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Membrane protein extraction and purification using styrene-maleic acid (SMA) co-polymer: Effect of variations in polymer structure

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Abstract:

The use of styrene maleic acid (SMA) co-polymers to extract and purify transmembrane proteins, whilst retaining their native bilayer environment, overcomes many of the disadvantages associated with conventional detergent based procedures. This approach has huge potential for the future of membrane protein structural and functional studies. In this investigation we have systematically tested a range of commercially available SMA polymers, varying in both the ratio of styrene to maleic acid and in total size, for the ability to extract, purify and stabilise transmembrane proteins. Three different membrane proteins (BmrA, LeuT and ZipA) which vary in size and shape were used. Our results show that several polymers can be used to extract membrane proteins comparably to conventional detergents. A styrene:maleic acid ratio of either 2:1 or 3:1, combined with a relatively small average molecular weight (7.5-10 kDa) is optimal for membrane extraction, and this appears to be independent of the protein size, shape or expression system. A subset of polymers were taken forward for purification, functional and stability tests. Following a one-step affinity purification SMA 2000 was found to be the best choice for yield, purity and function. However the other polymers offer subtle differences in size and sensitivity to divalent cations that may be useful for a variety of downstream applications.

Summary statement:

Several commercially available SMA polymers are capable of extracting and purifying transmembrane proteins. SMA 2000 is the best choice in terms of solubilisation efficiency, purity and yield. However, alternative polymers may allow fine tuning of the method for downstream applications.

Short title: Effect of polymer variation on SMALP solubilisation of membrane proteins

Keywords: membrane protein, SMALP, SMA, solubilisation, purification, polymer, nanodisc, detergent

Introduction:

The study of the structure and function of transmembrane proteins lags significantly behind that of soluble proteins. In large part this is because, in contrast to soluble proteins, there is a need to extract or solubilise these proteins from their lipid bilayer environment. Conventionally this has been achieved using detergents/surfactants which can disrupt the bilayer structure and form micellar structures around the hydrophobic transmembrane regions of the protein. Whilst this approach has yielded many results the use of detergents is not without significant challenges. In the detergent solubilised state membrane proteins tend to have limited stability and often exhibit much reduced activity when compared to native forms. This is the result of the delicate balance that needs to be struck to achieve efficient extraction without denaturing the protein. This denaturation results from the detergent being unable to reproduce the complex physical environment of the membrane in which the protein has evolved to function, and detergents often strip all lipids from the protein.

Recently we and others have shown that a new approach to membrane protein extraction using a styrene maleic acid (SMA) co-polymer can overcome many of these problems [1-3]. The SMA polymer spontaneously inserts into membranes and forms small discs of bilayer surrounded by the polymer, termed SMA lipid particles (SMALPs). Membrane proteins extracted in these discs can be purified by affinity chromatography whilst retaining their annular lipid bilayer environment [4-9]. There are several examples in the literature showing proteins in SMALPs that are functional and more thermostable than detergent-solubilised proteins [4-6, 9]. The SMALP disc structure is well suited for many biophysical and spectroscopic techniques [4, 7, 10, 11], as well as structural studies using electron microscopy [4, 12]. There are of course some limitations. Firstly the size of the SMALPs is typically around 10 nm diameter [1, 5, 13], which can be too small for some large proteins or complexes [14]. Secondly the SMALP structure is sensitive to divalent cations such as Mg^{2+} , which above a certain concentration causes the SMA to precipitate out of solution [3, 4].

Many different SMA co-polymers are commercially available, which can vary in the ratio of styrene:maleic acid and in their total length and size. To date the polymers used to extract membrane proteins are SMA(2:1) [1, 4-7, 9, 12] and SMA(3:1) [8, 10, 11, 15, 16], which have styrene to maleic acid ratios of 2:1 and 3:1, and are relatively small with average molecular weights of 7.5 kDa and 10 kDa respectively. In this study we systematically assessed a selection of SMA co-polymers (Table 1), with varying ratios of styrene:maleic acid, and a range of molecular weights. We aimed to determine whether these polymers were effective for extracting and purifying membrane proteins, and if so whether they could overcome the limitations in size and Mg^{2+} sensitivity of the SMA copolymers currently in use.

Our study assessed the performance of these polymers in the extraction process for three different bacterial membrane proteins of varying size and structure, namely BmrA, LeuT and ZipA. BmrA is a multidrug efflux pump of the ABC (ATP Binding Cassette) superfamily from *Bacillus subtilis* [17]. It forms a homodimer, where each monomer provides 6 transmembrane alpha-helices and a cytosolic nucleotide binding domain. Thus it has a large transmembrane region and a large cytosolic region. LeuT, an amino acid:sodium symporter

of the NSS (neurotransmitter:sodium symporter) family from *Aquifex aeolicus*, comprised of 12 transmembrane helices is located almost entirely within the membrane [18]. Finally ZipA, a membrane tether involved in cell division in *E. coli*, has just a single transmembrane helix with a large cytosolic domain [19].

Experimental procedures:

SMA preparation

SMA 2000, SMA 1000 and SMA 3000 were obtained from Cray Valley (Exton, PA, USA). XZ09008, XZ09006, SZ25010, SZ40005, SZ42010, SZ33030, SZ28065 and SZ28110 were from Polyscope (Geleen, NL). The commercially available polymers are provided as styrene maleic anhydride co-polymer and need to be converted to the styrene maleic acid by hydrolysis in NaOH [20, 21]. A solution of each styrene maleic anhydride copolymer 10% (w/v) in 1M NaOH was refluxed for 2 hours and allowed to cool to room temperature. The polymers were precipitated by the addition of excess concentrated HCl and washed extensively with distilled water. Washed polymer was dissolved in 0.6M NaOH to give a pH of 8, and freeze-dried. Styrene maleic acid copolymer powder was stored at room temperature.

Protein production & membrane preparation

C41 (DE3) *E.coli* cells were transformed with the vector pET23b-BmrA containing the gene for BmrA expressed as a fusion to a C-terminal His₆ tag (kind gift from Prof. Jean-Michel Jault, IBCP, Lyon). BL21 (DE3) *E.coli* cells were transformed with the vector pET101-ZipA with a V5 epitope and C-terminal His₆ tag (kind gift from Dr David Roper, University of Warwick) or with pET16b-LeuT with an N-terminal His₈ Tag (kind gift from Prof. Harald Sitte, Medical University of Vienna).

Small overnight cultures (5 ml) were used to inoculate 1 litre flasks of Luria Broth supplemented with 100 µg/ml ampicillin and grown at 37°C, 200 rpm until OD₆₀₀ reached 0.6. Protein synthesis was induced by the addition of 0.5 mM IPTG and the temperature was reduced to 25°C. Cells were harvested 18-20 hours later by centrifugation (6000 g, 10 min). The *E. coli* cell pellets were re-suspended in buffer 1 (50mM Tris at a pH 7.4, 250 mM sucrose, 0.25 mM CaCl₂) supplemented with protease inhibitors (1 µM pepstatin, 1.3 µM benzamidine, 1.8 µM leupeptin). Cells were disrupted on ice using sonication. Unbroken cells and debris were removed by a low speed spin (650 g, 20 min, 4°C), then membranes harvested by ultracentrifugation (100,000 g 20 min, 4°C). Membranes were resuspended in buffer 2 (20 mM Tris pH8, 150 mM NaCl) at a final concentration of 60 mg/ml wet membrane weight, aliquoted and stored at -80°C.

Solubilisation trials

Initial trials compared SMA 2000 with the conventional detergents octyl-β-D-glucoside (OG) and dodecyl-β-D-maltoside (DDM). Membranes containing each target protein (30 mg/ml wet weight) in buffer 2 were mixed with 2.5% (w/v) SMA 2000, 2% (w/v) OG (Sigma) or 2% (w/v) DDM (VWR) for 1 hour at room temperature, with gentle shaking. Samples were then centrifuged (100,000g, 20 min) and the supernatant containing solubilised protein was retained. The pellet containing insoluble material was resuspended in the same total volume of buffer 2 supplemented with 2% (w/v) SDS. Samples of soluble and insoluble material were analysed by Western blot, probed with an anti-his 1° primary antibody (1:1000, R&D systems). Blots were visualised using a 2° antibody of either anti-mouse Alkaline phosphatase and BCIP/NBT (Sigma) or anti-mouse HRP and SuperSignal West chemiluminescent kit (ThermoFisher). The percentage of total protein solubilised was determined from the Western blots using densitometry (ImageJ).

A similar procedure was followed to screen the different SMA polymer variations. Each polymer was used at a final concentration of 2.5% (w/v), and the amount of protein solubilised was normalised to that obtained with SMA 2000.

Ni-NTA affinity Purification

Solubilised protein was mixed with HisPur Ni²⁺-NTA resin (ThermoFisher) at a ratio of 100 µl resin per ml of solubilised protein, at 4°C overnight with gentle rotation. The sample was transferred to a gravity flow column (Machery-Nagel) and the flow through containing unbound material collected. The resin was washed 5 times with 10 bed volumes (BV) of buffer 2 supplemented with 20 mM imidazole, twice with 10 BV buffer 2 containing 40 mM imidazole and once with 2 BV buffer 2 supplemented with 60 mM imidazole. Proteins were eluted with buffer 2 supplemented with 200 mM imidazole and six fractions of 1 BV were collected. For purifications using DDM, all wash and elution buffers were also supplemented with 0.1 % (w/v) DDM. Fractions were analysed using 7.5% SDS-PAGE and InstantBlue stain (Expedeon). Elution fractions containing the target protein were pooled and stored at 4°C.

Purified protein quantification

The concentration of purified protein samples was determined from SDS-PAGE using BSA as a standard as described previously [22]. Unlike many colorimetric methods, this method does not suffer from interference from lipids, imidazole, or SMA. Briefly, samples (10 µl and 20 µl) of purified proteins were separated using 7.5% SDS-PAGE alongside BSA standards (0.25, 0.5, 0.75, 1, 1.25 µg), and stained with InstantBlue (Expedeon). Intensity of each band was analysed by densitometry (ImageJ), and a standard curve constructed for BSA. Using the standard curve the concentration of purified LeuT/ZipA/BmrA was calculated. From the concentration, the yield of purified protein per litre of culture could be calculated. To estimate the purity of each target protein, 1-2 µg of purified protein sample was loaded on 7.5% SDS-PAGE and stained with InstantBlue. The whole lane was analysed by densitometry, and the intensity of the protein of interest as a percentage of the total staining intensity was determined.

BmrA substrate binding assay

Substrate binding to BmrA was measured using a tryptophan fluorescence quenching assay as described previously [17]. Using centrifugal filter concentrators (Amicon Ultra, 30K cutoff) purified BmrA was concentrated and exchanged into buffer 2 to remove imidazole. Tryptophan fluorescence of BmrA (50 µg/ml) was monitored using a Perkin Elmer LS55 fluorimeter, with an excitation wavelength of 280 nm (slit width 10 nm), and emission measured at 310-400 nm (slit width 20 nm). Fluorescence quenching by successive additions (1 – 50 µM) of Hoechst 33342 or doxorubicin was measured at 335 nm (λ_{max}). Fluorescence intensities were corrected for the effects of dilution and the inner filter effect using N-acetyl tryptophanamide. Results were analysed by non-linear regression using Graphpad Prism (Graphpad Software Inc.) to fit a one-site binding hyperbola.

Thermostability

The aggregation of purified BmrA upon heating was measured using right angle light scattering. BmrA (50 µg/ml) was heated from 25°C - 95°C in increments of 10°C for 10 minutes each. After incubation at each temperature the light scattering was measured using a Perkin Elmer LS55 fluorimeter at a wavelength of 390 nm (slit width 10nm). Thermostability was also monitored by heating samples of purified BmrA (50 µg/ml) at specific temperatures for 10 minutes, followed by centrifugation to remove aggregates (10,000g, 10 min). Soluble protein remaining in the supernatant was analysed by SDS-PAGE.

Mg²⁺ sensitivity assay

SMA solubilised and purified ZipA (50 µg/ml) was mixed with various concentrations of MgCl₂ (0-10 mM) at room temperature for 10 min. Samples were centrifuged (100,000g, 20 min, 4 °C), the supernatant containing soluble protein was harvested, and the pellet containing insoluble material was resuspended in the same volume of buffer 2. Samples of both soluble and insoluble protein were run on SDS-PAGE, stained with InstantBlue, and the percentage of protein remaining in solution was calculated by densitometry.

Dynamic light scattering (DLS)

Purified LeuT was concentrated using Amicon Ultra filter concentrators to 100 µg/ml in 20 mM Tris HCl 150 mM NaCl pH 8 buffer. DLS data were recorded using a Malvern Zetasizer Nano S (633 nm) and 1.0 cm pathlength disposable plastic cuvettes (Brand BMBH, Germany). Measurements were taken at 20°C with 300 s equilibration time. Automated instrument parameters were used. Each measurement was repeated at least 11 times.

Statistical analysis

Statistical analysis was carried out using GraphPad Prism and used an ANOVA for multiple comparisons, with a Tukey post-hoc test; p<0.05 was considered significant.

Results:

Solubilisation of proteins using SMA 2000 and conventional detergents

The initial aim of the study was to determine if SMA 2000, the most widely used SMA polymer to date, was able to effectively solubilise three different membrane proteins: LeuT, ZipA and BmrA. It can be seen in Figure 1 that for each protein approximately 55% of the total was solubilised using 2.5% (w/v) SMA 2000. This was comparable to the solubilisation efficiency observed for each protein using the conventional detergent dodecylmaltoside (DDM). For ZipA a similar solubilisation efficiency was also observed for the shorter chain conventional detergent octyl glucoside (OG), however for LeuT and BmrA the solubilisation with OG was significantly less efficient ($p < 0.05$, ANOVA).

Screening SMA polymer variations for protein solubilisation

Having established that SMA 2000 could successfully solubilise all three proteins, the next step was to screen the new SMA polymers to establish if they were capable of extracting membrane proteins. The new SMA polymers tested vary in size from 5 to 110 kDa, and had ratios of styrene:maleic acid ranging from 3:1 to 1:1 (Table 1). As shown in Figure 2, three of the polymers, SMA 3000, XZ09008 and SZ25010, stand out as being effective at solubilising membrane proteins, giving results that are comparable to SMA 2000. SMA 3000 has been previously used to successfully solubilise several membrane proteins [8, 10, 11, 15, 16], so it is not surprising that it is effective. XZ09008 and SZ25010 have very similar properties to SMA 3000, with a 3:1 styrene:maleic acid ratio and a reported molecular weight of 10kDa. However most of the other polymers were not effective at all. The larger polymers, SZ33030, SZ28065 and SZ28110, with average molecular weights of 30-110 kDa were very poor at extracting proteins, despite having styrene:maleic acid ratios comparable to SMA 2000 and SMA 3000. The small polymers with a high percentage of maleic acid, SMA 1000, SZ40005 XZ09006 and SZ42010, were also not particularly effective at solubilising proteins, despite having similar molecular weights to SMA 2000 and SMA 3000. It is interesting to note that the results are strikingly similar for all three target proteins, despite their differences in size and shape. It appears that the effectiveness of each polymer is not protein specific. The target proteins tested here were all expressed in *E. coli*, however the same pattern of results are also observed for MRP4/ABCC4 expressed in *Sf9* insect cells and MRP1/ABCC1 in H69AR cancer cells (Supplementary Figure 1), showing the effectiveness of each polymer is not expression system-specific. Thus it seems that a styrene:maleic acid ratio of 2:1 or 3:1, and a molecular weight of 7.5-10 kDa is optimal for membrane protein solubilisation.

Purification of membrane proteins

The three polymers able to solubilise the membrane proteins effectively (SMA 3000, XZ09008 and SZ25010) were taken through to protein purification trials, alongside SMA 2000 and the conventional detergent DDM. Also included was the polymer XZ09006, which solubilised almost half as much protein as SMA 2000. Each protein was purified by a single step Ni^{2+} -NTA affinity chromatography procedure as shown in Figure 1A. The concentration and volume of purified protein were used to calculate the yield per litre culture for each protein with each solubilisation agent and the average results are shown in Figure 3B. The average yield using SMA 2000 was 1.2 ± 0.1 mg/l, 1.0 ± 0.1 mg/l and 0.8 ± 0.1 mg/l, for ZipA, BmrA and LeuT respectively. With the conventional detergent DDM, the yields were comparable to SMA 2000. However for each of the other polymers the yield of pure protein

was decreased compared to SMA 2000. XZ09006 gave the lowest yield for each protein. This is not surprising given that XZ09006 solubilised half as much protein as SMA 2000 or DDM. However, SMA 3000, XZ09008 and SZ25010 were able to solubilise almost the same amount of protein as SMA 2000. The decrease in yield observed must be due to a loss at some point during the purification procedure.

The purity of protein samples obtained with this single step affinity procedure was also analysed. It has previously been reported that SMA 2000 typically yields a more pure sample than conventional detergents [4]. This observation was reiterated with all three of the proteins studied here, and is illustrated in Figure 3C, where a sample of BmrA purified using DDM clearly contains many more contaminating proteins than the sample purified using SMA 2000. The average measurements for the degree of purity, measured using densitometric analysis of gels loaded with 1-2 μg total protein, are shown in Figure 3D. The purity of each protein obtained using DDM was significantly lower than when using SMA 2000. Thus despite achieving similar total yields with DDM and SMA 2000, the purity of the SMA 2000 encapsulated proteins was superior. The polymers XZ09008 and XZ09006 also gave samples of significantly lower purity than SMA 2000, however SMA 3000 and SZ25010 showed no significant differences to SMA 2000, despite the lower yields.

Ligand binding

We investigated whether the different polymers tested affected the membrane protein structure and function by measuring ligand binding to BmrA using a tryptophan fluorescence quenching assay [17]. Two substrates were tested: Hoechst 33342, and doxorubicin. The average binding affinity (K_d) and maximal quenching (%) parameters are shown in Table 2. It can be seen that BmrA in DDM micelles or SMA 2000 SMALPs has comparable ligand-binding properties, displaying a greater affinity and larger degree of quenching for Hoechst 33342 than doxorubicin, as has been previously observed [17]. XZ09006 purified BmrA also shows no differences to SMA 2000 or DDM. However for SMA 3000 (Figure 4D), XZ09008 and SZ25010 a significantly lower affinity is observed for Hoechst 33342. BmrA encapsulated within these polymers still appears to bind Hoechst 33342: the same maximal degree of quenching is obtained, but with a lower affinity. Interestingly only the Hoechst 33342 binding is affected, while all samples show comparable affinities and maximal degree of quenching with doxorubicin.

Influence of polymer type on SMALP Size

One important aspect of the SMALP method is the size of the particle that contains the protein. Larger particles could allow proteins with more extensive transmembrane domains to be solubilised (currently the limit seems to be 36 transmembrane helices [12]). Conversely, smaller particles might be of more use in techniques like NMR where rapid tumbling times are important.

To measure the size of the SMALPs produced with the various polymers we used dynamic light scattering (DLS), and the protein LeuT, since this protein is predominantly located within the membrane bilayer. This allowed the disc size to be measured with minimal interference from cytosolic regions of the protein. The results in Figure 4A show that SMA 2000 SMALPs display an average diameter of 8-9 nm, whereas SMA 3000, XZ09008 and SZ25010 give a smaller diameter of approximately 5 nm, despite these polymers having a larger molecular weight than SMA 2000.

Influence of polymer type on stability

A key observation of previous SMALP encapsulations has been an increase in stability of the encapsulated protein. In this experiment we examined whether different SMA types confer different stabilities. The thermostability of purified samples was assessed using a right-angle light scattering assay which was able to measure the thermally induced aggregation of the sample. As shown in Figure 4B, BmrA purified in DDM micelles showed a clear temperature dependent increase in aggregation, however this was not observed when BmrA was purified within SMA 2000 SMALPs. To confirm that the increase in light scattering was due to protein aggregation, samples were also analysed by centrifugation and SDS-PAGE (Supplementary Figure 2). A clear loss of DDM-solubilised BmrA from solution is seen at 65°C whereas this is not the case for BmrA in SMA 2000 SMALPs. Similar light scattering results to those obtained with SMA 2000 were also observed for BmrA within SMA 3000 and XZ09008 polymers. For SZ25010 and XZ09006 there was a slight increase in scattering at the highest temperatures, but this was very small when compared to the DDM sample.

Susceptibility of SMALPs to Mg²⁺

One of the deficiencies of the SMALP is its susceptibility to Mg²⁺. [4]. Magnesium ions are thought to bind to the SMA polymer surrounding a SMALP causing the polymer and the protein it encapsulates to precipitate out of solution. We examined the sensitivity to magnesium (Mg²⁺) of protein-encapsulated by other SMA types. As shown in Figure 4C, for ZipA purified within SMA 2000 SMALPs, a concentration of MgCl₂ below 4 mM is tolerated, but at concentrations above this, a dose-dependent loss of solubility is observed. ZipA within XZ09006 SMALPs displays a similar response to Mg²⁺. However ZipA within SMA 3000, X09008 and SZ25010 SMALPs is even more sensitive to Mg²⁺, with concentrations of 1-2 mM causing complete precipitation.

Discussion:

The recent application of SMA polymers for solubilisation and purification of membrane proteins has the potential to revolutionise the field of membrane protein structural and functional studies, resolving many issues currently posed by the use of conventional detergents. The SMALPs provide a solubilisation method that preserves the lipid bilayer environment of membrane proteins, making it possible to maintain near native structure, function and stability.

In this study we aimed to screen a variety of commercially available SMA polymer variations, to establish which features of the polymer are important for efficient protein extraction, purification and downstream analysis. We used a set of three different membrane proteins, with varying size and shape to reduce protein specific issues.

The first step was to assess the solubilisation efficiency for each protein with conventional detergents compared to SMA 2000. The solubilisation of both LeuT and BmrA was less effective with the shorter chain conventional detergent OG. This is not surprising since it is well established that short chain detergents are less effective for initial solubilisation, but better for crystallography. Notably a published LeuT crystal structures utilised DDM to solubilise the protein, before switching to OG for crystallization [18]. For each of the three proteins in this study, the solubilisation efficiency with the conventional detergent DDM was just over 50%. This is comparable to that reported previously for BmrA with DDM [17]. SMA 2000 solubilised approximately 55% of each protein, thus SMA 2000 is comparable to or better than conventional detergents. This concurs with a previous study of other protein targets [4].

During the solubilisation screening, none of the alternative polymers offered an improvement upon the solubilisation efficiency achieved with SMA 2000, though SMA 3000, XZ09008 and SZ25010 did give results comparable to SMA 2000. These three polymers each have a styrene:maleic acid ratio of 3:1, and an average molecular weight of 10 kDa, compared to SMA 2000, which has a 2:1 ratio of styrene:maleic acid and is a little smaller at 7.5 kDa. It should be considered however that all of the commercially available polymers used here have high polydispersity indexes and the molecular weights detailed in Table 1 are the average of a mixture of different sizes. This results from the methods used to synthesise the polymers, which yield samples containing a distribution of sizes, and also variations in the sequence of styrene and maleic acid groups [3]. We do not currently know if all of the different polymers within each distribution are functional in solubilisation and forming SMALPs or if it is a small subset. However given that more defined preparations are not easily available it was reasonable to test the commercially available polymers. The results showed clearly that a styrene:maleic acid ratio of either 2:1 or 3:1, combined with an average molecular weight of 7.5-10 kDa was required for efficient solubilisation of membrane proteins, and any deviation from this resulted in poor protein extraction.

Ni²⁺-NTA affinity purification of proteins within SMA 2000 SMALPs gave yields that were comparable to those obtained using DDM, however the purity achieved using SMA 2000 was significantly higher than that achieved using DDM, as has been reported previously for other proteins [4]. Surprisingly the yields of protein achieved using SMA 3000, XZ09008 and SZ25010 were lower than with SMA 2000, despite showing comparable solubilisation

efficiencies, suggesting that with these polymers protein was lost during the purification procedure. One possibility is less efficient binding to the Ni²⁺-NTA resin, as it is known that the binding interaction between the SMALP-protein and the Ni²⁺-NTA resin can be of low affinity [3]. An alternative could be that SMALPs formed from these polymers are less stable than SMA 2000 SMALPs, leading to loss of protein from solution during washing steps. It remains to be determined at which stage(s) the protein is lost, however under the standard conditions tested here it would seem that SMA 2000 gives the greatest yield and purity of purified membrane protein.

The size of SMALPs formed using SMA 2000 reported in the literature shows considerable variation [3]. This may be the result of differences in the method used to make the measurement. DLS [1, 7, 23], small angle neutron scattering [13] and electron microscopy [1, 5, 7] have all been used. For some proteins these measurement are complicated by the presence of large cytosolic domains which will contribute to the overall dimensions. To overcome this we used LeuT-SMALPs, where the protein is predominantly located within the bilayer region. Our results using DLS gave an average diameter for LeuT-SMA 2000 of 8-9nm, which is smaller than typical reports of 10-12nm. However it compares well with the diameter of 9 nm reported for PagP-SMALPs in Knowles et al [1], and the small angle neutron scattering studies of lipid-only SMALPs in Jamshad et al [13]. Surprisingly we found that LeuT-SMALPs formed from the larger polymers SMA 3000, XZ09008 and SZ25010 gave smaller diameters of approximately 5 nm. This is in contrast to previous reports where SMA 3000 was used and diameters of 12 nm were reported [10, 15]. Possible explanations for this could include the formation of SMALP-SMALP interactions to form dimers, as has been observed previously [12], methodological and sample preparation differences for DLS and electron microscopy that have previously been shown to give different measurements for the same sample [1, 13], or that the polymer can form different size particles depending on the size of the protein encapsulated. It has also been shown using lipids only that if less SMA is used during solubilisation, larger SMALPs can actually be formed, and the size of the particle formed depends on the ratio of SMA/lipid [24, 25]. In our study we have used an excess of SMA polymer at 2.5% (w/v) with membranes at 30 mg/ml (wet pellet weight). We have shown previously that lower concentrations of SMA, down to approximately 1% (w/v) are still effective for protein solubilisation under these conditions, but below this solubilisation is less efficient [21]. It will be interesting in the future to establish if conditions can be tailored to produce larger SMALPs whilst still extracting sufficient protein.

Interestingly we also observed differences in BmrA ligand binding and ZipA Mg²⁺ sensitivity for the same three polymers, SMA 3000, XZ09008 and SZ25010, compared to SMA 2000. BmrA within SMA2000 SMALPs bound both Hoechst 33342 and doxorubicin in a manner comparable to DDM solubilised BmrA, and gave parameters similar to previous published studies [17]. However, although BmrA within SMA 3000, XZ09008 and SZ25010 polymers was able to bind the substrate Hoechst 33342, a lower binding affinity was observed. This was not the case for the substrate doxorubicin. It is not totally clear why this difference was observed, but BmrA is a multidrug transporter, and possesses multiple distinct drug binding sites. Like with its relative P-glycoprotein/ABCB1, it is likely Hoeschst 33342 and doxorubicin bind to different sites [26]. The different physical properties of the various polymer SMALPs seem to have different effects on the two drug binding sites.

SMALPs formed from SMA 2000 are known to be sensitive to divalent cations such as Mg^{2+} . It is plausible to think that the two carboxyl groups of a maleic acid chelate Mg^{2+} , possibly inducing strain or conformational change in the SMA surrounding a SMALP. If this occurs to too many of the maleic acid groups protruding from a single SMALP it causes the SMA to precipitate. Without the SMA belt surrounding the lipid disc, it becomes unstable and the encapsulated proteins and lipids also precipitate. For ZipA within SMA 2000 SMALPs this occurs at concentrations exceeding 5 mM $MgCl_2$ (Figure 4C). However for ZipA in SMA 3000, XZ09008 and SZ25010 this precipitation occurs at lower concentrations (≤ 1 mM). It is possible this difference is due to the smaller size discs obtained with these polymers, as tighter wrapping around a smaller disc might be perturbed more significantly by chelating Mg^{2+} .

To summarize, this study has shown that several commercially available SMA polymers are capable of extracting and purifying membrane proteins. However, our results suggest that SMA 2000 is the best choice in terms of solubilisation efficiency, purification yield, purity and protein function. None of the other polymers tested were able to overcome the current limitations of SMA 2000, such as sensitivity to divalent cations or limited size. In fact the other polymers were more sensitive to Mg^{2+} , and produced smaller diameter discs. Nonetheless there may be occasions when this would be beneficial. For example if you wanted to analyse the annular lipids associated with a protein, extraction with an SMA that gave a smaller disc would contain fewer total lipids, and only retain those most closely associated with the protein. A structure that is less stable or more easily disrupted may be helpful for reconstituting SMALP-encapsulated proteins into proteoliposomes or other bilayer systems for functional analysis. Alternatively for structural studies, a smaller disc size could be beneficial. For electron microscopy a smaller belt of lipids would mask the protein less, and for crystallization trials fewer lipids and a more easily disrupted structure may be beneficial for forming crystals, whereas for NMR a smaller disc size might be beneficial in giving more rapid tumbling times. Thus, while SMA 2000 remains the first choice for protein purifications, understanding the properties of alternative polymers may allow us to fine tune the method for specific downstream applications.

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References:

1. Knowles, T.J., et al., *Membrane proteins solubilized intact in lipid containing nanoparticles bounded by styrene maleic acid copolymer*. J Am Chem Soc, 2009. **131**(22): p. 7484-5.
2. Lee, S.C., et al., *Encapsulated membrane proteins: A simplified system for molecular simulation*. Biochim Biophys Acta, 2016.
3. Dorr, J.M., et al., *The styrene-maleic acid copolymer: a versatile tool in membrane research*. Eur Biophys J, 2016. **45**(1): p. 3-21.
4. Gulati, S., et al., *Detergent-free purification of ABC (ATP-binding-cassette) transporters*. Biochem J, 2014. **461**(2): p. 269-78.
5. Dorr, J.M., et al., *Detergent-free isolation, characterization, and functional reconstitution of a tetrameric K⁺ channel: the power of native nanodiscs*. Proc Natl Acad Sci U S A, 2014. **111**(52): p. 18607-12.
6. Jamshad, M., et al., *G-protein coupled receptor solubilization and purification for biophysical analysis and functional studies, in the total absence of detergent*. Biosci Rep, 2015. **35**(2).
7. Swainsbury, D.J., et al., *Bacterial reaction centers purified with styrene maleic acid copolymer retain native membrane functional properties and display enhanced stability*. Angew Chem Int Ed Engl, 2014. **53**(44): p. 11803-7.
8. Prabudiansyah, I., et al., *Characterization of the annular lipid shell of the Sec translocon*. Biochim Biophys Acta, 2015. **1848**(10 Pt A): p. 2050-6.
9. Logez, C., et al., *Detergent-free Isolation of Functional G Protein-Coupled Receptors into Nanometric Lipid Particles*. Biochemistry, 2016. **55**(1): p. 38-48.
10. Orwick-Rydmark, M., et al., *Detergent-free incorporation of a seven-transmembrane receptor protein into nanosized bilayer Lipodisq particles for functional and biophysical studies*. Nano Lett, 2012. **12**(9): p. 4687-92.
11. Sahu, I.D., et al., *DEER EPR measurements for membrane protein structures via bifunctional spin labels and lipodisq nanoparticles*. Biochemistry, 2013. **52**(38): p. 6627-32.
12. Postis, V., et al., *The use of SMALPs as a novel membrane protein scaffold for structure study by negative stain electron microscopy*. Biochim Biophys Acta, 2015. **1848**(2): p. 496-501.
13. Jamshad, M., et al., *Structural analysis of a nanoparticle containing a lipid bilayer used for detergent-free extraction of membrane proteins*. Nano Research, 2015. **8**(3): p. 774-789.
14. Bell, A.J., L.K. Frankel, and T.M. Bricker, *High Yield Non-detergent Isolation of Photosystem I-Light-harvesting Chlorophyll II Membranes from Spinach Thylakoids: IMPLICATIONS FOR THE ORGANIZATION OF THE PS I ANTENNAE IN HIGHER PLANTS*. J Biol Chem, 2015. **290**(30): p. 18429-37.
15. Long, A.R., et al., *A detergent-free strategy for the reconstitution of active enzyme complexes from native biological membranes into nanoscale discs*. BMC Biotechnol, 2013. **13**: p. 41.
16. Li, D., et al., *Nano-size uni-lamellar lipodisq improved in situ auto-phosphorylation analysis of E. coli tyrosine kinase using (19)F nuclear magnetic resonance*. Protein Cell, 2015. **6**(3): p. 229-33.
17. Steinfels, E., et al., *Characterization of YvcC (BmrA), a multidrug ABC transporter constitutively expressed in Bacillus subtilis*. Biochemistry, 2004. **43**(23): p. 7491-502.
18. Yamashita, A., et al., *Crystal structure of a bacterial homologue of Na⁺/Cl⁻-dependent neurotransmitter transporters*. Nature, 2005. **437**(7056): p. 215-23.
19. Hale, C.A. and P.A. de Boer, *Direct binding of FtsZ to ZipA, an essential component of the septal ring structure that mediates cell division in E. coli*. Cell, 1997. **88**(2): p. 175-85.
20. Lee, S.C., et al., *A method for detergent-free isolation of membrane proteins in their local lipid environment*. Nat Protoc, 2016. **11**(7): p. 1149-62.
21. Rothnie, A.J., *Detergent-Free Membrane Protein Purification*. Methods Mol Biol, 2016. **1432**: p. 261-7.
22. Rothnie, A., et al., *The topography of transmembrane segment six is altered during the catalytic cycle of P-glycoprotein*. J Biol Chem, 2004. **279**(33): p. 34913-21.

23. Scheidelaar, S., et al., *Molecular model for the solubilization of membranes into nanodisks by styrene maleic Acid copolymers*. Biophys J, 2015. **108**(2): p. 279-90.
24. Cuevas Arenas, R., et al., *Influence of lipid bilayer properties on nanodisc formation mediated by styrene/maleic acid copolymers*. Nanoscale, 2016. **8**(32): p. 15016-26.
25. Vargas, C., et al., *Nanoparticle self-assembly in mixtures of phospholipids with styrene/maleic acid copolymers or fluorinated surfactants*. Nanoscale, 2015. **7**(48): p. 20685-96.
26. Martin, C., et al., *Communication between multiple drug binding sites on P-glycoprotein*. Mol Pharmacol, 2000. **58**(3): p. 624-32.

Table 1. Properties of the polymers. Values for the % maleic acid content, and the mass-average molecular weights for each polymer, as specified by the suppliers.

Polymer name	% maleic acid content	Average molecular weight (kDa)
<u>Cray Valley</u>		
SMA 1000	50	5.5
SMA 2000	33	7.5
SMA 3000	25	9.5
<u>Polyscope</u>		
XZ09006	40	7.5
XZ09008	25	10
SZ40005	42	5
SZ25010	25	10
SZ42010	42	10
SZ33030	33	30
SZ28065	28	65
SZ28110	28	110

Table 2. Binding assay parameters for BmrA. Quenching of the intrinsic tryptophan fluorescence of BmrA (50µg/ml), purified with each polymer or DDM, upon binding of substrates Hoechst 33342 or doxorubicin was measured. Data are mean±sem and n≥3. Data were analysed using an ANOVA, * p<0.05, ** p<0.01 significantly different to SMA 2000.

Polymer/detergent	Hoechst 33342		Doxorubicin	
	K _d (µM)	Maximal quenching (%)	K _d (µM)	Maximal quenching (%)
DDM	5.9 ± 1.4	55 ± 4	16 ± 5	22 ± 4
SMA 2000	4.0 ± 0.8	52 ± 5	16 ± 2	39 ± 5
SMA 3000	16 ± 2 *	49 ± 6	16 ± 5	39 ± 10
XZ09008	18 ± 3**	51 ± 9	12 ± 5	31 ± 8
SZ25010	15 ± 5 *	47 ± 5	21 ± 9	30 ± 10
XZ09006	4.5 ± 1.2	44 ± 5	15 ± 6	29 ± 7

Figure 1. Comparison of SMA2000 with conventional detergents for

extraction/solubilisation efficiency. Membranes (30mg/ml wet weight) from *E.coli* cells overexpressing each protein, were solubilised with either 2.5% (w/v) SMA 2000, 2% (w/v) octyl glucoside (OG) or 2% (w/v) dodecylmaltoside (DDM) in 20 mM Tris pH8, 150 mM NaCl for 1 hour at room temperature, whilst gently shaking. Samples were centrifuged at 100,000g for 20 min at 4°C. The supernatant containing solubilised protein was harvested, and the pellet containing insoluble material was resuspended in an equal volume of 2% (w/v) SDS. Samples of both solubilised and insoluble material were run on a Western blot using an anti-his 1° antibody (1:1000), and the % of total protein solubilised determined using densitometry. Data are mean±sem, n≥3.

Figure 2. Screen of SMA polymer variants for extraction of membrane proteins compared

to SMA 2000. Membranes (30mg/ml wet weight) from *E.coli* cells overexpressing each protein, were incubated with 2.5% (w/v) of each polymer for 1 hour at room temperature, whilst gently shaking. Samples were centrifuged at 100,000g for 20 min at 4°C. The supernatant containing solubilised protein was harvested. Samples were run on a Western blot using an anti-his 1° antibody (1:1000), and the degree of solubilisation relative to SMA2000 determined by densitometry. Data are mean±sem, n≥3.

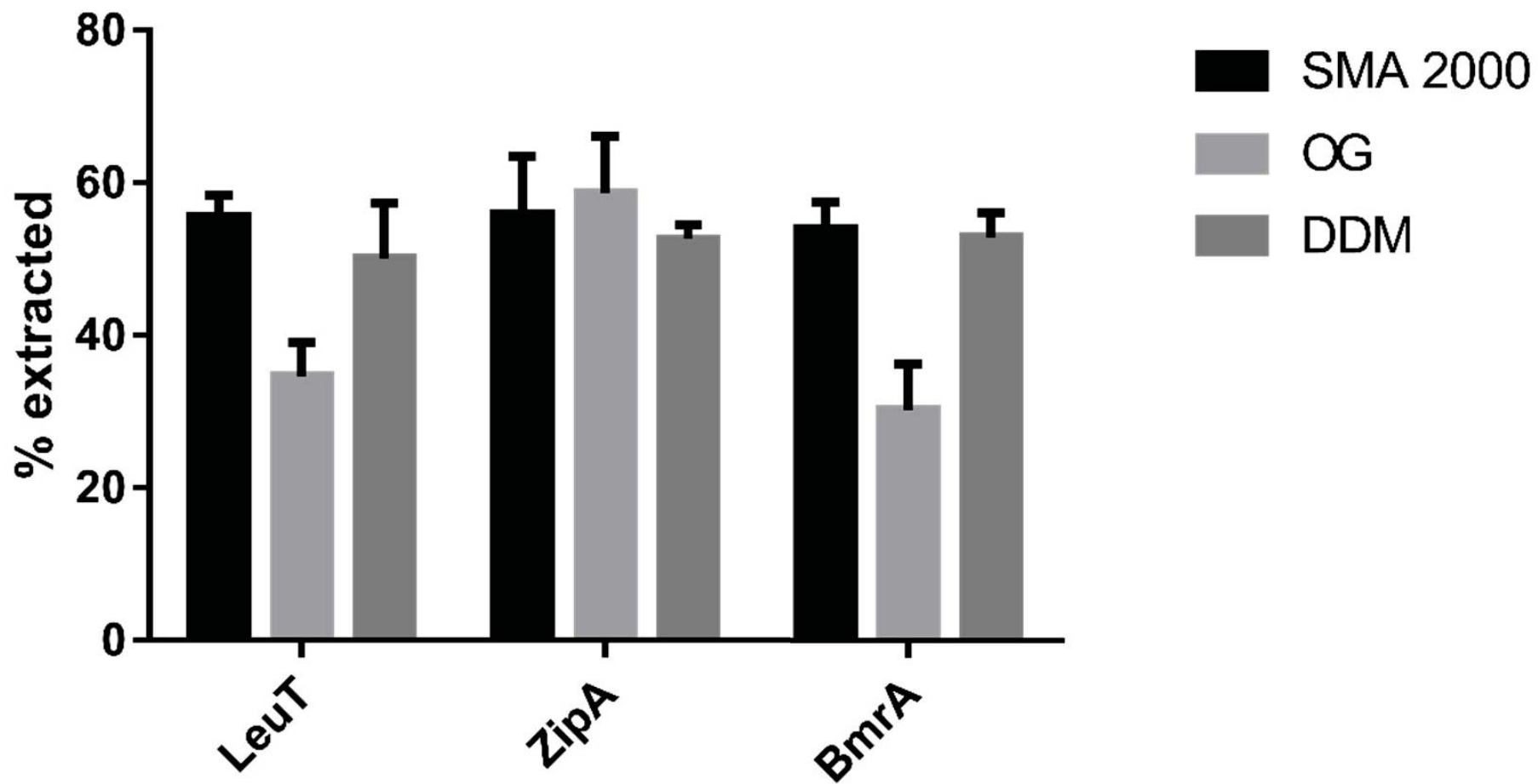
Figure 3. Yield and purity obtained using SMA polymer variants.

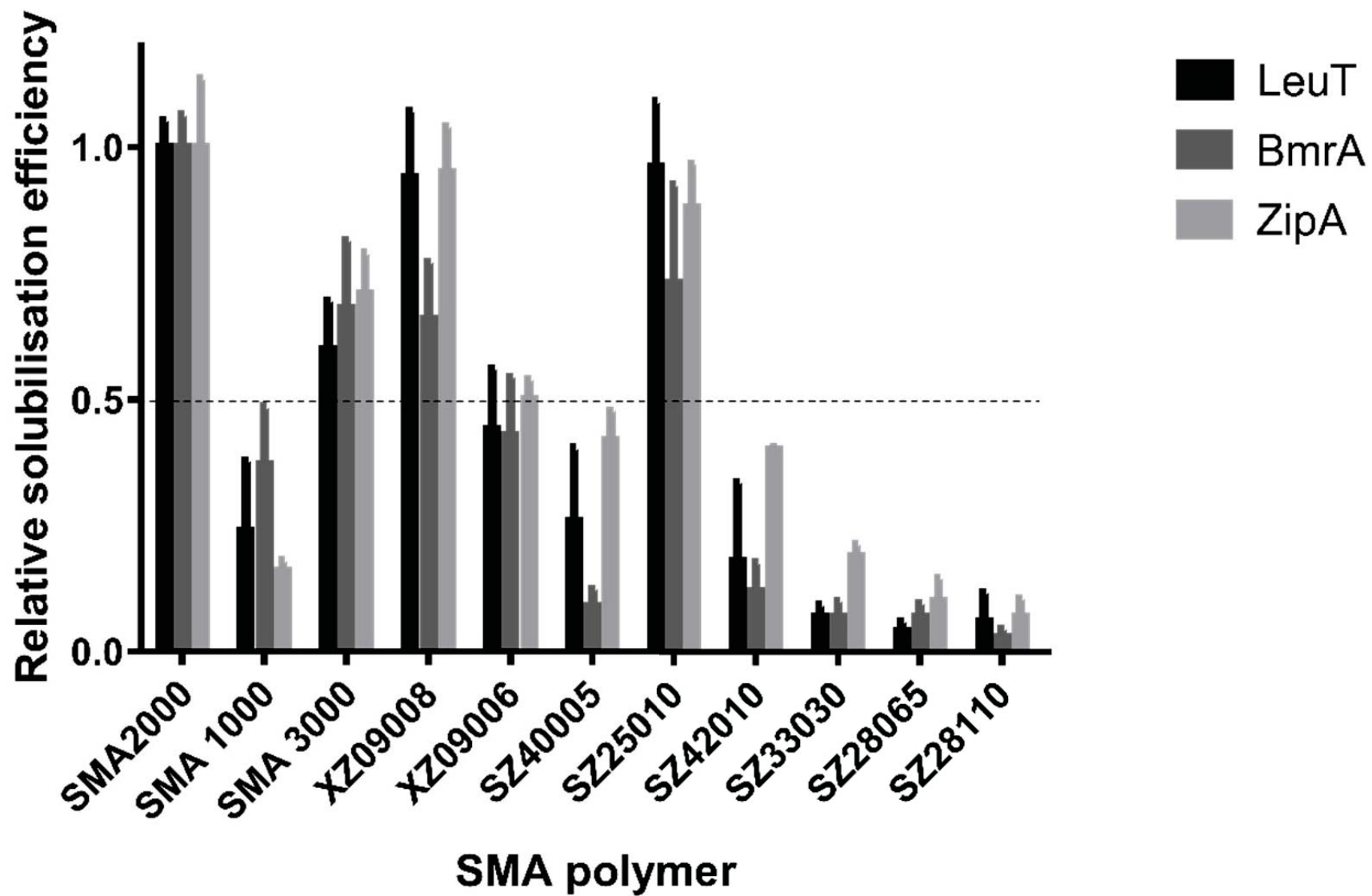
Solubilised proteins were mixed with Ni-NTA resin (100 µl resin per ml solubilised protein) overnight at 4°C, transferred to a gravity flow column and washed extensively with buffer supplemented with 20-60 mM imidazole. Proteins were eluted using buffer supplemented with 200 mM imidazole. A; example purification gel for ZipA solubilised using SZ25010, showing solubilised protein (sol), flowthrough (FT), 20 mM imidazole washes and 200 mM imidazole elution fractions on a 7.5% SDS-PAGE stained with InstantBlue. B; The concentration of purified protein was determined and total yield per L culture calculated. Data are mean±sem, n≥3. C; SDS-PAGE stained with InstantBlue highlighting the difference in purity obtained for BmrA solubilised and purified using SMA 2000 compared to DDM. D; Degree of purity for each protein preparation was determined using densitometry. Data are mean±sem, n≥3. Data were analysed using an ANOVA with a Tukey post-hoc test, * p<0.05, ** p<0.01 yield/purity is significantly lower than that obtained using SMA2000.

Figure 4. Size & stability of SMALPs formed from the different polymers.

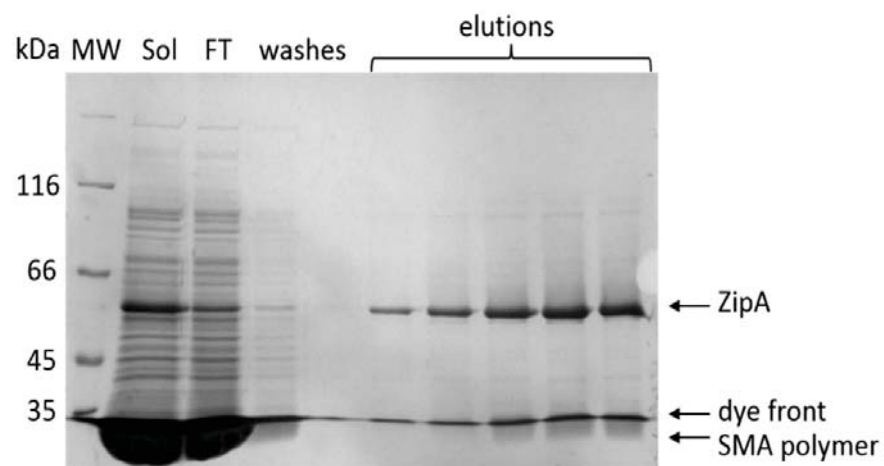
A; LeuT solubilised and purified with each polymer and DDM was analysed using dynamic light scattering. Results are presented as number-weighted particle size distributions. B; Thermal aggregation of BmrA purified using each polymer or DDM was monitored using right angle light scattering at a wavelength of 390nm. C; Magnesium sensitivity of purified ZipA with each polymer was assessed by centrifugation at 100,000g 20 min, and samples of both supernatant and pellet were run on SDS-PAGE and analysed by densitometry. D; Binding of Hoechst 33342 to BmrA extracted and purified with either SMA 2000 (closed circles) or SMA

3000 (open circles) as monitored by tryptophan fluorescence quenching. Data are mean \pm sem, n \geq 3.

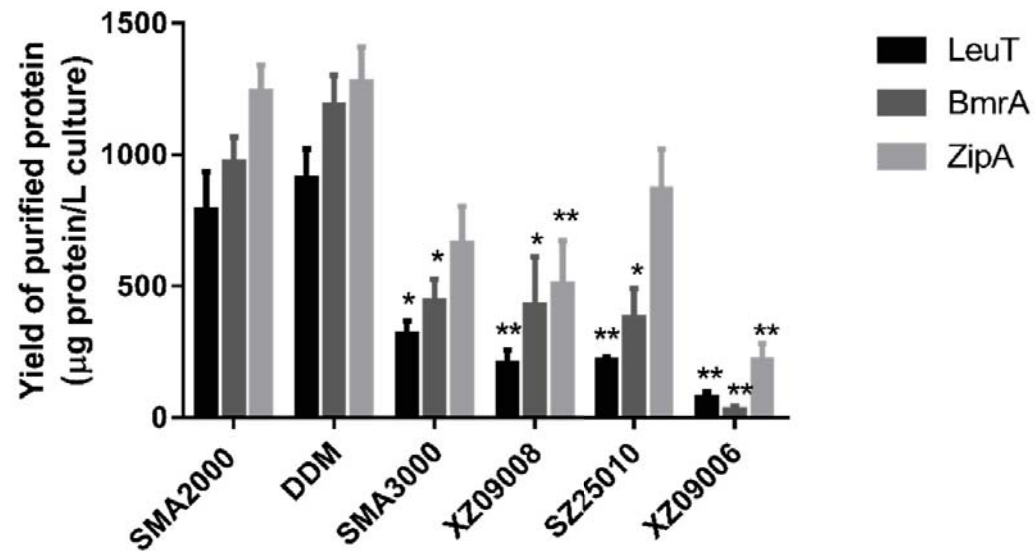




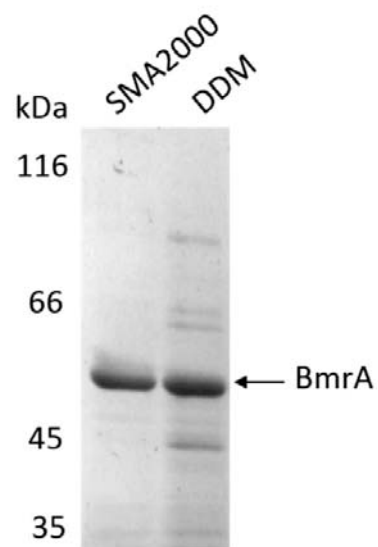
A.



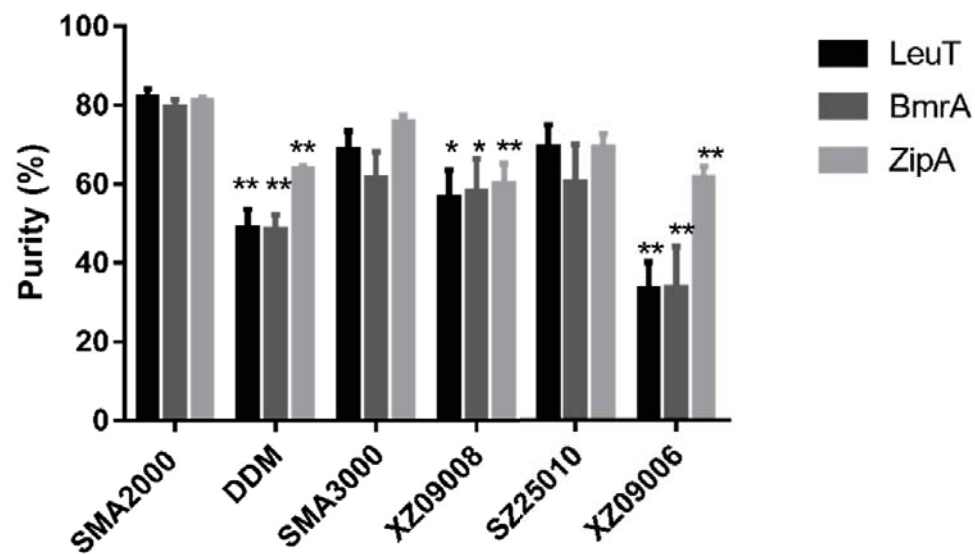
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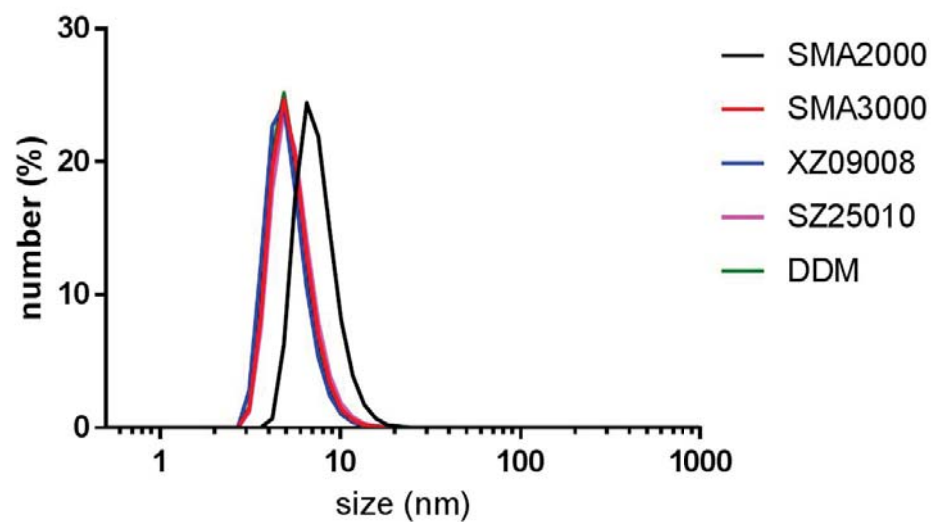
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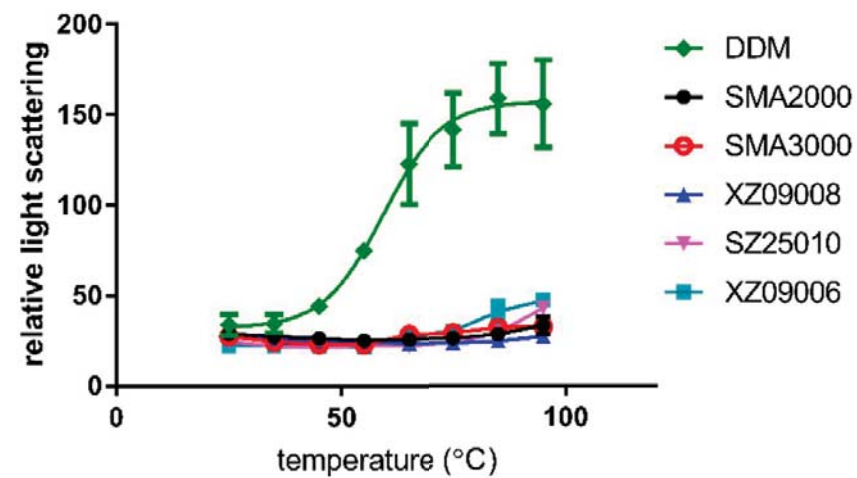
D.



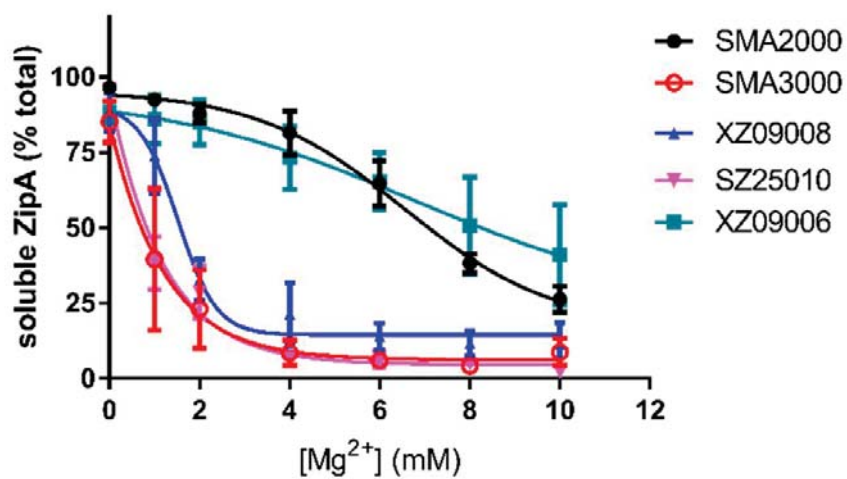
A.



B.



C.



D.

