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Two unique star-like [Mn^{IV}Mn^{III}₂Ln^{III}] clusters: magnetic relaxation phenomena

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Thomais G. Tziotzi,^{*a*} Milosz Siczek,^{*b*} Tadeusz Lis,^{*b*} Ross Inglis^{*,*c*} and Constantinos J. Milios^{*,*a*}

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Employment of H_3L^2 (= 2-(β -naphthalideneamino)-2-hydroxyethyl-1-propanol) in manganese-lanthanide cluster chemistry has led to the isolation of two new isostructural tetrametallic [$Mn^{III}_2Mn^{IV}Ln^{III}$] complexes (Ln = Gd, Dy), with the Dy analogue displaying temperature and frequency dependent out-of-phase signals, thus indicating possible single molecule magnetism behaviour.

The last decade has witnessed a major boost in the field of molecular magnetism. Since the discovery of the prototype single molecule magnet (SMM) [Mn₁₂OAc] that could retain its magnetization once magnetized at temperatures below ~3 K,^[1] various significant achievements have been accomplished: i) in 2007 a hexanuclear [Mn^{III}₆] cluster was reported with an energy barrier for the reorientation of the magnetization of $U_{\rm eff}$ = 86.4 K, breaking the record of U_{eff} = 60-64 K held by the prototype SMM for ~15 years,^[2] ii) in 2009 a [Dy^{III}₄] cluster was reported elevating the energy barrier to 170 K^[3] iii) in 2011 Long *et al.* synthesized a radical-bridged [Dy^{III}₂] complex with a blocking temperature of 8.3 K and $U_{\rm eff}$ = 178 K,^[4] while the terbium analogue displayed a blocking temperature of ~14 K and U_{eff} = 326 K,^[5] iv) in 2013 Winpenny *et al.* reported polynuclear lanthanide alkoxide complexes with $U_{\rm eff}$ values higher than 800 K,^[6] and v) very recently a number of pentagonal bipyramidal mononuclear Dy^{III} complexes have been reported with blocking temperatures reaching 20 K^[7] and $U_{\rm eff}$ values > 1000 K.^[8] From these recent benchmarks, it is apparent that the use of lanthanide ions in molecular magnetism has become a key element towards the improvement of the magnetic properties of discrete clusters and their potential technological applications as magnetic memory devices.

⁺ Footnotes relating to the title and/or authors should appear here. Electronic Supplementary Information (ESI) available: Full details of the experimental microanalyses and crystallographic data. See DOI: 10.1039/x0xx00000x

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We previously reported the use of the naphthalene-based triol ligand 2-(β -naphthalideneamino)-2-hydroxymethyl-1-propanol, H₃L^{1,[9]} in Co(II/III), Ni(II) and Cu(II) chemistry,^[10] and recently expanded our studies to mixed-metal Mn/Ln chemistry reporting a family of octanuclear [Mn^{III}₆Ln^{III}₂] complexes,^[11] and two dodecanuclear [Mn^{III}₆Ln^{III}₆] clusters (Ln = Gd, Dy).^[12] Herein, we present our efforts towards the use of H₃L² (=2-(β -naphthalideneamino)-2-hydroxyethyl-1-propanol) in Mn/Ln cluster chemistry, and report the synthesis, structures and magnetic properties of two tetrametallic [Mn^{IVI} $_{2}$ Ln^{III}] clusters (Ln = Gd, Dy).



 $\label{eq:Scheme 1.} \mbox{ The structures of H_3L^1 and H_3L^2 (top), and the coordination modes of H_3L^2 in 1 and 2 using Harris notation (bottom).}$

From the reaction of $Mn(ClO_4)_2 \cdot 6H_2O$, $Ln(NO_3)_3 \cdot 6H_2O$ (Ln = Gd, Dy), H₃L² and NH₄SCN in 1:1:1:3 ratio in the presence of base, NEt₃, in MeOH we were able to isolate and characterize two new heterometallic tetranuclear clusters with the general formula $[Mn_3Ln(L^2)_2(HL^2)_2(naph)(NCS)(H_2O)(MeOH)_{1.8}](0.5NO_3)(0.5ClO_4) \cdot 1.8$ MeOH·0.6H₂O (naph: naphthaldehyde; Ln: Gd, 1·1.8MeOH·0.6H₂O; Dy, **2**·1.8MeOH·0.6H₂O). The structure of **1** was solved by singlecrystal X-ray crystallography, while complex **2** is isostructural with **1** based on elemental analyses, IR spectra and powder XRD patterns (Figs S1 and S2). Cluster **1** crystallizes in the triclinic P-1 space group

^{a.} Department Of Chemistry, University of Crete, Voutes 71003, Herakleion, Greece. Fax: +30-2810-545001; Tel: +30-2810-545099; E-mail: <u>komil@uoc.gr</u>

^{b.} Faculty of Chemistry, University of Wroclaw, Joliot-Curie 14, Wroclaw 50-383, Poland.

^c EaStCHEM School of Chemistry, The University of Edinburgh, David Brewster Road, EH9 3FJ, Edinburgh, UK. Email: <u>r.inglis@ed.ac.uk</u>

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(Figure 1); its metallic core describes a central Gd^{III} ion bridged to three peripheral manganese centers via six μ -OR⁻ groups from the four deprotonated/partially-deprotonated naphthalene-triol ligands present in the structure. Two of the ligands are fully deprotonated, L³⁻, adopting a 3.2211 coordination mode (Harris notation),^[13] while the remaining two are partially deprotonated, HL²⁻, with a 2.2101 binding mode. The lanthanide center completes its eight-coordinate sphere with a chelating naphthaldehyde group, arising upon the in situ breaking of the Schiff-base triol ligand. All three manganese centers are six-coordinate, completing their coordination environment with the presence of solvate molecules (for Mn2) and one terminal thiocyanate ligand (for Mn3). Mn2 and Mn3 are in the 3+ oxidation state as confirmed by bond-valence sum calculations^[14] (BVS values: 3.14 and 3.06, for Mn2 and Mn3, respectively), with both of them adopting a JT-distorted octahedral geometry, while Mn1 is found in the 4+ oxidation level (BVS value: 3.94). The coordination sphere of the lanthanide center was found following SHAPE^[15] analysis to be square antiprismatic (Table S1). The metallic cores of 1 and 2 have a star-like topology, which is extremely rare in Mn/Ln chemistry, with only two previous examples reported,^[16] while this is the first case in which the 3d centers are found in a mixed-valent state.



Figure 1. The molecular structure of the cation of **1**, highlighting the orientation of the two *Jahn-Teller* axis (bold line). Solvent molecules and H atoms are omitted for clarity. Color code: Mn^{IV} = purple, Mn^{III} = red, Gd^{III} = dark-yellow, O = green, S = yellow, N = blue, C = grey.

Variable temperature *dc* magnetic susceptibility data were collected for both complexes in the temperature range 5-300 K under an applied field of 0.1 T, and are plotted as $\chi_{\rm M}T$ versus *T* plots in Figure 2. For **1**, the room temperature $\chi_{\rm M}T$ value of 15.30 cm³ K mol⁻¹ is very close to the expected value of 15.75 cm³ K mol⁻¹ for two non-interacting Mn^{III} (with *g* = 2.0), one Mn^{IV} (*g* = 2.0) and one Gd^{III} ions (*g* = 2.00). Upon cooling the value of $\chi_{\rm M}T$ remains almost unchanged until ~100 K, below which it decreases to a minimum value of 5.09 cm³ K mol⁻¹ at 5 K. For **2**, the room temperature $\chi_{\rm M}T$ value of 21.75 cm³ K mol⁻¹ is very close to the expected value of 21.97 cm³ K mol⁻¹ for two non-interacting Mn^{III} (with *g* = 2.0), one Mn^{IV} (*g* = 2.0) and one Dy^{III} ions (*S* = *5/2*, *L* = *5*, *J* = 1*5/2*, *g_j* = 4/3). Upon cooling the value of $\chi_{\rm M}T$ remains fairly constant until ~120 K, below which it decreases to a final value of 12.01 cm³ K mol⁻¹ at 5 K.



Figure 2. $\chi_M T$ vs. T plot for complexes **1** ([Mn₃Gd]) and **2** ([Mn₃Dy]) under an applied *dc* field of 0.1 T. The solid line represents a simulation of the magnetic susceptibility data for **1** (see text for details).



Figure 3. M vs. H plot for complex 1 ([Mn₃Gd]) in the 1 – 7 T and 2.0 – 6.0 K field and temperature range. The solid lines represent a simulation of the magnetization isotherms in the 2-6 K temperature range (see text for details).

In Figure 3, the magnetization data is presented for complex **1** as a *M vs. H* plot. We were able to successfully simulate both the magnetic susceptibility and magnetization data, assuming a 3-*J* model with: i) J_1 between Mn1-Mn2 and Mn1-Mn3, ii) J_2 between the trivalent Mn centers (Mn2-Mn3), and iii) J_3 between the Gd center and its peripheral manganese centers. Using the program PHI^[17] and employing the Hamiltonian in eqn (1)

$$\hat{H} = -2J_1(\hat{S}_1\hat{S}_2 + \hat{S}_1\hat{S}_3) - 2J_2(\hat{S}_2\hat{S}_3) - 2J_3(\hat{S}_1\hat{S}_4 + \hat{S}_2\hat{S}_4 + \hat{S}_3\hat{S}_4)$$
(1)

afforded the parameters $J_1 = -0.52 \text{ cm}^{-1}$, $J_2 = 0.12 \text{ cm}^{-1}$, $J_3 = -0.36 \text{ cm}^{-1}$, $g_{Mn(IV),Mn(III),Gd(III)} = 2.00$. These parameters lead to a spin-frustrated system with a band of near degenerate *S* states of value *S* = 0, *S* = 1 and *S* = 2, all within ~1.5 cm⁻¹, as expected due to the very weak interactions present within the cluster. The magnitude and nature of parameters J_2 and J_3 are in agreement with the previously reported star-like [Mn^{III}₃Gd] complex ($J_{Mn(III)}$.

 $Mn(III) = 0.56 \text{ cm}^{-1} \text{ vs. } 0.12 \text{ cm}^{-1} \text{ in } \mathbf{1}, J_{Mn-Gd} = -0.23 \text{ cm}^{-1} \text{ vs. } -0.36$ cm⁻¹ in 1),^[16a] and the Mn-Gd interaction, J_3 , also falls within the range for previously reported values.^[18] It is worth noting that to avoid over-parameterization problems, we treated the system with only one Mn-Gd interaction, J₃. Given the expected small magnitude of both Mn^{III}-Gd and Mn^{IV}-Gd interactions due to the core-like nature of the 4f electrons, we chose not to distinguish between the two distinct interactions present in 1. Given the relatively large magnetic moment of cluster 2, ac magnetic susceptibility measurements were performed on a polycrystalline of **2** in the 1.8-8 K range in zero applied *dc* field and a 3.5 G ac field oscillating at 200-1000 Hz frequency. The inphase signal decreases upon decreasing temperature, indicating the presence of low-lying excited states with larger "S" values than the ground-state (Figure 4, top), while in addition frequency-dependent out-of-phase (χ_{M}) "tails" are observed below ~3.7 K (Figure 4, bottom, and Fig. S3), suggesting possible single molecule magnetism behaviour. On the contrary, complex 1 displays no out-of-phase signals, clearly showing that the magnetic relaxation of cluster 2 is mainly due to the presence of the Dy^{III} center and not the Mn^{III} centers.



Figure 4. Plot of the in-phase χ_M 'T (top) and out-of-phase χ_M " signals (bottom) for **2** in *ac* susceptibility studies vs. *T* in a 3.5 G oscillating field at the indicated frequencies.

Conclusions

Two new heterometallic tetranuclear $[Mn^{IV}Mn^{III}_2Ln]$ (Ln = Gd, Dy) star-like clusters have been synthesized with the use of the H_3L^2 (= 2-(β -naphthalideneamino)-2-hydroxyethyl-1-propanol) ligand in mixed-metal Mn/Ln chemistry. They represent the first examples of a mixed-valent Mn/Ln complex with a star-like topology, and the magnetic properties of the Dy analogue, **2**, suggest possible single molecule magnetism behaviour.

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Notes and references

§ Crystal data for 1: $(C_{77.80}H_{80.4}GdMn_3N_5O_{16.8}S)\cdot 0.5(ClO_4)\cdot 0.5(N O_3)\cdot 1.8(CH_3OH) \cdot 0.6(H_2O), M = 1857.61, triclinic, space group P-1, a = 16.077(7) Å, b = 16.201(7) Å, c = 18.740(8) Å, a = 70.68(4)^{\circ}, \beta = 84.13(4)^{\circ}, \gamma = 68.67(3)^{\circ}, V = 4290(3)Å^3$, Z = 2, T = 100 K, R1 (I > 2\sigma) = 0.064 and wR2 (all data) = 0.201 for 32407 reflections collected, 14826 observed reflections (I > 2 σ (I)) of 21792 (Rint = 0.032) unique reflections and 1171 parameters, GOF = 1.07. CCDC reference number: 1470482.

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