University of Nebraska - Lincoln DigitalCommons@University of Nebraska - Lincoln

Patrick Dussault Publications

Published Research - Department of Chemistry

2010

$B(C_6F_5)_3$ -promoted tandem silulation and intramolecular hydrosilulation: diastereoselective synthesis of oxasilinanes and oxasilepanes

Roman Shchepin University of Nebraska–Lincoln

Chunping Xu University of Nebraska-Lincoln

Patrick Dussault University of Nebraska-Lincoln, pdussault1@unl.edu

Follow this and additional works at: http://digitalcommons.unl.edu/chemistrydussault

Shchepin, Roman; Xu, Chunping; and Dussault, Patrick, "B(C_6F_5)₃-promoted tandem silvlation and intramolecular hydrosilvlation: diastereoselective synthesis of oxasilinanes and oxasilepanes" (2010). *Patrick Dussault Publications*. 31. http://digitalcommons.unl.edu/chemistrydussault/31

This Article is brought to you for free and open access by the Published Research - Department of Chemistry at DigitalCommons@University of Nebraska - Lincoln. It has been accepted for inclusion in Patrick Dussault Publications by an authorized administrator of DigitalCommons@University of Nebraska - Lincoln.



NIH Public Access Author Manuscript

Org Lett. Author manuscript; available in PMC 2011 November

Published in final edited form as:

Org Lett. 2010 November 5; 12(21): 4772–4775. doi:10.1021/ol1018757. Copyright © 2010 American Chemical Society

B(C₆F₅)₃-promoted tandem silylation and intramolecular hydrosilylation: diastereoselective synthesis of oxasilinanes and oxasilepanes

Roman Shchepin[†], **Chunping Xu[†]**, and **Patrick Dussault^{*}** Department of Chemistry, University of Nebraska–Lincoln, Lincoln, NE 68588-0304

Abstract



 $B(C_6F_5)_3$ promotes regio- and stereoselective cyclizations of unsaturated alkoxysilanes to generate oxasilinanes and oxasilepanes. The same products are available directly from alkenols via tandem silylation and hydrosilylation.

Intramolecular hydrosilylation of alkenes is an important transformation in organic synthesis.1 Initially investigated for unsaturated silanes,2 the methodology is now often applied to unsaturated alkoxy- and aminosilanes,3 where stereospecific oxidative cleavage of the newly formed C-Si bond enables stereodefined synthesis of diols and aminoalcohols. 4,5 The majority of examples involve metal-catalyzed 5-*endo* or 5-*exo* ring closures, although six-membered cyclizations have been reported.1,3,6 We now report regio- and stereoselective formation of oxasilinanes and oxasilepanes via formation and cyclization of unsaturated alkoxysilanes in the presence of a nonmetal catalyst.

In the course of investigations into the influence of Lewis acids on the ozonolysis of unsaturated silanes, we found that addition of $B(C_6F_5)_3$ to a solution of unsaturated alkoxysilane **1-Pr** resulted in regioselective formation of oxasilinane **2-Pr** with high 3,5-*trans* diastereoselectivity (Table 1).7.8 The cyclization proceeded efficiently at -78 °C or RT and in the presence of either stoichometric or catalytic $B(C_6F_5)_3$. Cyclization was also observed for the dimethylsilyl ether (not shown),9 but the hydrolytic instability of this class of reactants led us to abandon this thread following the discovery of the tandem cyclizations discussed later.

pdussault1@unl.edu.

[†]These authors contributed equally to this work.

Supporting Information Available. Details regarding preparation and characterization of new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

The cyclization, apparently the first intramolecular example of a known intermolecular hydrosilylation, 10 was investigated further using alkoxysilanes prepared as illustrated in Scheme 1. 6-*Endo* cyclization onto an α -substituted styrene (**3-Pr**) proceeded slowly but in high yield and with high *trans* selectivity (Scheme 2). Cyclization onto a cyclobutene (**5-Pr**) proceeded much more slowly through a 6-*endo* pathway to furnish a modest yield of the *cis*-fused adducts (**6-Pr**) as a 5:1 mixture of sidechain epimers. A bishomoallyl substrate, **7-Pr**, reacted very slowly through a 6-*exo* pathway to furnish a *trans*-3,6-disubstituted-2-oxa-1-silinane (**8-Pr**).

 $B(C_6F_5)_3$ also catalyzes the reductive silylation of alcohols,11 and we became intrigued by the possibility of tandem silylation/hydrosilylation (Table 2). $B(C_6F_5)_3$ -promoted reaction of alkenol **1** with stoichometric Et_2SiH_2 or Ph_2SiH_2 generated oxasilinanes **2-Et** or **2-Ph** with very similar regio- and stereoselection as observed in the stepwise cyclizations. Although alcohols **3** and **5** decomposed under the tandem conditions, cyclohexenol **9** reacted to selectively furnish the 3,5-*trans* diastereomer of *cis*-fused octahydrobenzooxasilinanes **10-Et** and **10-Ph**; the lower yield for the Et_2SiH_2 reaction is likely related to undesired reductive deoxygenations (vida infra). Alkenol **11**, which generates an intermediate siloxane capable of undergoing cyclization through elecronically comparable 5-*exo* or 6-*endo* pathways, reacted only through the latter. Bishomoallyl alcohol **13** underwent selective reaction through a 7-*endo* pathway to furnish oxasilepane **14-Et** as a 62:38 cis/trans mixture.

Reactions employing Et_2SiH_2 often furnished a significant amount of byproducts appearing to result from alcohol deoxygenation.12 For example, reaction of benzylic alcohol **15** produced oxasilane **16-Et** along with a byproduct identified as a disiloxane on the basis of mass spectrometry and oxidative desilylation (Scheme 3).13^{,14} Application of the one-pot conditions to allylic alcohol **17** resulted only in rapid formation of the diethyl silyl ether. In general, reactions employing Ph_2SiH_2 proceeded more slowly but generated fewer byproducts; this can be seen, for example in the formation of **10-Et** vs. **10-Ph** (Table 2). The exception was cyclobutene **5**, where decomposition was observed for either silane.

Oxidative desilylation of the hindered siloxanes was initially attempted under Tamao conditions (KF, KHCO₃, aq. H₂O₂, MeOH/THF).5 However, as illustrated in Scheme 4, the oxidations were found to proceed in higher yield using a procedure developed by Woerpel (*t*-BuOOH, CsOH•H₂O, *n*-Bu₄NF, DMF).5 The stereochemistry of diols **19**15 and **21**16 was determined by comparison with literature reports, establishing (**14-Et**) or confirming (**16-Et**) the stereochemistry of cyclizations.

The cyclizations, clearly related to intermolecular $B(C_6F_5)_3$ -mediated hydrosilylations,10 and potentially related to cyclizations of unsaturated silanes in the presence of triphenylmethyl cation,17 almost certainly involve electrophilic attack on an alkene by a silylium-like species derived from interaction of $B(C_6F_5)_3$ with the Si-H (Scheme 5).18,19 Reduction of the resulting carbocation by the hydridoboron species would furnish the cyclized product and regenerate the Lewis acid catalyst. The selective formation of 3,5*trans*-disubstituted oxasilinanes can be rationalized by hyperconjugation of the newly formed C-Si bond with the carbocation,20 with the resulting conformation dictating approach of the hydride. Analogous stereoselectivity has been observed in formation of siloxanes through hydrogen atom deliver to carbon-centered radicals.21

Although 5-*exo* cyclizations are well-established for Pt-or Rh-catalyzed hydrosilylations,1^{,3} we observed selective 6-*endo* vs. 5-*exo* cyclization with a substrate where either mode would proceed via a secondary carbocation (Table 2, substrate **11**). We also observed very different rates for 6-*exo* and 6-*endo* cyclizations involving electronically similar carbocation intermediates (**7-Pr** vs. **1-Pr**). These results point to the importance of interactions between

Org Lett. Author manuscript; available in PMC 2011 November 5.

the alkene and the developing silylium-like species. The *cis* selectivity observed for sixmembered ring annelations, which complements results from metal-catalyzed cyclizations, 1[,]3[,]22 presumably reflects stereoelectronic requirements for trapping of the β -silyl cations. 23 The stereoselectivity of sidechain introduction results from cyclization through the lowenergy conformer of a chair-like transition state (eq 1).



Several lines of evidence indicate that the tandem reactions and stepwise processes involve a common hydrosilylation step. Both processes proceed with nearly identical regio- and diastereoselectivity. Furthermore, dialkylsilyl ethers are observed (TLC) as intermediates in some of the slower reactions, and become the only product when cyclization is disfavored, as for allylic alcohol **17** (Scheme 3). Finally, a diene substrate reacts selectively across the homoallyl alcohol (eq 2).



The formation of deoxygenated byproducts is observed mainly in the tandem reactions. The chemoselective deoxygenation of unhindered alcohols by trialkylsilane and $B(C_6F_5)_3$ has been postulated to involve attack of a silylium ate complex on intermediate silyl ethers,12 suggesting the deoxygenations observed here result from intermolecular reductions directly competing with cyclization.

Overall, the transformation provides a new method for the regio- and stereoselective synthesis of cyclic siloxanes and derived diols. Given that $B(C_6F_5)_3$ has been reported to catalyze the hydrosilylation of ketones and aldehydes,23 it is likely the method could be extended to allow the synthesis of oxasilacycles from unsaturated aldehydes and ketones.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

Research support and funding for NMR instrumentation was provided by NSF (CHE-0749916, MRI 0079750, and CHE 0091975). Research was conducted in facilities remodeled with NIH support (RR016544-01). We are grateful for advice from Prof. V. Gevorgyan (Univ. of Illinois at Chicago) and for advice or technical support from S. DiMagno, J. Dumais, J. Belot, C. Schwartz, and S. Basiaga (Univ. of Nebraska-Lincoln).

Org Lett. Author manuscript; available in PMC 2011 November 5.

(1)

(2)

References

- a) Yamamoto, K.; Hayashi, T. Transition Metals for Organic Synthesis. 2. Beller, M.; Bolm, C., editors. Wiley-VCH; Weinheim: 2004. p. 167-191.b) Marciniec, B. Hydrosilylation: A Comprehensive Review on Recent Advances. Marciniec, B., editor. Springer; New York: 2009. p. 3-51.
- 2. Benkeser RA, Mozden EC, Muench WC, Roche RT, Siklosi MP. J Org Chem. 1979; 44:1370.
- 3. Varchi G, Ojima I. Curr Org Chem. 2006; 10:1341.
- See, for example: a) Mak SYF, Curtis NR, Payne AN, Congreve MS, Francis CL, Burton JW, Holmes AB. Synthesis. 2005:3199. b) Li F, Roush WR. Org Lett. 2009; 11:2932. [PubMed: 19507846]
- Oxidative cleavage: a) Jones GR, Landais Y. Tetrahedron. 1996; 52:7599. b) Smitrovich JH, Woerpel KA. J Org Chem. 1996; 61:6044.
- Cyclizations to form six- or seven-membered rings: a) Widenhoefer RA. Acc Chem Res. 2002; 35:905. [PubMed: 12379143] b) Xin S, Harrod JF. J Orgmet Chem. 1995; 499:181. c) DeCamp AE, Mills SG, Kawaguchi AT, Desmond R, Reamer RA, DiMichele L, Volante RP. J Org Chem. 1991; 56:3564.d) Sibi, MP. e-EROS Encyclopedia of Reagents for Organic Synthesis. John Wiley & Sons; 2001. e) Widenhoefer RA, Krzyzanowska B, Webb-Wood G. Organometallics. 1998; 17:5124.
- 7. Reviews of B(C₆F₅)₃: a) Piers WE. Adv Organomet Chem. 2005; 52:1. b) Erker G. Dalton Trans. 2005:1883. [PubMed: 15909033]
- 8. Stereochemical assignments are based upon ${}^{3}J_{H}$ couplings, nOe correlations, and a literature correlation for diol 21.
- 9. Tamao K, Nakajima T, Sumiya R, Arai H, Higuchi N, Ito Y. J Am Chem Soc. 1986; 108:6090.
- 10. Rubin M, Schwier T, Gevorgyan V. J Org Chem. 2002; 67:1936. [PubMed: 11895414]
- 11. Blackwell JM, Foster KL, Beck VH, Piers WE. J Org Chem. 1999; 64:4887. [PubMed: 11674566]
- 12. Gevorgyan V, Rubin M, Benson S, Liu JX, Yamamoto Y. J Org Chem. 2000; 65:6179. [PubMed: 10987957]
- B(C₆F₅)₃-promoted formation of disiloxanes: Chojnowski J, Rubinsztajn S, Cella JA, Fortuniak W, Cypryk M, Kurjata J, Kazmierski K. Organometallics. 2005; 24:6077.
- 14. Lebel H, Ladjel C. J Organomet Chem. 2005; 690:5198.
- 15. Zhu G, Negishi E-i. Org Lett. 2007; 9:2771. [PubMed: 17583343]
- 16. Nájera C, Yus M, Seebach D. Helv Chim Acta. 1984; 67:289.
- 17. Steinberger H-U, Bauch C, Mueller T, Auner N. Can J Chem. 2003; 81:1223.
- 18. Larson GL, Fry JL. Org React. 2008; 71:1.
- 19. Reed CA. Acc Chem Res. 1998; 31:325.
- 20. Lambert JB. Tetrahedron. 1990; 46:2677.
- 21. Cai Y, Roberts BP. J Chem Soc, Perkin Trans 1. 1998:467. Ibid., 3653. In contrast to the results described in this work, we observed a single predominant conformation of oxasilinanes.
- 22. Formation of cis-oxasilinanes through 6-exo cyclization of α -silyloxymethyl radicals: Koreeda M, Hamann LG. J Am Chem Soc. 1990; 112:8175.
- 23. Parks DJ, Blackwell JM, Piers WE. J Org Chem. 2000; 65:3090. [PubMed: 10814201] Gevorgyan V, Rubin M, Liu JX, Yamamoto Y. J Org Chem. 2001; 66:1672. [PubMed: 11262111]

Shchepin et al.



^aInseparable 3:1 mixture with 1-chloro-1-alkylcyclobutane

Scheme 1.

Preparation of alkoxysilanes ^aInseparable 3:1 mixture with 1-chloro-1-alkylcyclobutane Shchepin et al.



Scheme 2. Additional cyclizations



Scheme 3. Byproduct formation ^{*a*} 0.6 equivalents. ^{*b*} 0.25 equivalents.



Scheme 4. Oxidative desilylation

Org Lett. Author manuscript; available in PMC 2011 November 5.



Scheme 5. Proposed mechanism

Table 1

Cyclization of **1-Pr**^a

iPr, iPr H ^{Si} O Hex 1-Pr	∬_Me B(C	₆ F ₅) _{3,} ene ►	iPr O Hex H	iPr Si 5 .H Me P-Pr
BAr ₃ (equiv)	temp (°C) ^b	<i>t</i> (h)	yield (%)	trans %
1.0	-78	< 0.1	82	nd
0.4	rt	< 0.1	88	nd
0.1	-78	< 0.1	93	nd

^aPrepared as illustrated in Scheme 1.

 b Final temperature; reactants mixed at -78 °C.

^c5% of the *cis*-diastereomer isolated.

NIH-PA Author Manuscript

NIH-PA Author Manuscript

Tandem silylation/hydrosilylation

Hex h_{n}	Hex h_{n} H_{1} H_{2} H_{1} H_{2} H_{1} H_{2} H_{1} H_{2} H_{1} H_{2} H_{1} H_{1} H_{1} H_{2} H_{1} H_{2} H_{1}	HexImage for the transmetric of		HO-	R ⁻	X ₂ SiH B(C ₆ F	2 (1.1 € 5)3 (0.1	equiv) I-0.5 equi	ر م	× × Si ×
subs $\mathbf{R}^1, \mathbf{R}^2$ \mathbf{X} $t(\mathbf{h})$ prod yield $trans$ δs 1 1 Me,H Et 0.1 2 -Et 47 % >90 1 1 Me,H Ph 0.15 2 -Ph 80 % >90 3 1 Ph,H Ph $ -$ decomp 5 1 (CH ₂) ₂ Ph $ -$ decomp 9 1 (CH ₂) ₄ Et 0.5 10 -Ph/b 39 % 90 11^{c} 1 (CH ₂) ₄ Ph 1 10 -Ph/b 73 % 84 11^{c} 1 H,Me Et 0.1 10 -Ph/b 73 % 90 11^{c} 1 H,Me Ph 1 10 -Ph/b 73 % 90 11^{c} 1 H,Me Ph 1 10 -Ph/b 73 % 90 11^{c} 1 H,PEt	subs $\mathbf{R}^1, \mathbf{R}^2$ \mathbf{X} \mathbf{f} (h) prod yield trans % 1 1 Me,H Et 0.1 2 -Et 47 $\mathbf{>00}$ 1 1 Me,H Et 0.1 2 -Et 47 $\mathbf{>00}$ 3 1 Me,H Ph 0 2 -Ph 80 $\mathbf{>00}$ 3 1 Ph,H Ph 0 2 -Ph 80 $\mathbf{>00}$ 5 1 0 -15 2 -Ph 80 $\mathbf{>00}$ 9 1 0 -15 \mathbf{P} 0 0 9 1 0 -15 1 0 -16 0 0 1 1 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	subs \mathbf{R} , \mathbf{R}^2 , \mathbf{X} \mathbf{f} (h)prodyield \mathbf{trans} $\mathbf{y}_{\rm ed}$ 11Me.HEt0.1 2 -Et 47% >9031Me.HPhPh 2 -Et 47% >9031Ph.HPhPh 2 -Et 47% >9041Me.HPh 2 -Et 47% >9031Ph.HPh 2 2 -decomp41(CH ₂) ₂ Ph 2 2 -decomp91(CH ₂) ₄ Ph 10 -Et/b 39% 90 11^{c} 1H.MePh 10 -Et/b 73% 84 11^{c} 1H.MePh 10 -Et/b 73% 50 11^{c} 1H.MePh 10 -Et/b 73% 50 11^{c} 1H.MePh 10 -Et/b 73% 50 13^{c} 2H.MePh 91 10 -Et/b 50 13^{c} 2H.MePh 91 10 -Et/b $\mathbf{73\%}$ 50 13^{c} 2H.MePh 91 10 10^{c} $\mathbf{73\%}$ 51 13^{c} 2H.MePh 91 10^{c} 10^{c} 10^{c} 50^{c} 13^{c} 2Ph 91^{c} 91^{c} 91^{c} 50^{c} 50^{c} 13^{c} 2 <th>Hex</th> <th>\prec</th> <th>n ™1</th> <th>t</th> <th>oluene,</th> <th>, 0 °C</th> <th>Hex</th> <th></th>	Hex	\prec	n ™1	t	oluene,	, 0 °C	Hex	
1 1 Me,H Et 0.1 2 -Et 47% >90 1 1 Me,H Ph 0.15 2 -Ph 80% >90 3 1 Ph,H Ph 0.15 2 -Ph 80% >90 5 1 Ph,H Ph - - decomp 6 1 (CH ₂) ₂ Ph - - decomp 9 1 (CH ₂) ₄ Et 0.5 10 -Et decomp 9 1 (CH ₂) ₄ Ph 1 10 -Et 39% 90 9 1 (CH ₂) ₄ Ph 1 10 -Et 73% 84 $11c$ 1 H,Me Pt 11 12 -Et 16% 60 $11c$ 1 H,Me Pt 1 12 -Et 24% $\sim1:1$ $11c$ 1 H,Me Pt 1 12 -Et $\sim1:1$ $\sim1:1$ </th <th>I Me,H Et 0.1 2-Et 47 % >90 I Me,H Ph 0.15 2-Ph 80 % >90 3 I Ph,H Ph 0.15 2-Ph 80 % >90 3 I Ph,H Ph - - decomp 5 I (CH₂)₂ Ph - - decomp 9 I CH₂ Ph - - decomp 9 I CH₂ Ph - - - decomp 9 I OD-Et 39 % 80 % 90 9 I OD-Etb 73 % 84 11c I HMe Pi I 12-Pi 73 % \sim 13 2 H.Me Et 0.1 14-Et 73 % 38</th> <th>1 Me,H Et 0.1 2-Et 47% >90 1 Me,H Ph 0.15 2-Ph 80% >90 3 1 Ph,H Ph 0.15 2-Ph 80% >90 4 1 Ph,H Ph 2 - decomp >90 4 1 Ph,H Ph - - - decomp 5 1 (CH₂)₂ Ph - - - decomp 9 1 (CH₂)₄ Et 0.5 10-Et/b 33% 90 9 1 (CH₂)₄ Ph 1 10-Ph/b 73% 84 11c 1 H,Me Ph 1 12-Et 16% \sim1:1 13 2 H,Me Pt 0.1 14-Et 73% 38 3.5 stereochemistry: 3.5 3.5 3.5 3.5 3.5 3.5</th> <th>sqns</th> <th>u</th> <th>R¹, R²</th> <th>X</th> <th><i>t</i> (h)</th> <th>prod</th> <th>yield</th> <th>trans %a</th>	I Me,H Et 0.1 2-Et 47 % >90 I Me,H Ph 0.15 2-Ph 80 % >90 3 I Ph,H Ph 0.15 2-Ph 80 % >90 3 I Ph,H Ph - - decomp 5 I (CH ₂) ₂ Ph - - decomp 9 I CH ₂ Ph - - decomp 9 I CH ₂ Ph - - - decomp 9 I OD-Et 39 % 80 % 90 9 I OD-Etb 73 % 84 11c I HMe Pi I 12-Pi 73 % \sim 13 2 H.Me Et 0.1 14-Et 73 % 38	1 Me,H Et 0.1 2-Et 47% >90 1 Me,H Ph 0.15 2-Ph 80% >90 3 1 Ph,H Ph 0.15 2-Ph 80% >90 4 1 Ph,H Ph 2 - decomp >90 4 1 Ph,H Ph - - - decomp 5 1 (CH ₂) ₂ Ph - - - decomp 9 1 (CH ₂) ₄ Et 0.5 10-Et/b 33% 90 9 1 (CH ₂) ₄ Ph 1 10-Ph/b 73% 84 11c 1 H,Me Ph 1 12-Et 16% \sim 1:1 13 2 H,Me Pt 0.1 14-Et 73% 38 3.5 stereochemistry: 3.5 3.5 3.5 3.5 3.5 3.5	sqns	u	R ¹ , R ²	X	<i>t</i> (h)	prod	yield	trans %a
1 1 Me,H Ph 0.15 2-Ph 80% >90 3 1 Ph,H Ph - - decomp 5 1 (H2)2 Ph - - decomp 9 1 (CH3)2 Ph - - decomp 9 1 (CH3)4 Et 0.5 10-Etb 39% 90 9 1 (CH3)4 Ph 1 10-Phb 73% 84 11c 1 HMe Et 0.1 12-Et 16% 60 11c 1 HMe Ph 1 12-Eth 24% ~11 13 2 H,Me Et 0.1 14-Et 73% 35	1 1 Me,H Ph 0.15 2-Ph 80 % >90 3 1 Ph,H Ph - - decomp 5 1 $(CH_2)_2$ Ph - - decomp 9 1 $(CH_2)_4$ Et 0.5 10 -Et $39 %$ 90 9 1 $(CH_2)_4$ Ph 1 10 -Et $39 %$ 90 1 $(CH_2)_4$ Ph 1 10 -Et $39 %$ 90 1 $(CH_2)_4$ Ph 1 10 -Et $39 %$ 90 1 $(CH_2)_4$ Ph 1 10 -Et $16 %$ $73 %$ 84 11c 1 H,Me Ph 1 12 -Ph 24% -11 13 2 H,Me Pt 0.1 14 -Et 73% 38	1 1 Me,H Ph 0.15 2-Ph 80 % >90 3 1 Ph,H Ph - - decomp 5 1 (CH ₂) ₂ Ph - - decomp 9 1 (CH ₂) ₄ Et 0.5 10-Eh 39 % 90 9 1 (CH ₂) ₄ Ph 1 10-Eh 39 % 90 11c 1 H,Me Et 0.1 12-Eh 16 % 60 11c 1 H,Me Ph 1 12-Eh 73 % 33 13 2 H,Me Et 0.1 12-Eh 73 % 33 3.5 stereochemistry: 3 0.1 14-Et 73 % 33	1	-	Me,H	Εt	0.1	2-Et	47 %	>90
3 1 Ph,H Ph $ -$	3 1 Ph,H Ph $ decomp$ 5 1 $(CH_2)_2$ Ph $ decomp$ 9 1 $(CH_2)_4$ Et 0.5 $10-Et^4b$ 39 <% 90 9 1 $(CH_2)_4$ Ph 1 $10-Ph_b$ 73 <% 84 $11c$ 1 HMe Et 0.1 $12-Et$ 16 60 $11c$ 1 HMe Et 0.1 $12-Et$ 16 61 $11c$ 1 HMe Et 0.1 $12-Et$ 16 73 81 13 2 HMe Et 0.1 $14-Et$ 73 83	3 1 Ph,H Ph $ decomp$ 5 1 $(CH_2)_2$ Ph $ decomp$ 9 1 $(CH_2)_4$ Et 0.5 10 - Et b 39% 90 9 1 $(CH_2)_4$ Et 0.5 10 - Et b 39% 90 1 1 10 - Et b 73% 84 $11c$ 1 H,Me Et 0.1 12 - Et 16% 60 $11c$ 1 H,Me Et 0.1 12 - Et 16% 73% 31 13 2 H,Me Et 0.1 14 - Et 73% 33 3.5 stereochemistry: 3.5 stereochemistry: 3.5 3.5 3.5 3.5 3.5	1	1	Me,H	Ρh	0.15	2-Ph	80 %	>90
5 1 $(CH_{2})_2$ Ph - - decomp 9 1 $(CH_{2})_4$ Et 0.5 10 - $\mathbf{E}t^b$ 39% 90 9 1 $(CH_{2})_4$ Ph 1 10 - $\mathbf{E}t^b$ 39% 90 9 1 $(CH_{2})_4$ Ph 1 10 - $\mathbf{P}t^b$ 73% 84 $11c$ 1 14 - $\mathbf{P}t$ 16% 60 61 $11c$ 1 14 - $\mathbf{R}t$ 24% $\sim 1:1$ 13 2 $\mathbf{H}, \mathbf{M}e$ $\mathbf{E}t$ 0.1 14 - $\mathbf{E}t$ 73% 38	5 1 $(CH_2)_2$ Ph - - decomp 9 1 $(CH_2)_4$ Et 0.5 10 -Et 39% 90 9 1 $(CH_2)_4$ Ph 1 10 -Et 39% 90 9 1 $(CH_2)_4$ Ph 1 10 -Ph 73% 84 $11c$ 1 H.Me Et 0.1 12 -Et 16% 60 $11c$ 1 H.Me Ph 1 12 -Ph 24% $\sim11:1$ 13 2 H.Me Et 0.1 14 -Et 73% 38	5 1 $(CH_2)_2$ Ph - - decomp 9 1 $(CH_2)_4$ Et 0.5 10 -Et 39% 90 9 1 $(CH_2)_4$ Ph 1 10 -Et 39% 90 9 1 $(CH_2)_4$ Ph 1 10 -Et 73% 84 $11c$ 1 H,Me Et 0.1 12 -Et 16% 60 $11c$ 1 H,Me Ph 1 12 -Et 73% 38 $11c$ 1 H,Me Ph 0.1 12 -Et 73% 38 13 2 H,Me Et 0.1 14 -Et 73% 38 3.5 stereochemistry: 3.5 stereochemistry: 3.5 3.5 3.5 3.5	e	1	Ph,H	Ph	ī		Ō	lecomp
9 1 (CH ₂) ₄ Et 0.5 10-Et 39 % 90 9 1 (CH ₂) ₄ Ph 1 10-Ph 73 % 84 11 1 H,Me Et 0.1 12-Et 16 % 60 11 1 H,Me Ph 1 12-Ph 24 % ~1:1 13 2 H,Me Et 0.1 14-Et 73 % 38	9 1 (CH ₂) ₄ Et 0.5 10 - \mathbf{Et} 39 % 90 9 1 (CH ₂) ₄ Ph 1 10 - \mathbf{Ph} 73 % 84 11 1 H.Me Et 0.1 12 - \mathbf{Et} 16 % 60 11 1 H.Me Ph 1 12 - \mathbf{Et} 16 % ~01 11 1 H.Me Ph 1 12 - \mathbf{Ph} 24 % ~11 13 2 H.Me Et 0.1 14 - \mathbf{Et} 73 % 38	9 1 $(CH_2)_4$ Et 0.5 10 -Et/b 39% 90 9 1 $(CH_2)_4$ Ph 1 10 -Et/b 73% 84 $11c$ 1 H,Me Et 0.1 12 -Et 16% 60 $11c$ 1 H,Me Ph 1 12 -Et 16% 61 $11c$ 1 H,Me Ph 1 12 -Et 73% 38 13 2 H,Me Et 0.1 14 -Et 73% 38 3.5 stereochemistry: S.5 stereochemistry: 36.7 38.7 38.7 38.7	ŝ	-	(CH ₂) ₂	Ph	·	ı	Ō	lecomp
9 1 (CH ₂) ₄ Ph 1 10-Ph b 73 % 84 11c 1 H,Me Et 0.1 12-Et 16 % 60 11c 1 H,Me Ph 1 12-Ph 24 % ~1:1 13 2 H,Me Et 0.1 14-Et 73 % 38	9 1 (CH ₂) ₄ Ph 1 10-Ph b 73 % 84 11c 1 H.Me Et 0.1 12-Et 16 % 60 11c 1 H.Me Ph 1 12-Ph 24 % ~1:1 13 2 H.Me Et 0.1 14-Et 73 % 38	9 1 $(CH_2)_4$ Ph 1 10 - Phb 73 % 84 $11c$ 1 H,Me Et 0.1 12 - Et 16 % 60 $11c$ 1 H,Me Ph 1 12 - Et 16 % 60 $11c$ 1 H,Me Ph 1 12 - Ph 24% $\sim 1:1$ 13 2 H,Me Et 0.1 14 - Et 73% 38 $3,5$ stereochemistry: S. stereochemistry: 13 14 - Et 73% 38	6	-	(CH ₂) ₄	Εt	0.5	$10-Et^b$	39 %	06
I1c 1 H.Me Et 0.1 12-Et 16 % 60 I1c 1 H.Me Ph 1 12-Ph 24 % ~1:1 13 2 H.Me Et 0.1 14-Et 73 % 38	I1c 1 H.Me Et 0.1 12-Et 16 % 60 I1c 1 H.Me Ph 1 12-Ph 24 % ~1:1 13 2 H.Me Et 0.1 14-Et 73 % 38	11c 1 H.Me Et 0.1 12-Et 16 % 60 11c 1 H.Me Ph 1 12-Ph 24 % ~1:1 13 2 H.Me Et 0.1 14-Et 73 % 38 3.5 stereochemistry;	6	-	(CH ₂) ₄	Ph	1	$10-Ph^b$	73 %	84
I1c 1 H,Me Ph 1 12-Ph 24 % ~1:1 13 2 H,Me Et 0.1 14-Et 73 % 38	I1 ^c 1 H.Me Ph 1 12-Ph 24 % ~1:1 13 2 H.Me Et 0.1 14-Et 73 % 38 S S spreachemistry 38 38 38 38 38	11c 1 H.Me Ph 1 12-Ph 24 % ~1:1 13 2 H.Me Et 0.1 14-Et 73 % 38 3.5 stereochemistry; .5 .5 .5 .5 .5	11 ^{<i>c</i>}	-	H,Me	Et	0.1	12-Et	16 %	60
13 2 H,Me Et 0.1 14-Et 73 % 38	13 2 H.Me Et 0.1 14-Et 73 % 38	13 2 H,Me Et 0.1 14-Et 73 % 38 3,5 stereochemistry;	11 ^c	-	H,Me	Ph	1	12-Ph	24 %	~1:1
	\$ 5 stereorchemistry.),5 stereochemistry;	13	7	H,Me	Et	0.1	14-Et	73 %	38

Org Lett. Author manuscript; available in PMC 2011 November 5.

 $c_{3.3:1}$ mixture of E/Z isomers.

Shchepin et al.

B(C₆F₅)₃-promoted tandem silylation and intramolecular hydrosilylation: diastereoselective synthesis of oxasilinanes and oxasilepanes

OL 2010-018757

Roman Shchepin, Chunping Xu, and Patrick Dussault* pdussault1@unl.edu

Supporting Information - Experimental Procedures

General Experimental Procedures:	2
Alcohol precursors	2
2-Methyldec-1-en-4-ol (1):	2
2-Phenyldec-1-en-4-ol (3):	3
1-Cyclobutenyloctan-2-ol (5):	3
2-Methyl-2-dodecen-6-ol (7):	3
1-Cyclohexenyloctan-2-ol (9):	3
Undec-2-en-5-ol (11):	3
2-Methylundec-1-en-5-ol (13):	4
3-Methyl-1-phenylbut-3-en-1-ol (15):	4
2-Methylnon-1-en-3-ol (17):	4
Diisopropylsilyl ethers	4
Diisopropyl-(2-methyldec-1-en-4-oxy)silane (1-Pr):	4
Diisopropyl-(2-phenyldec-1-en-4-oxy) silane (3-Pr):	4
Diisopropyl-(1-cyclobutenyloctyl-2-oxy)silane (5-Pr):	5
Diisopropyl-(2-methyl-2-dodecen-6-oxy)silane (7-Pr):	5
$B(C_6F_5)_3$ -catalyzed cyclizations of diisopropylsilyl ethers (illustrated for 1-Pr)	5
(trans)-1,1-Diisopropyl-3-hexyl-5-methyl-2,1-oxasilinane (trans-2-Pr) and cis-2-Pr	6
(trans)- 1,1-Diisopropyl-3-hexyl-5-phenyl-2,1-oxasilinane (trans-4-Pr):	7
$(3\alpha, 5\alpha, 6\alpha)$, $(3\beta, 5\alpha, 6\alpha)$ -1.1-Diisopropyl-3-hexyl-2-oxa-1-sila[4.2.0] bicyclooctane (6-Pr):	7
(<i>trans</i>)-1,1-Diisopropyl-3-hexyl-6-isopropyl-2,1-oxasilinane (8-Pr):	8
$B(C_6F_5)_3$ -catalyzed tandem silvlation/hydrosilvlation (general procedure)	8
1.1-Diethyl-3-hexyl-5-methyl-2.1-oxasilinane (2-Et):	9
1.1-Diphenyl-3-hexyl-5-methyl-2.1-oxasilinane (2-Ph):	9
$(3\beta 5\alpha 6\alpha)$ and $(3\alpha 5\alpha 6\alpha)$ -2 1-Benzoxasilin octahydro-1 1-diethyl-3-hexyl (10-Et)	9
$(3\beta,5\alpha,6\alpha) = 21$ Benzovasilin octahydro 1 1-dinhenyl-3 heyyl ((3\beta,5\alpha,6\alpha) = 10. Ph) and	4
$(3\alpha, 5\alpha, 6\alpha) = 2,1$ Benzovasilin, octanyuro, 1,1-diphenyl 2 hexyl $((3\alpha, 5\alpha, 6\alpha) = 10$ H) and $(3\alpha, 5\alpha, 6\alpha) = 2,1$ Benzovasilin, octahydro, 1,1 diphenyl 2 hexyl $((3\alpha, 5\alpha, 6\alpha) = 10$ H)	10
(30, 50, 00)-2,1-Delizoxasilili, octaliyulo, 1,1-ulpiteliyi-5-liexyi ((30, 50, 00)-10-Fii)	10
(5,0-trans) and $(5,0-trs)-1,1-Dietry 1-3-nexy 1-0-metry 1-2,1-0x as minimize (trans 12 Et and (sis) 12 Et)$	11
(<i>Hulls</i> -12-EL allu (<i>Cls</i>)-12-EL (2.6. turns) and (2.6. sis) 1.1 Dimbonyl 2 howel 6 mathed 2.1 sussiling as	11
(5,0-irans) and $(5,0-cis)-1,1-D$ phenyi-5-nexyi-0-methyi-2,1-oxasiiinane (trans 12 Db and (ais) 12 Db):	10
(ITAHS-12-TH and (Cls)-12-TH):	12
(5,0- <i>irans)</i> and (5,0- <i>cis)</i> -1,1-Diethyi-5-nexyi-6-methyi-2-oxa-1-silepane	

(<i>trans</i> - and <i>cis</i> -14-Et).	12
(3,5- <i>trans</i>)- 1,1-Diethyl-5-methyl-3-phenyl-2,1-oxasilinane (16-Et):	13
Diethyl (2-methylnon-1-en-3-oxy)silane (18-Et):	
Oxidative Desilylation:	13
(Tamao oxidation) 2-Methylundecane-1,5-diol (19):	14
(Woerpel oxidation) 2-(2-Hydroxyoctyl)-cyclohexanol (20):	14
3-Methyl-1-phenyl-1,4-butanediol (21)	15
References	15

General Experimental Procedures:

Tetrahydrofuran (THF) was distilled from Na/Ph₂CO under N₂. Dichloromethane (CH₂Cl₂) was distilled from CaH₂. Stock solutions of $B(C_6H_5)_3$ were prepared either by: A) Transferring a freshly opened commercial sample (typically 1.0 g) into a oven-dried flask under N₂ followed by dissolution (0.2 M) in freshly distilled (Na/Ph₂CO) toluene; or B) Working inside a glove box under inert atmosphere, dividing a 1.0 g commercial sample of $B(C_6F_5)_3$ into individual vials (ca. 200 mg/vial). The vials were removed from the glove box to prepare stock solutions in toluene (0.2 M) that were used immediately and then discarded. ⁱ $B(C_6H_5)_3$ dissolves completely in toluene at 0.2M; the solubility of the hydrate is significantly lower. All other reagents and solvents were used as purchased unless otherwise noted. Thin layer chromatography (TLC) was performed on 0.25 mm hard-layer silica G plates; developed plates were visualized with a handheld UV lamp and/or by staining with one of the following: 1% ceric sulfate and 10% ammonium molybdate in 10% H₂SO₄ (general stain, after charring) or 1% aq. KMnO₄ (for alkenes). Analytical and preparative HPLC werer performed on a 4.6 mm x 25 cm Si column (5 μm) or 21.4 mm x 25 cm Si column (8 μm); both employed RI detection. NMR spectra were recorded at 400 MHz (¹H) or 100 MHz (¹³C) in CDCl₃ unless otherwise indicated. ¹H NMR signals are reported as: [chemical shift (multiplicity, integration, J couplings in Hz, other information). Infrared spectra were recorded as neat films (ZnSe crystal or NaCl plates) with selected absorbances reported in wave numbers (cm⁻¹). High resolution mass spectrometry was conducted at the Nebraska Center for Mass Spectrometry.

Preparation of Alcohols:

2-Methyldec-1-en-4-ol (1): Into a 0 °C solution of heptanal (3.5 mL, 25 mmol) in THF (10 mL) was added dropwise a solution of 2-methylallyl magnesium chloride in THF (50 mL, nominally 0.5 *M*). After 20 min, the reaction was quenched with water (20 mL), acidified with conc. HCl (~

3 mL) and extracted with 10% EA/Hex (250 mL x 2). The combined extracts were sequentially washed with 10% aq. HCl and water. A standard workup and purification (5% EA/Hex) furnished 3.53 g (82%) of a compound with spectral properties matching literature reports.ⁱⁱ

2-Phenyldec-1-en-4-ol (**3**)ⁱⁱⁱ was prepared by ene reaction of heptanal with α -methylstryrene by the procedure of Snider: ^{iv} ¹H δ 7.44-7.41 (2H); 7.38-7.33 (2H); 7.32-7.27 (2H); 5.43 (d, 1H, 1.6); 5.18 (bs, 1H); 3.66 (m, 1H), 2.84 (ddd, 1H, 14, 4, 1; AB with 2.67), 2.67 (dd, 1H, 14, 9); 1.69 (d, 1H, 3); 1.4-1.5 (3H), 1.2-1.3 (6 H), 0.88 (t, 3H, 6.5); 13C δ 145.5, 140.5, 128.4, 127.7, 126.2, 115.2, 69.4, 43.8, 37.0, 31.8, 29.3, 25.6, 22.6, 14.1; IR 3368 (s, b); 2927, 2856, 1626, 1444, 898, 705 cm⁻¹; HRFAB MS calc. For C₁₆H₂₄OLi (M+Li)⁺: 239.1984; found 239.1984.

1-Cyclobutenyloctan-2-ol (**5**) was prepared from methylenecyclobutane (1.0 g, 15 mmol) and heptanal (2.7 mL, 1.3 equiv) by a similar procedure as for **3** to afford 0.99 g of alcohol **5** as an inseparable 3:1 mixture with 1-chlorocyclobutyl-2-octanol. The spectra of the product ($R_f = 0.3$, 10% EA) matched a literature report.^v

2-Methyl-2-dodecen-6-ol (7) was prepared (1.39 g, 71%) by reaction of the Grignard reagent derived from 5-bromo-2-methyl-2-pentene (2.0 mL, 15 mmmol) with a slight exces of heptanal: R_f =0.4 (5% EA/hex); ¹H δ 5.14 (bt, 1H, 6); 3.60 (m, 1H); 2.09 (m, 2H); 1.69 (bs, 3H); 1.63 (bs, 3H); 1.55-1.38 (6H); 1.35-1.23 (6H); 0.89 (t, 3H, 7); ¹³C δ 132.0; 124.2; 71.8; 37.5, 37.3, 31.8, 29.4, 25.7, 25.6, 24.4, 22.6, 17.6, 14.1; IR: 3377 (b, s, OH); 3328, 2924, 2855, 1454, 1377 cm-1; 2928, 2864, 2092, 1463, 1379, 1056, 1001, 837 cm⁻¹; HRFAB calculated for C₁₃H₂₅O (M-H)⁺: 197.1905; found 197.1912 (-5.4 ppm); M⁺ also observed at 196.1813.

1-Cyclohexenyloctan-2-ol (9) was prepared from the reaction of heptanal (1.12 mL, 8.0 mmol), methylenecyclohexane (1.12 mL, 10 mol) and Me₂AlCl (12 mL, nominally 1 M solution in hexanes) by a similar procedure as for **3**. The product (1.23 g, 73%) displayed spectra consistent with literature reports. ^{vi} $R_f = 0.4$ (10% EA/Hex

(*E*,*Z*)-**Undec-2-en-5-ol** (**11**) was prepared (1.17 g, 86%) from prop-1-enylmagnesium bromide (24 mL, nominally 0.5M in THF), epoxyoctane (1.2 mL, 8 mmol) and CuI(0.152g, 0.8 mmol).

The product was a 3.3:1 mixture of *E*- and *Z*-isomers based upon integration of the ¹H signals at δ 1.64 and 1.69 ppm. Spectral properties matched literature reports.^{vii} R_f =0.4 (10% EA/Hex).

2-Methylundec-1-en-5-ol (13) ^{viii} was prepared (1.45 g, 98%) from 2-methylallyl magnesium chloride (24 mL, nominally 0.5M in THF), 2-hexyloxirane (1.2 mL, 8 mmol) and CuI(0.152g, 0.8 mmol) by a similar manner as 15. $R_f = 0.3$ (10% EA/Hex).

3-Methyl-1-phenylbut-3-en-1-ol (15) was prepared (1.30 g, quant.) from reaction of benzaldehyde (0.85 mL, 8.0 mmol) and 2-methylallyl magnesium chloride (20.8 mL, nominally 0.5M in THF) by a procedure similar to that applied for **1**. Spectral properties matched a literature report.^{ix} $R_f = 0.3$ (10% EA/Hex)

2-Methylnon-1-en-3-ol $(17)^{x}$ was prepared (1.05 g, 84%) from heptanal (1.12 mL, 8.00 mmol) and allylmagnesium chloride (20.8 mL, nominally 0.5M solution in THF) by a similar procedure as used for **1**. R_f =0.2 (10% EA/Hex). Spectral properties matched a literature report.

Alkoxysilanes:

Diisopropyl(2-methyldec-1-en-4-oxy)silane (1-Pr)

Into a THF (15 mL) solution of 2-methyldec-1-en-4-ol (0.724 g, 4.30 mmol) was added sodium bis(trimethylsilyl)amide (4.3 mL, nominally 2*M*) followed by SiClH(*i*-Pr)₂ (1.1 mL) . After 4 h the reaction was quenched with brine and extracted with hexane (2 x 200 mL). The combined organic extracts were concentrated *in vacuo* and the residue was purified by flash chromatography in hexane to afford 1.09 g (89%) of the silyl ether: $R_f = 0.4$ (hexane); ¹H (600 MHz) δ 4.78 (app. s, 1H), 4.78(app. s, 1H), 4.21(s, 1H), 3.85(p, 1H, 5.7), 2.27(dd, 6 and 13.2), 2.17(dd, 1H, 7.2 and 13.2), 1.75(s, 3H), 1.43 (m, 10H), 1.05(m, 12H), 0.95(m, 2H), 0.90(t, 3H, 6.0); ¹³C (150 MHz) δ 142.9, 112.8, 73.0, 45.7, 36.7, 31.9, 29.4, 25.2, 22.9, 22.6, 17.60, 17.57, 17.46, 14.1, 12.73, 12.71; IR 2927, 2862, 2097, 1642, 1463, 1377 cm-1. HRMS (CI) calc. for C₁₇H₃₅OSi (M-H)⁺: 283.2457; found 283.2469 (4.2 ppm); M⁺ (284.2543) observed in lower abundance.

Diisopropyl-(2-phenyldec-1-en-4-oxy)silane (**3-Pr**) was prepared (0.435 g, 70%) from alcohol **3** (0.428 g, 1.8 mmol) by a procedure similar to that applied to the synthesis of **1-Pr**: $R_f = 0.9$ (5% EA/hex); ¹H δ 7.42 (bd, 2H, 8), 7.34 (bt, 2H, 8), 7.72 (app tt, 1H, 8, 1); 5.32 (d, 1H, 1.6), 5.13 (bs, 1H); 4.18 (bt, 1H, 1.6), 3.76 (m, 1H), 2.81 (ddd, 1H, 14, 5.8, 1); 2.64 (ddd, 1H, 14, 6.4, 1); 1.55-1.36 (3H), 1.33-1.18 (7H); 1.04-0.98 (12H, overlapping Me doublets); 0.98 -0.91 (m, 2H); 0.88 (t, 3H, 6.4); ¹³C δ 145.7, 141.2, 128.2, 127.3, 126.3, 114.9, 72.9, 43.2, 36.5, 31.8, 29.4, 24.9, 22.6, 17.54, 17.48, 14.41, 14.07, 12.64, 12.61; IR: 3031, 2954, 2865, 3095, 1462, 1254 cm⁻¹; HRFAB Calc. For C₂₂H₃₇OSi (M-H)⁺: 345.2613; found 345.2605 (2.5 ppm).

Diisopropyl(1-cyclobutenyloctyl-2-oxy)silane (5-Pr) was prepared in 55% yield (485 mg) from **5** (546 mg, estimated 2.25 mmol based upon purity) by a similar porocedure as for **1-Pr**: $R_f = 0.3$ (hexane); ¹H δ 5.72 (s, 1H); 4.20 (s, 1H); 3.81 (apparent pentet, 1H, 5-6); 2.45 (m, 2H), 2.35 (bs, 2H); 2.22 (m, 2H); 1.5-1.23 (10H); 1.07-1.02 (12H, isopropyl groups); 1.02-0.95 (2H); 0.895 (t, 3H, 6.5); ¹³C δ 147.3, 129,3, 73.1, 38.8, 36.9, 32.0, 31.9, 29.4, 27.0, 25.3, 22.6, 17.6, 17.5, 17.4, 14.1, 12.7; IR 2926, 2864, 2089, 1462, 1055, 837 cm⁻¹; HRFAB Calc. For C₁₈H₃₅OSi (M-H)⁺: 295.2457; found 295.2452 (1.7 ppm).

Diisopropyl-(2-methyl-2-dodecen-6-oxy)silane (7-Pr) was prepared (0.707g, 73%) from alcohol 7 (617 mg, 3.1 mmol) by a similar procedure as for 1-Pr: $R_f = 0.3$ (hexane); ¹H δ 5.12 (bt, 1H, 6-7), 4.21 (s, 1H), 3.69 (pentet, 1H, 6.4); 2.08 & 1.98 (ABXY, 2H), 1.69 (s, 3H); 1.62 (s, 3H); 1.53-1.45 (4H), 1.35-1.25 (8H), 1.08-1.02 (12H, isopropyl), 1.02-0.95 (2H); 0.90 (t, 3H, 6); ¹³C δ 131.4, 124.5, 74.3, 36.9, 36.8, 31,9, 29.5, 25.7, 25.3, 24.0, 22.6, 17.64, 17.59, 17.5, 14.1, 12.7; IR 2929, 2864, 2088, 1463, 1377, 1063, 1002, 841, 800 cm⁻¹; HRFAB Calc. For C₁₉H₃₉OSi (M-H)⁺: 311.2770; found 311.2773 (1.0 ppm).

<u>General Procedure for intramolecular hydrosilylation</u> (illustrated for **1-Pr**). To an anhydrous toluene solution (6 mL) of **1-Pr** (0.285g, 1.00 mmol), either at -78, 0 °C, or rt, was added $B(C_6F_5)_3$. The amount of catalyst ranged from 0.1 to 1.0 eq, as a 0.2M solution in toluene. After the reaction was complete (TLC), the reaction was quenched with sat. aq. NaHCO₃ (5 mL) and the resulting mixture extracted with hexane (2 X 100 mL). The combined organic layers were concentrated *in vacuo* and the residue was purified by flash chromatography (hexane) to afford

trans-**2-Pr** (236 mg, 83%) followed by a small amount of *cis*-**2-Pr** (18 mg, 6%). Analysis of the crude reaction mixtures by GC/MS generally found 91-95% of the *trans* isomer; the minor (syn) byproduct eluted first on GC. Both diasteromers displayed a predominant fragment at m/z 241, [M-iPr]⁻. The stereochemistry was assigned based upon the relative strength of nOe transfer in the *trans* and *cis* isomers (see Scheme below), and by the magnitude of the axial/axial and axial/equatorial couplings for ${}^{3}J_{5-6}$ couplings. The stereochemical assignment was supported by a correlation of the ${}^{3}J_{H}$ of the minor (*cis*) byproduct with a literature report for similar molecules .^{xi}

(3,5-trans)- 1,1-Diisopropyl-3-hexyl-5-methyl-2,1-oxasilinane (trans-2-Pr)

 $R_f = 0.2$ (hexane); ¹H δ 3.99(m, 1H), 1.96(m, 1H), 1.57(m, 1H), 1.38(m 11H), 0.99(m, 12H), 0.95(m, 4H), 0.88[m, 4H, includes 0.89(t, 3H, 6.8), and peak at 0.90], 0.80(ddd, 1H, J₁=1.6, J₂=4.8, J₃=14.8), 0.28(dd, 1H, 10.4 and 14.8); ¹³C δ 72.0, 41.3, 37.5, 31.9, 29.4, 26.3, 26.1, 23.8, 22.6, 17.28, 17.27; 17.19, 17.16, 15.1, 14.1, 13.7, 13.1; IR (2942,



2931, 2864, 1464 cm⁻¹; HRFABMS (3-NBA) calc. for $C_{14}H_{29}OSi [M-(i-Pr)^+]$: 241.1988; found 241.1992 (1.7 ppm). Diaxial couplings and nOe excitations are summarized in the accompanying graphic.

(3,5-cis)- **1,1-Diisopropyl-3-hexyl-5**methyl-2,1-oxasilinane (cis-2-Pr):

 $R_f = 0.4$ (hexane); ¹H δ 3.76(m, 1H), 1.77(m, 1H), 1.51(d of q., 1H, 2 and 13.6), 1.38(m 11H), 1.04-0.94 [m, 15H, peak at 1.00 (d, 6.5) visible nOe upon irradiation at 0.21], 0.89[m, 5H,



includes 0.89(t, 3H, 6.8), and other peaks], 0.72(ddd, 1H, $J_1=2.4$, $J_2=4.0$, $J_3=14.4$), 0.21(dd, 1H, 12.8 and 14.4); ¹³C δ 74.6, 44.3, 39.1, 31.9, 29.7, 29.4, 27.5, 25.3, 22.7, 17.71, 17.68, 17.14, 17.10, 15.7, 14.1, 13.1, 12.3; IR identical to *anti-2-Pr*. Diaxial couplings and nOe excitations are summarized in the accompanying graphic.

The ¹H NMR spectra of *cis*-3,5-disubstituted 2,1-oxasilacyclohexanes display H_3 (axial) as a ddd between 3.45 and 3.7 ppm and with individual coupling contants of up to 11 Hz.¹¹ The same work found the ²J coupling for H_6/H_6 ' to be 14 Hz, and the axial/axial and equatorial/axial ³J ₅₋₆ couplings to be 13.3 and 3.5 Hz, respectively. These values agree closely with our observations for *cis*-**2-Pr**.

(3,5-trans)- 1,1-Diisopropyl-3-hexyl-5-phenyl-2,1-oxasilinane (trans-4-Pr):

By a procedure similar to that described for **1-Pr**, cyclization of silane **3-Pr** (0.299 g, 0.862 mmol) furnished 0.257 g (86% yield) of **4-Pr**: $R_f = 0.3$ (5% EA/hex); ¹H δ 7.34 (t, 2H, 7.5); 7.27 (bd, 2H, 7.5); 7.22 (bd, 1H, 7.5); 4.16 (m, 1H); 3.05 (bt, 1H, 12.5), 1.96 (near dt, 12, 6; on same CH₂ as 1.75; 1.77 (m, 1H, on same CH₂ as 1.55); 1.75 (m, 1H, on same CH₂ as 1.96); 1.55 (m, 1H, on same CH₂ as 1.78), 1.5 (m, 1H, on same CH₂ as peak buried at 1.32), 1.4-1.26 (7H, includes portion of CH₂ shared with 1.5 as well as three CH₂-related spin systems), 1.12 (d, 3H, 7), 1.10 (3H, obscured), 1.09 (d, 3H, 7), 1.02 (6H, broad s), 0.98 (m, 1H, part of CH₂ with 0.90); 0.92 (t, 3H, 7), 0. 90 (partially obscured dd, 20, 14). Through-space (nOe) correlations: Excitation of 4.16: collapses 1.96 to dt (6,14) as H₄ equatorial); enhances 1.77, 1.75, 1.5, 1.32; Excitation of 3.05 enhances 7.27, 1.77, 1.5, (shows evidence of direct coupling to 1.96); enhances d at 1.11, 1.09, and m/bs at 1.00 and 0.08; enhances methyl at 0.89? ¹³C 149.8, 128.5, 126.4, 125.9, 73.0, 40.7, 37.0, 34.4, 31.9, 29.4, 26.5, 22.7, 17.34, 17.27, 17.21, 17.16, 15.7, 14.1, 13.9, 13.0; HR-FAB calcd. C₂₂H₃₈OSi (M-H)⁺: 345.2613; found: 345.2605 (2.5 ppm)

1,1-Diisopropyl-3-hexyl-2-oxa-1-sila[**4.2.0**] **bicyclooctane** (**6-Pr**) was prepared (65 mg, 32%) as a separable mixture of diasteromers by cyclization of **5-Pr** (200 mg, 0.67 mmol).

Diastereomer 1 (3α , 5α , 6α); 54 mg; R_f = 0.3 (hexane); ¹H δ 3.58 (m, 1H, methine, cross speaks to spin system centered on 1.5 ppm); 2.51 (m, 1H, methine, coupled to 2.38, 1.95, 1.57); 2.38 (apparent pentet, 1H, part of methylene, coupled to 2.51, 1.99, 1.95, 1.68, 1.57); 1.99 (m, 1H, methine); 1.95 (m, 1H, part of methylene); 1.68 (dd, 1H, 13, 6); 1.57 (m, 1H); 1.55-1.25 (11H);

1.04-9.97 (12H, 4 Me in isopropyl); 0.898 (t, 3H, 6.4), 0.88 (m, 1H), noE from 3.58 reveals as apparent pentet); ¹³C δ 73.18, 29.17, 38.61, 36.28, 21.97, 30.92, 29.41, 25.32, 26.69, 21.96, 18.57, 17.64, 17.53, 17.47, 17.07, 14.12, 13.44, 12.99; IR 2926, 2802, 1463, 1131, 1040, 882 cm⁻¹.

Diastereomer 2 (3β,5α,6α): 11.6 mg; $R_f = 0.2$ (hexane); ¹H δ 4.11 (m or apparent heptet, 1H); 2.825 (m, 1H; coupled into 2.2, 1.92, 1.47; correlates with methine C at 32.6 ppm); 2.24 (m, 1H; correlates with methylene C at 26 ppm), 2.13 (m, 1H, correlates with methylene at 26 ppm)), 2.06 (m, 1H, correlates wth methylene at 20 ppm), 1.92 (m, 2H correlates with methylene at 26 and methine at 17); 1.47 (m, 4H), 1.4-1.25 (9H), 1.15-1.05 (1H); 1.07 (m, 3H0, 1.03 (app d, 3H, 6.4), 0.997 (d, 3H, 6.7); 0.94 (d, 3H, 6.7); 0.83 (m, 1H); ¹³C δ 69.55, 38.79, 37.78, 32.64, 31.97, 29.44, 26.28, 25.71, 22.68, 20.07, 17.53, 17.43, 17.33, 17.13, 16.24, 14.10, 13.18, 13.12; IR 2927, 2803, 1464, 1092, 993, 882 cm-1; HRMS calcd. for C₁₈H₃₅OSi (M-H)⁺: 295.2457; found: 295.2456 (6.4 ppm).

(*trans*) **1,1-Diisopropyl-3-hexyl-6-isopropyl-2,1-oxasilinane (8-Pr)** was prepared (87 mg, 20% yield) from from **7-Pr** (419 mg, 1.34 mmol) by a similar procedure (0.2 equiv B(C₆F₅)₃) as for **2-Pr**, except that the reaction was warmed to 0° C and held at that temperature for 16 h. Following a careful flash chromatography to remove a large amount of byproduct, the product was isolated as a single product by NMR and GC/MS: $R_f = 0.5$ (hexane); ¹H δ 3.68 (m, 1H; C3-axial), 2.05 (dtd, 1H, 12.8, 5, 2.4; H₅-*eq*, HMQC shows relationships to 1.35; COSY shows couplings to H₅-axial, H₄-axial, H₆); 1.69 (m, 2H; 1H includes H₄-axial; linked by COSY to H₄-*eq* at 1.15); 1H is CH of C₆ sidechain, with correlations to C₆ and sidechain methylenes); 1.5-1.25 (9H, includes: 1.35 m for H₅-axial; multiple spin systems from sidechain CH₂ groups); 1.15 (m, 1H, C₄-eq); 1.11 (d, 3H, 6-7, iPrSi); 1.09 (d, 3H, 6-7, iPrSi); 1.10 (m, 1H, CH), 1.05 (d, 3H, -7, iPrSi); 1.00 (d, 3H, ~7, iPrSi); 0.96 and 0.93 (each d, 3H, J ~6.5, Me₂CHC₆); 0.89 (t, 3H, 4-5, Me); 0.63 (ddd, 13, 9, 5, H₆, COSY to iPrCH at 1.7); GC-MS: single major peak at 28.17 min (269, [M-iPr]); ¹³C δ 74.6, 39.2, 36.4, 32.2, 31.6, 29.6, 28.8, 27.6, 25.7, 24.8, 22.9, 22.0. 19.8, 18.6, 18.1, 17.8, 14.4, 13.8, 13.1; IR 2927, 2865 (s); 1464, 1382, 1068 cm⁻¹; HRFAB calc for C₁₉H₃₉OSi (M-H)⁺: 311.2770; found 311.2783 (3.9 ppm); (M+H)⁺ at 312.2818 also observed. <u>General procedure for tandem silylation/hydrosilylation:</u> Into a solution of unsaturated alcohol (typically 1 mmol) in 6 mL anhydrous toluene was added diethylsilane or diphenylsilane (1.2 mmol). The solution was cooled to 0 °C and $B(C_6F_5)_3$ was added (typically 0.1-0.5 equiv) from a 0.2-0.3 M stock solution in anhydrous toluene, resulting in vigorous bubbling. Once the alkene had largely disappeared (TLC), the reaction was quenched with 10% aq. NaHCO₃ (30 mL). The resulting mixture was extracted with hexane (2 x 50 mL) and the crude products were purifed by flash or column chromatography.

(3,5-trans)-1,1-Diethyl-3-hexyl-5-methyl-1-oxa-2-silinane (2-Et)

Using the tandem procedure described above, alcohol **1** (0.34 g, 2.0 mmol) was reacted with diethylsilane(0.33 mL, 2.6 mmol). TLC indicated that the reaction was completed within 5 minutes. Column chromatography using 0-5% EA/hex as the eluting solvent afforded 0.24g (47%) of the silacyclohexane. A small portion of the product was purified by semi-preparative HPLC (21 x 250mm, 5 mL/minute of 1% EA/hex): $R_f = 0.58$ (5% EA/hex); ¹H δ 3.93-3.99(1H), 1.98-2.06(1H), 1.55-1.59(1H), 1.36-1.50(4H), 1.27(7H, m), 1.01(3H, d, 6.7), 0.94(3H, t, 6.4), 0.94(3H, t, 7.9), 0.88(3H, t, 6.8), 0.72(1H, ddd, 1.4, 4.6, 14.5), 0.57(4H, q, 7.5), 0.33(1H, dd, 10, 14.5); ¹³C δ 72.0, 41.4, 37.3, 31.9, 29.4, 26.3, 25.7, 24.1, 22.6, 17.7, 14.1, 7.6, 6.9, 6.7, 6.5; IR: 2953, 2925, 2874, 1458, 1413, 1156, 1047, 1003, 762 cm⁻¹. HR-FABMS calcd. for $C_{15}H_{32}O[M+H]^+$: 257.2301; Found: 257.2300

(*3*,*5*-*trans*)-**1**,**1**-**Diphenyl-3**-**hexyl-5**-**methyl-1**-**oxa-2**-**silinane** (**2**-**Ph**) was prepared (0.59 g, 84%) from alcohol **1** (0.34 g, 2.0 mmol) and diphenylsilane (0.41 mL, 2.2 mmol) using the tandem procedure described above. The reaction was conducted for 10 min and the crude product wa purified by gradient flash chromatography (0-5% EA/hex). A small portion of the product was purified by semi-preparative HPLC (21x250 mm, 5 mL/min of 1% EA/hex). The major product was assigned by comparison with **1**-**Pr**: $R_f = 0.23$ (2% EA/hex); ¹H δ 7.51-7.61(5H), 7.30-7.42(5H), 4.16-4.21(1H), 2.24-2.28(1H), 1.27-1.63(15H), 1.03(3H, d, 6.8), 0.87(3H, t, 6.8), 0.79-0.95(2H); ¹³C δ 137.26, 137.24, 134.2, 134.1, 129.63, 129.57, 127.78, 127.73, 72.2, 41.6, 37.5, 31.8, 29.2, 26.1, 25.0, 24.8, 22.6, 19.0, 14.1; IR: 3068, 3049, 3000, 2954, 2925, 2856, 1454, 1428, 1151, 1116, 1041, 997, 821, 756, 731, 699 cm⁻¹; HR FABMS calcd. for C₂₃H₃₃OSi [MH]+: 353.2307; found: 353.2300 (1.7 ppm).

(3 β ,5 α ,6 α) and (3 α ,5 α ,6 α) **2,1-Benzoxasilin, octahydro-1,1-diethyl-3-hexyl (10-Et)**: Using the tandem procedure described above, alcohol **9** (0.21g, 1.00 mmol) was reacted with diethylsilane (0.17 mL, 1.3 mmol) for 30 min, to furnish, after standard workup and chromatography, 0.11 g (39%) of the oxasilane. A small portion of the product was purified by semi-preparative HPLC (21x250 mm, 5 mL/min of 1% EA/hex) to furnish a 5:1 mixture of C₃ epimers. Traces of several minor components were visible (RI detection) just befor elution of the major product: R_f =0.34 (2% EA/hex); ¹H δ 3.87(1H, m), 1.95-1.99(1H), 1.61-1.78(3H), 1.27-1.52(17H), 1.13(1H, q, 5.2), 0.97(3H, t, 8.0), 0.95(3H, t, 8.0), 0.88(3H, t, 6.8), 0.69-0.78(1H), 0.55-0.66(3H). ¹³C δ 70.0, 38.6, 38.1, 33.5, 31.9, 31.5, 29.4, 26.0, 25.4, 25.0, 24.7, 24.5, 22.7, 14.1, 6.8, 6.7, 6.2; IR: 2852, 1459, 1413, 1377, 1237, 1187, 1127, 1097, 1059, 1004, 972, 934, 802, 724 cm⁻¹; MS: HR-FAB: calcd. for C₁₈H₃₆O₂[M-H]⁺: 295.2456; found: 295.2448. The stereochemistry of the major product was assigned in analogy with **10-Ph** (below) and by the chemical shifts for the axial H₃-axial (3.7 ppm) in the trans/cis isomer (major) vs. the equatorial H₃ (3.9 ppm) in the cis/cis isomer (minor).

$(3\beta,5\alpha,6\alpha)$ and $(3\alpha, 5\alpha,6\alpha)$ -2,1-Benzoxasilin, octahydro, 1,1-diphenyl-3-hexyl (10-Ph)

Using the tandem procedure described above, alkenol **9** (0.22g, 1.1 mmol) was reacted with in diphenylsilane(0.21 mL, 1.1 mmol) for 1 h, to furnish, after standard workup and a gradient flash chromatography (0-5% EA/hex), 0.30 g (73%) of the cyclic oxasilane as a 1:5 mixture (NMR) of the *cis/cis* and *trans/cis* isomers, differing in the stereochemistry at C₃. $R_f = 0.50$ (2% EA/hex); HR FABMS calc. for C₂₆H₃₇OSi [MH]⁺: 393.2613; found: 393.2629 (3.8 ppm). A small portion of the product was further purified by semi-preparative HPLC (21 x 250mm, 5 mL/min of 1% EA/hex); the minor product eluted first.

cis/cis $(3\alpha, 5\alpha, 6\alpha)$ - (minor) ¹H δ 7.65-7.69(2H), 7.49-7.51(2H), 7.28-7.45(6H), 3.84-3.89(1H; COSY correlation with spin systems at δ 2.1, 1.96; weak correlation with δ 1.5; nOe observed to 2.1 and 1.2); 2.05-2.10(1H, correlates only with δ 1.96), 1.93-1.99(1H, correlates with δ 3.9, 2.05, 1.2), 1.42-1.71(12H), 1.19-1.39(10H), 1.2 (1H, obscured t or dd, correlates with 1.97, 1.7); 0.90(3H, t, 6.8); ¹³C δ 135.6, 134.5, 134.3, 129.6, 129.5, 127.9, 127.5, 74.7, 38.8, 35.3, 34.5, 33.5, 32.0, 29.4, 27.9, 25.4, 22.9, 22.7, 22.3, 21.2, 14.2; IR 3068, 3048, 3000, 2925, 2855, 1447, 1428, 1142, 1116, 1092, 1055, 1009, 970, 924, 801, 736, 710, 699 cm⁻¹. HRFAB calc. for C₂₆H₃₇OSi (MH)⁺: 393.2613; found: 393.2629 (3.8 ppm).

trans,cis (3 β ,5 α ,6 α) (major) ¹H δ 7.65 (m, 2H; nOe to 1.7), 7.55 (2H; modest nOe to 4.3), 7.31-7.41(6H), 4.31(1H, m, H₃; correlates with 1.6, 1.5; significant noE to d 7.7, 2.1; this proton appears to be significantly <u>de</u>shielded by the edge of the neighboring arene; this assumption is supported by the observation of mutual nOes involving the arene as well as by MM2 calculations; 2.20 (m, 1H; COSY crosspeaks wth 2.1, 1.5-1.6; nOE to 1.73, 1.5, 1.4), 2.07 (ddd, 1H, J values estimated as 13-14, 8, 3-4;COSY with 2.2, 1.5; nOE to peaks at d 4.3, 1.4), 1.83(m, 1H; correlates to 1.45; nOe with d 1.5, 1.3), 1.73(dt, 1H; weak COSY with 1.83; nOE to 2.2, 1.5), 1.17-1.48(20H), 0.87(3H, t, 6.8); ¹³C δ 137.8, 136.7,134.6, 134.3, 134.2, 129.5, 129.3, 127.8, 127.7, 127.6, 72.4, 37.7, 37.6, 32.6, 31.8, 29.5, 29.3, 26.1, 25.3, 24.8, 24.5, 24.0, 22.6, 14.1; IR 3068, 3048, 3022, 2999, 2920, 2851, 1590, 1486, 1447, 1428, 1376, 1187, 1114, 1057, 997, 938, 916, 821, 801, 772, 699 cm⁻¹.

(3,6-*trans and* 3,6-*cis*- **1,1-Diethyl-3-hexyl-6-methyl-2-oxa-1-silinane** (*trans*- and *cis*-**12-Et**): Using the tandem procedure describe above, alkenol **11** (0.34 g, 2.0 mmol) was reacted with Et₂SiH₂ (0.33 mL, 2.6 mmol) for 5 minutes to furnish, following standard workup and chromatography (0-5% EA/hex), a 2.7: 1 mixture of *trans*- and cis-**12-Et** (81.9 mg, 16%): $R_f = 0.41$ (2% EA/hex); HREIMS calc. for $C_{15}H_{31}OSi$ (M-H)⁺: 255.2244; found 255.2142 (0.9 ppm). A small portion of the product was purified by semi-preparative HPLC (21 x 250 mm, 5 mL/min 1% EA/hex); the major (trans) and minor (cis) isomers elute at 16 and 17 min, respectively. The assignment of *cis*- and *trans* oxasilanes was based upon the upfield ¹H chemical shift for the axial H₃.

trans-12: ¹H δ 3.60-3.65(1H), 1.83-1.89(1H), 1.61-1.66(1H), 1.20-1.47(12H), 0.99(3H, t, 7.9), 0.98(3H, t, 7.9), 0.91(3H, t, 6.8), 0.88(3H, t, J-6.8), 0.49-0.84(4H); ¹³C δ 74.5, 38.7, 35.8, 32.7, 31.9, 29.4, 25.5, 22.7, 17.5, 15.7, 14.1, 6.7, 5.0, 1.7; IR 2954, 2927, 2876, 2858, 1461, 1377, 1236, 1087, 1042, 1014, 836, 724 cm⁻¹.

cis-**12**: ¹H δ 3.70-3.76(1H), 1.82-1.90(1H), 1.62-1.69(1H), 1.24-1.56(14H), 1.02(3H, d, 7.6), 0.97(3H, t, 7.9), 0.96(3H, t, 7.9), 0.88(3H, t, 6.8), 0.51-0.74(4H); ¹³C δ 74.2, 38.1, 31.9, 30.8, 29.8, 29.3, 25.8, 22.7, 15.0, 14.4, 14.1, 6.8, 6.3, 4.2, 4.1; IR 2953, 2927, 2874, 1460, 1413, 1377, 1235, 1161, 1138, 1088, 1044, 1005 cm⁻¹.

(*trans*) and (*cis*)- **1,1-Diphenyl-3-hexyl-6-methyl-1-oxa-2-silacyclohexane** (*trans*- and *cis*-**12-Ph**). Using the tandem procedure described above, alcohol **11** (0.34g, 2.0 mmol) was reacted with Ph₂SiH₂ (0.41 mL, 2.2 mmol) for 1 h. The crude products were subjected to column chromatography using 0-5% EA/hex as the eluting solvent to afford 0.17g (24%) of the oxasilacyclohexane. A small portion of the product was further purified by semi-preparative HPLC (21x250mm, 1% EA/hexane, 5 mL/min), which partially resolved the major and minor isomers. The predominant isomer was assigned as trans on the basis of the 3.85 ppm chemical shift for the axial H₃: R_f =0.32 (2% EA/hex); ¹H (mixture of diastereomers) δ 7.65-7.68(m), 7.53-7.55(m), 7.30-7.44(m) 4.02-4.07(m), 3.85-3.90(m), 1.94-2.04(m), 1.78-1.86(m), 1.39-1.74(m), 1.13(d, 3H, 7.2), 1.07(d, 3H, 8), 0.87-0.92(m); ¹³C (mixture of two diastereomers) δ 136.0, 135.1, 134.5, 134.4, 134.3, 129.8, 129.6, 129.5, 127.9, 127.8, 127.7, 127.6, 75.8, 74.8, 38.8, 38.6, 35.9, 32.3, 31.9, 30.2, 30.1, 29.39, 29.36, 25.5, 25.4, 22.7, 19.4, 16.3, 15.3, 14.1, 13.9; IR: 3069, 3048, 2927, 2857, 1457, 1428, 1117, 1042, 987, 933, 736, 700 cm⁻¹; HRFAB calc. for C₂₃H₂₃OSi (M+H)⁺: 353.2300; found: 353.2293 (2.1 ppm).

(3,6-*trans*) and (3,6-*cis*)-**1,1-Diethyl-3-hexyl-6-methyl-2-oxa-1-silepane** (*trans-* and *cis-***14-Et**). Using the tandem procedure described above, alcohol **13** (0.360 g, 1.95 mmol) was reacted with diethylsilane (0.28 mL, 2.2 mmol) for 5 min, to furnish, following standard workup and chromatography, 0.29 g (55%) of a 38:62 mixture of 3,6-*trans-* and 3,6-*cis*-oxasilacycloheptanes accompanied by 0.16 g of an unknown side product. A small portion of the product was further purified by semi-preparative HPLC (21 x 250mm, 5 mL/min 1% EA/hex); the major and minor diastereomers eluting at 17.0 and 18.0 min, respectively: $R_f = 0.2$ (hexane); IR: 2982, 2953, 2925, 2874, 1461, 1377, 1237, 1090, 1007, 850, 757 cm⁻¹; HREIMS calcd. for C₁₆H₃₄OSi_:[M-C₂H₅]⁺: 241.1764; found: 241.1986.

Diastereomer 1((*trans*, minor): ¹H δ 3.59 (td, 1H, 8.6, 4), 1.73(m, 3H), 1.26-1.48(11H), 1.08 (m, 1H), 0.99(3H, d, 6.6), 0.95(3H, t, 7.9), 0.94(3H, t, 7.9), 0.88(3H, t, 6.8), 0.69 (dt, 1H, 15, 2) 1H), 0.48-0.63(5H). ¹³C δ 75.0, 39.9, 39.3, 38.4, 32.0, 31.0, 29.3, 28.8, 26.1, 23.3, 22.7, 14.1, 7.00, 6.8, 6.1, 5.9

Diasteromer 2 ((*cis*, major): ¹H δ 3.76 (m, 1H), 1.96 (m, 1H), 1.34-1.73(17H), 1.01(3H, d, 6.7), 0.94(6H, t, 7.9), 0.88(3H, t, 6.8), 0.73 (m,1H), 0.51-0.63(5H). ¹³C δ 73.8, 38.4, 35.2, 34.6, 31.9, 29.44, 29.37, 26.2, 25.6, 22.7, 22.1, 14.1, 7.3, 6.9, 6.8, 6.3

(*trans-*)-**1,1-Diethyl-5-methyl-3-phenyl-2-oxa-1-silinane** (*trans-***16-Et**) was prepared as a single diastereomer from **15** (0.32g, 2.0 mmol) and Et_2SiH_2 (0.33 mL, 2.6 mmol) using procedure "B" described above except that the reaction temperature was held between 5 and 10 °C. The product was assigned as the *trans*-isomer in analogy with **2-Pr**; this was confirmed by a correlation via diol **21** (vida infra). The crude product was purified by column chromatography (0-5% EA/Hex) to furnish 0.25 g of a diethylsiloxysiloxane byproduct followed by 0.25 g (51%) **16-Et**, predominantly as the trans diastereomer. A small portion of the product was further purified by semi-preparative HPLC (21 x 250mm, 5 mL/min of 1% EA/hex).

16-Et: $R_f = 0.2$ (hexane) or 0.4 in 5% EA/hex); ¹H δ 7.24-7.28(1H), 7.34-7.40(4H), 5.19(1H, dd, 3.9, 7.5), 2.19 (m, 1H), 1.85(1H, ddd, 14.1, 7.5, 3), 1.73(1H, ddd, 14.1, 7.1, 4), 1.17(3H, d, 6.9), 1.05(3H, t, 7.9), 1.05(3H, t, 7.9), 0.94(1H, dd, 5.7, 12.2), 0.74(2H, q, 7.9), 0.69(2H, q, 7.9), 0.55(1H, dd, 7.0, 14.7); ¹³C δ 145.7, 128.1, 126.6, 125.4, 71.9, 43.7, 25.0, 23.5, 16..5, 8.0, 7.3, 6.7, 6.6; IR: 3087, 3063, 3028, 2953, 2874, 1494, 1453, 1412, 1377, 1354, 1236, 1207, 1137, 1088, 1066, 1005, 907, 853, 809, 737, 699 cm⁻¹; HR-FABMS calcd. for C₁₅H₂₄OSi₂[M-H]⁺: 247.1517; found: 247.1527.

Byproduct: $R_f = 0.8$ in 5% EA/hex. The byproduct displayed major ions at 343 (M-H)⁺ and 189 (M- Si(Et)₂OSi(Et)₂H)⁺ in the GC/MS spectra, and was tentatively assigned as 1,1-diethyl-1-(diethylsiloxy)-2-methyl-4-phenylbutyl silane. This assignment was supported by the lack of a carbinol HC and the presence of silane (4.5 ppm, narrow pentet) and multiple Et₂Si spin systems (1.1-0.87 for methyl groups; 0.7-0.45 for ethyl) in a complicated ¹H NMR spectrum. ¹³C: 143.1, 128.4, 128.3, 125.5, 42.7, 33.7, 28.6, 22.9, 22.8, 7.4, 7.3, 7.1, 6.8, 6.6. Oxidative cleavage (Tamao oxidation, below) furnished 2-methyl-4-phenyl-1-butanol:^{xii} $R_f = 0.3$ (20% EA/hex); ¹H δ 7.30 (app t, 2H, 7); 7.22-7.18 (3H); 3.54 (dd, 1H, ABX, 10.8, 6); 3.48 (dd, 1H, ABX, 10.8, 6.4); 2.73 (ddd, ABXY, 1H, 13.6, 10, 5.6); 2.62 (ddd, ABXY, 13.6, 10, 6); 1.79 (m, 1H); 1.68 (apparent hextet, 1H, 6-7); 1.58 (1H, bs, OH); 1.46 (m, 1H); 1.008 (d, 3H, 7.2). IR (ATR crystal) 3346 (s, broad), 2926, 2873, 1454, 1037 cm-1; HREI: calcd. for C₁₁H₁₆O: [M]⁺164.1204; Found: 164.1204 (1.6 ppm).

Diethyl (2-methylnon-1-en-3-oxy) silane (18-Et): Attempted one-pot reaction of 2-methyl non-1-en-3-ol (**17**, 0.22 g, 1.4 mmol) with diethylsilane (0.20 mL, 1.1 equiv) and $B(C_6F_5)_3$ (0.2 g, n toluene, ~0.4 mmol) as described for the synthesis of **2-Et** furnished the corresponding diethylsilyl ether, **18-Et** as an inseparable mixture with small amounts of one or more siloxanes: $R_f = 0.8$ (hexane); ¹H: δ 4.85 (bs, 1H); 4.75 (bs, 1H); 4.51 (app pentet, 2.4, residual diethylsilane); 4.16 (t, 1H, 6.4); 1.68 (s, 3H); 1.45 (s, 0.7H, residual Si-H); 1.5 (m, 2H), 1.33-1.18 (8H); 1.0 - 0.92 (19 H, including some silane); 0.88 (t, 3H); 0.62 (dq, 4H, 7, 2); 0.53 (m, 8H, residual silane and siloxane); ¹³C: δ 147.4; 110.49; 76.71, 36.02, 31.9, 29.4, 25.5, 22.7, 17.1, 14.1, 7.47, 7.24, 6.97, 6.64, 6.61, 6.55; IR 2954, 2116, 1450 cm-1. HRMS was attempted but gave no recognizable fragments.

Oxidation to diols

Tamao oxidation^{xiii} (**illustrated for** (2R*,5S*)-**2-methylundecane-1,5-diol** (**19**): 135 mg (0.50 mmol) of **14-Et** wa reacted with KF (0.058 g, 2 eq), KHCO₃ (0.100 g, 2 eq), 30% H₂O₂ (1 mL, 20 eq, ca. 9M in H₂O) in MeOH/THF for 48 h to afford 0.030 g (30%) of diol **19** as an inseparable mixture of diasteromers. $R_f = 0.2$ (30% EA/hex); ¹H δ 3.58 (m, 1H); 3.47 (t, 2H, 6; or bd, 1H, ~6, depending upon sample concentration), 2.3 (broad, 2H, varies with concentration); 0.89 (app t, 3H, 8); 0.857 (appt t, 3H, 8);1.8-1.2 (15H); 0.85-0.92 (6H); ¹³C δ 72.4, 72.0, 67.8, 67.6, 37.53, 37.49, 35.8, 35.3, 34.3, 34.0, 31.8, 29.3, 28.9, 28.6, 25.7, 25.6, 22.6. 20.8, 16.8, 16.4, 14.1; The major diasteromer was assigned as 2R*,5S* based upon comparision with ¹³C data reported for a similar diol (major: 71.8, 67.8 ppm; minor 72.3, 67.6 ppm.^{xiv} IR 3330 (b, OH), 2926, 2856, 1458, 1030 cm⁻¹; HRFAB calc. For C₁₂H₂₇O₂ [MH]⁺: 203.2011; found: 203.2014 (1.3 ppm).

Oxidation of 88 mg (0.33 mmol) of the major $(2^{nd}$ eluting isomer) of **14-Et** using the Woerpel procedure described below afforded 36.8 mg (56%) of **19**. Spectral details were identical to those reported above.

Woerpel oxidation: ^{xv} 2-(2-Hydroxyoctyl)-cyclohexanol (20): To a solution of *tert*-butyl hydroperoxide (0.73 mL, 5-6M in decane) in 3 mL DMF at 0 $^{\circ}$ C was added cesium hydroxide

(0.52g, 3.1 mmol). The reaction mixture was allowed to warm to 25 °C, whereupon a solution of oxasilinane **10-Ph** (0.10g, 0.26 mmol) in 2 mL DMF was added dropwise. After 10 minutes, tetrabutylammonium fluoride (1.3 mL, 1M in THF) was added. The reaction solution was stirred at RT for 2 h and then quenched with 10 mL of 10% aq. sodium bisulfite. The mixture was extracted with ether (2 x 20 mL) and the combined organic layers were dried and concentrated. The residue was subject to column chromatography using 40% EA/hex as eluting solvent to afford 22.8 mg (38%) of diol **20**: $R_f = 0.50$ (50% EA/hex); IR (same except where noted): 3329-31, 2927-8, 2856, 1450, 1071 (diast 2), 1039 (diast 1), 976 cm⁻¹; HRMS calc. for $C_{14}H_{29}O_2$ (MH)⁺: 229.2168; found: 229.2159 (3.5 ppm).

Diastereomer 1: ¹H δ 3.95(1H, m), 3.73-3.78(1H), 2.19(2H, s), 1.65-1.72(4H), 1,48-1.61(3H), 1.29-1.47(14H), 0.88(3H, t, 6.6) ¹³C δ 69.2, 69.1, 40.0, 38.1, 38.0, 33.1, 31.8, 29.3, 27.1, 25.8, 25.4, 22.6, 20.5, 14.1

Diastereomer 2: ¹H δ 3.90-3.93(1H), 3.65-3.70(1H), 1.72-1.75(1H), 1.25-1.61(18H), 0.88(3H, t, 6.6); ¹³C δ 70.9, 70.0, 39.4, 39.1, 38.7, 32.4, 31.8, 29.3, 28.2, 25.6, 24.3, 22.6, 21.4, 14.1 MS:

(1R*,3S*) 3-Methyl-1-phenyl-1,4-butanediol (21) was prepared in 69% by oxidation of oxasilane 16-Et using the Woerpel procedure described above: R_f =0.2 (40% EA/hex); The product was assigned by comparison with literature reports.^{xvi} R_f = 0.2 (40% EA/hex); 1H δ 4.89 (dd, 1H, 7.6, 4.7); 3.57 (dd, ABX, 1H, 10.4, 4.4); 3.52 (dd, ABX, 1H, 10.4, 6.4); 1.9-1.7 (3-4H, includes both ABX and a multiplet); 0.97 (d, 3H, 6.8); 13C δ 144.7, 128.4, 127.4, 125.8, 71.8, 67.9, 43.5, 32.2, 17.2.

References:

- ⁱ Charoenchaidet, S.; Chavadej, S.; Gulari, E. J. Mol. Catal. A: Chem. 2002, 185, 1-2, 167-177.
- ⁱⁱ Wilson, S. R.; Guazzaroni, M. E. J. Org. Chem. 1989, 54, 3087-91; Cai, M., Huang, Y., Zhao,
 H., Zhang, R., J. Orgmet. Chem. 2004, 689, 2436-2440
- ⁱⁱⁱ Anwar, U.; Grigg, R.; Rasparini, M.; Savic, V.; Sridharan, V. Chem. Comm. 2000, 645.

^v Wilson, S. R.; Phillips, L. R.; Natale, K. J. Jr. J. Am. Chem. Soc. 1979, 101, 3340-3344.

^{iv} Snider, B. B.; Rodini, D. J.; Kirk, T.C.; Cordova, R. J. Am. Chem. Soc. 1982, 104, 555-563.

- ^{vi} Arnold, R.T.; Smolinsky, G. J. Am. Chem. Soc. 1959, 81, 6443-5; Eddy, C. R., Showell, J. S.;
 Zell, T. E. J. Am. Oil Chem. Soc. 1963, 40, 92-6.
- ^{vii} Tirado, R.; Prieto, J. A.; *J. Org. Chem.* **1993**, *58*, 5666-73; Hojo, M.; Sakuragi, R.; Okabe S.;
 Hosomi A. *Chem. Commun.* **2001**, 357-358.
- viii Tang, S.; Kennedy, R. M. Tetrahedron Lett. 1992, 33, 5303-6.
- ^{ix} Denmark, S. E.; Yang, S. Tetrahedron, 2004, 60, 9695-9708.
- ^x Katzennellenbogen, J. A.; Christy, K. J. J. Org. Chem. **1974**, 39, 3315. Fernández-Mateos, A., Burón, L. M., Clemente, R., Silvo, A. I. R., González, R. R. Synlett, **2004**, 1011-1014.
- ^{xi} For a report on the formation of *cis*-3,5-disubstituted 1,2-oxasilinanes through 6-*exo* cyclizations of α-silyloxymethyl radicals, see: Koreeda, M.; Hamann, L. G. *J. Am. Chem. Soc.* **1990**, *112*, 8175-8177.
- xii Lebel, H; Ladjel, C. J. Organomet. Chem. 2005, 690, 5198.
- ^{xiii} Tamao, K.; Ishida, N.; Tanaka, T.; Kumada, M. Organometallics **1983**, 2, 1694–1696. For a review of this area, see: Jones, G. R.; Landais, Y. *Tetrahedron* **1996**, *52*, 7599–7662
- xiv Zhu, G.; Negishi, E-i.; Org. Lett. 2007, 9, 2771.
- ^{xv} Smitrovich, J. H.; Woerpel, K. A. J. Org. Chem. **1996**, 61, 6044.
- ^{xvi} Najera, C.; Yus, M.; Seebach, D. *Helv. Chim. Acta.* 1984, 67, 289-300; Matsumoto, K.; Aoki,
 Y.; Oshima, K.; Utimoto, K.; Rahman, N. A. *Tetrahedron*, 1993, 49, 8487.

B(C₆F₅)₃-promoted tandem silylation and intramolecular hydrosilylation: diastereoselective synthesis of oxasilinanes and oxasilepanes

Roman Shchepin, Chunping Xu, and Patrick Dussault* pdussault1@unl.edu

Supporting Information: ¹H and ¹³C NMR spectra

Alcohols	pages
2-Methyldec-1-en-4-ol (1): ${}^{1}H {}^{13}C$	3-4
2-Phenyldec-1-en-4-ol (3): ${}^{1}H$ ${}^{13}C$	5-6
1-Cyclobutenyloctan-2-ol (5): ${}^{1}\text{H}^{-13}\text{C}$	7-8
2-Methyl-2-dodecen-6-ol (7): ${}^{1}H$, ${}^{13}C$	9-10
1-Cyclohexenvloctan-2-ol (9): 1 H. 13 C	11-12
(E,Z)-Undec-2-en-5-ol (11): ¹ H, ¹³ C	13-14
2-Methylundec-1-en-5-ol (13): ${}^{1}H$, ${}^{13}C$	15-16
3-Methyl-1-phenylbut-3-en-1-ol (15): ${}^{1}H$, ${}^{13}C$	17-18
2-Methylnon-1-en-3-ol (17): ¹ H, ¹³ C	19-20
Alkoxysilanes	pages
Diisopropyl-(2-methyldec-1-en-4-oxy)silane (1Pr): ¹ H, ¹³ C	21-22
Diisopropyl-(2-phenyldec-1-en-4-oxy) silane (3-Pr): ¹ H, ¹³ C	23-24
Disopropyl-(1-cyclobutenyloctyl-3-oxy)silane (5-Pr): ¹ H, ¹³ C	25-26
Diethyl-(2-methylnon-1-en-3-oxy) silane (18-Et): ¹ H, ¹³ C	27-28
Cyclic Siloxanes	pages
	20.00
(trans)-1,1-Diethyl-3-hexyl-5-methyl-2,1-oxasilinane $(trans-2-Et)$: ¹ H, ¹³ C	29-30
(trans)-1,1-Diphenyl-3-hexyl-5-methyl-2,1-oxasilinane $(trans-2-Ph)$: ^H , ^S C	31-32
(trans)-1,1-Diisopropyl-3-hexyl-5-methyl-2,1-oxasilinane $(trans-2-Pr)$ ¹ H, ¹³ C	33-34
(<i>trans</i>)- 1,1-Diisopropyl-3-nexyl-5-pnenyl-2,1-oxasilinane (<i>trans</i> -4-Pr): ¹ H, ² C	35-30
$(3\alpha 5\alpha 6\alpha) \in \mathbf{D}_{\mathbf{n}}^{-1} \mathbf{U}^{-13}C$	27 28
(3a, 5a, 6a) 6 D : ¹ H ¹³ C	30.40
$(3\rho, 3\alpha, 0\alpha)$ - U-FT . H , C (<i>trans</i>) 1.1 Diigeopropyl 2 heyyl 6 igeopropyl 2.1 eyesilinene (<i>trans</i> 8 Dr); ¹ H ¹³ C	39-40 41 42
(I'ans)-1,1-Diisopiopyi-3-nexyi-0-isopiopyi-2,1-oxasimiane (<i>I'ans</i> - o-FT). If, C 2.1 Benzovasilin, octabudro 1.1 dietbyl 3 beyyl (10 Ft)	41-42
$(38.5\alpha, 6\alpha)$ and $(3\alpha, 5\alpha, 6\alpha)$ 10Ft : ¹ H ⁻¹³ C	13 11
$(3\beta,5\alpha,6\alpha) = 2$ 1 Benzovagilin, actabudro 1 1 diphenyl 3 hevyl (10 Ph)	43-44
$(38.5\alpha.6\alpha) - (10-Ph) \cdot {}^{1}H {}^{13}C$	<u>45</u> 16
(3p, 5a, 6a) (10 Ph): ¹ H ¹³ C	45-40
(30, 30, 00)- $(10$ -FH). II, $C(trans) 1.1 Disthyl 3 havyl 6 methyl 2.1 avasilinana (12 Et). 1U 13C$	4/-40
$(\mu \alpha \mu \omega)^{-1}$, 1^{-1} $D(\alpha \mu \gamma)^{-3}$ - $\Pi c \lambda \gamma ^{-0}$ - $\Pi c (\mu \gamma)^{-2}$, 1^{-0} $\lambda a \beta \Pi \Pi a \Pi c (\mu 2^{-1} \Delta \mu)$. Π , C	42-20

(trans)-1,1-Diphenyl-3-hexyl-6-methyl-2,1-oxasilinane (12-Ph): ¹ H, ¹³ C	51-52
1,1-Diethyl-3-hexyl-6-methyl-2-oxa-1-silepane (14-Et):	53
(<i>trans</i>)- 14-Et : ¹ H	
(<i>cis</i>)- 14-Et : 1 H, 13 C (mixture)	54-55
(3,5- <i>trans</i>)-1,1-Diethyl-5-methyl-3-phenyl-2,1-oxasilinane (16-Et): ¹ H, ¹³ C	56-57
Diols	pages
	pages
2-Methylundecane-1,5-diol (19): ¹ H, ¹³ C	<u>58-59</u>
2-Methylundecane-1,5-diol (19): ¹ H, ¹³ C 2-(2-Hydroxyoctyl)-cyclohexanol (20):	58-59
2-Methylundecane-1,5-diol (19): ¹ H, ¹³ C 2-(2-Hydroxyoctyl)-cyclohexanol (20): diasteromer 1: ¹ H, ¹³ C	58-59 60-61
2-Methylundecane-1,5-diol (19): ¹ H, ¹³ C 2-(2-Hydroxyoctyl)-cyclohexanol (20): diasteromer 1: ¹ H, ¹³ C diasteromer 2: ¹ H, ¹³ C	58-59 60-61 62-63



Norman and Adding to the second s	ran Bangka, Badda Jasa Ju	(1) (1) (1) (1) (1) (1) (1) (1) (1) (1)	, Lida ala La di J. La dan Ma Ing kata Patra panganan panganan pa	, militar (and a stand or first start of a	ด้ ไปสู่ปล่างไป ได้ ได้ 1995 การส่างไป เป็นสืบคราม เป็นไปสาวารัณ เพราะการการการการใจ 1997 ให้ ๆ ปลาย เราะการการการการการการการการการการการการการก	Alderstadelige og en at den som det som som det som	611 ghtines fallefiller, all seally 1977 - 9 spail age migger 1997 175	Maka an ka ka ka wa sa	F2 - P F2 - P F2 - P F2 - P SI SF WDW SSB LB GB PC	70.00 usec -3.35 dB 13.34 dB 13.34 dB 400.1316005 MHz rocessing parameters 32768 100.6127525 MHz EM 0 1.00 Hz 0 1.40
Nacional a stillar con a versional a traverse service	nan Bergina di sila jara se Nan persida	նենուլ է ծ. ստղ է լթեմին էն է վինթորն ու սկրություն թունու	Alden Males Brody Marson Mar Propriet The Propriet Street Propriet Street		n. besteletetetetetetetetetetetetetetetetete	didentaderaa jalomad dina si Najamayongo aya mohara oo ka a	को को कि कि को की कि अन्य की रहती का सुरे रहन के स्वताय स्वतन का सुर प्रभाव से क	18.64.(1.41).10.24.0.000.04.04 9.9.9.19.10.10.20.000.000.00000000000000	PCPD2 PL2 PL12 PL13 SFO2 F2 - P SI SF SF THE MERICAN SF SSB	70.00 usec -3.35 dB 13.34 dB 13.34 dB 400.1316005 MHz rocessing parameters 32768 100.6127525 MHz EM 0
									PCPD2 PL2 PL12 PL13 SFO2 F2 - P	70.00 usec -3.35 dB 13.34 dB 13.34 dB 400.1316005 MHz rocessing parameters
									PCPD2 PL2 PL12 PL13	70.00 usec -3.35 dB 13.34 dB 13.34 dB
									PCPD2 PL2	70.00 usec -3.35 dB
									NUC2	1H
									===== CPDPRG	== CHANNEL f2 ======= 2 waltz16
									P1 PL1 SF01	0.50 dB 100.6228298 MHz
									====== NUC1	== CHANNEL f1 ======= 13C
			I						D11 TD0	0.03000000 sec 1
									DE TE D1	6.50 usec 295.2 K 2.00000000 sec
									AQ RG DW	1.3664756 sec 812.7 20.850 usec
								I	SWH FIDRES	23980.814 Hz 0.365918 Hz
						1			SOLVEN NS DS	T CDCl3 33 4
						1			PROBHD PULPRO TD	5 mm QNP 1H/13 G zgpg30 65536
							I		Date_ Time INSTRU	20100805 16.44 M spect
1					1				F2 - A	cquisition Parameters
	όн								NAME EXPNO	cx-6-24
\sim	\sim	\checkmark			\mathbb{V}				Curron	
			143	- 113		- βα. - 46.	225. 225.	14.	BF	RUKER
	1	OH 1	Control of the second s	T T T T T T T T T T T T T T T T T T T						$\begin{array}{c c c c c c c c c c c c c c c c c c c $














file: ...S_paper\More spectra\cx-6-23\1\fid expt: <zg30> transmitter freq.: 400.132471 MHz time domain size: 65536 points width: 8278.15 Hz = 20.6885 ppm = 0.126314 Hz/pt number of scans: 16

freq. of 0 ppm: 400.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 128.818 ppm/cm: 0.32194





	SpinWorks 2.5: 13C NM	IR							
		OH							
	1	11							
						l			
		na an a	an and the second s				, 		avisorus (ditteration)
PPM 120 110 100 90 80 70 60 50 40 30 20 10 0	PPM 120 11	0 100 90	80 70	60 50	40	30	20	10	Ó
file: D:\Roman's\rvs-4-31\2\fid expt: <zgpg3< td=""> freq. of 0 ppm: 100.612769 MHz transmitter freq.: 100.622830 MHz processed size: 65536 complex points time domain size: 65536 points LB: 1.000 GB: 0.0000 width: 22075.06 Hz = 219.384162 ppm = 0.336839 Hz/pt Hz/cm: 562.193 ppm/cm: 5.58714</zgpg3<>	file: D:\Roman's\rvs-4-31\2\fid expt: <zg transmitter freq.: 100.622830 MHz time domain size: 65536 points width: 22075.06 Hz = 219.384162 ppm =</zg 	gpg3 = 0.336839 Hz/pt		freq. of 0 ppm: 100.612769 MHz processed size: 65536 complex pr LB: 1.000 GB: 0.0000 Hz/cm: 562.193 ppm/cm: 5.58	pints 714				



SpinWork	s 2.5: 13C	NMR								
	\sim	ОН								
çərli əsələtin ittərəfəri dəri ittəriyi	المعلومية والمراجع	antike . Jan dil kukulata suration bi na onos shaka	n den diet beson en site dieten die state die see site (seite	ge de Martin Martin de Las les es an actives andre solle a martin a des les les sous andres solle anna de las l	tede at well the to a fine to a television of the second state	L bow. At fire out we about the distribution	neral metabolis (h. 1. a. sele la constance non la seconda de selección de selección de seconda de seconda de s	a distance in the set. As a set of the set of the set	to the Royan de Terre Lorenzi attres Lorenzi	والمرافع والمتعادين والمتعارين
and the state of the set of the s	ક્રમ્પ્રે Medice All કે તેમ તો કરકી કરી ને પિઝેટિંગ (જ્યાલ કે જાવી કે ક્રમ્પ્રે પ્રાપ્ત	a ta ann an an an an ann an an an an an an	na ferda a faile a faile an	an fi fi gu d fad galangi na gang kan	n ta ta a sing na ann ga taon tan tan ta ka t	an palland to set and flat to 1995. It is a professional data and the	and here is a second	led i v dd. a boulond "ski fadi i donen oddi. A shi i	and an	a fa anti-change at a statut
	-1		1 1 1		· · · · ·					1 1
PM	180	160	140	120	100	80	60	40	20	0
le: D:\CX_RVS	_paper\More spectr	a\cx-6-19-sm\2\fid	expt: <zgpg30></zgpg30>		freq. of 0	opm: 100.612774 MH	Ηz			
ansmitter freq.: me domain size	: 100.622830 MHz e: 65536 points				processe LB: 0.30	1 size: 65536 comple 00 GB: 0.0000	x points			
/idth: 22075.06	Hz = 219.384162 p	pm = 0.336839 Hz/j	ot		Hz/cm: 8	82.914 ppm/cm: 8	8.77449			



S		0H 15													
		Ι.													
			<u>homituda an aigidan</u> 			-*************************************			**************************************	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		//////////////////////////////////////	ар ж ^а (ал а _{ай} а а адаа ал а 	m,main,dawims(M	
PPM file: D:\Ro transmitte time doma width: 220 number o	140 man's\rvs-4-30\\ r freq.: 100.6228 ain size: 65536 p 075.06 Hz = 219 f scans: 77	130 21fid 330 MHz points .384162 ppm	120 = 0.336839	110 Hz/pt	100	90	80	70 freq. of 0 ppm: processed size LB: 1.000 C Hz/cm: 621.86	60 100.612769 M : 65536 comp GB: 0.0000 2 ppm/cm:	50 /Hz lex points 6.18013	40	30	20	10	




































































































SpinWorks	\bigcirc	он 21	он											
						1								
						,,			1)-1					**************************************
PPM 1 file: D:\CX_RVS_p transmitter freq.: 1 time domain size: width: 22075.06 H number of scans:	35 12 aper/More spec 00.622830 MHz 65536 points z = 219.384162 75	25 11 stra\cx-6-25\2\f ppm = 0.3368	5 10 id expt: <zgp 39 Hz/pt</zgp 	5 95 g30>	85	75	freq. of 0 pp processed LB: 0.300 Hz/cm: 608	55 50 51 51 51 51 51 51 51 51 51 51 51 51 51	55 774 MHz complex poin 00 /cm: 6.0492	45 ts 4	35	25	15	5