

Spring 4-13-2016

# Optimizing Chemical & Rheological Properties of Rejuvenated Bitumen

Dominic Nguyen

University of Nebraska-Lincoln, dominicnguyen@huskers.unl.edu

Hamzeh Haghshenas Fatmehsari

University of Nebraska-Lincoln, h.haghshenas@huskers.unl.edu


Santosh Kommidi

University of Nebraska-Lincoln, santosh.kommidi@gmail.com

Yong-Rak Kim

University of Nebraska-Lincoln, yong-rak.kim@unl.edu

Follow this and additional works at: <https://digitalcommons.unl.edu/civilengdiss>

 Part of the [Materials Chemistry Commons](#), [Other Chemical Engineering Commons](#), [Structural Materials Commons](#), and the [Transportation Engineering Commons](#)

Nguyen, Dominic; Haghshenas Fatmehsari, Hamzeh; Kommidi, Santosh; and Kim, Yong-Rak, "Optimizing Chemical & Rheological Properties of Rejuvenated Bitumen" (2016). *Civil Engineering Theses, Dissertations, and Student Research*. 93.  
<https://digitalcommons.unl.edu/civilengdiss/93>

This Article is brought to you for free and open access by the Civil Engineering at DigitalCommons@University of Nebraska - Lincoln. It has been accepted for inclusion in Civil Engineering Theses, Dissertations, and Student Research by an authorized administrator of DigitalCommons@University of Nebraska - Lincoln.





# Optimizing Chemical & Rheological Properties of Rejuvenated Bitumen

Dominic Nguyen<sup>1</sup>, Hamzeh Haghshenas<sup>2</sup>, Santosh Kommidi<sup>2</sup>, and Dr. Yong-Rak Kim<sup>2</sup>

1 – Department of Chemical & Biomolecular Engineering, University of Nebraska-Lincoln

2 – Department of Civil Engineering, University of Nebraska-Lincoln

## Introduction

Bitumen has long been a material used in the construction of roadways, yet new pavement only consists of 15% of recycled materials due to poor compatibility of aged bitumen and new materials.

Chemical additives such as rejuvenators have been used in an attempt to re-balance the chemical composition and restore the physical properties of aged bitumen back to its virgin state. However, a fundamental understanding of how rejuvenators revitalize bitumen is needed before developing the optimum rejuvenator.

## Objectives

- Use Fourier-transform infrared (FTIR) spectroscopy to determine the changes in chemical properties of virgin, aged, and rejuvenated bitumen.
- Employ a linear amplitude sweep (LAS), a procedure using a dynamic shear rheometer (DSR), to investigate rheological properties.
- Relate resulting chemical evolution to changes in macroscopic mechanical properties of the revitalized bitumen.

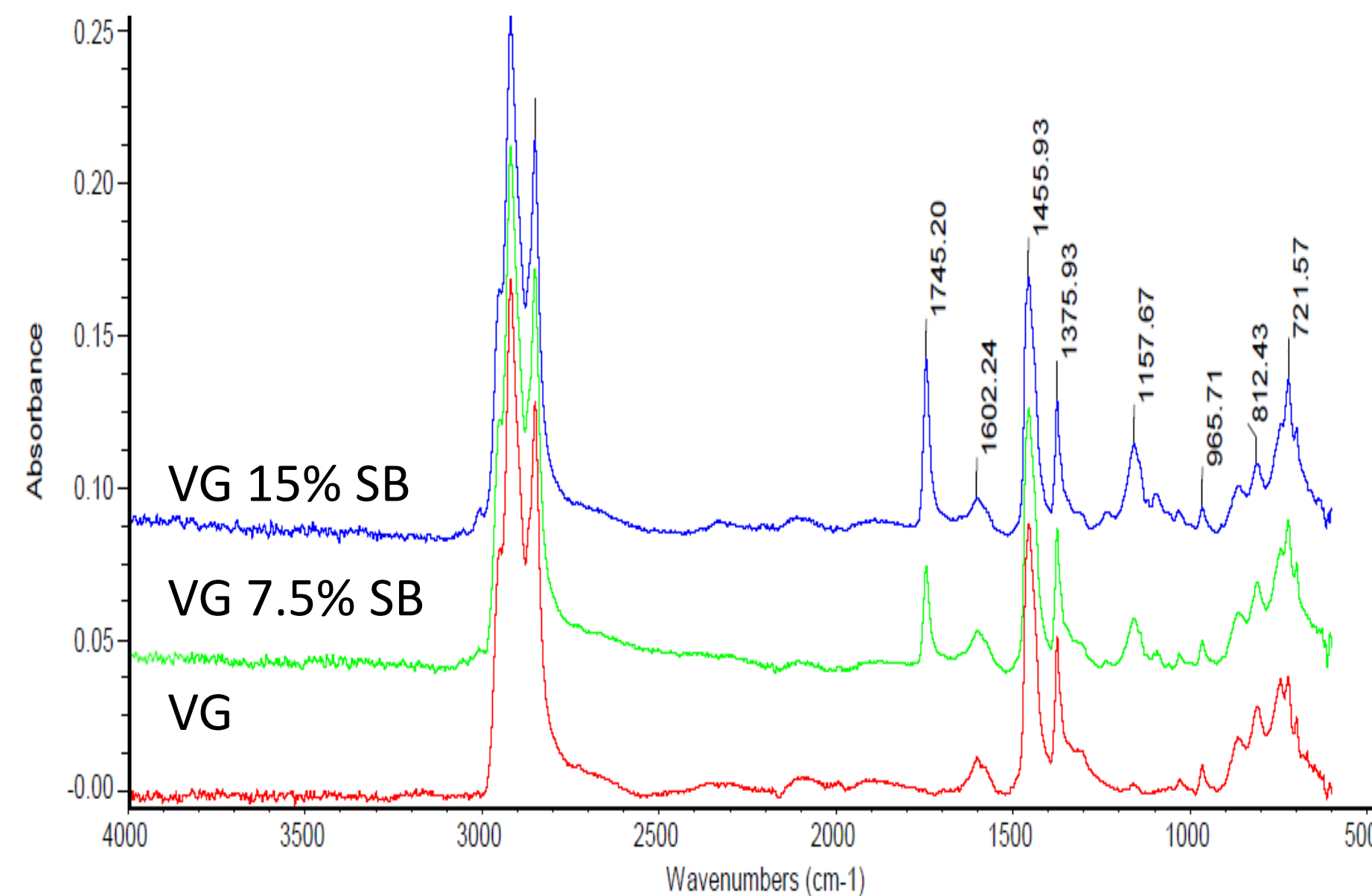
## FTIR Index Data

| INDEX                         | Carboxylic Acid           | Ether                   | Carbonyl                 | Sulfoxide                | Aliphatic              | Aromatic                |
|-------------------------------|---------------------------|-------------------------|--------------------------|--------------------------|------------------------|-------------------------|
| Approximate Wavelength (cm-1) | I <sub>COOH</sub><br>1745 | I <sub>Et</sub><br>1156 | I <sub>C=O</sub><br>1700 | I <sub>S=O</sub><br>1032 | I <sub>B</sub><br>1377 | I <sub>Ar</sub><br>1601 |
| VG                            | -0.00772                  | 0.00650                 | 0.00135                  | 0.00589                  | 0.18629                | 0.06106                 |
| VG 7.5%                       | 0.07572                   | 0.05225                 | -0.00166                 | 0.00631                  | 0.17887                | 0.04926                 |
| VG 15%                        | 0.12309                   | 0.07404                 | -0.00408                 | 0.00565                  | 0.17091                | 0.03995                 |
| RTFO                          | -0.00157                  | 0.01337                 | 0.00241                  | 0.01024                  | 0.18527                | 0.06205                 |
| RTFO 7.5%                     | 0.07039                   | 0.04935                 | -0.00022                 | 0.00848                  | 0.17826                | 0.04968                 |
| RTFO 15%                      | 0.11658                   | 0.07793                 | -0.00305                 | 0.00788                  | 0.17167                | 0.04149                 |
| PAV                           | -0.00398                  | 0.00989                 | 0.00639                  | 0.01689                  | 0.18502                | 0.06538                 |
| PAV 7.5%                      | 0.04455                   | 0.04387                 | 0.00456                  | 0.01584                  | 0.17936                | 0.05789                 |
| PAV 15%                       | 0.08780                   | 0.05971                 | 0.00224                  | 0.01399                  | 0.17399                | 0.04668                 |
| Soybean Oil                   | 0.36656                   | 0.21857                 | -0.00817                 | 0.00296                  | 0.10430                | -0.01074                |

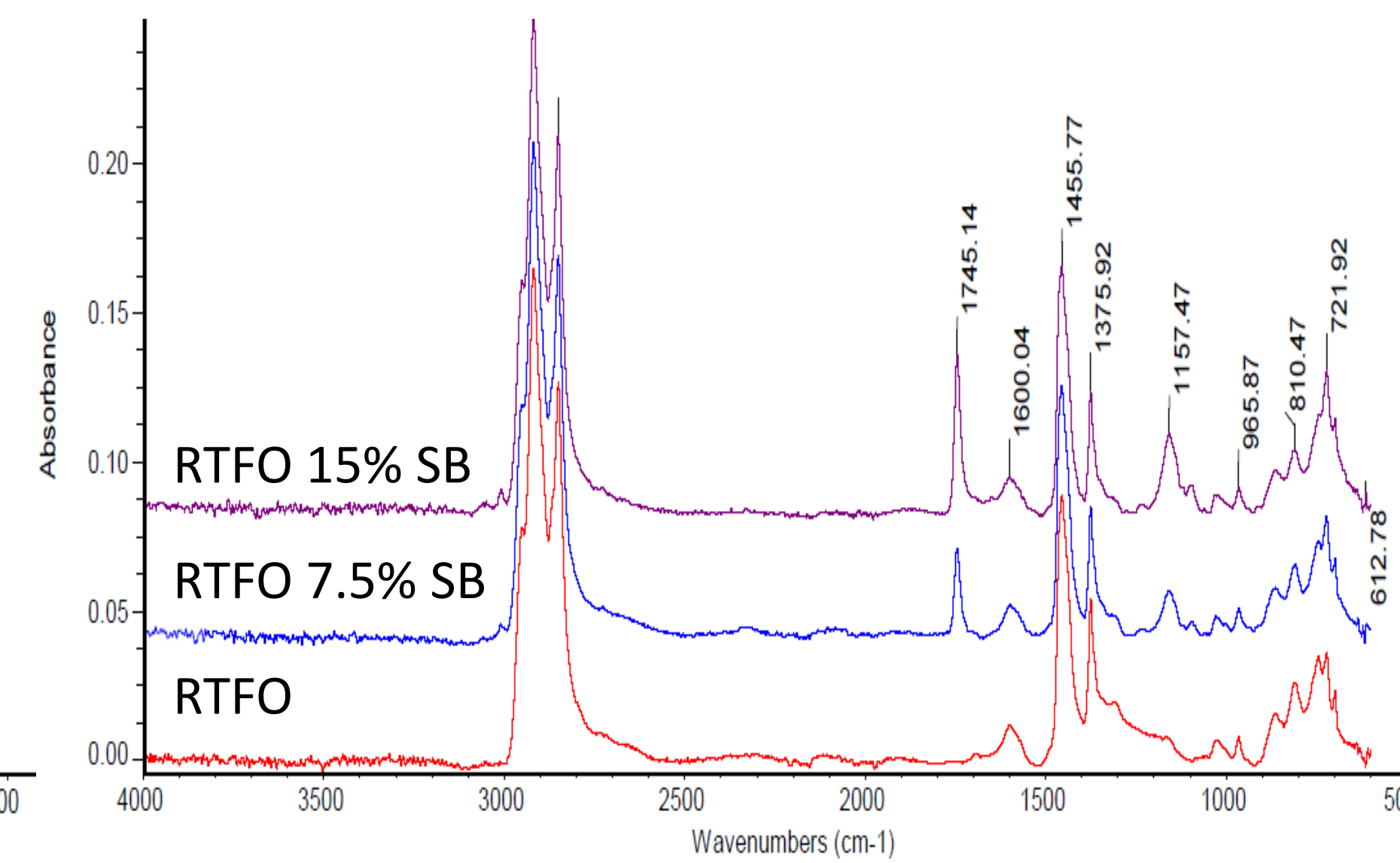
**Table 1:** Absorbance of characteristic functional groups in virgin (VG), rolling thin film oven (RTFO) aged, pressure aging vessel (PAV) aged, and rejuvenated bitumen.

Where  $I_{C=O} = A_{1700cm^{-1}} / \sum A$   
 $\sum A = \text{Total Peak Areas}$

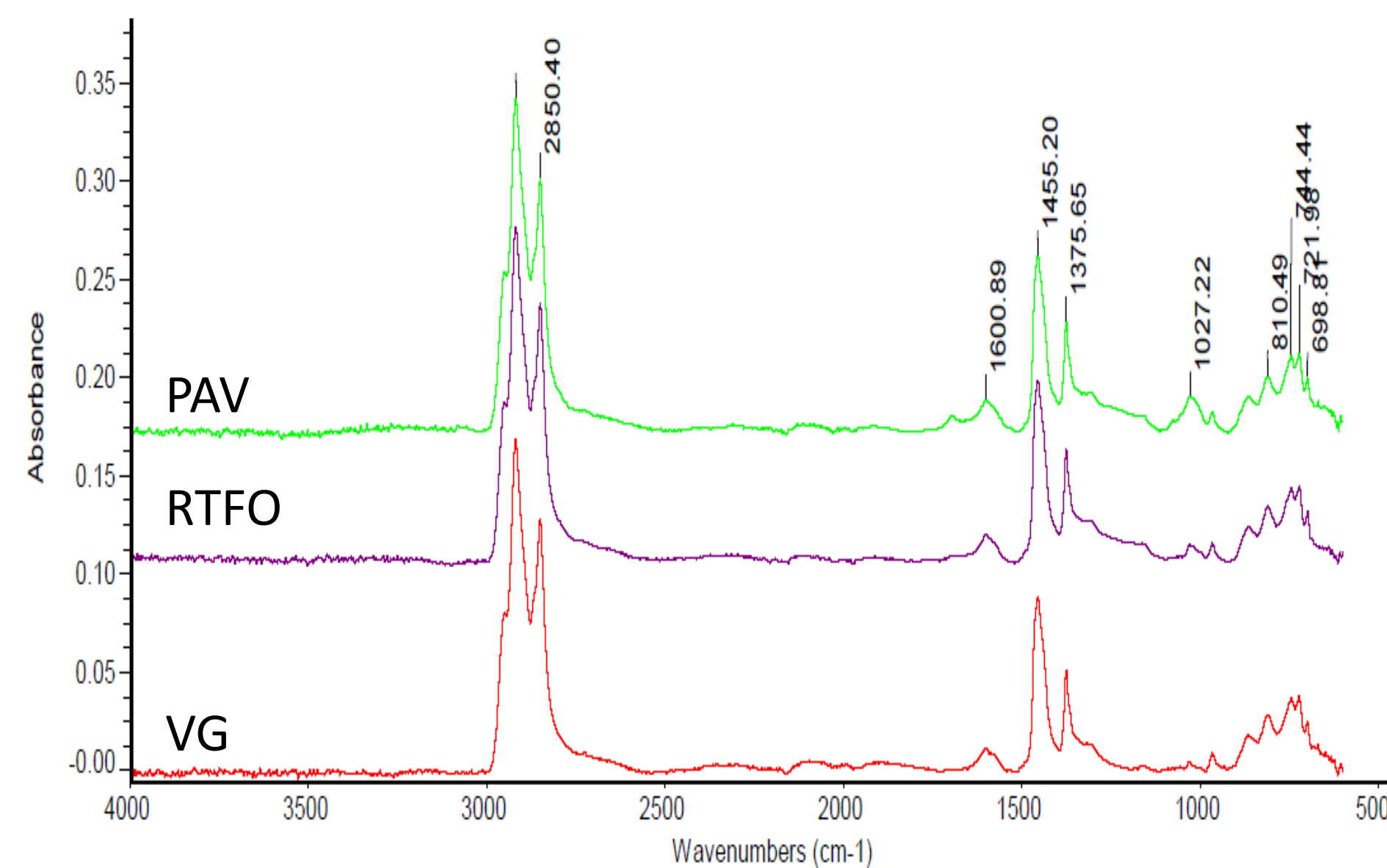
## FTIR Analysis



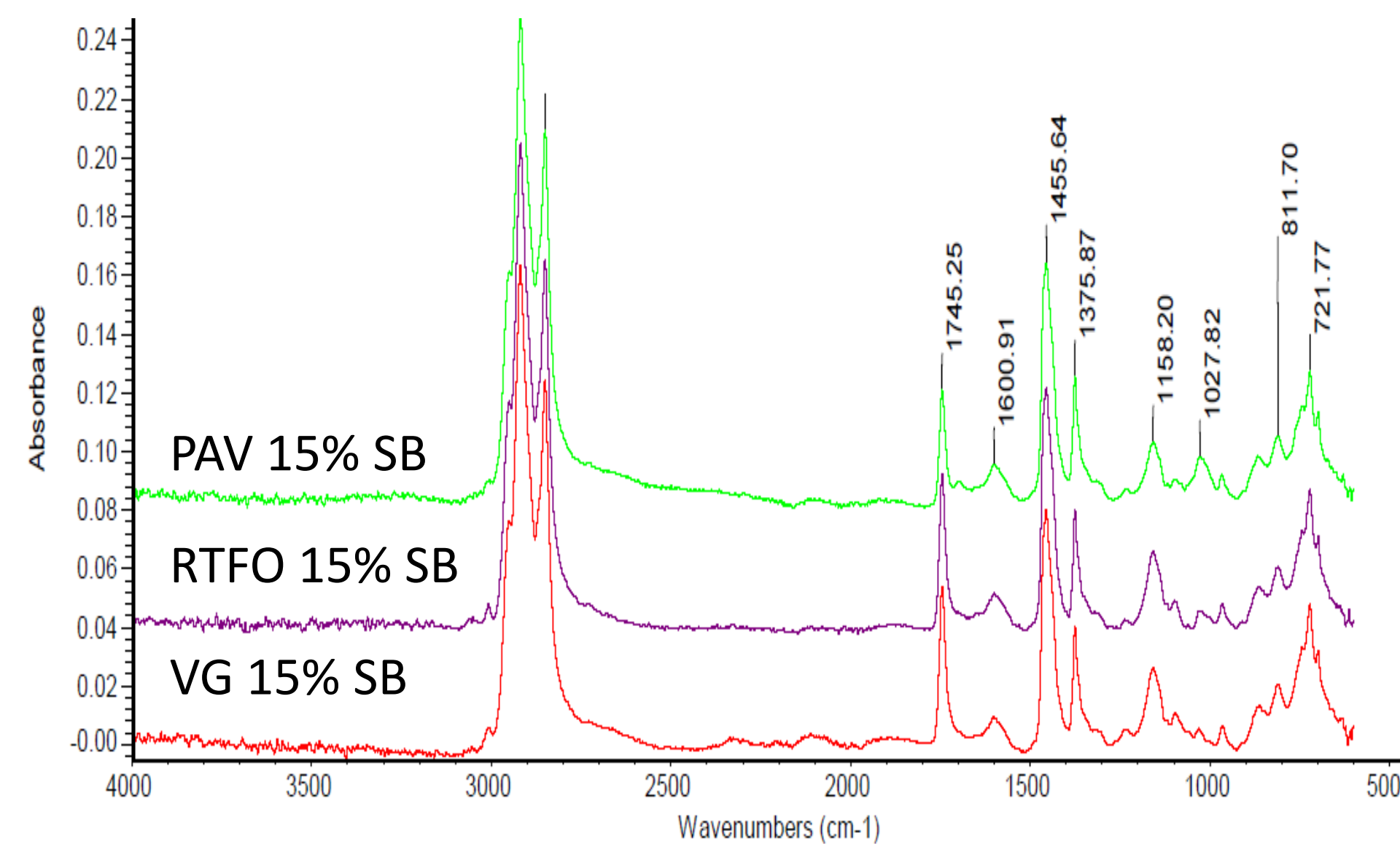
**Fig 1:** FTIR spectra of VG, VG 7.5% SB, and VG 15% SB bitumen samples.



**Fig 2:** FTIR spectra of RTFO, RTFO 7.5% SB, and RTFO 15% SB bitumen samples.



**Fig 3:** FTIR spectra of unmodified VG, RTFO, and PAV bitumen samples.



**Fig 4:** FTIR spectra of VG 15% SB, RTFO 15% SB, and PAV 15% SB bitumen samples.

## LAS Analysis

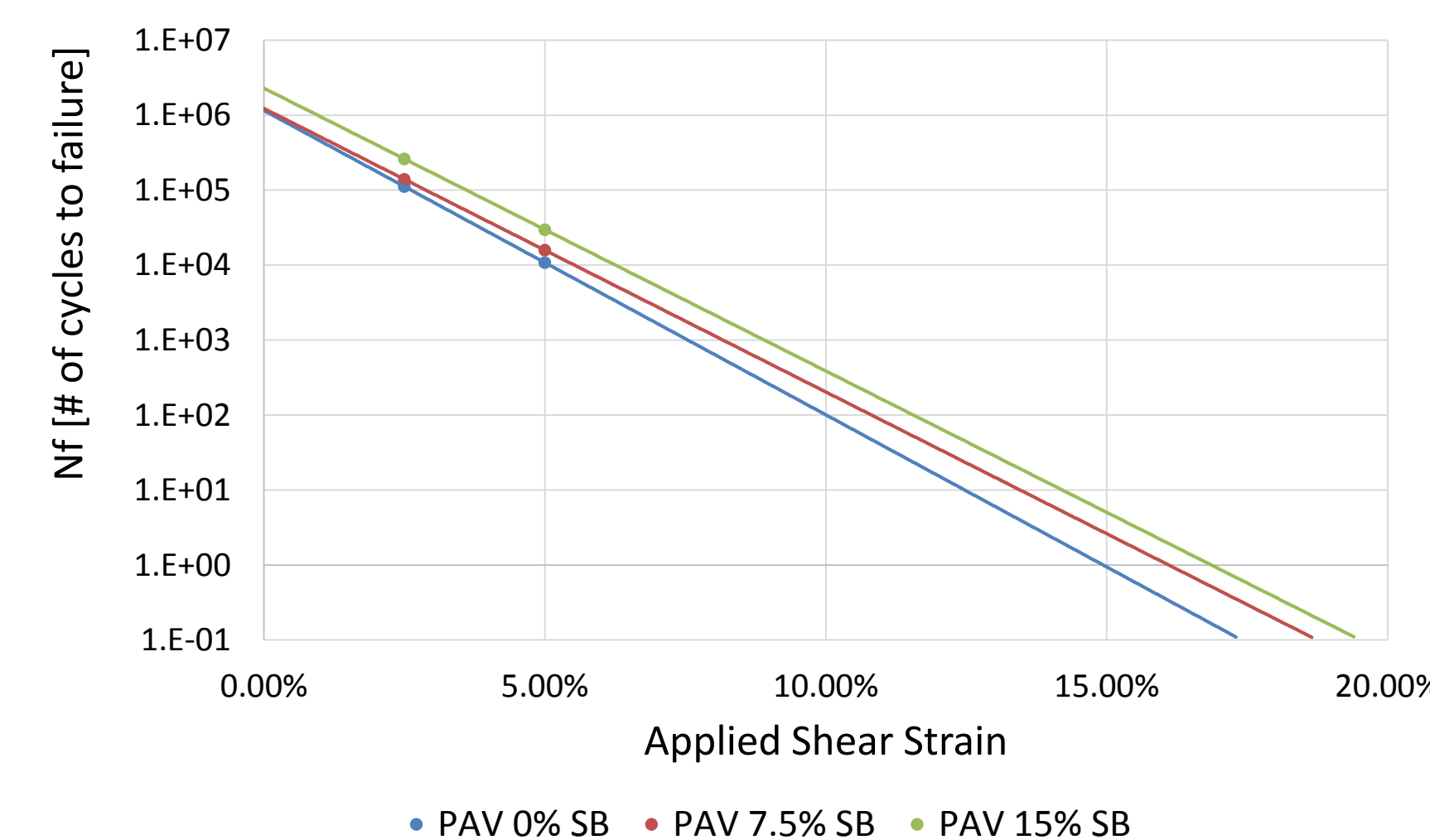
A frequency sweep test followed by a strain sweep test with linear increasing amplitude were used to calculate important binder parameters, A and B, used to determine fatigue performance ( $N_f$ ).

$$A = \frac{f(D_f)^k}{k(\Pi C_1 C_2)^\alpha} \quad B = -2\alpha \quad N_f = A(Y_{max})^B$$

### LAS Data of PAV Samples

|                       | PAV 0%   | PAV 7.5% | PAV 15%  |
|-----------------------|----------|----------|----------|
| A                     | 2443248  | 2471710  | 4586733  |
| B                     | -3.37037 | -3.14031 | -3.13159 |
| N <sub>f</sub> (2.5%) | 111,369  | 139,105  | 260,207  |
| N <sub>f</sub> (5.0%) | 10,769   | 15,777   | 29,690   |

### Bitumen Fatigue Curves



## Conclusions

FTIR analysis of  $I_{COOH}$  and  $I_{Et}$  confirms that soybean oil has been introduced to bitumen in the rejuvenation process.  $I_{Et}$  indicates soybean oil may have already been partially oxidized.

$I_{C=O}$  and  $I_{S=O}$  decrease in RTFO and PAV samples suggesting the aging process in the aged bitumen has been reversed from rejuvenation with soybean oil.  $I_B$  and  $I_{Ar}$  also decrease due to rejuvenation, indicating chain scission and aromatization that occurs during aging has been reversed.

LAS analysis of PAV samples manifests fatigue resistances ( $N_f$ ) of bitumen samples increases at every applied shear strain as a result of increasing concentration of rejuvenator.

The relation of FTIR and LAS results indicates rejuvenation of aged bitumen with soybean oil reverses the aging process at a molecular level and as a result, increases the fatigue life of the bitumen.

## References

- [1] Feng, Z., Bian, H., Li, X., and Yu, J. (2016). "FTIR analysis of UV aging on bitumen and its fractions." *Materials and Structures*, 10.1617/s11527-015-0583-9, 1381-1389.
- [2] Lu XH, Isacson U (2002) Effect of ageing on bitumen chemistry and rheology. *Constr Build Mater* 16:15–22
- [3] Kim, Y., H. J. Lee, D. N. Little, and Y. R. Kim. A simple testing method to evaluate fatigue fracture and damage performance of asphalt mixtures. *Journal of Association of Asphalt Paving Technologists*, Vol. 75, 2006, pp. 755–788.
- [4] Hintz, C., R. Velasquez, C. Johnson, and H. Bahia. Modification and Validation of the Linear Amplitude Sweep Test for Binder Fatigue Specification. In *Transportation Research Record: Journal of the Transportation Research Board*. Washington, DC., Vol. 2207, 2011, pp. 99–106.

## Acknowledgements

The authors would like to thank Dr. Martha Morton and Undergraduate Instrumentation Center of the Department of Chemistry of the University of Nebraska-Lincoln for assistance and use of their FTIR instrument.