

## Article

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Tew, S.T., Soon, Jan Mei, Benjakul, S, Prodran, T, Vittayanont, M and Tongnuanchan, P

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## Nutrition & Food Science

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Sue Teng Tew Jan Mei Soon Soottawat Benjakul Thummanoon Prodran Manee Vittayanont Phakawat Tongnuanchan

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# Development of gelatine-based bio-film from chicken feet incorporated with sugarcane bagasse

## Introduction

Plastics are used worldwide in everyday lives and in different forms such as food packaging, spoons, bottles, pens, shopping plastic bags, chairs and containers.

Recycling rates for most plastic packaging are low although recyclable packaging materials has increased (Hopewell *et al.*, 2009). Plastics take a long period of time for complete degradation as they do not degrade naturally to a large degree when released into the environment due to the many polymers that are exceptionally stable and durable (Webb *et al.*, 2012). In order to substitute these plastics, biodegradable plastics have been developed with the same function that are comparable to traditional petrochemical-based plastics for packaging applications (Song *et al.*, 2009). Biodegradable plastics are plastics in which the degradation mechanism is characterized by the full breakdown of the organic chemical compound by micro-organisms into water, carbon dioxide, methane, biomass and inorganic compounds under aerobic or anaerobic conditions and the action of living organisms (Deconinck and Wilde, 2013). The objectives in the development of biodegradable plastics are to utilize renewable and sustainable sources of raw materials by using crops instead of crude oil and to approach integrated waste management to reduce landfill (Davis and Song, 2006).

In the case of food packaging, edible film from natural polymer is important as an alternative to replace synthetic polymer as it can help to enhance food quality by acting as moisture, gas, aroma and lipid barriers as well as acting as a protection to a food product after the primary package is opened (Rattaya *et al.*, 2009). Generally, edible films are thin, continuous layer of edible material which is renewable such as proteins, lipids and carbohydrates (Jongjareonrak *et al.*, 2006). Examples of edible protein films had been developed from whey (Ramos *et al.*, 2012, 2013), soy (Otoni *et al.*, 2016) and sesame (Sharma and Singh, 2016), gelatin films from skin of cuttlefish (Jridi *et al.*, 2013) and fish (Kaewruang *et al.*, 2013; Nikoo *et al.*, 2014; Weng *et al.*, 2014) while other materials focused on polysaccharides such as carrageenan (Soni *et al.*, 2016), cassava starch (Bergo *et al.*, 2008) and methylcellulose (Rubilar *et al.*, 2015).

32           Although protein-based films have good gas barrier characteristics compared to  
33 synthetic films, they have poor mechanical properties and high water vapor permeability  
34 which are the main drawbacks of protein films acting as a packaging material (Hoque *et al.*,  
35 2011). Moreover, the main sources of commercial gelatin production are from skin  
36 and bones of swine and cattle but the usage of swine skin and bone is considered *haram*  
37 (unlawful) for Muslim and Judaism and beef gelatin is only acceptable if prepared  
38 according to religious requirements (Badii and Howell, 2006). There is also risk of  
39 contamination with bovine spongiform encephalopathy (BSE) if infected cattle skin and  
40 bones were used (Grommuang *et al.*, 2006). Properties of fish gelatine from skins of Nile  
41 Tilapia *Oreochromis niloticus*) and channel catfish (*Ictalurus punctatus*) (Zhang *et al.*,  
42 2016), tilapia (*Tilapia zillii*) scales (Weng *et al.*, 2014), unicorn leatherjacket (*Aluterus*  
43 *monoceros*) (Kaewruang *et al.*, 2013) and Amur sturgeon (*Acipenser schrenckii*) (Nikoo  
44 *et al.*, 2014) had been carried out as the demand for non-mammalian gelatine increases.  
45 Fish gelatine is acceptable for Islam but persisting residual odour in fish gelatin can cause  
46 problems if the film is intended for use in mildly flavoured products (Rafieian *et al.*, 2015;  
47 Sae-Leaw and Benjakul, 2015). Chicken by-products such as chicken deboner residue  
48 (CDR) (Rafieian *et al.*, 2015), chicken feet to replace cowhides for jokpyun (traditional  
49 Korean gel-type food) (Jun *et al.*, 2000), chicken bones (Lim *et al.*, 2010), chicken skin  
50 (Sarbon *et al.*, 2013)

51           Sugarcane bagasse is available abundantly in sugar production and beverage  
52 industry and 1 tonne of sugarcane produces 280 kg of bagasse (Cerqueira *et al.*, 2007).  
53 Although once considered a low value agricultural residue, sugarcane bagasse can be  
54 potentially utilized for its cellulose which contributes to stiffness (Afra *et al.*, 2013),  
55 reinforcing potential (Abraham *et al.*, 2011) and biodegradability (Chen *et al.*, 2011).  
56 Sugarcane produces maximum surplus residue (Hiloidhari *et al.*, 2014) and provides 40-  
57 50% cellulose content (Sun *et al.*, 2004). To our knowledge, this is the first study that  
58 incorporates hydrolyzed sugarcane bagasse to study the potential mechanical benefits in  
59 protein based bio-film. The aim of this work was to analyse the effect of hydrolyzed  
60 sugarcane bagasse incorporation on mechanical and water vapor barrier properties of bio-  
61 film derived from chicken feet extract to utilize agricultural by-products as potential food  
62 packaging materials.

63

## 64 **Materials and methods**

### 65 *Chemicals*

66 Phosphoric acid and hydrogen peroxide were purchased from Fisher Scientific  
67 (Loughborough, UK) while sodium hydroxide and sulfuric acid were purchased from  
68 RCI Labscan Limited (Bangkok, Thailand) and glycerol from Quality Reagent Chemical  
69 (QReC™) (New Zealand). All chemicals were of analytical grade.

70

### 71 *Raw materials*

72 Chicken feet produced by Sahafarm Co., LTD were purchased at Tesco Lotus, Hat Yai,  
73 Thailand. Sugarcane bagasse was obtained from the wet market in Songkla, Thailand.

74

### 75 *Extraction of gelatine from chicken feet sample*

76 Preparation of chicken feet sample was carried out according to Grommuang *et al.* (2006).  
77 The chicken feet were ground with meat grinder (4 mm mesh size) and washed several  
78 times with cold water. The ground chicken feet were then centrifuged at room  
79 temperature for 5 minutes and stored at -20°C for further use. Extraction of gelatine from  
80 chicken feet sample was done by pre-swelling the ground chicken feet first with 2.14%  
81 phosphoric acid at 20°C for 48 hours as described by Grommuang *et al.* (2006). It was  
82 then washed thoroughly with tap water until the pH reached 6 - 7. Extraction was done  
83 with distilled water for 5 hours at 70°C in water bath. The extract was concentrated at  
84 70°C with vacuum evaporator, chilled to set gel, ground and air dried overnight at 40°C  
85 before further grinding if necessary. Kjeldhal method (AOAC, 2000) was used to  
86 determine the protein content in the extracted chicken feet gelatine. The protein content  
87 of the extracted chicken feet gelatine was carried out in triplicate and the average value  
88 was calculated.

89

### 90 *Purification of cellulose from SCB*

91 Purification of cellulose from SCB was carried out as described by Teixeira *et al.* (2011)  
92 with slight modification. Oven dried SCB was blended to pass through 40 mesh screen.  
93 Five grams of dried SCB was then digested with 6% NaOH solution for 4 hours in 60°C

94 water bath. It was then stirred with magnetic stirrer while 100 mL hydrogen peroxide  
95 solution (11% v/v) was added slowly to the flask and stirred vigorously for 90 mins. The  
96 SCB was filtered and washed with distilled water until neutral pH.

97

#### 98 *Preparation of hydrolyzed SCB*

99 Hydrolyzed SCB was prepared according to Teixeira *et al.* (2011) with slight  
100 modification. SCB was dispersed in 100 mL of 6M H<sub>2</sub>SO<sub>4</sub> at 50°C. It was stirred  
101 vigorously for 2 hrs 500 ml cold distilled water (4°C) was added to stop the reaction. The  
102 pH of the solution was adjusted to pH 6 - 7 through dialysis in tap water with cellulose  
103 membrane before storing the suspension in refrigerator. Moisture content of the  
104 hydrolyzed SCB suspension was carried out in triplicates (AOAC, 2000).

105

#### 106 *Preparation of gelatine film with different percentage of glycerol*

107 Film forming solution (FFS) was prepared as described by Tongnuanchan *et al.* (2012,  
108 2013) Gelatine powder was mixed with distilled water to obtain the protein concentration  
109 of 3.5% (w/v). The mixture was heated at 70°C until completely dissolved. . Glycerol  
110 which acts as plasticizer was added at concentrations of 25% and 35% (w/w) of protein  
111 content. The film was then prepared by casting 4.0 g FFS onto a rimmed silicone resin  
112 plate (50 x 50 mm<sup>2</sup>) and air-blown for 12 hrs at 25°C. The film was further dried at 25°C  
113 and 50±5% relative humidity for 24 h in an environmental chamber (WTB Binder,  
114 Tuttlingen, Germany) (Prodpran *et al.*, 2007). The resulting films were peeled off  
115 manually and subjected to analyses.

116

#### 117 *Preparation of gelatine film incorporated with different weight percentage of hydrolyzed* 118 *SCB*

119 To incorporate the hydrolyzed SCB, modification of methods by Nagarajan *et al.* (2014)  
120 and Gilfillan *et al.* (2014) were applied. Gelatine powder was mixed with distilled water  
121 to obtain the protein concentration of 3.5% (w/v). The mixture was heated at 70°C until  
122 completely dissolved. Then, glycerol was added at concentrations of 35% (w/w) of  
123 protein content as a plasticizer. Hydrolyzed SCB suspension of 0.00, 0.131, 0.262, 0.393  
124 and 0.524 g (dry basis) to produce 0, 2.5, 5, 7.5 and 10% (w/w, on dry protein basis) were

125 prepared by homogenizing for 20 secs at 11,000 rpm (IKA Labortechnik homogenizer,  
126 Selangor, Malaysia). The hydrolyzed SCB suspensions were added to the film forming  
127 solution slowly and the mixtures were homogenized for another 1 min at 11,000 rpm. The  
128 final volume of the film forming suspensions were made up to 150 ml and were  
129 sonicated for 30 mins using sonicating bath (Elmasonic S 30 H, Singen, Germany) and  
130 stirred gently for 30 mins at room temperature in order to obtain a homogeneous  
131 suspension. Before casting the film forming suspensions, they were degassed for 10 mins  
132 using sonicating bath. The film was then prepared by casting 4.0 g film forming  
133 suspension onto a rimmed silicone resin plate (50 x 50 mm<sup>2</sup>) and air-blown for 12 hrs at  
134 room temperature before drying in an environmental chamber (WTB Binder, Tuttlingen,  
135 Germany) for 24 hrs at 25°C and 50 ± 5% RH. The resulting films were peeled off  
136 manually and subjected to analyses. Gelatine film without SCB (control) is named SCB 0  
137 and those incorporated with 2.5, 5, 7.5 and 10% SCB were named SCB 2.5, SCB 5.0,  
138 SCB 7.5 and SCB 10.0 respectively. Prior to testing, film samples were conditioned for  
139 48 h at 25°C and 50 ± 5% RH (Ahmad et al., 2012).

140

#### 141 *Determination of film properties*

##### 142 *Film thickness*

143 The thickness of films were measured using a micrometer (Mitutoyo, Model ID-C112PM,  
144 Serial No. 00320, Mitutoyo Corp., Kawasaki-shi, Japan) as described by Fazilah and  
145 Maizura (2010). Measurements were taken at fifteen random positions around each film  
146 of 10 film samples and average value was calculated.

147

##### 148 *Mechanical properties*

149 Tensile strength (TS) and elongation at break (EAB) of the films were determined as  
150 described by Iwata *et al.* (2000) using Universal Testing Machine (Lloyd Instruments,  
151 Hamsphire, UK). Five film samples (2 x 5 cm<sup>2</sup>) were first conditioned for 48 hrs at 25°C  
152 and 50 ± 5% RH before testing. The film samples were clamped under tensile loading  
153 using a 100 N load cell with initial grip length of 3 cm and cross-head speed at 30  
154 mm/min. Tensile strength (MPa) was calculated by dividing the maximum load (N)  
155 needed to pull the sample film apart by the cross-sectional area of the sample. Percentage

156 of elongation at break was calculated by the film elongation at the moment of rupture  
 157 divided with the initial grip length of samples multiplied by 100%.

158

#### 159 *Water Vapor Permeability (WVP)*

160 WVP of the films were determined using American Society for Testing and Materials  
 161 (ASTM) method (ASTM, 2004) as described by Rattaya *et al.* (2009). The film was  
 162 sealed on an aluminum permeation cup containing dried silica gel (0% RH) with silicone  
 163 vacuum grease and a rubber gasket was used to hold the film in place. The cups were  
 164 placed in a desiccator containing distilled water at 30°C. The aluminum permeation cups  
 165 were weighed at every 1 hr intervals for 8 hrs period. WVP of film was calculated as  
 166 follows:

167

$$168 \text{ WVP ( gm}^{-1} \text{ s}^{-1} \text{ Pa}^{-1} \text{ )} = wxA^{-1} t^{-1} ( P_2 - P_1 )^{-1};$$

169

170 where,  $w$  = weight gain of the cup (g);  $x$  = film thickness (m);  $A$  = area of exposed film  
 171 ( $\text{m}^2$ );  $t$  = time of gain (s), and  $( P_2 - P_1 )$  = vapor pressure difference across the film (Pa).

172

#### 173 *Color measurement*

174 Color of each different film was determined using a CIE colorimeter (Hunter Associates  
 175 Laboratory Inc., USA). Color of the film is expressed as  $L^*$  - (lightness/brightness),  $a^*$  -  
 176 (redness/greenness) and  $b^*$  - (yellowness/blueness) values. The total difference in color  
 177 ( $\Delta E^*$ ) was calculated according to the equation of Gennadios *et al.* (1996) as follows:

178

$$179 \Delta E^* = \sqrt{(\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2}$$

180 where,  $\Delta L^*$ ,  $\Delta a^*$ , and  $\Delta b^*$  are the differences between the color parameter of the film  
 181 samples and the color parameter of the white standard,

182 ( $L^* = 93.63$ ,  $a^* = -0.88$ , and  $b^* = 0.33$ ) when test done on films with different  
 183 glycerol

184 percentage, and

185 ( $L^* = 93.59$ ,  $a^* = -0.95$ , and  $b^* = 0.44$ ) when test done on films incorporated with  
 186 hydrolyzed sugarcane bagasse.



187

188 *Light transmittance and transparency value*

189 Light transmittance of the films was measured in ultraviolet (UV) and visible range from  
190 200 nm to 800 nm using a UV-Visible spectrophotometer (model UV-1800, Shimadzu,  
191 Kyoto, Japan) (Shiku *et al.*, 2004). The transparency value of film sample was calculated  
192 based on the equation of Han and Floros (1997) as shown below :

193

194 Transparency value =  $\frac{(-\log T_{600})}{x}$

195 where,  $T_{600}$  = the fractional transmittance at 600 nm, and  $x$  = the film thickness (mm). The  
196 higher the transparency value indicates the lower transparency of film.

197

198 *Scanning Electron Microscopy (SEM)*

199 Microstructure of surface and cross-section of film samples were determined as described  
200 by Tongnuanchan *et al.* (2013) using scanning electron microscopy (SEM) (Quanta 400,  
201 FEI, Eindhoven, the Netherlands). Film samples were fractured under liquid nitrogen  
202 before visualization for cross-section. The film samples were mounted on bronze stub  
203 and sputtered with gold using Sputter coater (SPI-Module, West-Chester, PA, USA) in  
204 order to make the sample conductive. Photographs were taken at an acceleration voltage  
205 of 15 kV.

206

207 *Statistical analysis*

208 Data were subjected to analysis of variance (ANOVA) and mean comparisons were  
209 carried out by Duncan's multiple range test. For pair comparison, T-test was used (Steel  
210 and Torrie, 1980). Results are presented as mean  $\pm$  standard deviation and the probability  
211 value of  $p < 0.05$  is considered as significant. Where relevant, an asterisk (\*) is used to  
212 indicate which values are presented as mean  $\pm$  standard deviation. Statistical analysis  
213 was performed using the Statistical Package for Social Sciences version 22.0 (IBM Corp.  
214 Released 2013. IBM SPSS Statistics for Windows, NY).

215

216 **Note:**

217 Films from gelatine with different percentage of glycerol were first produced, tested and  
218 analyzed in order to determine which film is suitable to proceed to form films  
219 incorporated with different weight percentage of SCB.

220

## 221 **Results and discussion**

### 222 *Protein content in extracted chicken feet gelatine and moisture content of hydrolyzed* 223 *SCB suspension*

224 The chicken feet gelatine contains about 74.22 % of protein. The moisture content of the  
225 hydrolyzed SCB suspension was 98.63%. The dry basis of the SCB was calculated by  
226 subtracting 98.63% with 100 % which resulted in 1.37 g. This means that there was 1.37  
227 g of SCB for every 100 ml of the hydrolyzed SCB suspension.

228

229

### 230 *Properties of gelatine film with different percentage of glycerol*

#### 231 *Thickness*

232 The thickness of films with different percentage of glycerol is as shown in Table 1. It is  
233 not significantly different) between the films containing 25% (0.058 mm) and 35%  
234 glycerol (0.060 mm). The glycerol did not affect the film thickness as glycerol was  
235 dissolved with the gelatine during preparation of FFS. Negligible differences in thickness  
236 of gelatine-based films with different levels of glycerol were also mentioned by Vanin *et*  
237 *al.*(2005), Kokoszka *et al.* (2010), Tongnuanchan *et al.*(2012) and Chamnanvatkatit *et al.*  
238 (2014).

239

240 **Table 1.** Properties of films from chicken feet gelatine with different percentage of  
241 glycerol

242

#### 243 *Mechanical Properties*

244 TS and EAB of the film with different percentage of glycerol are as shown in Table 1.  
245 There is significant difference ( $p < 0.05$ ) for both TS and EAB. It can be seen that TS of  
246 the film decreased from 44.86 MPa to 34.20 MPa when the glycerol percentage increased  
247 10%. As for the EAB of the film, the value increased about two-fold; from 15.99% to

248 33.30%. Glycerol concentration affects the film properties by improving the film  
249 extensibility and reducing its resistance as reported by Jouki *et al.* (2013). Glycerol  
250 improves the flexibility of gelatine-based film but decreases its stiffness.

251 Chamnanvatkatit *et al.* (2014) stated that glycerol gives plasticizing effect  
252 because it decrease the inter- and intra molecular attractive forces resulting TS to  
253 decrease and EAB to increase with the increasing of glycerol concentration. Plasticizer  
254 can be easily inserted between polymer chains to produce a “cross-linker” effect that  
255 decreases the free volume of the polymer and at the same time improves the extensibility  
256 of the films and diminishes mechanical strength (Jouki *et al.*, 2013). Other studies  
257 showed similar result concerning the effect of glycerol as plasticizer on protein-based  
258 films which include muscle proteins of Thai tilapia (Sobral *et al.*, 2005), whey protein  
259 (Ramos *et al.*, 2013) and bovine gelatine (Chamnanvatkatit *et al.*, 2014).

260

#### 261 *Water Vapor Permeability (WVP)*

262 WVP of the film prepared from chicken feet gelatine with 25% and 35% glycerol are  
263 shown in Table 1. There is no significant difference between the gelatine film with 25%  
264 and 35% glycerol. The WVP for 25% glycerol gelatine film is  $2.04 \times 10^{-11} \text{ gm}^{-1} \text{ s}^{-1} \text{ Pa}^{-1}$  and  
265  $2.14 \times 10^{-11} \text{ gm}^{-1} \text{ s}^{-1} \text{ Pa}^{-1}$  for 35% glycerol gelatine film. WVP increases as the glycerol  
266 percentage increases. This is due to lower water barrier in higher content of glycerol.

267 Glycerol enhances the water vapor permeability as it modifies the molecular  
268 organization of the protein network and increases the free volume leading to lesser dense  
269 network hence, films are permeable to water as it ease the water diffusion (Al-Hassan and  
270 Norziah, 2012). Arvanitoyannis *et al.* (1998) stated that the water vapor transfer rate  
271 increases proportionally with the increasing of the total plasticizer content (water and  
272 polyols) in the polymer matrix.

273 Chamnanvatkatit *et al.* (2014) with similar results also stated that glycerol is  
274 hydrophilic in nature which led to the hygroscopic characteristics of the films thus  
275 increases the moisture content of the film as well as the WVP of the film.

276

277 *Color measurement*

278 Differences in color between the gelatine film with 25% and 35% glycerol are presented  
 279 in Table 2. As mentioned,  $L^*$  is the lightness/brightness and  $a^*$  is redness/greenness  
 280 whereas  $b^*$  is the yellowness/blueness values. The values of  $L^*$ ,  $a^*$  and  $\Delta E^*$  have  
 281 significant difference ( $p < 0.05$ ) between the films from gelatine with 25% and 35%  
 282 glycerol. However, the  $b^*$  values showed no significant difference between the two types  
 283 of films. Based on the study carried out by Chamnanvatkatit *et al.* (2014), addition of  
 284 different concentrations of glycerol to bovine protein films does not impact the color of  
 285 the resulting films. However, Jouki *et al.* (2013) reported otherwise, when different  
 286 glycerol concentration were added to cress seed gum films. All of the color parameters  
 287 except  $a$ -value of the films were significantly changed when glycerol concentration  
 288 increased.

289

290 **Table 2.** Film colors made from chicken feet gelatine with different percentage of  
 291 glycerol

292

293 *Light transmittance and transparency value*

294 The light transmission in the UV range (200-280 nm) for film with 25% glycerol is from  
 295 0.02% to 21.54% while film with 35% glycerol is from 0.03% to 19.25%. As for visible  
 296 range (350-800 nm), the light transmittance ranges from 72.48% to 87.58% and 66.75%  
 297 to 85.62% for 25% and 35% glycerol gelatine-based film respectively (Table 3). This  
 298 conveys that there is a slight decrease in light transmission with the increase of  
 299 percentage of glycerol.

300

301 **Table 3.** Light transmittance and transparency values of films from chicken feet gelatine  
 302 with different percentage of glycerol

303

304 In addition, the increased in percentage of glycerol had no significant differences  
 305 on the transparency value between the two types of film. The transparency value differs  
 306 by 0.02 indicating the increased in glycerol percentage do not affect the transparency  
 307 value of the films. The resulting gelatine films were transparent and also clear which is

308 suitable for use as see-through packaging. Gelatine has low content of tyrosine and  
309 phenylalanine; aromatic amino acids that are sensitive to chromophores which absorb  
310 light at wavelength below 300 nm (Li *et al.*, 2006). The aromatic amino acids are  
311 important as an UV barrier property of protein films as gelatine film without glycerol has  
312 higher barrier for light transmission and UV range compared to film added with glycerol.

313

#### 314 *Analysis*

315 Based on the results for thickness, mechanical properties, water vapor permeability, color  
316 and light transmittance as well as transparency value tests of the films from chicken feet  
317 gelatine with different percentage of glycerol, film with 35% glycerol was chosen to be  
318 incorporated with different weight percentage of hydrolyzed SCB. Film with 35% of  
319 glycerol has lower TS but higher EAB. By incorporating hydrolyzed SCB, it was hoped  
320 that the TS increases and WVP of the film can further be lowered.

321

#### 322 *Properties of films from chicken feet gelatine incorporated with different percentage of* 323 *dry weight SCB*

##### 324 *Thickness*

325 Thickness of the film incorporated with different percentage of dry weight SCB is shown  
326 in Table 3. Generally, thickness of a film increases as the amount of weight percentage of  
327 SCB increases ( $p < 0.05$ ). The hydrolyzed SCB is likely distributed on the gelatine film  
328 and increase the thickness of the film. However, the thickness of the film is the same  
329 between the control film with 0% and 2.5% of dry weight SCB. There is no effect on the  
330 thickness of the film as the amount of SCB is not significant.

331

332 **Table 4.** Properties of films from chicken feet gelatine incorporated with different  
333 percentage of dry weight SCB

334

##### 335 *Mechanical Properties*

336 The mechanical properties of films incorporated with different percentage of dry weight  
337 SCB are presented in Table 4. Incorporating SCB in the gelatine film is supposed to  
338 increase the TS of the film. However, as shown in Table 4, the TS increased slightly from

339 22.50 MPa to 23.07 MPa for the film with 0 wt % and 5.0 wt % SCB respectively. The  
340 TS then decreased to 20.88 MPa and 19.76 MPa with 7.5 wt % and 10.0 wt % SCB  
341 incorporated respectively. This is in agreement with Gilfillan *et al.* (2012) and  
342 Prachayawarakorn *et al.* (2010) where fiber overloading resulted in decreasing tensile  
343 strength.

344 EAB of films decreases as the amount of percentage of dry weight SCB increases  
345 as shown in Table 4. The EAB of film decreased steadily from 59.97% to 24.82% for the  
346 film with 0 wt % and 10.0 wt % of SCB respectively. Slavutsky and Bertuzzi (2014)  
347 reinforced starch films with cellulose nanocrystals obtained from sugarcane bagasse and  
348 stated high value of TS as the sugarcane bagasse was dispersed properly in the matrix  
349 structure. In addition, EAB value decreases due to the rigid nature of the sugarcane  
350 bagasse. Another similar study was conducted by Gilfillan *et al.* (2014) where starch was  
351 incorporated with sugarcane bagasse nanofibres. The TS doubled but started to decrease  
352 at fibre loadings above 10 wt % while EAB decreased by up to 70% compared to film  
353 with no nanofibres.

354

#### 355 *Water Vapor Permeability (WVP)*

356 WVP of films from chicken feet gelatine incorporated with different weight percentage of  
357 SCB showed significant differences ( $p < 0.05$ ) between the films (Table 4). WVP of films  
358 decreased with the increasing levels of weight percentage of SCB incorporated in the film.  
359 The WVP of the films decreased from  $2.18 \times 10^{-11} \text{ gm}^{-1} \text{ s}^{-1} \text{ Pa}^{-1}$  to  $1.56 \times 10^{-11} \text{ gm}^{-1} \text{ s}^{-1} \text{ Pa}^{-1}$   
360 which is the SCB 0 (control) to SCB 10.0 (10 wt % SCB), which are the highest and  
361 lowest WVP of the films respectively. However, there is no significant difference  
362 between SCB 7.5 and SCB 10.0. This may due to the uneven dispersion of SCB on the  
363 film samples for 7.5 wt % and 10.0 wt %.

364 A high WVP of film is not desirable due to its usage and performance (Pereda *et*  
365 *al.*, 2011). From the results in this experiment, addition of hydrolyzed SCB improved the  
366 water vapor barrier properties of the film slightly. Rawdkuen *et al.* (2012) reported  
367 similar results by adding catechin-lysozyme combination (CLC) in fish gelatine film. The  
368 barrier properties improved as the moisture transfer between the food and the surrounding  
369 atmosphere is lowered when the film was applied to heterogeneous food product. Ahmad

370 *et al.* (2012) stated that the water vapor transfer process in films depends on the  
371 hydrophilic-hydrophobic ratio of the film constituents. In addition, film thickness also  
372 influences the water vapor permeability as thicker film can absorb more water from the  
373 environment (Rawdkuen *et al.*, 2010). In order to utilise the gelatine based bio film  
374 incorporated with SCB as a potential food packaging film, resistance of the film to water  
375 is desirable if the film is to be used for the preservation of intermediate or high moisture  
376 foods (Ozdemi and Floros, 2008). Films with good solubility had been proposed as  
377 packaging material for instant noodle seasoning bags and instant beverages or as casing  
378 for sausages, biscuits and candy (Wan *et al.*, 2015). The water solubility and swelling of  
379 the bio film should be determined in future studies.

380

#### 381 *Color measurement*

382 The color properties,  $L^*$  (lightness/brightness),  $a^*$  (redness/greenness) and  $b^*$   
383 (yellowness/blueness) values of the films from chicken feet gelatine incorporated with  
384 different levels of weight percentage of SCB are shown in Table 5. It can be concluded  
385 that all the color parameters were affected by the amount of weight percentage of SCB  
386 being incorporated in the film. The value increases proportionally with the weight  
387 percentage of SCB and there is significant difference ( $p < 0.05$ ) for the three parameters.  
388 The total color differences ( $\Delta E^*$ ) also showed significant difference ( $p < 0.05$ ). Control  
389 (SCB 0) showed the lowest value while the highest weight percentage (SCB 10.0)  
390 showed the highest value with 3.48 and 3.61 respectively.

391

392 **Table 5.** Film colors of chicken feet gelatine incorporated with different percentage of  
 393 dry weight SCB

394

395 *Light transmittance and transparency value*

396 Generally, films often exhibit lower light transmission in the UV range than in the visible  
 397 range (Rawdkuen *et al.*, 2012). Transmission of UV light of the film from control film to  
 398 incorporation of sugarcane bagasse (SCB 10.0) in chicken feet gelatine film at 280 nm  
 399 decreased from 22.20 to 9.95%. Hence, the films are successful in preventing the UV  
 400 light and possibly retard lipid oxidation induced by the UV light. The light transmittance  
 401 of the films at different wavelengths decreases as the weight percentage of the SCB  
 402 incorporated in the film increases (Table 6).

403

404 **Table 6.** Light transmittance and transparency of films from chicken feet gelatine  
 405 incorporated with different percentage of dry weight SCB

406

407 There is significant difference ( $p < 0.05$ ) on the transparency among all the films  
 408 with different weight percentage of SCB. The transparency value increases as the amount  
 409 of weight percentage of SCB incorporated increases. Transparency value increased from  
 410 0.99 (SCB 0) to 2.37 (SCB 10.0) with higher transparency value indicating that the films  
 411 have lower transparency. The increase of transparency value is most probably due to the  
 412 hydrolyzed SCB incorporated as the hydrolyzed SCB is solid and not transparent which  
 413 made the film not entirely clear.

414

415 *Scanning Electron Microscopy (SEM)*

416 SEM micrographs of the surface and cross-section of films from chicken feet gelatine  
 417 incorporated with different levels of weight percentage of hydrolyzed SCB are illustrated  
 418 in Figure 1.

419

420 **Figure 1.** Scanning Electron Microscopy micrographs of surface (magnification: 500x)  
 421 and cross section (magnification: 1800x) of films from chicken feet gelatine incorporated  
 422 with different levels of weight percentage of SCB. The SCB 0 which is the control film



423 showed smooth and homogeneous surface. The cross-section of the control film also  
424 showed smooth surface. As the weight percentage of hydrolyzed SCB increases, the  
425 surface of the films showed increment in white spots. The white spots are believed to be  
426 the hydrolyzed SCB.

427

428 The SCB 0 which is the control film showed smooth and homogeneous surface. As the  
429 weight percentage of hydrolyzed SCB increases, the surface of the films showed  
430 increment in white spots. The white spots are believed to be the hydrolyzed SCB. The  
431 cross-section of control film also showed smooth surface. The surface became rougher  
432 with the increase of weight percentage of SCB. However, through the micrographs, it can  
433 be deduced that the hydrolyzed SCB did not form a strong matrix with the protein matrix  
434 of gelatine. There is a weak bond between the SCB and the gelatine film and the  
435 hydrolyzed SCB merely formed a layer on top of the gelatine film. It can be seen that the  
436 hydrolyzed SCB did not disperse homogeneously on the gelatine film but agglomerate  
437 instead. Hence, further treatment of the SCB should be applied for a better dispersion of  
438 the SCB on the gelatine-based film.

439 Gilfillan *et al.* (2014) reported that the sugarcane nanofibres are well attached to  
440 the starch matrix based on the SEM micrographs of the composite from starch with SCB  
441 nanofibres. The SEM micrograph of starch film reinforced with cellulose nanocrystals  
442 obtained from SCB showed that the dispersion of the cellulose nanocrystals are  
443 homogeneous within the polymer matrix (Slavutsky and Bertuzzi, 2014). In this study,  
444 the SCB used were chemically hydrolyzed. The SCB particle size is still noticeably large  
445 (Figure 1). This may have affected the dispersion of the SCB on the gelatine film. The  
446 structure of cellulose fibers can be damaged by excessive hydrolysis (Gilfillan *et al.*,  
447 2014). It is suggested that further treatments be carried out on hydrolyzed SCB. Pre-  
448 treatment (Salehudin *et al.*, 2013) and combination of mechanical refining and enzymatic  
449 treatment were found to produce homogenous nanofibrils from sugarcane bagasse  
450 (Santucci *et al.*, 2016).

451

452 **Conclusion**

453 A higher percentage of glycerol used in the gelatine-based film, resulted in lower TS and  
454 higher EAB. Film containing 35% glycerol in gelatine extracted from chicken feet were  
455 further incorporated with different weight percentage (0, 2.5, 5.0, 7.5 and 10.0 wt %) of  
456 SCB. Although the mechanical strength of the film could not be improved by  
457 incorporating SCB, there was only slight improvement in the WVP barrier properties. As  
458 the weight percentage of SCB increases, the WVP of the film decreases. In addition, film  
459 from gelatine extracted from chicken feet incorporated with 5.0 wt % of SCB has the best  
460 properties when all the tests were taken into consideration. The thickness, color and  
461 transparency value of the film with 5.0 wt % of SCB were similar to the control film.  
462 However, the TS of SCB 5.0 film is increased and the WVP is lowered slightly. This  
463 limits the application of the film as biomaterial and further research to treat the  
464 hydrolysed SCB is recommended. The bio-film developed in this study incorporates  
465 sugarcane bagasse into the film derived from chicken feet and demonstrated an increment  
466 in tensile strength and reduction of water vapor permeability. This study is of value to  
467 food practitioners looking into utilising agricultural wastes (e.g. animal by-product and  
468 sugarcane bagasse).

469

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683

Table 1 Properties of films from chicken feet gelatine with different percentage of glycerol

<b>Glycerol (%)</b>	<b>Thickness (mm)</b>	<b>TS (MPa)</b>	<b>EAB (%)</b>	<b>WVP (<math>\times 10^{-11} \text{gm}^{-1} \text{s}^{-1} \text{Pa}^{-1}</math>)</b>
25%	$0.058 \pm 0.003^a$	$44.86 \pm 1.66^a$	$15.99 \pm 6.24^a$	$2.04 \pm 0.29^a$
35%	$0.060 \pm 0.003^a$	$34.20 \pm 0.97^b$	$33.30 \pm 6.79^b$	$2.14 \pm 0.11^a$

Results are presented as mean  $\pm$  sd. Different superscript letters in the same column indicate significant difference by independent samples T-test ( $p < 0.05$ ).

TS - Tensile strength

EAB - Elongation at break

WVP - Water vapor permeability



Table 2 Film colors made from chicken feet gelatine with different percentage of glycerol

Glycerol (%)	$L^*$	$a^*$	$b^*$	$\Delta E^*$
25%	90.77 ± 0.06 <sup>a</sup>	-1.30 ± 0.04 <sup>a</sup>	3.01 ± 0.27 <sup>a</sup>	3.94 ± 0.14 <sup>a</sup>
35%	91.29 ± 0.10 <sup>b</sup>	-1.40 ± 0.03 <sup>b</sup>	3.18 ± 0.07 <sup>a</sup>	3.73 ± 0.08 <sup>b</sup>

Results are presented as mean ± sd. Different superscript letters in the same column indicate significant difference by independent samples T-test ( $p < 0.05$ )

Table 3 Light transmittance and transparency values of films from chicken feet gelatine with different percentage of glycerol.

Glycerol (%)	Light transmittance (%) at different wavelength (nm)							Transparency value*	
	200	280	350	400	500	600	700		800
25 %	0.02	21.54	72.48	79.94	84.30	85.96	86.89	87.58	1.08 ± 0.05 <sup>a</sup>
35 %	0.03	19.25	66.75	74.95	80.54	83.02	84.53	85.62	1.10 ± 0.14 <sup>a</sup>

\*Mean ± SD

Different superscript letters in the same column indicate significant difference by independent samples T-test ( $p < 0.05$ ).

Table 4 Properties of films from chicken feet gelatine incorporated with different percentage of dry weight SCB

<b>Film sample</b>	<b>Thickness (mm)</b>	<b>TS (MPa)</b>	<b>EAB (%)</b>	<b>WVP (<math>\times 10^{-11} \text{gm}^{-1} \text{s}^{-1} \text{Pa}^{-1}</math>)</b>
SCB 0	$0.066 \pm 0.002^a$	$22.50 \pm 1.97^{bc}$	$59.97 \pm 5.83^c$	$2.18 \pm 0.08^d$
SCB 2.5	$0.066 \pm 0.003^a$	$22.68 \pm 1.14^c$	$41.67 \pm 4.95^b$	$2.06 \pm 0.04^c$
SCB 5.0	$0.073 \pm 0.004^b$	$23.07 \pm 0.67^c$	$35.75 \pm 3.59^b$	$1.85 \pm 0.08^b$
SCB 7.5	$0.085 \pm 0.004^c$	$20.88 \pm 1.36^{ab}$	$27.99 \pm 3.46^a$	$1.61 \pm 0.07^a$
SCB 10.0	$0.087 \pm 0.007^c$	$19.76 \pm 0.67^a$	$24.82 \pm 4.50^a$	$1.56 \pm 0.05^a$

Results are presented as mean  $\pm$  sd. Different superscript letters in the same column indicate significant difference by Duncan's multiple range tests ( $p < 0.05$ ).

TS - Tensile strength

EAB - Elongation at break

WVP - Water vapour permeability

SCB: Sugarcane bagasse

Table 5 Film colors of chicken feet gelatine incorporated with different percentage of dry weight SCB

<b>Film sample</b>	<b><i>L</i>*</b>	<b><i>a</i>*</b>	<b><i>b</i>*</b>	<b><math>\Delta E^*</math></b>
SCB 0	90.85 $\pm$ 0.07 <sup>a</sup>	-1.30 $\pm$ 0.06 <sup>a</sup>	2.61 $\pm$ 0.21 <sup>a</sup>	3.48 $\pm$ 0.06 <sup>a</sup>
SCB 2.5	90.86 $\pm$ 0.09 <sup>a</sup>	-1.25 $\pm$ 0.04 <sup>b</sup>	2.75 $\pm$ 0.12 <sup>b</sup>	3.57 $\pm$ 0.05 <sup>b</sup>
SCB 5.0	91.01 $\pm$ 0.16 <sup>b</sup>	-1.24 $\pm$ 0.03 <sup>b</sup>	2.90 $\pm$ 0.21 <sup>c</sup>	3.58 $\pm$ 0.04 <sup>b</sup>
SCB 7.5	91.26 $\pm$ 0.02 <sup>c</sup>	-1.23 $\pm$ 0.03 <sup>b</sup>	3.22 $\pm$ 0.08 <sup>d</sup>	3.59 $\pm$ 0.06 <sup>b</sup>
SCB 10.0	91.55 $\pm$ 0.08 <sup>d</sup>	-1.21 $\pm$ 0.07 <sup>b</sup>	3.40 $\pm$ 0.06 <sup>e</sup>	3.61 $\pm$ 0.07 <sup>b</sup>

Results are presented as mean  $\pm$  sd. Different superscript letters in the same column indicate significant difference by Duncan's multiple range tests ( $p < 0.05$ ).

SCB: Sugarcane bagasse

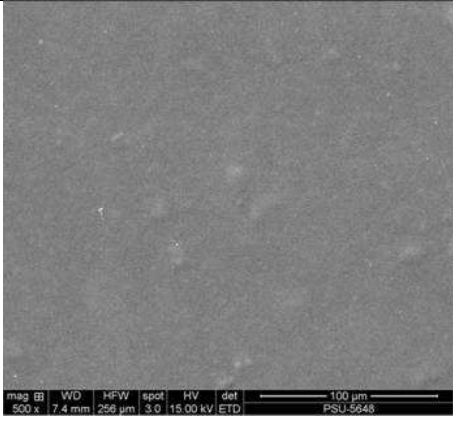
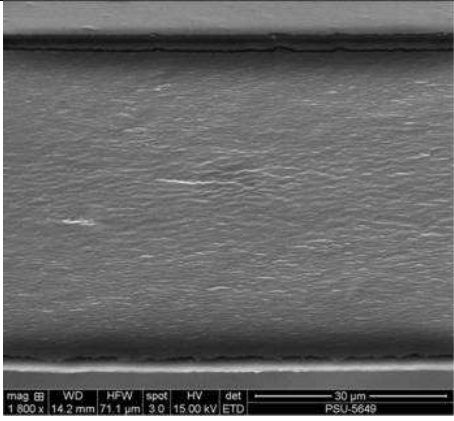
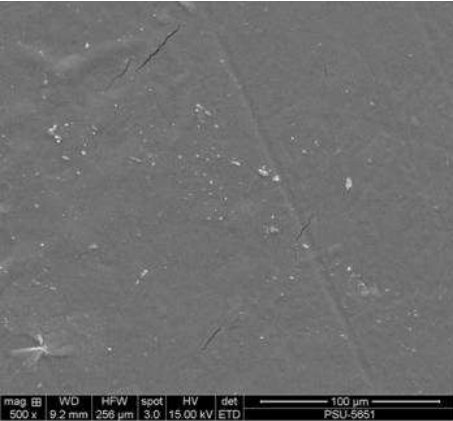
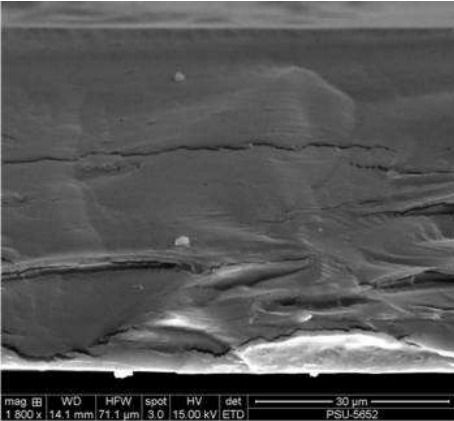
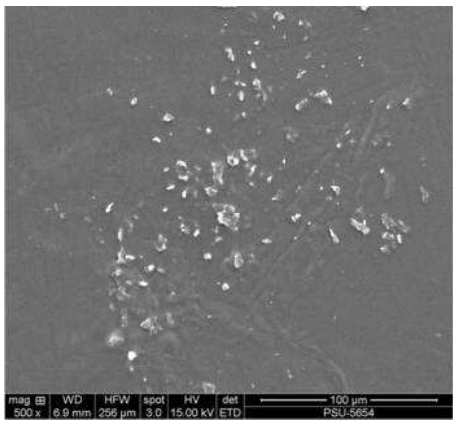
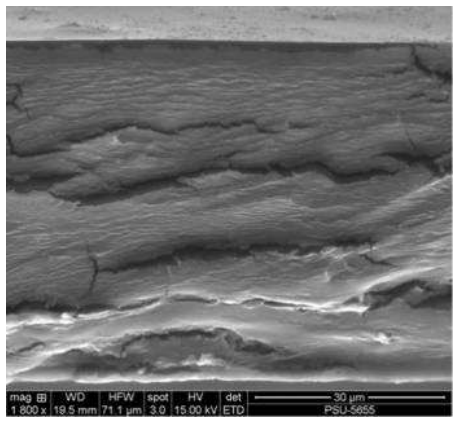
Table 6 Light transmittance and transparency of films from chicken feet gelatine incorporated with different percentage of dry weight SCB.

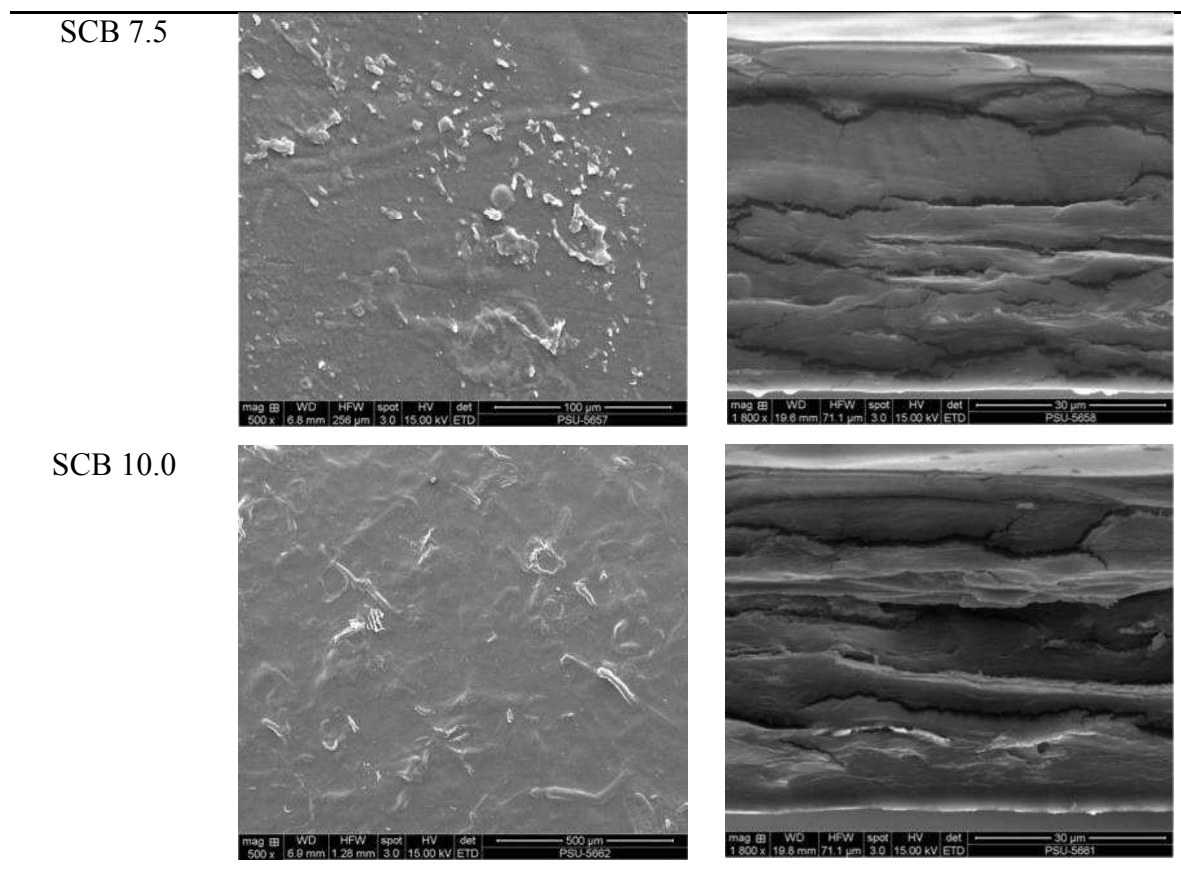
Film sample	Light transmittance (%) at different wavelength (nm)								Transparency value*
	200	280	350	400	500	600	700	800	
SCB 0	0.02	22.20	73.13	80.35	84.35	85.94	86.83	87.43	0.99 ± 0.09 <sup>a</sup>
SCB 2.5	0.02	19.88	68.94	76.91	81.08	82.62	83.54	84.23	1.06 ± 0.09 <sup>a</sup>
SCB 5.0	0.04	17.23	63.44	72.65	77.07	78.66	79.62	80.32	1.48 ± 0.13 <sup>b</sup>
SCB 7.5	0.02	12.57	51.79	61.16	65.37	67.03	68.03	68.81	1.84 ± 0.19 <sup>c</sup>
SCB 10.0	0.02	9.95	47.52	58.21	62.90	64.79	65.92	66.80	2.37 ± 0.14 <sup>d</sup>

\*Mean ± SD

Different superscript letters in the same column indicate significant difference by Duncan's multiple range test ( $p < 0.05$ ).

SCB: Sugarcane bagasse

Weight Percentage of SCB (%)	Surface	Cross-section
SCB 0	 <p>mag 500 x WD 7.4 mm HFW 256 µm spot 3.0 HV 15.00 kV det ETD 100 µm PSU-5648</p>	 <p>mag 1 800 x WD 14.2 mm HFW 71.1 µm spot 3.0 HV 15.00 kV det ETD 30 µm PSU-5649</p>
SCB 2.5	 <p>mag 500 x WD 9.2 mm HFW 256 µm spot 3.0 HV 15.00 kV det ETD 100 µm PSU-5651</p>	 <p>mag 1 800 x WD 14.1 mm HFW 71.1 µm spot 3.0 HV 15.00 kV det ETD 30 µm PSU-5652</p>
SCB 5.0	 <p>mag 500 x WD 6.9 mm HFW 256 µm spot 3.0 HV 15.00 kV det ETD 100 µm PSU-5654</p>	 <p>mag 1 800 x WD 18.5 mm HFW 71.1 µm spot 3.0 HV 15.00 kV det ETD 30 µm PSU-5655</p>



SCB: Sugarcane bagasse

**Figure 1.** Scanning Electron Microscopy micrographs of surface (magnification: 500x) and cross section (magnification: 1800x) of films from chicken feet gelatine incorporated with different levels of weight percentage of SCB. The SCB 0 which is the control film showed smooth and homogeneous surface. The cross-section of the control film also showed smooth surface. As the weight percentage of hydrolyzed SCB increases, the surface of the films showed increment in white spots. The white spots are believed to be the hydrolyzed SCB.