7	127301	-8, 0	-6, 8	130, 6	120134	-10, 0
235, 9	127276	-8, 7		130, 3	120134	-10, 3
235, 5	127276	-8, 4	-7, <del>4</del>	129, 9	120110	-10, 3 -10, 1
234, 5	127236	-9, 3	-7, 3	129, 6	120056	-10, 1
234, 5	127236	−8, 0	-6, 9	129, 5	120049	-10, 1
233, 7	127213	−9, 3	-7, 6	129, 5	120049	-10, 2
200, 1		0, 0	., .	120,0		10,2

233, 5	127207	-7, 7	-6, 8	128, 6	119981	-9, 9
232, 9	127190	-8, 0	-6, 9	128, 5	119974	-10, 1
232, 8	127187	-7, 7	<b>−7,</b> 2	128, 5	119974	-9, 9
232, 4	127175	-7, 2	-7, 1	128, 3	119959	-10, 1
232, 4	127175	-7, 5	-7, 1	127, 8	119923	-9, 5
231, 8	127157	-7, 8	-7, 4	127, 7	119915	-10, 1
231, 8	127157	-7, 4	-6, 9	127, 5	119901	-10, 0
231, 5	127148	-7, 4	-7,6	127, 4	119894	-9, 9
231, 3	127142	-7, 5	<b>−7,</b> 2	126, 9	119858	-10, 0
231, 0	127133	-8, 3	-7, 0	126, 9	119858	-10, 1
230, 8	127127	-7, 8	-7, 4	126, 7	119844	-10, 1
230, 5	127118	-7, 5	<b>−7,</b> 3	126, 4	119824	-10, 3
230, 2	127108	-7, 4	-6, 9	126, 2	119810	-10, 0
229, 8	127096	-7, 8	<b>−7,</b> 5	125, 8	119783	-9, 9
229, 7	127093	-7, 5	<b>−7,</b> 3	125, 7	119776	-10, 1
229, 0	127071	-7, 9	-6, 9	125, 5	119762	-10, 3
228, 6	127058	-7, 4	-7, 0	125, 0	119729	-10, 2
228, 1	127042	-7, 5	-6, 5	124, 8	119723	-9, 8
227, 9	127036	-7, 8	-7, 2	124, 7	119710	-10, 2
227, 5	127023	-7, 1	-6, 8	124, 5	119697	-10, 1
227, 4	127020	-7, 9	-6, 9	124, 0	119665	-10, 3
227, 2	127013	-8, 3	-6, 8	123, 8	119653	-10, 0
226, 6	126994	-7, 7	-6, 9	123, 5	119634	-9, 9
226, 3	126984	-8, 0	-6, 8	123, 3	119622	-10, 1
225, 8	126967	-7, 7	-6, 5	123, 2	119615	-10, 2
225, 6	126960	-8, 1	-6, 9	122, 8	119591	-9, 9
225, 2	126947	-7, 9	-6, 8	122, 7	119585	-10, 0
224, 8	126937	-7, 9	-6, 8	122, 4	119568	-10, 1
224, 4	126920	-7, 6	-6, 9	122, 4	119568	-9, 9
223, 8	126903	-7, 8	-6, 4	122, 4	119568	-10, 0
223, 6	126892	-8, 2	-6, 9	121, 9	119538	-10, 0
223, 1	126875	-7, 7	-6, 8	121, 6	119521	-10, 0
223, 0	126871	-7, 2	-7, 7	121, 5	119516	-9, 9
223, 0	126871	-8, 0	-6, 7	121, 4	119510	-9, 9
223, 0	126871	-7, 8	-6, 8	121, 1	119493	-10, 1
222, 0	126836	-8, 4	-7, 0	120, 5	119460	-9, 8
222, 0	126836	-8, 8	-6, 6	120, 5	119460	-9, 7
221, 9	126832	-8, 5	-6, 9	120, 4	119454	-9, 9
221, 7	126825	-9, 3	<b>−7,</b> 5	120, 1	119438	-9, 9
221, 1	126803	-8, 5	-6, 7	119, 8	119428	-10, 1
220, 6	126785	-8, 9	-7, 8	119, 4	119401	-9, 9
220, 4	126778	-8, 2	-7, 4	119, 2	119391	-9, 9
219, 8	126755	-8, 9	-7, 6	118, 4	119350	-9, 8
219, 2	126733	-8, 5	-6, 9	117,8	119320	-9, 9
218, 9	126721	-8, 6	-7, 0	117, 3	119296	-9, 5
217, 9	126687	-8, 4	-6, 8	116, 6	119262	-9, 8
217, 7	126675	-8, 5	-6, 9	116, 2	119243	-9, 8
216, 9	126644	-9, 1	-7, 7	115, 6	119216	-10, 0
216, 7	126636	-8, 8	-7, 0	115, 2	119197	-9, 8
216, 0	126608	-8, 9	-7, 9	114, 9	119184	-9, 6
•		•	•	,		,

215, 8	126599	-8, 4	<b>−7,</b> 5	114, 9	119184	-9, 8
215, 4	126583	-8, 5	-7, 8	114, 3	119157	-9, 9
215, 0	126567	-8, 6	-7, 0	114, 2	119153	-10, 0
214, 5	126546	-8, 1	-7, 4	114, 0	119144	-10, 0
213, 9	126521	-8, 7	-7, 2	113, 6	119127	-10, 1
213, 5	126504	-7, 9	-7, 3	113, 4	119118	-10, 0
213, 1	126486	-8, 0	<b>−7,</b> 2	113, 3	119114	-10, 1
212, 8	126473	-8, 1	-7, 1	113, 1	119105	-9, 8
212, 1	126443	-8, 3	-7, 2	112, 5	119079	-9, 7
211, 6	126420	-8, 3	<b>−7,</b> 2	112, 1	119062	-9, 7
211, 1	126398	-8, 1	− <del>7</del> , 2	112, 1	119062	-10, 1
210, 9	126389	-8, 5	-7, 0	112, 1	119062	-10, 0
210, 2	126357	-7, 6	-6, 8	111, 3	119029	-9, 9
209, 7	126333	-8, 6	-6, 9	111, 1	119021	-9, 8
209, 1	126305	-8, 7	− <del>7</del> , 1	111, 0	119017	-9, 9
208, 8	126291	-8, 6	-7,0	110, 9	119012	-9, 8
208, 2	126262	-8, 2	-6, 9	109, 8	118971	-9, 8
207, 9	126247	-8, 7	-6, 6	109, 5	118955	-9, 8
207, 4	126223	-7, 7	-6, 5	108, 7	118923	-9, 9
206, 9	126198	-8, 7	-6, 6	108, 6	118919	-10, 0
206, 6	126183	-8, 2	-6, 7	107, 8	118887	-9, 8
205, 8	126142	-8, 6	-6, 7	107, 7	118883	-9, 8
205, 5	126126	-8, 2	-6, 7	107, 0	118854	-9, 7
205, 0	126100	-8, 4	-6, 7	106, 9	118850	-9, 9
204, 8	126090	-8, 6	-6, 7	106, 0	118814	-10, 0
204, 4	126069	-8, 5	-7, 0	105, 8	118806	-10, 0
203, 9	126042	-8, 2	-7, 0	105, 0	118774	-9, 7
203, 8	126036	-8, 7	-7, 0	104, 8	118766	-10, 0
203, 5	126020	-8, 2	-7, 1	103, 6	118716	-9, 8
203, 3	126009	<b>−8,</b> 2	-7, 2	103, 6	118716	-9, 5
202, 9	125987	-8, 2	-7, 2	102, 5	118671	-9, 6
202, 9	125987	-7, 8	<b>7</b> 0	102, 1	118654	-9, 1
202, 5	125965	-8, 5	-7, 1	101, 6	118633	-9, 5
202, 5	125965	<b>−8,</b> 1	-7, 7	101, 3	118620	-9, 2
202, 0	125937	-8, 4	<b>−7,</b> 3	100, 3	118577	-9, 2
201, 9	125931	-9, 1	-7, 4	100, 2	118572	-8, 4
200, 7	125863	-7, 7	<b>−7,</b> 3	99, 4	118537	-8, 7
200, 7	125863	-8, 8	-7, 2	99, 3	118533	-8, 5
199, 7	125804	-8, 8	-7, 4	98, 3	118489	-8, 2
199, 7	125804	-8, 0	-7, 4	97, 9	118471	-8, 5
198, 8	125750	-7, 7	-7, 4	97, 3	118444	-8, 2
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195, 2	125528	-7, 9	<b>−7,</b> 2	91, 2	118158	-8, 5
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194, 3	125471	-8, 3	-7, 1	88, 5	118024	-8, 3
194, 3	125471	-7, 8	-7, 3	85, 6	117877	-8, 6
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189, 6	125163	-7, 8	-6, 7	77, 9	117466	-7, 9
189, 5	125156	-8, 0	-7, 0	77, 7	117455	-8, 9
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177, 8	124358	-8, 9	-6, 9	56, 0	116125	-7, 7

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176, 4	124267	-8, 9	-7, 1	51, 3	115753	-6, 5
175, 6	124203	-8, 7	-7, 1	51, 1	115736	-6, 6
175, 3	124182	-9, 0	<b>−7,</b> 1	50, 5	115685	-6, 5
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174, 3	124110	-9, 0	-7, 1	48, 4	115497	-6, 5
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172, 0	123941	-8, 8	-6, 7	45, 6	115221	-7, 4
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167, 9	123632	-8, 9	-7, 3	39, 7	114536	-5, 7
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166, 0	123468	-9, 5	-7, 3	37, 4	114231	-6, 5
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163, 8	123280	<b>−9,</b> 3	-7, 0	35, 2	113922	-6, 1
163, 3	123236	-9, 5	-7, 1	34, 2	113792	-6, 3
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157, 9	122733	-8, 7	-6, 8	28, 7	112936	-6, 0
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157, 6	122704	-8, 7	-7, 0	27, 8	112792	<b>-6,</b> 2
156, 7	122616	-9, 2	-6, 8	27, 4	112728	-5, 5
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154, 1	122357	-8, 6	-6, 6	25, 5	112420	-6, 0
153, 6	122306	-9, 1	-6, 6	24, 5	112272	-5, 1
153, 2	122266	<b>−9,</b> 2	-6, 6	24, 4	112239	<b>−6,</b> 2
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152, 2	122165	-9, 1	-6, 6	22, 7	111958	-5, 5
151, 4	122084	-9, 1	<b>−7,</b> 2	22, 2	111875	-6, 0
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150, 4	121992	-9, 1	-6, 9	21, 2	111709	-6, 6
149, 8	121921	-8, 7	-6, 9	20, 1	111525	-5, 8
148, 9	121830	-8, 8	-7, 0	20, 1	111525	-6, 0
148, 5	121789	-8, 5	-6, 9	19, 3	111392	-6, 1
147, 9	121728	-8, 6	-7, 4	19, 2	111375	-6, 1
147, 4	121688	-9, 0	-7, 1	18, 0	111174	-7, 1
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145, 5	121486	-8, 9	-7, 4	3, 4	108779	<b>-4,</b> 2
145, 4	121476	<b>−8,</b> 2	-7, 0			

V-PDB) of drilled calcite samples, and water contents, hydrogen and oxygen i stance from top means distance from hiatus 1 (see Fig. 1).

	BAR-II#L c	olumn			BAR-II#B (
$\delta^{18}0$	distance	Age	$\delta^{13}$ C	$\delta^{18}$ 0	distance
(‰)	from top (mm)	(years BP)	(‰)	(%)	from top (mm)
-7, 6				rsity of Lausanne	25, 3
-6, 9	240, 0	129068	-6, 6	-7,9	24, 9
			-8, 0		
−7, 1 −6, 6	235, 8 231, 6	128883	-3, 0 -7, 4	-7, 6 -7, 0	24, 4 22, 8
		128702		-7, 0	
-7, 1	227, 4	128534 128385	-7, 8	-7, 6	21, 9
-6, 8	223, 2		-7, 2	-7, 5	20, 0
-6, 7	219, 0	128264	-7, 5	-6, 9	19, 1
-6, 5	214, 8	128168	-7, 9	-7, 3	18, 5
-6, 6	210, 6	128082	-7, 5	-7, 3	17, 9
-6, 4	206, 4	127991	-7, 9	-7, 5	17, 3
-6, 7	202, 2	127878	-7, 7	-7, 0	16, 0
-6, 7	198, 0	127728	-7, 2	-6, 8	15, 3
-6, 4	193, 8	127538	-7, 1	-7, 0	14, 8
-6, 6	189, 6	127314	-7, 2	-6, 8	13, 7
-6, 8	185, 4	127061	-7, 2	-7, 1	11,8
-6, 9	181, 2	126785	-7, 9	-6, 7	10, 3
-6, 7	177, 0	126491	-8, 2	-6, 5	8, 1
-6, 8	172, 8	126182	-7, 8	<b>−6,</b> 7	6, 0
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-6, 9	160, 2	125197	-7, 4	-6, 9	1, 1
-6, 8	156, 0	124857	-7, 8	-7, 0	0, 1
-7, 2	151,8	124512	-7, 7	-7, 0	
-7, 1	147, 6	124164	-7, 6	− <b>7</b> , 1	
-7, 1	143, 4	123812	-7, 7	-6, 8	
-6, 8	139, 2	123458	-9, 0	<b>−7,</b> 2	
-7, 1	135, 0	123100	-8, 9	<b>−7,</b> 2	
-7, 4	130, 8	122741	-8, 9	-7, 0	
-7, 1	126, 6	122379	-8, 5	-6, 8	
-7, 1	122, 4	122017	-8, 7	-6, 6	
-7, 2	118, 2	121654	-9, 1	-6, 6	
-7, 3	114, 0	121290	-9, 3	-6, 7	
-7, 2	109, 8	120927	-9, 4	-6, 9	
<b>−7,</b> 2	105, 6	120565	-9, 6	-6, 9	
-7, 3	101, 4	120204	-9, 7	-6, 7	
-7, 0	97, 2	119845	-9, 6	-6, 7	
-7, 0	93, 0	119488	-9, 7	-6, 8	
-7, 1	88, 8	119131	-8, 7	<b>−6,</b> 2	
-7, 2	84, 6	118772	-8, 5	-6, 7	
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-7, 1	76, 2	118048	-7, 9		
-7, 0	72, 0	117682	-8, 5		
-7, 0	67, 8	117315	-8, 4		
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-7, 1	high-resol	ution analy	ses, IGGR	Budapest
-7, 1	27, 4	113755	-6, 3	-7, 9
-7, 1	26, 7	113654	-5, 2	-8, 1
-7, 1	26, 2	113561	-5, 4	-7, 7
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-7, 0	24, 4	113265	-7, 1	-8, 0
<b>−7,</b> 2	23, 9	113181	-6, 5	-7,8
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-7, 1	20, 0	112522	-6, 5	-8, 0
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- -8, 0
- -8, 0
- -8, 4
- -8, 0
- -7, 8
- -7, 9
- -8, 4
- -8, 0
- -8, 1
- -8, 1
- -7, 7 -8, 3
- -8, 8
- -9,8

# sotope compositions

column				BAR-II#L column
Age (years BP)	H <sub>2</sub> O mg/gr	δ <sup>18</sup> 0 (‰)	δD (‰)	distance Age $H_2O$ from top (years BP) $mg/gr$
(years bi)	шg/ gт	(700)	(700)	(mm)
127699	0, 44	-8,8	-70	22, 9 128618 0, 94
127612	0, 31	-11,2	-76	20, 8 128040 0, 75
127513	1, 12	-9,5	-65	17, 8 126517 0, 74
127073	1,00	-11,6	-73	17, 1 126035 0, 55
126764	1, 43	-12,2	-73	15, 8 124967 0, 07
125812	1, 32	-11,9	-69	14, 9 124315 0, 13
125282	1, 36	-11,2	-68	13, 6 123183 0, 63
124848	1, 26	-10,6	-70	12, 3 122107 2, 16
124438	0, 19	-9,2	-67	11,8 121631 no data
123979	0, 18	-13,8	-81	10, 4 120422 2, 13
122944	0, 35	-11,9	-74	9, 5 119656 1, 73
122191	0,68	-10,3	-68	8, 8 119062 2, 16
121671	0, 71	-10,8	-70	8, 6 118849 1, 62
120612	2, 32	-10,2	-64	4, 3 115154 0, 24
119314	1, 95	-10,7	-67	0, 4 111498 0, 27
118658	2, 13	-10,6	-64	
117658	0,86	-11,5	-68	
116379	0,61	-11,5	-69	
114592	0,60	-12,3	-78	
113534	0, 19	-10,7	-75	
109988	0, 31	-12,4	-77	
108518	0, 73	-16,4	-90	

$\delta^{18}0$	δD
(‰)	(‰)
-12,5	-80
-13,0	-75
-12,9	-75
-12,7	-75
-15,0	-79
-12,4	-76
-12,5	-80
-10,5	-71
-9,8	-64
-9,7	-64
-10,5	-65
-10,4	-65
-9,8	-65
-13,6	-83
-12,8	-83

Supplementary Table 4. Precisions of dating analyses of selected stalagmite records

stalagmite	publication	average	n
BAR-II	this study	864	23
Han-9	Vansteenberge et al. (2016)	415	23
BDinf	Couchoud et al. (2009)	1103	18
TKS	Meyer et al. (2008)	2568	22
SCH-5	Moseley et al. (2015)	571	23
HÖL−10	Moseley et al. (2015)	744	13
D4	Regattieri et al. (2014)	2579	9
CC5	Drysdale et al. (2005)	1194	18
CC28	Drysdale et al. (2007)	1050	16
S0-17a	Badertscher et al. (2011)	305	9
K1-2010	Nehme et al. (2015)	1293	17
Peqin	Bar-Matthews et al. (2003)	1282	5
Soreq	Bar-Matthews et al. (2003)	2471	15
SCH-7	Boch et al. (2011)	527	20
D3	Kelly et al. (2006)	1018	17
D4	Kelly et al. (2006)	1116	19

# s (see text) for their last interglacial p

min	max
683	1288
212	731
860	1650
860	5000
413	1736
453	1613
1100	5360
900	1700
700	1700
230	413
490	3060
810	2700
1220	7710
440	790
700	1400
800	1600

## Highlights

- Stable isotope compositions of a last interglacial (LIG) stalagmite, Central Europe
- A sequence of climate change events during the LIG for European speleothems
- Hydrogen isotope data of inclusion water reveal a major event at about 125 ± 2 ka
- Temperature and winter precipitation changes around the Mediterranean at 125  $\pm$  2 ka