

Antique stone quarries in Turkey: a case study on tuffs in the Temple of Apollon Smintheus

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Abstract: All types of stones have been used as building stones, depending on their durability, visual harmony with the intended construction and availability. In the Hellenistic period, as in other periods, tuff was preferred as a building stone due to its convenience – it was easy to extract, transport and use for building. In the present study, three ancient quarries that were the possible tuff sources for the Apollon Smintheus Temple in Çanakkale are investigated by comparing the durability properties of stones in the temple and in the quarries. These properties are determined using physical and physico-mechanical tests, comparing fresh and artificially weathered samples. Microstructural and elemental correlations were found using optical microscopy, stereomicroscopy, X-ray diffraction, scanning electron microscopy, methylene blue adsorption and X-ray fluorescence analyses. The results indicate that temple tuffs and two of three quarries have similar geological engineering and microstructural properties with the strong claim that those two quarries could have been the source of building stone for the temple.



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Historically, only high-status stone blocks were transported for longer distances, whereas local quarries were used for larger constructions (Gómez-Heras & Fort González 2004; Török & Příkryl 2010). Since local stones do not entail high transportation costs they can be extracted whenever is suitable (Bell 1990). What matters most is that the rock must not include closely spaced joints, cracks or other planes of weakness in large blocks extracted by stone quarrying (Rapp 2002). In Hellenistic architecture all types of stones – marble, gneiss, limestone, sandstone, granite, andesite, tuff, breccias, serpentinite, porphyry and trachyte – were used. Among all other stones tuffs were easy to get, carry and process, and were therefore relatively cheap and could be used in any part of a building on condition that it was sheltered from the weather (Tucci 2015). In Turkey, tuffs that have low unit weight, high to very high porosity and poor to good durability properties (Topal & Doyuran 1997, 1998; Topal 2002; Topal & Sözmen 2003; Yavuz 2012; Çelik *et al.* 2014) are seen in antique structures as well as in historic monuments of the Seljuk–Ottoman

period that have survived to the present day. Unremarkably, antique quarries can be found with the masons' marks and shear ledges of dimension stones; however, it is not possible today to observe the tuff quarries as they appeared in ancient times, because tuffs deteriorate over time when exposed to weathering and/or because of the urban construction mass (Příkryl 2006; Příkryl & Török 2010). In the case of lack of knowledge of the original quarry, the most contiguous geological resources are examined through mineralogical, chemical, and physical analyses and *in situ* observation and testing to find replacement stones that are compatible in terms of characteristics and appearance. This approach may yield misapplication of stones.

The purpose of this study is to determine the provenance of the tuffs used in the Temple of Apollon Smintheus, architecture dating from the Hellenistic period, by comparison of physical, mechanical, chemical and mineralogical features. To that end, samples taken from the temple and samples taken from three potential quarry regions were compared using several analytical techniques to

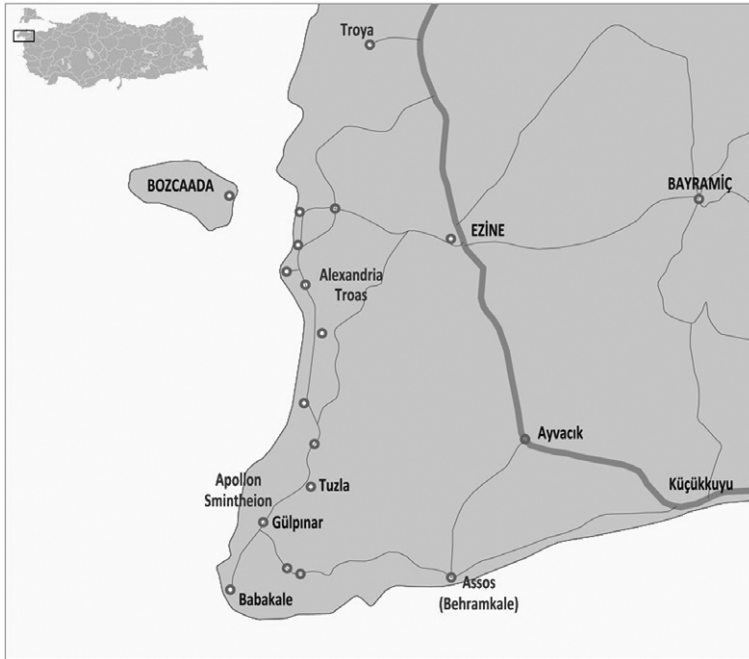


Fig. 1. Location map of the study area.

determine the source quarry for the temple tuffs. The results of the analyses may serve for surface deterioration studies and durability assessment as well.

Site description

The Apollon Smintheus Temple, is situated on the southwest corner of the Biga Peninsula (Troas)



Fig. 2. Temple of Apollon Smintheus.

and was constructed during the second century BC. It is located within the boundaries of the city of Çanakkale and in the municipality of Gülpınar, which was called Külahlı until the 1920s (Fig. 1). With its reliefs illustrating stories from *The Iliad* of Homer, the temple of Apollon Smintheus is a unique example of the Ionic style of the Troas region of Turkey (Fig. 2) (Özgünel 2001). The tuff was used as the foundation stone as well as for filling. Basaltic andesite was placed over the tuff, and the upper parts and columns were made of marble (Gökçe 2000).

Geological setting around the temple

The study area is surrounded by geological formations that include sedimentary and volcanic rocks belonging to the Tertiary period (Öngür 1973; Kayan 1994; Ilgar *et al.* 2008). The temple is located at the borders between sedimentary and volcanic units. While the western part of the temple rests on a sedimentary–alluvial unit, the east and south sides rest on Miocene volcanic units. In the area where Gülpınar town is located, pumice and ignimbritic tuffs, basaltic andesites, and andesite and other pyroclastic rocks are observed. The unit where the three selected quarries are located is represented by tile-red welded tuffs with big white and grey pumice fragments (Fig. 3).

Materials and methods

Sampling

The first quarry was selected on the road towards Babakale, southwest of the temple (Fig. 3). Archaeologists consider that the formations near the old Babakale Road were used in ancient times because of their likeness to the parapet blocks from the Roman bridge (Kaplan 2012). The second quarry was selected southeast of the temple, in Fatma Gerdan, where vitric tuff similar to the ones in Cappadocia are present. The third one, Kızılkeçili, was located in the east of the hieron in a steep valley. The same block dimensions and similar colour/texture as the temple tuff blocks in three regions supported the argument that one of these served as the antique quarry. Finally, grey vitric tuff samples were taken from the stairs and the northeast façade of the temple.

Physical properties

Durability is the resistance of a stone to weathering. It is one of the most significant aspects of natural stones. Various physico-mechanical properties of stones are used for durability assessment. Durability may be assessed by accelerated laboratory tests, complex environmental testing and exposure site

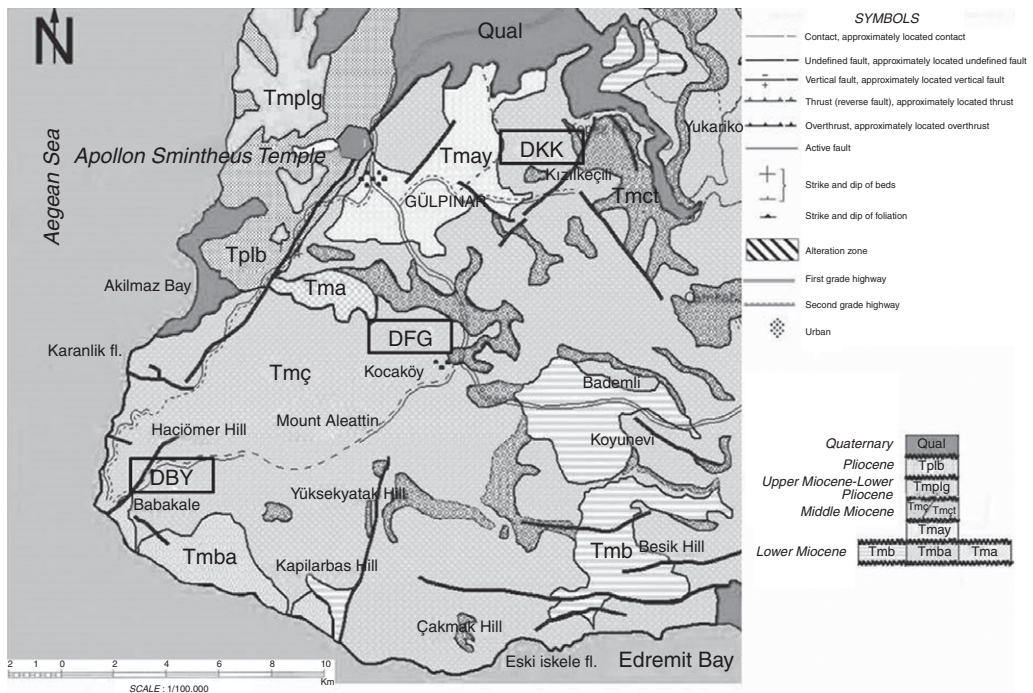


Fig. 3. Geological map of the study area.

Table 1. Basic physical and physico-mechanical test values of temple and fresh quarry samples*

Tuff samples	P (%)	D (g/cm ³)	WAC (%)	Real D	Sat. coeff.	Dry unit weight	Sat. unit weight
DTS	54 ± 7.87	1.30 ± 0.12	41.15 ± 2.20	2.87 ± 0.75	0.77 ± 0.07	12.76 ± 1.19	18.02 ± 1.96
DBY	16 ± 2.15	2.47 ± 0.17	6.48 ± 0.85	2.95 ± 0.23	0.4 ± 0.03	24.24 ± 1.62	25.69 ± 1.70
DFG	48 ± 4.12	1.24 ± 0.08	38.39 ± 2.85	2.40 ± 0.35	0.8 ± 0.05	12.18 ± 0.80	16.72 ± 1.09
DKK	45 ± 5.31	1.40 ± 0.14	32.39 ± 5.36	2.58 ± 0.41	0.7 ± 0.07	13.78 ± 1.36	17.91 ± 1.42

*P, porosity; D, density; WAC, water absorption capacity; Sat. coeff., saturation coefficient; Sat. unit weight, saturated unit weight.

testing (Přikryl 2013). In this study, physico-mechanical properties of the tuffs were used for durability assessment. Physico-mechanical properties together with durability assessment were studied for the test specimens before and after artificial

weathering tests that encompassed salt-crystallization and wetting/drying cycles. Effective porosity, dry and saturated unit weights, bulk density, water absorption, ultrasonic pulse velocity (UPV), capillary and moisture absorption, uniaxial compressive

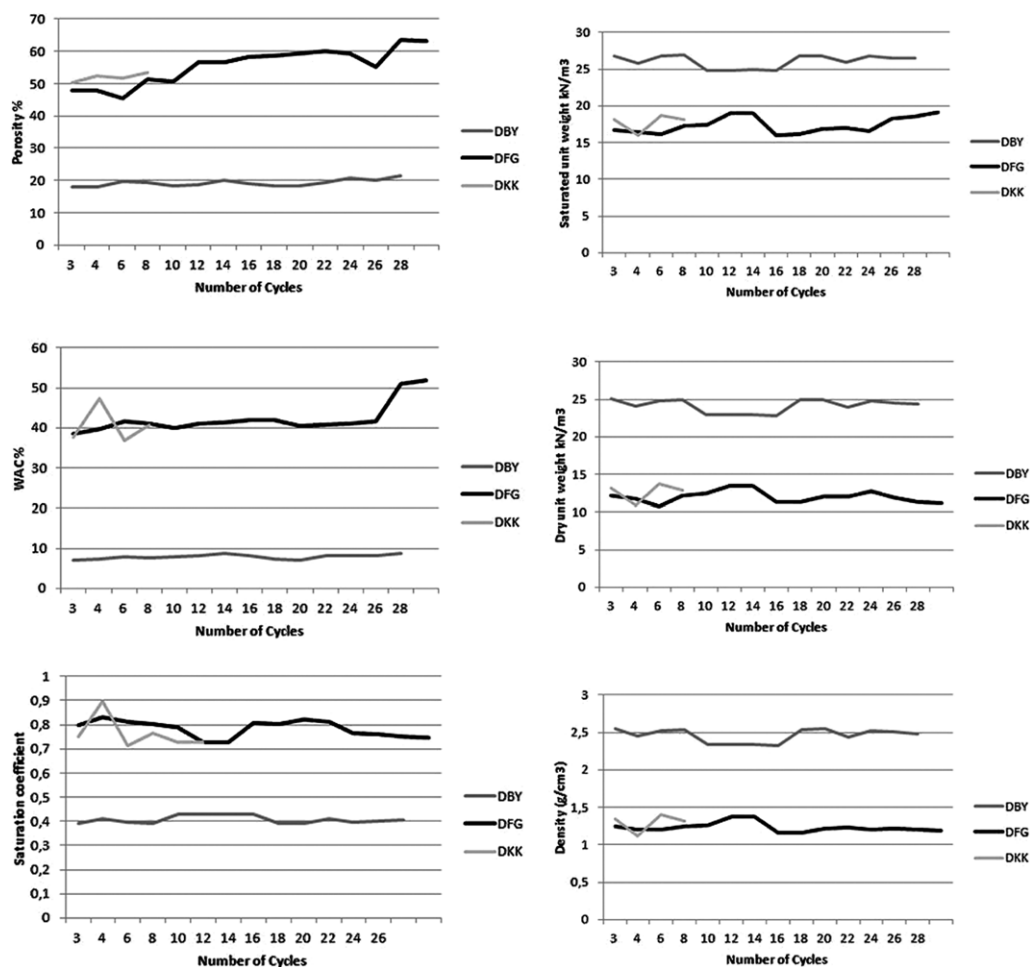


Fig. 4. Physical properties of the samples after the wetting/drying cycles.

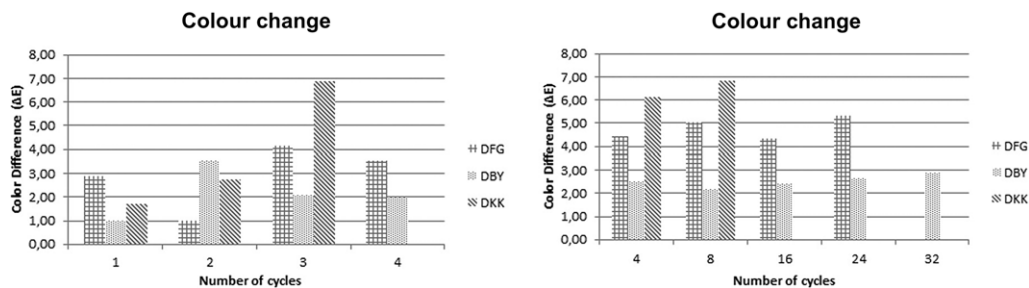


Fig. 5. Colour change of quarry samples after the salt-crystallization cycles (on the left) wetting/drying cycles (on the right).

strength (UCS) and pore size distribution were measured first on two cubic tuff samples of temple stone (DTS1, DTS2), 155 cubic tuff samples from Babakale Yolu (DBY), 52 cubic tuff samples from Fatma Gerdan (DFG) and 26 cubic tuff samples from Kızılkeçili (DKK).

The salt-crystallization test was performed following RILEM (1980) on 26 cubic tuff samples from DBY, 26 cubic tuff samples from DFG and 13 cubic tuff samples from DKK. Samples were immersed in $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ solution for 1 h. They were then dried in a 60°C oven for 24 h. After that, they were cooled to room temperature. The test was repeated six times. Later, the samples were washed every day for 1.5 months to clean up

the salt. Analyses carried out on the fresh samples were repeated on the weathered samples.

The wetting/drying test was performed following ASTM (2012) on 26 cubic tuff samples from DBY, 26 cubic tuff samples from DFG and 13 cubic tuff samples from DKK. For the wetting/drying tests, the samples were immersed for 24 h in distilled water at $15\text{--}20^\circ\text{C}$ and then dried in an oven at 60°C , after which they were cooled to room temperature. This procedure was repeated 32 times. As in the salt-crystallization test, analyses were repeated on the weathered samples.

For the colour measurements a Spectrophotometer CM-2600d/2500d Konica Minolta was used. The modulus of elasticity values of samples were

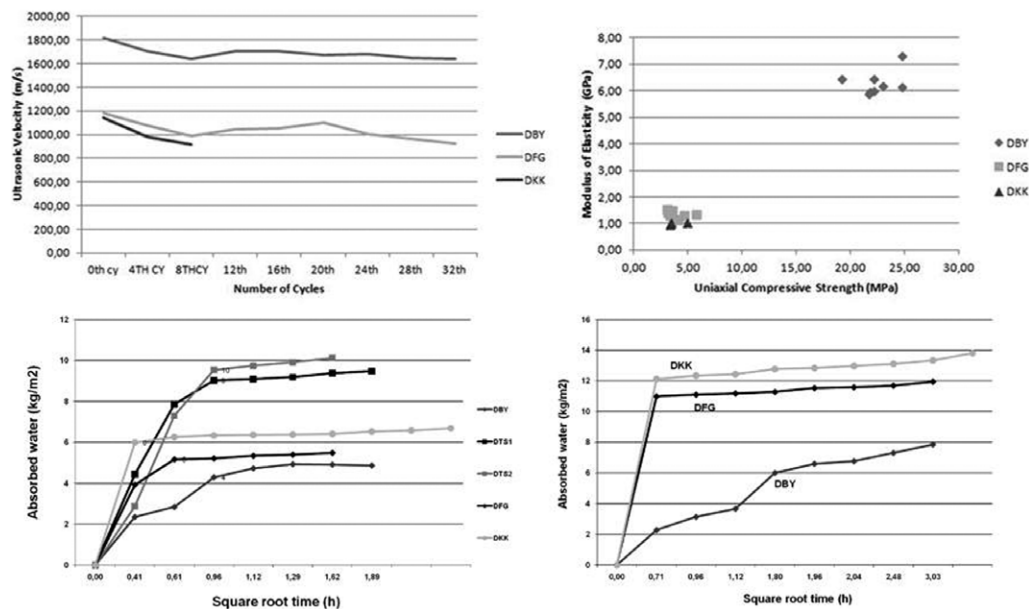


Fig. 6. UPV (on the top, left) and modulus of elasticity (on the top, right) values after the wetting/drying cycles and capillary absorption values of the fresh samples (on the bottom, left) and after the salt-crystallization cycles (on the bottom, right).

determined directly by means of equations described in ASTM D 2845-90 (1990) and RILEM (1980), using their UPV and density values. The UPV values were measured on both saturated and dry series of the samples in the direct transmission mode (cross-direction) by using a portable Pundit Plus CNS Farnell Instrument with 220 kHz transducers.

The capillary absorption test was performed following RILEM (1980) by measuring the weight difference over time. Although this long procedure is time-consuming, it indicates pore size distribution (Caner-Saltık 1998). The samples were suspended in the distilled water and measurements were taken every 5 min for the first 30 min and every half hour thereafter. The difference between

the initial weights gave the capillary absorption coefficient. Moisture absorption implied the finest pore ($<0.5 \mu$). In order to achieve an 80% relative humidity environment, 10% CaCl_2 solution was put into desiccators. A couple of days later, when the environment had achieved 80% relative humidity and was at 20°C , the samples were placed in it, and weight difference was measured periodically until constant weight was achieved.

An ELE point-load instrument was also used on the samples. The point-load strength index was used to calculate UCS (Broch & Franklin 1972; Bieniawski 1975; ISRM 1985; Topal 2000). The tests were performed on the samples from the temple and the quarries under both dry and saturated conditions.

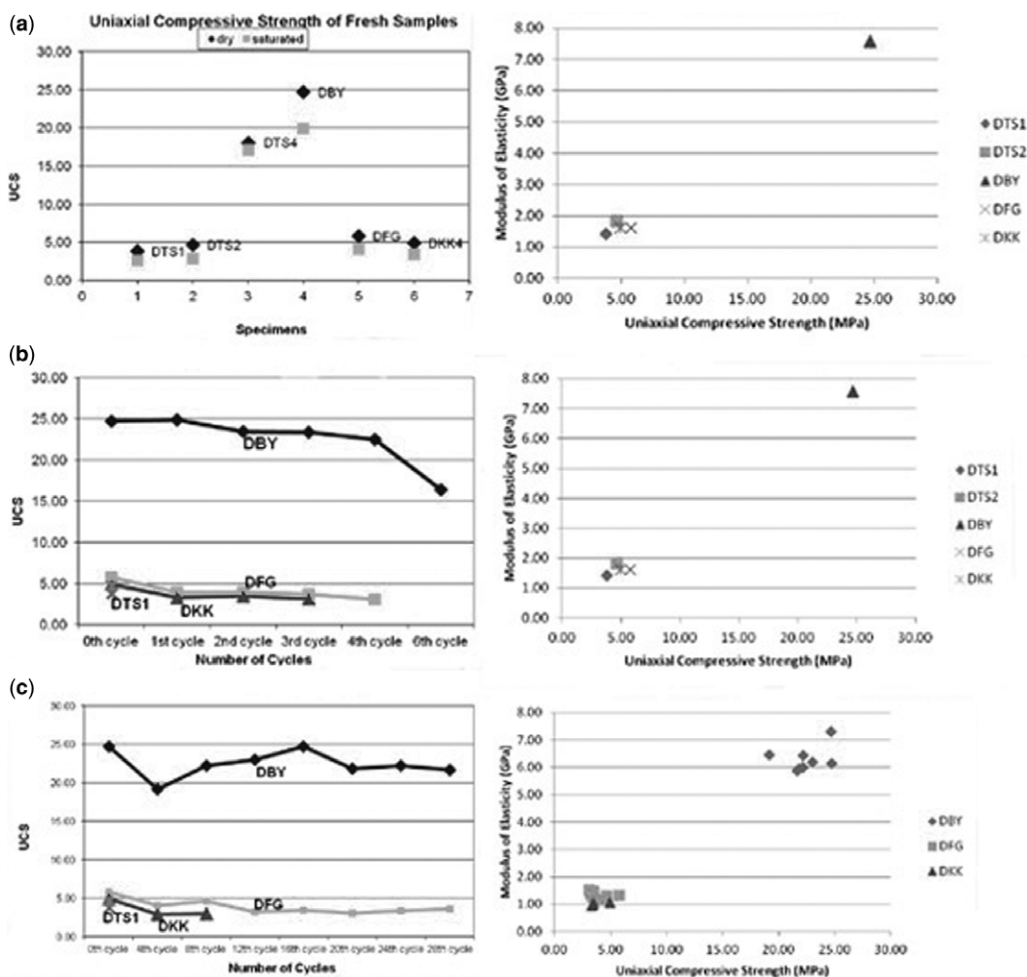


Fig. 7. (a) UCS of the fresh samples (on the left), modulus of elasticity v. UCS trend (on the right), (b) after the salt-crystallization cycles, and (c) after the wetting/drying cycles.

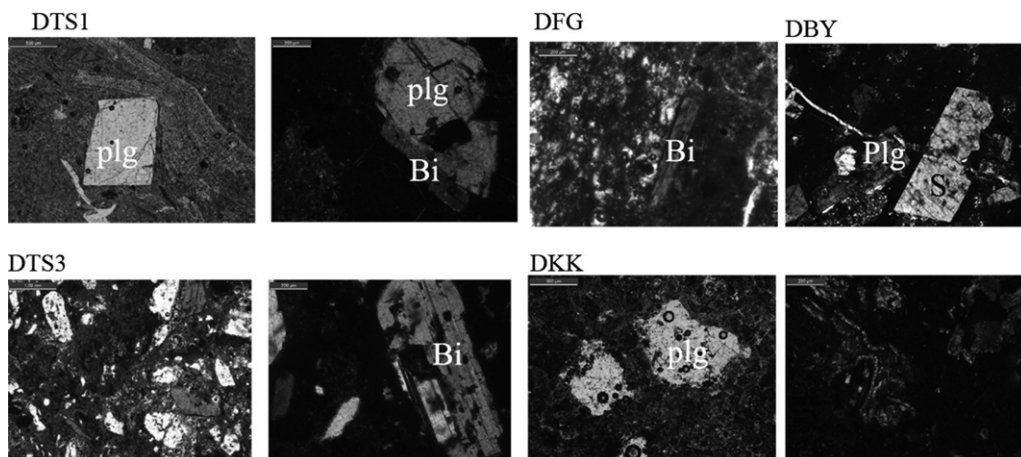


Fig. 8. Photomicrographs of the samples (DTS1: single Nicol 2.5×; Plg, plagioclase; VG: volcanic glass cross Nicol 10×; Plg, plagioclase; Bi, Biotite; DFG: single Nicol 10×, biotite; DBY: cross Nicol 5×; S, sanidine; Plg, plagioclase; DTS3: single Nicol 5×, hematite coatings are seen; DTS2: cross Nicol, 10×, biotite; DKK: single Nicol, cross Nicol 10×.

Mineralogical, petrographical and chemical properties

Thin and cross-sections of the samples were analysed under an optical microscope and polarizing microscopes from Nikon AFX-512A and Carl Zeiss. X-ray diffraction (XRD) analysis was performed using a Bruker D8 Advance Diffractometer, Sol-X detector, with CuK α radiation, adjusted to 40 kV and 40 mA, and EVA software. X-ray fluorescence (XRF) analysis was carried out with the Rigaku ZSX Primus II device. For scanning electron microscopy (SEM-EDX) a Tescan Vega II XMU was used. Because all of the samples were non-conductive, a sputter coater was

used to coat surfaces with a conducting gold layer. Methylene blue adsorption (MBA) tests were performed using a spectrophotometer (SP3000 Plus OPTIMA and 1 cm long silica cells) to gain information about the existence/amount of clay minerals.

Results and discussion

Physical/mechanical properties of the tuffs

After the first measurements, it was observed that the temple stones possess low unit weight and high porosity and water-absorption capacity values (Table 1). After artificial weathering cycles, a little

Table 2. Mineralogical and petrographic properties of the samples

	DTS	DBY	DFG	DKK
Groundmass	Volcanic glass matrix with flow pattern	Ferrous cubic-shaped opaque minerals and ignimbritic flow patterns from place to place	Tuffaceous matrix with flow pattern	Volcanic glass matrix with flow pattern
Crystals	Quartz, plagioclase, biotite, pyroxene, and circular opaque minerals	Plagioclase, sanidine, biotite and opaque minerals	Quartz, plagioclase, biotite and opaque minerals, hematite and clay minerals	Quartz, plagioclase, biotite and ferrous cubic-shaped opaque minerals
Rock fragments	Microlithic porphyritic andesitic rock fragments	Andesite rock fragment with phenocrysts, plagioclase and chlorite	Microlithic porphyritic volcanic rock fragments	Microlithic porphyritic andesitic rock fragments

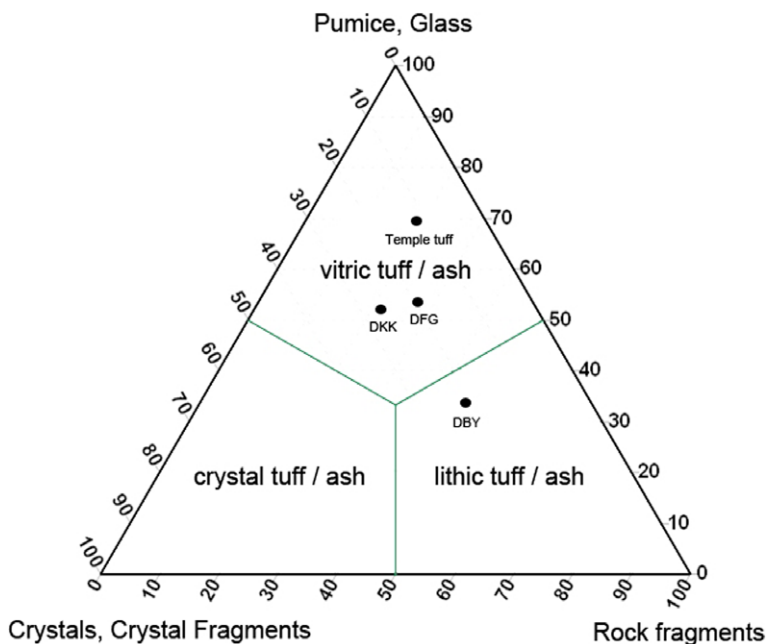


Fig. 9. Subdivision of the studied tuffs according to their fragment compositions.

decrease in the bulk density values was observed, and effective porosity values increased. DKK was the most affected. Dry unit weights were not influenced by the weathering tests (Fig. 4).

The colours of the tuffs became browner after the weathering tests. Change of L^*a^*b values of DBY after the wetting/drying cycle showed values similar to DTS1; the L^*a^*b values of DFG and

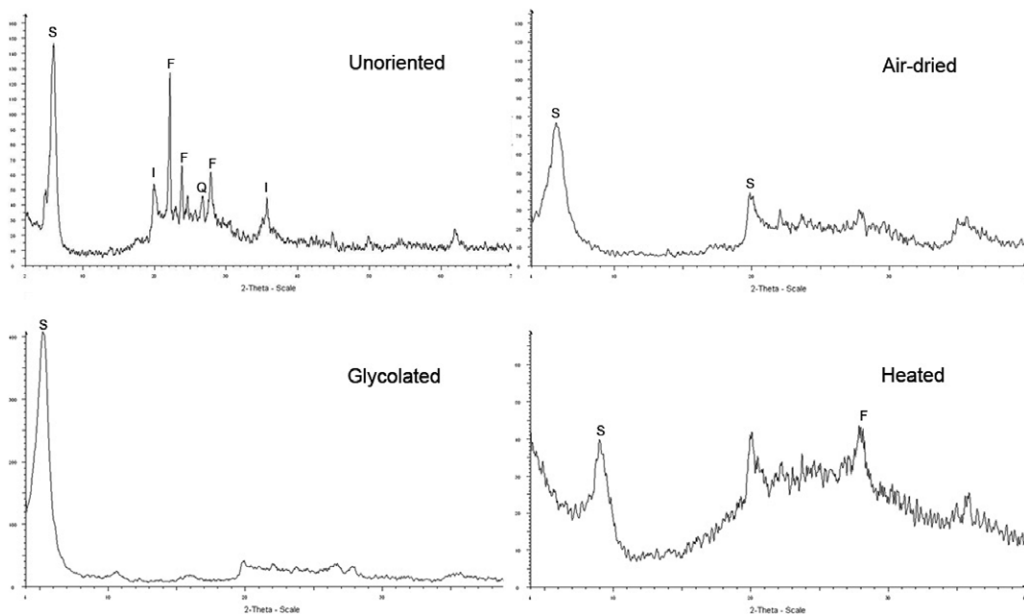


Fig. 10. XRD of DFG as oriented, air-dried, ethylene-glycolated and heated. S, smectite; F, feldspar; Q, quartz.

DKK were closer to DTS2. The colour of DKK was more affected by weathering (Fig. 5). (*CIE L*a*b** is a colour space specified by the International Commission on Illumination. It describes all the colours visible to the human eye and was created to be used as a reference. *L* is lightness whereas *a* represents red/green opponent and *b* represents yellow/blue opponent.)

UPV values were close to each other, except for DBY, which had a value of 1813 m/s (Fig. 6). Modulus of elasticity (Emod) which is calculated from the results of UPV and UCS of the samples followed a trend. A sudden decrease in the UPV values of the samples was observed after the third salt-crystallization cycle, whereas the wetting/drying cycles resulted in a gradual decrease.

In the analyses of the fresh samples, DBY had the lowest ($4.6 \text{ kg/m}^2/\text{s}^{1/2}$) and DKK had the highest ($11.4 \text{ kg/m}^2/\text{s}^{1/2}$) capillary coefficients. Capillary coefficients of DFG and DKK were increased by the salt-crystallization cycles. The values came closer to the capillary coefficients of the temple stone. The temple samples, DFG and DKK had low moisture absorption values. Results showed that pore size distribution of the temple stones, DFG and DKK was similar (Fig. 6).

The temple stone samples, DKK and DFG had low UCS values in the same range. DBY had the highest strength value (24.7 MPa). The modulus of elasticity v. UCS trend showed that, excepting DBY, all other rocks had low durability (Fig. 7a). According to ISRM (1981) and Deere & Miller (1966), the samples have low and very low UCS values.

Artificial weathering by the salt-crystallization cycles caused a drop of the UCS of all specimens. DFG and DKK were in the unsafe zone after the decrease (Fig. 7b). Reduction of the mechanical properties after the wetting/drying test was less pronounced than was seen with salt crystallization. However, UCS of DFG and DKK specimens still fit to the non-durable part of the Emod/UCS graph (Fig. 7c).

Furthermore, the strength of stone can be notably reduced due to the presence of moisture, because clays in a stone tend to attract water when exposed to moisture (Topal 1995). The general stone qualities, depending on the wet/dry strength ratios proposed by Winkler (1997), are: 80–90, good and safe; 70–80, further testing required; 60–70, unsafe for frost and hygric forces; <60, very poor quality, clay present. The DTS, DFG and DKK samples have similar wet/dry strength ratios of 61, 71 and 69%, respectively. Except for DBY (81%), they are all within the unsafe category, that is to say, they are likely to break down.

The Apollon Smintheus Temple tuff is brownish-grey vitric tuff having very high porosity and

Table 3. Major element composition of samples determined by XRF

	DTS1	DTS2	DBY	DFG	DKK
SiO ₂	64.5	63.2	66.7	65.8	65.2
TiO ₂	0.55	0.57	0.49	0.55	0.56
Al ₂ O ₃	17.8	17.3	16.8	17.8	19.3
Fe ₂ O ₃	3.29	3.69	2.91	3.29	3.38
MnO	0.07	0.11	0.12	0.1	0.07
MgO	1.77	1.78	0.5	1.74	2.42
CaO	2.78	4.37	1.88	2.28	2.31
Na ₂ O	1.63	2.34	4.11	2.05	1.46
K ₂ O	6.07	5.34	5.7	5.56	4.5
P ₂ O ₅	0.78	0.41	0.11	0.15	0.13
Total	99.94	100.53	99.99	100.03	100.43
Al/(Fe + Al)	0.84	0.82	0.85	0.84	0.85
Al/(Al + Mg)	0.91	0.91	0.97	0.91	0.89

very low unit weight. The tuff, which has high capillary suction and low UCS and modulus of elasticity, is evaluated as ‘moderately weak’ and ‘weak’ in relation to rock mechanical standards (ISRM 1981; ASTM D5313/D5313M 2012). Although samples had low durability values, it should be remembered that the stone has been present in the temple foundation for 2000 years.

Mineralogical, petrographical and chemical properties of the tuffs

Observations in optical microscopy confirmed that all samples are volcanic and have volcanic rock fragments in various forms. The DFG and DKK samples include volcanic glass fragments with ignimbritic flow patterns. The DBY thin section is different from the others, as it has sanidine instead

Table 4. Minor element composition of samples determined by XRF

	DTS1	DTS2	DBY	DFG	DKK
F	0.2480	0.2470	0.1710	0.2570	0.2100
SO ₃	0.0787	0.1790	0.0616	0.0438	0.0382
Cl	0.0788	0.1680	0.0302	0.0695	0.0453
CuO	0.0075	0.0078	0.0051	0.0077	0.0084
ZnO	0.0109	0.0090	0.0059	0.0090	0.0205
Ga ₂ O ₃	0.0040			0.0035	
Rb ₂ O	0.0362	0.0298	0.0245	0.0320	0.0292
SrO	0.0504	0.0565	0.0585	0.0557	0.0633
Y ₂ O ₃	0.0135	0.0116		0.0135	
ZrO ₂	0.0530	0.0517	0.0495	0.0491	0.0519
BaO	0.1130	0.1240	0.2440	0.1620	0.1450
PbO	0.0107	0.0257	0.0107	0.0081	0.0151
Br		0.0015			
ThO ₂		0.0053	0.0033	0.0054	
Nb ₂ O ₅			0.0027		0.0032

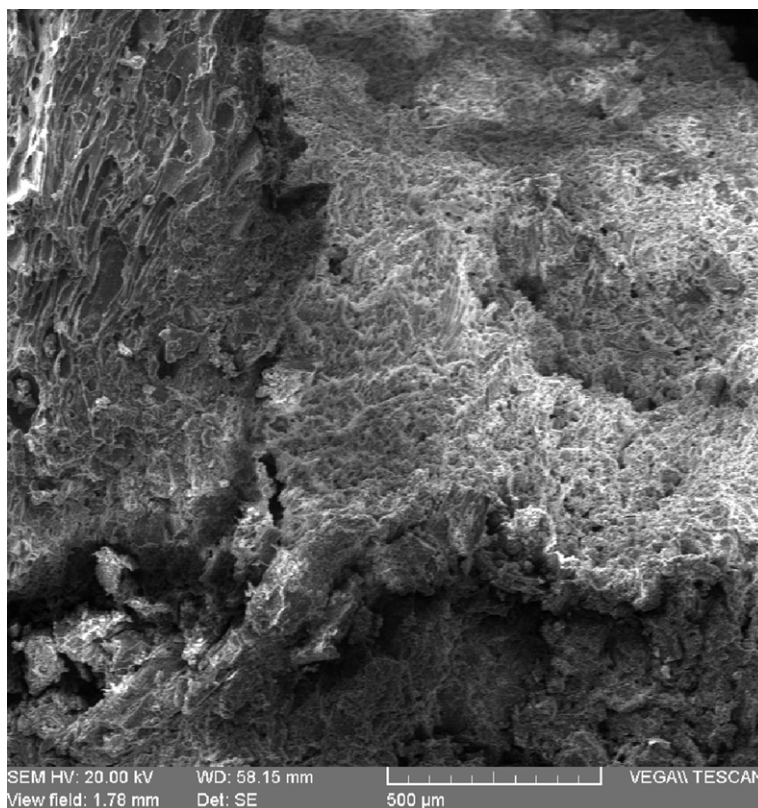


Fig. 11. SEM image of DFG showing the honeycomb morphology of smectite.

of quartz and its matrix/crystal ratio is higher than the others (Fig. 8 & Table 2). The percentage of modal composition of the tuffs using pumice/glass, crystals/crystal fragments and rock fragments according to Schmid (1981) reveal that they are vitric tuff/ash (Fig. 9).

In addition, XRD analysis shows that all samples had quartz and feldspar as main minerals. The DBY and temple stone had biotite as a secondary mineral. X-ray diffractograms of all other samples show smectite (001) basal diffraction and amorphous phase, namely, volcanic glass. In the oriented XRDs, it is seen that in the temple stone, DFG and DKK graphs, the smectite (001) basal diffraction

occurred in the range of 10–18 Å in air-drying, ethylene glycol and heat treatments. The basal spacings of the air-dried smectites were in the range of 12–15 Å. However, with the ethylene glycol treatment, it expanded uniformly to 17.2 Å, and the first d-spacing collapsed to 10 Å upon heat treatment. In DBY, it was observed in air-drying, ethylene glycol and heat treatments that the illite was visible at 10 Å and chlorite had basal reflections of 14 and 7 Å (Fig. 10).

With regard to elemental analysis results by XRF, it is considered that samples that have low SiO₂ values have low strength values, as seen in (Tables 3 & 4). Without DBY all samples have low values.

In the SEM analyses, honeycomb structures with a chemical composition consistent with smectite of 10–200 μm (Fig. 11) were present in two different arrangements. Some were seen in ball-shaped units and covered the inner surfaces of holes. The quantitative analysis of the MBA test provided the amount of clay in the sample. Assuming that other minerals do not adsorb the methylene blue dye, cation-exchange capacity (CEC)

Table 5. MBA and CEC values of samples

Sample code	MBA (g/100 g)	CEC (meq/100 g)
DTS1	2.5	7.82
DFG	3.25	10.16
DKK	5.5	17.20
DBY	0.6	1.88

values revealed the clay content, and DBY had the lowest clay content with the lowest CEC values (Table 5).

Conclusion

These analyses demonstrate that Fatma Gerdan Quarry (DFG) and Kızılkeçili Quarry (DKK) samples have visual and structural properties similar to the temple stone. They are brownish-grey, welded vitric tuff having very high porosity and very low unit weights, while the Babakale Yolu (DBY) tuff is grey latitic tuff having lower porosity and a higher unit weight. The DTS, DFG and DKK tuffs having high capillary suction, low UCS and modulus of elasticity are evaluated as 'moderately weak' and 'weak', like the temple tuff. The low durability is due to the presence of smectite-type swelling clay that is the alteration product of volcanic glass. Having low capillary suction, high UCS and modulus of elasticity, DBY was classified as 'medium' strength, unlike the temple tuffs. Although the stones of DFG, DKK and the temple do not show high values of engineering properties, very high durability properties are not necessary in practice. So these stones are considered appropriate for building stones provided they are used under shelter and not exposed to rainwater. Briefly, it can be concluded that those two quarries (DFG and DKK) are possibly the ancient tuff quarries from which stones for the temple were taken.

This research demonstrates that the deep interpretation of the physical and mechanical properties of tuff after artificial weathering tests is useful for the assessment of stone deterioration. UCS and UPV should be used in all studies related to tuff, because the inner damage to the rock can best be determined by those values. The adverse effect of water on the samples is obviously seen in this study. Having considered the presence of clay minerals in the microstructure of the tuffs, their swelling and contraction during the wetting/drying cycles might have caused the disintegration of the tuff samples after certain number of wetting/drying cycles. Smectite-type clay minerals swell considerably upon wetting and shrink upon drying. Therefore, for conservation purposes, the direct contact of the tuff with water from any source should be avoided. Further studies related to the quarries are also needed.

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