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**THE ABUNDANCE AND ISOTOPIC COMPOSITION OF WATER IN HOWARDITE-EUCRITE-DIOGENITE METEORITES.**

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**Introduction:** Volatile elements play a fundamental role in planetary formation and evolution through their influence on melting, silicate melt viscosity, magma crystallization and eruption processes [1]. The Howardite-Eucrite-Diogenite (HED) suite of meteorites represents the largest suite of crustal rocks available from a differentiated basaltic asteroid and account for between 2-3% of all meteorites collected globally [2].

Apatite is a widely distributed mineral, albeit in trace amounts, in planetary materials which acts as a recorder of volatile abundances in magmas and magmatic source regions [3] and is the most common volatile-bearing phase in lunar rocks [4, 5] and eucrites [3]. We are currently undertaking a detailed study of apatite in HED meteorites using Secondary Ion Mass Spectrometry (SIMS) techniques to better constrain the volatile inventory and evolutionary history of their putative parent body, 4 Vesta [6, 7].

**Methods:** We used the Cameca NanoSIMS 50L at the Open University to measure OH abundances and D/H ratios in apatite grains from two Eucrites (DaG 945, DaG 844) using the protocol described in [8]. In total, 13 measurements were made on 10 different apatite grains. As the cosmic ray exposure (CRE) ages for these samples are unknown, the measured D/H ratios were corrected for spallation processes using a CRE age of 38 Ma - the oldest cluster of CRE ages for HED meteorites [9]. This correction had little effect on the final D/H ratio.

**Results:** Apatite H<sub>2</sub>O abundances range from ~50 to ~222 ppm in DaG 945 and ~1080 to ~3446 ppm in DaG 844, with weighted average  $\delta D$  values of  $17 \pm 160$  ‰ and  $-88 \pm 72$  ‰ ( $2\sigma$ ), respectively.

**Discussion:** Our results are within error of and extend the range of data reported by [10]. DaG 945 contains less water and is believed to have undergone granulitic metamorphism and at least some partial melting [11], which could explain the low water contents measured in apatite in this sample. The average  $\delta D$  values of these two samples are also similar to the  $\delta D$  values of carbonaceous chondrites, the Earth and the Moon, and are consistent with a common source of water.

**References:** [1] Saal A. E. et al. 2008. *Nature* 454(7201): 192-195. [2] Janots E. et al. 2012. *Meteoritics & Planetary Science* 47(10):1558-1574. [3] Sarafian A. R. et al. 2013. *Meteoritics & Planetary Science* 48(11):2135-2154. [4] McCubbin F. M. et al. 2011. *Geochimica et Cosmochimica Acta* 75(17):5073-5093. [5] McCubbin F. M. et al. 2010. *American Mineralogist* 95(8-9):1141-1150. [6] McCord et al. 1970. *Science* 168:1445-1447. [7] Binzel R. P. and Xu S. 1993. *Science* 260:186-191. [8] Barnes J. J. et al. 2013. *Chemical Geology* 337-338(0):48-55. [9] Eugster O. and Michel T. 1995. *Geochimica et Cosmochimica Acta* 59(1):177-199. [10] Sarafian A. R. et al. 2014. Abstract #2106. 45th Lunar & Planetary Science Conference. [11] Yamaguchi A. et al. 2009. *Geochimica et Cosmochimica Acta* 73(23):7162-7182.