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Temporal variability in bioassays of the stomatal ammonia compensation point in relation to plant and soil nitrogen parameters in intensively managed grassland

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Abstract. The exchange of ammonia between crop canopies and the atmosphere depends on a range of plant parameters and climatic conditions. However, little is known about effects of management factors. We have here investigated the stomatal ammonia compensation point in response to cutting and fertilization of a grass sward dominated by *Lolium perenne*. Tall grass had a very low NH₃ compensation point (around 1 nmol mol⁻¹), reflecting the fact that leaf nitrogen (N) concentration was very low. During re-growth after cutting, leaf tissue concentrations of NO₃⁻, NH₄⁺, soluble N and total N increased along with apoplastic NH₄⁺ concentrations. In contrast, apoplastic pH decreased resulting in largely unaltered NH₃ compensation points. Nitrogen fertilization one week after cutting caused the apoplastic NH₄⁺ concentration of the newly emerging leaves to increase dramatically. The NH₃ compensation point peaked between 15 and 25 nmol mol⁻¹ the day after the fertiliser was applied and thereafter decreased over the following 10 days until reaching the same level as before fertilisation. Ammonium concentrations in leaf apoplast, bulk tissue and litter were positively correlated (P=0.001) throughout the experimental period. Bulk tissue NH₄⁺ concentrations, total plant N and soil

NH₄⁺ concentrations also showed a positive correlation. A very high potential for NH₃ emission was shown by the plant litter.

1 Introduction

Ammonia is emitted from plants when the atmospheric NH₃ concentration is lower than the NH₃ compensation point, the latter being equal to the NH₃ concentration in the substomatal cavity (Farquhar et al., 1980; Husted et al., 1996). In the opposite situation, i.e. when the atmospheric NH₃ concentration exceeds the NH₃ compensation point, deposition of NH₃ occurs. The quantity of NH₃ exchanged between crop canopies and the atmosphere may vary between seasons, depending on climatic conditions (Schjoerring and Mattsson, 2001; Sommer et al., 2004). In particular, temperature is known to have a major effect on the NH₃ exchange under controlled environmental conditions (Husted and Schjoerring, 1996; Mattsson et al., 1997) as well as in the field (van Hove et al., 2002; Trebs et al., 2006).

The NH₃ emission potential of grasslands may vary with species composition (Horvath et al., 2005) because grass species differ in NH₃ compensation point (Hanstein et al., 1999; Herrmann et al., 2001; Mattsson and Schjoerring, 2002; Mattsson et al., 2008). In a non-fertilized managed



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grassland in The Netherlands, NH_3 emission fluxes were frequent, covering about 50% of the time in a warm, dry summer period (Wichink Kruit et al., 2007). In contrast, during a wet, cool autumn period, deposition fluxes dominated (80% of the time) due to small canopy compensation points caused by low temperatures and a generally wet surface (Wichink Kruit et al., 2007). Nitrogen fertilisation is one of the major management factors of grasslands and NH_3 volatilisation can be influenced by the form, timing and dosage of N fertiliser (Riedo et al., 2002). Measurements of NH_3 volatilisation under controlled laboratory conditions have shown that high amounts of N supplied to the roots increase NH_3 emission (Mattsson et al., 1998; Mattson and Schjoerring, 1996) and NH_3 compensation points (Mattsson and Schjoerring, 2002). Increasing the N availability to plant roots leads to elevated steady state levels of different N pools within the plant tissue. In a field experiment over two years, the NH_3 losses from wheat, oilseed rape and barley increased under conditions of high N concentration in the foliage (Schjoerring and Mattsson, 2001). In a Scottish experiment, a higher NH_3 compensation point of the grass was seen after only one of the two cuttings and fertilisations (Loubet et al., 2002). Little is known about the NH_3 emission potential of grasslands where repeated cuttings and N fertilisations are normal management practice. A better understanding of the component parameters influencing the NH_3 emission potential is needed in order to model NH_3 exchange between grasslands and the atmosphere.

The aim of the present study was to estimate the NH_3 emission potential of grassland in relation to common management practice. In order to do this, the temporal variation in the NH_3 compensation point and its underlying components of grass leaves and soil were followed at a field site (Sutton et al., 2008), starting with tall grass and spanning subsequent events of cutting, lifting and N-fertilization.

2 Materials and methods

The investigation took place as part of the GRAMINAE integrated experiment conducted on a field near Braunschweig from 22 May to 15 June 2000. The main field was 600×300 m in size and consisted of a mixed sward dominated by *Lolium perenne* (around 60% abundance), *Phleum pratense* (~15% abundance) and *Festuca pratensis* (~12% abundance; Mattsson et al., 2008). The data presented for leaves of tall grass plants are mean values of these 3 most abundant species, weighted by their relative abundance in the field. The grass was cut on 29 May and lifted for silage on 31 May. An area of 10×10 m was left uncut for additional sampling of tall grass. Fertilizer (100 kg N ha⁻¹ in calcium ammonium nitrate) was applied on the main field on the 5 June. A 10×10 m plot was left unfertilized and another plot of the same size received 200 kg N ha⁻¹ in calcium ammo-

nium nitrate. Growth and development of the grass were as described in Sutton et al. (2008).

2.1 Sampling of plant material

Throughout the entire experiment, plants were sampled almost every day between 12:00 and 03:00 p.m. (GMT). Cut green leaves were immediately taken to the field laboratory where the apoplastic solution was extracted using a vacuum infiltration technique (Husted and Schjoerring, 1995). Whole leaves were infiltrated in isotonic sorbitol solution (280 mM) at a pressure of 16 bar under vacuum for 5 s. The procedure was repeated 5 times in order to ensure full infiltration. Infiltrated leaves were carefully blotted dry and kept in plastic bags to equilibrate for 15 min in daylight. Leaf apoplastic solution was extracted by centrifugation at 800 g for 10 min at 4°C. After extraction, pH of the apoplastic samples was measured with a micro-combination pH electrode (9810, Orion, Beverly, USA) and samples were frozen at -18°C. Leaf samples for bulk tissue NH_4^+ and NO_3^- analysis were also frozen down at the same time for later extraction. Samples of litter (senescent leaves) and stubble (cut stems) were frozen every day after the grass was cut. For total N concentration, samples of leaves, litter and stubbles were taken daily and immediately dried in an oven (70°C) over night. Guttation droplets were collected on the main field and the high fertilized plot between 03:00 and 06:00 a.m. and immediately frozen.

2.2 Plant analysis

Ammonium in apoplastic extracts was determined by fluorometry on an HPLC system (Waters Corp. Milford, USA) equipped with a pump, a column oven with a 3.3 m stainless steel reaction coil, an autosampler cooled to 2°C and a scanning fluorescence detector. The reaction between NH_4^+ and *o*-phthaldehyde (OPA) to form an alkylthioisoin-dole fluorochrome was performed at neutral pH with 2-mercaptoethanol as reducing agent. This fluorochrome was detected at an excitation wavelength of 410 nm and an emission wavelength of 470 nm (Husted et al., 2000a).

The plant leaves, litter and stubble were homogenised in 10 mM formic acid in a cooled mortar with a little sand. The homogenate was centrifuged at 25000 g (2°C) for 10 min and the supernatant was transferred to polysulphone centrifugation filters (Size 500- μl , mesh 0.45 μm ; Micro VectraSpin, Whatman Ltd., Maidstone, UK) and spun at 5000 g (2°C) for 5 min. The filtered solution was used for analysis of NO_3^- and NH_4^+ concentrations on a flow injection system (Quick Chem instrument, Lachat Instruments INC, Milwaukee, USA). Tissue extracts were also analysed for total soluble N concentration (so-called substrate N) using an ANCA-SL Elemental Analyser coupled to a 20-20 Tracermass Mass Spectrometer (SerCon Ltd., Crewe, UK). The same equipment was used for analysis of total N and C concentrations

in oven dried plant material ground to a fine powder. For bulk tissue pH measurements, 0.2 g sample of leaf material was homogenized in 2 ml of deionized water in a cooled mortar with a little sand. The homogenate was centrifuged at 14 000 g (4°C) for 10 min and pH in the supernatant measured with a microelectrode (Metrohm, Herisau, Switzerland).

2.3 Soil sampling and analysis

Soil samples were taken at least every third to fourth day with a soil auger at random positions over the field. Soil cores were separated into two layers (0–10 cm and 10–30 cm) and frozen at -18°C . A sub-sample was analysed for moisture content by calculating % weight loss after drying the soil for 24 h at 108°C . Another sub-sample (10 g) was used for pH measurements after extraction for 1 h in 25 ml 0.01 M CaCl_2 . Plant available NH_4^+ and NO_3^- were analysed with flow injection after extraction of 25 g of soil in 50 ml 2M KCl.

2.4 Calculation of the NH_3 compensation point

The stomatal NH_3 compensation point (χ_{NH_3} mol NH_3 mol $^{-1}$ air) at 25°C was calculated by use of Eq. (1) derived from Husted and Schjoerring (1996) taking into account that $K_d \ll [\text{H}^+]_{\text{apoplast}}$ within the range of apoplastic pH values:

$${}^{25}\chi_{\text{NH}_3} = K_{\text{H},25} \times K_{\text{d},25} \times \Gamma = 10^{-11.01} \times \Gamma \quad (1)$$

Γ is the dimensionless ratio between the apoplastic NH_4^+ and H^+ concentrations, and K_{H} and K_{d} are thermodynamic constants of $10^{-1.76}$ atm l mol $^{-1}$ and $10^{-9.25}$ mol l $^{-1}$ at 25°C , respectively. Equation (1) literally calculates the pressure of NH_3 (unit: atm), which according to Dalton's law of partial pressures is equal to the mol fraction (or volume fraction) at a given atmospheric pressure.

The calculated χ_{NH_3} at 25°C (T_{ref}) was adjusted to the actual canopy temperature T_c by the following equation derived from Husted and Schjoerring (1996):

$$\ln(T_c \chi_{\text{NH}_3} / T_{\text{ref}} \chi_{\text{NH}_3}) = (\Delta H_{\text{dis}}^0 + \Delta H_{\text{vap}}^0) / R \times (1/T_{\text{ref}} - 1/T_c) = 34.868 - 10395.91 / T_c \quad (2)$$

$T_c \chi_{\text{NH}_3}$ is the requested NH_3 compensation point at the actual canopy temperature T_c (K), ΔH_{dis}^0 the enthalpy of NH_4^+ dissociation (52.21 kJ mol $^{-1}$), ΔH_{vap}^0 the enthalpy of vaporization (34.18 kJ mol $^{-1}$), and R the gas constant (0.00831 kJ K $^{-1}$ mol $^{-1}$).

3 Results

3.1 Apoplastic parameters

The apoplastic NH_4^+ concentration was below $50 \mu\text{M}$ in the tall grass both before and after the main field was cut

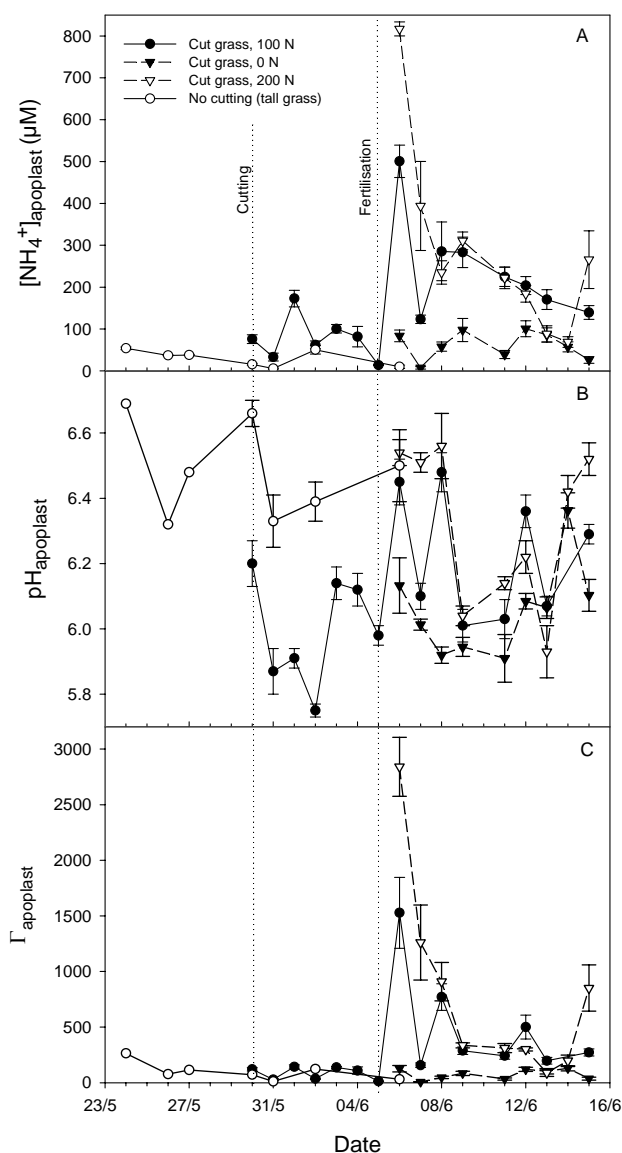


Fig. 1. Temporal variation in A) apoplastic $[\text{NH}_4^+]_{\text{apoplast}}$, B) apoplastic pH, and C) the ratio of $[\text{NH}_4^+]_{\text{apoplast}}$ to $[\text{H}^+]_{\text{apoplast}}$ (Γ ; dimensionless) in four treatments of a *Lolium perenne* dominated sward. The grass was cut on 29 May and lifted for silage on 31 May. An area of 100 m^2 was left uncut for additional sampling of tall grass. Fertilizer (100 kg N ha^{-1} in calcium ammonium nitrate) was applied on the main field on 5 June. A 100 m^2 plot was left unfertilized and another plot of same size was applied 200 kg N ha^{-1} in calcium ammonium nitrate. Vertical dotted lines indicate times of cutting and fertilisation, respectively. Values are means of three replicates \pm S.E.

(Fig. 1a). New grass leaves emerging after cutting showed slightly elevated apoplastic NH_4^+ concentrations compared to leaves of the tall grass (Fig. 1a). Following application of 100 kg N ha^{-1} to the main field 6 days after cutting of the grass, apoplastic NH_4^+ concentrations rapidly

Table 1. Tissue extracts of green leaves, stems and senescent leaves (litter) of the main field analysed for pH, NH_4^+ concentration and the ratio Γ between $[\text{NH}_4^+]$ and $[\text{H}^+]$. Means of 3 replicates \pm SE.

	pH	$[\text{NH}_4^+]$, mM	Γ
Green leaves	6.33 ± 0.02	1.79 ± 0.01	3827 ± 171
Stems	6.37 ± 0.04	1.15 ± 0.14	2696 ± 282
Senescent leaves	7.03 ± 0.05	16.2 ± 1.2	$173\,586 \pm 12\,917$

peaked at around $500 \mu\text{M}$, but thereafter decreased over the next 10 days until reaching almost the same level as before fertilisation (Fig. 1a). Plants on a plot receiving 200 kg N ha^{-1} attained a maximum apoplastic NH_4^+ concentration around $800 \mu\text{M}$ (Fig. 1a). When no nitrogen was applied (0 N plot) apoplastic NH_4^+ concentrations remained below $100 \mu\text{M}$ throughout the experimental period.

Apoplastic pH was higher in the tall grass compared to the cut grass (Fig. 1b). Fertilisation caused a transient increase in apoplastic pH without showing any difference between plants receiving 100 or 200 kg N ha^{-1} (Fig. 1b). The ratio between apoplastic NH_4^+ and H^+ concentrations (Γ_{apoplast}) ranged from 10 to 150 before fertilisation (Fig. 1c). The slight increase in apoplastic NH_4^+ following cutting was counteracted by decreasing pH (Fig. 1b). Accordingly, Γ_{apoplast} hardly changed between cutting and fertilisation (Fig. 1c). During the first 2 days after fertilisation, Γ_{apoplast} increased to above 1000, but thereafter started to decrease in parallel with the NH_4^+ concentration (Fig. 1c). In the 0 N plot, Γ_{apoplast} remained below 150 throughout the experiment.

The calculated stomatal NH_3 compensation point, χ_{NH_3} , corrected for temperature differences between different days (Eq. 2), was $1\text{--}2 \text{ nmol mol}^{-1}$ before cutting (data not shown). This level was maintained for unfertilised grass after cutting. After N fertilisation, χ_{NH_3} peaked at $15\text{--}25 \text{ nmol mol}^{-1}$ but decreased to $3\text{--}4 \text{ nmol mol}^{-1}$ already 4 days after fertilisation (data not shown).

3.2 Bulk tissue nitrogen status

Bulk tissue NH_4^+ concentrations of the tall grass as well as the cut grass prior to fertilisation were lower than $2 \mu\text{mol g}^{-1}$ fresh weight (Fig. 2a). After fertilisation, bulk tissue NH_4^+ increased rapidly and substantially, peaking around $14 \mu\text{mol g}^{-1}$ FW with little difference between 100 and 200 kg N ha^{-1} treatments (Fig. 2a). Plants not receiving N fertilizer (0 N treatment) maintained a bulk tissue NH_4^+ level below $4 \mu\text{mol g}^{-1}$ FW.

Bulk tissue NO_3^- concentrations were extremely low in the tall grass (Fig. 2b), while in the new leaves developing after cutting, NO_3^- increased considerably. Fertilisation caused a dramatic increase (4 to 5 fold) in bulk tissue NO_3^- and the

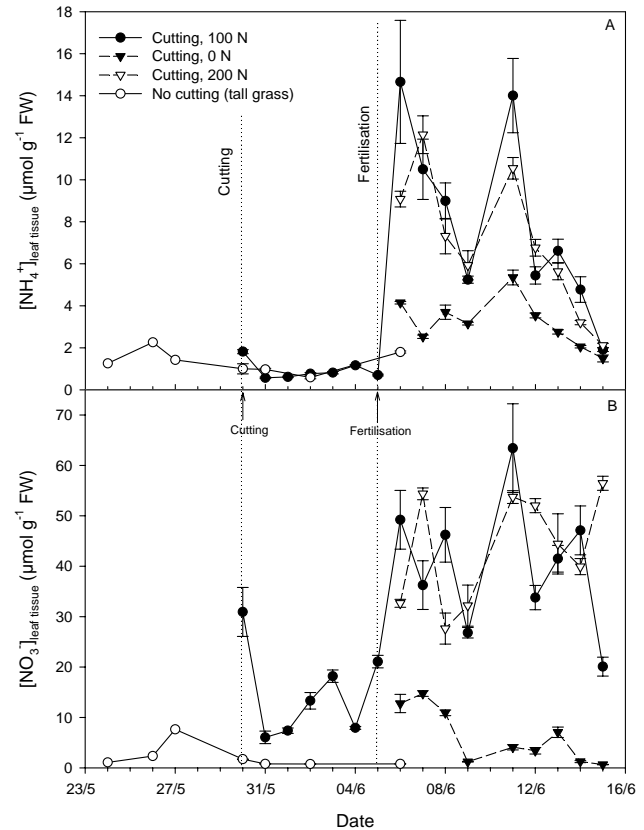


Fig. 2. Temporal variation in (A) leaf $[\text{NH}_4^+]$ and (B) leaf $[\text{NO}_3^-]$ on a fresh weight basis in four treatments of a *Lolium perenne* dominated sward. Details on experimental treatments are given in Fig. 1. Values represent means of three replicates \pm S.E.

high level remained until the end of the experiment (Fig. 2b). In unfertilised grass, NO_3^- concentrations decreased towards the end of the experiment to values similar to those of the tall grass.

Total N concentration in the tall grass leaves decreased from 3% (dry weight basis) before cutting to about 2% 9 days later (Fig. 3b). The remaining part of the cut stems (stubble) also had a total N concentration of ca. 2% throughout the rest of the experimental period. In the newly produced leaves of the 100 N treatment (main field), the total N concentration increased from around 3% just after cutting to around 5% at the end of the experiment (Fig. 3b). The final foliar N concentration in the 0 and 200 N treatments were 3.5 % and 5.5 %, respectively (data not shown).

Tissue extracts were also analysed for total soluble N concentration which can be interpreted as a dynamic N pool available for plant growth. This so-called “substrate N” was very high in leaves remaining or developing after cutting (Fig. 3a). Following fertilisation, plants in the 100 and 200 N treatments had significantly higher substrate N than unfertilised grass (0 N treatment). Substrate N constituted between 10 and 40% of total leaf N.

Table 2. Correlation coefficient table for all the different parameters measured and calculated in grass plants and soil from the 100 N treatment (main field). r^2 values with level of significance (*=0.05; **=0.01, ***=0.001). χ_{NH_3} is the stomatal compensation point of green leaves.

	$[\text{NH}_4^+]_{\text{apo}}$	pH_{apo}	$[\text{NH}_4^+]_{\text{tissue}}$	χ_{NH_3}	$[\text{NH}_4^+]_{\text{litter}}$	$[\text{NH}_4^+]_{\text{guttation}}$	SubstN	TotN	$[\text{NH}_4^+]_{\text{soil}}$
$[\text{NH}_4^+]_{\text{apo}}$	0.08	—	0.60***	0.84***	0.50***	0.22	0.48***	0.27	0.23
pH_{apo}	0.08	—	0.02	0.21	0.07	0.02	0.16	0.15	0.004
$[\text{NH}_4^+]_{\text{tissue}}$	0.60***	0.02	—	0.48***	0.73***	0	0.35**	0.42**	0.29
χ_{NH_3}	0.84***	0.21	0.48***	—	0.46***	0.01	0.25*	0.15	0.21
$[\text{NH}_4^+]_{\text{litter}}$	0.50***	0.07	0.73***	0.46***	—	0.34	0.40	0.21	0.47*
$[\text{NH}_4^+]_{\text{guttation}}$	0.22	0.02	0	0.01	0.34	—	0.19	0.06	0.19
SubstN	0.48***	0.16	0.35**	0.25*	0.40	0.19	—	0.23	0.44
TotN	0.27	0.15	0.42**	0.15	0.21	0.06	0.23	—	0.68*
$[\text{NH}_4^+]_{\text{soil}}$	0.23	0.004	0.29	0.21	0.47*	0.19	0.44	0.68*	—

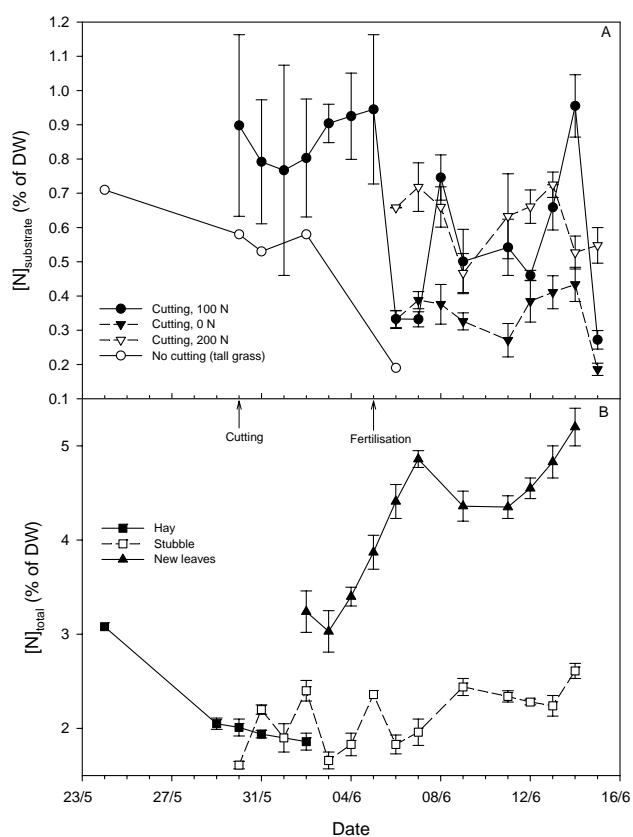


Fig. 3. Temporal variation in (A) concentration of total soluble N ($[\text{N}]_{\text{substrate}}$) in leaf tissue, and (B) total N in new leaves, stubble and hay of a *Lolium perenne* dominated sward. DW=dry weight. Details on experimental treatments are given in Fig. 1. Arrows indicate times of cutting and fertilisation, respectively. Values represent means of three replicates \pm S.E.

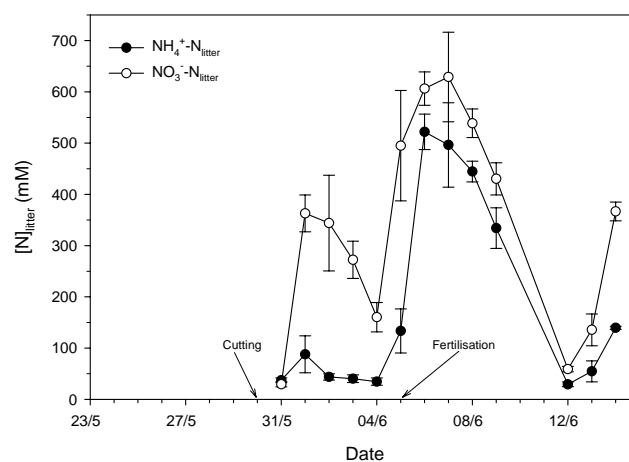


Fig. 4. Temporal variation in litter NH_4^+ and NO_3^- concentrations in a *Lolium perenne* dominated sward. Details on experimental treatments are given in Fig. 1. Arrows indicate times of cutting and fertilisation, respectively. Values represent means of three replicates \pm S.E.

The litter component of the grassland consisting of senescent plant leaves either attached to the lower part of the stems or lying on the ground constituted about 20% of the total above-ground biomass before cutting (data not shown). Prior to cutting, litter concentrations of NH_4^+ and NO_3^- were below 50 mM (Fig. 3), while 3 days after cutting the NO_3^- concentration in the litter had increased to about 350 mM, while litter NH_4^+ remained below 50 mM (Fig. 4). After fertilisation (100 kg N ha⁻¹), litter NH_4^+ increased to around 500 mM but started to decrease again already after a few days (Fig. 4). Nitrate concentrations were slightly higher than NH_4^+ concentrations after fertilisation but followed the same temporal pattern (Fig. 4).

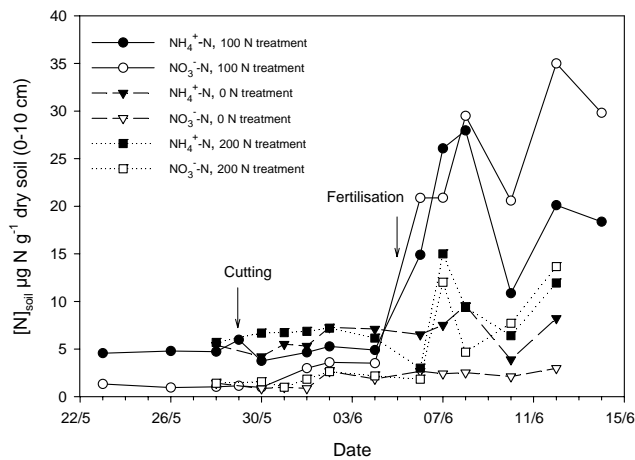


Fig. 5. Temporal variation in $[\text{NH}_4^+]$ and $[\text{NO}_3^-]$ of the top layer of soil in three treatments of a *Lolium perenne* dominated sward. Details on experimental treatments are given in Fig. 1. Arrows indicate times of cutting and fertilisation, respectively.

In parallel to apoplastic extracts, NH_4^+ and H^+ concentrations in bulk tissue extracts can be used to derive Γ values for different plant components of the sward in order to compare their potential NH_3 exchange (Table 1). Bulk pH values of green leaves and stems were similar to those in the apoplastic solution ranging between 6.2 and 6.6 (Table 1; Fig. 1b). Due to higher NH_4^+ concentrations in the tissue extracts (Table 1) than in the apoplastic solution (Fig. 1a), the resulting Γ value in green leaves was around 40 times higher than Γ_{apoplast} (Fig. 1c). Extracts of plant litter showed both higher pH and, particularly, NH_4^+ concentrations compared to leaves and stems which resulted in a Γ_{litter} value of around 173 500, i.e. 45–60 times higher than Γ_{leaf} and Γ_{stem} (Table 1).

3.3 Soil parameters

Soil NH_4^+ and NO_3^- concentrations (Fig. 5) were low before fertilisation with NH_4^+ concentrations being higher than NO_3^- concentrations in both the top soil fraction (0–10 cm) and the deeper soil fraction (10–30 cm, not shown). The top soil concentration of inorganic N increased slightly after cutting of the grass (Fig. 5). Application of 100 kg N ha^{-1} caused a dramatic increase in both NH_4^+ and NO_3^- (Fig. 5), while soil inorganic N increased less in the 200 N treatment. This difference may reflect the fact that the N fertiliser in the 200 N plot was applied by hand a few hours later and under drier conditions compared to that in the 100 N treatment of the main field where rainfall followed within a couple of hours after the application.

3.4 Correlation analysis

The NH_4^+ concentrations in apoplastic solution, leaf tissue and litter were mutually positively correlated (Table 2). The

NH_3 compensation point derived from the apoplastic measurements was also positively correlated with leaf tissue NH_4^+ , but not with total leaf N content. Apoplastic pH and the NH_4^+ concentration in guttation droplets were not significantly correlated with any of the other parameters (Table 2).

4 Discussion

Before cutting, the tall grass had low NH_4^+ concentrations in both leaf apoplast and bulk tissue (Fig. 1a, 2a). This resulted in NH_3 compensation points so low that the grass was not likely to emit NH_3 before cutting which is in agreement with atmospheric NH_3 concentration gradients above the canopy showing predominantly deposition fluxes (Milford et al., 2008). Extremely low tissue NO_3^- concentrations (Fig. 2b) as well as low soil NH_4^+ and NO_3^- levels (Fig. 5) also indicated that the small amounts of inorganic N available to the plants were efficiently taken up and utilised for growth and seed development at this stage. Before fertilisation, the soil content of NH_4^+ was 4 times higher than that of NO_3^- (Fig. 5), which is not unusual for grassland soil (Whitehead, 1995).

Between cutting and fertilisation there was a re-growth period of one week in which the grass leaves first showed increased NO_3^- concentrations (Fig. 2b) and 5 days later also increased NH_4^+ concentrations (Fig. 2a). Ryegrass has been shown to rapidly accumulate NO_3^- in both leaves and stubble after cutting as NO_3^- is involved in the osmotic adjustment (Ourry et al., 1989). Also the soluble N and total N concentrations of the leaves showed higher values during re-growth after cutting compared to the tall grass (Fig. 3a, b). The increase in these N pools was paralleled by slightly increasing NO_3^- and NH_4^+ concentrations in the soil. Several authors have reported that during the first days after cutting, uptake of N is inhibited (Bakken et al., 1998; Ourry et al., 1988). In such case plants respond by allocating N from reserves in root and stubble to the developing leaves. This inhibition of uptake could explain the increasing levels of soil NO_3^- and NH_4^+ after cutting. In *Lolium perenne* and *Bromus erectus* grown in nutrient solution, tissue NH_4^+ concentrations of expanding leaves did not start to increase until 6 days after cutting (Sutton et al., 2001). In the same experiment, apoplastic NH_4^+ concentrations increased in the new expanding leaves 3–6 days after cutting. Also in the present study, apoplastic NH_4^+ concentrations showed slightly higher values in expanding leaves during re-growth compared to the leaves of the tall grass (Fig. 1a). Increasing tissue concentrations of NH_4^+ and NO_3^- can also result from shortening of the leaf growth zone and smaller dilution of the N transported to this zone after defoliation compared to the fully expanded leaves before cutting (Schäufele and Schnyder, 2001).

Grass cutting has in several cases been reported to lead to NH_3 volatilization from grassland (Milford et al., 1999, 2002; Loubet et al., 2001). The emitted NH_3 may originate from the plants as a consequence of increased N pools during the period of leaf expansion. However, the potential for NH_3 emission did not seem to increase after cutting in the present work since Γ_{apoplast} (the ratio between apoplastic NH_4^+ and H^+) were unaltered due to counteracting effects of decreased pH and increased NH_4^+ concentrations (Fig. 1c). Another source of NH_3 emission could be the litter, i.e. senescent leaves attached to the stems or lying on the ground surface. Ammonium concentrations were considerably higher in the litter material compared to green leaves (Fig. 4) due to the protein degradation processes going on in the litter (Mattsson and Schjoerring, 2003). Senescence-related processes were probably also enhanced after cutting when both the climatic conditions and the proportion of litter out of total biomass were changed at the bottom of the canopy (David et al., 2008). In a tall canopy, NH_3 emitted from the litter can be taken up by leaves positioned higher above the ground (Husted et al., 2000b; Nemitz et al., 2000) while in the absence of a tall canopy, the litter NH_3 may escape to the atmosphere. High NH_4^+ concentrations and relatively high pH values in the litter also resulted in an extremely high Γ_{litter} value (Table 1) indicating a strong potential for NH_3 emission. In a non-fertilized grassland in the Netherlands, Wichink Kruit et al. (2007) observed an average canopy value of 2200, which is in line with that recorded for stems and young leaves in the present work (Table 1).

After fertilisation, all plant N pools increased with peak values already on the first day after fertilisation. Micrometeorological measurements also showed high NH_3 emissions after fertilisation with some contribution from the fertiliser itself during the first 2 days (Milford et al., 2008). The fertiliser was rapidly dissolved in the soil solution since it was raining the same afternoon as the main field was fertilised (Sutton et al., 2008). Consequently, the fertiliser contamination was restricted to a very short period. The fertilisation was also reflected in higher NH_4^+ concentrations in guttation droplets collected at the leaf tips in the early mornings after fertilisation (not shown).

Ammonium concentrations in both leaf tissue and apoplast started to decrease again already a few days after fertilisation (Fig. 1a, 2a) while leaf tissue NO_3^- concentrations remained high for the rest of the experiment (Fig. 2b). This may partly reflect declining soil NH_4^+ levels (Fig. 5) and partly rapid assimilation of NH_4^+ in the plant cells, while NO_3^- was stored in the leaf cell vacuoles for later use. It has previously been shown that when NH_4NO_3 is supplied to plant roots, NH_4^+ is absorbed more readily than NO_3^- (Bloom, 1981; Clarkson et al., 1986). Nitrate accumulates in grass herbage when the rate of uptake by the roots exceeds the rate of conversion to organic N (Whitehead, 1995). The NH_4^+ concentrations in apoplast and bulk tissue were obviously sensitive parameters

responding rapidly to fluctuations in soil nitrogen availability. Also in a laboratory experiment with *Lolium perenne* and *Bromus erectus* both leaf tissue and apoplastic NH_4^+ concentrations were shown to respond rapidly to changing NH_4^+ concentrations in the nutrient solution (Mattsson and Schjoerring, 2002).

A correlation analysis revealed that the stomatal NH_3 compensation point calculated on the basis of apoplastic parameters was positively correlated with the bulk tissue NH_4^+ concentration in leaves and litter (Table 2). Some previous investigations have likewise shown good correlation between apoplastic and leaf tissue NH_4^+ concentrations (Mattsson et al., 1998; Mattsson and Schjoerring, 2002), suggesting that the tissue NH_4^+ concentration may be used as an indicator of the NH_3 compensation point. Other studies have found the correlation to depend on growth conditions (Herrmann et al., 2008) or not to be present (Hill et al., 2002).

5 Conclusions

We conclude that the management practice has a major impact on the potential plant-atmosphere NH_3 exchange in grassland by influencing both plant and soil N parameters. The NH_3 compensation point derived from apoplastic measurements was positively correlated with bulk tissue NH_4^+ concentrations in leaves and litter. This suggests that measurements of NH_4^+ and pH in bulk extracts of plant material in grassland can be used as a simple indicator of the NH_3 exchange potential.

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