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1 Performance testing of a cross-flow membrane-based liquid desiccant 2 dehumidification system

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11 Abstract

12 A membrane-based liquid desiccant dehumidification system is one of high energy efficient 13 dehumidification approaches, which allows heat and moisture transfers between air stream and 14 desiccant solution without carryover problem. The system performance is investigated 15 experimentally with calcium chloride, and the impacts of main operating parameters on 16 dehumidification effectiveness (i.e. sensible, latent and total effectiveness) are evaluated, which 17 include dimensionless parameters (i.e. solution to air mass flow rate ratio m^* and number of 18 heat transfer units NTU) and solution properties (i.e. concentration C_{sol} and inlet temperature 19 $T_{sol,in}$). The sensible, latent and total effectiveness reach the maximum values of 0.49, 0.55, 20 and 0.53 respectively at $m^* = 3.5$ and NTU = 12, and these effectiveness are not limited by m^* 21 and NTU when $m^* > 2$ and NTU > 10. Both the latent and total effectiveness increase with 22 C_{sol} , while almost no variation is observed in the sensible effectiveness. All effectiveness can 23 be improved by decreasing $T_{sol.in}$. The experimental data provide a full map of main design 24 parameters for the membrane-based liquid desiccant air conditioning technology. 25

Keywords: liquid desiccant, membrane-based, dehumidification, performance testing,
effectiveness
effectiveness
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37 Nomenclature

Α	membrane surface area (m ²)
AH	absolute humidity (kg/m ³)
<i>c</i> _p	specific heat capacity (J/kgK)
С	concentration (%)
C_r^*	capacitance ratio
d	width of the rectangular channel (m)
h	convective heat transfer coefficient (W/m^2K)
Н	height of the rectangular channel (m)
H^*	operating factor
k	thermal conductivity (W/mK)
L	characteristic length of the rectangular channel (m)
m^*	solution to air mass flow rate ratio
'n	mass flow rate (kg/s)
Nu	Nusselt number
NTU	number of heat transfer units
NTU _m	number of mass transfer units
Р	atmospheric pressure (pa)
P_{v}	equilibrium vapour pressure of desiccant solution (pa)
RH	relative humidity (%)
Т	temperature (°C)
U	overall heat transfer coefficient (W/m^2K)
<i>Ϋ</i>	volumetric flow rate (l/min)
W	humidity ratio (kg/kg)

Greeks

ε	effectiveness
δ	thickness of membrane (m)
ρ	density (kg/m ³)

Subscripts

air	air flow
crit	critical value
in	inlet
lat	latent

тет	membrane
min	minimum value
out	outlet
sen	sensible
sol	solution flow
tol	total

39 1. Introduction

40 Buildings consume a significant part of the global total energy, particularly heating, ventilation 41 and air-conditioning (HVAC) systems are responsible for around 50% of the energy consumed 42 in buildings [1]. As a matter of fact, the energy consumption for dehumidification process 43 accounts for 20-40% of the total energy used in HVAC systems, and it can be higher when 100% 44 fresh air ventilation is required for better indoor environment [2]. Without proper air 45 dehumidification, occupants would feel uncomfortable and mildew would grow on building 46 interior walls in the humid region. Furthermore, production safety and quality would be 47 seriously affected by high humidity level [2]. It has been shown that the building energy 48 consumption could be decreased by 20-64% with efficient dehumidification technologies [3]. 49 Currently, cooling coil is mostly preferred for dehumidification [4], which adopts cooled water 50 as the cold medium generated from vapour compression system (VCS). The conventional VCS 51 has advantages of good stability in performance, long life and a reasonable electrical COP 52 (between 2 and 4) [5]. However, the working fluids used in VCS such as R-22, R-410A and R-53 134A with the high global warming potential are harmful to the environment. Furthermore, 54 VCS consumes substantial amount of electrical energy [6]. In the traditional cooling coil, air 55 dehumidification is undertaken simply by cooling air below its dew point for condensation in 56 order to reduce its moisture content. Normally, this type of dehumidification is followed by

reheating the dehumidified air to a desired temperature. Consequently, this combined process
consumes a considerable amount of energy to cool (typically using a VCS) and heat (using hot
water or electricity) the supply air [7].

60 In the traditional desiccant system, the vapour pressure gradient between humid air and 61 desiccant results in heat and moisture transfers [8, 9]. The system operates using either solid or 62 liquid desiccant. Solid desiccant system is compact, simple and less subject to desiccant carryover and corrosion problems, while liquid desiccant system has lower regeneration 63 64 temperature, higher dehumidification capacity and lower air side pressure drop [10]. Liquid 65 desiccants can be regenerated using low-grade heat sources such as solar energy, and the 66 regenerated solution can be used as energy storage medium as well [11]. In such way, the liquid 67 desiccant system has been well developed recently.

The traditional liquid desiccant system commonly adopts the packed bed, where air and desiccant are in direct contact. Comprehensive researches have been conducted on the direct contact system [12-15], and it has been found that air conditioning energy consumption reduces by up to 26-80% in the hot and humid climate. However, in the direct contact system, small desiccant droplets are carried over by the supply air to the indoor environment, which badly affects the occupant health, building structure and furniture [2].

74 Recently, selectively permeable membrane has been used to replace the packed bed as the heat 75 and mass transfer medium to overcome the desiccant droplet carryover problem. Semipermeable membrane is able to prevent the solution from carrying over into the supply air, 76 77 while selectively permitting heat and moisture transfers between the liquid desiccant and supply 78 air [2, 16-20]. The selectively permeable membrane can be classified into two types: parallel 79 plate [21-33] and hollow fiber [34-38]. Several researches have been carried out to investigate 80 the membrane-based dehumidifier performance. For example, Moghaddam et al. [21] 81 experimentally and numerically studied different parameter influences on the steady state 82 performance of a small-scale counter-flow liquid-to-air membrane energy exchanger (LAMEE), 83 these parameters include thermal capacity ratio (Cr^*) , number of heat transfer units (NTU) and 84 number of mass transfer units (NTU_m) . Hemingson et al. [22, 23] developed a model of 85 moisture transfer resistance between the membrane and solution for a counter-flow LAMEE. 86 and conducted experimental tests under a range of outdoor weather conditions. Fan et al. [24, 87 25] built a mathematical model for a single cross-flow LAMEE, which is applied to a run-88 around LAMEE system consisting of both dehumidifier and regenerator. The impacts of Cr^* , 89 NTU and NTU_m on both sensible and latent effectiveness of the run-around system are 90 evaluated. Seyed-Ahmadi et al. [26, 27] developed a mathematical model to simulate the 91 transient behaviours of a single cross-flow LAMEE and a run-around LAMEE, which is also 92 compared with Fan's steady state model. Apart from counter and cross flows, an innovative 93 flow configuration, counter-cross flow, has been investigated. Vali et al. [28, 29] modelled a 94 run-around LAMEE system using the counter-cross flow exchangers as dehumidifier and 95 regenerator, and assessed the steady state system performance. Moghaddam et al. [30] studied 96 the effect of the direction of heat and mass transfer inside the counter-cross flow LAMEE 97 through experiment and numerical simulation. However, in the above researches, the 98 fundamental data required for mathematical modelling such as Nusselt number (Nu) and 99 Sherwood number (Sh) are simply borrowed from well-known books, which are generally 100 obtained under uniform temperature or heat flux boundary condition. Thus they are unable to 101 reflect the real heat and mass transfer properties. To solve this problem, Huang et al. [31] 102 proposed a mathematical model for the cross-flow parallel-plate membrane module to 103 conjugate heat and mass transfer in a cross-flow LAMEE under a fully developed flow

104 condition. The fundamental data of *Nu* and *Sh* under various aspect ratios are calculated.
105 However, the assumption of a fully developed flow is not reasonable in this model. Accordingly,
106 they [32] improved this model by considering the effect of the developing entrance length on
107 the fluid flow pattern.

- Most of the researches in literatures focus on numerical modelling of heat and mass transfer in 108 109 LAMEE. Some of them experimentally assess the LAMEE performance for different heat and 110 mass transfer directions or liquid desiccant types [3] [30]. Some researchers analyse the impacts of NTU, solution to air mass flow rate ratio (m^*) , and solution inlet temperature $(T_{sol.in})$ on 111 112 whole liquid desiccant air-conditioning system [39]. A few studies investigate the membranebased dehumidifier performance with regard to NTU, m^* and solution inlet concentration (C_{sol}) 113 114 [21-25][40]. Thus in order to get a full map of the operating characteristics of a LAMEE, a 115 series of experimental tests are carried out in this study to evaluate the performance of a fullscale membrane-based cross-flow liquid desiccant dehumidifier. The experimental results are 116 presented with regard to the four important operating parameters: NTU, m^* , $T_{sol,in}$ and C_{sol} . 117 This work provides a comprehensive parametric study on the dehumidifier performance 118 119 through experimental investigations, which supplies valuable data for liquid desiccant air 120 conditioning system design.
- 121

122 2. Test Apparatus and Instrumentation

A test facility is designed and built in the laboratory to assess the performance of a cross-flow membraned-based liquid desiccant dehumidification system under different operating conditions. The test rig mainly consists of a dehumidifier, a regenerator, two solution tanks and three heat exchange units. The schematic diagram of the test rig is shown in Fig. 1, and the dehumidifier specifications and membrane physical properties are given in Table 1.

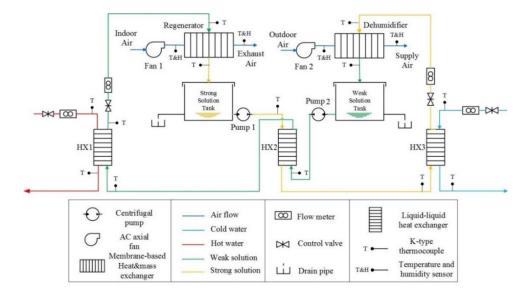


Fig. 1. Schematic diagram of the laboratory test rig

Table 1

Symbol	Unit	Value
L^*	m	0.23
W*	m	0.41
Н	m	0.21
d _{air}	m	0.0077
d_{sol}	m	0.0043
δ_{mem}	m	1.05×10^{-4}
k _{mem}	W/mK	0.3

131 Dehumidifier specifications and membrane physical properties

2.1 Air loop

The outdoor air flows into the dehumidifier where both its moisture content and temperature are reduced by cold desiccant solution, then it leaves the dehumidifier unit at dry and cool state. Its flow rate is controlled by adjusting an AC axial fan rotation speed (ebm-papst Mulfingen GmbH & Co. KG). An air conditioning unit and a humidifier are used to simulate the hot and humid weather condition. The dehumidifier structure is illustrated in Fig. 2. The dehumidifier has a dimension of 410mm (L) x 230mm (W) x 210mm (H) with 11 air channels and 11 solution channels. As can be seen in Fig. 2, wavy polyethylene sheets are used to support the air channels. Air and desiccant solution flows are in a cross configuration. Heat and mass transfer takes place in semi-permeable membranes that separate the air and solution channels. Three gauze layers are paved on the top surface of the dehumidifier unit to ensure even solution distribution.

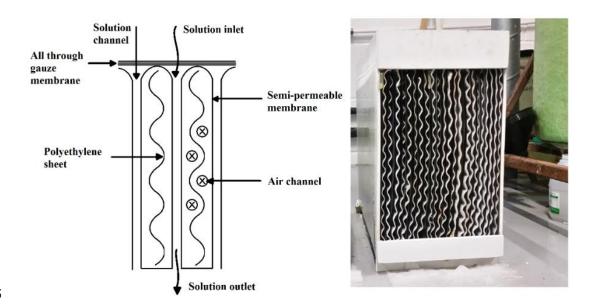


Fig. 2. Schematic diagram of the dehumidifier

150 2.2 Liquid loop

151 Calcium chloride (CaCl₂) solution is circulated in the system by two identical pumps (15W) 152 centrifugal magnetically driven type with flow rate range of 0-10L/min) and their flow rates are 153 measured by two liquid flow indicators (Parker UCC PET 1-15 L/min). Before entering the 154 dehumidifier, the strong solution is pre-cooled in a brazed plate heat exchanger HX2 by the 155 weak solution, and further cooled in HX3 by cold water. Afterwards, the strong solution is 156 pumped into the dehumidifier and sprayed through a nozzle. Then the strong solution flows 157 downwards in the solution channels, absorbs the moisture from the air and becomes weak 158 solution. The weak solution is then pumped into HX2 for pre-heating, followed by further-159 heating in HX1 by hot water. The heated weak solution is pumped into the regenerator. The re-160 concentrated solution from the regenerator is collected by a stainless steel solution tank. Then 161 the strong solution is pumped out of the strong solution tank to HX2 and a whole circuit is 162 completed. The desiccant solution and air transport properties are listed in Table 2.

163 **Table 2**

164 Air and desiccant solution transport properties

Symbol	Unit	Value
k _{air}	W/mK	0.03
k _{sol}	W/mK	0.5
D _{air}	m^2/s	2.46×10^{-5}
D _{sol}	m^2/s	0.892×10^{-2}
$c_{p,air}$	J/kgK	1020
C _{p,sol}	J/kgK	3200
$ ho_{air}$	kg/m^3	1.29

165

166 2.3 Instrumentation

167 Air velocities through the dehumidifier and regenerator are measured at the air duct outlets by a thermo-anemometer (Testo 405) with a measuring range up to 10m/s. All fans at the inlets of 168 169 the dehumidifier and regenerator are equipped with infinitely variable speed controllers to 170 adjust air flow rates. All air inlets and outlets are instrumented with humidity and temperature 171 sensors (Sensirion Evaluation KIT EK-H4). Water and desiccant solution temperatures are 172 measured with K-type thermocouples, and all sensors are connected to a DT500 data logger. 173 The dehumidifier, regenerator, heat exchangers, storage tanks and pipes are well insulated to 174 reduce the environment influence.

175 A correlation based on Melinder's work [41] is used to determine the solution concentration,

176 which is a function of solution density and temperature. The correlation is given as:

177 $C_{sol} = -253.147703 + 0.0443853996T_{sol} + 0.000163666247T_{sol}^2$

178 $+0.331709855\rho_{sol} - 0.000079370267\rho_{sol}^2$

(1)

- 179 Where C_{sol} is solution concentration (%). T_{sol} is solution temperature (°C) and ρ_{sol} is solution
- 180 density (g/ml). The solution density is measured by Brannan hydrometers. All measurement
- 181 devices and their accuracies are listed in Table 3

182 **Table 3**

183 Measurement devices and uncertainties

Device	Measurement	Range	Uncertainty
Testo thermos-anemometer	Air velocity	0-10 m/s	±5%
Sensiron Evaluation KIT EK-H4	Temperature	-40-125 °C	±0.4%
	Relative humidity	0-100 %	±3%
K-type thermocouple probe	Temperature	0-1100 °C	±0.75%
DT500 Datalogger	Data acquisition	-	±0.15%
Parker UCC PET liquid flow indicator	Solution flow rate	1-15 L/min	±5%
Parker liquid flow indicator	Water flow rate	2-22 L/min	±2%
Brannan hydrometer	Density	1-1.2 g/ml	±2%
Brannan hydrometer	Density	1.2-1.4 g/ml	±2%

184

185 2.4 Uncertainty analysis

186 Uncertainty analysis provides a measure of the errors during a measurement associated with a 187 calculated value. Thus it is of vital importance to estimate uncertainties during the experiment. 188 Based on a method of propagation of uncertainties introduced by Taylor [42], when the 189 computed value q is any function of several variables x, \dots, z , the uncertainty of q can be 190 obtained by:

191
$$\delta q = \sqrt{\left(\frac{\partial q}{\partial x}\delta x\right)^2 + \dots + \left(\frac{\partial q}{\partial z}\delta z\right)^2}$$
(2)

Based on Eq. (2), the absolute uncertainty of a calculated value can be derived. Error bars are
included in the graphs for experimental result analyses. The detail uncertainties for all target
measurements are given in Appendix.

195

196 **3 Experimental methodology**

197 The system performance indicators and relevant parameters are defined in this section, and the

198 experimental procedures for dimensionless parameter and solution property tests are presented.

- 199 3.1 Dehumidifier performance evaluation
- 200 3.1.1 Operating parameters
- 201 3.1.1.1 Capacitance ratio (C_r^*)

Heat capacity rate is defined as the product of specific heat capacity and mass flow rate (W/K).

203 Thus the heat capacities of desiccant solution and air are expressed by Eqs. (3)-(4) [43].

$$204 \qquad C_{sol} = \dot{m}_{sol} c_{p,sol} \tag{3}$$

$$205 \qquad C_{air} = \dot{m}_{air} c_{p,air} \tag{4}$$

- 206 Where \dot{m}_{sol} is solution mass flow rate (kg/s), \dot{m}_{air} is air mass flow rate (kg/s), $c_{p,sol}$ is
- solution specific heat capacity (J/kgK) and $c_{p,air}$ is air specific heat capacity (J/kgK).
- 208 Then the capacitance ratio (or heat capacity rate ratio) C_r^* is given by Eq. (5) [11].

$$209 \qquad C_r^{\ *} = \frac{c_{sol}}{c_{air}} = \frac{\dot{m}_{sol}c_{p,sol}}{\dot{m}_{air}c_{p,air}} \tag{5}$$

- 210 3.1.1.2 Solution to air mass flow rate ratio (m^{*})
- Solution to air mass flow rate ratio is a measurement of relative flow rate of two heat exchanging fluids. In this experiment, the solution to air mass flow rate ratio (m^*) is used since it is a more straight forward parameter. The solution to air flow rate ratio is defined as:

$$214 \qquad m^* = \frac{\dot{m}_{sol}}{\dot{m}_{air}} \tag{6}$$

- 215 3.1.1.3 Operating factor (H^*)
- 216 Operating factor is a dimensionless number defined as the ratio between the latent energy
- 217 difference and sensible energy difference for the air and desiccant solution at the inlets [29].

218
$$H^* = \frac{\Delta H_{lat}}{\Delta H_{sen}} \approx 2500 \frac{W_{air,in} - W_{sol,in}}{T_{air,in} - T_{sol,in}}$$
(7)

- 219 Where $T_{air,in}$ and $T_{sol,in}$ are air and solution temperatures respectively (°C), $W_{air,in}$ is air
- humidity ratio (kg/kg) and $W_{sol,in}$ is solution equilibrium humidity ratio (kg/kg).
- 221 3.1.1.4 Number of heat transfer units (NTU)
- 222 Effectiveness-NTU method is one of the most commonly used ways for heat exchanger analysis.
- 223 Compared with *log-mean-temperature-difference* method, it provides a superior way to analyse
- heat exchanger in terms of non-dimensional variables [44].

$$225 \qquad NTU = \frac{UA}{C_{min}} \tag{8}$$

226
$$U = \left[\frac{1}{h_{air}} + \frac{\delta}{k_{mem}} + \frac{1}{h_{sol}}\right]^{-1}$$
(9)

227 Where U is the overall heat transfer coefficient (W/m^2K) , A is membrane surface area (m^2) ,

228 C_{min} is the minimum value of air and desiccant solution heat capacity rates (W/K), h_{air} is air 229 convective heat transfer coefficient (W/m^2K) , h_{sol} is solution convective heat transfer 230 coefficient (W/m^2K) , δ is membrane thickness (m) and k_{mem} is membrane thermal 231 conductivity (W/mK).

- 232 3.1.1.5 Number of mass transfer units (NTU_m)
- 233 The number of mass transfer units is defined as following:

$$234 \qquad NTU_m = \frac{U_m A}{\dot{m}_{min}} \tag{10}$$

235
$$U_m = \left[\frac{1}{h_{m,air}} + \frac{\delta}{k_m} + \frac{1}{h_{m,sol}}\right]^{-1}$$
 (11)

- 236 Where U_m is the overall mass transfer coefficient (kg/m^2s) , \dot{m}_{min} is the minimum mass flow
- rate of air and desiccant solution (kg/s), $h_{m,air}$ is convective mass transfer coefficient of air
- 238 $(kg/m^2s), h_{m,sol}$ is convective mass transfer coefficient of desiccant solution $(kg/m^2s), \delta$ is
- thickness of membrane (m), k_m is membrane water permeability (kg/m s). It has been showed
- 240 the convective mass transfer coefficient of desiccant solution is much higher than that of the
- 241 air, thus $\frac{1}{h_{m sol}}$ can be neglected for the simplicity.
- 242 3.1.2 Effectiveness
- 243 Effectiveness is the most important parameter used to evaluate the performance of a heat and 244 mass exchanger [45]. Three types of effectiveness have been defined in this study: sensible 245 effectiveness (ε_{sen}), latent effectiveness (ε_{lat}) and total effectiveness (ε_{tot}). ε_{sen} is the ratio 246 between the actual and maximum possible rates of sensible heat transfer in a heat exchanger. 247 ε_{lat} is the ratio between the actual and the maximum possible moisture transfer rates in a mass exchanger. ε_{tot} is the ratio between the actual and maximum possible energy (enthalpy) 248 249 transfer rates in a heat and mass exchanger. The capacity rate of desiccant solution is higher 250 than that of the air, which means $Cr^* \ge 1$, then the sensible, latent and total effectiveness are 251 defined by Eqs. (12) - (14). [46].

252
$$\varepsilon_{sen} = \frac{T_{air,in} - T_{air,out}}{T_{air,in} - T_{sol,in}}$$
(12)

253
$$\varepsilon_{lat} = \frac{W_{air,in} - W_{air,out}}{W_{air,in} - W_{sol,in}}$$
(13)

254
$$\varepsilon_{tol} = \frac{\varepsilon_{sen} + H^* \varepsilon_{lat}}{1 + H^*}$$
(14)

Where $T_{air,out}$ is air temperature at the outlet (°C) and $W_{air,out}$ is air humidity ratio at the outlet (kg/kg).

257

258 *3.2 Experimental procedure*

259 3.2.1 Dimensionless parameter tests

At first, the experiment aims to explore the impacts of number of heat transfer units (NTU) and solution to air mass flow rate ratio (m^*) on the dehumidifier performance. The air inlet condition is set at a temperature of 30°C and relative humidity (RH) of 70%, and the solution concentration is 39%. *NTU* is set in the range of 4 to 12. For each *NTU*, seven tests are conducted with m^* set as 0.5, 1, 1.5, 2, 2.5, 3 and 3.5. Because air heat capacity rate is always lower than desiccant solution's, thus Eq. (8) can be written as:

$$266 \qquad NTU = \frac{UA}{c_{p,air}\dot{m}_{air}} \tag{15}$$

In order to determine the required air mass flow rate for a corresponding NTU, the overall heat transfer coefficient (U value) needs to be decided at first. According to Eq. (9), δ and k_{mem} are physical properties of the membrane material, so h_{air} and h_{sol} need to be determined. In this

- 270 experiment, these two parameters are obtained from air side Nusselt number (Nu_{air}) and
- 271 solution side Nusselt number (Nu_{sol}).
- 272 Many literatures have investigated Nu with different channel aspect ratios based on a constant
- temperature or heat flux boundary condition. However, according to Huang's comments [31],
- these values are unable to accurately reflect heat and mass transfer properties in the membrane
- 275 module since membrane surface boundary condition is neither uniform temperature
- 276 (concentration) nor uniform heat flux (mass flux). In literature [31], the natural formed
- 277 boundary layer has been simulated and the values of Nu under different channel aspect ratios
- are derived as given in Table 4.

279 **Table 4**

- Fully developed Nusselt numbers ($Nu_{C,a}$ for air side and $Nu_{C,s}$ for solution side) in the parallel-
- 281 plate membrane channel for various aspect ratios [31]

Aspect ratio	Nu _{C,a}	Nu _{C,s}
1.0	3.12	3.41
1.43	3.23	3.64
2	3.48	4.05
3	4.15	4.74
4	4.61	5.35
8	5.79	6.41
50	7.54	7.91
100	7.7	8.08
œ	_	_

282

283 The air and solution side aspect ratios are 27 and 47 respectively in this study, thus the 284 corresponding Nusselt numbers can be calculated: $Nu_{air} = 6.58$, $Nu_{sol} = 7.74$ referred to 285 Table 2. The characteristic length of a rectangular channel can be obtained by applying L= 286 (4dH)/[2(d+H)], where d is the channel width (m) and H is the channel height (m), which are 287 given in Table 1. For the dehumidifier, the air side and solution side characteristic length are 288 0.015 m and 0.008 m respectively. Subsequently, h_{air} and h_{sol} can be derived as 13.16 $W/m^2 K$ and 532.13 $W/m^2 K$ respectively. Then the U value is calculated as 12.78W/m K. For 289 290 a given NTU, the required air mass flow rate can be derived from Eq. (15), correspondingly a 291 series of m^* values are obtained. Based on Eq. (6), once the air mass flow rate is determined, a 292 series of solution mass flow rates corresponding to different m^* can be obtained as well. All 293 target measurements are shown in Table 5.

- 294
- 295

296

298 **Table 5**

Ν	ITU		4		6		8	1	10		12
m^*	Cr*	ṁ _{sol} (kg∕s)	V _{sol} (l∕min)	ṁ _{sol} (kg∕s)	∛ _{sol} (l/min)	ṁ _{sol} (kg∕s)	∛ _{sol} (l/min)	ṁ _{sol} (kg∕s)	∛ _{sol} (l/min)	ṁ _{sol} (kg∕s)	V _{sol} (l∕min)
0.5	1.55	0.030	1.292	0.020	0.861	0.015	0.646	0.012	0.517	0.010	0.431
1	3.1	0.061	2.584	0.040	1.722	0.030	1.292	0.024	1.033	0.020	0.861
1.5	4.65	0.091	3.875	0.061	2.584	0.045	1.938	0.036	1.550	0.030	1.292
2	6.2	0.121	5.168	0.081	3.445	0.061	2.584	0.048	2.067	0.040	1.722
2.5	7.75	0.151	6.459	0.101	4.306	0.076	3.230	0.061	2.584	0.050	2.153
3	9.3	-	-	0.121	5.167	0.091	3.875	0.073	3.100	0.061	2.584
3.5	10.85	-	-	0.141	6.028	0.106	4.521	0.085	3.617	0.071	3.014

299 Target measurements for dimensionless parameter tests

300

301 3.2.2 Solution property tests

The next stage of experiment aims to investigate the dehumidifier performance variations with solution inlet temperature (T_{sol}) and solution concentration (C_{sol}). In this stage, the air inlet condition is set as 30°C and 70% RH, and *NTU* and m^* are set to be 8 and 2 respectively. The testing range of the solution temperature is from 18°C to 23°C. For each *NTU*, three solution concentrations are tested: 33%, 36% and 39%. Since *NTU* and m^* are kept constant, the air and solution flow rates are unchanged. The air mass flow rate is calculated to be 0.030 kg/s and the solution mass flow rate is 0.061 kg/s (volume flow rate 2.583 *l/min*).

309 For analysis, the air specific humidity or humidity ratio (kg/kg) needs to be determined. A

310 correlation between RH (%) and absolute humidity (AH) (kg/m^3) is derived by Mander [47]:

311
$$AH = \frac{6.112 \times e^{\left[\frac{17.67 \times T}{T+243.5}\right] \times RH \times 2.1674}}{1000(273.15+T)}$$
(16)

312 Where *T* is air temperature (°C). Then air specific humidity W_{air} (kg/kg) can be calculated 313 by:

314
$$W_{air} = \frac{AH}{\rho_{air}}$$
(17)

315 Where ρ_{air} is air density (kg/m³).

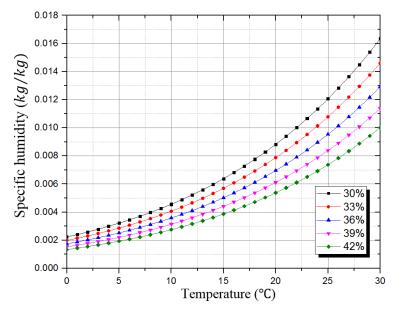
The equilibrium specific humidity (W_{sol}) is used to calculate both the sensible and latent effectiveness, the relationship between the specific humidity and vapour pressure is given by [40]:

319
$$W_{sol} = 0.62198 \frac{P_v}{P - P_v}$$
 (18)

Where *P* is the atmospheric pressure (*Pa*) and *P_v* is vapour pressure of desiccant solution (*Pa*). The equilibrium vapour pressure of desiccant solution is a function of T_{sol} and C_{sol} (*P_v* = $f(T_{sol}, C_{sol})$), the correlation is given by [49]:

323
$$Log P_{\nu} = KI \left[A - \frac{B}{T - E_s} \right] + \left[C - \frac{D}{T - E_s} \right]$$
(19)

- 324 Where P_{ν} is solution equilibrium vapour pressure (kPa), K is an electrolyte parameter relating
- 325 to solute $(CaCl_2)$; A, B, C, D and E_s are parameters regarding to solvent (water). Accordingly,
- 326 a psychrometric chart of $CaCl_2$ is plotted and shown in Fig. 3.



327 328

328 329

Fig. 3. Psychometric chart of CaCl₂

330 *3.3 Experiment validation based on analytical solution*

Experimental results are validated by comparing to Zhang and Niu's analytical solution for enthalpy exchanger with membrane cores. According to their research, the sensible effectiveness is a function of two dimensionless parameters, NTU and C_r^* . For unmixed cross flow, the function can be presented as [50]:

335
$$\varepsilon_s = 1 - exp\left[\frac{\exp\left(-NTU^{0.78}C_r^{*-1}\right) - 1}{NTU^{-0.22}C_r^{*-1}}\right]$$
 (20)

336 Similar to sensible effectiveness, the latent effectiveness can be calculated:

337
$$\varepsilon_l = 1 - exp\left\{\frac{NTU_m^{0.22}}{m^{*-1}}\left[exp\left(-m^{*-1}NTU_m^{0.78}\right) - 1\right]\right\}$$
 (21)

338

339 4. Results and Discussion

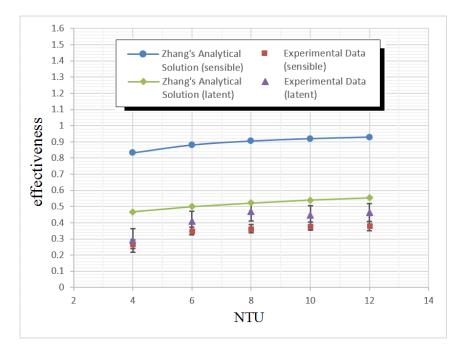
Forty five experimental tests have been conducted to achieve the objectives of this study. Based
on the experimental results, the influences of main operating parameters on the system
performance are analysed.

343 4.1 Effects of dimensionless parameters

344 Two dimensionless parameters, m^* and NTU, are examined to identify their influences on the

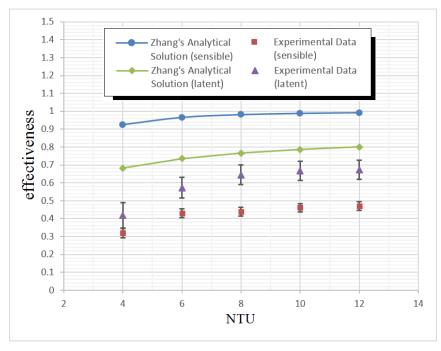
345 dehumidifier performance, experimental results are compared to Zhang's analytical solution

- [50]. The effectiveness experimental and analytical results under $m^* = 0.5$ and 1 are shown in
- 347 Fig.4 and Fig. 5.



348

Fig. 4. Experimental and analytical results of sensible and latent effectiveness under $m^* = 0.5$



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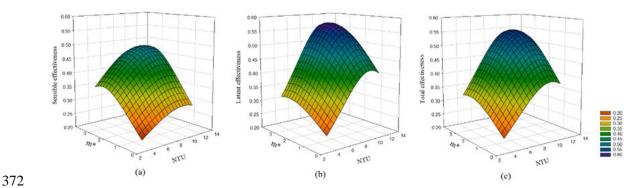
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Fig. 5. Experimental and analytical results of sensible and latent effectiveness under $m^* = 1$

The variation trends of experimental data are similar to that of the analytical results for both sensible and latent effectiveness under $m^* = 0.5$ and 1. However, the sensible effectiveness discrepancies between them are significant, and the analytical results are higher for both m^* . The discrepancies between experimental and analytical results are caused by the following assumptions. Firstly, membrane frosting, membrane fouling, and mal-distribution effects are neglected in the analytical models. Secondly, the inhomogeneous membrane properties, such as thickness and thermal conductivity, are not considered in the analytical models. Last but not least, the laminar flow is assumed for the air stream in the models to calculate convective heat and mass transfer coefficients. However the amount of heat and mass transfer enhancements are not considered, which could be another source of discrepancy between experimental and analytical results.

The variations of the sensible, latent and total effectiveness with m^* and NTU are shown in Fig.6. Comparatively the sensible effectiveness is the lowest one among these three effectiveness at the same m^* and NTU, while the latent effectiveness is the highest one. The maximum values of the sensible, latent and total effectiveness are 0.478, 0.561 and 0.539

- 369 respectively when $m^* = 3.5$ and NTU = 12. Oppositely, the lowest values of these effectiveness
- are 0.167, 0.181, and 0.177 when $m^* = 0.5$ and NTU = 4. The separate effects of m^* and NTU



are addressed in the following sections.

Fig. 6. Variations of effectiveness: (a) sensible effectiveness (b) latent effectiveness and (c)
total effectiveness with *m**and *NTU*

375

376 4.1.1 Effect of mass flow rate ratio m^{*}

The effectiveness under each testing condition can be obtained on the basis of the theories in section 3.1. One example of the effectiveness at NTU = 6 is given in Table 6, the variations of the sensible, latent and total effectiveness with m^* under different NTUs are shown in Figs. 7– 9.

Table 6 381

382 Sensible, latent and total effectiveness at NTU = 6 and $C_{sol} = 39\%$

								501			
m*	Cr*	<i>T_{air,in}</i> (°С)	T _{air,out} (℃)	T _{sol,in} (°C)	€ _{sen}	H^*	$W_{air,in}$ (kg/kg)	$W_{air,out}$ (kg/kg)	W _{sol,in} (kg/kg)	ε _{lat}	€ _{total}
0.5	1.565	30.293	28.710	19.909	0.249	2.874	0.0180	0.0152	0.0061	0.3093	0.2937
1	3.13	30.045	27.788	20.162	0.329	3.050	0.0182	0.0146	0.0062	0.3728	0.3622
1.5	4.695	30.416	27.952	21.178	0.375	3.173	0.0183	0.0145	0.0066	0.3972	0.3919
2	6.26	29.981	27.178	20.564	0.404	3.158	0.0182	0.0138	0.0063	0.4389	0.4305
2.5	7.825	29.652	26.894	20.804	0.425	3.265	0.0180	0.0137	0.0064	0.4446	0.4399
3	9.3	29.468	26.835	20.890	0.423	3.416	0.0182	0.0139	0.0065	0.4361	0.4332
3.5	10.85	29.790	26.701	20.369	0.434	2.981	0.0175	0.0133	0.0063	0.4435	0.4411

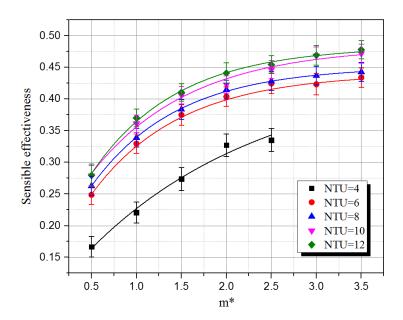


Fig. 7. Sensible effectiveness variations with m^* under different *NTUs*

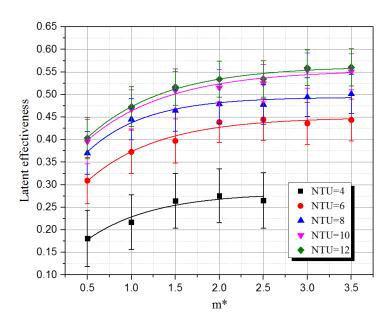
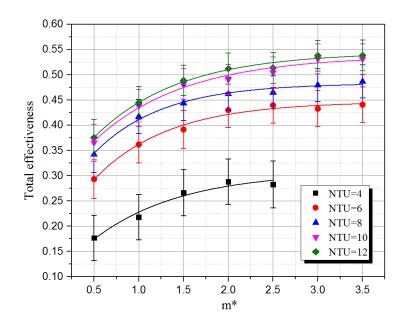




Fig. 8. Latent effectiveness variations with m^* under different NTUs



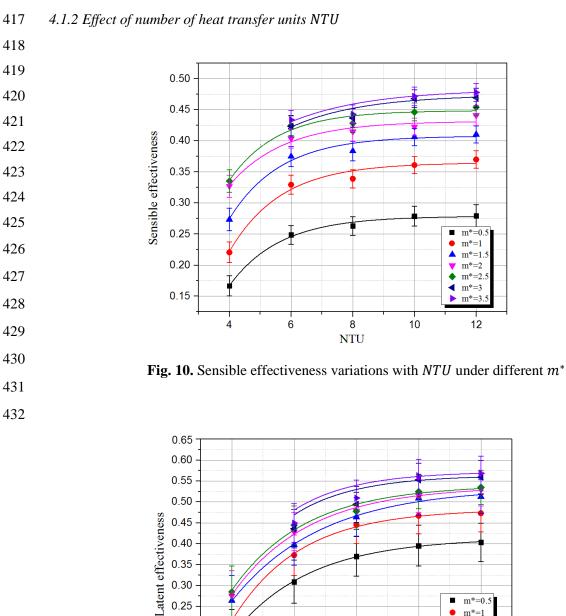


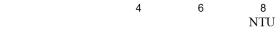
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Fig. 9. Total effectiveness variation with *m*^{*} under different *NTUs*

393 It is evident that the sensible, latent and total effectiveness increase with m^* . For instance, at 394 NTU= 6, the sensible effectiveness increases from 0.249 to 0.434 as m^* changes from 0.5 to 395 3.5. In the meanwhile, the latent and total effectiveness increase from 0.309 to 0.444 and from 396 0.294 to 0.441 respectively. However, the gradients of their increases become less moderate 397 when m^* is in the range of 0.5 to 2, and a slight variation is observed once m^* is over 2. Similar 398 effects of mass flow rate ratio are presented in literature [29], these effectiveness (sensible, 399 latent and total) reach the maximum values and the dehumidification system has the highest efficiency when heat capacitance ratio Cr^* reaches a critical value Cr^*_{crit} (around 6.26). These 400 effectiveness increase with Cr^* and is more sensitive to Cr^* at lower Cr^* [51, 52]. As the heat 401 402 capacitance ratio Cr^* is proportional to the mass flow rate ratio m^* , the effectiveness variations 403 with m^* are similar to that with Cr^* . Therefore a similar critical value of m^* is defined as m^*_{crit} , 404 which is 2 in this study. A similar trend obtained from numerical modelling is found in 405 literatures [24, 25], both the sensible and latent effectiveness increase with m^* when $m^* < 1$, 406 but they are nearly constant when $m^* \ge 1$. So in most cases, it is desirable to maintain the 407 dehumidification system operating at a condition where m^* is equal to m^*_{crit} . It is also worth 408 mentioning that at a low NTU, all effectiveness are very low especially for the latent 409 effectiveness. For instance, at NTU = 4, the latent effectiveness is in the range of 0.181 to 0.265. Thus there is hardly benefit to increase m^* or Cr^* for performance improvement at low NTU. 410 411 On the other hand, the increase rate of the sensible effectiveness is more significant compared 412 with that of the latent effectiveness at the same NTU. For instance, at NTU = 6, the sensible effectiveness increases by 74% when m^* increases from 0.5 to 3.5, while the latent 413

effectiveness only rises by 43%. Similarly, at NTU = 8, the sensible effectiveness increases by 414 415 68% in the same mass flow rate ratio range, whereas the latent effectiveness rises only by 35.4%.





0.30

0.25

0.20

0.15

0.10

0.05

433 434

Fig. 11. Latent effectiveness variations with NTU under different m^*

m*=0.5

10

 $m^{*=1}$ 4

m*=1.5 $m^{*=2}$

m*=2.5 4 • m*=3

m*=3.

12

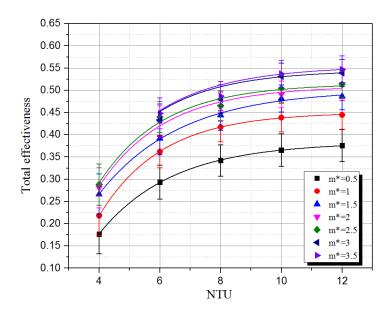
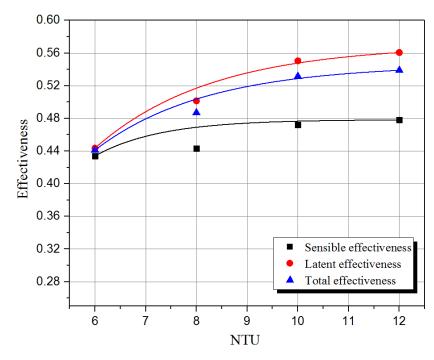






Fig. 12. Total effectiveness variations with NTU under different m^*

The variations of three effectiveness with NTU under different m^* are presented in Figs. 10-12. 439 Compared to the flow rate, the non-dimensional group NTU is a comprehensive indicating 440 441 parameter because it eliminates the impact of channel geometric properties. Significant 442 increases of the sensible, latent and total effectiveness with NTU can be found when NTU is in 443 the range of 4 to 6, and the associated gradients reduce from NTU = 8 and at the end the 444 gradients are becoming negligible. These trends indicate that at a high NTU, the effectiveness 445 improvements are no longer limited by NTU, in other words, increasing NTU will not enhance 446 the system efficiency. Similar to Cr_{crit}^* and m_{crit}^* mentioned previously, a critical NTU exists 447 and is defined as NTU_{crit} . All effectiveness reach the maximum as NTU reaches the critical 448 value NTU_{crit} , which is 8 in this experiment. These results show a good agreement with 449 numerical simulation data of a cross-flow air-to-air enthalpy exchanger with hydrophilic 450 membrane cores in literature [53]. The effectiveness of the enthalpy exchanger increase with 451 NTU when NTU is in the range of 0 to 5 and the increase gradients become moderate when 452 NTU is greater than 5. A similar effectiveness variation trend is indicated in literature [39] as 453 well, both the supply air humidity ratio and temperature decrease as NTU increases from 1 to 454 10. However, less significant effects on the system performance are noted when NTU > 10.





458

Fig. 13. Effectiveness variations with NTU under $m^* = 3.5$

459 The variations of the sensible, latent and total effectiveness with NTU at $m^*=3.5$ are plotted in 460 Fig. 13. The maximum sensible effective is around 0.478 at NTU = 12, while the maximum 461 latent and total effectiveness are approximately 0.561 and 0.539 respectively. At the same m^* , 462 the latent effectiveness is higher than the sensible effectiveness. One reason restricting the 463 sensible effectiveness is high cold water temperature. The sensible effectiveness is limited seriously by the inlet solution temperature, which depends on the cold water temperature in the 464 465 system. Another reason is that the solution cannot be evenly spayed to the membrane surface, 466 especially at high NTU and low m^* , for example when the solution mass flow rate is very low. 467 Therefore, spray nozzle with a larger volumetric spray distribution pattern should be used to 468 improve dehumidification performance. The latent effectiveness is strongly affected by the 469 membrane vapour diffusion resistance, which is related to membrane water permeability. Thus 470 the latent effectiveness can be improved by increasing the membrane water permeability [54]. 471 This can be implemented by utilizing porous membranes or increasing membrane surface area 472 [28, 29]. However, the porous membrane that has lower vapour diffusion resistance may lead 473 to the problem of droplets carryover. Additionally, the bigger size membrane results in higher 474 air pressure drop, and thus more fan power is required. Moreover, the crystallization of the 475 desiccant would considerably affect heat and mass transfer in the dehumidifier by changing the 476 membrane water permeability [28]. As a result, investigations in the optimum membrane 477 vapour diffusivity with considerations of latent effectiveness, carryover and fan power are 478 needed for further research.

480 To sum up, both the sensible and latent effectiveness of dehumidifier reach their maximum 481 values at $m_{crit}^* = 2$ and $NTU_{crit} = 8$, and the gradients of their increases hardly change as m^* is 482 over m_{crit}^* and NTU is over NTU_{crit} . The sensible effectiveness can be improved by utilizing 483 spray nozzle with a larger volumetric spray distribution pattern, and the latent effectiveness can 484 be enhanced by increasing membrane water permeability.

485

486 *4.2 Effects of solution properties*

487 The influences of desiccant solution properties on the system performance are investigated, the 488 main parameters of the solution properties are solution concentration C_{sol} and inlet temperature 489 $T_{sol,in}$. The variations of the sensible, latent and total effectiveness with C_{sol} and $T_{sol,in}$ are 490 presented in Fig. 14, with NTU and m^* set at 8 and 2 respectively. The sensible effectiveness 491 reaches the maximum value of 0.446 when $C_{sol} = 33\%$ and $T_{sol,in} = 18$ °C, and its minimum 492 value is 0.424 when $C_{sol} = 39\%$ and $T_{sol,in} = 23$ °C. By contrast, the maximum latent effectiveness is 0.538 at $C_{sol} = 39\%$ and $T_{sol,in} = 18$ °C, and its minimum value is 0.372 at C_{sol} 493 494 = 33% and $T_{sol,in}$ = 23°C. For the total effectiveness, its maximum value is 0.510 at C_{sol} = 39% 495 and $T_{sol,in} = 18^{\circ}$ C, and the minimum one is 0.389 when $C_{sol} = 33\%$ and $T_{sol,in} = 23^{\circ}$ C. The 496 effects of C_{sol} and $T_{sol,in}$ on the effectiveness are analysed separately in the following sections.

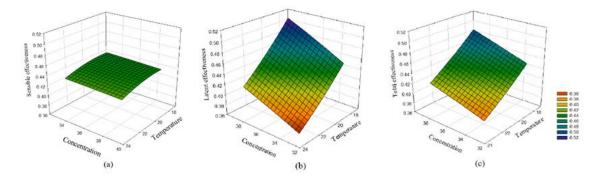






Fig. 14. Variations of effectiveness: (a) sensible effectiveness (b) latent effectiveness and (c) total effectiveness with C_{sol} and $T_{sol,in}$

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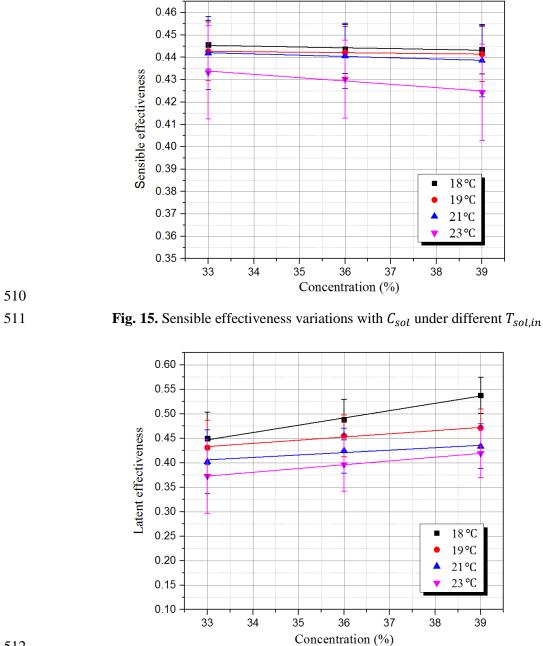




Fig. 16. Latent effectiveness variations with C_{sol} under different $T_{sol,in}$

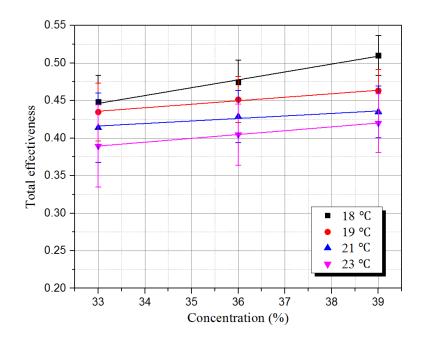




Fig. 17. Total effectiveness variations with C_{sol} under different $T_{sol,in}$

517 The solution concentration has a significant effect on the system performance since it is directly 518 related to the surface vapour pressure. The variations of the effectiveness with C_{sol} are shown 519 in Figs. 15-17, it can be seen that increasing concentration has different impacts on the sensible, 520 latent and total effectiveness. The sensible effectiveness is negatively related to C_{sol} , while the 521 latent and total effectiveness are positively related to C_{sol} . At the inlet solution temperature of 522 21°C, the sensible effectiveness decreases from 0.442 to 0.439 as the solution concentration 523 increases from 33% to 39%, while the latent and total effectiveness increase from 0.402 to 524 0.434 and from 0.414 to 0.435 respectively. The sensible effectiveness is insensitive to the 525 solution concentration as only a slight decrease with the concentration can be seen. This is 526 because the increase of latent effectiveness would lead to more latent heat to be released to the 527 air channel during condensation on the solution side membrane surface. In the meanwhile, the 528 convective heat transfer coefficient on the air side is relatively low, as a result, the sensible 529 effectiveness would be slightly decreased. For instance, at $T_{sol,in} = 18$ °C, the sensible effective 530 decreases by 0.67% when C_{sol} increases from 33% to 39%. Meanwhile, the latent and total 531 effectiveness increase by 19.6% and 13.8% respectively.

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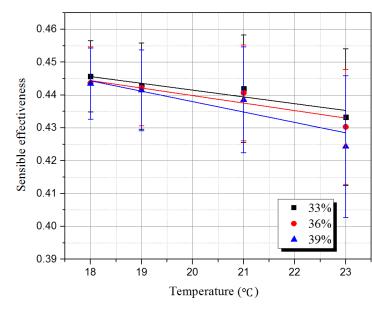


Fig. 18. Sensible effectiveness variations with $T_{sol,in}$ under different C_{sol}

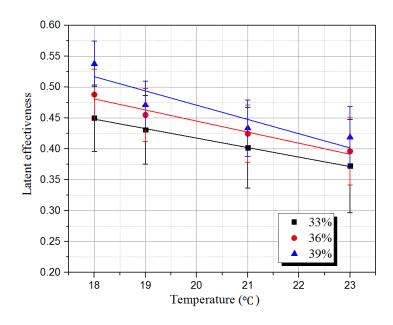


Fig. 19. Latent effectiveness variations with $T_{sol,in}$ under different C_{sol}

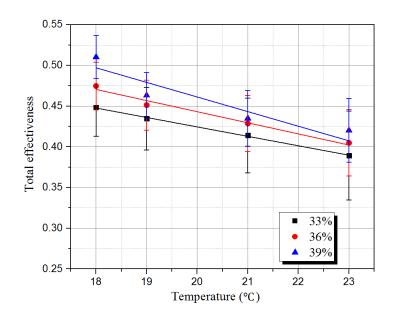




Fig. 20. Total effectiveness variations with $T_{sol,in}$ under different C_{sol}

The solution inlet temperature $T_{sol,in}$ is another key parameter influencing the system performance as it is related to the surface vapour pressure as well. It is clearly reflected in Figs. 18-20 that unlike the impacts of C_{sol} , these effectiveness decrease accordingly with the solution inlet temperature. This is attributed to the reduction of vapour pressure at the solution side. Similar effects are stated in literature [53], the lower solution inlet temperature leads to the lower conditioned air temperature and humidity ratio, which contributes to the higher sensible and latent effectiveness.

The impact of $T_{sol,in}$ on the sensible effectiveness is far less than that on the latent one. This means that the sensible effectiveness is also insensitive to $T_{sol,in}$. Similar to the impact of C_{sol} , this is mainly due to the fact that the increasing of latent effectiveness would contribute to more latent heat to be released to the air channel during the process of condensation. As a result, the sensible effectiveness is weakened. For instance, at $C_{sol} = 33\%$, the sensible effectiveness decreases by 2.9% as the solution inlet temperature increases from 18°C to 23°C, while the latent and total effectiveness reduce by 17.3% and by 13.1% respectively.

It is also found that at different concentrations, the solution temperature has different effects on the effectiveness. The higher the solution concentration, the more significant effect the solution temperature has. As the solution temperature reduces from 23 °C to 18 °C, the sensible effectiveness increases by 4.5% at the solution concentration of 39%, by 3.3% and 3.0% at the solution concentrations of 36% and 33% respectively. In terms of the latent effectiveness, when the solution temperature decreases from 23 °C to 18 °C, the latent effectiveness rises by 28.4% at the solution concentration of 39%, by 23.2% at the concentration of 36% and by 21% at the

- 570 concentration of 33%. Thus decreasing solution temperature would be a more effective way for571 the performance improvement with the high concentration solution.
- 572 To sum up, the system can achieve higher latent effectiveness at lower solution temperature 573 and higher concentration, which is clearly noted in Fig. 14 (b). The solution temperature and 574 concentration have more significant influences on the latent effectiveness compared with the 575 sensible effectiveness. As shown in Fig. 14 (a), the sensible effectiveness hardly varies with the 576 solution temperature and concentration. A similar statement is presented in literature [53], 577 which indicates the sensible effectiveness of a cross-flow membrane-based enthalpy exchanger 578 is not sensitive to its operating conditions. The system dehumidification performance can be 579 improved by increasing the solution concentration and lowering the solution inlet temperature. 580 Comparatively, increasing the solution concentration is preferred in the liquid desiccant system, 581 because more energy is needed to achieve lower solution inlet temperature. However, the use 582 of highly concentrated solution could cause the crystallization problem, which leads to fluid 583 mal-distribution, blockage of the channels, high pumping pressure and membrane fouling. 584 Therefore, the solution properties need to be assessed to avoid crystallization risk [54].
- 585

586 **5. Conclusions**

587 The performance evaluation of a cross-flow membrane-based dehumidification system with 588 CaCl₂ desiccant solution is carried out experimentally in this study. The influences of main 589 operating parameters on dehumidification effectiveness (sensible, latent and total effectiveness) 590 have been assessed, which include number of heat transfer units (*NTU*), solution to air mass 591 flow rate ratio (m^*), solution inlet temperature ($T_{sol,in}$) and concentration (C_{sol}). Following key 592 points can be concluded based on the experimental results:

- The sensible, latent and total effectiveness increase with m* and NTU. However, the
 increase gradients hardly change when m* and NTU are over m^{*}_{crit} and NTU_{crit}
 respectively.
- 596 597
- It is desirable to operate the system at the critical condition where $m_{crit}^* = 2$ and $NTU_{crit} = 8$.
- The sensible effectiveness is the lowest one among the three effectiveness at the same 599 m^* and NTU, while the latent effectiveness is the highest one. The increase rate with 600 NTU in the sensible effectiveness is more significant compared to that in the latent 601 effectiveness.
- The sensible effectiveness can be improved by utilizing spray nozzle with a larger
 volumetric spray distribution pattern, while the latent effectiveness can be increased by
 enhancing membrane water permeability.

- Both the latent and total effectiveness increase with the solution concentration while
 the sensible effectiveness nearly has no variation. All effectiveness can be improved
 by decreasing the solution inlet temperature.
- Increasing solution concentration is a preferable way to improve dehumidification
 efficiency with less energy consumption. However, the operating condition needs to be
 assessed to avoid crystallization risk for high concentrated solution.
- 611 Future research work will be conducted to explore the impacts of various liquid desiccants on
- 612 the system performance.
- 613

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- 617

618 Appendix

619 Absolute uncertainties for dimensionless parameter tests are given in Table A.1.

620 **Table A.1**

621 Absolute uncertainties for dimensionless parameter tests

<i>m</i> *	NTU	δ_{sen}	δ_{lat}	δ_{tot}	m^*	NTU	δ_{sen}	δ_{lat}	δ_{tot}
0.5	4	0.01615	0.06213	0.044883	3	8	0.014702	0.042904	0.03229
1	4	0.016724	0.060976	0.044846	3.5	8	0.014907	0.042307	0.032089
1.5	4	0.018229	0.060499	0.045483	0.5	10	0.015979	0.049164	0.036836
2	4	0.017755	0.059658	0.044874	1	10	0.013815	0.043305	0.031994
2.5	4	0.018229	0.061614	0.046308	1.5	10	0.013669	0.041306	0.030547
0.5	6	0.014997	0.051354	0.038294	2	10	0.0141	0.041523	0.030913
1	6	0.015058	0.048103	0.036417	2.5	10	0.014249	0.041133	0.03077
1.5	6	0.016149	0.04873	0.037254	3	10	0.01429	0.039403	0.029659
2	6	0.015396	0.045433	0.034703	3.5	10	0.015076	0.039804	0.030384
2.5	6	0.016223	0.046097	0.035492	0.5	12	0.017679	0.0459	0.035811
3	6	0.016694	0.046315	0.036027	1	12	0.014121	0.044403	0.03252
3.5	6	0.015134	0.046261	0.03485	1.5	12	0.013709	0.04044	0.030115
0.5	8	0.014984	0.048837	0.03626	2	12	0.016475	0.041759	0.032289
1	8	0.014602	0.045189	0.033609	2.5	12	0.014109	0.040556	0.030278
1.5	8	0.015736	0.045647	0.034502	3	12	0.015603	0.039519	0.030438
2	8	0.014731	0.04392	0.032928	3.5	12	0.014506	0.040724	0.030244
2.5	8	0.01483	0.044008	0.033099					

622 Absolute uncertainties for solution property tests are given in Table A.2.

623 **Table A.2**

624 Absolute uncertainties for solution property tests

625

$T_{sol,in}$ (°C)	C_{sol} (%)	δ_{sen}	δ_{lat}	δ_{tot}
18.142	33	0.01084	0.053881	0.035335
18.058	36	0.010978	0.04145	0.029275
18.103	39	0.010862	0.037063	0.026455
19.498	33	0.013142	0.055662	0.038496
18.903	36	0.011601	0.042919	0.030657
19.396	39	0.012323	0.038406	0.028218
21.395	33	0.016322	0.065111	0.045988
20.975	36	0.014607	0.046138	0.034533
21.190	39	0.016092	0.04561	0.034219
23.220	33	0.020806	0.075518	0.054722
22.999	36	0.017421	0.054272	0.04077
23.388	39	0.02154	0.049277	0.039237

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