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AN EVALUATION OF  
ACCIDENTAL WATER-REACTIONS WITH LITHIUM  
COMPOUNDS IN FUSION REACTOR BLANKETS

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3. D. A. Dube, M. S. Kazimi and L. M. Lidsky, "Thermal Response of Fusion Reactor Containment to Lithium Fire," 3rd Top. Meeting in Fusion Reactor Technology, May 1978.
4. R. W. Sawdye and M. S. Kazimi, "Application of Probabilistic Consequence Analysis to the Assessment of Potential Radiological Hazards of Potential Hazards of Fusion Reactors," MITNE-220, Dept. of Nuclear Engineering, M.I.T., July 1978.
5. D. A. Dube and M. S. Kazimi, "Analysis of Design Strategies for Mitigating the Consequences of Lithium Fire within Containment of Controlled Thermonuclear Reactors," MITNE-219, Dept. of Nuclear Engineering, M.I.T., July 1978.
6. R. W. Sawdye and M. S. Kazimi, "Fusion Reactor Reliability Requirements Determined by Consideration of Radiological Hazards," Trans. Am. Nucl. Soc. 32, 66, June 1979.
7. R. W. Green and M. S. Kazimi, "Safety Considerations in the Design of Tokamak Toroidal Magnet Systems," Trans. ANS 32, 69, June 1979.
8. R. W. Green and M. S. Kazimi, "Aspects of Tokamak Toroidal Magnet Protection," PFC/TR-79-6, Plasma Fusion Center, M.I.T., July 1979.
9. S. J. Piet and M. S. Kazimi, "Uncertainties in Modeling of Consequences of Tritium Release from Fusion Reactors," PFC/TR-79-5, Plasma Fusion Center, M.I.T., July 1979.
10. M. J. Young and S. J. Piet, "Revisions to AIRDOS-II," PFC/TR-79-8, Contract #K-1702, Plasma Fusion Center, M.I.T., August 1979.
11. S. J. Piet and M. S. Kazimi, "Implications of Uncertainties in Modeling of Tritium Releases from Fusion Reactors," Proc. Tritium Technology in Fission, Fusion and Isotopic Applications, April 1980.
12. M. S. Tillack and M. S. Kazimi, "Development and Verification of the LITFIRE Code for Predicting the Effects of Lithium Spills in Fusion Reactor Containments," PFC/RR-80-11, Plasma Fusion Center, M.I.T., July 1980.

Publications Under Contract #K-1702 (continued)

13. M. S. Kazimi and R. W. Sawdye, "Radiological Aspects of Fusion Reactor Safety: Risk Constraints in Severe Accidents," J. of Fusion Energy, Vol. 1, No. 1, pp. 87-101, January 1981.
14. P. J. Krane and M. S. Kazimi, "An Evaluation of Accidental Water-Reactions with Lithium Compounds in Fusion Reactor Blankets," PFC/RR-81-26, Plasma Fusion Center, M.I.T., July 1981.
15. D. R. Hanchar and M. S. Kazimi, "Tritium Permeation Modelling of a Conceptual Fusion Reactor Design," PFC/RR-81-27, Plasma Fusion Center, M.I.T., July 1981.

AN EVALUATION OF ACCIDENTAL WATER-REACTIONS  
WITH LITHIUM COMPOUNDS IN FUSION REACTOR BLANKETS

ABSTRACT

Efforts to mitigate potential problems of lithium-based blankets for fusion reactors include the use of lithium compounds for breeding purposes. This report investigates the safety aspects of these alloys relative to the use of pure lithium in a water-cooled blanket. Included in the study is a modification of the LITFIRE computer code to predict the thermal response of an internal blanket breeder-water interaction.

For the problem analyzed, results indicate that some of the lithium-lead alloys may pose safety problems approximate to those associated with the use of liquid lithium.  $\text{Li}_2\text{O}$  is shown to be significantly safer than liquid lithium, while results using  $\text{LiAl}$  are similar to those of the lithium-lead alloys.

In addition, the study provides an overview of this safety question, signaling areas that require further development.

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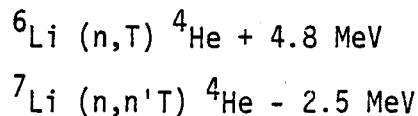
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## CHAPTER 1. INTRODUCTION

Advancements in plasma physics research, together with a growing concern for the risks of energy production in the public sector, has led to an increasing number of detailed fusion safety studies. Topics, including routine and accidental releases of tritium, activation of structural material by neutron bombardment, and the consequences of lithium fires, are currently under various degrees of investigation. The findings of these studies are incorporated in subsequent fusion reactor designs, answering some questions and creating still more.

This work is the product of a research program whose objective is to minimize the potential problems of a lithium-based blanket for fusion reactors. Such a blanket is practically forced upon us by the choice of a D-T fuel mixture for first generation fusion power plants [1]. The needed tritium is bred via the reactions:



Initially, natural lithium (92.58%  ${}^7\text{Li}$ , the rest  ${}^6\text{Li}$ ), a liquid at operating blanket temperatures, was the primary candidate for blanket breeder and/or coolant materials, due to its excellent breeding and heat transfer qualities, effectiveness in neutron moderation and relatively low pumping power need as compared to other liquid metals [2]. However, with time and study, serious disadvantages in the use of liquid lithium emerged.

Pure lithium is highly reactive with air, water and concrete; all materials that will be in abundant supply in the reactor environment. Experimentation at the Hanford Engineering Development Laboratory (HEDL) and computer modelling studies at MIT using the computer code LITFIRE (to be discussed in more detail later in this report) indicate that temperatures and pressures in the reactor containment area, in the event of a sizable lithium spill, can attain critically high values [3]. Figure 1.1 shows such a possibility. Such an event could provide a pathway for release of tritium or structural activation products, providing a hazard to plant personnel or the outside world.

This problem and others, including corrosion, difficulties in tritium recovery, and magnetohydrodynamic instabilities [4], have led designers to consider alternative materials for fusion blankets. Among such considerations are lithium-lead alloys.

This report will provide a preliminary analysis of lithium-lead alloys for use as breeding materials from the safety point of view. While it is thought that these materials provide less of a hazard than the use of liquid lithium, little has been demonstrated. Thus, there is the need to formulate some framework for a comparison.

Before actual calculations can be made, a basis must be established. The NUWMAK reactor design by the University of Wisconsin (1978) was chosen for this purpose, due to its use of  $\text{Li}_{62}\text{Pb}_{38}$  eutectic as the tritium breeder. The primary hazard here involves interaction between the lithium-lead alloy breeder and the boiling water coolant, inside the blanket. Specifics of this design are further discussed in Chapter 2.

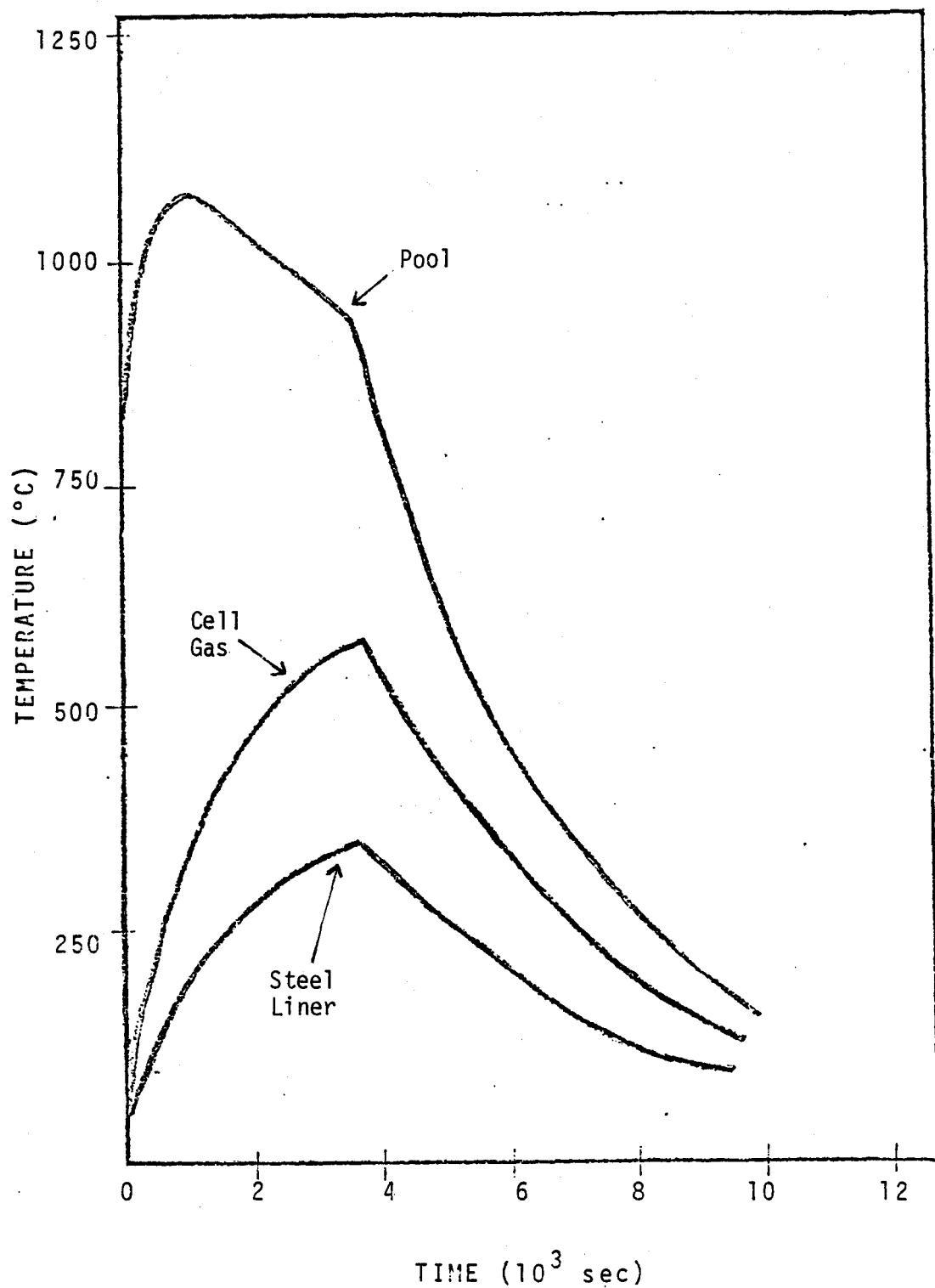


Figure 1.1 LITFIRE predictions for consequences of lithium spill in UWMAK III containment (Reference 3).

Using this basis, two separate studies are performed. The first is a static calculation: the breeder and coolant are allowed to interact immediately and the subsequent equilibrium final temperature of the blanket materials is determined. This is presented in Chapter 3. The second study is a dynamic calculation, using LITFIRE, of the temperature histories at various points in the blanket, if some accident allows breeder and coolant to come into contact. This is presented in Chapter 4. It should be stressed that in both studies, the values obtained are not sufficient evidence in themselves. Rather, these values must be compared with similar calculations employing the use of pure lithium. In this way, a measure of the relative hazards of the alternate breeders can be assessed.

## CHAPTER 2. BLANKET DESIGN BASIS DESCRIPTION

### 2.1 Introduction

NUWMAK, a conceptual thermonuclear reactor designed by the Fusion Engineering Program of the University of Wisconsin in March 1979, is one of a number of second generation studies aimed at maximizing the strengths of fusion while minimizing the weaknesses. This work builds upon the findings of a number of first generation designs aimed at identifying the important problems of fusion power.

The design philosophy of this study was to search for an "end product that has the potential to be reliable, maintainable, environmentally acceptable, and reasonably economic [4]." To do this, a number of changes were made to the preceding reactor concept, UWMAK III, including: an increase in the magnetic field to increase power density, a simplification of design to facilitate maintainence (as well as eliminate large cost items), the selection of structural materials to minimize resource requirements and costs, and a change in blanket breeder and coolant materials to reduce thermal fatigue and thereby increase reliability. The resulting structure is shown in Figures 2.1 and 2.2

The new elements in blanket breeder and coolant materials are of particular interest to those concerned with fusion safety studies. Utilized in NUWMAK are: (a)  $\text{Li}_{62}\text{Pb}_{38}$  eutectic for the breeding material, used because its melting point is very near the blanket operational temperature and thus, the latent heat of melting can provide energy storage, and (b) boiling water for the coolant, the "perfect choice," discarded previously with the presence of pure lithium as the breeder.

CROSS-SECTIONAL VIEW OF  
NUWMAK

Figure VIII-A-1

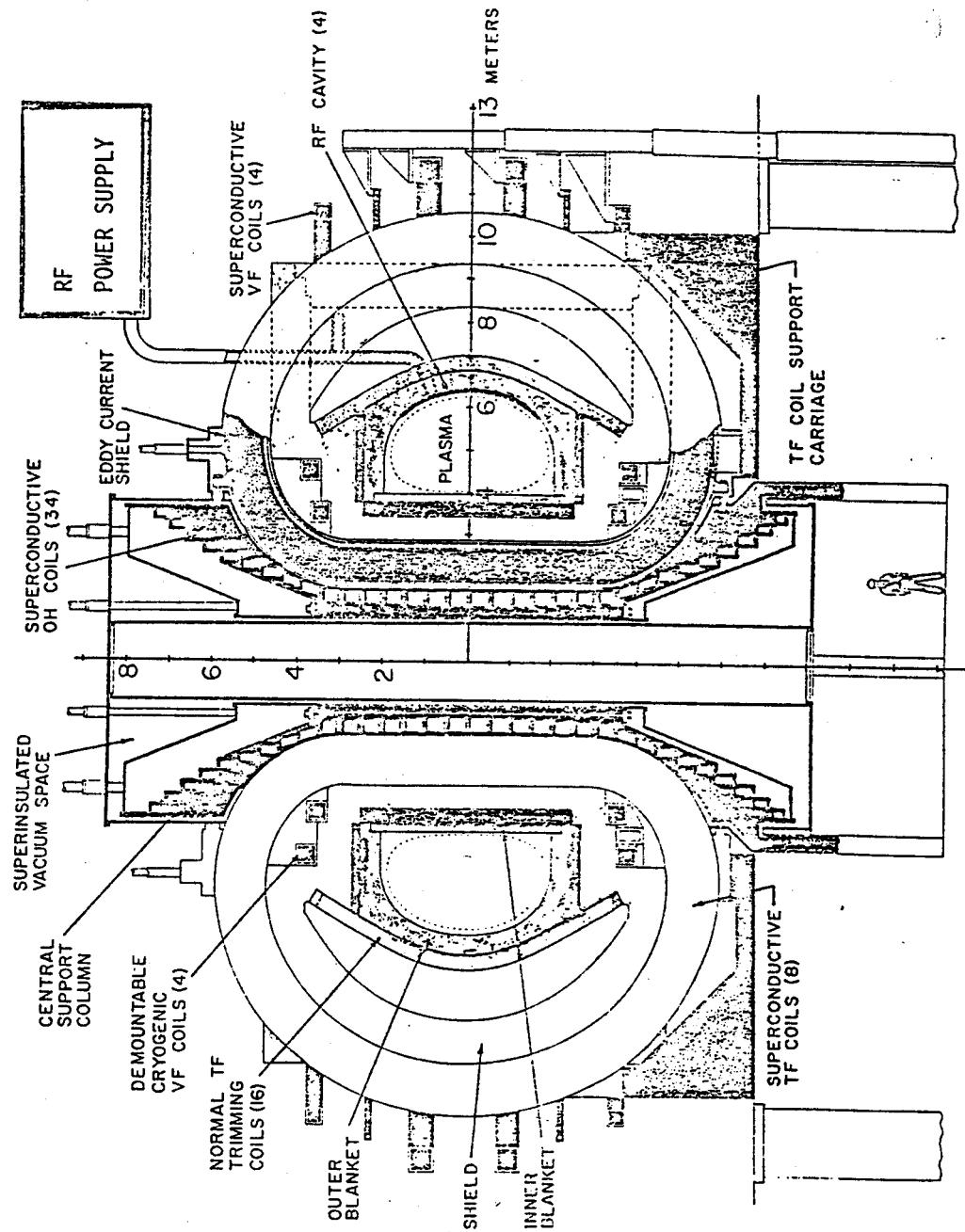


Figure 2.1 Cross-sectional View of NUWMAK (Reference 4).

TOP VIEW OF NUWMAK

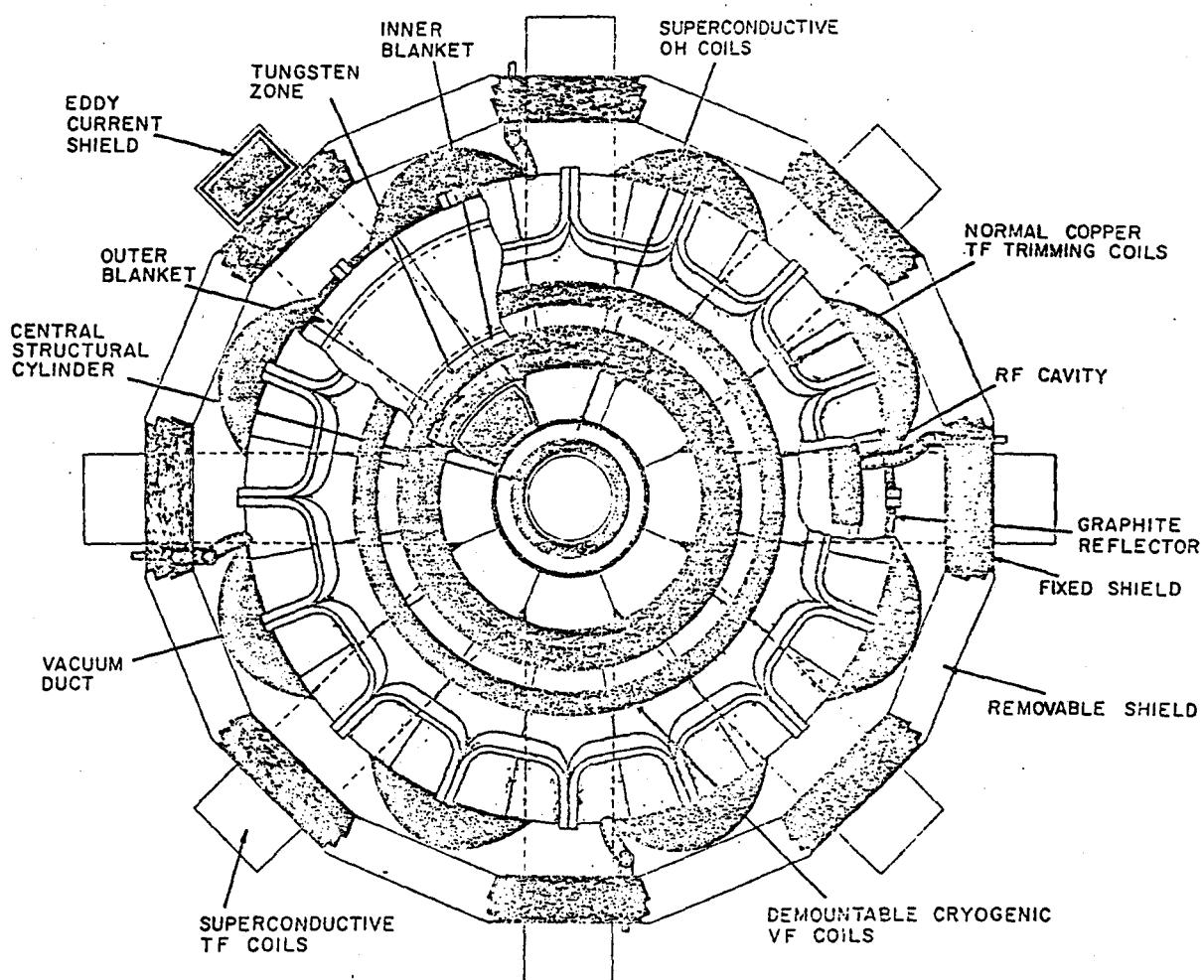


Figure 2.2  
(Reference 4)

A preliminary survey of the hazards of lithium-lead alloys indicates that interaction with water is by far the more severe problem, as these materials are relatively inert in air. NUWMAK, in addition to its use of  $\text{Li}_{62}\text{Pb}_{38}$ , provides an opportunity for the breeder and coolant to come into contact, specifically a breach in the cooling system inside the blanket. Thus, NUWMAK is the logical first choice for a basis for an investigation of the relative safety of the lithium-lead alloys.

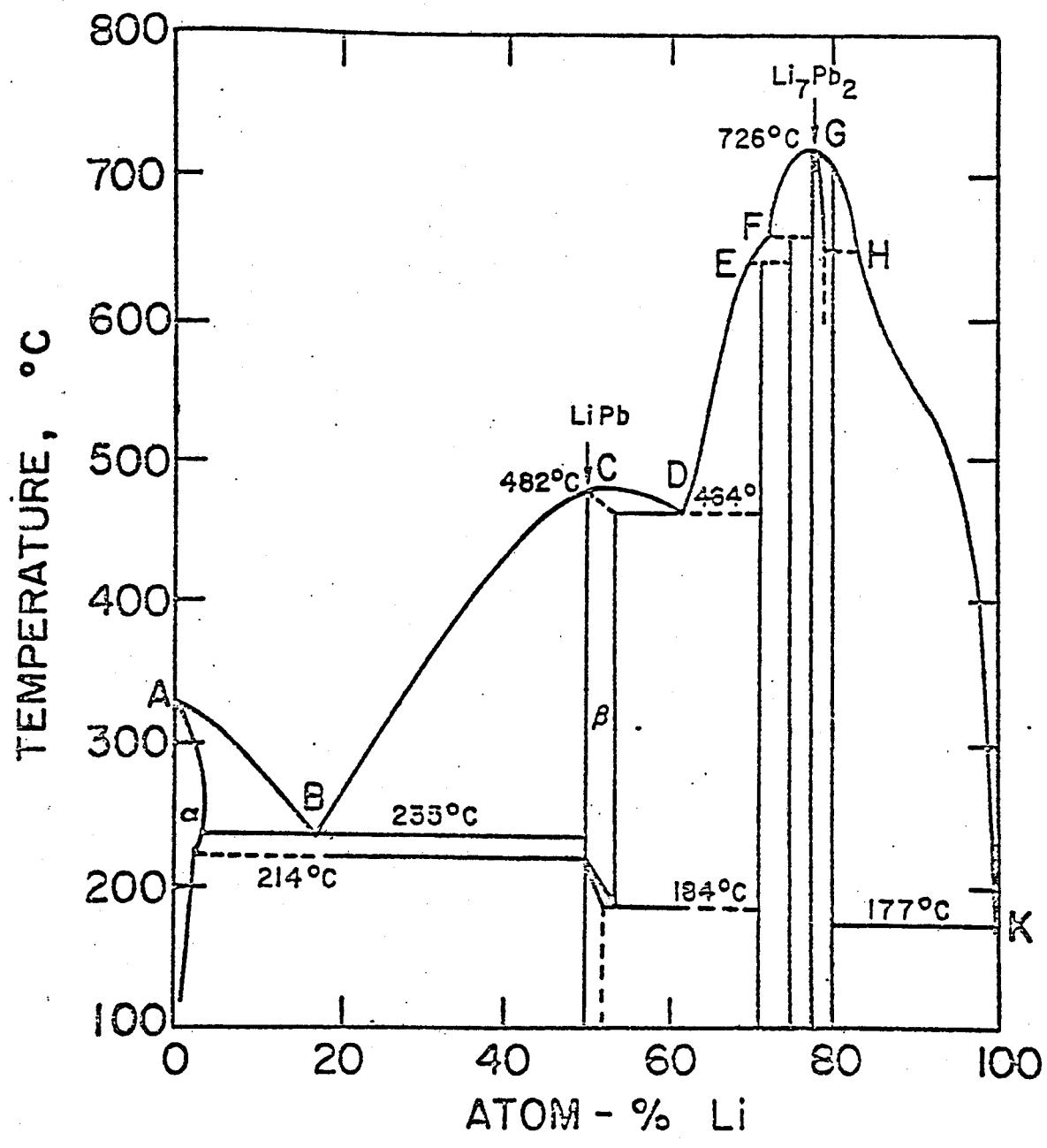
## 2.2 Breeding Materials

### 2.2.1 Lithium-lead alloys

The lithium compound selected as the tritium breeding material must satisfy many requirements. It must have desirable neutronic and irradiation characteristics, chemical stability at blanket operating temperatures, and be compatible with other blanket materials. More importantly, the compound must breed and release tritium at sufficient rates to fuel the reactor, yet limit the tritium inventory in the blanket to reasonable levels.

Lithium-lead compounds are interesting materials. Figure 2.3 shows the phase diagram of a Li-Pb two component mixture. However, beyond  $\text{Li}_7\text{Pb}_2$ , considered the most attractive lithium-lead alloy for breeding purposes, little else on the subject of Li-Pb physical properties is certain. An accumulation of data relevant to this study through a literature search and "data synthesis" is presented in Appendix A.

Two neutronic characteristics of the lithium-leads, breeding capability and long-lived activation products, have been studied. In the latter only  $^{205}\text{Pb}$  presents any problem in the long term unless significant amounts of impurities exist. This is not expected to be serious [4]. Breeding



Pb-Li PHASE DIAGRAM

Figure 2.3  
(Reference 4)

capability depends on the amount of lead present.

$\text{Li}_{17}\text{Pb}_2$  has been examined in detail and exhibits an excellent breeding property. This is due to the presence of lead, which acts as a neutron multiplier. Figure 2.4 shows the effect of lithium concentration in the lithium-lead breeder on the total breeding ratio. The total bulk shield thickness required to protect toroidal field coils is shown for reference. It is evident that a reduction of the lithium concentration will cause a substantial loss of fuel multiplication. However, the reduction of lithium also tends to enhance magnet protection by virtue of more effective nuclear radiation shielding by the added lead [5].

A diffusion study by Wiswall [6] shows that the solubility of tritium in Li-Pb is much lower than that in pure lithium, as the activity of lithium is very low due to the presence of lead. Thus, the tritium inventories can be much smaller, tritium diffusivities relatively higher, and tritium recovery much easier. Specifically, Fig. 2.5 shows that  $\text{Li}_{17}\text{Pb}_2$  has the lowest inventory of any of a number of proposed breeders. A problem here could be the fact that  $\text{Li}_{17}\text{Pb}_2$  becomes a "chunk" at high temperature. This would increase the diffusion path of tritium and make recovery difficult [4].

A more serious problem of the lithium-leads involves their compatibility with the blanket environment. All react to some extent with water. Table 2.1 shows the results of experimentation to investigate this at Argonne National Laboratory. It can be seen that Li-Pb alloys can react vigorously with water, more so at elevated temperatures. However, a significant result is that  $\text{Li}_{17}\text{Pb}_{83}$  exhibited only moderate reaction with water. Also, insufficient hydrogen was evolved by the alloy reactions to attain ignition conditions, unlike the case with liquid lithium.

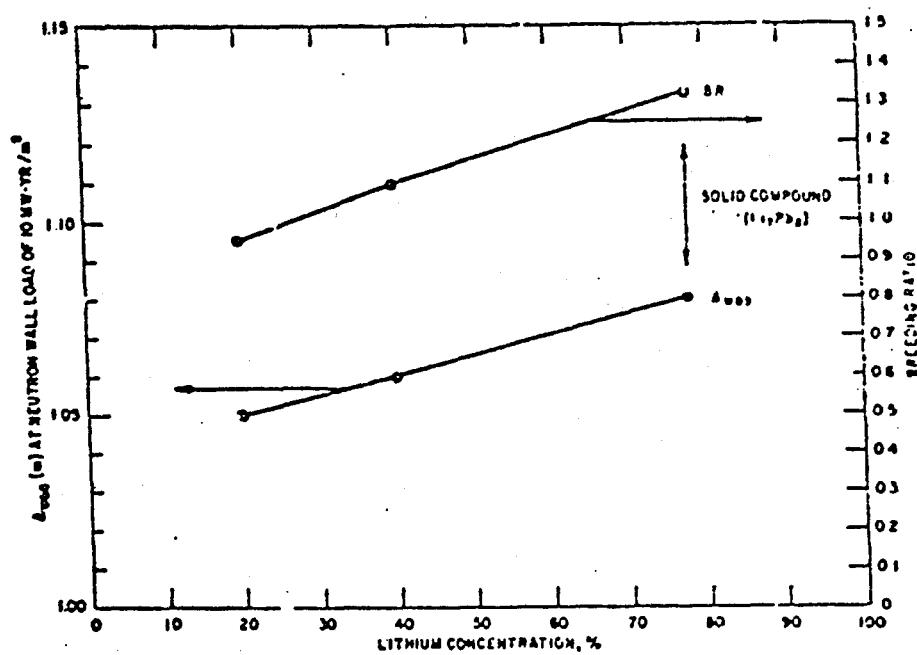


Figure 2.4 Effects of lithium concentration in a lithium-lead breeder on shielding requirement and tritium breeding (Reference 5).

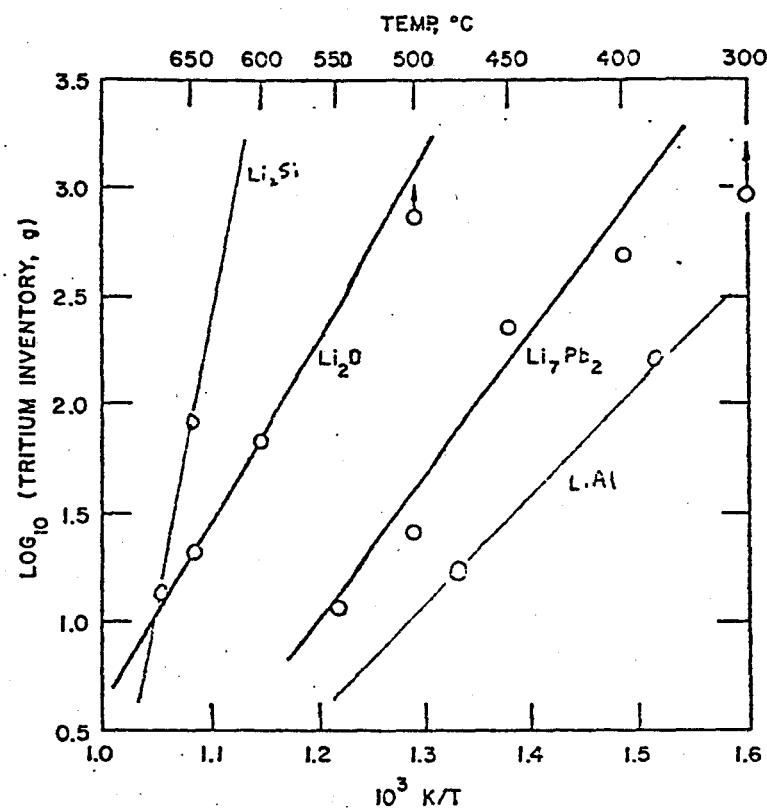


Figure 2.5 Estimated tritium inventory in alternative breeder blankets for a 3000 MW<sub>th</sub> reactor (Reference 9).

TABLE 2.1

Reactions of Li-Pb Alloys and Lithium  
with Water

Case	Sample			Water Temp/°K	Reaction
	Composition	State	Temp/°K		
1	Li <sub>7</sub> Pb <sub>2</sub>	s	773	298	Modest
2	Li <sub>7</sub> Pb <sub>2</sub>	s	773	369	Vigorous
3	Li <sub>7</sub> Pb <sub>2</sub>	s	873	368	Vigorous
4	Li <sub>7</sub> Pb <sub>2</sub>	l	1103	368	Very Vigorous
5	Li <sub>0.62</sub> Pb <sub>0.38</sub>	l	773	368	Vigorous
6	Li <sub>0.17</sub> Pb <sub>0.83</sub>	l	773	368	Very modest
7	Li	l	773	368	H <sub>2</sub> Detonation
8	Li <sup>a</sup>	l	773	368	Detonation

<sup>a</sup> Injected under water

(Reference 7)

In all reactions, LiOH (melting point at 470 °C) is formed. Liquid LiOH is an extremely corrosive substance and would degrade the integrity of the activated structural materials [7]. Fortunately, Li-Pb is relatively inert in air. It was reported [8] that "LiPb (50-50 mixture) resembles Pb in every respect except density. LiPb would not ignite, even when exposed to the flame of a gas-air torch." Therefore, though a Li-Pb - H<sub>2</sub>O reaction could be very serious, it is not expected to be as severe as an accident involving pure lithium. Table 2.2 provides a summary of the advantages and disadvantages of the lithium-lead alloys.

### 2.2.2 Alternative Breeders

For completeness, LiAl and Li<sub>2</sub>O, also candidates for the tritium breeding material, will be analyzed with regard to safety in this study. Other strong candidates, specifically Li<sub>2</sub>SiO<sub>3</sub> and LiAlO<sub>2</sub>, have been ignored in this study since they have no appreciable reaction with water.

LiAl is in many ways akin to Li<sub>7</sub>Pb<sub>2</sub>, such as a similarity in reactivity with water. However, the latter material is preferred by designers due to a superior tritium breeding capability. This is because there is a lower lithium-atom density in LiAl and no neutron multiplier, with the absence of Pb. Tritium extraction characteristics are also poorer than that of Li<sub>7</sub>Pb<sub>2</sub> [9]. The primary activation product is <sup>26</sup>Al, more of a problem than <sup>205</sup>Pb. Physical properties of this breeder are also unexplored and are discussed in Appendix A. The phase diagram is shown in Fig. 2.6.

Li<sub>2</sub>O has a marked advantage over the lithium-lead alloys in the matter of reaction with water. Because it is in oxide form, this compound does not evolve hydrogen upon reaction and produces less heat [9]. It is

TABLE 2.2  
Summary of Favorable and Unfavorable Features of  
Lithium-Lead Breeders

Lithium-Lead Alloys

<u>Advantages</u>	<u>Disadvantages</u>
1. High breeding ratio attainable	1. Poor technology base: high degree of uncertainty in properties
2. Probably less reactive with water than liquid Li	2. Reactive with water coolant
3. Tritium recovery appears feasible with low-pressure helium	3. High blanket weight
4. Lead helps shield magnets	4. Uncertain radiation damage effects
	5. Uncertain tritium release mechanism
	6. Requires blanket changeout during reactor life
	7. Activation product: $^{205}\text{Pb}$

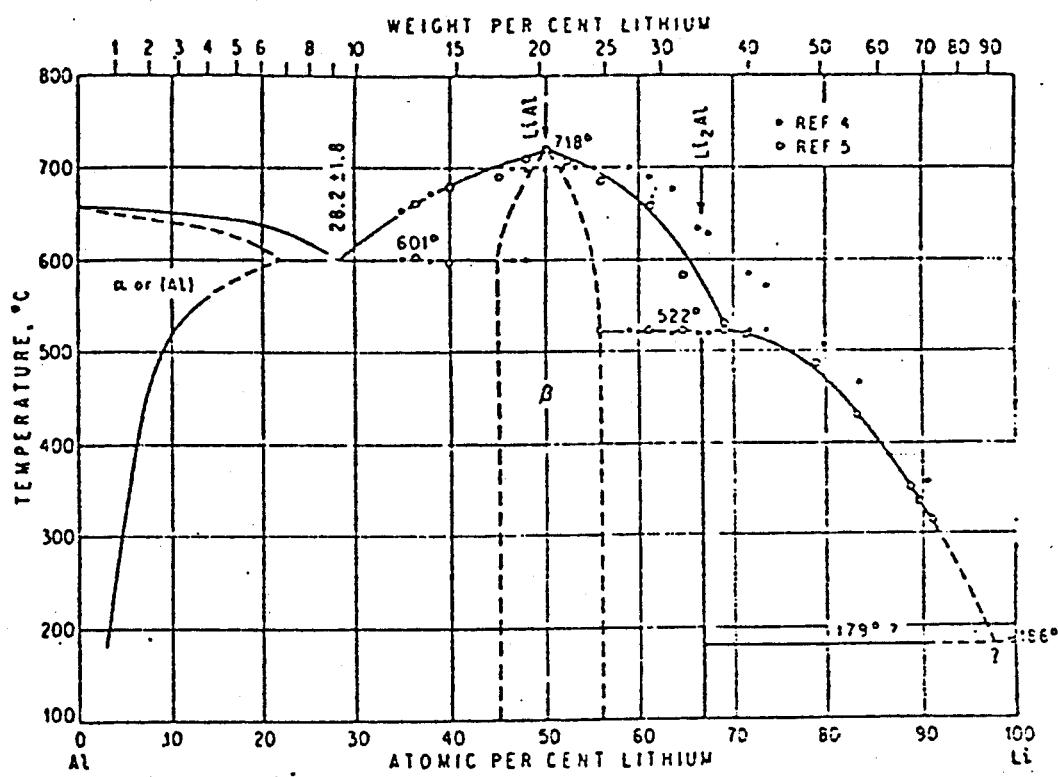


Figure 2.6 Phase Diagram for LiAl System.  
(Reference 9)

therefore expected that the use of this breeder will not pose a major safety problem. However,  $\text{Li}_2\text{O}$  has fallen into disfavor because it has a high lithium-atom density. This, combined with a poor diffusion rate, leads to excessive tritium inventories. Physical properties of this breeder are also discussed in Appendix A.

### 2.3 Coolant

Many factors affect the choice of coolant. Unique in power production to fusion is the interrupted burn cycle of the plasma. This produces the problem of thermal fatigue in the first wall and blanket, caused by the fluctuating temperature of these structures with the plasma burn cycle. The temperature change is a combination of three effects:

1. Coolant temperatures rise,  $T_{c,out} - T_{c,in}$
2. Film temperature drop,  $T_{wall} - T_c (=Q/h)$
3. Temperature difference across the first wall,  $\Delta T = Q \chi/k$  [4].

While the third effect depends on structural materials, the first two can be greatly reduced by choice of the proper coolant, one which has a small coolant temperature rise that simultaneously provides a large heat transfer coefficient. This describes a boiling liquid and the first logical choice is boiling water.

A considerable technology has been developed through the years for the use of water as a coolant in energy production. There exist many advantages. Water is an excellent heat-transfer fluid, costs little and is readily available. It is easy to pump and is non-corrosive with conventional structural materials.

It is interesting to note, however, that up to this point, few designers have considered water for the primary coolant. There exist a number of concerns and questions including neutronics, tritium, and safety considerations. In addition, the use of a boiling water or steam coolant will require high pressure containment. The NUWMAK design calls for cooling water at 300 °C and 8.6 MPa (1250 psi). This poses additional design and safety problems.

The safety concerns of a water-cooled blanket have been discussed. It should be noted that this problem effectively prohibited the use of water as a coolant until alternative breeders to lithium were suggested. Even so, current designs using water stress the use of strong cladding materials for coolant channels.

High-integrity cladding is also the prescription to minimize tritium diffusion into the cooling water. Tritiated water is a safety hazard and recovery of the tritium is difficult and expensive. However, recent studies using permeation rates for stainless steel cooling tubes show tritium losses could be less than 1 Ci/day, assuming the formation of oxide films inside the tubes [9]. Irregardless, it is obvious that the coolant loop cannot be used for tritium recovery, necessitating some recirculation of the breeder.

The necessity of avoiding contact between breeding materials and coolant tends inherently, to increase structural material content in the blanket. From a neutronics point of view, this increase tends to degrade the tritium breeding ability, due to increased parasitic neutron absorption. However, the strong neutron slowing-down power of water improves breeding performance by increasing the  $^{6}\text{Li} (\text{n},\text{T}) ^{4}\text{He}$  reaction rate for low energy neutrons. This decreases parasitic absorption by

the structural materials [5]. Therefore, proper design and choice of materials can leave the tritium breeding capability of a water-cooled blanket virtually unchanged.

Figure 2.7 shows this phenomenon for various breeders, with 316 stainless steel used as structural material. It is important to note that the addition of water significantly improves the breeding ratio in the  $\text{Li}_7\text{Pb}_2$ , more so than the other breeders. This is due to reduced parasitic absorption in lead as well as the stainless steel.

Table 2.3 summarizes the advantages and disadvantages of the water-cooled blanket concept. Chief among the assets is the use of available technology in its construction. The major liabilities appear to be the safety problem and limited choices in compatible breeding and structural materials. Further study in both areas is necessary before a final decision can be made.

## 2.4 NUWMAK Blanket Design

### 2.4.1 Structural Materials

Many structural materials have been considered for a water-cooled blanket design, including: austenitic stainless steel, high nickel alloys and selected titanium, vanadium and niobium alloys [9]. Table 2.4 summarizes an assessment of these candidates with regard to properties associated with the blanket environment.

One of the liabilities of the water-cooled design is apparent. The vanadium and niobium materials, which respond well to neutron bombardment, are eliminated due to water corrosion problems. Also noted is a general lack of data regarding the compatibility of these structural materials with the lithium-lead alloys. Since decomposition of these alloys is not

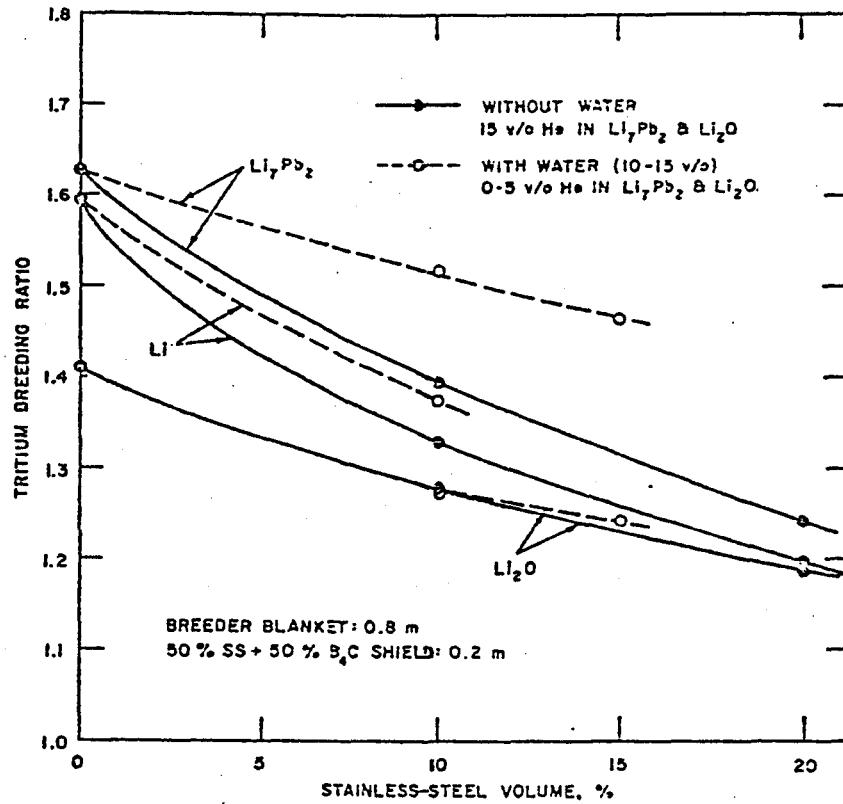


Figure 2.7 Impact of Structural Material Content (316 SS) on Tritium Breeding (Reference 6).

TABLE 2.3  
Summary of Favorable and Unfavorable Features of  
the Water-Cooled Blanket Concept

<u>Water Coolant</u>	
<u>Advantages</u>	<u>Disadvantages</u>
1. Excellent heat-transfer fluid	1. Highly reactive with candidate breeding materials
2. Well-developed technology base	2. Reaction product (LiOH) is very corrosive
3. Low cost and readily available	3. Requires high-pressure containment
4. Relatively low temperature ( $\sim 320^{\circ}\text{C}$ ) operation	4. Cannot be used for tritium recovery
5. Compatible with conventional structural materials	5. Expensive to remove tritium from $\text{H}_2\text{O}$ (safety)
6. Low pumping power need	6. Water tends to be a sink for tritium
7. Enhances tritium production	7. Nb and V, candidate structure materials are incompatible
8. Liquid at room temperature	

TABLE 2.4  
Summary of Structural Material Assessment for the  
Water-Cooled Blanket Concept

Property Requirement	Rating*				
	Fe	Ni	Ti	V	Nb
Bulk Radiation Effects	2	2	?	1	1
Compatibility with H <sub>2</sub> O	1	1	1	4	4
Compatibility with Liquid Li	3	5	3	1	1
Compatibility with Solid Li <sub>2</sub> O and Li <sub>7</sub> Pb <sub>2</sub>	3	3	3	3	3
Compatibility with H(DT) Environment	1	1	3	1	1

\* Rating numbers defined as follows:

1. Compares favorably with other candidate structural materials.
2. Limits operating life but probably acceptable under certain conditions.
3. Little data available but may be a limiting factor.
4. Probably not viable for conditions of interest.

desirable, this question should be studied.

The NUWMAK design utilizes Ti-6Al-4V alloy for the first wall and coolant tube material due to its high strength-to-weight-ratio, good fatigue resistance, fabricability, low long term residual activity and well established industry [4]. Physical properties of this alloy are shown in Table 2.5. The NUWMAK shield is more conventional, primarily B<sub>4</sub>C.

#### 2.4.2 Mechanical Design

The blanket of NUWMAK is shown in Fig. 2.8. The blanket structure is Ti-4Al-4V which operates at a temperature of approximately 350 °C. The coolant is boiling water at 300 °C and 1250 psi. The breeder is Li<sub>62</sub>Pb<sub>38</sub> eutectic, operating at approximately 400 °C. The design life for each blanket module is two years.

The blanket is divided into eight modules in the reactor. Each module is fed and discharged coolant and breeding materials separately. There are two blanket units in each module; the inner blanket near the machine axis and the outer blanket, as seen in Fig. 2.8. These two units are completely separate from each other.

The first wall consists of a continuous bank of tubes running in the vertical direction. Beyond this, the blanket is cooled with rows of vertical tubes on a triangular pitch. The spacing between rows of tubes is progressively increased towards the back of the blanket to account for the radially decreasing nuclear heating [4]. Radial struts are spaced at 20 cm intervals, reinforcing the first wall against the hydrostatic pressure of the breeding material, which fills the space between the coolant tubes.

TABLE 2.5  
Physical Properties of Ti-6-4

Atomic Weight	45.9
Melting Point	1668 °C
Mass Density	4.4 g/cm <sup>3</sup>
Yield Strength	530 MPa
Modulus of Elasticity	85 GPa
Yield-to Weight Ratio	120 N-m/g
Thermal Conductivity	.12 W/cm-K
Coefficient of Thermal Expansion	10 x 10 <sup>-6</sup> /°C
Heat Capacity	668.8 J/kg-K

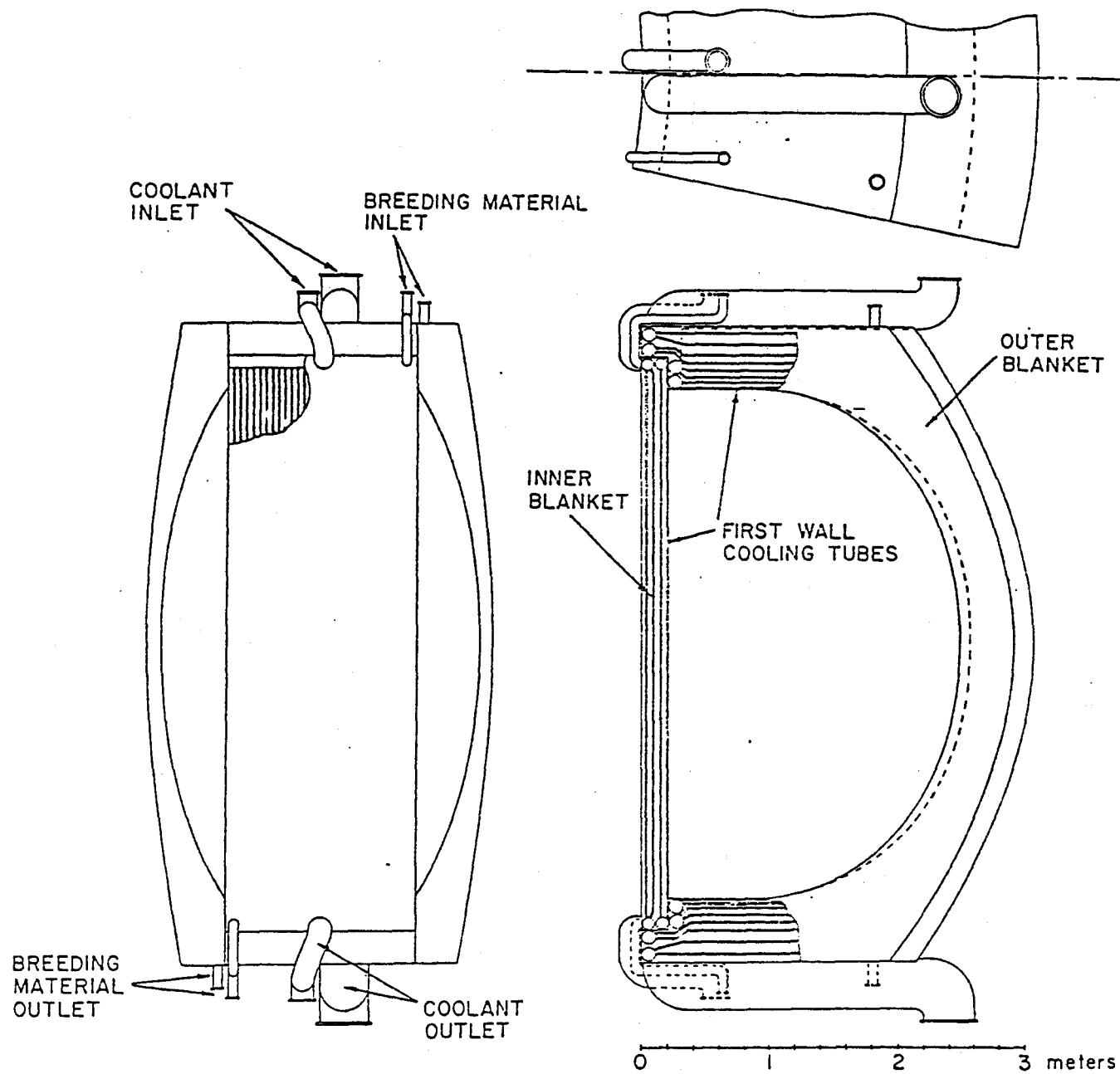


Figure 2.8 Cross-Sectional View of Blanket (Reference 4).

At the blanket's edge is a thin graphite reflector. Beyond this is a shield to protect the cryogenic magnet coils. The shield is primarily  $B_4C$ , operating at approximately 150 °C. Figure 2.9 shows a schematic diagram of this system.

#### 2.4.3 Summary of Important Parameters

The major features of the NUWMAK design are given in Table 2.6. In addition, important blanket parameters pertinent to this study are given in Table 2.7. These values are used where appropriate in subsequent calculations.

## SCHEMATIC OF THE BLANKET AND SHIELD FOR NUWMAK

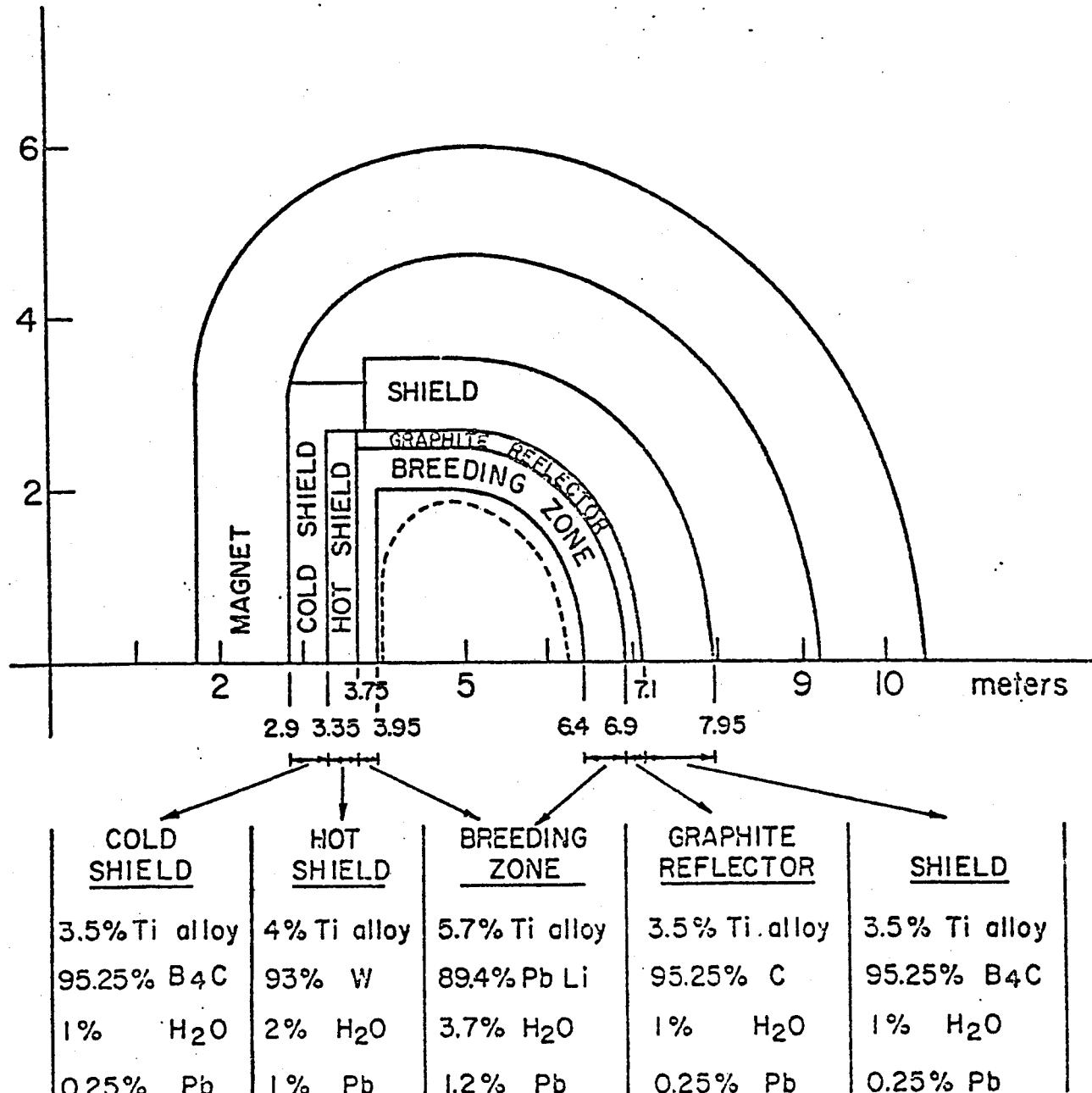


Figure 2.9  
(Reference 4)

TABLE 2.6  
Major Features of NUWMAK Design

Power

Total Thermal Power	2283 MW <sub>th</sub>
Net Electric Power	660 MW <sub>e</sub>

Plasma

Major Radius	5.13 m
Minor Radius	1.13 m
Plasma Height to Width Ratio (b/a)	1.64
Plasma Current	7.2 MA
Toroidal Beta	6%
$n_e \tau_E$	$2 \times 10^{14} \text{ cm}^{-3}\text{-sec}$
q(a)	2.64

Magnet

On-Axis Toroidal Field	6.05 Tesla
Toroidal Field at NbTi Conductor	11.5 Tesla
Stabilizer	Aluminum
Number of Toroidal Field Coils	8
Number of Cu Trim Coils	16

Blanket

Structural Material	Titanium Alloy
Coolant	Boiling Water
Breeding Material	$\text{Li}_{62}\text{Pb}_{38}$
Average Neutron Wall Loading	$4.34 \text{ MW/m}^2$

TABLE 2.7  
Summary of Important Blanket Parameters

Plasma Burn Time	225 sec
Plasma Down Time	20 sec
Coolant Temperature	300 °C
Coolant Pressure	8.6 MPa
Total Coolant Flow Rate	1500 kg/sec
Total Coolant Tube Surface Area	4350 m <sup>2</sup>
Heat Transfer Coefficient of Boiling Water	20000 Btu/hr-ft <sup>2</sup> - °F
Coolant Tube OD	1.3 cm
Coolant Tube ID	1.0 cm
N. Tubes in Outer Blanket Module	475
Pitch Length	12 cm
Space for Breeder in Outside Blanket Module	17.72 m <sup>3</sup>
Breeder Temperature	400 °C
Shield Temperature	150 °C



## CHAPTER 3. EQUILIBRIUM $T_f$ CALCULATION

### 3.1 Introduction

Lithium-lead alloys are considered less of a safety hazard due to the presence of lead, which is thought to slow down the water reaction, decrease the heat of reaction and help absorb what heat is released. However, a number of physical properties are altered with the addition of lead. Since most of these properties have direct bearing on an interaction with water, the consequences of such an interaction are not directly predictable.

For this reason, a preliminary analysis of the Li-Pb - H<sub>2</sub>O reaction is performed using a static calculation. In this case, the breeder inside one blanket module is allowed to interact completely with varying amounts of the water available to that module. Assuming the heat of reaction is contained within the blanket, the equilibrium temperature of the reaction products, unreacted breeder and blanket structural materials is then determined.

Such a scenario is unrealistic, but this calculation is important for two reasons. First, it serves as a reference for further study. Second, with its assumptions, such a calculation may indicate the maximum attainable blanket temperature in a particular module in the case of an internal blanket water interaction.

### 3.2 Assumptions and Methodology

The outer blanket section of an individual module is chosen for consideration. The inner blanket section contains only a nominal amount of breeder and the consequences of an accident in that section do not

appear as severe. Data from the NUWMAK design (Table 7.H.2) indicates that an outer blanket module contains 38.4 tonnes of titanium structural material, 106.0 tonnes of graphite and  $17.7 \text{ m}^3$  of space to contain the breeder. Therefore, the amount of breeder present can be determined with knowledge of the density.

The initial temperature of breeder and graphite is 400 °C. The structural materials are at a temperature of 350 °C and the coolant is at 300 °C and 1250 psi. The reaction between breeder and coolant is assumed to be immediate and complete at 400 °C, the heat of reaction helping to raise the water temperature to that point. The heat of reaction can be determined using:

$$\Delta H_r = \Delta H_{25}^\circ + \Sigma \Delta H_{\text{prod}} - \Sigma \Delta H_{\text{react}}, \text{ cal/g breeder} \quad (3.1)$$

where  $\Delta H_{25}^\circ$  is the standard heat of hydrolysis at 25 °C and  $\Delta H_{\text{prod}}$  and  $\Delta H_{\text{react}}$  are the enthalpy changes of reaction products and reactants, respectively, as they are heated from 25 °C to 400 °C.

The amount of coolant water available to the outer blanket module can be determined by analysis of NUWMAK's steam generating unit. In this respect, NUWMAK is very much like a fission boiling water reactor [4]. Examination of the Dresden BWR reveals that the total coolant volume in the 3411 MW<sub>th</sub> plant's cooling system is 11,695 ft<sup>3</sup> [10]. Scaling this down to NUWMAK's 2283 MW<sub>th</sub> output and assuming the average density of water in the coolant loop to be 62.37 lb/ft<sup>3</sup> (an overestimate), it can be shown that approximately 15,260 pounds of water are available to interact with the breeder in one module.

It is assumed that a fixed percentage of this cooling water interacts with the breeder. Thereafter, no loss of heat is allowed from the blanket. The resulting final equilibrium temperature can be calculated using the expression

$$T_f = T_0 + \frac{Q_R}{(M_s \bar{C}_s + M_b \bar{C}_b + \sum M_{Ri} \bar{C}_{Ri})}, \quad (3.2)$$

where  $T_f$  = final equilibrium blanket temperature

$T_0$  = initial blanket temperature

$Q_R$  = reduced heat of reaction

$M_s$  = structural mass

$M_b$  = remaining breeder mass after reaction

$M_R$  = reaction product mass after reaction

$\bar{C}_i$  = mean specific heat for each component

Reaction products include hydrogen gas, LiOH and the alloy element.

A proper evaluation of the  $M_i \bar{C}_i$  terms in the above equation varies with each component. For example, LiOH is evaluated above its melting point as

$$\frac{M_{LiOH}}{\bar{C}_{LiOH}} = \frac{M_{LiOH} [\bar{C}_s (T_{melt} - T_0) + \Delta H_{melt} + \bar{C}_L (T_f - T_{melt})]}{T_f - T_0} \quad (3.3)$$

where  $\bar{C}_s$  = mean solid specific heat of LiOH

$\bar{C}_L$  = mean liquid specific heat of LiOH at 470 °C

$\Delta H_{melt}$  = heat of melting for LiOH at 470 °C

$T_{melt}$  = melting point of LiOH (470 °C)

Similar expressions can be written for the other components.

The reduced heat of reaction can be written as

$$Q_R = x M_c [a_0 \Delta H_R - \bar{C}_c (T_o - T_c)] - M_{Ti} \bar{C}_{Ti} (T_o - T_{Ti}), \quad (3.4)$$

where  $M_c$  = mass of total available coolant

$M_{Ti}$  = mass of titanium structural material

$\bar{C}_c$  = mean specific heat of coolant

$\bar{C}_{Ti}$  = mean specific heat of titanium structural material

$T_c$  = coolant temperature

$T_{Ti}$  = titanium structural material temperature

$x$  = fraction of available coolant reacting

$a_0$  = reaction stoichiometric combination constant

This is done to raise the coolant and titanium alloy structure to the initial blanket temperature of 400 °C.

### 3.3 Results and Discussion

Figure 3.1 shows the resulting equilibrium blanket temperatures for the various breeders under consideration, plotted against the reacting percentage of available water. A number of interesting results are noted.

First, as expected, the pure lithium breeding blanket reached the highest temperature upon reaction with water. The dotted line signifies that with a high percentage of available water reacting, some of the unreacted lithium will begin to vaporize at a temperature also close to the melting point of steel. This indicates a potential for further problems if a steel blanket liner is used.

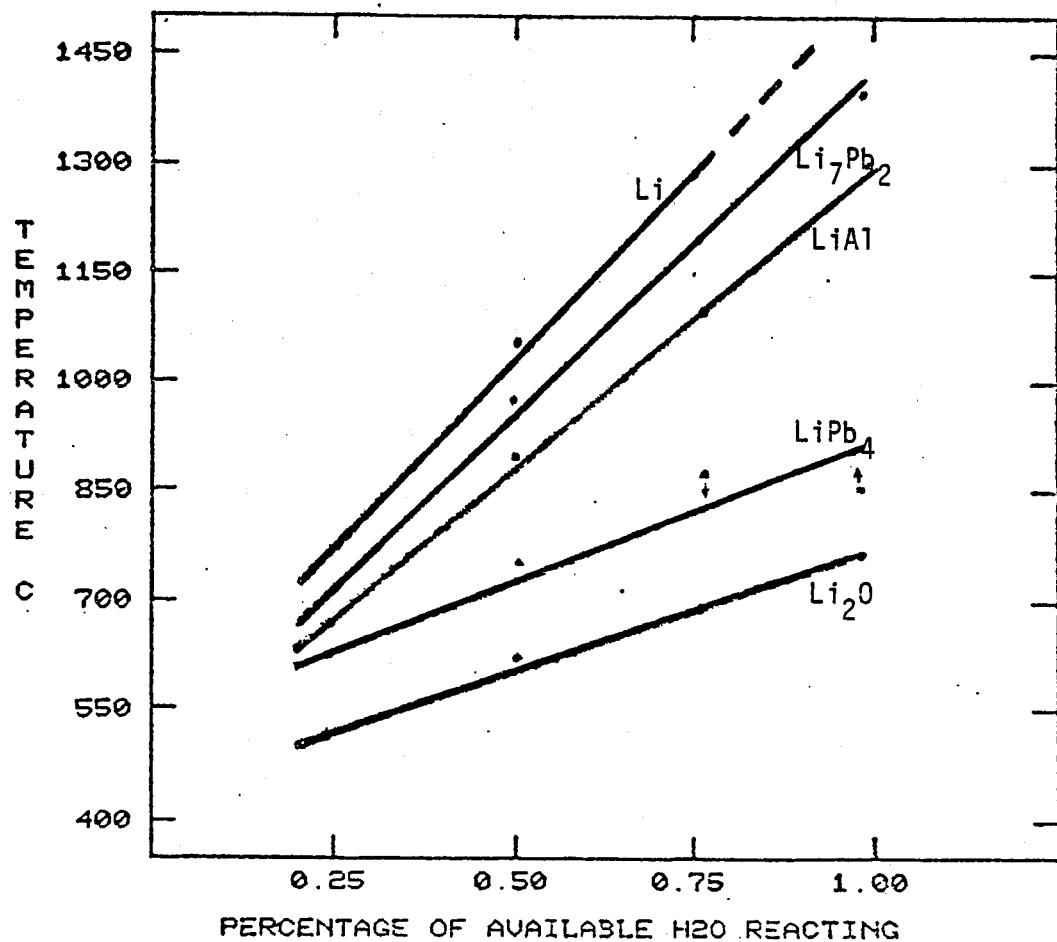


Figure 3.1 Equilibrium Final Temperature Profiles for Various Breeders in the Static Calculation.

$\text{Li}_7\text{Pb}_2$  and LiAl are very much alike. Though lower equilibrium temperatures are exhibited than those of the pure lithium breeder, the difference is not very large to be significant. At low percentages of reacting available water, there is no difference.

$\text{Li}_2\text{O}$  and  $\text{LiPb}_4$ , on the other hand, appear to be significantly "cooler" than the pure lithium case. In the case of  $\text{Li}_2\text{O}$ , the key difference is a very low heat of reaction with water.  $\text{LiPb}_4$ , with a relatively high density, is the only case in which there exists more available water than breeder. Thus, a limited heat of reaction and large residue of lead leads to reduced equilibrium temperatures.

## CHAPTER 4. DYNAMIC CALCULATION USING LITFIRE

### 4.1 Introduction

Though valuable as a reference, the calculations of Chapter 3 have little to do with a plausible internal blanket breeder-coolant interaction. It is incorrect to assume that these materials will react instantly at a constant temperature; it is imprudent to declare that the flow of cooling water will cease and that all heat will be retained within the blanket perimeter. Clearly, a dynamic formulation is needed.

To this end, the LITFIRE computer code is modified to estimate the thermal response of the NUWMAK blanket to possible accidents. In this modification, called the internal blanket accident option, the breeder and water react in a zone located in the middle of the breeder mass. The leakage of water into this "reaction zone" is determined by the number of broken coolant tubes, set small enough to justify the assumption that this is the limiting effect on the reaction rate. The heat of reaction is transferred to the breeder mass by conduction and free convection, to the blanket liner and shield by further conduction, and out of the blanket via forced convective cooling by unbroken coolant tubes. Figure 4.1 shows the heat flow diagram for this system.

It is hoped this model presents a truer picture of what will happen within the blanket in the event of a cooling system leak. Again, due to uncertainties concerning data and some assumptions, this study can only provide a measure of relative safety, compared with the trials utilizing liquid lithium.

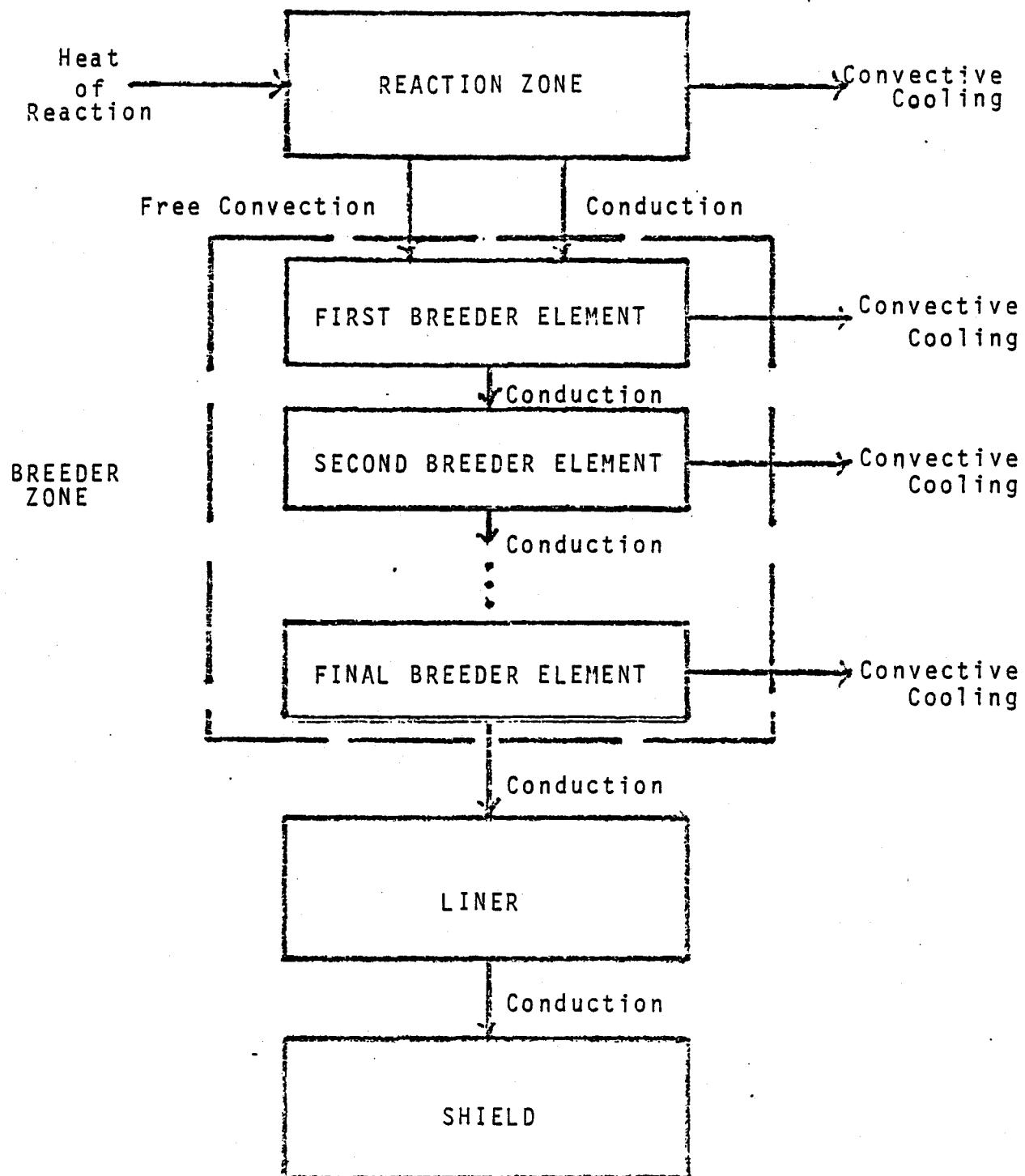


Figure 4.1 Internal Blanket Accident Option Heat Flow Diagram.

#### 4.2 LITFIRE Description

LITFIRE is a computer code developed at MIT [11] to predict the consequences of a hypothetical lithium spill in a fusion reactor containment. It was first written in 1977 as a modification of the Argonne National Laboratory code SPOOLFIRE, used to model the consequences of sodium fires. LITFIRE was later modified and improved in 1980 [3], utilizing the experimental results of small-scale lithium spill tests performed at the Hanford Engineering Development Laboratory.

In this code, the flow of heat is traced from the lithium reaction zone source to reactor containment components, and eventually out to the ambient. This system is simulated by a nodal network in which each node has a heat capacity and temperature equal to that of its physical counterpart. Heat flows between nodes are calculated using standard heat transfer correlations.

To provide the reactor containment thermal and pressure response, LITFIRE solves a set of coupled heat and mass transfer equations. This is done by using the method of finite differences for the spacial dimensions, and either Simpson's rule or a Runge-Kutta method in the time domain [3]. Properties are computed at each time step from the integral equation

$$Y(t) = Y(t_0) + \int_{t_0}^t dt' dY/dt',$$

where the rates of change  $dY/dt$  are given for each node by finite difference solution of the heat transfer relations.

#### 4.3 Internal Blanket Accident Option

##### 4.3.1 Assumptions and Structural Model

The internal blanket accident option models an accidental interaction of coolant water and lithium-based breeder in the center of an outside blanket module. This interaction is caused by a breach of several neighboring coolant tubes, while the reactor as a whole undergoes normal operation.

It is assumed that this event is undetected, thus assuring continuance of the plasma burn and coolant recirculation. It is felt that the three monitored parameters relevant to an accident of this type, namely bulk breeding material temperature, coolant temperature, and coolant flow rate, will not change appreciably until later stages of the accident.

The reaction rate is immediate and limited by the leakage of water into the breeding material. The leakage rate, dictated by the number of broken coolant tubes, is set very low, 0.6 kg/sec (three broken tubes), to make this assumption reasonable. Although there is a suspicion that the water reaction rate of the lithium-lead alloys is slow at low temperatures, no data exists. It is also assumed that the reaction zone pressure at high temperatures does not significantly retard the leakage rate of water into the zone.

The reaction zone is very difficult to characterize. However, certain assumptions can be made. First, the zone can be considered spherical, as boiling water at 8.6 MPa will disperse equally in all directions upon tube rupture. The reaction zone must be large enough to accomodate the influx of breeder and coolant, thus the zone radius

should be large compared to the coolant tube pitch length. However, to accomodate the assumption that the water reaction is instantaneous, the reaction zone volume should be small, compared to that of the blanket module.

The initial radius of the reaction zone for three ruptured coolant tubes is set at one foot. This is over six times the characteristic distance separating the three coolant tubes in question. The volume of this zone is less than 2% of the total blanket volume. This seems reasonable for the small leakage rate. Further research in this area would aid the accuracy of the model.

The reaction products (LiOH and alloy element, if any) remain in the reaction zone, thus increasing the radius of this zone with time. The heat capacity of this expanded zone is calculated summing the products of the individual heat capacity of each component multiplied by the weight percent.

The nodal structure of this system is shown in Figure 4.2. It can be seen that the reaction zone and breeder mass are broken into a number of sections. This is to more accurately account for heat transfer by conduction. The number of sections in each zone is selected so as to keep the element widths to approximately 6 inches. These elements increase and decrease in width with time in the reaction zone and breeder zone, respectively, as the reaction zone expands.

The blanket elements are spherical, like the reaction zone, to facilitate computation. The outer element is therefore irregularly shaped to account for the NUWMAK geometry. This is not expected to create any difficulty, as it is doubtful that much heat will be

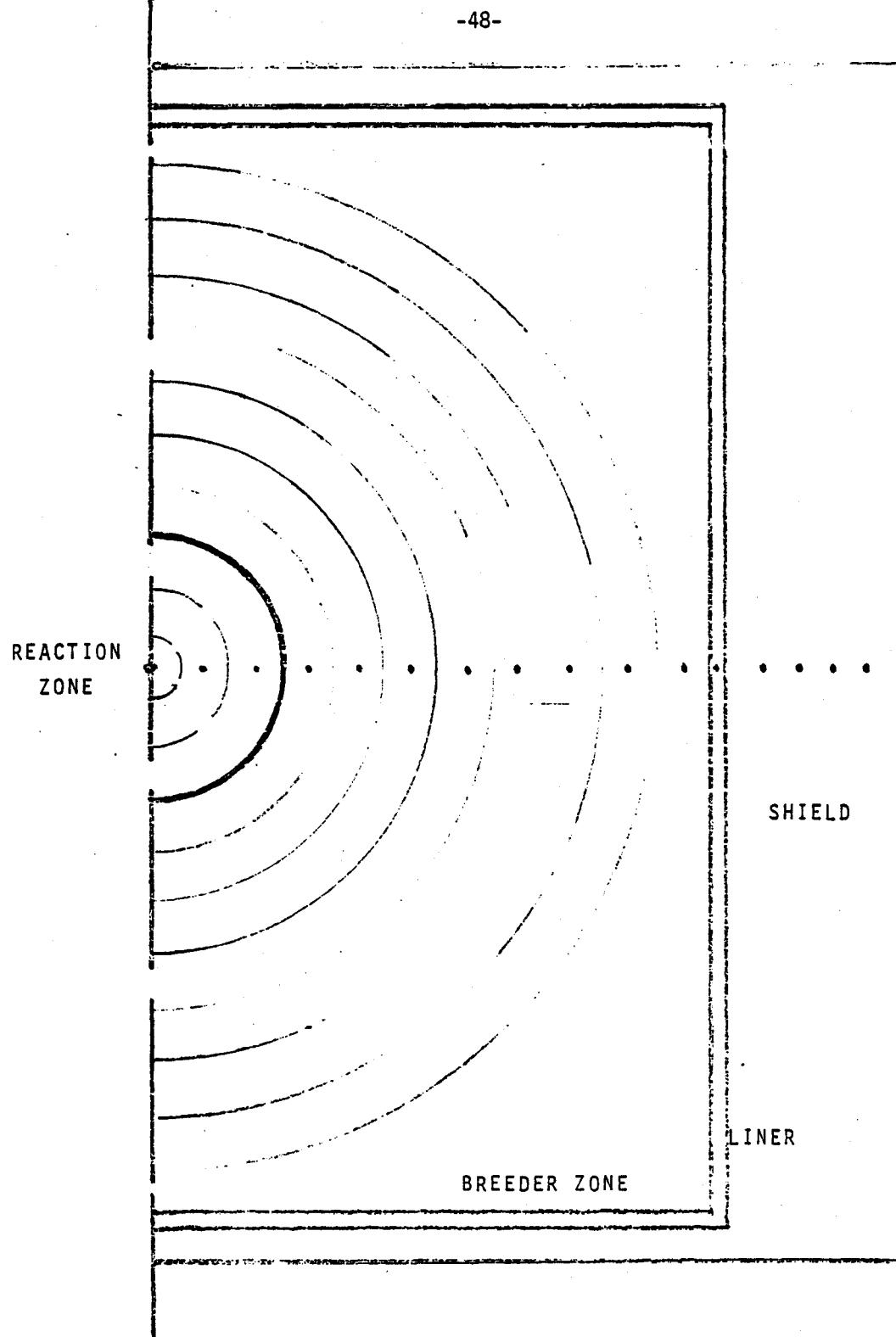


Figure 4.2 Internal Blanket Accident Option  
Node Structure.

transferred to this region. For completeness, the steel liner and  $B_4C$  shield are also monitored in the code. The conduction of heat is assumed to stop at the far edge of the shield.

The heat of reaction is distributed evenly throughout the reaction zone, heating the reactants and reaction products. Heat is removed by conduction and free convection to the breeder zone, and via forced convection by unbroken coolant tubes as shown in Fig. 4.1. The surface area of cooling tubes in each element is calculated as a volume percentage of the total.

Finally, it should be noted that the densities and thermal conductivities of the lithium-lead alloys and other alternate breeders are held constant with temperature in this analysis. As discussed in the appendix, this data has only been determined at one temperature. Rather than increase uncertainties with various correlations, the values are unaltered.

#### 4.3.2 Heat Transfer Mechanisms

##### A. Heat of Reaction

All of the alternate breeders considered in this study react to some extent with water. Table 4.1 shows the reactions of interest. Other reactions also take place, such as the production of  $Li_2O_2$ , but are discarded as they play a very minor role. For example, the peroxide is unstable above 250 °C [11] and is not produced above that temperature.

Reaction occurs at the reaction zone temperature  $T_{cz}$ . The heat of reaction can be calculated using

$$\Delta H_R = H_{25}^\circ + \sum \Delta H_{prod} - \sum \Delta H_{react}, \quad \text{Btu/lb breeder} \quad (4.1)$$

TABLE 4.1  
Breeder-Coolant Reactions of Interest

	$\Delta H_{hyd}$ (kJ/g-atom of Li)
$Li + H_2O \rightarrow LiOH + 1/2 H_2$	205
$1/7 Li_7Pb_2 + H_2O \rightarrow LiOH + 1/2 H_2 + 2/7 Pb$	200
$LiPb_4 + H_2O \rightarrow LiOH + 1/2 H_2 + 4Pb$	170
$LiAl + H_2O \rightarrow LiOH + 1/2 H_2 + Al$	200
$Li_2O + H_2O \rightarrow 2LiOH$	64

where  $H_{25}^o$  is the standard heat of hydrolysis at 25 °C and  $\Delta H_{prod}$  and  $\Delta H_{react}$  are the enthalpy changes of the reaction products and reactants, respectively, as they are heated from 25 °C to  $T_{cz}$ .

Since the reaction is immediate, limited by the leakage of water into the reaction zone, the reaction rate is the leakage rate. This can be written as

$$R_w = \dot{m} N_T \quad \text{lb H}_2\text{O/sec} \quad (4.2)$$

where  $\dot{m}$  is the mass flow rate of water through one tube and  $N_T$  is the number of ruptured tubes. Thus, the total heat generation rate inside the reaction zone can be given by

$$Q = a_0 \Delta H_R \dot{m} N_T \quad \text{BTU/sec} \quad (4.3)$$

where  $a_0$  is the stoichiometric combination constant for the breeder and water in the given reaction.

#### B. Sensible Heat Addition to Reactants in the Reaction Zone

A portion of the heat of reaction is used to heat the inflowing coolant water and breeder to the reaction zone temperature. This can be written as

$$Q_s = N_T \dot{m}_w c_w (T_{cz} - T_c) + \dot{m}_b c_b (T_{cz} - T_L) \quad \text{BTU/sec} \quad (4.4)$$

where  $\dot{m}_w$  is the mass flow rate of water in a coolant tube  
 $c_w$  is the mean specific heat of the coolant

$\dot{m}_b$  is the mass influx of breeder to the reaction zone

$c_b$  is the mean specific heat of the breeder

$T_c$  is the coolant temperature

$T_L$  is the bulk breeder temperature.

In this case, the influx of breeder into the reaction zone is considered equal to the leakage rate of the coolant into the zone, as reaction is immediate. This mass transfer is further discussed in the free convection section.

### C. Forced Convective Cooling

Forced convection, due to the continued coolant recirculation through undamaged tubes, is an important heat transfer mechanism. Because only a small number of the 475 coolant tubes in an outside blanket module are damaged, cooling will take place in both the breeding and reaction zones. The cooling tube surface area in each element can be determined as a volume percentage of the blanket as a whole. For example, the initial reaction zone, approximately 2% of the blanket by volume, comes into contact with roughly 2% of the total cooling tube surface area. This total area can be determined using the information in Table 2.7.

This total heat flow can be computed for each element using

$$Q_c = \left[ \frac{\delta}{k_{Ti} A_{so}} + \frac{1}{h A_{sI}} \right]^{-1} (T_i - T_c) \quad \text{BTU/sec} \quad (4.5)$$

where  $\delta$  is the coolant tube thickness

$k_{Ti}$  is the thermal conductivity of the titanium alloy

$A_{so}$  is the outer coolant tube surface area

$A_{si}$  is the inner coolant tube surface area

$h$  is the boiling water heat transfer coefficient

$T_i$  is the bulk temperature of the element in question.

#### D. Conduction

Conduction plays a major role in the transfer of heat from the reaction zone to the breeder mass. The heat conduction term between two elements can be expressed as

$$Q_{cond_{ij}} = A_i \left[ \frac{k_i k_j}{k_i + k_j} \right] (T_i - T_j) / d_{ij} \quad \text{BTU/sec} \quad (4.6)$$

where  $A_i$  is the inner element surface area

$k_i$  is the inner element thermal conductivity

$k_j$  is the outer element thermal conductivity

$T_i$  is the inner element bulk temperature

$T_j$  is the outer element bulk temperature

$d_{ij}$  is the separation distance between the elements.

The surface area assigned to each element is at its outer perimeter. In the above expression, it is assumed that the inner element is at a higher temperature, as is the case at all times in the LITFIRE option.

#### E. Free Convection

A preliminary order of magnitude analysis indicates that the free convective enhancement to conduction is  $\text{Pr} \text{Gr}^{1/2}$ , where Pr and Gr are the Prandtl and Grashof numbers. For a 400 °C temperature

difference between the reaction and breeder zones in the lithium case, this enhancement is better than a factor of ten. Thus, free convection is an important mode of heat transfer in the model.

As mentioned before, there will be mass transfer in the breeding zone due to the influx of this material into the reaction zone. It is this movement that allows convective cooling of the reaction zone by the first breeder zone element.

Given the spherical shape of the reaction zone, the semi-empirical relation

$$\overline{Nu} = 2.0 + 0.60 \text{ Gr}^{1/4} \text{ Pr}^{1/3} \quad (4.7)$$

is useful to find the average heat transfer coefficient for  $\text{Gr}^{1/2}$ .  $\text{Pr}^{1/3} < 200$ .  $\overline{Nu}$  is the average Nusselt number and is related to the average heat transfer coefficient  $\overline{h}_m$  by

$$\overline{Nu} = \overline{h}_m L/k \quad (4.8)$$

where  $L$ , the characteristic distance, is in this case the reaction zone diameter.

Thus, the heat flow due to free convection can be described by

$$Q'_c = \frac{A_{cz} k_b}{D} (2.0 + 0.60 \text{ Gr}^{1/4} \text{ Pr}^{1/3}) (T_{cz} - T_L) \frac{\text{BTU}}{\text{sec}} \quad (4.9)$$

where  $A_{cz}$  is the reaction zone surface area and  $k_b$  is the bulk breeder thermal conductivity.  $\text{Gr} = \frac{D^3 \rho^2 g \beta \Delta T}{\mu^2}$  and  $\text{Pr} = \frac{c_p \mu}{k}$  are applicable to

the fluid breeders.

#### F. Radiation

Using an order of magnitude analysis, the radiative heat flow is related to the conductive heat flow by

$$Q_{\text{rad}} = \frac{\sigma T^3 L}{k} Q_{\text{cond}} \quad (4.10)$$

where  $\sigma$  is the Stefan-Boltzmann constant. At 1500 °F, this radiative heat flow is a factor of ten less than that of conduction in a lithium breeder. Therefore, radiation is neglected in the model.

#### 4.3.3 The Numerical Scheme

The temperature of a thermal element may be found from the solution to

$$mc \frac{dT}{dt} = q_1 + q_2 + q_3 + \dots, \quad T = T_0 \text{ at } t = t_0, \quad (4.11)$$

where  $mc$  is the element's heat capacity and  $q_1, q_2, q_3 \dots$  are heat flows into the element, shown in Fig. 4.1. This may also be expressed as

$$T = \int_{t_0}^t \frac{1}{mc} (q_1 + q_2 + q_3 + \dots) dt + T_0. \quad (4.12)$$

In LITFIRE, this is expressed as

$$T = \text{INTGRL}(T_0, dt/dt). \quad (4.13)$$

A set of sub-routines is used to perform the integrations using either Simpson's Rule or a Runge-Kutta method.

For example, heat flows into the reaction zone are the heat of reaction, sensible heat addition to the reactants, conduction, free convection and forced convective cooling. Therefore, the temperature of the reaction zone at time  $t$  can be determined in LITFIRE by using Eq. (4.13) and

$$\frac{dT}{dt} = \frac{1}{mc} [Q - Q_s - Q_{cR} - Q_{cond_{RL1}} - Q'_c] \quad (4.14)$$

where the subscript R denotes the reaction zone and L1 denotes the first breeder element.  $Q$ ,  $Q_s$ ,  $Q_{cR}$ ,  $Q_{cond_{RL1}}$ , and  $Q'_c$  can be determined using Equations (4.3), (4.4), (4.5), (4.6), and (4.9), respectively.

Similar equations can be written for each thermal element in the model.

#### 4.4 Results and Discussion

Figure 4.3 shows the thermal response of the reaction zone (TCZ), first breeder element (TLI1) and middle breeder element (TLI4) over the first 1000 seconds for a lithium breeder accident, as described earlier in this chapter. Similar graphs are plotted in Figures 4.4 through 4.7 for  $\text{Li}_7\text{Pb}_2$ ,  $\text{LiPb}_4$ ,  $\text{LiAl}$  and  $\text{Li}_2\text{O}$  breeders, respectively. A number of interesting points are noted.

First, the general shapes of the curves in each case are similar, although different maximum temperatures are attained. A change of time step shows no appreciable difference. The reaction zone rises rapidly, reaching a maximum value within the first two minutes of the coolant tube breaks. Thereafter, the temperature decreases monotonically,

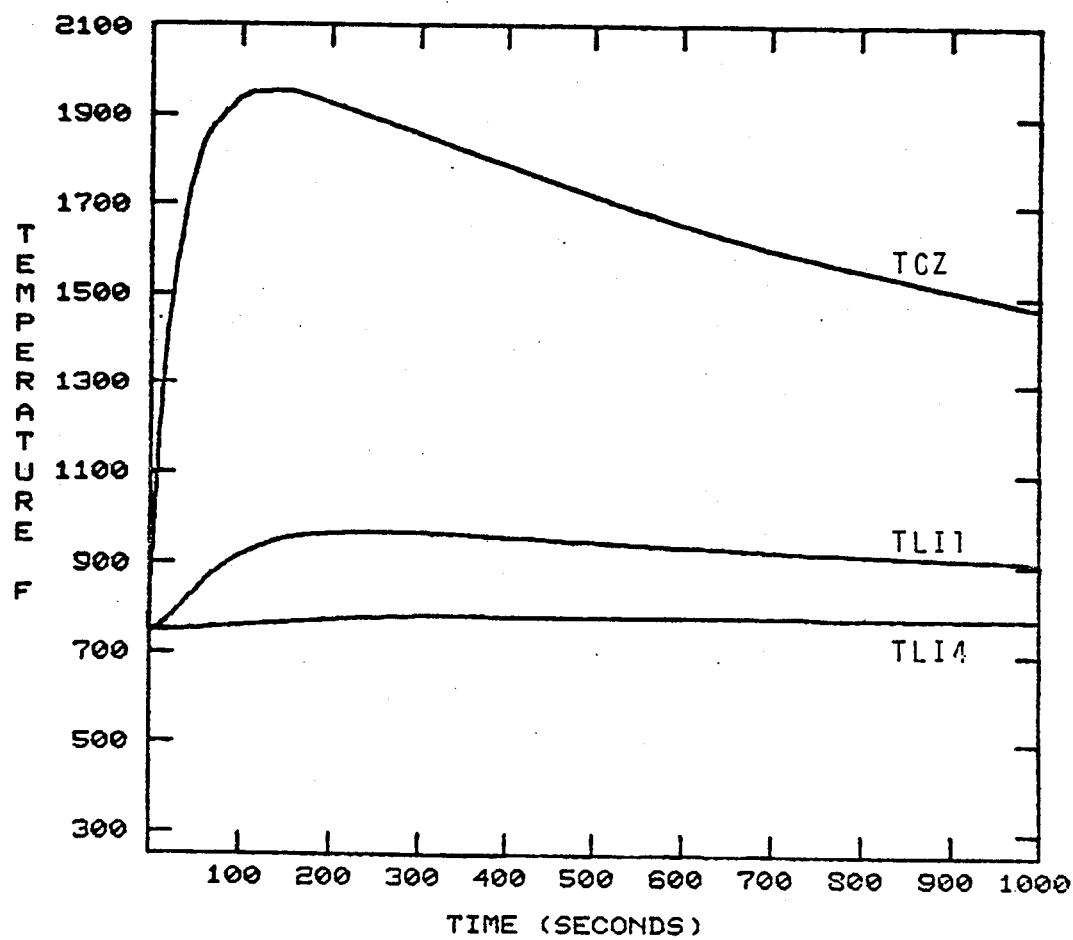


Figure 4.3 Lithium Breeder Thermal Response to Water Interaction.

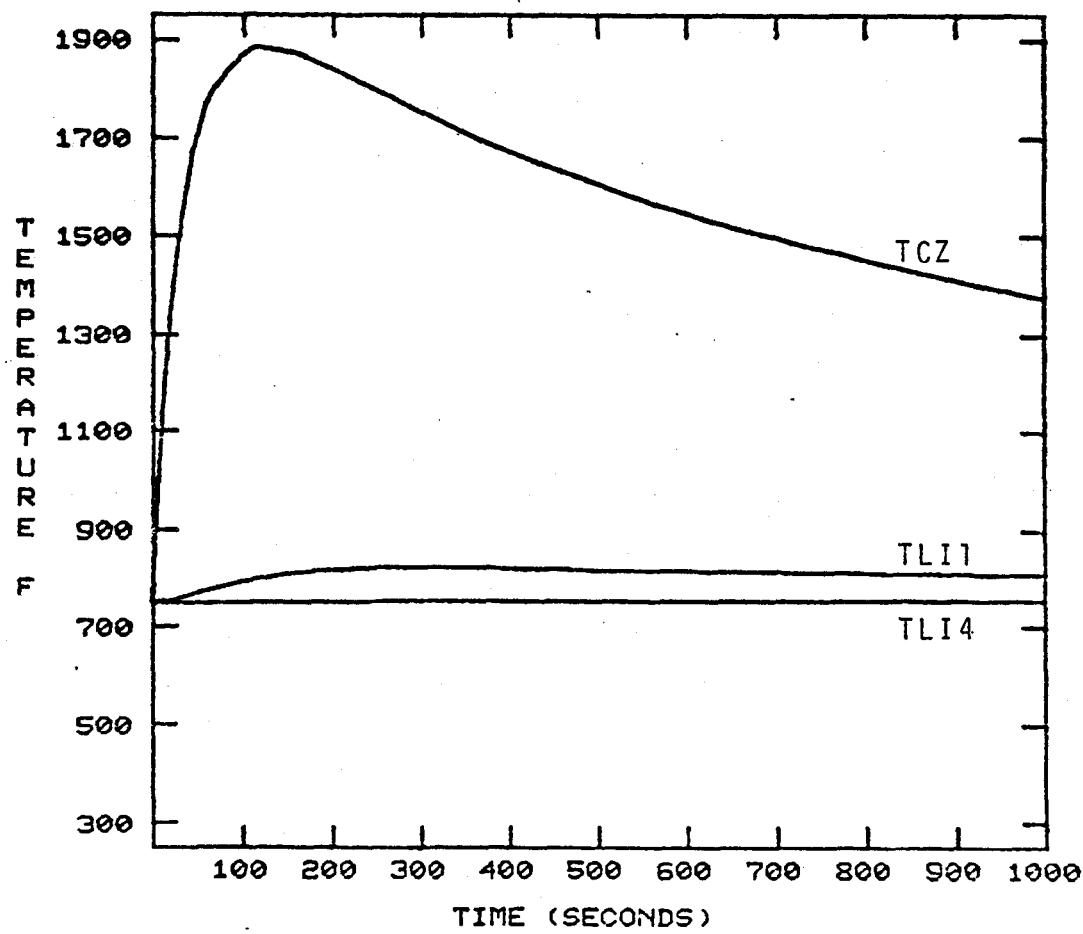


Figure 4.4  $\text{Li}_7\text{Pb}_2$  Breeder Thermal Response to Water Interaction.

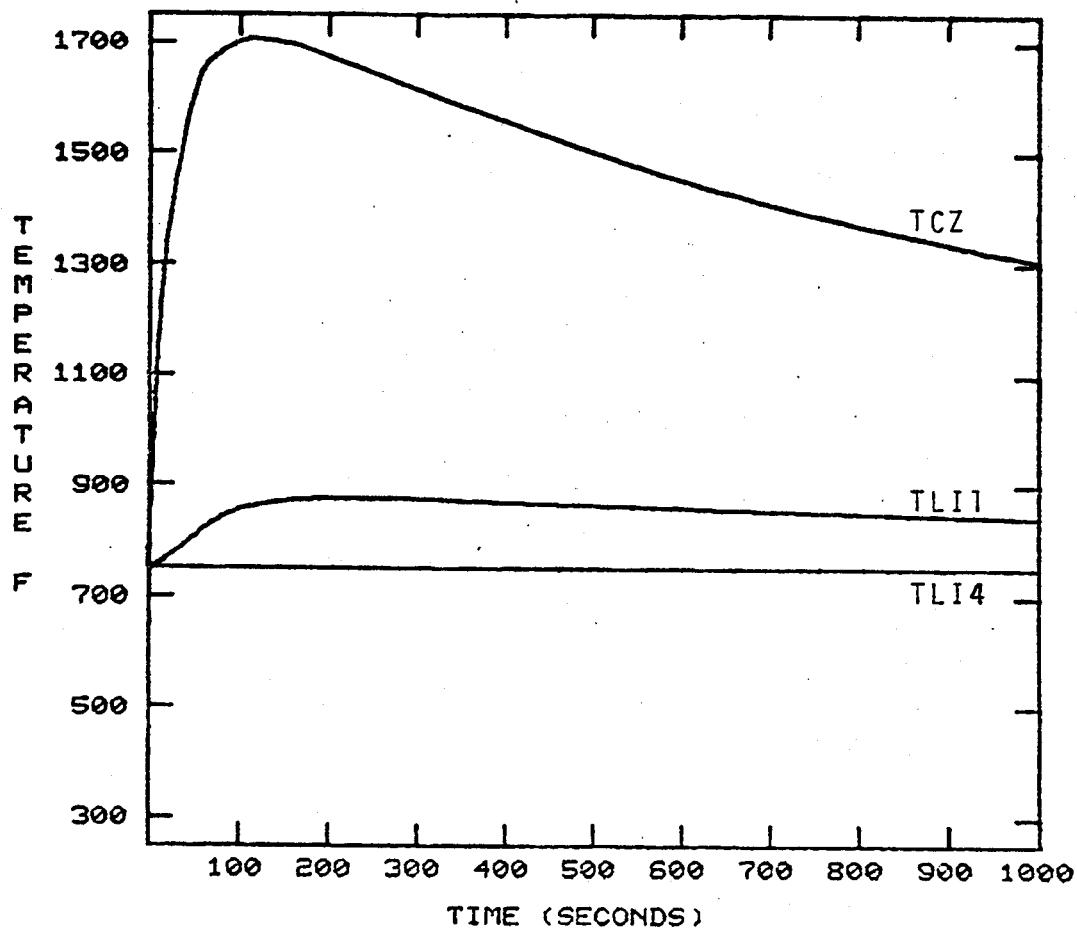


Figure 4.5  $\text{LiPb}_4$  Breeder Thermal Response to Water Interaction.

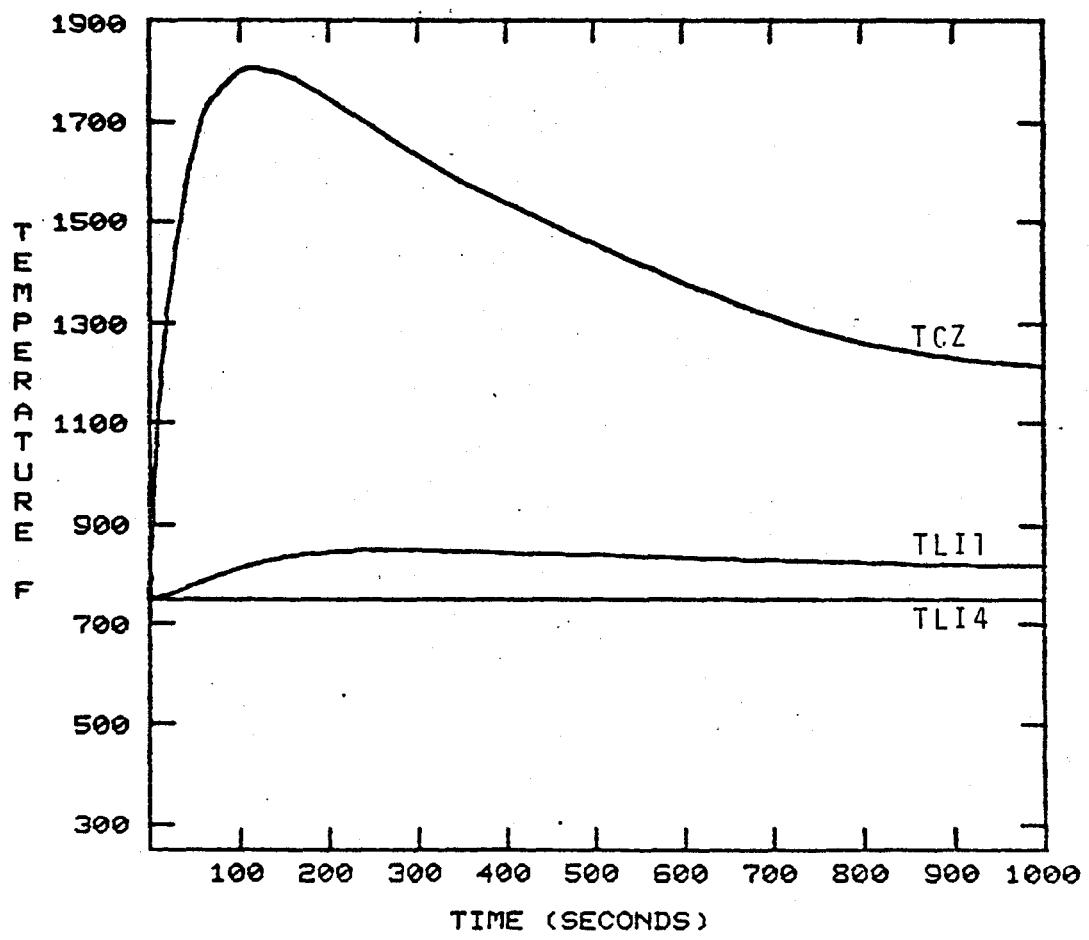


Figure 4.6 LiAl Breeder Thermal Response to Water Interaction.

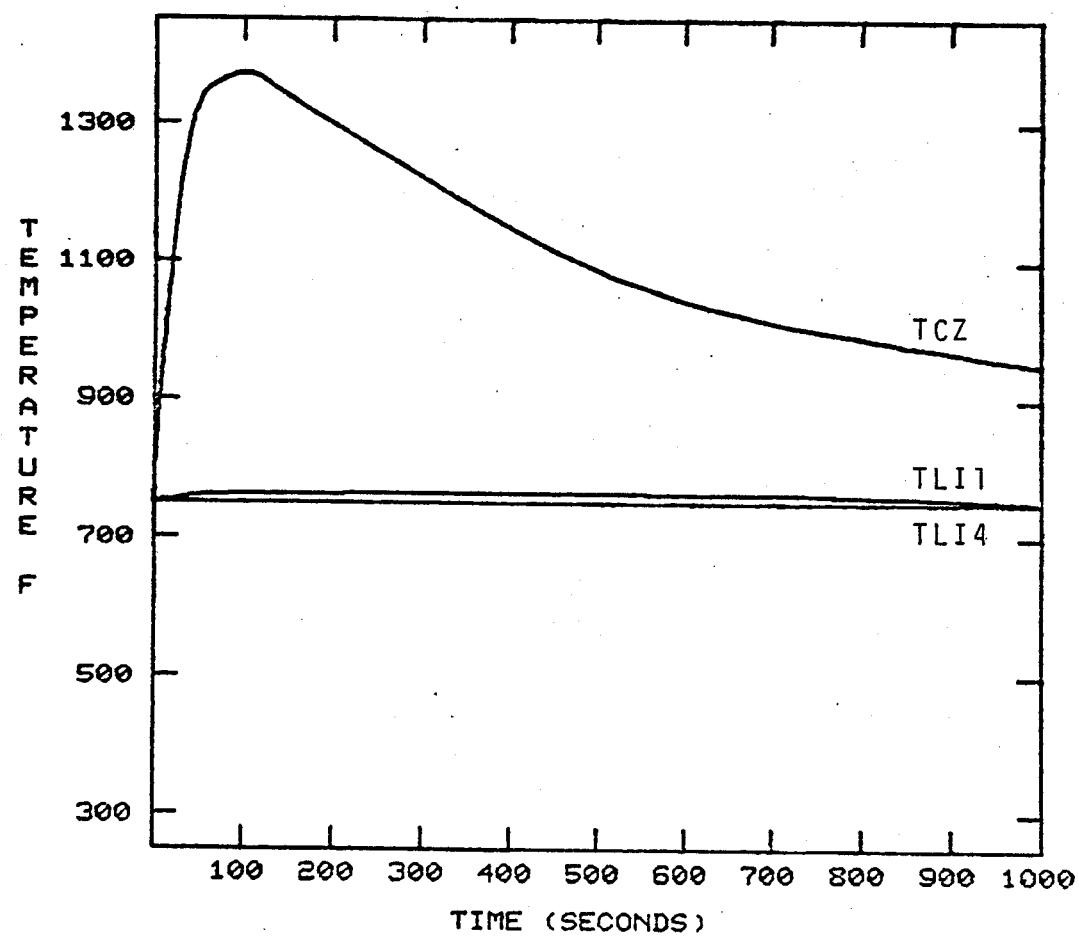


Figure 4.7  $\text{Li}_2\text{C}$  Breeder Thermal Response to Water Interaction.

more gradually as time progresses. The first blanket element exhibits similar behavior, but at lower temperatures and lagging over a minute behind. The middle blanket element is relatively unchanged, slowly increasing 50 °F in the lithium case and nearly constant in the alternate breeders.

This behavior may be due to different characteristic times for the various heat transfer processes. The heat of reaction in the reaction zone is instantaneous and relatively large compared to the conductive and convective flows, thus a rapid initial rise. Eventually, as the temperature difference between the reaction zone and first breeder element increases, so do the conductive and convective flows and the temperature profile flattens and decreases. Thereafter, as this temperature difference decreases, the reaction zone temperature decreases more gradually. This continues until the breeder (or water, in the LiPb<sub>4</sub> case) is depleted, eliminating the heat of reaction. Thereafter, all zones are eventually recooled to 752 °F. The first 1000 seconds are shown as the entire process takes upwards to 11 hours ( $4 \times 10^4$  seconds).

The relatively small volume of the reaction zone (less than 2% of the blanket module volume) may account for both the high temperatures attained in the reaction zone and the large temperature difference between this zone and the first breeder element (approximately 1000 °F in the lithium run). A small volume implies a small mass and surface area. The small mass is very sensitive to the heat of reaction and the small surface area impedes conductive and convective flows out. As the reaction zone expands, this effect is diminished.

The blanket liner, shield and other breeder elements were also monitored. All zones outside of the fourth breeder element showed no significant change. Thus, the accident appears to be effectively localized. The second and third breeder elements showed similar behavior as the first, with progressively shorter time lags.

Figure 4.8 presents a comparison of the reaction zone temperature profiles for the various breeders. As expected, the liquid lithium breeder produces the highest temperatures. Also expected, using the results of Chapter 3 as reference, is that the  $\text{Li}_2\text{O}$  breeder temperatures are significantly less. This breeder has a definite safety advantage compared to pure lithium.

The same can not be said, however, of the lithium-lead breeders and  $\text{LiAl}$ . Though less, the resulting temperatures are well within the range of the lithium case, as close as 70 °F ( $\text{Li}_7\text{Pb}_2$ ) and not further apart than 250 °F ( $\text{LiPb}_4$ ). These differences are insignificant at a base temperature of 1950 °F.

This result was surprising in the case of  $\text{LiPb}_4$ . Equation (4.14) indicates that the temperature rate of change in the reaction zone is inversely proportional to the reaction zone mass. Since the reaction zone volume is initially fixed and there is a factor of 20 increase in the density of  $\text{LiPb}_4$  over that of Li (at 400 °C, the density of Li is 0.51 g/cm<sup>3</sup>; the density of  $\text{LiPb}_4$  is 9.9 g/cm<sup>3</sup>), it was felt that the temperature rise would be proportionally decreased.

However, a closer look at Eq. (4.14) indicates that the temperature rate of change is also inversely proportional to the reaction zone specific heat. In this case, the specific heat of  $\text{LiPb}_4$ , approximately

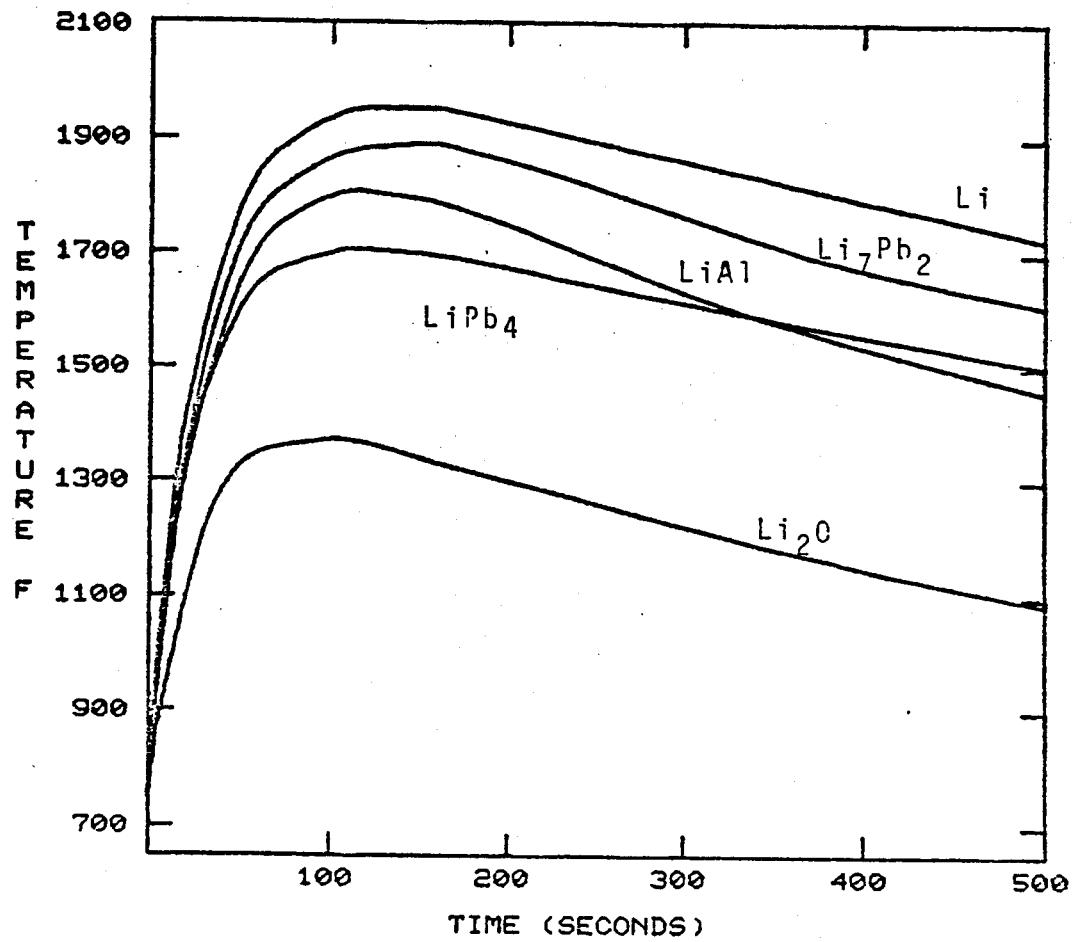


Figure 4.8 Comparison of Reactor Zone Temperature Profiles of the Various Breeders

that of Pb, is nearly a factor of 20 less than that of pure lithium (at 400 °C, the specific heat of lithium is 1.01 cal/g - °F; that of  $\text{LiPb}_4$  is 0.041 cal/g - °F). Thus, the terms in the denominator of Eq. (4.14) effectively cancel each other out and the slightly different reaction zone temperature profiles of Li and  $\text{LiPb}_4$  are simply a reflection of the slightly different heats of reaction of the two breeders with water.

A comparison of the thermal responses of the first breeder elements for the various breeders, shown in Fig. 4.9, is also interesting. Again, the lithium case produces the highest temperatures and the  $\text{Li}_2\text{O}$  case produces the lowest. The lithium-lead alloy and LiAl results are again similar. Here the  $\text{Li}_7\text{Pb}_2$  element is cooler overall due to a very low thermal conductivity, experienced in the lithium-lead system at a 20% lithium atom percentage [13]. However, in this comparison, the differences between these alloys and the lithium case are more pronounced. This indicates that the use of liquid lithium will produce more widespread accidental consequences.

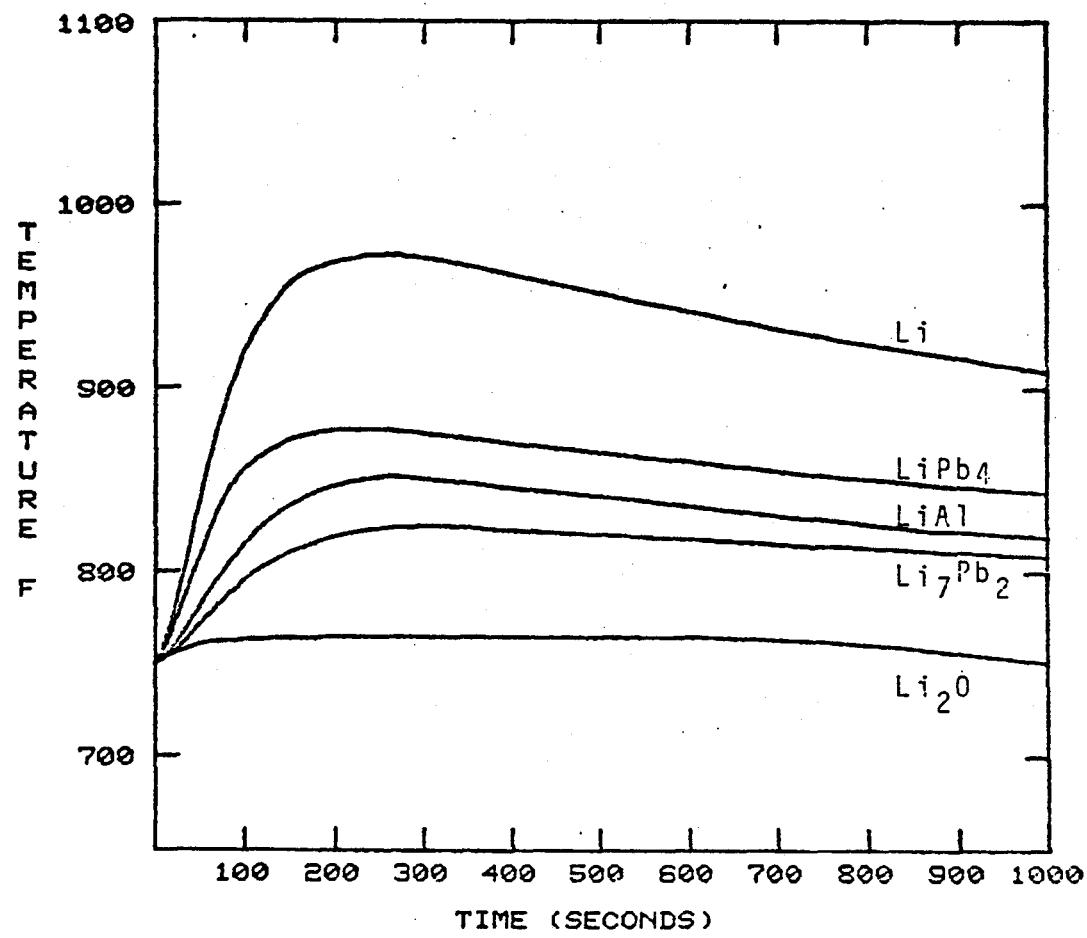


Figure 4.9 Comparison of First Breeder Element Temperature Profiles of the Various Breeders.

## CHAPTER 5. CONCLUSIONS AND RECOMMENDATIONS

Results indicate that the lithium-lead alloys may not be significantly safer than pure lithium as a fusion reactor breeding material, utilizing the NUWMAK geometry. In both calculations, short term temperatures resulting from interaction of  $\text{Li}_7\text{Pb}_2$  and  $\text{LiPb}_4$  with water, though lower, are within a few hundred degrees of those associated with the use of liquid lithium.

However, a proper conclusion to this study is that a conclusion cannot yet be made. The calculations of Chapters 3 and 4 provide an overview of the safety question, raising some interesting observations and determining areas that need to be explored in more detail.

First, the lithium-lead alloys and LiAl can pose safety problems approximate to those of liquid lithium, as noted above. This can be quite serious, as shown in Chapter 3, with temperatures reaching to the melting point of steel in a water-cooled blanket. For this reason alone, further study of these alternate breeders is required.

Results of Chapter 4 indicate that such an interaction could go undetected, as a continuance in coolant recirculation may keep the consequences localized. Measurable quantities, like bulk breeder temperature and coolant temperature and flow rate are practically unperturbed. Thus, further design work might include features to mitigate this problem.

Uncertainties in characterizing the reaction zone in the dynamic model render the resulting temperature profiles less meaningful. Further work in this area, perhaps experimental, would make the model more accurate.

Finally, better physical properties data on the alternate breeders is required. In particular, an understanding of the water reaction rates is central to the study of these materials. If it can be proved, as surmized, that these rates are significantly lower than those of liquid lithium, a significant reduction in safety hazards may be assured.

REFERENCES

1. A. J. Impink, Jr. and W. G. Homeyer, "Tritium Regeneration in Proposed Fusion Power Reactors," *Transactions of the American Nuclear Society*, S(1): 100, June 1962.
2. J. L. Ballif, et al., "Lithium Literature Review: Lithium's Properties and Interactions," HEDL report TC-1000, January, 1978.
3. Mark S. Tillack, "Development and Verification of the LITFIRE Model for Predicting the Effects of Lithium Spills in Fusion Reactor Containments," Master's Thesis, MIT, June 1980.
4. B. Badger, et al., "NUWMAK", University of Wisconsin, UWFDM-330, March 1979.
5. "Special Purpose Materials - Annual Progress Report," DOE/ET-0095, May 1979.
6. R. H. Wiswall and E. Wirsing, "Tritium Recovery from Fusion Blankets Using Solid Lithium Compounds," Radiation Effects and Tritium Technology for Fusion Reactors Conference, Gatlinburg, Tenn., 1975.
7. R. G. Clemmer, et al., "Assessment of Solid Breeding Blanket Options for Commercial Tokamak Reactors," ANL-CEN/205, 1980.
8. N. A. Frigerio and L. L. LaVoy, "The Preparation and Properties of LiPb, A Novel Material for Shields and Collimators," *Nuclear Technology*, 10, 322, 1971.
9. D. L. Smith, et al., "Fusion Reactor Blanket/Shield Design Study," ANL/FPP-79-1, July 1979.
10. "Dresden Nuclear Power Station Units 2 and 3, Safety Analysis Report," Commonwealth Edison Company, 1968.
11. D. A. Dube and M. S. Kazimi, "Analysis of Design Strategies for Mitigating the Consequences of Lithium Fire within Containment of Controlled Thermonuclear Reactors," MITNE-219, July 1978.
12. R. Bird, W. Stewart and E. Lightfoot, Transport Phenomenon, John Wiley & Sons, Inc., 1960.
13. Marie-Louis Sabourgi, et al., "Thermodynamic Properties of a Quasi-ionic Alloy from Electromotive Force Measurements: The Li-Pb System," The Journal of Chemical Physics, 68, 4, February 15, 1978.
14. John H. Perry, et al., Chemical Engineers' Handbook, McGraw-Hill Book Company, Inc., 1963.



**APPENDIX A**  
**Physical Property Data**



Table A.1 summarizes the important physical property data of the various alternate breeders analyzed in this report. Except as indicated, all of these values are gathered from the previously identified references. It is important to realize that, aside from the lithium and melting point data, all numbers are estimates at best.

For reasons detailed earlier, only the physical properties of the lithium breeder are allowed to vary with temperature in calculations of Chapters 3 and 4. The correlations used are:

$$\rho = 0.5368 - 1.0208 \times 10^{-4} T \text{ (g/cm}^3\text{)}$$

$$k = 10.48 + 4.98 \times 10^{-3}(T - 180.6) - 0.58 \times 10^{-6}(T - 180.6)^2 \text{ (cal/sec-m-}^\circ\text{C)}$$

$$c_p = 1.0037 - 0.01063x + 0.00564x^2 - 0.001279x^3 \text{ (cal/g }^\circ\text{C)}$$

where  $x = .004938T - 6.20741$

Here,

$\rho$  = density of lithium

$k$  = thermal conductivity of lithium

$c_p$  = specific heat of lithium

$T$  = lithium temperature in  $^\circ\text{C}$

The latent heats of melting for the alloy breeders are determined using the correlation

$$H_{melt}/T_{melt} \sim 2.2$$

This is an average value for metallic alloys [14].

The thermal conductivity of  $\text{LiPb}_4$  is estimated with the correlation

$$k_m = k_1 w_1 + k_2 w_2 - 0.72(k_2 - k_1)(w_1 w_2)$$

This is appropriate for a binary liquid mixture. Here, the weight fraction  $w_2$  refers to the component having the larger value of  $k$  [14].

The specific heat data of the alloy breeders is determined using the relation

$$c_{p_{\text{alloy}}} = x_{\text{Li}} c_{p_{\text{Li}}} + x_A c_{pA}$$

where

$x_{\text{Li}}$  = weight fraction of lithium

$x_A$  = weight fraction of the alloy element

$c_{pA}$  = specific heat of the alloy element

$c_{p_{\text{Li}}}$  = specific heat of lithium

The specific heat values for the alloy elements, lead and aluminum, can be found in the literature.

TABLE A.1  
Summary of Physical Properties of Candidate  
Breeding Materials at Standard Conditions

<u>Properties</u>	<u>Lithium</u>	<u>Li<sub>7</sub>Pb<sub>2</sub></u>	<u>LiPb<sub>4</sub></u>	<u>LiAl</u>	<u>Li<sub>2</sub>O</u>
Melting Point, °K	453	999	508	973	1970
ΔH <sub>melt</sub> , cal/g mole	722.4	2200 <sup>a</sup>	1120 <sup>a</sup>	2140 <sup>a</sup>	—
Density, g/cm <sup>3</sup>	0.51	4.59	9.9	1.76	2.01
Li atom density, g/cm <sup>3</sup>	0.51	0.49	0.08	0.37	0.93
Thermal conductivity w/m-k	50	~20	~35 <sup>a</sup>	30	1.73
Specific Heat cal/g °K	1.01	0.14 <sup>b</sup>	0.041 <sup>b</sup>	0.44 <sup>b</sup>	0.35
Heat of Reaction with Water kJ g-atom Li	245	200	170 <sup>c</sup>	200	64
Atomic weight	7	463	835	34	30

a. correlation

b. interpolated value

c. estimated



APPENDIX B  
Complete Listing of LITFIRE



LITFIRE COMPUTER CODE IS A MODIFICATION OF THE CODE SPOOL-FIRE DELIT00010  
 AT ARGONNE NATIONAL LABORATORY. LITFIRE DESCRIBES THE TEMPERATURELIT00020  
 PRESSURE HISTORY OF A FUSION REACTOR CONTAINMENT TO LITHIUM FIRE LIT00030  
 AND WAS DEVELOPED IN THE NUCLEAR ENGINEERING DEPARTMENT AT MIT INITIALLY  
 LIT00040  
 THIS VERSION OF LITFIRE HAS A MANDATORY ONE NODE IN THE LITHIUMLIT00050  
 POOL, WITH A SUSPENDED (INSULATED) PAN OPTION  
 THE CONTAINMENT IS SINGLE CELL  
 DEFINITION OF VARIABLES AND UNITS  
 THIS IS A RELATIVELY COMPLETE LISTING OF THE VARIABLES, CONSTANTS, LIT00110  
 PARAMETERS USED IN THE COMPUTER CODE LITFIRE II, ACCURATE AS OF LIT00120  
 SEPTEMBER 30, 1980. FURTHER DOCUMENTATION AND SOME DERIVATIONS LIT00130  
 MAY BE FOUND IN THE LITFIRE PROGRAMMER'S MANUAL  
 LIT00140  
 LIT00150  
 AINS OUTSIDE EXPOSED AREA OF INSULATING LAYER ON PAN (FT2) LIT00160  
 AKLI THERMAL CONDUCTIVITY OF LITHIUM BTU/FT.-SEC. F LIT00170  
 INPUT AS BTU/FT. HR. DEG. F LIT00180  
 AK1, AK2 PROD. OF THERMAL COND. AND PRANDTL NO. BTU/SEC-FT-DEG. LIT00190  
 AMLI ATOMIC MASS OF BREEDER  
 AMPB ATOMIC MASS OF ALLOY METAL  
 APAN PAN EXTERNAL AREA FOR HEAT TRANSFER  
 ARE SURFACE AREA OF BREEDER ELEMENT  
 ASLI SURFACE AREA OF LITHIUM FT2 LIT00230  
 ATI INNER SURFACE AREA OF COOLANT TUBES IN ELEMENT  
 ATO OUTER SURFACE AREA OF COOLANT TUBES IN ELEMENT  
 AW EXPOSED WALL AREA FT2 LIT00240  
 AWB AREA OF THE STEEL CONTAINMENT BELOW THE SPILL PAN  
 B USED IN CALC. THERMAL RESIST. OF LINER-GAP-CONC. FT.  
 BB ANALOGOUS TO B, ONLY FOR FLOOR CONCRETE  
 B1 & B2 COEFFICIENT OF GAS EXPANSION 1/DEG. F LIT00250  
 BLIN TIME AFTER SPILL AT WHICH INERT GAS FLOODING AND  
 EXHAUST BEGINS SEC LIT00260  
 BLOUT TIME AFTER SPILL AT WHICH FLOODING AND EXHAUST STOPS SEC  
 BLOWR INERT GAS INPUT RATE LB/SEC LIT00270  
 BLOWV INERT GAS INPUT RA. FT3/MIN LIT00280  
 BREDTH LENGTH AROUND THE SIDE OF THE SPILL PAN IN FEET  
 \*\*\* "C" IS THE INITIAL USED FOR INDICATING A THERMAL DIFFUSIVITY.  
 CONDUCTIVITY BETWEEN TWO NODES DIVIDED BY THE HEAT CAPACITY OF ONE  
 THOSE NODES  
 C1 CONTAINMENT GAS TO WALL STEEL IN GAS  
 C2 PAN TO CGNT GAS IN GAS  
 C3 STEEL LINER TO CONCRETE WALL IN WALL CONCRETE  
 C4(1) CONCRETE NODE I TO NODE I+1 IN CONCRETE  
 C5 CONCRETE WALL TO AMBIENT IN CONCRETE  
 C6 CONTAINMENT GAS TO WALL STEEL IN STEEL  
 C7 STEEL LINER TO CONCRETE WALL IN STEEL  
 C8 STEEL LINER TO CONCRETE FLOOR IN STEEL  
 C9 STEEL LINER TO CONCRETE FLOOR IN CONCRETE  
 C10(1) CONCRETE FLOOR NODE I TO NODE I+1 IN FLOOR CONCRETE  
 C11 STEEL TO AMBIENT (NO CONCRETE OPTION) IN STEEL  
 C12 PAN TO GAS IN PAN IN STEEL  
 C13 PAN TO GAS IN PAN  
 CCZG COMBUSTION ZONE TO CONTAINMENT GAS IN GAS

C CCZP POOL TO COMBUSTION ZONE IN POOL LIT00540  
 C CGCZ COMBUSTION ZONE TO CONTAINMENT GAS IN COMBUSTION ZONE LIT00550  
 C CGLI POOL TO CONTAINMENT GAS (NO COMBUSTION) IN POOL LIT00560  
 C CH CONTAINMENT HEIGHT FT LIT00570  
 C CIN12 INNER TO OUTER INSULATION IN INNER INSULATION LIT00580  
 C CIN21 INNER TO OUTER INSULATION IN OUTER INSULATION LIT00590  
 C CINPN STEEL PAN TO INNER INSULATION IN INSULATION LIT00600  
 C CLIG POOL TO COMBUSTION GAS (NO COMBUSTION) IN GAS LIT00610  
 C CLIPN POOL TO SPILL PAN IN POOL (SUSP PAN OPTION) LIT00620  
 C CLST LITHIUM POOL TO FLOOR STEEL IN LITHIUM LIT00630  
 C CPANL1 POOL TO PAN IN PAN LIT00640  
 C CPCZ LITHIUM POOL TO COMBUSTION ZONE IN COMBUSTION ZONE LIT00650  
 C CPN1N1 STEEL PAN TO INNER INSULATION IN PAN LIT00660  
 C CSBL1 LITHIUM POOL TO FLOOR STEEL IN STEEL LIT00670  
 C CCZ AMOUNT OF HEAT BEING DEVELOPED IN THE COMB. ZONE (BTU/LB SEC) LIT00680  
 C CMBR TOTAL COMBUSTION RATE LB. LI/SEC.-FT2 LIT00690  
 C CMBRH INITIAL COMBUSTION RATE LB. LI/HR.-FT2 LIT00700  
 C CMBRH1 CMBRN COMB. RATE FOR NITROGEN REACTION LB. LI/SEC.-FT2 LIT00710  
 C CMBRN1 CMBR0 COMB. RATE FOR OXYGEN REACTION LB. LI/SEC.-FT2 LIT00720  
 C CMBRW CMBR1 COMB. RATE FOR WATER VAPOR REACTION LB. LI/SEC.-FT2 LIT00730  
 C CPA1 INERT GAS SPECIFIC HEAT BTU/LB.-DEG. F LIT00740  
 C CPAB SPEC. HEAT OF FLOODING GAS BTU/LB-DEG.F LIT00750  
 C CPCON HEAT CAPACITY OF FLOOR AND WALL CONCRETE LIT00760  
 C CPH2 HEAT CAPACITY OF HYDROGEN GAS LIT00770  
 C CPINS SPECIFIC HEAT OF INSULATION BTU/LB DEG F LIT00780  
 C CPLIS SPECIFIC HEAT OF LI BTU/LB.-DEG. F LIT00790  
 C CPLI1 MEAN HEAT CAPACITY OF BREEDER AS SOLID BTU/LB MOLE LIT00800  
 C CPLIN SPEC. HEAT OF LITHIUM NITRIDE BTU/LB-DEG. F LIT00810  
 C CPLIO SPECIFIC HEAT OF LITHIUM OXIDE BTU/LB.-DEG. F LIT00820  
 C CPLIOH SPECIFIC HEAT OF LIOH BTU/LB-MOLE F LIT00830  
 C CPMCZ EFFECTIVE HEAT CAPACITY OF COMB. ZONE BTU/DEG F LIT00840  
 C CPMNH2 HEAT CAPACITY OF HYDROGEN IN CONTAINMENT BTU/DEG. F LIT00850  
 C CPMLIH HEAT CAP. OF LITH. HYDROXIDE IN CONT. BTU/DEG. F LIT00860  
 C CPMLIN HEAT CAP. OF LITH. NITRIDE IN CONT. BTU/DEG. F LIT00870  
 C CPMLO1 HEAT CAP. OF LITHIUM OXIDE IN CONTAINMENT BTU/DEG. F LIT00880  
 C CPMNI HEAT CAPACITY OF NITROGEN IN CONTAINMENT BTU/DEG. F LIT00890  
 C CPMOX HEAT CAPACITY OF OXYGEN IN CONTAINMENT BTU/DEG. F LIT00900  
 C CPMWA HEAT CAP. OF WATER VAP. IN CONTAINMENT BTU/DEG. F LIT00910  
 C CPM2 HEAT CAPACITY OF NITROGEN GAS BTU/LB-DEG F LIT00920  
 C CPPAN SPECIFIC HEAT OF SPILL PAN BTU/LB-DEG F LIT00930  
 C CPPB HEAT CAPACITY OF ALLOY METAL IN BREEDER ZONE BTU/LB-F LIT00940  
 C CPPB1 MEAN HEAT CAPACITY OF ALLOY METAL SOLID BTU/LB MOLE LIT00950  
 C CPPL LIQUID HEAT CAPACITY OF ALLOY METAL BTU/LB R LIT00960  
 C CPPZ HEAT CAPACITY OF ALLOY METAL IN REACTION ZONE BTU/LM F LIT00970  
 C CP1 USED TO CALCULATE CP CHANGE OF ALLOY METAL BTU/LB R LIT00990  
 C CP2 USED TO CALCULATE CP CHANGE OF ALLOY METAL BTU/LB R LIT01000  
 C CPSTL HEAT CAPACITY OF STEEL LINER (BTU/LB-DEG F) LIT01010  
 C CPWA SPEC. HEAT OF WATER VAPOR BTU/LB.-DEG. F LIT01020  
 C CF THERMAL IMPEDANCE BETWEEN BREEDER ELEMENTS IN INNER ELEMENT LIT01030  
 C CT THERMAL IMPEDANCE BETWEEN BREEDER ELEMENTS IN OUTER ELEMENT LIT01040  
 C DELH STANDARD HEAT OF HYDROLYSIS OF BREEDER BTU/LB MOLE LIT01050  
 C DDELUT OUT TIME STEP SEC. LIT01060

C DELT LITHIUM VAPOR FILM THICKNESS FT (OUTPUT IN INPUT) 1070  
 C DFILM DIFFUSION COEFF. TO COMB. ZONE FT2/SEC. LIT01080  
 C DIFFL LITHIUM VAPOR DIFFUSION COEFFICIENT FT2/SEC. LIT01090  
 C DPROD ENTHALPY CHANGE OF REACTION PRODUCTS IN REACTION ZONE LIT01100  
 C DP1,DP2,DP3 PSIA INCREASE IN CONTAINMENT PRESSURE DUE TO EACH LIT01110  
 C DREAC POOL TIME STEP (TEMP./RATE OF CHANGE OF TEMP.) LIT01120  
 C DT1 CONT. GAS TIME STEP LIT01130  
 C DT2 STEEL WALL TIME STEP LIT01140  
 C DT3 STEEL WALL TIME STEP LIT01150  
 C DT4 COMBUSTION RATE TIME STEP LIT01160  
 C DT5 COMBUSTION ZONE TEMP. TIME STEP LIT01170  
 C DBDT(1) CONC. FLOOR TEMP. RATE OF CHANGE, NODE 1 DEG. F/SEC. LIT01180  
 C DTCDF(1) CONC. WALL TEMP. RATE OF CHANGE, NODE 1 DEG. F/SEC. LIT01190  
 C DTMIN MINIMUM TIME STEP TO BE USED SEC. LIT01200  
 C DT1...DT4 USED IN CALCULATING TIME STEP SEC. LIT01210  
 C DYNAMI SUBROUTINE USED IN CONTROLLING INTEGRATION LOOPS LIT01220  
 C D1,D2 KINEMATIC VISCOSITY OF CELL GAS (SQUARED) FT4/SEC2 LIT01230  
 C EFILM FILM DEPTH OF DEPLETED ZONE ABOVE COMB. ZONE (IN INCH) LIT01240  
 C EMCCONC THERMAL EMISSIVITY OF CONCRETE LIT01250  
 C EMG THERMAL EMISSIVITY OF COMBUSTION ZONE LIT01260  
 C EMIN THERMAL EMISSIVITY OF INSULATION AROUND PAN LIT01270  
 C EMLI THERMAL EMISSIVITY OF LITHIUM POOL LIT01280  
 C EMSTL THERMAL EMISSIVITY OF STEEL LINER LIT01290  
 C ESCR HEAT REMOVAL RATE BY EMERGENCY SPACE COOLING BTU/SEC LIT01300  
 C ESCIN TIME AFTER SPILL WHEN ESCR BEGINS SEC LIT01310  
 C EXHSTR RATE OF CONTAINMENT GAS EXHAUST LB/SEC LIT01320  
 C EXHSTV RATE OF CONTAINMENT GAS EXHAUST FT3/SEC LIT01330  
 C EXX USED IN CALC. MASS & HEAT TRANSF. COEFF. 1/FT3 LIT01340  
 C EX1,EX2 USED IN CALCULATING MASS & HEAT TRANSF. COEFF. 1/FT LIT01350  
 C FCT1,FCT2,FCT3 FRACTION OF NITROGEN PRESENT IN EACH INJECTION LIT01360  
 C FF1,FF2 USED IN HEAT BALANCE EQS. FOR SPRAY FIRE BTU LIT01370  
 C FMLEAK FRACT. OF MASS LEAKED OUT OF CONTAINMENT LIT01380  
 C FMLEFT FRACTION OF MASS STILL WITHIN CONTAINMENT LIT01390  
 C FN1 WT. FRACTION OF NITROGEN IN CELL GAS LIT01400  
 C FOUT LOSS RATE OF CONT. GAS WHICH EITHER LEAKS OR IS EXHAUST LIT01420  
 C FOX WT. FRACTION OF OXYGEN IN CELL GAS LIT01430  
 C FRA FRACTION OF COMBUSTION PRODUCTS EVOLVED INTO CELL GAS LIT01440  
 C FWA WT. FRACTION OF WATER VAPOR IN CELL GAS LIT01450  
 C G AIR GAP BETWEEN STEEL LINER AND CONCRETE FLOOR FT. LIT01460  
 C GIN GRAVITATIONAL CONSTANT 32.2 FT/SEC2 LIT01470  
 C H INTERIOR FILM COEF. BTU/SEC-FT\*\*2-DEG. F LIT01480  
 C HA EXTERIOR FILM COEF. BTU/SEC. FT\*\*2 DEG. F LIT01490  
 C (INPUT AS BTU/HR FT\*\*2 DEG F LIT01500  
 C HB HEAT TRANSFER COEFFICIENT TO POOL BTU/SEC-FT2-DEG F LIT01510  
 C HBINF EQUILIBRIUM VALUE OF HB LIT01520  
 C HCO HEAT TRANSFER COEFFICIENT OF SOILING WATER BTU/SEC-FT2 LIT01530  
 C HF GAS TRANSPORT COEFF. TO POOL FT/SEC. LIT01540  
 C HFINF EQUILIBRIUM VALUE OF HF LIT01550  
 C HIN COEFFICIENT FOR HEAT TRANSFER CORRELATIONS (H, HB, HF) LIT01560  
 C HTCAPG HEAT CAPACITY OF CONTAINMENT ATMOSPHERE BTU/DEG.F LIT01570  
 C

C 1 GENERAL PURPOSE DO LOOP INDEX LIT01600  
 C IB DO LOOP INDEX USED FOR FLOOR CONCRETE ITERATIONS LIT01610  
 C INITIALIZING SUBROUTINE FOR INTEGRATION CALCULATIONS LIT01620  
 C INJEC1,INJEC2,INJEC3 FLAGS FOR GAS INJECTION \*\*\* INJEC=1 INDICATES THAT THE PARTICULAR INJECTION HAS OCCURRED LIT01630  
 C INGR1 ARITHMETIC STATEMENT FUNCTION FOR FINDING INTEGRALS LIT01640  
 C J1=1 IF LITHIUM IS BREEDER LIT01650  
 C J2=1 IF HYDROGEN IS EVOLVED LIT01660  
 C K LEAK RATE CONSTANT FROM CONTAINMENT 2.58E-09 LIT01670  
 C KCON THERMAL CONDUCTIVITY OF THE FLOOR AND WALL CONCRETE LIT01680  
 C KGAP THERMAL COND. OF THE AIR GAP BETWEEN THE LINER AND CONCRETE LIT01690  
 C KFILM THERM. COND. OF LI POOL/COMB. ZONE FILM BTU/SEC-FT-F LIT01700  
 C KIN1 THERMAL CONDUCTIVITY OF INNER INSULATION - CALC. IN PROGLIT01710  
 C KIN2 THERMAL CONDUCTIVITY OF OUTER INSULATION - CALC. IN PROGLIT01720  
 C KPAN THERMAL CONDUCTIVITY OF LI PAN BTU/HR-FT-DEG F LIT01730  
 C KSIL THERMAL CONDUCTIVITY OF THE STEEL LINER (BTU/HR-FT-DEG) F LIT01740  
 C L CONCRETE WALL ELEMENT THICKNESS FT. LIT01750  
 C LBW DISTANCE BETWEEN TWO BREEDER ELEMENTS LIT01760  
 C LEAK CELL GAS LEAKAGE RATE FROM CONTAINMENT 1/SEC. LIT01770  
 C LEAKO INITIAL CELL GAS LEAKAGE RATE FROM CONTAINMENT 1/SEC. LIT01780  
 C LIBP LITHIUM BURNED IN POOL FIRE LB. LIT01790  
 C LIC MASS OF LITHIUM IN CONTAINMENT LB. LIT01800  
 C LIL AMOUNT OF LI LEFT IN POOL BUT NOT ALLOWED TO BE LESSTHAN LIT01820  
 C LILP THAN LIT/10 FOR NUMERICAL STABILITY IN HEAT TRANSFER LIT01830  
 C LIS TRUE AMOUNT OF LITHIUM IN POOL (LB) LIT01840  
 C LIT LITHIUM USED IN SPRAY FIRE LB. LIT01850  
 C LIT MASS OF LITHIUM IN POOL INITIALLY LB. LIT01860  
 C L1 CONCRETE FLOOR ELEMENT THICKNESS FT. LIT01870  
 C MA WT. OF INERT GAS IN CELL LB. LIT01880  
 C MAIR INITIAL MASS OF INERT GAS IN CONTAINMENT (LB) LIT01890  
 C MB MASS OF CELL GAS LB. LIT01900  
 C MCZ INITIAL REACTION ZONE MASS LB MOLES LIT01910  
 C MCZ1 INITIAL REACTION ZONE MASS LB MOLES LIT01920  
 C NH2 WT. OF HYDROGEN IN CONT. CELL GAS LB. LIT01930  
 C MLIH WT. OF LITHIUM HYDROXIDE IN CONT. CELL GAS LB. LIT01940  
 C MLIN WT. OF LITHIUM NITRIDE IN CONT. GAS CELL LB. LIT01950  
 C MNINI MASS OF LITHIUM NITRIDE IN CONT. INITIALLY LB. LIT01960  
 C MLJO WEIGHT OF LITHIUM OXIDE IN CELL GAS. PRESENTLY ALL OF THE LIT01970  
 C SPRAY FIRE PRODUCT REMAINS IN THE CELL GAS. A FRACTION LIT01980  
 C OF THE PRODUCTS FROM THE POOL FIRE IS ADDED LB. LIT01990  
 C NLIO1 MASS OF LITHIUM OXIDE IN CONT. INITIALLY LB. LIT02000  
 C MLIOH MASS OF LiOH PRODUCT IN LB MOLES LIT02010  
 C MNII WEIGHT OF NITROGEN IN CONTAINMENT CELL GAS LB. LIT02020  
 C MNINI RATE OF INJECTION OF NITROGEN DURING A 60 SEC INTERVAL LIT02030  
 C MNINI1,MNINI2,MNINI3 USED TO MODEL HEAT PROCEDURE (LB/SEC) LIT02040  
 C NOX WEIGHT OF OXYGEN IN CELL GAS LB. LIT02050  
 C MOXI INITIAL WEIGHT OF OXYGEN IN CONTAINMENT LB. LIT02060  
 C MOXINU RATE OF INJECTION OF OXYGEN USED TO MODEL HEAT EXPERIMENTAL (LB.) LIT02070  
 C MOXINU1,M0XIN2,M0XIN3 OCCURS DURING A 60 SEC. INTERVAL(LB./SEC)LIT02110  
 C MOXINU MASS OF OXYGEN INJECTED (LBS.) LIT02120

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C MPB MASS OF ALLOY METAL PRODUCT IN LB MOLES LIT02130  
 C MWA WEIGHT OF WAT. VAP. IN CONTAINMENT CELL GAS LB. LIT02140  
 C MWA1 MASS OF WATER VAPOR IN CON. CELL GAS INITIALLY LB. LIT02150  
 C NA NUMBER OF ELEMENTS IN BREEDER ZONE LIT02160  
 C NAME (1) INPUT CONTAINING PROGRAM TITLE AND HEADING LIT02170  
 C NL = NUMBER OF CONCRETE WALL NODES LIT02180  
 C NL1 = NUMBER OF CONCRETE FLOOR NODES LIT02190  
 C NLM1 ,NL1M1 WALL AND FLOOR CONCRETE NUMBER OF NODES MINUS ONE LIT02200  
 C NT NUMBER OF COOLANT TUBES DAMAGED LIT02210  
 OUTINT LEAK FRACTION LIT02220  
 C OVERP CONTAINMENT OVER PRESSURE PSIG LIT02230  
 C OXLB OXYGEN BURNED LB. LIT02240  
 C OXLBI OXYGEN BURNED INITIALLY LB. LIT02250  
 C OXLFS OXYGEN LEFT AFTER SPRAY FIRE LB. LIT02260  
 PA. GAS PRESSURE IN CELL PSIA LIT02270  
 C PAZERO INITIAL CELL PRESSURE PSIA LIT02280  
 C PBMLT MELTING POINT OF ALLOY METAL R. (VS. MONOXIDE) FORMED LIT02290  
 PERCENT BY NUMBER OF PEROXIDE (V.S. MONOXIDE) FORMED LIT02300  
 C PLIV PARTIAL PRESSURE OF LITHIUM VAPOR PSIA LIT02310  
 C PZERO CONTAINMENT PRESSURE AFTER SPRAY FIRE PSIA LIT02320  
 C 'QC FORCED CONVECTIVE COOLING HEAT FLOW LIT02330  
 C QCN HEAT OF COMB. FOR NITROGEN REACTION BTU/LB. LI LIT02340  
 C QCO HEAT OF COMBUSTION FOR OXYGEN REACTION BTU/LB. LI LIT02350  
 C QCO1 HEAT OF COMBUSTION FOR MONOXIDE REACTION BTU/LB. LI LIT02360  
 C QCO2 HEAT OF COMBUSTION FOR PEROXIDE REACTION BTU/LB. LI LIT02380  
 C QCW HEAT OF COMB. FOR REACTION WITH WATER VAPOR BTU/LB. LI LIT02390  
 C QIN HEAT ADDITION TO CELL GAS FROM SPRAY FIRE BTU LIT02400  
 C QLIQH LATENT HEAT OF MELTING FOR LIQUID BTU/LB-MOLE LIT02410  
 C QMELT HEAT OF FUSION OF BREEDER BTU/LB MOLE LIT02420  
 C QMELTP HEAT OF FUSION OF ALLOY METAL BTU/LB MOLE LIT02430  
 C QOUT1,2,3,4 USED IN HEAT BALANCE EQS. FOR SPRAY FIRE BTU LIT02440  
 C QRAD INDICATES A RADIATIVE HEAT FLOW BTU/SEC LIT02450  
 C QRADB FROM STEEL FLOOR (PAN) TO FLOOR CONC. OR TO AMBIENT LIT02460  
 C QRADC FROM STEEL WALL TO WALL CONCRETE OR TO AMBIENT LIT02470  
 C QRADCG FROM SPILL PAN TO CELL GAS LIT02480  
 C QRADG FROM LI POOL TO GAS (NO COMB.) OR FROM COMB. ZONE TO CELL LIT02490  
 C QRADP FROM COMB. ZONE TO LITHIUM POOL (COMB. ZONE MODEL ONLY) LIT02500  
 C QRADS FROM SPILL PAN TO STEEL FLOOR LIT02510  
 C QRADW FROM COMB. ZONE TO WALL STEEL OR FROM LI POOL TO WALL ST LIT02520  
 C QVAP HEAT OF VAPORIZATION OF LITHIUM BTU/LB LIT02530  
 C QWA HEAT OF REACTION OF BREEDER WITH WATER LIT02540  
 C PAREA SURFACE AREA OF REACTION ZONE LIT02550  
 C THE SYMBOL "R" DESIGNATES A TEMPERATURE RATE OF CHANGE IN SOME NODE LIT02560  
 C TO RADIATION HEAT TRANSFER BETWEEN THAT NODE AND SOME OTHER NODE LIT02570  
 C RADB IN FLOOR STEEL DUE TO RAD. TO FLOOR CONC. OR TO AMBIENT LIT02580  
 C RADC IN WALL STEEL DUE TO RAD. TO CONCRETE OR TO AMBIENT LIT02590  
 C RADCB IN FLOOR CONCRETE FROM STEEL FLOOR (PAN) LIT02600  
 C RADCC IN WALL CONCRETE FROM STEEL WALL LIT02610  
 C RCZG IN GAS FROM COMBUSTION ZONE LIT02620  
 C RCZP IN LITHIUM POOL FROM COMBUSTION ZONE LIT02630  
 C RCZW IN WALL STEEL FROM COMBUSTION ZONE LIT02640  
 C RGASPA IN PAN DUE TO RAD. TO CONTAINMENT GAS LIT02650

C RG11 IN POOL DUE TO RAD. TO GAS (NO COMB) LIT02660  
 C RLIG IN GAS DUE TO RAD. FROM POOL (NO COMBUSTION) LIT02670  
 C RLIW IN WALL STEEL FROM LITHIUM POOL (NO COMB) LIT02680  
 C RPAGAS IN CELL GAS DUE TO RAD. FROM LI PAN LIT02690  
 C RPANST IN WALL STEEL DUE TO RAD. FROM LITHIUM PAN LIT02700  
 C RSTPAN IN PAN DUE TO RAD. TO FLOOR STEEL LIT02710  
 C RWL1 IN LITHIUM POOL FROM RAD. TO WALL STEEL (NO COMB) LIT02720  
 C RA MEAN RADIUS OF COMBUSTION PRODUCT PARTICLES MICRONS LIT02730  
 C RCMBH2 STOICH. COMB. RATIO FOR H<sub>2</sub>O VAPOR REACT. LB. L1/LB. H2L1T02740  
 C RCMBN STOICH. COMB. RATIO OF NITROGEN REACT. LB. L1 / LB. N LIT02750  
 C RCMBD STOICH. COMB. RATIO FOR OXYGEN REACTION LB. L1/LB. O LIT02760  
 C RCMB01 STOICH. COMB. RATIO FOR MONOXIDE REACTION LB. L1/LB. OLIT02770  
 C RCMBD2-STOICH. COMB. RATIO FOR PEROXIDE REACTION LB. L1/LB. QLIT02780  
 C RCMBW-STOICH. COMB. RATIO FOR WAT. VAP. REACT. LB. L1/LB. H2O LIT02790  
 C RELERR MAXIMUM ALLOWABLE FRACTIONAL TEMP. CHANGE ACROSS A SINGUL LIT02800  
 C RHCON DENSITY OF FLOOR AND WALL CONCRETE LIT02810  
 C RHLI DENSITY OF LITHIUM LB. / FT<sup>3</sup> LIT02820  
 C RHQA DENSITY CELL GAS LB/FT<sup>3</sup> LIT02830  
 C RHQA1 INITIAL DENSITY OF CELL GAS LB/FT<sup>3</sup> LIT02840  
 C RHOL1H DENSITY OF LITHIUM HYDROXIDE LB/FT<sup>3</sup> LIT02850  
 C RHOL1N DENSITY OF LITHIUM NITRIDE LB/FT<sup>3</sup> LIT02860  
 C RHOL1O DENSITY OF LITHIUM OXIDE LB/FT<sup>3</sup> LIT02870  
 C RHOLIV LITHIUM VAPOR DENSITY ABOVE POOL LB/FT<sup>3</sup> LIT02880  
 C RH PAN DENSITY OF LI SPILL PAN LBS/FT\*<sup>2</sup> LIT02890  
 C RHPB DENSITY OF ALLOY METAL LB-MOLE/FT<sup>3</sup> LIT02910  
 C RHSTL DENSITY OF STEEL LINER (LB/FT<sup>3</sup>) LIT02920  
 C RIFCZG RADIATIVE INTERCHANGE FACTOR BETWEEN COMB. ZONE AND THE CELL GAS LIT02930  
 C RIFCZP RADIATIVE INTERCHANGE FACTOR BETWEEN COMB. ZONE AND THE PCOL SURFACE LIT02950  
 C RIFCZW RADIATIVE INTERCHANGE FACTOR BETWEEN COMB. ZONE AND CONTAINMENT WALLS LIT02960  
 C RIFFAG RADIATIVE INTERCHANGE FACTOR - PAN TO GAS LIT02970  
 C RIFFAS RADIATIVE INTERCHANGE FACTOR - PAN TO STEEL FLOOR LIT03000  
 C RIFPG RAD. INT. FAC. BETWEEN POOL AND CELL GAS LIT03010  
 C RIFPW RAD. INT. FACT. BETWEEN POOL AND WALL LIT03020  
 C RIFSLC RADIATIVE INTERCHANGE FACTOR BETWEEN STEEL LINER LIT03030  
 C RIN UNIVERSAL GAS CONSTANT 1545 FT. LBF / LB. MOLE-DEG. F LIT03040  
 C RN1LB RATE OF NITROGEN CONSUMPTION LB./ SEC LIT03050  
 C RN2 DEGREE TO WHICH NITROGEN-LI REACTION OCCURS. VALUE IS LIT03070  
 C BETWEEN ZERO AND ONE (0 FOR NO REACTION, \*1 FOR COMPLETE LIT03080  
 C ROXLB RATE OF OXYGEN CONSUMPTION BY POOL FIRE LB./SEC. LIT03090  
 C RRAD INITIAL RADIUS OF REACTION ZONE FT LIT03100  
 C RTLI,RTG,RADB,RADW,RADCB,RADW,VARIOUS RATES OF TEMP. LIT03110  
 C CHANGE OF NODES DEG. F/SEC. LIT03120  
 C RVOL INITIAL REACTION ZONE VOLUME FT<sup>3</sup> LIT03130  
 C RVOL1 REACTION ZONE VOLUME FT<sup>3</sup> LIT03140  
 C RWLB RATE OF WATER VAPOR CONSUMPTION LB./SEC LIT03150  
 C RWCZ,RCZW,RCZG,RADB,RADW,RADCB,RADW,RLIW,RWL1,RGL1,RLIG ---- LIT03160  
 C C VARIOUS RATES OF TEMP. CHANGE OF NODES DEG. F/SEC LIT03170  
 C R1 COEFFICIENT OF BREEDER IN WATER REACTION EQUATION LIT03180

C R2 COEFFICIENT OF ALLOY METAL IN WATER REACTION EQUATION LIT03190  
 C SFCLR HEAT REMOVAL RATE BY EMERGENCY COOLING OF STEEL LIT03200  
 C FLOOR LINER BTU/SEC LIT03210  
 C SFLIN TIME AFTER SPILL WHEN SFCLR BEGINS. SEC LIT03220  
 C SIGMA STEPHAN-BOLTZMAN CONSTANT .113E-8 BTU/FT\*2(HR/R)\*\*2 LIT03230  
 C SPILL TOTAL WEIGHT OF LITHIUM SPILLED LB. LIT03240  
 C SPRAY WEIGHT FRACTION OF LITHIUM CONSUMED IN THE SPRAY FIRE LIT03250  
 C TA AMBIENT TEMPERATURE DEG. F LIT03260  
 C TAU TIME CONSTANT FOR TRANSIENT NATURAL CONVECTION LIT03270  
 C TONE,TTWO,THREE TIME IN SECONDS AT WHICH EACH INJECTION OCCURS LIT03280  
 C TB(I) TEMP. OF 11TH NODE OF CONCRETE FLOOR DEG. R LIT03290  
 C TB(1) INITIAL TEMP. OF 11TH NODE OF CONCRETE FLOOR DEG. R LIT03300  
 C TBF, TCF, TGF, ETC. CORRESPONDING TEMP. IN DEGREES FAHRENHEIT LIT03310  
 C TBLOW INERT GAS INLET TEMP. DEG. R LIT03320  
 C TC(I) TEMP. OF 11TH NODE OF CONCRETE WALL DEG. R LIT03330  
 C TCIC(1) INITIAL TEMP. OF 11TH NODE OF CONCRETE WALL DEG. R LIT03340  
 C TCZ COMBUSTION ZONE TEMPERATURE DEG. R LIT03350  
 C TCZF COMBUSTION ZONE TEMP. IN FAHRENHEIT DEG. R LIT03360  
 C CZ1 INITIAL VALUE OF COMB. ZONE TEMP. DEG. R LIT03370  
 C TE EQUILIBRIUM TEMP. RESULTING FROM SPRAY FIRE DEG. R LIT03380  
 C TFEFF NORMALIZED TEMP. OF LI POOL/COMB. ZONE FILM LIT03390  
 C TG GAS TEMP. AFTER SPRAY FIRE DEG. R LIT03400  
 C TGF CONTAINMENT GAS TEMP. IN FAHRENHEIT DEG. R LIT03410  
 C GZERO INITIAL CELL GAS TEMP. DEG. R LIT03420  
 C THF, THW CONCRETE FLOOR AND WALL THICKNESSES DEG. R LIT03430  
 C INPUT AS INCHES. USED IN FT. FOR PROGRAM LIT03440  
 C THK1N1 INNER INSULATION THICKNESS INPUT AS INCHES LIT03450  
 C THK1N2 OUTER INSULATION THICKNESS INPUT AS INCHES LIT03460  
 C THKPAN SPILL PAN THICKNESS IN FEET (INPUT AS INCHES) LIT03470  
 C TIME TIME AFTER SPILL HAS OCCURRED SEC. LIT03480  
 C TIMEF STOP INTEGRATION TIME SEC. LIT03490  
 C TIMEO OUTPUT TIME INDICATOR SEC. LIT03500  
 C TINS1 TEMP. OF INNER NODE OF INSULATION DEG. R LIT03510  
 C TINS2 TEMP. OF OUTER NODE OF INSULATION DEG. R LIT03520  
 C TLII LITHIUM TEMP. IN POOL DEG. R LIT03530  
 C TLIF LITHIUM POOL TEMP. IN FAHRENHEIT DEG. R LIT03540  
 C TLLI INITIAL LITHIUM POOL TEMP. (DEG R) LIT03550  
 C TLIO INITIAL LITHIUM POOL TEMP. DEG. R LIT03560  
 C TMELT MELTING TEMP. OF LITHIUM DEG. R LIT03570  
 C TN TEMP. OF BREEDER ZONE ELEMENT DEG. R LIT03580  
 C TO TEMP. OF CELL GAS AFTER SPRAY FIRE DEG. R LIT03590  
 C TPAN LITHIUM PAN TEMP (DEG R) SUSP PAN OPTION LIT03600  
 C TPANF LITHIUM PAN TEMP (DEG F) LIT03610  
 C TPANZO INITIAL PAN TEMPERATURE IN DEGREES R LIT03620  
 C TS STEEL WALL LINER TEMP. DEG. R LIT03630  
 C TSB STEEL FLOOR LINER TEMP. DEG. R LIT03640  
 C TSBF FLOOR STEEL LINER TEMPERATURE IN DEGREES FAHRENHEIT LIT03650  
 C TSB1 INITIAL STEEL FLOOR LINER TEMP. DEG. R LIT03660  
 C TSF WALL STEEL LINER TEMPERATURE IN FAHRENHEIT LIT03670  
 C TSZERO INITIAL STEEL WALL LINER TEMP. DEG. R LIT03680  
 C TVAP BOILING POINT OF LITHIUM DEG. R LIT03690  
 C T1 FILM TEMP. BETWEEN CELL GAS AND POOL DEG. R LIT03700  
 C T2 FILM TEMP. BETWEEN CELL GAS AND STEEL WALL LINER DEG. R LIT03710

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C USUBA HEAT TRANSF. COEFF. : CONTAINMENT-AMBIENT BTU/SEC-FT2-DEGLIT03720  
 C V CELL FREE VOLUME FT<sup>3</sup> LIT03730  
 C VOL VOLUME OF BREEDER ELEMENT LIT03740  
 C W THICKNESS OF STEEL POOL LINER FT. LIT03750  
 C (INPUT AS INCHES) LIT03760  
 C WA WT. FRACTION OF INERT GAS IN ATMOSPHERE LIT03770  
 C WAB WEIGHT FRACTION OF INERT GAS IN FLOODING GAS LIT03780  
 C WN2 WEIGHT FRACTION OF NITROGEN IN ATMOSPHERE IN FLOODING GAS LIT03790  
 C WN2B WEIGHT FRACTION OF NITROGEN IN ATMOSPHERE IN FLOODING GAS LIT03800  
 C WO2 WEIGHT FRACTION OF OXYGEN IN ATMOSPHERE IN FLOODING GAS LIT03810  
 C WO2B WEIGHT FRACTION OF OXYGEN IN FLOODING GAS LIT03820  
 C WWA WEIGHT FRACTION OF WATER VAPOR IN CONTAINMENT ATMOSPHERE LIT03830  
 C WWAB WT. FRACTION OF WATER VAPOR IN FLOODING GAS LIT03840  
 C XBLOW USED IN CONJUNCTION WITH IBLW LIT03850  
 C XESC USED IN CONJUNCTION WITH IESC LIT03860  
 C XMAIR AMOUNT OF GAS IN CONTAINMENT AFTER SPRAY LB.-MOLES LIT03870  
 C XLI WEIGHT FRACTION OF LITHIUM IN LIPB ALLOY LIT03880  
 C XPB WEIGHT FRACTION OF ALLOY METAL LIT03890  
 C XMOL MOL. WEIGHT OF CONTAINMENT GAS LB./LB.-MOLE LIT03900  
 C XMOLA MOLECULAR WT. OF INERT GAS LB./LB.-MOLE LIT03910  
 C XMOLAB MOL. WT. OF INERT FLOODING GAS LIT03920  
 C XSFL INDICATES EMERGENCY COOLING OF FLOOR STEEL LIT03930  
 C XSP1=0. FOR NO COOLING , XSFL=1. FOR COOLING LIT03940  
 C (1/SEC.) LIT03950  
 C XSFLC USED IN CONJUNCTION WITH ISFLC LIT03960  
 C YALICZ EFFECTIVE THERMAL ADMITTANCE, FILM-COMB. ZONE BTU/SEC-DEGLIT03970  
 C YALIG EFFECTIVE THERMAL ADMITTANCE, POOL-CELL GAS BTU/SEC-DEGLIT03980  
 C YAPCZ EFFECTIVE THERMAL ADMITTANCE, POOL-COMB. ZONE BTU/SEC-DEGLIT03990  
 C ZLI THICKNESS OF LITHIUM NODE FT. LIT04000  
 C ZZ TEMPERATURE RATE OF CHANGE IN BREEDER ELEMENT LIT04010  
 C Z21 POOL TEMP. RATE OF CHANGE DEG. F/SEC. LIT04020  
 C Z22 LI SPILL PAN TEMP. RATE OF CHANGE (DEG R/SEC) LIT04030  
 C Z24 CELL GAS TEMP. RATE OF CHANGE DEG. F/SEC. LIT04040  
 C Z25 STEEL WALL LINER TEMP. RATE OF CHANGE DEG. F/SEC. LIT04050  
 C Z26 COMB. ZONE TEMP. RATE OF CHANGE DEG. F/SEC. LIT04060  
 C Z27 FLOOR STRUCTURE TEMP. RATE OF CHANGE DEG. F/SEC. LIT04070  
 C Z28 INNER INSULATION TEMP. RATE OF CHANGE (SUSP. PAN OPTION) LIT04080  
 C Z29 OUTER INSULATION TEMP. RATE OF CHANGE (SUSP. PAN OPTION) LIT04090  
 C ZZ99 USED TO ENSURE POSITIVE COMBUSTION RATE LIT04100  
 C  
 C PROGRAM DECISION FLAGS  
 C  
 C IBLW = 1 FLOOD CONTAINMENT WITH INERT GAS.  
 C = 0 NO CONTAINMENT FLOODING.  
 C ICMB = 0 NO OXYGEN LEFT AFTER SPRAY FIRE.  
 C = 1 THERE IS STILL OXYGEN LEFT AFTER SPRAY FIRE.  
 C SET INITIALLY TO 1 AND THEN RESET TO 0 WHEN THE  
 C PROGRAM CALCULATES THAT THE OXYGEN HAS RUN OUT.  
 C  
 C ICNI = 1 NITROGEN REACTIONS POSSIBLE.  
 C = 0 NITROGEN REACTIONS NOT POSSIBLE.  
 C  
 C ICZ = 1 COMBUSTION ZONE MODEL USED  
 C = 0 COMBUSTION ZONE MODEL NOT USED  
 C  
 C IESC = 1 EMERGENCY SPACE COOLING OPTION  
 C  
 C

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C      = 0    NO EMERGENCY SPACE COOLING          LIT04250
C      LIT = 0    NO LITHIUM LEFT TO BURN           LIT04260
C      = 1    LITHIUM LEFT TO BURN (INITIAL CONDITION).
C      IMETH = 1    RUNGE-KUTTA METHOD OF INTEGRATION USED.
C      = 3    SIMPSON'S RULE METHOD OF INTEGRATION USED.
C      IPAGE   NO. OUTPUT LINES PER PAGE        LIT04290
C      ISFLC = 1    EMERGENCY COOLING OF STEEL FLOOR LINER OPTION
LIT04300
C      = 0    NO EMERGENCY COOLING OF STEEL FLOOR LINER
LIT04310
C      FLAG = .TRUE.    NO WALL CONCRETE          LIT04320
C      FLAG1 = .TRUE.    NO FLOOR CONCRETE         LIT04330
C      FLAGJ = .TRUE.    INJECTIONS OF DRY GAS DURING RUN
LIT04340
C      FLAGL = .TRUE.    LILP IS FIXED AT A MINIMUM
LIT04350
C      FLAGS = .TRUE.    YES ON SUSPENDED PAN GEOMETRY
LIT04360
C      FLAGSI = .TRUE.    IF USER WISHES INPUT/OUTPUT IN SI UNITS
LIT04370
C      IOPTB=1    IF INTERNAL BLANKET ACCIDENT OPTION IS TO BE USED
LIT04380
C
C      IMPLICIT REAL (K,L,M)                      LIT04400
LIT04410
0001   DIMENSION TCF(20),TBF(20),TC(20),DICTD(20),DTBDT(20),
       *          TCIC(20),IBIC(20),NAME('100'),L(20),C4(20),C10(20),
       *          TN(20),RAD(20),ARE(20),VOL(20),CF(20),CT(20),ZZ(20),
       *          TNF(20),QC(20),ATO(20),AT(20),
       *          LOGICAL FLAG,FLAGI,FLAGL,FLAGS,FLAGJ,FLAGSI,FLAGIO
LIT04420
C      REAL INTBL
LIT04430
0002   REAL NT,MLI,MCZ,MLIOH,MLIO,MLIT,MLIV
       *          REAL NA,LBN,LBR,LB2,MB(20)
LIT04440
0003   COMMON /UNITS/ AN,V,CH,G,KGP,W,K,HA,HIN,THW,TMEL,TVP,TCZI,
       *          COMMON /IMETH/ ICOUNT,ISTORE,INOIN,IPASS,DELT,XIC(101),ZZZ(501),
LIT04450
0004   COMMON /NAME/ QPAN,RPHAN,CPPAN,TPANZ,APE,AWB,THPAN,BREDT,ASLI,SPILL,QCO1,
       *          QCO2,OCN,OCW,OCAP,OCAP,FLAGJ,FLAGS,FLAGIO,IBLW,IESC,ISFLC,RHOLIN,
LIT04460
0005   *          RHOLIO,RHOLIH,CPA,CPLI,RHLI,AKLI,CPSL,RHSL,KSTL,PCON,RHCON,
       *          KCON,PZERO,TZERO,TBFL,TATLII,DP1,DP2,DP3,ESCR,SFLCR,
LIT04470
0006   *          CPAB,TBLOW,EXHSTV,BLOW,BF,TCF,THKIN2,THKIN1,AINS,CPINS,RHINS,
       *          IOPTB,NT,RRAD,AMLI,HCO,QLIOH,PBMELT,CP1,CP2,RHPLB,
LIT04480
0007   *          .AMPB,XLI,CPLL,CPLI1,CPPB1,R1,R2,J1,J2,QMELT,QMELD,DELH
LIT04490
0008   COMMON /EXTRA/ TINS1F,TINS2F,TB,TC,TG,TGF,TSB,TSBF,TS,TSF,YCZ,
       *          TCZF,TLI,TLIF,TPAN,TPANF,OVERP,MOX,MNI,LILP,HF,HB,EFILM,DFILM,
LIT04500
0009   *          TINS1,TINS2
LIT04510
C*****INPUT SECTION*****                                *****
C*****READ IN TITLE AND HEADINGS*****                *****
C*****CONTINUE READ IN FILE OPTIONS*****             *****
0010   998 CONTINUE READ(5,707,END=999) (NAME(I),I=1,100)
LIT04520
C      707 FORMAT(20A4)
LIT04530
C      **** READ IN FLAGS AND OPTIONS ****
C      **** READ (5,705) FLAG,FLAG1,FLAGS,FLAGJ,FLAGSI,IBLOW,IESC,ISFLC,IOPTB
LIT04540
0011   705 FORMAT (5(L1,1X),4(I1,1X))                  *****
LIT04550
C*****READ IN CONTAINMENT SPECIFICATIONS*****        *****
LIT04560
0012   READ(5,702) AW,V,CH,W02,WNA,W
LIT04570
0013
0014
0015
0016

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 0017 READ(5,702) G,KGAP,W,K,EMGF,BETA LIT04780  
 0018 READ(5,702) THW,THF,AIB LIT04790  
 0019 READ(5,709) NL,(L(I)) I=1,NL LIT04800  
 0020 READ(5,709) NL,(L(I)),I=1,NL1 LIT04810  
 709 FORMAT(12,15F5.3) LIT04820  
 0022 IF (FLAGS) READ(5,702) TPANZO,APAN, AINS,BREDF LIT04830  
 0023 IF (FLAGS) READ(5,702) THKPN,THKIN, THKIN2 LIT04840  
 0024 IF (FLAGS) READ(5,702) KPN,RHPN,CPPN,RHNS,CPINS,EMINS LIT04850  
 0025 READ(5,702) TEHCZD,XMEHG,AEHC,CPEHC LIT04860  
 C  
 C\*\*\*\*\* READ IN HEAT TRANSFER CORRELATION COEFFICIENTS \*\*\*\*\*  
 0026 READ(5,702) HIN,HINEHC,HINGS,HINSAM,TAUZ LIT04880  
 C  
 C\*\*\*\*\* READ IN SPILL PARAMETERS \*\*\*\*\*  
 0027 READ(5,702) ASLI,SPILL,SPRAY,FRA,RA,XMOLA LIT04910  
 0028 702 FORMAT(6F12.4) LIT04920  
 C  
 C\*\*\*\*\* READ IN REACTION CONSTANTS \*\*\*\*\*  
 0029 READ(5,702) QCO1,QCO2,QCN,QCW LIT04950  
 0030 READ(5,702) RCMBO1,RCMBO2,RCMBN,RCMBW,RCMBH2 LIT04960  
 0031 READ(5,702) TMELT,TVAP,TVAP,EMCZ,TCZ1,PERCEN LIT04970  
 C  
 C\*\*\*\*\* READ IN PHYSICAL CONSTANTS \*\*\*\*\*  
 0032 READ(5,702) RHOL0,RHOL1,RHOLH,CPA LIT05000  
 0033 READ(5,702) CP11,RHL1,AKL,EMSL1,EML1,EMCNC LIT05020  
 0034 READ(5,702) CPST1,RHST1,KST1,CPCON,RHCON,KCON LIT05030  
 C  
 RHST1 IS CORRECTED FOR EXTRANEous FLANGES, ETC. LIT05040  
 C  
 C\*\*\*\*\* READ IN INITIAL CONDITIONS \*\*\*\*\*  
 0035 READ(5,702) PAZERO,TZERO,TSBI,TA,TLII LIT05060  
 C  
 C\*\*\*\*\* READ IN INTEGRATION CONTROL PARAMETERS \*\*\*\*\*  
 0036 READ(5,706) IMETH, DTMIN, TIMEF, RELERR, DELOUT LIT05070  
 0037 706 FORMAT(14,5F12.4) LIT05080  
 C  
 C\*\*\*\*\* READ IN INJECTION PARAMETERS \*\*\*\*\*  
 0038 1030 FORMAT(14) LIT05090  
 0039 IF (FLAGS) READ(5,1030)TONE,DP1,FCT1,THREE,DP3,FCT3,LIT05100  
 C  
 C\*\*\*\*\* CONTAINMENT FLOODING WITH INERT GAS OPTION \*\*\*\*\*  
 0040 XBLOW=0. LIT05120  
 0041 IF (IBLOW.EQ.1) READ(5,702) W026,WWAB,WN2B,XMOLAB,CPAB,TBLOW,BLOWN,EXHSTV,BLIN,BLOUT LIT05130  
 C  
 C\*\*\*\*\* GO TO 437 \*\*\*\*\*  
 0042 708 FORMAT(14) LIT05140  
 0043 IF (IBLOW.EQ.1) GO TO 437 LIT05150  
 0044 BLIN=0. LIT05160  
 0045 BLOUT=0. LIT05170  
 0046 W02B=0. LIT05230  
 0047 WAB=0. LIT05240  
 0048 WN2B=0. LIT05250  
 0049 WMAB=0. LIT05260  
 0050 XMOLAB=1. LIT05270  
 0051 CPAB=1. LIT05280  
 LIT05290  
 LIT05300

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0052      TBLOW=1.
0053      BLOWV=0.
0054      EXHSTV=0.
0055      CONTINUE
0056      WAB1 = -WD2B-WN2B-WMAB
0057      BLDWR = 1.35E-03 * BLOWV
0058      EXHSTR = 1.35E-03 * EXHSTV
C*****EMERGENCY SPACE COOLING OF CONTAINMENT OPTION*****
C*****XESC=0.
C*****IF(IESC.EQ.1) READ(5,702) ESCR,ESCTIN
C*****IF(IESC.EQ.1) GO TO 439
C*****IF(SFLC.EQ.1) READ(5,702) SFLCR,SFLTIN
C*****ASF1=0.
C*****SFLCR=0.
C*****ESCR=0.
C*****ESCTIN=0.
C      439 CONTINUE
C*****IF(SFLC.EQ.1) GO TO 440
C*****EMERGENCY STEEL FLOOR LINER COOLING OPTION
C*****SFLTIN=0.
C      440 CONTINUE
C*****INTERNAL BLANKET ACCIDENT OPTION
C*****IF (LOPTB .EQ. 1) READ (5,702) NT,RRAD,AMLI,HCO,QLIOH,CPLTOH
C*****IF (LOPTB .EQ. 1) READ (5,702) PBMLT,CP1,CP2,RHPB,AMPB,XLI
C*****IF (LOPTB .EQ. 1) READ (5,702) CPPL,CPB1,R1,R2
C*****IF (LOPTB .EQ. 1) READ (5,702) J1,J2,QMELT,QMELTP,DEIH
C*****IF (LOPTB .EQ. 1) GO TO 441
C      441 CONTINUE
C*****PRINT OUT THE INPUT
C*****C
C      RRAD=0.
C      AMLI=7.0
C      HCO=1.
C      QLIOH=0.
C      CPLTOH=1.
C      CONTINUE
C*****PRINT OUT THE INPUT
C*****C
C      RCMBO=((100.-PERCENT)*RCMBO1+PERCENT*RCMBO2)/100.
C      QCO=((100.-PERCENT)*RCMBO1*QCO1+PERCENT*RCMBO2*QCO2)/(RCMBO*100.)
ZLI=SPILL/RHLI/ASLI
0083      WRITE (6,802) (NAME(I), I=1,60)
0084      FORMAT ('1',3'(20A4,/),/')
0085      802      WRITE (6,142) CP1,RHLI,RHOL10,AKLI,EMLI,CPS1L,RHSTL,RHOLIN,KSTL,
0086          EMSTL,CPCON,RHCON,RHOLIN,KCON,EMCONC,
0087          WRITE (6,143) AW,V,TIW,K,CH,W,THF,G
0088          WRITE (6,144) ASLI,SPILL,SPRAY,W0,ZLI
0089          WRITE (6,153) FLAGS,FLAGJ,FLAGS1
0090          WRITE (6,155) HIN,HINERC,HINGS,HINSAM
0091          WRITE (6,155) QCO,RCMBO,TVAP,RCMBH2,PERCEN,QCIN,RCMBN,TMELT,FRA,
0092
0093

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        OCW, RCMBW, QVAP, RA
      WRITE(6,146) TGZERO,TLII,TSBI,WA,PAZERO,TSZERO,TCZI,TA,WMA
      WRITE(6,147) IMETH,DTMIN,TIMEF,RELERR,DELOUT
      WRITE(6,148) IBLOW,IESC,SFLC,FLAG,FLAG,IOPTB
      WRITE(6,149) WO2B,BLOW,ESCR,SFLC,CPA,WWAB,BLOUT,ESCTIN,SFLTIN,
      CPAB,WN2B,BLIN,EXHSTV,TBLOW,XMOLAB
      IF (FLAGS) WRITE(6,151) TPANZO,APAN,AWB,THKPAN,BREDTH,KPAN,RHPAN,
      CPPAN,EMINS
      IF (FLAGS) WRITE(6,152) THKIN1,THKIN2,AINS,RHINS,CPIINS
      WRITE(6,154) XMEHC,XMEHC,AEHHC,CPEHC
      IF (IOPTB .EQ. 1) WRITE(6,101) NT,RAD,AMLI,HCO,QLIOH,CPLIOH,
      PBMLT,CP1,CP2,RHPC,CP1,CP2,XLI,CP1,CP2,R1,R2,
      J1,J2,QMELT,QMELTP,DELH
      FORMAT(' PHYSICAL PROPERTIES ','1X,19(1H-)/T10','CPLI = ',F12.4,
      T35,'RHLI = ',F12.4,T60,'RHLID = ',F12.4,T85,'AKLI = ',F12.4,
      T110,'EMLI = ',F12.4/T10,'CPSTL = ',F12.4,T35,'RHSTL = ',F12.4,
      T60,'RHOLIN = ',F12.4,T85,'KSTL = ',F12.4,T110,'EMSTL = ',F12.4/
      ,T10,'CPCON = ',F12.4,T135,'RHCN = ',F12.4/T60,'RHOLIH = ',F12.4,
      T85,'KCON = ',F12.4/T110,'EMCN = ',F12.4/
      ,T60,'KGAP = ',F12.4,T85,'EMCZ = ',F12.4//)
      FORMAT(' INNER CONTAINMENT DIMENSIONS ','/1X,28(1H-)//T10','AW = ',
      F12.4,T35,V = ,F12.4,T60,THW = ,F12.4,T85,K = ,F12.4//T10,
      CH = ,F12.4,T35,W = ,F12.4,T60,THF = ,F12.4,T85,G = ,
      F12.4//)
      FORMAT(' SPILL PARAMETERS ','/1X,16(1H-)/T10','ASLI = ',F12.4,T35,
      SPILL = ,F12.4,T60,'SPRAY = ',F12.4,T85,W0 = ,F12.4,T110,
      ZLI = ,F12.4//)
      FORMAT(' HEAT TRANSFER CORRELATION COEFFICIENTS ','/1X,38(1H-)//T10,
      HIN = ,F12.4,T35,'HINEHC = ',F12.4,T60,'HINGS = ',F12.4,
      T85,'HNSAM = ',F12.4//)
      FORMAT(' COMBUSTION PARAMETERS ','/1X,21(1H-)/T10','QCO = ',F12.4,
      T35,'RCMBO = ',F12.4,T60,'TVAP = ',F12.4,T85,'RCMBH2 = ',F12.4,
      T110,'PERCENT = ',F12.4//T10,'QCN = ',F12.4,T35,'RCMBN = ',F12.4,
      TGO,'TMELT = ',F12.4,T85,FRA = ,F12.4/T10,'QCN = ',F12.4//T10,
      RCMBW = ,F12.4,T60,'QVAP = ',F12.4,T85,'RA = ',F12.4//)
      FORMAT(' INITIAL CONDITIONS ','/1X,8(1H-)/T10','TGZERO = ',F12.4/T35,
      T35,'TLII = ',F12.4,T60,'TSBI = ',F12.4,T85,'WA = ',F12.4,T110,
      PAZERO = ,F12.4//T10,'TSZERO = ',F12.4,T35,'TCZI = ',F12.4,T60,
      ITA = ,F12.4,T85,WA = ,F12.4//)
      FORMAT(' INTEGRATION CONTROL PARAMETERS ','/1X,30(1H-)/T10',
      IMETH = ,I4,T35,DTMIN = ,F12.4,T60,'TIMEF = ',F12.4,T85,
      RELERR = ,F12.4,T110,'DELOUT = ',F12.4//)
      FORMAT(' OPTIONS IN EFFECT ','/1X,7(1H-)/T10','IBLOW = ',I4,T35,
      IESC = ,I4,T60,'ISFLC = ',I4,T85,'FLAG = ',L4,T110,'FLAG1 = ',
      L4//T10,IOPTB = ,I4//)
      FORMAT(' MISCELLANEOUS INPUT ','/1X,19(1H-)/T10,'W02B = ',F12.4,T35,LIT06290
      BLOW = ,F12.4,T60,'ESCR = ',F12.4,T85,'SFLC = ',F12.4,T110,
      CPA = ,F12.4//T10,'WWAB = ',F12.4,T35,'BLOUT = ',F12.4,T60,LIT06300
      ESCTIN = ,F12.4,T85,'SFLTIN = ',F12.4,T110,'CPAB = ',F12.4//T10,LIT06310
      WN2B = ,F10.4,T30,'BLIN = ',F10.4,T50,'EXHSTV = ',F10.4,T70,LIT06320
      TBLOW = ,F10.4,T90,'XMOLA = ',F10.4,T110,'XMOLAB = ',F10.4,LIT06330
      0111 151 FORMAT('// DATA FOR SUSPENDED PAN OPTIONAL GEOMETRY: ','/1X,
      41(1H-),//T10,'TPANZO = ',F12.4,T35,'APAN = ',F12.4,T60,'AWB = LIT06360
      
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        ,F12.4,T85,'THKPN=' ,F12.4,T110,'BREDFH =' ,F12.4//T10,
        'KPAN =' ,F12.4,T35,'RHPAN =' ,F12.4,T60,'CPAN =' ,F12.4,T85, LIT06370
        'EMINS =' ,F12.4)
0112    152    FORMAT(//T10,'THKIN1 =' ,F12.4,T35,'THKIN2 =' ,F12.4,T60,'AINS =' ,LIT06380
        'F12.4,T85,'RHINS =' ,F12.4,T110,'CPINS =' ,F12.4) LIT06390
0113    153    FORMAT( T60,'FLAGS =' ,L1,'85,'FLAGU =' ,L1,T110,'FLAGS1 =' ,L1//),
        L1,T110,'EXTANEOUS HEAT CAPACITY NODE DATA /X,33(1H-)//T10, LIT06400
        1,T1EHC20 =' ,F12.4,T35,'XMEHC =' ,F12.4,T60,'AEHC =' ,F12.4, LIT06410
        2,T85,'CPEHC =' ,F12.4)
0114    154    FORMAT(//,INTERNAL BLANKET OPTION INPUT,'/X,29(1H-)//T10,
        .NT =' ,F10.1,T35,'RRAD =' ,F10.2,T60,'AMLI =' ,F10.2, LIT06420
        .T85,'HCO =' ,F10.2,T110,'QLIGH =' ,F10.2//T10,'CPLIOH =' ,F10.2, LIT06430
        .T35,'POMELT =' ,F10.2,T60,'CR1 =' ,F10.3,T85,'CP2 =' ,F10.5, LIT06440
        .T110,'RHPB =' ,F10.2//T10,'AMPB =' ,F10.2,T35,'XL1 =' ,F10.3, LIT06450
        .T60,'CPPL =' ,F10.3,T85,'CPLI1 =' ,F10.2,T110,'CPPB1 =' ,F10.2//, LIT06460
        .T10,'R1 =' ,F10.3,T35,'R2 =' ,F10.3,T60,'U1 =' ,F10.1, LIT06470
        .T85,'U2 =' ,F10.1,T110,'QMELT =' ,F10.2//T10,'QMELTP =' ,F10.2, LIT06480
        .T110,'DELM =' ,F10.1) LIT06490
C***** C INITIALIZE PROGRAM VARIABLES LIT06500
C***** C
C
0115
0116    EMF=0.9
0117    GAMMA=1.4
0118    PG=1.
0119    FPM=1.
0120    IF(FLAGS)FPG=0.23
0121    IF(FLAGS)FPW=0.384
0122    C3=0.
0123    C5=0.
0124    C7=0.
0125    C8=0.
0126    C9=0.
0127    CMBRHI=0.
0128    LIBP=0.
0129    LILOX=0.
0130    LIINI=0.
0131    LEAKD=0.
0132    MLIN=0.
0133    MLINI=0.
0134    MLIH=0.
0135    MH2=0.
0136    OXLB=0.
0137    OXLB1=0.
0138    HTCAPG=1.
0139    OUTINT=0.
0140    TIME=0.
0141    RADCC=0.
0142    RADCB=0.
0143    RADB=0.
0144    ZZ2=0.

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0145      ZZB=0.          LIT06900
0146      ZZ9=0.          LIT06910
0147      ZZ6=0.          LIT06920
0148      CMBRD=0.        LIT06930
0149      CMBRN=0.        LIT06940
0150      CMBRW=0.        LIT06950
0151      RNLLB=0.        LIT06960
0152      RXLB=0.          LIT06970
0153      RWALB=0.        LIT06980
0154      IG=0.            LIT06990
0155      J=10PTB         LIT07000
0156      XPB=1.-XLI     LIT07010
0157      CGZ=0.          LIT07020
0158      MLIY=0.          LIT07030
0159      MLIDH=0.         LIT07040
0160      RVOL=4.189*RRAD**3 LIT07050
0161      MCZ1=RVOL/(AMLI/RHLI) LIT07060
0162      MLI1=625.87*RHLI/AMLI-MCZ1 LIT07070
0163      MLI1T=0.          LIT07080
0164      HA=1.            LIT07090
0165      H=1.              LIT07100
0166      NN=8.            LIT07110
0167      NA=8.            LIT07120
0168      ND=NN-1.          LIT07130
0169      DO 31 J=1,NN      LIT07140
0170      TN(J)=TLI1       LIT07150
0171      ZZ(J)=0.          LIT07160
0172      ZZ(1)=1.          LIT07170
0173      ZZ=1.             LIT07180
0174      CPLI=CPLI*AMLI   LIT07190
0175      CPMLI=CPMLI      LIT07200
0176      CPZ=1.             LIT07210
0177      DO 185 IAM=1,20    LIT07220
0178      C4(IAM)=0.        LIT07230
0179      C10(IAM)=0.        LIT07240
0180      DICDT(IAM)=0.      LIT07250
0181      DIBDT(IAM)=0.      LIT07260
0182      IF(FLAGSI) CALL SI LIT07270
0183      DELT=DTMIN          LIT07280
0184      HF=0.              LIT07290
0185      HB=0.              LIT07300
0186      RCMBO=((100.-PERCENT)*RCMBO1+PERCENT*RCMBO2)/100. LIT07310
0187      QC0=((100.-PERCENT)*RCMBO1*QC01+PERCENT*RCMBO2*QC02)/(RCMBO1+100.) LIT07320
0188      FLAGIO= FALSE       LIT07330
0189      IF (.NOT. FLAGS) TPANZO=0. LIT07340
0190      TPAN=TPANZO          LIT07350
0191      TINS1=0.5*(TPANZO+TGZERO) LIT07360
0192      TINS2=1-TGZERO        LIT07370
0193      TINS1=TINS1          LIT07380
0194      TINS2=TINS2          LIT07390
0195      TEHC=TEHCZ0          LIT07400
0196      DFILM=0.             LIT07410
0197      INJEC1=0.             LIT07420

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0198      INJEC2=0          LIT07430
0199      INJEC3=0          LIT07440
0200      TAU=120.          LIT07450
0201      SIGMA=4.7611E-13  LIT07460
0202      IF (IOPTB .EQ. 1) SIGMA=0.  LIT07470
0203      NLMI=NL-1          LIT07480
0204      NLMI=NL-1          LIT07490
0205      FLAG=.FALSE.       LIT07500
0206      FLAG=.FALSE.       LIT07510
0207      IF (IOPTB .EQ. 1) GO TO 25  LIT07520
0208      FLAG=.FALSE.       LIT07530
0209      IF (THW .LT. 0.01) FLAG=.TRUE.  LIT07540
0210      IF (THF .LT. 0.01) FLAG1=.TRUE.  LIT07550
0211      LIS=SPILL*SPRAY  LIT07560
0212      LIS=SPILL-LIS    LIT07570
0213      WN2=-1.-WN2-WWA-WA  LIT07580
0214      XMOL=1./(W02/32.+WN2/28.+WMA/18.+WVA/XMOLA)  LIT07590
0215      RIN=545./XMOL  LIT07600
0216      TIME0=-.001        LIT07610
0217      TI=TLII          LIT07620
0218      LI=LP=LIT          LIT07630
0219      LI=LILP          LIT07640
0220      ZLI=LILP/RHLLIASLI  LIT07650
0221      TCZ=TCZI          LIT07660
0222      FILEFT=1.0         LIT07670
0223      GIN=32.2          LIT07680
0224      TS=TSZERO         LIT07690
0225      TSB-TSBI          LIT07700
0226      DO 1001 I=1,NL    LIT07710
0227      TCIC(I)=TSZERO   LIT07720
0228      TC(I)=TSZERO     LIT07730
0229      LI(I)=THW*LI(I)/12.  LIT07740
0230      DO 1002 I=1,NL1    LIT07750
0231      TBIC(I)=TSBI     LIT07760
0232      TB(I)=TSBI       LIT07770
0233      LI(I)=THF*LI(I)/12.  LIT07780
0234      ICZ=1             LIT07790
0235      ICMB=1            LIT07800
0236      ILIT=1             LIT07810
0237      ICNI=1             LIT07820
0238      RHDA=PAZERO*I44./RIN/TGZERO  LIT07830
0239      RHDAI=RHDA        LIT07840
0240      IPAGE=50           LIT07850
0241      MNII=WN2*RHOAI*V  LIT07860
0242      MNII=MNII         LIT07870
0243      MCXI=W02*RHOAI*V  LIT07880
0244      MLIOI=LIS*(1.+RCMBO)/RCMBO  LIT07890
0245      MWAI=WWA*RHOAI*V  LIT07900
0246      MWAI=MWAI         LIT07910
0247      MAI=W*A*RHOAI*V  LIT07920
0248      MAI=MAI           LIT07930
0249      LIC=LIS           LIT07940
                                         LIT07950

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C*****      MAIN      CONVERSION TO FT. - LB. - SEC.    *****
C          W=W/12.          LIT07960
C          G=G/12.          LIT07980
C          THW=THW/12.      LIT07990
C          THF=THF/12.
C          IF (FLAGS) THKPN=THKPN/12.
C          IF (FLAGS) THKIN1=THKIN1/12.
C          IF (FLAGS) THKIN2=THKIN2/12.
C          AKL1=AKL1/3600.
C          KSTL=KSTL/3600.
C          KCN=KCN/3600.
C          KGAP=KGAP/3600.
C          IF (.NOT. FLAGS) THKPN=W
C          IF (.NOT. FLAGS) KPN=KPN/3600.
C          IF (IOPTB .EQ. 1) GO TO 850
C
C          ***** CHECK THAT ENOUGH OXYGEN IS LEFT FOR POOL FIRE AFTER SPRINKLER *****
C
C          OXLFS=WD2*RCMBO*LIS/RCMBO
C          IF (OXLFS.LT.0.0) LIS=RCMBO*WD2*RHOAI*V
C          IF (OXLFS.LT.0.0) OXLFS=0.0
C
C          **** SPRAY FIRE COMPUTATION STARTED *****
C          ****
C          IF (LIS.LE.0.0) GO TO 627
C          TC=TGZERO
C          QIN=LIS*(QCO+CPLI*(TLI-T0))
C          FF2=0 IN
C          TE=TGZERO+1.
C          140 CONTINUE
C          **** SPECIFIC HEAT FOR DILITHIUM OXIDE *****
C          CP = .0602*T**.326 T = DEG. R
C          IF A DIFFERENT REACTION PRODUCT IS DESIRED, THE INTEGRAL OF THE
C          DESIRED PRODUCT MUST BE SUBSTITUTED IN QOUT1.
C
C          QOUT1=(1.+RCMBO)/RCMBO*LIS*(0.0602/1.326)*(TE**1.326-T0**1.326)
C          QOUT2=WH2*RHUA*V*(.172*(TE-T0)+8.57E-6/2.0*(TE**2.-T0**2.))+1.02E-9
C          /3. *(TE**3.-T0**3.)
C          QOUT3=OXLFS* (.184*(TE-T0)+3.2E-6/2.0*(TE**2.-T0**2.))+1.36E04 *
C          (1./TE-1./T0)
C          QOUT4=NWA*RHOAV*(0.44*(TE-T0))+WA*RHOAV*CPA*(TE-T0)
C          FF1=QIN-QOUT1-QOUT2-QOUT3-QOUT4
C          IF (FF1*FF2.LT.0.) GO TO 150
C          TE=TE+1.
C          IF (TE.GT.1.0E06) GO TO 910
C          FF2=FF1
C          GO TO 140
C
C          PORTION OF PROGRAM FOR GETTING INITIAL GAS TEMP. AND PRESS.    ***LIT08480
C
0250
0251
0252
0253
0254
0255
0256
0257
0258
0259
0260
0261
0262
0263
0264
0265
0266
0267
0268
0269
0270
0271
0272
0273
0274
0275
0276
0277
0278
0279
0280
0281
0282
0283

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0284      627 CONTINUE          LIT08490
0285        TE=1GZERO          LIT08500
0286      150 CONTINUE          LIT08510
0287        TG=IE              LIT08520
0288        MOX-MOXI-LIS/RCMBO  LIT08530
0289        MOXI=MOX            LIT08540
0290      MLI0=MLI01           LIT08550
0291        XMAIR=MNI1/2.8.+MOX/32.+MA/XMOLA+MNA/18.
0292        PZERO=1545.*XMAIR*TG/144./V          LIT08560
0293        PA=PZERO           LIT08570
0294        TGZERO=TG          LIT08580
0295        WRITE(6,726) TG,PZERO          LIT08590
0296        FORMAT(//,SPRAY FIRE RESULTS' /1X,1B(1H-)//5X,'TGZERO = ',F6.1,
0297          PZERO = ',F6.3//)
0297        IF (FLAGJ) WRITE(6,1090) TONE,DP1,FCT1,TTWO,DP2,FCT2,THREE,DP3,FCT3
0298        FORMAT(//,' GAS INJECTION DATA' ,/1X,1B(1H-)//3(5X,'AT TIME = ',
0298          'F0.0.5X,'DETA P = ',F8.4.5X,'FCT N2 = ',F8.4.5X,'/)//)
0299        C***** SPRAY FIRE COMPUTATION CONCLUDED *****          LIT08650
0300      200 CONTINUE          LIT08750
0301        C  ARTIFICIAL INJECTION OF OXYGEN AND NITROGEN          LIT08760
0302        C  TO MODEL HEIDI EXPERIMENTAL PROCEDURE          LIT08770
0303        MODINJ=0.          LIT08780
0304        MININJ=0.          LIT08790
0305        IF (.NOT. FLAGJ) GO TO 1060          LIT08800
0306        IF (TIME .LT. TONE .OR. TIME .GT. TONE+60.) GO TO 1040          LIT08810
0307        IF (INJEC1 .EQ. 0 .AND. DP1 .GT. 0.0) WRITE(6,1080) TONE,DP1          LIT08820
0308        MODINJ=MODINJ/60.          LIT08830
0309        MININJ=MININJ/60.          LIT08840
0310        MODINJ=MODINJ/60.          LIT08850
0311        MININJ=MININJ/60.          LIT08860
0312        IF (TIME .LT. TTWO .OR. TIME .GT. TTWO+60.) GO TO 1050          LIT08870
0313        IF (INJEC2 .EQ. 0 .AND. DP2 .GT. 0.0) WRITE(6,1080) TTWO,DP2          LIT08880
0314        INJEC2=1          LIT08890
0315        MODINJ2= 2.9822*V/TG+DP2*(1.-FCT1)          LIT08940
0316        MININJ2= 2.6094*V/TG+DP2*FCT2          LIT08950
0317        MODINJ=MODINJ/60.          LIT08960
0318        MININJ=MININJ/60.          LIT08970
0319        IF (TIME .LT. THREE .OR. TIME .GT. THREE+60.) GO TO 1060          LIT08980
0320        IF (INJEC3 .EQ. 0 .AND. DP3 .GT. 0.0) WRITE(6,1080) THREE,DP3          LIT09000
0321        INJEC3=1          LIT09010
0322        MODINJ3= 2.9822*V/TG+DP3*(1.-FCT3)          LIT09010

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 MNINJ3= 2.6094\*V/TG\*DP3\*FCT3 LIT09020  
 MOXINJ=MOXJ3/60. LIT09030  
 MNINJ=MNINJ3/60. LIT09040  
 1060 CONTINUE LIT09050  
 IF (IOPTB .EQ. 1) GO TO 21 LIT09060  
 C COMPUTE PHYSICAL PROPERTIES DEPENDENT ON TEMPERATURE LIT09070  
 C CALCULATE AIR COMPOSITION AND SPECIFIC HEAT AT CONST. VOLUME \*\*\* LIT09090  
 C MAIR MOX+MN1+MWA+MH2+MA LIT09100  
 RHOA=MAIR/V LIT09110  
 FOX=MOX/MAIR LIT09120  
 FVA=MWA/MAIR LIT09130  
 FN1=MN1/MAIR LIT09140  
 CPNOX= (0.184 + 3.2E-06\*TG - 1.36E04 /(TG\*\*2))\*MOX LIT09150  
 CPNM1= (0.172 + 8.57E-06\*TG + 1.02E-09\*TG\*\*2)\*MN1 LIT09160  
 CPWA=0.44 LIT09170  
 CPH2=3.76 LIT09180  
 CPLIH=0.67 LIT09190  
 CPN2= (0.172+8.57E-06\*TG+1.02E-09\*TG\*\*2) LIT09200  
 CPLIO=.0602\*TG\*.326 LIT09210  
 CPLIN=.3368+3.67\*-4\*TG LIT09220  
 CPLI0=CPLIO\*MN10 LIT09230  
 CPLIZ=CPLI LIT09240  
 AKLZ=AKLI LIT09250  
 TLI=1216. LIT09260  
 IF (J2 .EQ. 0.) GO TO 851 LIT09270  
 TLIK=(TLI-462.)/1.8+273. LIT09280  
 TCZK=(TCZ-462.)/1.8+273. LIT09290  
 CPPB=CP1+CP2\*TLIK LIT09300  
 CPPZ=CP1+CP2\*TCZK LIT09310  
 IF (TLI .GT. PBMLT) CPPB=CPPL LIT09320  
 IF (TCZ .GT. PBMLT) CPPZ=CPPL LIT09330  
 CPFFAC=0.004938\*TLI LIT09340  
 CPFAZ=0.004938\*TG -6.20741 LIT09350  
 CPLI=1.0037-.01063\*CPFFAC+.00564\*CPFFAC\*\*2-.001279\*CPFFAC\*\*3 LIT09360  
 IF (TCZ .GE. 1302.) CPLZ=0.995 LIT09370  
 IF (TCZ .GE. 1302.) GO TO 1255 LIT09380  
 CPLZ=1.0037-.01063\*CPFFAZ+.00564\*CPFFAZ\*\*2-.001279\*CPFFAZ\*\*3 LIT09390  
 CPLI=XLI+CPLI+XPB\*CPPB LIT09400  
 CPLZ=XLI+CPLZ+XPB\*CPPZ LIT09410  
 CPLI=CPLI\*AMLI LIT09420  
 CPLZ=CPLZ\*AMLI LIT09430  
 IF (J1 .NE. 1.) GO TO 851 LIT09440  
 RHLI=33.49-.0035\*(TLI -460.) / LIT09450  
 AKLI=(10.48+2.767E-03\*(TLI -817. )-0.322E-06\*(TLI -817. )\*\*2)/1488. LIT09460  
 AKLZ=(10.48+2.767E-03\*(TCZ -817. )-0.322E-06\*(TCZ -817. )\*\*2)/1488. LIT09470  
 IF (IOPFB .EQ. 1) GO TO 851 LIT09480  
 CPLI=((LIT-LIBP)\*CPLI+LIL0X\*CPLIO+LILNI\*CPLIN)/LILP LIT09490  
 IF (TCZ .LT. TMELT) DREAC=19500.+8.2\*(TCZ-672.)+CPLZ\*(TCZ-537.) LIT09500  
 \*R1 LIT09510  
 IF (TCZ .GE. TMELT) DREAC=19500.+8.2\*(TCZ-672.)+R1\*CPLI1\* LIT09520  
 \*(TMELT-537. )+R1\*QMELT+R1\*CPLZ\*(TCZ-TMELT) LIT09530  
 0369 LIT09540

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0370      HYD=U2**3.5*(TCZ-537.)          LIT09550
0371      LI0HHL=(2.-J2)*12.5*(TCZ-537.)    LIT09560
0372      LI0HGH=(9070.+20.74*(TCZ-1338.))*(2.-J2)   LIT09570
0373      OML=CPPZ*(TCZ-537.)*R2           LIT09580
0374      OMG=R2*(CPBPB1*(PBMELT-537.))+QMELT*CPPZ*(TCZ-PBMELT)  LIT09590
0375      TEM1=AMIN1(PBMELT,1338.)        LIT09600
0376      TEM2=AMAX1(PBMELT,1338.)        LIT09610
0377      IF (TCZ .LT. TEM1) DPROD=HYD+LI0HHL+OML  LIT09620
0378      IF (TCZ .LT. TEM1) GO TO 1601  LIT09630
0379      IF (TCZ .GE. TEM2) DPROD=HYD+LI0HG+OMG  LIT09640
0380      IF (TCZ .GE. TEM2) GO TO 1601  LIT09650
0381      IF (PBMELT .GT. 1338.) DPROD=HYD+LI0HG+OML  LIT09660
0382      IF (PBMELT .LT. 1338.) DPROD=HYD+LI0HL+OMG  LIT09670
0383      QWA=DELH+DPROD-DREAC            LIT09680
0384      IF (IOPTB EQ. 1) GO TO 22       LIT09690
0385      HTCAP=CPMD0X+CPMNI+CPML10+CPA*MA+CPLIN*MLIN+CPLIH*MLIH+CPLH2*MH2  LIT09700
0386      HTCAPG=HTCAP/GAMMA             LIT09710
0387      CPA=0.175+6.68E-05*TG-1.05E-08*TG**2.          LIT09720
0388      C TWO MILLIMETERS ARE ASSUMED TO COVER THE POOL OPTICALLY  LIT09730
0389      C * * * * * THIS HEAT CAPACITY (CPA) IS GOOD FOR CARBON DIOXIDE ONLY  LIT09740
0390      C Z=(LILOX/RHOLIO+LIINI/RHOLIN)/ASLI           LIT09750
0391      C EMG=1.-EXP(-(MLIO/RHOLIO+MLIN/RHOLIN)+2.27E05*CH/V/RA) LIT09760
0392      C IF (EMG.LT.0.005) EMG=0.005          LIT09770
0393      C THE RIF'S FOR RADIATION FROM THE POOL USE TAUCZ INSTEAD OF (LIT09780
0394      C THE ALLOWS US TO MODEL POOL TO COMBUSTION ZONE COUPLING WITH LIT09790
0395      C FLEXIBILITY          LIT09800
0396      C RIFPW=1./((1.-EML1)/EML1+(1.-EMSTL)*ASLI/EMSTL/AN+1./((1.-EMG)  LIT09810
0397      C * (TCZ*(TAUCZ-1.)+1.)*FPW*EMG/(ASLI/AW+1./FPG/(ICZ*(TAUCZ-1.))+1.) LIT09820
0398      C RIFCW=1./((1.-EMCZ)/EMCZ+(1.-EMSTL)*ASLI/EMSTL/AN+1./((1.-EMG)  LIT09830
0399      C +EMG/(1.+ASL1/AW))          LIT09840
0400      C RIFPG=EML1*EMG*(EMG-EMG*EML1+FML1/FPG/(ICZ*(TAUCZ-1.))+1.)  LIT09850
0401      C RIFCG=(EMCZ*EMG/((1.-EMCZ)*EMG+EMCZ))          LIT09860
0402      C IF (FLAGS) RIFPS=1./((1.-EMINS)/EMINS+(1.-EMSTL)/EMSTL*AINS/AW+  LIT09870
0403      C (AINS/AW+1.)/(1.+AINS/AW+(1.-EMG))          LIT09880
0404      C RIFSLC=(EMSTL*EMCNC)/(EMSTL+EMCNC-EMSTL*EMCNC)  LIT09890
0405      C RIFCZP=(EML1*EMCZ)/(EMCZ+EML1-EMCZ*EML1),      LIT09900
0406      C * * * * * CALCULATING GAS HEAT TRANSFER COEFFICIENTS * * * * * LIT09990
0407      C IF (ICZ .EQ. 1) T1=0.5*(TG+TCZ)          LIT10000
0408      C IF (ICZ .EQ. 0) T1=0.5*(TG+TL1)          LIT10010
0409      C T2 = 0.5*(TG + TS)          LIT10020
0410      C T3E=0.5*(TG+TEHC)          LIT10030
0411      C IF (FLAG) T4H=(TS+TA)/2.          LIT10040
0412      C IF (.NOT. FLAG) T4H=(TC(NL)+TA)/2.  LIT10050
0413      C B1 = 1.0/T1          LIT10060
0414      C B2 = 1.0/T2          LIT10070

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0-09          B3E=1.0/T3E          LIT10080
0410         B4H=1.0/T4H          LIT10090
0411         D1 = ((4.94E-05*T1 + 0.0188)/(RHDA*3600.0))**2    LIT10100
0412         D2 = ((4.94E-05*T2 + 0.0188)/(RHDA*3600.0))**2    LIT10110
0413         D3E=((4.94E-05*T3E+0.0188)/(RHDA*3600.0))**2    LIT10120
0414         D4H=((4.94E-05*T4H+0.0188)/(-.074*3600.0))**2    LIT10130
0415         AK1=(0.014+1.92E-05*(T1-460.))/3600.                LIT10140
0416         AK2=(0.014+1.92E-05*(T2-460.))/3600.                LIT10150
0417         AK3E=(0.014+1.92E-05*(T3E-460.))/3600.              LIT10160
0418         AKH=(0.014+1.92E-05*(T4H-460.))/3600.              LIT10170
0419         EX2 = (GIN*B2*ABS(TG - TS)/D2)*0.3333            LIT10180
0420         EX3E=(GIN*B3E*ABS(TG-TEHC)/D3E)*0.3333            LIT10190
0421         IF (FLAG) EX4H=(GIN*B4H*ABS(TA-TS)/D4H)*0.3333      LIT10200
0422         IF (.NOT. FLAG) EX4H=(GIN*B4H*ABS(TA-TC(NL))/D4H)*0.3333  LIT10210
0423         H=MINGS*AK2*EX2                                     LIT10220
0424         HEHC=HINENC*AK3E*EX3E                           LIT10230
0425         HA=HINSAM*AK4H*EXH                           LIT10240
C*****          CALCULATING SOME PRELIMINARY THERMAL DIFFUSIVITIES   LIT10250
          C BETWEEN NODES (MORE TO COME)                                LIT10260
          C2        B=L(1)/(KC0N*2.)+G/KGAP+W/(KSTL*2.)           LIT10270
          C3        BB=L(1)/(KC0N*2.)+G/KGAP+W/(KSTL*2.)           LIT10280
          C4        USUBA=KC0N*HA/(KC0N*1.HA*LNL)/2.             LIT10290
          C5        C=KSTL*HA/W/HICPG/(W*H/2.+KSTL)*( -1)       LIT10300
          C6        C11=KSTL*HA/(RHSTL*W*(KSTL+W*HA/2.))        LIT10310
          C7        IF (FLAG) GO TO 779                         LIT10320
          C8        C3=1./ (B*L(1)*RHCON*CPCON)                   LIT10330
          C9        DO 1010 I=1,NLM1                            LIT10340
          C10       C4(1)=2.*KC0N/(RHCON*CPCON*L(I)*(L(I)+L(I+1)))  LIT10350
          C11       C5=USUBA/(RHCON*CPCON*L(NL))               LIT10360
          C12       C7=1./ (BW*RHSTL*CPSTL)                  LIT10370
          C13       CE=KSTL*H/(RHSTL*W*(W*H/2.+KSTL))        LIT10380
          C14       IF (LOPTB .NE. 1) GO TO 878                 LIT10390
          C15       C1=0.                                         LIT10400
          C16       C6=0.                                         LIT10410
          C17       CB=1./ (0.5*W/KSTL/ASLI+0.43/KCON/ASLI) /    LIT10420
          C18       *(CPSTL*RHSTL*ASLI*W)                      LIT10430
          C19       CS=1./ (0.5*W/KSTL/ASLI+0.43/KCON/ASLI) /    LIT10440
          C20       *(CPCON*RHCON*ASLI+0.86)                   LIT10450
          C21       DO 1020 I=1,NLM1                            LIT10460
          C22       C10(I)=2.*KC0N/(RHCON*CPCON*L(I)*(L(I)+L(I+1)))  LIT10470
          C23       IF (LOPTB .EQ. 1) GO TO 524                 LIT10480
          C24       IF (FLAG) GO TO 780                         LIT10490
          C25       CB=1./ (BB*W*RHSTL*CPSTL)                  LIT10500
          C26       CS=1./ (BB*L(1)*RHCON*CICON)                LIT10510
          C27       IF (ICZ.EQ.1) EXX=(GIN*BI*ABS(TCZ-TG)/DD)     LIT10520
          C28       IF (ICZ.EQ.0) EXX=(GIN*BI*ABS(TL1-TG)/DD)     LIT10530
          C29       IF (EXX .LE. 0.0) GO TO 300                 LIT10540
          C30       EX1 = (EXX)**0.3333                         LIT10550
          C31       CEHCGS=HEHC*AEHC/HTCAPG                    LIT10560
          C32       CGSEHC=HEHC*AEHC/XMERC/CPEHC                  LIT10570
          C33       *****          CALCULATING GAS CONVECTION COEFFICIENT  LIT10580
          C34       DIFF=241.57/(132.0+T1/1.8)*(T1/493.2)**2.5/3600.  LIT10590
          C35       *****          *****          *****          *****          LIT10600

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0456      HFINF=HIN* DIFF*X1          LIT10610
0457      HBINF=HIN*A.M.!.*EX1       LIT10620
0458      IF (TAU .LT. DELT) TAU=DELT   LIT10630
0459      HF=HF+(HFINF-HF)*DELT/TAU   LIT10640
0460      HBHB=(HBINF-HB)*DELT/TAU   LIT10650
0461      IF(TIME.GT.ESCTIN) XESC=1.   LIT10660
0462      IF(TIME.GT.SFLTN) XSFL=1.    LIT10670
C      **** TEST FOR COMBUSTION *****
0463      ICNI=0                      LIT10680
0464      TEZ=(TCZ+TLI)/2.            LIT10690
0465      IF (TEZ.LE.2340. .AND. FOX.LE.0.28 .AND. MN1.GT.0.0) ICN1=1
0466      IF (.NOT.(ILIT.EQ.0 .OR. (ICMB.EQ.0 .AND. ICN1.EQ.0) .OR. TLJ.LT.
0467      *     TIMELT)) GO TO 525        LIT10730
0468      IF(ICZ.EQ.1) WRITE(6,529) ICZ,ICNI,ILIT,ICMB,TCZ,FOX,TLI   LIT10740
0469      FORMAT(' COMBUSTION HAS JUST STOPPED. PARAMETERS ARE: ICZ= ',I1,
0470      *     ' ICNI= ',I1, ' ILIT= ',I1, ' ICMB= ',I1, ' TCZ= ',F8.2, ' FOX= ',
0471      *     ' F7.3, TLJ= ',F8.2)        LIT10750
0472      IF(ICZ.EQ.1) IPAGE=IPAGE+2   LIT10760
0473      GO TO 522                    LIT10790
0474      CONTINUE                   LIT10800
0475      IF (10PTB .EQ. 1) GO TO 23   LIT10810
C      **** COMPUTATIONS USING COMBUSTION ZONE MODEL *****
C      **** COMPUTING RATE OF LITHIUM COMBUSTION *****
525      RN2=0                      LIT10830
C      **** COMPUTATION OF LITHIUM VAPOR DIFFUSION *****
0476      ICZ=1                      LIT10840
0477      IF(TEZ.LT.1900. .AND. FOX.LE.0.28) RN2=   LIT10850
0478      *(1.0-FOX/0.28)/(1.0-(TEZ-1900.)/440.)**2)   RN2=
0479      IF(TEZ.GE.1900. .AND. TEZ.LE.2340. .AND. FOX.LE.0.28) RN2=   LIT10860
0480      *(1.0-FOX/0.28)*(1.0-(TEZ-1900.)/440.)**2)   RN2=
0481      CMBRN=HF*FOX*RHOA+RCMBD           LIT10870
0482      CMBRW=HF*FWA*RHOA+RCMBW           LIT10880
0483      CMBR = CMBRO + CMBRN + CMBRW   LIT10890
0484      RN1LB=CMBRN*ASL/RCMBN           LIT11000
0485      ROXLB=CMBRO*ASL/RCMBO           LIT11010
0486      RWALB=CMBRW*ASL/RCMBW           LIT11020
C      **** COMPUTATION OF LITHIUM VAPOR DIFFUSION *****
0487      PLIV=(10.**(4.3831-14180.2/TLI))*14.7   LIT11030
0488      RHOLIV=PLIV*144./RINTTLI           LIT11040
0489      DIFFLI=3.56E-01*(TLI/460.)*1.81)/PA      LIT11050
0490      DFILM=DIFLLI*RHOLIV/CMBR           LIT11060
0491      EFLIM=DFILM*12.                  LIT11070
C      THE FILM THERMAL CONDUCTIVITY IS A WEIGHTED AVERAGE OF NITROGLIT11100
C      LITHIUM VAPOR, WITH PLIV AS THE WEIGHTING FACTOR      LIT11110
0492      TFEFF=(TCZ+TLI)/2.+459.67        LIT11120
0493      KNIT=.0432+TFEFF*(.0078-TFEFF*(8.2E-04+TFEFF*2.08E-04))  LIT11130

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0491      KLIT=0.55+T*EFF*(-4.99E-04+T*EFF*+1.206E-07)          LIT11140
0492      KFLIM=(PLI*(KLIT-KUNIT)+PA*KNITY)/14.7/3600.           LIT11150
0493      C***** COMPUTATION OF HEAT TRANSFER COEFFICIENTS      ****
0494      C THIS HEAT CAPACITY IS SHEER GUESSTWORK THE 0.1 1S FOR VERY LOW COMBILIT11160
0494      CPMCZ=AKLI*(1.+RCMBN)/RCMBN+CMBR0+CPLIO+(1.+RCMBN)/RCMBN+CMBRN*   LIT11170
0494      CPLIN+(1.+RCMBW)/RCMBN-(1./RCMBH2)-CMBRN+CWA+(1.+RCMBH2)/   LIT11180
0494      .RCMBH2-CMBRN*CPLI+RN2-HF*FN1+RHOA+CPLI)*300.+1.0          LIT11190
0495      IF (CPMCZ*AKLI.LE. 0.001) CPMCZ=0.001*AKLI               LIT11200
0496      CGCZ=HB*AKLI/CPMCZ                                         LIT11210
0497      CGCG=HB*AKLI/HTCPAG                                         LIT11220
0498      CPZ= YAPCZ/CPMCZ                                           LIT11230
0499      CCZP= YAPCZ/(CPLI*LIL)                                     LIT11240
0500      CCZ=( CMBRN*QCO+CMBRN*QCN+CMBRW*QCW)*AKLI               LIT11250
0501      CLIST=2.*AKLI*KSTL*(LIL*CPLI*(ZLI*KSTL+U*AKLI))       LIT11260
0502      CSBLI=2.*AKLI*KSTL*(RHSTL*W*CPSTL*(ZLI*KSTL+U*AKLI))  LIT11270
0503      23     IF (LOPTB .EQ. 0) GO TO 852                           LIT11280
0504      CMRR=1.                                                 LIT11290
0505      YAPCZ=1.                                               LIT11300
0506      TG=[LIL]                                              LIT11310
0507      RVOL1=RVOL+MLICH+C15*(2.-J2)+MPB/RHPS                LIT11320
0508      RRAD=(3.*RVOL1/4./3.14)*(1./3.)                         LIT11330
0509      RAREA=12.57*RRAD**2                                      LIT11340
0510      LBR=(5.3*RRAD)/2.                                         LIT11350
0511      LBN=LBR/NA                                             LIT11360
0512      LB2=LBN/2.                                              LIT11370
0513      DO 32 J=1,NN                                            LIT11380
0514      RAD(J)=RRAD+J*1BN                                       LIT11390
0515      ARE(J)=12.57*RAD(J)**2                                    LIT11400
0516      VOL(J)=4.19*RAD(J)**3                                    LIT11410
0517      ATOCZ=ATOIZ*2.34                                       LIT11420
0518      ATICZ=ATOIZ/1.3                                         LIT11430
0519      QCCZ=1./((5.33/ATOIZ+1.)/(HC0*ATICZ))                 LIT11440
0520      ATO(1)=(VOL(1)-RVOL1)*RHLI/AMLI                      LIT11450
0521      MBL(1)=(VOL(1)-RVOL1)*RHLI/AMLI                      LIT11460
0522      DO 37 J=2,NN                                            LIT11470
0523      ATO(J)=(VOL(J)-VOL(J-1))*2.34                         LIT11480
0524      MBL(J)=(VOL(J)-VOL(J-1))*RHLI/AMLI                  LIT11490
0525      DO 24 J=1,NN                                            LIT11500
0526      ATI(J)=ATO(J)/1.3                                       LIT11510
0527      QC(J)=1./(5.33/ATO(J)+1./(HC0*ATI(J)))              LIT11520
0528      A-R1/(2.-R2)                                           LIT11530
0529      MLI=MLI1-*ML10H                                         LIT11540
0530      MCZ=MCZ+A*ML10H-MPB                                     LIT11550
0531      IF (1G.EQ. 1) CPLIOH=20.74                            LIT11560
0532      CPMCZ=MCZ1*CPLZ/MCZ+MPB*CPPZ/MCZ+A*ML10H*CPLIOH/MCZ  LIT11580
0533      CCZ=(NT*0.025)*((-1.)*QWA-18.*((TCZ-612.)*CPLI*(TCZ-T11)*R1)/MCZ  LIT11600
0534      IF (1CZ .EQ. 0) CCZ=0.                                 LIT11610
0535      CF(1)=1. / ((LB2/(AKLZ*RAREA)+LB2/(AKL1*RAREA))*MCZ*CPMCZ)  LIT11620
0536      CT(1)=1. / ((LB2/(AKLZ*RAREA)+LB2/(AKL1*RAREA))*MB(1)*CPLI)  LIT11630
0537      DO 33 J=2,NN                                          LIT11640
0538      CF(J)=(AKLI*ARE(J-1))/LBN/(CPLI*MB(J-1))            LIT11650
0539                                         LIT11660

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0539      33    CT(J)=AKL1*ARE(J-1)/LBN/(CPLI*MR(J))          LIT11670
0540      CLIS=1./((LB2/(AKL1*ARE(NN))+0.5*W/(KSTL*ARE(NN)))*CPLI*MB(NN))   LIT11680
0541      CSBLI=1./((LB2/(AKL1*ARE(NN))+0.5*W/(KSTL*ARE(NN)))           LIT11690
0542      *CPSTL*RHSTL*W*ARE(NN))                                         LIT11700
0543      TDIF=TCZ-TN(1)                                               LIT11710
0544      GGCZ=0.                                                       LIT11720
0545      CGCZ=0.                                                       LIT11730
0546      IF (TDIF .LT. 50.) GO TO 26
0547      IF (TLI .LT. TMELT) GO TO 26
0548      CGCZ=AREA*AKL1/(RRAD*2.)*(2.0+0.001*((2.65E-04*(TCZ-TL1)
     *RRAD**3)**0.25)/(CPMCZ*MCA))
0549      CCGZ=(AREA*AKL1/(RRAD*2.))*(2.0+0.001*((2.65E-04*(TCZ-TL1)
     *RRAD**3)**0.25)/(CPLI*MB(1)))
0550      IF (1CZ .GE. 1325.) GO TO 860
0551      IF (LG .NE. 0) GO TO 862
0552      LG =1
0553      Q=CPMCZ*MCAZ*Z26
0554      MLIO=MLIOH
0555      TIM=MLIOH*QLIOH/Q
0556      MLIT=TIM*NT*0.025
0557      MLIV=MLIT+MLIO
0558      862    IF (MLIOH .LE. MLIV) GO TO 863
0559      861    CGZ=0.
0560      GO TO 852
0561      863    CGZ=1.
0562      852    IF (LOPTB .EQ. 1) GO TO 2452
0563      QRAD=SIGMA*ASLI*(TCZ**4-TL1**4)*RIFCZP
0564      QRAD=SIGMA*ASLI*(TCZ**4-TS**4)*RIFCZM
0565      QRAD=SIGMA*ASLI*(TCZ**4-TG**4)*RIFCZG
0566      RCZP=QRADW/(W*AW*RHSTL*CPSTL)
0567      RCZG=QRADG/HTCAPG
0568      IF (.NOT. FLAG1) QRADB=SIGMA*ASLI*(TSB**4-TB(1)**4)*RIFSLC
0569      IF (FLAG1) QRADB=SIGMA*ASLI*(TSB**4-TA**4)*EMSTL
0570      IF (.NOT. FLAG) QRADC=SIGMA*AW*(TS**4-TC(1)**4)*RIFSLC
0571      IF (FLAG) QRADC=SIGMA*AW*(TS**4-TA**4)*EMSTL
0572      RADB=QRADB/(W*ASLI*RHSTL*CPSTL)
0573      RADC=QRADC/(W*AW*RHSTL*CPSTL)
0574      RADC=QRADC/(W*AW*RHSTL*CPSTL)
0575      IF (.NOT. FLAG1) RADCB=QRADB/(TL1(1)+ASLI*RHCON*CPCON)
0576      IF (.NOT. FLAG) RADCC=QRADC/(L((1)-AW+RHCON*CPCON))
0577      RADB=SIGMA*ASLI*(TL1**4-TS**4)*RIFPW
0578      RADC=SIGMA*ASLI*(TL1**4-TG**4)*RIFPG
0579      RLW=QRADY/(W*AW*RHSTL*CPSTL)
0580      RWL1=QRADY/CPLI/LIL
0581      RGL1=QRADZ/CPLI/LIL
0582      RLG=QRADZ/HTCAPG

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C\*\*\*\*\*  
C\* CALCULATING TEMPERATURE RATES OF CHANGE  
C\* (WITH COMBUSTION)  
C\*\*\*\*\*

IF(LOPTB .EQ. 0) GO TO 1100

0583

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0584      CONTINUE
0585      ZZ6=CCZ/CPMCZ-CF(1)*(TC2-TN(1))-CGCZ*(TC2-TN(1))
0586      -(QCCZ*(TCZ-1032.)*QCCZ*180.)/(CPMCZ*MCZ)
0587      ZZ(1)=CT(1)*(TC2-TN(1))+CCG*(TC2-TN(1))-CF(2)*(TN(1)-TN(2))
0588      -(QC(1)*(TN(1)-1032.)/Q(1)*180.)/(CPLI+MB(1))
0589      DO 34 J=2,NO
0590      ZZ(J)=CT(J)*(IN(J-1)-TN(J))-CF(J+1)*(TN(J)-TN(J+1))
0591      -(QC(J)*(TN(J)-1032.)/Q(1)*180.)/(CPLI+MB(J))
0592      ZZ(NN)=CT(NN)*(TN(NN-1)-TN(NN))-CLIST*(TN(NN)-TSB)
0593      +CLIST*450.-QC(NN)*(TN(NN)-1032.)-QC(NN)*180.)/(CPLI+MB(NN))
0594      IF (LOPTB .EQ. 1) GO TO 523
C      CALCULATE COMBUSTION ZONE TEMPERATURE RATE OF CHANGE (DEG. R / SEC.)
0595      1100 ZZ6=CCZ-(QRADP+QRADG)/CPMCZ-(J-1)*QVAP*CMBR*ASLI/CPMCZ
0596      -CPCCZ*(TCZ-TG)
0597      IF (TCZ .LT. TL1) ZZ6=(TL1-TCZ)/DELT
C      CALCULATE LITHIUM TEMPERATURE RATE OF CHANGE (DEG. R / SEC.)
0598      ZZ=CCZ*(TCZ-TL1)-RCZP-CLIST*(TL1-TSB)+(J-1)*QVAP*CMBR*ASLI/CCZP
0599      *YAPCZ*RWL-RGLI+J*CCZG*(TCZ-TL1)
0600      +47800./(CPMLI*ML1)*J
C      CALCULATE CELL GAS TEMPERATURE RATE OF CHANGE (DEG. R / SEC.)
0601      ZZ4=CC1*(TG-TS)+CCZG*(TCZ-TG)+RCZGX*BLWR*BLWR*(TBLOW-TG)
0602      * /HTCAPG-ESCR*XESC/HTCAPG+CEHCGS* (TEHC-TG)+RLLIG
C      CALCULATE WALL STEEL TEMPERATURE RATE OF CHANGE (DEG. R / SEC.)
0603      IF (.NOT. FLAG) ZZ5=CC6*(TG-TS)-C7*(TS-TC1)+RCZW*RADC+RLLW
0604      IF (FLAG) ZZ5=CC6*(TG-TS)-C11*(TS-TA)+RCZW*RADC+RLLW
0605      GO TO 523
C      **** COMPUTATIONS WITHOUT COMBUSTION ZONE MODEL ****
0606      522 CONTINUE
0607      ICZ=0
0608      CMBR=0.0
0609      RN2=0.0
0610      VALG=AKL*HB*ASLI/(AKL+HB*ZL1/2.)
0611      CLIG=YALIG/HTCAPG
0612      QRADW=SIGMA*ASLI*(TL1**4-TS**4)*RIFPW
0613      QRADG=SIGMA*ASLI*(TL1**4-TG**4)*RIFPG
0614      RLW=QRADW/(W*AN*RHSTL*CPSTL)
0615      RLLI=QRADW/CPLI/LIL
0616      RGLI=QRADG/CPLI/LIL
0617      RLIG=QRADG/HTCAPG
0618      IF (.NOT. FLAG1) QRADB=SIGMA*ASLI*(TSB**4-TA**4)*EMSTL
0619      IF (FLAG1) QRADB=SIGMA*ASLI*(TSB**4-TA**4)*EMSTL
0620      RLIG=SIGMA*ASLI*(TS**4-TA**4)*RIFSLC
0621      IF (.NOT. FLAG) QRADC=SIGMA*AW*(TS**4-TC1**4)*RIFSLC
0622      IF (FLAG) QRADC=SIGMA*AW*(TS**4-TA**4)*EMSTL
0623      RADB=QRADB/(W*ASLI*RHSTL*CPSTL)
0624

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0615      RADC=ORADC/(W*AW*RHSTL*CPSTL)          LIT12730
        IF(.NOT. FLAG1) RADCB=ORADB/(L1(1)*ASLI*RHCON*CPCON)    LIT12740
        IF(.NOT. FLAG) RADCC=ORADC/(L(1)*AW*RHCON*CPCON)     LIT12750
        CGLI=YALIG/(LIL*CPLI)                         LIT12760
        CLIST=2.*ASLI*AKLI*KSTL/(LIL *CPLI*(ZLI *KSTL+W*AKLI))   LIT12770
        CSBLI=2.*AKLI*KSTL/(RHSTL*W*CPSTL*(ZLI *KSTL+W*AKLI))  LIT12780
        C* *****
        C* CALCULATING TEMPERATURE RATES OF CHANGE *   LIT12790
        C* (WITHOUT COMBUSTION) *                         LIT12800
        C* *****
        C* CALCULATE LITHIUM TEMPERATURE RATE OF CHANGE (DEG. R / SEC.)  LIT12810
        864      Z21=CGLI*(TG-TLI)-CLIST*(TLI-TS9)-RWLI-RGLI   LIT12820
        C LET COMBUSTION ZONE FOLLOW POOL TEMPERATURE FOR POSSIBLE REIGNITION  LIT12830
        Z22=(TLI-TCZ)/DELT                           LIT12840
        C
        C CALCULATE CELL GAS TEMPERATURE RATE OF CHANGE (DEG. R / SEC.)  LIT12850
        Z24=C1*(IG-TS)+CLIG*(TLI-TG)+RLIG*XBLW*BLDR*CPAB*(TBLOW-TG)  LIT12860
        . /HTCAPG-ESCRVXSC/HTCAPG+CHNGS*(TEHC-TG)           LIT12870
        C
        C CALCULATE WALL STEEL TEMPERATURE RATE OF CHANGE (DEG. R / SEC.)  LIT12880
        IF(.NOT. FLAG) Z25=C6*(TG-TS)-C7*(TS-TC(1))+RLIW-RADC  LIT12890
        IF(FLAG) Z25=C6*(TG-TS)-C11*(TS-TA)+RLIW-RADC
        523 CONTINUE
        C
        C *****
        C* COMPUTATIONS VALID WITH EITHER MODEL *               LIT12900
        C* *****
        C
        C ZZE=CGSEHC* (TG-TEHC)
        CALCULATE FLOOR STEEL TEMPERATURE RATE OF CHANGE (DEG. R / SEC.)  LIT12910
        IF(.NOT. FLAG1) Z27=CSBLI*(TLI-TSB)-CB*(TSB-TB(1))-RADB-
        . XSFSLFLCR*12./ (W*ASLI*RHSTL*CPSTL)             LIT12920
        IF(FLAG) Z27=CSBLI*(TLI-TSB)-C11*(TSB-TA)-RADB-XSFL*SFLCR*12./
        . (W*ASLI*RHSTL*CPSTL)
        IF(FLAG) GO TO 777
        C
        C CALCULATE WALL CONCRETE TEMPERATURE CHANGE
        DICDT(1)=C3*(TS-TC(1))+C4(1)*(TC(2)-TC(1))+RADCC  LIT12930
        DICDT(NL)=C4(NLM1)*(TC(NLM1)-TC(NL))-C5*(TC(NL)-TA)  LIT12940
        DO 5 I=2,NLM1
        5 DICDT(I)=C4(I)*(TC(I+1)-TC(I))+C4(I-1)*(TC(I-1)-TC(I))
        777 CONTINUE
        C
        IF(FLAG) GO TO 778
        C
        C CALCULATE FLOOR CONCRETE TEMPERATURE CHANGE
        DBDT(1)=C9*(TSB-TB(1))+C10(1)*(TB(2)-TB(1))+RADCB  LIT12950
        DBDT(NL)=C10(NLM1)*(TB(NLM1)-TB(NL))
        DO 50 IB=2,NLM1
        50 DBDT(IB)=C10(IB)*(TB(IB+1)-TB(IB))+C10(IB-1)*(TB(IB-1)-TB(IB))
        778 CONTINUE
        0636

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 0642      IF (IOPTB .EQ. 1) GO TO 854      LIT13260  
 0643      IF (.NOT. FLAGS) GO TO 527      LIT13270  
 C      CALCULATIONS WITH SUSPENDED LITHIUM SPILL PAN      LIT13280  
 C      HPAN=0.714\*HB      LIT13290  
 0644      AHT=A\$LI+T2L1\*BREDFTH      LIT13300  
 0645      TET1=0.0025\*(TINS1-460.)-2.5      LIT13310  
 0646      KIN1=(-.70892+.36584\*TET1+.01565\*TET1\*\*3)/43200.      LIT13320  
 0647      TET2=0.0025\*(TINS2-460.)-2.5      LIT13330  
 0648      KIN2=(-.70892+.36584\*TET2+.01565\*TET2\*\*3)/43200.      LIT13340  
 0649      YPAGS=A\$INS/(THKIN2/2./KIN2+1./HPAN)      LIT13350  
 0650      C2=YPAGS/HTCAPG      LIT13360  
 0651      QRADS=SIGMA\*A\$INS\*(TINS2\*\*4-TSB\*\*4)\*RIFPAS      LIT13370  
 0652      QRADCG=SIGMA\*A\$INS\*(TINS2\*\*4-(IG\*.4)\*RIFPAG      LIT13380  
 0653      RPANS=QRADS/(RHSTL\*AMB\*W\*CPSTL)      LIT13390  
 0654      RSTPAN=QRADS/(RHINS\*A\$INS\*THKIN2\*CPINS)      LIT13400  
 0655      RGASP=QRADCG/(RHINS\*A\$INS\*THKIN2\*CPINS)      LIT13410  
 0656      RPAGA=QRADCG/HTCAPG      LIT13420  
 0657      C13=YPAGS/(RHINS\*A\$INS\*THKIN2\*CPINS)      LIT13430  
 0658      CPAN1=2.\*AHT/(RHPAN\*APAN\*THKPAN\*CPAN)/(TLI/AKLI+THKPAN/KPAN)      LIT13440  
 0659      CPNIN1=2./(RHPAN\*APAN\*THKPAN\*CPNIN)/(THKPAN/KPAN/APAN+THKIN1/      LIT13450  
 0660      \* KIN1/A\$INS)      LIT13460  
 0661      CIN1PN=2./(RHINS\*A\$INS\*THKIN1\*CPINS)/(THKPAN/KPAN/APAN+THKIN1/      LIT13470  
 0662      \* KIN1/A\$INS)      LIT13480  
 0663      CIN12=2./(RHINS\*CPINS\*THKIN1)/(THKIN1/KIN1+THKIN2/KIN2)      LIT13490  
 0664      CIN21=CIN12\*THKIN1/THKIN2      LIT13500  
 C      MODIFYING LITHIUM TEMPR. RATE OF CHANGE      LIT13510  
 0665      ZZ1=ZZ1+CLIST\*(TLI-TSB)-CLIPAN\*(TLI-TPAN)      LIT13520  
 C      MODIFYING CELL GAS TEMP. RATE OF CHANGE      LIT13530  
 0666      ZZ4=ZZ4+C2\*(TINS2-TG)+RPAGAS      LIT13540  
 C      MODIFYING FLOOR STEEL TEMP. RATE OF CHANGE      LIT13550  
 0667      ZZ7=ZZ7-CSBLI\*(TLI-TSB)+C6\*(TG-TSB)+RPANST      LIT13560  
 C      CALCULATE LITHIUM SPILL PAN TEMP. RATE OF CHANGE (DEG R/SEC)      LIT13570  
 0668      ZZZ=CPANL1\*(TLI-TPAN)+CPNIN1\*(TINS1-TPAN)      LIT13580  
 C      CALCULATE INSULATION TEMPERATURE RATE OF CHANGE      LIT13590  
 0669      ZZ8=CIN1PN\*(TPAN-TINS1)+CIN12\*(TINS2-TINS1)      LIT13600  
 0670      ZZ9=CIN21\*(TINS1-TINS2)+C13\*(TG-TINS2)-RSTPAN-RGASP      LIT13610  
 0671      527      CONTINUE      LIT13620  
 C      \*\*\*\*\*CALCULATING OVERPRESSURE\*\*\*\*\*      LIT13630  
 C      \*\*\*\*\*CALCULATING OVERPRESSURE\*\*\*\*\*      LIT13640  
 C      XMAIR=MOX/32.+MNI/38.+MA/7MOLA+MWA/18.      LIT13650  
 0672      PA=1545.\*XMAIR\*TG/144./V      LIT13660  
 0673      OVERP=PA-PAZERO      LIT13670  
 0674                LIT13760  
 C      \*\*\*\*\*CALCULATING OVERPRESSURE\*\*\*\*\*      LIT13770  
 C      \*\*\*\*\*CALCULATING OVERPRESSURE\*\*\*\*\*      LIT13780

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0675      IF (TIME.GT.BLIN) XBLOW=1.
0676      IF (TIME.GT.RLOUT) XBLOW=0.
C*****CALCU. TOTAL LEAKAGE*****
C*****AEROSOL REMOVAL COMPUTATION FOR EMG ....
C     BETA IS THE INVERSE STICKING COEFFICIENT (IN SECONDS) FOR AEROSOLS
C     IMPACTING THE WALL - STICK GIVES THE RATE AT WHICH AEROSOL IS REMOVED
C     THROUGH STICKING TO THE WALL
C     STICK=AW/(V*BETA*12.)
0677
C     IF (OVERP) 10,10,11
0678      10 LEAK=0.
0679      GO TO 12
0680      11 LEAK = K*OVERP**0.5
0681      12 CONTINUE
0682      FOUT=EXHSTR/MAIR*XBLOW+LEAK
0683      FMLEFT= EXP(-OUTINT)
0684      FMLEAK = 1. - FMLEFT
0685      DO INTEGRATIONS *
C*****DO INTEGRATIONS *
C*****INTGRLL(0.,CMBR*ASLI)
C*****LIBP=INTGRLL(0.,CMBR*ASLI)
0686      LIBD=INTGRLL(0.,(1.+RCMBO)/RCMB0+CMBRO*ASLI*(1.-FRA))
0687      LILDX=INTGRLL(0.,(1.+RCMBO)/RCMB0+CMBRO*ASLI*(1.-FRA))
0688      LILNL=INTGRLL(0.,(1.+RCMBN)/RCMBN+CMBRN*ASLI*(1.-FRA))
0689      OXLB=INTGRLL(OXLB, RDXLB)
0690      MDX=INTGRLL(MOXI, MOXI, MO2B*XBLWR*XBLWR-MOX*XBLWR-MOXINJ)
0691      MN1=INTGRLL(MN1I, MN1I, MN2B*XBLWR*XBLWR-MN1*XBLWR-MN1INJ)
0692      MA=INTGRLL(MAI, WAB*XBLWR*XBLWR-MA*FOUT)
0693      MWA=INTGRLL(MWA1, WWWB*XBLWR*XBLWR-MWA*XBLWR-MWA*FOUT-RWALB)
0694      MLI0=INTGRLL(MLI0I, -MLIO*XBLWR*XBLWR-MLI0*XBLWR-MLI0*FOUT+(1.+RCMBO)/RCMB0+CMBRO*ASLI*FRA-
*MLIO*STICK)
0695      MLIN=INTGRLL(MLINI, -MLIN*XBLWR+(1.+RCMBN)/RCMBN+CMBRN*ASLI*FRA-
*MLIN*STICK)
0696      MLIH=INTGRLL(O., -MLIH*XBLWR+((1.+RCMBW)/RCMBW-1./RCMBH2)*CMBRW
*ASLI*FRA-MLIH*STICK)
0697      MH2=INTGRLL(O.,-MH2*FOUT+(1.+RCMBH2)/RCMBH2*CMBRH2*ASLI)
0698      854      IF (CGZ .EQ .1.) GO TO 8541
0699      TC2=INTGRLL(TCZI,ZZ26)
0700      GO TO 8542
0701      TC2=1338.
0702      IOPTB =1
0703      8542     IF (IOPTB .EQ. 0) GO TO 855
0704      DO 35 J=1,NN
0705      35 TN(J)=INTGRLL(TLII,ZZ(J))
0706      TLII=N(4)
0707      TSB=INTGRLL(TZERO,ZZ7)
0708      GO TO 856
0709      855      TG=INTGRLL(TZERO,ZZ7)
0710      TLII=INTGRLL(TLII,ZZ1)
LIT13790
LIT13800
LIT13810
LIT13820
LIT13830
LIT13840
LIT13850
LIT13860
LIT13870
LIT13880
LIT13890
LIT13900
LIT13910
LIT13920
LIT13930
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LIT13960
LIT13970
LIT13980
LIT13990
LIT14000
LIT14010
LIT14020
LIT14030
LIT14040
LIT14050
LIT14060
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LIT14080
LIT14090
LIT14100
LIT14110
LIT14120
LIT14130
LIT14140
LIT14150
LIT14160
LIT14170
LIT14180
LIT14190
LIT14200
LIT14210
LIT14220
LIT14230
LIT14240
LIT14250
LIT14260
LIT14270
LIT14280
LIT14290
LIT14300
LIT14310

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      IF(TLI.GE.TVAP) GO TO 820
      TS=INTEGR(TSZERO,Z25)
      TPAN=INTEGR(TPANZO,Z22)
      TINS1=INTEGR(TINS1I,Z22)
      TINS2=INTEGR(TINS2I,Z22)
      TEHC=INTEGR(TEHCZO,Z22)
      DO 750 I=1,NL
      TC(I)=INTEGR(TCIC(I),DTCDT(I))
      750 CONTINUE
      OUTGR=INTEGR(LEAKO,LEAK)
      TSB=INTEGR(TSBI,Z27)

      856 DO 751 I=1,NL
      751 TB(I)=INTEGR(TBIC(I),DTBDT(I))
      752 CONTINUE
      C CALL DYNAMI(TIME,&200)
      C ****
      C* POST INTEGRATION SECTION
      C* CHECK OVER AND TLI FOR STOP CONDITION
      C* CHECK AND CORRECT FOR LITHIUM AND OXYGEN SUPPLY
      C ****
      C
      IF (10PB .EQ. 0) GO TO 180
      MPB=R2*NT*0.025*TIME
      MLIOH=(2.-J2)*N1*0.025*TIME
      IF (MLIOH/(2.-J2) .GE. 847.8) ICZ=0
      IF (MLIOH/(2.-J2) .GE. 847.8) WRITE(6,879) TIME,MLIOH
      879 FORMAT(1X,'WATER DEPLETED',1X,F6.2,1X,F6.2)
      IF (ICZ .EQ. 0 .AND. TCZ .LE. 1340.) GO TO 858
      GO TO 503
      180 CONTINUE
      LILP=LIT-LIBP+LILDX+LILNI
      IF(LILP.LE.0.) LILP=0.0
      ZLI=LILP/RHL1/ASLI
      ALPHA=AKLI/(RHL1*CPLI)
      IF((LILP .LT. 0.1*LIT) .AND. (ALPHA*DELT .GT. ZLI*ZLI .OR. LILP
      * LT. 1.0)) FLAG=TRUE.
      IF (.NOT. FLAG) LILP=LILP
      IF (TG.LT.500. .AND. OVERP.LT.1.) GO TO 745
      IF (TLI.LT. TWELT) GO TO 743
      IF (ICMB.EQ.C .OR. MDX.GE.0.0) GO TO 201
      OXLB=OXLS
      ICMB=0
      CMBRD=0.0
      RXLB=0.0
      201 CONTINUE
      IF (TLI.EQ.0. .OR. (LIT-LIBP).GE.0.01) GO TO 500
      OXLB=LIT/RCMBD
      LLIT=0
      LIT=LIBP
      CMBR=0.0
      CMBRD=0.0
      0754
      0753
      0752
      0751
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      0711

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0755      CMBRN=0.0          LIT14850
0756      CMBRW=0.0          LIT14860
0757      RDXLB=0.0          LIT14870
0758      RNILB=0.0          LIT14880
0759      RWALB=0.0          LIT14890
0760      CONTINUE           LIT14900
0761      IF(MNI.GE.0.0) GO TO 202 LIT14910
0762      MNI=0.0             LIT14920
0763      ICN1=0              LIT14930
0764      CMBRN=0.             LIT14940
0765      RNILB=0.0           LIT14950
0766      CONTINUE           LIT14960
0767      IF(MWA.GE.0.0) GO TO 502 LIT14970
0768      MWA=0.0             LIT14980
0769      CMBRW=0.0           LIT14990
0770      RNILB=0.0           LIT15000
0771      CONTINUE           LIT15010
0772      CMBRH=3600.* (CMBRO+CMBRN+CMBRW) LIT15020
0773      IF(CMBRH .GE.0.01 .OR. TIME.LE.10.) GO TO 503 LIT15030
0774      CMBRO=0.0           LIT15040
0775      CMBRN=0.0           LIT15050
0776      CMBRW=0.0           LIT15060
0777      CMBRH=0.0           LIT15070
0778      RDXLB=0.0           LIT15080
0779      RNILB=0.0           LIT15090
0780      RWALB=0.0           LIT15100
0781      503 CONTINUE           LIT15110
C***** C***** C***** C***** C***** C***** C***** C*****
C*   CONVERT TEMP. TO DEG. F   *
C***** C***** C***** C***** C***** C***** C***** C*****
C
0782      IF (FLAGS1) CALL SI LIT15120
0783      IF(FLAGS1) GO TO 8 LIT15130
0784      DO 6 I=1,2,0          LIT15140
0785      TBF(I)=TB(I)-460.    LIT15150
0786      6 TCF(I)=TC(I)-460.  LIT15160
0787      TCF=TG-460.          LIT15170
0788      DO 36 I=1,N,N        LIT15180
0789      TNF(I)=N(I)-460.    LIT15190
0790      TSBF=TSB-460.        LIT15200
0791      TSF=TS-460.          LIT15210
0792      TCZF=TCZ-460.        LIT15220
0793      TLIF=TLI-460.        LIT15230
0794      TPANF=TPAN-460.      LIT15240
0795      TEHC=TEHC-460.       LIT15250
0796      TINS1F=TINS1-460.     LIT15260
0797      TINS2F=TINS2-460.     LIT15270
0798      8 CONTINUE           LIT15280
C***** C***** C***** C***** C***** C***** C***** C*****
C*   TIME STEP CONTROL   *
C***** C***** C***** C***** C***** C***** C***** C*****

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0798 IF (IOPTB .EQ. 0) GO TO 865
0799 DT1=ABS(RELERR*TN(1)/TZ(1))
0800 DT3=ABS(RELERR*TN(4)/TZ(4))
0801 DT5=1.0E06
0802 IF (TCZ .EQ. 0) GO TO 915
0803 DT5=ABS(RELERR*TCZ/TZ6)
0804 BILGE=AMIN1(DT1,DT3,DT5)
0805 BIL=BILGE-(DELT)/DELT
0806 IF (ABS(BIL) .GT. 0.1) DELT=BILGE
0807 IF (TIME .LT. 8000.) DELOUT=50.
0808 IF (TIME .LT. 20.) DELOUT=1.
0809 IF (TIME .LE. 1.0) DELOUT=0.1
0810 IF (TIME .GE. 8000.) DELOUT=600.
0811 ALPHA1=RHLI*CPMLI/AKLI
0812 ALPHA2=RHSTL*CPSTL/AKSTL
0813 ALPHA3=RHC0N*CPCON/KCON
0814 T1=0.3*LBN*2/ALPHA1
0815 T2=100.
0816 T3=0.075*W*2/ALPHA2
0817 T4=0.3*0.43*2/ALPHA3
0818 BIT= AMIN1(T1,T2,T3,T4)
0819 IF (DELT .GT. BIT) DELT=BIT
0820 IF (DELT .LT. DMIN) DELT=DMIN
0821 IF (DELT.GT. DELOUT) DELT=DELOUT
0822 GO TO 866
0823 DT1=ABS(RELERR*T1/TZ1)
0824 DT2=ABS(RELERR*TG/TZ4)
0825 DT3=ABS(RELERR*TS/TZ5)
0826 IF (ILIT.EQ.0 .OR. ICZ.EQ.0) GO TO 735
0827 DT5=ABS(RELERR*TCZ/TZ6)
0828 Z299=(CMBRH-CMBRH1)/DELT
0829 IF (Z299.EQ.0.) GO TO 735
0830 DT4=ABS(RELERR*CMBRH/(CMBRH-CMBRH1)*DELT)
0831 CMBRH1=CMBRH
0832 IF (IPASS.EQ.1) DT4=1.E06
0833 GO TO 736
0834 735 CONTINUE
0835 DT4=.0E06
0836 DT5=1.0E06
0837 736 CONTINUE
0838 BILGE=AMIN1(DT1,DT2,DT3,DT4,DT5)
0839 BIL=BILGE-(DELT)/DELT
0840 C THIS CONDITION IS TO REMOVE INSTABILITY DUE TO STEEP NITROGEN REACTION
0841 IF (TCZ.GT.1900..AND.ABS(BIL).GT.0.1) DELT=DELT+(BILGE-DELT)/10.
0842 IF ((NOT (TCZ.GT.1900..AND.ABS(BIL).GT.0.1))DELT=BILGE
0843 IF (TIME .LT. 8000.) DELOUT=50.
0844 IF (TIME .LT. 20.) DELOUT=1.
0845 IF (TIME .GE. 8000.) DELOUT=0.1
0846 C*** TESTING CONDUCTION LIMIT ON TIME STEP ***
0847 ALPHA2=((THKPAN+ZLI)/(ZLI*AKL1+THKPAN/KPAN))/((RHLI*CPMLI+RHSTL*CPSTL/AKSTL)/(THKPAN+ZLI)*2/ALPHA2)
0848 PY0=0.075*((THKPAN+ZLI)*2/ALPHA2)
0849

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0848 PYU=0.075*(THKPAN+ZLI)*2/ALPHA2
0849 IF(DEL.T.GT.-PYU)DEL.T=PYU
0850 IF(DEL.T.LT.DTMIN) DELT=DTMIN
0851 IF(DEL.T.GT.DELT) DELT=DELOUT
0852 IF(TIME.LT.1.0) DELT=.10
0853 IF(TCZ.GT.1.900. .AND. DELT.GT.20.) DELT=.20.

C **** OUTPUT SECTION ***
C ****
C
866 IF (TIME .LT. TIME0) GO TO 810
0854 TIME=TIME+DELOUT
0855 IF (I PAGE .GE. .50) WRITE (6,803) (NAME(I),I=1,92)
0856 FORMAT ('1',3(20X,20A4./),//,.32A4)
0857 IF (I PAGE .GE. .50) I PAGE=0
0858 I PAGE=I PAGE+1
0859 IF (IOPTB .EQ. 0) GO TO 857
0860 WRITE (6,882) TIME,TCZF,TNF(1),TNF(2),TNF(4),TSBF,TBF(1),MLIH
0861 FORMAT (7(F9.2,1X),F9.4)
0862 GO TO 810
0863
0864 RCC2P=CC2P*(TCZ-TLI)
0865 WRITE (6,804) TIME,DELT,CMBRH,LILP ,OVERP,EMG,EMLI,LIBP,TSF,TGF,
0866 RN2,TCZF,TLIF,TANF,MOX,MNI
0867 FORMAT (2(F7.1,X,4(F6.2,1X),2(F7.2,1X),F8.2,1X,2(F7.2,1X),F6.2,
0868 3(1X,F8.2),2(1X,F9.2))
0869 CONTINUE
0870 IF (TIME .GT. TIMEF) GO TO 900
0871 C RETURN TO TOP OF DYNAMIC CYCLE
0872 GO TO 200
C **** ERROR POINTERS ***
C ****
C
0873 CONTINUE
0874 WRITE (6,744)
0875 744 FORMAT (' POOL TEMP. HAS DROPPED TO LITHIUMS MELTING TEMP.')
0876 GO TO 900
0877 745 CONTINUE
0878 WRITE (6,746)
0879 746 FORMAT (' CELL GAS TEMP. AND PRESS. HAVE RETURNED TO NORMAL')
0880 GO TO 900
0881 820 CONTINUE
0882 WRITE (6,821)
0883 821 FORMAT (' LITHIUM TEMP. ABOVE BOILING POINT')
0884 GO TO 900
0885 910 CONTINUE
0886 WRITE (6,725)
0887 725 FORMAT ('X, NO ROOT FOUND FOR SPRAY FIRE FOR TEMP.S LESS THAN '.
0888 '1 MILLION DEG. R')
0889 GO TO 900
0890 WRITE (6,859) TIME

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```
01107    059  FORMAT ('DREEDER IS COOLING. TIME IS ',F7.1)
        GO TO 900
01108    300  CONTINUE
        WRITE(6,710)
01109    710  FORMAT('EXX IS NEGATIVE--CANNOT TAKE ROOT')
        WRITE(6,711) TCZ,CMDR,ZZ6,ZZ5,RN2
01110    711  FORMAT('MESSED UP VARIABLES',SE10.3)
        900  CONTINUE
        WRITE(6,713)
01111    713  FORMAT('PROGRAM EXECUTION STOPPED BY PROGRAM')
        WRITE(6,101) DT1,DT2,DT3,DT4,DT5
01112    101  FORMAT('VALUES',SE10.3)
01113    C   RETURN TO BEGINNING OF PROGRAM AND READ IN NEW DATA. IF NO NEW DATA
        C   AVAILABLE, PROGRAM EXECUTION WILL AUTOMATICALLY BE STOPPED BY END
        C   PARAMETER IN THE FIRST STATEMENT.
01114    999  CONTINUE
        GO TO 998
01115    998  CALL EXIT
        END
01116    0901
        0902
```

C THESE 3 SUBROUTINES ARE DESIGNED TO BE USED IN A MAIN PROGRAM WHICH LIT16630  
 C SIMULATES A DYNAMIC SYSTEM EXPRESSED AS A SET OF ODE'S. THESE ODE'S SLIT16640  
 C MAY BE REEXPRESSED AS A SET OF INTEGRALS WHICH MUST BE INTEGRATED. LIT16650  
 C SIMULTANEOUSLY THROUGH THE DOMAIN OF INTEREST STARTING WITH THE APPROPRIATE LIT16660  
 C INITIAL CONDITIONS. FOR EXAMPLE, THE FUNCTION Y MAY BE FOUND FROM LIT16670  
 C SOLUTION OF DY/DT = RATE \* F(Y,T) AND Y=0 AT T=0. THIS MAY BE LIT16680  
 C REWRITTEN Y = INTGRL(Y0, RATE), THE OPEN INTEGRAL OF RATE OVER T SLIT16690  
 C AT Y0. A SET OF ODE'S MAY BE TREATED IN A SIMILAR MANNER. LIT16700  
 C THE MAIN PROGRAM SHOULD CONSIST OF TWO MAIN PARTS, THE INITIAL LIT16710  
 C SECTION AND THE DYNAMIC SECTION. THE DYNAMIC SECTION IS FURTHER DIVIDED LIT16720  
 C INTO INTEGRATION AND POST-INTEGRATION SECTIONS. LIT16730

C THE INITIAL SECTION SHOULD BE USED FOR INPUT, CALCULATION OF NECESSARY LIT16740  
 C CONSTANTS, AND FOR CALCULATING AND SETTING OF INITIAL CONDITIONS. LIT16750  
 C SHOULD CONTAIN THE REAL INTGRL, COMMON, AND CALL INIT STATEMENTS. LIT16760

C THE INTEGRATION SECTION SHOULD START WITH A NUMBERED CONTINUE LIT16770  
 C STATEMENT AND END WITH THE CALL DYNAMI STATEMENT. IT SHOULD CONTAIN LIT16780  
 C ALL CALCULATIONS OF PROGRAM VARIABLES AND NON-CONSTANT RATES. ALL LIT16790  
 C FUNCTION STATEMENTS SHOULD APPEAR IN A GROUP IMMEDIATELY PRECEDING LIT16800  
 C CALL DYNAMI STATEMENT. LIT16810

C THE INTEGRATION SECTION WILL BE LOOPED SEVERAL TIMES DURING EACH LIT16820  
 C INTEGRATION STEP (SIMPSON'S RULE USES 4 LOOPS PER STEP. RUNGE-KUTTA LIT16830  
 C 5 LOOPS PER STEP). DYNAMI CONTROLS THE INTEGRATION BY TELLING THE LIT16840  
 C INTGRL FUNCTION WHAT STEP IT SHOULD PERFORM NEXT. THE INTEGRATION LIT16850  
 C VARIABLE TIME IS ALSO CONTROLLED BY DYNAMI. IT MAY OR MAY NOT BE INITIALIZED LIT16860  
 C ED DURING EACH LOOP. TIME SHOULD BE INITIALIZED IN THE INITIAL SECTION LIT16870  
 C DYNAMI UTILIZES MULTIPLE RETURNS TO CONTROL PROGRAM FLOW. THE STATISTICS LIT16880  
 C NUMBER PASSED TO DYNAMI SHOULD BE THAT OF THE FIRST STATEMENT IN THE LIT16890  
 C INTEGRATION SECTION. THIS CAUSES THE PROPER INTEGRATION LOOPING. AT LIT16900  
 C END OF EACH INTEGRATION STEP A NORMAL RETURN IS EXECUTED AND CONTROLLED LIT16910  
 C RETURNS TO THE FIRST STATEMENT FOLLOWING CALL DYNAMI. THIS SHOULD BLIT16920  
 C THE FIRST STATEMENT OF THE POST-INTEGRATION SECTION. LIT16930

C BECAUSE VARIABLE VALUES MAY DIFFER FROM THEIR TRUE VALUE DURING LIT16940  
 C INTEGRATION LOOPING, ALL PROGRAM LOGIC AND VARIABLE TIME STEP CALCULUS LIT16950  
 C EXECUTED ONCE AT THE END OF EACH INTEGRATION STEP. TIME AND ALL VARIABLES LIT16960  
 C CONTAINED WITHIN THE INTEGRATION SECTION WILL BE UPDATED TO THEIR LIT16970  
 C VALUES BEFORE CONTROL IS TRANSFERRED TO THE POST-INTEGRATION SECTION LIT16980  
 C THIS SECTION SHOULD CONTAIN AT LEAST ONE IF STATEMENT WHICH STOPS PLIT16990  
 C EXECUTION. AND THE LAST STATEMENT SHOULD BE A GO TO ST NO. WHERE SLIT17000  
 C IS THE STATEMENT NUMBER OF THE FIRST STATEMENT IN THE INTEGRATION SLIT17010  
 C APPROXIMATELY 100 INTEGRATIONS MAY BE PERFORMED SIMULTANEOUSLY. LIT17020

C VARIABLE LIST

C A MATRIX WHICH STORES THE INTERMEDIATE VALUES CALCULATED DURING ELLIT17030  
 C DELT INTEGRATION TIME STEP. TIME AND ALL VARIABLES LIT17040  
 C DXDT RATE BEING INTEGRATED. CALCULATED USING INTEGRAL VALUE AS LIT17050  
 C RETURNED BY INTGRL DURING THE PREVIOUS LOOP AND TIME SET BLIT17060  
 C DYNAMI. USED BY INTGRL AS CALLED FOR BY ICOUNT. LIT17100  
 C ICOUNT TELLS INTGRL WHICH INTEGRATION LOOP IS PRESENTLY BEING DIVIDED LIT17110  
 C IMETH = 1 USE RUNGE-KUTTA METHOD LIT17120  
 C \* 3 USE SIMPSON'S RULE LIT17130  
 C INOIN TELL DYNAMI HOW MANY INTGRL STATEMENTS THERE ARE IN THE MAIN LIT17140  
 C PROGRAM. LIT17150

```

C IPASS TELLS INGR TO DO TWO SPECIAL FUNCTIONS DURING THE FIRST LIT17160
C EXECUTIONS OF THE INTEGRATION SECTION. LIT17170
C ISTORE TELLS INGR WHERE TO STORE THE RESULT OF ITS INTERMEDIA LIT17180
C CALCULATION IN MATRIX A. LIT17190
C XIC MATRIX WHICH STORE INITIAL CONDITIONS AND THEN IS UPDATED LIT17200
C PRESENT INTEGRAL VALUE AT THE END OF EACH INTEGRATION STEP. LIT17210
C XINC INITIAL CONDITION LIT17220
C
C SUBROUTINE DYNAMIC(TIME,*) LIT17230
COMMON IMETH,ICOUNT,ISTORE,INDIN,IPASS,DELT,XIC(101),A(501) LIT17240
      IF(IPASS.EQ.0) GO TO 40 LIT17250
      IF(IMETH.EQ.1) GO TO 10 LIT17260
LIT17270
C SIMPSON'S RULE (DEFAULT) IMETH>2 LIT17280
C
C IF((ICOUNT.EQ.4) GO TO 4 LIT17290
IF((ICOUNT.EQ.3) GO TO 3 LIT17300
TIME=TIME+DELT/2. LIT17310
ICOUNT=ICOUNT+1 LIT17320
RETURN 1 LIT17330
LIT17340
LIT17350
LIT17360
LIT17370
LIT17380
LIT17390
LIT17400
LIT17410
LIT17420
LIT17430
LIT17440
LIT17450
LIT17460
LIT17470
LIT17480
LIT17490
LIT17500
LIT17510
LIT17520
LIT17530
LIT17540
LIT17550
LIT17560
LIT17570
LIT17580
LIT17590
LIT17600
LIT17610
LIT17620
LIT17630
LIT17640
C
C RUNGE-KUTTA METHOD - FIXED STEP - IMETH=1
C
C 10 CONTINUE
      IF((ICOUNT.EQ.5) GO TO 4 LIT1740
      IF((ICOUNT.EQ.4) GO TO 14 LIT1741
      IF((ICOUNT.EQ.2) GO TO 12 LIT1742
TIME=TIME+DELT/2. LIT1743
ICOUNT=ICOUNT+1 LIT1744
RETURN 1 LIT1745
LIT1746
LIT1747
LIT1748
LIT1749
LIT1750
LIT1751
LIT1752
LIT1753
LIT1754
LIT1755
LIT1756
LIT1757
LIT1758
LIT1759
LIT1760
LIT1761
LIT1762
LIT1763
LIT1764
C
C 0019
C 0020
C 0021
C 0022
C 0023
C 0024
C 0025
C 0026
C 0027
C 0028
C 0029
C 0030
C 0031
C 0032
C 0033
C 0034
C 0035
      10 CONTINUE
      IF((ICOUNT.EQ.5) GO TO 4
      IF((ICOUNT.EQ.4) GO TO 14
      IF((ICOUNT.EQ.2) GO TO 12
TIME=TIME+DELT/2.
ICOUNT=ICOUNT+1
RETURN 1
LIT1740
LIT1741
LIT1742
LIT1743
LIT1744
LIT1745
LIT1746
LIT1747
LIT1748
LIT1749
LIT1750
LIT1751
LIT1752
LIT1753
LIT1754
LIT1755
LIT1756
LIT1757
LIT1758
LIT1759
LIT1760
LIT1761
LIT1762
LIT1763
LIT1764
      12 CONTINUE
      ICOUNT=3
      RETURN 1
      14 CONTINUE
      ICOUNT=5
      RETURN 1
      40 CONTINUE
      IPASS=1
      RETURN
END

```

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MAIN

C THIS SUBROUTINE INITIALIZES VARIABLES USED BY THE INTEGRATION ROUTINE LIT117650  
C IT SHOULD BE PLACED IN THE INITIALIZATION SECTION OF THE MAIN PROGRAM LIT117660  
C BEFORE THE FIRST STATEMENT OF THE DYNAMIC SECTION. SEE DYNAMI FOR LIT117670  
C LIST AND INTEGRATION DESCRIPTION.

C

SUBROUTINE INIT  
COMMON IMETH,ICOUNT,ISTORE,INOUT,IPASS,DELT,XIC(101),A(501)

0001 LIT117700  
0002 IPASS=0 LIT117710  
0003 ISTORE=0 LIT117720  
0004 ICOUNT=1 LIT117730  
0005 INOUT=0 LIT117740  
0006 RETURN LIT117750  
0007 END LIT117760  
0008 LIT117770

```

C FUNCTION INTGRL PERFORMS THE ACTUAL INTEGRATIONS. IN THE MAIN
C PROGRAM, ALL INTGRL STATEMENTS SHOULD BE PLACED IN A GROUP AT THE ELIT11790
C OF THE INTEGRATION SECTION. ALL RATE CALCULATIONS SHOULD PRECEDE LIT11780
C GROUP AND IT SHOULD BE IMMEDIATELY FOLLOWED BY THE CALL DYNAMI STATLIT117810
C FOR VARIABLE LIST AND DESCRIPTIONS SEE DYNAMI.
C
0001    REAL FUNCTION INTGRL(XXIC,DXDT)
0002      COMMON IMETH,ICOUNT,ISTORE,INODIN,IPASS,DELT,XIC(101),A(501)
0003      IF(IPASS.EQ.0) GO TO 40
0004      ISTORE=ISTORE+1
0005      IF(IMETH.EQ.1) GO TO 10
C
C SIMPSON'S RULE (DEFAULT) IMETH GREATER THAN 2
C
0006      IF(ICOUNT.EQ.4) GO TO 4
0007      IF(ICOUNT.EQ.3) GO TO 3
0008      IF(ICOUNT.EQ.2) GO TO 2
0009      1 CONTINUE
0010      INODIN=INODIN+
0011      IF(IPASS.EQ.1) XIC(INODIN)=XXIC
0012      A(ISTORE)=DXDT
0013      INTGRL=XIC(INODIN)+DELT*DXDT/2.
0014      A(500-ISTORE)=INTGRL
0015      RETURN
0016      2 CONTINUE
0017      A(ISTORE)=DXDT
0018      INTGRL=A(500+INODIN-ISTORE)+DELT*DXDT/2.
0019      RETURN
0020      3 CONTINUE
0021      INTGRL=XIC(ISTORE-2*INODIN)+DELT/6.* (A(ISTORE-2*INODIN)+4.* *
0022          A(ISTORE-2*INODIN)+DXDT)
0023      XIC(ISTORE-2*INODIN)=INTGRL
0024      RETURN
0025      4 CONTINUE
0026      INTGRL=XIC(ISTORE-3*INODIN)
0027      RETURN
C
C RUNGE-KUTTA METHOD - FIXED STEP- IMETH=1
C
0028      10 CONTINUE
0029      IF(ICOUNT.EQ.5) GO TO 15
0030      IF(ICOUNT.EQ.4) GO TO 14
0031      IF(ICOUNT.EQ.3) GO TO 13
0032      IF(ICOUNT.EQ.2) GO TO 12
0033      11 CONTINUE
0034      INODIN=INODIN+
0035      IF(IPASS.EQ.1) XIC(INODIN)=XXIC
0036      A(ISTORE)=DELT*DXDT
0037      INTGRL=XIC(INODIN)+5.*A(ISTORE)
0038      RETURN
0039      12 CONTINUE
0040      A(ISTORE)=DELT*DXDT
0041      INTGRL=XIC(ISTORE-INODIN)+.5*A(ISTORE)
0042      RETURN

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```
0042      13 CONTINUE  
          A(ISTORE)=DELT*DXT  
          INTGRL=XIC(ISTORE-2*INOIN)+A(ISTORE)  
          RETURN  
0044      14 CONTINUE  
          AA=DELT*DXT  
          INTGRL=XIC(ISTORE-3*INOIN)+1./6.*  
          *(A(ISTORE-3*INOIN)+1./6.*  
          *(A(ISTORE-2*INOIN)+2.*A(ISTORE-INOIN)+AA)  
          XIC(ISTORE-3*INOIN)=INTGRL  
          RETURN  
0046      15 CONTINUE  
          INTGRL=XIC(ISTORE-4*INOIN)  
          RETURN  
0048      40 CONTINUE  
          INTGRL=XXIC  
          RETURN  
0049      END  
0050  
0051  
0052  
0053  
0054  
0055  
0056  
0057
```

C SI IS USED AT THE OPTION OF THE USER TO CONVERT LITFIRE SO THALIT18480  
 C IT ACCEPTS INPUT AND DELIVERS OUTPUT IN SI UNITS LIT18490  
 C USE JOULES / METERS / SEC / PASCALS LIT18500  
 C SPECIAL CARE SHOULD BE TAKEN THAT ALL OUTPUT VARIABLES WHICH LIT18510  
 C MUST BE CONVERTED APPEAR IN THIS SUBROUTINE LIT18520  
 C  
 SUBROUTINE SI LIT18530  
 IMPLICIT REAL (K,L,M) LIT18540  
 DIMENSION TBF(20),TC(20),TB(20),TC(20) LIT18550  
 LOGICAL FLAGJ,FLAGS,FLAGIO LIT18560  
 COMMON /UNITS/ AW,CH,G,KGAP,W,K,HA,HIN,THW,THF,TMELT,TVAP,TCZI, LIT18570  
 K PAN,RHPAN,CPPAN,TPANZO,APAN,AWB,THKPN,BREDFL,ASLI,SPILL,QCO1, LIT18580  
 QCO2,QCN,QCW,QFL,QAP,FLAGI,FLAGJ,FLAGK,FLAGL,FLAGM,FLAGN,ISFLC,RHOLIN, LIT18590  
 RHOLI,RHOLIH,CRA,CPLI,RHLI,AKL,I,CSTL,RHSTL,KSTL,CPCON,RHCON, LIT18600  
 KCON,PAZERO,TZERO,TBSI,TA,TLII,DP,DP2,DP3,ESCR,SFLCR, LIT18610  
 CPAB,TBLOW,EXHSIV,BLOW,TB,TC,THKIN2,THKIN1,AINS,CPINS,RHINS, LIT18620  
 COMMON /EKTRA/ TINS1F,TINS2F,TB,TC,TG,IGF,TSB,TSF,TCZ, LIT18630  
 TCZF,TLII,TLIF,TFAN,TPANF,OVERP,MOX,MNI,LILP,HF,HB,EPFLM,DFLM, LIT18640  
 TINS1,TINS2 LIT18650  
 IF(FLAGIO) GO TO 1 LIT18660  
 0007 C  
 INPUT VARIABLES - CHANGE TO ENGLISH LIT18670  
 RHOLI0=RHOLI0\*0.06243 LIT18680  
 RHOLIN=RHOLIN\*0.06243 LIT18690  
 RHOLIH=RHOLIH\*0.06243 LIT18700  
 CPA=CPA\*0.000239 LIT18710  
 CPLI=CPLI\*0.000239 LIT18720  
 RHII=RHLI\*0.06243 LIT18730  
 AKLI=AKL\*0.5778 LIT18740  
 CPSTL=CPSTL\*0.000239 LIT18750  
 RHSTL=RHSTL\*0.06243 LIT18760  
 KSTL=KSTL\*0.5778 LIT18770  
 CPCON=CPCON\*0.000239 LIT18780  
 RHCON=RHCON\*0.06243 LIT18790  
 KCON=KCON\*0.5778 LIT18800  
 PAZERO=PAZERO\*101325. LIT18810  
 TGZERO=TGZERO\*1.8 LIT18820  
 TSZERO=TSZERO\*1.8 LIT18830  
 TSBI=TSBI\*1.8 LIT18840  
 TA=TA\*1.8 LIT18850  
 TLII=TLII\*.8 LIT18860  
 HA=HA\*0.176 LIT18870  
 IF(.NOT. FLAGJ) GO TO 2 LIT18880  
 0021 DP1=DP1\*101325. LIT18890  
 0022 DP2=DP2\*101325. LIT18900  
 0023 DP3=DP3\*101325. LIT18910  
 0024 IF((ESCR.EQ. 1) ESCR=ESCR\*9.478E-04 LIT18920  
 0025 IF((SFLCR.EQ. 1) SFLCR=SFLCR\*9.478E-04 LIT18930  
 0026 IF((IBLOW.EQ. 0) GO TO 3 LIT18940  
 0027 CPAB=CPAB\*0.000239 LIT18950  
 0028 TBLOW=TBLOW\*.8 LIT18960  
 0029 EXSTV=EXHSIV\*35.315 LIT18970  
 0030 TBLOW=BLOWV\*2119. LIT18980  
 0031 2 LIT18990  
 0032 0033 0034 0035 0036 0037 0038 LIT19000

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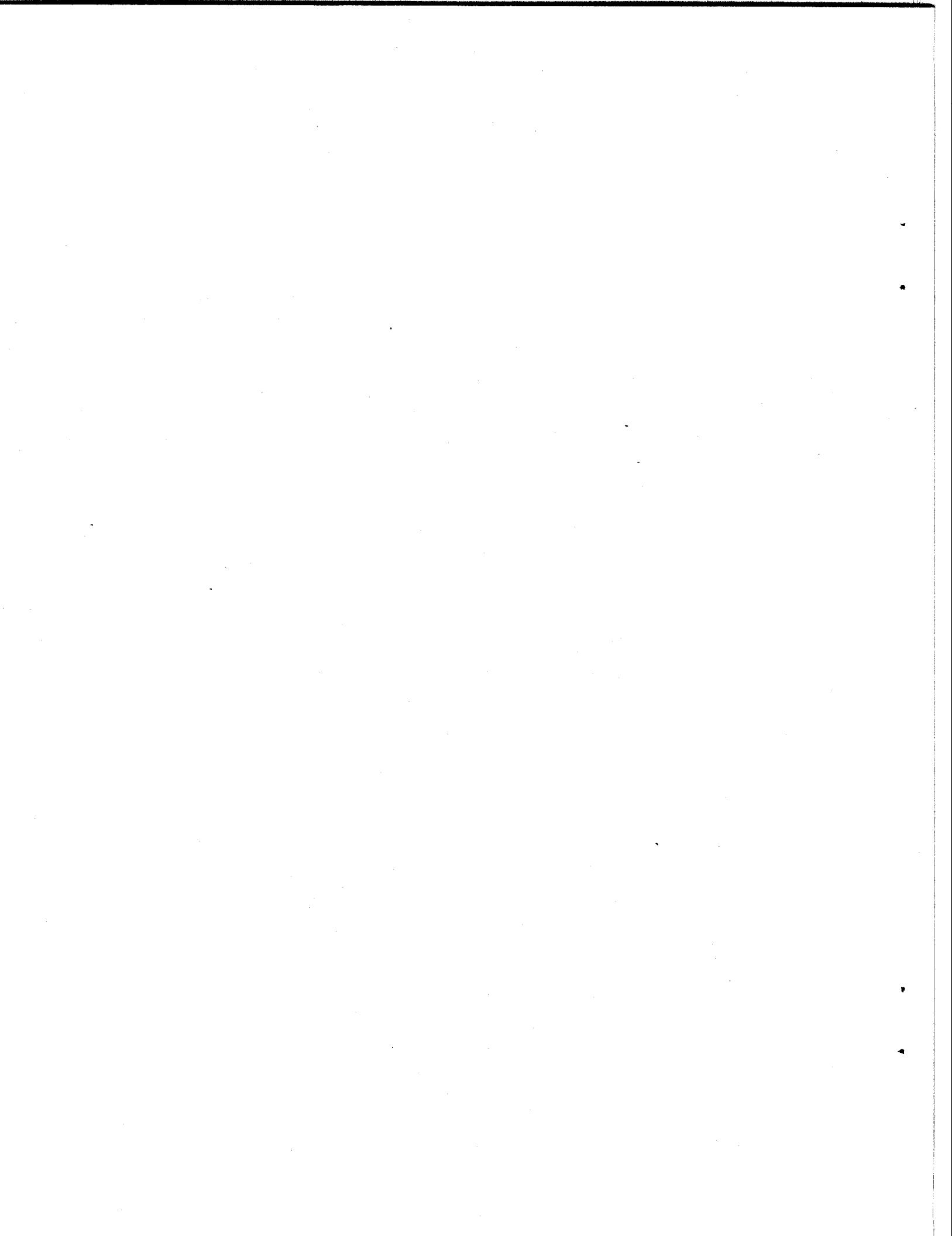
0039      3      AW=AW*10.764          LIT19010
          V=V*35.315          LIT19020
          CH=CH*3.281          LIT19030
          G=G*39.37           LIT19040
          KGAP=KGAP*0.5778     LIT19050
          W=W*39.37           LIT19060
          THW=THW*39.37         LIT19070
          THF=THF*39.37         LIT19080
          TMELT=TMELT*1.8       LIT19090
          TVAP=TVAP*1.8         LIT19100
          TCZI=TCZI*1.8         LIT19110
          ASLI=ASLI*10.764      LIT19120
          SPILL=SPILL*2.205     LIT19130
          QCO1=QCO1*0.00043     LIT19140
          QCO2=QCO2*0.00043     LIT19150
          QCN=QCN*0.00043      LIT19160
          QCW=QCW*0.00043      LIT19170
          QVAP=QVAP*0.00043     LIT19180
          IF (.NOT. FLAGS) GO TO 4   LIT19190
          KPAN=KPAN*0.5778       LIT19200
          RH PAN=RH PAN*0.06243   LIT19210
          CPPAN=CPPAN*0.000239    LIT19220
          TPANZO=TPANZO*1.8       LIT19230
          APAN=APAN*10.764        LIT19240
          AWB=AWB*10.764          LIT19250
          THKPAN=THKPAN*39.37     LIT19260
          BREDTB=BREDTB*3.281     LIT19270
          THKIN1=THKIN1*39.37     LIT19280
          THKIN2=THKIN2*39.37     LIT19290
          AINS=AINS*10.764         LIT19300
          RHINS=RHINS*0.06243     LIT19310
          CPINS=CPINS*0.000239     LIT19320
          CONTINUE                 LIT19330
          RETURN                   LIT19340
          4
          1      CONTINUE
          0072
          0073      C      OUTPUT VARIABLES - CHANGE TO SI
          C      FLAG1=.TRUE.
          DO 7  I=1,20
          0075      7      TB(I)=TB(I)*0.5556-273.
          0076      7      TC(I)=TC(I)*0.5556-273.
          0077      7      TG=G*0.5556-273.
          0078      7      TSBF=TSB*0.5556-273.
          0079      7      TSFTS*0.5556-273.
          0080      7      TCZF=TC*0.5556-273.
          0081      7      TLIF=TLI*0.5556-273.
          0082      7      IF (FLAGS) TPAN=TPAN*0.5556-273.
          0083      7      TINS1F=TINS1*0.5556-273.
          0084      7      TINS2F=TINS2*0.5556-273.
          0085      7      OVERP=OVERP+6893.
          0086      7      MX=MX*0.454
          0087      7      MN1=MN1*0.454
          0088      7      LLIP=LILP*0.454
          0089      7

```

FORTRAN IV G1 RELEASE 2.0 S1  
0090 HF=HF\*0.3048  
0091 HB=HB\*5.68  
0092 EFILM=DFILM\*0.3048  
0093 RETURN  
0094 END

PAGE 0003  
18/07/48  
DATE = 80270  
LIT19540  
LIT19550  
LIT19560  
LIT19570  
LIT19580

APPENDIX C  
Sample Input to LITFIRE



PHYSICAL PROPERTIES

CPLI =	0.0340	RHLI =	617.7000	RHOLIO =	1.00000	AKLI =	20.0200	EMLI =	1.0000
CPSTL =	0.1350	RHSTL =	487.0000	RHOLIN =	1.00000	KSTL =	25.0000	ENSTL =	1.0000
CPCON =	0.2300	RHCN =	156.0000	RHOLIH =	1.00000	KCON =	15.0000	ENCON =	1.0000
				KGAP =	1.0000	EMCZ =	1.0000		

INNER CONTAINMENT DIMENSIONS

AW =	1.0000	V =	1.0000	THW =	0.0	K =	1.0000		
CH =	1.0000	W =	0.1260	THF =	41.3000	G =	0.0		

SPILL PARAMETERS

ASLI =	354.0000	SPILL =	19700.0000	SPRAY =	0.0	WD2 =	0.0	ZLI =	0.0901
						FLAGU = F		FLAGSI = F	

HEAT TRANSFER CORRELATION COEFFICIENTS

HIN =	1.0000	HINEHC =	0.0	HINGS =	0.0	HINSAM =	0.0		
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COMBUSTION PARAMETERS

QCO =	0.0	RCMBO =	1.00000	TVAP =	4000.0000	RCMBH2 =	1.00000	PERCN =	0.0
QCN =	0.0	RCMBN =	1.00000	TMELT =	915.0000	FRA =	0.0		
QCW =	6960.0000	RCMBW =	0.3900	QVAP =	1718.0000	RA =	0.0		

INITIAL CONDITIONS

TGZERO =	1032.0000	TLII =	1212.0000	TSBI =	762.0000	WA =	1.0000	PAZERO =	1250.0000
TSZERO =	762.0000	TCZI =	1212.0000	TA =	700.0000	WWA =	1.0000		

INTEGRATION CONTROL PARAMETERS

IMETH =	3	DTMIN =	0.2900	TIMEF =	8000.0000	RELR =	0.0060	DELOUT =	2000.0000
---------	---	---------	--------	---------	-----------	--------	--------	----------	-----------

OPTIONS IN EFFECT

I BLOW = 0      IESC = 0      ISFLC = 0      FLAG = T      FLAG1 = F  
IOPTB = 1

MISCELLANEOUS INPUT

W02B = 0.0      BLOWV = 0.0      ESCR = 0.0      SFLCR = 0.0      CPA = 1.0000  
WNAB = 0.0      BLOUT = 0.0      ESCTIN = 0.0      SFLTIN = 0.0      CPAB = 1.0000  
WN2B = 0.0      BLIN = 0.0      EXHSTV = 0.0      TBLOW = 1.0000      XMOLA = 1.0000      XMOLAB = 1.0000

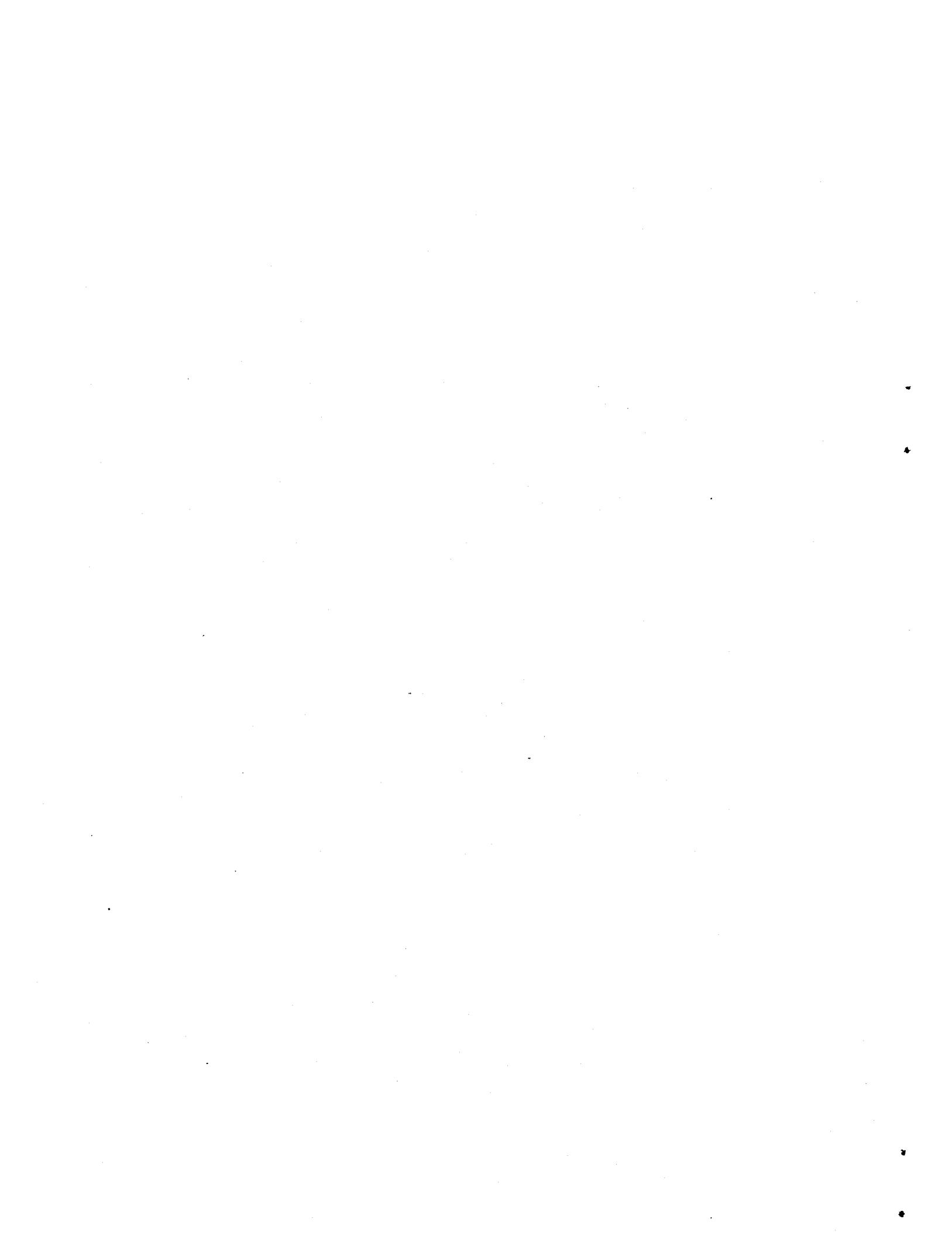
EXTANEOUS HEAT CAPACITY NODE DATA

TEHCZO = 1300.0000      XMEHC = 0.0      AEHC = 0.0      CPEHC = 1.0000

INTERNAL BLANKET OPTION INPUT

NT = 3.0      RRAD = 1.00      AMLI = 835.00      HCD = 5.56      QLIQH = 5029.00  
CPLIOH = 12.50      PBMLET = 1081.50      CP1 = 0.028      CP2 = 0.00001      RHPB = 710.00  
AMPB = 207.00      XLI = 0.008      CPPL = 0.033      CPLI1 = 34.35      CPPB1 = 6.93  
R1 = 1.000      R2 = 4.000      J1 = 0.0      J2 = 1.0      QMELT = 2011.70  
QMELTP = 2203.00      DELH = -71560.0

APPENDIX D  
Sample Output of LITFIRE



Time	T <sub>112</sub>	T <sub>111</sub>	T <sub>110</sub>	T <sub>109</sub>	T <sub>108</sub>	T <sub>107</sub>
0.0	752.00	752.00	752.00	752.00	752.00	302.00
0.10	757.33	752.00	752.00	752.00	752.00	304.63
0.20	762.65	752.00	752.00	752.00	752.00	307.24
0.30	767.93	752.01	752.00	752.00	752.00	309.83
0.40	773.19	752.01	752.00	752.00	752.00	312.41
0.50	776.43	752.01	752.00	752.00	752.00	314.96
0.60	783.65	752.02	752.00	752.00	752.00	317.50
0.70	786.83	752.03	752.00	752.00	752.00	320.02
0.80	794.00	752.04	752.00	752.00	752.00	322.52
0.90	799.14	752.05	752.00	752.00	752.00	325.00
1.00	804.26	752.06	752.00	752.00	752.00	327.47
1.10	809.35	752.08	752.00	752.00	752.00	329.91
2.10	859.03	752.32	752.00	752.00	752.00	353.44
3.10	900.54	753.62	751.82	752.00	752.00	373.95
4.10	938.93	754.08	751.83	752.00	752.00	394.43
5.10	975.22	754.64	751.84	752.00	752.00	413.48
6.10	1009.55	755.29	751.85	752.00	752.00	431.22
7.10	1042.04	756.02	751.86	752.00	752.00	447.72
8.10	1072.82	756.84	751.87	752.00	752.00	463.08
9.10	1102.05	757.73	751.89	752.00	752.00	477.38
10.10	1129.66	758.68	751.91	752.00	752.00	490.68
11.10	1155.91	759.68	751.92	752.00	752.00	503.06
12.10	1180.84	760.75	751.95	752.00	752.00	514.58
13.10	1204.51	761.85	751.97	752.00	752.00	525.30
14.10	1227.02	763.01	752.00	752.00	752.00	535.28
15.10	1248.41	764.20	752.02	752.00	752.00	544.57
16.10	1269.77	765.43	752.05	752.00	752.00	553.21
17.10	1288.15	766.69	752.09	752.00	752.00	561.26
18.10	1306.60	767.98	752.12	752.00	752.00	568.74
19.10	1324.18	769.29	752.16	752.00	752.00	575.71
20.10	1340.93	770.62	752.20	752.00	752.00	582.20
20.29	1672.56	833.53	756.51	752.00	752.00	667.63
22.0.29	1703.40	864.77	761.45	752.03	752.03	671.56
22.0.29	1693.60	876.10	764.68	752.09	752.09	673.23
22.0.29	1666.21	878.83	766.37	752.15	752.15	674.79
22.0.29	1635.69	878.09	767.14	752.21	752.21	676.28
32.0.18	1605.05	876.00	767.43	752.25	752.25	677.71
37.0.11	1575.35	873.41	767.48	752.27	752.27	679.09
42.0.24	1546.87	870.66	767.42	752.28	752.28	680.43
47.0.16	1519.94	867.91	767.30	752.29	752.29	681.72
52.0.29	1494.34	865.21	767.16	752.30	752.30	682.97
57.0.22	1470.23	862.60	767.00	752.30	752.30	684.18
62.0.14	1447.43	860.09	766.83	752.31	752.31	685.35
67.0.27	1425.78	857.66	766.67	752.32	752.32	686.49
72.0.20	1405.36	855.35	766.51	752.33	752.33	687.59
77.0.12	1365.02	853.13	766.35	752.33	752.33	688.66
82.0.25	1367.61	850.99	766.19	752.34	752.34	689.71
87.0.18	1350.22	848.95	766.04	752.35	752.35	690.72
92.0.10	1333.69	847.00	765.89	752.35	752.35	691.70
97.0.23	1317.92	845.12	765.75	752.36	752.36	692.67

T <sub>inc</sub>	T <sub>c2</sub>	T <sub>L1</sub> 1	T <sub>L1</sub> 2	T <sub>L1</sub> 4	T <sub>c3</sub>	T <sub>c4</sub>
1020.16	1302.98	843.32	765.61	752.37	693.60	76.5117
1070.28	1288.69	841.59	765.47	752.37	694.52	80.2712
1120.21	1275.12	839.93	765.34	752.38	695.41	84.0157
1170.14	1262.18	838.34	765.21	752.39	696.27	87.7603
1220.26	1249.78	836.81	765.09	752.39	697.13	91.5197
1270.19	1237.97	835.34	764.97	752.40	697.95	95.2642
1320.12	1226.68	833.93	764.84	752.41	698.75	99.0087
1370.24	1215.83	832.57	764.72	752.41	699.54	102.7682
1420.17	1205.48	831.26	764.60	752.42	700.31	106.5127
1470.30	1195.52	829.99	764.48	752.43	701.05	110.2722
1520.22	1186.00	828.77	764.37	752.44	701.79	114.0167
1570.15	1176.87	827.61	764.27	752.44	702.51	117.7612
1620.28	1169.06	826.47	764.18	752.45	703.21	121.5207
1670.20	1159.63	825.37	764.08	752.46	703.89	125.2652
1720.13	1151.51	824.33	763.99	752.46	704.56	129.0097
1770.26	1143.67	823.30	763.90	752.47	705.22	132.7692
1820.18	1136.16	822.33	763.82	752.48	705.86	136.5137
1870.11	1128.91	821.37	763.73	752.49	706.48	140.2582
1920.24	1121.90	820.45	763.65	752.49	707.09	144.0177
1970.16	1115.16	819.55	763.58	752.50	707.70	147.7622
2020.29	1103.63	818.69	763.50	752.51	708.29	151.5217
2070.22	1102.34	817.84	763.42	752.51	708.86	155.2662
2120.14	1096.27	817.05	763.36	752.52	709.43	166.20
2170.27	1090.36	816.25	763.28	752.53	709.98	168.77
2220.20	1084.70	815.49	763.22	752.54	710.53	171.28
2270.12	1079.20	814.76	763.15	752.55	711.06	173.76
2320.25	1073.96	814.02	763.09	752.55	711.58	176.21
2370.18	1068.71	813.33	763.02	752.56	712.09	177.7632
2420.10	1063.71	812.66	762.96	752.57	712.58	181.5077
2470.23	1058.85	811.98	762.90	752.58	713.07	185.2672
2520.16	1054.15	811.34	762.84	752.58	713.56	189.0117
2570.28	1049.57	810.72	762.78	752.59	714.04	192.7712
2620.21	1045.15	C10.11	762.72	752.60	714.50	196.5157
2670.14	1040.85	809.50	762.66	752.61	714.95	492.18
2720.26	1036.65	808.93	762.60	752.62	715.40	494.33
2770.19	1032.59	808.38	762.54	752.62	715.83	496.43
2820.12	1028.65	807.83	762.49	752.63	716.26	498.50
2870.24	1024.83	807.28	762.44	752.64	716.69	500.54
2920.17	1021.06	806.76	762.39	752.65	717.11	502.56
2970.30	1017.41	806.26	762.35	752.66	717.52	504.53
3020.22	1013.87	805.77	762.31	752.67	717.92	506.48
3070.15	1010.42	805.28	762.26	752.68	718.31	508.39
3120.28	1007.05	804.79	762.22	752.68	718.69	510.29
3170.20	1003.78	804.34	762.18	752.69	719.07	512.14
3220.13	1000.58	803.90	762.14	752.70	719.44	513.97
3270.26	997.46	803.46	762.10	752.71	719.81	515.79
3320.18	994.42	803.04	762.07	752.72	720.18	517.56
3370.11	991.46	802.61	762.03	752.73	720.54	519.33
3420.24	988.56	802.18	761.99	752.74	720.88	521.05
3470.16	985.73	801.78	761.96	752.75	721.23	522.76

TIME	T(°C)	T(°C)	T(°C)	T(°C)	T(°C)
3520.29	982.96	801.40	761.93	752.76	721.56
3570.22	980.27	801.02	761.90	752.77	721.89
3620.14	977.63	800.65	761.86	752.78	722.22
3670.27	975.05	800.23	761.83	752.79	722.54
3720.20	972.54	799.91	761.80	752.80	722.86
3770.12	970.08	799.55	761.77	752.81	723.17
3820.25	967.66	799.19	761.74	752.82	723.48
3870.18	965.31	798.86	761.71	752.83	723.73
3920.10	963.00	798.53	761.69	752.84	724.09
3970.23	960.73	798.21	761.66	752.86	724.38
4020.16	958.53	797.89	761.64	752.87	724.67
4070.28	956.37	797.58	761.61	752.88	724.95
4120.26	954.22	797.27	761.59	752.89	725.23
4170.26	952.10	796.96	761.56	752.90	725.51
4220.26	950.04	796.65	761.53	752.91	725.79
4270.26	948.03	796.34	761.50	752.92	726.06
4320.26	946.05	796.05	761.48	752.94	726.33
4370.26	944.12	795.77	761.46	752.95	726.59
4420.26	942.24	795.50	761.44	752.96	726.85
4470.26	940.38	795.23	761.42	752.97	727.11
4520.26	938.57	794.96	761.40	752.98	727.36
4570.26	936.79	794.70	761.38	753.00	727.61
4620.26	935.04	794.45	761.36	753.01	727.86
4670.26	933.34	794.20	761.34	753.02	728.11
4720.26	931.65	793.95	761.32	753.03	728.34
4770.26	930.02	793.70	761.30	753.05	728.58
4820.26	928.39	793.45	761.28	753.06	728.82
4870.26	926.81	793.21	761.26	753.07	729.04
4920.26	925.25	792.98	761.24	753.08	729.27
4970.26	923.72	792.75	761.23	753.10	729.49
5020.26	922.22	792.53	761.21	753.11	729.72
5070.26	920.73	792.31	761.20	753.12	729.93
5120.26	919.30	792.10	761.18	753.14	730.14
5170.26	917.86	791.89	761.17	753.15	730.36
5220.26	916.46	791.69	761.16	753.17	730.57
5270.26	915.09	791.48	761.14	753.18	730.78
5320.26	913.71	791.28	761.13	753.19	730.98
5370.26	912.40	791.06	761.12	753.21	731.18
5420.26	911.08	790.89	761.10	753.22	731.38
5470.26	909.77	790.69	761.09	753.24	731.57
5520.26	908.52	790.51	761.08	753.25	731.77
5570.26	902.46	790.32	761.07	753.27	731.96
5620.26	900.02	790.13	761.05	753.28	732.15
5670.26	894.83	789.94	761.04	753.30	732.34
5720.26	903.64	789.76	761.03	753.31	732.53
5770.26	902.46	789.57	761.02	753.33	732.72
5820.26	901.30	789.39	761.00	753.34	732.90
5870.26	900.18	789.23	761.00	753.36	733.06
5920.26	899.05	789.06	760.99	753.37	733.26
5970.26	897.93	788.89	760.98	753.39	733.43

TIME	PROGRAM EXECUTION STOPPED BY	PROGRAM	VALUES 0.845E+04 0.0	0.100E+03 0.0	0.114E+04
6020.26	896.86	788.73	760.97	753.40	587.16
6070.26	895.79	788.57	760.96	753.42	733.77
6120.26	894.73	788.41	760.95	753.44	733.94
6170.26	893.67	788.25	760.94	753.45	734.11
6220.26	892.66	788.10	760.93	753.47	734.28
6270.26	891.66	787.95	760.93	753.49	734.44
6320.26	890.66	787.80	760.92	753.50	734.60
6370.26	889.66	787.65	760.91	753.52	734.76
6420.26	888.69	787.50	760.90	753.54	734.92
6470.26	887.76	787.36	760.90	753.55	735.08
6520.26	886.82	787.22	760.89	753.57	735.23
6570.26	885.88	787.08	760.89	753.59	735.39
6620.26	884.94	786.93	760.88	753.61	735.54
6670.26	884.05	786.80	760.87	753.62	735.70
6720.26	883.18	786.67	760.87	753.64	735.84
6770.26	882.30	786.53	760.86	753.66	735.99
6820.26	881.43	786.40	760.86	753.68	736.13
6870.26	880.55	786.27	760.85	753.70	736.28
6920.26	879.70	786.13	760.85	753.71	736.42
6970.26	878.89	786.01	760.84	753.73	736.56
7020.26	878.07	785.88	760.84	753.75	736.70
7070.26	878.00	785.84	760.84	753.77	736.85
7120.26	878.00	785.91	760.86	753.79	736.99
7170.26	878.00	785.99	760.91	753.82	737.12
7220.26	878.00	786.08	760.96	753.84	737.25
7270.26	878.00	786.18	761.01	753.86	737.38
7320.26	878.00	786.27	761.07	753.89	737.51
7370.26	878.00	786.37	761.12	753.91	737.64
7420.26	878.00	786.46	761.17	753.93	737.77
7470.26	878.00	786.56	761.23	753.96	738.90
7520.26	878.00	786.65	761.28	753.98	738.03
7570.26	878.00	786.74	761.33	754.03	738.16
7620.26	878.00	786.84	761.39	754.03	738.29
7670.26	877.99	786.93	761.44	754.06	738.42
7720.26	878.00	787.03	761.50	754.08	738.54
7770.26	878.00	787.12	761.55	754.11	738.67
7820.26	877.97	787.21	761.60	754.13	738.78
7870.26	877.83	787.27	761.65	754.16	738.90
7920.26	877.60	787.32	761.69	754.19	739.02
7970.26	877.27	787.33	761.72	754.21	739.14
					618.48
					597.7695