# Crystal and Molecular Structure of Chelidonine 

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The crystal and molecular structure of benzophenantridine alkaloid chelidonine, $\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{NO}_{5}(\mathrm{M}=353.37)$ was solved by direct methods and refined by the least-squares technique to a conventional $R$ index of 0.117 for 1888 unique reflections. The crystals are monoclinic, space group $P 2_{1}$. Both, B and C rings have half chair conformation and are in cis junction. The $N$-methyl and C14-hydroxyl groups are in equatorial and axial positions, respectively. The OH-group forms intramolecular hydrogen bond with N atom. Conformation of the molecule is quite different from that found in chelidonine-p-brombenzoate ${ }^{1}(\mathrm{CBB})$.

## INTRODUCTION

In the course of our studies of natural products we investigated the crystal and molecular structure of benzophenantridine alkaloid chelidonine isolated from Chelidonium majus L. (Papaveraceae). In Europe and in Asia it has been used from time immemorial as medicinal plant. The aerial parts of the plant were collected at a locality of Fruska Gora in the flowering phase. Chelidonine was isolated as a dominant alkaloid. The plant material was extracted with methanol ${ }^{2}$ and the alkaloids were separated by column chromatography ${ }^{3}$. Chelidonine has physiological effects similar to papaverin because it affects spasmodically the path of billious acids and bronchi. ${ }^{4}$ The relative configuration of chelidonine was determined by Šantovy
et al. ${ }^{5}$ while its total synthesis was performed by Oppolzer and Keller. ${ }^{10}$. Chelidonine has been applied in the homeopat treatment of hepatitic diseases. ${ }^{11}$

## EXPERIMENTAL

## Crystal Data

From single crystal diffractometry: $a=0.8964(3), b=0.9115(3), c=1.0622(3) \mathrm{nm}, \beta=$ $93.32(3)^{\circ}, V=0.866 \mathrm{~nm}^{3}, Z=2, D_{\mathrm{x}}=1.354 \mathrm{Mgm}^{-3}$, space group $P 2_{1}, \mathrm{~F}(000)=372$, $\mu($ Mo$K \alpha)=0.91 \mathrm{~cm}^{-1,} \lambda=0.07107 \mathrm{~nm}$, crystal size: $0.219 \times 0.219 \times 0.292 \mathrm{~mm}$.

## Intensity Data, Structure Determination and Refinement

Intensity data were collected on an Enraf-Nonius CAD-4 diffractometer (Berne) equipped with graphite monochromator using $\omega-2 \Theta$ scan in the range $2 \Theta<53.9^{\circ}$. Cell constants were determined by least-squares from setting angles of 20 reflections collected in the range of $4.4^{\circ}<\Theta<8.4^{\circ}$ scanned by $M o K \alpha$ radiation. The systematic absences are $k=2 n+1$ in OkO. $h_{\max }=11, k_{\max }=11, l_{\max }=13$. A standard reflection $(\overline{2} \overline{2} 0)$ was monitored every 200 min . but no intensity variations were recorded. Of the 2007 measured reflections, 1888 with $F_{\mathrm{o}}>3 \sigma\left(F_{\mathrm{o}}\right)$ were taken as observed. No correction for absorption was applied.


Figure 1. A perspective view of the molecule with the numbering sheme. Numbers refer to C atoms unless otherwise indicated.

The structure was solved by direct methods using the SHELX86 program. ${ }^{6}$ An E-map computed by using the phase set for 291 normalized structure factors having $E>1.20$ revealed all 26 non-hydrogen atoms $(R=0.240)$. Refinement was performed using the SHELX76 program $^{7}$ by the least-squares procedure, minimizing $\Sigma w\left(\left|F_{\mathrm{o}}\right|-\left|F_{\mathrm{c}}\right|\right)^{2}$ for 235 parameters, assuming unit weights for all reflections. Hydrogen atoms were generated from assumed geometry. They were not refined but included in the structure factor calculations. Final $R$ is 0.117 for the 1888 reflections observed. The high $R$-value is due to the poor quality of the crystal. In the final stage, the maximum ratio of shift to error was 0.09 . Max. and min. heights in the final difference Fourier synthesis are $425 \mathrm{e} / \mathrm{nm}^{3}$ and $-363 \mathrm{e} / \mathrm{nm}^{3}$, respectively. Scattering factors were taken from SHELX76. All calculations* were carried out on a PC AT computer.

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## DESCRIPTION OF THE STRUCTURE AND DISCUSSION

Figure 1 shows a perspective view of the molecule of the title compound computed from the fractional atomic coordinates given in Table I and II. Figure 2 shows a packing of the molecules. Bond distances and bond angles are listed in Table III.


Figure 2. Packing of the molecules.

In the five-membered rings, the C-O single bonds are of two types: the first type involves the C -atom of an aromatic ring and varies in length between 0.137 and 0,138 nm , and the second type involves the C -atom of a $\mathrm{CH}_{2}$ group and ranges from 0.143


Figure 3. Newman projection perpendicular to C9-C10 bond.
to 0.146 nm . The mean value of three C-N-C angles $\left(111^{\circ}\right)$ indicates $\mathrm{sp}^{3}$ hybridization of N -atom. The dihedral angle between A and D benzene rings is $47.9(3)^{\circ}$. The corresponding angles in two symmetrically independent CBB molecules are $84^{\circ}$ and $90^{\circ}$.

TABLE I
Atomic Coordinates $\left(\times 10^{4}\right)$ of non-H Atoms and Equivalent Isotropic Thermal Parameters $\left(\mathrm{pm}^{2}\right)$ with Estimated Standard Deviation in Parentheses.

$$
\begin{aligned}
U_{\mathrm{eq}}= & \left.\frac{1}{3} \right\rvert\, U_{11}\left(a a^{*}\right)^{2}+U_{22}\left(b b^{*}\right)^{2}+U_{33}\left(c c^{*}\right)^{2}+2\left(U_{12} a b a^{*} b^{*} \cos \gamma+U_{13} a c a^{*} c^{*} \cos \beta+\right. \\
& \left.+U_{23} b c b^{*} c^{*} \cos \alpha\right) \mid
\end{aligned}
$$

|  | $x / a$ | $y / b$ | $z / c$ | $U_{\text {eq }}$ |
| :---: | :---: | :---: | :---: | :---: |
| N | 1766(7) | 440(12) | 6632(6) | 330(10) |
| 01 | 6283(9) | 534(14) | 10988 (7) | 620(20) |
| 02 | 3857(8) | -172(13) | 10440(6) | 520 (20) |
| 03 | 217 (8) | 387(12) | 734 (6) | 490(20) |
| 04 | -54(9) | -1788(12) | 1858 (6) | $480(20)$ |
| 05 | 2754(7) | 3226(0) | 6260(6) | 370(10) |
| C1 | 5038(13) | -304(19) | 11429(11) | 640(30) |
| C2 | 4526(10) | 271(13) | 9379(9) | 370(20) |
| C3 | 6010(10) | 679(15) | 9706(9) | 420(20) |
| C4 | 6940(12) | 1168(15) | 8819(10) | $500(30)$ |
| C5 | $6303(10)$ | 1243(13) | $7559(9)$ | $360(20)$ |
| C6 | 4860(9) | 835(12) | 7233(8) | - 300(20) |
| C7 | 3889 (9) | 369 (12) | 8173(8) | 310(20) |
| C8 | 2291(9) | -66(14) | 7877(8) | $360(20)$ |
| C9 | 2813(9) | 27(11) | 5694(8) | $270(20)$ |
| C10 | 4241(10) | 984(12) | 5910 (8) | $310(20)$ |
| C11 | 2163(9) | 190(13) | 4339 (8) | 310(20) |
| C12 | 2405(10) | 1483(13) | 3651(9) | $360(20)$ |
| C13 | 3420 (11) | 2691(13) | 4142 (9) | 390 (20) |
| C14 | 3889(10) | 2608(12) | 5538(9) | 340(20) |
| C15 | 1334(10) | -972(13) | 3789 (8) | 320 (20) |
| C16 | 770 (10) | -775 (13) | 2567 (9) | $35020)$ |
| C17 | 953(10) | 477(15) | 1894(9) | $380(20)$ |
| C18 | 1752(11) | 1644(15) | 2422(9) | $400(20)$ |
| C19 | -299(13) | -1093(16) | 625(11) | $520(20)$ |
| C20 | 202(9) | -108(15) | 6394(9) | 430(20) |

TABLE II
Atomic Coordinates $\left(\times 10^{3}\right)$ of H-Atoms $\mid U=510(70) \mathrm{pm}^{2}$

|  | $x / a$ | $y / b$ | $z / c$ |
| :--- | ---: | ---: | ---: |
| H | 184 | 269 | 610 |
| H1A | 468 | 18 | 1229 |
| H1B | 160 | 42 | 856 |
| H4 | 809 | 147 | 905 |
| H5 | 698 | 162 | 682 |
| H8A | 160 | 42 | 856 |
| H8B | 219 | -125 | 791 |
| H9 | 305 | -112 | 586 |
| H10 | 506 | 59 | 529 |
| H13A | 443 | 265 | 362 |
| H13B | 286 | 372 | 396 |
| H14 | 488 | 325 | 577 |
| H15 | 114 | -197 | 431 |
| H18 | 188 | 265 | 189 |
| H19A | 29 | -168 | -8 |
| H19B | -148 | -109 | 36 |
| H20A | -23 | 24 | 547 |
| H20B | -48 | 34 | 710 |
| H20C | 18 | -129 | 645 |
|  |  |  |  |

TABLE III
Interatomic Distances (nm) and Angles $\left({ }^{\circ}\right)$ with Estimated Standard Deviation in Parentheses

| C1-01 | 0.1452(14) | C10-C14 | $0.1560(12)$ |
| :---: | :---: | :---: | :---: |
| C1-02 | 0.1453 (13) | C11-C12 | 0.1410 (13) |
| C2-02 | 0.1368 (11) | C11-C15 | 0.1402(12) |
| C2-C3 | 0.1405 (13) | C12-C13 | $0.1502(13)$ |
| C2-C7 | $0.1374(12)$ | C12-C18 | $0.1408(13)$ |
| C3-01 | $0.1377(12)$ | C13-C14 | 0.1519 (13) |
| C3-C4 | $0.1368(15)$ | C14-05 | $0.1426(11)$ |
| C4-C5 | 0.1425 (14) | C15-C16 | $0.1377(13)$ |
| C5-C6 | 0.1371 (12) | C16-04 | 0.1379 (11) |
| C6-C7 | $0.1427(11)$ | C16-C17 | $0.1362(14)$ |
| C6-C10 | 0,1487(12) | C17-03 | $0.1366(11)$ |
| C7-C8 | $0.1502(11)$ | C17-C18 | $0.1382(15)$ |
| C8-N | 0.1453 (11) | C19-03 | 0.1429 (15) |
| C9-N | 0.1457 (11) | C19-04 | 0.1460 (13) |
| C9-C10 | $0.1555(11)$ | C20-N | $0.1496(11)$ |
| C9-C11 | $0.1528(11)$ |  |  |
| 02-C1-01 | 105.4(9) | C9-C11-C15 | 118.9(8) |
| C1-01-C3 | 105.8(8) | C10-C6-C7 | 118.7 (7) |
| 01-C3-C2 | 109.3 (9) | C10-C9-C11 | $110.5(7)$ |
| C3-C2-02 | 109.2 (8) | C10-C14-05 | 112.3 (7) |
| C1-02-C2 | 106.5 (7) | C10-C14-C13 | 109.6(7) |
| C2-C3-C4 | $121.5(9)$ | C11-C12-C13 | 122.7(8) |
| 01-C3-C4 | 129.2 (9) | C11-C15-C16 | 116.8(8) |
| C3-C2-C7 | 123.3 (8) | C11-C9-N | 113.2 (6) |
| 02-C2-C7 | 127.5 (8) | C11-C12-C18 | 120.0 (9) |
| C3-C4-C5 | 115.8 (9) | C12-C13-C14 | 115.5 (7) |
| C4-C5-C6 | 123.0 (9) | C12-C11-C15 | 120.6(8) |
| C5-C6-C7 | 120.6 (8) | C12-C18-C17 | 118.3(9) |
| C5-C6-C10 | 120.6 (8) | C13-C12-C18 | 117.2 (9) |
| C6-C7-C2 | 115.7 (8) | C13-C14-05 | 109.8 (8) |
| C2-C7-C8 | $121.2(7)$ | C14-C10-C6 | 112.6 (7) |
| C6-C7-C8 | 123.0 (7) | C15-C16-C17 | 123.7 (9) |
| C7-C8-N | $111.2(7)$ | C15-C16-04 | 125.7 (9) |
| $\mathrm{C} 8-\mathrm{N}-\mathrm{C} 9$ | 110.7 (6) | C16-C17-C18 | 120.5(8) |
| $\mathrm{C} 8-\mathrm{N}-\mathrm{C} 20$ | 107.4 (7) | C16-C17-03 | 110.9 (9) |
| $\mathrm{N}-\mathrm{C} 9-\mathrm{C} 10$ | 108.1 (7) | C16-04-C19 | 104.3 (8) |
| C9-N-C20 | 115.6 (7) | C17-C16-04 | 110.6 (8) |
| C9-C10-C6 | 110.3 (7) | C17-03-C19 | 105.5(8) |
| C9-C10-C14 | 110.0 (7) | C18-C17-03 | 128.6(10) |
| C9-C11-C12 | 120.5 (8) | 03-C19-04 | 107.8(8) |

Both, B and C rings have half-chair conformation and are fused by cis-junction. Puckering parameters of Cremer and Pople ${ }^{8}$ are: $Q=0.054(1) \mathrm{nm}, ~ \varnothing=218(2)^{\circ}$, $\Theta=48(1)^{\circ}$ for ring B and $Q=0.051(1) \mathrm{nm}, \varnothing=346(2)^{\circ}, \Theta=46(1)^{\circ}$ for ring C. Contrary to the CBB , the N -methyl group is in equatorial position. In accordance with the antiperiplanar ${ }^{9}$ torsion angle, $\mathrm{C} 6-\mathrm{C} 10-\mathrm{C} 9-\mathrm{C} 11=179.4(8)^{\circ}$ (Figure 3). (in CBB molecules they are synclinal, $74^{\circ}, 81^{\circ}$ ), the C14-hydroxyl group assumes axial position, whereas the bromobenzoyl moiety in CBB is in equatorial position. The OH group is bent over ring B donating only an intramolecular hydrogen bond to N -atom with the parameters $\mathrm{O} 5-\mathrm{H}: 0.096 \mathrm{~nm}, \mathrm{~N} \ldots \mathrm{H}: 0.213 \mathrm{~nm}, \mathrm{O} 5 \ldots \mathrm{~N}: 0.273 \mathrm{~nm}, \not \subset \mathrm{O} 5-\mathrm{H} . . \mathrm{N}:$ $119^{\circ}$. The molecules are linked together by van der Waals interactions.

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## SAŽETAK

## Kristalna i molekulska struktura helidonina

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Kristalna i molekulska struktura helidonina, $\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{NO}_{5}$, odredena je direktnim metodama i utačnjena do $R=0.117$ za 1888 refleksa. Kristali pripadaju monoklinskom sistemu, a prostorna grupa je P21. Prstenovi B i C poseduju polustoličastu konformaciju, $N$-metilna grupa je u ekvatorijalnom, C14-hidroksilna grupa u aksijalnom položaju. OH grupa gradi intramolekulsku vodoničnu vezu sa $N$-atomom. Konformacija molekula bitno se razlikuje od one u he-lidonin- $p$-brombenzoatu.


[^0]:    * Lists of the observed and calculated structure factors and anisotropic thermal parameters are obtainable from the authors on request.

