# Characterisation and surface profiling techniques for composite particles produced by dry powder coating in pharmaceutical drug delivery 

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## Highlights:

- Dry powder coating is a promising one-step process to produce composite particles with improved functionalities.
- Blend characterisation is based on the use of proper sampling techniques
- The selection of the right surface characterisation technique is essential to determine resultant functionality and optimisation of process parameters


#### Abstract

Production of composite particles using dry powder coating is a one-step environmentally-friendly process for the fabrication of particles with targeted properties and favourable functionalities. Diverse functionalities such flowability-enhancement, content-uniformity and dissolution can be developed from dry particle coating. The overarching aim of this review is to enable a holistic understanding of particle functionalities that can be tailored and the selection of relevant characterisation techniques to understand their molecular basis. Key features in powder blend sampling process will be initially addressed, followed by exploring the relevant characterisation techniques within two domains. The first part discusses the functionality delivered by dry coating. The second section will focus on surface profiling that explores the dynamics and surface characteristics of the composite blends.


Key Words: Composite particles, dry-powder coating, guest, host, powder sampling, surface profiling.

Powder coating methods to improve particle properties is increasing in popularity especially in the production of functionalised particle surfaces. Currently employed methods which include; film coating; electrospraying; phase inversion nanoencapsulation and emulsion polymerisation are useful in producing particles with desired properties however require multiple steps and can often result in the chemical alteration of the materials during processing or within the resulting particles [1-4]. Dry particle coating is a one-step high shear process whereby small guest particles are adhered on to the surface of a larger host particle. The process is environmentally friendly as it does not require addition of solvent or a drying process and primarily utilises the attractive forces generated between fine guest particles and host particles (mainly Van der Waals, electrostatic or hydrogen bonding) to improve properties or produce new functionalities, resulting in a cost effective method of producing high quality stable functionalised particles [5].

Dry powder coating or hybrid mixing is a process where fine guest particles are adsorbed on to the surface of host particles under high impaction and shear forces [6-15] and Beach, L., PhD Thesis, New Jersey Institute of Technology, 2011]. The process generates forces of attraction between guest and host particles that are stronger than the weight of an individual fine guest particle resulting in deposition of the fine particles and synthesis of composite particles with improved functionalities or new properties. The forces of attraction range between van der Waals, electrostatic, hydrogen bonding and capillary forces (only generated in blends with some moisture content) [16-20]. Several applications in the pharmaceutical industry have been reported for dry coated particles and include enhancement of flowability and dispersibility of cohesive actives $[10,12,16,21]$, improvement in content uniformity and modified drug dissolution [22-24]. These applications can produce significant advantages for enhancing process and manufacture of pharmaceuticals as well as address clinical challenges encountered in drug delivery [22-24]. A thorough overview of technologies used in dry coating was reported in our recent review [5]. When compared to engineering, the application of dry coating in pharmaceutical industry is underutilised owing to the limitations encountered with the current technologies such as heat generation (as in mechanofusion), particle attrition (as in hybridiser) or contamination (as in magnetically assisted impaction coater) [5, 15, 23, 25-27]. Nevertheless, the
process is promising being a one-step, environmentally friendly, solvent-less process that does not require additional drying or produce by-products $[6,9,10,15,28,29]$. As there is currently little information available on the characterisation of the composite particles produced using dry coating, this review discusses powder sampling techniques which are key in pharmaceutical processing followed by a comprehensive evaluation of relevant characterisation techniques. The present work will highlight powder blend sampling process followed by evaluation of the different functionalities that can be introduced through dry coating. The last section will discuss the different surface profiling techniques to determine the extent and efficiency of surface coating. The overall objective of this article is to inform the reader of the various potential functionalities that can be introduced through the different dry coating processes (completely solvent free as opposed to techniques which utilise dry deposition as a part of the process) and the recent advances in surface characterisation tools to study changes in particle surface properties.

## Powder sampling

## Poor blending or inaccurate sampling?

More than $75 \%$ of the pharmaceutical dosage forms are formulated as tablets or capsules that are processed using powder blends. Therefore, blend homogeneity is vital and reliable sampling and characterization methods are critical processes [30, 31]. Content uniformity in the final dosage form cannot be attained without a uniform mix that does not undergo any rearrangement between blending and compression-filling operations [32,33]. The ultimate purpose of sampling is to collect a specific quantity of the powder that is expected to be representative of the entire powder bed. It entails a high level of accuracy to ensure content uniformity and physical homogeneity (e.g., particle size). Representative sampling encapsulates the process of identifying the appropriate samples with respect to location and time from within the wider powder bed as well as the selection of suitable sampling techniques [30, 34] and P.M Portillo, PhD thesis, The State University of New Jersey, 2008]. The two most important factors in accurate characterisation of the blend are sampling procedures and sampling tools as depicted in Figure 1 [31, 35].

## Sampling technique

Although the best sampling technique is to test the final dosage form for content uniformity, in process testing provides the formulator with the option of identifying any intervention if needed [36, 37]. Sampling locations and the number of samples to be collected are the two main variables in sampling procedure. To develop an effective sampling procedure, proper understanding of the basic dynamics of the blending equipment and the limitations of powder sampling should be initially addressed. Effective sampling requires collection of adequate number of representative samples from the entire mix taking into consideration areas of poor mixing (identified through prior evaluation of blending device) whilst obtaining enough samples to create accurate and reproducible results without the need for over-sampling. Although, greater number of samples usually led to more accurate results, the objective of any sampling scheme is to gather reliable results with the fewest number of samples [36, 38-40].

Prior to commencing sampling, the following needs to be determined as summarised in Figure 1: appropriate blending time, speed range, dead spots in the device as well as the location where segregation could occur particularly in the intermediate bulk containers. To measure the true uniformity of the blend, sample size should be identified and the effect of sample size should be studied [41]. Sample size could vary between 1-10 times the final dosage units and the Food and Drug Administration (FDA) guidance states that justification for sample size needs to be provided when the size exceeds 3 times the final unit dosage form [38]. However, in dry powder coating, the process may need to be tuned as the time to obtain the sample is crucial and dependent on the duration for the formation of composite particles which in turn is dependent on the type of equipment used as well as the processing conditions.

The FDA recommendations for sampling process involve collection of three replicates from each location. A total of 10 locations need to be selected from different zones in the powder bed and assayed. Relative standard deviation (RSD) $\leq 5 \%$ is essential for all the samples tested while individual results need to be within $\pm 10 \%$ of the mean value of the final unit dosage form [38].

Variability within the sampling could be attributed to either the sampling technique, analytical method or the mixture homogeneity [38, 41]. For cohesive powder blends, the variation in content within different regions could be attributed to the presence of agglomerates [38].


Figure 1: Flow chart highlighting the powder sampling process elements. The main components are sampling technique and the sampling device.

The blending uniformity working group (BUWG) from Product Quality Research Institute (PQRI) set recommendations on the use of stratified sampling of blend and dosage units in order to demonstrate the adequacy of mixing within the blends [38]. Based on the report, the FDA established the guidance for industry on stratified in-process sampling. Stratified sampling is a sampling method for dosage units that is collected according to predefined intervals and targeted locations (anticipating the locations with greatest potential to produce non uniform content). Results from these tests are used to monitor manufacturing process (i.e., the selected process for monitoring is the one responsible for causing the greatest variability in the final dosage form) [38, 42]. A recent study has developed an inline NIR (near infra-red) spectroscopy coupled with fibre optic to measure content uniformity [41].

The fibre optic probe measures the intensity of the reflected light from the powder bed surface using photocells and can be used to determine sample homogeneity.

## Sampling devices

A lot of the generally used sampling devices lack accuracy and could result in disruption of the powder bed during sampling as concluded by Muzzio and colleagues [31]. Besides, sampling errors could result from poor flow of the cohesive mixtures into the sampling device. The most accurate and reliable results were produced when the core sampler was used. The core thief sampling device is based on the principle of enveloping a static portion of the powder bed and can be used to obtain sample without disturbing the bed structure. Withdrawal of the sample prevents segregation and further contamination/mixing with the other regions within the powder bed [30, 31]. Sampling tool and technique also vary according to the powder characteristics (cohesive versus granular free flowing). Although thief sampling remains the most commonly used sampling technique to assess powder mixing efficiency, other in-line analytical techniques were introduced that include lightinduced fluorescence, light reflectance, effusivity and NIR spectroscopy [30]. For dry coated particles, sampling would probably follow the requirements of free flowing powders as opposed to cohesive blends as dry coating has been shown to significantly enhance flowability of cohesive powders. However both the duration and the sample location need to be determined depending on the type of the equipment chosen to produce the dry coated particles.

## Characterisation techniques

## Functionality characterisation (characterisation of powder behaviour)

Dry powder coating has been extensively employed to improve various functionalities with a focus on enhancement of flowability, fluidisation, aerosolization and dispersion properties $[11,12,14,16,17$, $23,24,43,44]$. Inclusion of sub-micron fine guest particles on to the surface of micro-sized cohesive host particles resulted in the formation of surface asperities that reduce surface contact between cohesive particles thereby enhancing flowability, dispersibility or fluidisability $[11,17,19,45,46]$. Furthermore, production of homogenous blend using dry coating was reported in multiple studies [14,
$29,47]$. In other applications, research from [15, 29, 48-59] investigated the use of dry coating to enhance the dissolution behaviour of insoluble actives based on the principle of dispersion of agglomerated cohesive particles which in turn maximises the surface area and results in enhanced dissolution. Various different hydophilic polymers have been studied whereby the fine guest particles (drug) were depsoited on to the surface of wettable hostss which in turn produced a synergistic effect in drug dissolution because of the fine drug particles (larger surface area) and the wettability of the host particles. The different techniques to study the charaterisation of the resultant particles have been described in detail in the section covering techniques for dry coating charatetisation.

## Flow properties (flowability, dispersibility and fluidisability)

Owing to the importance of flow behaviour of powders, various characterisation techniques have been developed to describe the flow behaviour that relates to the surface interaction between components of the dry coated blend [60]. Upon dry coating, composite particles with improved flowability, dispersibility, and fluidisability were produced [61]. The following techniques were employed in characterising these properties. Each technique possesses its advantages and limitations as summarised in Table 1.

## Angle of repose (AoR)

AoR is a direct measure of flowability; the test reflects the resistance of material to movement as a result of interparticulate friction. It is one of the official compendial tests (USP <1174>) for flowability measurements. Although the results could vary according to the procedure used, it has its applications in pharmaceutical industry owing to the ease of equipment set up and use. Aor is a measure of the internal angle formed between the cone-like pile of the powder and the horizontal base (Figure 2) using either a static funnel height or fixed base diameter [60]. The diameter and height of the powder cone are measured and angle of repose $(\operatorname{AoR}=\theta)$ is calculated using equation 1 :

$$
\theta=\tan ^{-1} \times \frac{h}{r} \ldots \ldots \ldots \ldots \ldots . . \text { Eq. } 1
$$

Table 1: Summary of flow properties analysis techniques highlighting the main advantages and limitations



Figure 2: Schematic diagram for Angle of Repose (AoR) highlighting the angle formed between the horizontal surface and the cone like powder formed after passing through funnel.

Where (h) is the height of the powder cone and (r) is the radius. Values of AoR exceeding $45^{\circ}$ indicate poor flowability. When the AoR is $\leq 40^{\circ}$, the powder is said to have fair flow properties and inclusion of flow aid is not required. On the other hand, AoR $\leq 35^{\circ}$ but $\geq 31^{\circ}$ is considered as good flowing powder while values $\leq 30^{\circ}$ represent a powder with excellent flowability [60].

## Bulk and tapped densities / Carr Index and Hausner ratio

Bulk and tapped densities provide information on the packing properties of powder. Carr index and Hausner ratio are derived from both bulk and tapped density. Bulk and tapped densities are determined using graduated cylinder technique. A powder is poured into a graduated cylinder without compacting. The weight $(\mathrm{M})$ and the apparent bulk volume $\left(\mathrm{V}_{0}\right)$ are measured and used to calculate the apparent bulk density $\left(\mathrm{P}_{\mathrm{b}}\right)$ using equation 2 :

$$
\mathrm{P}_{b}=M / V_{0} \ldots \ldots \ldots \ldots \ldots \ldots . E q 2
$$

The bulk density gives an indication of the flowability with flowable powders demonstrating high bulk density [65]. Tapped density is the density change due to the mechanical tapping of the cylinder containing the powder mix. Upon tapping of the powder mix, tapped volume $\left(\mathrm{V}_{\mathrm{t}}\right)$ is recorded and tapped density $\left(\mathrm{P}_{\mathrm{t}}\right)$ is calculated form equation 3:

$$
\mathrm{P}_{t}=\frac{M}{V_{t}} \ldots \ldots \ldots \ldots \ldots . E q 3
$$

Carr index and Hausner ratio parameters are interrelated and results from tapped and bulk densities are used to calculate flowability [60, 66]. Carr index (I) is calculated using equation 4 :

$$
I=\frac{\left(\rho_{t}-\rho_{b}\right)}{\left(\rho_{t}\right)} \times 100 \ldots \ldots \ldots \ldots \ldots \ldots \ldots . E q 4
$$

Where $\boldsymbol{\rho}_{\boldsymbol{t}}$ is the tapped density, $\boldsymbol{\rho}_{\boldsymbol{b}}$ is the bulk density. Values less than $15 \%$ indicate powder with good flow characteristics. Whereas, (I) over $25 \%$ represents a powder with poor flowability [60]. Hausner ratio (HR) is an indirect representation of the flow properties that can be calculated from equation 5 :

$$
H R=\frac{P_{t}}{P_{b}} \ldots \ldots \ldots \ldots \ldots \ldots \ldots \ldots . E q 5
$$

Where $\left(\mathrm{P}_{\mathrm{t}}\right)$ is the tapped density and $\left(\mathrm{P}_{\mathrm{b}}\right)$ is the bulk density. USP-29 [77] specifies that a ratio of < 1.25 represents a powder with good flow properties. Ratio $>1.34$ is an example of a powder with poor flow properties. It is worth mentioning that angle of repose, Hausner ratio and Carr index are not intrinsic properties of the powder and are related to the procedure used to test them [60].

Research from [43] investigated the impact of dry coating particles to determine changes on their flowability. Three model cohesive drugs including salbutamol, triamcinolone and salmetrol were studied for flowability changes. All the three drugs had low bulk density possibly due to agglomeration of the fine particles. However, upon dry coating, the results showed that there was an increase of $50-100 \%$ in bulk density. The particles from flowable powder (upon dry coating) did not show a substantial change in tapped density when compared to bulk density as would be expected from a free flowing powder.

No single test can provide a comprehensive view of the flowability of powders [Beach, L., PhD Thesis, New Jersey Institute of Technology, 2011]. Overall the above tests provide an insight of the interparticulate adhesiveness [11, 67]. Fluidisability of the powder bed is usually measured using aeratibility using powder rheometer [68].

## Aeration test

Aeration test is used to measure changes in flowability/ fluidisability of the powder upon aeration. The procedure involves placing a vessel over a perforated base that is connected to air inlet with conditioning blade (Figure 3A). Increasing air pressure is introduced into the system coupled with measuring the force required to move the blade (aeration energy). The higher the aeration energy needed for fluidisation, the poorer the flow properties of the powder (Figure 3B). The ratio between aeration energy to air velocity is termed the basic flowability energy (BFE), and the aeration energy at air velocity of $10 \mathrm{~mm} / \mathrm{s}$ is known as aeration ratio (AR) [24]. Zhou et al. [24] used the test to determine the enhancement of fluidisation behaviour of dry coated micronized lactose with magnesium stearate as guest particles.


Figure 3: Schematic diagram showing the aeration test (A) where air is introduced from the bottom of the perforated chamber containing the tested powder while testing the aeration energy (B) diagram showing the difference in aeration energy between cohesive and non-cohesive powders.

## Vibrated packing density

Similar to aeration test, this technique measures bulk density for cohesive powders. The test has been employed to identify the improvement in powder flowability upon dry coating process. The process is
built on initial fluidisation of the powder bed followed by allowing adequate time to settle. Ultrasound is used to measure the volume of a predetermined weight of powder to assess the solid fraction. Flowable powders tend to have higher solid fraction than less flowable counterparts [Beach, L., PhD Thesis, New Jersey Institute of Technology, 2011].

## Shear Cell test

The USP <1174> describes shear cell method for providing a thorough and precise assessment of the powder flow behaviour [60]. There are various methods employed to measure the shear test and the fundamental principle is based on powder mechanics. The uniaxial compression test is executed when a cylinder is filled with the sample powder followed by application of vertical pressure on to the sample (consolidation stress $\left(\sigma_{1}\right)$ ) as depicted in Figure 4. Upon the presentation of consolidation load, the bulk density and powder strength increases. Then the cylinder is removed leaving the consolidated powder. This is followed by the application of increasing vertical compression strength on the sample until the splitting of sample bed. The stress responsible for breaking the consolidated powder is called the unconfined yield stress $\left(\sigma_{c}\right)$ [69] and see: http://nordicrheologysociety.org/files/2013/10/16-Schulze-Shear-Testing-of-Powders-for-Process-Optimization.pdf.]. Both bulk density and yield stress increase with the increase in consolidation stress $\left(\sigma_{1}\right)$. The value representing $\sigma_{1} / \sigma_{c}$ provides flow function coefficient (FFC). As the strength of the powder needs to be overcome to initiate flow, the higher the value of FFC the better the flowability of the material [see: http://nordicrheologysociety.org/files/2013/10/16-Schulze-Shear-Testing-of-Powders-for-ProcessOptimization.pdf.]. Values of FFC range from 1-10, where a value <4 represents a cohesive powder whereas FFC exceeding 10 is an indication of free flowing powder $[14,61,68]$ and Beach, $\mathrm{L} ., \mathrm{PhD}$ Thesis, New Jersey Institute of Technology, 2011].


Figure 4: Schematic diagram showing the process of shear test. Consolidated shear stress is applied to a powder bed followed by vertical compression to the consolidated powder till yield stress is obtained to produce unconsolidated powder.

Yield limit is a measure of the shear stress needed to initiate powder flow. Shear test is carried out as described above where the consolidated powder bed is split using an increasing normal stress. Shear stress is measured and plotted against the normal stress [see: http://nordicrheologysociety.org/files/2013/10/16-Schulze-Shear-Testing-of-Powders-for-ProcessOptimization.pdf.]. Cohesion is calculated from the graph when the shear stress is zero (the higher the cohesion value the higher the interparticulate forces). A study by [43] using shear test revealed that a reduction in cohesive forces upon dry coating was obtained as demonstrated by a higher shear stress value at each normal stress point for uncoated compared with dry coated particles as a result of reduced interparticulate forces.

The difference between AoR and shear testing is that AoR is carried out without subjecting the powder bed to any consolidation (zero external consolidation force) unlike the shear test where the test is carried out after initial consolidation of the powder bed [60, 61].

## Particle size analysis/ aerosizer

The use of particle size analyser (aerosizer) to determine the dispersibility of composite particles upon dry coating has been reported in various studies [20, 43, 61, 68]. Studying the change in particle size distribution as a function of dispersion pressure using particle size analyser with laser diffraction was employed to determine the extent of agglomeration and de-agglomeration behaviour of the powder blend after dry coating $[61,68]$. The extent of dispersibility was calculated from the gradient obtained upon plotting pressure titration curve (plotting the D90 (maximum size for $90 \%$ of the sample) value
versus dispersion pressure) with low values indicating good dispersibility [64]. It was also noted that for dry coated particles the particle size ( D 90 value) did not reduce upon increase of dispersion pressure whereas uncoated particles showed a drastic reduction in particle size upon the increase in dispersion pressure (therefore producing higher gradient value). This difference was attributed mainly to the breakdown of agglomerates of the uncoated particles upon increase of dispersion pressure. These results provide an insight of dispersibility of the material as reduced/lowering of dispersion pressure translates into higher dispersibility underpinned by reduced cohesivity which enhances flowability. Besides, use of particle size analysis also provides vital quantitative data on the degree of particle attrition which is one of the important factors which requires optimisation in dry coating [43, 61]. Real time aerosolisation characterisation of dry coated powders using laser diffraction systems has also been used to investigate optimum coating percentage of magnesium stearate guest particles over salbutamol sulphate host particles. $2 \%$ magnesium stearate was calculated as the optimum percentage for dry coating that resulted in de-agglomeration of cohesive agglomerates causing improved aerosolisation [70].

## Near infrared spectroscopy (NIR)

NIR characterisation technique was introduced as a means to quantify flowability of powders. It is an in-line process based on capturing the noise from powder flowing from a funnel with the principle that less noise is produced from free flowing material. A flow intensity index is used to evaluate the consistency of powder flow (it is the inverse of the noise spectra). Consistency of powder flow was determined upon plotting the flow intensity index versus time with an aim to understand the flow properties [64] and Beach, L., PhD Thesis, New Jersey Institute of Technology, 2011].

## Porosity

The measurement of porosity using gas displacement technique for dry coated powder bed was reported to provide an indication of the packing properties [68]. Results from [68] demonstrated that dry coated blends have lower porosity compared with physically mixed particles. This difference was attributed to the increase in packing of particles upon the reduction of interparticulate adhesive forces
producing more compact particles. Additionally, smaller fine particles get lodged into the pores of the host particles resulting in reduction in porosity.

## Content uniformity and dissolution studies

The determination of blend homogeneity for the dry coated composite particles is carried out according to the compendial requirement for actives and excipients [71, 72]. Furthermore, dissolution studies that relate to modification of drug release for model compounds upon dry coating were studied with a view to developing functionalised particles with modified release rate [19]. The methods described in compendia (USP methods using either apparatus I or II) were the most utilised techniques to study the changes in release behaviour of various functionalised particle formulations [71].

## Crystallinity

During dry coating, particles are exposed to high degree of compression and shear forces. Some devices produce particle attrition (e.g., high force devices like hybridizer, mechanofusion or Fluid energy mill) that might result in the formation of an amorphous form or induce polymorphic transformations of the host or guest particles [22, 29, 73]. In their research Höckerfelt and Nystorm [74] demonstrated that crystalline material subjected to mechanical force (without particle attrition or micronisation) can transform into an amorphous state which can impact on the properties of the resultant particles. The change in material crystallinity will influence its solubility, bioavailability as well as stability $[22,73]$. Ishizaka $[6,29]$ used X-ray powder diffraction analysis (XRD) to examine crystallinity of processed material (particularly APIs). The results demonstrated a change of the API (oxyphenbutazone) from crystalline to amorphous state upon dry powder coating which further upon storage over a seven month period produced variable results (degree of crystallinity of oxyphenbutazone was reduced to $12.8 \%$ after 4 days of dry coating and was increased to $43.3 \%$ after 7 months). Furthermore, Raman spectroscopy as well as differential scanning calorimeter (DSC) were utilised to study and examine crystallinity. Using DSC, the degree of crystallinity was calculated from the ratio between heat of fusion of micronized particle and the heat of fusion of original particles [22].

## Wettability

Production of dry coated composite particles with either enhanced hydrophilicity or hydrophobicity has been reported [23]. To verify the change in hydrophilic nature, wettability test was carried out using the rate penetration method. A reduction in the amount of water absorbed was noted when cornstarch $(15 \mu \mathrm{~m})$ host particles were coated with $1 \%$ silica $(0.3 \mu \mathrm{~m})$ using MAIC (Magnetically assisted impaction coating) dry coating. Untreated corn-starch particles absorbed almost $60 \%$ of their total weight of water, whereas dry coating dropped it to $18 \%$ of its weight [23].

## Particle dynamics and Surface profiling characterisation for dry coating

The movement of particles within the powder bed during mixing and dry coating process plays a fundamental role in the dynamics of mixing. Therefore, developing knowledge of the particle dynamics can enable the understanding of the dry coating mechanism and therefore, helps in enhancing the design of the dry coating device [75]. Besides, characterisation of a thin coat of the guest particles over the host is challenging due to the need for high resolution analytical techniques that can distinguish changes in surface properties [76]. Table 2 summarises the techniques that have been studied in characterisation and understanding surface coating process in dry coating.

## Positron Emission Particle Tracking (PEPT)

Positron Emission Particle Tracking (PEPT) has been employed to identify the mobility of dry coated versus uncoated blends [Beach, L., PhD Thesis, New Jersey Institute of Technology, 2011]. PEPT offers thorough quantitative information on the internal flow patterns and the dynamics of the dry coating process [41]. An increase in powder flowability is related to the increase in its mobility. The test is based on the principle of tracing the movement of a single radioactive moiety as a function of time. Emission of gamma radiation from the radioactive candidate during its movement within the device can be used to track particle flow with a resolution of one tenth of a mm [41] and Beach, L., PhD Thesis, New Jersey Institute of Technology, 2011]..

Similarly radioactive particle tracking (RPT) is another technique that measures the movement of particles in circulating fluidised beds. It uses Sc46 (Scandium) as the source for gamma-ray emissions but consists of different detector system when compared to PEPT [75].

## X-ray Photoelectron Spectroscopy (XPS)

XPS is a characterisation technique that maps particle topography ( $2-10 \mathrm{~nm}$ depth) and provides specific chemical information on coating quality of guest particles. It not only reveals the elemental composition present on the surface but also provides vital information on particle bonding. The principle is based on an x-ray beam bombarding the surface of a sample and exciting electrons that escape from the atom as portrayed in figure 5 . These events are captured by electron energy analyser that produces an energy spectrum [77]. Research from [76, 78] investigated dry coating of lactose and ibuprofen particles with magnesium stearate using XPS. The elemental composition of the untreated samples did not show the presence of magnesium on the surface of lactose particles, while an increasing amount of magnesium was detected with increasing the concentration of magnesium stearate with a concomitant reduction in the intensity of oxygen atoms that represented the surface of lactose particles. Additionally, the intensity for $\mathrm{C}-\mathrm{C}$ bond (characterising magnesium stearate) increased with the increase in coverage while that of $\mathrm{C}-\mathrm{O}$ bond (characterising Lactose host) was reduced. It is worth mentioning that the accuracy of the


Figure 5: Schematic diagram showing the XPS process, whereby the sample surface is bombarded with an x-ray beam resulting in the excitation of electrons from the atom. The electron energies are detected and converted to an XPS spectrum for the sample.
technique is limited since the guest particles occupy up to 2 nm of the guest surface whereas the surface mapping penetrates up to 10 nm of the particle surface. Nevertheless, this technique can be used to distinguish changes in surface morphology upon dry coating and provides a reasonable quantitative estimate of surface loading of the guest particles [77, 78].

Time- of -flight secondary ion mass spectrometry (TOF-SIMS)
TOF-SIMS is an analytical method that is used for surface profiling and provides chemical information pertinent to elemental, isotopic and molecular structure of the first 1-2 monolayers of small particles [76]. The basic principle of TOF-SIMS is based on energy rich ion beam bombarding the primary targeted atom that generates a secondary ion on the surface of the sample which in turn is detected by a mass spectrometer [79]. TOF-SIMS signal mapping for both lactose (host) and functionalised lactose (with magnesium stearate) upon dry coating was investigated by [76]. Initial experiments focused on extracting individual mass spectra for both the host as well as the guest
particles (for e.g., $\mathrm{C}_{6} \mathrm{H}_{11} \mathrm{O}_{5}$ ring to represent lactose and magnesium signals to represent magnesium stearate). Uncoated lactose particle signal map revealed the ring $\left(\mathrm{C}_{6} \mathrm{H}_{11} \mathrm{O}_{5}\right)$ for lactose, however, when magnesium stearate was adsorbed to the surface of lactose by dry coating, the signal map showed an increase in the density of magnesium signal with increase in the concentration of magnesium stearate (with a concurrent reduction in the density of signal for $\mathrm{C}_{6} \mathrm{H}_{11} \mathrm{O}_{5}$ ). One of the key outcomes from these studies was the application of TOF-SIMS to quantify the percentage of guest material that provides optimal surface coverage thereby significantly influencing the resultant functionality of the dry coated particles [76]. Additionally, TOF-SIMS was investigated to profile surface coating of ibuprofen host particles with magnesium stearate that was used to improve lubrication, compressibility and flowability of cohesive ibuprofen [78].

## Scanning Electron Microscopy (SEM)

SEM can be used to qualitatively determine particle morphology as well as the quality and extent of dry coating. It enables the investigation of the changes in particle size and shape which provides vital information on the suitability of processing technology (particle attrition - major or minor) as well as quality of the final finished product. SEM has been extensively utilised in multiple studies to evaluate the morphology of dry coated particles $[19,24,29,43,61,80]$. SEM coupled with energy dispersive X-ray analysis (EDX) has been used for qualitative evaluation of coating efficiency [19, 61, 68]. The work from [61] using SEM to demonstrate the change in cohesiveness of powder upon dry coating revealed that the coating of fine particles over the host particles reduces particle agglomeration. Elemental mapping using SEM-EDX confirmed the deposition of fine leucine particles over potassium chloride host particles which were used as the basis for elucidating the enhancement of flowability of the resultant dry coated particles.

## Fourier transform infrared spectroscopy (FTIR)

FTIR spectrum usually represents the fingerprints of materials where absorption peaks represent the frequency of vibrations between the bonds within the molecule. Each spectrum is formed when the energy absorbed from a particular frequency of infrared radiation leads to the excitation of a specific
bond within the molecule to a higher state of vibration by stretching or bending. Specific bonds can easily be identified at different spectral regions and the peak size in each spectrum is a direct indication of the amount of material available within the sample [81, 82] [29, 83]. Pfeffer and coworker [23] used FTIR spectrum to identify the chemical reaction upon dry coating using MAIC where model host (corn-starch or cellulose) and guest (silica) particles upon dry coating resulted in the reduction of the intensity of FTIR spectra of OH group. This reduction in surface OH group intensity was attributed to the reaction between acidic silanol group $(-\mathrm{Si}(\mathrm{OH})-$ ) on the surface of silica and the neutral hydroxyl groups on corn-starch/cellulose surfaces forming hydrophobic (-O-) groups with a release of water molecule.

## Specific surface area

Determination of specific surface area of powder is carried out according to USP <846> using gas (nitrogen) adsorption followed by quantitative analysis of the amount of adsorbed gas using BET theory (Brunauer, Emmett and Teller) [60]. Dry coating of particles results in changes in particle surface characteristics which in turn influences the total surface area [76]. A study evaluating the reduction in specific surface area upon mechanofusion of lactose host particles with ( $0.5 \%$ and $1 \%$ ) magnesium stearate was attributed to the reduction in surface asperities (irregularities) of lactose where magnesium stearate (guest) fills the gap producing a smoother surface for the resultant composite particles (also supported by SEM images). The reduction in specific surface area varied according to the percentage of added magnesium stearate (MS); untreated ( $1.079 \mathrm{~m}^{2} / \mathrm{g}$ ), at $0.5 \% \mathrm{MS}$ ( 0.826 ); $1 \%$ MS, ( 0.77 ) and at $5 \%$ MS ( 0.913 ). This technique also enabled the determination of the optimal percentage of MS to produce complete coverage. An increase in specific surface area with 5\% MS indicated that the amount of MS was greater than that was needed for complete surface coverage of the host [76].

## Surface free energy

The total surface energy is the sum of dispersive and specific surface energies with the former related to van der Waals forces that are universally to all particles surfaces. The specific surface energy on
the other hand relates to the interactions including polar, acid-base, hydrogen or ionic bonding. Inverse Gas Chromatography (IGC) was used to study the changes in surface energy of dry coated particles [11, 80]. It enabled the evaluation of changes in surface energy upon dry coating and its resultant effect on particle functionality such as enhancement of flowability [80]. The method developed by [80] using IGC was used to compare dry coated lactose (with magnesium stearate) using mechanofusion with regular mixing in turbula blender. Hepatane, octane and nonane were used to determine the non-polar (dispersive) surface energy whereas dichlormethane and ehtylacetate were used as solvents to evaluate the polar (specific) surface energy. The results revealed that the total surface energy of lactose with $5 \%$ magnesium stearate was significantly lower following dry coating compared with conventional mixing which provided the evidence for the enhancement of flowability and dispersibility of dry coated mixture compared to conventionally mixed powder blend. Additionally, AFM (atomic force microscopy) and contact angle methods were used to determine the free surface energy [80]. The use of atomic force microscopy (AFM) to study particle properties such as surface roughness provided the evidence for flowability enhancement as dry coating of sub-micron particles with nano particles showed an increase in roughness which reduced the contact area between particles [11, 24, 65]. Chen and colleagues [11] used images from AFM to estimate the interparticulate adhesion. Saharan [54] reported the use of AFM to investigate the increase in affinity of guest API (zanamivir) to Lactose (host) when the molecular arrangement in zanamivir changed from crystalline to amorphous form.

## Numerical Simulation using DEM (Discrete Element Method)

Numerical simulation of dry powder coating devices has been reported [23, 84, 85]. Chen and coworkers [84] conducted numerical simulation for mechanofusion using DEM to visualise air flow pattern and dynamics of the systems, and offered an understanding of the influence of various operating parameters on the resultant particle characteristics. Similarly, a three-dimensional DEM was used to study the compression and shear forces involved in theta composer [27]. It was found that the shear force and hence coating is proportional to the speed of the rotor tip and mixing time but inversely proportional to the distance between the rotor and the wall. In another study by Nakamura
and colleagues [86] numerical simulation of dry coating in RFB (rotating fluidised bed) using DEMCFD (Discrete Elemental Modelling coupled with Computational Fluid Dynamics) enabled the researchers to simulate the three-dimensional fluidization of particles within the system. Mass distribution was visualised and the effect of different processing parameters were also investigated. The results were used to provide additional evidence of the coating process using RFB compared with conventional fluidised bed. Recently [87] a hybrid DEM with PBM (population balanced modelling) was employed to develop a better understanding of the mixing process using conventional mixers. DEM simulation provides information at particle level including its velocity, while PBM is a reflection of blending dynamics affecting the mixing process which includes RSD (relative standard deviation) and blend composition.

Table 2: Summary of particle dynamics and surface profiling techniques for dry coated composite particles highlighting the used techniques, their advantages and limitations

| Technique | Advantages | Limitations | Reference |
| :---: | :---: | :---: | :---: |
| PEPT | - Quantitative analysis of dynamics of the particles within the dry coating device. <br> - Can be used for opaque devices entailing metals with multiphase components | - High cost tag <br> - Could not fit all dry coating devices <br> - Requires specialised personnel | [88] |
| RPT | - Quantitative analysis of dynamics of the particles within the dry coating device. |  | [75] |
| XPS | - Elemental analysis of upper most layer $2-10 \mathrm{~nm}$ <br> - High accuracy and specificity | - The depth up to 10 nm allows for lower layer to be detected therefore no precise quantification <br> - Distinctive differences in elemental composition between guest and host is prerequisite to enable the identification of the coat <br> - High cost tag <br> - Requires specialised personnel <br> - Not suitable for thermolabile material <br> - Slight contaminations might alter the results | [77, 78, 89] |
| TOF-SIMS | - Elemental analysis and quantification of upper most layer $2-5 \mathrm{~nm}$ <br> - Ultrahigh sensitivity and specificity <br> - Very small sample size | - High cost tag <br> - Requires specialised personnel particularly for interpretation and analysis of data <br> - Slight contaminations might alter the results | $\begin{array}{ll} {[78,} & 79, \\ 89] & \end{array}$ |
| SEM | - Qualitative analysis of the extent of coating <br> - Particle attrition could be detected. <br> - Simple techniques when compared to XPS and TOFSIMS <br> - Very small sample size | - Requires conditioning for the sample by coating to enhance the conductivity. | [61] |
| FTIR | - Identify physical or chemical bonding between host and guest particles by changes in the intensity of specific spectra | - Requires specialised personnel to analyse spectra | [29, 83] |
| BET | - Identify the reduction in specific surface area upon coating. Used to identify optimum percentage of guest material to produce complete coverage. | - Not suitable for thermolabile material. <br> - Require conditioning for the sample | [60] |
| IGC | - Identify the change in surface energy (dispersive and specific) upon coating. <br> - Good for thermolabile material | - Requires specialised personnel to analyse results <br> - High price tag | [11, 80] |
| AFM | - Determine surface roughness at submicron level. <br> - Can be used to determine surface energy, enhancement in flowability and particle affinity. | - Requires specialised personnel to analyse results <br> - Only for small particle size (submicron host and nano scale guest) <br> - High price tag | [11, 54] |
| DEM | - Numerical simulation of particles within the device to understand the dynamics of the | - Requires specialised personnel <br> - Extensive research owing to the big difference between guests to host particles | $[23, \quad 84$, $85]$ |

## Conclusion

Dry powder coating is an emerging technique in pharmaceutical and chemical industries with diverse applications. A one step process produces composite particles with improved functionalities. The gap in this domain is due to the lack of user friendly processing equipment and characterisation techniques that enable understanding of the process and evaluation of the resultant functionalities. In this review, key characterisation techniques were addressed and adapted to understand particle functionalities to discuss coating performance. The knowledge of characterisation technique will enable formulators and researchers to select the optimal method for their formulation. The use of commonly employed flowability characterisation techniques like angle of repose, bulk and tapped densities are still of value for screening changes in flowability. However, emerging techniques such as aeration, shear cell test and particle size analyses for dispersibility provide an added value of enabling the quantitative analysis of the extent of dry coating necessary to achieve the target functionality. Complete coverage may not be required for specific functionalities as in the case such as flowability, dispersibility, and fluidisability. Hence correlating functionality assessment with the degree of coating should be targeted in formulation optimisation. Also, processing conditions could vary according to the desired functionality and physical properties of components. Therefore, advanced surface profiling techniques like XPS, TOF-SEMS, AFM, BET and SEM contribute to better understanding of the mechanism of coating. Evaluating the changes in surface energy, area and elemental analysis will facilitate formulation and device performance optimisation.

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## References

1. Stange, U. et al. (2014) Taste masking of naproxen sodium granules by fluid-bed coating. Pharm Dev Technol. 19, 137-47.
2. Almería, B. et al. (2011) A multiplexed electrospray process for single-step synthesis of stabilized polymer particles for drug delivery. J.Control. Release. 154, 203-210.
3. Jules, S.J. and Edith, M. (2004) A novel mechanism for spontaneous encapsulation of active agents: Phase inversion nanoencapsulation. In Carrier-based drug delivery. American Chemical Society, pp. 214-223.
4. El-Actsser, M. et al. (2001) Advances in emulsion polymerization for coatings applications: Latex blends and reactive surfactants. J Coating. Technol. 73, 51-63.
5. Dahmash, E.Z. and Mohammed, A.R. (2015) Functionalised particles using dry powder coating in pharmaceutical drug delivery: Promises and challenges. Expert Opin. Drug Deliv. 1-13.
6. Ishizaka, T. et al. (1988) Complexation of aspirin with potato starch and improvement of dissolution rate by dry mixing. Chem Pharm Bull. 36, 2562-2569.
7. Alderborn, G. and Nyström, C. (1996) Pharmaceutical powder compaction technology. Edited by Gran Alderborn, Christer Nystrm. Drugs and the pharmaceutical sciences: 71: New York : Marcel Dekker .
8. Alonso, M. and Alguacil, F. (1999) Dry mixing and coating of powders. Rev Metalu. 35, 315328.
9. Ramlakhan, M. et al. (2000) Dry particle coating using magnetically assisted impaction coating: Modification of surface properties and optimization of system and operating parameters. Powder Technol. 112, 137-148.
10. Yang, J. et al. (2005) Dry particle coating for improving the flowability of cohesive powders. Powder Technol. 158, 21-33.
11. Chen, Y. et al. (2010) Characterization of particle and bulk level cohesion reduction of surface modified fine aluminum powders. Colloids Surf., A. 361, 66-80.
12. Zhou, Q. et al. (2010) Effect of host particle size on the modification of powder flow behaviours for lactose monohydrate following dry coating. Dairy Sci Technol. 90, 237-251.
13. Mullarney, M.P. et al. (2011) Applying dry powder coatings. Pharm Technol. 35, 94-102.
14. Mullarney, M.P. et al. (2011) Applying dry powder coatings to pharmaceutical powders using a comil for improving powder flow and bulk density. Powder Technol. 212, 397-402.
15. Han, X. et al. (2013) Dry coating of micronized api powders for improved dissolution of directly compacted tablets with high drug loading. Int J Pharm. 442, 74-85.
16. Zhou, Q. et al. (2011) Effect of mechanical dry particle coating on the improvement of powder flowability for lactose monohydrate: A model cohesive pharmaceutical powder. Powder Technol. 207, 414-421.
17. Chen, Y. et al. (2007) Fluidization of coated group c powders. AIChE J. 54, 104-121.
18. Smikalla, M. et al. (2011) Impact of excipients on coating efficiency in dry powder coating. Int J Pharm. 405, 122-131.
19. Chen, Y. et al. (2009) Fluidized bed film coating of cohesive geldart group C powders. Powder Technol. 189, 466-480.
20. Jallo, L.J. et al. (2012) Improvement of flow and bulk density of pharmaceutical powders using surface modification. Int J Pharm. 423, 213-25.
21. Honda, H. et al. (1994) Preparation of monolayer particle coated powder by the dry impact blending process utilizing mechanochemical treatment. Colloids Surf., A. 82, 117-128.
22. Han, X. et al. (2011) Simultaneous micronization and surface modification for improvement of flow and dissolution of drug particles. Int J Pharm. 415, 185-195.
23. Pfeffer, R. et al. (2001) Synthesis of engineered particulates with tailored properties using dry particle coating. Powder Technol. 117, 40-67.
24. Zhou, Q.T. et al. (2010) Understanding the influence of powder flowability, fluidization and de-agglomeration characteristics on the aerosolization of pharmaceutical model powders. Eur J Pharm Sci. 40, 412-421.
25. Gera, M. et al. (2010) Mechanical methods for dry particle coating processes and their applications in drug delivery and development. Recent Pat Drug Deliv Formul. 4, 58-81.
26. Quevedo, J. et al. (2006) Fluidization of nanoagglomerates in a rotating fluidized bed. AIChE J. 52, 2401-2412.
27. Endoh, S. et al. (2004) Experimental and theoretical analysis of mechanical coating process of particles with the theta composer. Chem Eng Commun. 191, 1259-1270.
28. Koishi, M. Ishizaka, T. and Nakajima, T. (1984) Preparation and surface properties of encapsulated powder pharmaceuticals. In Microencapsulation and artificial cells (Chang, T.M.S., eds) pp259-262, Humana Press
29. Ishizaka, T. et al. (1989) Preparation of drug-diluent hybrid powders by dry processing. $J$ Pharm Pharmacol. 41, 361-368.
30. Susana, L. et al. (2011) Development and characterization of a new thief sampling device for cohesive powders. Int J Pharm. 416, 260-267.
31. Muzzio, F.J. et al.(2003) Sampling and characterization of pharmaceutical powders and granular blends. Int J Pharm. 250, 51-64.
32. Prescott, J.K. and Garcia, T.P. (2001) A solid dosage and blend content uniformity troubleshooting diagram. Pharm. Technol. 25, 68-88.
33. Hancock, B.C. and Garcia-Munoz, S. (2012) How do formulation and process parameters impact blend and unit dose uniformity? Further analysis of the product quality research institute blend uniformity working group industry survey. J Pharm Sci. 102, 982-986.
34. Portillo, P.M. et al.(2008) Quality by design methodology for development and scale-up of batch mixing processes. J Pharm Innov. 3, 258-270.
35. Muzzio, F.J. et al. (1997) Sampling practices in powder blending. Int J Pharm. 155, 153-178.
36. Kræmer, J., et al. (1999) Sampling bias in blending validation and a different approach to homogeneity assessment. Drug Development \& Industrial Pharmacy. 25, 217.
37. Mahato, R.I. and Narang, A.S., eds (2012) Pharmaceutical dosage forms and drug delivery. Crc press pharmacy education series.
38. FDA. (2003) Guidance for industry: Powder blends and finished dosage units-stratified inprocess dosage unit sampling and assessment. Pharmaceutical Manufacturing Handbook: Regulations and Quality. Hoboken: Wiley.
39. Portillo, P.M. et al. (2006) Characterizing powder mixing processes utilizing compartment models. Int J Pharm. 320, 14-22.
40. Paul, E. L. Antimo-Obeng, V. A., and Kresta, S. M. eds. (2004) Handbook of industrial mixing: Science and practice. John Wiley and Sons-Interscience.
41. Bridgwater, J. (2012) Invited review: Mixing of powders and granular materials by mechanical means-a perspective. Particuology. 10, 397-427.
42. Boehm, G. et al.(2003) The use of stratified sampling of blend and dosage units to demonstrate adequacy of mix for powder blends. PDS J Pharm Sci Tech. 57, 59-74.
43. Zhou, Q.T. et al. (2010) Improving aerosolization of drug powders by reducing powder intrinsic cohesion via a mechanical dry coating approach. Int. J. Pharm. 394, 50-59.
44. Ghoroi, C. et al. (2013) Dispersion of fine and ultrafine powders through surface modification and rapid expansion. Chem Eng Sci, 85, 11-24.
45. Zhou, Q.T. and Morton, D.A. (2012) Drug-lactose binding aspects in adhesive mixtures: Controlling performance in dry powder inhaler formulations by altering lactose carrier surfaces. Advanced drug delivery reviews. 64, 275-284.
46. Zhou, Q.T. et al. (2013) Effect of surface coating with magnesium stearate via mechanical dry powder coating approach on the aerosol performance of micronized drug powders from dry powder inhalers. AAPS PharmSciTech. 14, 38-44.
47. Song, M. and De Villiers, M.M. (2004) Effect of a change in crystal polymorph on the degree of adhesion between micronized drug particles and large homogenous carrier particles during an interactive mixing process. Pharm Dev Technol. 9, 387-398.
48. Nystrom, C. and Westerberg, M. (1986) The use of ordered mixtures for improving the dissolution rate of low solubility compounds. J Pharm Pharmacol. 38, 161-165.
49. Stewart, P. and Alway, B. (1995) Aggregation during the dissolution of diazepam in interactive mixtures. Particul Sci Technol. 13, 213-226.
50. Du, J.P. and Hoag, S.W. (2003) Characterization of excipient and tableting factors that influence folic acid dissolution, friability, and breaking strength of oil- and water-soluble multivitamin with minerals tablets. Drug Dev Ind Pharm. 29, 1137-1147.
51. Nagai, Y. et al. (2006) Improvement in dissolution property of poorly water-soluble drugs by using mechanofusion system. J Soc Powder Technol. 43, 20-27
52. Shaw, L.R. et al. (2005) The effect of selected water-soluble excipients on the dissolution of paracetamol and ibuprofen. Drug Dev Ind Pharm. 31, 515-525.
53. Saharan, V.A. and Choudhury, P.K. (2011) Dissolution rate enhancement of gliclazide by ordered mixing. Acta pharmaceutica (Zagreb, Croatia). 61, 323-34.
54. Saharan, V.A. et al. (2008) Ordered mixing: Mechanism, process and applications in pharmaceutical formulations. Asian J Pharm Sci. 3, 240-259.
55. Ouabbas, Y. et al. (2007) Effect of mechanical dry coating on the flowability and the wettability of silica gel powder. Proceedings XI ${ }^{\circ}$ Congrès de la Société Française de Génie des Procédés. Saint Etienne, 96, 2-910239-70-5. 96.
56. Supabphol, R. and Stewart, P.J. (1996) Aggregation during the dissolution of diazepam in interactive and granulated mixtures. Pharm Pharmacol Commun. 2, 233-236.
57. Allahham, A. and Stewart, P.J. (2007) Enhancement of the dissolution of indomethacin in interactive mixtures using added fine lactose. Eur. J.Pharm. Biopharm: 67, 732-742.
58. Westerberg, M. and Nystrom, C. (1993) Physicochemical aspects of drug release .17. The effect of drug surface-area coverage to carrier materials on drug dissolution from ordered mixtures. Int J Pharm. 90, 1-17.
59. Qu, L. et al. (2014) Investigation of the potential for direct compaction of a fine ibuprofen powder dry-coated with magnesium stearate. Drug DevInd Pharm. 41, 1-13.
60. USP-29. The United States Pharmacopeia : The National Formulary : USP 29 NF 24 : By authority of the united states pharmacopeial convention, inc. Meeting at washington, d.C., march 22-24, 2009/prepared by the committee of revision and published by the board of trustees. 2009: /Rockville, Md. : United States Pharmacopeial Convention, 2006. 32nd revision.
61. Ghoroi, C. et al. (2013) Dispersion of fine and ultrafine powders through surface modification and rapid expansion. Chem Eng Sci. 85, 11-24.
62. Geldart, D. et al. (2006) Characterization of powder flowability using measurement of angle of repose. China Particuology. 4, 104-107.
63. Castellanos, A. (2005) The relationship between attractive interparticle forces and bulk behaviour in dry and uncharged fine powders. Adv in Phys. 54, 263-376.
64. Ropero, J. et al. (2009) Near-infrared spectroscopy for the in-line characterization of powder voiding part i: Development of the methodology. J Pharm Innov. 4, 187-197.
65. Castellanos, A. (2005) The relationship between attractive interparticle forces and bulk behaviour in dry and uncharged fine powders. Advances in Physics. 54, 263-376.
66. Senthil, A. et al. (2011) Development and evaluation of orally disintegrating tablets of metoprolol tartarate by direct compression method using different diluents. IRJP. 2, 118-125.
67. Davé, R.N. et al. (2013) Special issue on pharmaceutical powders: Towards developing understanding of the influence of materials and processes on product performance. Powder Technol. 236, 1-4.
68. Ghoroi, C. et al., (2013) Multi-faceted characterization of pharmaceutical powders to discern the influence of surface modification. Powder Technol-Lausanne-. 236, 63-74.
69. Schwedes, J. (2003) Review on testers for measuring flow properties of bulk solids. GM. 5, 143.
70. Shi, J. et al. (2015) The kinetics of de-agglomeration of magnesium stearate dry-coated salbutamol sulphate powders. KONA Powder and Particle Journal. 32, 131-142.
71. USP-35 (2011) The United States Pharmacopeia. Usp 35 : The national formulary. Nf 30 / by authority of the united states pharmacopeial convention ; prepared by the council of experts and its expert committees. Rockville, Md. : United States Pharmacopeial Convention, c2011.
72. BP (2012), The British pharmacopoeia 2012. London : Stationery Office, 2012.
73. Romero, A.J. et al. (1993) Monitoring crystal modifications in systems containing ibuprofen. Int J. Pharm. 99, 125-134.
74. Höckerfelt, M.H. et al. (2009) Dry mixing transformed micro-particles of a drug from a highly crystalline to a highly amorphous state. Pharm Dev. Technol. 14, 233-239.
75. Doucet, J. et al. (2008) An extended radioactive particle tracking method for systems with irregular moving boundaries. Powder Technol. 181, 195-204.
76. Zhou, Q.T. et al. (2011) Investigation of the extent of surface coating via mechanofusion with varying additive levels and the influences on bulk powder flow properties. Int J Pharm. 413, 36-43.
77. Gurker, N. et al. (1983) Imaging XPS—a new technique, i-principles. Surf Interface Anal. 5, 13-19.
78. Qu, L. et al. (2015) Investigation of the potential for direct compaction of a fine ibuprofen powder dry-coated with magnesium stearat Drug Dev Ind Pharm. 41, 825-837.
79. Stephan, T. (2001) Tof-sims in cosmochemistry. Planetary and Space Science. 49, 859-906.
80. Das, S.C. and Stewart, P.J. (2012) Characterising surface energy of pharmaceutical powders by inverse gas chromatography at finite dilution. J Pharm Pharmacol. 64, 1337-1348.
81. Watson, D.G. and Edrada-Ebel, R., eds (2012) Pharmaceutical analysis : A textbook for pharmacy students and pharmaceutical. Churchill Livingstone.
82. Bhalekar, M.R. et al. (2010) Synthesis of mcc-peg conjugate and its evaluation as a superdisintegrant. AAPS PharmSciTech. 11, 1171-8.
83. Khan, F.L.A. et al. (2008) Ftir study of hydrogen bonding interactions between alkyl esters and hexanol, p-cresol in carbon tetrachloride. Indian J Pure AP Phy. 46, 12-19.
84. Chen, W. et al. (2004) Numerical simulation of mechanofusion system. Powder Technol. 146, 121-136.
85. Portillo, P.M. et al. (2007) Hybrid dem-compartment modeling approach for granular mixing. Aiche Journal. 53, 119-128.
86. Nakamura, H. et al. (2006) Numerical simulation of film coating process in a novel rotating fluidized bed. Chem Pharm Bull. 54, 839-46.
87. Sen, M. et al. (2013) Mathematical development and comparison of a hybrid pbm-dem description of a continuous powder mixing process. Journal of Powder Technology. DOI: http://dx.doi.org/10.1155/2013/843784 (http://www.hindawi.com/)
88. Pérez-Mohedano, R. et al. (2015) Positron emission particle tracking (pept) for the analysis of water motion in a domestic dishwasher. Chem Eng J. 259, 724-736.
89. Mukhopadhyay, S.M. (2003) Sample preparation for microscopic and spectroscopic characterization of solid surfaces and films. Sample Preparation Techniques in Analytical Chemistry. 162, 377-411.
