

| 1 | Potential impact of dairy yeasts on the typical flavour of traditional ewes' |
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| 2 | and goats' cheeses |
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25 Abstract

The contribution of Debaryomyces hansenii, Kluyveromyces lactis and 26 Kluyveromyces marxianus strains to the typical flavour of traditional ewes' and 27 goats' cheeses was assessed. Fourteen yeast strains were grown in liquid 28 medium mimicking cheese composition and volatile compounds were identified 29 30 by GC-MS. Yeasts were able to produce key volatile compounds characteristic of the cheeses from which they were isolated. Inter-species and inter-strain 31 variations were observed. Under the conditions tested D. hansenii produced the 32 lowest levels of volatile compounds, with large intra-strain variations. 33 Kluyveromyces strains primarily produced esters and alcohols. K. marxianus 34 strains were associated with the production of acids, ethyl decanoate, 1-35 propanol and benzaldehyde, whereas K. lactis was correlated with the presence 36 of ketones, ethyl acetate and secondary alcohols. In conclusion, this study 37 38 shows the heterogeneous potential of dairy yeasts to contribute to final cheese flavour. 39

40 **1. INTRODUCTION**

Yeasts play an important role in proteolysis, lipolysis, fermentation of 41 residual lactose, and assimilation of lactic and citric acid during the ripening 42 of some cheeses, contributing to aroma development and to the rheological 43 properties of the final dairy product (McSweeney, 2004). Moreover yeasts 44 have been recovered from all stages of cheesemaking, as well as from milk, 45 brine and dairy process equipment among others (Corbo, Lanciotti, 46 Albenzio, & Sinigaglia, 2001; Delavenne et al., 2011; Gardini et al., 2006; 47 Seiler & Busse, 1990). 48

49 Debaryomyces hansenii is the dominant yeast species found in most cheese varieties (Fleet, 1990; Fox, Guinee, Cogan, & Mc.Sweeny, 2000). 50 D. hansenii possesses the ability to grow at high salt concentrations, low pH 51 52 and low water activity, as well as metabolising lactic and citric acids, which makes cheese a suitable environment for its proliferation (Breuer & Harms, 53 54 2006). Lactose-fermenting yeasts *Kluyveromyces lactis* and *Kluyveromyces* marxianus are also regularly found in dairy products and milk. Their lactose-55 fermenting ability promotes their growth in the cheese, where other yeasts 56 are scarce. Besides these species, cheeses may often contain other yeast 57 species. such as Yarrowia lipolytica, Geotrichum candidum 58 and Saccharomyces cerevisiae (Fleet, 1990). 59

60 Cheese flavour is one of the most relevant attributes influencing consumers' 61 acceptance and preference (Arora, Cormier, & Lee, 1995), and is the result 62 of a complex balance between various volatile and non volatile compounds, 63 which individually do not reflect the overall odour and taste (Fox & Wallace, 64 1997). Many volatile compounds have been implicated in cheese aroma,

such as acids, esters, ketones, aldehydes, alcohols or sulphur compounds,
and each dairy product has a characteristic and unique composition of
volatile components (Plutowska & Wardencki, 2007).

The contribution of yeasts to development of cheese aroma is considered 68 positive in some instances, creating commercial interest in using selected 69 strains as ripening cultures (Frohlich-Wyder, 2003; Romano, Capace, & 70 Jespersen, 2006). Several studies have shown that, in different cheeses, 71 72 relevant yeast species contribute differently to volatile production. G. candidum and Y. lipolytica are known to produce considerable amounts of 73 various volatile sulphur compounds; K. lactis, K. marxianus and S. 74 cerevisiae have been found to produce primarily esters; and D. hansenii 75 mainly produced branched-chain aldehydes and alcohols (Arfi, Spinnler, 76 77 Tache, & Bonnarme, 2002; Martin, Berger, Le Du, & Spinnler, 2001; Spinnler, Berger, Lapadatescu, & Bonnarme, 2001; Leclercq-Perlat, 78 79 Corrieu, & Spinnler, 2004; Sørensen, Gori, Petersen, Jespersen, & Arneborg, 2011). However, these studies emphasised inter-species aroma 80 production, with few surveys focussing on strain variation. Berger, Khan, 81 Molimard, Martin, and Spinnler (1999) reported the production of different 82 yields of sulphur compounds by G. candidum, depending on the strain 83 selected, and Gori, Sørensen, Petersen, Jespersen, and Arneborg (2012) 84 recently showed large strain variations in the production of flavour 85 compounds by D. hansenii. 86

Iberian traditional cheeses made from ewes' and goats' milk have high
intrinsic value, arising from their unique sensory characteristics, which
makes them highly appreciated by consumers (Freitas & Malcata, 2000). In

previous studies, yeasts present during the ripening process of ewes' and 90 goats' raw milk cheeses produced in a small traditional dairy in the 91 Mediterranean area of Spain were identified (Padilla, Manzanares, & 92 Belloch, 2014). D. hansenii and K. lactis were the yeast species most 93 frequently isolated from both kind of cheeses, and the former predominated 94 at the end of ripening period. K. marxianus, although less frequent, was 95 present during the first weeks of maturing. Moreover, results demonstrated 96 genetic heterogeneity present in the isolates (Padilla et al., 2014), and their 97 strain-dependent ability to generate bioactive compounds (García-Tejedor, 98 Padilla, Salom, Belloch, & Manzanares, 2013; Padilla et al., 2012). 99 However, there is little knowledge about the impact of the yeast isolates on 100 the final quality of the cheeses. 101

102 The objective of the present study was to further characterise both the aforementioned raw milk cheeses and their yeast microbiota, to gain a 103 104 better understanding of the relationship between yeast ripening strains and 105 cheese flavour. For this purpose, the volatile profile of the cheeses was characterised. Volatile compounds were extracted by Solid Phase Micro 106 (SPME) and analysed 107 Extraction by Gas Chromatography-Mass 108 Spectrometry (CG-MS). Moreover, the ability of 14 yeast strains belonging to D. hansenii, K. lactis and K. marxianus species to grow in a defined 109 medium and produce volatile compounds also present in ripened cheeses 110 111 was assessed.

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113 2. MATERIALS AND METHODS

114 2.1 Cheese samples

Commercial semi-hard ewes' and goats' cheeses produced in an artisanal 115 dairy farm sited in the rural Castello province (Spain) were analysed for 116 volatile compounds. The cheeses were made from raw milk coagulated with 117 the addition of mesophilic lactic acid bacteria starters and plant (Cynara 118 cardunculus) rennet (Abiasa Company, Pontevedra, Spain). After 119 precipitation of proteins, the curd was cut with vertical and horizontal knives 120 and crumbled manually. The remaining whey was removed first manually 121 and afterwards using a press. After salting, cheeses were air-dried until the 122 rind was formed and ripened in wooden shelves at 10-12°C and a relative 123 humidity of 85-90% for 60 days. 124

Three cheeses from the same batch and from ewes' milk and goats' milk were analysed at the end of ripening period (3 batches x 2 cheeses = 6 samples). After the rind was removed, cheeses were cut in pieces and ground with 0.75 mg butylated hydroxytoluene/20 g sample, wrapped in aluminium foil, vacuum-packed and stored at -20° C until GC analysis.

130 2.2 Yeast strains

Fourteen yeast strains belonging to the species *K. marxianus* (Km1-Km4), *K. lactis* (Kl1-Kl5) and *D. hansenii* (Dh1-Dh5) isolated during the ripening process from the artisanal cheeses described above and with different genetic characteristics were used in this study (Padilla et al., 2014). Yeast strains were maintained on GPYA medium (2 % glucose, 0.5 % peptone, 0.5 % yeast extract and 2 % agar, pH 5.5).

137 2.3 Culture conditions and media

138 Cheese-like medium (CLM; casamino acids 15 g L⁻¹, sodium lactate 19 ml 139 L⁻¹, yeast extract 1 g L⁻¹, CaCl₂ g L⁻¹, MgSO₄ 0.5 g L⁻¹, KH₂PO₄ 6.8 g L⁻¹,

NaCl 10 g L⁻¹ and lactose 28 g L⁻¹) was prepared according to Kagkli et al. 140 (2006) without addition of L-methionine. Flasks (100-mL) containing 50 mL 141 of CLM were inoculated with 10⁶ cells mL⁻¹ from overnight pre-cultures 142 grown in GPY medium (GPYA without agar) at 28°C and 150 rpm. CLM 143 cultures were incubated over 48h at 28°C and 150 rpm. At the end of the 144 incubation period, samples were taken for OD₆₀₀ measurement. Yeast cells 145 were removed by centrifugation (3220 x g, 10 min) and culture pH was 146 measured. Lactose and L-lactic acid were guantified in the supernatants 147 using Roche enzymatic kits (Darmstadt, Germany). For each strain, three 148 replicate cultures were analysed and a control without yeast inoculation was 149 also included. 150

151 2.4 Analysis of headspace volatile compounds by SPME GC–MS

152 An Agilent HP 7890 series II GC (Hewlett- Packard, Palo Alto, CA, USA) with an HP 5975C mass selective detector (Hewlett-Packard) equipped with 153 154 Gerstel MPS2 multipurpose sampler (Gerstel, Mülheim an der Ruhr, 155 Germany) was used in all experiments. The volatile components of the samples were extracted by SPME. All extractions were carried out using a 156 DVB/CAR/PDMS (divinylbenzene/carboxen/polydimethylsiloxane) fibre of 157 158 50/30 mm film thickness (Supelco, Bellefonte, PA, USA). The fiber was conditioned as indicated by the manufacturer prior to use in order to remove 159 any possible contaminants. For cheeses, 5 g of product was placed in a 20 160 mL headspace vial sealed with a PTFE-faced silicone septum. The vial was 161 maintained at 50°C for 15 min to equilibrate the headspace, and then the 162 fiber was exposed over 30 min at the same temperature. Before each 163 injection, the fiber was baked at 250°C for 10 min. Each sample was 164

analysed in triplicate. For CLM yeast cultures, 7 mL of supernatant plus 1.4 165 g of NaCl were added to a 20 mL headspace vial sealed with a PTFE-faced 166 silicone septum. The vial was kept at 50°C for 15 min to equilibrate the 167 headspace. The SPME fiber was then exposed to the headspace while 168 maintaining the sample at 30°C for 15 min. During extraction, the sample 169 was agitated continuously in pulses of 10 sec at 250 rpm. Before and after 170 each injection, the fiber was baked at 250°C for 10 and 5 min, respectively. 171 172 Each sample was analysed twice.

After the extraction step, the analytes were thermally desorbed for 5 min 173 from the fiber into the injector port of the GC-MS operating at 240°C in 174 splitless mode. The compounds were then separated using a DB-624 175 capillary column J & W Scientific (Agilent Technologies, Santa Clara, CA, 176 177 USA) (30m, 0.25mm i.d., film thickness 1.4 µm). For volatile analysis, the GC oven temperature program began at 40°C, where it was held for 5 min, 178 then ramped to 100°C at 3 °C min⁻¹ and maintained for 5 min, then to 150°C 179 180 at 3°C min⁻¹ and to 210°C at 4°C min⁻¹, and, finally, held at 210°C for 5 min. Mass spectra were obtained by electron impact at 70 eV, and data were 181 acquired across the range 29-400 amu (scan mode). 182

Compounds were identified by comparison with mass spectra from the library database (Nist'05), Kovats retention index (Kovats, 1965) and by comparison with authentic standards. The quantification of volatile compounds was done in SCAN mode using total ion chromatograms (TIC). The results were expressed as abundance units (AU x 10^{-6}). Volatile compounds quantitated from CLM control were subtracted from each yeastinoculated medium.

190 2.5 Statistical evaluation

The effect of yeasts on the generation of volatile compounds in CLM was tested by one-way analysis of variance (ANOVA). Differences between sample means were analysed according to Fisher's least significant difference (LSD) test. Principal component analysis (PCA) was used to test relationships among yeast species, pH, lactose and lactate consumption and main volatile compounds. Statistical analysis was performed using the statistic software XLSTAT, 2009.4.03 (Addinsoft, Barcelona, Spain).

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199 **3. RESULTS**

200 3.1 Volatile compounds in ewes and goats milk cheeses

Sixty-five volatile compounds were quantitied in the headspace of 201 202 Mediterranean ewes' and goats' cheeses (Table 1). They were classified into acids (14), esters (18), ketones (9), aldehydes (5), alcohols (17), 203 204 terpenes (1) and sulphur compounds (1). Four of the sixty-five compounds 205 were not present in the ewe's cheese, while eight of them were not present in the headspace of goat's cheese. As expected, most of the volatiles found 206 in these cheeses have been previously reported in other varieties of ewes' 207 208 and goats' raw milk cheeses (Table 1).

Esters and alcohols were the most abundant chemical families identified in the headspace of the Mediterranean cheeses studied, whereas, quantitatively, carboxylic acids were the most abundant volatiles. Among short and medium-chain carboxylic acids, the most abundant were acetic, butanoic, hexanoic, octanoic and decanoic acids, although branched-chain fatty acids such as 3-methylbutanoic and 2-methylbutanoic acids were also

found in both cheeses. Among esters, ethyl esters were the most abundant, although propyl- and branched-chain esters were also identified. Methyl ketones were the most abundant ketones detected in these products while aldehydes were not major components in these cheeses.

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3.2 Yeast growth in CLM and production of volatile compounds

Growth and aromatic profile from pure cultures of yeast strains belonging to 221 222 D. hansenii, K. lactis and K. marxianus were determined. All yeast strains were able to grow in a liquid medium mimicking cheese composition (CLM). 223 Lactose and lactate concentrations and pH values after 48 h of growing in 224 CLM were determined. Kluyveromyces strains depleted the available 225 lactose almost completely. K. lactis consumed around 5 % lactate, while 226 227 lactate consumption by K. marxianus strains was around 16 %. When grown in CLM, Kluyveromyces strains increased the pH from 5 to 5.2-5.8. 228 229 D. hansenii strains grew in CLM, consumed around 25 % of lactose and 5 230 % of lactate, and the pH value increased to 5.6-6.6.

Volatile compounds detected in the headspace of CLM are summarised in Tables 2 and 3. Only compounds which were also found in the Mediterranean cheeses (Table 1) are listed in Table 2, while Table 3 shows other volatile compounds detected in the headspace of CLM.

As observed in this study, yeasts were able to produce 27 compounds of those compounds found in the cheeses, including 6 acids, 7 esters, 3 ketones, 2 aldehydes and 9 alcohols (Table 2). Interestingly, the volatile composition of the headspace of CLM showed inter-species and inter-strain variations. General variations can be seen in Fig. 1, which shows volatile

compounds classified by chemical groups and yeast species. *K. marxianus*and *K. lactis* were the best producers of esters and alcohols, without
significant differences between the two species. Similar production of
aldehydes was found for *K. marxianus* and *D. hansenii*, while the former
was the best acid producer. In general, *D. hansenii* produced the lowest
levels of volatile compounds in the conditions tested. Moreover, standard
deviations indicated large strain variations for *D. hansenii*.

247 Among D. hansenii strains (Table 2), Dh1 produced the highest levels of total esters and alcohols. K. marxianus species was prominent due to the 248 production of acetic acid. Production of octanoic acid was restricted to K. 249 marxianus species, with Km3 being the best producer strain. K. marxianus 250 Km2 stood out as the leading producer of total esters, due to the high 251 252 production of 3-methyl-1-butanol acetate, the level of which was almost tenfold higher than that produced by K. lactis strains. Among esters, K. lactis 253 254 strains produced primarily ethyl acetate. Interestingly, ethyl octanoate 255 production was restricted to K. marxianus Km1 and Km2 under the conditions tested. With the exception of 2-pentanone production by Dh4, 256 ketone production was restricted to K. lactis strains. In contrast, none of the 257 258 K. lactis strains were able to produce aldehydes. Only two aldehydes were detected after yeast growth in CLM: benzaldehyde produced by K. 259 marxianus Km3 and Km4 and 3-methylbutanal produced by D. hansenii 260 Dh1 and Dh5. Regarding alcohol production, K. lactis produced 9 different 261 volatiles, although K. marxianus strains Km1, Km2 and Km3 stood out as 262 263 the best total alcohol producers, given the production of phenylethyl alcohol. Neither K. marxianus nor D. hansenii species were able to produce 2-264

heptanol and 2-nonanol. Moreover 2,3-butanediol was not detected after growth of *D. hansenii* in CLM, while *K. marxianus* strains were the best producers of such compounds (Table 2).

Apart from these compounds, 23 more volatiles compounds were identified 268 in the headspace of CLM after yeast growth (Table 3). Those compounds 269 mainly comprised esters (16), although 1 ketone, 3 aldehydes, 2 alcohols 270 and methionol were also detected; table 3 also shows the percentage of 271 272 yeast strains able to produce each compound. Some of these, 3 esters, 2 aldehydes and methyl isobutyl ketone, were only detected in CLM after 273 growth of D. hansenii. In contrast, none of the D. hansenii strains tested 274 was able to produce propyl and 2-phenylethyl propanoate. It is also 275 worthwhile to note that none of the K. marxianus strains were producers of 276 277 aldehydes or 3-methyl-pentanol. Although these compounds were not detected in the cheeses, most of them have been described as typical 278 279 cheese volatiles (Curioni & Bosset, 2002).

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- 281 3.3 Principal component analysis

Finally, a PCA model was developed using a dataset with 14 yeast strains 282 283 and 30 variables, comprising 27 volatiles (those present in cheeses and CLM, Table 2), lactose and lactate consumption and pH of the medium (Fig. 284 2). Two principal components were able to explain 70.5% of the total 285 variance observed. Principal component 1 (PC1) accounted for 39.6% of 286 the variance while PC2 accounted for 30.9%. PC1 differentiated the 287 288 incubations by the yeast genera inoculated. Kluyveromyces strains appeared in the positive part of PC1, while *Debaryomyces* was situated in 289

the negative side. Kluyveromyces strains were related to the maximum 290 production of volatile compounds and to the highest lactose consumption. 291 292 On the other hand, growth of *D. hansenii* was associated with the highest increase in pH value, and with the presence of 3-methylbutanal. PC2 293 differentiated the inoculations within Kluyveromyces strains. K. lactis was 294 related to the presence of volatiles compounds such as ketones, ethyl 295 acetate and secondary alcohols; on the other side, K. marxianus strains 296 297 were associated with the highest consumption of lactate and with the production of acids (acetic, propanoic and octanoic acids), ethyl decanoate, 298 299 1-propanol and benzaldehyde, among others.

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301 **4. DISCUSSION**

This study provides a characteristic fingerprint of volatiles present in Mediterranean cheeses and indicates the metabolic potential of ripening yeast strains to impact on cheese flavour. The proportion of volatile compounds depends on the extraction method used, and in this case a SPME technique with DVB/CAR/PDMS fibres was employed. The method used allowed comparisons among the different yeast strains, since the volatile compounds were obtained on a semi-quantitative basis.

The present research demonstrates the ability of *K. marxianus* and *K. lactis,* and to a lesser extent *D. hansenii* strains, to produce key volatile compounds characteristic of the cheeses from which they were isolated (Table 1 and 2). All the strains tested in this study were able to grow in a defined cheese-like medium (CLM) containing lactose, lactate and casamino acids and generate volatile compounds. Although these

conditions differ from real cheese, this medium has been successfully used
for screening purposes of yeast species and strains with potential use in
cheese ripening (Kagkli et al., 2006; Spinnler et al., 2001).

D. hansenii strains consumed 318 As expected. less lactose than 319 Kluyveromyces, and this might account for the lower production of aroma compounds in CLM. The prevalence of *D. hansenii* during ripening in 320 different kind of cheeses has been reported by several authors (Fleet, 1990; 321 322 Fox & Wallace, 1997; Fox et al., 2000) and it is considered as an obvious candidate for starter cultures (Bockelmann, 2002). Recently, Gori et al. 323 (2012) reported the potential of D. hansenii strains to increase the 324 nutty/malty flavour of cheese due to the production of aldehydes, although 325 large strain variations were found. In this study and under the conditions 326 327 tested, 3 branched-chain aldehydes (2-methylpropanal, 3-methylbutanal and 2-methylbutanal) were only produced by D. hansenii, with a large inter-328 329 strain variation. 2-Methylpropanal and 2-methylbutanal derived from the 330 catabolism of valine and isoleucine, respectively, were only detected in CLM, whereas 3-methylbutanal derived from leucine was also detected in 331 cheeses characterised here. Aldehydes are potent odorants in several 332 333 cheese varieties, although they are considered transitory compounds because they are quickly reduced to primary alcohols (Curioni & Bosset, 334 2002). In fact, the corresponding alcohols derived from the three branched-335 336 aldehydes (2-methyl-1-propanol, 3-methyl-1-butanol and 2-methyl-1butanol) were detected in the cheeses and in D. hansenii CLM, as also 337 reported by Sørensen et al. (2011). 338

Ester formation in cheese is mainly related to yeast metabolism (Molimard 339 & Spinnler, 1996) although some lactic acid bacteria and Micrococcaceae, 340 as well as chemical reactions, can be responsible (Gripon, Monnet, 341 Lamberet, & Desmazeaud, 1991). Esters come from a reaction between an 342 alcohol, derived from lactose fermentation or amino-acid catabolism, and a 343 fatty acid or amino acid catabolite intermediate. Most esters detected in 344 cheese are described as having sweet, fruity and floral notes. Although a 345 346 fruity flavour is traditionally regarded as a defect in cheese varieties such as Cheddar (Horwood, Stark, & Hull, 1987), it is a positive attribute of other 347 cheese varieties such as Parmigiano Reggiano (Meinhart & Schreier, 348 1986). Ester production by *Kluyveromyces* strains has been reported by 349 several authors (Arfi et al., 2002; Jiang, 1993; Leclercq-Perlat et al., 2004; 350 351 Martin et al., 2001). Ethyl acetate was the main ester produced, although ethyl propanoate, propyl acetate, butyl acetate, ethyl butanoate and ethyl 352 353 octanoate were also detected after growth of Kluyveromyces (Arfi et al., 354 2002; Leclercq-Perlat et al., 2004). Moreover, 2-phenylethyl acetate, 3methylbutyl ethanoate and 2-methylpropyl ethanoate are also produced by 355 Kluyveromyces strains (Jiang, 1993; Leclercq-Perlat et al., 2004), but strain-356 357 specific variations were not addressed. The present results confirm ethyl acetate as one of the primarily esters formed by Kluyveromyces strains, 358 together with the production of 3-methyl-1-butanol acetate by two strains of 359 K. marxianus. In total, K. marxianus and K. lactis strains respectively 360 produced 20 and 16 different kinds of esters, respectively, (Table 2 and 3) 361 362 highlighting the capability of the genus *Kluyveromyces* for ester production. Production of ethyl octanoate, restricted to K. marxianus strains under the 363

conditions tested, was also reported by Leclercq-Perlat et al. (2004). These
authors also observed that the ester production efficiency of *K. marxianus*was higher than that of *D. hansenii*, in agreement with the results obtained
here. With the exception of two ester compounds, the five *K. lactis* strains
tested produced the same ester profile, whereas *K. marxianus* strains
differed in seven esters. These results suggest a lower inter-strain variation
in *K. lactis* than in *K. marxianus*.

Several of the potential alcohols which may be precursors of the 371 aforementioned esters were also identified in yeast CLM. Those alcohols 372 were also detected after the growth of *D. hansenii* strains, where production 373 of esters was negligible. It has been suggested that a highly hydrolytic 374 activity towards esters in *D. hansenii* strains might be the reason for the 375 limited accumulation of ester compounds (Besancon, Ratomahenina, & 376 Galzy, 1995). The D. hansenii strains tested in this study have also been 377 378 characterised as having hydrolytic activity towards fatty acid esters (Padilla 379 et al., 2014).

Interestingly this study shows that only K. lactis strains were able to 380 2-pentanone, 2-heptanone and 2-nonanone, 381 produce which were 382 characteristic compounds of those cheeses from which they were isolated (Table 1 and 2). A previous study has shown the ability of K. lactis to 383 produce other kinds of ketones, such as 3-hydroxy-2-butanone and 1-384 385 hydroxy-2-propanone, from a medium containing glucose, yeast extract and vitamins (Jiang, 1993). However, to the best of our knowledge, ketones 386 generation by yeasts in a medium mimicking cheese composition has not 387 been reported. Methyl ketones are associated with fruity, floral and musty 388

notes, and their synthesis has been related to the enzymatic activity of
 moulds in surface-ripened cheeses (Curioni & Bosset, 2002).

Short-chain free fatty acids, predominant components of the flavour of many 391 cheeses such as those described here, were mainly characteristics of K. 392 393 marxianus CLM. As milk fat was not present in CLM, those acids may originate from the degradation of lactose and free amino acids or by 394 oxidation of ketones, esters and aldehydes (Molimard & Spinnler, 1996). 395 396 Branched-chain fatty acids, such as 2-methylpropanoic, 2- and 3methylbutanoic acids, are characteristic compounds of goat and ewe 397 398 cheeses, and they are probably derived from valine, isoleucine and leucine respectively (Kuzdzal-Savoie, 1980). The potential of selected yeast strains 399 to produce fatty acids when grown in a fat-containing medium deserves 400 401 further study.

Sulphur compounds were not abundant volatiles either in cheeses or CLM. 402 403 López del Castillo-Lozano, Delile, Spinnler, Bonnarme and Landaud (2007) 404 reported the necessity of methionine supplementation in culture media for the production of volatile sulphur compounds by yeasts. Since only one 405 sulphur compound was detected in cheeses under the conditions tested, we 406 407 did not consider supplementing casamino acids present in CLM with methionine. The only sulphur compound generated in CLM was methionol 408 (Table 3), a stable end product of methionine metabolism by yeasts (Liu 409 and Crow, 2010). In the conditions tested, methionol was produced by 410 Kluyveromyces strains, and only to a small extent by Debaryomyces 411 412 strains.

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414 **5. CONCLUSIONS**

This study has confirmed the potential of dairy yeasts to contribute to the 415 final cheese flavour. Moreover, species and strain variations were 416 significant, indicating a heterogeneous contribution to volatile compound 417 production and the feasibility of strain selection to modulate cheese flavour 418 and aroma. However, the development of suitable yeast starters requires 419 further studies, since complex interactions among cheese microbiota should 420 be taken into account. Characterization of enzyme activities involved in 421 flavour formation by dairy yeasts is in progress. 422

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577 Legends to Figures

Fig. 1. Total volatile compounds abundance by chemical group (expressed as AU x 10^6) in the headspace of CLM supernatants after yeast growth: Km: *Kluyveromyces marxianus*; KI: *Kluyveromyces lactis*; Dh: *Debaryomyces hansenii*. Data are mean ± SD of levels of volatile compounds produced by the different strains tested. Different letters in the same chemical group indicate significant differences (p < 0.05) among yeast species.

Fig. 2. Loadings of the first two principal components (PC1-PC2) of the
analysed parameters (pH, percentage of lactose and lactate consumption
and volatile compounds) of CLM after growth of different yeast strains:
Km1-Km4 (*Kluyveromyces marxianus*), Kl1-Kl5 (*Kluyveromyces lactis*),
Dh1-Dh5 (*Debaryomyces hansenii*).

Table 1. Abundance of volatile compounds (expressed as AU×10⁶ extracted by

591 SPME) in the headspace of the raw milk cheeses^a.

| Compound | LRI⁵ | RI⁰ | Goats' cheese | Ewes' cheese | Previously reported ^d |
|---|------|-----|---------------|------------------|-------------------------------------|
| Acids | | | | | |
| Acetic acid | 709 | А | 196.2 ± 13.6 | 224.6 ± 31.6 | 1-7 |
| Propanoic acid | 815 | А | 3.2 ± 0.1 | 9.5 ± 1.6 | 3-7 |
| 2-Methylpropanoic acid | 852 | А | 27.3 ± 5.3 | 12.5 ± 0.1 | 1,3-6 |
| Butanoic acid | 891 | А | 482.7 ± 15.0 | 216.3 ± 15.6 | 1-7 |
| 3-Methylbutanoic acid | 932 | А | 30.2 ± 7.1 | 20.8 ± 1.6 | 1,3,5-7 |
| 2-Methylbutanoic acid | 939 | А | 31.8 ± 7.8 | 15.1 ± 2.0 | 7 |
| Pentanoic acid | 971 | А | 2.7 ± 0.3 | 1.3 ± 0.1 | 3,4,7 |
| Hexanoic acid | 1080 | А | 676.6 ± 19.5 | 219.5 ± 9.6 | 1-4,6,7 |
| Heptanoic acid | 1165 | А | 4.7 ± 0.2 | 2.0 ± 0.1 | 3,4 |
| Octanoic acid | 1264 | А | 271.4 ± 7.9 | 67.1 ± 1.7 | 2-4 |
| Benzenecarboxylic acid | 1283 | А | 2.0 ± 0.3 | 1.4 ± 0.1 | 2 |
| Nonanoic acid | 1357 | А | 2.3 ± 0.1 | nd | 2,4 |
| Decanoic acid | 1453 | А | 128.2 ± 9.5 | 33.0 ± 1.8 | 2-4 |
| Dodecanoic acid | 1646 | В | 3.1 ± 0.4 | 0.9 ± 0.1 | 2,3 |
| Esters | | | | | |
| Ethyl acetate | 641 | А | 7.3 ± 2.1 | 25.4 ± 5.2 | 1-7 |
| Propyl acetate | 743 | А | 1.1 ± 0.1 | 4.1 ± 2.0 | 1,4,5,7 |
| 1-Methylpropyl acetate | 787 | В | nd | 10.8 ± 7.4 | 1,7 |
| Ethyl butanoate | 828 | А | 18.9 ± 6.3 | 11.2 ± 2.5 | 1-5,7 |
| Butyl acetate | 844 | А | nd | 0.6 ± 0.1 | 1,5,7 |
| 3-Methyl-1-butanol acetate | 907 | А | 5.6 ± 0.7 | 3.8 ± 0.3 | 1,7 |
| Propyl butanoate | 923 | А | 4.6 ± 0.4 | 2.8 ± 0.7 | 3-5,7 |
| 1-Methylpropyl butanoate 2-Methylpropyl 2-methyl | 960 | В | 4.9 ± 0.6 | 7.5 ± 0.3 | - |
| | 979 | в | 0.6 ± 0.1 | na 15 o x 1 o | - |
| Ethyl hexanoate | 1027 | A | 28.9 ± 13.5 | 15.6 ± 1.0 | 1-7 |
| 3-iviethylbutyl butanoate | 1084 | В | 6.2 ± 0.1 | nd | - |
| 2-Propenyl hexanoate | 1111 | ъ | 0.5 ± 0.0 | nd | - |
| Propyl hexanoate | 1123 | A | 6.3 ± 0.1 | 2.1 ± 0.4 | 5, 7 |
| 1-Methylbutyl butanoate | 1175 | В | 1.4 ± 0.0 | 0.9 ± 0.1 | - |
| Etnyl 2-methyl-propanoate | 1182 | ъ | 1.4 ± 0.1 | nd | 1 |
| Etnyl octanoate | 1225 | A | 16.8 ± 1.4 | 3.9 ± 0.4 | 1,3-7 |
| Propyl octanoate | 1322 | ъ | 1.0 0.0 | nd | - |
| Etnyl decanoate | 1425 | A | 7.5 ± 0.8 | 1.5 ± 0.1 | 1,2,4-6 |
| Ketones | F00 | | | | ~ |
| Acetone | 529 | A | 1.2 ± 0.1 | 1.1 ± 0.3 | 2 |
| 2-Butanone | 635 | A | 43.2 ± 10.7 | 529.3 ± 26.3 | 1,3-7 |
| 2-Pentanone | 729 | A | 41.7 ± 2.8 | 14.0 ± 6.2 | 1,4-7 |
| 3-Hydroxy-2-butanone | 779 | A | 3.1 ± 0.6 | 31.8 ± 7.9 | 5,6 |
| 2-Hexanone | 833 | Α | 1.6 ± 0.0 | nd | 5-7 |
| 2-Heptanone | 932 | Α | 42.1 ± 3.2 | 3.7 ± 1.6 | 1,2,4-7 |

| 8-Nonen-2-one | | 1135 | В | 3.4 ± 0.5 | nd | - |
|----------------------|-----|------|---------|---------------|---------------|------------------|
| 2-Nonanone | | 1139 | А | 147.4 ± 22.2 | 7.8 ± 2.9 | 1,2,4-7 |
| 2-Undecanone | | 1344 | А | 3.2 ± 0.6 | 0.6 ± 0.1 | - |
| Aldehydes | | | | | | |
| 2-Propenal | | 519 | А | 1.4 ± 0.1 | 0.5 ± 0.1 | 1, 5, 7 |
| 3-Methylbutanal | | 691 | А | 1.2 ± 0.2 | 1.0 ± 0.1 | 1-7 |
| Hexanal | | 838 | А | 0.8 ± 0.1 | nd | 1, 2, 5, 7 |
| Benzaldehyde | | 1017 | А | 1.1 ± 0.1 | 1.6 ± 0.2 | 2,3 |
| Benzeneacetaldehyde | | 1107 | А | 1.4 ± 0.2 | 0.8 ± 0.2 | 3 |
| Alcohols | | | | | | |
| Ethyl alcohol | | 511 | А | 62.4 ± 9.2 | 32.5 ± 6.5 | 1,5-7 |
| Isopropyl alcohol | | 538 | А | 3.0 ± 0.5 | 1.6 ± 0.3 | 5,7 |
| 2-Propen-1-ol | | 610 | В | 3.8 ± 0.4 | 2.7 ± 1.0 | 1,5-7 |
| 1-Propanol | | 615 | А | 16.7 ± 1.5 | 20.6 ± 6.8 | 1,4-6 |
| 2-Butanol | | 647 | А | 133.2 ± 13.4 | 461.4 ± 32.5 | 1,3-5 |
| 2-Methyl-1-propanol | | 682 | А | 1.0 ± 0.1 | 0.4 ± 0.1 | 1,5,6 |
| 1-Methoxy-2-propanol | | 718 | В | 1.3 ± 0.2 | 1.4 ± 0.3 | 3, 5, 6 |
| 1-Butanol | | 719 | А | 3.2 ± 0.5 | 3.8 ± 0.7 | 1,2,4-7 |
| 2-Pentanol | | 747 | А | 68.2 ± 5.3 | 10.7 ± 3.9 | 1,4-7 |
| 3-Methyl-1-butanol | | 793 | А | 18.3 ± 0.8 | 6.6 ± 0.9 | 1,3-7 |
| 2-Methyl-1-butanol | | 796 | А | 2.8 ± 0.2 | 1.2 ± 0.1 | 7 |
| 2,3-Butanediol | | 879 | А | 37.9 ± 3.6 | 59.4 ± 6.4 | - |
| 1-Hexanol | | 920 | А | 4.9 ± 0.3 | 2.8 ± 0.7 | 1,2,5,6 |
| 2-Heptanol | | 944 | А | 71.2 ± 6.9 | 14.2 ± 5.9 | 1,4-7 |
| 1-Heptanol | | 1022 | А | nd | 0.8 ± 0.1 | 1,7 |
| 2-Nonanol | | 1147 | А | 12.1 ± 1.1 | 2.7 ± 0.7 | 1, 4 |
| Phenylethyl alcohol | | 1193 | А | 5.8 ± 0.6 | 2.3 ± 0.3 | - |
| Terpenes | | | | | | |
| D-Limonene | | 1043 | А | nd | 39.2 ± 1.4 | 1, 2, 4, 5, 7 |
| Sulphur compounds | | | | | | |
| Dimethyl sulfone | - | 1057 | Α | 3.1 ± 0.5 | 1.9 ± 0.3 | - |
| ALL: Abundance units | the | | A CO | 3.1±0.3 | 1.9 ± 0.3 | - (TIC) for e |

592 AU: Abundance units, the result of counting the total ion chromatogram (TIC) for each 593 compound.

594 ^aValues are mean \pm SD (n=3).

^bLinear retention indices (LRI) of the compounds eluted from the GC-MS using a DB-624 capillary column (J&W Scientific 30 m×0.25 mm i.d.×1.4 μm film thickness).

^cReliability of identification (RI): A, mass spectrum and retention time identical with an authentic
 standard; B, tentative identification by mass spectrum.

^dCompounds previously reported in ewes' and goats' raw milk cheeses. Reference numbers are
as follows: (1) Carbonell, Núñez, & Fernández-García, 2002; (2) Condurso, Verzera, Romeo,
Ziino, & Conte, 2008; (3) Delgado, González-Crespo, Cava, García-Parra, & Ramírez, 2010; (4)
Delgado, González-Crespo, Cava, & Ramírez, 2011; (5) Fernández-García, Carbonell, Gaya, &
Nuñez, 2004; (6) Izco & Torre, 2000 and (7) Larráyoz, Addis, Gauch, & Bosset, 2001.
nd: Not detected.

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| | | Kluyveromyd | es marxianus | | | Kluyveromyces lactis | | | | | Debaryomyces hansenii | | | | |
|----------------------------|------------------------|-----------------------|----------------------|------------------------|-----------------------|------------------------|-----------------------|-----------------------|------------------------|---------------------|-----------------------|----------------------|--------------------|--------------------|--|
| Compound | Km1 | Km2 | Km3 | Km4 | KI1 | KI2 | KI3 | KI4 | KI5 | Dh1 | Dh2 | Dh3 | Dh4 | Dh5 | |
| Acids | | | | | | | | | | | | | | | |
| Acetic acid | 214.81ª | 206.69ª | 247.29ª | 97.49 ^b | 5.57° | nd | 4.19° | 49.31 ^{bc} | 17.97° | nd | nd | 22.70° | nd | nd | |
| Propanoic acid | nd | nd | nd | 6.24ª | nd | nd | nd | nd | nd | 2.71ª | nd | nd | nd | nd | |
| 2-Methylpropanoic acid | 21.33 ^f | 19.76 ^f | 39.16 ^{de} | 25.23 ^{ef} | 68.89 ^b | 59.98 ^{bc} | 55.01 ^{bcd} | 91.20ª | 104.68ª | 23.03 ^{ef} | 0.11 ^h | 45.83 ^{cd} | 0.56 ^h | 0.93 ^{gh} | |
| 3-Methylbutanoic acid | 11.49ª | 12.52ª | 1.41 ^{bc} | 2.53 ^b | nd | nd | nd | nd | nd | 1.51 ^{bc} | nd | 2.46 ^b | 0.60 ^c | 0.50 ^c | |
| 2-Methylbutanoic acid | 52.52ª | 54.62ª | 35.17° | 29.28 ^{cd} | 38.15 ^{bc} | 34.90° | 34.82° | 51.31ª | 50.65ª | 20.70 ^d | 2.93° | 46.13 ^{ab} | 0.88 ^e | 2.28 ^e | |
| Octanoic acid | 1.32 ^b | nd | 3.18ª | 0.50 ^b | nd | nd | nd | nd | nd | nd | nd | nd | nd | nd | |
| Total acids | 301.47ª | 293.59ª | 326.21ª | 161.27 ^b | 112.61 ^{cd} | 94.88 ^{de} | 94.02 ^{de} | 191.82 ^b | 173.30 ^b | 47.95 ^{ef} | 3.04 ^f | 117.12 ^{cd} | 2.04 ^f | 3.71 ^f | |
| Esters | | | | | | | | | | | | | | | |
| Ethyl acetate | 770.82 ^d | 914.69 ^d | 909.30 ^d | 904.95 ^d | 1418.81° | 1612.86 ^{abc} | 1519.04 ^{bc} | 1702.68 ^{ab} | 1795.43ª | 185.98 ^e | 1.86 ^e | 58.67° | 10.29 ^e | 0.80 ^e | |
| Propyl acetate | 15.06cd ^e | 22.94ª | 19.76 ^{ab} | 12.86 ^{de} | 10.35 ^e | 16.25 ^{bcd} | 11.84 ^{de} | 18.41 ^{abc} | 16.00 ^{bcd} | 0.68 ^f | nd | nd | nd | nd | |
| Butyl acetate | nd | nd | 11.88ª | 11.33ª | 0.53 ^b | 0.84 ^b | 0.34 ^b | 0.55 ^b | 0.49 ^b | nd | nd | nd | nd | nd | |
| 3-Methyl-1-butanol acetate | 1290.58 ^b | 1561.42ª | 545.40° | 550.74° | 120.26 ^{de} | 148.23 ^{de} | 144.69 ^{de} | 188.40 ^d | 163.81 ^{de} | 4.96 ^e | 0.30 ^e | 0.91 ^e | nd | nd | |
| 3-Methylbutyl butanoate | nd | 0.79ª | nd | 0.74ª | nd | 0.74ª | 0.68ª | 0.65ª | 0.79 ^a | nd | nd | nd | nd | nd | |
| Ethyl octanoate | 2.23ª | 3.19ª | nd | nd | nd | nd | nd | nd | nd | nd | nd | nd | nd | nd | |
| Ethyl decanoate | 2.00 ^{bc} | 1.38 ^{cd} | 2.85 ^b | 5.78ª | nd | nd | nd | nd | nd | 2.13 ^{bc} | nd | 0.91 ^d | nd | nd | |
| Total esters | 2080.69 ^b | 2504.41ª | 1489.20 ^d | 1486.40 ^d | 1549.95 ^d | 1778.92 ^{bcd} | 1676.59 ^{cd} | 1910.69 ^{bc} | 1976.52 ^{bc} | 193.75 ^e | 2.16 ^e | 60.49 ^e | 10.29 ^e | 0.80 ^e | |
| Ketones | | | | | | | | | | | | | | | |
| 2-Pentanone | nd | nd | nd | nd | 1.24 ^b | 0.95b ^c | 0.52 ^{de} | 0.76 ^{cd} | 1.95ª | nd | nd | nd | 0.24 ^e | nd | |
| 2-Heptanone | nd | nd | nd | nd | 8.22 ^d | 11.28° | 10.97° | 15.90ª | 13.51 ^b | nd | nd | nd | nd | nd | |
| 2-Nonanone | nd | nd | nd | nd | 4.54° | 6.36° | 5.26° | 11.02ª | 8.80 ^b | nd | nd | nd | nd | nd | |
| Total ketones | nd | nd | nd | nd | 14.00 ^d | 18.59° | 16.75° | 27.68ª | 24.26 ^b | nd | nd | nd | 0.24 ^e | nd | |
| Aldehydes | | | | | | | | | | | | | | | |
| 3-Methylbutanal | nd | nd | nd | nd | nd | nd | nd | nd | nd | 1.62ª | nd | nd | nd | 2.13ª | |
| Benzaldehyde | nd | nd | 21.59 ^b | 38.96ª | nd | nd | nd | nd | nd | nd | nd | nd | nd | nd | |
| Total aldehydes | nd | nd | 21.59 ^b | 38.96ª | nd | nd | nd | nd | nd | 1.62° | nd | nd | nd | 2.13° | |
| Alcohols | | | | | | | | | | | | | | | |
| Ethyl alcohol | 1075.15 ^{cde} | 1054.34 ^{de} | 1273.58ª | 1134.08 ^{bcd} | 1215.53 ^{ab} | 1011.79 ^e | 994.41 ^e | 1157.85 ^{bc} | 1140.75 ^{bcd} | 543.92 ^f | nd | 252.55 ⁹ | 10.85 ^h | 12.37 ^h | |

Table 2. Volatile compounds (expressed as AU×10⁶ extracted by HS-SPME) identified in the headspace of CLM after yeast growth^a.

| 1-Propanol | 29.90° | 26.35° | 55.70 ^b | 66.05ª | 16.34 ^d | 14.92 ^{de} | 16.13 ^d | 12.09 ^{de} | 10.85 ^{ef} | 7.15f ^g | 0.29 ^h | 2.95g ^h | 0.52 ^h | 0.33 ^h |
|---------------------|-----------------------|-----------------------|-----------------------|------------------------|------------------------|------------------------|------------------------|-----------------------|-----------------------|-----------------------|----------------------|----------------------|---------------------|--------------------|
| 2-Methyl-1-propanol | 193.25 ^d | 193.23 ^d | 187.82 ^d | 156.24° | 283.99 ^b | 277.63 ^{bc} | 256.03° | 285.64 ^b | 337.74ª | 193.06 ^d | 10.53 ^g | 66.77 ^f | 9.97 ^g | 3.02 ^g |
| 3-Methyl-1-butanol | 1570.08 ^{cd} | 1517.17 ^{cd} | 1619.72 ^{bc} | 1440.63 ^d | 1748.11 ^{ab} | 1658.89 ^{abc} | 1633.48 ^{abc} | 1578.36 ^{cd} | 1789.07ª | 591.30 ^{ef} | 702.24 ^e | 458.81 ^f | 87.48 ^g | 31.85 ^g |
| 2-Methyl-1-butanol | 942.22 ^e | 932.17° | 1228.13ª | 1181.01 ^{abc} | 1144.75 ^{bcd} | 1130.80 ^{cd} | 1110.76 ^d | 1111.02 ^d | 1110.01 ^d | 1213.12 ^{ab} | 191.44 ⁹ | 825.22 ^f | 28.01 ^h | 22.46 ^h |
| 2,3-Butanediol | 13.98 ^b | 11.58 ^{bc} | 31.72ª | 14.40 ^b | 8.91 ^{cd} | 2.03 ^e | 4.06 ^e | 7.66 ^d | 7.88 ^d | nd | nd | nd | nd | nd |
| 2-Heptanol | nd | nd | nd | nd | 6.11 ^b | 6.61 ^b | 5.78 ^b | 7.64ª | 8.44ª | nd | nd | nd | nd | nd |
| 2-Nonanol | nd | nd | nd | nd | 8.37 ^b | 8.34 ^b | 7.65 ^b | 11.72ª | 13.13ª | nd | nd | nd | nd | nd |
| Phenylethyl alcohol | 1301.50ª | 1291.65ª | 745.37 ^b | 578.62° | 282.76 ^{de} | 251.70 ^{de} | 286.85 ^d | 230.59 ^{de} | 218.21° | 47.08 ^f | 230.40 ^{de} | 49.93 ^f | 15.12 ^f | 28.15 ^f |
| Total alcohols | 5126.08ª | 5026.49 ^{ab} | 5142.04ª | 4571.03 ^{cd} | 4714.87 ^{bc} | 4362.71 ^d | 4315.15 ^d | 4402.59 ^{cd} | 4636.08 ^{cd} | 2595.63° | 1134.90 ^g | 1656.23 ^f | 151.95 ^h | 98.18 ^h |

AU: Abundance units, the result of counting the total ion chromatogram (TIC) for each compound. ^aValues are mean from n=3. Volatile compounds from CLM control were subtracted to each yeast CLM. Means followed by different letters in the same row indicate significant differences among yeast strains (p<0.05; one-way ANOVA with Fisher's LSD test)

nd: Not detected.

| | | | | Yeast ^a | |
|-----------------------------------|------|-----|------------------|--------------------|-------------|
| Compound | LRI♭ | RI⁰ | K. marxianus | K. lactis | D. hansenii |
| Esters | | | | | |
| Ethyl propanoate | 738 | А | 100 ^c | 100 | 80 |
| Ethyl 2-methyl-propanoate | 785 | А | 100 | 100 | 40 |
| 2-Methylpropyl acetate | 804 | А | 100 | 100 | 20 |
| Propyl propanoate | 837 | А | 50 | 100 | 0 |
| Ethyl 2-methyl-butanoate | 876 | А | 100 | 100 | 40 |
| Ethyl 3-methyl-butanoate | 879 | А | 0 | 0 | 40 |
| 2-Methylpropyl propanoate | 895 | А | 75 | 100 | 20 |
| 2-Methyl-1-butanol acetate | 909 | А | 100 | 100 | 40 |
| 3-Methyl-1-butanol propanoate | 995 | А | 100 | 100 | 20 |
| 2-Methyl-1-butanol propanoate | 999 | А | 100 | 80 | 20 |
| 2-Methylbutyl 2-methyl-propanoate | 1044 | А | 50 | 100 | 40 |
| 3-Methylbutyl 2-methyl-butanoate | 1128 | А | 0 | 0 | 40 |
| 2-Methylbutyl 2-methyl-butanoate | 1133 | А | 0 | 0 | 40 |
| 2-Phenylethyl acetate | 1317 | А | 25 | 100 | 20 |
| 2-Phenylethyl propanoate | 1407 | А | 100 | 100 | 0 |
| Phenylethyl butyrate | 1451 | А | 100 | 100 | 20 |
| Ketones | | | | | |
| Methyl isobutyl ketone | 781 | В | 0 | 0 | 80 |
| Aldehydes | | | | | |
| Acetaldehyde | 469 | А | 0 | 100 | 60 |
| 2-Methylpropanal | 595 | А | 0 | 0 | 60 |
| 2-Methylbutanal | 700 | А | 0 | 0 | 60 |
| Alcohols | | | | | |
| 3-Methyl-pentanol | 899 | А | 0 | 100 | 80 |
| 3,7-Dimethyl-6-octen-1-ol | 1285 | В | 100 | 100 | 20 |
| Sulphur compounds | | | | | |
| Methionol | 1060 | А | 100 | 100 | 20 |

Table 3. Generation of volatile compounds (not found in the Mediterranean cheeses) in the headspace of CLM inoculated with yeasts.

^a Percentage of strains producing volatile compound in CLM media. ^b Refer to foodnote ^a in Table 1. ^c Refer to foodnote ^b in Table 1.

Figure 1 Padilla et al., 2013





PC1 (39.60 %)