Performance, emission and combustion characteristics of an indirect injection (IDI) multi-cylinder compression ignition (CI) engine operating on neat jatropha and karanj oils preheated by jacket water

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ABSTRACT

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11 12 13 14 Renewable non-edible plant oils such as jatropha and karanj have potential to substitute fossil diesel fuels in CI engines. A multi-cylinder water cooled IDI type CI engine has been tested with jatropha 15 and karanj oils and comparisons made against fossil diesel. The physical and chemical properties of 16 17 the three fuels were measured to investigate the suitability of jatropha and karanj oils as fuels for CI 18 engines. The engine cooling water circuit and fuel supply systems were modified such that hot jacket 19 water preheated the neat plant oil prior to injection. Between jatropha and karanj there was little 20 difference in the performance, emission and combustion results. Compared to fossil diesel, the brake specific fuel consumption on volume basis was around 3% higher for the plant oils and the brake 21 22 thermal efficiency was almost similar. Jatropha and karanj operation resulted in higher CO₂ and NO_x emissions by 7% and 8% respectively, as compared to diesel. The cylinder gas pressure diagram 23 24 showed stable engine operation with both plant oils. At full load, the plant oils gave around 3% higher 25 peak cylinder pressure than fossil diesel. With the plant oils, cumulative heat release was smaller at 26 low load and almost similar at full load, compared to diesel. At full load, the plant oils exhibited 5% shorter combustion duration. The study concludes that the IDI type CI engine can be efficiently 27 28 operated with neat jatropha (or karanj) oil preheated by jacket water, after small modifications of the 29 engine cooling and fuel supply circuits.

Keywords: Non-edible plant oil, CI engine, Performance, Combustion, Emissions, Biofuel

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37 38 **1. Introduction**

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40 The total estimated world greenhouse gas (GHG) emissions in the year 2005 amounted to 44 billion 41 tonnes of CO₂ equivalent, of which 66.5% were associated with energy services. Of the total emissions, the share from transportation, electricity and heat was 39.2% – corresponding to 59% of 42 the emissions related to energy services [1]. Internal combustion engines, gas turbines and steam 43 44 turbines are widely used in the provision of these services. Use of internal combustion engines, and especially CI engines, is rapidly growing for small-to-medium scale decentralised power, irrigation, 45 46 combined heat and power, and tri-generation applications due to a number of advantages e.g. 47 robustness, high efficiency, fuel flexibility. Therefore a substantial reduction in GHG emissions is 48 possible by replacing fossil diesel in CI engines with biomass derived renewable fuels. Different types 49 of liquid biofuels have been discussed in the literature [2-5]. One of these is non-edible or edible plant oil, either neat or blended with fossil diesel [6-12]. Plant oils can also be transesterified to produce 50 51 biodiesel [13-15]. However, neat (unprocessed) plant oils offer considerable life cycle energy and 52 GHG emission savings compared to both biodiesel and fossil diesel [16,17]. Use of neat oils also 53 offers economic advantages due to the fewer requirements for machinery, chemicals and processes. 54

55 Plant oil, when blended with fossil diesel or transesterified into biodiesel, does not generally require any engine modification [18,19]. Modification is recommended, however, if the plant oil is used 56

57 directly in neat form, mainly due to higher viscosity of the plant oils compared to fossil diesel which 58 leads to incomplete combustion and coke formation inside the cylinder and in fuel supply systems including the injectors [16,20,21]. For example, in one study, Bari [21] reported that, after 500 hours 59 of running with crude palm oil, maximum power decreased by 20% and minimum brake specific fuel 60 consumption increased by 26%. Heavy carbon deposits in the combustion chamber and wear in the 61 injector pump plunger and delivery valve were also observed [21]. Plant oils are usually preheated to 62 reduce the viscosity before injection for better engine performance and reduced emissions. Typically 63 64 engine exhaust gas [22-24] or electricity [7,8,10,25,26] has been used for preheating. Use of engine 65 exhaust gas is better as it utilises waste heat and therefore involves no additional running costs; however, it requires close monitoring and control of the exhaust gas flow and temperature of the 66 preheated oil. Overheating due to either malfunction or sudden overloading of the engine could cause 67 68 the oil to flash. Preheating by exhaust gas therefore requires automation in order to meet health and safety requirements, which can be expensive. An alternative is to use the engine jacket water for 69 70 preheating. In one recent study, for example, Basinger et al. [20] designed and installed a V-shaped 71 plug type oil preheater by modifying the Change Over Valve (COV) of the Lister engine, which 72 collected heat from the cooling water jacket. The heat transfer modelling and identification of the 73 channel geometry of the preheater enabled the authors to maintain the oil temperature before the injector at about 90°C across all engine loads [20]. 74

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76 To avoid direct competition with food crops, non-edible plant oils are an attractive substitute for fossil 77 diesel fuel. Of the non-edible plant oils, jatropha (Jatropha curcas) and karanj (Pongamia pinnata) 78 have received considerable attention [27-31] mainly due to their availability and ease of cultivation. 79 The jatropha tree is notable for its tolerance of poor soils and dry climates. Karanj is a forest tree that is especially interesting as a nitrogen fixing species. Their oils have usually been evaluated separately 80 81 in different engines and under different test conditions [22-24, 28, 32-35]. Moreover, most studies 82 were conducted in only single cylinder engines of mainly direct injection (DI) type with compression ratios in the range of 16.5–18.5 [22-24, 28, 32-35]. Only a few studies investigated engine combustion 83 84 characteristics [23, 32]. Chauhan et al. [22] tested preheated and unheated jatropha oil in a 5.9 kW 85 single cylinder DI engine; jatropha oil was preheated to 100°C and they reported 3% decrease in 86 thermal efficiency when compared to fossil diesel operation. These authors reported that NOx emission was lower than fossil diesel in the case of unheated jatropha operation, but increased by 45% 87 88 when preheated jatropha was used [22]. Preheated neat karanj oil gave around 40% increase in Brake Specific Fuel Consumption (BSFC) and 25% decrease in brake thermal efficiencies as compared to 89 90 fossil diesel operation [23]. Khatri et al. [24] investigated the effect of injection timing on engine performance when operated on preheated karanj oil-diesel blends using a single cylinder 3.7 kW 91 92 Kirloskar DI engine with compression ratio of 16.5 and injection timing of 23° BTDC. The thermal efficiency of the 40:60 karanj-diesel blend was found to be similar with fossil diesel operation at 93 optimum injection timing of 19° BTDC. In another similar study, 23% higher BSFC, 14% lower 94 brake thermal efficiency and 15% higher CO₂ emission were reported with preheated neat jatropha oil 95 96 used instead of fossil diesel in a single cylinder DI engine with compression ratio of 17.5 [35].

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98 As interest in jatropha and karanj is growing, it is important to investigate these oils in CI engines of 99 types not prominent in the current literature. Furthermore, for valid comparison, the oils need to be tested in the same engine and under the same conditions. Given the high viscosity of plant oils, use of 100 an IDI type engine may be advantageous over DI, because mixing of the plant oil with air should be 101 better due to the turbulence created by the partial burning in the secondary combustion chamber (pre-102 chamber) of the IDI engine. Consequently, it is expected that, the combustion of high viscous plant 103 104 oils will be better in IDI type engine, and the coke formation in the combustion chamber will be minimum. Furthermore, NO_X emission may be lower in the IDI than in the DI engine, due to the lack 105 106 of air in the secondary combustion chamber and lower temperature in the main combustion chamber [36]. With the aim of verifying these anticipated benefits, an experimental study of the performance, 107 108 exhaust emissions and combustion parameters of a multi-cylinder IDI type CI engine running on neat 109 plant oils is presented in this study. Jatropha and karanj will be compared against each other and against fossil diesel operation for reference. The engine cooling circuit will be modified so that the hot 110

jacket water (which is normally wasted) preheats the plant oil. The physical and chemical characteristics of the fuels will first be measured for use in the performance calculations and evaluation of the results. The study includes combustion analysis by means of cylinder pressure measurements, and a preliminary investigation of the coke formation inside the cylinder, piston and injectors.

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118 2. Materials and methods

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121 2.1. Fuel Characterisation

Jatropha oil produced in Ghana was obtained from a UK supplier (Matrix Biofuels Ltd.), whereas 123 124 karanj oil produced in India was obtained from an Indian supplier (A-Mark Procon Ltd.). The oils were cleaned in the laboratory using 10, 5 and 1 µm sock filters. A Parr 6100 bomb calorimeter was 125 126 used to measure the higher heating values (HHV). Canon Fenski u-tube viscometers and thermostatic water bath ($\pm 0.1^{\circ}$ C) were used to measure the kinematic viscosities at different temperatures 127 according to ISO 3104 having an accuracy of $\pm 0.22\%$. The pH values were measured using a HACH 128 HQ40d meter and densities using a standard hydrometer. Flash point was measured using a Setaflash 129 series 3 plus closed cup flash point tester (model 33000-0). A Shatox OPLCM instrument was used to 130 measure the cetane number, pour point temperature and filterability temperatures. Elemental analysis 131 to investigate the carbon, hydrogen, nitrogen, oxygen and sulphur content (% wt) was performed by 132 an accredited laboratory using a Thermo EA1108 series elemental analyser. The lower heating value 133 134 (LHV) was calculated from the HHV and hydrogen content.

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137 2.2. *Experimental set-up*138

The specifications of the IDI water cooled CI engine selected for the study are given in Table 1. The
experimental set-up, including modifications carried out to the engine fuel supply and cooling water
systems, is shown in Fig. 1.

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143 2.2.1. Fuel supply and preheat144

145 A dual-fuel supply system was designed so that the engine can be started and warmed up with fossil diesel before switching to plant oil after about 10 minutes (Fig. 1). At the end of each test run 146 147 reversion to diesel operation flushed the remaining plant oil from the fuel supply and engine injection system. This avoided clogging of the fuel filters and injector holes due to thickening of the oil once 148 the system cools. The fuel supply lift pump supplied with the engine was disconnected, and the spill 149 150 from the injector was connected to the main fuel supply line using a T-connector. Additional filters were connected into the fuel supply system. The fuel supply tanks were placed at 3 m height for 151 reliable fuel flow and to overcome the pressure losses in the additional fuel filters and heat exchanger 152 153 (HX). Note that Bari et al. [37] have already reported the advantages as regards power output of 154 connecting additional fuel filters and raising the fuel supply tank in plant-oil fuelled engine operation. 155 Two 12V DC fuel transfer pumps were installed to transfer the fuels from the ground tank to the 156 overhead tanks (not shown). Stainless steel piping and valves were used to avoid any corrosion or 157 erosion.

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159 At full load, engine jacket water temperature is around 100° C with a flow of 33 l/min (Table 1); this 160 means that a considerable amount of the heat energy is available in the hot jacket water. To preheat

161 the plant oil with this water, a plate type HX was connected to the engine jacket water cooling system

162 via two manually operated valves that were opened when using plant oils but kept closed for diesel.

163 Another manually operated valve controlled the flow of the hot jacket water to the plate HX and

hence the temperature of the preheated plant oils (Fig. 1).

165 In a standard engine, hot jacket water is cooled by a radiator and fan and then returned to the cooling 166 circuit. To facilitate testing within a standard test cell, however, the radiator was by-passed by a 167 header tank type HX (Bowman UK) supplied with tap water. Although the radiator water pipes were 168 disconnected from the engine, the radiator itself and the fan were not disconnected to enable 169 comparisons of the engine performance with the rated figures provided by the manufacturer. 170 Moreover the radiator fan was needed to cool the engine accessories.

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172 2.2.2. Instrumentation

An eddy current dynamometer (Froude Hofmann AG80HS) was used to measure and adjust the 174 175 engine load and speed (Fig. 1). The torque and speed accuracies of the dynamometer were ± 0.4 Nm 176 and ± 1 rpm respectively. A five-gas emission analyser (Bosch BEA 850) and smoke opacity meter 177 (Bosch RTM 430) were used to analyse the exhaust gas components and to measure the smoke intensity respectively. A graduated cylinder and stop watch were used to measure the fuel 178 179 consumption rate (Fig. 1). K-type thermocouples were used to measure the temperatures of the exhaust gas, fuel inlet, lub oil and engine jacket water. Fuel temperature was measured immediately 180 after the plate HX and after the fuel injection pumps ie. just before the injector, thus approximating as 181 closely as possible the temperature of the injected fuel (Fig. 2). A LabVIEW data acquisition system 182 183 was used to log the temperatures at the various locations. The dynamometer was calibrated and the 184 calibration curve (not shown) was used to obtain the brake torque applied at the engine shaft. The 185 engine was operated at different loads at a constant speed of 1500 rpm. Brake thermal efficiency was 186 calculated based on LHV.

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A pressure sensor (Kistler 6125C11) and charge amplifier (Kistler 5064B11) were used to measure 188 189 pressure in the cylinder nearest the radiator end of the engine. A suitable location in the main combustion chamber was identified and a mounting sleeve with locking mechanism was designed and 190 191 fabricated for this installation. Machining of the cylinder head through the water jacket was performed 192 accordingly (Fig. 2). The sensor was not installed flush to the combustion chamber surface; rather it 193 was recessed by 2 mm to avoid damage to the diaphragm and to aid maintenance. Another pressure sensor (Kistler 4065A500A0) and amplifier (Kistler 4618A0) were used to measure the fuel line 194 195 injection pressure (Fig. 2). This sensor was installed on the fuel delivery line of the same cylinder. An optical encoder (Kistler 2614A) was installed for detection of the crank angle position. The amplifiers 196 and the encoder electronics were connected to the 'KiBox' (Kistler, model 2893AK8) for data 197 198 logging, which was connected to a PC through the ethernet port (Fig. 1). KiBoxCockpit software was 199 used to measure and analyse the various engine combustion parameters.

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202 3. Results and discussion

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205 **3.1.** Fuel properties

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With regard to LHV, density and cetane number, Table 2 shows only very small differences between karanj and jatropha and minor differences of both plant oils compared to fossil diesel. The LHVs of the plant oils were lower by about 11% and their densities higher by about 12%. Their cetane numbers were almost 10% lower than for fossil diesel. In contrast, their viscosities and flash point temperatures were considerably higher than those of diesel. Viscosity of the plant oils is temperature dependent and can be reduced by 80-90% through preheating to 90° C (Fig. 3). Note that high flash point is advantageous as regards safety in storage, handling and preheating.

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The carbon content in the fossil diesel was 7–8% higher than in the plant oils. Nitrogen content in jatropha oil was much higher than in karanj oil and in fossil diesel (Table 2). Hydrogen content was comparable to fossil diesel but oxygen content was 6 to 7 times higher in the plant oils. For all three

fuels sulphur was at trace levels. The properties of any plant oil depend on the plant variety and

conditions of cultivation; however, the properties of the oils measured in our study are in line withthose found in the literature [16].

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223 **3.2.** Engine performance and exhaust emissions

Preheating raised the temperature of the oils to 58–75°C, the higher temperatures being achieved at higher brake power outputs due to increasing jacket water temperature (Fig. 4). Although the HX was installed close to injection pump, the temperature of the fuel dropped by 12–15°C by the time it reached the injectors (Fig. 4). The level of preheating achieved by using the jacket water reduced the viscosity of the oil considerably by 60–70% (Fig. 3).

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231 At full load, BSFC was 16% and 3% higher on weight and volume bases respectively, when the engine was fuelled with preheated neat jatropha (or karanj) oil, as compared to unheated fossil diesel 232 233 (Fig. 5a, 5b). Because the calorific value of the jatropha or karanj oil is lower than that of fossil diesel, 234 more fuel is needed for the same engine output. In addition, although the plant oils were preheated to 235 60–75°C, they remained 10 times more viscous than diesel leading to less efficient mixing of fuel and air. These factors explain the somewhat higher fuel consumptions. Comparison on a volume rather 236 237 than weight basis showed smaller differences in BSFC due to the higher densities of the plant oils 238 (Table 2). Between jatropha and karanj, BSFC results were very close on both weight and volume 239 bases (Fig. 5a, 5b). Brake thermal efficiencies from both plant oils were close to those from diesel at 240 high load, but 10% lower than from diesel at low load condition (Fig. 5c).

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242 At low loads, CO and CO_2 emissions were almost the same for all fuels (Fig. 6a, 6b). The amount of 243 intake air (i.e. oxygen content) was constant throughout the engine test – so, at higher loads the fuel to air ratio decreased (lower lambda values) and hence CO emission increased sharply due to the lower 244 245 oxygen content in the mixture. In the case of jatropha and karanj oil operation and at higher load 246 conditions, CO and CO₂ emissions were higher than with fossil diesel as higher amounts of the plant oils were burned for the same engine output (Fig. 6a, 6b). In addition, it is probable that the higher 247 oxygen content in the plant oils helped to combust the jatropha (or karanj) oil and caused higher CO₂ 248 249 emission than with fossil diesel. At full load condition, CO₂ emission was 7% higher for plant oil 250 operation (Fig. 6b). It was also observed that at higher load condition, CO emission was slightly 251 higher with jatropha than karanj. This may be explained by the lower oxygen content in the jatropha 252 oil.

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Higher combustion temperatures in CI engines generally give higher NO_x emission. For jatropha and karanj operation, NO_x emissions were 8% higher than fossil diesel in these tests (Fig. 6c). This is consistent with the higher peak cylinder pressures observed with the plant oils (section 3.3), as higher pressures are normally associated with higher combustion temperatures. Oxygen emissions were almost similar under low load conditions for all fuels; but at high loads O_2 emissions were lower for jatropha (or karanj) oil operation (Fig. 6d). Higher level of fuel-borne oxygen in the jatropha (or karanj) oil probably caused this difference.

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Exhaust temperature is important for co-generation applications (CHP and tri-generation). Little difference was observed in exhaust temperatures among the three fuels tested (Fig. 7a). At high load condition, the smoke opacity values were slightly higher than fossil diesel when the engine was operated either on jatropha or karanj oil (Fig. 7b). Lambda values were found to be mostly similar for all fuels (Fig. 7c). However, at higher loads, lower lambda values were observed for plant oil operation corresponding to the higher BSFC.

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At low load, an additional means to preheat the oil is desirable, as the jacket water temperature is low. An off-the-shelf glow plug type electrical heater could be used for this purpose. This type of heater is powered by the engine alternator and can be switched on during low load operation. Overall, the performance and exhaust emission results obtained in the current study compare favourably to those with additional fuel filters and raised fuel tank, caused this better engine performance. Previous studies were mostly conducted in single cylinder DI engines with compression ratios in the range of 16.5–18.5, whereas the compression ratio of the engine used in this study was 22. Although the engine was not tested at other compression ratios, it is believed that this high compression ratio could be another reason for better engine performance and emission.

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281 **3.3.** Combustion characteristics

Stable engine operation was obtained regardless of fuel and load, as evident from the smooth 283 284 pressure-crank angle diagram (Fig. 8). No significant differences in peak cylinder pressure were noted 285 between jatropha and karanj oils over a range of loads. Compared to fossil diesel, however, peak cylinder pressures were about 3% higher (Fig. 8). The cylinder pressure profiles before the peak 286 pressure point were almost similar for all three fuels, and only minor variations were observed after 287 288 the peak cylinder pressure point. For all fuels and load conditions, the peak cylinder pressure occurred between 10° and 12.5° crank angle (CA) after top dead centre (ATDC) position. In general, jatropha 289 and karanj oil operation resulted in peak cylinder pressure occurring close to the top dead centre 290 291 position, earlier than with diesel (Fig. 8).

At low loads, integral heat release from fossil diesel combustion was higher (Fig. 9). This can be explained by the lower preheating temperature and higher viscosities of the plant oils at lower loads (Fig. 3, Fig. 4) leading to incomplete combustion. At higher loads, the heat release behaviour was similar to fossil diesel (Fig. 9). For any specific load condition, peak integral heat release values were almost similar between jatropha and karanj. In general, the fuel line injection peak pressure values were higher in the case of the plant oil operation, due to the higher viscosity values of the plant oil even after preheating (Fig. 10); at full load it was almost 9% higher than for fossil diesel.

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301 Total combustion duration is defined as the duration of the crank angle between 5% and 90% 302 combustion. Ignition delay is another important combustion parameter, which is related to the ignition quality (ie. cetane number) of the fuel. Engine parameters affecting the ignition delay period are: 303 304 compression ratio, engine speed, cylinder gas pressure, temperature of the air intake, and quality of fuel spray [34]. In the case of plant oil operation, the start of combustion was delayed compared to 305 306 fossil diesel operation which is consistent with the lower cetane number (Table 2). Total combustion duration increased with engine load for all fuels. The duration was shorter for the plant oils than for 307 diesel (Fig. 11a, 11b). At full load, combustion duration of the jatropha (or karanj) operation was 308 almost 5% lower than for fossil diesel; no significant differences were observed between jatropha and 309 310 karanj in this respect.

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It was observed that in the case of plant oils, the maximum heat release rates were advanced by about 1 to 2.5°CA (figure not shown) relative to fossil diesel. The ignition delay was longer, but the combustion duration was shorter than fossil diesel (Fig. 11). Quick burning in a short period of crank angle duration (i.e. in smaller cylinder volume) caused higher peak cylinder pressures in the case of plant oils. Higher peak cylinder pressures in the case of plant oils have also been reported in the literature [20,21,26,37].

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319 In case of plant oils, the start of combustion was delayed; but once combustion started, the oils burnt 320 quickly resulting in shorter combustion duration than fossil diesel. It is likely that the presence of fuelborne oxygen in the plant oils helped complete combustion, and together with the high ignition delay, 321 322 this caused shorter combustion duration. Furthermore, it was observed that the peak cylinder pressures 323 and maximum heat release rates were advanced in case of jatropha (or karanj) oil relative to fossil diesel, due to the shorter combustion duration. The higher ignition delay period and shorter 324 combustion duration in the case of plant oils (or blends) resulted in higher peak cylinder pressures 325 than with fossil diesel [38]. The combustion duration or ignition of fuel is a complex system and may 326 327 depends on many variables, for example: engine type (IDI/DI) and speed, compression ratio, geometry of combustion chamber, injection parameters (IOP, injection timing), injector geometry, 328

329 temperature of air intake and fuel properties. Generally, plant oils or biofuels have lower cetane number than fossil diesel; therefore, longer ignition delays were observed [38-39]. In contrast to this, 330 shorter ignition delay period has also been reported in the literature [23, 26, 37]. Some studies found 331 332 that fuels with high cetane number have shorter ignition delay and longer combustion duration [40-41]. However, both longer [23, 26] and shorter [42-43] combustion duration were reported for plant 333 oils and other biofuels. This is an area where further research work is recommended. 334

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337 3.4. Engine durability: coke formation 338

339 Coke deposition inside the engine cylinder and in the fuel supply systems, piston ring sticking, 340 leakage of the plant oils into the crank case, clogging of fuel filters and injectors, engine wear and other engine durability problems were reported in the literature when neat plant oils were used in 341 standard CI engines [16,21,37,44-46]. Although a complete endurance study was not performed 342 343 during this study, a visual observation of the piston, cylinder, fuel supply system, and injector 344 conditions was carried out after 50 hours of engine operation with preheated non-edible plant oils. A 345 very small amount of coke formation occurred on to the piston surfaces and cylinder heads (Fig. 12). Injectors were inspected and no corrosion or erosion of the injector was observed. The fuel filters 346 347 were not clogged. Neither engine knocking nor any instability in the engine operation was noticed. 348 Adoption of a preventive maintenance schedule and some changes in the routine maintenance strategy 349 should be conducive to longer term engine operation when the engine is fuelled with these oils.

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352 4. Conclusions

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A three-cylinder indirect-injection compression ignition engine, with nominal output 9.9 kW and 354 355 compression ratio 22, has been tested with non-edible plant oils from jatropha and karanj trees. Performance, emissions and combustion characteristics were compared against fossil diesel operation. 356 The engine was modified with a heat exchanger such that the plant oils were preheated by the engine 357 358 jacket water to temperatures of 46-60°C prior to injection. The three fuels were characterised and found to have properties similar to those reported in the literature. 359

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Between the jatropha and karanj oils, there were few differences in the results from the engine tests. 361 362 Compared to fossil diesel, however, there were a number of small but significant differences when 363 using the plant oils:

- At full load the BSFC was increased by 3% on a volume basis and 16% on a weight basis, 364 i. while brake thermal efficiency was similar. 365
- 366 ii. CO₂ and NO_x emissions were higher by 7% and 8% respectively; CO emissions were similar to fossil diesel operation at low load but at higher loads the plant oils gave slightly 367 higher CO emissions; the O₂ emissions were either similar to or less than those from 368 fossil diesel. 369
- 370 iii. Peak cylinder pressures were slightly higher ($\sim 3\%$).
- Peak integral heat release values were smaller at low load, translating into a lower 371 iv. 372 efficiency than with fossil diesel, but at full load they were almost the same. 373
 - At full load, fuel line injection pressure was around 9% higher. v.
- 374 Total combustion duration was lower under all load conditions. vi.

375 376 The cylinder gas pressure diagrams with jatropha and karanj oils indicated stable engine operation with these alternative fuels. This study concludes that, an IDI type CI engine used in the current study 377 378 combusted the plant oils more efficiently and produced better engine performance, emission and combustion characteristics, compared to results typically reported in the literature using jatropha (or 379 karanj) oil in DI engines with lower compression ratio. This preliminary study reports that neat non-380 edible jatropha or karanj oils can be used effectively (without blending with fossil diesel or 381 382 conversion to biodiesel) in a suitably selected CI engine, provided minor modifications to the standard engine are made. However, full endurance tests are required to assess the engine durability issues andare recommended as further work.

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387 Acknowledgements

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This work was supported by the Research Councils (UK) and DST (Government of India) under the Science Bridges programme (grant number EP/G039992/1). The authors would like to thank Mr David Smith for providing technical assistance in various stages of the study, and Mr Asrar Razzaq (project student, Aston University) and Mr Dhrubatara Mitra (SRI, Ranchi, India) for their help during the experiments.

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Table 1

Specification of the experimental engine

Lister Petter (UK)		
LPWS Bio3 water cooled		
3		
86/80 mm		
1500 rpm		
9.9 kW		
10.9 kW		
Indirect injection. Self-vent fuel system with individual fuel-injection pumps		
20° BTDC		
Natural		
1.395		
1:22		
1-2-3		
1500		
75		
3.19 litre/hr (fossil diesel)		
Combustion-chamber glow plugs		
41.4 litres/sec at full loads at 1500 rpm		
33 litres/min (at 1500 rpm)		
99 - 102°C		

Table 2

Physical and chemical properties	Jatropha oil	Karanj oil	Fossil diesel
Kinematic viscosity (cSt) at 25°C	56.01	79.21	3.37
Pour point temperature (°C)	-6.05	-7.65	-6.17
Cetane number	44.6	45.6	50.2
Filterability temperature (°C)	-7.2	-	-8.03
Flash point temperature (°C)	245	257	70
pH value @ 22°C	6.21	5.82	6.85
Density (kg/m ³) @ 18°C	916	934	834
Higher heating value (HHV) (MJ/kg)	39.87	39.10	44.67
Lower heating value (LHV) (MJ/kg)	37.25	36.61	41.87
Carbon (% wt)	77.88	78.10	84.73
Hydrogen (% wt)	12.38	11.74	13.20
Nitrogen (% wt)	1.19	< 0.10	< 0.10
Sulphur (% wt)	< 0.10	< 0.10	< 0.10
Oxygen (% wt)	8.75	10.30	1.40

Measured properties of jatropha, karanj and fossil diesel



1: Engine; 2: Dynamometer; 3: Smoke meter; 4: Exhaust analyser; 5: Exhaust data acquisition; 6: Exhaust gas discharge; 7: Dynamometer controller; 8: NI data acquisition for temperature; 9, 10: Kistler combustion analyser; 11: Jatropha/Karanj tank; 12: Diesel tank; 13: 3-way valve; 14: Vent screw; 15: Additional fuel filter; 16: Valve; 17: Fuel measurement; 18: Cold water flow to HX; 19: HX to cool jacket water; 20: Plant oil preheater; 21: Crank angle encoder; 22: Cylinder pressure transducer; 23, 24: Amplifier; 25: Injection pressure sensor; 26: NI DAQ; 27: Thermocouple

Fig. 1. Schematic diagram of the engine modifications and experimental rig



Fig. 2. Measurement of cylinder pressure, and fuel line injection temperature and pressure



Fig. 3. Viscosity vs. temperature of jatropha, karanj and fossil diesel



Fig. 4. Preheating temperatures of neat jatropha and karanj oils at different engine load conditions



Fig. 5c - Brake thermal efficiency vs. load

Fig. 5. Brake specific fuel consumption and brake thermal efficiency vs. engine load



Fig. 6. Comparison of exhaust emissions at different engine load conditions: (a) CO emission, (b) CO_2 emission, (c) NO_x emission, and (d) O_2 emission



7c-Lambda values vs. load

Fig. 7. Exhaust gas temperature, smoke opacity and lambda values vs. engine load



Fig. 8. Cylinder pressure vs. crank angle at different engine loads



Fig. 9. Integral heat release vs. crank angle for different fuels and at different loads



Fig. 10. Maximum fuel injection pressure vs. engine load



(a) 90% combustion



(b) Total combustion duration

Fig. 11. Combustion of fuels at various engine load conditions



Fig. 12. Coke deposits in the cylinder head and intake/exhaust valves