

1 **Obtaining microbial communities with exoelectrogenic activity from** 2 **anaerobic sludge using a simplified procedure**

3 **Short title: A simplified procedure to obtain exoelectrogenic activity**

4 Ribot-Llobet E.¹, Montpart N.¹, Ruiz Y.¹, Rago L.¹, Lafuente, J.^{1,2}, Baeza J.A.^{1*},
5 Guisasola A.¹

6 Edgar Ribot-Llobet

7 ¹ Departament d'Enginyeria Química. Escola d'Enginyeria.

8 Universitat Autònoma de Barcelona, 08193, Bellaterra (Barcelona), Spain.

9 email: edgar.ribot@uab.cat

10
11 Núria Montpart

12 ¹ Departament d'Enginyeria Química. Escola d'Enginyeria.

13 Universitat Autònoma de Barcelona, 08193, Bellaterra (Barcelona), Spain.

14 E-mail: nuria.montpart@uab.cat

15
16 Yolanda Ruiz

17 ¹ Departament d'Enginyeria Química. Escola d'Enginyeria.

18 Universitat Autònoma de Barcelona, 08193, Bellaterra (Barcelona), Spain.

19 E-mail: yolanda.ruiz@uab.cat

20
21 ¹ Laura Rago

22 Departament d'Enginyeria Química. Escola d'Enginyeria.

23 Universitat Autònoma de Barcelona, 08193, Bellaterra (Barcelona), Spain.

24 E-mail: laura.rago@uab.cat

25
26 Javier Lafuente

27 ¹ Departament d'Enginyeria Química. Escola d'Enginyeria.

28 Universitat Autònoma de Barcelona, 08193, Bellaterra (Barcelona), Spain.

29 E-mail: javier.lafuente@uab.cat

30 ² MATGAS Research Centre, Campus Universitat Autònoma de Barcelona,

31 08193, Bellaterra (Barcelona), Spain.

32
33 Juan A. Baeza

34 ¹ Departament d'Enginyeria Química. Escola d'Enginyeria.

35 Universitat Autònoma de Barcelona, 08193, Bellaterra (Barcelona), Spain.

36 email: juanantonio.baeza@uab.cat

37 Tlf: 34 935811587, Fax: 34 935812013

38 *Corresponding author: Juan A. Baeza

39
40 Albert Guisasola

41 ² Departament d'Enginyeria Química. Escola d'Enginyeria

42 Universitat Autònoma de Barcelona, 08193, Bellaterra (Barcelona), Spain.

43 email: albert.guisasola@uab.cat

1 HIGHLIGHTS

2

- 3
- Low-cost increase of exoelectrogenic activity from anaerobic sludge.
- 4
- Development of a new sediment-based MFC with simplified configuration.
- 5
- The brush inoculated for 30 days in the new MFC achieves 0.9W/m^2 in an AC-
- 6 MFC.
- 7
- The new procedure has comparable performance to more complex techniques.

8

9

1 **ABSTRACT**

2 **BACKGROUND**

3 Microbial Fuel Cells (MFCs) is a technology used to transform the chemical energy
4 present in substrates into electricity. The starting-up of these systems, i.e. enriching the
5 anodic community in exoelectrogenic bacteria is usually long or requires expensive
6 equipment.

7 **RESULTS**

8 An easy and low-cost procedure based on sediment MFC was developed to select
9 microbial communities with exoelectrogenic activity from anaerobic sludge of a waste
10 water treatment plant (WWTP). The configuration was based on a simple vessel
11 working as a single chamber MFC with a cathode of stainless steel wool in the liquid
12 surface and a submerged graphite fibre brush as anode. In 30 days of operation, a
13 biofilm with remarkable exoelectrogenic activity was grown on the anode of the MFC.
14 This graphite fibre brush anode was able to supply 0.9W/m^2 when working in an air-
15 cathode MFC (AC-MFC) during 45 days of operation.

16 **CONCLUSION**

17 The presented procedure was demonstrated as a successful, low-cost and low-
18 maintenance procedure to obtain exoelectrogenic activity and had comparable
19 performances to other more costly and complex inoculation procedures. The Sed-MFC
20 does not require potentiostat, external aeration, stirring, membranes or an enriched
21 inoculum in exoelectrogenic biomass.

22

23 **Keywords:** anaerobic sludge, exoelectrogenic bacteria, microbial fuel cell (MFC),
24 sediment MFC, stainless steel cathode.

25

1 INTRODUCTION

2 Bioelectrochemistry is an emerging technology to transform the chemical energy
3 present in substrates into electricity (in microbial fuel cells, MFC) or other products of
4 interest (in microbial electrolysis cells, MEC) using microorganisms as catalysts.¹⁻⁴
5 These microorganisms, known as exoelectrogens or anode respiring bacteria (ARB), are
6 able to bring electrons out of the microbial cell and transfer them to a solid anode
7 without any external chemical mediator.⁵ ARB comprise many bacteria genera, which
8 can be found in different natural environments from marine sediments to anaerobic
9 systems, as *Geobacter*⁶⁻⁹, *Shewanella*¹⁰⁻¹¹ or *Rhodoferax*¹².

10 The starting-up of a bioelectrochemical system, i.e. enriching the anodic community in
11 ARB, usually takes weeks. The most common inoculum consists of using either the
12 liquid effluent or some scrapped biofilm from the anode of an existing
13 bioelectrochemical system. Another common start-up technique is controlling the anode
14 at a fixed potential: i.e. the anode inoculated with anaerobic sludge is immersed into a
15 substrate solution and poised at a certain potential. The choice of the optimal anode
16 potential is a controversial issue¹³⁻¹⁴. An anode with a more positive potential would
17 theoretically result in a higher microbial diversity because different microorganisms
18 would obtain a high yield of energy transferring their electrons to the anode. However,
19 lower anode potential would select those more specialized bacteria able to use a
20 minimal amount of energy to grow releasing electrons to an anode. In any case, working
21 at a fixed anode potential requires an expensive potentiostat, which can be particularly
22 costly if high amounts of ARB are needed. A cheaper option would be using an MFC
23 with a selected external resistor resulting in a desired potential range. Kim et al.¹⁵
24 studied several inoculation techniques using a classical two-chamber MFC

1 configuration. They were able to increase the power from 22 to 30 mW/m² using ferric
2 iron-coated carbon electrodes. Liu et al.¹⁶ demonstrated that the performance of mixed
3 culture microbial biofilms could be improved by a consecutive, purely electrochemical
4 selection and biofilm acclimatization procedure. Their method was shown to be very
5 efficient but it also required a multipotentiostat.

6 Despite these advances recently made, MFCs and MECs still face significant challenges
7 for large-scale real-world applications¹⁷. For example, when moving bioelectrochemical
8 systems into pilot or industrial scale,¹⁸⁻¹⁹ the development of a low-cost and reliable
9 procedure to obtain ARB-enriched biofilms on large anodes will be essential. The
10 selected procedure should not require either ARB-enriched cultures or expensive
11 equipment as for example potentiostats or selective membranes. In this sense, the aim of
12 this study was to develop an efficient (simplified, successful and scalable) technique to
13 select ARB in a graphite fibre brush anode suitable for different bioelectrochemical
14 systems (MFC or MEC). The developed method is based on sediment/benthic MFC and
15 uses anaerobic sludge as inoculum. In short, a benthic MFC harvests energy from
16 natural environments by placing an electrode in the sediment (anode) and connecting it
17 with an electrical circuit to another electrode (cathode) situated on the overlying water
18 layer.²⁰⁻²⁴ This work proposes the adaptation of the benthic MFC concept to a simplified
19 lab-configuration (hereafter named Sed-MFC). In short, the Sed-MFC configuration
20 consists of a single chamber MFC where the anode, a brush graphite, is buried into
21 settled anaerobic sludge meanwhile the cathode, a stainless steel wool mesh, floats on
22 the upper layer of the cell, thus in contact with the medium and the atmosphere. Then,
23 the Sed-MFC corresponds to an air cathode configuration.

1 To the best of our knowledge, this is the first report of a methodology based on benthic
2 MFC to obtain anodes with increased exoelectrogenic activity from raw anaerobic
3 sludge.

4

5 **MATERIALS AND METHODS**

6

7 *Sed-MFC construction and operation*

8 The proposed Sed-MFC consisted of a conventional plastic vessel (1L) with an anode, a
9 cathode and an electrical wire connection (Fig. 1). The anode was a graphite fibre brush
10 (70 mm diameter x 70 mm length) made with fibres of diameter 7.2 μm (type
11 PANEX33 160K, ZOLTEK, Hungary) and titanium wire. The brush was thermally
12 treated at 440°C for 30 minutes to increase further microbial adhesion.²⁵ The cathode
13 was commercial SSW placed in the air/liquid interface and connected to a copper wire
14 over the water surface to avoid undesired copper corrosion that could affect MFC
15 performance²⁶. This low cost cathode provided high specific area, which balanced the
16 overpotential losses.²⁷

17 Acetate was selected as electron donor and 2-bromoethanesulfonate (BES) was added to
18 prevent methanogenesis. The cell inoculation comprised 500 mL of anaerobic sludge,
19 125 mL of acetate solution, 0.38 mL of micronutrient solution and 100 mL of phosphate
20 buffer solution (PBS). Then, it was filled up with deionized water up to 1000 mL. The
21 anaerobic sludge was obtained from an anaerobic digester of an urban WWTP
22 (Manresa, Barcelona). The PBS stock solution consisted of (g/L): 80 NaCl, 2 KCl, 14.4
23 Na_2HPO_4 , 2.4 KH_2PO_4 (0.1M, pH 7.4) The acetate solution was (g/L): 11.33
24 $\text{NaCH}_3\text{COO}\cdot 3\text{H}_2\text{O}$, 0.19 $\text{CaCl}_2\cdot 2\text{H}_2\text{O}$, 1.2 $\text{MgSO}_4\cdot 7\text{H}_2\text{O}$, 1.02 NH_4Cl and the

1 micronutrient solution was (g/L): 1.5 FeCl₃·6H₂O, 0.15 H₃BO₃, 0.03 CuSO₄·5H₂O, 0.18
2 KI, 0.12 MnCl₂·H₂O, 0.06 Na₂MoO₄·2H₂O, 0.12 ZnSO₄·7H₂O, 0.15 CoCl₂·6H₂O, 0.12
3 AlCl₃, 0.12 NiCl₃ and 10 EDTA.²⁸ The final concentration of acetate in the MFC was
4 1.4 g/L. 10mM of BES were added to suppress methanogenic activity as in other
5 works.²⁹⁻³⁰

6 The conductivity and the pH were corrected to be around 10-15 mS/cm and 7.0-7.5
7 respectively. Cells were kept at room temperature (around 21°C) during all the
8 operational period. The SSW cathode was immersed 50% in the liquid, and the other
9 50% exposed to the atmosphere. Then the circuit was closed connecting the titanium
10 wire from the brush and the copper wire from the steel wool through a 560 Ω resistance.

11

12 *Air cathode MFC (AC-MFC) description*

13 Power and polarisation curves could not be done in Sed-MFCs due to their lack of
14 homogeneity (i.e. the liquid was not stirred). For this reason, when these curves were
15 needed, the brush from the Sed-MFC was slightly rinsed to remove all the non-attached
16 bacteria and was placed in a conventional AC-MFC using a fresh medium with the
17 desired initial acetate concentration.

18 The AC-MFC (Fig. 1) consisted of a 400 mL glass vessel with a lateral 7 cm diameter
19 aperture where the cathode was assembled. The cathode was made with carbon cloth
20 coated with carbon powder and platinum suspension on the inner side, whereas the
21 outer side was coated with a polytetrafluoroethylene (PTFE, Teflon) solution.³¹⁻³² The
22 anode was the carbon fibre brush coming from the Sed-MFC. Both electrodes were
23 connected through a 560 Ω resistance and voltage evolution was monitored.

1 Thus, the Sed-MFC has the anode buried in anaerobic sludge and a SSW-based cathode
2 while the AC-MFC has the enriched anode and a Pt-based cathode. The main goal of
3 the Sed-MFC is to enrich the anode in exoelectrogenic bacteria for its posterior use in
4 another MFC.

5

6 *Chemical analyses and monitoring*

7 Acetate was analysed by gas chromatography (Agilent Technologies, 7820-A) using a
8 flame ionization detector (FID) and helium as carrier gas. The voltage across the
9 external resistance in the Sed-MFC and AC-MFC was monitored using a 16-bit data
10 acquisition card (Advantech PCI-1716, Taiwan) connected to a personal computer with
11 software developed in LabWindows CVI 2010 for data acquisition and monitoring. Cell
12 intensity and power were calculated according Ohm's law (equations 1, 2).

$$13 \quad I = V / R_e \quad (\text{eq. 1})$$

$$14 \quad P = V \cdot I \quad (\text{eq. 2})$$

15 where V is the voltage drop in the resistance (V), R_e is the external resistance (Ω), I is
16 the current intensity (A) and P is the power (W). Intensity, as well as power, was
17 normalized with respect to the projected cathode area for comparison purposes for both
18 Sed-MFC and AC-MFC. The cathodic projected areas were around $6.4 \cdot 10^{-3} \text{ m}^2$ and
19 $3.9 \cdot 10^{-3} \text{ m}^2$ for the stainless steel wool and the platinum-coated carbon cloth,
20 respectively. Power and polarization curves were obtained with a multi-resistance board
21 which allowed changing the external resistance between 25 and 470000 Ω . A 10
22 minutes period was used for the voltage stabilization at each resistance. Coulombic
23 efficiency (CE), i.e. fraction of electrons recovered as current versus that in the initial
24 organic matter, was calculated as equation 3.

1
$$CE = \frac{\int Idt}{F \cdot b \cdot \Delta S \cdot V_R} \quad (\text{eq. 3})$$

2 where t is time (s), F is Faraday's constant (96485 C/mol-e⁻), b is the stoichiometric
3 number of electrons produced per mol of substrate (8 mol-e⁻/mol acetate), ΔS is the
4 substrate consumption (mol/L) and V_R the liquid volume (L).

5

6 *Electrochemical analyses*

7 Low-scan cyclic voltammetry (LSCV) was performed using a μAutolab type II
8 potentiostat in three-electrode mode in the AC-MFC. The anode was used as working
9 electrode and the cathode as the auxiliary one. An Ag/AgCl, KCl 3M electrode (+210
10 mV vs. SHE) was used as reference electrode. The system was under open circuit
11 conditions for one hour just before the LSCV started. LSCV was recorded at 0.1 mV/s
12 from the anode open circuit potential -0.50 V, to 0.3 V vs. Ag/AgCl.

13

14 *Scanning Electron Microscopy*

15 Samples of graphite fibre brush were collected and fixed with a solution of 2.5%
16 glutaraldehyde and 2% paraformaldehyde. Samples were treated with osmium
17 tetraoxide, dehydrated with ethanol and dried at critical point with carbon dioxide
18 (BAL-TEC CPD030; Bal-Tec). Then, the samples were coated with few nanometers of
19 Au-C (E5000 Sputter Coater, BIO-RAD, California, USA) to increase signal detection
20 and visualized on a Scanning Electron Microscope (Hitachi S-70, Japan).

21

22 **RESULTS AND DISCUSSION**

23 *Sed-MFC development and performance*

1 Anaerobic WWTP sludge was inoculated in three identical Sed-MFCs with an external
2 resistance of 560 Ω . Fig. 2a shows the voltage profiles obtained during more than 30
3 days. The initial voltage around 150 mV decreased during the first days of operation
4 due to the acclimatization period. Moreover, residual oxygen presence in the water and
5 the sediments also favoured this initial slow response. After approximately 4 days, the
6 voltage increased linearly (around 0.6 mV/h) which led to an increase in intensity,
7 indicating the development of exoelectrogenic activity. This linear increase period
8 reached fairly high voltage values, up to 300 mV (0.94 A/m² and 0.3 W/m²). A constant
9 water loss was observed due to evaporation, which was detrimental for the Sed-MFC
10 operation, since low water levels prevented the correct contact between the cathode and
11 the medium (i.e. the cathode surface in contact with water decreased). To avoid
12 complete substrate depletion and ensure good contact between the water and the
13 cathode, fresh medium was periodically added causing some oxygen diffusion and
14 partial ARB inhibition. The systems recovered their working voltage some days after
15 the medium addition.

16 These Sed-MFCs also allow inoculating at different external resistance and thus
17 providing different external conditions that can induce the growth of different microbial
18 communities in the anode. For example, Fig. 2b shows the voltage profiles obtained in
19 another experiment with three cells under the same operational conditions except for the
20 different external resistances used. As can be observed, the potential increases when the
21 external load increases, in agreement with the theoretical background. In this case, the
22 cells with higher external resistances gave similar power results (around 0.17 mW).

23 The time needed to develop a significant amount of exoelectrogenic biofilm is an
24 essential parameter for the design of Sed-MFCs. To this aim, five graphite fibre brushes

1 were developed in Sed-MFCs for different time periods. Each brush was transferred
2 directly to an AC-MFC, where power density curves were determined (Figure 3). After
3 these evaluations, it was concluded that a 30-day operational period ensured an
4 acceptable biofilm development.

5 Another experiment was designed to determine if the Sed-MFC could be further
6 simplified. Ensuring microbial adhesion is essential and thus, a thermal treatment of the
7 graphite fibres was initially performed. Thermal treatments are recommended to
8 enhance microbial adhesion since i) solvents and lubricants (from the anode
9 manufacturing) are washout from the anode surface and ii) active area is increased due
10 to microfractures generation²⁵, but this treatment increases the construction costs of
11 MFC. Considering that our Sed-MFC architecture was different from other reported
12 MFC (volume, distance between electrodes and electrodes surface are higher), an
13 experiment was performed to study if the positive effect of the thermal treatment was
14 significant in the Sed-MFC configuration. Then, a thermally treated graphite fibre brush
15 and an untreated brush were inoculated in a Sed-MFC for 25 days. After this period,
16 both anodes were placed in two different AC-MFCs. Fig. 4 shows the power and
17 polarisation curves obtained with both brushes. The maximum powers reached by the
18 untreated and treated graphite fibre brush were 312mW/m^2 and 903mW/m^2 ,
19 respectively. The thermal treatment resulted in not only three times higher power but
20 also in a significant internal resistance decrease: 362Ω for the untreated brush versus
21 151Ω for the treated brush. Therefore, these results corroborate the better performance
22 of the thermally treated brush and hence this treatment is recommended for the Sed-
23 MFC. In this sense, the SEM microphotographies (Fig. 4b) for treated fibres corroborate
24 the good colonization of the brush anode.

1

2 *From Sed-MFC to AC-MFC*

3 The extent of exoelectrogenic activity obtained in the Sed-MFC was evaluated by
4 moving an anodic brush which had been placed in a Sed-MFC for 30 days into a
5 conventional AC-MFC under the same operational conditions (Fig. 5a). The first cycle
6 (from day 0 to 14) corresponds to an acclimation cycle whereas the results from the
7 second cycle (from 14 to 17.5 days) onwards were already promising. A high coulombic
8 efficiency (51%) was achieved, the voltage reached 480 mV and the cycle length was
9 2.5 days. The experimental voltage ranged between 370 and 450 mV and an average
10 coulombic efficiency of 55% was obtained. Then, only one cycle was needed to adapt
11 the anode brush from the Sed-MFC to an AC-MFC operation. The AC-MFC system
12 performance was very satisfactory, achieving maximum values up to 0.134 A/m^2 ($P =$
13 0.07 W/m^2) with a reasonably fair coulombic efficiency.

14 The exoelectrogenic activity was also evaluated through LSCV by comparing an anodic
15 brush obtained from a Sed-MFC and stabilized in an AC-MFC for 48 hours to a non-
16 inoculated brush (Fig. 5b). The inoculated brush exhibited one order of magnitude
17 higher exoelectrogenic activity than the obtained with the non-inoculated brush, which
18 showed negligible activity. The inoculated anode showed one typical oxidation peak at -
19 $0.25 \text{ V vs Ag/AgCl}$. The value of the anode potential giving half of the maximum
20 current density, known as E_{kA} , was around -0.37 V , which is in agreement with the
21 results found for acetate-fed *Geobacter* pure culture systems.³³ The LSCV also showed
22 a high capacitive current for the inoculated anode, indicating the presence of a
23 conductive biofilm attached to the anode surface.³⁴

24

1 *Comparison with other works*

2 The proposed inoculation procedure is based on placing a graphite fibre brush in a
3 Sed-MFC with an anaerobic sludge blanket during 30 days. Anaerobic WWTP sludge is
4 a good candidate for inoculation because it is easy to obtain and contains a high
5 diversity of bacterial communities, including electrochemically active strains of
6 bacteria.¹⁵ The Sed-MFC methodology has several advantages with respect to other
7 MFC configurations. No external aeration is required, as the cathode is directly exposed
8 to air resulting in significant aeration savings. The internal resistance is minimised
9 because the electrodes can be located nearby. The system has low maintenance
10 requirements, as only the level of liquid must be supervised with low periodicity.
11 Neither stirring nor proton exchange membrane (PEM) are required which decreases the
12 operational costs. The main purpose of the PEM is to avoid oxygen entering to the
13 anode. With the proposed configuration, the amount of oxygen in contact with the
14 sludge blanket is negligible, particularly taking into account that the system is not
15 stirred. Moreover, if some oxygen entered, it would be consumed in the upper layer of
16 the blanket, maintaining the lower layer (where the brush is placed) under the required
17 anaerobic conditions.

18 Reported configurations in the literature¹⁴ propose an initial polarisation period where a
19 certain potential is applied to the cell in order to enhance ARB growth on the anode.
20 This external voltage is reported to increase the ARB growth at the expense of
21 increasing the cost. However, the proposed Sed-MFC does not consider the polarisation
22 period since the objective is to develop an efficient (i.e. simplified, successful and
23 scalable) procedure to obtain anodic microbial communities with exoelectrogenic
24 activity using anaerobic sludge. The total cost of the cell materials is practically due to

1 the titanium wire (around 166€/m, 0.5 mm diameter titanium) used to build the anode
2 brush. Table 1 compares the performance of the presented procedure with other reported
3 works. This comparison is not a straightforward issue since a wide range of reactor
4 types, volumes, inoculum sources and substrates are found in the literature. In our case,
5 we compare the experimental results obtained in the first batch with the AC-MFC when
6 the anodic brush was transferred. In this study, a maximum power of 0.9 W/m² (Fig. 4)
7 was reached, which is a fairly good result for a reactor volume of 400mL. Wang et al.³⁵
8 presented a selection strategy able to reach half the power output of this study in about
9 the same time, 35 days, and using a similar reactor volume, 480 mL. However they used
10 a potentiostat, what increases considerably the cost of the inoculation process. Other
11 studies where inoculation time was high, such as Logan et al.²⁵, obtained very high
12 power output, nevertheless the volume was much lower, which obviously reduces
13 power losses. Kim et al.,¹⁵ who worked with a similar reactor volume of 620mL, stated
14 that 50 hours were needed for inoculation when anaerobic sludge was used as inoculum;
15 however, power output was thirty times lower than the one observed in this study.
16 Finally, Wang et al.³⁶ also presented a work where inoculation time was very fast (60
17 hours) and power output was of the same order of magnitude as ours. However, the
18 inoculum was coming from a previous working MEC with an already enriched
19 exoelectrogenic environment that could have expedited the inoculation process.
20 Thus, our system, in comparison with others, seems to provide a fair amount of
21 exoelectrogenic activity in a relatively high reactor volume when starting up from a
22 poor ARB environment like anaerobic sludge from an anaerobic digester in a reasonable
23 time frame.

24

1 **CONCLUSIONS**

2 A simplified and efficient procedure to increase the exoelectrogenic activity of anodic
3 microbial communities from anaerobic WWTP sludge was developed. The Sed-MFC
4 configuration was demonstrated as a successful, low-cost and low-maintenance
5 procedure to obtain exoelectrogenic activity. The anode graphite fibre brush developed
6 in a Sed-MFC for 30 days provided good results and showed comparable performances
7 to other more costly and complex inoculation procedures. The Sed-MFC does not
8 require potentiostat, external aeration, stirring or membranes. The electrodes can be
9 located nearby decreasing the internal resistance and the anaerobic sludge blanket
10 allows maintaining strict anaerobic conditions in the anode.

11

12 **ACKNOWLEDGEMENTS**

13 Discussions with S. Guri and L. Vega from MATGAS and Carburos Metálicos-Air
14 Products Group, are gratefully acknowledged. Financial support was provided by
15 Carburos Metálicos- Air Products Group and the Spanish Government, under the
16 project BIOSOS (CDTI, program Ingenio 2010). The authors are members of the
17 GENOCOV group (Grup de Recerca Consolidat de la Generalitat de Catalunya, 2009
18 SGR 815).

19

20 **REFERENCES**

- 21 1. Bennetto HP, Microbes come to power. *New Sci* **114**:36-40 (1987).
22 2. Bennetto HP, Electricity generation by microorganisms. *Biotechnol Educ* **1**:163-168
23 (1990).

- 1 3. Liu H, Cheng S, Logan BE, Power generation in fed-batch microbial fuel cells as a
2 function of ionic strength, temperature, and reactor configuration. *Environ Sci*
3 *Technol* **39**:5488-5493 (2005).
- 4 4. Rozendal RA, Hamelers HVM, Euverink GJW, Metz SJ, Buisman CJN, Principle
5 and perspectives of hydrogen production through biocatalyzed electrolysis. *Int J*
6 *Hydrogen Energ* **31**:1632-1640 (2006).
- 7 5. Logan BE, Exoelectrogenic bacteria that power microbial fuel cells. *Nat Rev*
8 *Microbiol* **7**:375-381 (2009).
- 9 6. Dumas C, Basseguy R, Bergel A, Microbial electrocatalysis with *Geobacter*
10 *sulfurreducens* biofilm on stainless steel cathodes. *Electrochim Acta* **53**:2494-2500
11 (2008).
- 12 7. Kim M, Lee Y, Optimization of culture conditions and electricity generation using
13 *Geobacter sulfurreducens* in a dual-chambered microbial fuel-cell. *Int J Hydrogen*
14 *Energ* **35**:13028-13034 (2010).
- 15 8. Lovley DR, Microbial fuel cells: novel microbial physiologies and engineering
16 approaches. *Curr Opin Biotech* **17**:327-332 (2006).
- 17 9. Yi, H., Nevin, K.P., Kim, B.-., Franks, A.E., Klimes, A., Tender, L.M., Lovley, D.R.,
18 2009. Selection of a variant of *Geobacter sulfurreducens* with enhanced capacity for
19 current production in microbial fuel cells. *Biosens. Bioelectron.* 24, 3498-3503.
- 20 10. Biffinger JC, Byrd JN, Dudley BL, Ringeisen BR, Oxygen exposure promotes fuel
21 diversity for *Shewanella oneidensis* microbial fuel cells. *Biosens Bioelectron* **23**:820-
22 826 (2008).

- 1 11. Kim HJ, Park HS, Hyun MS, Chang IS, Kim M, Kim BH, A mediator-less
2 microbial fuel cell using a metal reducing bacterium, *Shewanella putrefaciens*.
3 *Enzyme Microb Technol* **30**:145-152 (2002).
- 4 12. Liu Z, Li H, Effects of bio- and abio-factors on electricity production in a
5 mediatorless microbial fuel cell. *Biochem Eng J* **36**:209-214 (2007).
- 6 13. Torres CI, Marcus AK, Parameswaran P, Rittmann BE, Kinetic experiments for
7 evaluating the nernst-monod model for anode-respiring bacteria (ARB) in a biofilm
8 anode. *Environ Sci Technol* **42**:6593-6597 (2008).
- 9 14. Wagner RC, Call DF, Logan BE, Optimal set anode potentials vary in
10 bioelectrochemical systems. *Environ Sci Technol* **44**:6036-6041 (2010).
- 11 15. Kim JR, Min B, Logan BE, Evaluation of procedures to acclimate a microbial fuel
12 cell for electricity production. *Appl Microbiol Biotechnol* **68**:23-30 (2005).
- 13 16. Liu Y, Harnisch F, Fricke K, Sietmann R, Schröder U, Improvement of the anodic
14 bioelectrocatalytic activity of mixed culture biofilms by a simple consecutive
15 electrochemical selection procedure. *Biosens Bioelectron* **24**:1006-1011 (2008).
- 16 17. Zhou M, Wang H, Hassett DJ, Guc T, Recent advances in microbial fuel cells
17 (MFCs) and microbial electrolysis cells (MECs) for wastewater treatment, bioenergy
18 and bioproducts. *J Chem Technol Biot*, **88**:508-518, (2013).
- 19 18. Cusick RD, Bryan B, Parker DS, Merrill MD, Mehanna M, Kiely PD, Liu G, Logan
20 BE, Performance of a pilot-scale continuous flow microbial electrolysis cell fed
21 winery wastewater. *Appl Microbiol Biotechnol* **89**:2053–2063 (2011).
- 22 19. Logan BE, Scaling up microbial fuel cells and other bioelectrochemical systems.
23 *Appl Microbiol Biotechnol* **85**:1665–1671 (2010).

- 1 20. Dumas C, Mollica A, Féron D, Basséguy R, Etcheverry L, Bergel A, Marine
2 microbial fuel cell: Use of stainless steel electrodes as anode and cathode materials.
3 *Electrochim Acta* **53**:468-473 (2007).
- 4 21. Hong SW, Chang IS, Choi YS, Chung TH, Experimental evaluation of influential
5 factors for electricity harvesting from sediment using microbial fuel cell. *Bioresource*
6 *Technol* **100**:3029-3035 (2009).
- 7 22. Seok WH, Hyung JK, Yong SC, Tai HC, Field experiments on bioelectricity
8 production from lake sediment using microbial fuel cell technology. *B Kor Chem Soc*
9 **29**:2189-2194 (2008).
- 10 23. Zhao J, Li XF, Ren YP, Wang XH, Jian C, Electricity generation from Taihu Lake
11 cyanobacteria by sediment microbial fuel cells. *J Chem Technol Biotechnol* **87**:1567-
12 1573 (2012).
- 13 24. Ren Y, Pan D, Li X, Fu F, Zhao Y, Wang X, Effect of polyaniline-graphene
14 nanosheets modified cathode on the performance of sediment microbial fuel cell. *J*
15 *Chem Technol Biot*, DOI: 10.1002/jctb.4146 (2013).
- 16 25. Logan B, Cheng S, Watson V, Estadt G, Graphite fiber brush anodes for increased
17 power production in air-cathode microbial fuel cells. *Environ Sci Technol* **41**:3341-
18 3346 (2007).
- 19 26. Zhu X, Logan BE, Copper anode corrosion affects power generation in microbial
20 fuel cells. *J Chem Technol Biot*, DOI 10.1002/jctb.4156 (2013).
- 21 27. Ribot-Llobet E, Nam JY, Tokash JC, Guisasola A, Logan BE, Assessment of four
22 different cathode materials at different initial pHs using unbuffered catholytes in
23 microbial electrolysis cells. *Int J Hydrogen Energ* **38**:2951-2956 (2013).

- 1 28. Lovley DR, Phillips EJP, Novel mode of microbial energy metabolism: organic
2 carbon oxidation coupled to dissimilatory reduction of iron or manganese. *Appl*
3 *Environ Microbiol* **54**:1472-1480 (1988).
- 4 29. Chidthaisong A, Conrad R, Specificity of chloroform, 2-bromoethanesulfonate and
5 fluoroacetate to inhibit methanogenesis and other anaerobic processes in anoxic rice
6 field soil. *Soil Biol Biochem* **32**:977-988 (2000).
- 7 30. Nielsen JL, Juretschko S, Wagner M, Nielsen PH, Abundance and phylogenetic
8 affiliation of iron reducers in activated sludge as assessed by fluorescence in situ
9 hybridization and microautoradiography. *Appl Environ Microbiol* **68**:4629-4636
10 (2002).
- 11 31. Cheng S, Liu H, Logan BE, Increased performance of single-chamber microbial fuel
12 cells using an improved cathode structure. *Electrochem Commun* **8**:489-494 (2006).
- 13 32. Cheng S, Liu H, Logan BE, Power densities using different cathode catalysts (Pt
14 and CoTMPP) and polymer binders (Nafion and PTFE) in single chamber microbial
15 fuel cells. *Environ Sci Technol* **40**:364-369 (2006).
- 16 33. Torres CI, Krajmalnik-Brown R, Parameswaran P, Marcus AK, Wanger G, Gorby
17 YA, Rittmann BE, Selecting anode-respiring bacteria based on anode potential:
18 Phylogenetic, electrochemical, and microscopic characterization. *Environ Sci*
19 *Technol* **43**:9519-9524 (2009).
- 20 34. Srikanth S, Marsili E, Flickinger MC, Bond DR, Electrochemical characterization of
21 *Geobacter sulfurreducens* cells immobilized on graphite paper electrodes. *Biotechnol*
22 *Bioeng* **99**:1065-1073 (2008).

- 1 35. Wang X, Feng Y, Ren N, Wang H, Lee H, Li N, Zhao Q, Accelerated start-up of
2 two-chambered microbial fuel cells: Effect of anodic positive poised potential.
3 *Electrochim Acta* **54**:1109-1114 (2009).
- 4 36. Wang A, Sun D, Ren N, Liu C, Liu W, Logan BE, Wu W, A rapid selection strategy
5 for an anodophilic consortium for microbial fuel cells. *Bioresource Technol*
6 **101**:5733-5735 (2010).
- 7

1 **Tables**

2

3

4

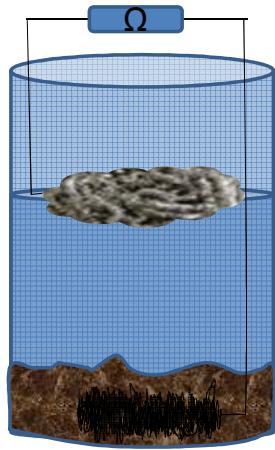
Table 1. Comparison of different procedures aiming at increased exoelectrogenic activity.

References	[36]	[28]	[16]	[14]	This study
Volume (mL)	420	26	620	480	400
Internal resistance (Ω)	N.D.	8	N.D.	91.84	133
Maximum power (W/m^2)	0.23	2.4	0.008-0.03	0.45	0.9
Reactor type	H-type	Cube air cathode	H-type	Cube-type	Air cathode
Inoculum origin	Previous MEC	Previous MFC	Anaerobic sludge	Anaerobic sludge	Anaerobic sludge
Cathode catalyst	Platinum	CoTMMPP	Platinum	Ferricyanide	Platinum
Substrate	Acetate	Acetate	Acetate	Glucose	Acetate
Polarization period	No	No	No	Yes	No
Inoculation time	60 hours	>6 month	50 hours	35 days	30 days

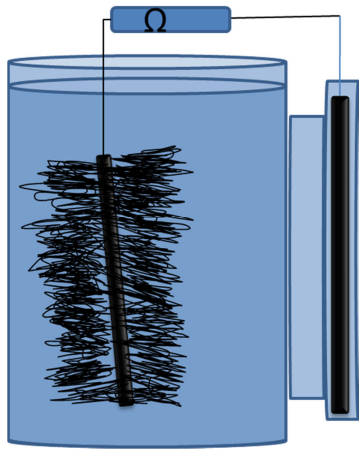
5

6

1



2

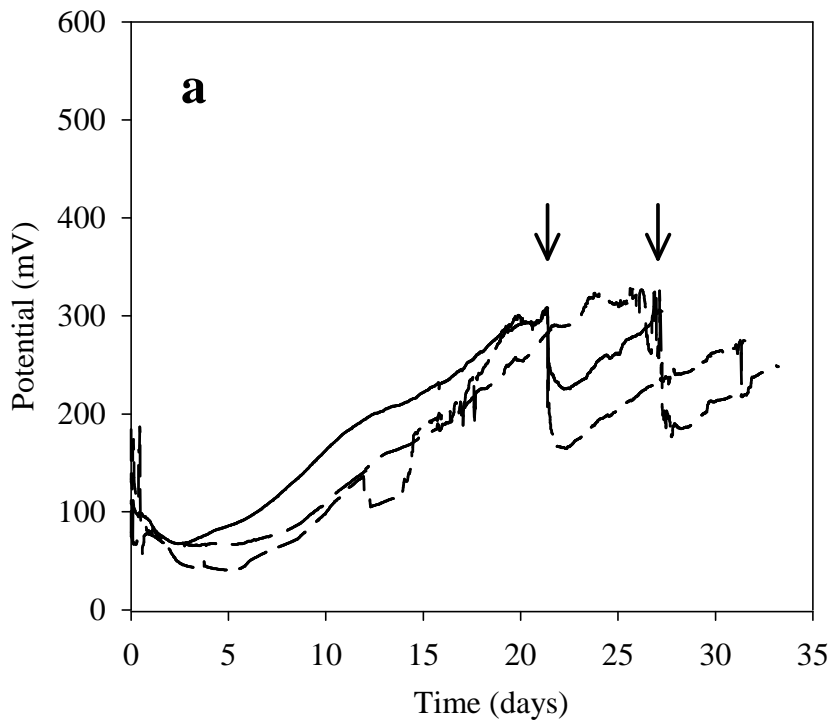


3 **Fig. 1.** Schematic representations (Left) and pictures (Right) of the Sed-MFC (Top) and

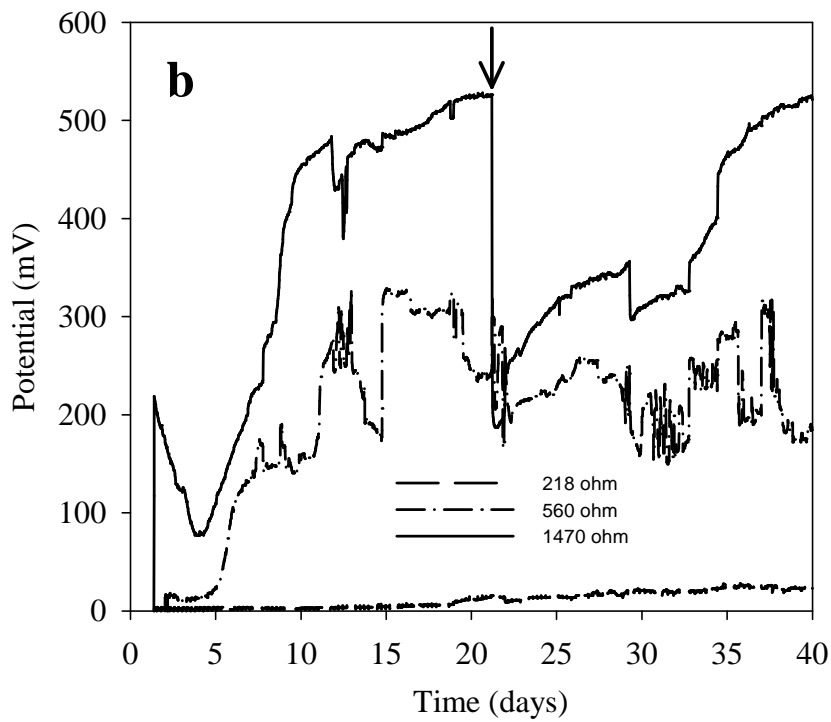
4

the AC-MFC (Bottom)

5

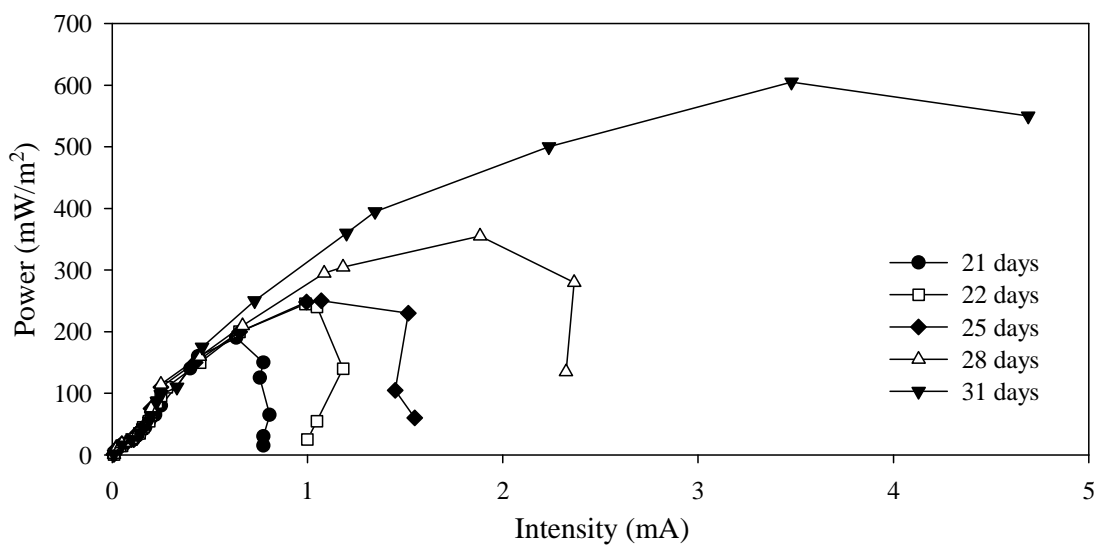


1



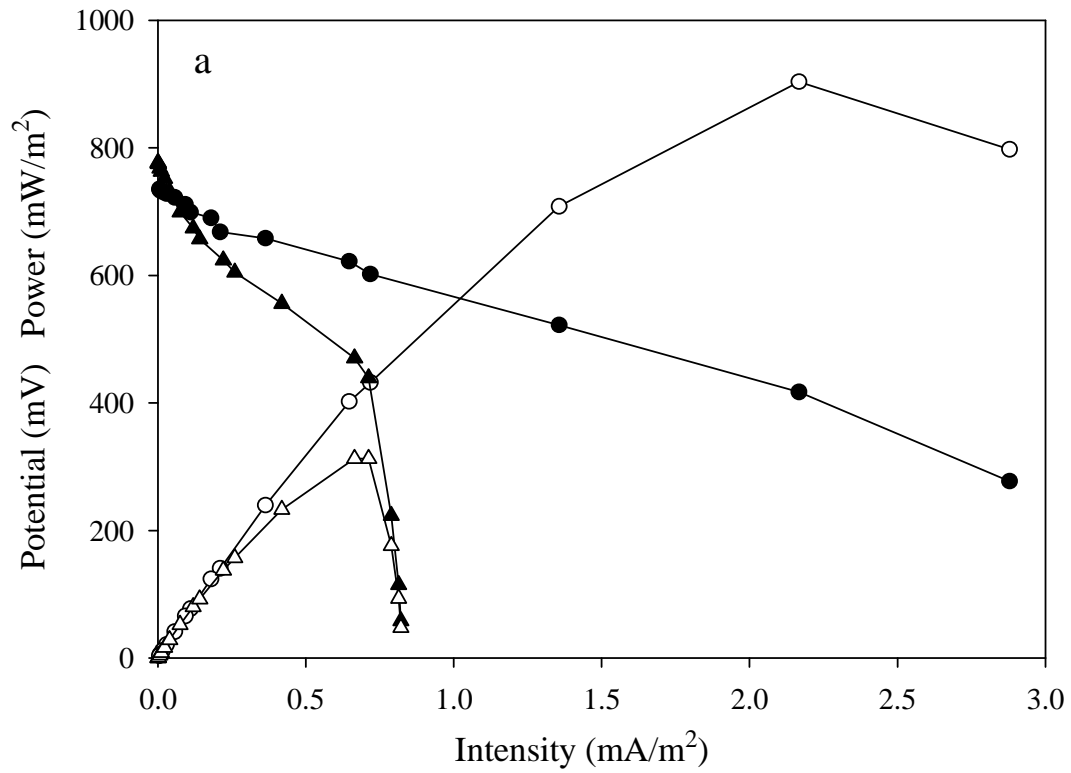
2

3 **Fig. 2.** a) Monitored voltage across 560Ω resistance for three different Sed-MFC with
 4 identical inoculation. b) Experimental profiles for three different Sed-MFC with
 5 different external resistances. Arrows indicate substrate addition.

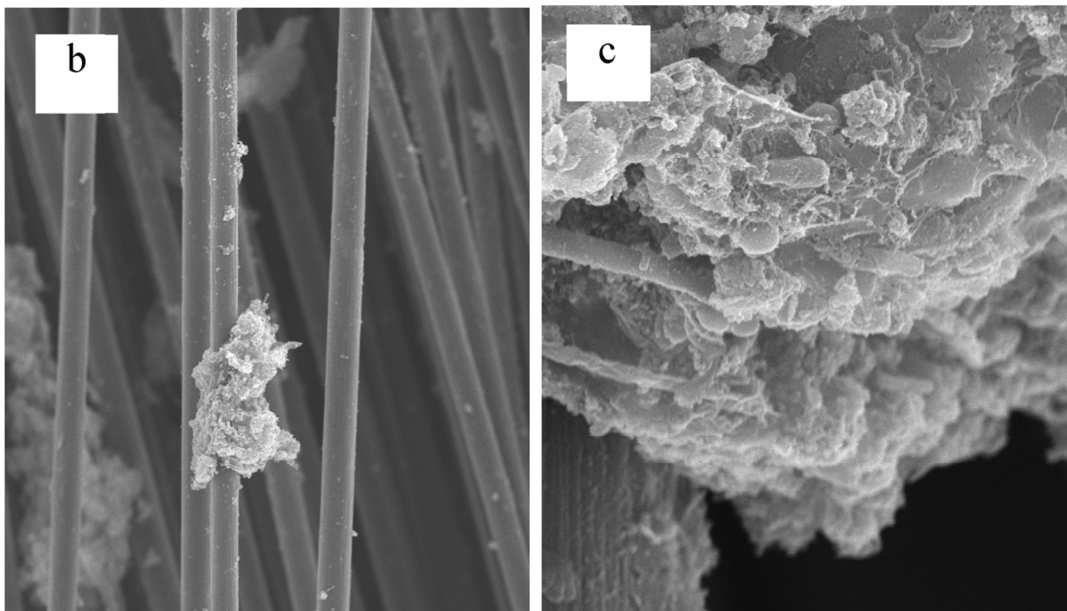


1
2
3
4
5

Fig. 3. Power curves in AC-MFC of anodes developed in Sed-MFCs with different inoculation periods.

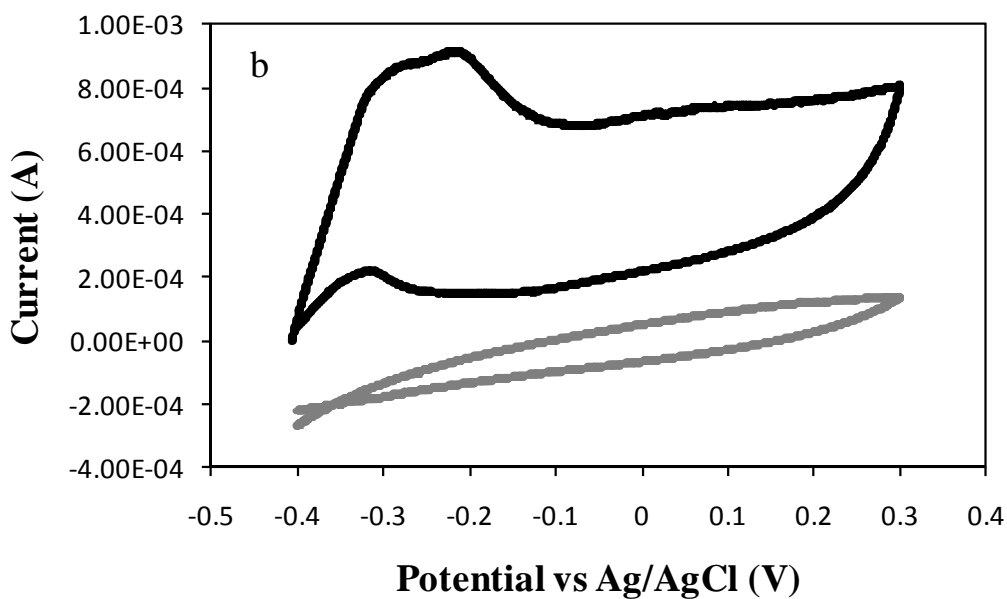
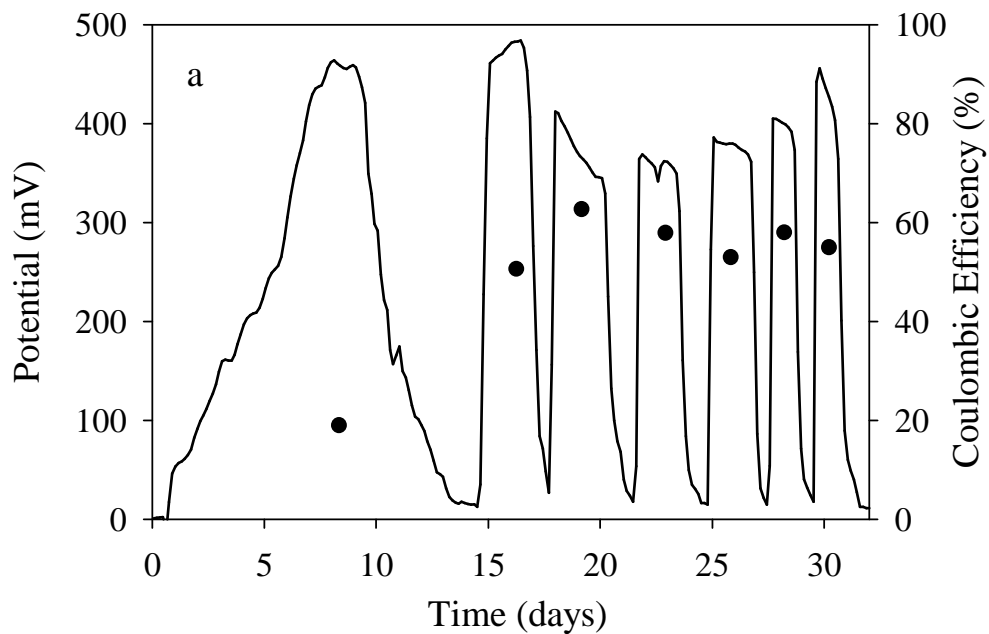


1



2

3 **Fig. 4.** a) Power (white symbols) and polarization (black symbols) curves for thermally
 4 treated (circles) and untreated (triangles) graphite fibre brush. b) and c) SEM photos of
 5 a colonized treated fibre.



1

2

3 **Fig. 5.** a) Experimental voltage profiles (line) and Coulombic efficiencies (dots) for
 4 each batch cycle of an AC-MFC using an anode brush inoculated for 30 days in a Sed-
 5 MFC. b) LSCV of an anode brush in an AC-MFC, 48 hours after being removed from a
 6 Sed-MFC (black) and anode brush without bacteria (grey).

