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Enantioselective Copper-Catalyzed Reductive Coupling of Vinylazaarenes with *N*-Boc Aldimines

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Abstract: The diastereo- and enantioselective reductive coupling of vinylazaarenes with *N*-Boc aldimines is described. The reactions proceed using chiral Cu–bisphosphine complexes in the presence of TMDS as a hydride source to give reductive coupling products in moderate to high enantioselectivities.

Key words: asymmetric catalysis, copper, enantioselectivity, imines, vinylazaarenes

The broad significance of aromatic nitrogen heterocycles (azaarenes) in chiral biologically active compounds and other functional molecules has prompted our group to investigate the potential of C=N-containing azaarenes as activating groups in new catalytic enantioselective processes. During this program, we have reported enantioselective Cu-catalyzed reductions of β , β -disubstituted alkenylazaarenes and Cu-catalyzed reductive couplings of alkenylazaarenes with ketones (Scheme 1a). This latter process enables the synthesis of products containing an azaarene and two stereogenic centers, including a tertiary alcohol. The ability to employ other types of electrophiles would be beneficial to expand the range of accessible products.

Krische and co-workers have described the racemic Rh-catalyzed hydrogenative coupling of vinylazines with N-sulfonylaldimines (Scheme 1b),³ and we envisaged that a related enantioselective variant employing chiral copper hydride chemistry⁴ could be developed. Although the intermolecular reductive aldol reaction⁵ of α , β -unsaturated carbonyl compounds, catalyzed by chiral copper–hydride complexes, has been successful,⁶ the corresponding reductive Mannich reactions⁷ have been less well studied.⁸ To date, the only report of enantioselective copper-catalyzed reductive Mannich reactions is from Kanai, Shibasaki, and co-workers, who described the reductive coupling of α , β -unsaturated esters with N-phosphinoyl ketimines.^{7f}

Herein, we describe the enantioselective Cu-catalyzed reductive coupling of vinylazaarenes with *N*-Boc aldimines. Both vinylazines and vinylazoles are effective substrates, and the Boc-protection in the products is advantageous for subsequent deprotection.

Our investigations began with an evaluation of chiral

a) Enantioselective reductive coupling of alkenylazaarenes with ketones (ref. 2d)

b) Racemic reductive coupling of vinylazines with N-sulfonylaldimines (ref. 3)

Scheme 1 Existing catalytic reductive coupling reactions of alkenylazaarenes.

bisphosphines, reductants, and reaction conditions for copper-catalyzed reductive coupling of vinylquinoline (1a) with N-Boc aldimine 2a (1.1 equiv), which led to the identification of (S)-DTBM-SEGPHOS (L1) as an effective chiral ligand. In the presence of Cu(OAc)₂·H₂O (5 mol%), L1 (5 mol%) and 1,1,3,3tetramethyldisiloxane (TMDS, 1.2 equiv), the reaction proceeded smoothly in THF at room temperature to give reductive coupling product 3a as the anti-diastereomer in 63% yield, >19:1 dr, and 85% ee (Scheme 2). 9,10 Under these conditions, various other vinylazaarenes 1b-1f were also effective in reactions with imine 2a, which gave products **3b–3f** in 58–74% yield and 75–92% ee. Besides quinolone (3a), azaarenes that were tolerated included quinoxaline (3b), a bromopyridine (3c), a benzoxazole phenylpyridazine (3d), (**3e**), and benzothiazole (**3f**). High diastereoselectivities (≥10:1 dr) observed with vinylazaarenes containing benzannulation (3a, 3b, 3e, and 3f), while 2-bromo-6vinylpyridine and 3-phenyl-6-vinylpyridazine resulted in more modest diastereoselectivities (3c and 3d). In the former case, the diastereomers were difficult to separate completely, and the minor isomer was formed in a much lower enantiomeric excess (20% ee) compared with the major isomer (80% ee).

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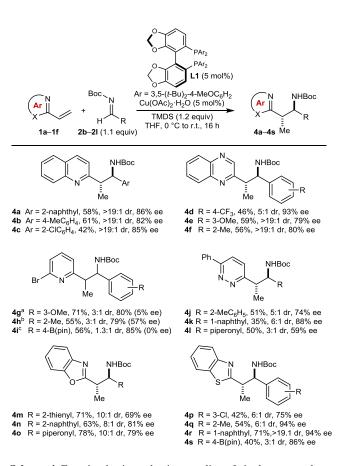
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Scheme 2 Enantioselective reductive coupling of vinylazaarenes 1a–1f with *N*-Boc imine 2a. Reactions were conducted using 0.30 mmol of 1a–1f. Unless stated otherwise, yields are of pure isolated major diastereomers. Diastereomeric ratios were determined by ¹H NMR analysis of the unpurified reaction mixtures. Enantiomeric excesses were determined by chiral HPLC analysis. ^a Yield of a 4.3:1 mixture of diastereomers. The enantiomeric excess of the minor diastereomer is indicated in parentheses.

The reductive coupling of 4-phenyl-2-vinylthiazole (1g) with imine 2a proceeded with low enantioselectivity using ligand L1. Fortunately, (R,R)-Ph-BPE (L2) provided improved results, and gave 3g in 88% yield, >19:1 dr, and 87% ee (Scheme 3).

Scheme 3 Reductive coupling of 1g with 2a using ligand L2.

Various other (hetero)aromatic *N*-Boc aldimines **2b–21** also underwent reductive coupling with vinylazaarenes **1a–1f**, giving products **4a–4s** in 59–94% ee for the major diastereomers (Scheme 4). The diastereoselectivities of these reactions ranged from 1.3:1 dr (**4i**) to >19:1 dr (**4a–4c**, **4e**, **4f**, and **4r**). Compared with the results shown in Scheme 2 and with similar reactions using ketones as electrophiles, these reactions often provided lower yields of the reductive coupling products due to more prevalent side-reactions, such as simple reduction of both reaction partners without C–C bond formation. Although no definitive trends could be deduced from the particular combinations of substrates that resulted in the highest enantioselectivities, values of 85% ee or higher were observed in several cases (**4a**, **4c**, **4d**, **4i**, **4k**, and **4q–4s**). With respect to the imine, a range of substituents



Scheme 4 Enantioselective reductive coupling of vinylazaarenes 1a–1f with various *N*-Boc imines. Reactions were conducted using 0.30 mmol of 1a–1f. Unless stated otherwise, yields are of pure isolated major diastereomers. Diastereomeric ratios were determined by ¹H NMR analysis of the unpurified reaction mixtures. Enantiomeric excesses were determined by chiral HPLC analysis. Where measured, the enantiomeric excess of the minor diastereomer is indicated in parentheses. ^a Yield of a 4.7:1 mixture of diastereomers. ^b Yield of a 3:1 mixture of diastereomers.

(methyl, trifluoromethyl, chloro-, methoxy, dioxolane, or boronate esters) at various positions on the phenyl ring were tolerated. Furthermore, reactions of imines containing 1-naphthyl, 2-naphthyl, or 2-thienyl groups were also successful. As observed previously (Scheme 1), the reactions of 2-bromo-6-vinylpyridine and 3-phenyl-6-vinylpyridazine generally gave lower diastereoselectivities (4g–4l) compared with the other vinylazaarenes. Lower diastereoselectivities were also obtained with an imine containing a *p*-pinacol boronic ester (4i and 4s). Interestingly, the minor diastereomers obtained for products 4g and 4i were obtained in low or non-existent enantiomeric excesses.

Once again, (R,R)-Ph-BPE (**L2**) was a superior ligand compared with **L1** in a reductive coupling involving 4-phenyl-2-vinylthiazole (Scheme 5).

Scheme 5 Reductive coupling of 1g with 2f using ligand L2.

The reaction of 2-vinylbenzothiazole (1f) with an imine 2m (2.0 equiv) containing an aliphatic substituent (cyclohexyl) was also studied (Scheme 6). Although low stereoselectivities were obtained using ligands L1 or L2, the Josiphos ligand SL-J006-1 (L3) gave improved results, and provided two diastereomers of 4u in 73% and 16% yields, in 82% and 75% ee for the major and minor isomers, respectively.

Scheme 6 Reductive coupling of **1f** with an aliphatic *N*-Boc imine **2m** using ligand **L3**. ^a Values in parentheses refer to the yield and enantiomeric excess of the minor diastereomer.

Finally, removal of the Boc group from **4a** was achieved under acidic conditions (using HCl generated by the reaction of TMSCl with MeOH), which provided amine **5** in 90% yield with no loss of enantiopurity (Scheme 7).

Scheme 7 Deprotection of 4a.

In summary, the results presented herein demonstrate the ability of chiral copper–bisphosphine complexes to catalyze the enantioselective reductive coupling of vinylazaarenes with hetero(aryl) *N*-Boc imines. The reactions provide reductive coupling products with moderate to high enantioselectivities (up to 94% ee).

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Data for 3f: $R_f = 0.32$ (20% EtOAc/petroleum ether); mp 142-145 °C (EtOAc/petroleum ether); [α]_D²⁴ +55.0 (*c* 1.00, CHCl₃); IR (film): 2979, 2928, 1713 (C=O), 1498, 1390, 1365, 1170, 1022, 759, 700 cm⁻¹; ¹H NMR (400 MHz, (CD₃)₂CO): δ = 7.98 (t, J = 8.9 Hz, 2 H), 7.54-7.45 (m, 1 H), 7.45-7.35 (m, 3 H), 7.30 (t, J = 7.4 Hz, 2 H), 7.23 (t, J = 7.5 Hz, 1 H), 6.80 (d, J = 7.5 Hz, 1 H), 5.15-4.90 (m, 1 H), 3.88-3.65 (m, 1 H), 1.37 (d, J = 6.8 Hz, 3 H), 1.28 (s, 9 H); ¹³C NMR (125.8 MHz, (CD₃)₂CO): δ = 174.7, 155.9, 154.1, 142.9, 135.6, 129.1, 128.0, 127.8, 126.8, 125.7, 123.4, 122.6, 78.9, 60.3, 45.1, 28.5, 19.5; HRMS (ESI) m/z calcd for C₂₁H₂₅N₂O₂S [M+H]⁺: 369.1631, found: 369.1634; HPLC: Chiralpak IC column (98:2 hexane:*i*-PrOH, 0.8 mL/min, 280 nm, 25 °C); t_R (major) = 18.9 min, t_R (minor) = 27.9 min; 88% ee.

(10) Where indicated, the relative and absolute stereochemistries of the products were assigned by analogy with those of products **3f**, **3g**, **4d**, **4k**, **4q**, and **4t**, which were determined by X-ray crystallography (see Supporting Information for details). CCDC 1019731–1019736 contain the supplementary crystallographic data for this paper These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif or by writing to the Cambridge Crystallographic Data Centre, 12, Union

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