

LOW ENERGY COMPLEX-FORMING DYNAMICS ON ATOM-DIATOM REACTIONS



T. González-Lezana

IFF-CSIC, Serrano 123 Madrid (Spain)

The dynamics of atom-diatom reactions can vary with the collision energy, and direct and insertion pathways can compete [1]. Recent work has treated low energy collisions by means of statistical methods thus revealing the possible complex-forming character of the process at such energy regimes [2,3]. We have been studying the H⁺+H₂ process [4,5] and isotopic variants, such as $D^+ + H_2$ [6,7], by means of a statistical quantum model (SQM) [8,9] in comparison with exact quantum mechanical (EQM) methods and experimental results. Interestingly the



main dynamical features are conveniently reproduced. A similar investigation for LiYb+Li \rightarrow Li₂+Yb [10] revealed the energy range at which the process corresponds mainly to a S-wave dynamics and the apparent limits of a statistical description.



 $^{6}Li^{174}Yb+^{6}Li \rightarrow ^{6}Li_{2}+^{174}Yb$ mediated by a deep potential well at the linear geometry



Cross sections have to be calculated with an appropriate description of the corresponding asymptotic region. Limits of the validity of the SQM cross sections depending on Rmax are shown at the left. Rate coefficients in comparison with EQM and experiment (right) shows good agreement.

Essentially S-wave process below 10⁻⁴ K. SQM cross sections in good agreement with EQM results



Rate coefficients for ${}^{6}Li^{174}Yb(v=j=0)+{}^{6}Li$



Behaviour of the rate constants at low temperature depends the also on calculation of the cross sections at the low energy On the left, regime. different comparison of approaches.

References

[1] F. Dayou et al., J. Chem. Phys. **128**, 174307 (2008). [2] M. L. González-Martínez et al., Phys. Rev. A. 90, 052716 (2014). [3] M. Lara *et al., Phys. Rev. A* **91**, 030701(R) (2015). [4] P. Honvault et al., Phys. Rev. Lett. **107**, 023201 (2011). [5] T. González-Lezana et al., Int. Rev. Phys. Chem.. 33, 371 (2014). [6] T. González-Lezana et al. J. Phys. Chem. A **118**, 6416 (2014). [7] T. González-Lezana et al. J. Chem. Phys. **139**, 054301 (2013). [8] E.J. Rackham *et al., J. Chem. Phys.* **119**, 12895 (2003). [9] T. González-Lezana et al., Int. Rev. Phys. Chem. 26, 29 (2015). [10] C. Makrides *et al., Phys. Rev. A* **91**, 012708 (2015). [11] L. Pagani *et al., AA* **551**, A38 (2013). [12] L. Pagani et al., Phil. Trans. Royal Soc. **370**, 5200 (2012).